1	Title: Profiling oil sands acid extractable organic mixtures from industrial developments and
2	natural groundwaters using high resolution mass spectrometry and comprehensive two
3	dimensional gas chromatography-mass spectrometry
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9	Supporting Information for Manuscript
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11 SI Geochemistry

In this section, a more thorough investigation is presented of the use of geochemical properties to distinguish natural groundwater from groundwater affected by oil sands process water (OSPW). Various criteria have been developed from past studies and these are briefly discussed and applied to the OSPW and background samples collected in this study. The rationale for using these specific properties is discussed in the main manuscript and will not be repeated here.

18 The geochemical composition of OSPW has generally been reported as being dominant 19 in sodium and bicarbonate, with varying degrees of chloride (Baker, 1999; Gibson et al., 2011; 20 Hunter, 2001; MacKinnon et al., 2005). The composition can vary over time and between the 21 various tailings ponds. Thus, it typically plots in the lower portion of the diamond on a Piper 22 diagram (Hunter, 2001), a region commonly classified as alkaline, or in the right region 23 (MacKinnon et al., 2005), classified as saline, or in between those two (Baker, 1999; Gibson et 24 al., 2011). The two OSPW samples collected for this study fit this pattern, being saline (but still 25 $\sim 40\%$ bicarbonate) (Figure 3B). Natural groundwater samples were predominantly fresh, as has 26 been reported for shallow groundwater in previous work (MacKinnon et al., 2005); however, one 27 or more of these samples plotted in each of the other three regions of a Piper plot (Figure 3A), 28 indicating that geochemical composition is not perfectly diagnostic of OSPW.

For sodium (Na) concentration as a distinguishing geochemical marker for OSPW, Baker
(1999) applied criteria of < 30 mg/L for natural groundwater and < 150 mg/L to indicate OSPW.
Hunter (2001) applied these conditions: over 1000 mg/L indicates natural saline groundwater, <
10 mg/L indicates background groundwater, and > 40 mg/L is possibly groundwater affected by

OSPW. The Na concentrations for our two OSPW samples were about 290 and 640 mg/L, which
fits with the above criteria. Of note, OSPW reported by MacKinnon et al. (2005) had a Na
concentration of about 900 mg/L, which is close to the criteria for natural saline groundwater
(Hunter, 2001). The Na concentrations for background groundwater samples in this study ranged
from < detection limit to nearly 1600 mg/L (Table 1). This variability encompasses the full range
of criteria, which also demonstrates that Na alone cannot be broadly applied to distinguish
OSPW impacts.

For applying the sodium to chloride (Na:Cl) molar ratio as an OSPW indicator, Hunter
(Hunter, 2001) applied the following conditions: background groundwater was generally about 1
for limestone and < 5 for sands; while a ratio >15 indicated OSPW affected groundwater. No
other criteria for Na:Cl have been reported. The 2 OSPW samples in this study had low Na:Cl
(2.5 and 1.0), and would have been classified as natural groundwater. Our natural groundwater
samples had ratios ranging between 0 and 22 (Table 1); thus covering both natural and OSPWaffected Na:Cl categories. Thus, the use of Na:Cl is not supported by this data.

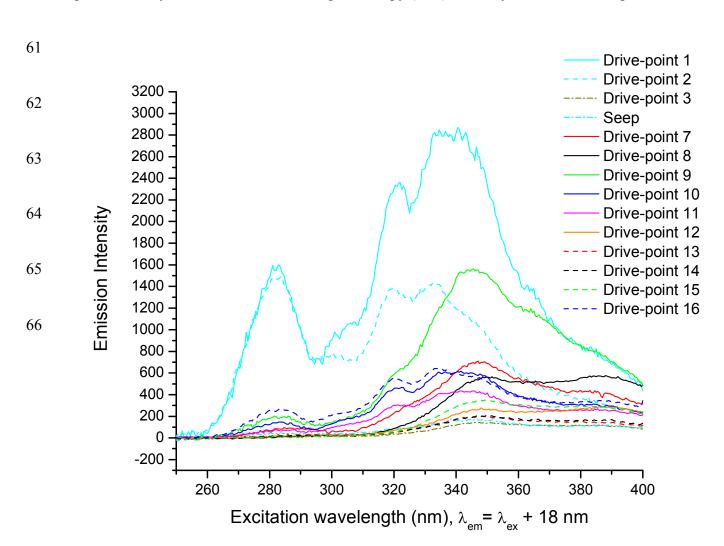
Specific criteria for boron (B) have not been suggested; however, (MacKinnon et al.,
2005) reported elevated levels, around 2.5 mg/L, in OSPW, compared to surface aquifer
groundwater at around 0.12 mg/L. Our two OSPW samples had B at 2.3 and 3.2 mg/L, similar to
that reported above for OSPW. However, several of the natural groundwaters were at similar or
greater (up to 4 mg/L; Table 1) levels.

Finally, ammonium concentrations in OSPW have been reported at ~ 10 mg/L
(MacKinnon et al., 2005) and > 15 mg/L (Hunter, 2001), with natural groundwater having lower
values. Ammonium for our 2 OSPW samples ranged broadly between 1.3-28.4 mg/L. Nearly all

- 55 natural groundwater samples had levels below 3 mg/L, with one exception, at 16 mg/L (Table 1).
- 56 These findings indicate ammonium also cannot be applied singly to distinguish natural from
- 57 OSPW-affected.

58

59 SI Figure 1



60 Spectra from synchronous fluorescence spectroscopy (SFS) for analyzed Far-field samples.

References

68	Baker, K.M., 1999. Identification of process water in a surficial aquifer at Syncrude's Mildred
69	Lake Site. BSc Thesis. Department of Earth Sciences, University of Waterloo. pp. 75.
70	
71	Gibson, J.J., Birks, S.J., Moncur, M., Yi, Y., Tattrie, K., Jasechko, S., Richardson, K., Eby, P.,
72	2011. Isotopic and geochemical tracers for fingerprinting process-affected waters in the
73	oil sands industry: A pilot study. Oil Sands Research and Information Network,
74	University of Alberta, School of Energy and the Environment, Edmonton, AB. OSRIN
75	Report No. TR-12. pp. 109.
76	
77	Hunter, G.P., 2001. Investigation of groundwater flow within an oil sands tailings impoundment
78	and environmental implications. MSc Thesis. Department of Earth Sciences, University
79	of Waterloo. pp. 190.
80	
81	MacKinnon, M.D., Kampala, G., Marsh, B., Fedorak, P.M., Guigard, S., 2005. Indicators for
82	assessing transport of oil sands process-affected waters. In: Thomson, N.R. (Ed.)
83	Bringing groundwater quality research to the watershed (Proceedings of GQ2004
84	International Conference of Groundwater Quality). Vol. 297. IAHS Press, pp. 71-80.
85	
86	
87	