

# Supporting Information

## Gold-Catalyzed Tandem [3,3] Propargyl Ester Rearrangement Leading to *E*-1*H*-Inden-1-ones

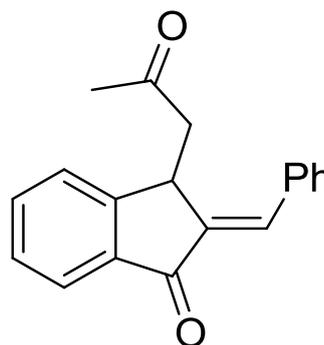
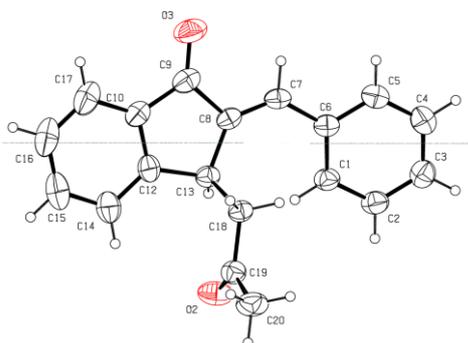
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## X-ray Diffraction Analysis of Compound 2a



**X-ray structure of 2a (CCDC 960618)**

Bond precision: C-C = 0.0073 Å Wavelength=0.71070

Cell: a=11.2513(11) b=16.593(2) c=8.0237(9)  
 alpha=90 beta=90 gamma=90

Temperature: 292 K

	Calculated	Reported
Volume	1498.0(3)	1498.0(3)
Space group	P c a 21	P c a 21
Hall group	P 2c -2ac	P 2c -2ac
Moiety formula	C19 H16 O2	C19 H16 O2
Sum formula	C19 H16 O2	C19 H16 O2
Mr	276.32	276.32
Dx, g cm <sup>-3</sup>	1.225	1.225
Z	4	4
Mu (mm <sup>-1</sup> )	0.078	0.078
F000	584.0	584.0
F000'	584.27	
h,k,lmax	14,20,10	14,20,10
Nref	3042[ 1639]	2648
Tmin,Tmax	0.978,0.991	0.325,1.000
Tmin'	0.975	

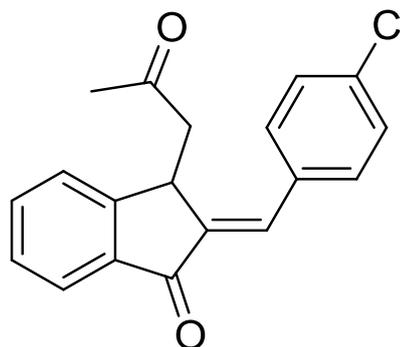
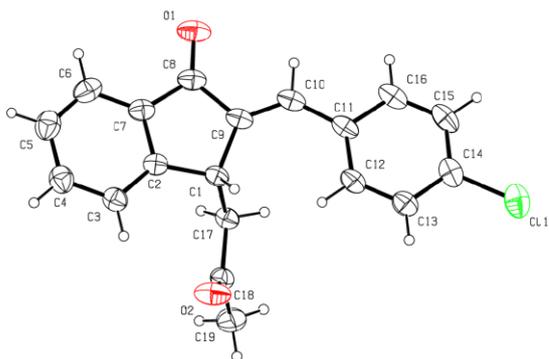
Correction method= MULTI-SCAN

Data completeness= 1.62/0.87 Theta(max)= 26.370

R(reflections)= 0.0699( 1419) wR2(reflections)= 0.1565( 2648)

S = 1.070 Npar= 192

## X-ray Diffraction Analysis of Compound 2d



**X-ray structure of 2d**

Bond precision: C-C = 0.0032 Å Wavelength=0.71073

Cell: a=5.9699(3) b=8.9969(6) c=28.9795(15)

alpha=90 beta=95.309(6) gamma=90

Temperature: 293 K

	Calculated	Reported
Volume	1549.83(15)	1549.81(16)
Space group	P 21/c	P 1 21/c 1
Hall group	-P 2ybc	-P 2ybc
Moiety formula	C19 H15 Cl O2	C19 H15 Cl O2
Sum formula	C19 H15 Cl O2	C19 H15 Cl O2
Mr	310.76	310.76
Dx, g cm <sup>-3</sup>	1.332	1.332
Z	4	4
Mu (mm <sup>-1</sup> )	0.251	0.251
F000	648.0	648.0
F000'	648.86	
h,k,lmax	7,11,35	7,11,35
Nref	3059	3045
Tmin,Tmax	0.909,0.921	0.540,1.000
Tmin'	0.909	

Correction method= MULTI-SCAN

Data completeness= 0.995

Theta(max)= 26.021

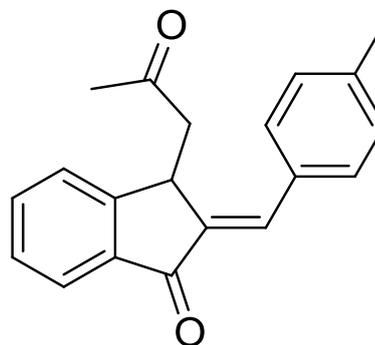
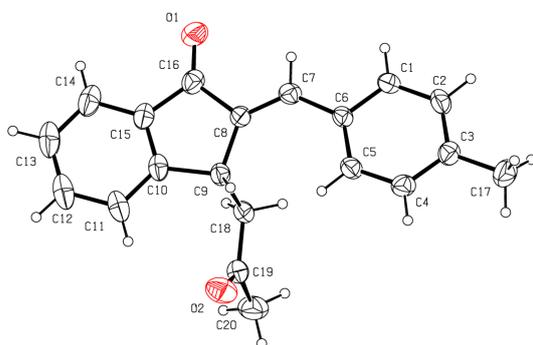
R(reflections)= 0.0506( 2151)

wR2(reflections)= 0.1376( 3045)

S = 1.049

Npar= 200

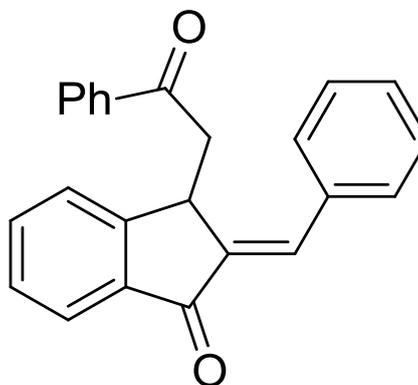
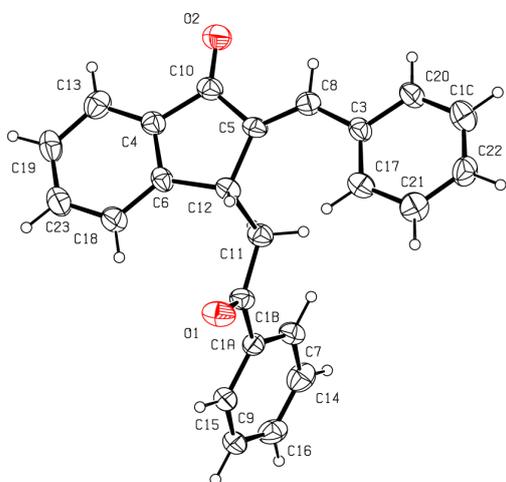
## X-ray Diffraction Analysis of Compound 2f



**X-ray structure of 2f**

Bond precision: C-C = 0.0059 Å Wavelength=0.71073  
Cell: a=5.5429(6) b=8.4278(9) c=34.013(3)  
alpha=90 beta=90 gamma=90  
Temperature: 293 K  
Calculated Reported  
Volume 1588.9(3) 1588.9(3)  
Space group P 21 21 21 P 21 21 21  
Hall group P 2ac 2ab P 2ac 2ab  
Moiety formula C<sub>20</sub> H<sub>18</sub> O<sub>2</sub> C<sub>20</sub> H<sub>18</sub> O<sub>2</sub>  
Sum formula C<sub>40</sub> H<sub>36</sub> O<sub>4</sub>  
Mr 290.34 290.34  
D<sub>x</sub>, g cm<sup>-3</sup> 1.214 1.214  
Z 4 4  
Mu (mm<sup>-1</sup>) 0.077 0.077  
F<sub>000</sub> 616.0 616.0  
F<sub>000</sub>' 616.28  
h,k,l<sub>max</sub> 6,10,41 6,10,41  
N<sub>ref</sub> 3125[ 1860] 2949  
T<sub>min</sub>, T<sub>max</sub>  
T<sub>min</sub>'  
Correction method= Not given  
Data completeness= 1.59/0.94 Theta(max)= 26.020  
R(reflections)= 0.0664( 1941) wR<sub>2</sub>(reflections)= 0.1653( 2949)  
S = 1.077 N<sub>par</sub>= 202

## X-ray Diffraction Analysis of Compound 2o



**X-ray structure of 2o**

Bond precision: C-C = 0.0062 Å Wavelength=0.71070

Cell: a=14.712(3) b=5.8398(14) c=20.996(6)

alpha=90 beta=105.27(3) gamma=90

Temperature: 293 K

Calculated Reported

Volume 1740.2(8) 1740.2(8)

Space group P 21/c P 1 21/c 1

Hall group -P 2ybc -P 2ybc

Moiety formula C<sub>24</sub> H<sub>18</sub> O<sub>2</sub> C<sub>24</sub> H<sub>18</sub> O<sub>2</sub>

Sum formula C<sub>24</sub> H<sub>18</sub> O<sub>2</sub> C<sub>24</sub> H<sub>18</sub> O<sub>2</sub>

Mr 338.38 338.38

D<sub>x</sub>, g cm<sup>-3</sup> 1.292 1.292

Z 4 4

Mu (mm<sup>-1</sup>) 0.081 0.081

F<sub>000</sub> 712.0 712.0

F<sub>000</sub>' 712.32

h,k,l<sub>max</sub> 18,7,25 18,7,25

N<sub>ref</sub> 3425 3399

T<sub>min</sub>, T<sub>max</sub> 0.982, 0.994 0.071, 1.000

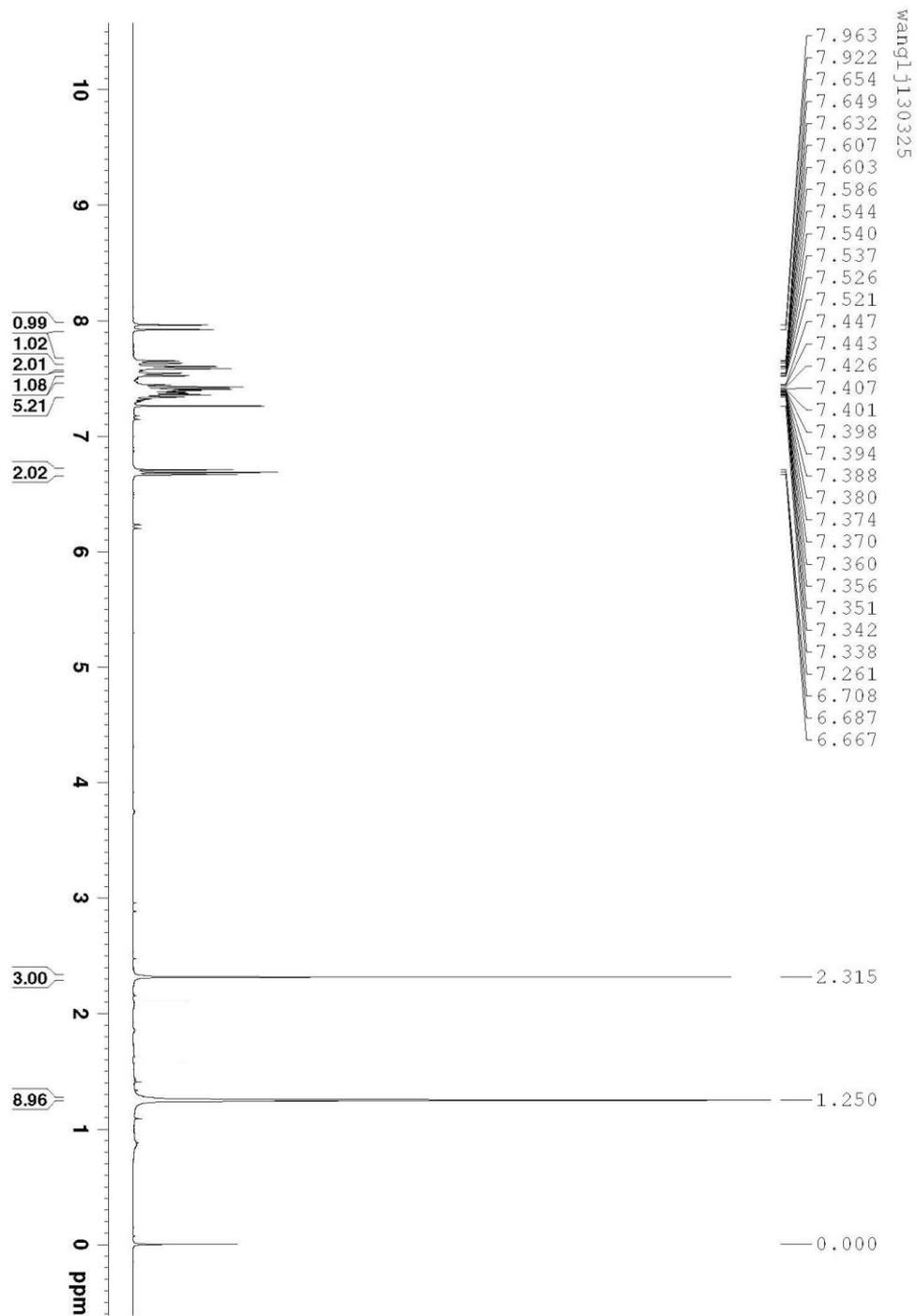
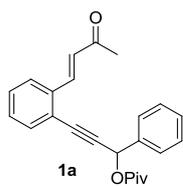
T<sub>min</sub>' 0.981

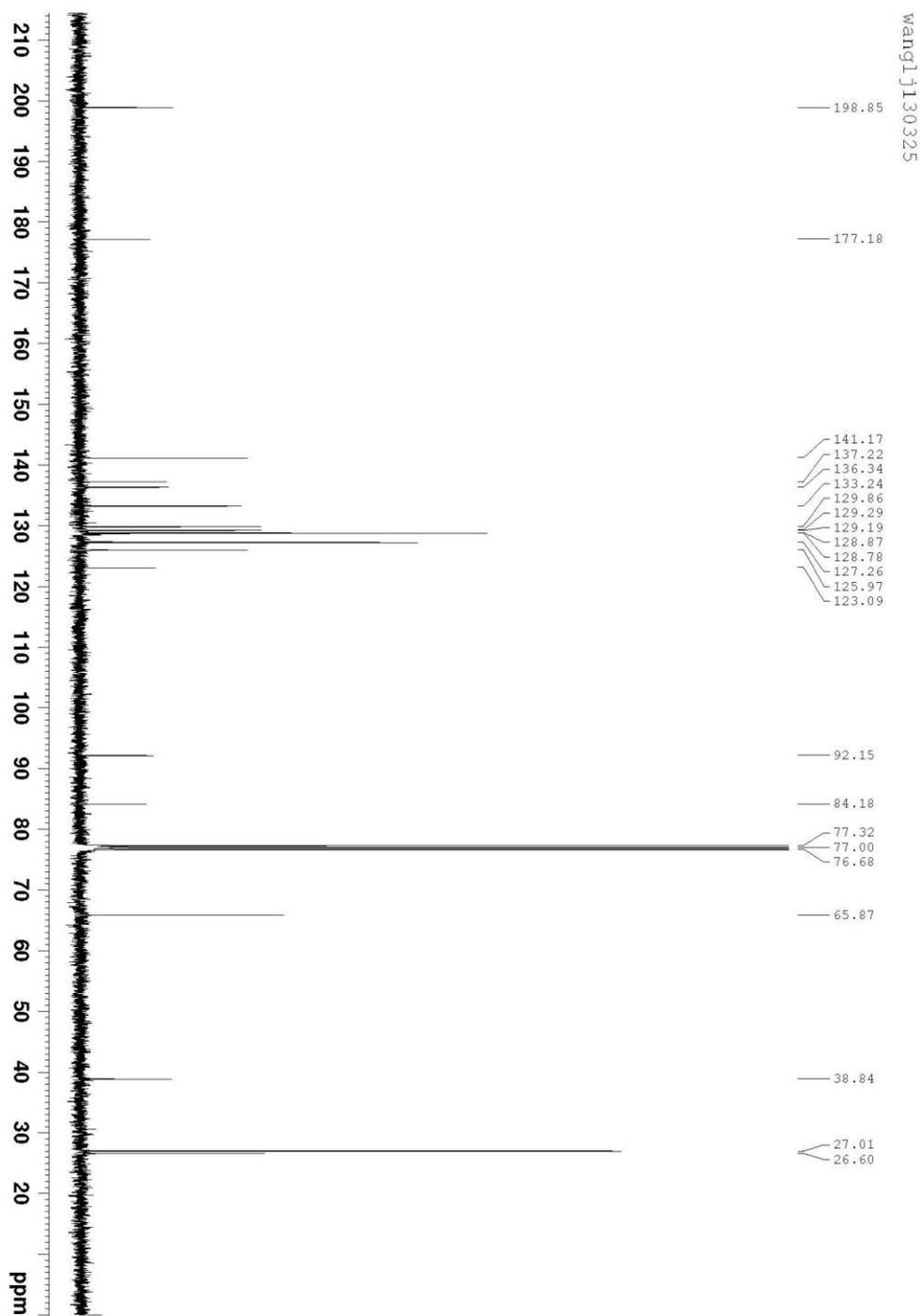
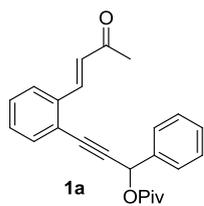
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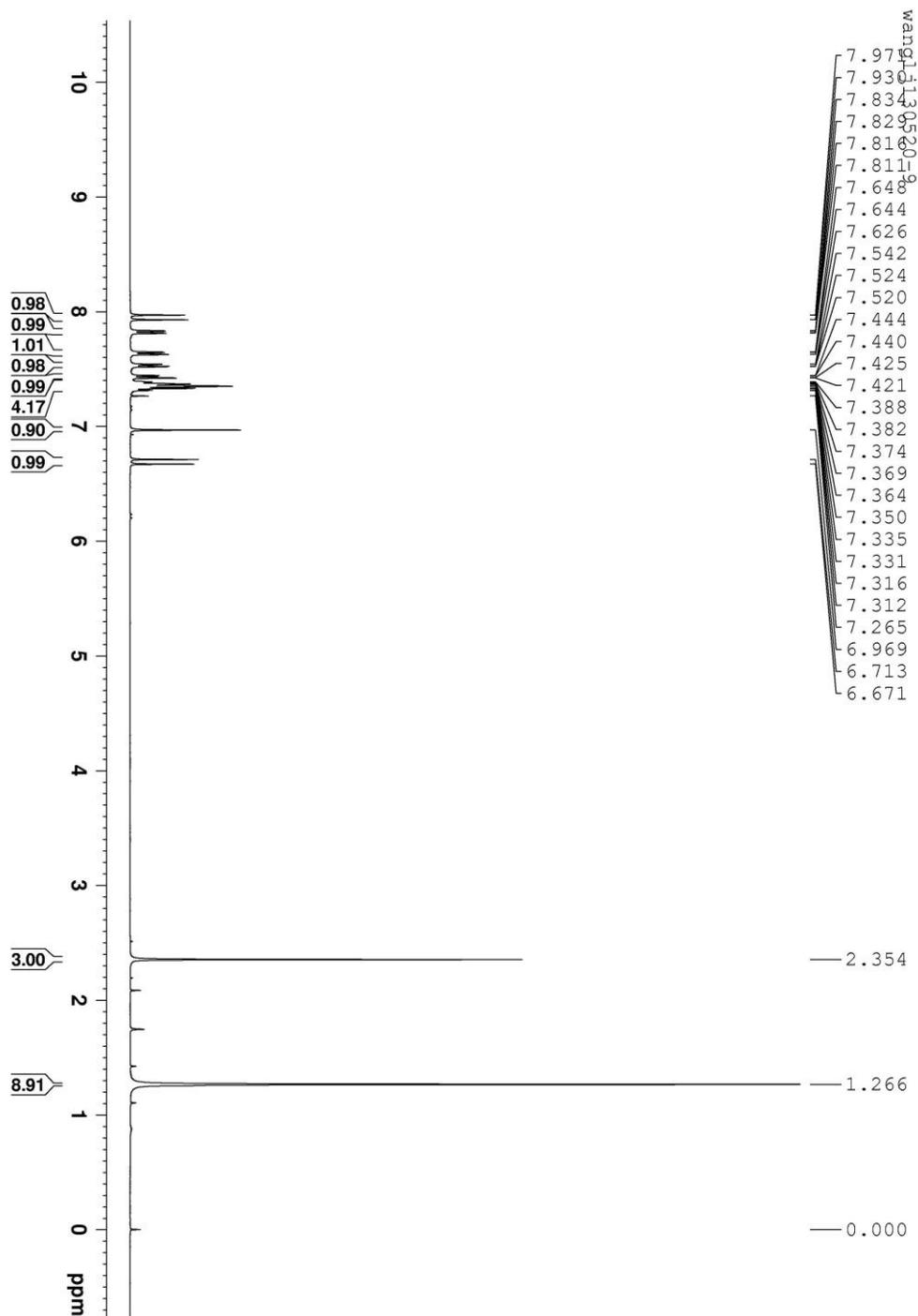
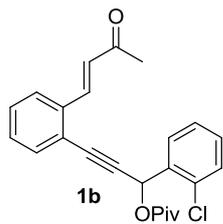
Data completeness= 0.992 Theta(max)= 26.020

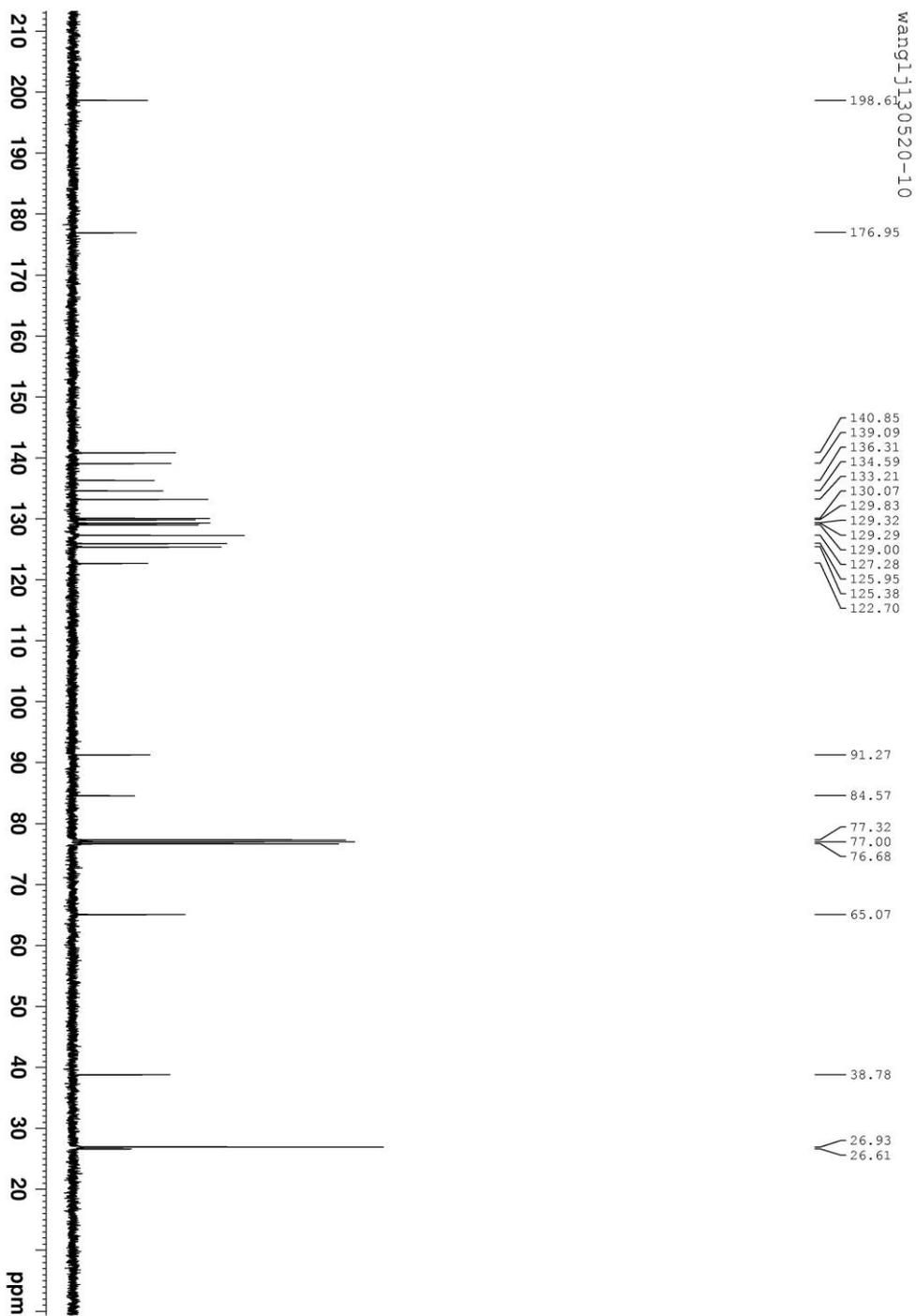
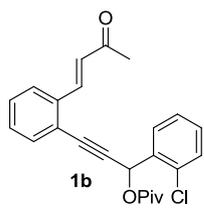
R(reflections)= 0.0755( 1295) wR<sub>2</sub>(reflections)= 0.1987( 3399)

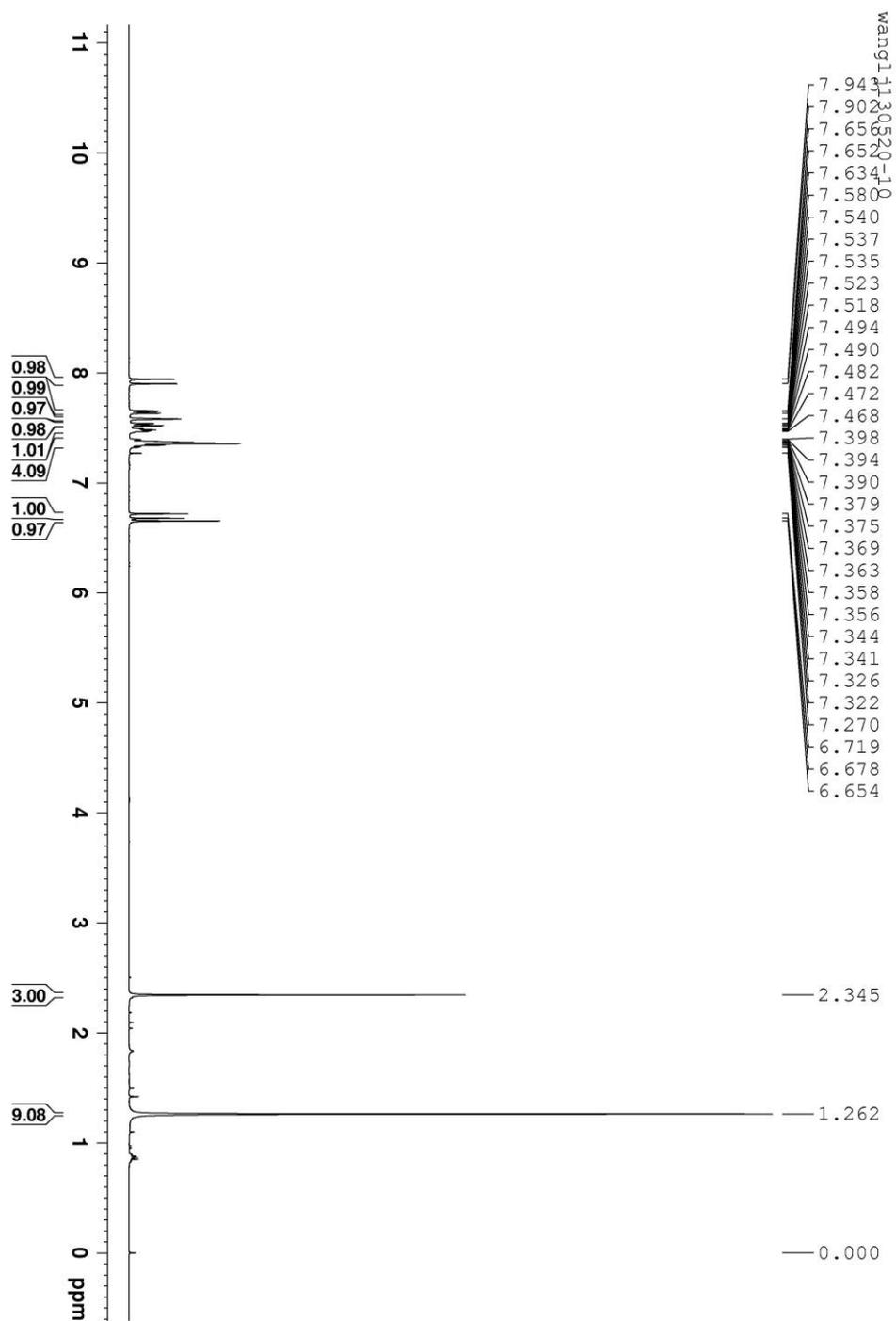
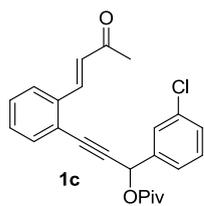
S = 0.964 N<sub>par</sub>= 236

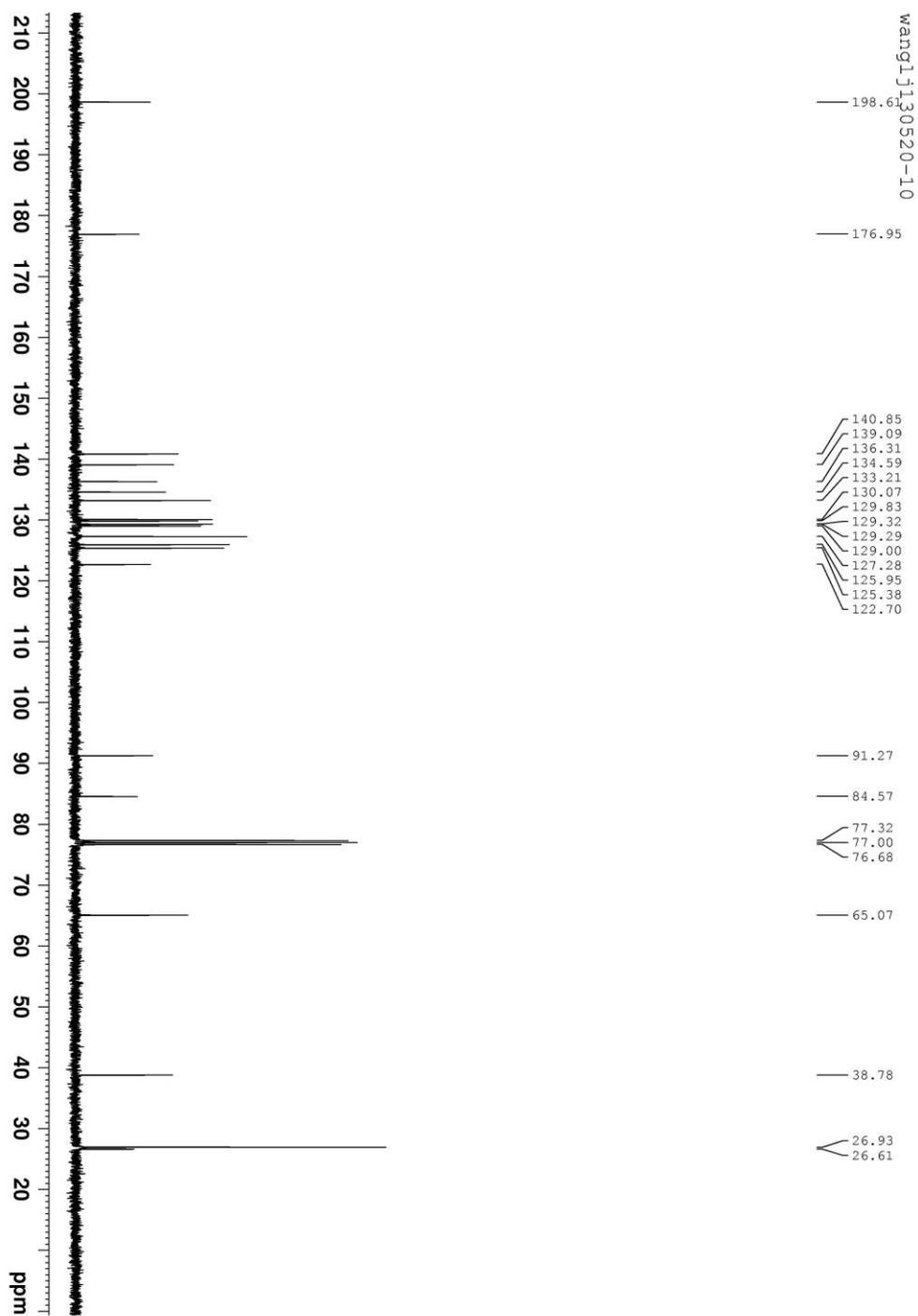
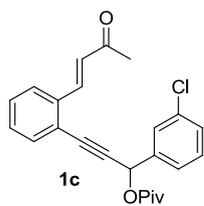


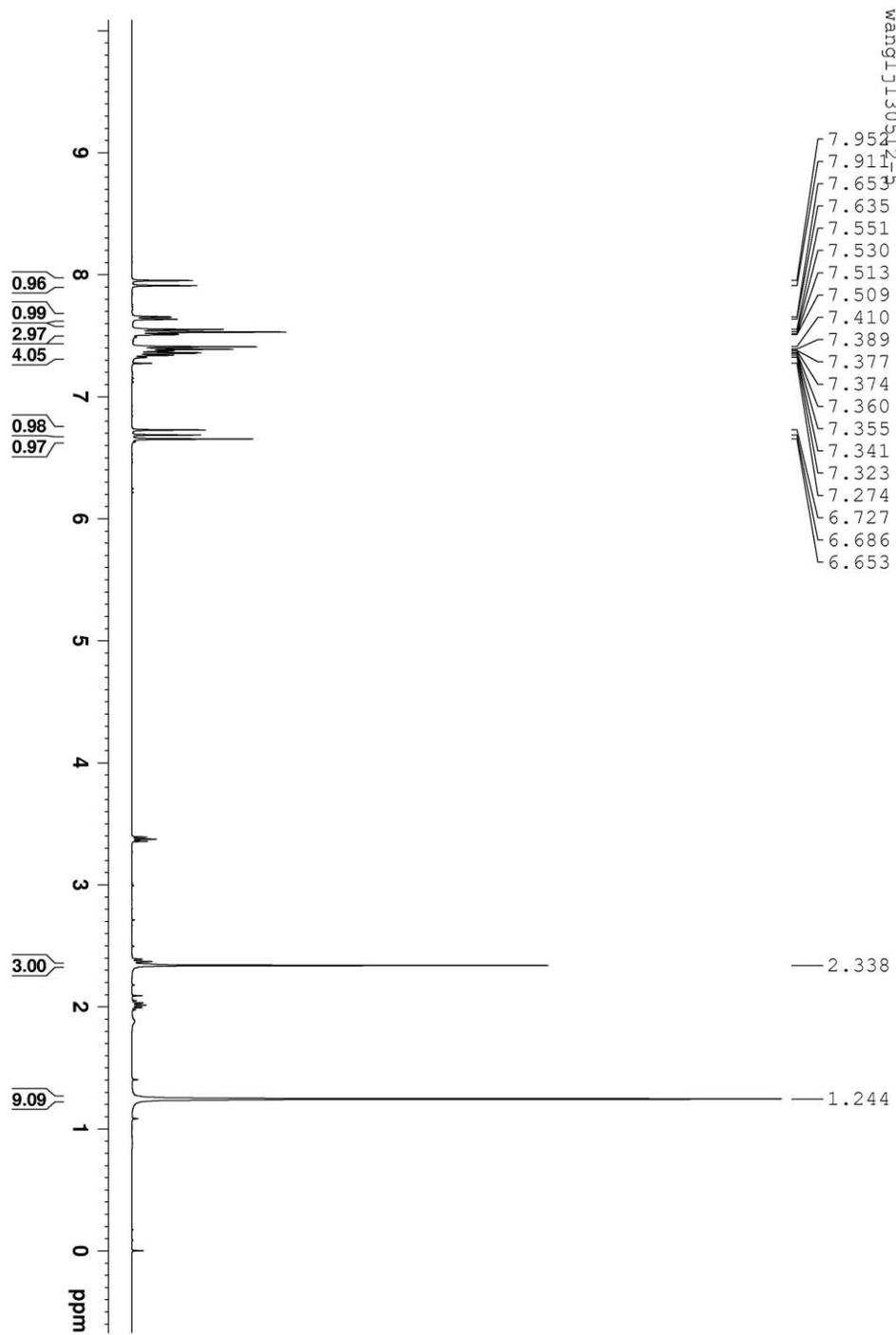
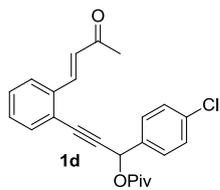


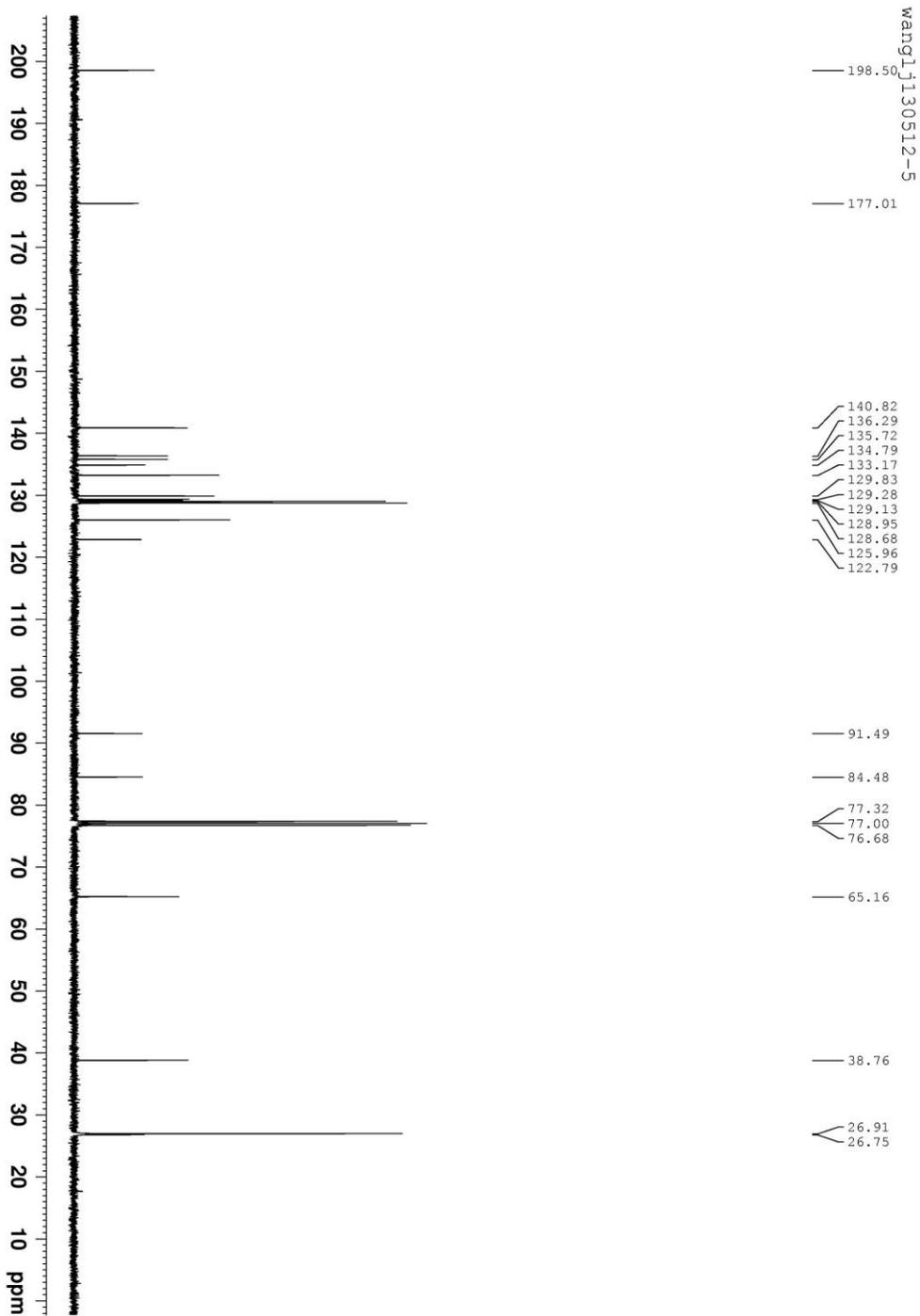
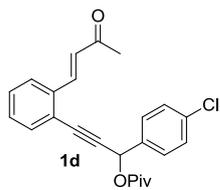


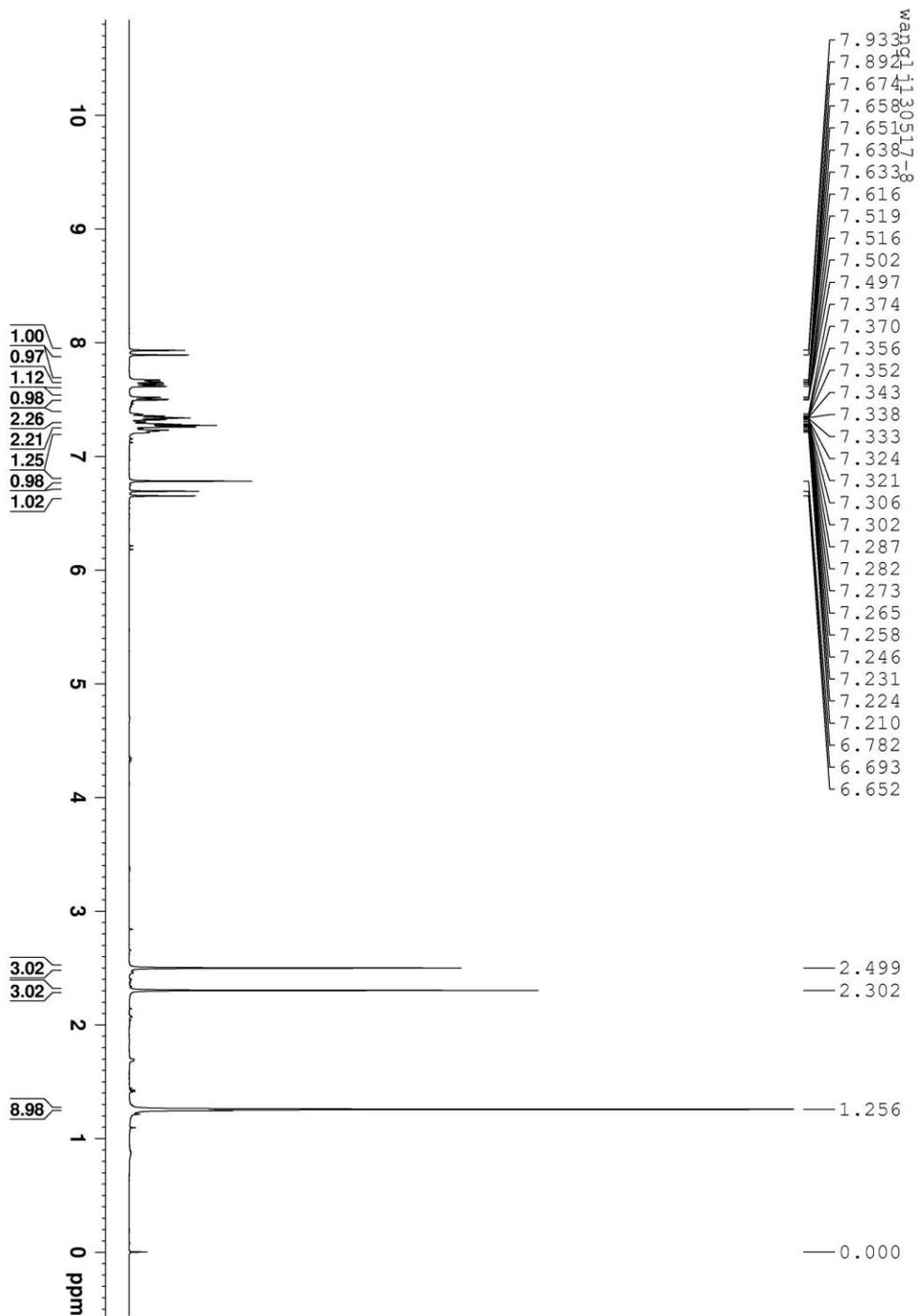
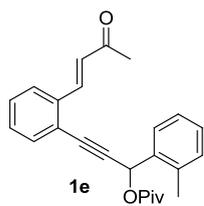


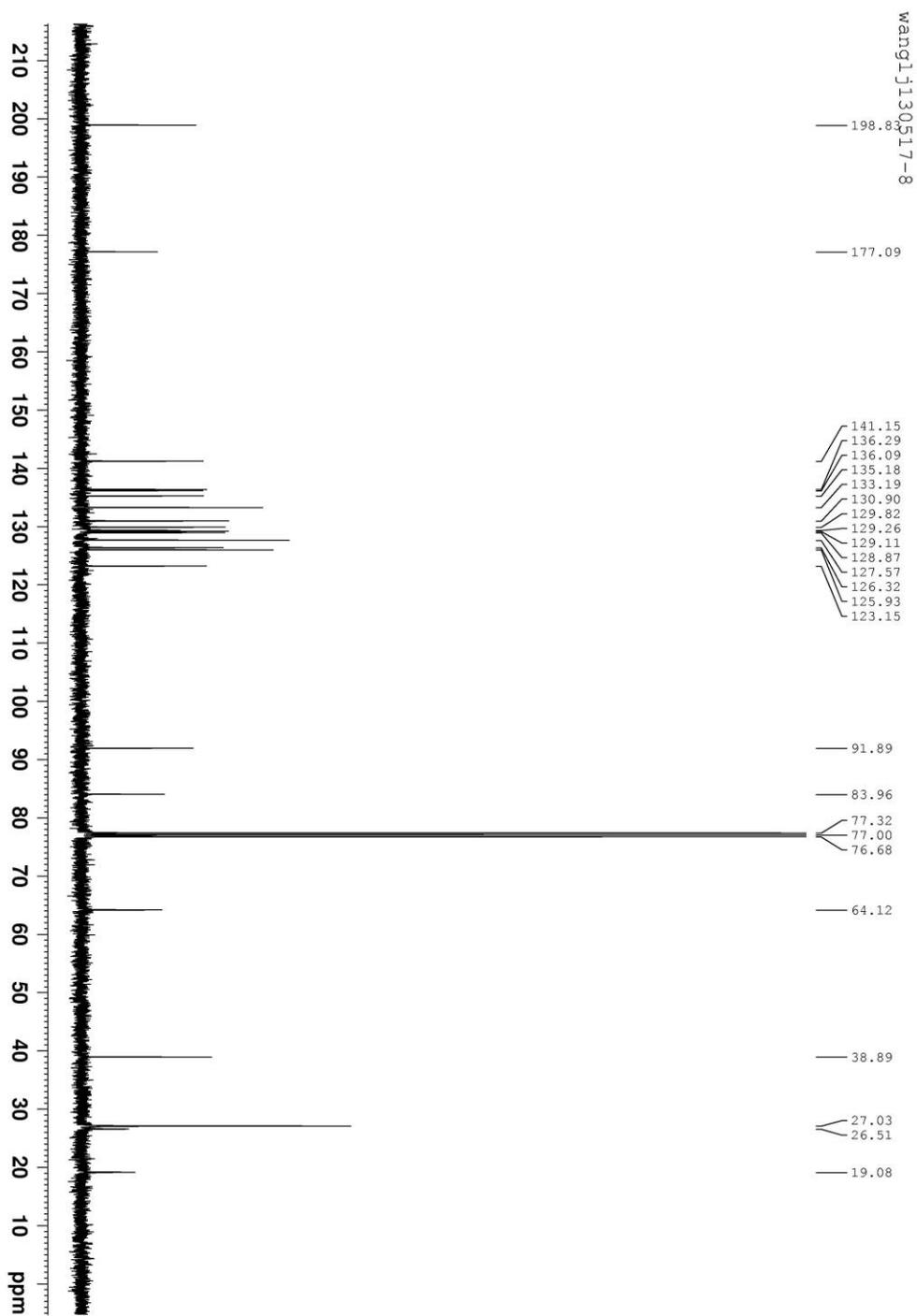
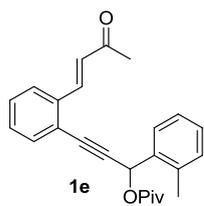


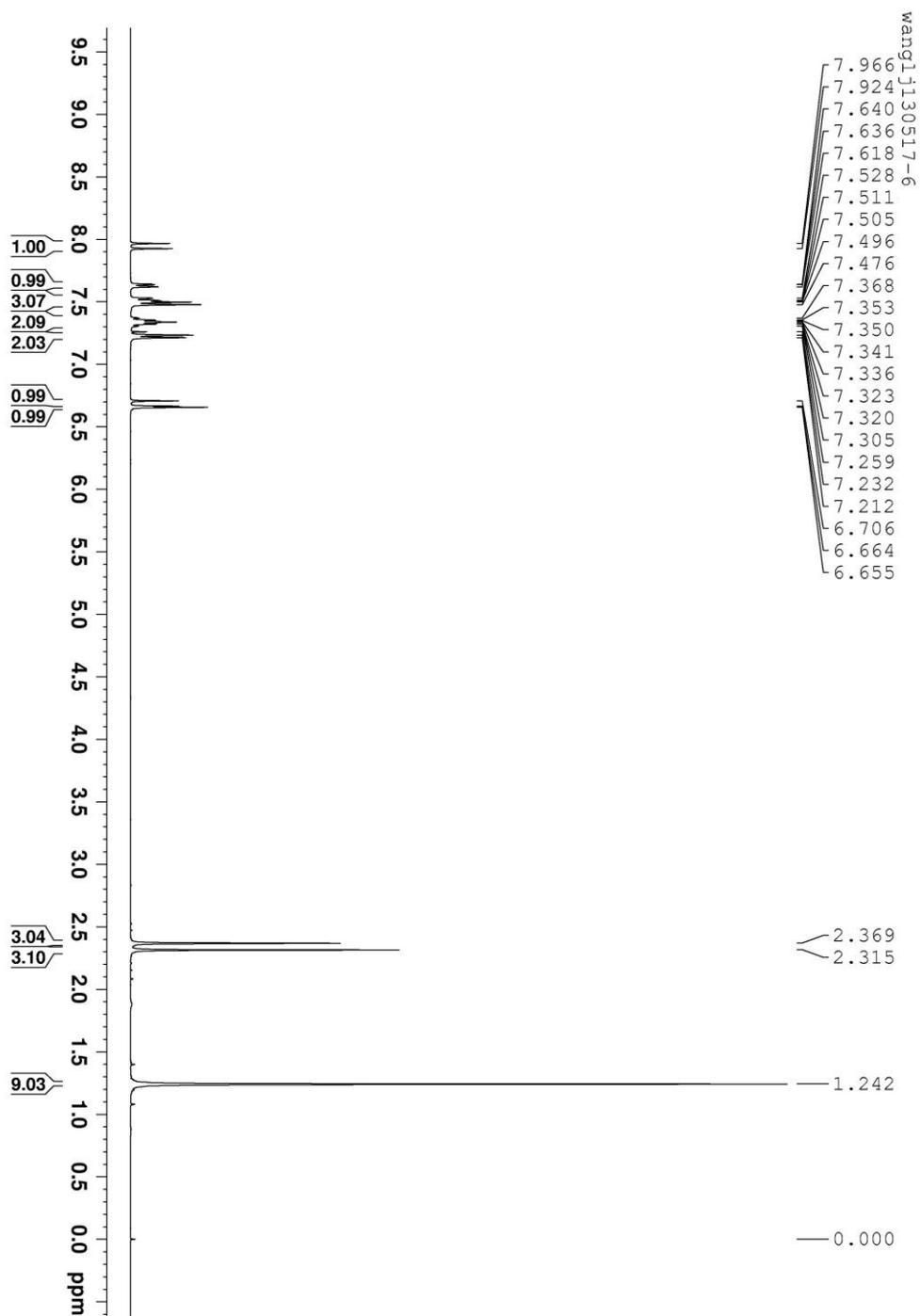
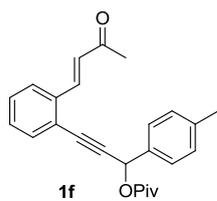


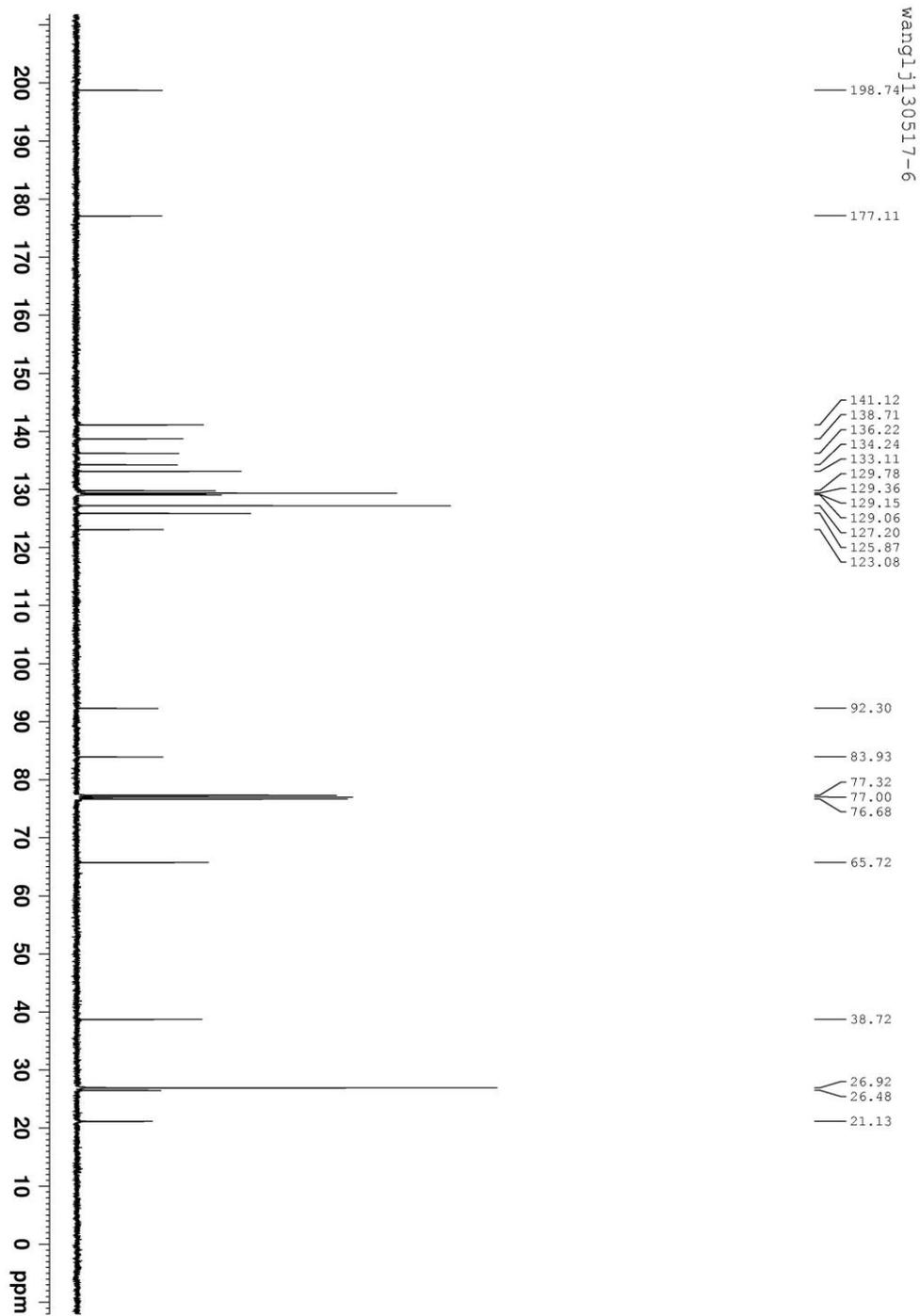
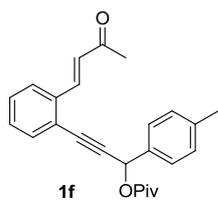


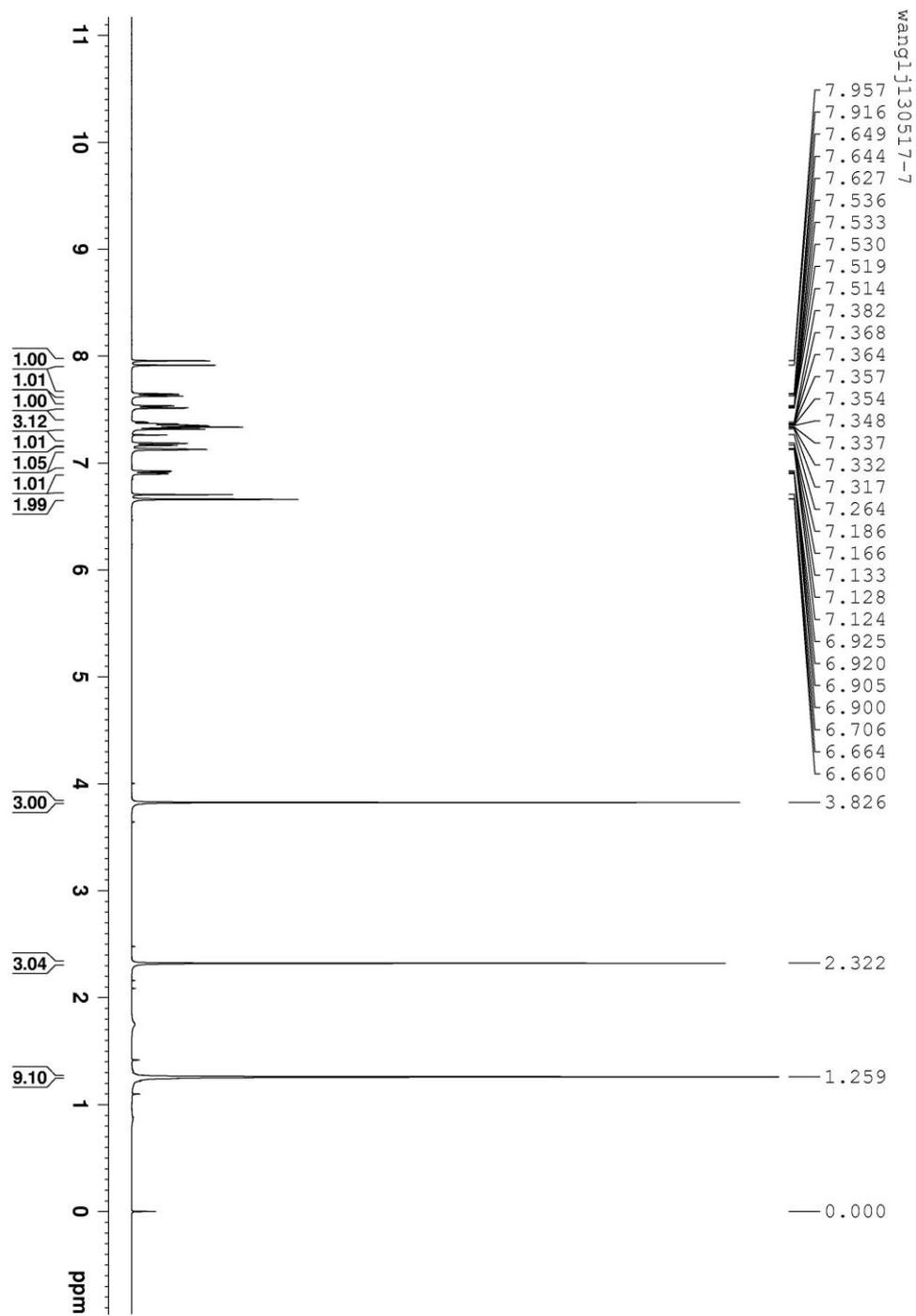
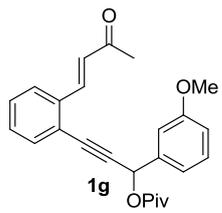


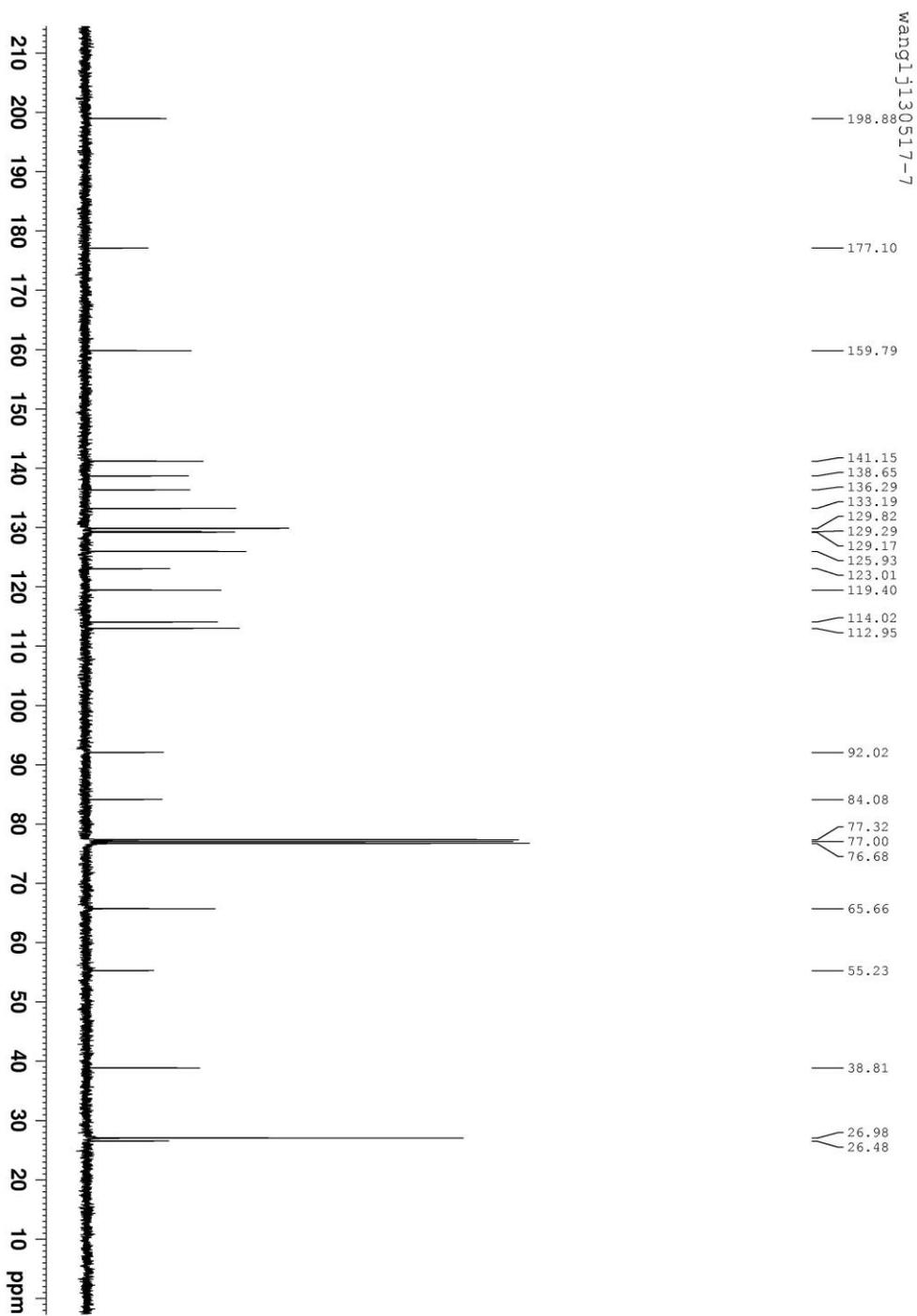
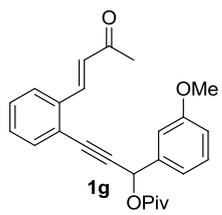


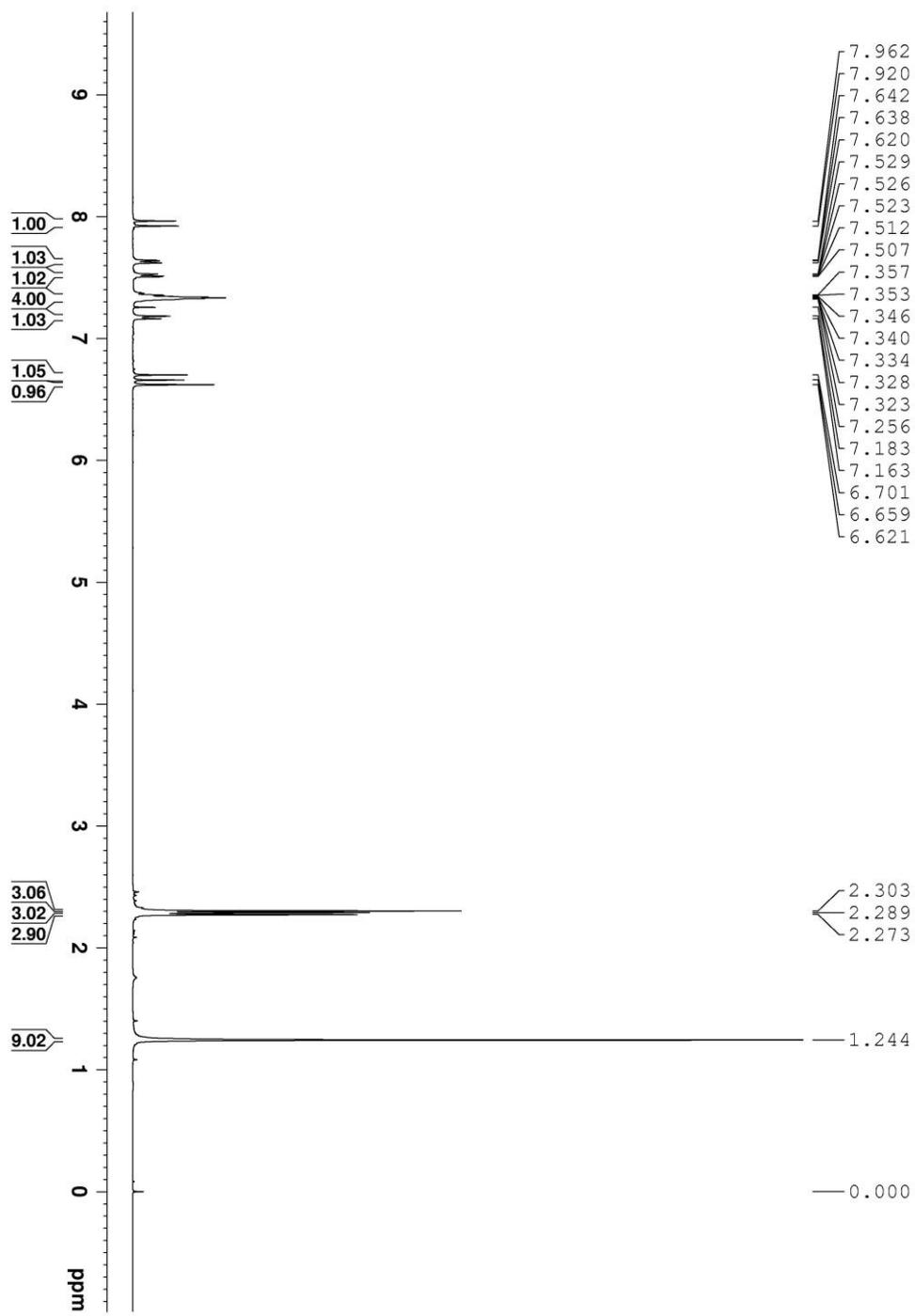
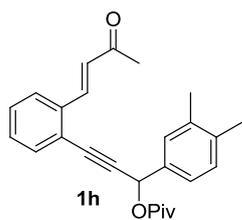


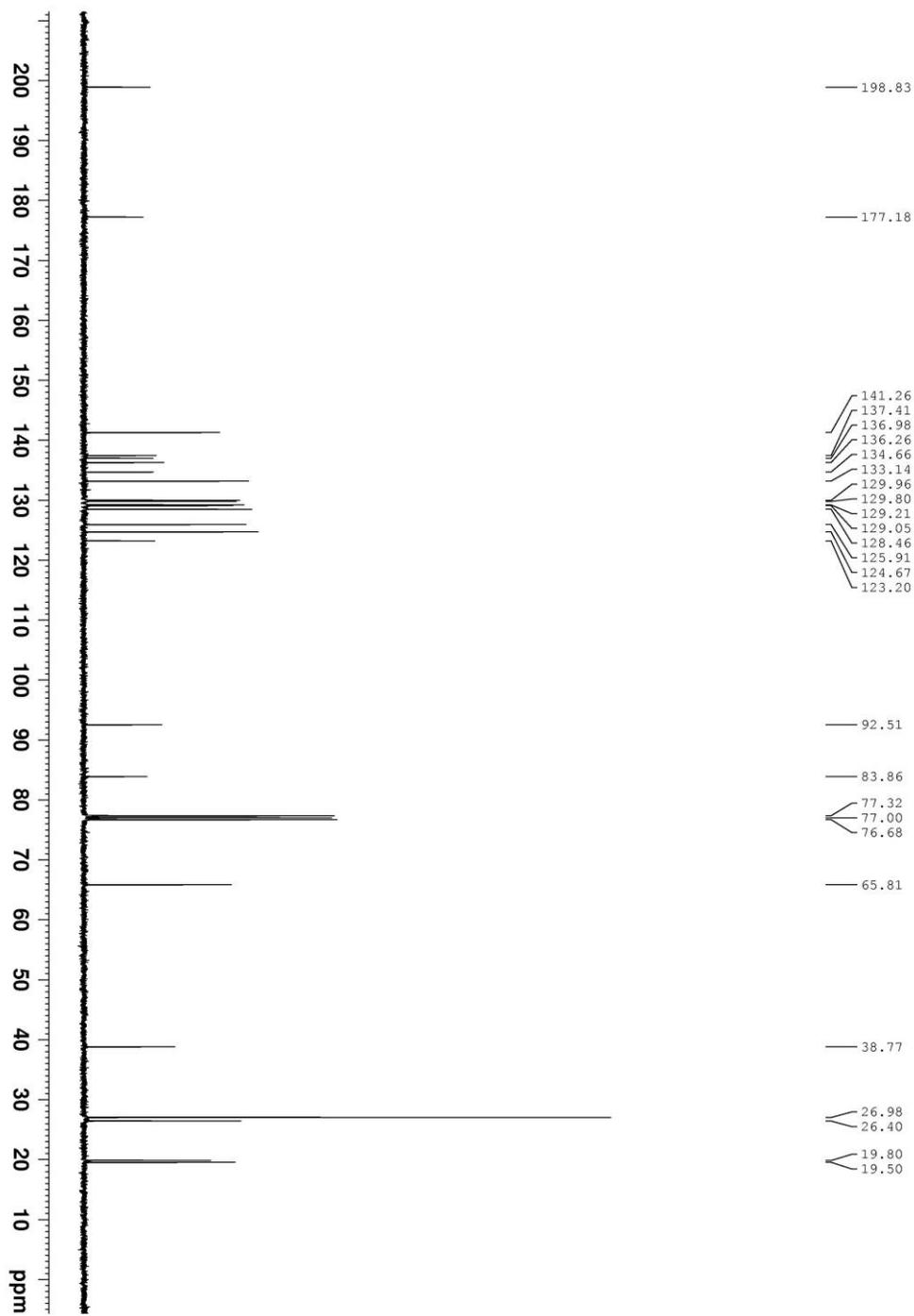
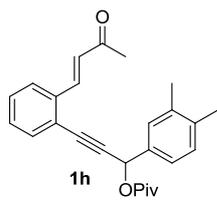


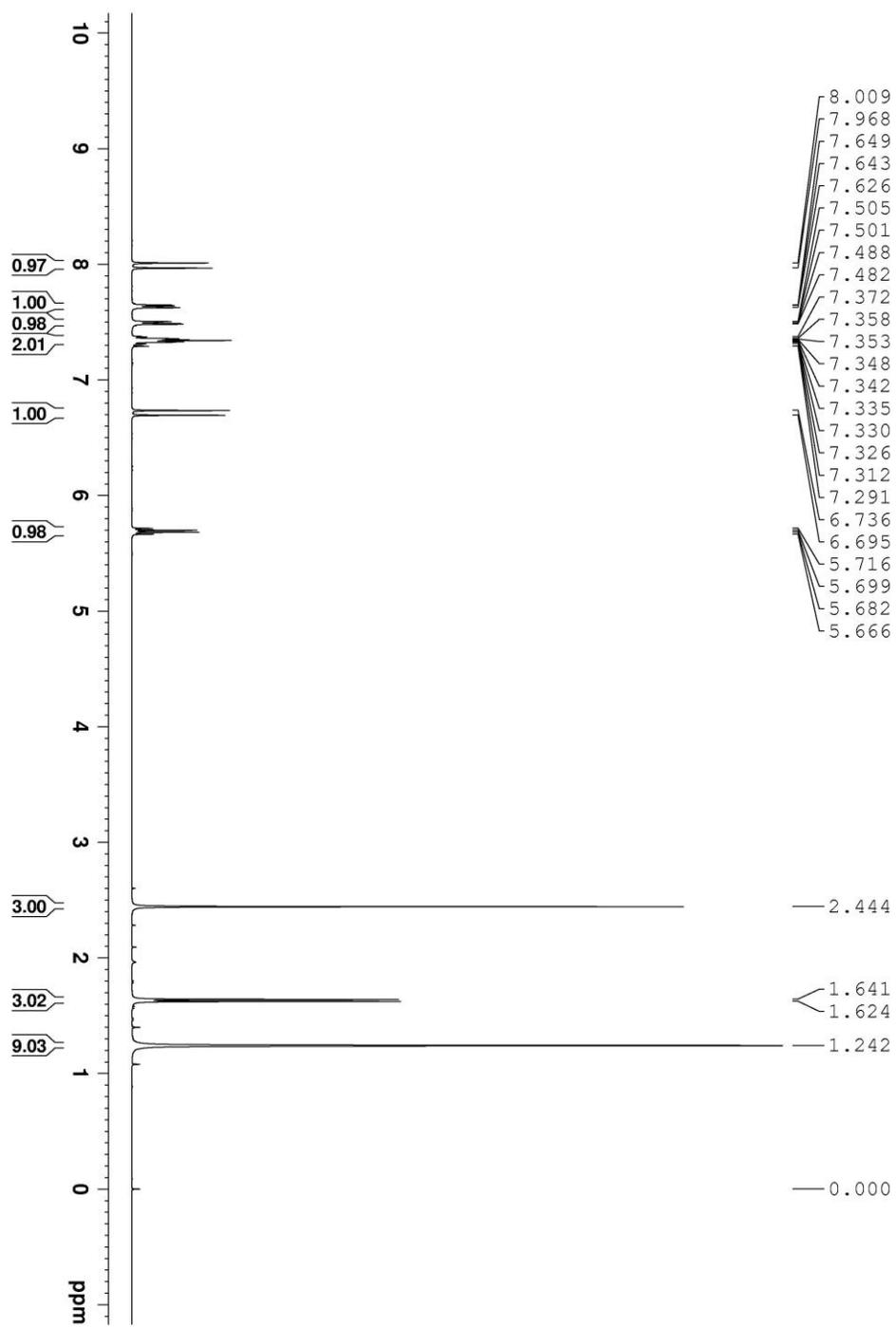
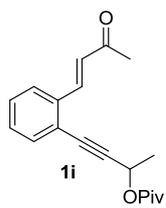


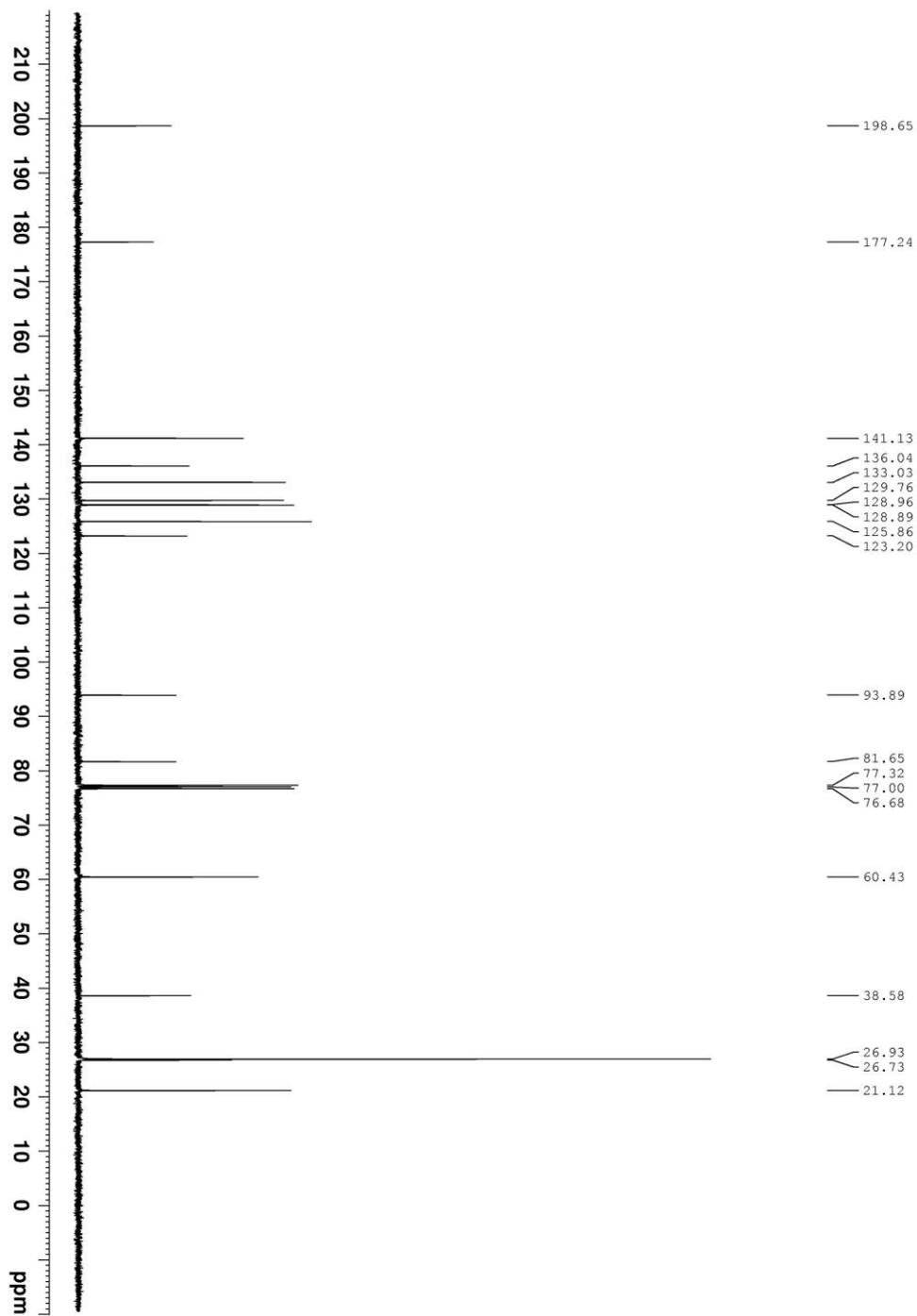
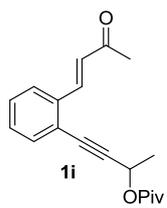


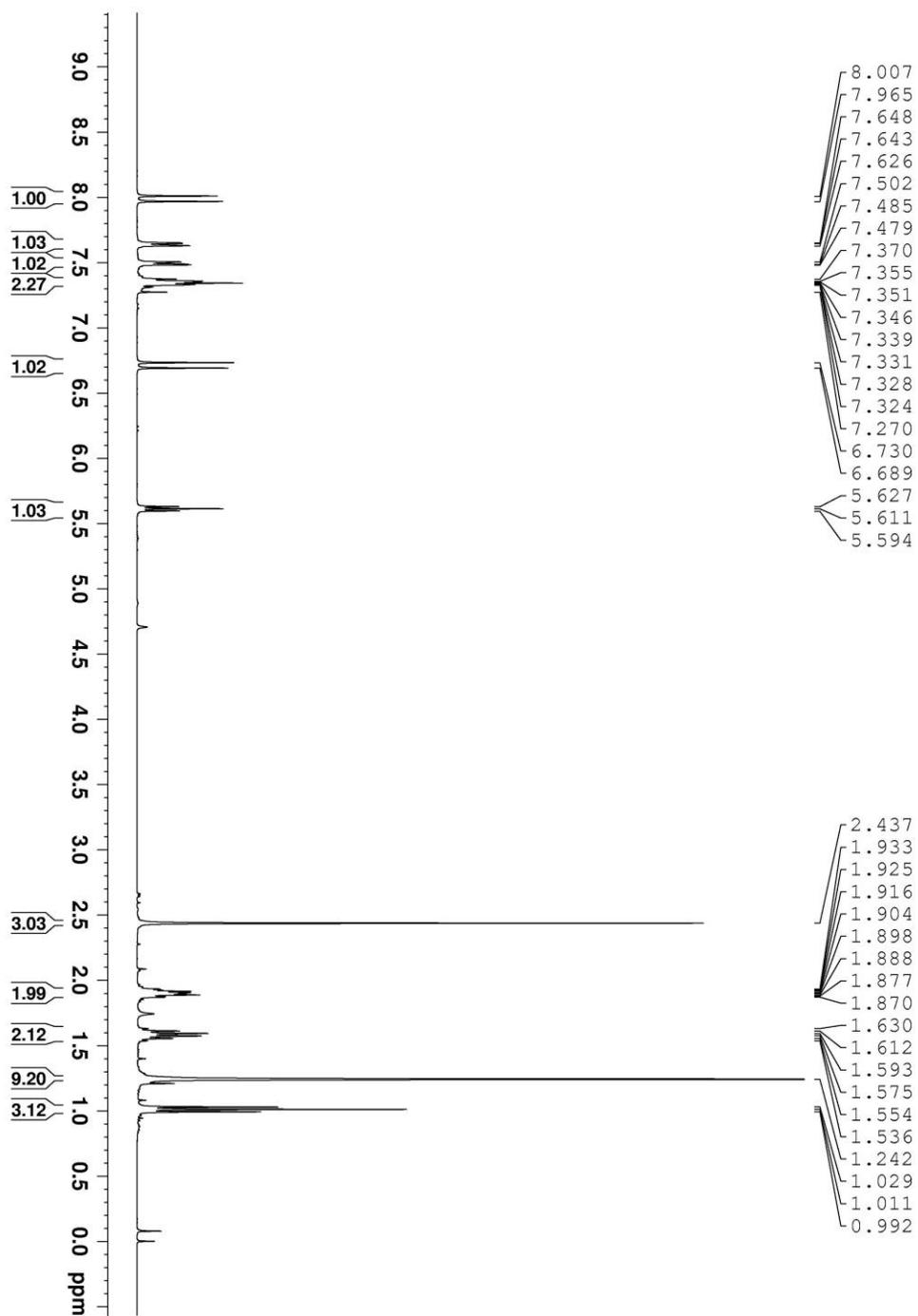
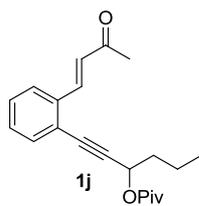


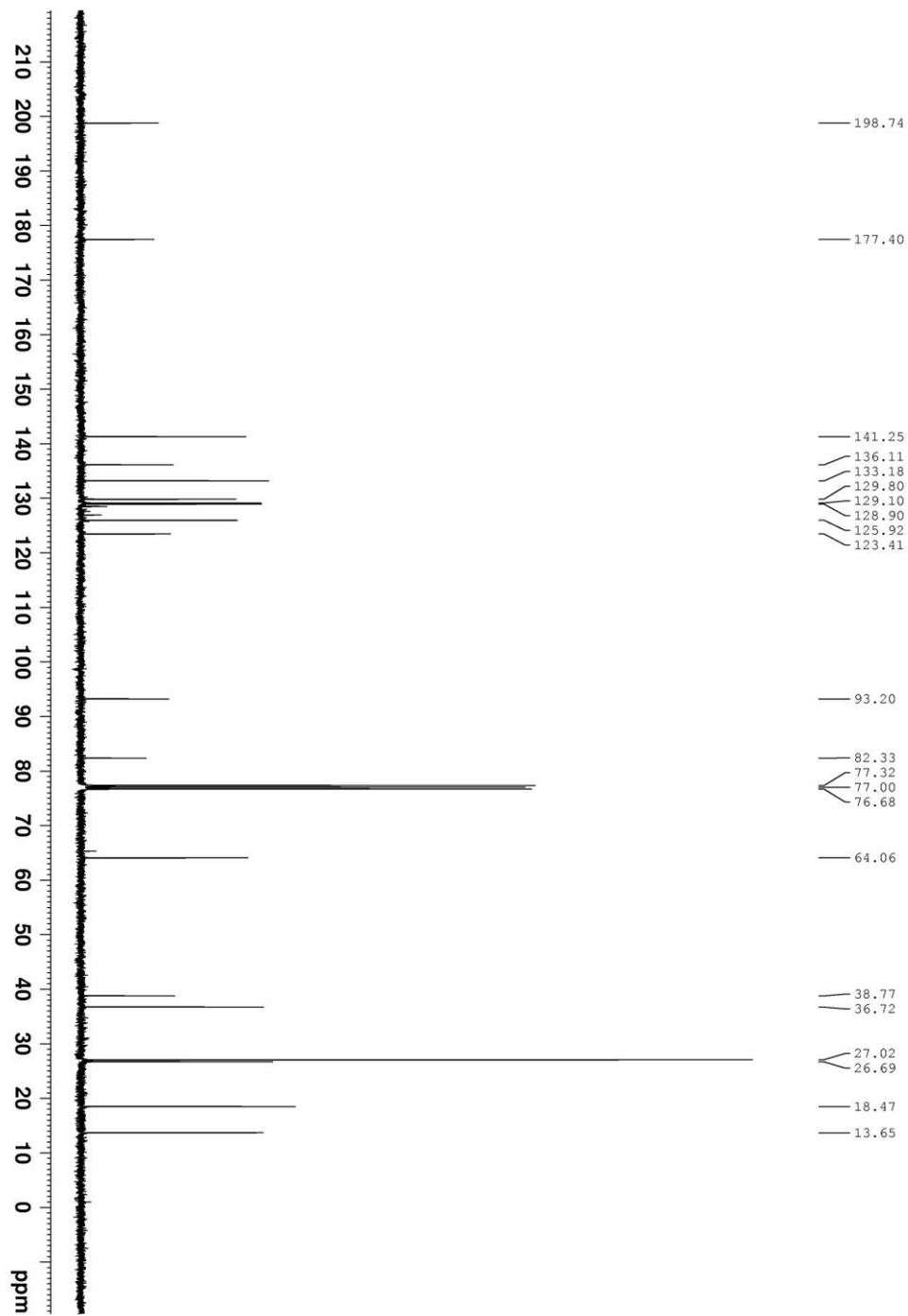
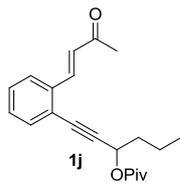


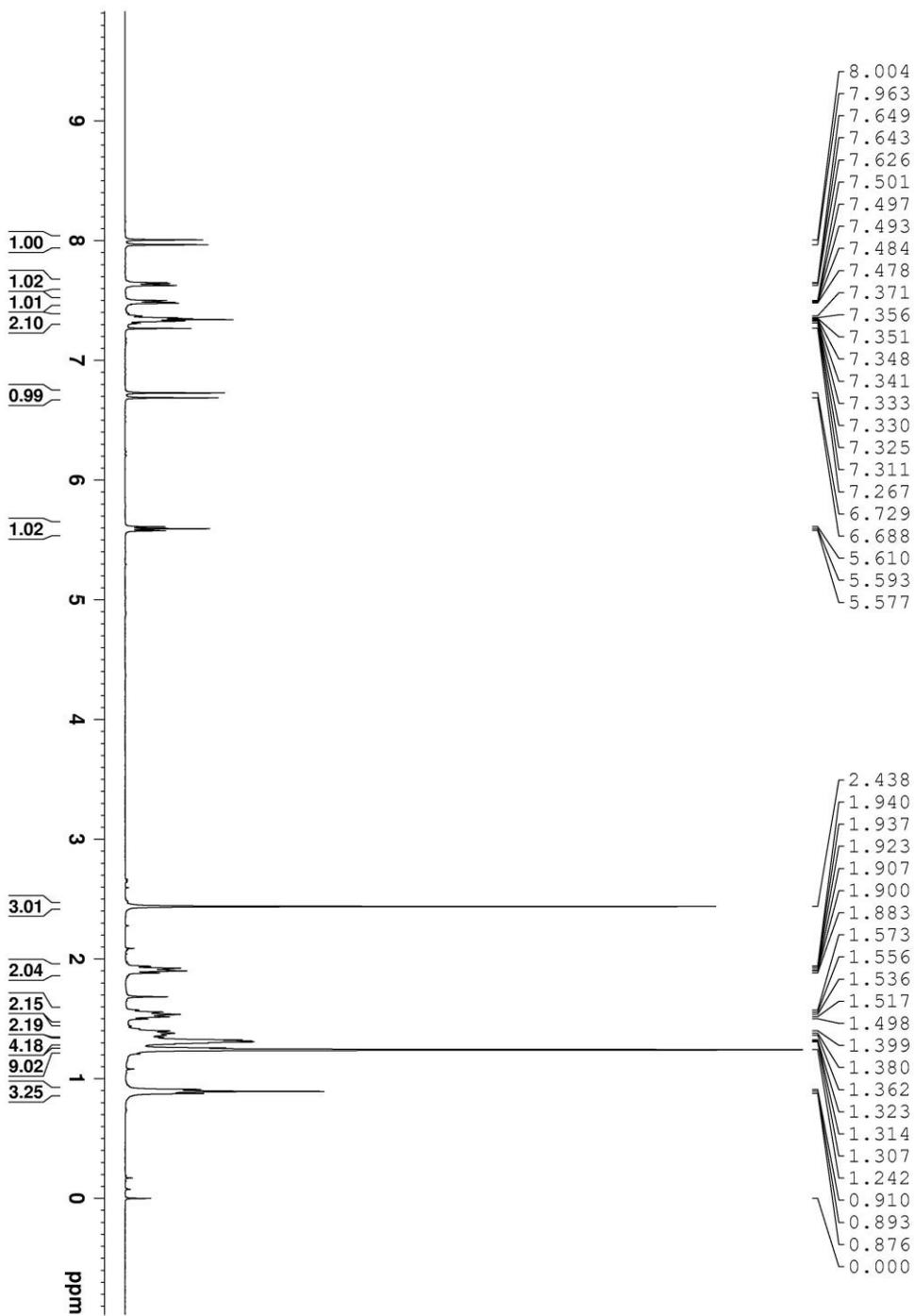
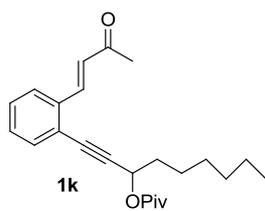


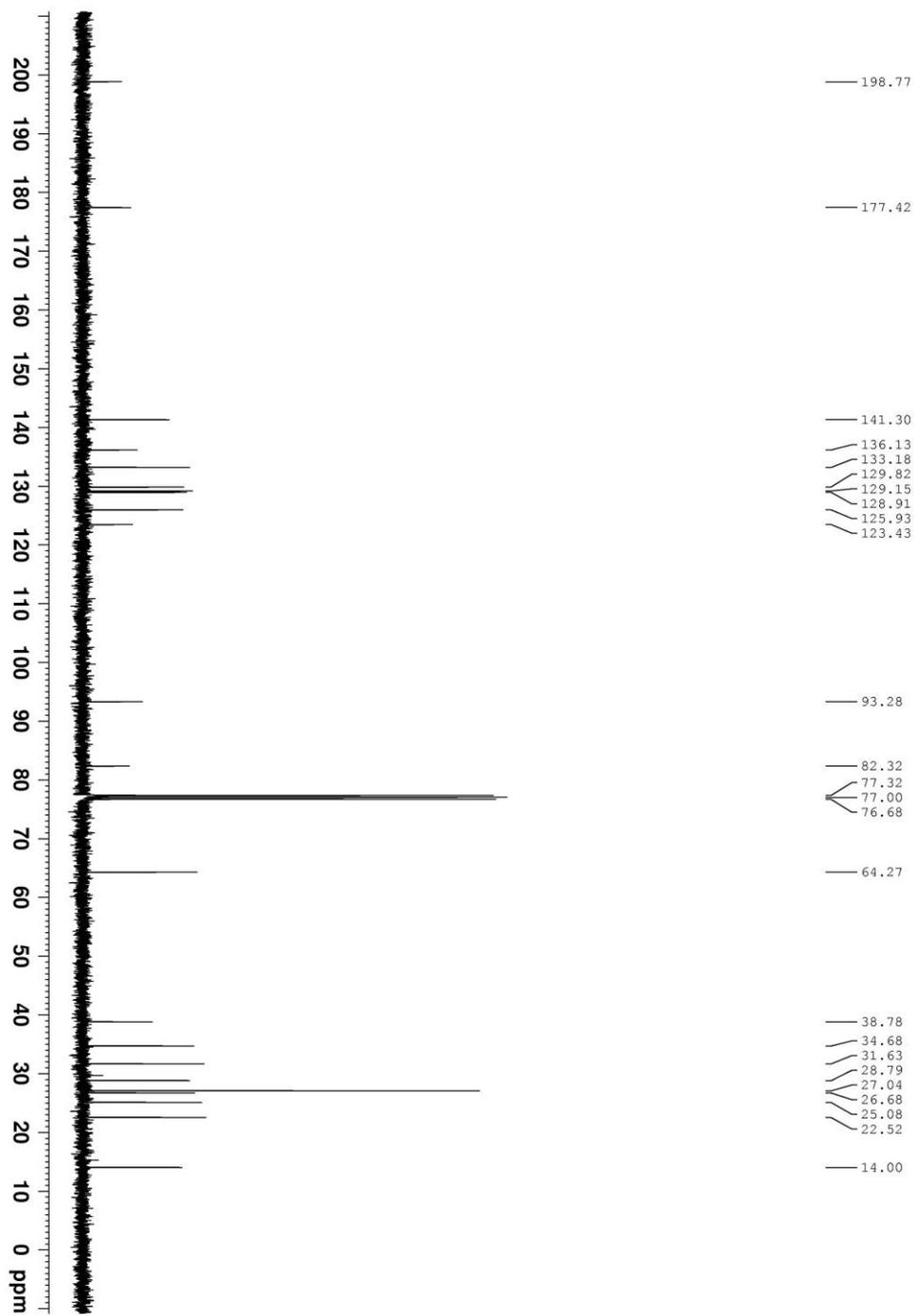
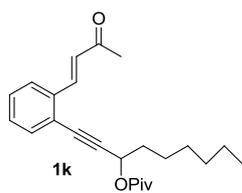


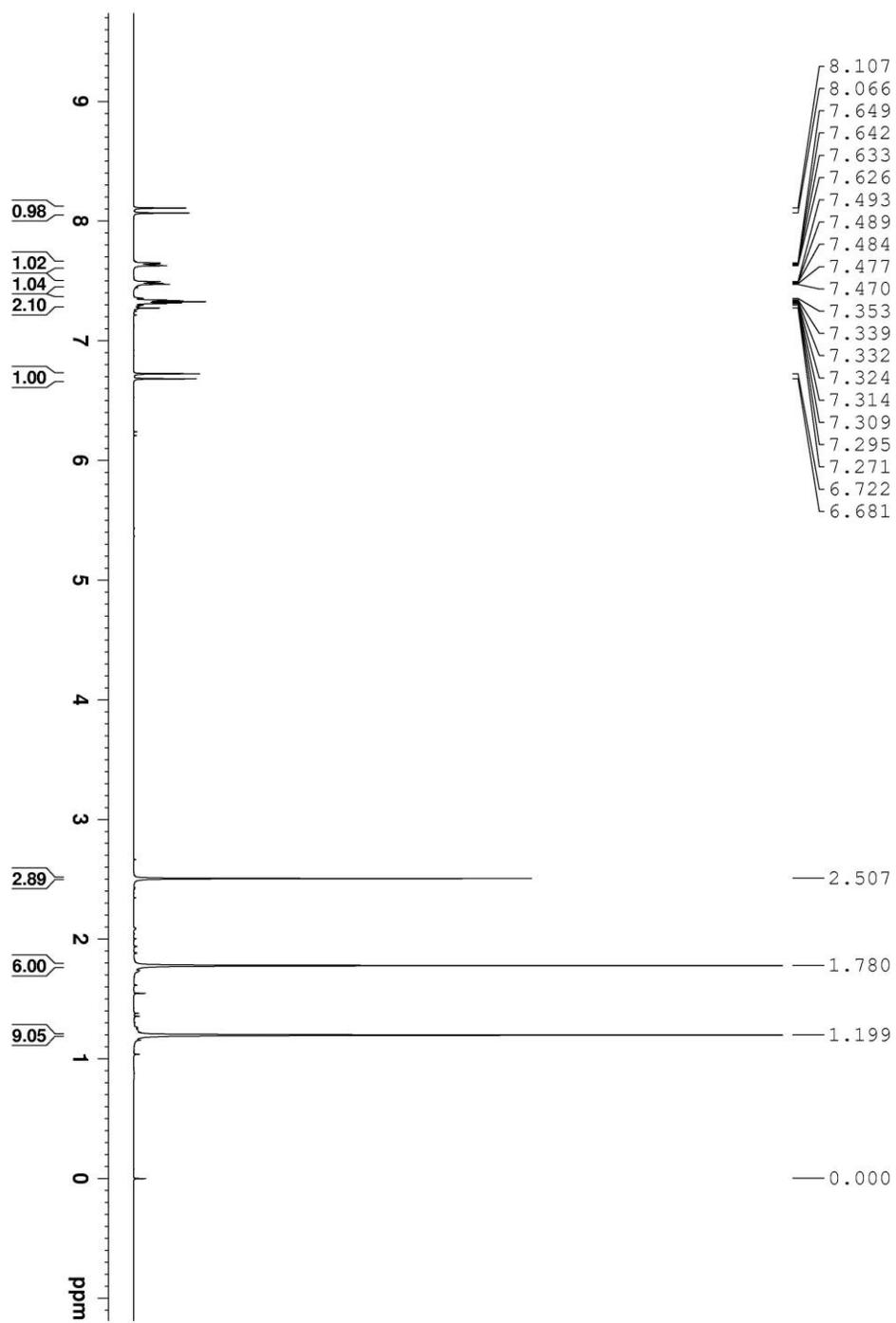
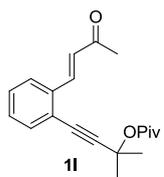


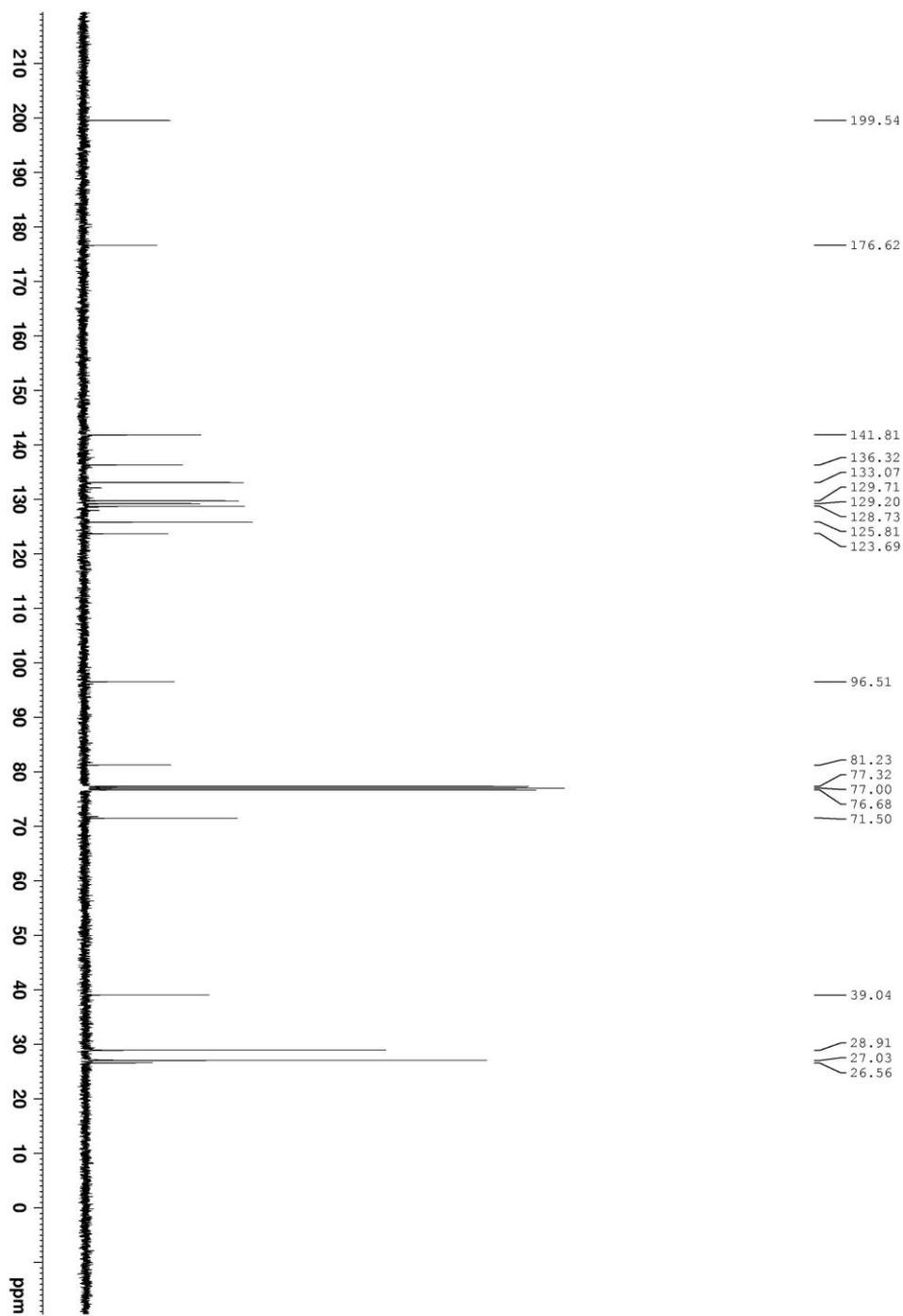
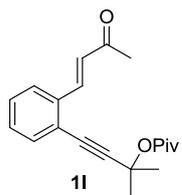


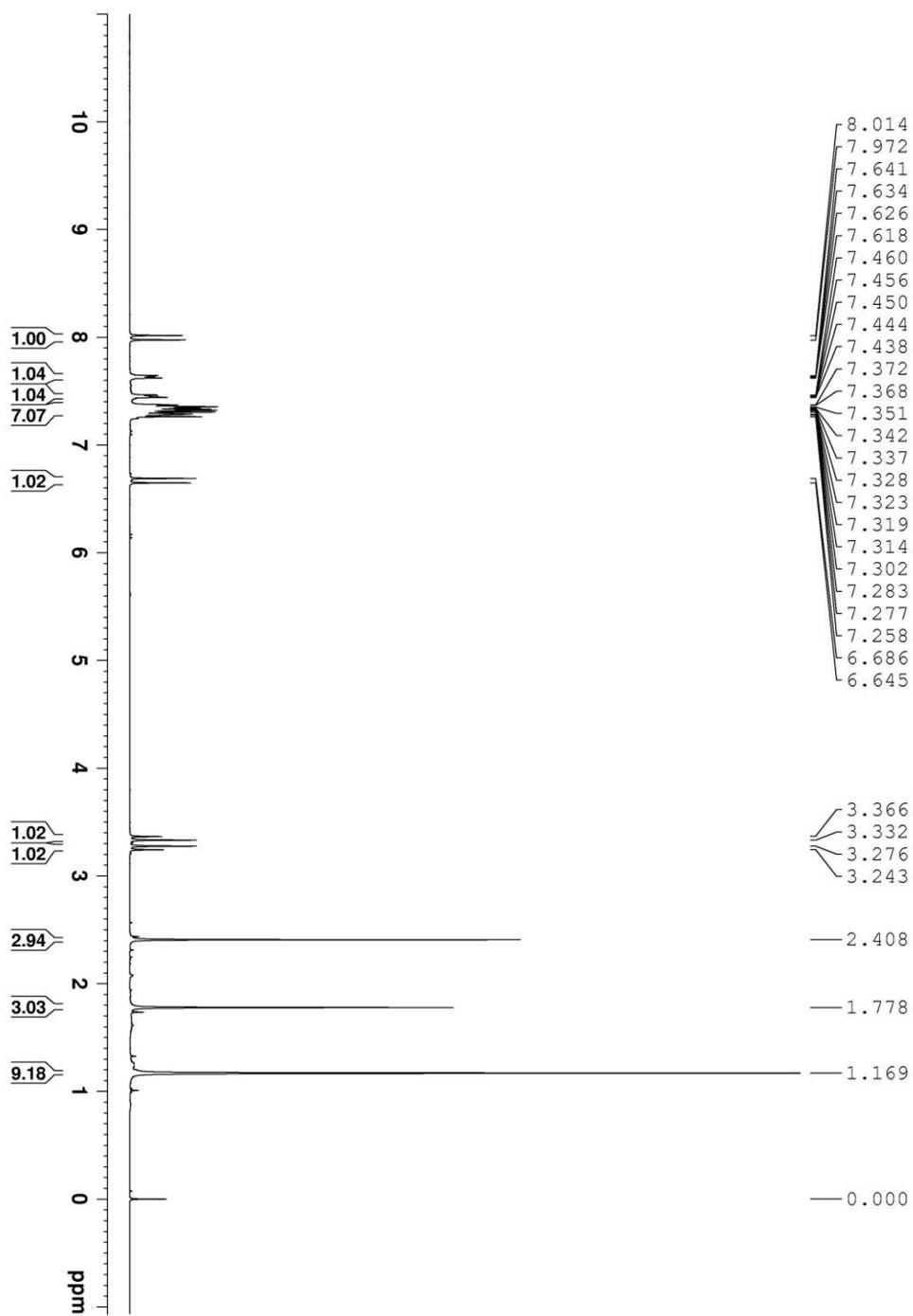
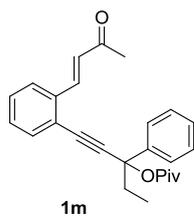


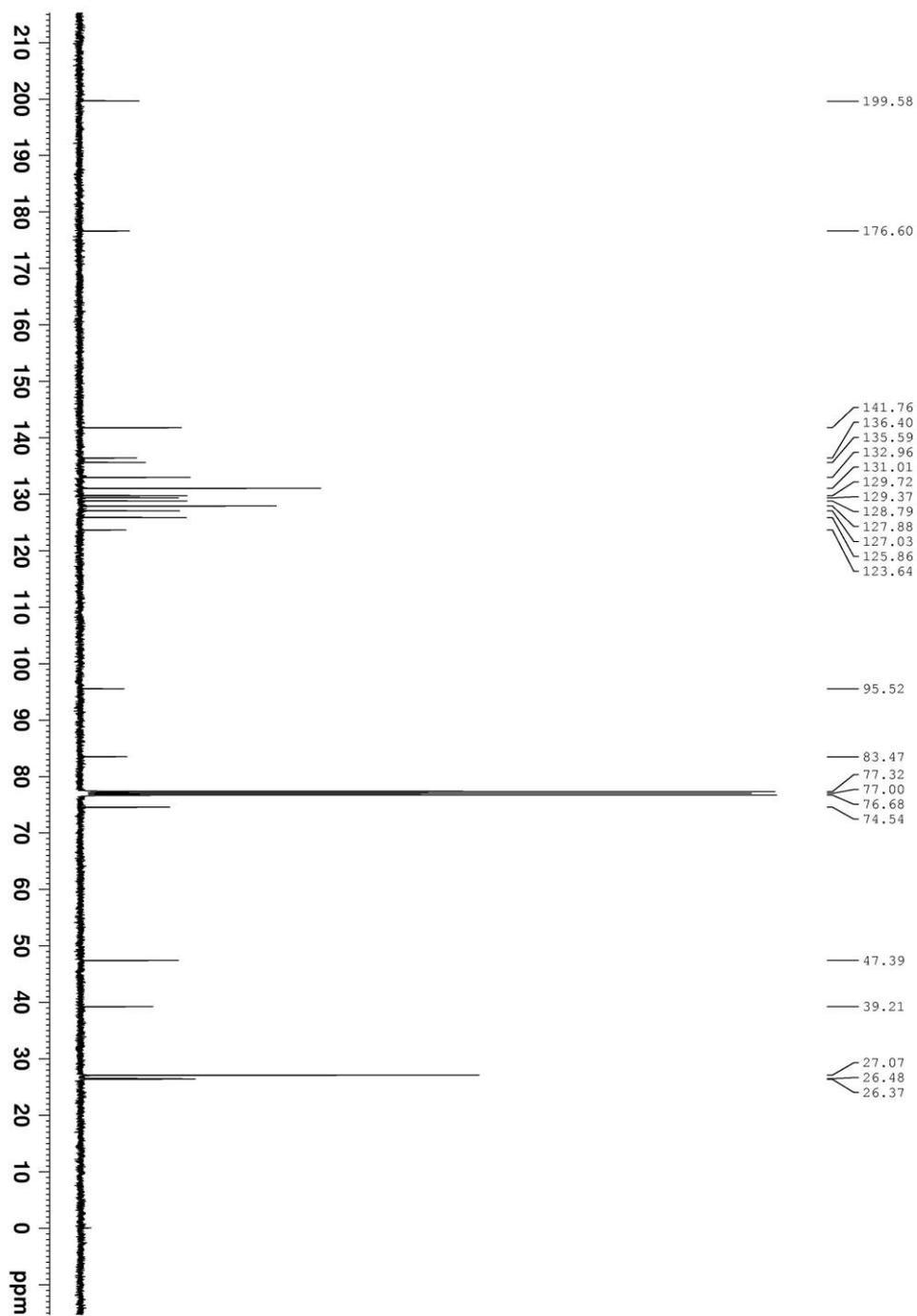
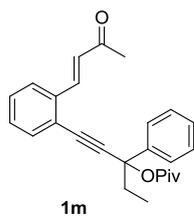


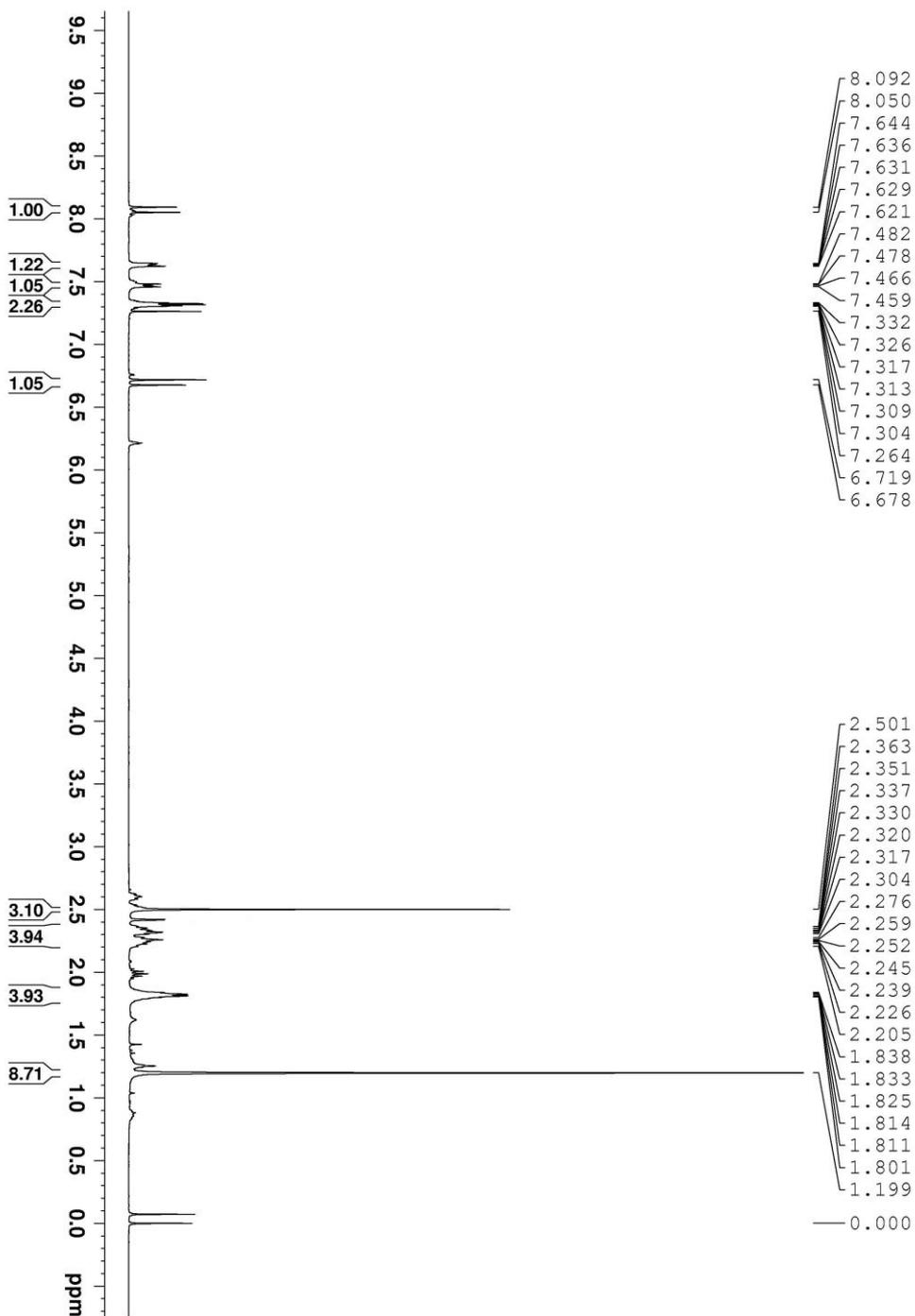
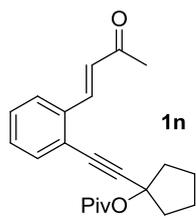


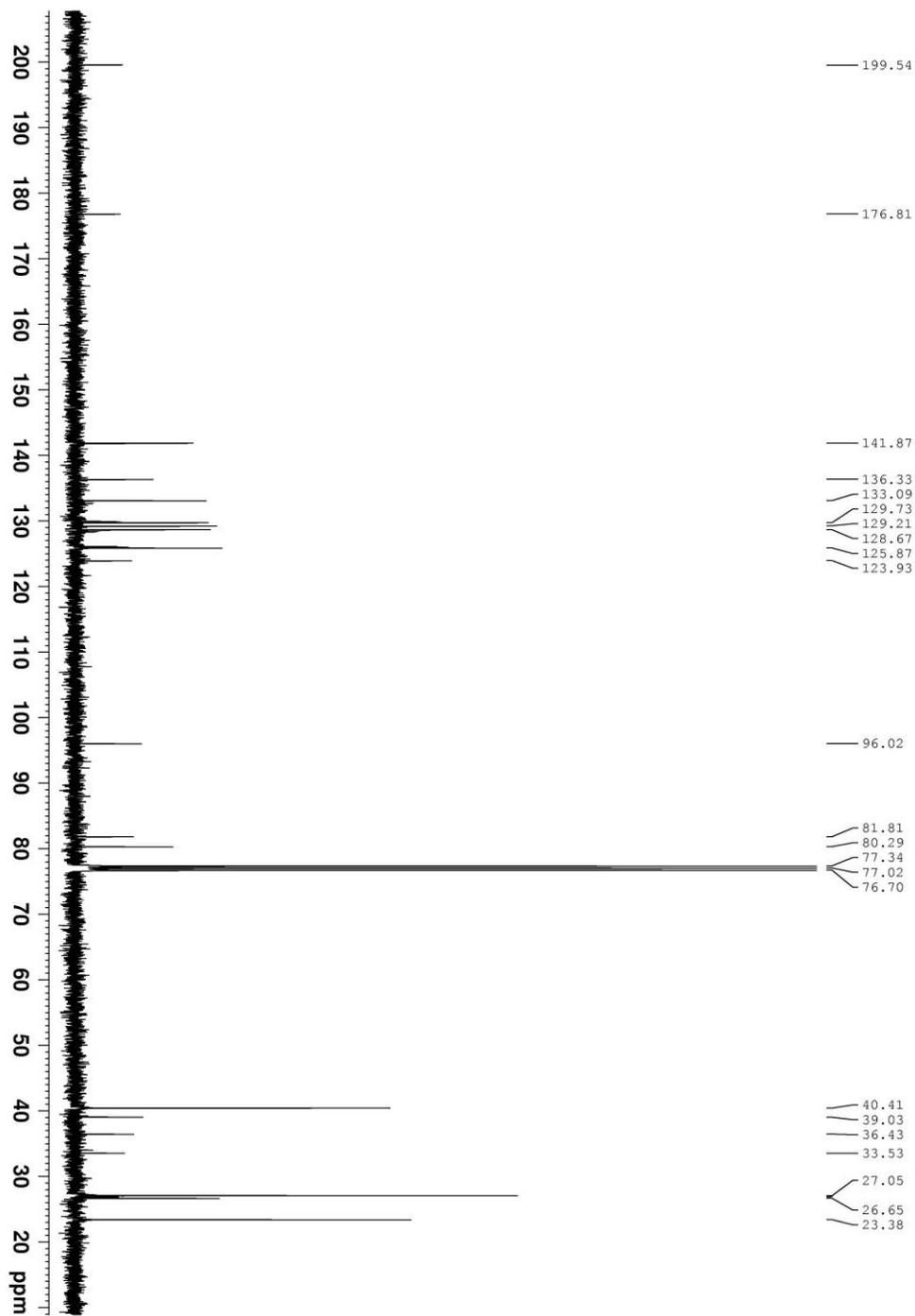
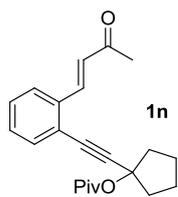


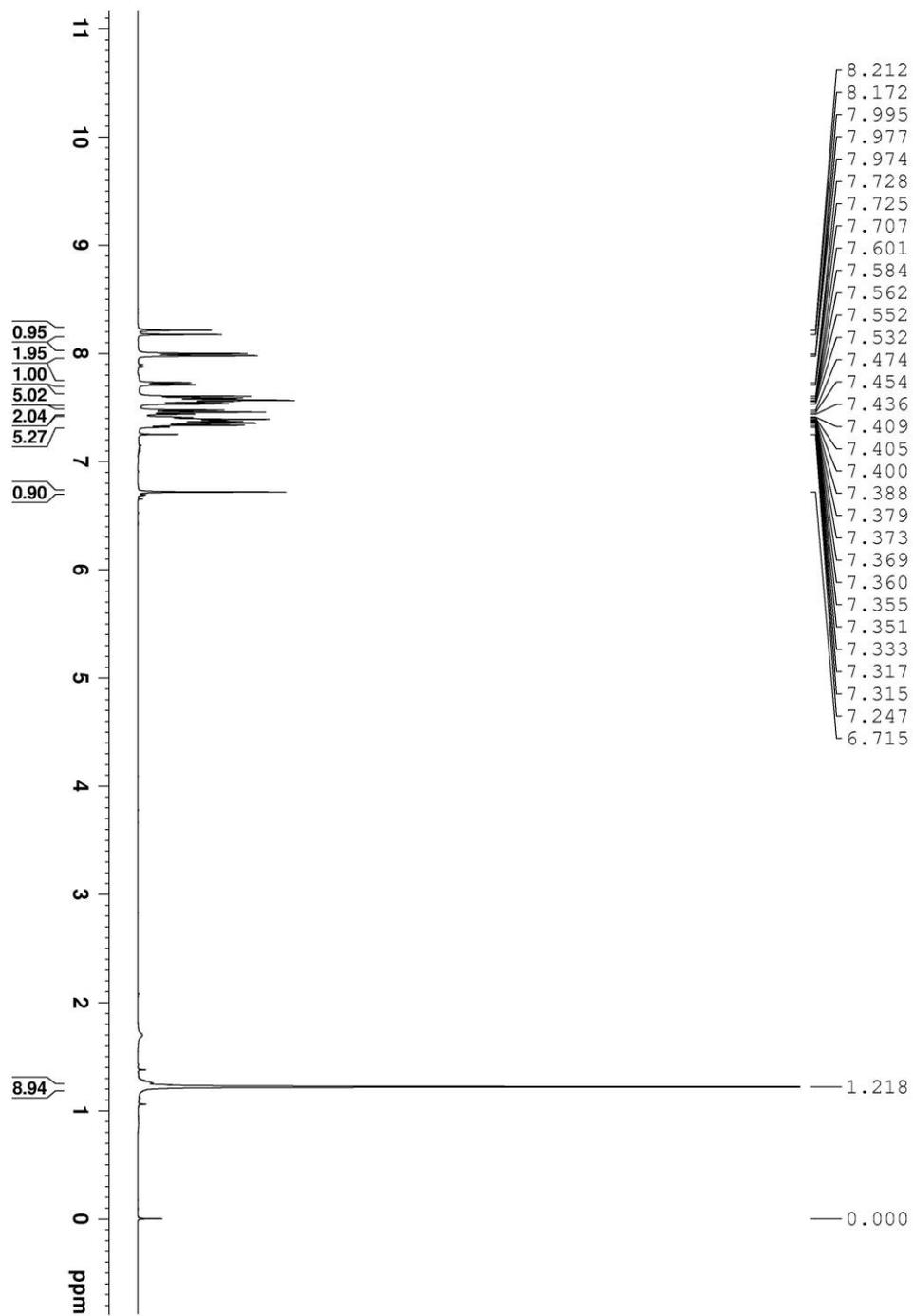
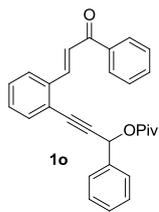


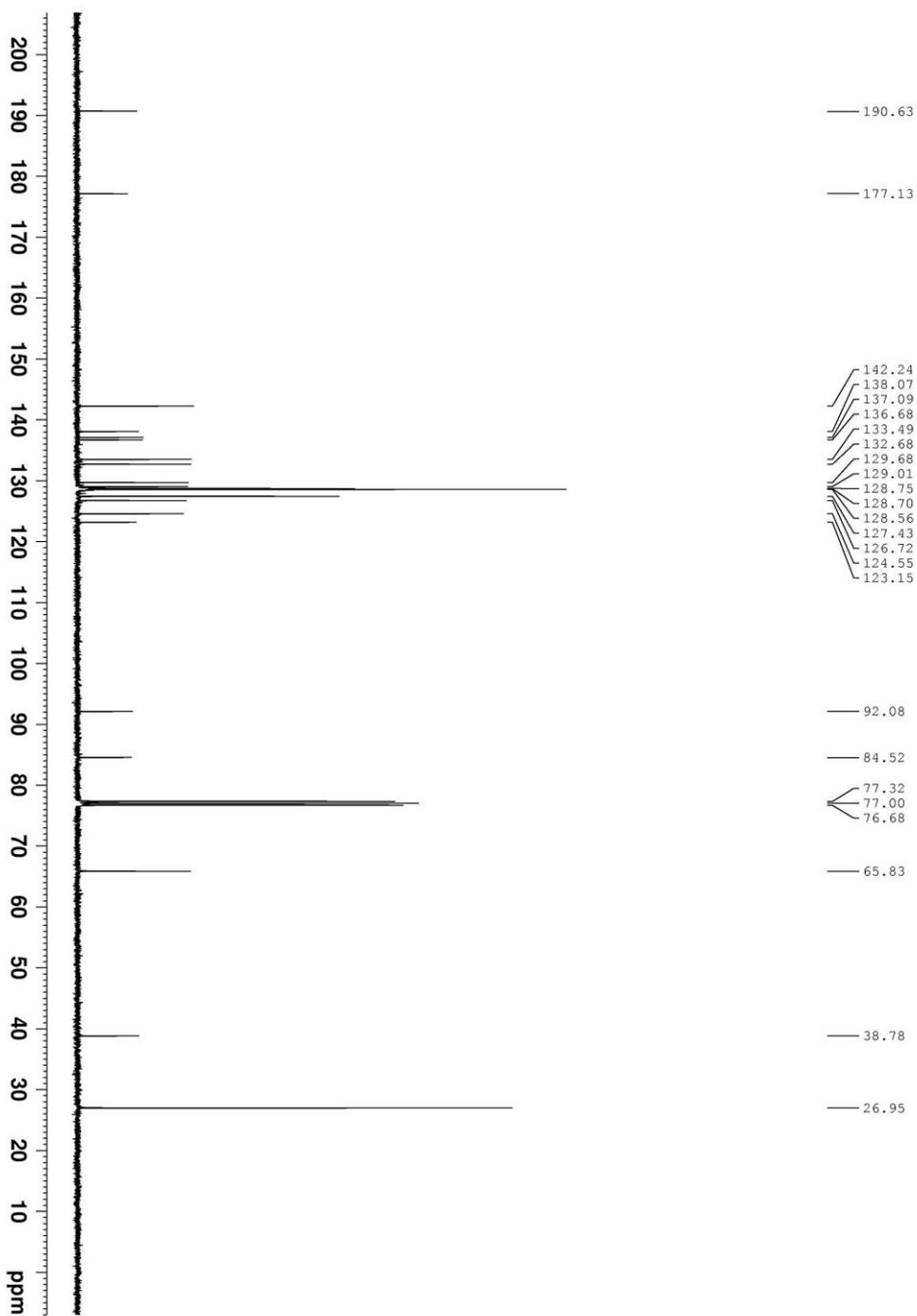
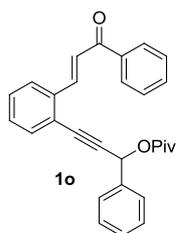


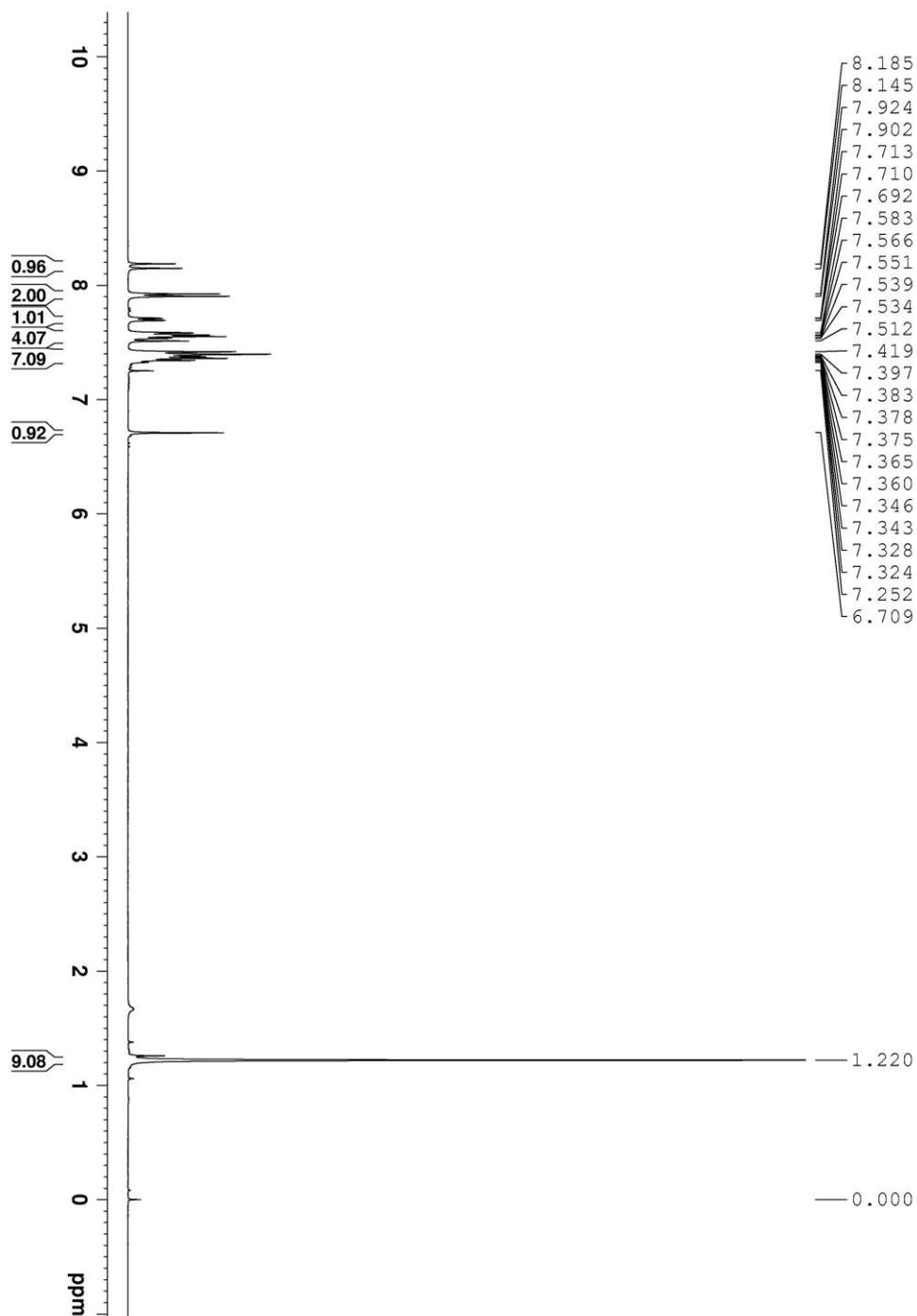
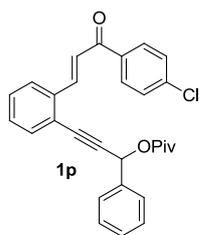


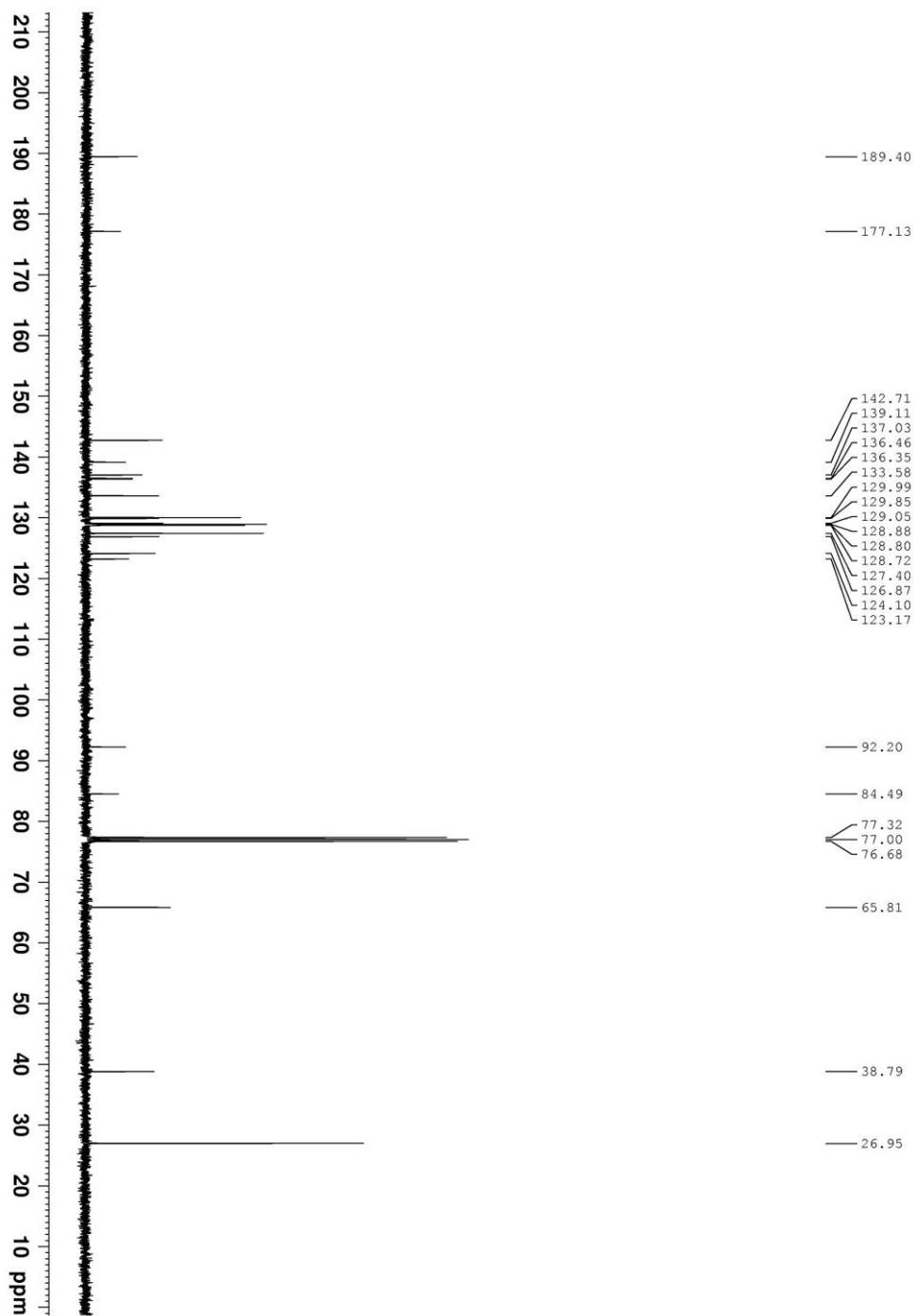
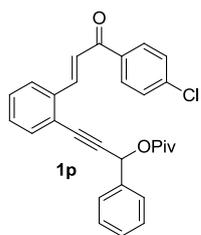


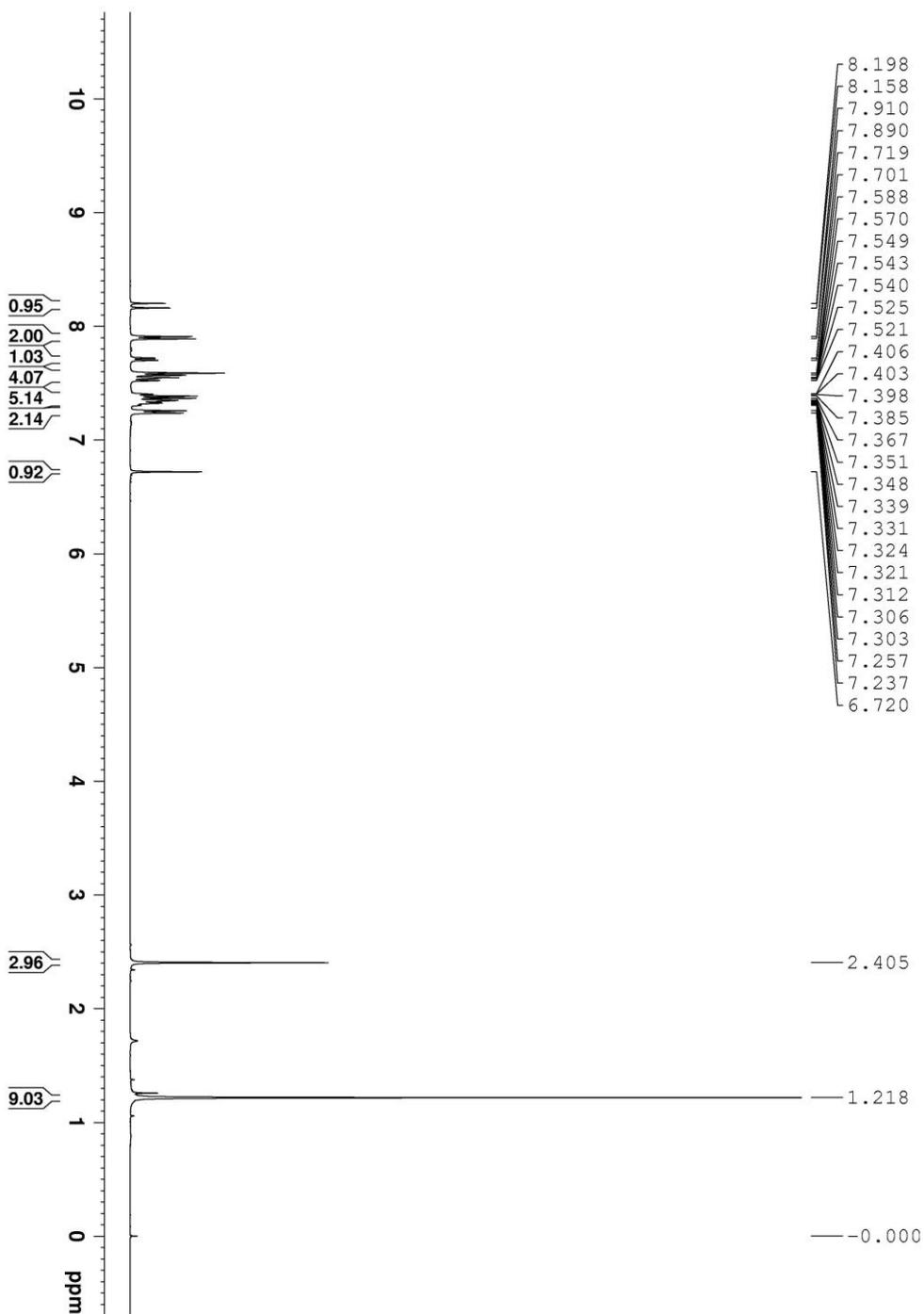
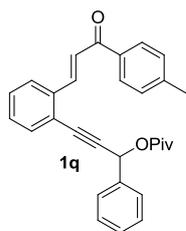


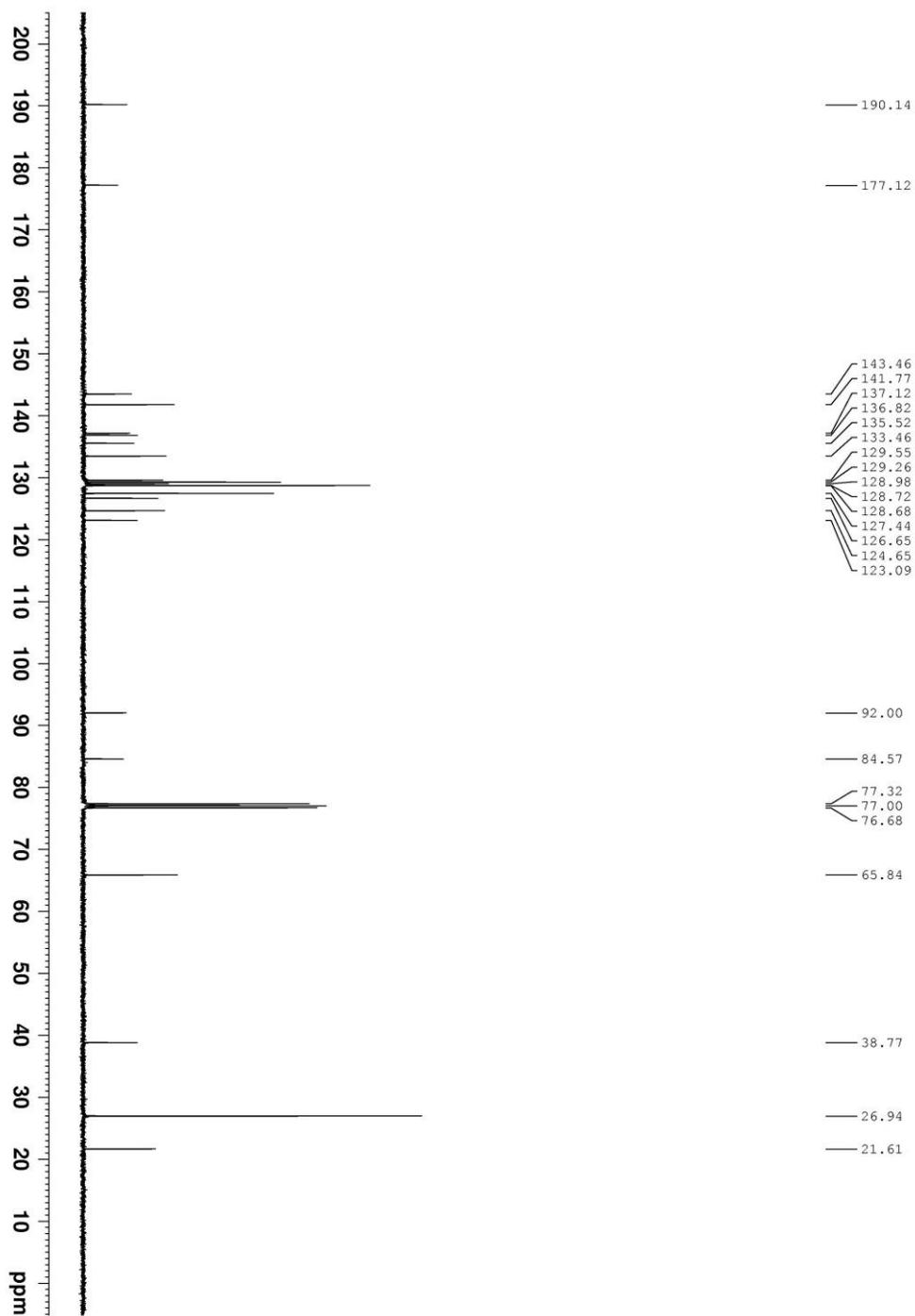
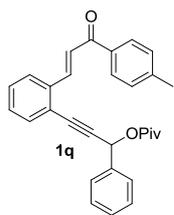


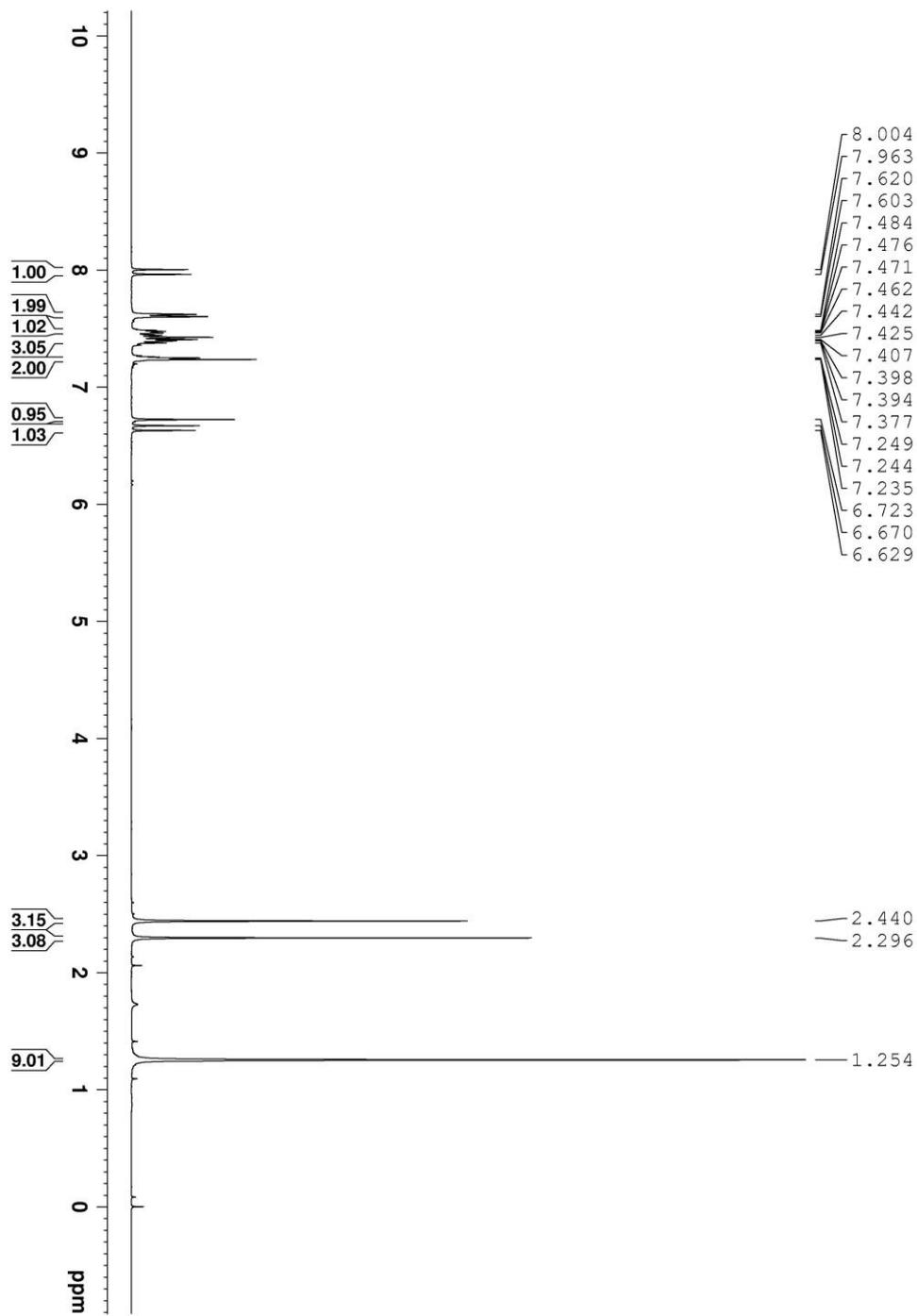
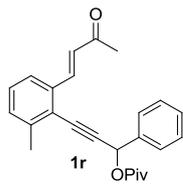


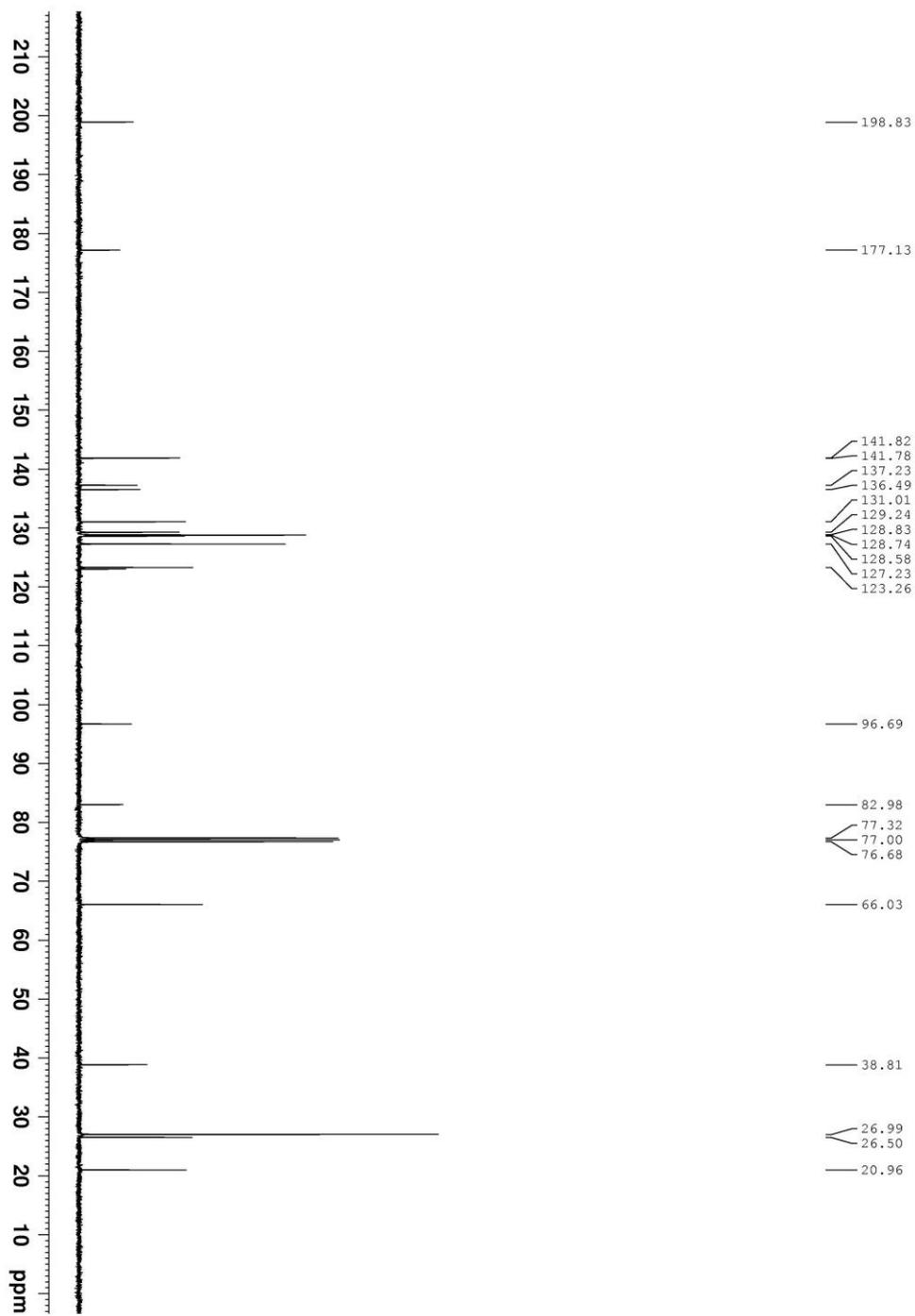
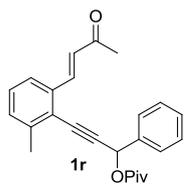


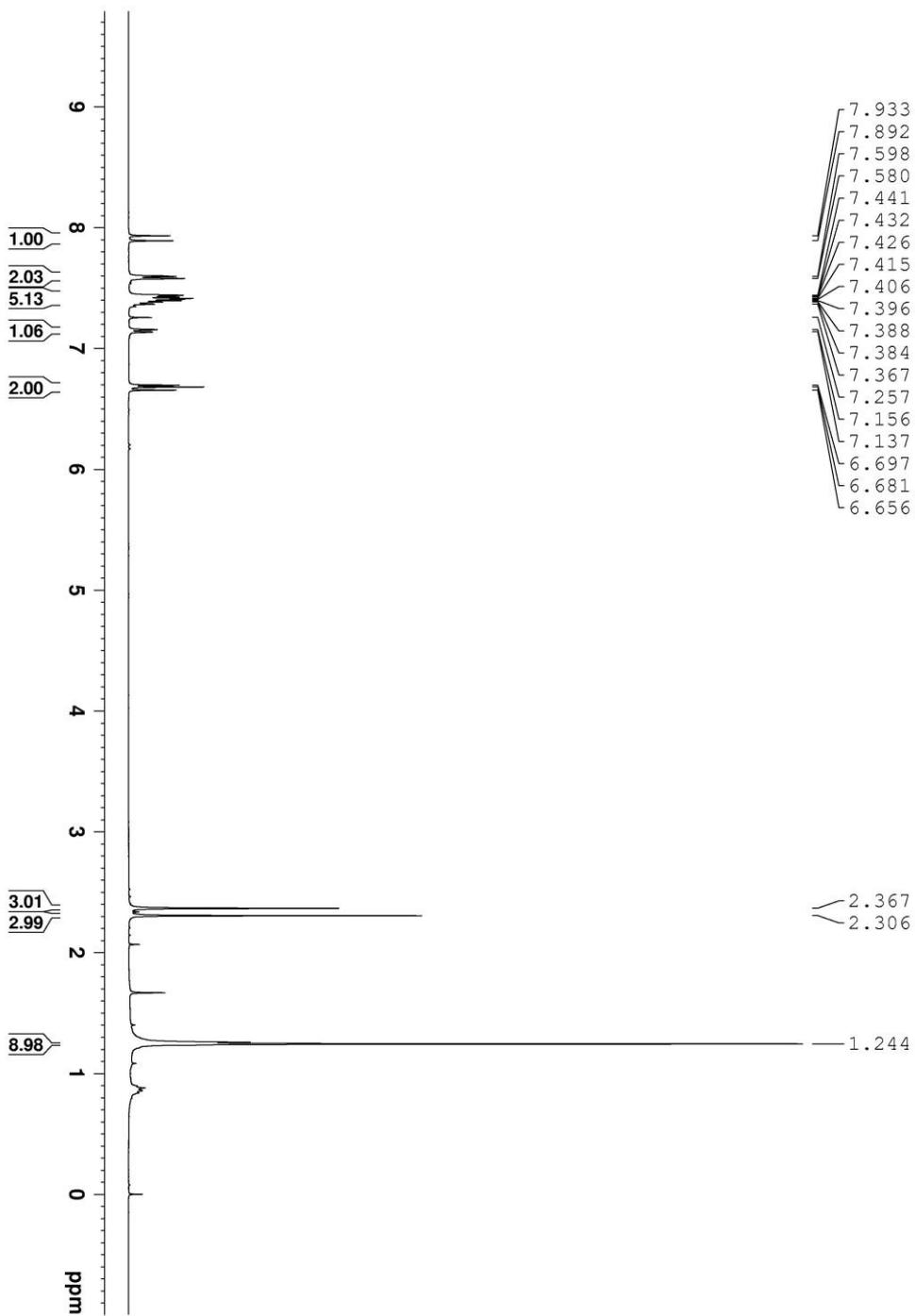
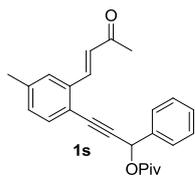


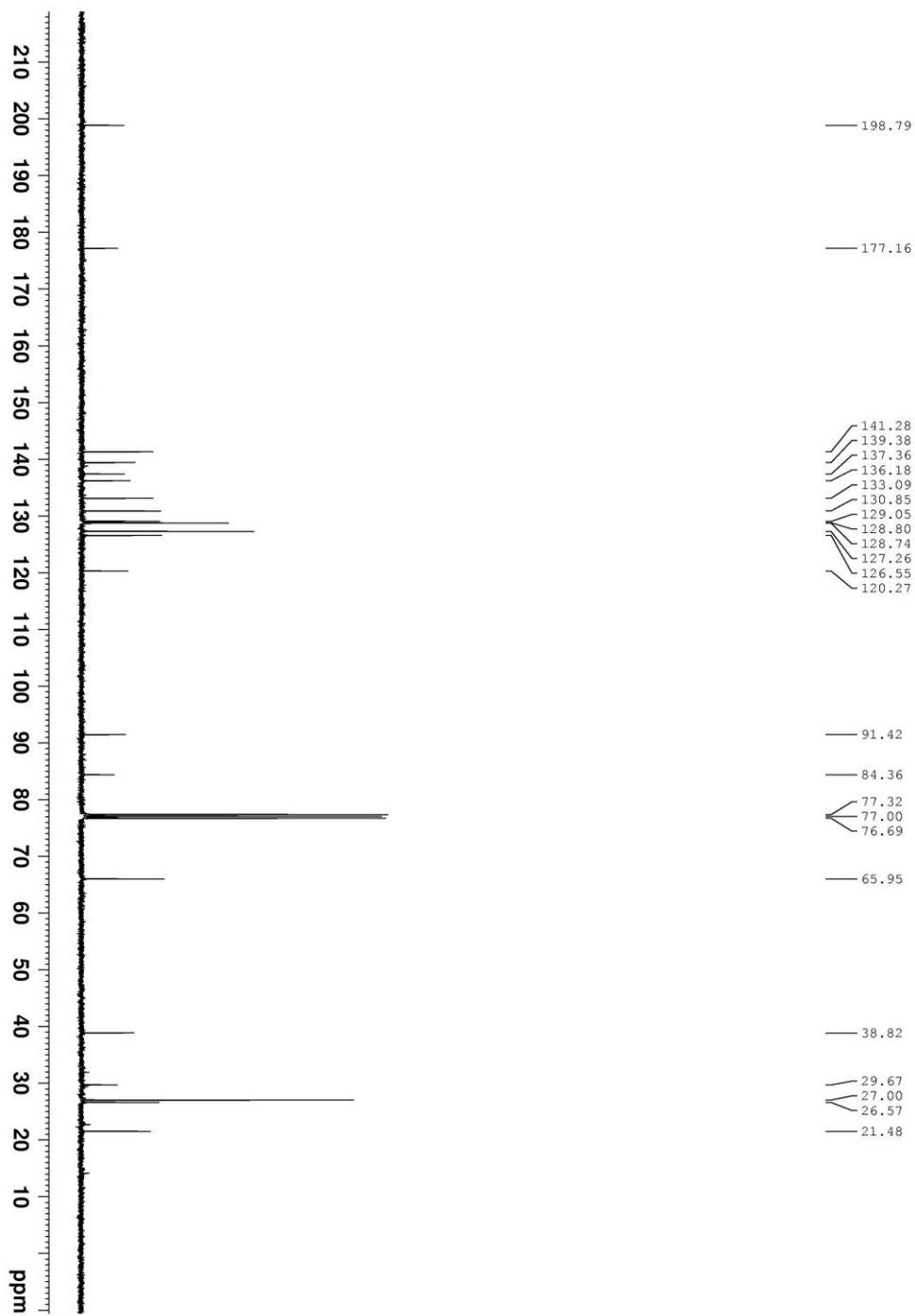
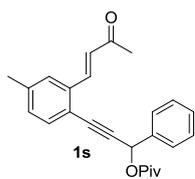


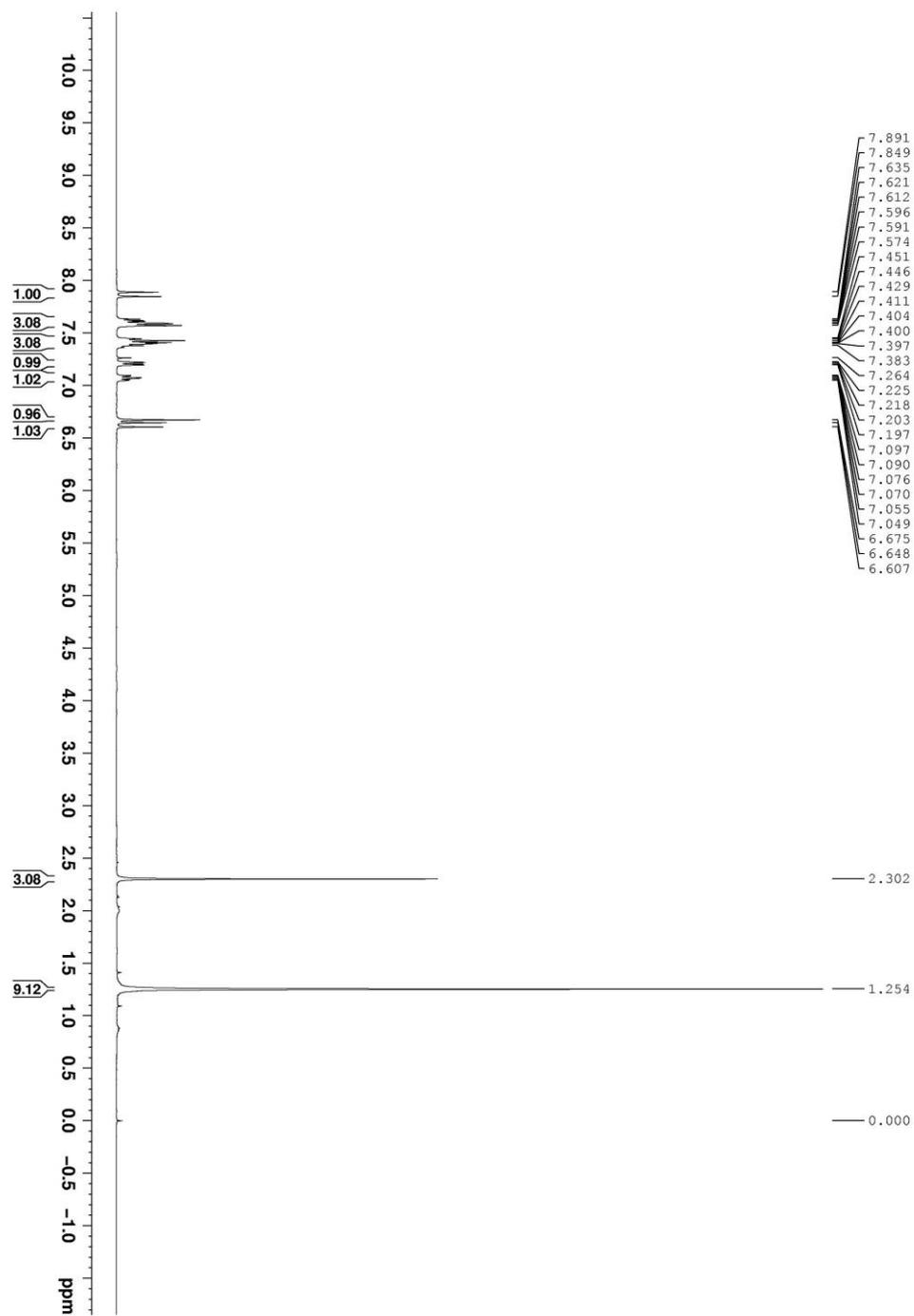
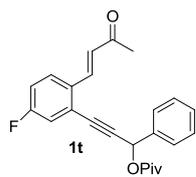


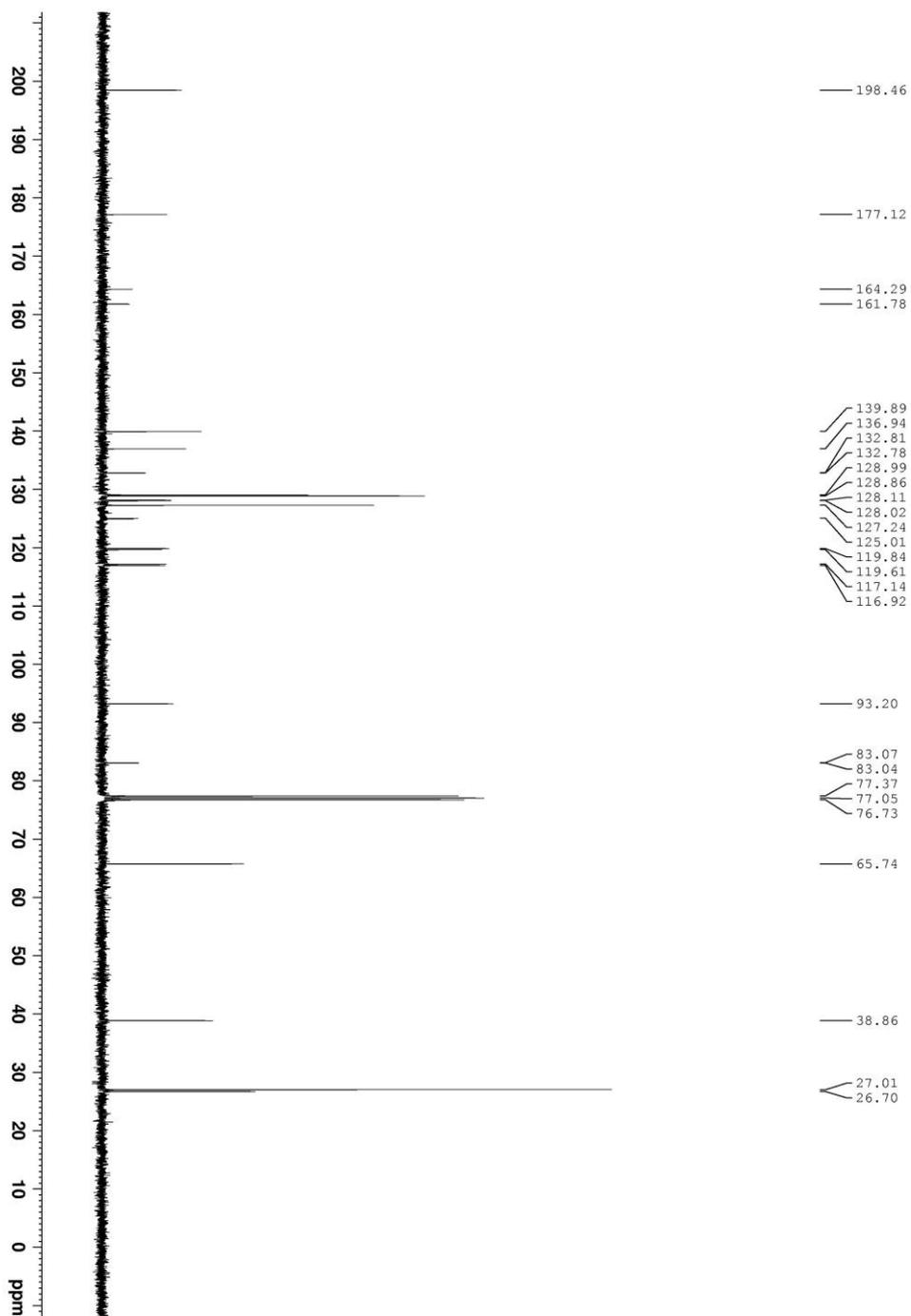
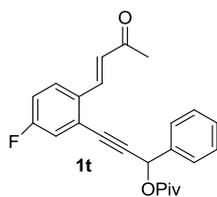


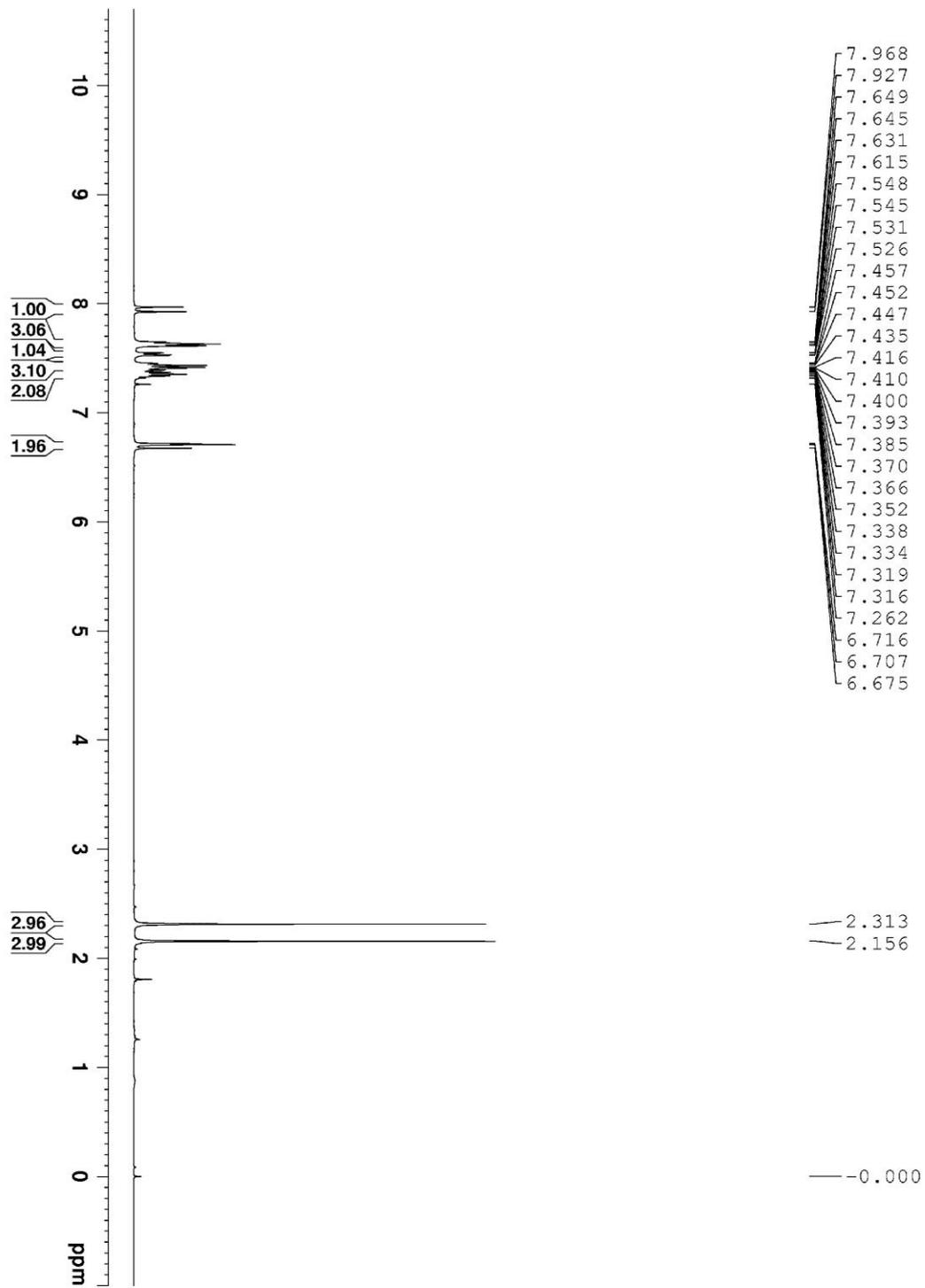
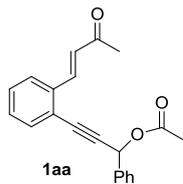


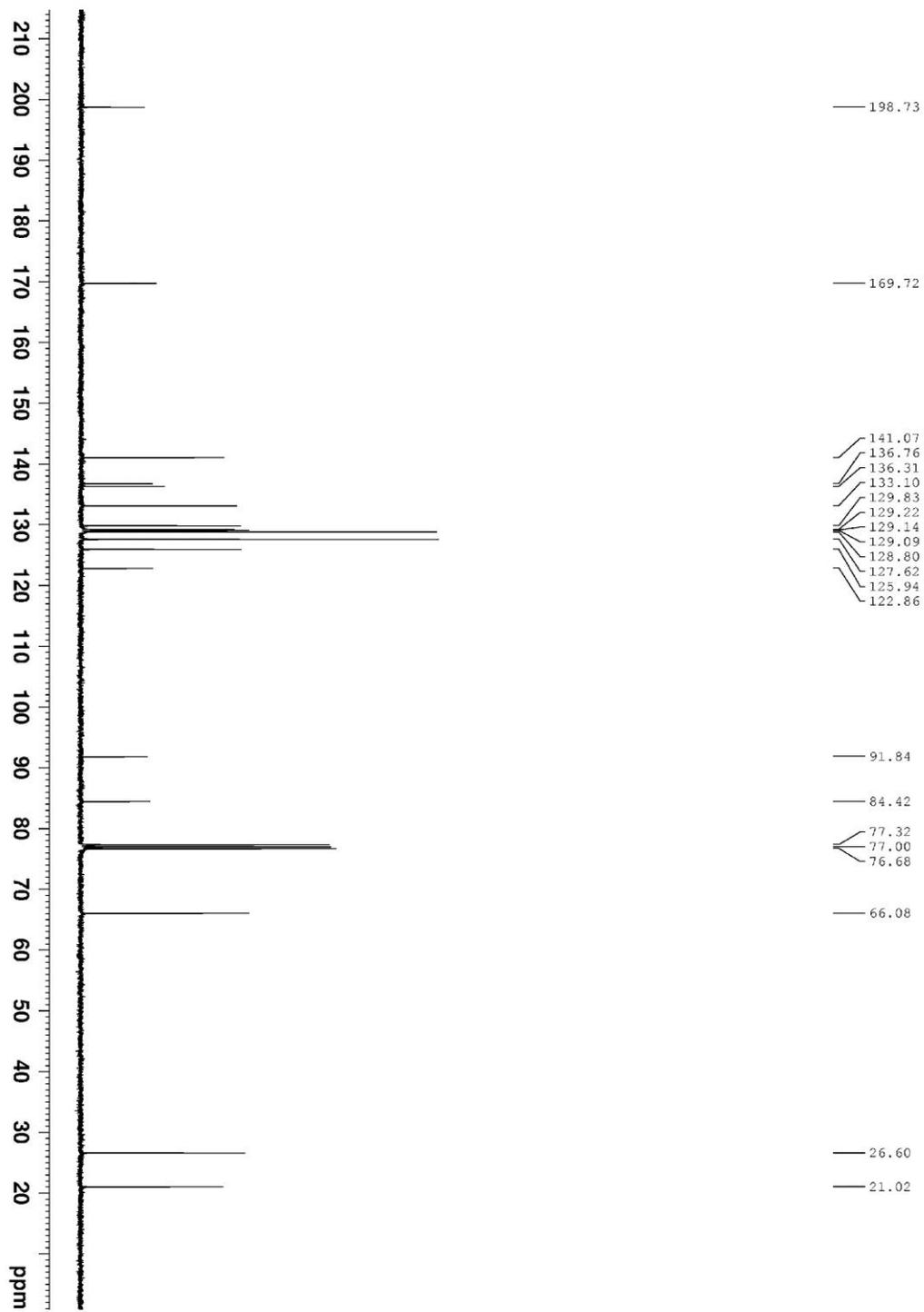
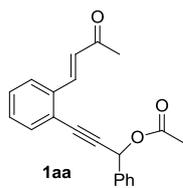


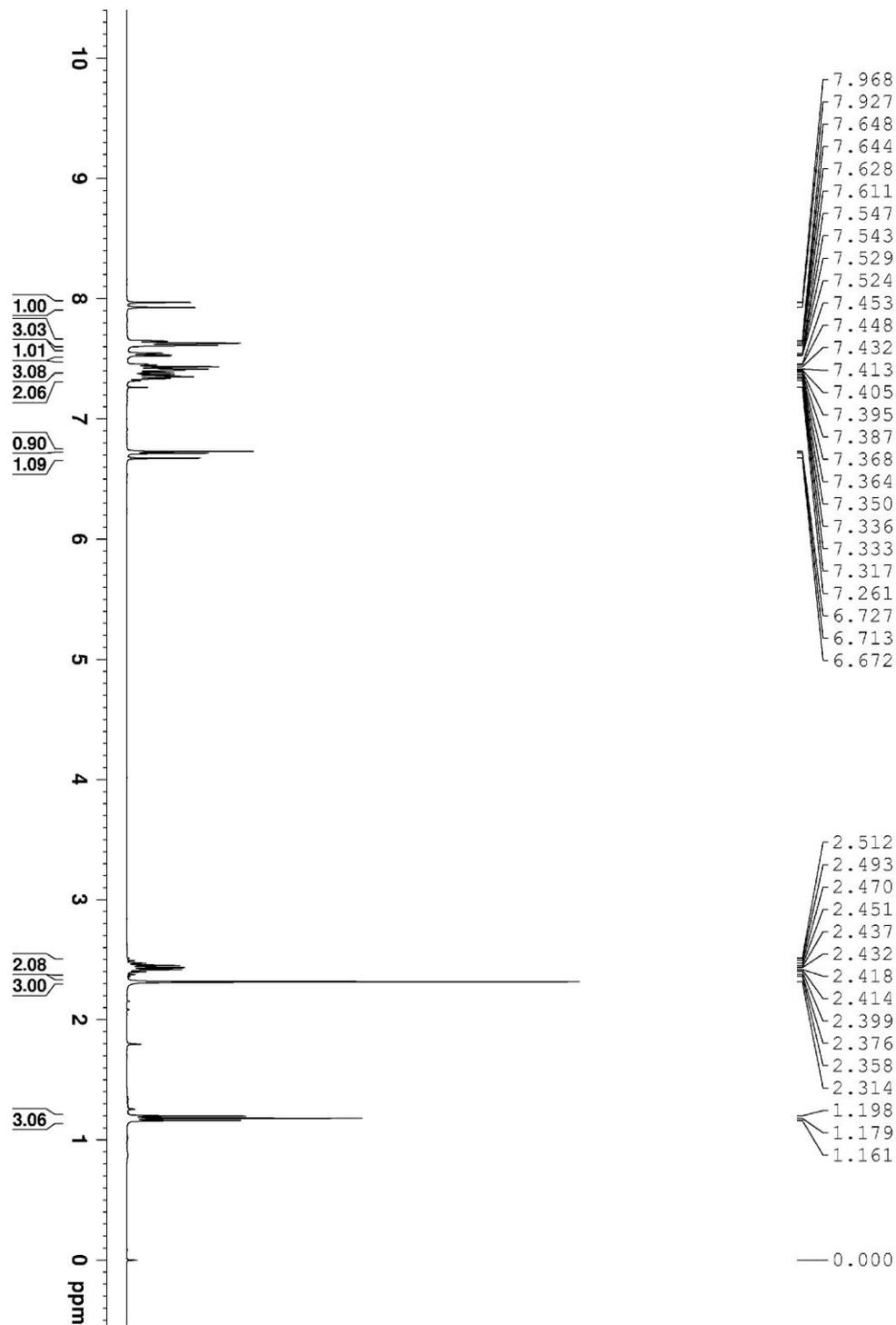
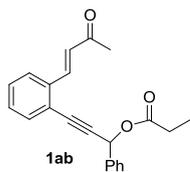


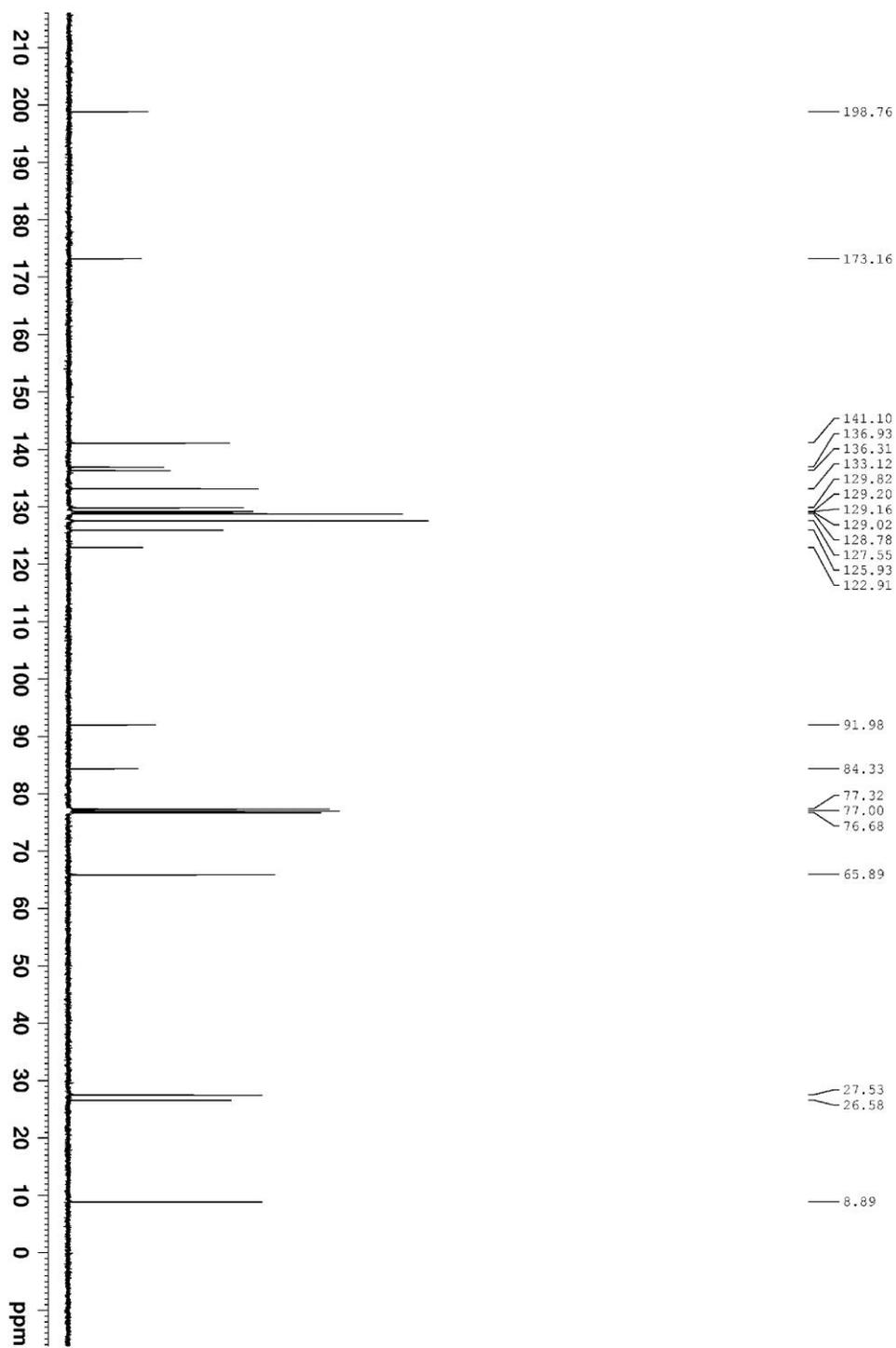
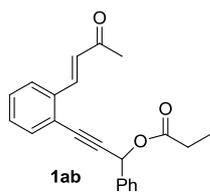


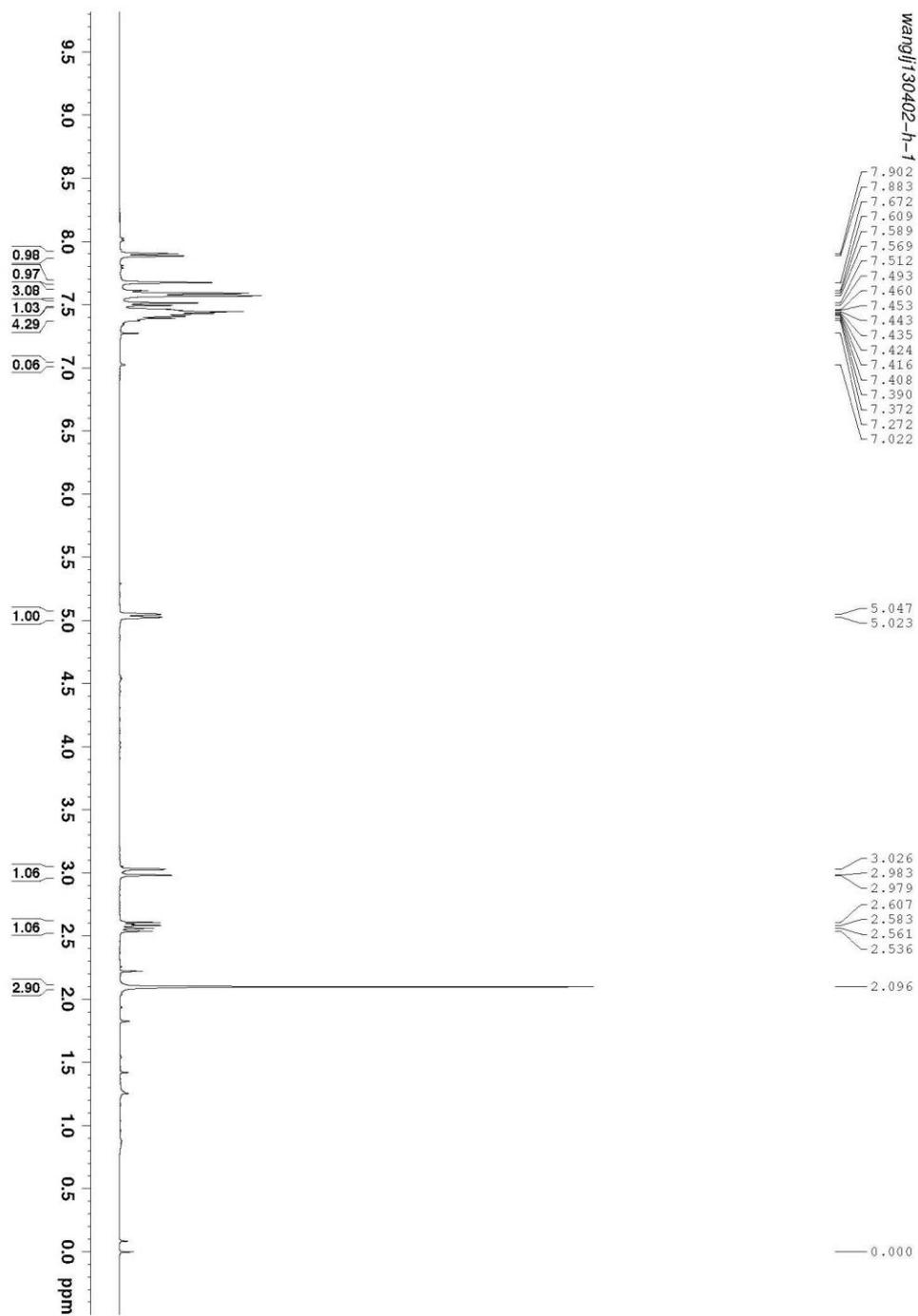
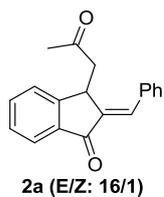


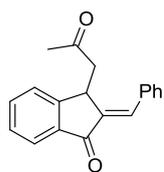




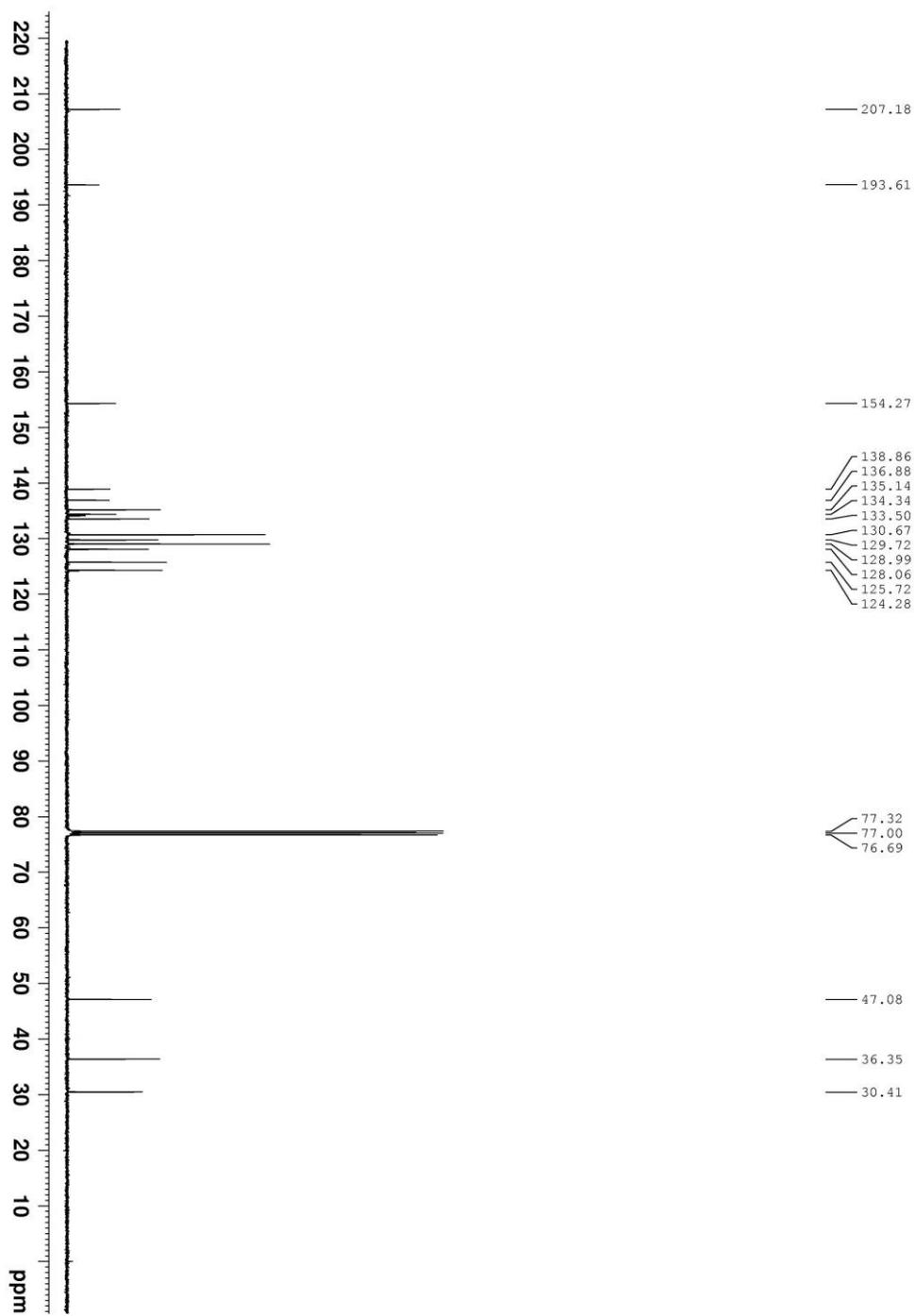


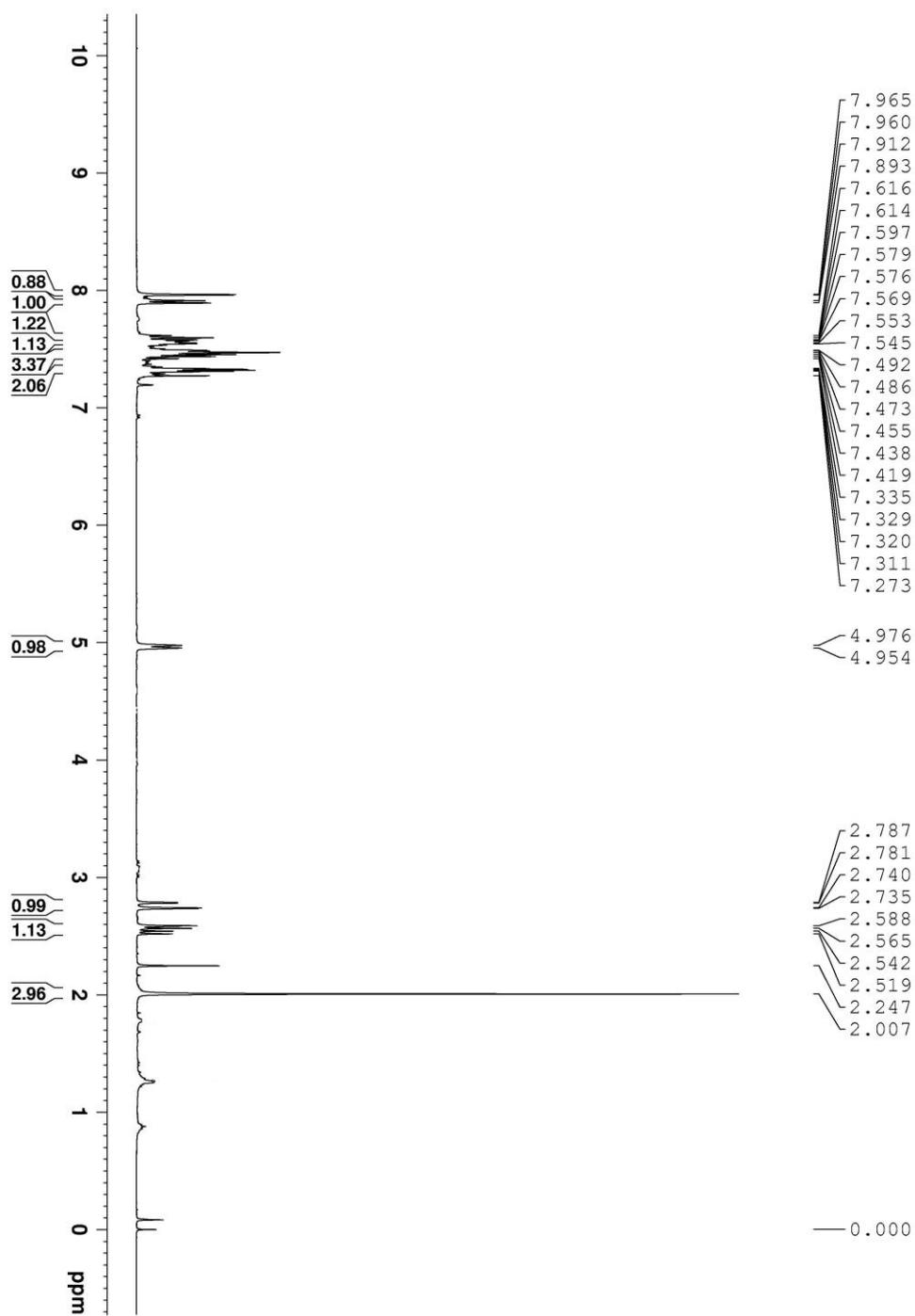
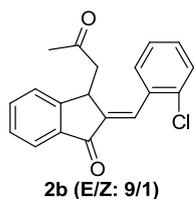


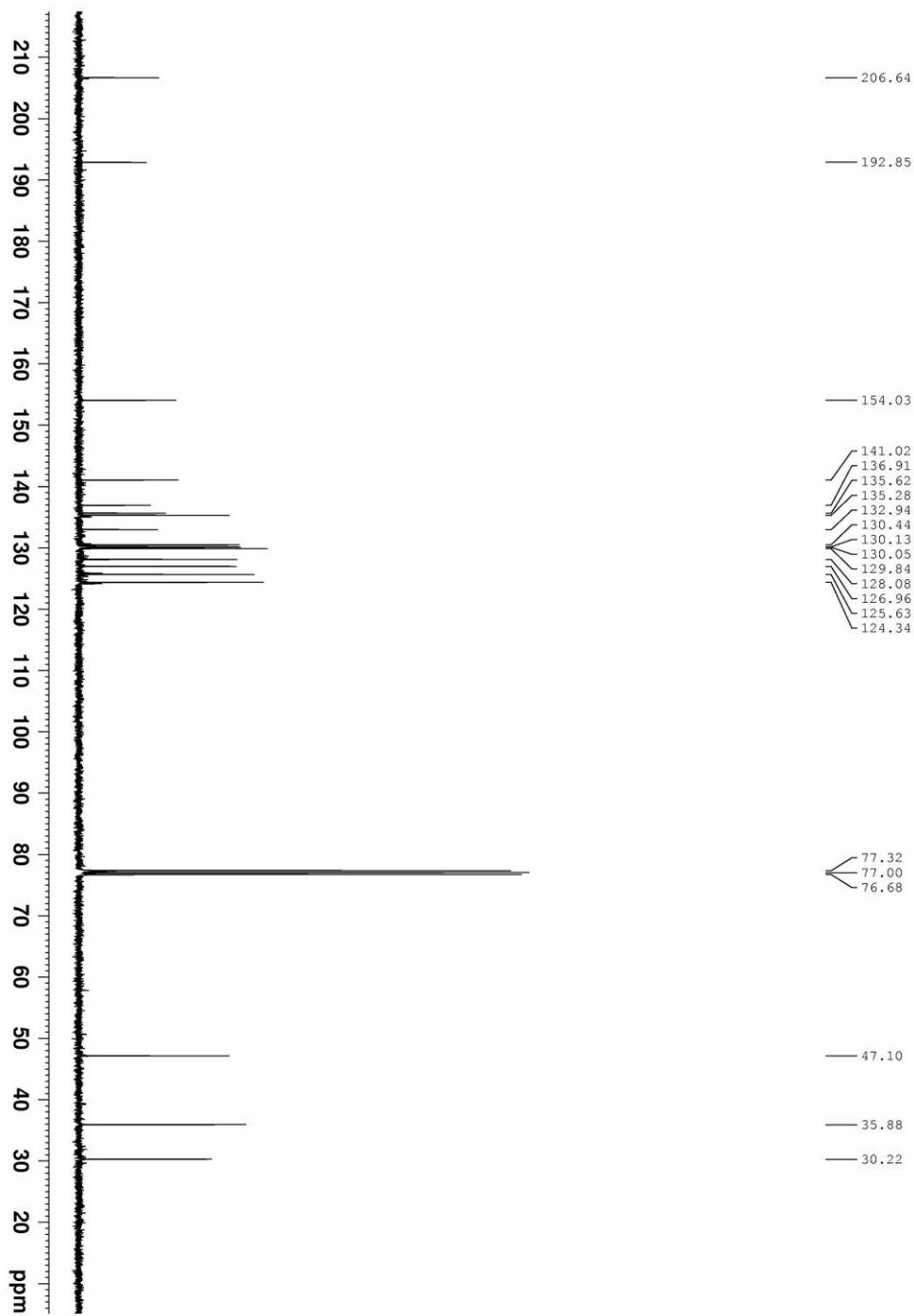
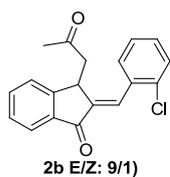


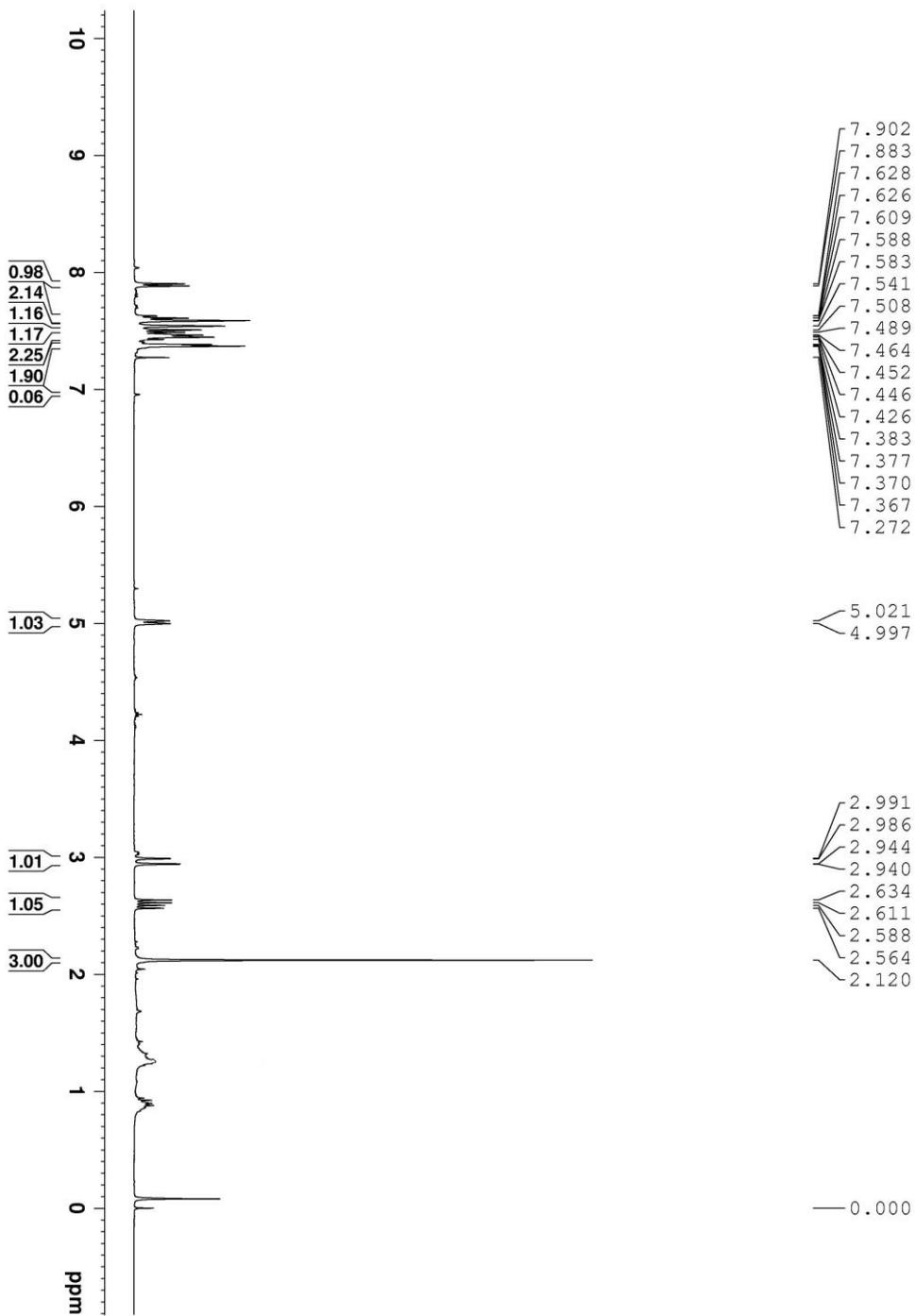
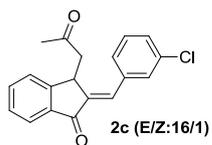


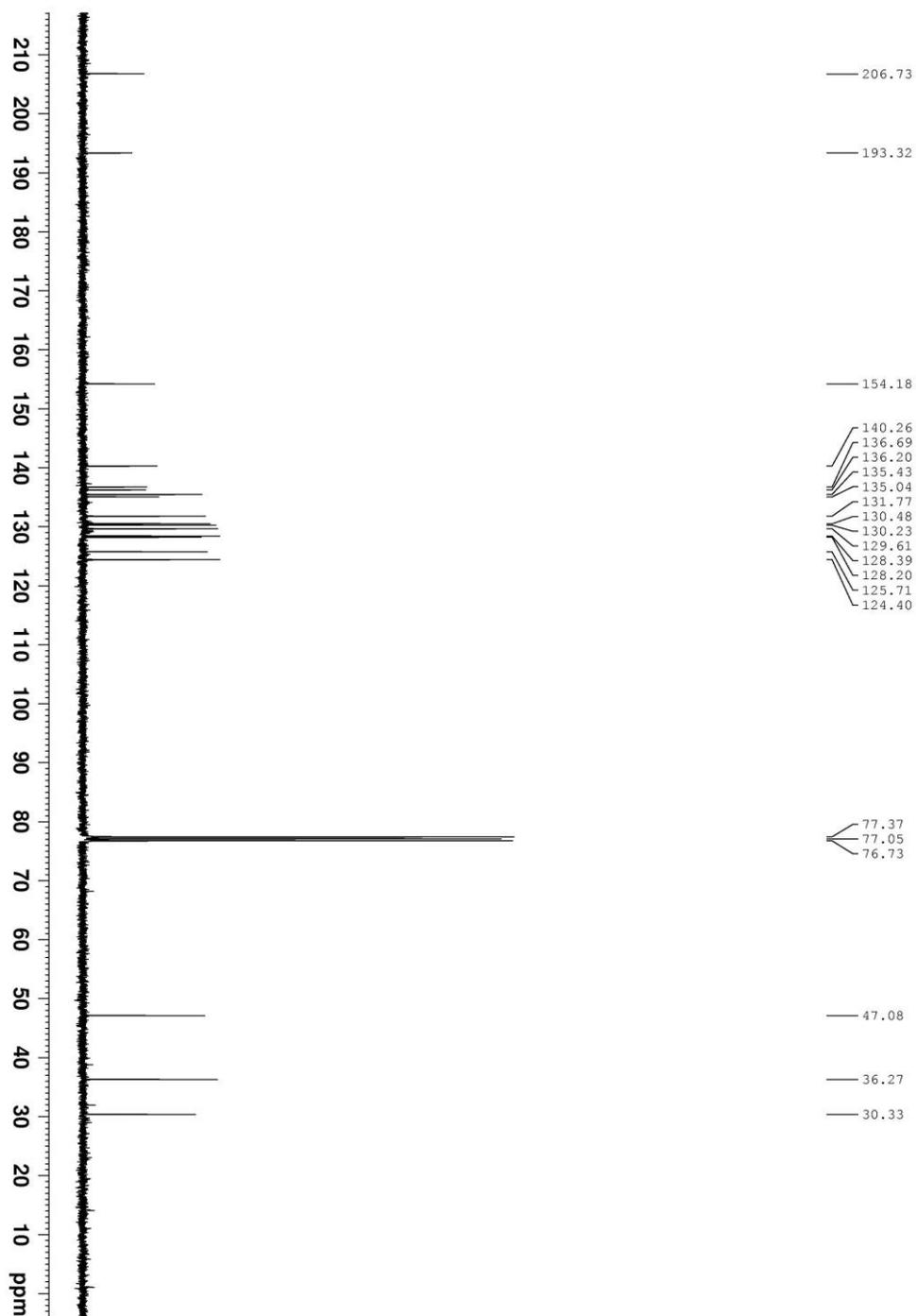
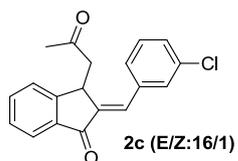
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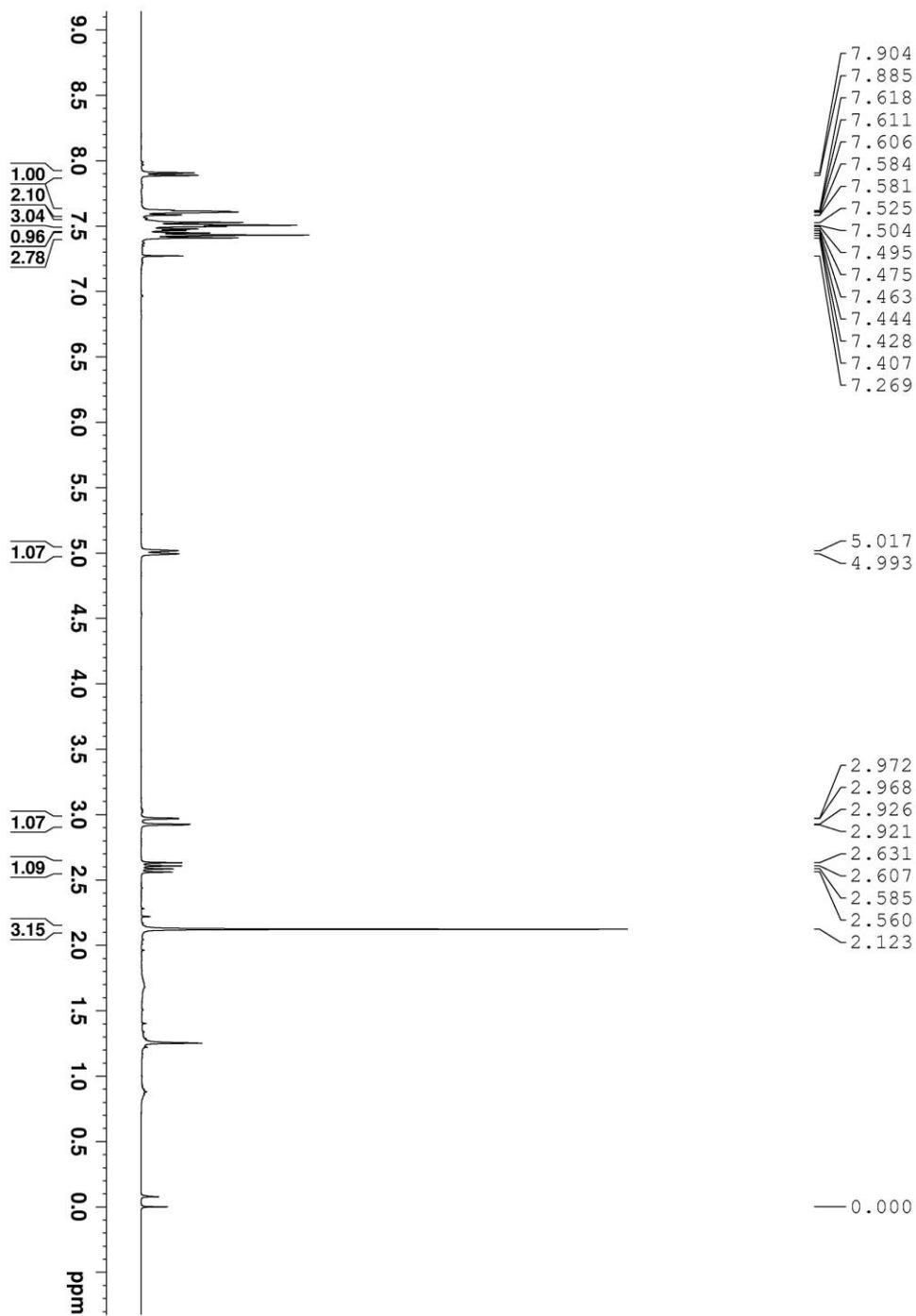
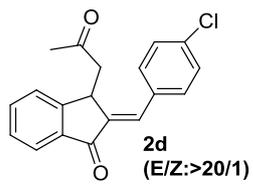


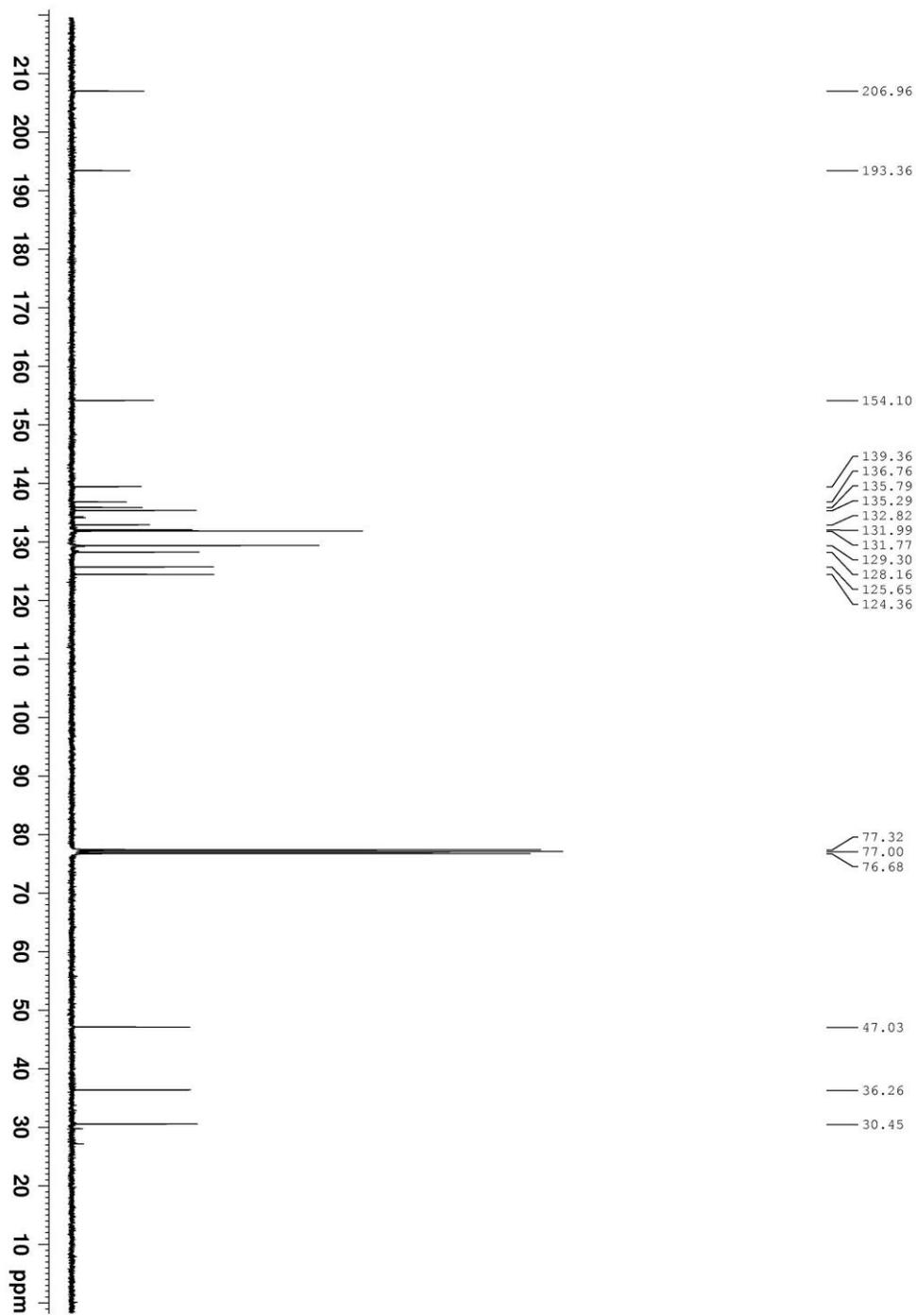
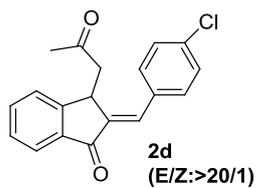


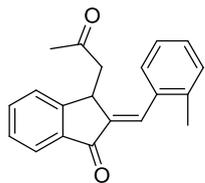




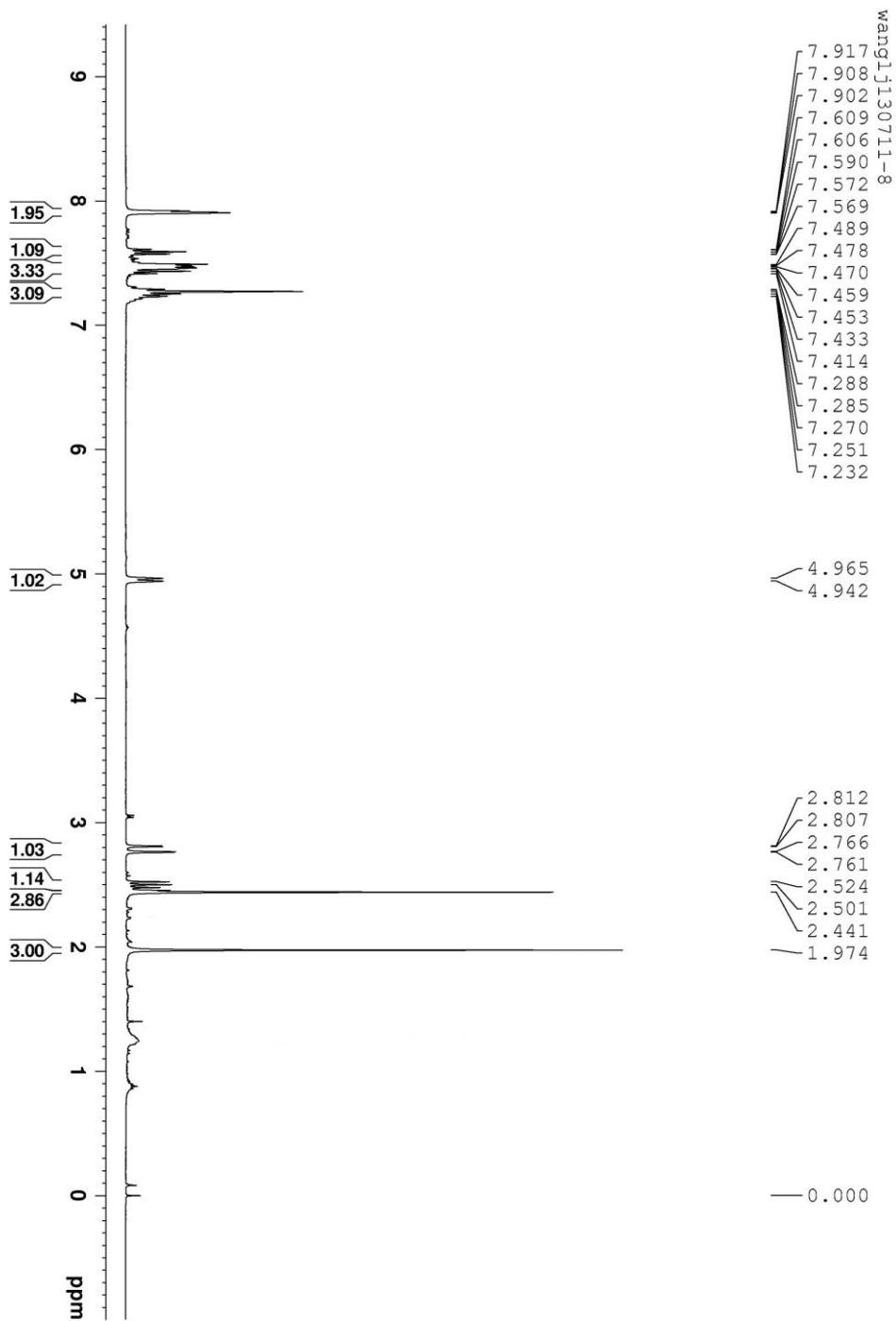


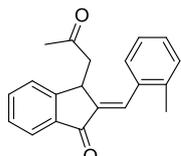




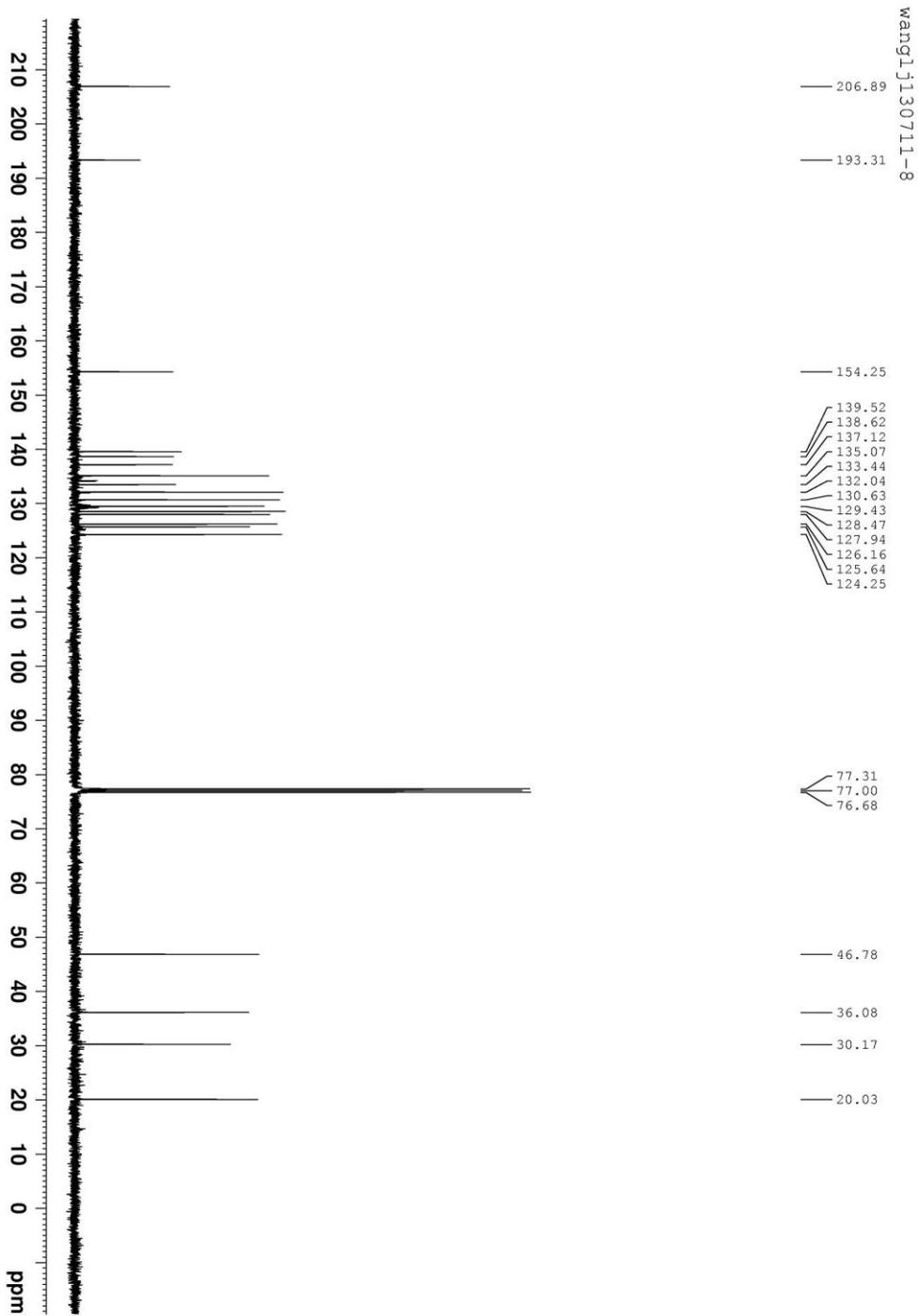


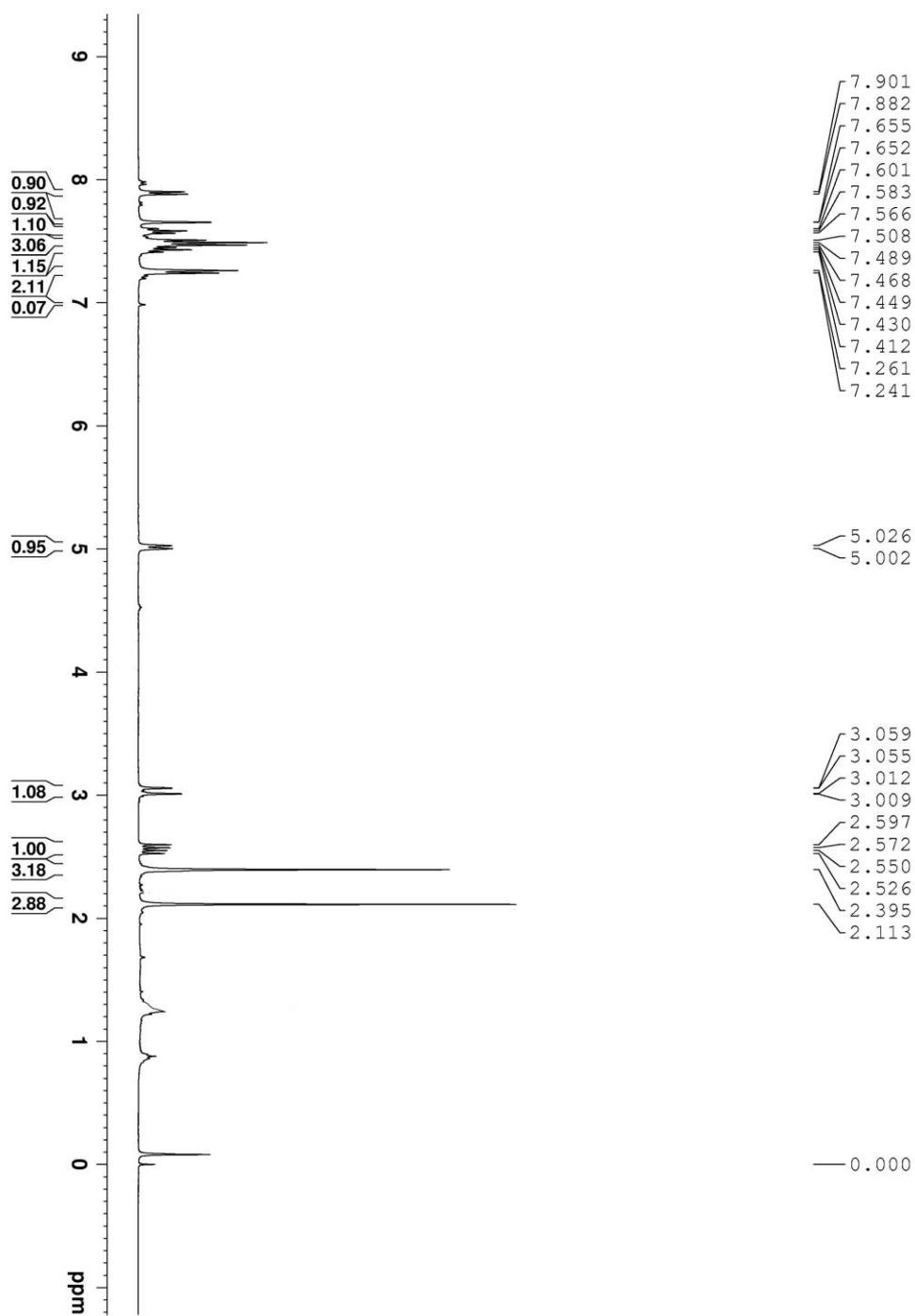
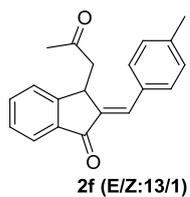
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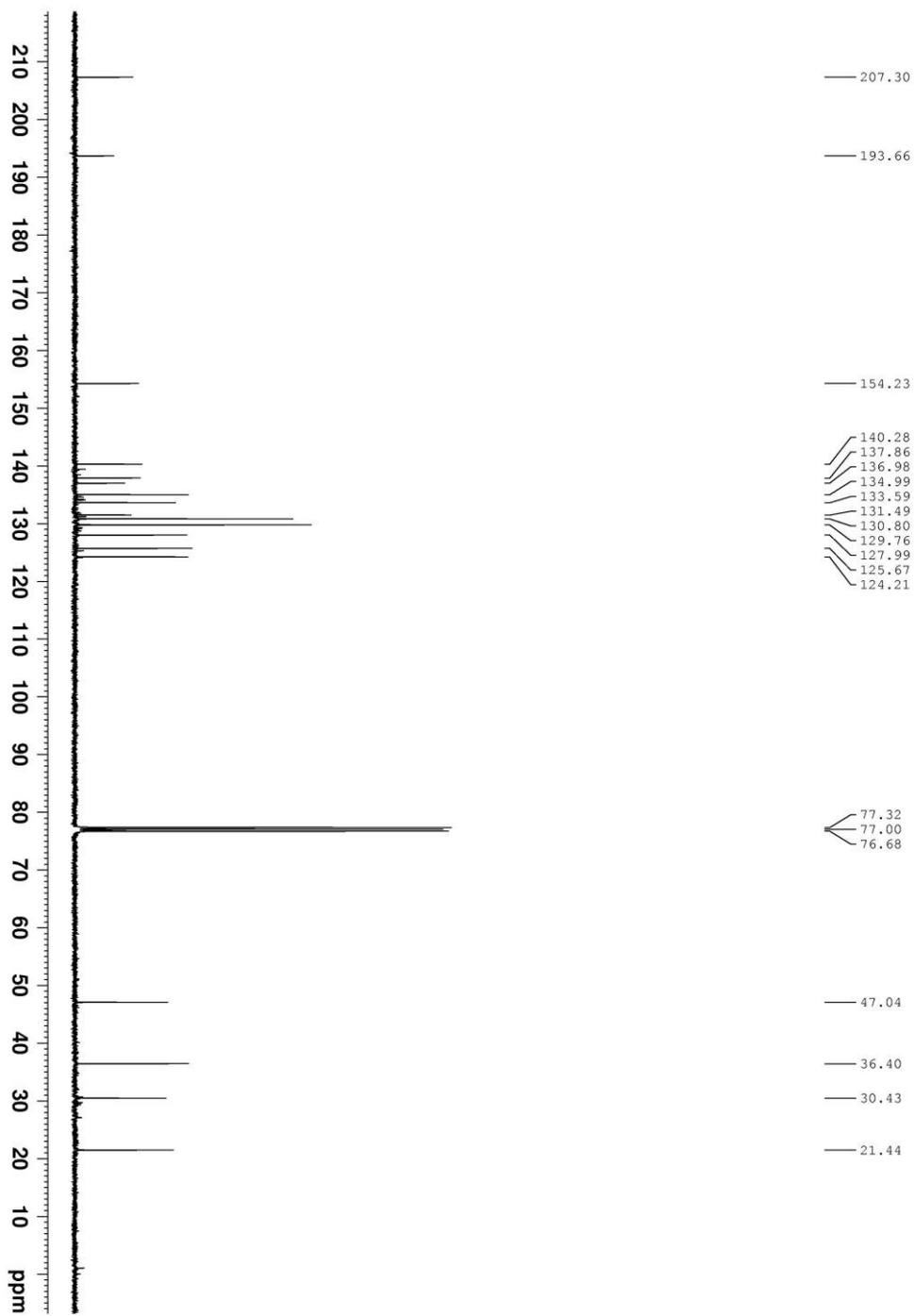
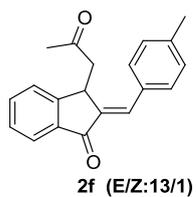


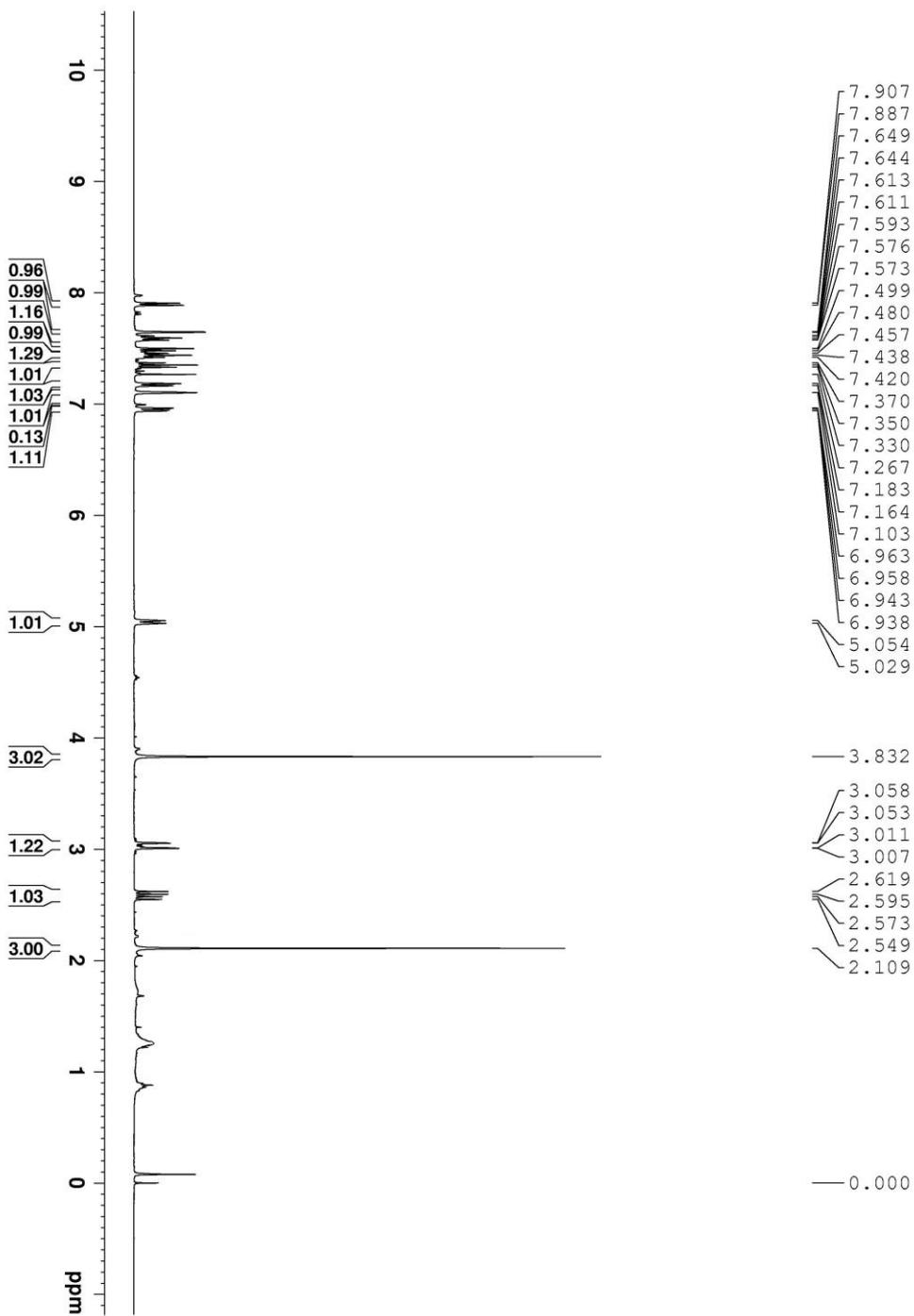
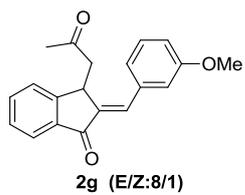


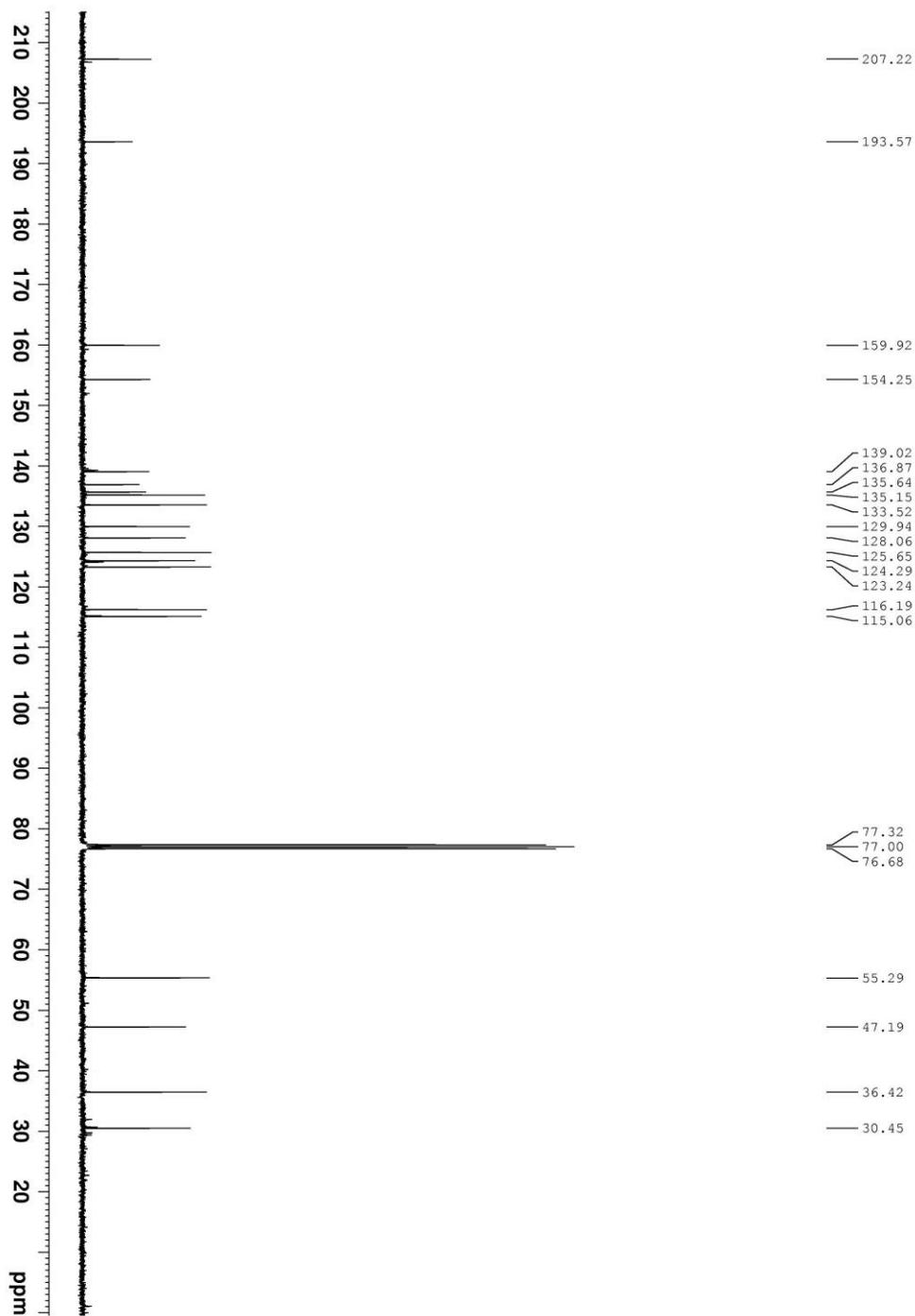
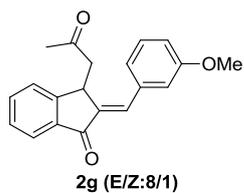
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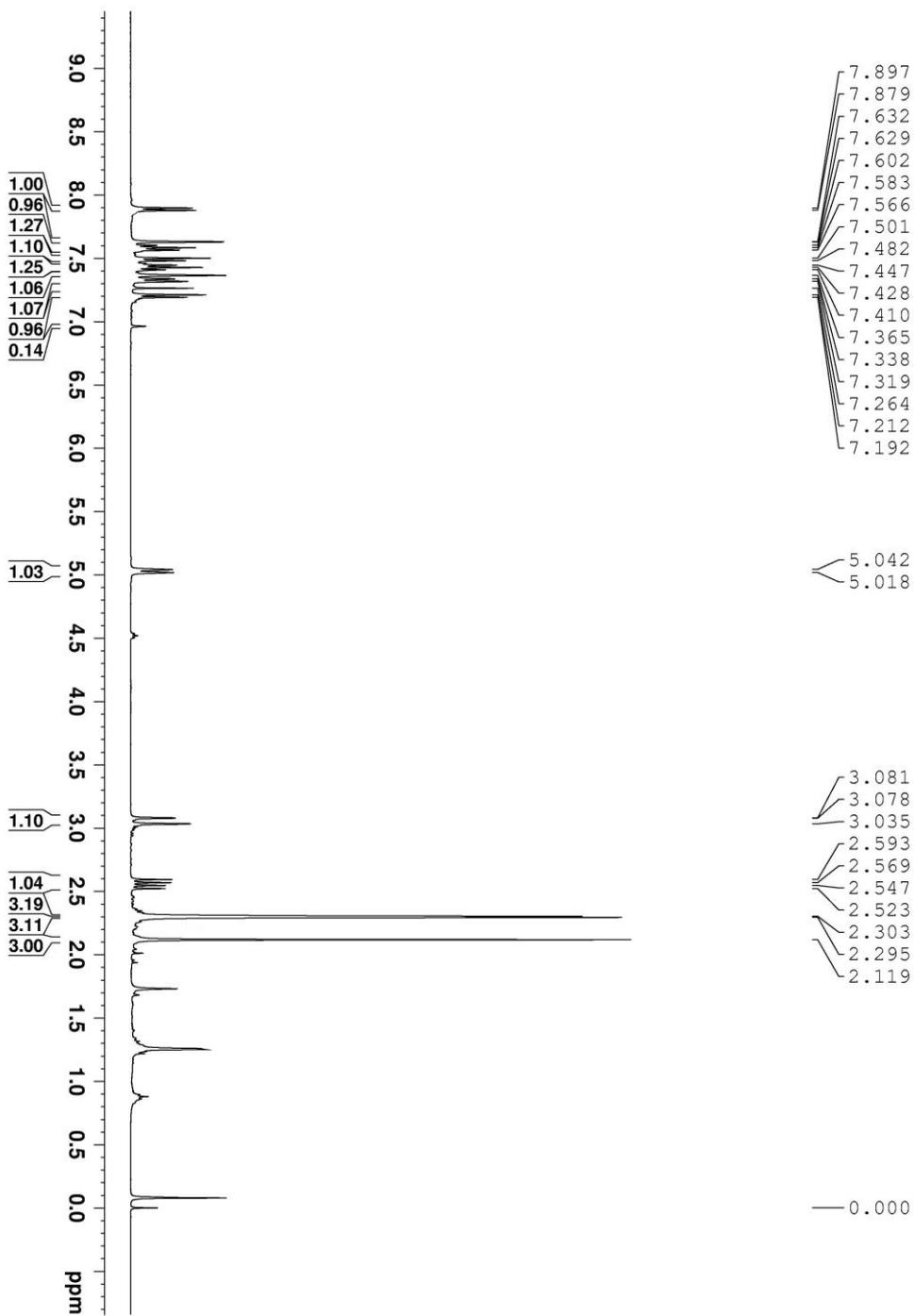
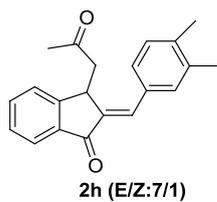


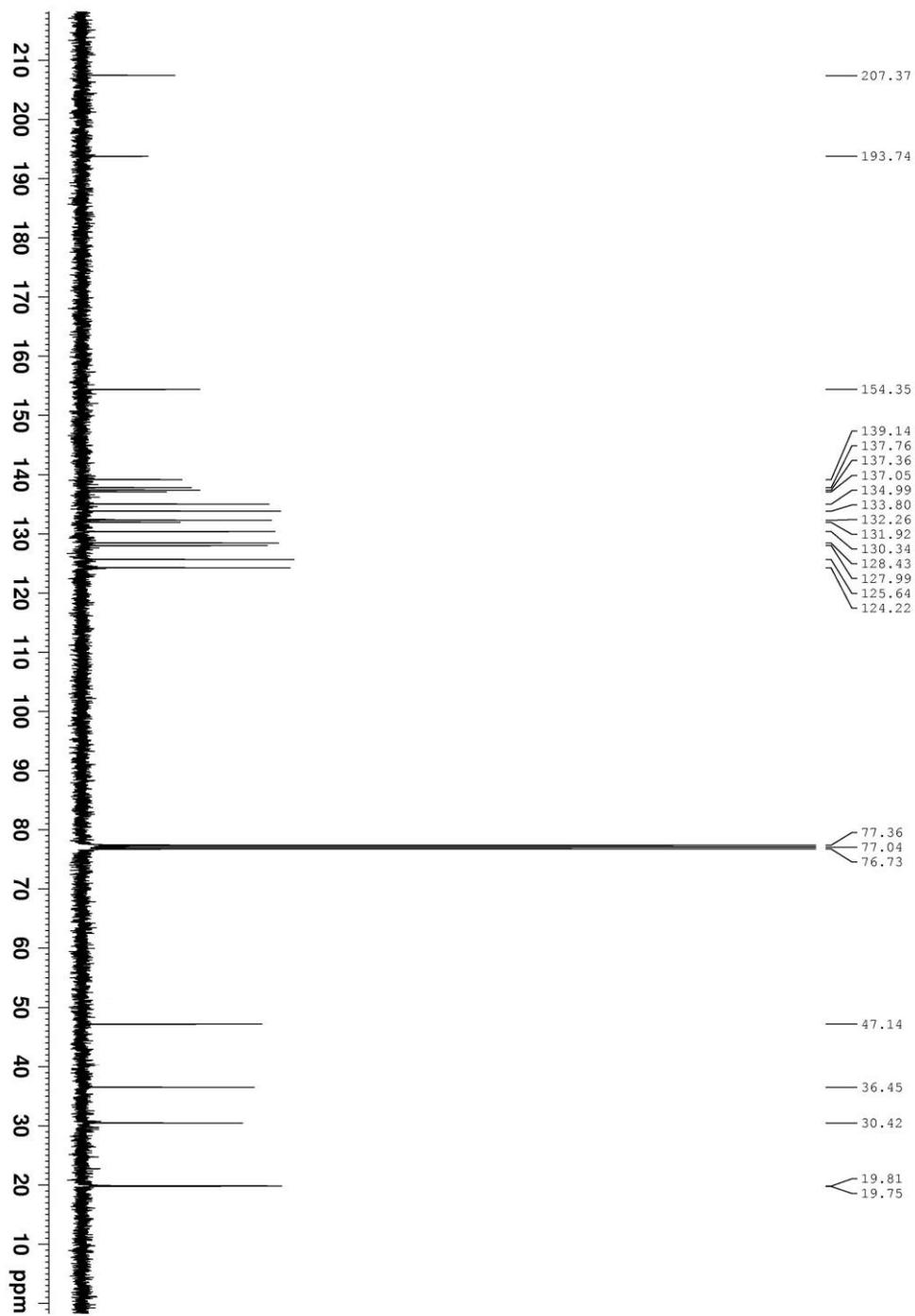
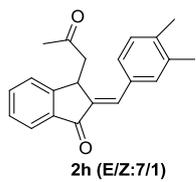


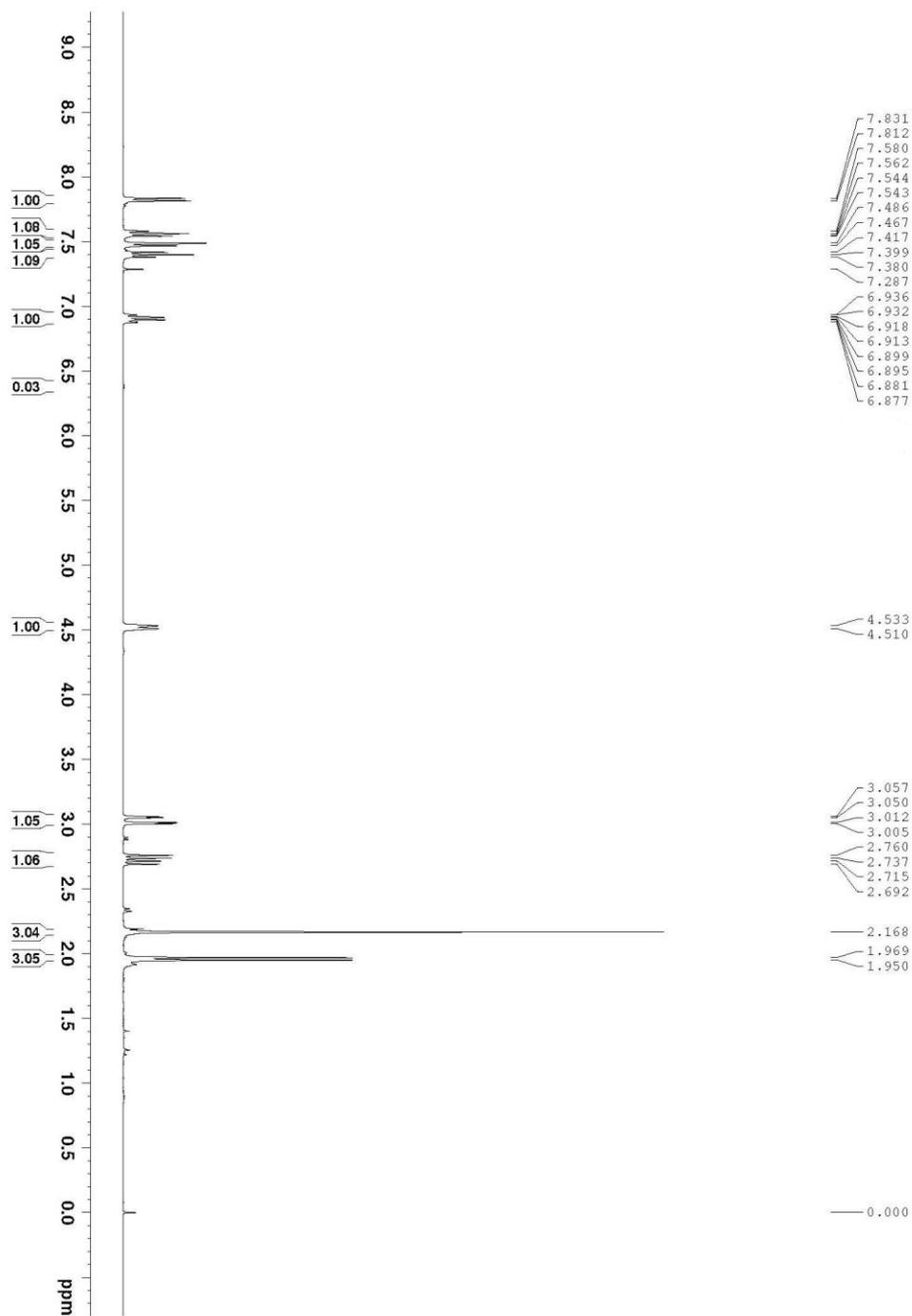
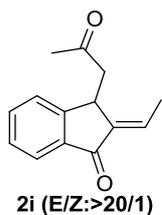


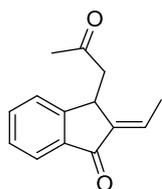






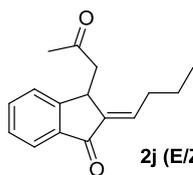






2i (E/Z:>20/1)





2j (E/Z:>20/1)

