

Supporting Information

CO-Induced Methyl Migration in a Rhodium Thiophosphoryl Pincer Complex and its Comparison with Phosphine-Based Complexes: The Divergent Effects of S and P Donor Ligands

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Part I: Experimental ^1H and ^{31}P NMR spectra of the SCS complexes (p. 1-14)

Part II: Computed atomic properties for SCS and PCP complexes (p. 15-42)

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Part I: Experimental ^1H and ^{31}P NMR spectra of the SCS complexes

Complex 3

$^{31}\text{P}\{^1\text{H}\}$ NMR (162 MHz, CD_3CN): 87.67 (d, $^2J_{\text{RHP}} = 1.4$ Hz). ^1H NMR (400 MHz, CD_3CN): 7.37 (m, $^3J_{\text{HH}} = 7.7$ Hz, 2H, Ar-H), 7.20 (m, $^3J_{\text{HH}} = 7.7$ Hz, 1H, Ar-H), 2.62 (m, $^3J_{\text{HH}} = 7.0$ Hz, 2H, $\text{PCH}(\text{CH}_3)_2$), 2.56 (m, $^3J_{\text{HH}} = 7.0$ Hz, 2H, $\text{PCH}(\text{CH}_3)_2$), 2.50 (s, 3H, COCH_3), 1.21 (dd, $^3J_{\text{PH}} = 17.8$ Hz, $^3J_{\text{HH}} = 6.9$ Hz, 6H, $\text{PCH}(\text{CH}_3)_2$), 1.21 (dd, $^3J_{\text{PH}} = 17.9$ Hz, $^3J_{\text{HH}} = 7.0$ Hz, 6H, $\text{PCH}(\text{CH}_3)_2$), 1.10 (dd, $^3J_{\text{PH}} = 17.7$ Hz, $^3J_{\text{HH}} = 7.0$ Hz, 6H, $\text{PCH}(\text{CH}_3)_2$), 1.09 (dd, $^3J_{\text{PH}} = 18.0$ Hz, $^3J_{\text{HH}} = 7.1$ Hz, 6H, $\text{PCH}(\text{CH}_3)_2$).

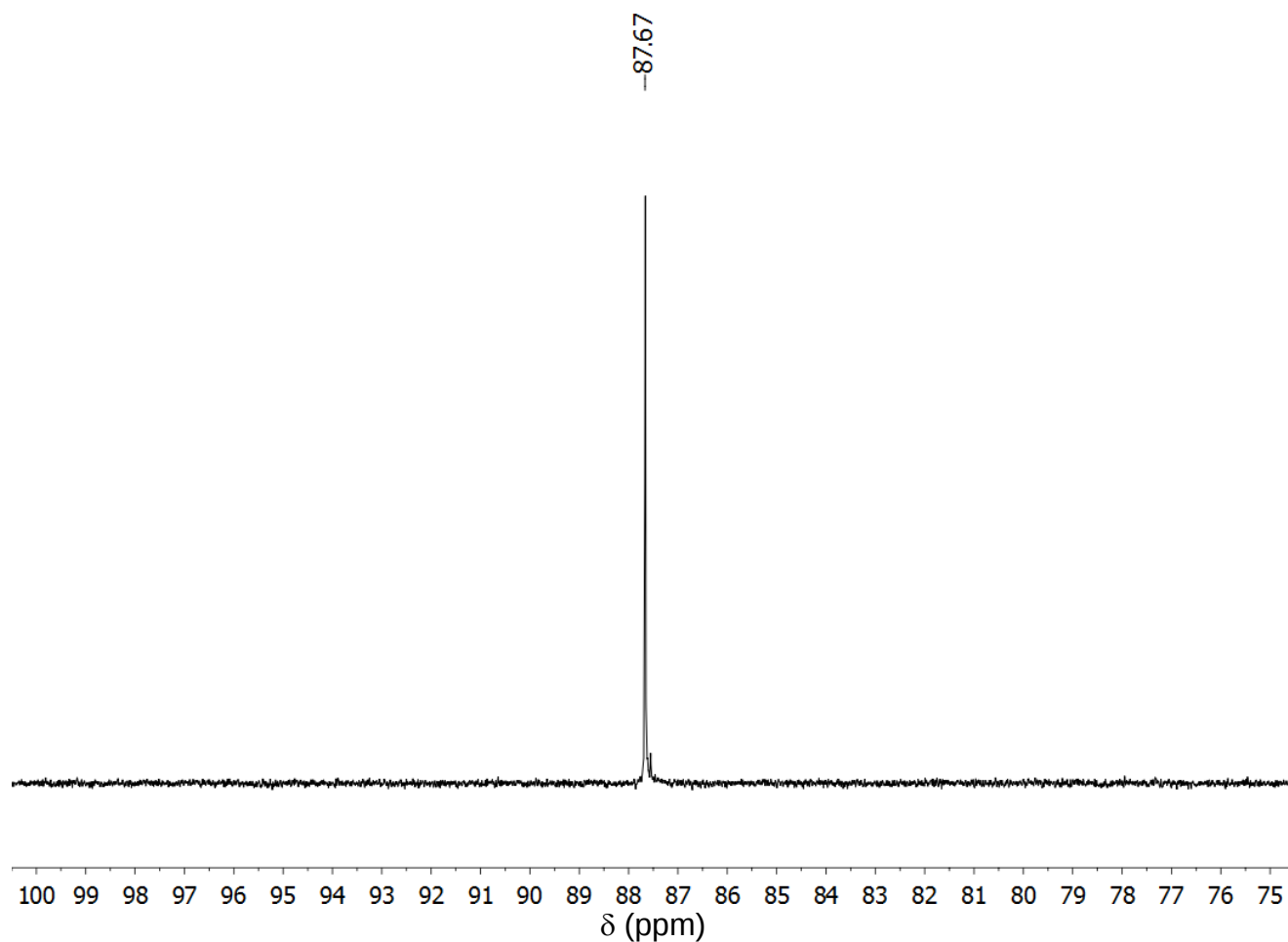


Figure S1. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of complex **3** in CD_3CN at 20°C .

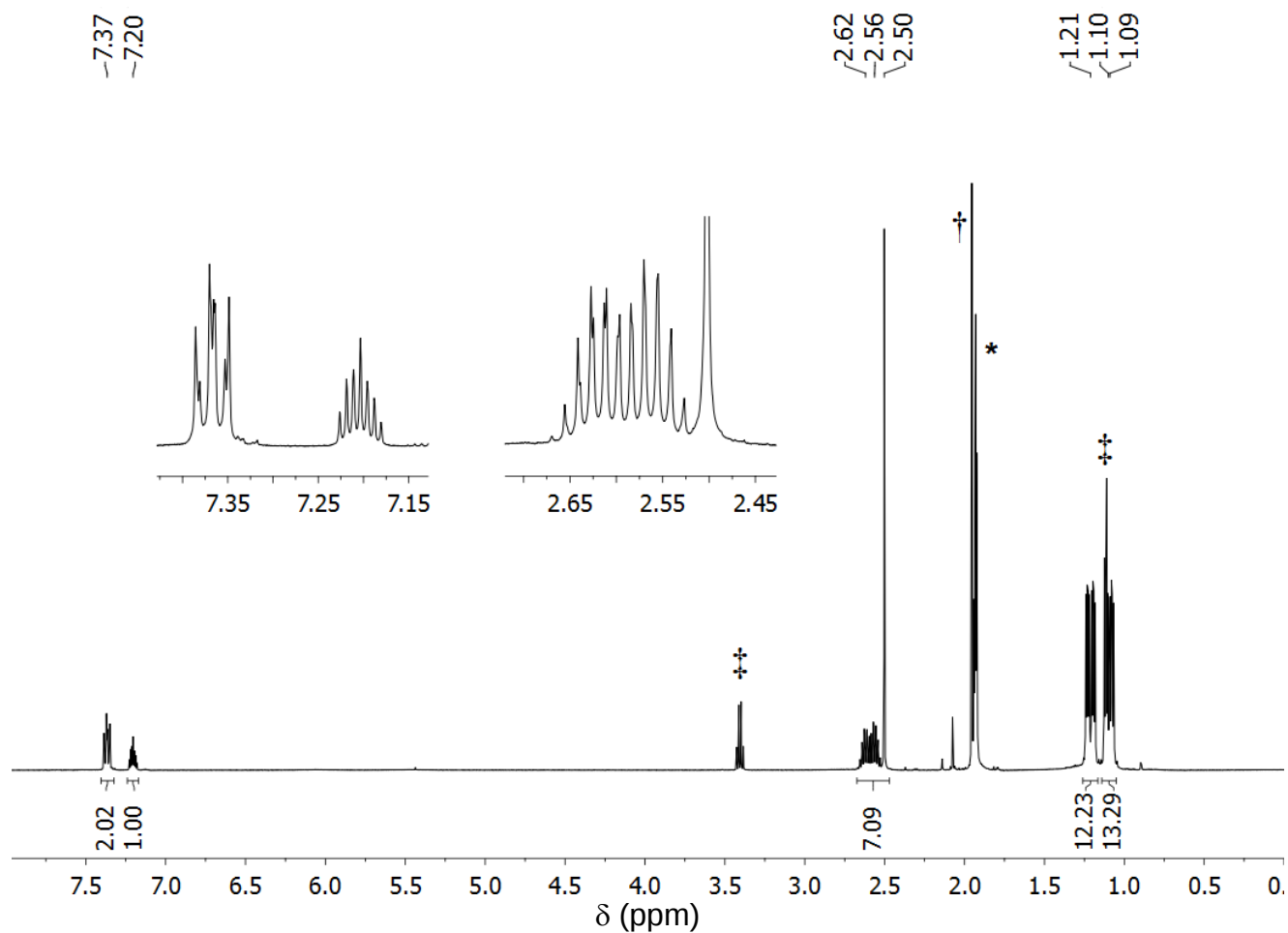


Figure S2. ^1H NMR spectrum of complex **3** in CD_3CN at 20°C . In addition to **3**, the following signals also appear in the spectrum: CHD_2CN from the solvent (*), CH_3CN from **3** (†), and residual ether (‡).

Complex 4

$^{31}\text{P}\{^1\text{H}\}$ NMR (202 MHz, acetone- d_6): 100.29 (d, $^2J_{\text{RHP}} = 2.8$ Hz). ^1H NMR (500 MHz, acetone- d_6): 7.96 (m, $^3J_{\text{HH}} = 7.6$ Hz, 2H, Ar-H), 7.57 (m, $^3J_{\text{HH}} = 7.6$ Hz, 1H, Ar-H), 3.12 (m, $^2J_{\text{PH}} = ^3J_{\text{HH}} = 7.0$ Hz, 2H, $\text{PCH}(\text{CH}_3)_2$), 2.81 (m, $^2J_{\text{PH}} = 8.5$ Hz, $^3J_{\text{HH}} = 7.0$ Hz, 2H, $\text{PCH}(\text{CH}_3)_2$), 2.49 (s, 3H, COCH_3), 1.39 (dd, $^3J_{\text{PH}} = 18.2$ Hz, $^3J_{\text{HH}} = 6.9$ Hz, 6H, $\text{PCH}(\text{CH}_3)_2$), 1.28 (dd, $^3J_{\text{PH}} = 18.5$ Hz, $^3J_{\text{HH}} = 7.0$ Hz, 6H, $\text{PCH}(\text{CH}_3)_2$), 1.19 (dd, $^3J_{\text{PH}} = 18.2$ Hz, $^3J_{\text{HH}} = 7.0$ Hz, 6H, $\text{PCH}(\text{CH}_3)_2$), 1.13 (dd, $^3J_{\text{PH}} = 18.2$ Hz, $^3J_{\text{HH}} = 7.1$ Hz, 6H, $\text{PCH}(\text{CH}_3)_2$).

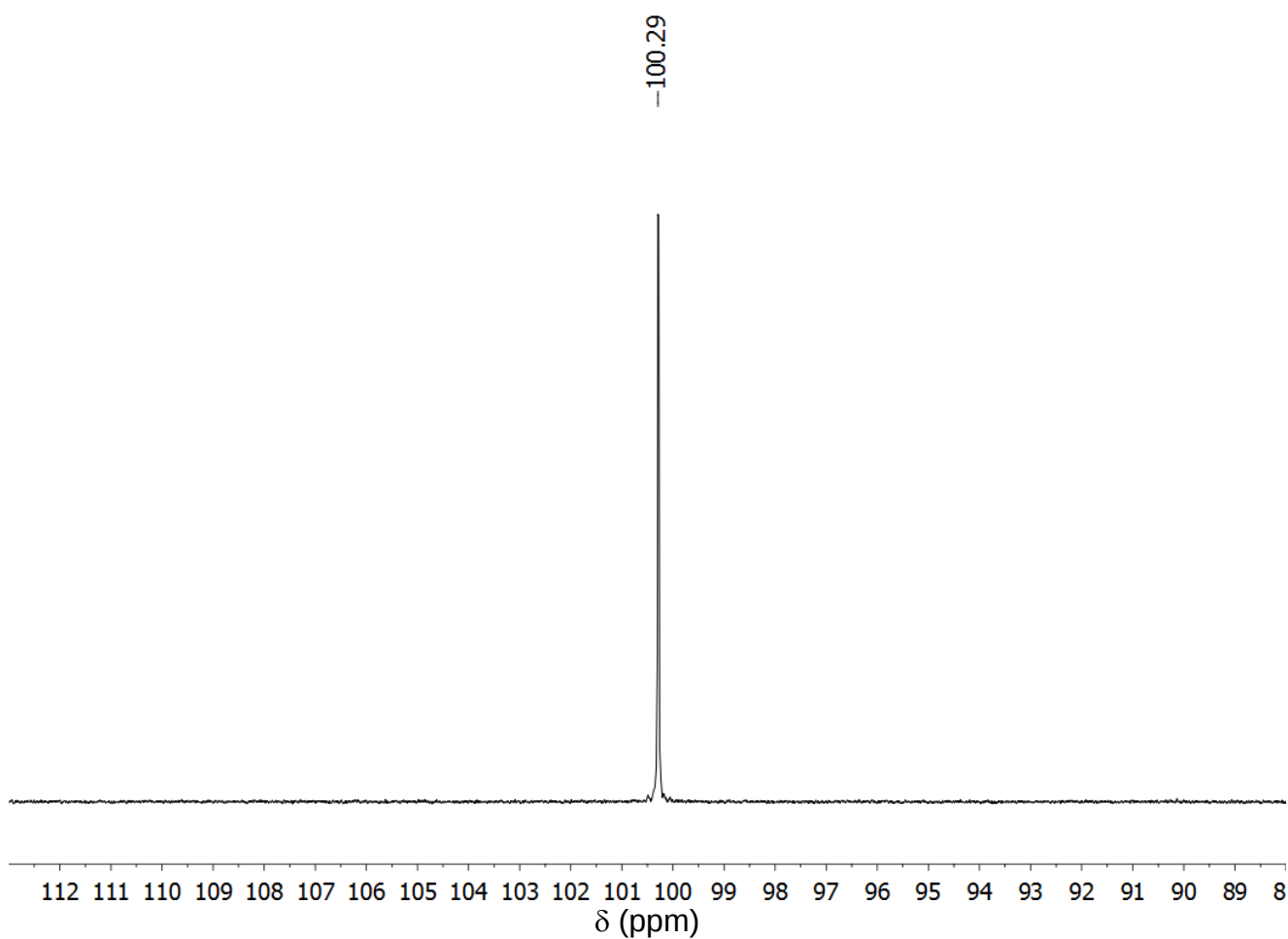


Figure S3. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of complex **4** in acetone- d_6 at 20°C.

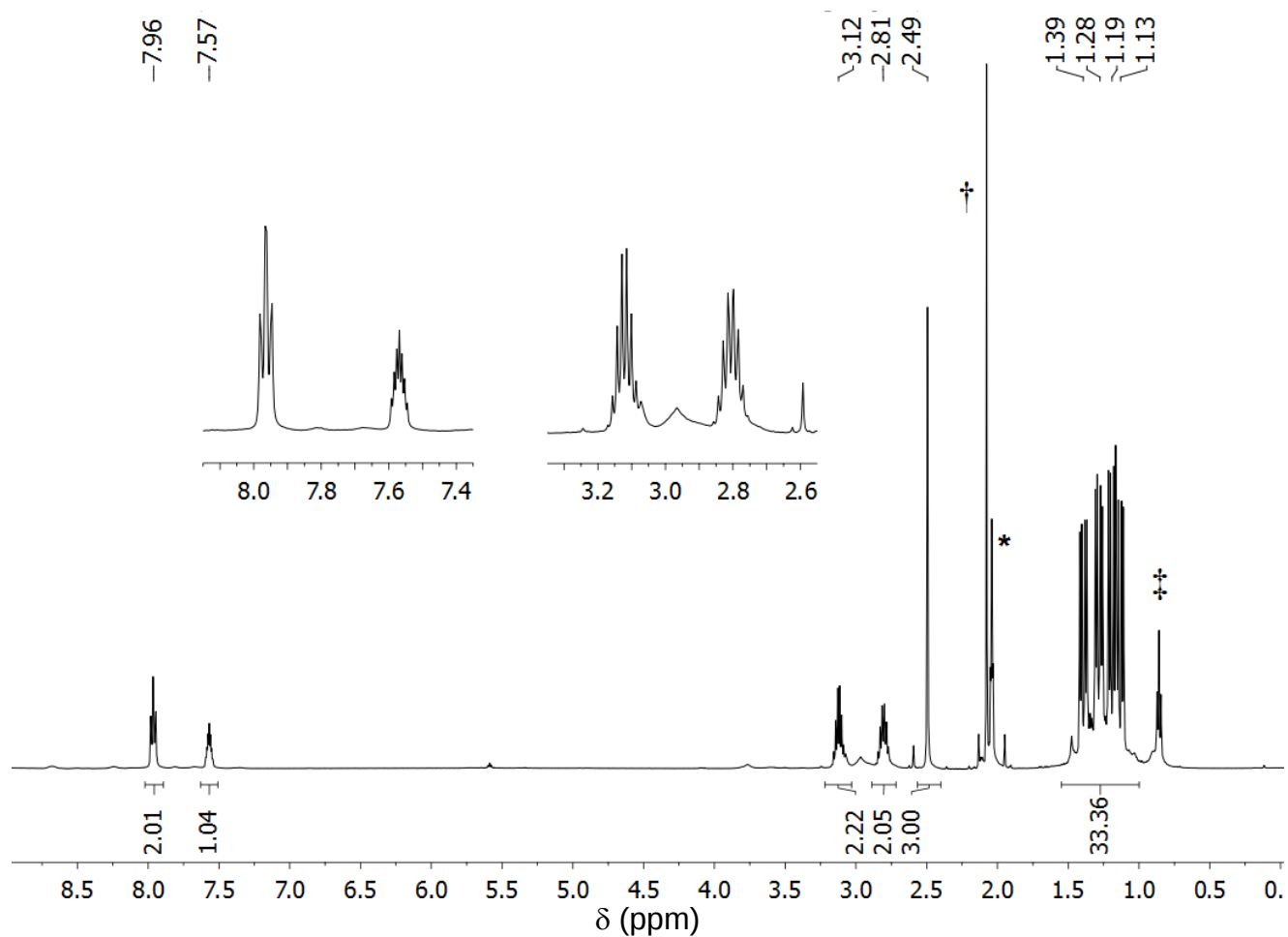


Figure S4. ^1H NMR spectrum of complex **4** in acetone-d_6 at 20°C . In addition to **4**, the following signals also appear in the spectrum: acetone- d_5 from the solvent (*), acetone from **4** ($\dagger$), and residual pentane ($\ddagger$).

Complex 6

$^{31}\text{P}\{^1\text{H}\}$ NMR (202 MHz, CD_3CN): 87.72 (s). ^1H NMR (500 MHz, CD_3CN): 7.33 (m, $^3J_{\text{HH}} = 7.5$ Hz, 2H, Ar-H), 7.16 (m, $^3J_{\text{HH}} = 7.7$ Hz, 1H, Ar-H), 2.69 (m, $^3J_{\text{HH}} = 7.2$ Hz, 2H, $\text{PCH}(\text{CH}_3)_2$), 2.47 (m, $^3J_{\text{HH}} = 7.2$ Hz, 2H, $\text{PCH}(\text{CH}_3)_2$), 1.28 (dd, $^3J_{\text{PH}} = 17.6$ Hz, $^3J_{\text{HH}} = 6.9$ Hz, 6H, $\text{PCH}(\text{CH}_3)_2$), 1.17 (dd, $^3J_{\text{PH}} = 11.1$ Hz, $^3J_{\text{HH}} = 7.3$ Hz, 6H, $\text{PCH}(\text{CH}_3)_2$), 1.14 (dd, $^3J_{\text{PH}} = 10.7$ Hz, $^3J_{\text{HH}} = 6.8$ Hz, 6H, $\text{PCH}(\text{CH}_3)_2$), 1.05 (dd, $^3J_{\text{PH}} = 17.3$ Hz, $^3J_{\text{HH}} = 7.2$ Hz, 6H, $\text{PCH}(\text{CH}_3)_2$), -20.31 (dt, $^1J_{\text{RhH}} = 15.5$ Hz, $^3J_{\text{PH}} = 1.6$ Hz, 1H, Rh-H).

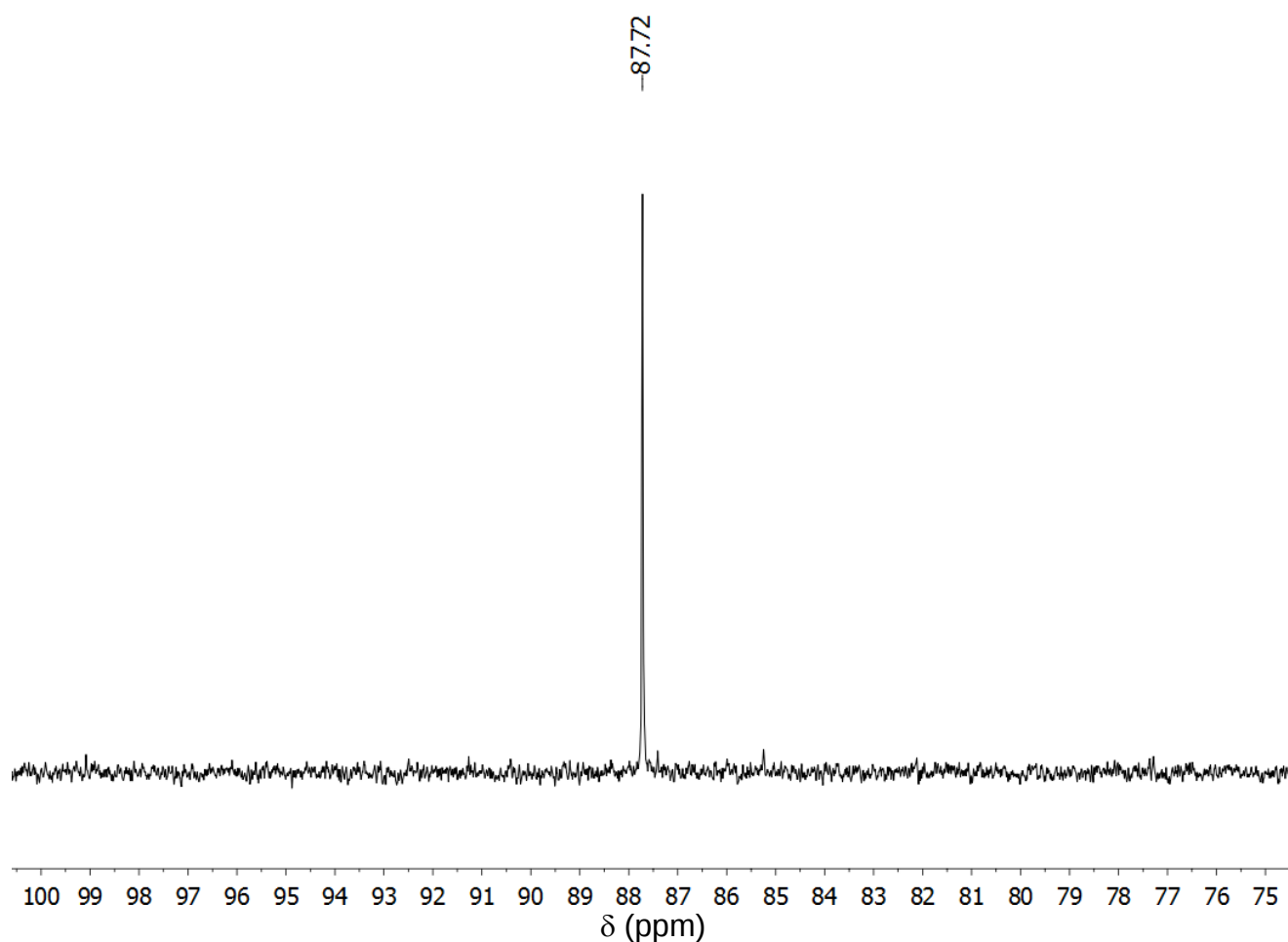


Figure S5. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of complex **6** in CD_3CN at 20°C.

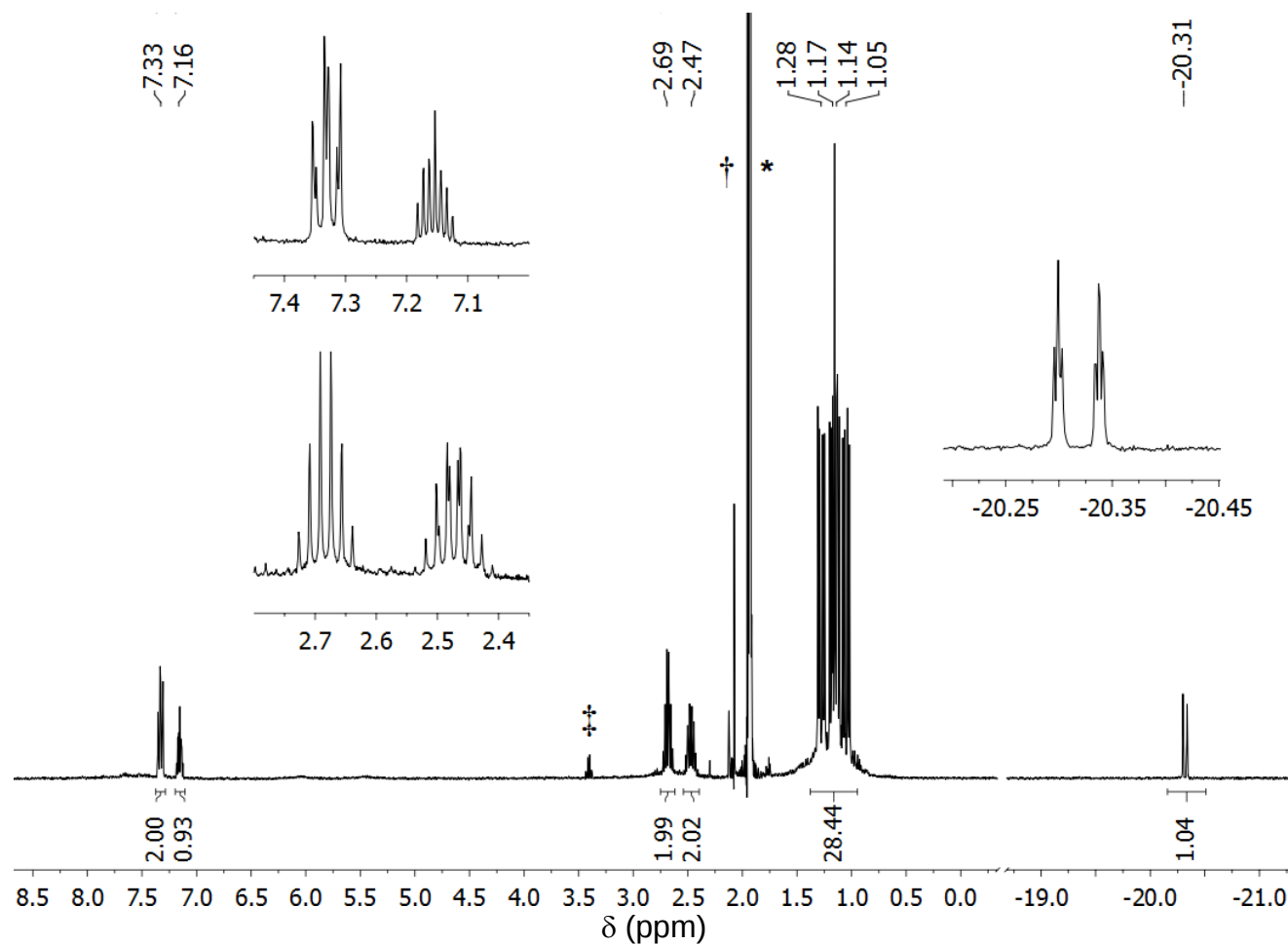


Figure S6. ^1H NMR spectrum of complex **6** in CD_3CN at 20°C . In addition to **6**, the following signals also appear in the spectrum: CHD_2CN from the solvent (*), CH_3CN from **6** ($\dagger$), and residual ether ($\ddagger$).

Complex 7

$^{31}\text{P}\{^1\text{H}\}$ NMR (202 MHz, CD_3CN , 20°C): 77.29 (d, $^3J_{\text{RhP}} = 4.5$ Hz). ^1H NMR (500 MHz, CD_3CN , 20°C): 6.82 (m, $^3J_{\text{HH}} = 7.5$ Hz, 2H, Ar-H), 6.65 (m, $^3J_{\text{HH}} = 7.4$ Hz, 1H, Ar-H), 2.40 (m, 4H, $\text{PCH}(\text{CH}_3)_2$), 1.30 (dd, $^3J_{\text{PH}} = 16.5$ Hz, $^3J_{\text{HH}} = 7.1$ Hz, 12H, $\text{PCH}(\text{CH}_3)_2$), 1.10 (dd, $^3J_{\text{PH}} = 16.6$ Hz, $^3J_{\text{HH}} = 7.0$ Hz, 12H, $\text{PCH}(\text{CH}_3)_2$).

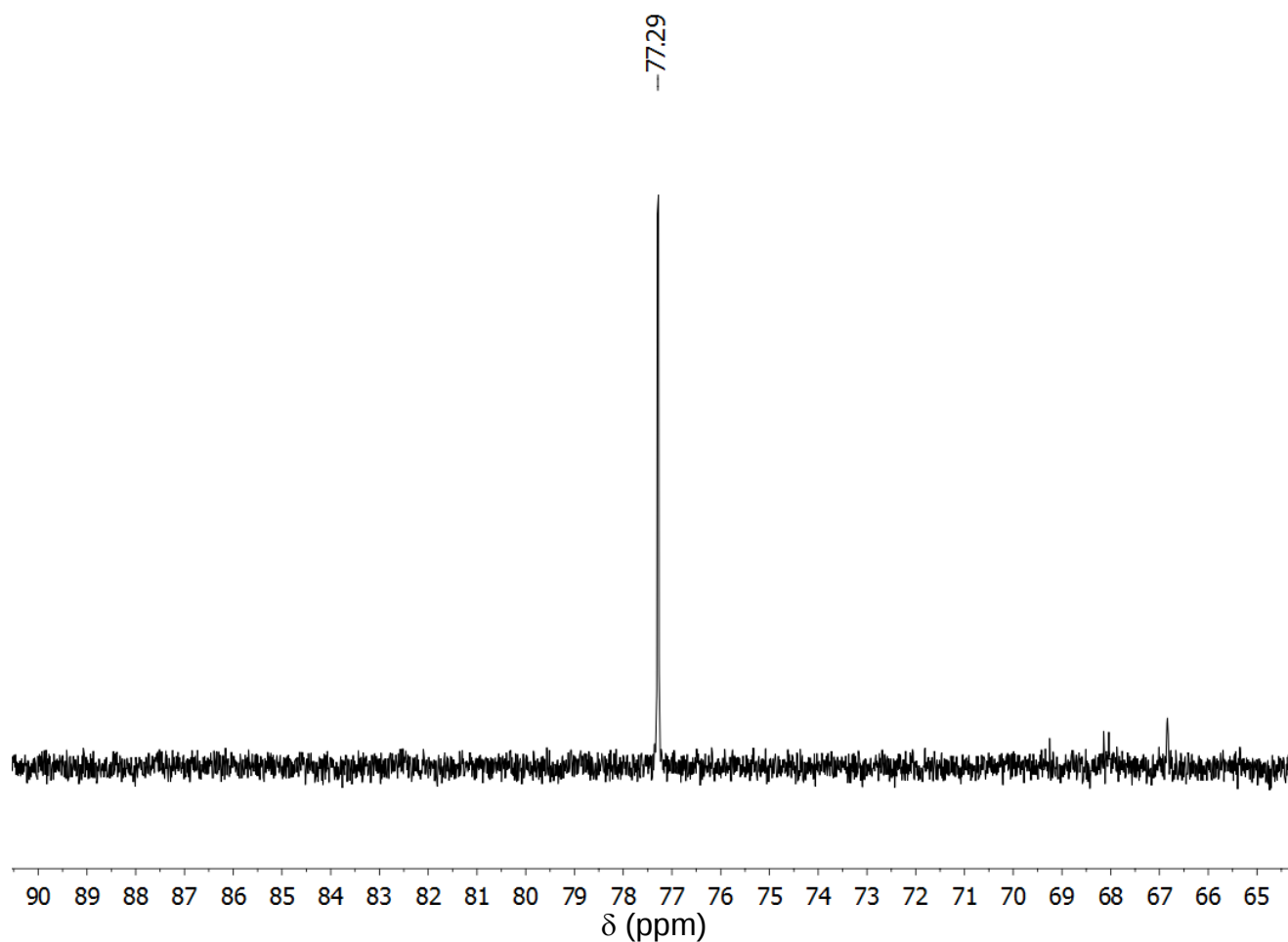


Figure S7. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of complex 7 in CD_3CN at 20°C .

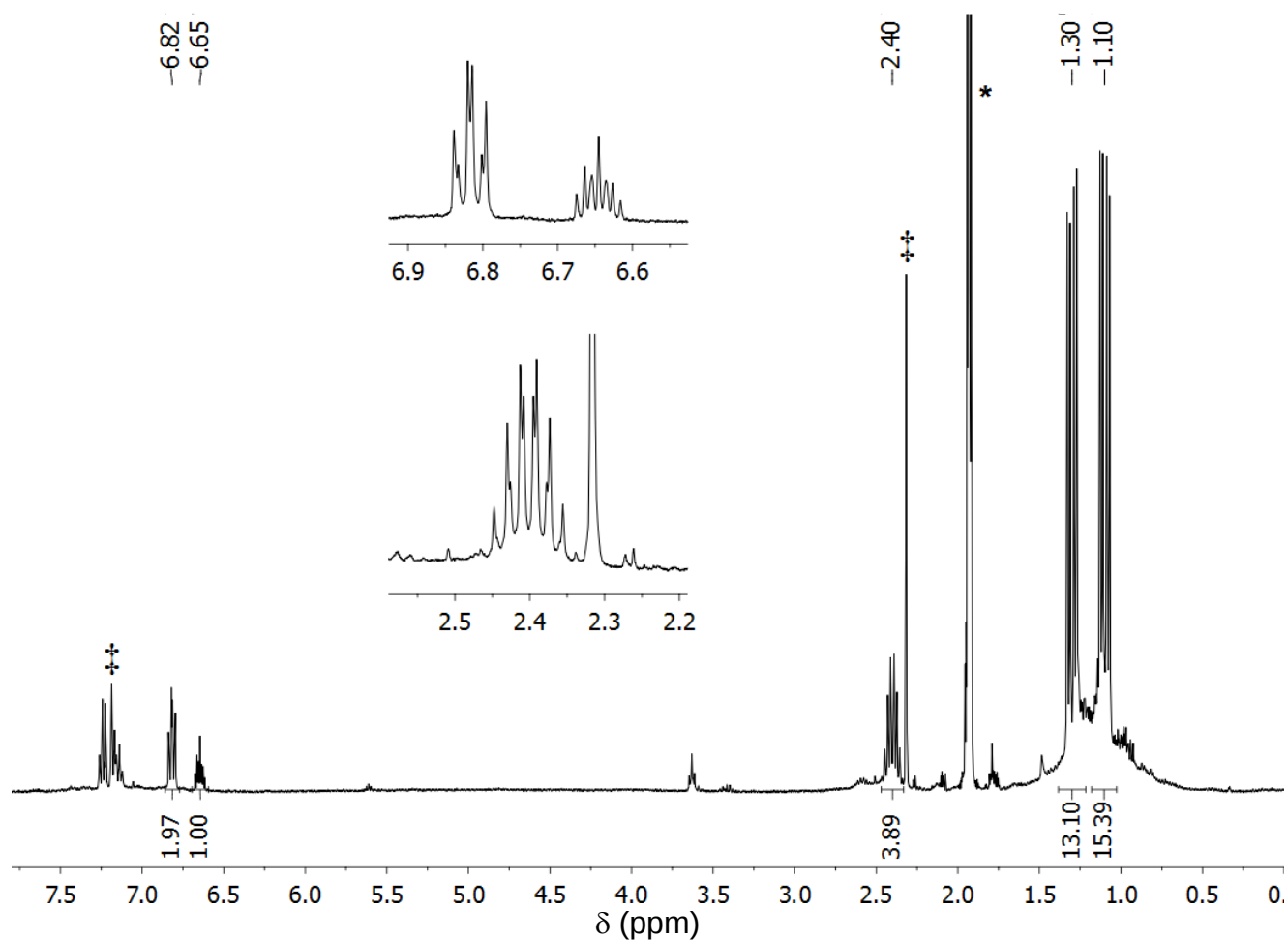


Figure S8. ^1H NMR spectrum of complex **7** in CD_3CN at 20°C . In addition to **7**, the following signals also appear in the spectrum: CHD_2CN from the solvent (*), and residual toluene ($\ddagger$) from the extraction step of the synthesis. Toluene could not be thoroughly removed under vacuum due to decomposition of the complex under such conditions.

Complex 8

$^{31}\text{P}\{^1\text{H}\}$ NMR (162 MHz, CD_2Cl_2 , 20°C): 98.33 (s). ^1H NMR (400 MHz, CD_2Cl_2 , 20°C): 7.33 (m, $^3J_{\text{HH}} = 7.6$ Hz, 2H, Ar-H), 7.12 (m, $^3J_{\text{HH}} = 7.6$ Hz, 1H, Ar-H), 2.54 (m, $^2J_{\text{PH}} = ^3J_{\text{HH}} = 7.0$ Hz, 2H, $\text{PCH}(\text{CH}_3)_2$), 2.43 (m, $^3J_{\text{HH}} = 7.0$ Hz, 2H, $\text{PCH}(\text{CH}_3)_2$), 1.69 (s, 3H, Rh- CH_3), 1.32 (dd, $^3J_{\text{PH}} = 17.5$ Hz, $^3J_{\text{HH}} = 6.9$ Hz, 6H, $\text{PCH}(\text{CH}_3)_2$), 1.24-1.09 (m, 18H, $\text{PCH}(\text{CH}_3)_2$).

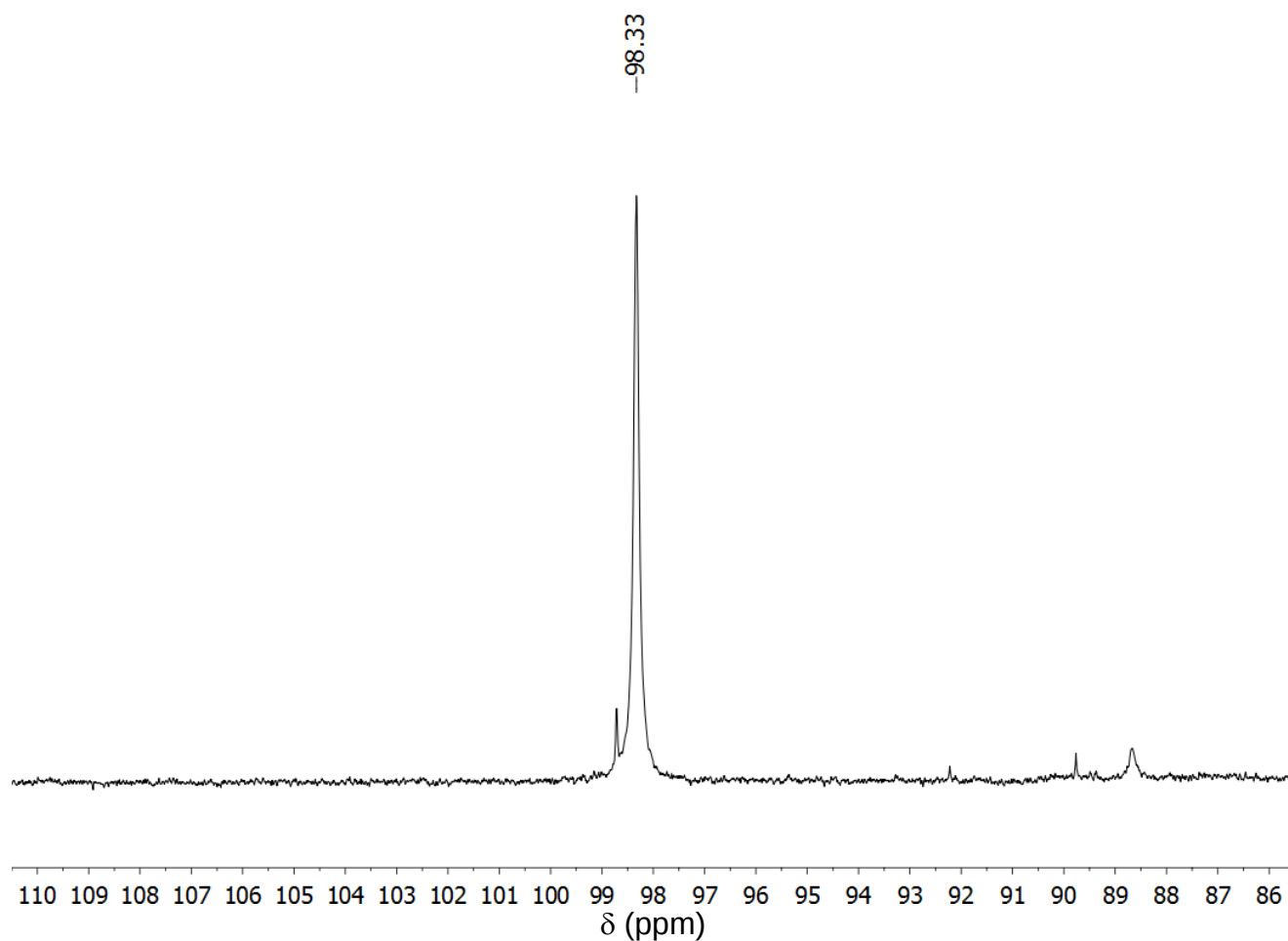


Figure S9. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of complex **8** in CD_2Cl_2 at 20°C. Residual impurities are also observed in this spectrum.

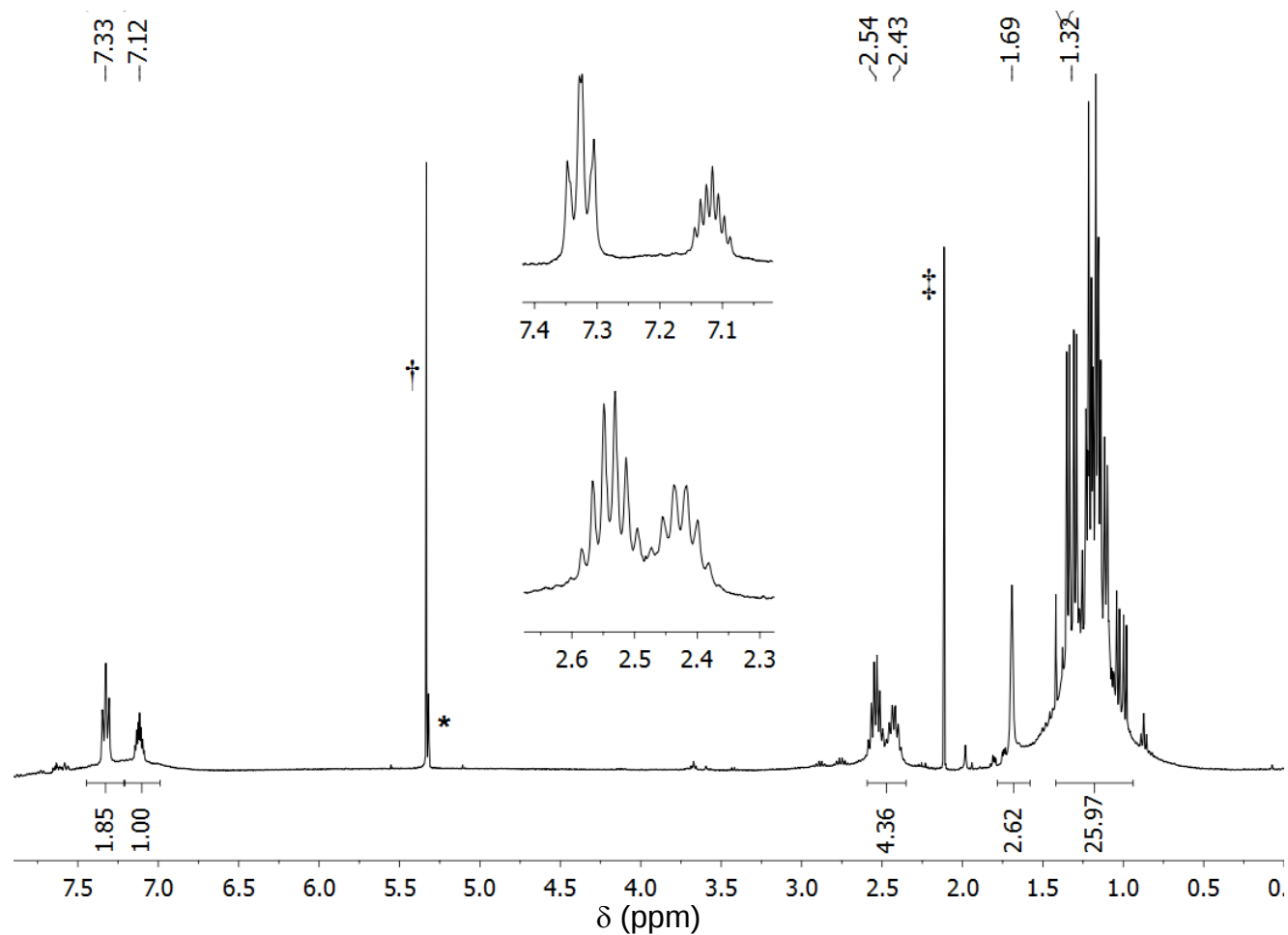


Figure S10. ^1H NMR spectrum of complex **8** in CD_2Cl_2 at 20°C . In addition to **8**, the following signals also appear in the spectrum: CHDCl_2 from the solvent (*), residual CH_2Cl_2 ($\dagger$), and residual CH_3CN ($\ddagger$).

Complex 9

$^{31}\text{P}\{^1\text{H}\}$ NMR (202 MHz, CD_3OD , -70°C): 81.36 (s). ^1H NMR (500 MHz, CD_3OD , -70°C): 7.57 (m, $^3J_{\text{HH}} = 7.8$ Hz, 2H, Ar-H), 7.30 (m, $^3J_{\text{HH}} = 7.7$ Hz, 1H, Ar-H), 2.86 (m, $^2J_{\text{PH}} = ^3J_{\text{HH}} = 6.7$ Hz, 2H, $\text{PCH}(\text{CH}_3)_2$), 2.73 (m, $^2J_{\text{PH}} = ^3J_{\text{HH}} = 7.2$ Hz, 2H, $\text{PCH}(\text{CH}_3)_2$), 1.35-1.21 (m, 18H, $\text{PCH}(\text{CH}_3)_2$), 1.26 (d, $^2J_{\text{RhH}} = 1.3$ Hz, 3H, Rh- CH_3), 1.14 (dd, $^3J_{\text{PH}} = 18.4$ Hz, $^3J_{\text{HH}} = 6.9$ Hz, 6H, $\text{PCH}(\text{CH}_3)_2$).

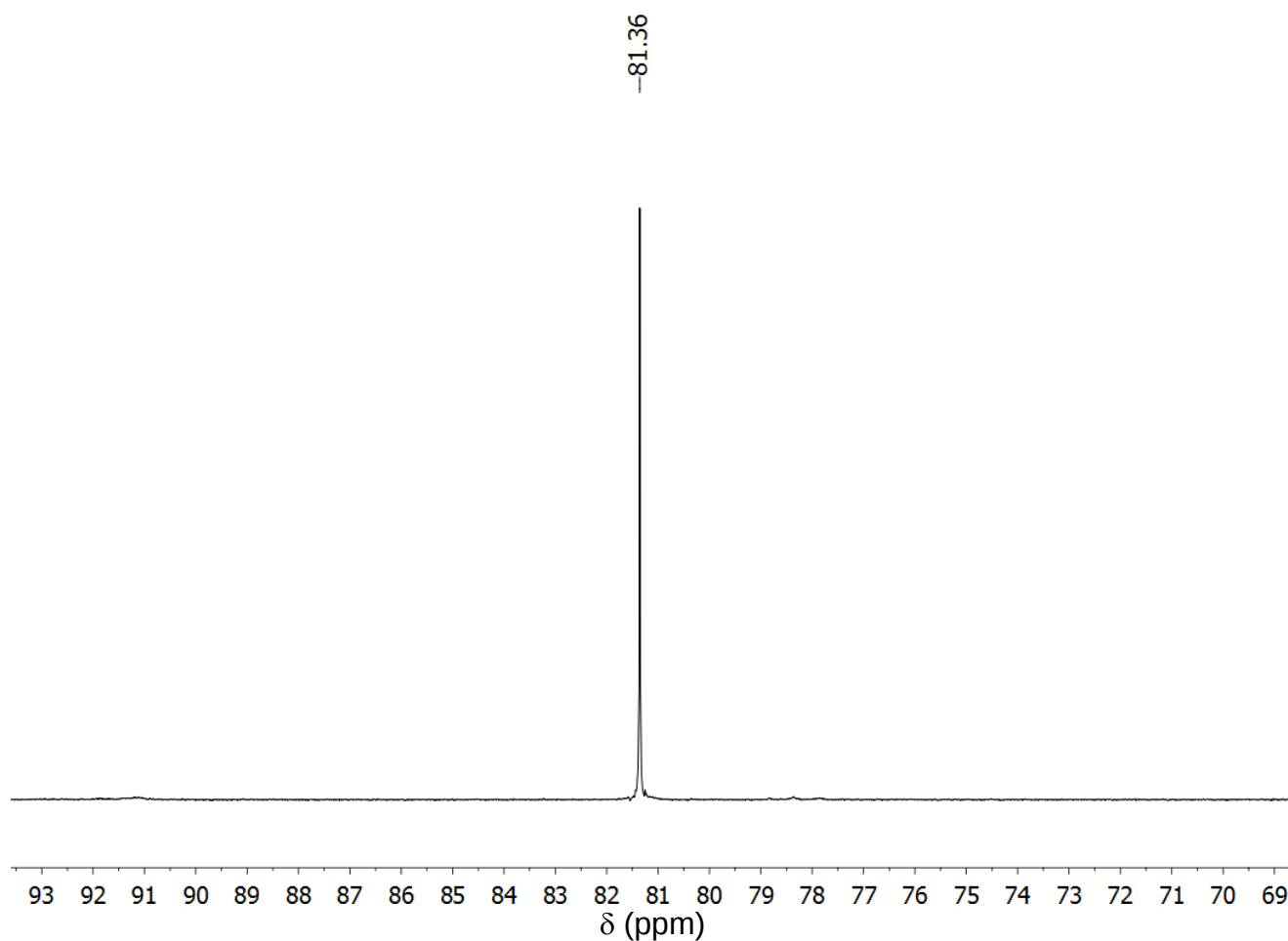


Figure S11. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of complex **9** in CH_3OD at -70°C .

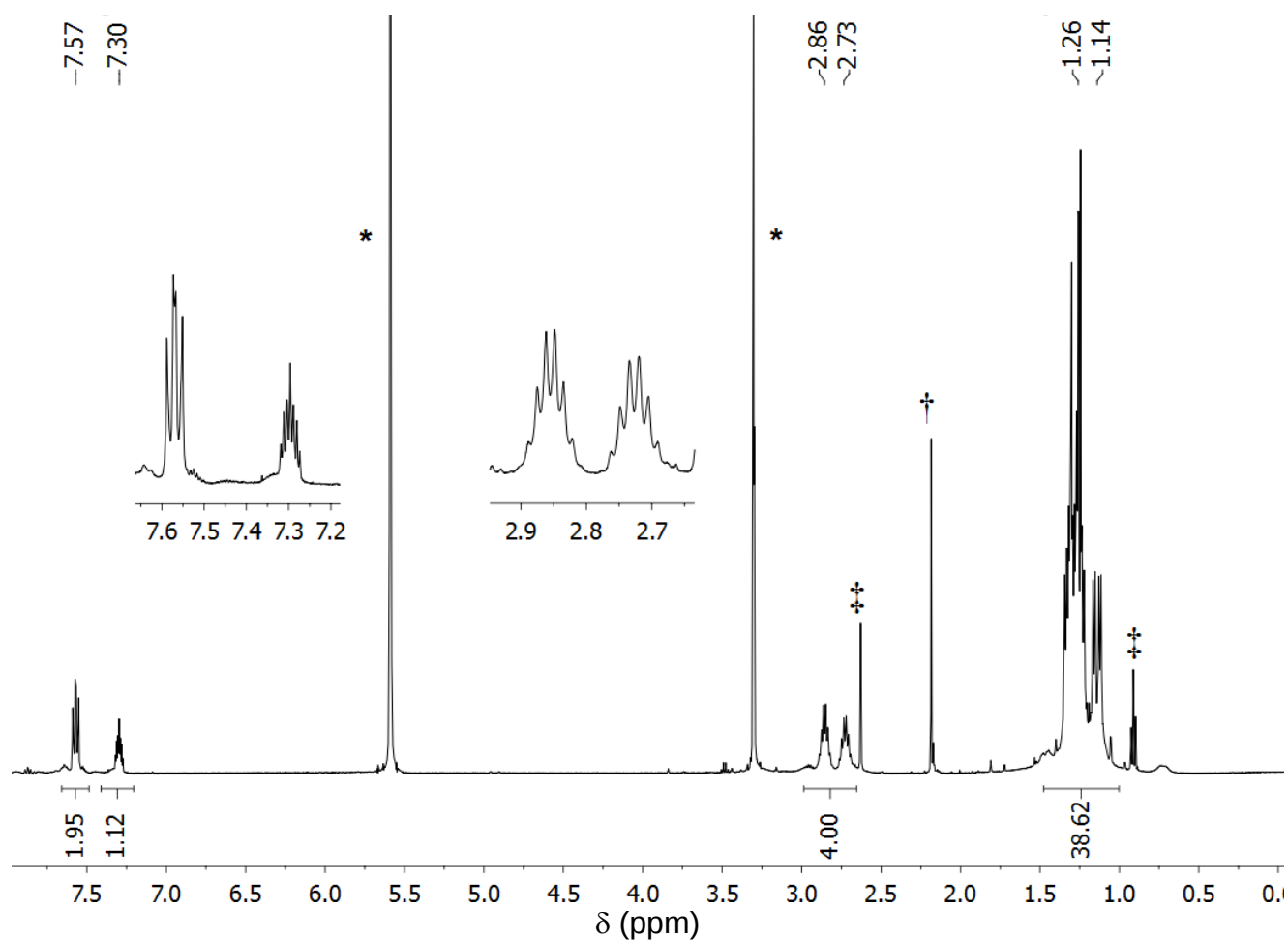


Figure S12. ^1H NMR spectrum of complex **9** in CH_3OD at -70°C . In addition to **9**, the following signals also appear in the spectrum: residual signals from the deuterated solvent (*), acetone from complex **2** used to prepare **9** (†), and unidentified impurities (‡).

Complex 10

$^{31}\text{P}\{^1\text{H}\}$ NMR (202 MHz, CD_3OD , -40°C): 100.08 (s). ^1H NMR (500 MHz, CD_3OD , -40°C): 7.81 (m, $^3J_{\text{HH}} = 7.9$ Hz, 2H, Ar-H), 7.44 (m, $^3J_{\text{HH}} = 7.7$ Hz, 1H, Ar-H), 2.85 (m, $^2J_{\text{PH}} = ^3J_{\text{HH}} = 7.1$ Hz, 2H, $\text{PCH}(\text{CH}_3)_2$), 2.70 (m, $^2J_{\text{PH}} = ^3J_{\text{HH}} = 7.2$ Hz, 2H, $\text{PCH}(\text{CH}_3)_2$), 1.31 (dd, $^3J_{\text{PH}} = 17.7$ Hz, $^3J_{\text{HH}} = 7.1$ Hz, 6H, $\text{PCH}(\text{CH}_3)_2$), 1.30 (dd, $^3J_{\text{PH}} = 17.9$ Hz, $^3J_{\text{HH}} = 7.0$ Hz, 6H, $\text{PCH}(\text{CH}_3)_2$), 1.20 (dd, $^3J_{\text{PH}} = 18.0$ Hz, $^3J_{\text{HH}} = 6.7$ Hz, 6H, $\text{PCH}(\text{CH}_3)_2$), 1.18 (dd, $^3J_{\text{PH}} = 18.3$ Hz, $^3J_{\text{HH}} = 6.6$ Hz, 6H, $\text{PCH}(\text{CH}_3)_2$), 0.99 (d, $^2J_{\text{RhH}} = 2.2$ Hz, 3H, Rh- CH_3).

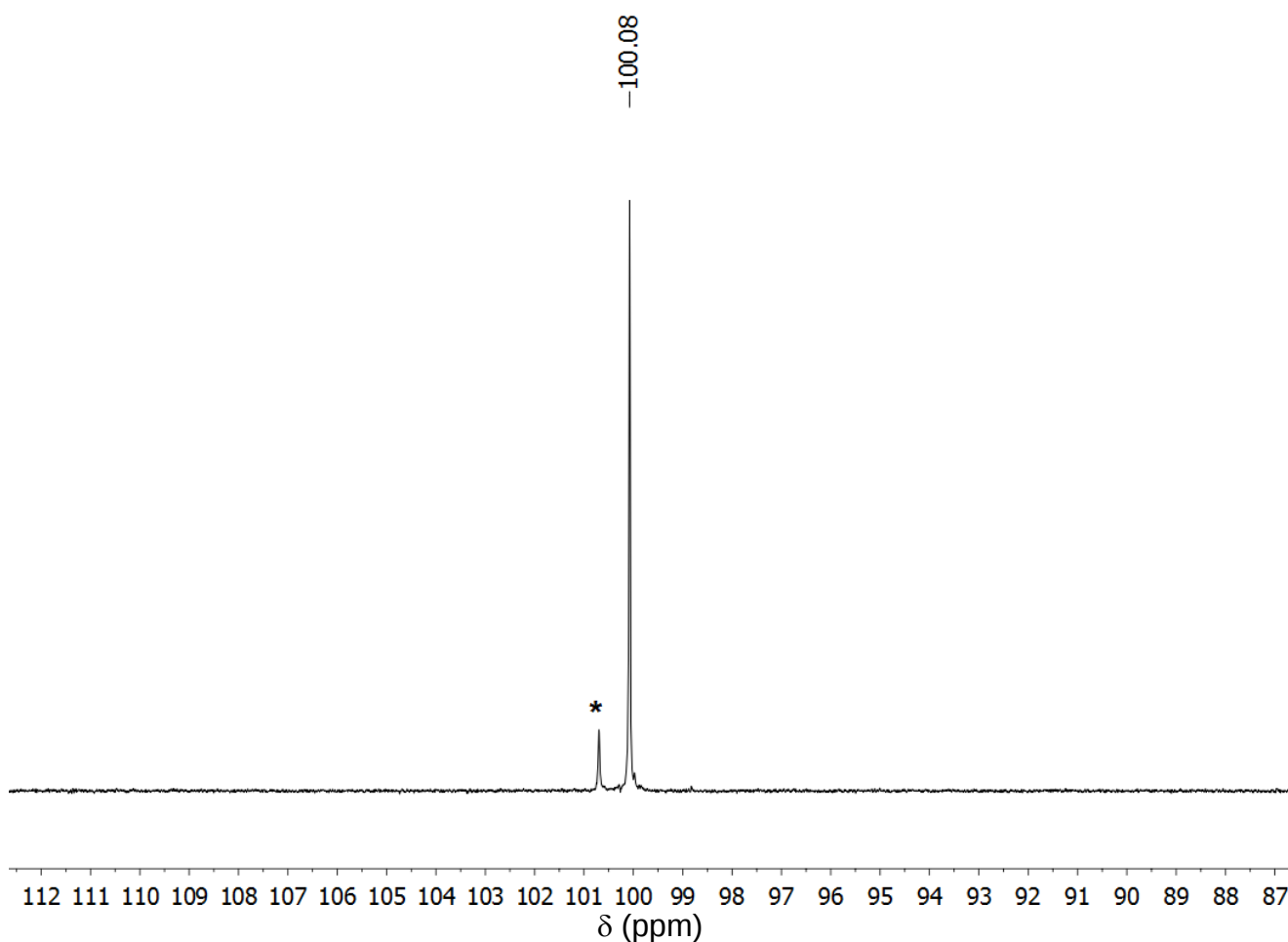


Figure S13. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of complex **10** in CH_3OD at -40°C . An unidentified impurity also appears in the spectrum (*).

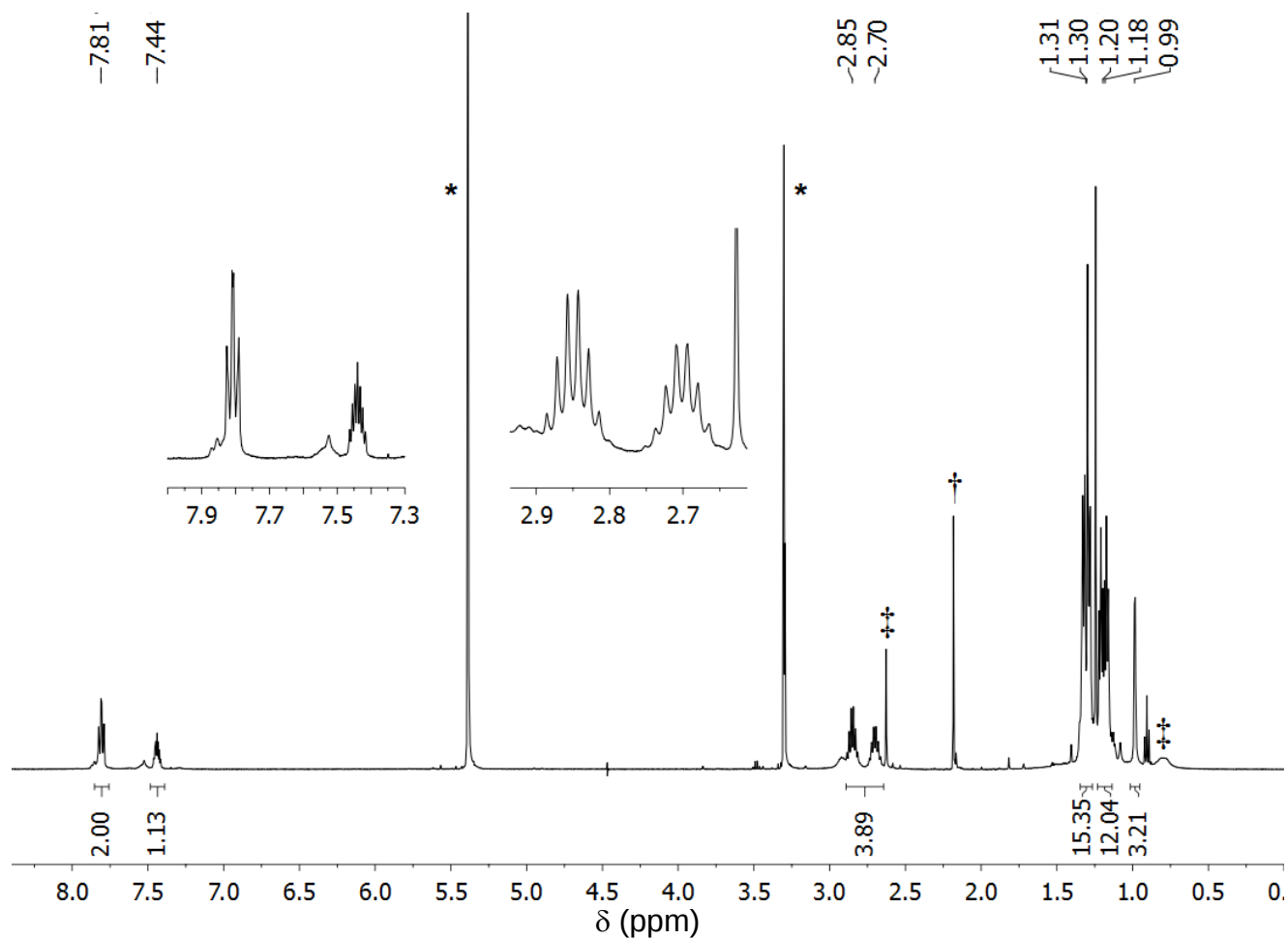


Figure S14. ^1H NMR spectrum of complex **10** in CH_3OD at -40°C . In addition to **10**, the following signals also appear in the spectrum: residual signals from the deuterated solvent (*), acetone from complex **2** used to prepare **10** (†), and unidentified impurities (‡).

Part II: Computed atomic properties for SCS and PCP complexes

Table S1. Atomic properties for selected atoms of the PCP and SCS complexes

Net atomic charges q(A)													
	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh	0.19	0.36	0.40	0.45	0.32	0.42	0.39	0.33	0.53	0.59	0.60	0.48	0.56
C _{ipso}	-0.13	-0.15	-0.24	-0.13	-0.08	-0.08	0.03	-0.03	-0.07	-0.15	-0.04	-0.04	-0.19
C _{ortho}	-0.02	-0.02	-0.02	-0.01	-0.65	-0.66	-0.65	-0.64	-0.69	-0.70	-0.68	-0.68	-0.70
C _{meta}	-0.02	-0.03	-0.01	-0.03	0.01	0.00	0.00	0.01	0.00	0.00	0.01	0.00	0.00
C _{para}	0.00	-0.02	0.01	-0.02	0.02	0.01	0.00	0.01	0.00	0.01	0.01	0.00	0.00
H _{phenyl}	0.05	0.03	0.03	0.04	0.08	0.07	0.06	0.08	0.06	0.03	0.03	0.06	0.06
H _{phenyl}	0.05	0.03	0.03	0.04	0.08	0.07	0.07	0.08	0.07	0.06	0.06	0.06	0.07
H _{phenyl}	0.06	0.04	0.04	0.04	0.09	0.08	0.07	0.08	0.06	0.03	0.03	0.07	0.06
P/S	1.80	1.81	1.85	1.87	-0.78	-0.79	-0.73	-0.76	-0.81	-0.69	-0.72	-0.77	-0.75
C _{CH₃}	0.00	-0.05	-0.04	-0.16	0.02	0.05	0.05	-0.03	-0.05	-0.04	-0.12	-0.15	-0.16
H _{CH₃}	0.05	0.00	0.01	-0.02	0.03	0.01	0.01	0.07	0.01	0.00	0.00	0.05	0.06
H _{CH₃}	0.05	0.01	0.04	-0.01	0.06	0.03	0.04	0.07	0.01	0.03	-0.03	0.05	0.05
H _{CH₃}	0.00	0.00	0.01	-0.04	0.03	0.03	0.01	0.01	0.01	0.03	0.00	0.05	0.04
C _{CO}	1.06		1.09	1.12	1.11	1.09	1.08	1.11		1.15	1.14	1.12	1.19
O _{CO}	-1.22		-1.17	-1.18	-1.21	-1.21	-1.22	-1.20		-1.15	-1.17	-1.19	-1.19

Lagrangian of Atom A L(A) = - ¹ / ₄ Atomic Integral of the Laplacian of the Electron Density													
	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh	0.00	40.69	40.43	40.43	40.74	40.65	40.68	40.72	40.69	40.42	40.46	0.00	0.00
C _{ipso}	0.00	37.90	37.91	37.90	37.90	37.89	37.86	37.87	37.86	37.87	37.84	0.00	0.00
C _{ortho}	0.00	37.90	37.90	37.91	38.18	38.19	38.17	38.17	38.22	38.22	38.21	0.00	0.00
C _{meta}	0.00	37.90	37.89	37.90	37.89	37.90	37.90	37.89	37.89	37.88	37.88	0.00	0.00
C _{para}	0.00	37.89	37.88	37.90	37.89	37.90	37.89	37.89	37.89	37.88	37.89	0.00	0.00
H _{phenyl}	0.00	0.59	0.60	0.59	0.58	0.58	0.59	0.58	0.58	0.60	0.60	0.00	0.00
H _{phenyl}	0.00	0.59	0.60	0.59	0.58	0.58	0.59	0.58	0.58	0.59	0.58	0.00	0.00
H _{phenyl}	0.00	0.59	0.59	0.59	0.57	0.58	0.58	0.57	0.58	0.60	0.60	0.00	0.00
P/S	0.00	340.4	340.4	340.3	398.2	398.2	398.1	398.1	398.2	398.1	398.2	0.00	0.00
C _{CH₃}	0.00	37.76	37.74	37.78	37.80	37.79	37.81	37.82	37.77	37.75	37.79	0.00	0.00
H _{CH₃}	0.00	0.61	0.61	0.60	0.60	0.61	0.61	0.58	0.60	0.61	0.60	0.00	0.00
H _{CH₃}	0.00	0.59	0.59	0.61	0.58	0.59	0.59	0.58	0.60	0.59	0.62	0.00	0.00
H _{CH₃}	0.00	0.61	0.61	0.62	0.60	0.60	0.61	0.61	0.59	0.59	0.60	0.00	0.00
C _{CO}	0.00		37.09	37.09	37.12	37.11	37.17	37.12		37.09	37.07	0.00	0.00
O _{CO}	0.00		75.82	75.84	75.82	75.83	75.81	75.82		75.83	75.84	0.00	0.00

Percentage of average number of electrons in atom A that are localized in atom A
 $\%Loc(A) = 100 * LI(A) / N(A)$

	PCP complexes				SCS complexes								
%Loc(A)	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh	85.89	85.96	83.56	83.78	87.01	85.74	86.85	86.81	86.36	84.28	84.50	84.88	84.15
C _{ipso}	64.65	65.19	66.06	65.01	64.48	64.37	64.25	64.21	64.81	66.19	64.60	64.10	66.35
C _{ortho}	64.59	64.77	64.76	64.73	68.18	68.21	68.29	68.23	68.50	68.43	68.44	68.37	68.42
C _{meta}	65.79	65.82	65.76	65.81	65.70	65.59	65.63	65.70	65.68	65.59	65.58	65.63	65.71
C _{para}	65.82	65.83	65.75	65.84	65.89	65.85	65.87	65.91	65.92	65.82	65.83	65.94	65.91
H _{phenyl}	42.26	42.93	43.33	42.89	40.47	40.86	40.93	40.66	41.72	42.91	42.72	41.70	41.77
H _{phenyl}	42.27	42.93	43.33	42.86	40.46	40.83	41.00	40.67	41.79	42.43	42.36	41.70	41.87
H _{phenyl}	42.38	43.01	43.01	43.03	41.23	41.68	41.88	41.37	41.72	42.91	42.78	41.82	41.76
P/S	86.04	86.25	86.04	86.08	92.26	92.28	92.27	92.34	92.54	92.17	92.30	92.39	92.29
C _{CH₃}	65.94	66.19	66.28	66.72	66.00	65.94	65.86	65.95	66.19	66.28	66.47	66.41	66.90
H _{CH₃}	41.56	43.23	42.97	43.90	41.08	41.61	41.23	40.43	42.84	42.89	42.91	40.92	40.29
H _{CH₃}	41.56	43.34	41.68	42.97	41.57	43.11	42.61	40.43	42.86	42.28	43.46	40.93	41.85
H _{CH₃}	40.31	43.23	42.97	43.53	40.69	41.55	41.76	38.73	43.32	42.28	43.60	39.46	41.85
C _{CO}	67.54		70.37	70.58	67.64	68.51	67.31	67.77		70.53	71.73	69.51	65.20
O _{CO}	90.21		90.07	90.10	90.22	90.23	90.21	90.22		90.09	90.11	90.21	89.96

Magnitudes of intra-atomic dipole moments $|\mu_{intra}(A)|$

	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh	0.34	0.05	0.09	0.12	0.29	0.40	0.40	0.34	0.16	0.11	0.19	0.26	0.26
C _{ipso}	0.13	0.25	0.47	0.19	0.18	0.26	0.09	0.09	0.29	0.55	0.25	0.21	0.56
C _{ortho}	0.09	0.06	0.07	0.06	1.03	1.04	1.02	1.03	1.04	1.03	1.05	1.05	1.03
C _{meta}	0.13	0.12	0.11	0.12	0.15	0.15	0.14	0.14	0.12	0.09	0.10	0.13	0.12
C _{para}	0.13	0.12	0.12	0.13	0.17	0.17	0.17	0.17	0.17	0.15	0.15	0.17	0.16
H _{phenyl}	0.13	0.13	0.13	0.13	0.12	0.12	0.12	0.12	0.12	0.12	0.12	0.12	0.12
H _{phenyl}	0.13	0.13	0.13	0.13	0.12	0.12	0.12	0.12	0.12	0.12	0.12	0.12	0.12
H _{phenyl}	0.13	0.13	0.12	0.13	0.12	0.12	0.12	0.12	0.12	0.12	0.12	0.12	0.12
P/S	1.44	1.49	1.41	1.41	0.57	0.60	0.40	0.48	0.54	0.49	0.58	0.52	0.53
C _{CH₃}	0.08	0.12	0.08	0.31	0.11	0.11	0.09	0.06	0.08	0.06	0.24	0.11	0.14
H _{CH₃}	0.13	0.14	0.13	0.14	0.12	0.13	0.12	0.12	0.14	0.14	0.15	0.14	0.14
H _{CH₃}	0.13	0.14	0.14	0.14	0.13	0.13	0.13	0.12	0.14	0.14	0.14	0.14	0.14
H _{CH₃}	0.11	0.14	0.13	0.13	0.13	0.11	0.11	0.09	0.14	0.14	0.14	0.14	0.14
C _{CO}	1.27		1.41	1.41	1.27	1.31	1.25	1.27		1.39	1.44	1.36	1.12
O _{CO}	0.77		0.87	0.85	0.79	0.79	0.76	0.79		0.89	0.89	0.83	0.75

Magnitudes of bonding dipole moment contributions |Mu_Bond(A)|

	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh	0.14	0.76	0.49	0.67	0.25	0.25	0.35	0.20	0.31	0.36	0.46	0.15	0.07
C _{ipso}	0.30	0.73	0.70	0.53	0.05	0.09	0.21	0.04	0.25	0.39	0.22	0.11	0.24
C _{ortho}	0.06	0.19	0.11	0.15	1.36	1.36	1.20	1.32	1.45	1.54	1.44	1.47	1.58
C _{meta}	0.04	0.10	0.04	0.09	0.12	0.11	0.11	0.12	0.06	0.01	0.03	0.06	0.07
C _{para}	0.06	0.01	0.03	0.01	0.13	0.11	0.10	0.11	0.08	0.06	0.07	0.08	0.08
H _{phenyl}	0.03	0.02	0.01	0.02	0.04	0.04	0.03	0.04	0.03	0.01	0.01	0.03	0.03
H _{phenyl}	0.03	0.02	0.01	0.02	0.05	0.04	0.04	0.04	0.04	0.03	0.03	0.03	0.04
H _{phenyl}	0.03	0.02	0.02	0.02	0.05	0.05	0.04	0.05	0.03	0.01	0.01	0.04	0.03
P/S	0.62	0.65	0.64	0.50	1.72	1.76	1.41	1.61	1.69	1.40	1.57	1.63	1.59
C _{CH₃}	0.14	0.25	0.14	0.65	0.07	0.03	0.16	0.09	0.23	0.15	0.45	0.14	0.17
H _{CH₃}	0.03	0.01	0.00	0.02	0.04	0.04	0.01	0.03	0.01	0.01	0.01	0.03	0.04
H _{CH₃}	0.03	0.00	0.02	0.03	0.03	0.01	0.03	0.06	0.01	0.01	0.02	0.03	0.02
H _{CH₃}	0.11	0.01	0.00	0.04	0.03	0.06	0.02	0.01	0.00	0.01	0.01	0.03	0.02
C _{CO}	1.18		1.05	1.04	1.09	1.12	1.16	1.08		0.91	0.93	1.04	0.94
O _{CO}	1.78		1.70	1.72	1.76	1.76	1.78	1.75		1.66	1.68	1.73	1.74

Average number of electron pairs formed in atom A $D2(A,A) = (1/2)[N(A) * N(A) - LI(A)]$

	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh	134.0	131.2	130.9	130.0	131.8	130.4	130.7	131.7	128.5	127.8	127.6	129.4	128.2
C _{ipso}	16.78	16.91	17.42	16.81	16.51	16.51	15.92	16.25	16.44	16.89	16.30	16.29	17.08
C _{ortho}	16.17	16.17	16.15	16.12	19.83	19.92	19.87	19.79	20.11	20.15	20.04	20.03	20.14
C _{meta}	16.14	16.19	16.08	16.17	16.00	16.03	16.04	16.00	16.04	16.01	15.97	16.02	16.05
C _{para}	16.05	16.14	15.98	16.15	15.94	15.98	16.01	15.98	16.02	15.95	15.94	16.05	16.02
H _{phenyl}	0.25	0.26	0.26	0.26	0.24	0.24	0.25	0.24	0.25	0.26	0.26	0.25	0.25
H _{phenyl}	0.25	0.26	0.26	0.26	0.24	0.24	0.25	0.24	0.24	0.24	0.24	0.25	0.24
H _{phenyl}	0.24	0.26	0.25	0.26	0.23	0.23	0.24	0.23	0.25	0.26	0.26	0.24	0.25
P/S	81.5	81.3	80.8	80.6	133.1	133.2	132.3	132.7	133.6	131.6	132.1	132.9	132.6
C _{CH₃}	16.04	16.32	16.23	16.94	15.93	15.72	15.76	16.22	16.27	16.24	16.71	16.87	16.93
H _{CH₃}	0.25	0.28	0.28	0.29	0.27	0.28	0.28	0.25	0.28	0.28	0.28	0.26	0.25
H _{CH₃}	0.25	0.28	0.26	0.30	0.25	0.26	0.26	0.25	0.28	0.27	0.31	0.26	0.26
H _{CH₃}	0.30	0.28	0.28	0.31	0.28	0.27	0.28	0.30	0.27	0.27	0.28	0.26	0.26
C _{CO}	10.51		10.32	10.21	10.31	10.36	10.45	10.30		10.05	10.06	10.20	10.02
O _{CO}	38.33		37.91	38.04	38.21	38.27	38.33	38.19		37.74	37.88	38.07	38.06

$D2(A, Mol) = D2(A, A) + D2(A, A')$, where $D2(A, A') = (1/2)[N(A) * N(A') - DI(A, A') / 2]$ is half of average number of electron pairs formed between atom a and other atoms of molecule

	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh	1336.1	1206.	1319.7	1315.4	1459.4	1600.4	1602.8	1458.8	1325.7	1436.2	1435.3	1445.1	1438.5
C _{ipso}	487.0	445.9	496.2	487.5	531.7	586.5	576.5	527.7	488.4	538.3	528.6	528.3	541.2
C _{ortho}	478.4	436.4	478.4	477.9	581.6	642.9	642.2	581.2	538.9	586.2	584.6	584.6	586.1
C _{meta}	478.6	437.1	477.8	479.0	524.5	578.9	579.2	524.5	483.2	524.6	524.1	524.8	525.3
C _{para}	477.4	436.5	476.4	478.7	523.7	578.2	578.8	524.3	483.0	523.9	523.8	525.4	524.9
H _{phenyl}	75.49	70.01	77.42	76.62	80.56	89.88	90.28	80.90	75.63	84.98	84.63	82.19	82.31
H _{phenyl}	75.49	70.01	77.42	76.58	80.68	89.96	90.20	80.91	74.67	82.49	82.40	82.19	81.33
H _{phenyl}	74.84	69.66	76.10	76.51	79.63	89.13	89.77	80.10	75.63	84.98	84.71	81.38	82.31
P/S	1049.	956.2	1045.	1044.	1468.	1620.	1615.	1466.	1354.	1460.	1463.	1468.	1466.
C _{CH₃}	477.2	438.9	480.1	490.0	523.6	573.9	574.5	528.0	486.7	528.6	535.8	538.1	539.3
H _{CH₃}	75.50	72.43	79.00	80.73	84.70	95.27	95.20	81.59	79.98	87.29	87.43	83.08	81.85
H _{CH₃}	75.51	71.74	76.55	80.59	82.49	93.78	92.68	81.55	80.02	85.20	90.16	83.12	83.56
H _{CH₃}	79.44	72.43	79.00	82.29	85.17	93.81	95.58	86.44	79.47	85.20	87.78	83.23	83.56
C _{CO}	392.4		390.3	388.3	428.0	473.5	474.8	427.9		424.3	425.1	426.8	421.2
O _{CO}	732.8		729.0	730.2	805.4	888.9	889.5	805.2		800.6	802.0	804.0	803.7

Electron localization index $LI(A) =$ average number of electrons localized in atom A

	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh	14.43	14.30	13.87	13.86	14.51	14.22	14.43	14.47	14.22	13.83	13.86	14.02	13.83
C _{ipso}	3.96	4.01	4.12	3.99	3.92	3.91	3.84	3.87	3.93	4.07	3.90	3.87	4.10
C _{ortho}	3.89	3.90	3.90	3.89	4.53	4.54	4.54	4.53	4.59	4.59	4.57	4.57	4.58
C _{meta}	3.96	3.97	3.95	3.96	3.94	3.93	3.94	3.94	3.94	3.93	3.93	3.94	3.94
C _{para}	3.95	3.96	3.94	3.96	3.94	3.95	3.95	3.95	3.96	3.94	3.94	3.96	3.95
H _{phenyl}	0.40	0.41	0.42	0.41	0.37	0.38	0.38	0.38	0.39	0.42	0.41	0.39	0.39
H _{phenyl}	0.40	0.41	0.42	0.41	0.37	0.38	0.38	0.38	0.39	0.40	0.40	0.39	0.39
H _{phenyl}	0.40	0.41	0.41	0.41	0.38	0.38	0.39	0.38	0.39	0.42	0.41	0.39	0.39
P/S	11.36	11.37	11.32	11.31	15.49	15.49	15.44	15.48	15.56	15.38	15.44	15.50	15.46
C _{CH₃}	3.96	4.01	4.00	4.11	3.95	3.92	3.92	3.98	4.00	4.00	4.07	4.08	4.12
H _{CH₃}	0.39	0.43	0.43	0.45	0.40	0.41	0.41	0.38	0.43	0.43	0.43	0.39	0.38
H _{CH₃}	0.39	0.43	0.40	0.44	0.39	0.42	0.41	0.38	0.43	0.41	0.45	0.39	0.40
H _{CH₃}	0.40	0.43	0.43	0.45	0.40	0.40	0.41	0.38	0.43	0.41	0.44	0.38	0.40
C _{CO}	3.33		3.45	3.45	3.31	3.36	3.31	3.31		3.42	3.49	3.39	3.14
O _{CO}	8.32		8.26	8.27	8.31	8.31	8.32	8.30		8.24	8.26	8.29	8.26

Average number of electrons in Vol(A)=0.001 au N(Vol(A)),0.001

	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh	44.79	44.62	44.59	44.53	44.66	44.58	44.59	44.65	44.45	44.40	44.39	44.50	44.43
C _{ipso}	6.12	6.14	6.24	6.13	6.07	6.08	5.97	6.03	6.06	6.15	6.04	6.04	6.18
C _{ortho}	6.01	6.01	6.01	6.00	6.64	6.65	6.65	6.63	6.69	6.69	6.67	6.68	6.69
C _{meta}	6.01	6.01	5.99	6.01	5.98	5.99	5.99	5.98	5.99	5.98	5.98	5.98	5.99
C _{para}	5.99	6.00	5.97	6.00	5.97	5.97	5.98	5.97	5.98	5.97	5.97	5.98	5.98
H _{phenyl}	0.93	0.95	0.95	0.94	0.91	0.92	0.92	0.91	0.92	0.96	0.95	0.92	0.93
H _{phenyl}	0.93	0.95	0.95	0.94	0.91	0.92	0.92	0.91	0.91	0.92	0.92	0.92	0.91
H _{phenyl}	0.92	0.94	0.94	0.94	0.89	0.90	0.91	0.90	0.92	0.96	0.95	0.91	0.93
P/S	13.20	13.19	13.15	13.13	16.73	16.74	16.69	16.70	16.76	16.64	16.67	16.72	16.70
C _{CH₃}	6.00	6.05	6.03	6.16	5.98	5.94	5.95	6.03	6.04	6.03	6.11	6.14	6.15
H _{CH₃}	0.93	0.98	0.98	0.99	0.96	0.98	0.98	0.92	0.98	0.98	0.98	0.94	0.92
H _{CH₃}	0.93	0.97	0.95	1.00	0.93	0.95	0.94	0.92	0.98	0.96	1.02	0.94	0.94
H _{CH₃}	0.99	0.98	0.98	1.02	0.96	0.96	0.98	0.98	0.97	0.96	0.98	0.94	0.94
C _{CO}	4.93		4.90	4.88	4.88	4.90	4.91	4.88		4.84	4.85	4.87	4.81
O _{CO}	9.17		9.13	9.15	9.16	9.17	9.18	9.16		9.11	9.13	9.15	9.15

Traceless form of quadrupole moment tensor of atom A's electronic charge density distribution, using the nucleus of atom A as origin Q(A)

$Q_{xx}^A = -[3\langle\rho_{x^2}\rangle^A - \langle\rho_{r^2}\rangle^A]$													
	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh	2.15	0.96	-0.08	-0.37	2.52	1.00	2.42	2.70	1.44	0.65	0.22	1.28	0.78
C _{ipso}	1.46	2.00	2.20	1.50	1.34	1.09	1.27	1.31	1.71	1.96	1.56	1.17	2.24
C _{ortho}	1.30	1.55	1.54	1.20	0.00	-0.40	-0.02	0.09	0.13	0.04	0.03	0.07	0.09
C _{meta}	1.57	1.67	1.61	1.30	1.27	1.24	1.33	1.30	1.36	1.45	1.42	1.33	1.38
C _{para}	1.82	1.92	1.81	1.61	1.82	1.65	1.90	1.90	1.95	1.89	1.92	1.99	1.93
H _{phenyl}	0.29	0.30	0.30	0.26	0.33	0.31	0.31	0.31	0.29	0.31	0.30	0.29	0.30
H _{phenyl}	0.29	0.30	0.30	0.27	0.30	0.30	0.33	0.31	-0.17	-0.17	-0.17	0.29	-0.17
H _{phenyl}	-0.16	-0.17	-0.17	-0.17	-0.16	-0.17	-0.16	-0.16	0.29	0.31	0.31	-0.17	0.30
P/S	-2.06	-2.09	-1.93	-1.88	1.37	0.99	1.09	1.73	1.90	1.66	1.74	1.97	1.95
C _{CH₃}	-0.23	0.11	0.05	0.38	-0.22	-0.23	-0.15	-0.30	0.14	-0.01	0.40	0.11	0.17
H _{CH₃}	0.27	0.29	0.29	-0.15	0.42	0.20	0.19	0.27	0.27	-0.23	0.12	0.24	-0.22
H _{CH₃}	0.27	-0.22	-0.24	0.17	-0.02	-0.21	-0.21	0.27	0.27	0.25	0.44	0.24	0.23
H _{CH₃}	-0.35	0.29	0.29	0.43	-0.09	0.31	0.40	-0.30	-0.22	0.24	-0.20	-0.21	0.23
C _{CO}	-0.97		-1.06	-0.87	-0.78	-0.79	-0.80	-0.79		-0.77	-0.57	-0.64	-1.25
O _{CO}	-0.07		-0.01	0.03	-0.04	-0.05	-0.09	-0.04		0.03	0.04	0.00	-0.10

$$Q_{xy}^A = -3\langle\rho_{xy}\rangle^A$$

	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh	0.00	0.00	0.01	0.02	-0.54	-1.33	-0.61	-0.01	0.02	0.00	-0.02	-0.01	-0.03
C _{ipso}	0.00	0.00	0.00	0.09	-0.13	-0.67	0.12	0.00	0.00	0.00	-0.02	0.00	0.00
C _{ortho}	0.32	0.06	0.01	0.15	-0.93	-1.27	1.14	-1.03	1.72	1.65	-1.67	1.63	-1.67
C _{meta}	0.36	-0.21	-0.17	-0.12	0.25	-0.01	-0.33	0.39	-0.22	-0.16	0.17	-0.27	0.20
C _{para}	0.00	0.00	0.00	0.09	-0.19	-0.67	0.09	0.00	0.00	0.00	-0.01	0.00	0.00
H _{phenyl}	-0.14	0.25	0.25	0.25	-0.07	-0.16	0.14	-0.09	0.24	0.24	-0.25	0.22	-0.24
H _{phenyl}	0.14	-0.25	-0.25	-0.23	0.05	0.03	-0.10	0.09	0.00	0.00	0.01	-0.22	0.00
H _{phenyl}	0.00	0.00	0.00	0.00	-0.02	0.01	-0.02	0.00	-0.24	-0.24	0.23	0.00	0.24
P/S	-1.70	-1.94	-1.78	-2.00	1.13	0.63	-1.12	1.20	1.03	1.34	-1.03	-1.00	-1.54
C _{CH₃}	0.00	0.00	0.00	0.05	-0.07	-0.03	-0.03	0.00	0.00	0.00	0.04	0.00	0.00
H _{CH₃}	0.15	0.18	0.28	0.18	-0.02	-0.31	0.30	0.16	0.23	0.00	-0.29	-0.26	0.00
H _{CH₃}	-0.15	0.00	0.00	-0.32	-0.22	0.06	-0.06	-0.16	-0.23	0.26	0.25	0.26	-0.26
H _{CH₃}	0.00	-0.18	-0.28	0.12	0.22	0.31	-0.30	0.00	0.00	-0.26	0.10	0.00	0.26
C _{CO}	0.00		0.00	0.02	-0.29	-0.60	-0.56	0.00		0.00	0.10	0.00	0.00
O _{CO}	0.00		0.00	0.01	-0.02	-0.02	-0.04	0.00		0.00	-0.02	0.00	0.00

$$Q_{xz}^A = -3\langle\rho_{xz}\rangle^A$$

	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh	0.00	-0.01	0.00	-0.29	-1.67	-1.94	0.06	0.00	-0.01	0.01	0.15	0.01	-0.01
C _{ipso}	0.00	0.00	0.00	-1.29	0.13	0.40	-0.01	0.00	0.00	0.00	0.06	0.00	0.00
C _{ortho}	0.41	-0.39	-0.39	-1.33	-1.57	-1.07	1.17	-1.39	0.13	0.06	-0.18	0.71	-0.18
C _{meta}	-0.01	-0.15	-0.16	-1.31	0.12	0.40	0.01	-0.01	0.00	0.02	0.08	0.13	-0.05
C _{para}	0.00	0.00	0.00	-1.35	0.15	0.73	-0.09	0.00	0.00	0.00	0.07	0.00	0.00
H _{phenyl}	0.20	0.00	0.01	-0.13	-0.20	-0.18	0.20	-0.21	0.00	-0.01	0.00	0.10	-0.01
H _{phenyl}	-0.20	0.00	-0.01	-0.16	0.24	0.26	-0.20	0.21	0.00	0.00	0.00	-0.10	0.00
H _{phenyl}	0.00	0.00	0.00	-0.03	-0.01	0.03	-0.02	0.00	0.00	0.01	0.05	0.00	0.01
P/S	0.07	-0.22	-0.45	0.35	-1.23	-1.66	0.93	-0.50	-0.67	-0.71	0.37	0.13	0.73
C _{CH₃}	0.00	0.00	0.00	-0.12	-0.04	-0.02	-0.05	0.00	0.00	0.00	0.00	0.00	0.00
H _{CH₃}	0.22	-0.23	0.11	0.04	-0.17	-0.05	-0.01	-0.25	0.22	0.00	0.16	0.13	0.00
H _{CH₃}	-0.22	0.00	0.00	0.15	0.18	-0.04	0.07	0.25	-0.22	-0.13	-0.09	-0.13	0.14
H _{CH₃}	0.00	0.23	-0.11	-0.17	0.07	0.11	-0.05	0.00	0.00	0.13	0.03	0.00	-0.14
C _{CO}	0.00		0.00	-0.14	-0.05	0.18	-0.49	0.00		0.00	-0.42	0.00	0.00
O _{CO}	0.00		0.00	0.00	0.00	0.02	-0.04	0.00		0.00	0.02	0.00	0.00

$$Q_{yy}^A = -[3\langle\rho_{y^2} \rangle^A - \langle\rho_{1^2} \rangle^A]$$

Q _{YY} (A)	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh	2.38	0.30	2.01	-1.07	3.32	-0.10	-1.89	-0.32	0.27	1.76	-0.80	-0.35	2.23
C _{ipso}	-1.39	1.56	1.08	1.36	-1.87	-1.62	-0.98	-1.34	1.22	0.55	0.99	0.36	0.75
C _{ortho}	-0.59	1.74	1.71	1.72	-0.70	0.54	0.48	-0.38	2.42	2.52	2.45	1.88	2.42
C _{meta}	-0.97	1.87	1.81	1.84	-1.45	-0.35	-0.01	-0.88	1.71	1.72	1.64	1.34	1.73
C _{para}	-0.94	1.49	1.35	1.52	-1.48	-0.25	-0.05	-1.01	1.42	1.44	1.39	1.14	1.41
H _{phenyl}	-0.16	0.00	0.01	0.02	-0.20	-0.11	-0.12	-0.17	0.00	0.00	0.00	-0.03	0.00
H _{phenyl}	-0.16	0.00	0.01	-0.01	-0.21	-0.18	-0.13	-0.17	0.44	0.45	0.44	-0.03	0.44
H _{phenyl}	0.07	0.45	0.45	0.46	-0.02	0.15	0.21	0.06	0.00	0.00	-0.01	0.39	0.00
P/S	0.97	0.78	0.77	0.54	1.24	0.76	1.26	1.68	0.91	1.43	0.39	0.94	0.95
C _{CH₃}	-0.06	-0.12	0.06	0.16	-0.20	-0.15	0.02	0.01	-0.11	0.10	0.08	0.28	-0.42
H _{CH₃}	-0.22	-0.19	-0.09	0.34	-0.25	0.03	0.05	-0.21	-0.16	0.39	0.01	-0.08	-0.15
H _{CH₃}	-0.22	0.44	0.37	0.01	0.05	0.23	0.04	-0.20	-0.16	-0.08	-0.20	-0.08	-0.06
H _{CH₃}	0.37	-0.19	-0.09	-0.23	0.26	-0.07	-0.11	0.37	0.44	-0.08	0.33	0.29	-0.06
C _{CO}	1.92		2.03	-0.97	2.01	0.12	0.28	0.67		1.72	-0.91	-0.33	-0.05
O _{CO}	0.12		0.02	-0.01	0.10	0.00	0.07	0.04		-0.05	0.02	-0.01	-0.02

$$Q_{yz}^A = -3\langle\rho_{yz} \rangle^A$$

	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh	5.22	1.14	0.07	0.05	-3.94	-3.89	5.88	-6.42	-0.74	-0.13	-0.37	-4.22	-2.42
C _{ipso}	-1.81	-0.17	0.16	0.29	1.13	0.88	1.53	1.62	0.15	-0.11	0.54	0.99	0.23
C _{ortho}	-2.18	0.22	0.46	0.32	1.86	2.04	2.27	2.25	0.30	0.04	0.59	1.50	0.32
C _{meta}	-2.63	0.22	0.46	0.39	2.27	2.35	2.25	2.39	0.28	0.02	0.56	1.31	0.25
C _{para}	-2.25	0.27	0.48	0.31	2.09	2.09	2.17	2.29	0.26	-0.01	0.53	1.27	0.19
H _{phenyl}	-0.15	0.00	0.02	-0.05	0.12	0.15	0.15	0.14	0.02	0.01	0.04	0.08	0.02
H _{phenyl}	-0.15	0.00	0.02	0.08	0.13	0.12	0.12	0.14	0.04	0.00	0.08	0.08	0.03
H _{phenyl}	-0.36	0.04	0.07	0.05	0.33	0.35	0.34	0.35	0.02	0.01	0.03	0.19	0.02
P/S	0.08	-0.06	-0.13	0.17	0.47	0.31	-1.47	-0.02	0.89	0.84	1.44	0.80	1.36
C _{CH₃}	-0.24	-0.14	0.08	-0.22	0.20	0.17	0.34	0.40	0.14	0.06	-0.14	0.13	-0.30
H _{CH₃}	0.06	-0.11	0.09	0.18	0.00	0.02	-0.04	-0.07	0.11	0.18	-0.17	-0.08	-0.14
H _{CH₃}	0.06	-0.02	-0.19	-0.19	-0.22	-0.28	-0.30	-0.08	0.11	-0.08	-0.04	-0.08	-0.08
H _{CH₃}	0.19	-0.11	0.09	-0.04	0.02	0.06	0.03	-0.20	-0.02	-0.08	0.24	0.11	-0.08
C _{CO}	1.76		-0.38	-0.43	-1.37	-1.95	1.96	-2.11		0.17	-0.64	-1.73	-1.40
O _{CO}	0.10		-0.01	-0.01	-0.07	-0.09	0.13	-0.10		0.01	0.00	-0.03	-0.07

$$Q_{zz}^A = -[3\langle\rho_{z^2}\rangle^A - \langle\rho_{r^2}\rangle^A]$$

	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh	-4.53	-1.26	-1.93	1.43	-5.84	-0.91	-0.53	-2.38	-1.71	-2.40	0.58	-0.93	-3.01
C _{ipso}	-0.06	-3.56	-3.28	-2.85	0.53	0.52	-0.29	0.03	-2.93	-2.51	-2.55	-1.52	-3.00
C _{ortho}	-0.71	-3.29	-3.25	-2.92	0.70	-0.14	-0.46	0.30	-2.56	-2.56	-2.48	-1.95	-2.51
C _{meta}	-0.60	-3.54	-3.42	-3.14	0.18	-0.89	-1.31	-0.41	-3.08	-3.16	-3.07	-2.67	-3.12
C _{para}	-0.88	-3.41	-3.16	-3.12	-0.34	-1.40	-1.85	-0.89	-3.37	-3.32	-3.31	-3.13	-3.34
H _{phenyl}	-0.13	-0.30	-0.30	-0.27	-0.13	-0.20	-0.20	-0.14	-0.29	-0.31	-0.31	-0.26	-0.30
H _{phenyl}	-0.13	-0.30	-0.30	-0.26	-0.08	-0.12	-0.20	-0.14	-0.27	-0.28	-0.27	-0.26	-0.27
H _{phenyl}	0.09	-0.29	-0.28	-0.28	0.18	0.01	-0.04	0.10	-0.29	-0.31	-0.30	-0.22	-0.30
P/S	1.09	1.32	1.15	1.35	-2.61	-1.75	-2.35	-3.41	-2.81	-3.09	-2.13	-2.91	-2.89
C _{CH₃}	0.29	0.01	-0.11	-0.54	0.42	0.38	0.13	0.29	-0.03	-0.09	-0.48	-0.39	0.25
H _{CH₃}	-0.05	-0.10	-0.20	-0.19	-0.17	-0.23	-0.24	-0.06	-0.11	-0.16	-0.13	-0.16	0.37
H _{CH₃}	-0.05	-0.23	-0.12	-0.18	-0.03	-0.02	0.17	-0.06	-0.11	-0.17	-0.24	-0.16	-0.17
H _{CH₃}	-0.03	-0.10	-0.20	-0.21	-0.18	-0.24	-0.29	-0.07	-0.22	-0.17	-0.13	-0.08	-0.17
C _{CO}	-0.95		-0.97	1.84	-1.22	0.68	0.52	0.12		-0.96	1.47	0.97	1.30
O _{CO}	-0.05		-0.01	-0.02	-0.05	0.04	0.02	0.01		0.02	-0.06	0.02	0.12

Area of interatomic surface of atom A whose electron density is greater than 0.002,
Area_IAS(A),0.002

	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh	180.4 0	189.0 5	254.4 8	259.2 7	176.0 7	215.4 4	265.1 2	190.0 4	36.32	243.3 6	245.7 8	207.3 2	196.1 9
C _{ipso}	122.7 7	97.23	103.2 7	101.4 2	82.55	159.3 5	88.35	134.4 1	100.5 3	106.6 4	105.7 8	129.9 2	98.47
C _{ortho}	115.4 9	104.3 7	105.2 0	104.3 6	114.1 1	115.6 6	116.9 8	134.4 5	122.4 6	124.2 5	123.2 9	128.9 8	121.4 7
C _{meta}	101.8 2	97.93	97.25	97.44	93.70	96.97	96.09	112.5 1	107.2 0	107.9 1	108.0 1	108.2 6	105.8 1
C _{para}	91.06	92.26	91.19	92.23	89.56	90.49	93.27	89.93	90.55	90.64	90.68	90.85	90.61
H _{phenyl}	27.77	27.45	27.37	27.27	34.59	33.92	34.92	34.34	31.32	31.91	32.01	31.64	30.78
H _{phenyl}	27.77	27.45	27.37	27.29	NA	34.69	35.36	34.31	24.56	24.73	24.73	31.64	24.61
H _{phenyl}	24.79	25.23	24.94	25.31	24.16	24.46	24.65	24.32	31.32	31.90	32.18	24.65	30.78
P/S	184.7 7	185.4 9	191.6 1	202.0 3	92.45	116.1 7	101.4 5	87.92	NA	91.16	94.48	90.67	88.78
C _{CH₃}	104.9 1	158.4 3	167.1 0	152.4 7	73.61	122.3 1	58.15	125.5 1	27.22	159.3 0	154.8 2	175.2 6	143.0 2
H _{CH₃}	33.46	35.04	34.47	29.09	15.95	49.14	9.46	36.60	NA	34.40	33.13	34.32	29.02
H _{CH₃}	33.47	27.25	31.77	41.44	30.58	26.72	26.85	36.54	NA	31.01	41.39	34.34	29.02
H _{CH₃}	46.53	35.05	34.47	42.60	56.72	50.94	47.67	64.54	27.22	31.00	28.41	32.54	28.98
C _{CO}	82.02		79.83	95.43	80.28	82.53	93.95	79.00		78.54	82.21	79.30	94.96
O _{CO}	32.95		32.23	37.26	32.49	32.77	36.07	32.21		31.61	37.28	32.30	35.80

Area of atom A's part of IsoDensity Surface (IDS) with electron density greater than 0.002,
Area_IDS(A),0.002

	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh	32.68	40.74	17.97	23.04	34.47	14.02	25.61	34.17	35.26	17.60	18.78	24.54	20.40
C _{ipso}	8.52	17.89	14.37	11.40	7.10	4.06	6.24	8.44	14.83	12.65	9.78	6.24	19.48
C _{ortho}	12.31	18.15	17.73	17.13	15.72	17.98	15.89	16.67	18.66	18.20	17.54	15.19	18.92
C _{meta}	26.64	29.75	29.46	30.03	26.61	24.28	23.46	26.86	26.87	26.75	26.62	25.92	27.20
C _{para}	30.78	31.15	30.72	31.38	31.22	30.77	30.02	31.74	32.18	31.93	32.15	32.63	31.96
H _{phenyl}	32.34	33.22	33.52	33.20	24.89	25.76	24.92	25.66	28.78	29.75	29.35	28.47	30.01
H _{phenyl}	32.28	33.25	33.28	33.18	24.84	25.13	25.34	25.41	34.30	34.76	34.65	28.94	34.49
H _{phenyl}	35.43	35.01	35.19	35.11	34.03	34.05	34.36	33.96	29.00	29.72	29.34	34.41	29.01
P/S	5.74	8.99	4.37	7.33	88.17	82.12	78.46	90.01	94.64	86.55	86.32	89.12	89.39
C _{CH₃}	13.44	16.67	15.56	13.45	11.72	11.51	10.70	11.75	16.51	14.32	14.49	13.00	15.64
H _{CH₃}	26.93	28.15	27.41	34.08	17.97	16.38	14.33	21.84	27.84	28.59	29.51	25.27	29.80
H _{CH₃}	26.90	34.85	28.97	23.95	28.87	34.51	33.96	21.92	27.94	30.02	22.38	24.97	30.96
H _{CH₃}	20.48	27.87	27.33	21.67	18.38	16.77	14.17	15.61	34.37	30.00	33.94	25.43	31.27
C _{CO}	25.85		25.53	16.70	24.02	24.50	14.53	25.46		22.02	21.96	25.67	11.92
O _{CO}	90.38		89.67	86.26	90.34	90.08	88.01	90.08		89.71	86.48	89.79	85.24

Atomic Surface Area, Area(A),0.002 = Area_IAS(A|Total) + Area_IDS(A)

	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh	213.0 9	229.7 9	272.4 5	282.3 1	210.5 4	229.4 7	290.7 2	224.2 1	NA	260.9 6	264.5 5	231.8 6	216.5 9
C _{ipso}	131.2 9	115.1 2	117.6 4	112.8 2	NA	163.4 1	NA	142.8 5	115.3 6	119.2 9	115.5 6	136.1 6	117.9 6
C _{ortho}	127.7 9	122.5 3	122.9 3	121.4 9	129.8 3	133.6 5	132.8 7	151.1 2	141.1 2	142.4 5	140.8 3	144.1 8	140.4 0
C _{meta}	128.4 7	127.6 8	126.7 1	127.4 7	120.3 1	121.2 4	119.5 5	139.3 8	134.0 7	134.6 6	134.6 3	134.1 8	133.0 1
C _{para}	121.8 3	123.4 2	121.9 1	123.6 1	120.7 7	121.2 6	123.2 9	121.6 7	122.7 3	122.5 7	122.8 3	123.4 8	122.5 7
H _{phenyl}	60.11	60.67	60.89	60.47	59.48	59.68	59.85	60.00	60.10	61.65	61.36	60.11	60.78
H _{phenyl}	60.06	60.70	60.65	60.47	NA	59.82	60.70	59.71	58.86	59.49	59.37	60.57	59.10
H _{phenyl}	60.21	60.24	60.13	60.43	58.19	58.51	59.01	58.28	60.32	61.63	61.52	59.06	59.79
P/S	190.5 1	194.4 7	195.9 8	209.3 6	180.6 2	198.3 0	179.9 1	177.9 3	NA	177.7 1	180.8 1	179.7 9	178.1 7
C _{CH₃}	118.3 5	175.1 0	182.6 7	165.9 2	NA	133.8 2	NA	137.2 5	NA	173.6 2	169.3 0	188.2 7	158.6 7
H _{CH₃}	60.40	63.20	61.88	63.18	NA	65.52	NA	58.45	NA	62.99	62.64	59.59	58.82
H _{CH₃}	60.37	62.10	60.74	65.39	59.46	61.24	60.80	58.46	NA	61.04	63.77	59.31	59.98
H _{CH₃}	67.01	62.93	61.80	64.27	75.10	67.72	61.84	80.15	61.58	61.00	62.35	57.97	60.25
C _{CO}	107.8 6		105.3 7	112.1 3	104.3 0	107.0 3	108.4 8	104.4 6		100.5 6	104.1 7	104.9 7	106.8 8
O _{CO}	123.3 3		121.9 1	123.5 2	122.8 3	122.8 5	124.0 8	122.2 9		121.3 2	123.7 6	122.1 0	121.0 4

N_IAS(A)													
	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh	3.96	3.93	5.11	5.15	3.55	3.98	4.21	3.81	1.04	4.81	4.79	4.38	4.45
C _{ipso}	4.52	3.77	3.75	3.80	2.07	5.09	1.84	4.46	3.79	3.73	3.83	4.25	3.66
C _{ortho}	4.13	4.01	4.01	4.03	3.98	4.00	3.96	4.09	3.99	4.02	4.00	4.04	4.01
C _{meta}	3.82	3.80	3.79	3.79	3.77	3.80	3.78	3.89	3.83	3.84	3.84	3.84	3.82
C _{para}	3.75	3.77	3.75	3.77	3.73	3.74	3.75	3.73	3.74	3.74	3.74	3.74	3.74
H _{phenyl}	0.97	0.97	0.97	0.97	1.01	1.01	1.02	1.01	0.98	0.99	0.99	0.98	0.98
H _{phenyl}	0.97	0.97	0.97	0.97	NA	1.02	1.02	1.01	0.94	0.95	0.95	0.98	0.94
H _{phenyl}	0.95	0.95	0.95	0.96	0.93	0.94	0.94	0.94	0.98	0.99	0.99	0.94	0.98
P/S	4.33	4.33	4.40	4.47	2.02	2.15	1.98	1.94	NA	1.98	1.97	1.96	1.96
C _{CH₃}	4.08	4.09	4.11	4.11	2.08	4.16	1.96	4.25	0.97	4.10	4.12	4.45	4.00
H _{CH₃}	1.02	1.02	1.02	0.98	0.11	1.13	0.06	1.02	NA	1.03	1.01	1.02	0.99
H _{CH₃}	1.02	0.97	1.01	1.07	0.97	0.96	0.95	1.02	NA	1.00	1.07	1.02	0.98
H _{CH₃}	1.27	1.02	1.02	1.10	1.19	1.13	1.13	1.60	0.97	1.00	0.98	1.04	0.98
C _{CO}	3.04		2.81	2.86	3.01	2.95	3.13	2.98		2.78	2.68	2.85	3.45
O _{CO}	1.69		1.71	1.73	1.69	1.70	1.70	1.69		1.71	1.73	1.70	1.73

G_IAS(A)													
	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh	2.75	2.67	3.51	3.57	2.60	2.92	3.23	2.83	0.76	3.49	3.43	3.24	3.23
C _{ipso}	3.05	2.52	2.50	2.54	1.42	3.45	1.23	2.97	2.55	2.51	2.57	2.88	2.43
C _{ortho}	2.73	2.63	2.63	2.65	2.63	2.65	2.58	2.71	2.64	2.65	2.65	2.68	2.64
C _{meta}	2.39	2.37	2.36	2.36	2.36	2.38	2.36	2.45	2.39	2.40	2.40	2.40	2.38
C _{para}	2.32	2.34	2.32	2.34	2.32	2.33	2.34	2.32	2.33	2.33	2.33	2.33	2.32
H _{phenyl}	0.52	0.52	0.52	0.52	0.56	0.56	0.57	0.56	0.53	0.54	0.54	0.53	0.53
H _{phenyl}	0.52	0.52	0.52	0.52	NA	0.56	0.57	0.56	0.50	0.50	0.50	0.53	0.50
H _{phenyl}	0.50	0.51	0.51	0.51	0.49	0.50	0.50	0.50	0.53	0.54	0.54	0.50	0.53
P/S	2.76	2.82	2.85	2.92	1.22	1.35	1.18	1.15	NA	1.19	1.19	1.19	1.19
C _{CH₃}	2.41	2.44	2.43	2.43	1.18	2.46	1.10	2.55	0.53	2.41	2.43	2.71	2.33
H _{CH₃}	0.58	0.57	0.57	0.54	0.08	0.66	0.05	0.58	NA	0.57	0.56	0.57	0.54
H _{CH₃}	0.58	0.53	0.55	0.61	0.54	0.53	0.53	0.58	NA	0.55	0.60	0.57	0.54
H _{CH₃}	0.78	0.57	0.57	0.63	0.70	0.68	0.67	1.08	0.53	0.55	0.54	0.59	0.54
C _{CO}	3.26		3.16	3.22	3.29	3.25	3.37	3.27		3.20	3.10	3.20	3.57
O _{CO}	2.22		2.30	2.33	2.25	2.26	2.23	2.25		2.33	2.35	2.29	2.24

Table S2. Electron density Bond Critical Point (BCP) analysis of molecular structure of the PCP and SCS complexes

Electron Density, ρ_b													
	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh-C _{ipso}	0.05	0.15	0.13	0.14	0.05	0.07	0.02	0.04	0.15	0.12	0.15	0.12	0.12
C _{ipso} -C _{ortho}	0.29	0.30	0.29	0.30	0.29	0.27	0.29	0.29	0.29	0.29	0.30	0.29	0.29
Rh-P/S	0.10	0.10	0.10	0.10	0.08	0.08	0.08	0.08	0.08	0.09	0.08	0.08	0.09
C _{ipso} -C _{CH₃}	0.10	0.14	0.12	0.12			0.25		0.14	0.13	0.13	0.09	0.08
Rh-C _{CO}	0.18		0.13	0.13	0.18	0.17	0.19	0.18		0.13	0.11	0.15	0.21
C-O	0.47		0.48	0.48	0.47	0.47	0.47	0.47		0.48	0.48	0.48	0.47

Hamiltonian form of kinetic energy density, K(r)													
	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh-C _{ipso}	0.01	0.07	0.06	0.06	0.01	0.02	0.00	0.01	0.08	0.05	0.07	0.05	0.05
C _{ipso} -C _{ortho}	0.26	0.27	0.26	0.28	0.25	0.23	0.26	0.26	0.26	0.26	0.26	0.25	0.26
Rh-P/S	0.04	0.04	0.04	0.04	0.02	0.02	0.02	0.02	0.02	0.02	0.02	0.02	0.03
C _{ipso} -C _{CH₃}		0.07	0.05	0.06					0.07	0.06	0.06	0.03	0.03
Rh-C _{CO}	0.11		0.06	0.05	0.10	0.09	0.12	0.10		0.06	0.04	0.07	0.13
C-O	0.84		0.86	0.87	0.84	0.85	0.84	0.84		0.87	0.87	0.86	0.84

Cremer and Kraka energy density H(r) = G(r) + V(r) = -K(r)													
	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh-C _{ipso}	-0.01	-0.07	-0.06	-0.06	-0.01	-0.02	0.00	-0.01	-0.08	-0.05	-0.07	-0.05	-0.05
C _{ipso} -C _{ortho}	-0.26	-0.27	-0.26	-0.28	-0.25	-0.23	-0.26	-0.26	-0.26	-0.26	-0.26	-0.25	-0.26
Rh-P/S	-0.04	-0.04	-0.04	-0.04	-0.02	-0.02	-0.02	-0.02	-0.02	-0.02	-0.02	-0.02	-0.03
C _{ipso} -C _{CH₃}		-0.07	-0.05	-0.06					-0.07	-0.06	-0.06	-0.03	-0.03
Rh-C _{CO}	-0.11		-0.06	-0.05	-0.10	-0.09	-0.12	-0.10		-0.06	-0.04	-0.07	-0.13
C-O	-0.84		-0.86	-0.87	-0.84	-0.85	-0.84	-0.84		-0.87	-0.87	-0.86	-0.84

Laplacian of electron density $\nabla^2\rho_b$													
	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh-C _{ipso}	0.15	0.08	0.13	0.06	0.14	0.15	0.05	0.13	0.12	0.18	0.10	0.13	0.16
C _{ipso} -C _{ortho}	-0.67	-0.69	-0.68	-0.71	-0.64	-0.59	-0.67	-0.67	-0.67	-0.67	-0.68	-0.66	-0.67
Rh-P/S	0.11	0.13	0.10	0.10	0.18	0.18	0.19	0.18	0.20	0.19	0.17	0.18	0.18
C _{ipso} -C _{CH₃}	0.11	0.04	0.03	0.08			-0.54		0.03	0.02	0.05	0.16	0.13
Rh-C _{CO}	0.46		0.46	0.45	0.46	0.47	0.49	0.45		0.45	0.42	0.46	0.30
C-O	1.00		1.12	1.13	1.08	1.06	0.96	1.09		1.21	1.17	1.12	1.00

Bond ellipticity $\varepsilon = \lambda_1/\lambda_2 - 1$ ($|\lambda_1| \geq |\lambda_2|$)

	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh-C _{ipso}	0.19	0.06	0.04	0.03	0.43	0.31	0.05	0.38	0.07	0.03	0.03	0.00	0.05
C _{ipso} -C _{ortho}	0.18	0.19	0.17	0.21	0.16	0.12	0.17	0.18	0.16	0.13	0.17	0.16	0.14
Rh-P/S	0.05	0.02	0.04	0.02	0.06	0.11	0.07	0.02	0.07	0.02	0.03	0.06	0.07
C _{ipso} -C _{CH₃}	0.05	0.01	0.01	0.01			0.03		0.01	0.03	0.01	0.23	0.33
Rh-C _{CO}	0.01		0.03	0.03	0.03	0.03	0.07	0.08		0.02	0.01	0.04	0.16
C-O	0.01		0.00	0.00	0.00	0.00	0.00	0.00		0.00	0.00	0.00	0.02

Electrostatic Potential from Nuclei V_{nuc}

	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh-C _{ipso}	47.79	48.03	49.76	50.16	49.67	53.52	48.28	48.95	50.65	51.94	52.87	52.14	51.43
C _{ipso} -C _{ortho}	40.74	39.33	40.59	41.47	43.40	45.80	44.77	43.24	42.36	43.47	44.38	44.22	43.28
Rh-P/S	48.42	46.00	48.41	48.32	49.32	52.69	51.96	49.22	47.20	49.66	49.41	49.52	49.73
C _{ipso} -C _{CH₃}	39.79	44.70	46.65	47.18					46.49	48.41	48.82	49.49	46.22
Rh-C _{CO}	49.59		48.40	49.22	50.53	53.40	54.06	50.43		49.45	49.95	50.53	51.91
C-O	40.81		40.06	41.39	41.32	43.96	44.67	41.37		40.67	42.26	41.99	43.13

Electron-nuclear attractive contribution to virial field V , V_{en}

	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh-C _{ipso}	-2.48	-7.16	-6.41	-6.84	-2.26	-3.52	-1.10	-2.18	-7.83	-6.42	-7.82	-6.51	-6.14
C _{ipso} -C _{ortho}	-11.90	-11.66	-11.94	-12.54	-12.43	-12.55	-13.17	-12.65	-12.43	-12.78	-13.11	-12.84	-12.77
Rh-P/S	-4.79	-4.58	-4.90	-4.66	-3.84	-4.17	-4.19	-3.95	-3.99	-4.24	-3.95	-4.04	-4.40
C _{ipso} -C _{CH₃}	-4.79	-6.15	-5.68	-5.83			-10.82		-6.58	-6.11	-6.47	-4.40	-3.83
Rh-C _{CO}	-9.16		-6.45	-6.35	-9.27	-8.93	-10.48	-9.24		-6.52	-5.63	-7.70	-10.70
C-O	-19.01		-19.18	-19.87	-19.45	-20.77	-20.81	-19.49		-19.69	-20.47	-20.07	-20.12

Repulsive contribution to virial field V , $V_{\text{rep}} = V - V_{\text{en}}$

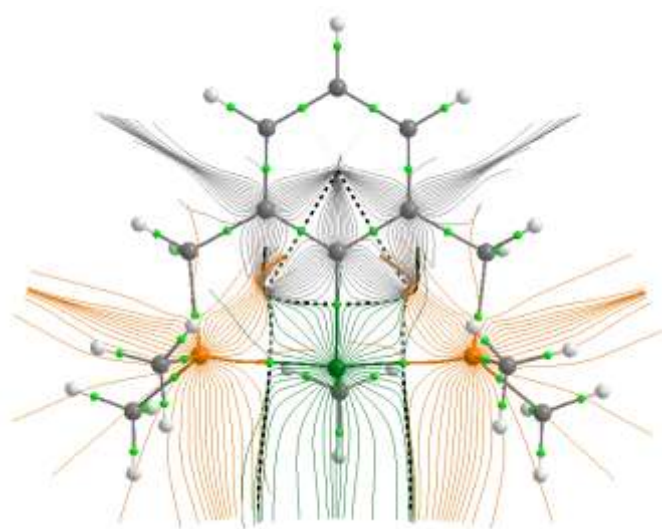
	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh-C _{ipso}	2.43	6.99	6.26	6.70	2.21	3.45	1.08	2.13	7.65	6.27	7.65	6.37	6.00
C _{ipso} -C _{ortho}	11.56	11.30	11.59	12.16	12.09	12.25	12.81	12.30	12.08	12.42	12.75	12.50	12.41
Rh-P/S	4.69	4.47	4.79	4.56	3.76	4.09	4.10	3.86	3.89	4.14	3.86	3.95	4.30
C _{ipso} -C _{CH₃}	4.69	6.01	5.57	5.70			10.56		6.43	5.99	6.33	4.30	3.74
Rh-C _{CO}	8.84		6.22	6.13	8.95	8.64	10.13	8.92		6.29	5.45	7.44	10.36
C-O	17.09		17.18	17.86	17.49	18.80	18.89	17.53		17.65	18.43	18.06	18.18

Laplacian of the electron-nuclear attractive contribution to virial field $V_{\text{en}}, \nabla^2 V_{\text{en}}$													
	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh-C _{ipso}	-7.20	-3.93	-6.30	-2.80	-6.73	-8.02	-2.64	-6.49	-5.92	-9.16	-5.22	-7.00	-8.13
C _{ipso} -C _{ortho}	27.34	27.07	27.48	29.51	27.80	27.15	30.12	28.83	28.41	29.26	30.01	29.03	29.18
Rh-P/S	-5.12	-5.76	-4.85	-5.06	-9.03	-9.29	-9.72	-9.03	-9.60	-9.27	-8.64	-8.89	-8.87
C _{ipso} -C _{CH₃}	-5.12	-1.79	-1.29	-3.76					-1.63	-1.14	-2.60	-8.06	-5.84
Rh-C _{CO}	-22.92		-22.29	-22.39	-23.42	-24.88	-26.60	-22.72		-22.48	-21.00	-23.15	-15.79
C-O	-40.77		-45.06	-46.75	-44.56	-46.81	-42.73	-44.98		-49.21	-49.65	-47.14	-43.03

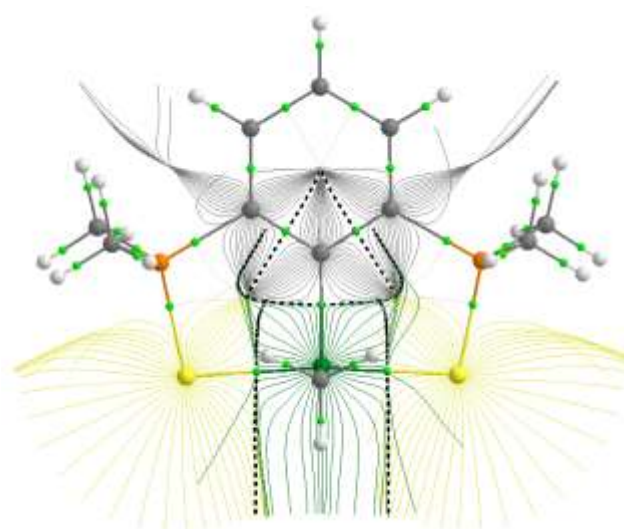
Laplacian of the repulsive contribution to virial field $V_{\text{rep}}, \nabla^2 V_{\text{rep}}$													
	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh-C _{ipso}	6.70	2.57	5.16	1.65	6.29	7.42	2.56	6.14	4.42	7.94	3.84	5.92	7.05
C _{ipso} -C _{ortho}	-30.24	-30.07	-30.43	-32.67	-30.61	-29.69	-33.13	-31.78	-31.38	-32.25	-33.05	-31.94	-32.22
Rh-P/S	4.37	4.95	4.09	4.33	8.22	8.50	8.89	8.20	8.68	8.37	7.80	8.03	7.97
C _{ipso} -C _{CH₃}	4.38	0.61	0.35	2.74					0.39	0.14	1.53	7.19	5.13
Rh-C _{CO}	20.72		20.70	20.85	21.23	22.93	24.21	20.54		20.86	19.65	21.36	13.31
C-O	21.27		24.07	25.60	24.29	26.49	23.61	24.63		27.44	27.96	26.22	23.90

Laplacian of the Hamiltonian form of kinetic energy density, $\nabla^2 K(\mathbf{r})$													
	PCP complexes				SCS complexes								
	23	15	17	16	19	19-S	19-S1	20	11	13	12	TS2	TS1
Rh-C _{ipso}	0.05	0.14	0.06	0.11	0.04	0.07	0.03	0.05	0.14	0.07	0.14	0.14	0.09
C _{ipso} -C _{ortho}	1.61	1.70	1.67	1.81	1.55	1.35	1.71	1.66	1.69	1.70	1.74	1.64	1.73
Rh-P/S	0.13	0.13	0.12	0.12	0.15	0.15	0.16	0.16	0.16	0.16	0.14	0.15	0.17
C _{ipso} -C _{CH₃}	0.13	0.10	0.07	0.04					0.12	0.09	0.04	0.07	0.05
Rh-C _{CO}	0.61		0.29	0.28	0.57	0.48	0.63	0.54		0.26	0.23	0.35	0.73
C-O	-28.23		-29.85	-30.15	-28.82	-29.18	-28.09	-28.87		-30.44	-30.69	-29.77	-27.86

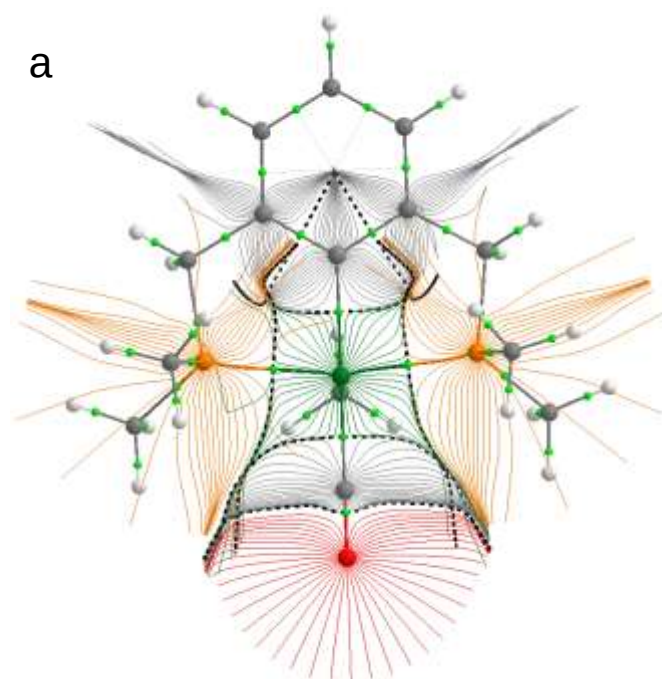
QTAIM pictures of aryl-methyl PCP and SCS complexes



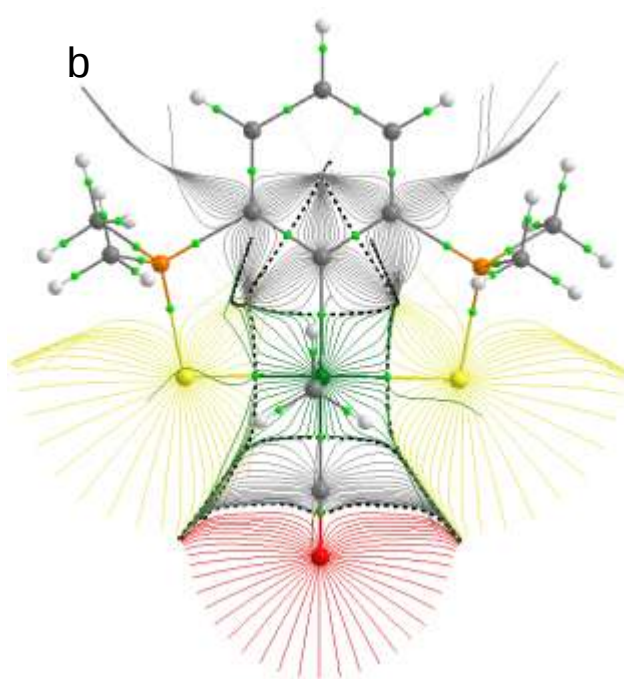
PCP-Rh-CH3



SCS-Rh-CH3

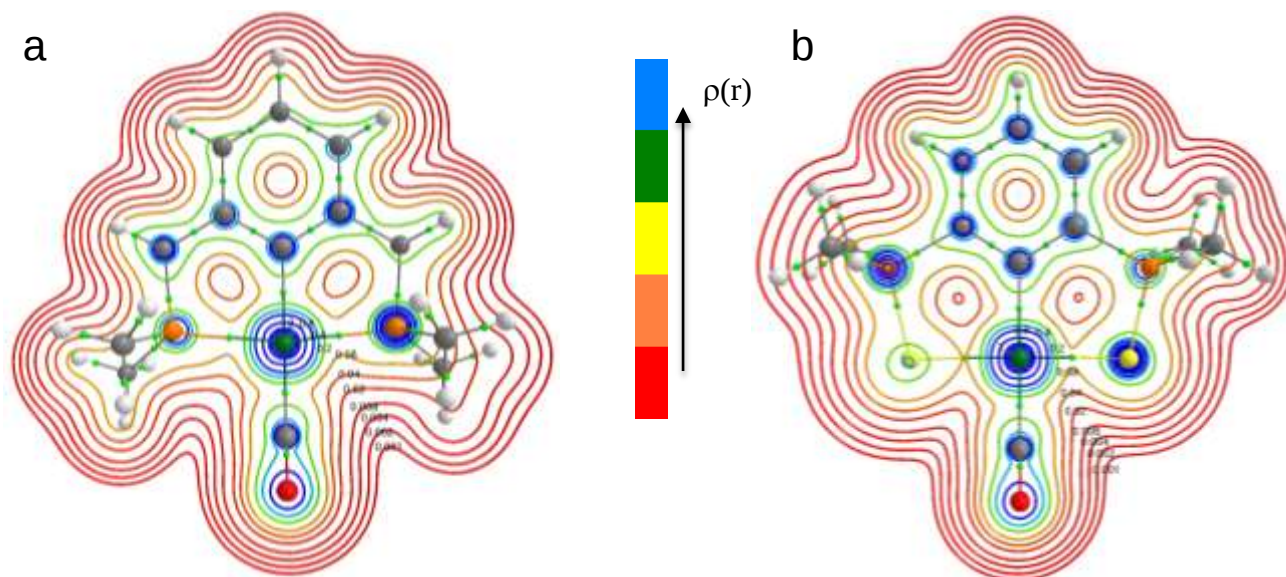


PCP-Rh-CH3-CO

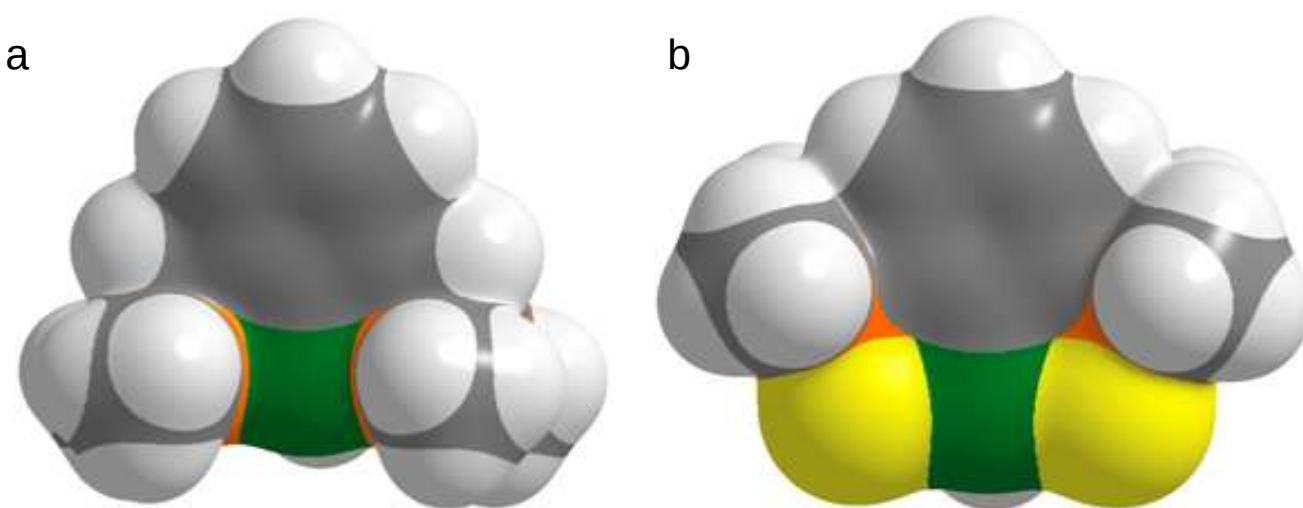


SCS-Rh-CH3-CO

Basin path + IAS EV path for complexes **17** (a) and **13** (b).

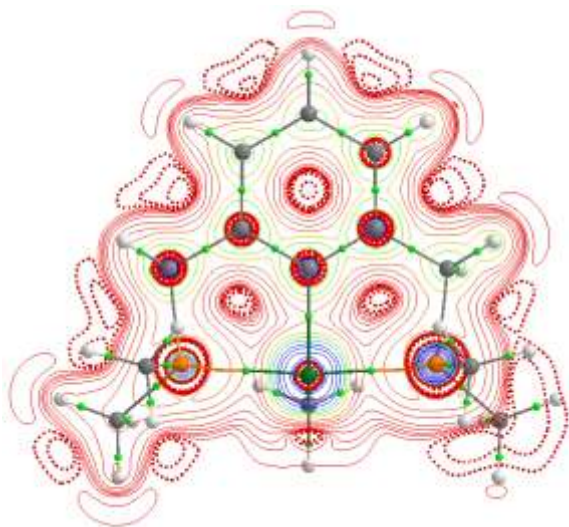


Counter maps of electron density $\rho(r)$ for complexes **17** (a) and **13** (b).

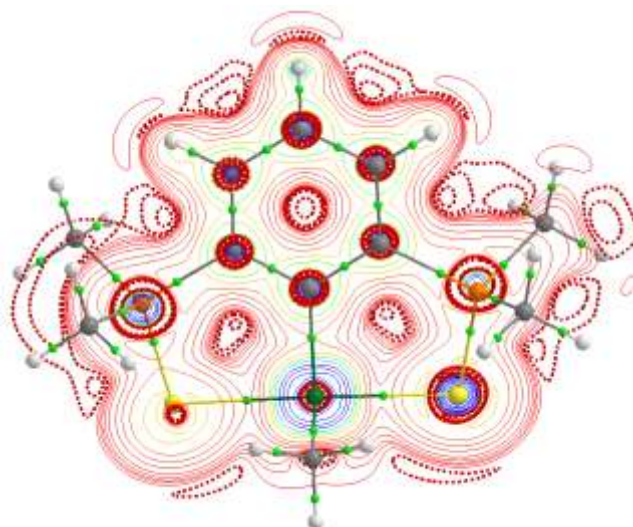


3D Laplasian of electron density for complexes **15** (a) and **11** (b). Resolution 0.05, isosurface 0.01

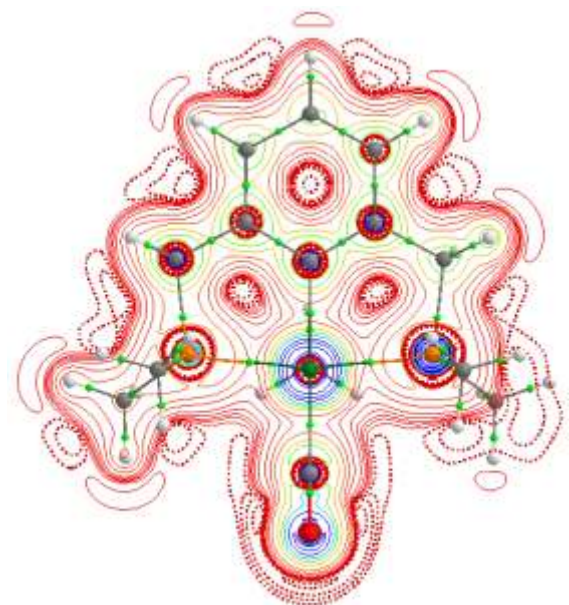
The Hamiltonian form of the electron kinetic energy density $K(=-\hbar^2/4m)\nabla^2\rho$. Min el. Dens=0.001, resolution 0.05, allow tubular; color scale 0-1



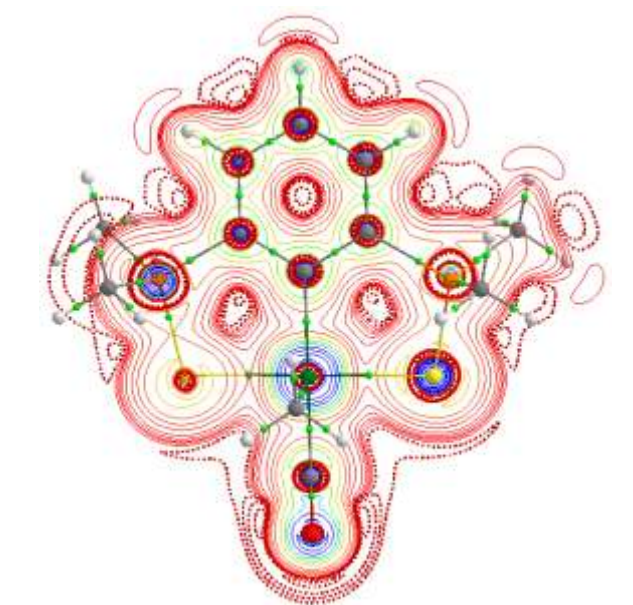
PCP-Rh-CH3



SCS-Rh-CH3



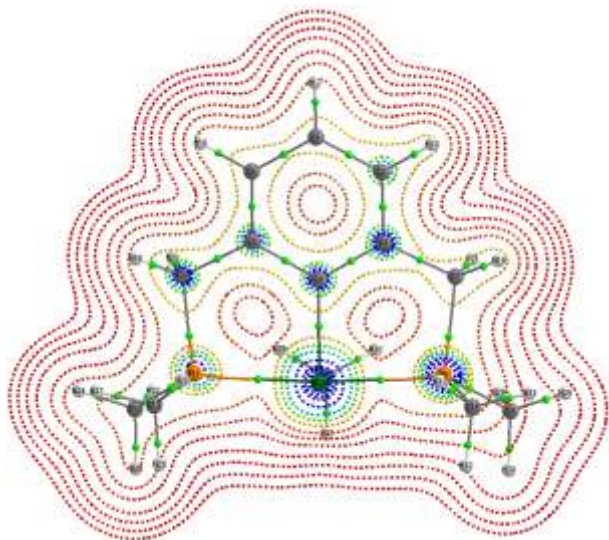
PCP-Rh-CH3-CO



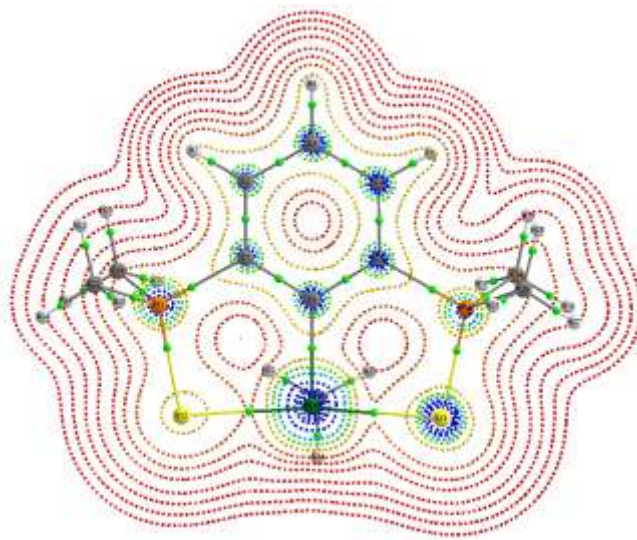
SCS-Rh-CH3-CO

Counter maps for molecular structure of the PCP and SCS complexes in three dimensions

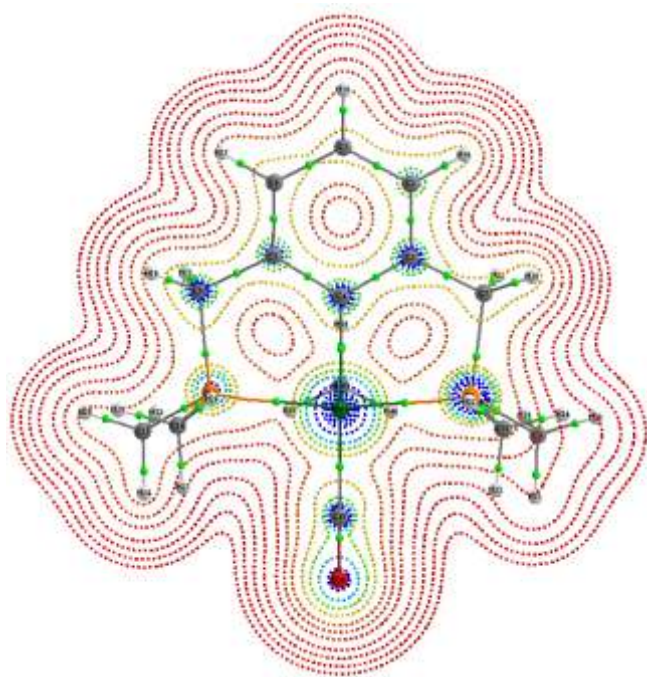
Rho, min el. Dens=0.001, resolution 0.05, allow tubular; color scale 0-1



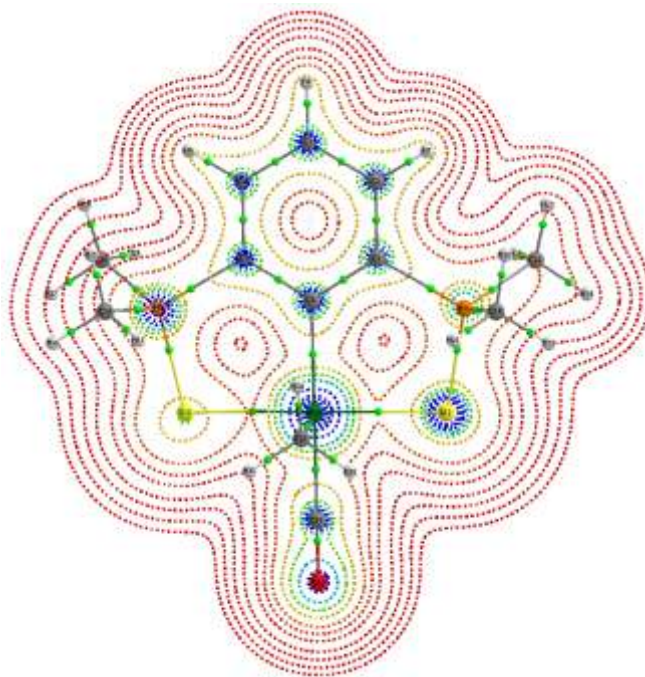
PCP-Rh-CH3



SCS-Rh-CH3

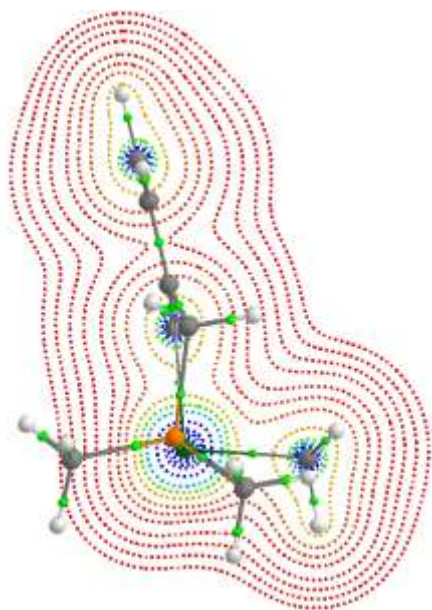


PCP-Rh-CH3-CO

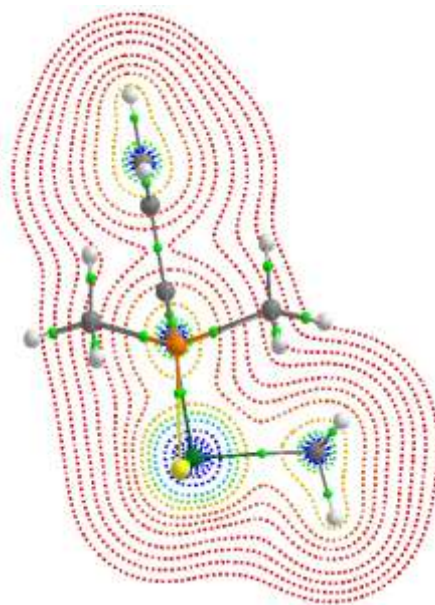


SCS-Rh-CH3-CO

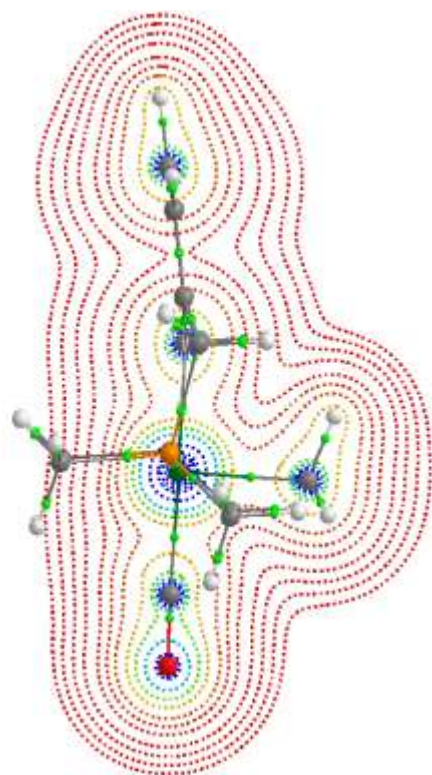
Rho, min el. Dens=0.001, resolution 0.05, allow tubular; color scale 0-1



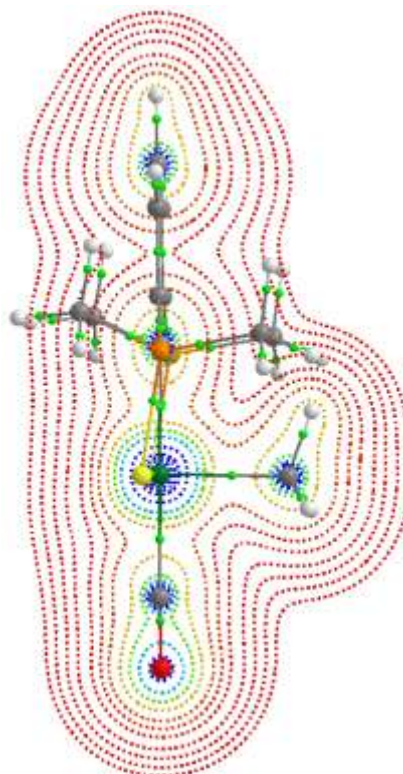
PCP-Rh-CH3



SCS-Rh-CH3

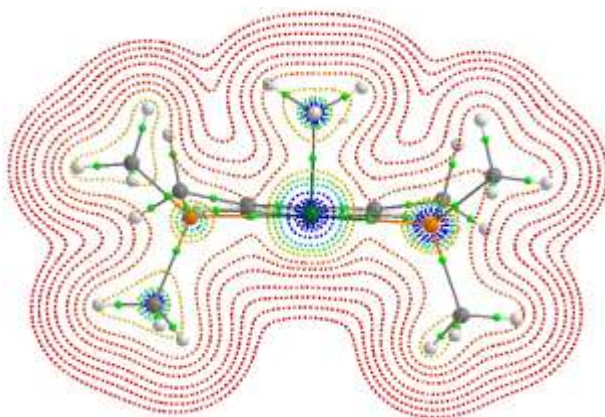


PCP-Rh-CH3-CO

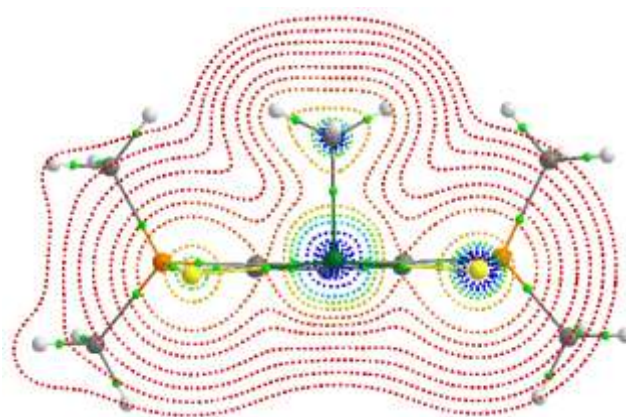


SCS-Rh-CH3-CO

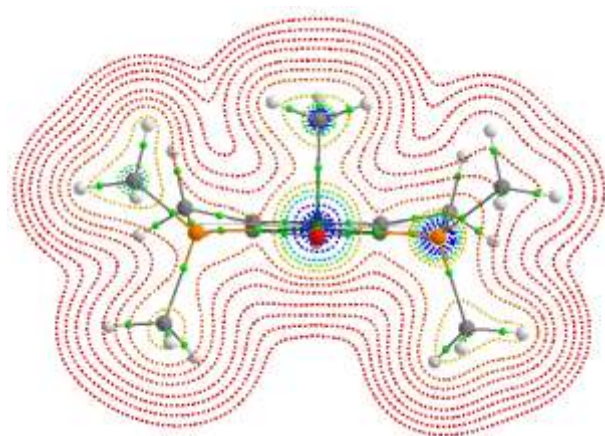
Rho, min el. Dens=0.001, resolution 0.05, allow tubular; color scale 0-1



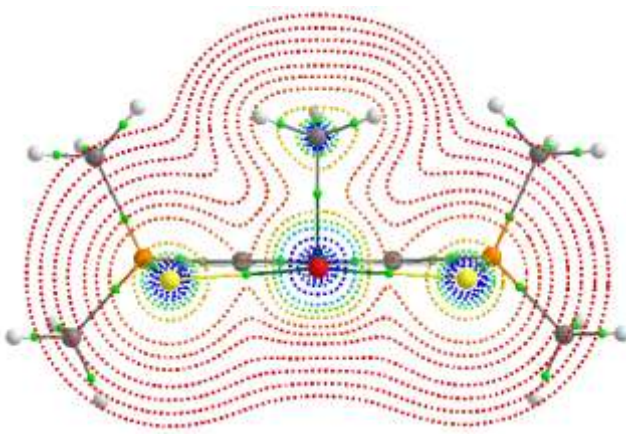
PCP-Rh-CH3



SCS-Rh-CH3

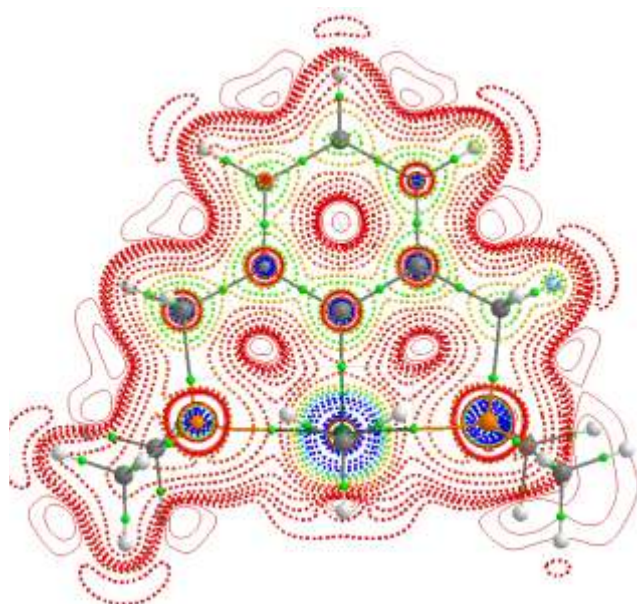


PCP-Rh-CH3-CO

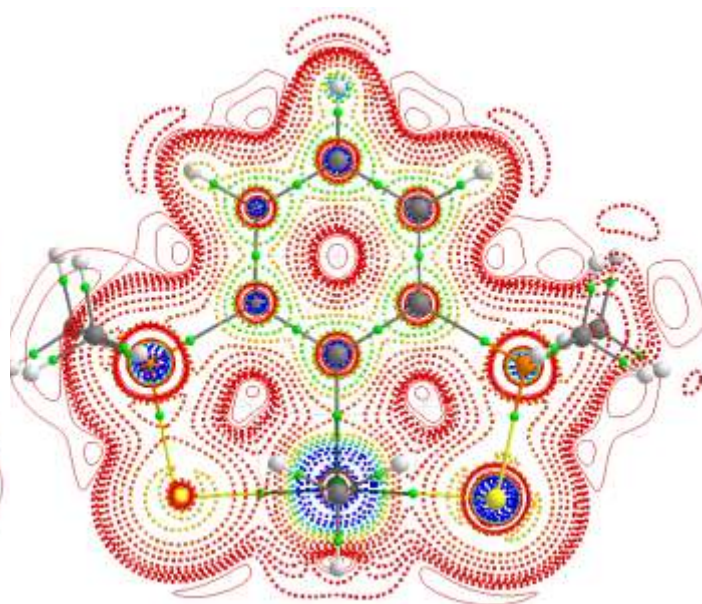


SCS-Rh-CH3-CO

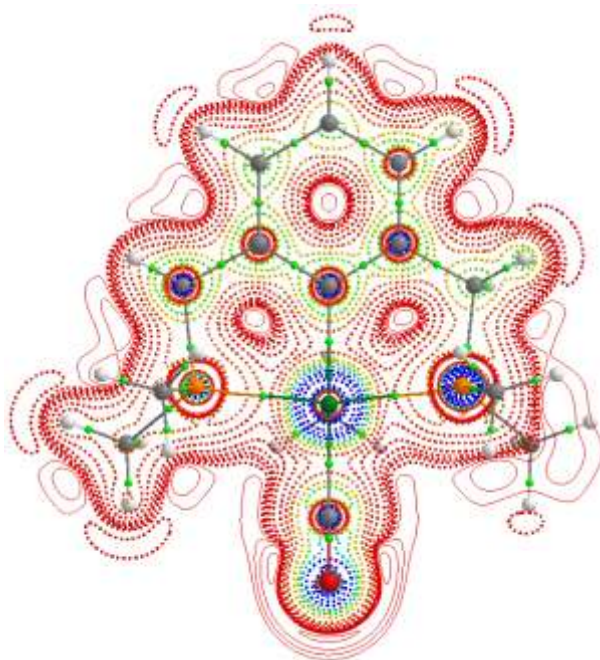
K, The Hamiltonian kinetic energy density; min el. Dens=0.001, resolution 0.05, allow tubular; color scale 0-1



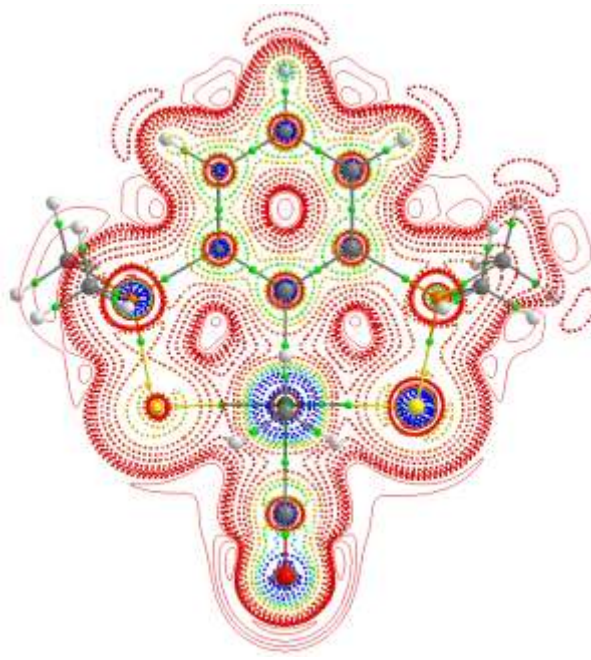
PCP-Rh-CH3



SCS-Rh-CH3

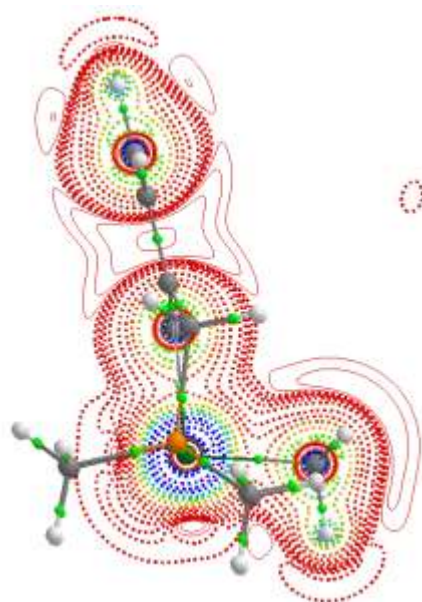


PCP-Rh-CH3-CO

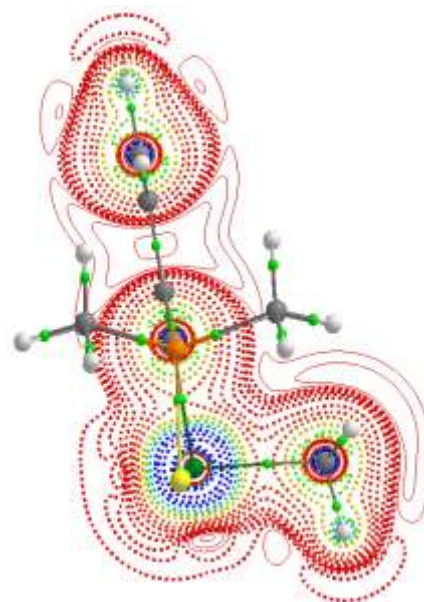


SCS-Rh-CH3-CO

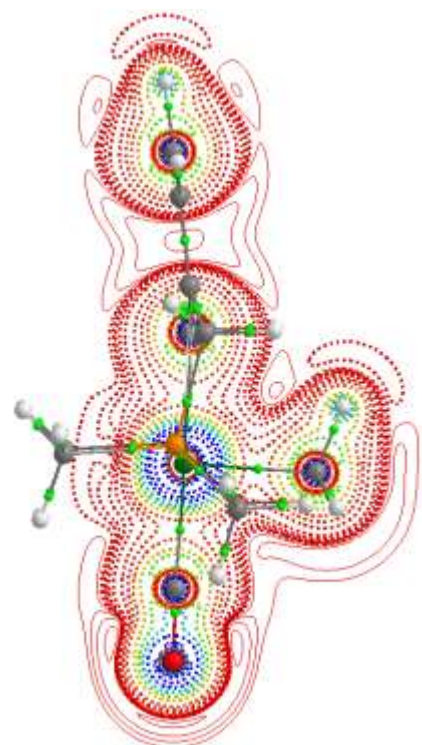
K, The Hamiltonian kinetic energy density; min el. Dens=0.001, resolution 0.05, allow tubular; color scale 0-1



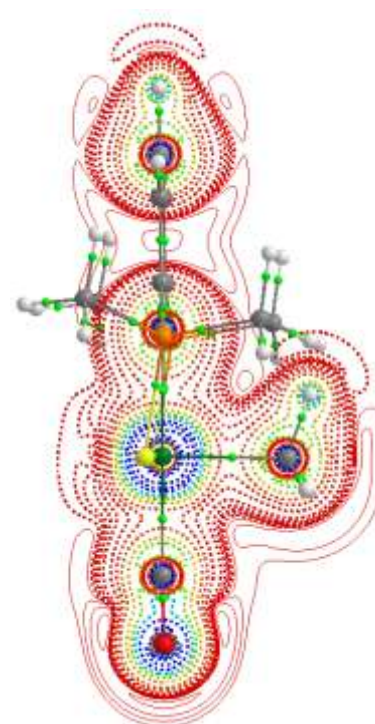
PCP-Rh-CH3



SCS-Rh-CH3

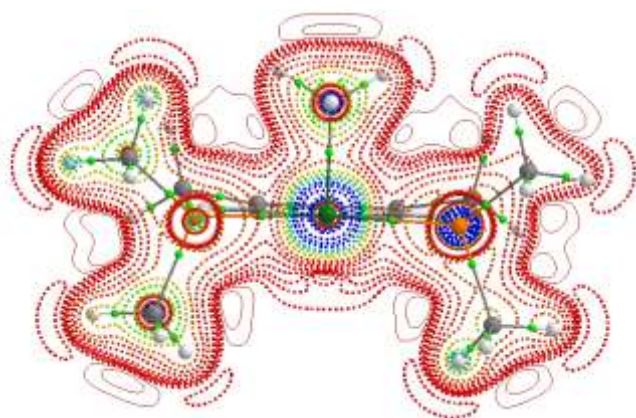


PCP-Rh-CH3-CO

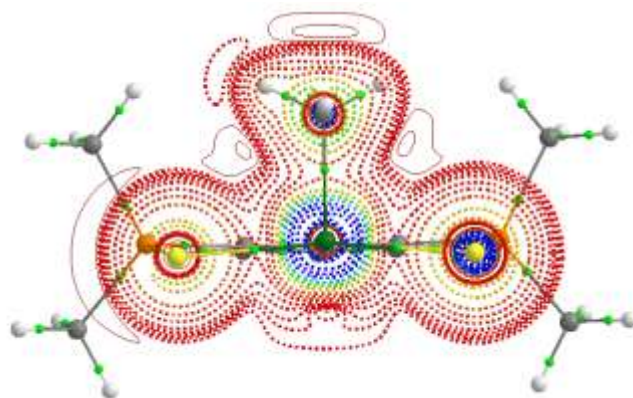


SCS-Rh-CH3-CO

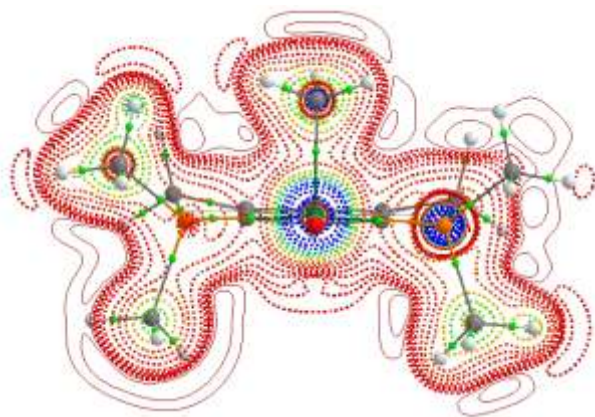
K, The Hamiltonian kinetic energy density; min el. Dens=0.001, resolution 0.05, allow tubular; color scale 0-1



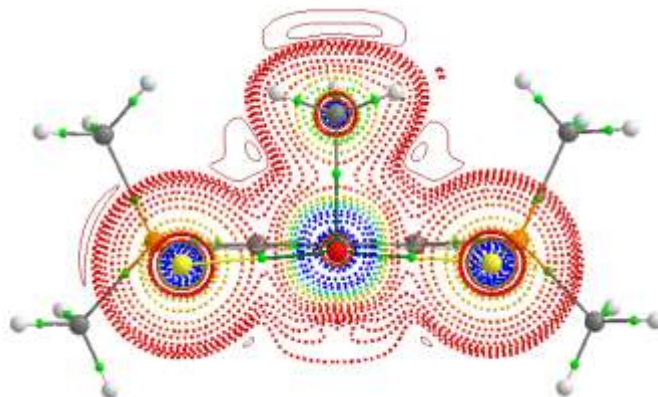
PCP-Rh-CH3



SCS-Rh-CH3

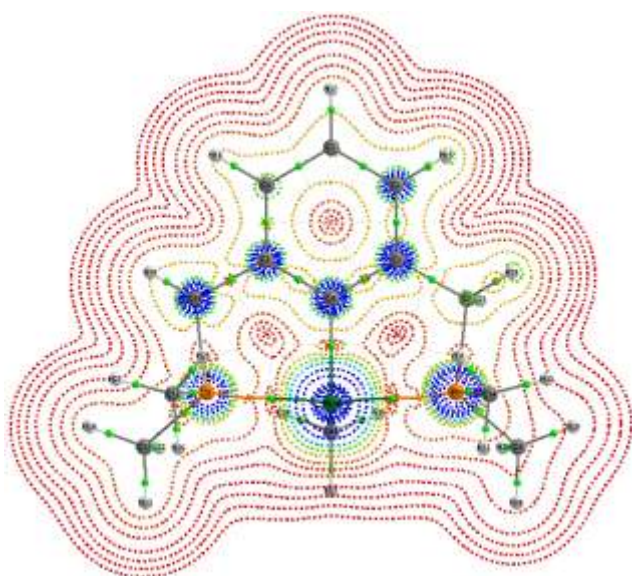


PCP-Rh-CH3-CO

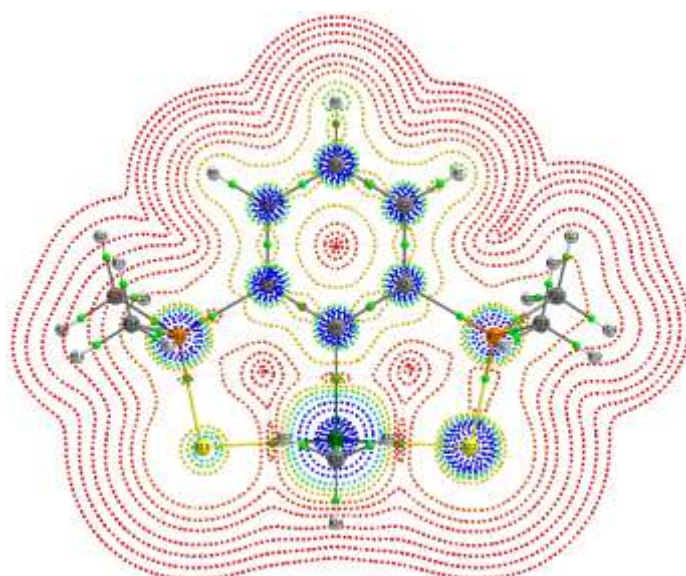


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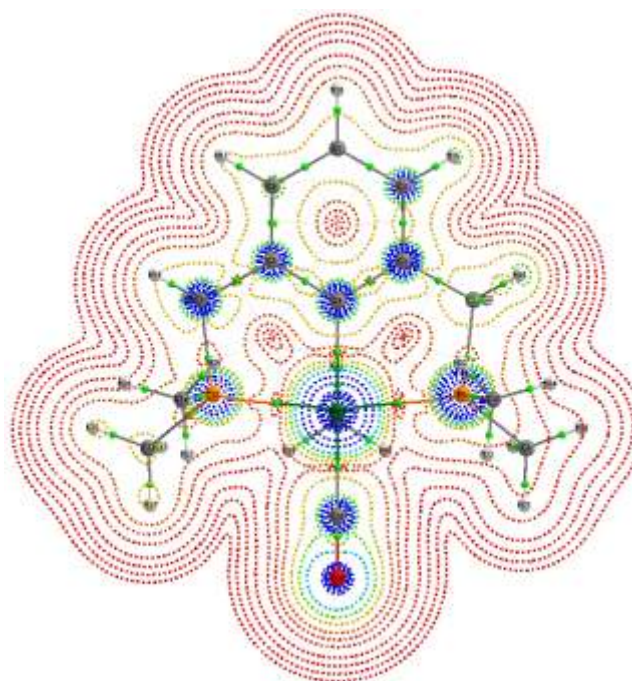
GRM, magnitude of the gradient of the electron density; min el. Dens=0.001, resolution 0.05, allow tubular; color scale 0-1



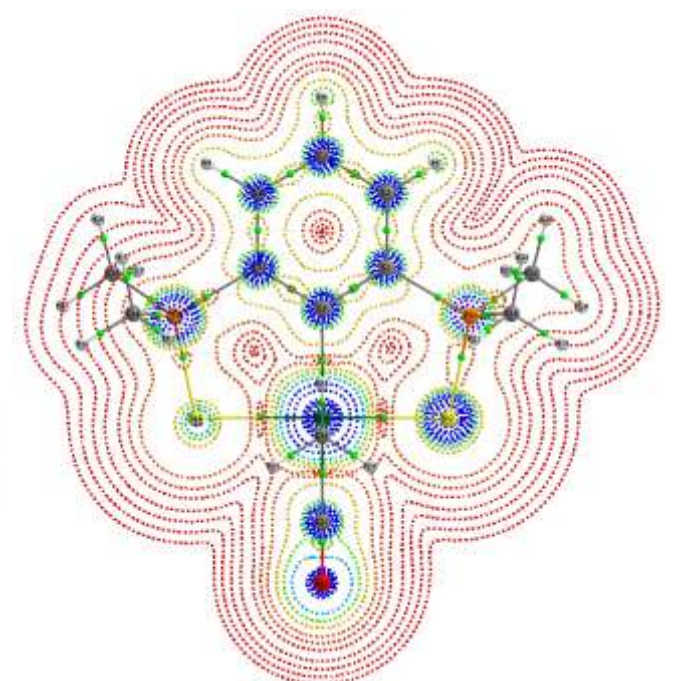
PCP-Rh-CH3



SCS-Rh-CH3

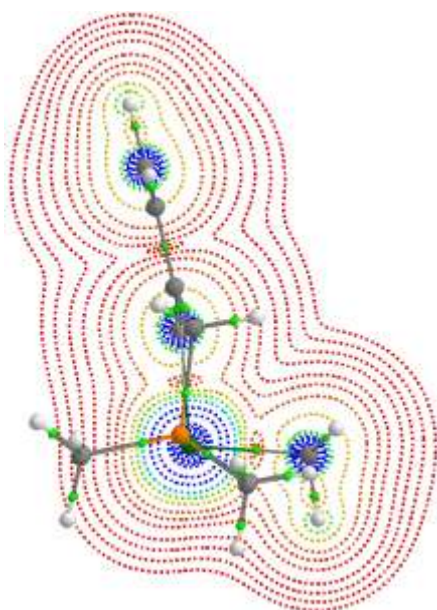


PCP-Rh-CH3-CO

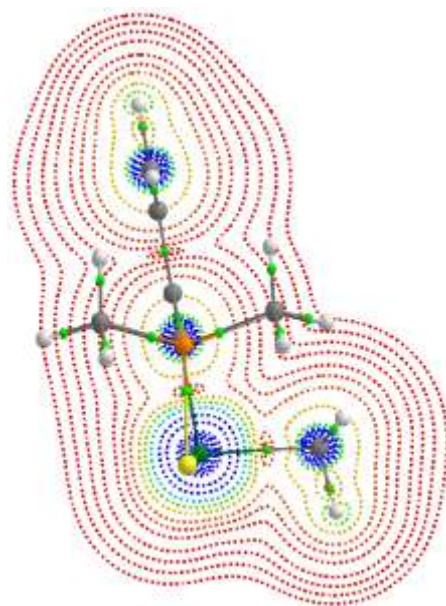


SCS-Rh-CH3-CO

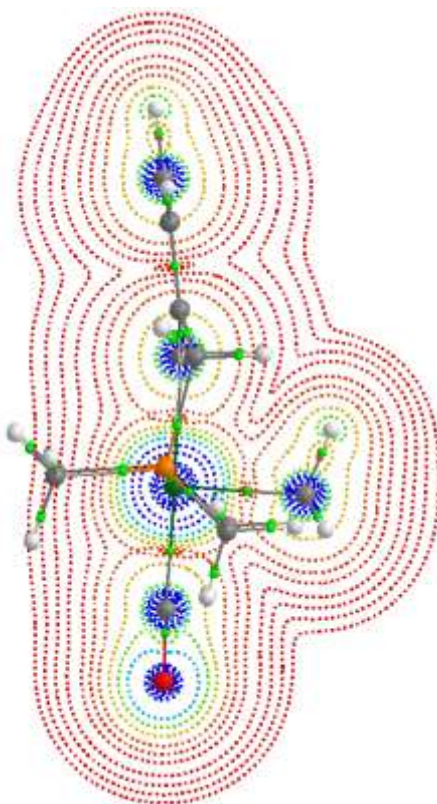
GRM, magnitude of the gradient of the electron density ($|\text{Grad}\rho|$); min el. Dens=0.001, resolution 0.05, allow tubular; color scale 0-1



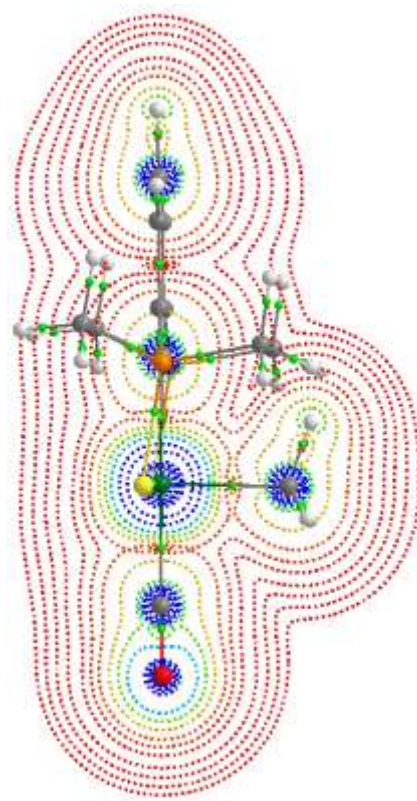
PCP-Rh-CH3



SCS-Rh-CH3

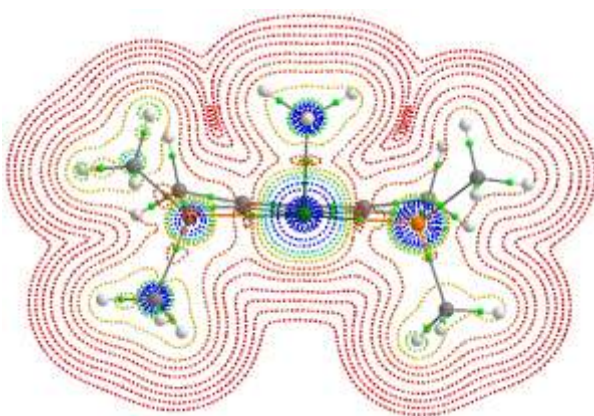


PCP-Rh-CH3-CO

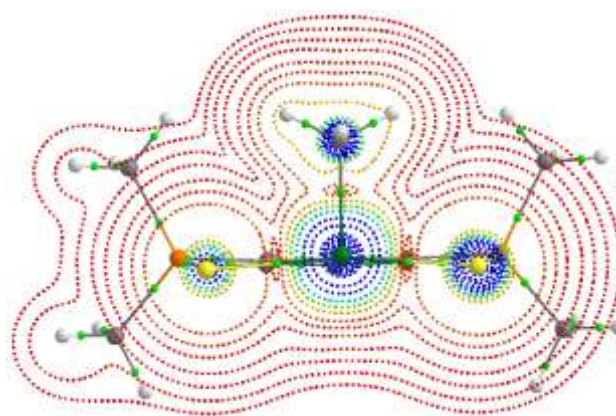


SCS-Rh-CH3-CO

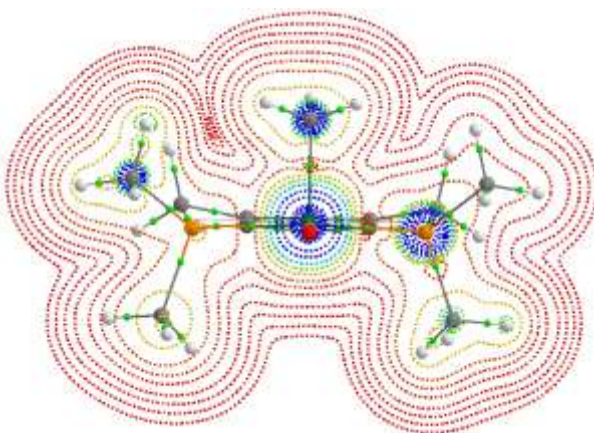
GRM, magnitude of the gradient of the electron density; min el. Dens=0.001, resolution 0.05, allow tubular; color scale 0-1



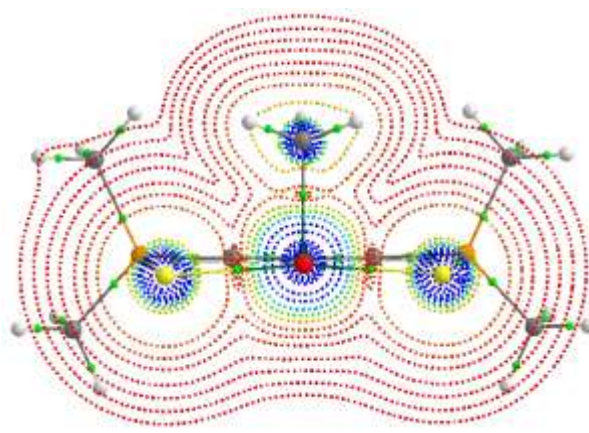
PCP-Rh-CH3



SCS-Rh-CH3

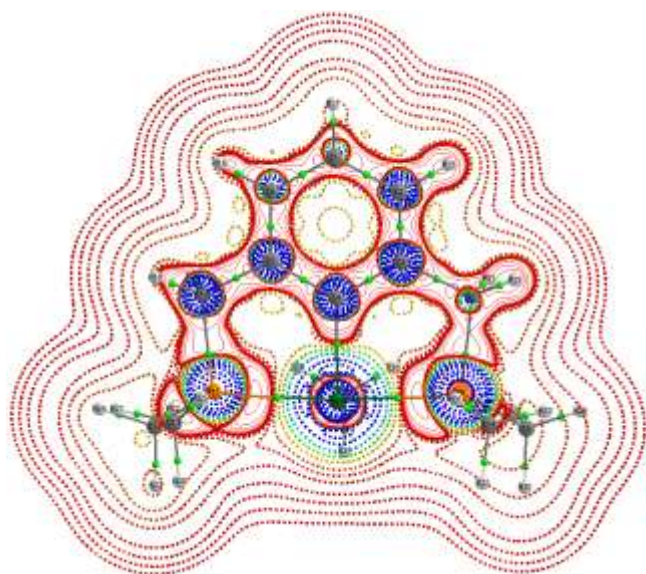


PCP-Rh-CH3-CO

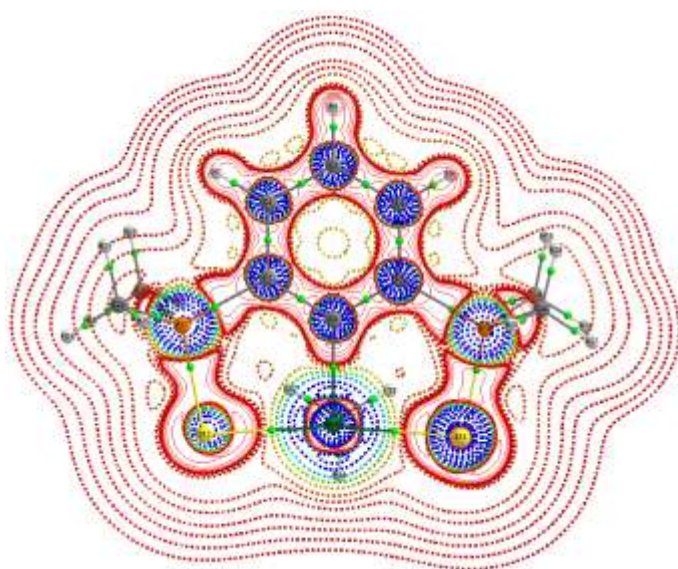


SCS-Rh-CH3-CO

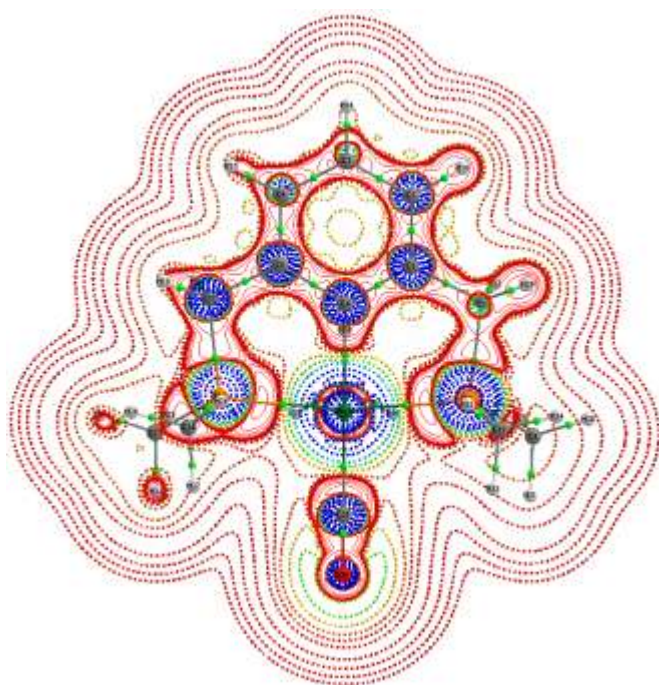
DelSqRho; min El. Dens=0.001; resolution 0.05; allow tubular; color scale 0-1



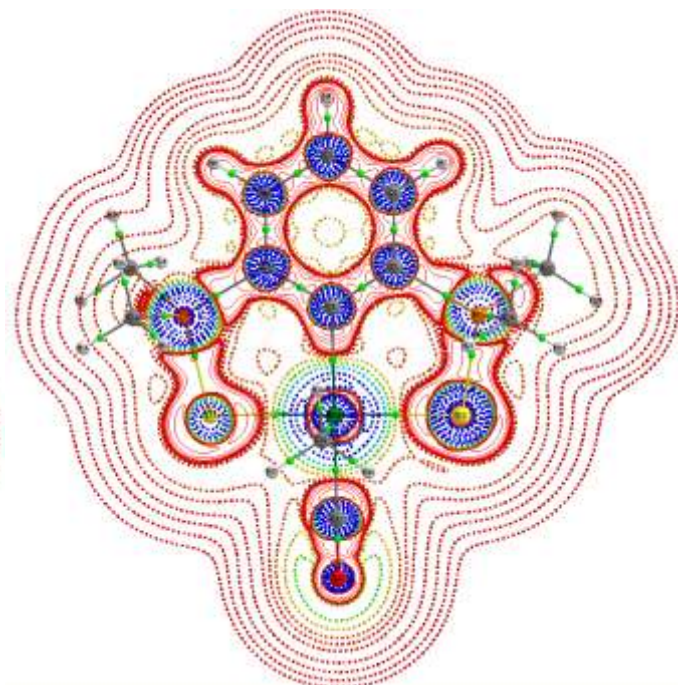
PCP-Rh-CH3



SCS-Rh-CH3

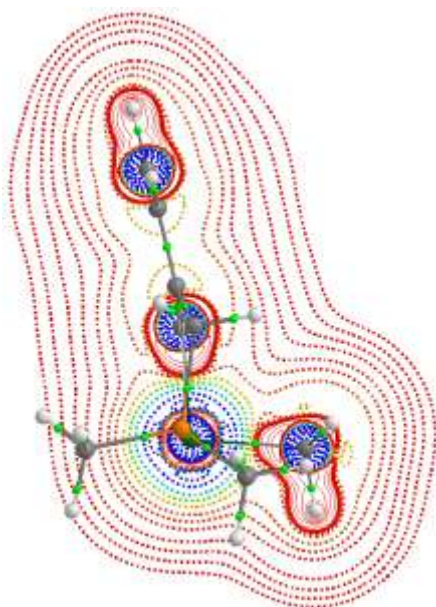


PCP-Rh-CH3-CO

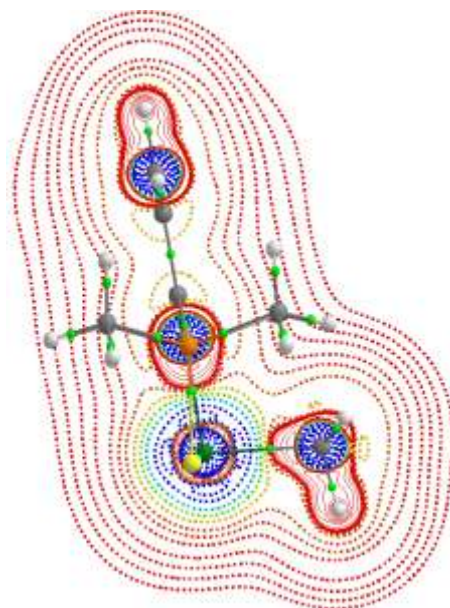


SCS-Rh-CH3-CO

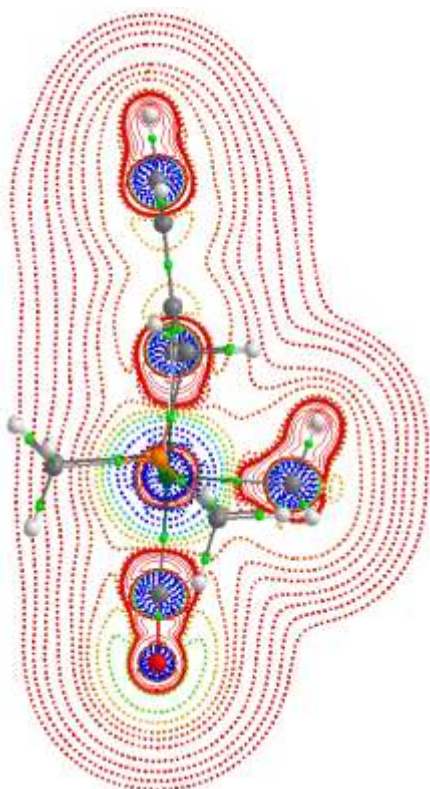
DelSqRho; min El. Dens=0.001; resolution 0.05; allow tubular; color scale 0-1



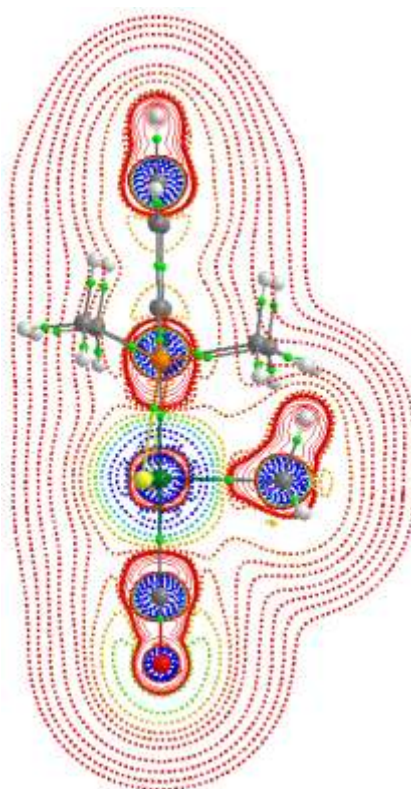
PCP-Rh-CH3



SCS-Rh-CH3

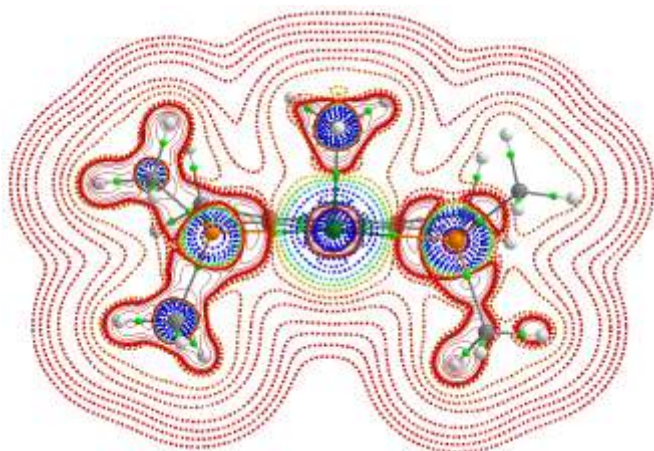


PCP-Rh-CH3-CO

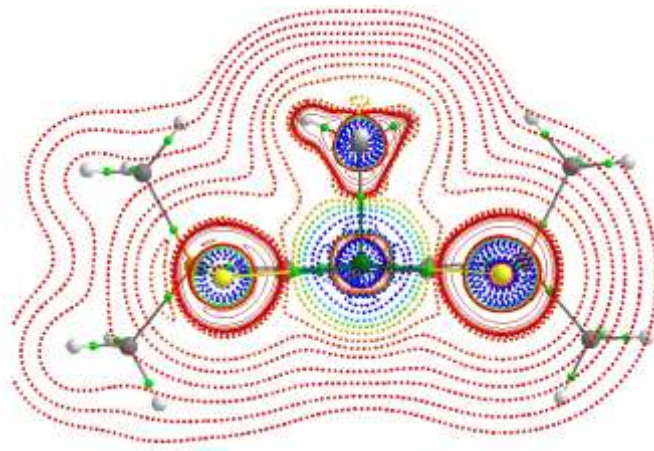


SCS-Rh-CH3-CO

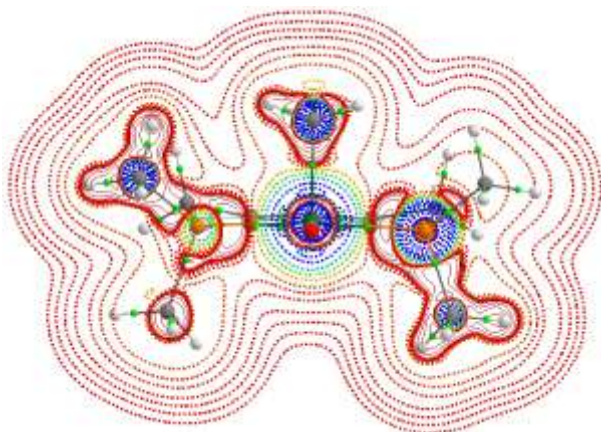
DelSqRho; min El. Dens=0.001; resolution 0.05; allow tubular; color scale 0-1



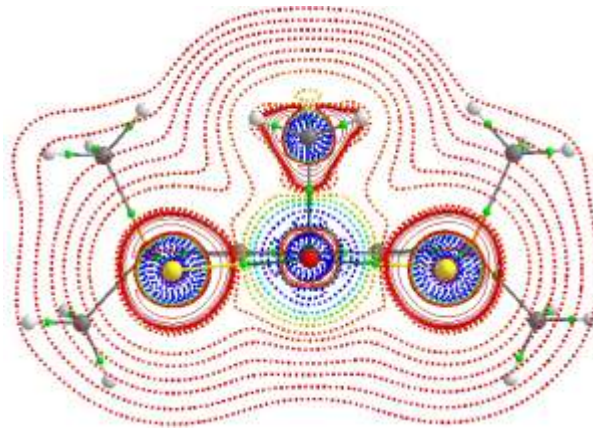
PCP-Rh-CH3



SCS-Rh-CH3



PCP-Rh-CH3-CO



SCS-Rh-CH3-CO