Supporting Information

Mechanism for Activation of C-CN Bond of Nitriles by a Cationic CpRh(III)-Silyl Complex: A Systematic DFT Study

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1. Additional Computational Results

a. Optimized structures for oxidative addition mechanism



Figure S1. Optimized transition state structure **TSoxd** for oxidative addition mechanism.

b. Relevant intermediates involved in radical mechanisms



Figure S2. Relevant intermediates in radical mechanisms.

CP5 is the critical intermediate involved in SET mechanism which commences with single electron transfer from Rh(III) center to the aryl cyanide ligand. Therefore, **CP5** should be in a triplet state with a Rh(IV) center. Optimization with triplet electron configuration was performed for the location of **CP5**. Intermediate **CP6** with formula **[CpRh(PMe₃)(SiPh₃)(CN)]⁺⁺** is the radical cation involved in cyanide

transfer from the PhCN to the Rh center, resulting in the formation of a Rh(IV) intermediate.

c. Y-shape transition state for the isomerization between Ph-CN and Ph-NC at Rh



Figure S3. Optimized transition state structure $TS_{isomerization}$ connecting cyanide and isocyanide.

d. Reaction coordinates and selected optimized structures for the favored cyano insertion/deisocyanide mechanism in solvent phase (optimizations were done in solvent)



Figure S4. Reaction pathway with optimization in dichloromethane phase.



Figure S5. Selected intermediates and transition state optimized in dichloromethane.

e. Reaction coordinates for the favored cyano insertion/deisocyanide mechanism using B97D functional taking dispersion effect into account



Figure S6. B97D-based reaction pathway in dichloromethane phase.

2. Optimized Cartesian Coordinates for Stationary Points

This information is summarized in a separate .xyz file called "coordinates.xyz",

which can be visualized by molecular modeling packages, such as Mercury.

3. Electronic Energies, Thermal Corrections and Solvation Energies

Table S1. Single Point Electronic Energies, Zero-Point Vibrational Energies, Thermal

Complex	SP (a.u.)	ZPE (a.u.)	ΔH (a.u.)	ΔG (a.u.)	Gsol		
					(kcal/mol)		
CP0	-2074.448389	0.578168	0.616771	0.505942	-12.63		
TS1	-2074.407937	0.577697	0.615188	0.508039	-13.54		
CP1	-2074.457519	0.577904	0.616366	0.503238	-9.95		
CP2	-2074.429268	0.577363	0.615834	0.504093	-13.47		
TS2	-2074.425065	0.57581	0.614132	0.501119	-12.48		
Prod	-2074.46352	0.576542	0.61556	0.500411	-12.22		
TSoxd	-2074.33117	0.576354	0.614608	0.507201	-17.77		
TS1'	-2074.35812	0.576209	0.614372	0.505079	-12.33		
CP3	-2074.394809	0.578098	0.616045	0.507557	-12.96		
CP4	-2074.39409	0.577131	0.615532	0.503768	-13.8		
TS3	-2074.37148	0.575163	0.61349	0.503062	-12.45		
Prod'	-2074.440812	0.576185	0.615508	0.49939	-11.32		
CP5	-2074.3898	0.574897	0.615387	0.493621	-11.17		
phenyl•	-231.622243	0.087653	0.092966	0.060264	-1.8		
CP6	-1842.686054	0.484978	0.518812	0.419258	-19.53		
TS isomerization	-2074.346706	0.574813	0.613385	0.502733	-12.84		
	Substrate effects						
CP0 ^{MeCN}	-1882.662714	0.524875	0.560557	0.457595	-15.02		
TS1 ^{MeCN}	-1882.621098	0.524248	0.558804	0.458874	-17.04		
CP1 ^{MeCN}	-1882.67134	0.524627	0.559978	0.455836	-13.79		
CP2 ^{MeCN}	-1882.643879	0.523397	0.558665	0.454363	-15.06		
TS2 ^{MeCN}	-1882.639736	0.522359	0.557454	0.451912	-15.29		
Prod ^{MeCN}	-1882.683739	0.523054	0.557934	0.453933	-14.55		
CP0 ^{iPrCN}	-1961.313574	0.581794	0.620246	0.508975	-13.04		
TS1 ^{iPrCN}	-1961.270017	0.581326	0.618548	0.512443	-15.01		
CP1 ^{iPrCN}	-1961.317186	0.581645	0.61976	0.508792	-10.83		
TS2 ^{iPrCN}	-1961.273295	0.579442	0.61659	0.508861	-12.91		
CP0 ^{tBuCN}	-2000.638794	0.61006	0.649754	0.537098	-12.26		
TS1 ^{tBuCN}	-2000.593376	0.60926	0.647834	0.53993	-13.35		
CP1 ^{tBuCN}	-2000.639988	0.60979	0.649116	0.535879	-10.43		
TS2 ^{tBuCN}	-2000.594217	0.607549	0.646753	0.533996	-12.39		
CP0 ^{para-OMe}	-2189.00987	0.610756	0.65208	0.53447	-12.46		
TS1 ^{para-OMe}	-2188.969495	0.610386	0.650547	0.53769	-13.61		
CP1 ^{para-OMe}	-2189.01969	0.610669	0.651797	0.532414	-9.67		
CP0 ^{para-CF3}	-2411.589299	0.582628	0.62406	0.506239	-13.03		
TS1 ^{para-CF3}	-2411.549364	0.582219	0.623491	0.505234	-14.14		

Corrections to Enthelpies and Cibbs Free Energies and Solvation Energies in CH	
- לטורפטוטווא וט דאונוזמוטופא מווע לווטטא דרפפ דאופוצופא. מווע סטועמוטוו דאופוצופא ווו לדוא	Cb.ª

CP1 ^{para-CF3}	-2411.599816	0.582595	0.624692	0.502863	-9.72

SP: Single point electronic energy; a. note Δ H: Thermal correction to enthalpy; Δ G: Thermal correction to Gibbs free energy; Gsol: Solvation energy in CH₂Cl₂.

ZPE: Zero-point vibrational energy;

4. Full Citation of Gaussian03 Program

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