Supporting Information

In Situ Modification of Semiconductor Surface by anEnzymatic Process: A General Strategy forPhotoelectrochemical Bioanalysis

Wei-Wei Zhao, Zheng-Yuan Ma, Jing-Juan Xu* and Hong-Yuan Chen*

State key Laboratory of Analytical Chemistry for Life Science, School of Chemistry and Chemical Engineering, Nanjing University, Nanjing 210093, China

E-mail address: xujj@nju.edu.cn, hychen@nju.edu.cn

^{*} Correspondence author. Tel/Fax: +86-25-83597294.

EXPERIMENTAL SECTION

Materials and Apparatus.

Indium tin oxide (ITO) with type of N-STN-S1-10 was purchased from China Southern Glass Holding Co., Ltd. (Shenzhen, China), with coating thickness of 180±20nm and sheet resistance of 8.1±0.6 Ωcm⁻². (NH4)₂TiF₆ were obtained from Adamas Reagent Co., Ltd (Shanghai, China). H₃BO₃ was purchased from Nanjing Chemical Reagent Co., Ltd (China). Alkaline phosphate (ALP), ascorbic acid 2-phosphate (AAP), 2, 4-dichlorophenoxyacetic, acid (2, 4-DA) were purchased from Sigma-Aldrich (St. Louis, MO). Other chemicals were of analytical reagent grade and used as received. The dilution and usage of ALP should with 0.01 M Tris-HCl (pH 8.12) containg 5 mM MgCl₂ and 0.1 mM ZnCl₂. The washing buffer solution was 0.01 M Tris-HCl (pH 8.12). All aqueous solutions were prepared using ultra-pure water (Milli-Q, Millipore).

PEC measurements were performed with a homemade PEC system equipped with a 500W Xe lamp and a monochromator. Photocurrent was measured on a CHI 750a electrochemical workstation (China) with a three-electrode system: a modified ITO electrode with a geometrical area of 0.25 ± 0.01 cm² as the working electrode, a Pt wire as the counter electrode and a saturated Ag/AgCl electrode as the reference electrode. All the photocurrent measurements were performed at a constant potential of 0 V (vs saturated Ag/AgCl) and under the illumination of 410 nm visible light. To avoid the inbibitory effect of inorganic phosphate on ALP, 0.1 M Tris-HCl (pH 8.1) Tris-HCl buffer system, rather than the commonly used phosphate buffer solution (PBS), was used as the supporting electrolyte for photocurrent measurements. The solutions were deaerated by bubbling highly pure nitrogen for at least 15 min before experiment and a nitrogen atmosphere was kept over the test solution during the PEC detections. UV-vis absorption spectra were obtained on a Shimadzu UV-3600 UV/vis spectrophotometer (Shimadzu Corporation,

Japan). Scanning electron microscopic (SEM) images were recorded by a Hitachi S4800 scanning electron microscope (Hitachi Co., Japan).

Fabrication of the Liquid Phase Deposited TiO₂ Nanoparticles (NPs) Modified ITO Electrodes and the PEC Bioanalysis Development.

The ITO slices were cleaned by immersion in 2 M boiling KOH solution solved in 2-propanol for 20 min, followed by washing copiously with water and dried at 120 °C for 2 h.

Liquid phase deposition (LPD) of TiO₂ was performed according to previous report² with slight modification. (NH₄)₂TiF₆ (2.0096 g) and H₃BO₃ (1.8642 g) were separately dissolved in deionized water (100 mL). 0.3 M H₃BO₃ solution was added to 0.1 M (NH₄)₂TiF₆ solution with the volume ratio of 1:1. Then the freshly cleaned ITO substrates were immersed vertically into the mixed solution immediately. After keeping the solution at room temperature for 48 h, the as obtained TiO₂ film was rinsed with distilled water and followed by annealing at 500 °C for 60 min in air atmosphere to transform the amorphous TiO₂ into anatase TiO₂ and finally naturally cooled down to room temperature.

Subsequently, 20 µL of 2.5 mg/mL ALP was spread onto the ITO/TiO₂ electrode surface at 4 °C in a moisture atmosphere to avoid evaporation of solvent. After incubation for 5 h, the electrode was rinsed with the washing buffer to remove physically absorbed ALP. After rinsing, the ITO/TiO₂/ALP electrode was allowed for incubation in the 0.1 M Tris-HCl buffer (pH 8.12) containing variable concentrations of AAP at 37 °C for 60 min. Thereafter, the electrodes were introduced for the respective absorbance or PEC measurements. For the detection of 2, 4-DA inhibitor, the incubation was performed in the 0.1 M Tris-HCl buffer (pH 8.12) containing 0.02 M AAP and variable concentrations of 2, 4-DA.

REFERENCES

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- (2) W. W. Zhao, C. Y. Tian, J. J. Xu and H. Y. Chen, Chem. Commun., 2012, 48, 895.