

Supporting Information

Decomposition of Potent Greenhouse Gas Sulfur Hexafluoride (SF₆) by Kirschsteinite-dominant Stainless Steel Slag

Jia Zhang ^a, Ji Zhi Zhou ^a, Zhi Ping Xu ^b, Yajun Li ^a, Jun Zhao ^a, Xiuxiu Ruan ^a, Qiang Liu ^{a*},
Guangren Qian ^{a**}

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Fe₂O₃ **(C)**

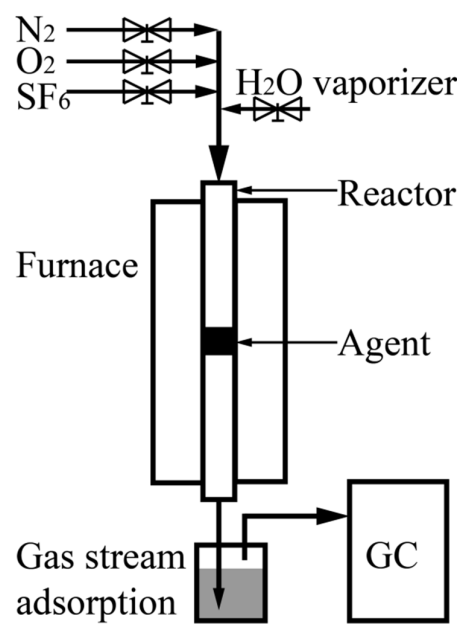


Figure S1 Schematic diagram of experimental setup

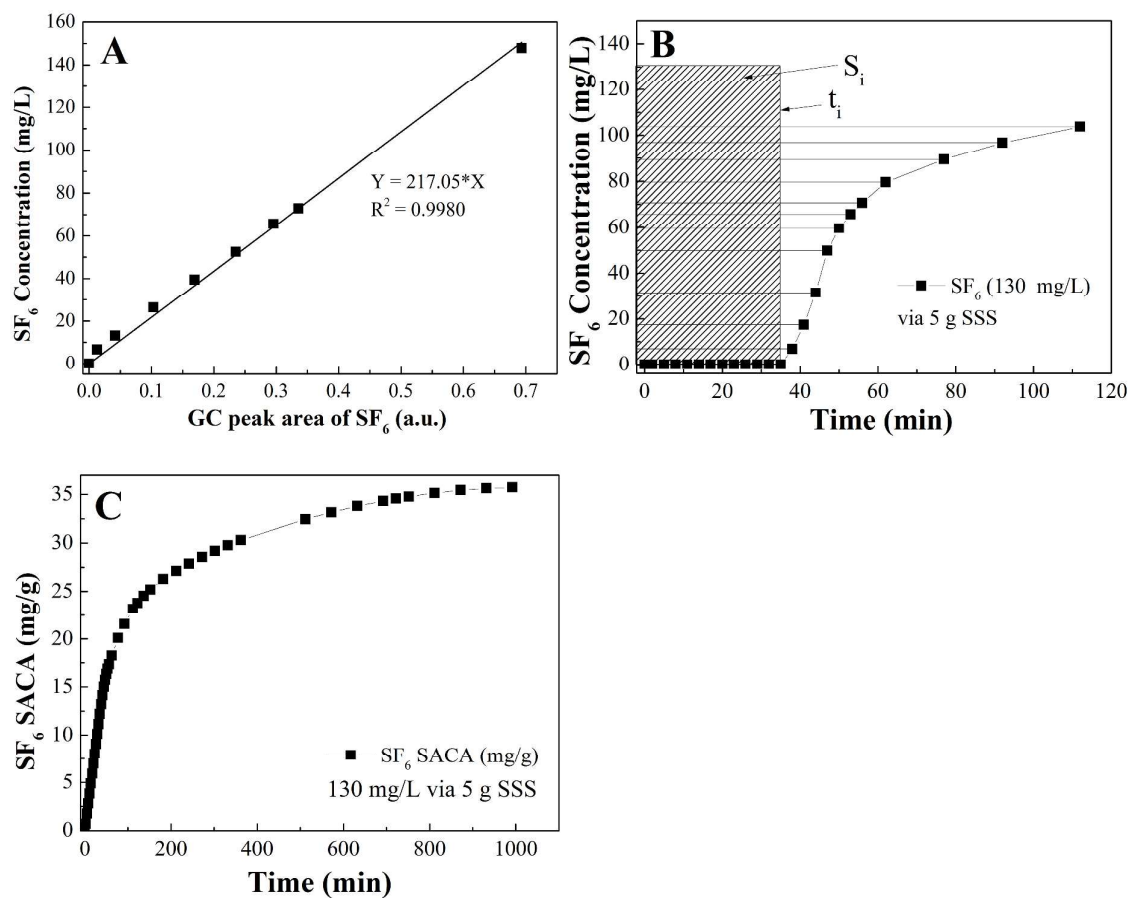


Figure S2. Data acquirement, treatment and presentation

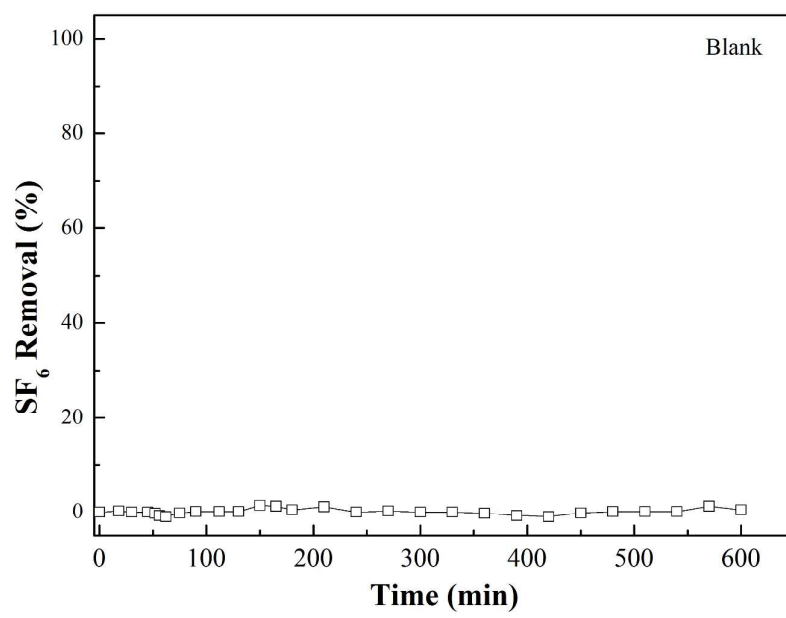


Figure S3. Blank test ran under no slag at 600 °C

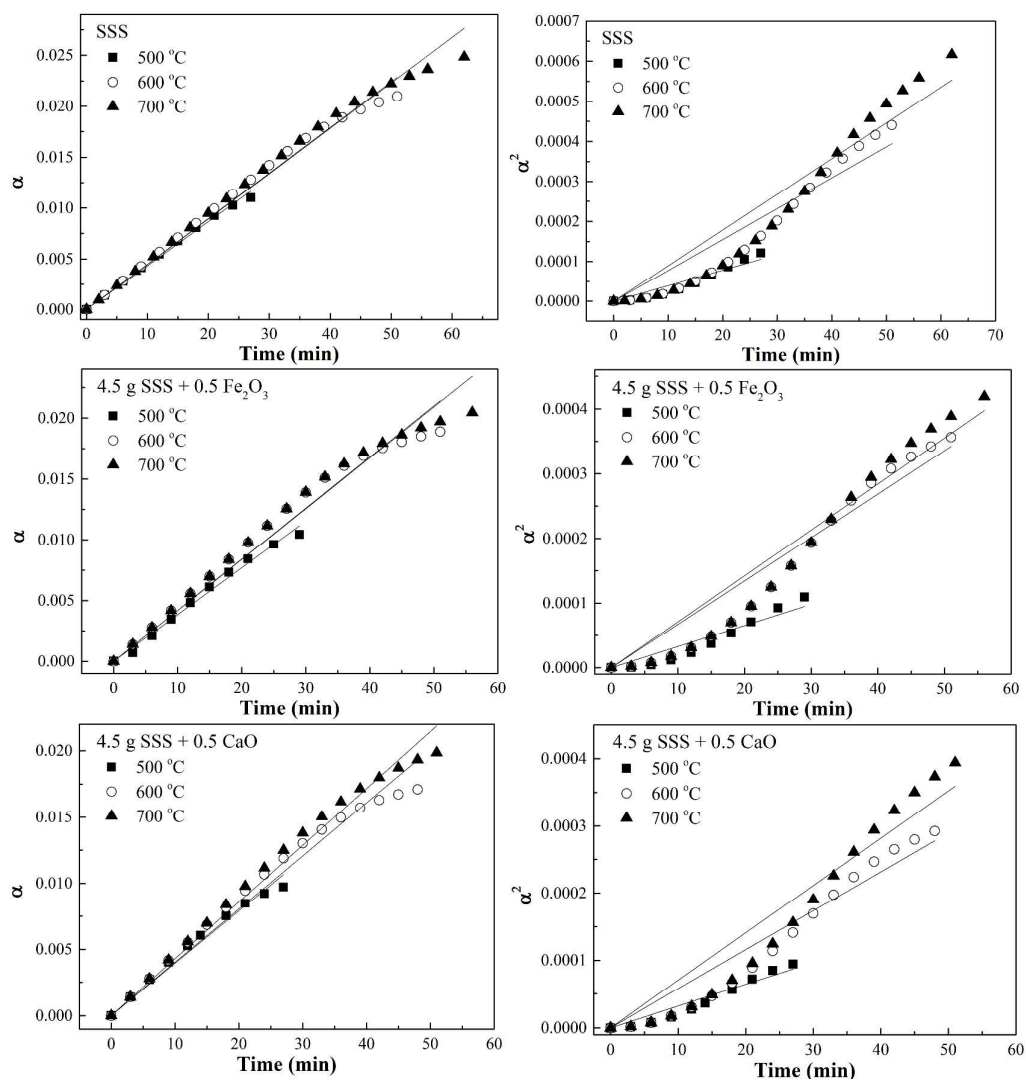


Figure S4 α and α^2 fitted with time

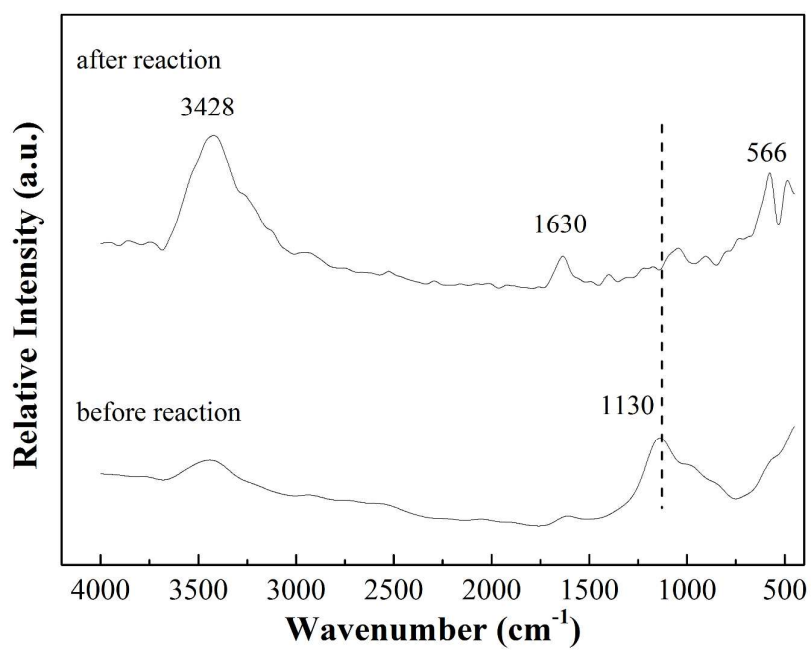


Figure S5 FTIR of slag before and after reaction

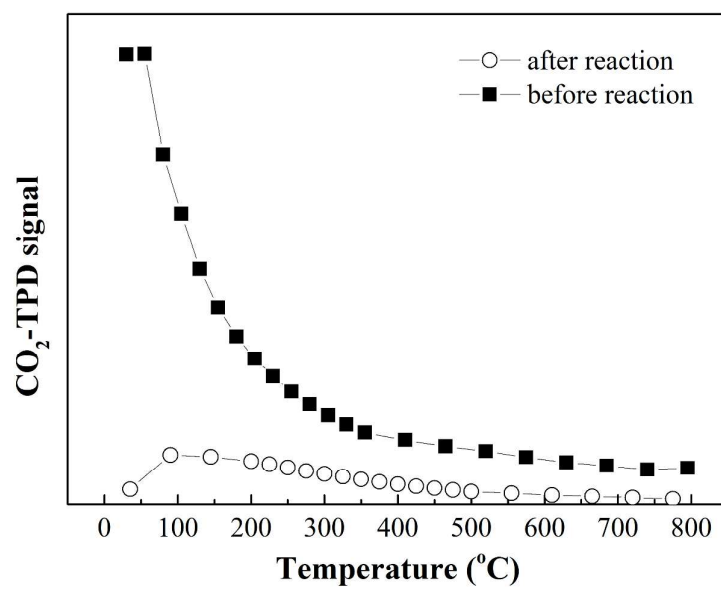


Figure S6 CO₂-TPD of slag before and after reaction

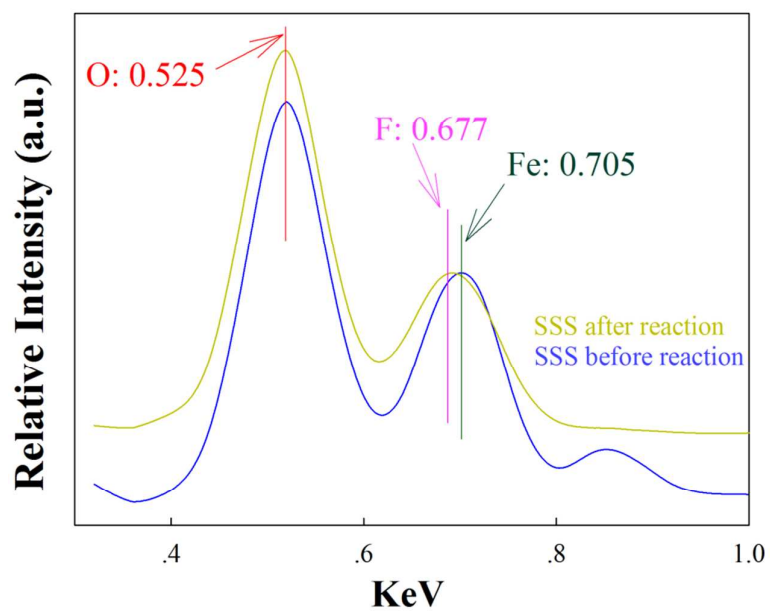


Figure S7 EDSs of SSS before and after reaction: partial comparison

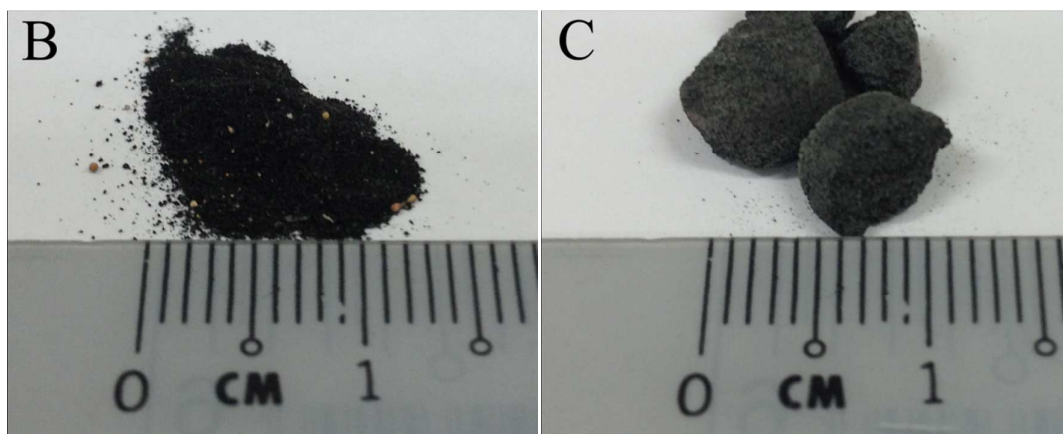
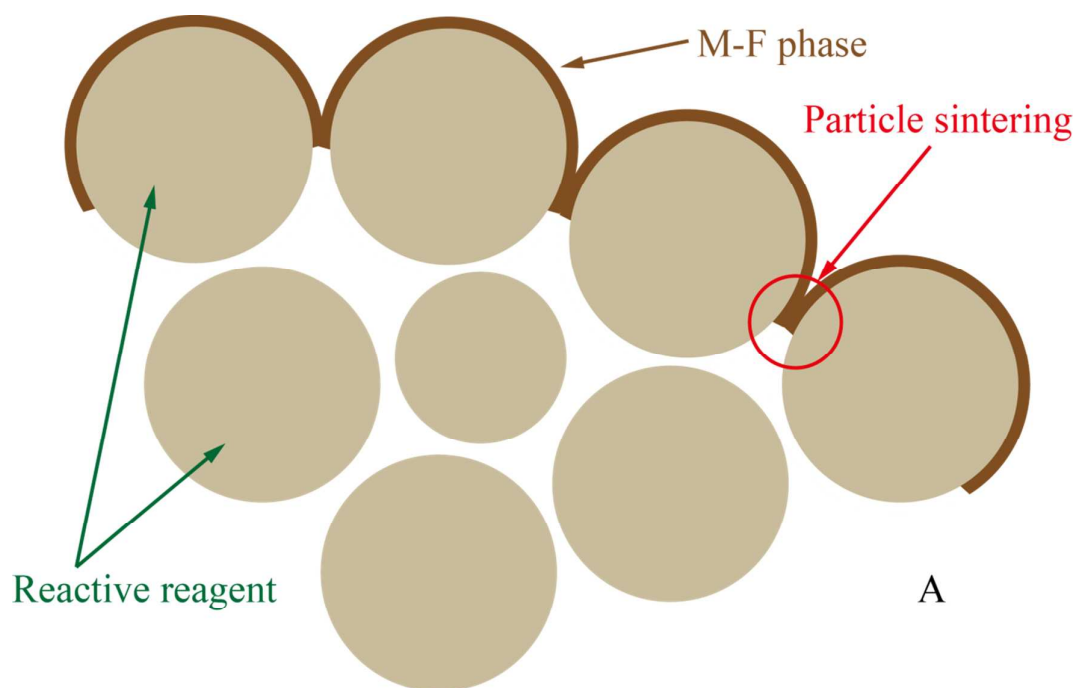


Figure S8 Scheme of sintering (A); reacted SSS powder (B); reacted SSS with 10 % Fe_2O_3 (C)