

A New Method to Predict the Thermal Degradation Behaviour of Polybenzoxazines from Empirical Data Using Structure Property Relationships

Ian Hamerton^{1*}, Scott Thompson¹, Brendan J. Howlin¹ and Corinne A. Stone²

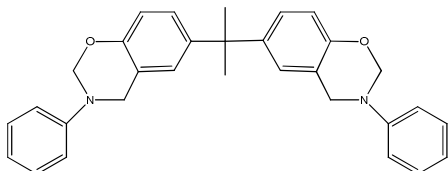
¹ Department of Chemistry, Faculty of Engineering and Physical Sciences, University of Surrey, Guildford, Surrey, GU2 7XH, UK

² Dstl, Porton Down, Salisbury, SP4 0JQ, UK

SUPPLEMENTARY CHARACTERISATION MATERIAL

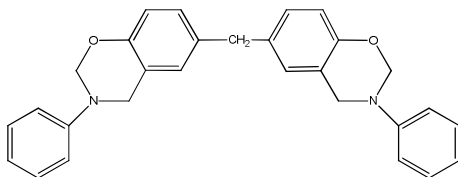
Materials Overview

MT 35600 Monomer- Bisphenol A based *bis*-benzoxazine BA-a



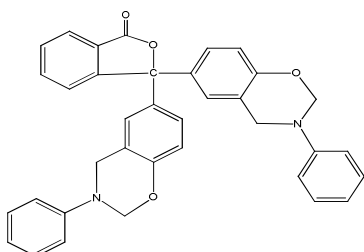
Chemical Formula: $C_{31}H_{30}N_2O_2$
Molecular Weight: 462.58
Elemental Analysis: C, 80.49; H, 6.54; N, 6.06; O, 6.92

MT 35700 Monomer- Bisphenol F based *bis*-benzoxazine BF-a



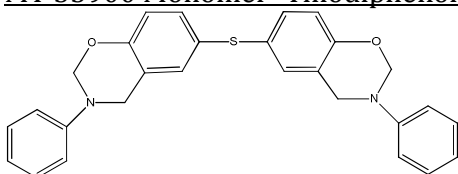
Chemical Formula: $C_{29}H_{26}N_2O_2$
Molecular Weight: 434.53
Elemental Analysis: C, 80.16; H, 6.03; N, 6.45; O, 7.36

MT 35800 Monomer- Phenolphthalein based *bis*-benzoxazine BP-a



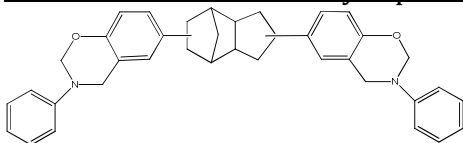
Chemical Formula: $C_{36}H_{28}N_2O_4$
Molecular Weight: 552.62
Elemental Analysis: C, 78.24; H, 5.11; N, 5.07; O, 11.58

MT 35900 Monomer- Thiodiphenol based *bis*-benzoxazine BT-a

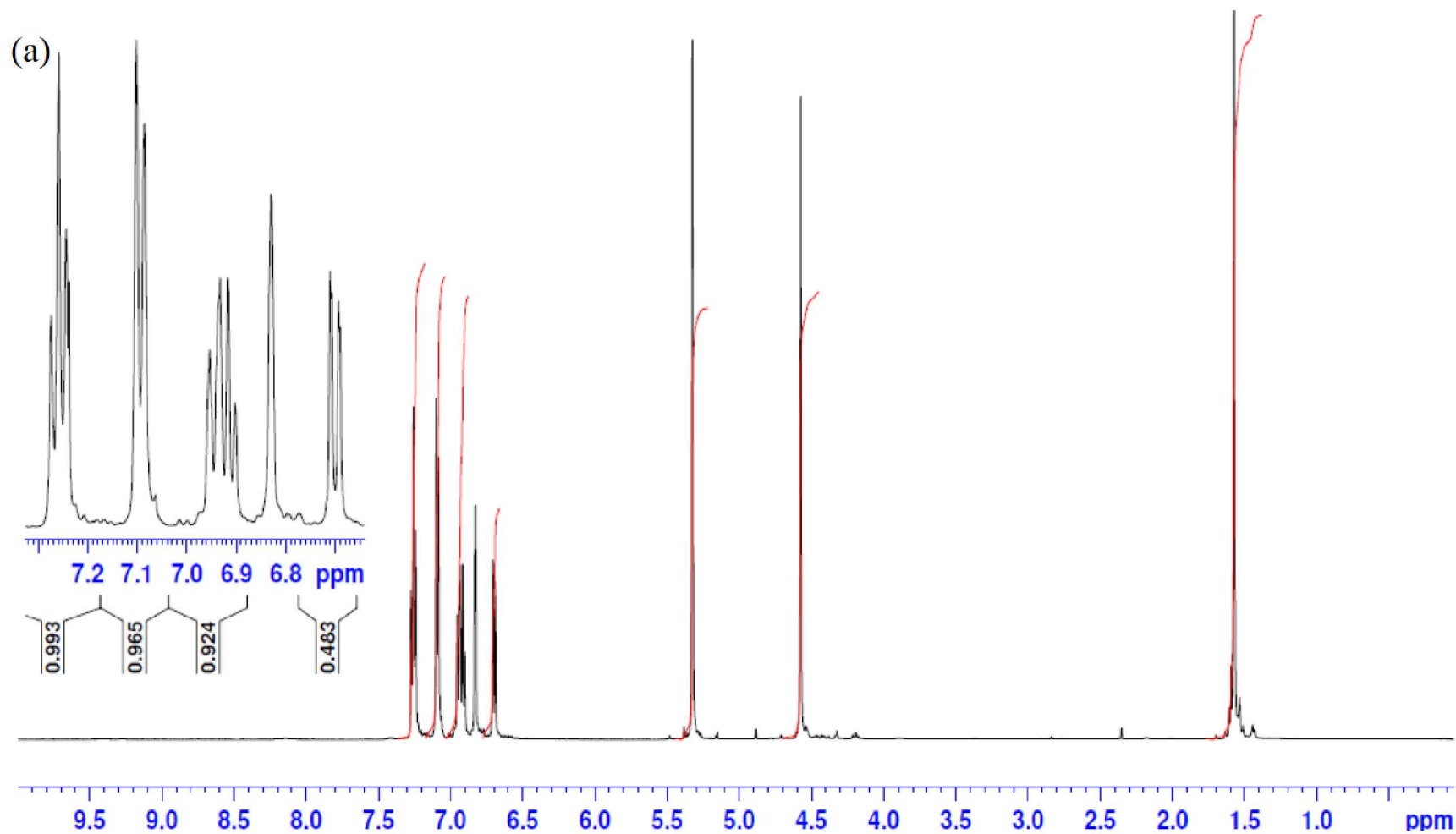


Chemical Formula: $C_{28}H_{24}N_2O_2S$
Molecular Weight: 452.57
Elemental Analysis: C, 74.31; H, 5.35; N, 6.19; O, 7.07; S, 7.09

LME 10140 Monomer- Dicyclopentadiene based *bis*-benzoxazine BD-a

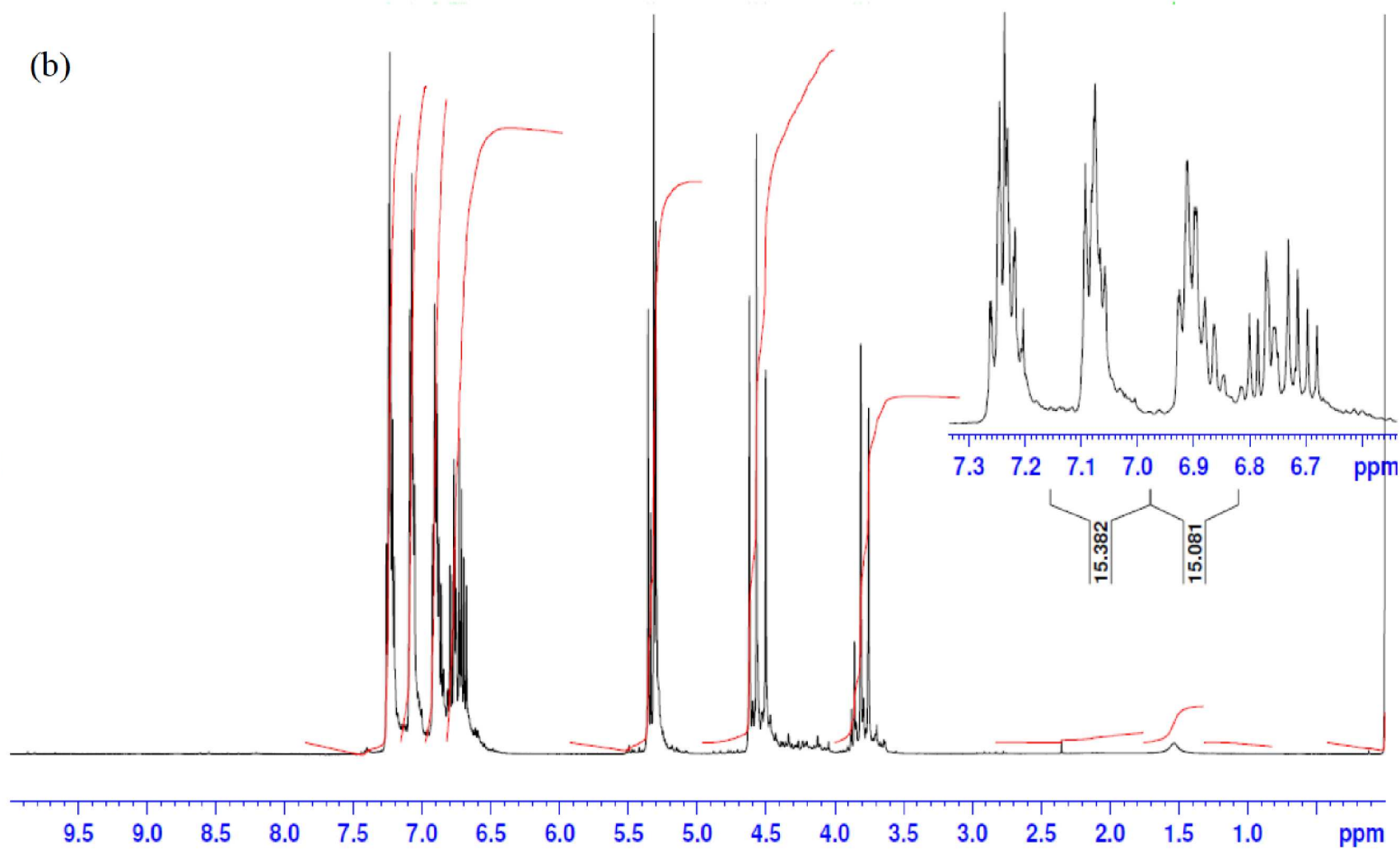


Chemical Formula: $C_{39}H_{42}N_2O_2$
Molecular Weight: 570.76
Elemental Analysis: C, 82.07; H, 7.42; N, 4.91; O, 5.61



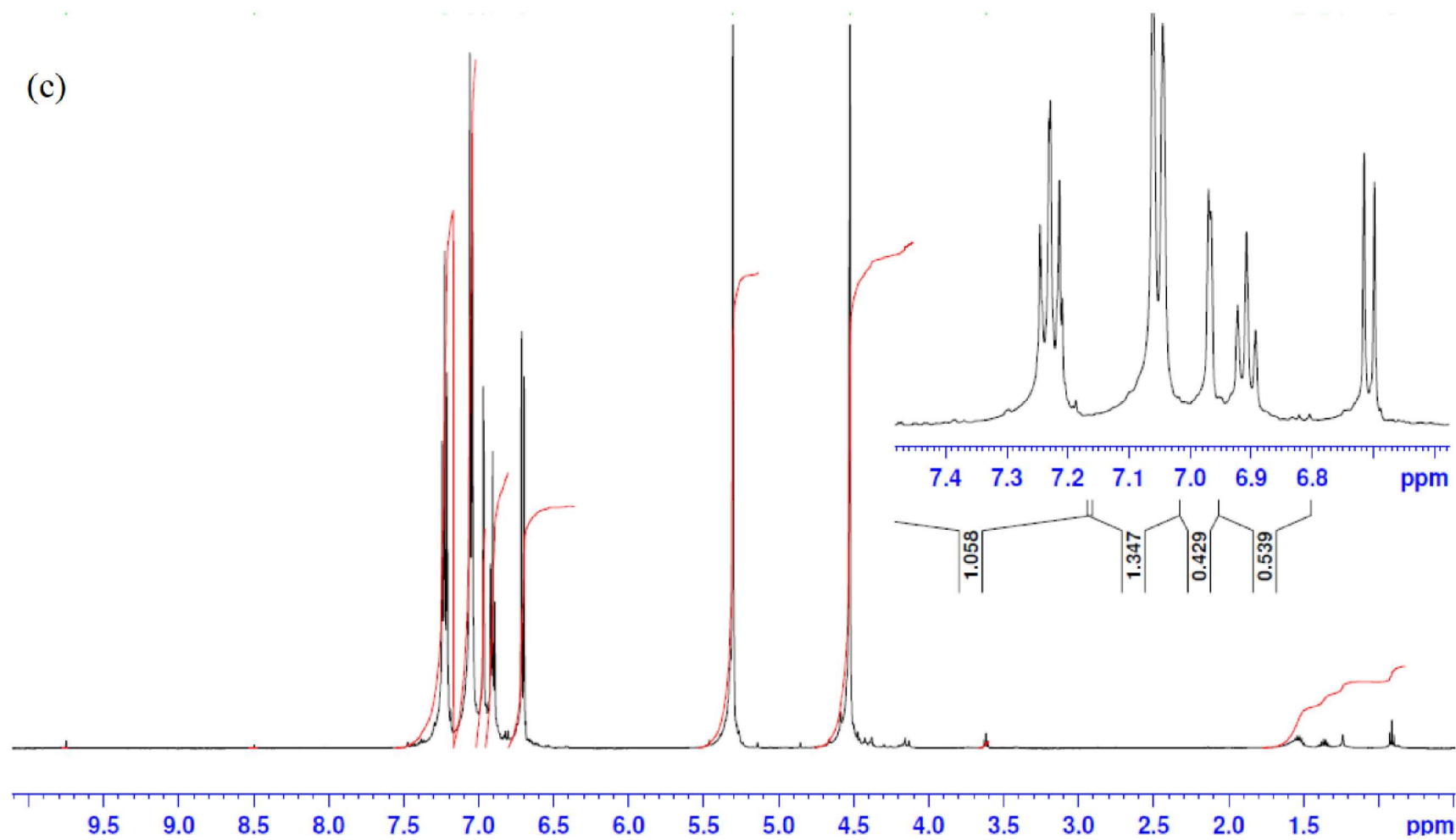
^1H NMR Spectrum (500 MHz, CDCl_3 , 298 K) of BA-a (expansion showing aromatic region)

(b)



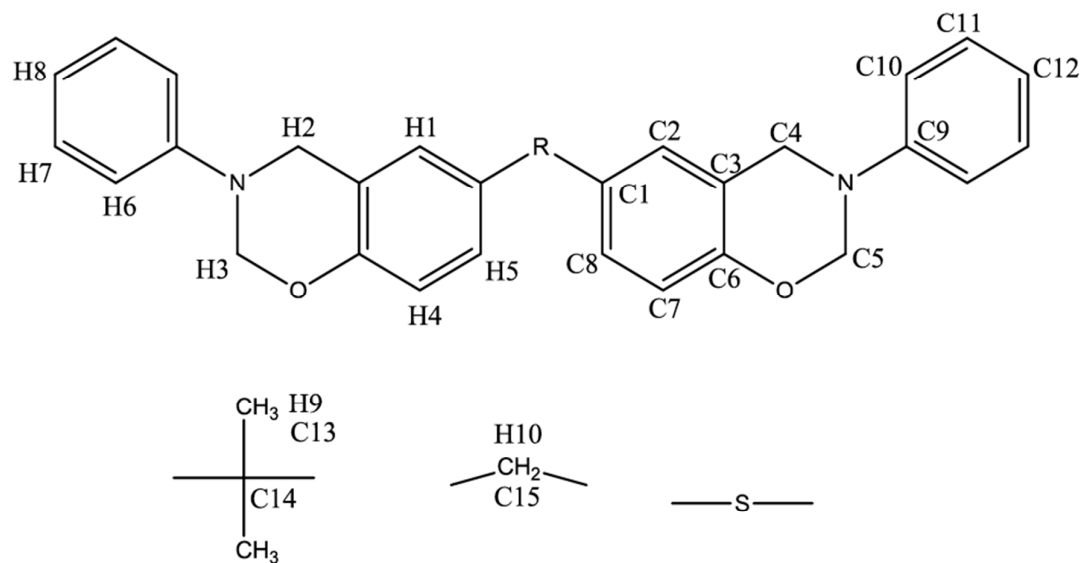
^1H NMR Spectrum (500 MHz, CDCl_3 , 298 K) of BF-a (expansion showing aromatic region)

(c)



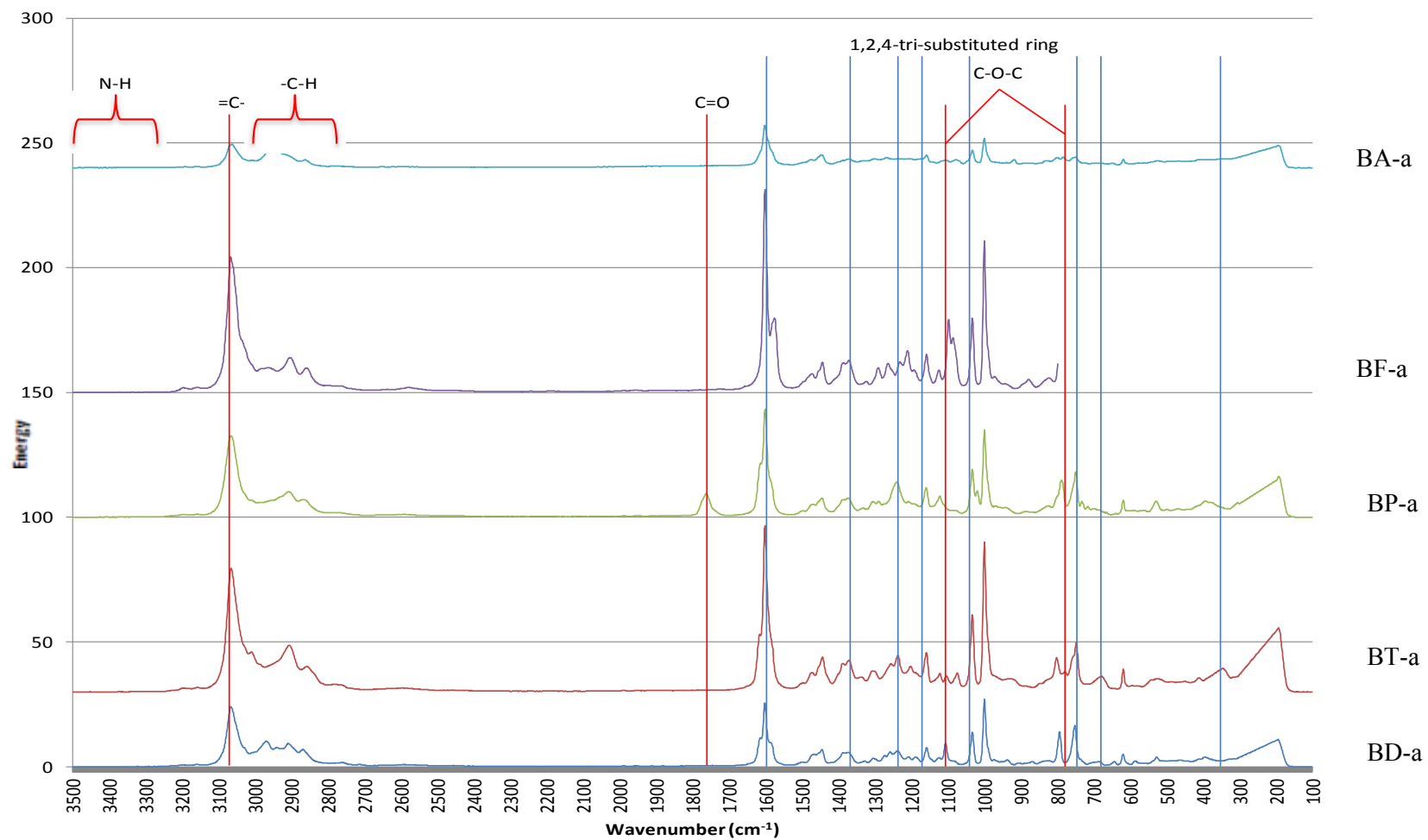
^1H NMR Spectrum (500 MHz, CDCl_3 , 298 K) of BT-a (expansion showing aromatic region)

NMR spectral assignments for BA-a, BF-a and BT-a. Only peaks that could be assigned have been included, some not possible to assign due to interference from unknown contaminants.



<i>Proton</i>	H1		H2	H3		H4		H5		H6		H7		H8		H9		H10
<i>BA-a</i>	6.83 (s)		5.32 (s)	4.57 (s)		6.71 (d)		6.95 (d)		6.95 (d)		7.24 (t)		7.08 (t)		1.61 (s)		-
<i>BF-a</i>	6.8-6.93		5.33 (s)	4.57 (s)		6.68-6.79		6.8-6.93		6.8-6.93		7.20-7.22		7.06-7.10		-		3.76 (s)
<i>BT-a</i>	6.85		5.23 (s)	4.54 (s)		6.71 (d)		6.949		6.925 (d)		7.22 (t)		7.07		-		-
<i>Carbon</i>	C1	C2	C3	C4	C5	C6	C7	C8	C9	C10	C11	C12	C13	C14	C15			
<i>BA-a</i>	143.24	126.40	124.77	79.13	50.70	152.21	116.39	121.21	148.52	120.09	129.27	118.03	41.80	31.08	-			
<i>BF-a</i>	135.17?	125.06	124.78	79.50	51.00	152.25	116.85	121.50	148.49	120.66	129.49	118.50	-	-	34.74			
<i>BT-a</i>	131.21?		127.37	79.59	50.04	153.87	117.92	121.70	148.21		129.74	118.34	-	-	-			

Raman of Monomers



Raman Spectra of Monomers

Carbon, Hydrogen and Nitrogen Content of Monomers (duplicate measurements)

Measured	BA-a	BF-a	BP-a	BT-a	BD-a
% C	80.25, 80.13	79.74, 79.91	77.57, 77.68	73.55, 73.96	75.38, 81.82
% H	6.46, 6.45	5.87, 5.88	5.30, 5.28	5.15, 5.33	5.51, 6.90
% N	5.79, 5.65	6.16, 6.14	4.55, 4.45	5.15, 5.33	5.22, 4.10

Theoretical	BA-a	BF-a	BP-a	BT-a	BD-a
% C	80.49	80.16	78.24	74.31	82.28
% H	6.54	6.03	5.11	5.35	6.92
% N	6.06	6.45	5.07	6.19	5.05