Supporting Information

for

Yttrium- and Aluminum-Bis(phenolate)pyridine Complexes : Catalysts and Model Compounds of the Intermediates for the Stereoselective Ring-Opening Polymerization

of racemic Lactide and β -Butyrolactone

Joice S. Klitzke, Thierry Roisnel, Evgeny Kirillov, Osvaldo Casagrande Jr and Jean-François Carpentier

Figure S1. ¹H NMR spectrum (300 MHz, CDCl₃, 298 K) of 1-(methoxymethyl)-4-methyl-2-(2-phenylpropan-2-yl)benzene.

Figure S2. ¹H NMR spectrum (500 MHz, CDCl₃, 298 K) of {ONO^{Me,Cumyl}}H₂.

Figure S3. ${}^{13}C{}^{1}H$ NMR spectrum (125 MHz, CDCl₃, 298 K) of {ONO^{Me,Cumyl}}H₂.

Figure S4. ¹H NMR spectrum (500 MHz, toluene- d_8 , 298 K) of {ONO^{Me,Cumy1}}Y[N(SiHMe₂)₂](THF)(Et₂O) (1).

Figure S5. ¹H NMR spectrum (500 MHz, pyridine- d_5 , 298 K) of {ONO^{Me,Cumyl}}Y[N(SiHMe₂)₂](THF)(Et₂O) (**1**)

Figure S6. ¹³C{¹H} NMR spectrum (125 MHz, pyridine- d_5 , 298 K) of {ONO^{Me,Cumyl}}Y[N(SiHMe₂)₂](THF)(Et₂O) (**1**).

Figure S7. Details of the aliphatic region of the VT ¹H NMR spectra (500 MHz, toluene- d_8 , 298–363 K) of {ONO^{Me,Cumyl}}Y[N(SiHMe₂)₂](THF)(Et₂O) (**1**).

Figure S8. ¹H NMR spectrum (500 MHz, toluene- d_8 , 258 K) of {ONO^{Me,Cumyl}}Y((*R*)-OCH(CH₃)CH₂COOMe) (**2**).

Figure S9. ¹³C{¹H} NMR (100 MHz, toluene- d_8 , 258 K) of {ONO^{Me,Cumyl}}Y((*R*)-OCH(CH₃)CH₂COOMe) (2).

Figure S10. ¹H–¹H NOESY NMR spectrum (400 MHz, toluene- d_8 , 258 K) of {ONO^{Me,Cumyl}}Y((*R*)-OCH(CH₃)CH₂COOMe) (**2**).

Figure S11. ¹H NMR spectrum (500 MHz, THF- d_8 , 298 K) of {ONO^{Me,Cumyl}}Y((*S*,*S*)-OCH(CH₃)OCH(CH₃)COOMe)(THF) (**3**).

Figure S12. ¹³C-¹H HMQC NMR spectrum (500 MHz, THF- d_8 , 298 K) of {ONO^{Me,Cumyl}}Y((*S*,*S*)-OCH(CH₃)OCH(CH₃)COOMe)(THF) (**3**).

Figure S13. ¹³C-¹H HMBC NMR spectrum (500 MHz, THF- d_8 , 298 K) of {ONO^{Me,Cumyl}}Y((*S*,*S*)-OCH(CH₃)OCH(CH₃)COOMe)(THF) (**3**).

Figure S14a. ¹H NMR (500 MHz, THF- d_8 , 298K) spectrum of the 1:1 reaction mixture of **1** and methyl (*S*,*S*)-lactyllactate at room temperature after 30 min (bottom spectrum) and after 18 h reaction, evaporation of volatiles and addition of fresh THF- d_8 (bottom spectrum).

Figure S15. ¹H NMR spectrum (500 MHz, CD₂Cl₂, 298 K) of {ONO^{Me,Cumyl}}AlMe (4).

Figure S16. ${}^{13}C{}^{1}H$ NMR spectrum (125 MHz, CD₂Cl₂, 298 K) of {ONO^{Me,Cumyl}}AlMe (4)

Figure S17. ¹H NMR spectrum (500 MHz, toluene- d_8 , 298 K) of {ONO^{Me,Cumyl}}Al(*i*Pr (*S*)-lactate) (5).

Figure S18. ¹³C{¹H} NMR spectrum (125 MHz, toluene- d_8 , 298 K) of {ONO^{Me,Cumyl}}Al(*i*Pr (*S*)-lactate) (**5**).

Figure S19. ¹H–¹H NOESY NMR spectrum (400 MHz, toluene- d_8 , 298 K) of {ONO^{Me,Cumyl}}Al(*i*Pr (*S*)-lactate) (**5**).

Figure S20. ¹H NMR spectrum (500 MHz, CD₂Cl₂, 298 K) of {ONO^{Me,Cumyl}}Al((*R*)-OCH(CH₃)CH₂COOMe) (**6**).

Figure S21. ¹³C{¹H} NMR spectrum (100 MHz, C_6D_6 , 298 K) of {ONO^{Me,Cumyl}}Al((*R*)-OCH(CH₃)CH₂COOMe) (6).

Figure S22. ¹H NMR spectrum (500 MHz, CD₂Cl₂, 298 K) of {ONO^{Me,Cumyl}}Al((*rac*)-OCH(CF₃)CH₂COOEt) (**7**).

Figure S23. ¹³C{¹H} NMR spectrum (125 MHz, CD₂Cl₂, 298 K) of $\{ONO^{Me,Cumyl}\}Al((rac)-OCH(CF_3)CH_2COOEt)$ (7).

Figure S23bis. DEPT 135 NMR spectrum (125 MHz, CD_2Cl_2 , 298 K) of $\{ONO^{Me,Cumyl}\}Al((rac)-OCH(CF_3)CH_2COOEt)$ (7).

Figure S24. ¹⁹F{¹H}NMR spectrum (185 MHz, CD_2Cl_2 , 298 K) of { $ONO^{Me,Cumyl}$ }Al((*rac*)-OCH(CF₃)CH₂COOEt) (7).

Figure S25. ${}^{1}\text{H}-{}^{1}\text{H}$ NOESY NMR spectrum (400 MHz, CD₂Cl₂, 298 K) of {ONO^{Me,Cumyl}}Al((*rac*)-OCH(CF₃)CH₂COOEt) (7).

Figure S26. ${}^{1}H-{}^{1}H$ COSY NMR spectrum (500 MHz, CDCl₃, 298 K) of a PLA produced from {ONO^{Me,Cumyl}}Y((*R*)-OCH(CH₃)CH₂COOMe) (**2**).

Figure S27. Detail of the MALDI-ToF mass spectrum of a PLA sample produced from **2** using IAA as matrix (Table 1, entry 10).

Figure S28. ${}^{1}\text{H}-{}^{1}\text{H}$ COSY NMR spectrum (500 MHz, CDCl₃, 298 K) of a PHB produced from {ONO^{Me,Cumyl}}Y((*R*)-OCH(CH₃)CH₂COOMe) (**2**).

Figure S29: Methine region of ¹H NMR and ¹H homo-decoupled NMR spectra of different PLAs.

Figure S30. Carbonyl (left) and methylene (right) regions of the ${}^{13}C{}^{1}H$ NMR spectrum (100 MHz, CDCl₃, 298 K) of a PHB produced from {ONO^{Me,Cumyl}}Y((*R*)-OCH(CH₃)CH₂COOMe) (**2**).

Figure S31. Detail of the MALDI-ToF mass spectrum of a PHB sample produced from **2** using IAA as matrix (Table 1, entry 20).

Table S1. Summary of crystal and refinement data for $\{ONO^{Me,Cumyl}\}H_2$, 1 and 6.

Figure S32. Molecular structure of proligand {ONO^{Me,Cumyl}}H₂.



Figure S1. ¹H NMR spectrum (300 MHz, CDCl₃, 298 K) of 1-(methoxymethyl)-4-methyl-2-(2-phenylpropan-2-yl)benzene.



Figure S2. ¹H NMR spectrum (500 MHz, CDCl₃, 298 K) of {ONO^{Me,Cumyl}}H₂.



Figure S3. ¹³C{¹H} NMR spectrum (125 MHz, CDCl₃, 298 K) of {ONO^{Me,Cumyl}}H₂ (*stands for residual solvent resonances).



Figure S4. ¹H NMR spectrum (500 MHz, toluene- d_8 , 298 K) of {ONO^{Me,Cumyl}}Y[N(SiHMe₂)₂](THF)(Et₂O) (1) (* stands for residual solvent resonances).



Figure S5. ¹H NMR spectrum (500 MHz, pyridine- d_5 , 298 K) of {ONO^{Me,Cumyl}}Y[N(SiHMe₂)₂](THF)(Et₂O) (**1**) (* stands for residual solvent resonance).



Figure S6. ¹³C{¹H} NMR spectrum (125 MHz, pyridine- d_5 , 298 K) of {ONO^{Me,Cumyl}}Y[N(SiHMe₂)₂](THF)(Et₂O) (1) (* stands for residual solvent resonance).



Figure S7. Details of the aliphatic region of the VT-¹H NMR spectra (500 MHz, toluene- d_8 , 298–363 K) of {ONO^{Me,Cumyl}}Y[N(SiHMe₂)₂](THF)(Et₂O) (1); from bottom to top, 298 K, 333 K, 353 K and 363 K. Markers **A** denote genuine compound **1**, and **B** and **C** unidentified species forming upon time and heating (* refers to residual solvent resonances).



Figure S8. ¹H NMR spectrum (500 MHz, toluene- d_8 , 258 K) of {ONO^{Me,Cumyl}}Y((*R*)-OCH(CH₃)CH₂COOMe) (2).



Figure S9. ¹³C{¹H} NMR (100 MHz, toluene- d_8 , 258 K) of {ONO^{Me,Cumyl}}Y((*R*)-OCH(CH₃)CH₂COOMe) (2) (* stands for residual solvent resonance).



Figure S10. Detail of the ${}^{1}H{-}^{1}H$ NOESY NMR spectrum (400 MHz, toluene- d_{8} , 258 K) of {ONO^{Me,Cumyl}}Y((*R*)-OCH(CH₃)CH₂COOMe) (2) (* stands for residual solvent resonance).



Figure S11. ¹H NMR spectrum (500 MHz, THF- d_8 , 298 K) of {ONO^{Me,Cumyl}}Y((*S*,*S*)-OCH(CH₃)OCH(CH₃)COOMe)(THF) (**3**) *in situ* generated upon addition of one equiv of methyl (S,S)-lactyllactate to **1** (after 30 min at room temperature) (* stands for residual solvent resonance).



situ generated upon addition of one equiv of methyl (S,S)-lactyllactate to 1 (after 30 min at room temperature).



Figure S13. ${}^{13}C{}^{-1}H$ HMBC NMR spectrum (500 MHz, THF- d_8 , 298 K) of {ONO^{Me,Cumyl}}Y((*S*,*S*)-OCH(CH₃)OCH(CH₃)COOMe)(THF) (**3**) *in situ* generated upon addition of one equiv of methyl (S,S)-lactyllactate to **1** (after 30 min at room temperature).



Figure S14a. Details of the aromatic region of the ¹H NMR (500 MHz, THF- d_8 , 298K) spectrum of the 1:1 reaction mixture of **1** and methyl (*S*,*S*)-lactyllactate at room temperature after 30 min (bottom spectrum) and after 18 h reaction, evaporation of volatiles and addition of fresh THF- d_8 (bottom spectrum).



Figure S14b. Details of the SiH region.



Figure S14c. Details of the aliphatic region.



Figure S15. ¹H NMR spectrum (500 MHz, CD₂Cl₂, 298 K) of {ONO^{Me,Cumyl}}AlMe (4) (*stands for residual solvent resonances).



Figure S16. ¹³C{¹H} NMR spectrum (125 MHz, CD₂Cl₂, 298 K) of {ONO^{Me,Cumyl}}AlMe (**4**) (*stands for residual solvent resonances).



Figure S17. ¹H NMR spectrum (500 MHz, toluene-*d*₈, 298 K) of {ONO^{Me,Cumyl}}Al(*i*Pr (*S*)-lactate) (**5**) (*stands for residual solvent resonances).



Figure S18. ¹³C{¹H} NMR spectrum (125 MHz, toluene- d_8 , 298 K) of {ONO^{Me,Cumyl}}Al(*i*Pr (*S*)-lactate) (5).



Figure S19. ${}^{1}\text{H}-{}^{1}\text{H}$ NOESY NMR spectrum (400 MHz, toluene- d_8 , 298 K) of {ONO}^{Me,Cumyl} Al(*i*Pr (*S*)-lactate) (5).



Figure S20. ¹H NMR spectrum (500 MHz, CD_2Cl_2 , 298 K) of { $ONO^{Me,Cumyl}$ }Al((*R*)-OCH(CH₃)CH₂COOMe) (**6**) (* stands for residual solvent resonances; a small amount of free ligand contaminates the product).



Figure S21. ¹³C{¹H} NMR spectrum (100 MHz, C_6D_6 , 298 K) of {ONO^{Me,Cumyl}}Al((*R*)-OCH(CH₃)CH₂COOMe) (**6**) (* stands for impurities and residual solvent resonances).



Figure S22. ¹H NMR spectrum (500 MHz, CD₂Cl₂, 298 K) of {ONO^{Me,Cumyl}}Al((*rac*)-OCH(CF₃)CH₂COOEt) (**7**) (* stands for residual solvent resonances).



Figure S23. ¹³C{¹H} NMR spectrum (125 MHz, CD₂Cl₂, 298 K) of { $ONO^{Me,Cumyl}$ }Al((*rac*)-OCH(CF₃)CH₂COOEt) (7) (*stands for residual solvent resonances).



Figure S23bis. DEPT 135 NMR spectrum (125 MHz, CD_2Cl_2 , 298 K) of { $ONO^{Me,Cumyl}$ }Al((*rac*)-OCH(CF₃)CH₂COOEt) (7).



Figure S24. ¹⁹F{¹H}NMR spectrum (185 MHz, CD₂Cl₂, 298 K) of { $ONO^{Me,Cumyl}$ }Al((*rac*)-OCH(CF₃)CH₂COOEt) (7).



Figure S25. $^{1}H^{-1}H$ NOESY NMR spectrum (400 MHz, CD₂Cl₂, 298 K) of {ONO^{Me,Cumyl}}Al((*rac*)-OCH(CF₃)CH₂COOEt) (7).



Figure S26. ${}^{1}H-{}^{1}H$ COSY NMR spectrum (500 MHz, CDCl₃, 298 K) of a PLA produced from {ONO^{Me,Cumyl}}Y((*R*)-OCH(CH₃)CH₂COOMe) (**2**) (Table 1, entry 10).



Figure S27. Detail of the MALDI-ToF mass spectrum of a PLA sample produced from **2** using IAA as matrix (Table 1, entry 10). Top: experimental spectrum; the observed two distributions at m/z = 4463 and 4479 Da correspond to MeOC(O)CH₂CH(CH₃)O–(LA)_n–H macromolecules ionized by Na⁺ and K⁺. Bottom: calculated isotopic distribution for macromolecules ionized with Na⁺.



Figure S28. ${}^{1}\text{H}-{}^{1}\text{H}$ COSY NMR spectrum (500 MHz, CDCl₃, 298 K) of a PHB produced from {ONO^{Me,Cumyl}}Y((*R*)-OCH(CH₃)CH₂COOMe) (**2**) (Table 1, entry 21).



Figure S29. Methine region of ¹H NMR and ¹H homo-decoupled NMR spectra of different PLAs.



Figure S30. Carbonyl (left) and methylene (right) regions of the ${}^{13}C{}^{1}H$ NMR spectrum (100 MHz, CDCl₃, 298 K) of a PHB produced from {ONO^{Me,Cumyl}}Y((*R*)-OCH(CH₃)CH₂COOMe) (**2**) (Table 1, entry 21).



Figure S31. Detail of the MALDI-ToF mass spectrum of a PHB sample produced from **2** using IAA as matrix (Table 1, entry 20). Top: experimental spectrum; the observed two distributions correspond to $MeOC(O)CH_2CH(CH_3)O-(BBL)_n-H$ macromolecules ionized by Na⁺ and K⁺. Bottom: calculated isotopic distribution for macromolecules ionized with Na⁺.

	$\{ONO^{Me,Cumyl}\}H_2$	1	6
Empirical formula	C ₃₇ H ₃₇ NO ₂	$C_{49}H_{67}N_2O_4Si_2Y$	$C_{84}H_{88}Al_2N_2O_{10}$
Formula weight	527.68	893.14	1339.52
Crystal system	Monoclinic	Monoclinic	Triclinic
Space group	$P 2_l/n$	$P 2_1/n$	Р <i>—1</i>
<i>a</i> , Å	13.1098(12)	12.896(2)	15.8935(5)
<i>b</i> , Å	8.0072(7)	25.405(3)	16.9816(4)
<i>c</i> , Å	27.619(3)	15.536(2)	18.2728(6)
α (°)	90	90	107.2690(10)
β (°)	90.292(4)	110.122(5)	107.1600(10)
γ (°)	90	90	114.6260(10)
Volume, Å ³	2899.2(5)	4779.3(11)	3750.09(19)
Z	4	4	2
Density, g.m ⁻³	1.209	1.241	1.186
m, mm^{-1}	0.074	1.314	0.098
F(000)	1128	1896	1424
Crystal size, mm	$0.57 \times 0.08 \times 0.05$	$0.55 \times 0.46 \times 0.38$	$0.6 \times 0.3 \times 0.23$
θ range, deg	2.94 to 27.48	2.94 to 27.48	1.33 to 27.52
Limiting indices	$\begin{array}{l} -16 \leq h \leq 16, -9 \leq k \\ \leq 10, -35 \leq l \leq 35 \end{array}$	$-16 \le h \le 16, -27 \le k \le 32, -30 \le l \le 20$	$\begin{array}{l} -20 \leq h \leq 17, -20 \leq \\ k \leq 22, -23 \leq l \leq 23 \end{array}$
Reflec. Collected	26552	42223	42735
R _{int}	0.07	0.0531	0.044
Unique Refl [I>2 σ (I)]	6546	10851	17008
Data/restrains/ param.	6546 / 0 / 369	10851 / 0 / 512	17008 / 0 / 913
Goodness-of-it on F ²	1.012	1.022	1.061
$R_1 [I > 2\sigma(I)]$ (all data)	0.0546 (0.12)	0.0518 (0.0742)	0.078 (0.119)
wR ₂ [I> $2\sigma(I)$] (all data)	0.1113 (0.144)	0.1261 (0.136)	0.2223 (0.2412)
Largest diff. e.A ⁻³	0.255 and -0.243	1.397 and -0.993	0.512 and -0.450

Table S1. Summary of crystal and refinement data for $\{ONO^{Me,Cumyl}\}H_2$, 1 and 6.



Figure S32. Molecular structure of proligand {ONO^{Me,Cumyl}}H₂ (all hydrogens atoms, except those of the hydroxyl groups, are omitted for clarity; thermal ellipsoids drawn at 50% probability). Selected bond distances (Å) and angles (deg): H(O(1))–N = 1.919; H(O(2))–N = 1.985; \angle Pyr–Ph(1) = 23.55; \angle Pyr–Ph(2) = 23.21.

Computational Details

The geometry optimizations have been performed with the program package TURBOMOLE using density functional theory (DFT).¹ The gradient corrected density functional BP86 in combination with the resolution identity approximation (RI)² was applied for the geometry optimizations of stationary points. A triple- ξ zeta valence quality basis set def2-TZVP³ was used for all atoms. In the calculations have been included solvation effects using COSMO model implemented in the TURBOMOLE program package.⁴ The default optimized atomic COSMO radii and the corresponding parameters for solvent toluene ($\varepsilon = 2.38$, radius = 3.48 Å) have been used.

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- ³ Weigend, F.; Ahlrichs. R. *Phys. Chem. Chem. Phys.* **2005**, 7, 3297.
- ⁴ (a) Klamt, A.; Eckert, F. *Fluid Phase Equilibria* 2000, 172, 43. (b) Schafer, A.; Klamt, A.; Sattel, D.; Lohrenz, J. C.; Eckert, F. *Phys. Chem. Chem. Phys.* 2000, *2*, 2187.

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H 4.6475284 -14.6180701 11.246710)7
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En	ergy = -2097.	356995094	
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0	0.1670266	-0.0065147	8.1493584
0	2.5495781	-1.2649090	9.1026515
0	4.1774218	0.2772797	9.2364246
С	2.8915682	-0.0789431	9.2458886
С	5.1508692	-0.7901317	9.0565629
Η	5.0466691	-1.5322246	9.8564937
Η	6.1246382	-0.2961515	9.1049826
Η	5.0045055	-1.2711449	8.0826278
С	-0.4416665	-4.8506599	13.1801104
Н	-1.5072910	-4.6510024	13.3097063

С	0.0060978	-6.1712670	13.1245332	С	-1.3947850	-2.4676433	10.9955513
Η	-0.7104677	-6.9895708	13.2183528	C	-2.8245651	-2.9466578	8.9165354
С	1.3649416	-6.4467464	12.9445930	C	-4.0176690	-3.6465362	8.6575241
Η	1.7184755	-7.4778983	12.8958047	Н	-4.7212200	-3.8188806	9.4691807
С	2.2625254	-5.3846265	12.8225556	C	-4.2635792	-4.1516113	7.3876309
Η	3.3262348	-5.5822615	12.6762394	Н	-5.1724391	-4.7208343	7.1857687
С	1.8074747	-4.0631701	12.8785952	C	-3.3236533	-3.9484858	6.3856585
Η	2.5309696	-3.2549850	12.7708751	Н	-3.4749584	-4.3596677	5.3898670
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С	-0.0831598	-2.3315702	13.2053893	С	-1.1825890	-3.0383790	5.5494559
С	-0.3137772	-2.1160285	14.7220179	C	-1.7306971	-2.7733306	4.2759725
Η	0.6260396	-2.3087361	15.2581257	Н	-2.8045033	-2.5884318	4.1969100
Η	-1.0743670	-2.7999313	15.1229232	С	-0.9506392	-2.7097749	3.1281233
Η	-0.6203315	-1.0824232	14.9413902	С	-1.5437431	-2.3639007	1.7840882
С	0.9434303	-1.2666537	12.7564062	Н	-1.2313718	-1.3613515	1.4495183
Η	1.8383877	-1.2803499	13.3960970	Н	-2.6417489	-2.3720909	1.8191680
Η	0.4906593	-0.2686578	12.8417611	Н	-1.2253296	-3.0747025	1.0063556
Η	1.2544804	-1.4267914	11.7175062	С	0.4235177	-2.9803051	3.2623432
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Η	-4.8336390	0.1767388	13.1347977	Н	2.9310199	-2.2248829	6.3136285
Η	-5.1674519	-1.3757352	13.9163041	Н	3.0117258	-1.4201971	4.7248371
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Η	-2.8045033	-2.5884318	4.1969100
С	-0.9506392	-2.7097749	3.1281233
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Η	-2.6417489	-2.3720909	1.8191680
Η	-1.2253296	-3.0747025	1.0063556
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Η	1.0310830	-2.9818374	2.3560000
С	1.0331776	-3.2368264	4.4889577
С	0.2317803	-3.2036047	5.6768761
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С	3.7811338	-5.0392689	6.2935315
Η	4.4004288	-4.1930871	6.5910589
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Η	4.7979339	-6.3836428	7.6405386
С	3.2289140	-7.3869523	6.5462571
Η	3.4023046	-8.3559818	7.0169331
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Η	1.5970482	-8.0856241	5.3110037
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Η	2.2633371	0.8070134	6.6103929

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Η	1.7557738	-13.8730924	10.9031997
С	0.6635482	-13.7800634	12.7607327
Η	0.9811848	-14.7599834	13.1404462
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Η	-1.0472403	-7.1967123	14.1719334
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Η	-4.4614987	-9.0925488	14.0870541
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Η	-6.3156526	-9.4849296	10.8380448
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Η	-1.8001309	-8.4232859	6.5116226
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С	2.5761718	-10.4934635	7.1981906
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С	4.9152035	-12.5025503	11.0712949
Η	4.3895373	-12.5828414	10.1191680
С	-0.1479980	-14.8300029	10.6155379
Н	0.2103137	-15.8215720	10.9306580

Η	-0.1059065	-14.7751703	9.5196201
Η	-1.1979561	-14.7171192	10.9245307

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Energy = -2301.410479498

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0	-0.4655085	-0.1505659	7.9807588
0	1.8893713	-1.2641458	8.8134203
0	3.4997088	0.2008448	9.3597909
С	2.2283051	-0.1065530	9.1047539
С	4.4614151	-0.8918936	9.3146237
Η	4.1403099	-1.7035277	9.9770786
Η	5.4035865	-0.4534748	9.6535948
Η	4.5511257	-1.2657969	8.2883640
С	-0.2083351	-4.9819629	12.4797345
Η	-1.2526529	-4.7486267	12.6962820
С	0.1584744	-6.3032807	12.2199523
Η	-0.5997495	-7.0884672	12.2387089
С	1.4889324	-6.6203755	11.9323703
Η	1.7785383	-7.6511742	11.7231511
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Η	3.4820896	-5.8288983	11.6768343
С	2.0663328	-4.2771765	12.1663295
Η	2.8301237	-3.5003427	12.1335559
С	0.7362833	-3.9425626	12.4554018
С	0.3072893	-2.5144035	12.8479525

С	0.2344548	-2.5035553	14.3957482
Η	1.2122715	-2.8012147	14.7992985
Η	-0.5135421	-3.2111372	14.7788809
Η	0.0015744	-1.5000835	14.7827174
С	1.3492780	-1.4471992	12.4450611
Η	2.2889932	-1.5807781	13.0008910
Η	0.9614795	-0.4482954	12.6911953
Η	1.5586116	-1.4947332	11.3718923
С	-1.0533074	-2.1759302	12.2057203
С	-2.1446668	-1.7983850	12.9858307
Η	-2.0197354	-1.7150784	14.0660880
С	-3.4122741	-1.5019615	12.4504814
С	-4.5283986	-1.0061408	13.3366191
Η	-4.3365103	0.0188417	13.6928159
Η	-4.6469188	-1.6388033	14.2293976
Н	-5.4875911	-0.9954696	12.8012315
С	-3.5896317	-1.6918522	11.0884317
Н	-4.5715685	-1.4944550	10.6546222
С	-2.5358308	-2.1020335	10.2415520
С	-1.2242938	-2.2388986	10.7846716
С	-2.8521968	-2.4864654	8.8535130
С	-4.1463819	-2.9484157	8.5536312
Н	-4.8833798	-3.0357485	9.3474898
С	-4.4569299	-3.3503545	7.2623264
Η	-5.4505333	-3.7367443	7.0299336
С	-3.4795797	-3.2956273	6.2790307
Η	-3.6908032	-3.6570930	5.2762072
С	-2.1906236	-2.8280572	6.5934495
С	-1.1616928	-2.8011806	5.5376507

С	-1.5774601	-2.7285333	4.1889266	С	2.7129416	-4.7720593	5.9667251
Η	-2.6262002	-2.5228914	3.9677078	С	3.6461704	-4.8606115	7.0083771
С	-0.6922700	-2.8782006	3.1322848	Н	4.2803632	-4.0069674	7.2464425
С	-1.1382766	-2.7485153	1.6965149	С	3.7759458	-6.0281919	7.7668736
Η	-0.7003174	-1.8592032	1.2157010	Н	4.5063097	-6.0652118	8.5771565
Η	-2.2308178	-2.6591229	1.6253985	С	2.9723584	-7.1374615	7.4993503
Η	-0.8315481	-3.6198358	1.0977692	Н	3.0684456	-8.0464041	8.0950227
С	0.6496782	-3.1649892	3.4464498	С	2.0357838	-7.0649654	6.4643614
Η	1.3372974	-3.3403062	2.6182160	Н	1.3946348	-7.9209058	6.2448998
С	1.1347436	-3.2176663	4.7514396	С	1.9125261	-5.8976061	5.7094709
С	0.2273204	-2.9466621	5.8263071	Н	1.1721296	-5.8535252	4.9082822
С	2.6134271	-3.5427274	5.0415885	С	0.2677540	1.0391086	8.0198919
С	3.3059041	-2.2916413	5.6272630	Н	-0.4284977	1.8734075	8.2325561
Η	2.8876750	-2.0309729	6.6051117	С	1.2808525	1.0447892	9.1999708
Η	3.1620029	-1.4438534	4.9422996	Н	1.8369717	1.9885139	9.2662921
Η	4.3900544	-2.4528419	5.7264641	Н	0.7093158	0.9169819	10.1340279
С	3.3870872	-3.9180725	3.7526691	С	0.9515758	1.3315629	6.6788703
Η	4.4177028	-4.1874280	4.0227275	Н	1.4608866	2.3073243	6.6908449
Η	3.4329746	-3.0736557	3.0486722	Н	0.1973160	1.3430469	5.8811252
Η	2.9410818	-4.7807749	3.2389752	Н	1.6882900	0.5527469	6.4298474