Supporting Information

One-Step Carbonization Synthesis of Hollow Carbon Nanococoons with Multimodal Pores and their Enhanced Electrochemical Performance for Supercapacitor

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Figures

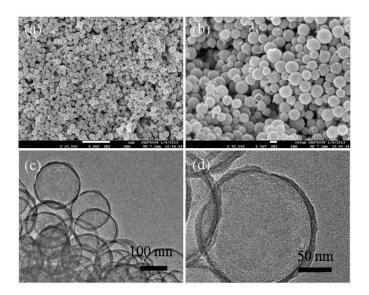


Fig. S1 SEM (a, b) and TEM (c, d) images of HCNs at different magnifications.

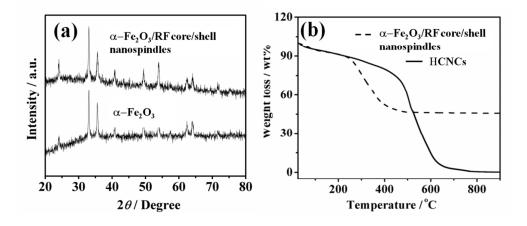


Fig. S2 (a) XRD patterns of spindle α -Fe₂O₃ and α -Fe₂O₃/RF core/shell nanospindles, respectively. (b) TGA curves of HCNCs and α -Fe₂O₃/RF core/shell nanospindles under air atmosphere with a heating rate of 10 °C min⁻¹, respectively.

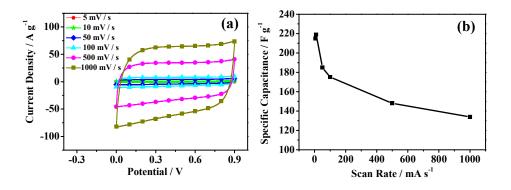


Fig. S3 Cyclic voltammograms (a) and specific capacitances (*vs.* scan rate) (b) of HCNCs. The test electrodes were prepared by loading a slurry consisting of 80 wt % active materials, 10 wt % carbon black, and 10 wt % poly(tetrafluoroethylene) (used as a binder, PTFE 60 wt% dispersion in H₂O, Sigma-Aldrich) on a nickel foam $(1 \times 1 \text{ cm}^2, 1.5 \times 1.5 \text{ cm}^2)$ and dried at 60 °C for 24 h. As-made electrodes were pressed at a pressure of 10 MPa for 1 min and further dried in a vacuum oven at 60 °C overnight. The loading mass of the active materials is 2 mg. The electrolyte was 2.0 M KOH aqueous solution.

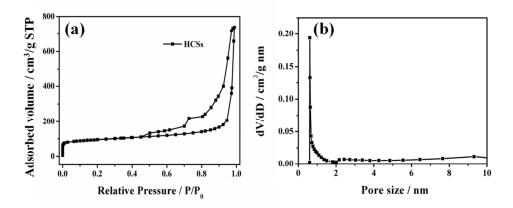


Fig. S4 (a) N₂ sorption isotherms and (b) the BJH pore size distribution curves of HCSs.

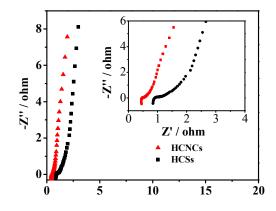


Fig. S5 Electrochemical impedance spectra of HCNCs and HCSs measured in the frequency range of 10 mHz to 10 kHz at the open circuit voltage with an alternate current amplitude of 5 mV.