### **Supporting Information**

## Synthesis and evaluation of the multi-target-directed ligands against

#### Alzheimer's disease based on the fusion of donepezil and ebselen

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1. Experimental procedures for the synthesis of amine intermediates amine(4a-4e). First, the reaction of 1-benzylpiperidin-4-one s1 with hydroxylamine hydrochloride, in the presence of potassium carbonate, produced oxime s2, which was reduced with LiAlH<sub>4</sub> to produce amine 4a. Separately, s1 reacted with diethyl cyanomethylphosphonate in the presence of K<sub>2</sub>CO<sub>3</sub> to produce unsaturated nitrile s3. The hydrogenation of compound s3 catalysed by Pt/C yielded nitrile s4, which was reduced with LiAlH<sub>4</sub> in THF at 0°C to produce amine 4c. The synthesis of amine 4e was similar to that of amine 4c. The Wittig reaction of s1 with triethyl phosphonoacetate produced unsaturated ester s5, which underwent hydrogenation, reduction by LiAlH<sub>4</sub> and oxidation to make aldehyde s6, successively. Following the same procedure used to prepare amine 4c, compound s6 was converted to amine 4e, which has four carbon spacers between the piperidine ring and the amino group. Using 1-benzylpiperidine-4-carbaldehyde as the starting material, amine 4d was obtained by the above method. Amine 4b was obtained by benzylation of piperidine-4-carboxamide and subsequent reduction with LiAlH<sub>4</sub>.

**Scheme S1**. Synthesis of amine intermediates **4a-4e**. Reagents and conditions: (a) Hydroxylamine hydrochloride,  $K_2CO_3$ , EtOH; (b) LiAlH<sub>4</sub>, THF, reflux; (c) Diethyl cyanomethylphosphonate,  $K_2CO_3$ , THF; (d) Pt/C, H<sub>2</sub>; (e) LiAlH<sub>4</sub>, THF,  $0^{\circ}C$ ; (f) Triethyl phosphonoacetate,  $K_2CO_3$ , THF; (g) Oxalyl chloride, DMSO, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, -78 $^{\circ}C$ ; (j) BnBr, NaHCO<sub>3</sub>, toluene.

**1-benzylpiperidin-4-one oxime (s2).** <sup>1</sup> 1-Benzyl-4-piperidone (4 g, 21.2 mmol) in dry EtOH (4 mL) was added to a mixture of hydroxylamine hydrochloride (2.95 g, 42.4 mmol) and K<sub>2</sub>CO<sub>3</sub> (5.85 g, 42.4 mmol) in EtOH (16 mL). The reaction mixture was refluxed for 1 h and was filtered after cooling. The residue was washed with EtOH, and the filtrate was evaporated to yield oxime (s2) (3.81 g, 88%) as a solid, which was used without further purification.

4-benzylcyclohexanamine (4a). 1-Benzylpiperidin-4-one oxime (s2) (2.04 g, 10

mmol) in dry THF (5 mL) was added to a suspension of LiAlH<sub>4</sub> (1.0 g, 26.3 mmol) in dry THF (20 mL) at 0 °C, and the mixture was refluxed for 16 h. After cooling to room temperature, 1.0 mL of water was added slowly, and the solution was stirred for 15 min at 0 °C. One millilitre of 15% NaOH was added, the solution was stirred for another 15 min and 3.0 mL water was added. The mixture was filtered and washed with EtOAc and H<sub>2</sub>O. The combined organic layers were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated, resulting in a colourless oil (4a) (1.42 g, 75%) that was used without further purification.

**1-benzylpiperidine-4-carboxamide** (s10). <sup>2</sup> Benzyl bromide (13.0 mL, 110 mmol) was added slowly to a mixture of commercial piperidine-4-carboxamide (s9) (12.8 g, 100 mmol), NaHCO<sub>3</sub> (15.12 g, 180 mmol) and toluene (200 mL). The reaction mixture was refluxed for 2 h and filtered after cooling. The residue was dissolved in MeOH and filtered, and the filtrate was evaporated to yield the crude product. The brown solid was crystallised from acetone and MeOH to produce pure product (s10) (16.3 g, 75%).

(1-benzylpiperidin-4-yl)methanamine (4b) 1-Benzylpiperidine-4-carboxamide (s10) (3.27 g, 15 mmol) in dry THF (10 mL) was added to a suspension of LiAlH<sub>4</sub> (1.0 g, 26.3 mmol) in dry THF at 0 °C. The mixture was refluxed for 5 h. In a procedure similar to the production of compound 4a, compound 4b was produced as a colourless oil (0.624 g, 67%) that was used without further purification.

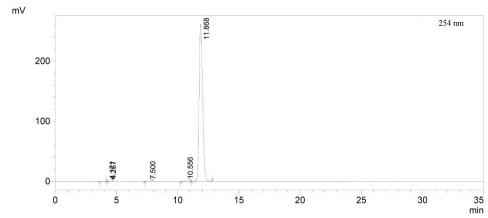
## General procedure for the synthesis of amine (4c, 4d, 4e) from an aldehyde or ketone. $^2$

- (1) A mixture of K<sub>2</sub>CO<sub>3</sub> (1 eq) and diethyl cyanomethylphosphonate (1.2 eq) in dry THF was stirred at room temperature for 15 min and refluxed for 20 min. After cooling, the aldehyde or ketone (1 eq) was added, and the mixture was refluxed for 12 h. After cooling, a 10% K<sub>2</sub>CO<sub>3</sub> solution was added, and the mixture was extracted with EtOAc. The organic phase was dried over Na<sub>2</sub>SO<sub>4</sub> and evaporated under reduced pressure. The crude product was purified by flash chromatography, and a white solid was obtained.
- (2) To a solution of the above product in MeOH, 10% Pt/C was added, filling the hydrogen to 300 psi. The reaction was stirred at room temperature for 10 h. The solution was then filtered, and the filtrate was concentrated to produce a colourless oil, which was used without further purification.
- (3) The product from step 2 in dry THF was added to a suspension of LiAlH<sub>4</sub> at 0 °C, and the mixture was stirred at 0 °C for 1 h. In a process similar to the production of compound 3, the amines were obtained and used without further purification.
- **2-(1-benzylpiperidin-4-yl)ethanamine (4c)** Commercial 1-benzyl-4-piperidone was used as the starting reactant to make oil 2-(1-benzylpiperidin-4-yl) ethanamine **(4c)**, yield: 61%.
- **3-(1-benzylpiperidin-4-yl)propan-1-amine (4d)** Commercial 1-benzyl-4-formylpiperidine was used as the starting reactant to make 3-(1-benzylpiperidin-4-yl)propan-1-amine, yield: 64%.
- **4-(1-benzylpiperidin-4-yl)butan-1-amine (4e)** 2-(1-Benzylpiperidin-4-yl) acetaldehyde <sup>3</sup> was used as the starting reactant to make

4-(1-benzylpiperidin-4-yl)butan-1-amine, yield: 56%.

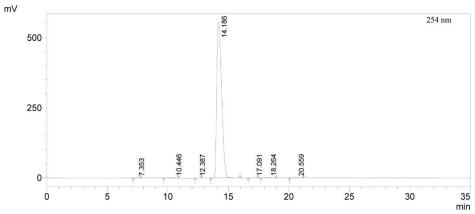
# 2. HPLC and FT-IR spectrum of the target compounds 2.1 HPLC spectrum



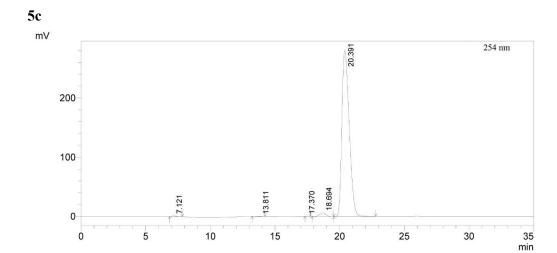


peak#	retention time	area	area %
1	4. 171	4533	0.088
2	4. 267	2666	0.052
3	7. 500	2055	0.040
4	10. 556	8260	0. 161
5	11.868	5119801	99.659
total		5137314	100, 000

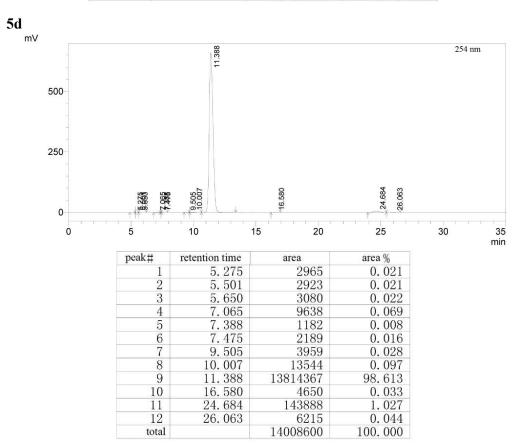




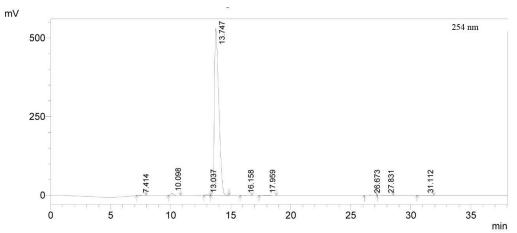
peak#	retention time	area	arca %
1	7. 353	2527	0.016
2	10, 446	28632	0. 183
3	12. 387	1552	0.010
4	14. 186	15549616	99. 582
-5	17.091	3438	0.022
6	18. 264	16570	0.106
7	20. 559	12546	0.080
total		15614881	100.000



peak#	retention time	area	area %
1	7. 121	16464	0. 147
2	13.811	7391	0.066
3	17. 370	692	0.006
4	18.694	180496	1.615
5	20. 391	10971566	98. 165
total		11176609	100.000

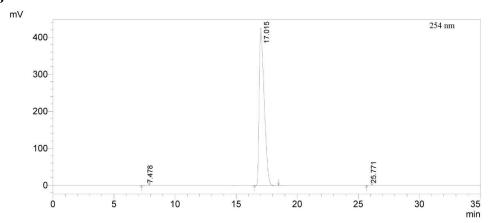




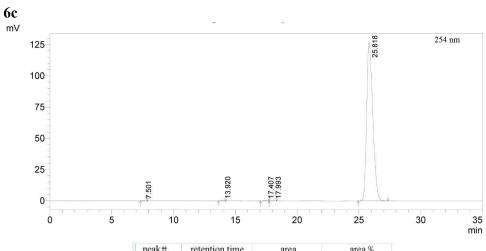


peak#	retention time	area	area %
1	7.414	8042	0.058
2	10.098	94358	0. 686
3	13. 037	19470	0. 142
4	13. 747	13597137	98. 904
5	16. 158	4952	0.036
6	17. 959	8191	0.060
7	26.673	2953	0.021
8	27.831	4372	0.032
9	31. 112	8358	0.061
total		13747833	100.000

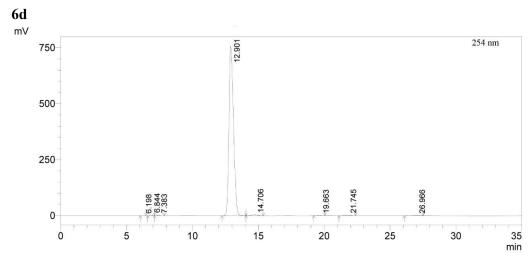
6b



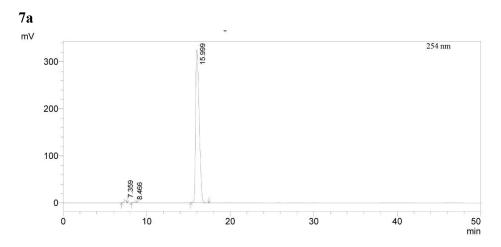
peak#	retention time	area	area %
1	7.478	2446	0.020
2	17.015	11983005	99. 965
3	25. 771	1730	0.014
total	330333030	11987180	100.000



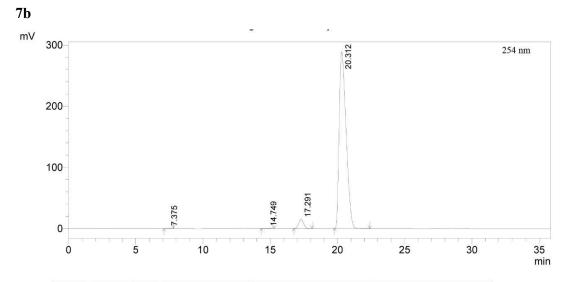
peak#	retention time	area	area %
1	7. 501	2271	0.052
2	13. 920	3006	0.068
3	17. 407	3225	0.073
4	17. 993	2294	0.052
5	25. 818	4396982	99. 755
total		4407777	100.000



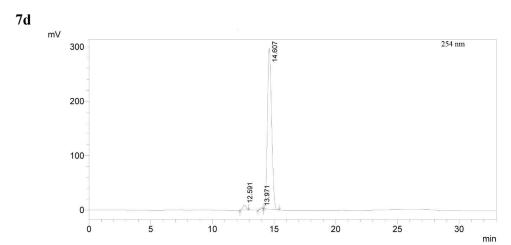
peak#	retention time	area	area %
1	6. 198	1462	0.008
2	6.844	33855	0. 187
3	7. 383	12794	0.071
4	12. 901	17904688	98. 761
5	14. 706	103394	0.570
6	19.663	8614	0.048
7	21.745	48330	0. 267
8	26. 966	16169	0.089
total		18129305	100.000



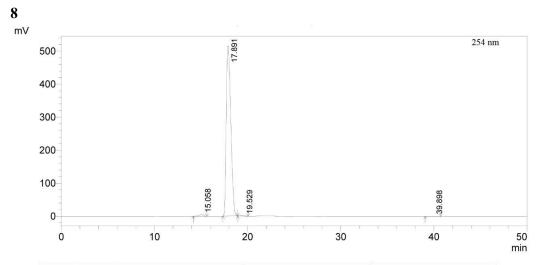
peak#	retention time	area	area %
1	7.359	101580	1.025
2	8. 466	35899	0.362
3	15. 999	9776487	98.613
total		9913966	100.000



peak#	retention time	area	area %
1	7. 375	14795	0. 139
2	14.749	11557	0.108
3	17. 291	386300	3.619
4	20. 312	10262837	96. 135
total		10675490	100.000

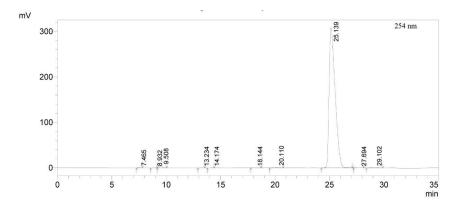


peak#	retention time	area	area %
1	12. 591	179211	2.501
2	13. 971	30754	0.429
3	14.607	6956598	97.070
total		7166564	100.000



peak#	retention time	area	area %
1	15. 058	166413	0.943
2	17.891	17442648	98.855
3	19. 529	19136	0. 108
4	39. 898	16456	0.093
total		17644653	100.000

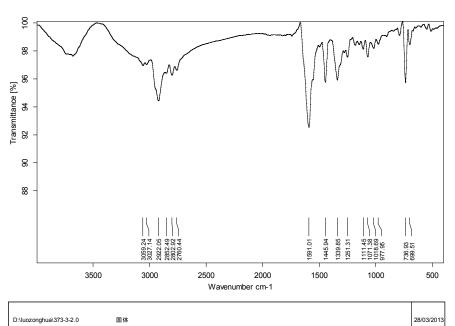




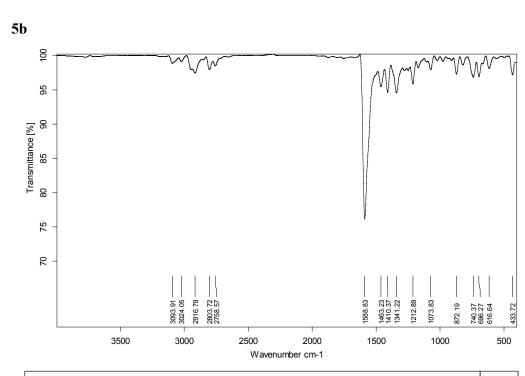
peak#	retention time	area	area %
1	7. 465	8305	0.070
2	8. 932	2272	0.019
3	9.508	68808	0.576
4	13. 234	2799	0.023
5	14. 174	1753	0.015
6	18. 144	5694	0.048
7	20. 110	48154	0.403
8	25. 139	11758557	98. 501
9	27. 694	4674	0.039
10	29. 102	36516	0.306
total		11937532	100.000

## 2.2. FT-IR spectrum

## 5a

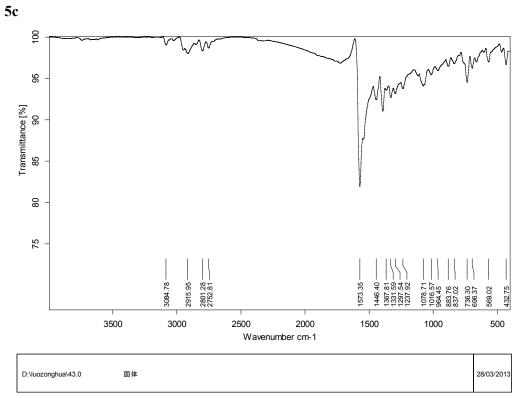


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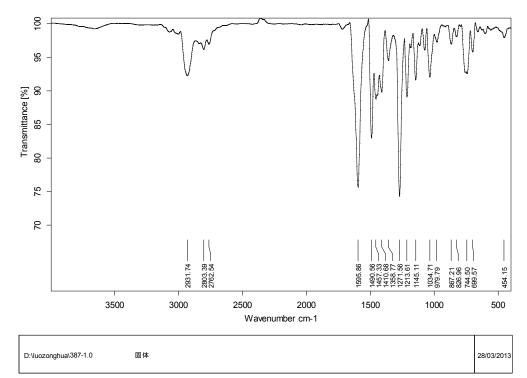
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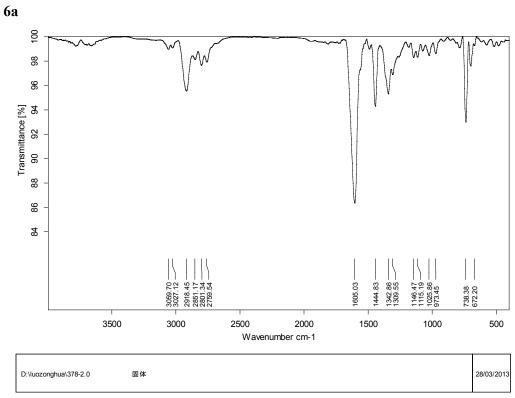


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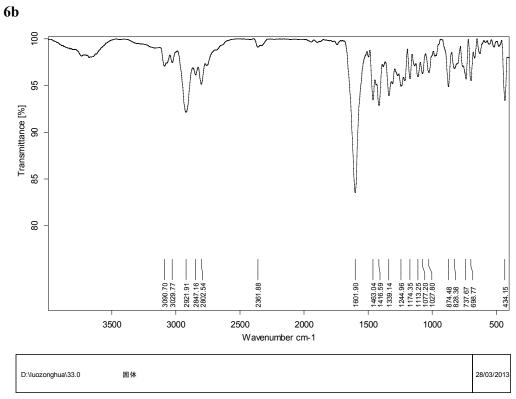




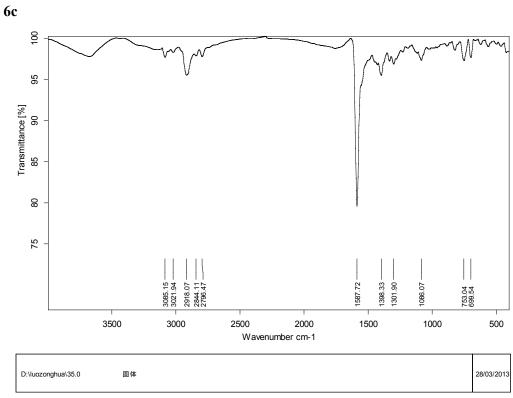
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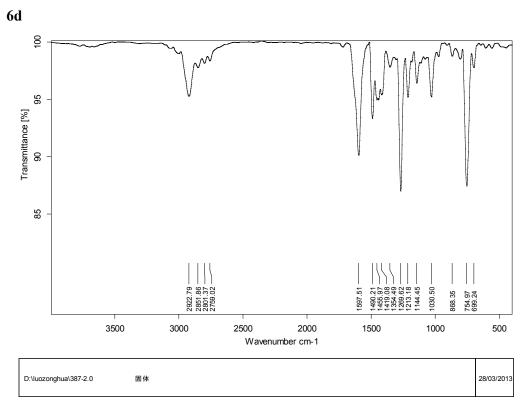
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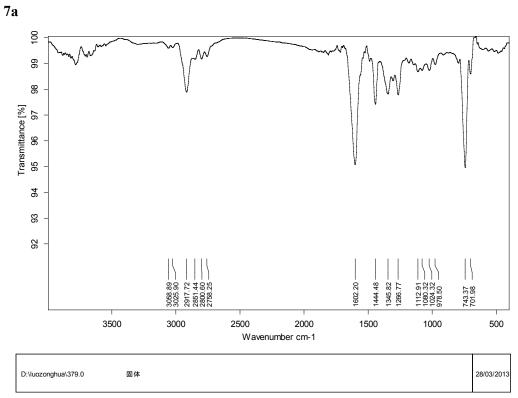
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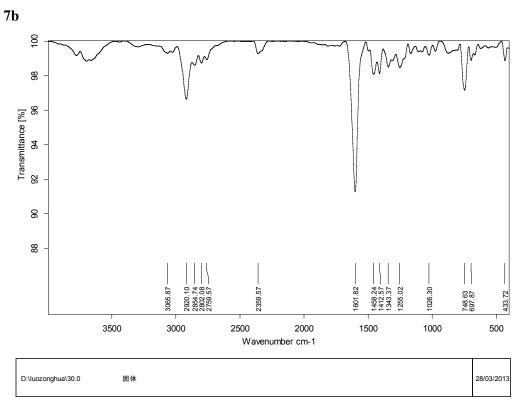
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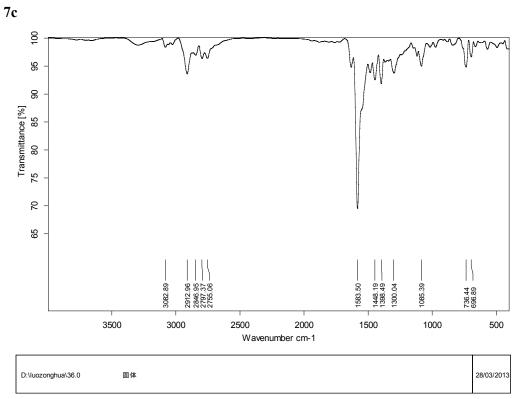
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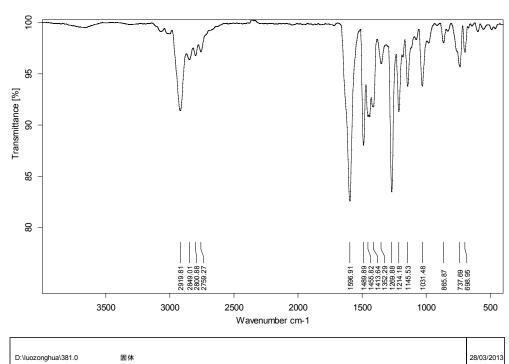


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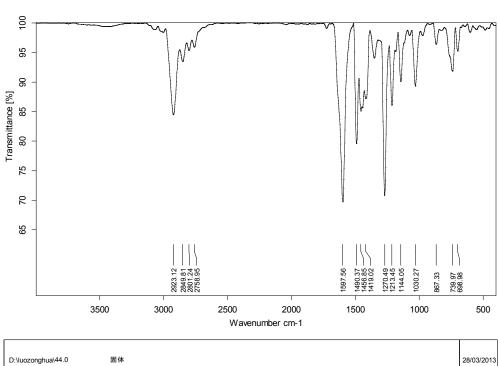
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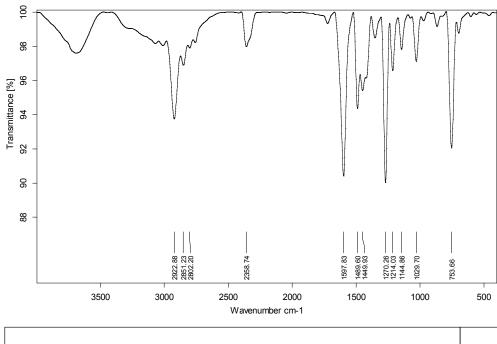
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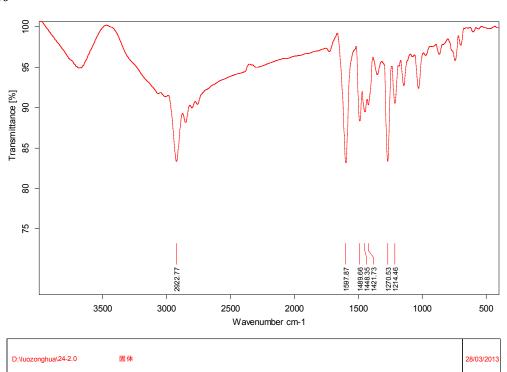




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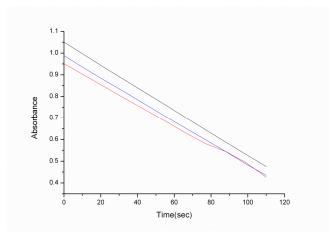


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#### 3. Coupled Reductase Assay

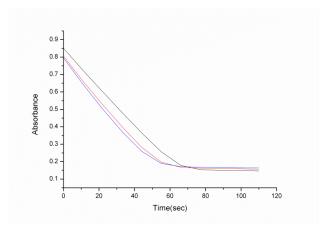
**Procedure:** Phosphate buffer solution of pH 7.5 was taken in 1 mL cuvette. GSH(2 mM), NADPH(0.4 mM), GR(1.3 U/mL) were added into cuvette contained buffer solution. Finally,  $H_2O_2(1.6 \text{ mM})$  was added to initiate the reaction in a cuvette having mixture of all and immediately start the experiment for the control values in absence of any catalyst. For the test samples(80  $\mu$ M), solution was made in MeOH and added into cuvette containing the mixture of buffer solution, GSH, NADPH and GR. Now,  $H_2O_2$  was added to initiate the reaction.

**Table S1**. Control values (in the absence of the catalyst) in the coupled reductase assay. Catalyzed reduction of  $H_2O_2$  by GSH: GSH (2 mM), NADPH (0.4 mM), GR (1.3 unit/mL), and  $H_2O_2$  (1.6 mM), at pH 7.5 in MeOH.



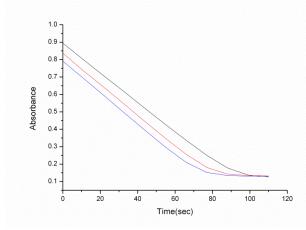
Sr. No.	ΔΑ	ΔA/min	$v_{\theta}(\mu M \cdot min^{-1})$	$v_{\theta}(\mu M \cdot min^{-1})$
1	0.0583	0.318	51.1254	
2	0.053	0.289091	46.47764	$49.5 \pm 2.6$
3	0.0579	0.315818	50.77463	

**Table S2**. Reduction rate  $(\nu_{\theta})$  of **7d** in the coupled reductase assay in the coupled reductase assay. Catalyzed reduction of  $H_2O_2$  by GSH: GSH (2 mM), NADPH (0.4 mM), GR (1.3 unit/mL),  $H_2O_2$  (1.6 mM) and **7d** (80  $\mu$ M), at pH 7.5 in MeOH.



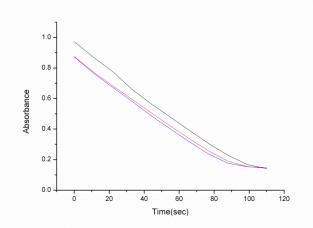
Sr. No.	ΔΑ	ΔA/min	$v_{\theta}(\mu M \cdot min^{-1})$	$v_{\theta}(\mu M \cdot min^{-1})$
1	0.1270	0.6930	111.37	
2	0.1445	0.7882	126.7173	123.5±10.9
3	0.1510	0.8236	132.4174	

**Table S3**. Reduction rate  $(v_0)$  of **8** in the coupled reductase assay in the coupled reductase assay. Catalyzed reduction of  $H_2O_2$  by GSH: GSH (2 mM), NADPH (0.4 mM), GR (1.3 unit/mL),  $H_2O_2$  (1.6 mM) and **8**(80  $\mu$ M), at pH 7.5 in MeOH.



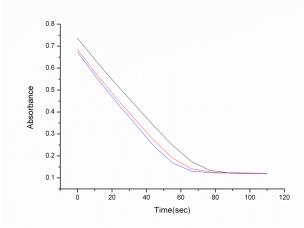
Sr. No.	ΔΑ	$\Delta A/min$	$v_{\theta}(\mu M \cdot min^{-1})$	$v_{\theta}(\mu M \cdot min^{-1})$
1	0.0944	0.514909	82.78281	
2	0.1011	0.551455	88.65829	$86.1 \pm 3.0$
3	0.0989	0.539455	86.72903	

**Table S4**. Reduction rate  $(\nu_0)$  of **9** in the coupled reductase assay in the coupled reductase assay. Catalyzed reduction of  $H_2O_2$  by GSH: GSH (2 mM), NADPH (0.4 mM), GR (1.3 unit/mL),  $H_2O_2$  (1.6 mM) and **9**(80  $\mu$ M), at pH 7.5 in MeOH.



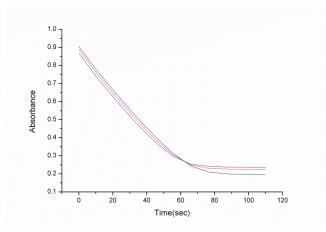
Sr. No.	ΔΑ	ΔA/min	$v_0(\mu M \cdot min^{-1})$	$v_0(\mu M \cdot min^{-1})$
1	0.0876	0.477818	76.81964	
2	0.0935	0.51	81.99357	$80.1 \pm 2.8$
3	0.0929	0.506727	81.46741	

**Table S5**. Reduction rate  $(\nu_{\theta})$  of **10** in the coupled reductase assay in the coupled reductase assay. Catalyzed reduction of  $H_2O_2$  by GSH: GSH (2 mM), NADPH (0.4 mM), GR (1.3 unit/mL),  $H_2O_2$  (1.6 mM) and **10**(80  $\mu$ M), at pH 7.5 in MeOH.



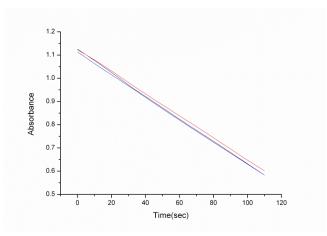
Sr. No.	ΔΑ	ΔA/min	$v_{\theta}(\mu M \cdot min^{-1})$	$v_{\theta}(\mu M \cdot min^{-1})$
1	0.1055	0.575455	92.51681	
2	0.1115	0.608182	97.77843	$97.1 \pm 4.3$
3	0.1152	0.628364	101.0231	

**Table S6**. Reduction rate  $(v_0)$  of ebselen in the coupled reductase assay in the coupled reductase assay. Catalyzed reduction of  $H_2O_2$  by GSH: GSH (2 mM), NADPH (0.4 mM), GR (1.3 unit/mL),  $H_2O_2$  (1.6 mM) and ebselen (80  $\mu$ M), at pH 7.5 in MeOH.



Sr. No.	ΔΑ	ΔA/min	$v_{\theta}(\mu M \cdot min^{-1})$	$v_{\theta}(\mu M \cdot min^{-1})$
1	0.1372	0.748364	120.3157	
2	0.1397	0.762	122.508	121.3±1.1
3	0.1382	0.753818	121.1926	

**Table S7**. Reduction rate ( $\nu_0$ ) of donepezil in the coupled reductase assay in the coupled reductase assay. Catalyzed reduction of  $H_2O_2$  by GSH: GSH (2 mM), NADPH (0.4 mM), GR (1.3 unit/mL),  $H_2O_2$  (1.6 mM) and donepezil (80  $\mu$ M), at pH 7.5 in MeOH.



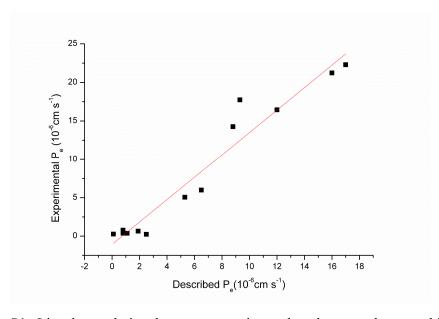
Sr. No.	ΔΑ	$\Delta A/min$	$v_{\theta}(\mu M \cdot min^{-1})$	$v_{\theta}(\mu M \cdot min^{-1})$
1	0.054	0.294545	47.35457	
2	0.0484	0.264	42.44373	46.1±3.2
3	0.0553	0.301636	48.49459	

#### 4. Tables of results for the PAMPA

**Table S8** Permeability ( $P_e \times 10^{-6}$  cm s<sup>-1</sup>) in the PAMPA-BBB assay for 13 commercial drugs, used in the Experiment Validation.

Commercial drugs	Bibl <sup>a</sup>	PBS: EtOH (70:30) <sup>b</sup>
testosterone	17	$22.3 \pm 1.4$
verapamil	16	$21.2 \pm 1.9$
desipramine	12	$16.4 \pm 1.2$
progesterone	9.3	$17.7 \pm 1.2$
promazine	8.8	$14.3 \pm 0.5$
chlorpromazine	6.5	$6.0\pm0.3$
clonidine	5.3	$5.1 \pm 0.3$
piroxicam	2.5	$0.24 \pm 0.01$
hydrocortisone	1.9	$0.65 \pm 0.01$
lomefloxacin	1.1	$0.37 \pm 0.02$
atnolol	0.8	$0.78 \pm 0.02$
ofloxacin	0.8	$0.37 \pm 0.02$
theophylline	0.1	$0.26\pm0.01$

<sup>&</sup>lt;sup>a</sup> Taken from reference 4. <sup>b</sup> Data are the mean ± SD of three independent experiments



**Figure S1.** Lineal correlation between experimental and reported permeability of commercial drugs using the PAMPA-BBB assay.  $P_e$  (exp.)=1.4574Pe (bibl.) -1.0773 ( $R^2$ =0.9427)

<b>Table S9.</b> Ranges of Permeability of PAMPA-BBB Assays ( $P_e$ , 10 <sup>-6</sup> cm s <sup>-1</sup> )				
Compounds of high BBB permeation (CNS+)	$P_{\rm e} > 4.7$			
	4 <b>.</b>			
Compounds of uncertain BBB permeation (CNS+/-)	$4.7 > P_e > 1.8$			
Common de effect DDD memoration (CNC)	n <1.0			
Compounds of low BBB permeation (CNS-)	$P_{\rm e} < 1.8$			

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