

Gradient Elution Moving Boundary Electrophoresis with Field Amplified Continuous Sample Injection

Alison A. Sikorsky^{1,2}, John T. Fourkas^{1,3}, and David Ross^{2*}*

¹Department of Chemistry & Biochemistry, University of Maryland, College Park, MD 20742, USA

²Material Measurement Laboratory, National Institute of Standards and Technology, Gaithersburg, Maryland 20878, USA

³Institute for Physical Science and Technology, Maryland NanoCenter, and Center for Nanophysics and Advanced Materials, University of Maryland, College Park, MD 20742, USA

*Correspondence to: sikorsky@umd.edu or david.ross@nist.gov

The following pages contain data from repetitions of the experiments presented in the main text, as well as raw data corresponding to all figures included in the main text, and the supporting figures. Calculations and conductivity measurements are also included.

Figure S-1 (Figure 2 repeated):

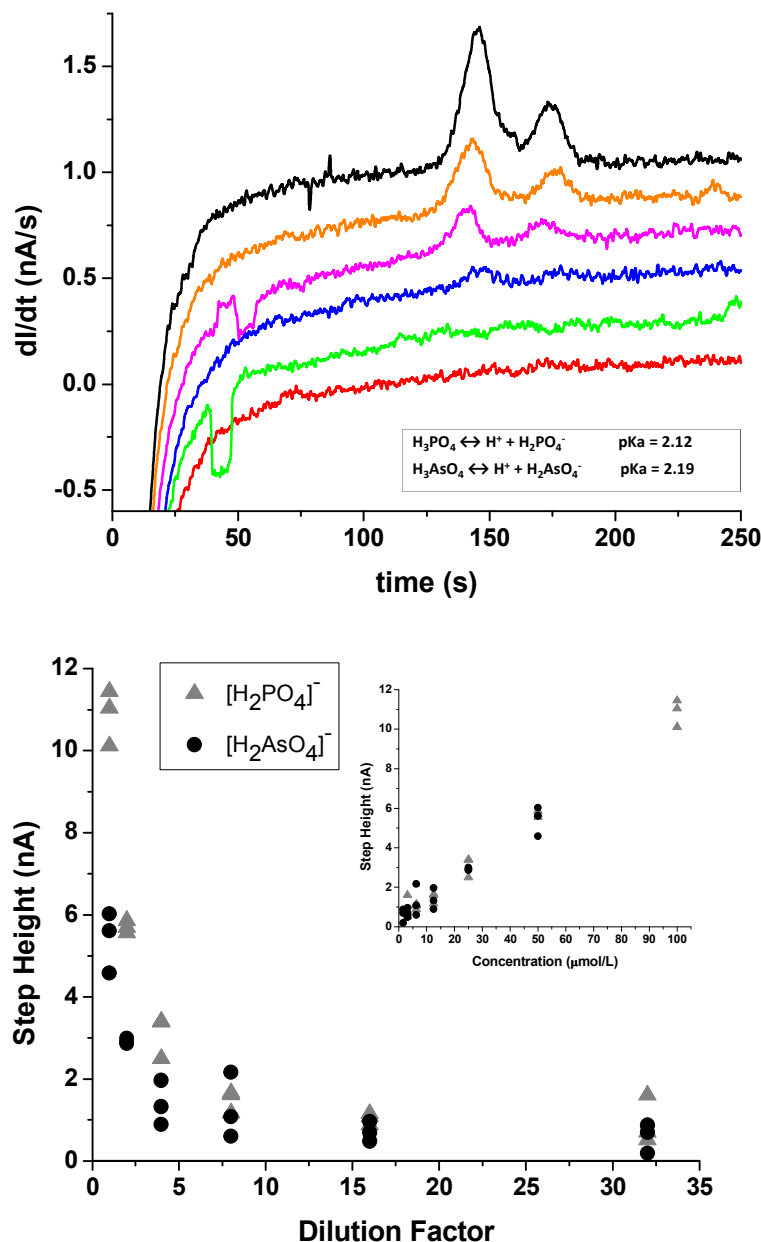


Figure S1. GEMBE data for samples with a serial 2-fold dilution of 100 μM NaH_2PO_4 (left peak), 50 μM Na_2HAsO_4 (right peak). Dilutions were performed with run buffer so that the sample conductivity was the same as the run buffer conductivity. (A) Time derivative of the current vs. time. The dilution factor increases from top to bottom starting with 1 x (100 μM NaH_2PO_4 , 50 μM Na_2HAsO_4), 2 x, 4 x, 8 x, 16 x, and ending with 32 x (3.125 μM NaH_2PO_4 , 1.5625 μM Na_2HAsO_4). (Inset – species of interest). (B) Step height (equivalent to peak area from A) vs. dilution factor. As the samples are progressively diluted and as the analyte concentration decreases, the signal decreases. The LOD for arsenate is approximately 12 $\mu\text{mol/L}$ (signal to noise ratio equal to 3). (Inset – step height from (A) vs. concentration).

Figure S-2 (Figure 4 repeated):

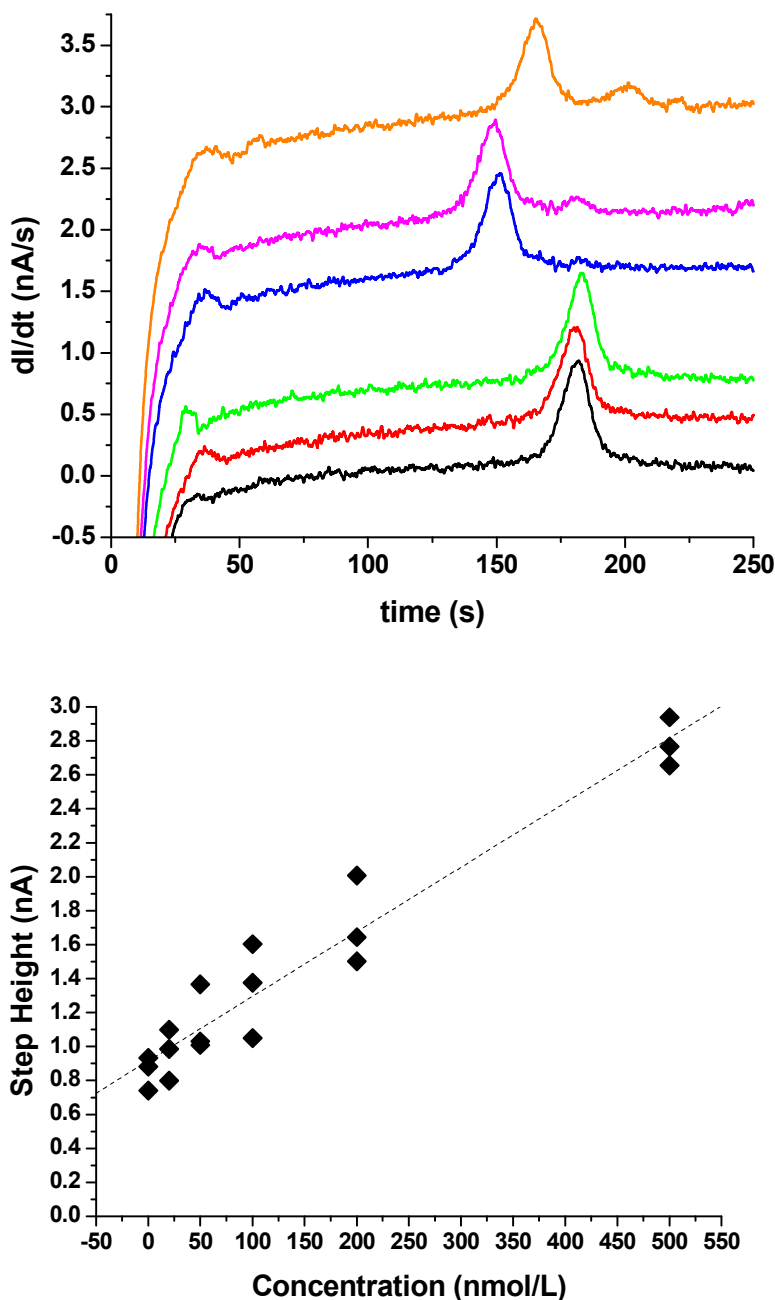


Figure S-2. FACS-GEMBE data with sample buffer diluted 10 \times (with DI water) relative to the run buffer. (A) Time derivative of the current vs. time. Phosphate concentration (left peak) is held constant at 2 μ mol/L and arsenate concentration (right peak) is decreased from top to bottom: 500 nmol/L, 200 nmol/L, 100 nmol/L, 50 nmol/L, 20 nmol/L, 10 nmol/L, 0 nmol/L. (B) Step height (equal to peak area from A) vs. concentration. The LOD for arsenate is reduced approximately 67 fold from 10 μ mol/L (Fig. 2) to 150 nmol/L. The dotted line represents a linear fit to all data points.

Raw Data (no smoothing): [Phosphate step on left, Arsenate step on right]

Figure S-3: Serial 2x Dilutions with Run Buffer – (Fig S-1)

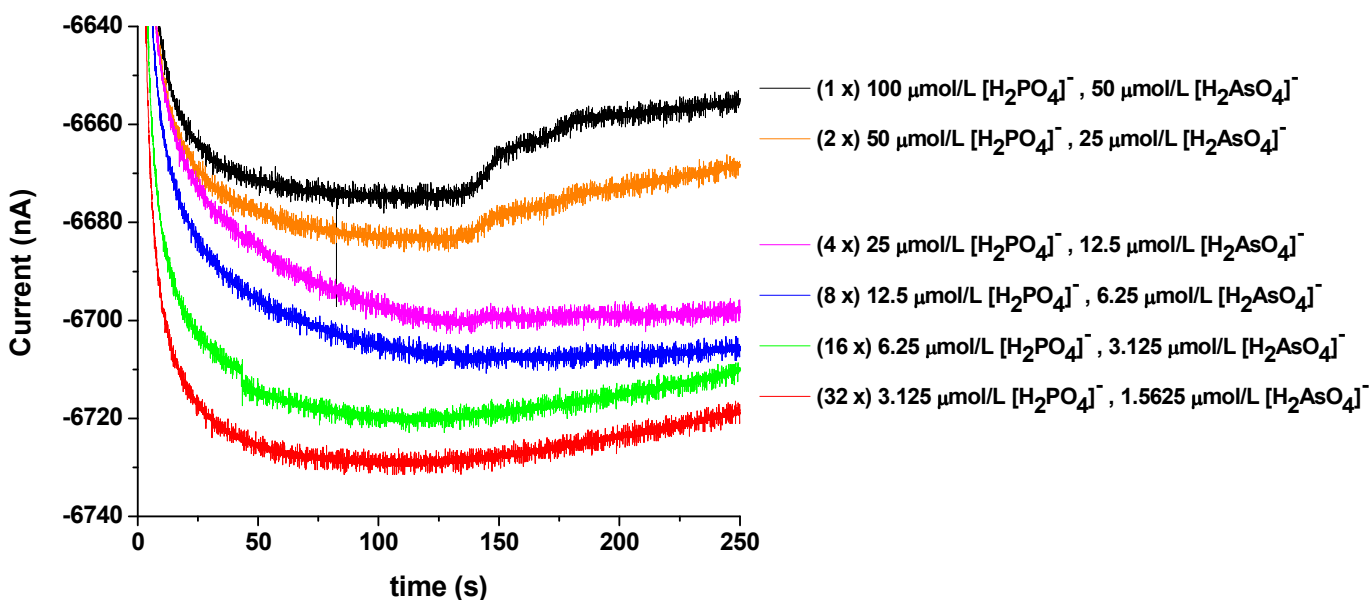


Figure S-4: Serial 2x Dilutions with Run Buffer – (Fig 2 in main text)

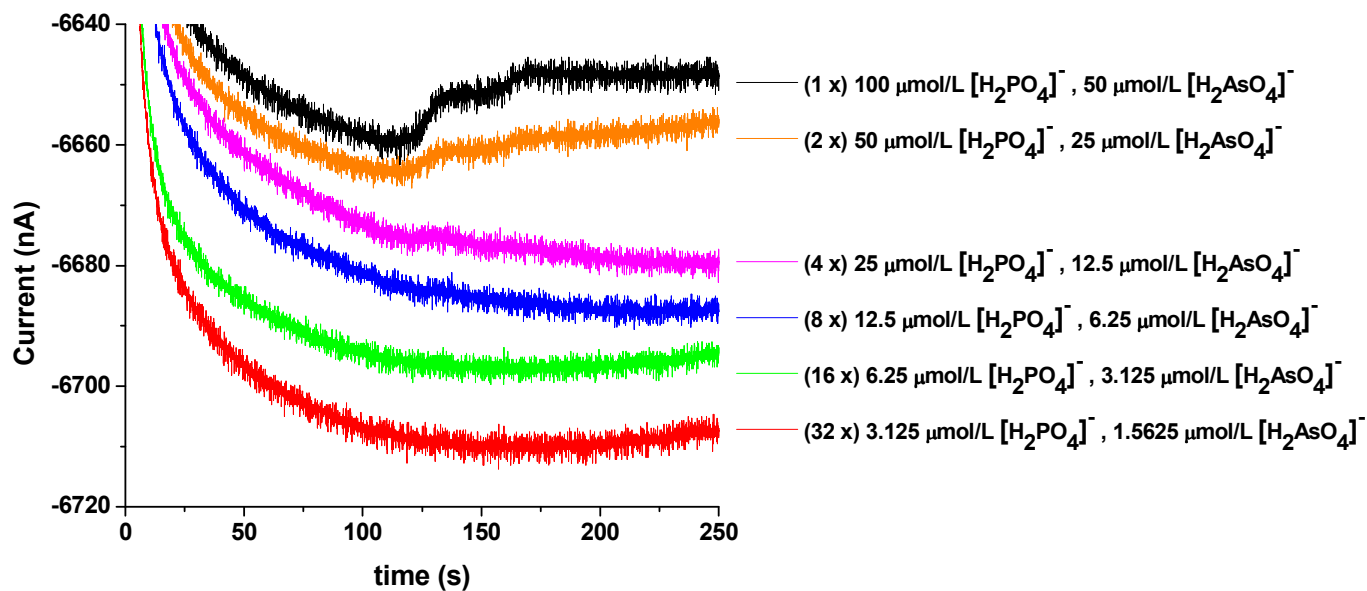


Figure S-5: Serial 2x Dilutions with DI water – (Figure 3 in main text)

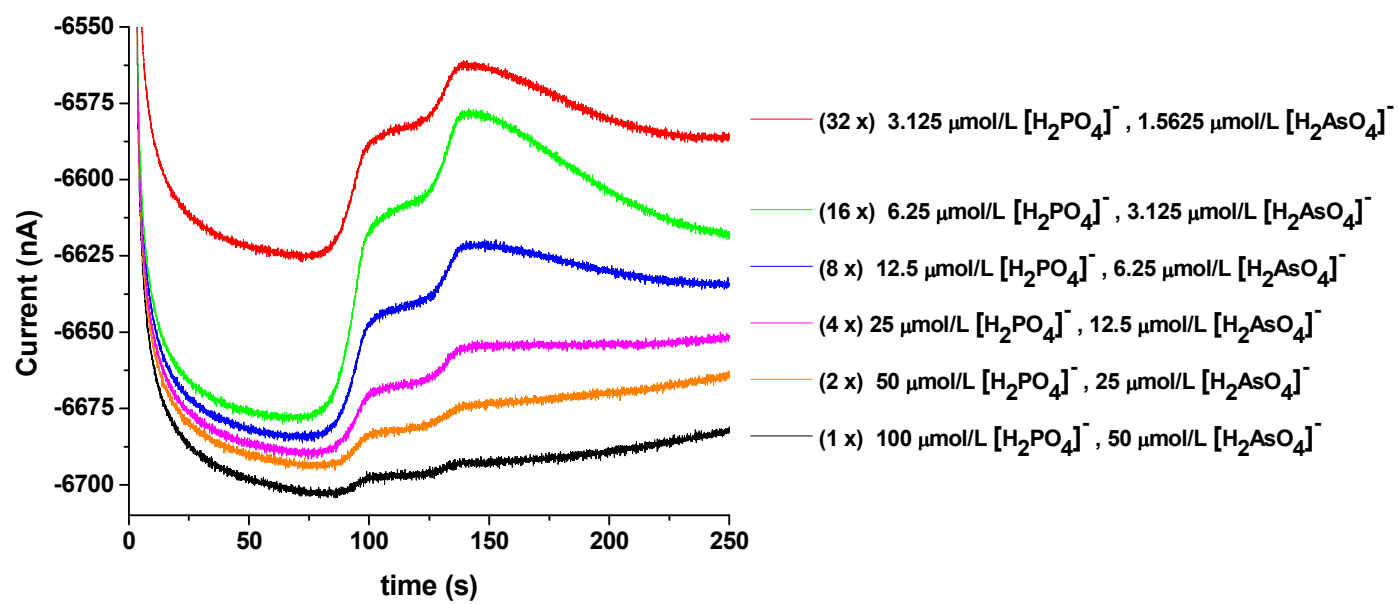


Figure S-6: FACSI-GEMBE with sample buffer diluted 10x – (Figure 4 in main text)

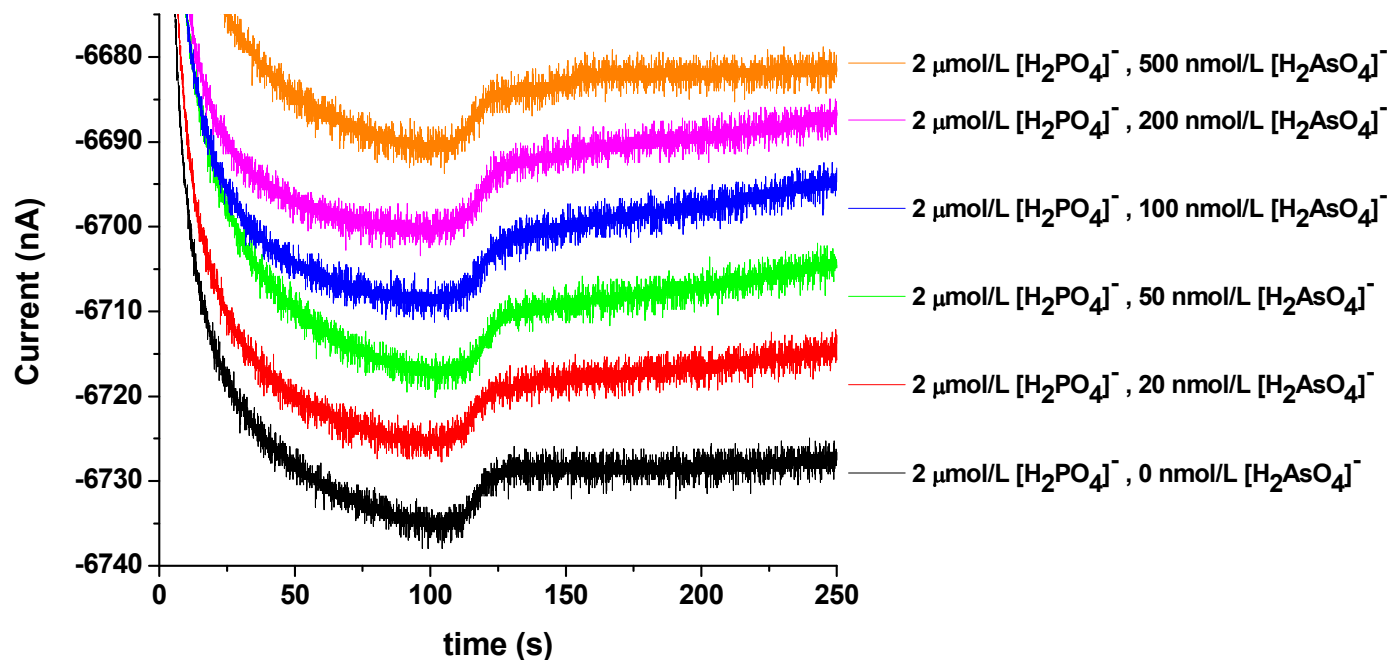


Figure S-7: FACSI-GEMBE with sample buffer diluted 10x – (Figure S-2)

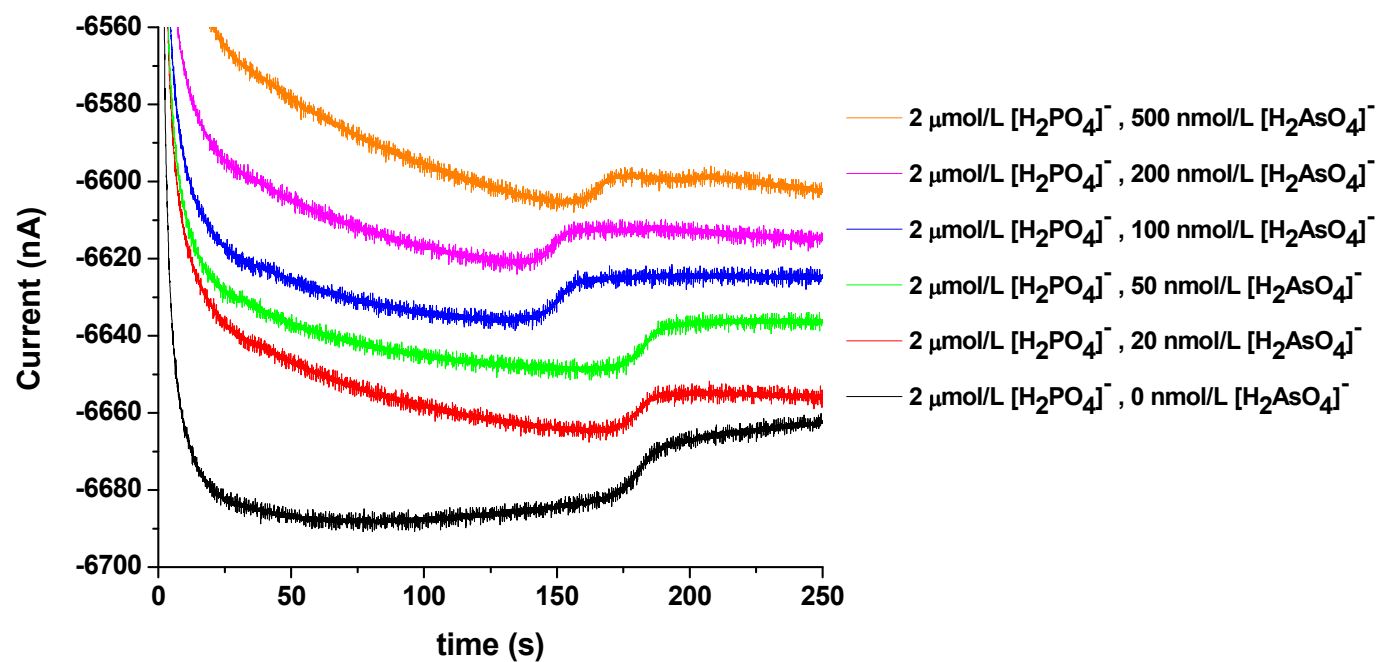


Figure S-8: FACSI-GEMBE of tap water samples – (Figure 5 in main text). The large peak around 150 seconds represents an unknown species in the tap water samples. Phosphate elutes around 250 seconds, followed by arsenate.

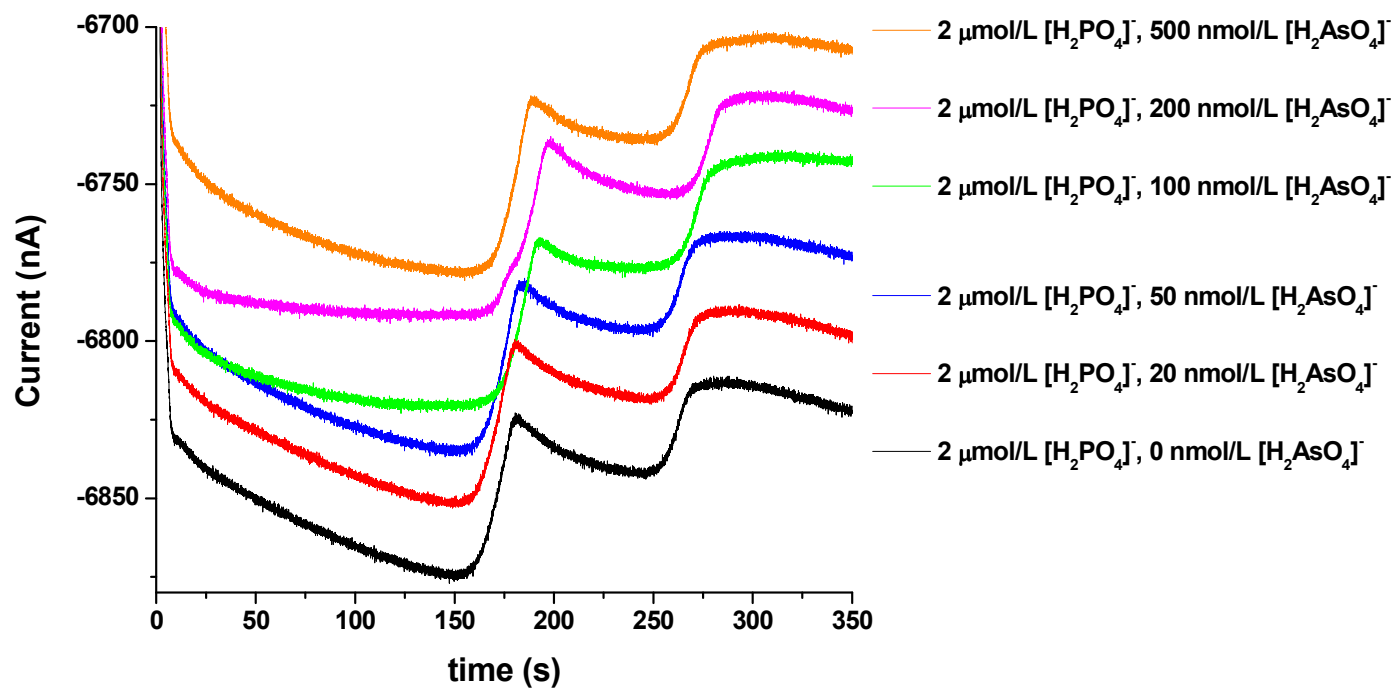


Figure S-9: Tap water – including unknown peak at multiple pHs: FACSII-GEMBE of tap water (sample = 9 parts tap water, 1 part run buffer). Run buffer: 100mM beta-alanine, 80mM HCl, pH = 3.0. The GEMBE data is cut off at 200 seconds to allow visualization of the arsenate shoulder eluting with the unknown peak.

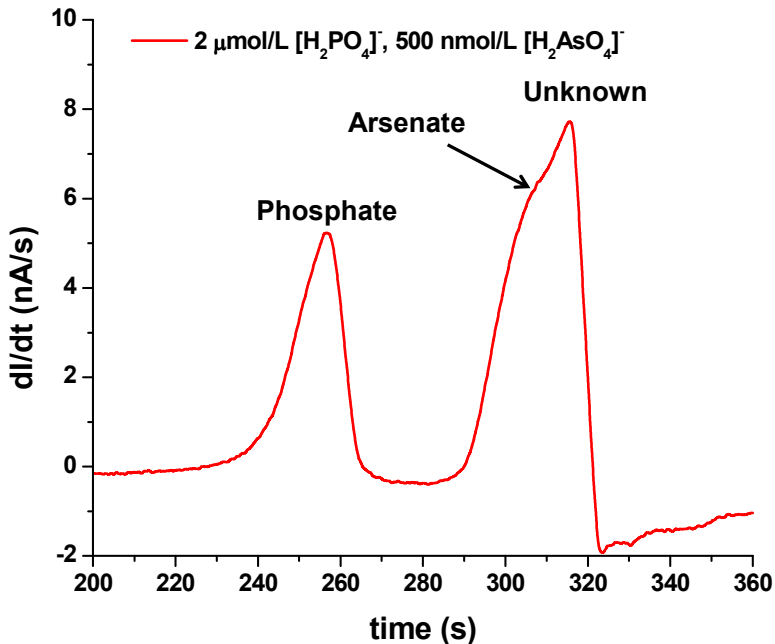
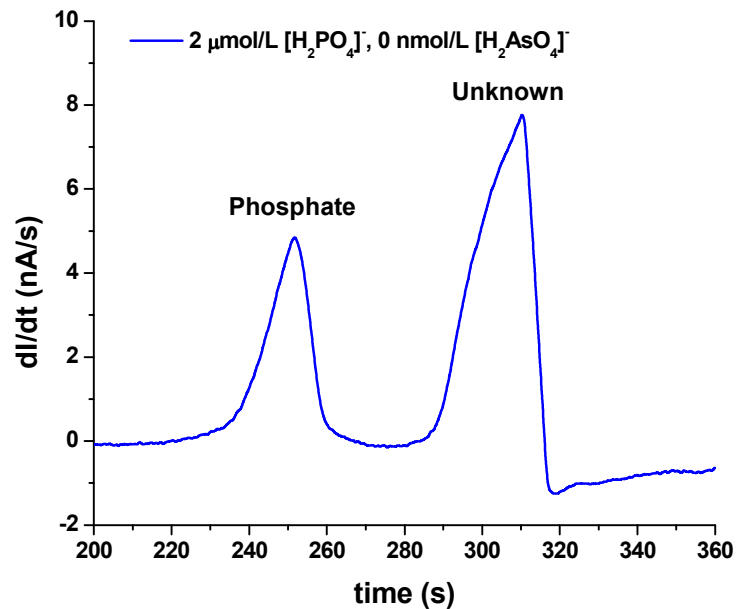
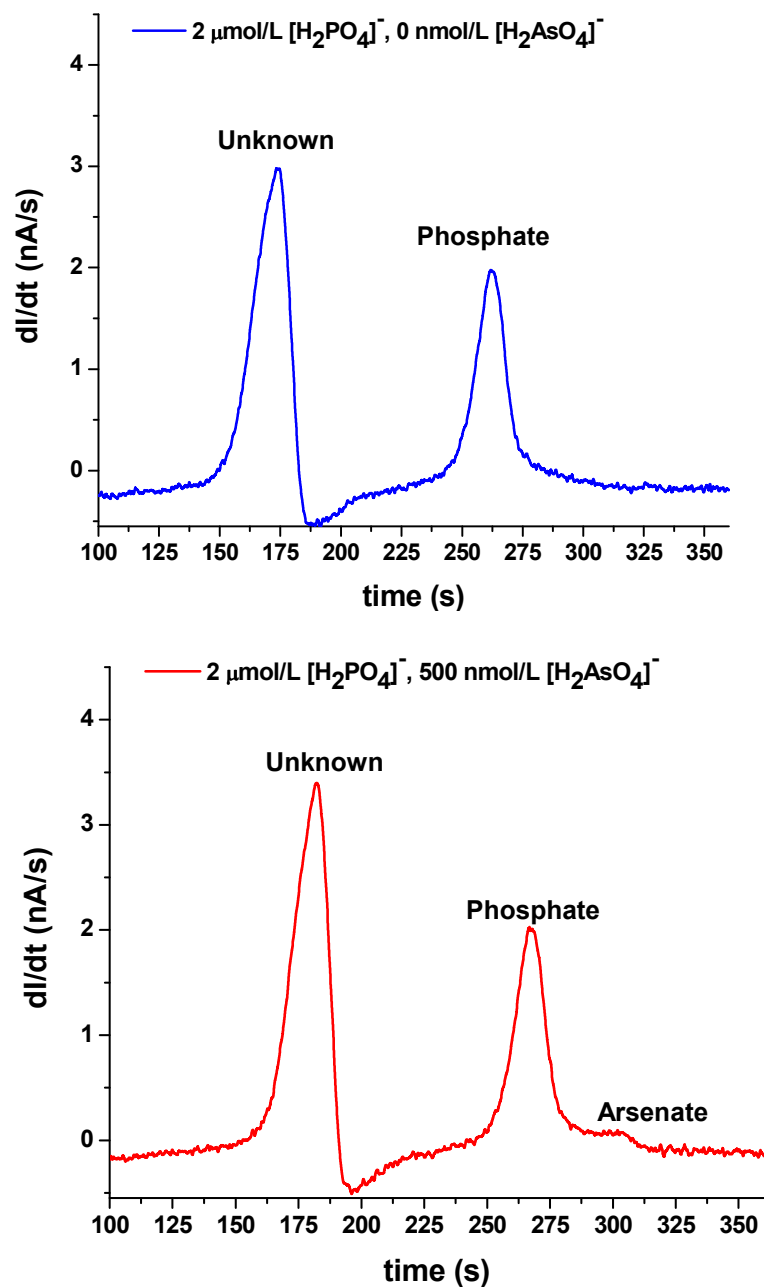


Figure S-10: FACSI-GEMBE of tap water (sample = 9 parts tap water, 1 part run buffer). . Run buffer: 100mM beta-alanine, 70mM HCl, pH = 3.2. The GEMBE data is cut off at 100 seconds to allow visualization of the arsenate peak eluting after the unknown and phosphate peaks.



Calculations

LOD calculation

Figure 2(B): LOD was determined using arsenate data in the concentration range from zero to 50 $\mu\text{mol/L}$. The slope of the fit line was 0.100 nAL/ μmol . The standard deviation (SD) of the fit residuals was 0.300 nA. The LOD was taken to be $3 \times \text{SD/slope}$.

Figure 4(B): LOD was determined using arsenate data in the concentration range from zero to 500 nmol/L. The slope of the fit line was 2.760 nAL/ μmol . The standard deviation (SD) of the fit residuals was 0.132 nA. The LOD was taken to be $3 \times \text{SD/slope}$.

Figure 5: LOD was determined using arsenate data in the concentration range from zero to 500 nmol/L. The slope of the fit line was 2.136 nAL/ μmol . The standard deviation (SD) of the fit residuals was 0.144 nA. The LOD was taken to be $3 \times \text{SD/slope}$.

Comparison of Figure 2 and Figure 3 data: Conductivity Ratio and corresponding estimated increase in signal.

For FACSI-GEMBE with sample buffer diluted (with DI water) 16 x relative to run buffer:

Measured conductivity:

1x run buffer = 6.54 mS/cm (no dilution with DI water)

16x run buffer = 0.796 mS/cm (16 x dilution with DI water)

Calculated conductivity Ratio: $(6.54 \text{ mS/cm}) / (0.796 \text{ mS/cm}) = 8.21$

Step height (phosphate signal) at final concentration of 6.25 $\mu\text{mol/L}$:

With FACSI (sample buffer diluted 16x with DI water) = 75.5795

Without FACSI* (sample buffer equal to run buffer) = $0.1097 \times 6.25 = 0.685625$

*signal extrapolated to low concentration using the slope of the calibration curve (see Figure 2 inset)

Signal enhancement = $75.5795 / 0.685625 = 110.23$

LOD improvement (Run buffer/Tap water comparison)

Measured conductivity:

Run buffer: 6.54 mS/cm

Run buffer diluted 10 x with tap water: 1.1023 mS/cm*

*determined from fit of conductivity data for run buffer diluted with tap water (See supplemental Figure S11)

Calculated conductivity ratio: $(6.54 \text{ mS/cm}) / (1.1023 \text{ mS/cm}) = 5.93$

LOD for arsenate without FACSI (samples prepared in run buffer) = $11.577 \text{ } \mu\text{mol/L}$

(Figure 2 and S1: 8.958 and $14.196 \text{ } \mu\text{mol/L}$)

LOD for arsenate with FACSI (samples = 9 parts tap water, 1 part run buffer) = 202.732 nmol/L
= $0.202732 \text{ } \mu\text{mol/L}$

(Figure 5: 202.732 nmol/L)

LOD improvement: $(11.577 \text{ } \mu\text{mol/L}) / (0.202732 \text{ } \mu\text{mol/L}) = 57.1\text{x}$ improvement

LOD improvement (Run buffer/DI water comparison)

Measured conductivity:

Run buffer: 6.54 mS/cm

Run buffer diluted 10 x with tap water: 1.0036 mS/cm^*

*determined from fit of conductivity data for run buffer diluted with DI water (See supplemental Figure S11)

Calculated conductivity ratio: $(6.54 \text{ mS/cm}) / (1.0036 \text{ mS/cm}) = 6.52$

LOD for arsenate without FACSI (samples prepared in run buffer) = $11.577 \text{ } \mu\text{mol/L}$

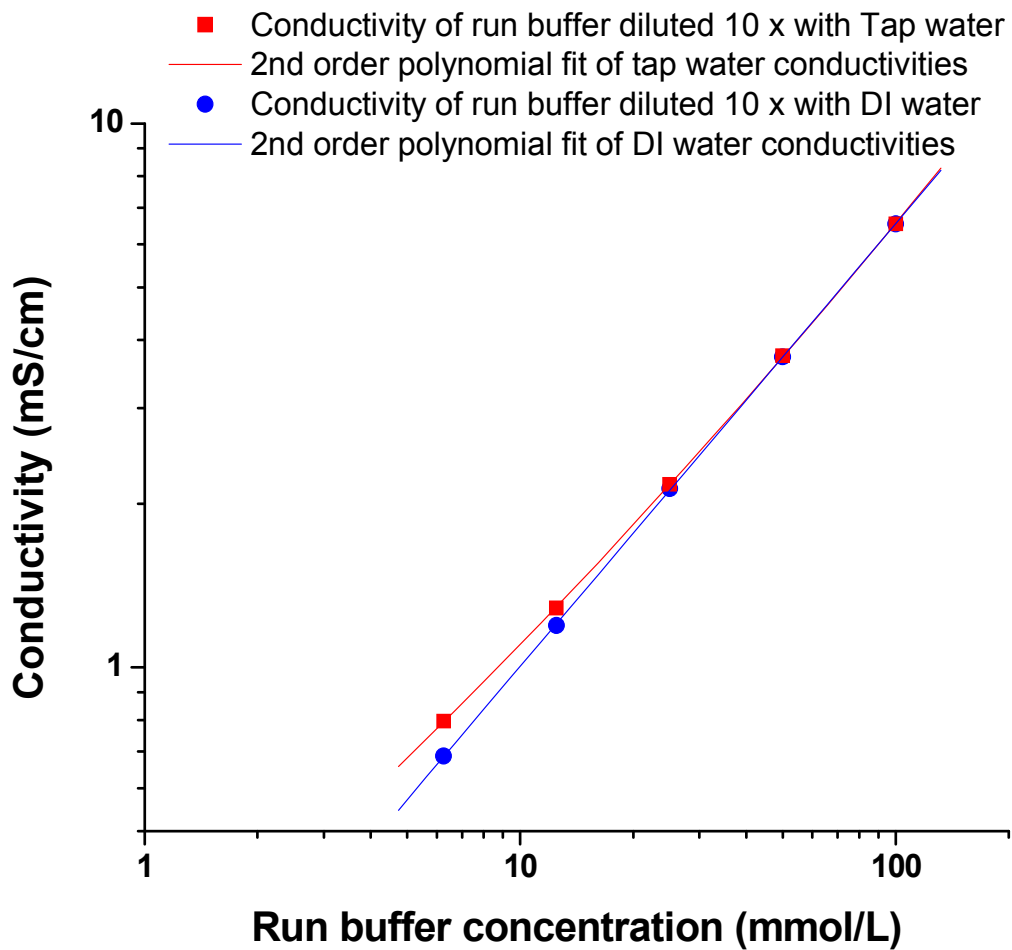
(Figure 2 and S1: 8.958 and $14.196 \text{ } \mu\text{mol/L}$)

LOD for arsenate with FACSI (samples = 9 parts tap water, 1 part run buffer) = 139.992 nmol/L
= $0.139992 \text{ } \mu\text{mol/L}$

(Figure 4 and S2: 143.32 and 136.664 nmol/L)

LOD improvement: $(11.577 \text{ } \mu\text{mol/L}) / (0.139992 \text{ } \mu\text{mol/L}) = 82.7\text{x}$ improvement

Figure S-11: Conductivity of tap water/DI water diluted buffer (pH = 3.0)



DI water fit:

$$\text{Log}(\text{conductivity}) = (8.87846 * 10^{-4} * (\text{Log}(\text{concentration}))^2) + (0.81177 * \text{Log}(\text{concentration})) - (0.81108)$$

Tap water fit:

$$\text{Log}(\text{conductivity}) = (0.06171 * (\text{Log}(\text{concentration}))^2) + (0.58929 * \text{Log}(\text{concentration})) - (0.60862)$$