

Supporting Information

for

Trends in the Athermal Entropy of Mixing of Polymer Solutions

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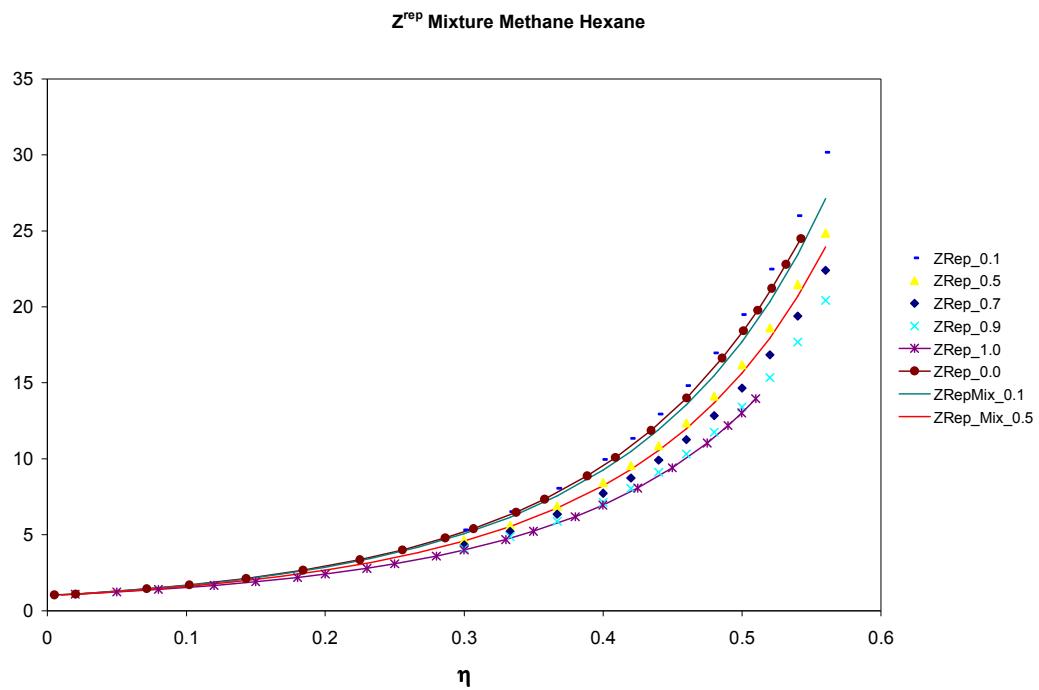


Figure S1. Mixture repulsive compressibility factor for methane + ethane system

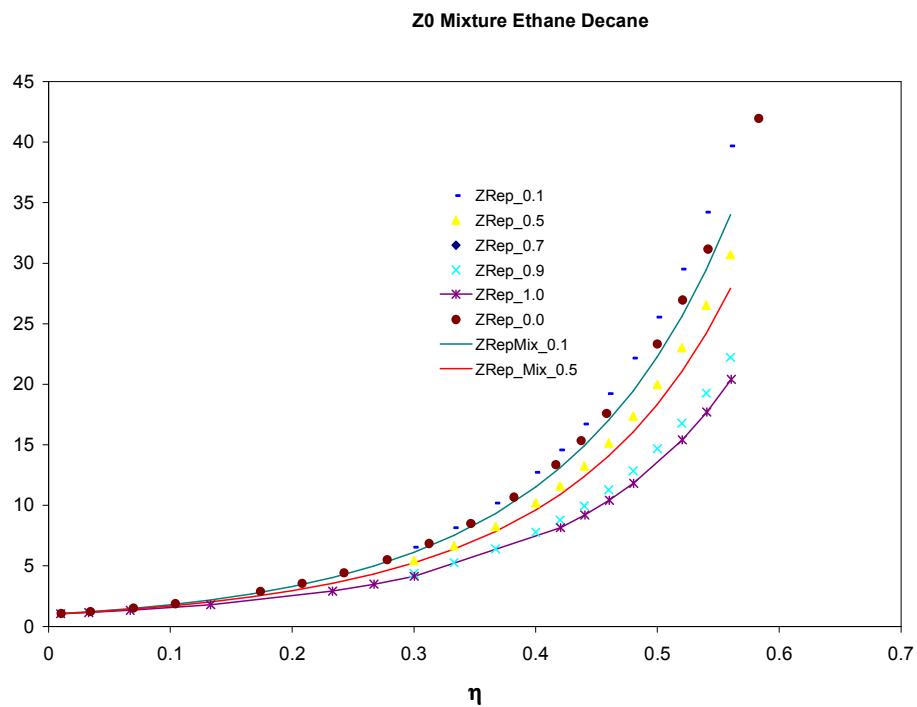


Figure S2. Mixture repulsive compressibility factor for ethane + decane system

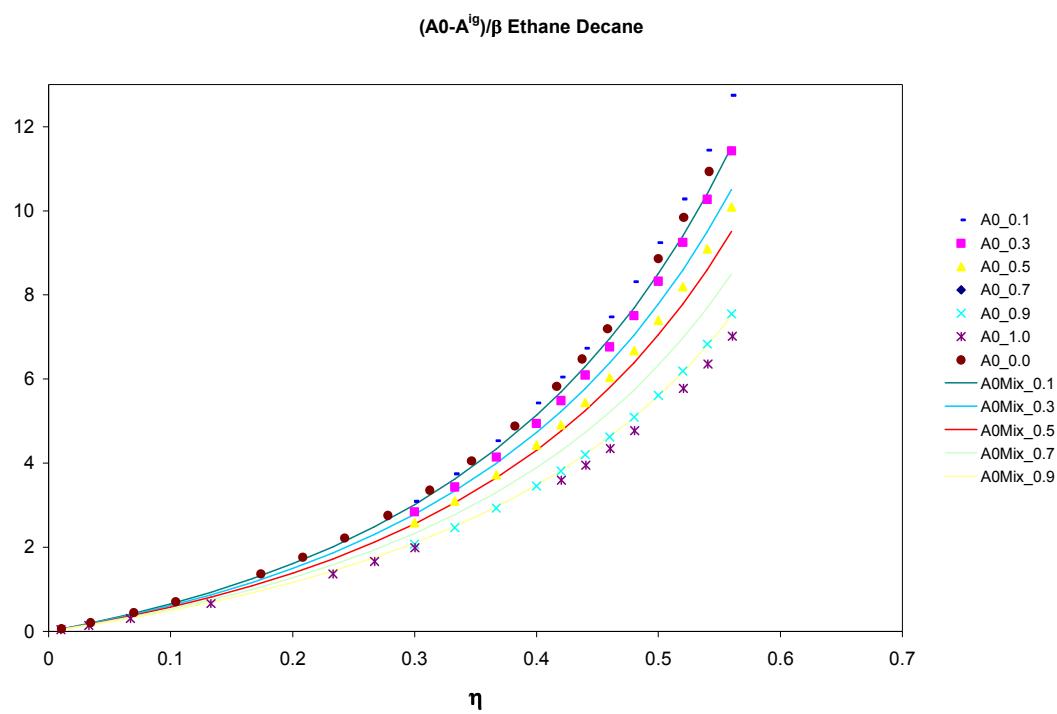


Figure S3. Repulsive Helmholtz free energy for ethane + decane mixture

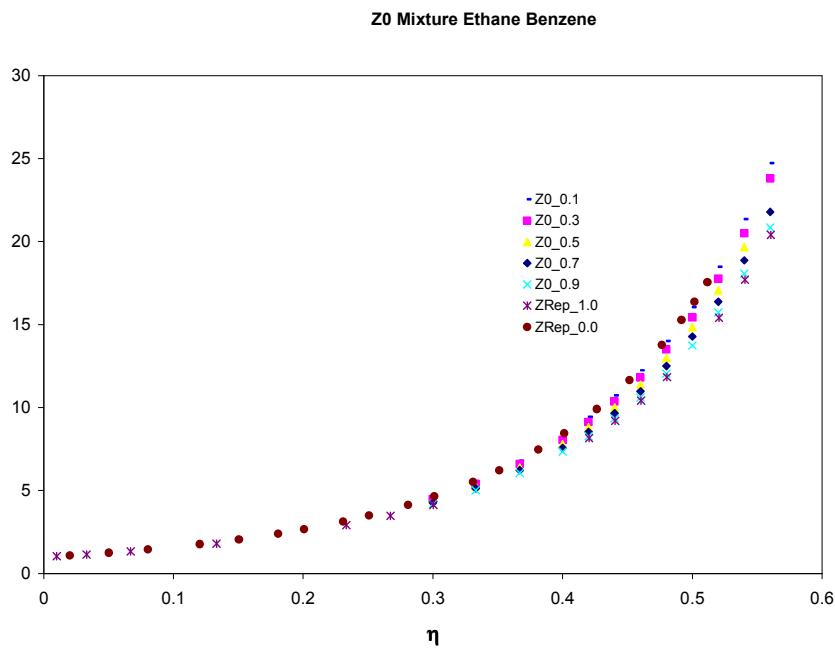


Figure S4. Mixture repulsive compressibility factor for ethane + benzene system

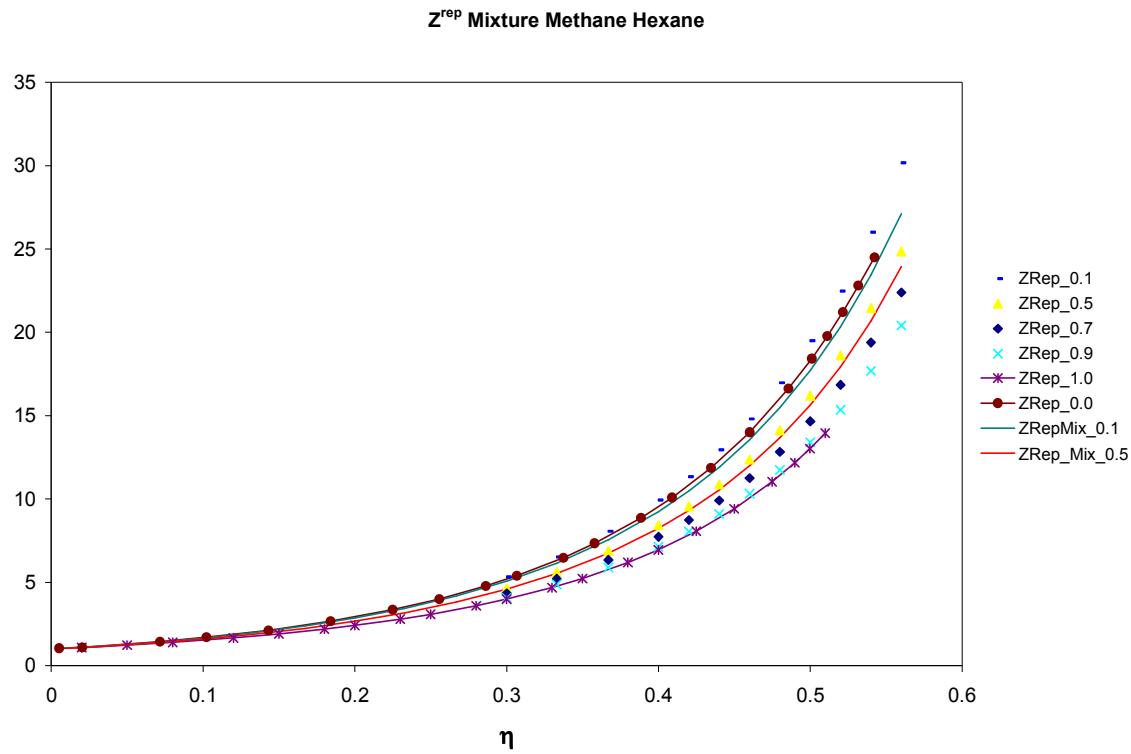


Figure S5. Mixture repulsive compressibility factor for methane +hexane system

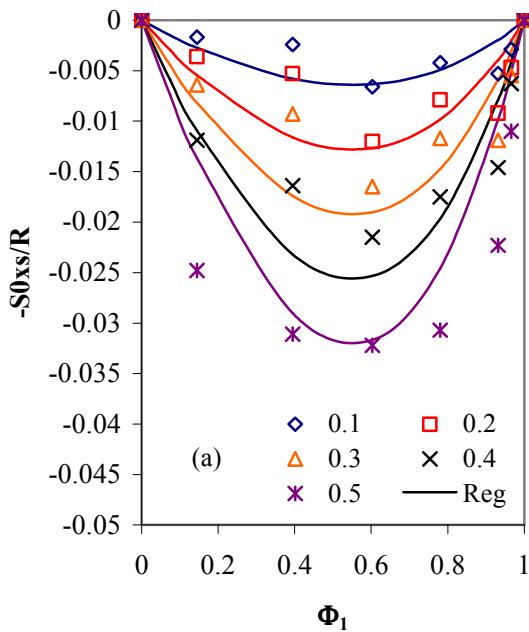


Figure S6. Excess entropy of *trans*-2-butene + ACN from molecular simulations and comparing with vdW-1, $\chi_{ij}^S = -4.77$.

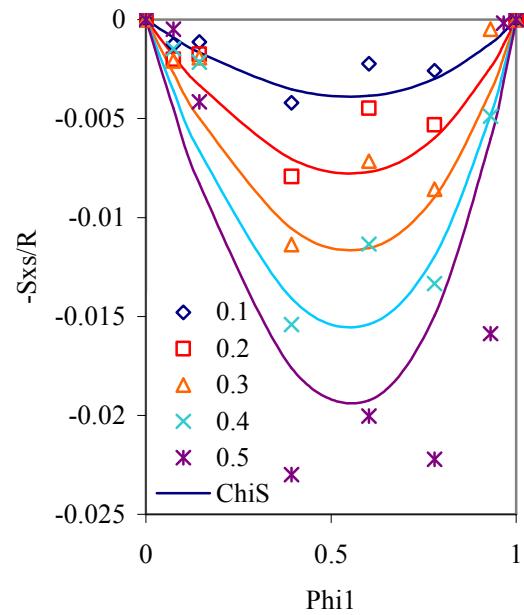


Figure S7. Excess entropy of *cis*-2-butene + ACN from molecular simulations and comparing with vdW-1, $\chi_{ij}^S = 2.90$.

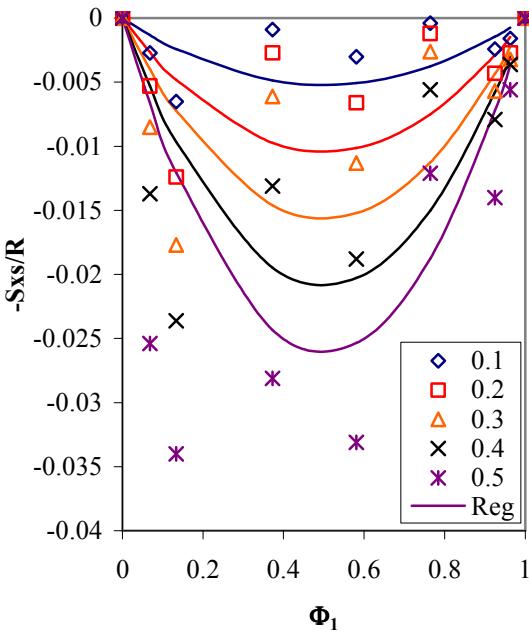


Figure S8. Excess entropy of 1,3-butadiene + ACN from molecular simulations and comparing with vdW-1, $\chi_{ij}^S = -0.21$.

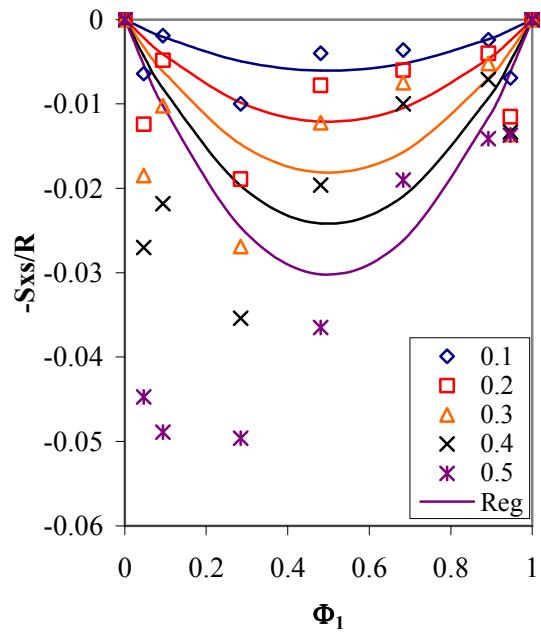


Figure S9. Excess entropy of *trans*-2-butene + DMF from molecular simulations and comparing with vdW-1, $\chi_{ij}^S = -0.24$.

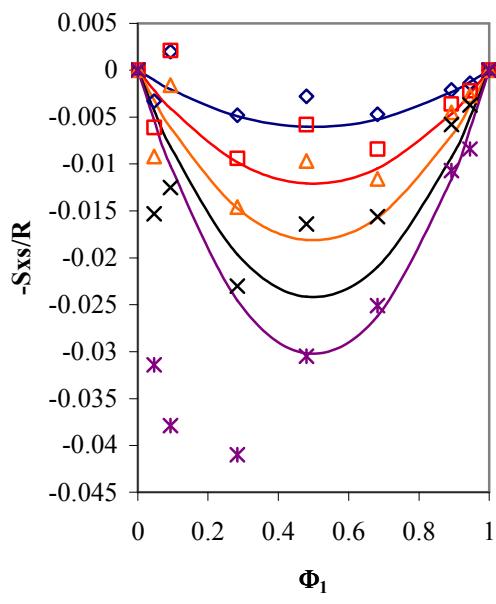


Figure S10. Excess entropy of *cis*-2-butene
+ DMF from molecular simulations and comparing with vdW-1, $\chi_{ij}^S = -0.24$.

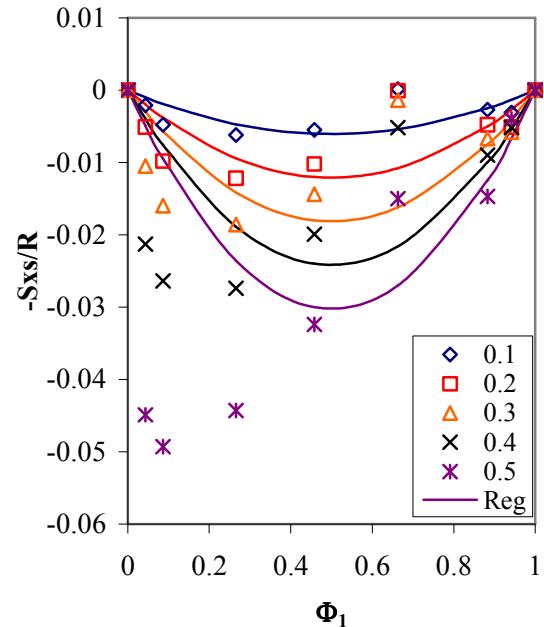


Figure S11. Excess entropy of 1,3-butadiene
+ DMF from molecular simulations and comparing with vdW-1, $\chi_{ij}^S = -0.24$.

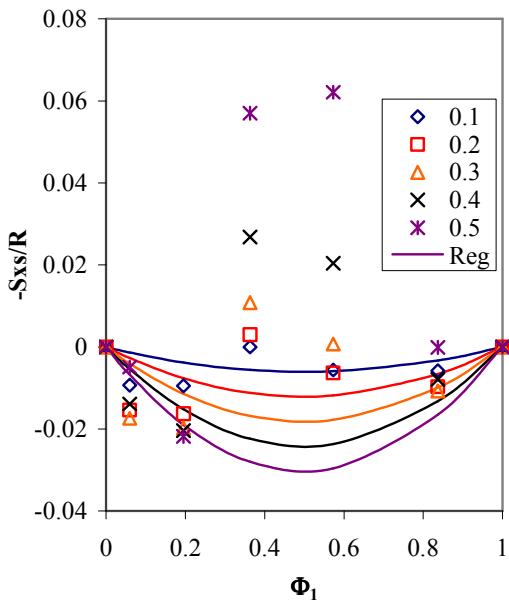


Figure S12. Excess entropy of ethanol
+ R-277 from molecular simulations and comparing with vdW-1, $\chi_{ij}^S = -0.24$.

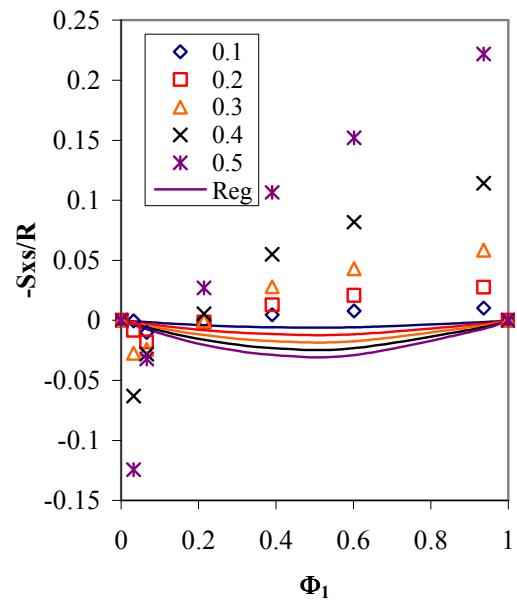


Figure S13. Excess entropy of butane
+ perfluoroC₄ from molecular simulations and comparing with vdW-1, $\chi_{ij}^S = -0.24$.

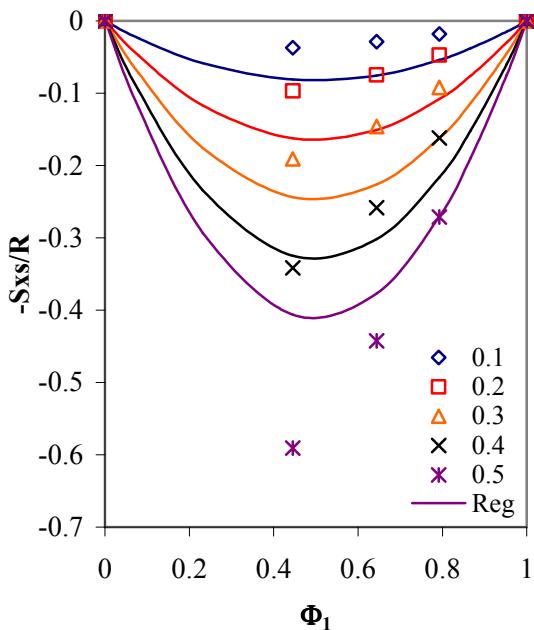


Figure S14. Excess entropy of water + butane from molecular simulations and comparing with vdW-1, $\chi_{ij}^S = -3.29$.

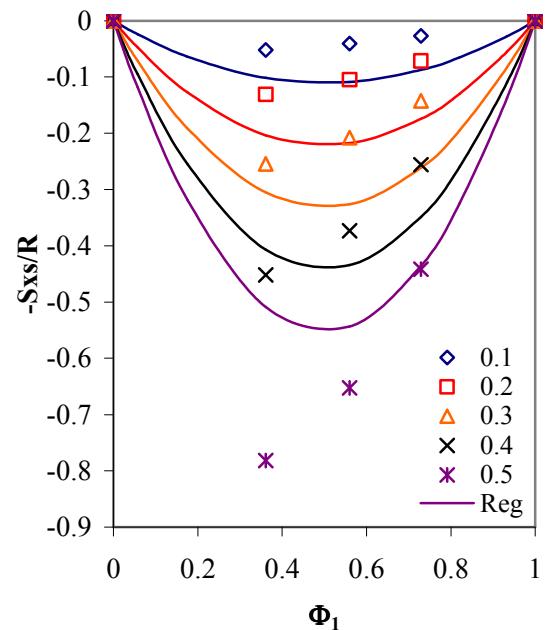


Figure S15. Excess entropy of water + hexane from molecular simulations and comparing with vdW-1, $\chi_{ij}^S = -4.41$.

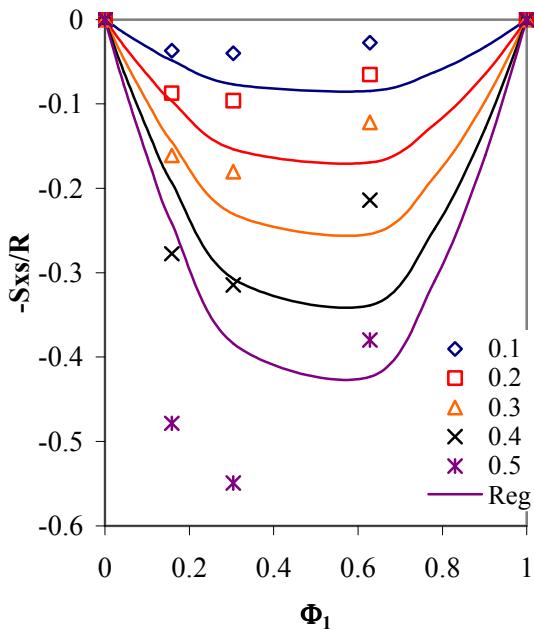


Figure S16. Excess entropy of water + butanol from molecular simulations and comparing with vdW-1, $\chi_{ij}^S = -3.62$.

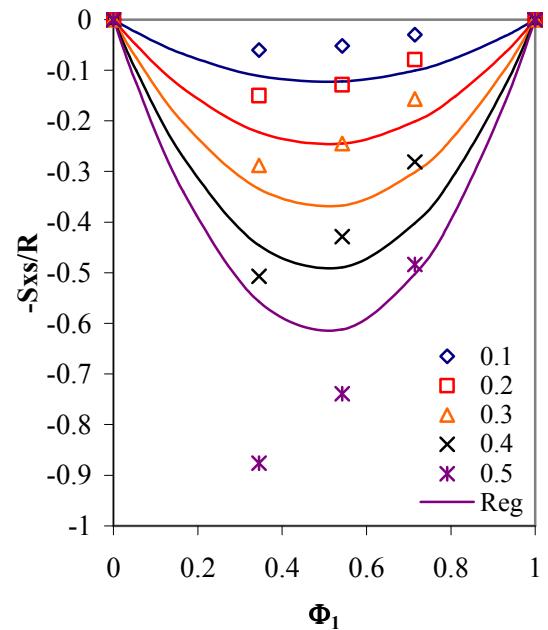


Figure S17. Excess entropy of water + 3MeC₅OH from molecular simulations and comparing with vdW-1, $\chi_{ij}^S = -4.93$.

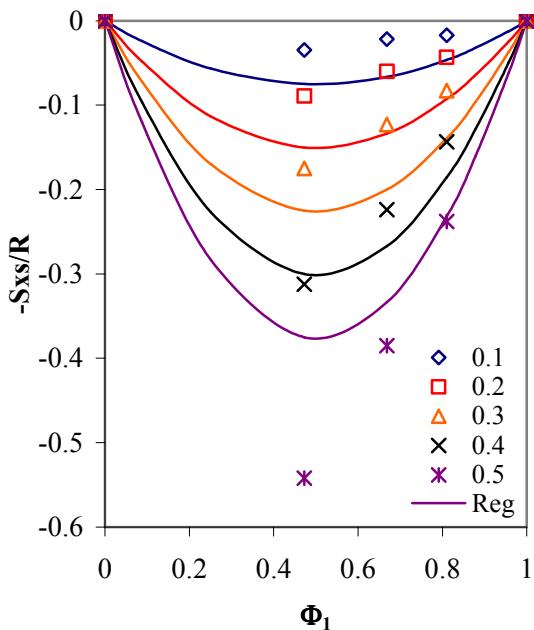


Figure S18. Excess entropy of water + C₃Amine from molecular simulations and comparing with vdW-1, $\chi_{ij}^S = -3.01$.

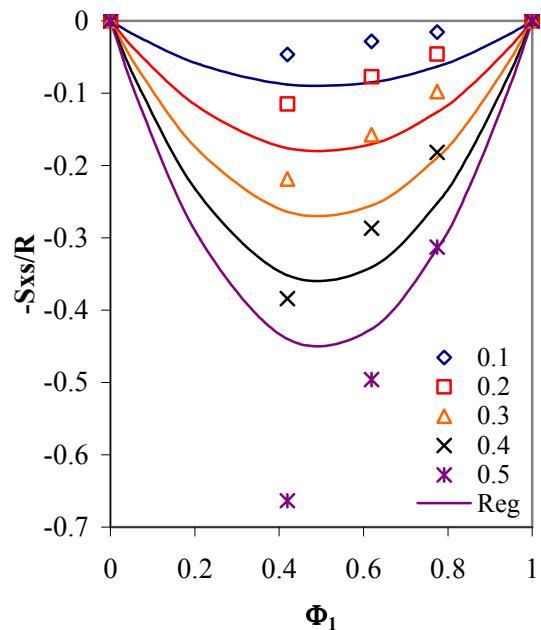


Figure S19. Excess entropy of water + C₄Amine from molecular simulations and comparing with vdW-1, $\chi_{ij}^S = -3.61$.

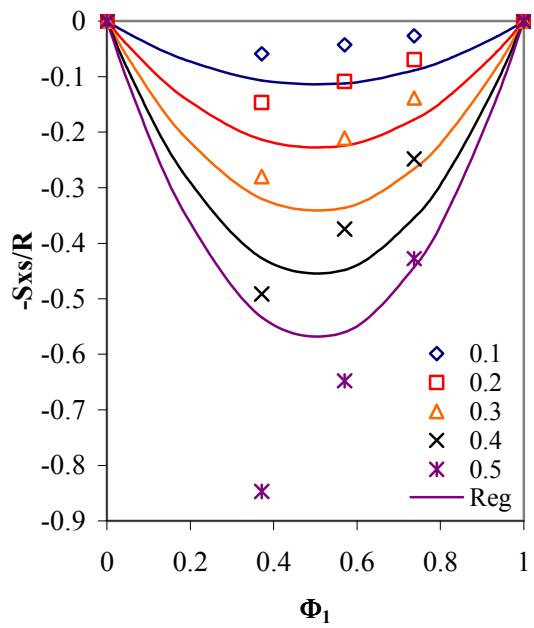


Figure S20. Excess entropy of water + 3C₅Amine from molecular simulations and comparing with vdW-1, $\chi_{ij}^S = -4.57$.

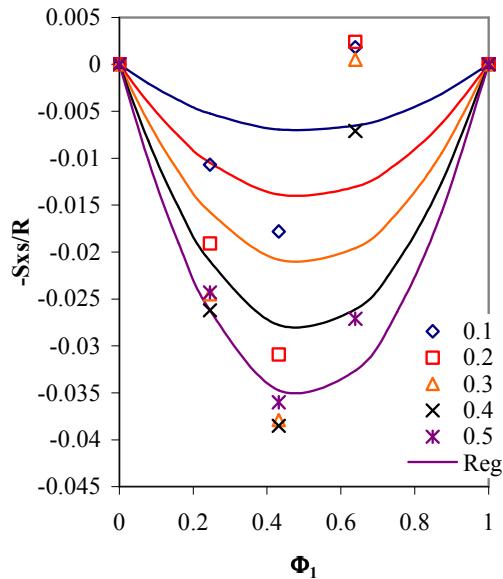


Figure S21. Excess entropy of hexane + butanol from molecular simulations and comparing with vdW-1, $\chi_{ij}^S = -0.28$.

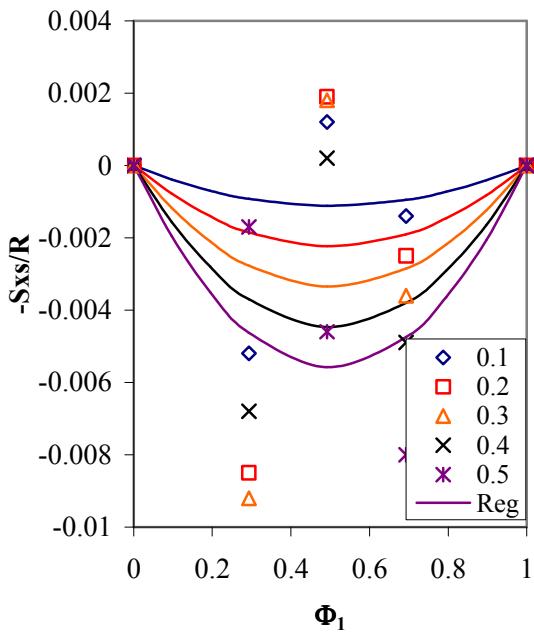


Figure S22. Excess entropy of C₄Amine + butanol from molecular simulations and comparing with vdW-1, $\chi_{ij}^S = -0.04$.

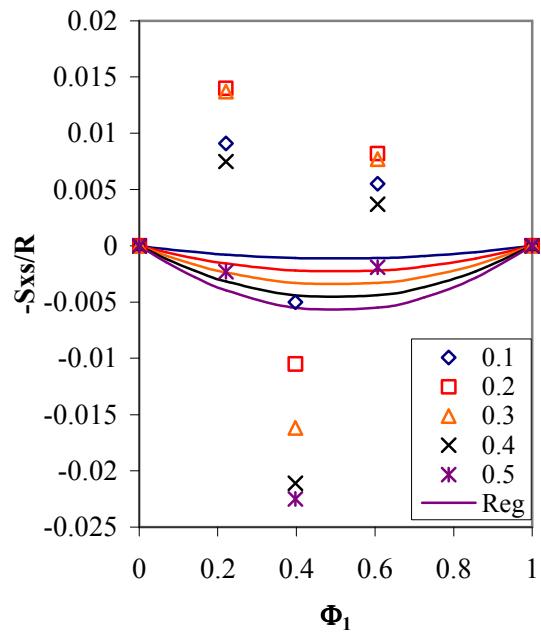


Figure S23. Excess entropy of C₅Amine + propanol from molecular simulations and comparing with vdW-1, $\chi_{ij}^S = -0.05$.

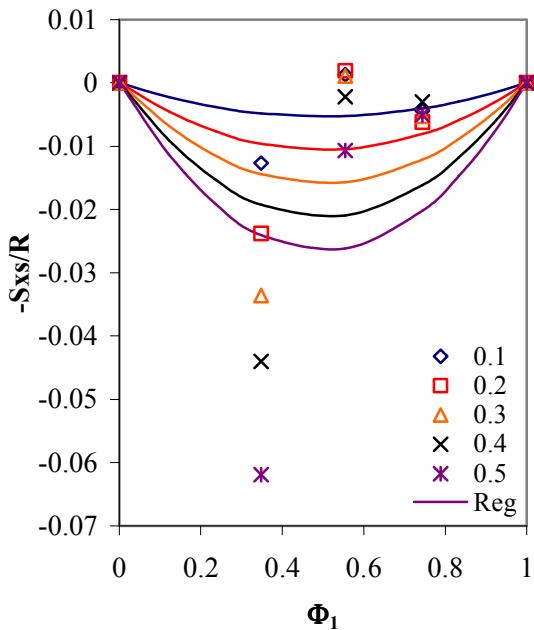


Figure S24. Excess entropy of octanol + 3MeC₅OH from molecular simulations and comparing with vdW-1, $\chi_{ij}^S = -0.21$.