

Understanding Methane Adsorption in Porous Aromatic Frameworks: An FTIR, Raman, and Theoretical Combined Study

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1. Solid-state ^{13}C MAS NMR
2. Powder XRD
3. TGA curve
5. SEM
6. GCMC adsorption isotherms of CH_4 on PAF-302
7. IR difference spectra of CD_4 adsorbed on PAF-302

Solid-state ^{13}C MAS NMR

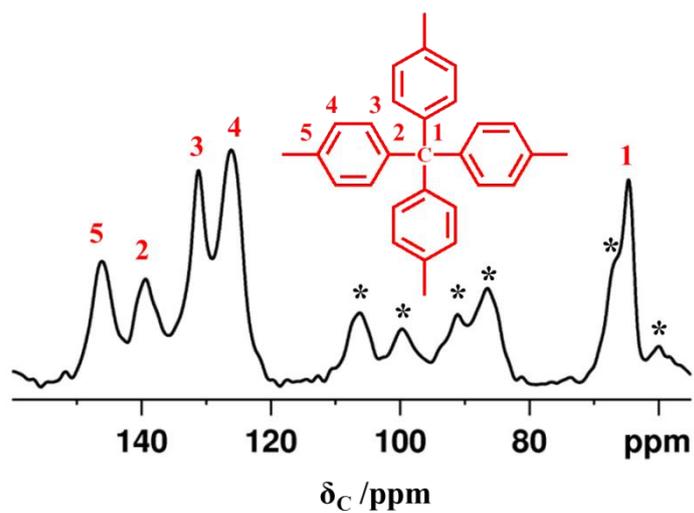


Figure S1 : ^{13}C solid state CPMAS NMR spectra of PAF302. A cross polarization contact time of 5 ms and a MAS spin rate of 5 kHz were used in the experiment. * denotes spinning sidebands.

Powder XRD

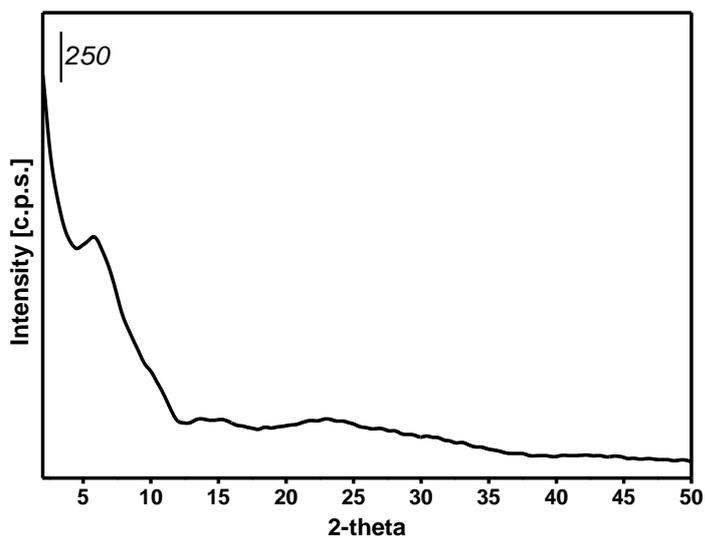


Figure S2 : Powder XRD pattern of PAF-302

Thermogravimetric analysis (TGA)

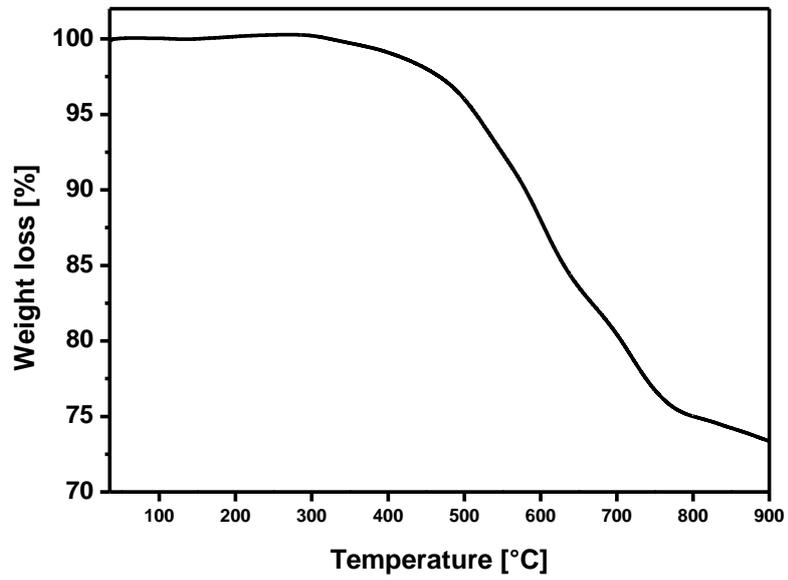


Figure S3 : Thermogram of PAF-302 under argon atmosphere

Scanning electron microscopy (SEM)

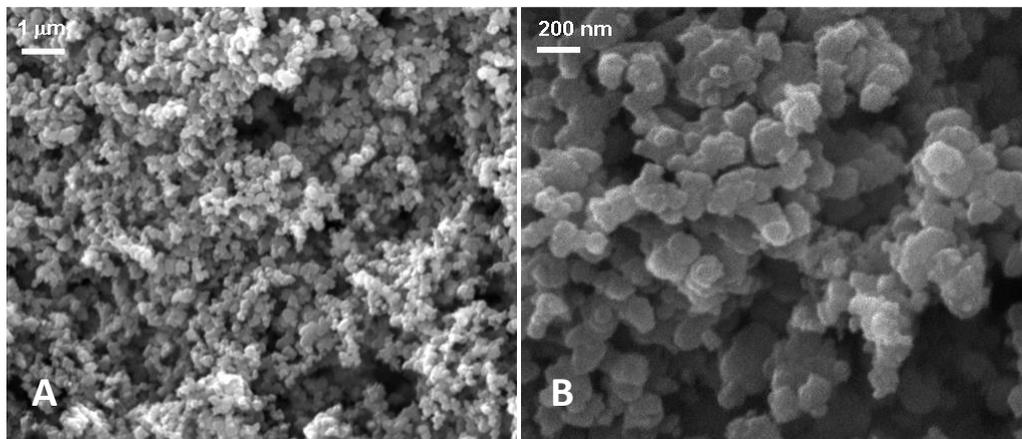


Figure S4 : SEM image of PAF-302 at 25000X A) and 100000X B)

GCMC adsorption isotherms of CH₄ on PAF-302

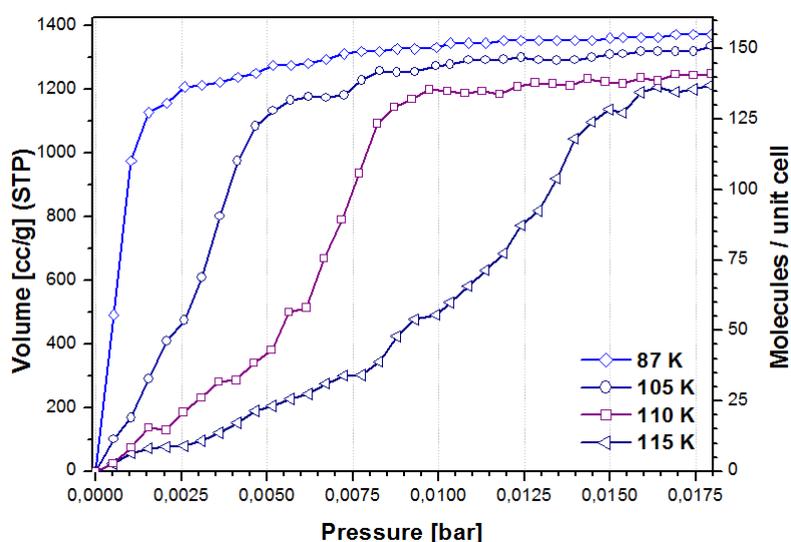


Figure S5 : Adsorption isotherms of CH₄ on PAF-302 at different temperatures: the isotherms were simulated with Grand Canonical Monte Carlo (GCMC) method using a force field specifically designed to study methane adsorption in PAF materials¹⁹

IR difference spectra of CD₄ adsorbed on PAF-302

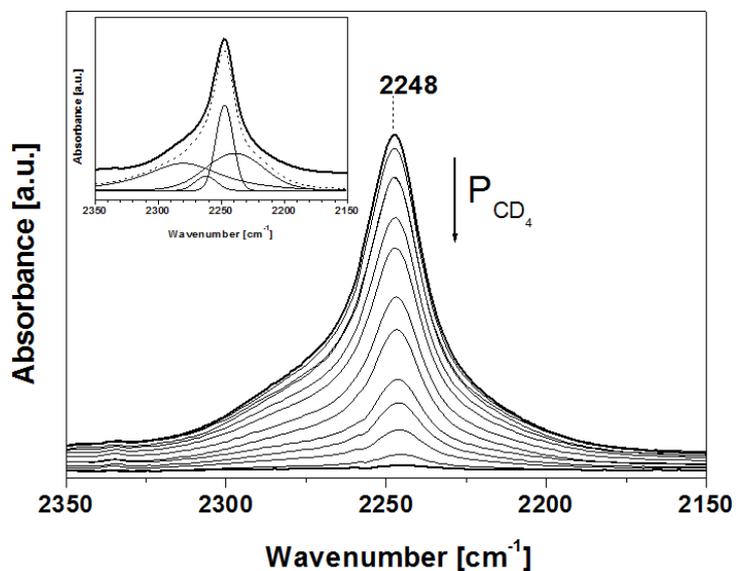


Figure S6 : IR difference spectra in the 2350-2150 cm⁻¹ range of CD₄ adsorbed at 110 K at different pressures (the equilibrium pressure was varied from 8.4, the most intense curve, to 0.21 mbar); The inset shows the spectrum at the highest CD₄ loading (top solid line) fitted by four Gaussian curves. The overall spectrum obtained by adding the four components (dotted line) is reported for comparison.