

The Effect of Additives on the Zinc Carbenoid-Mediated Cyclopropanation of a Dihydropyrrole

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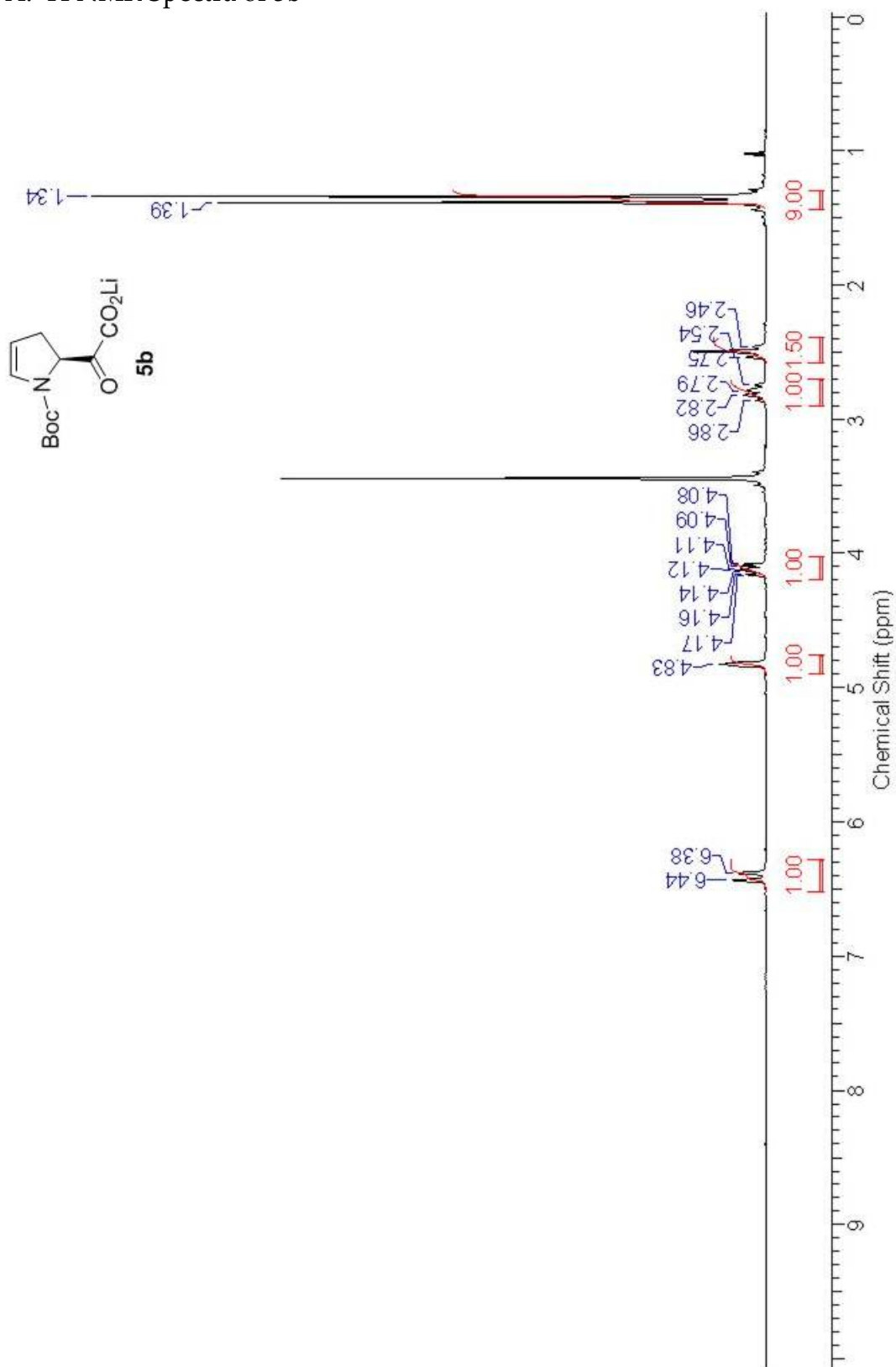
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Supporting Information

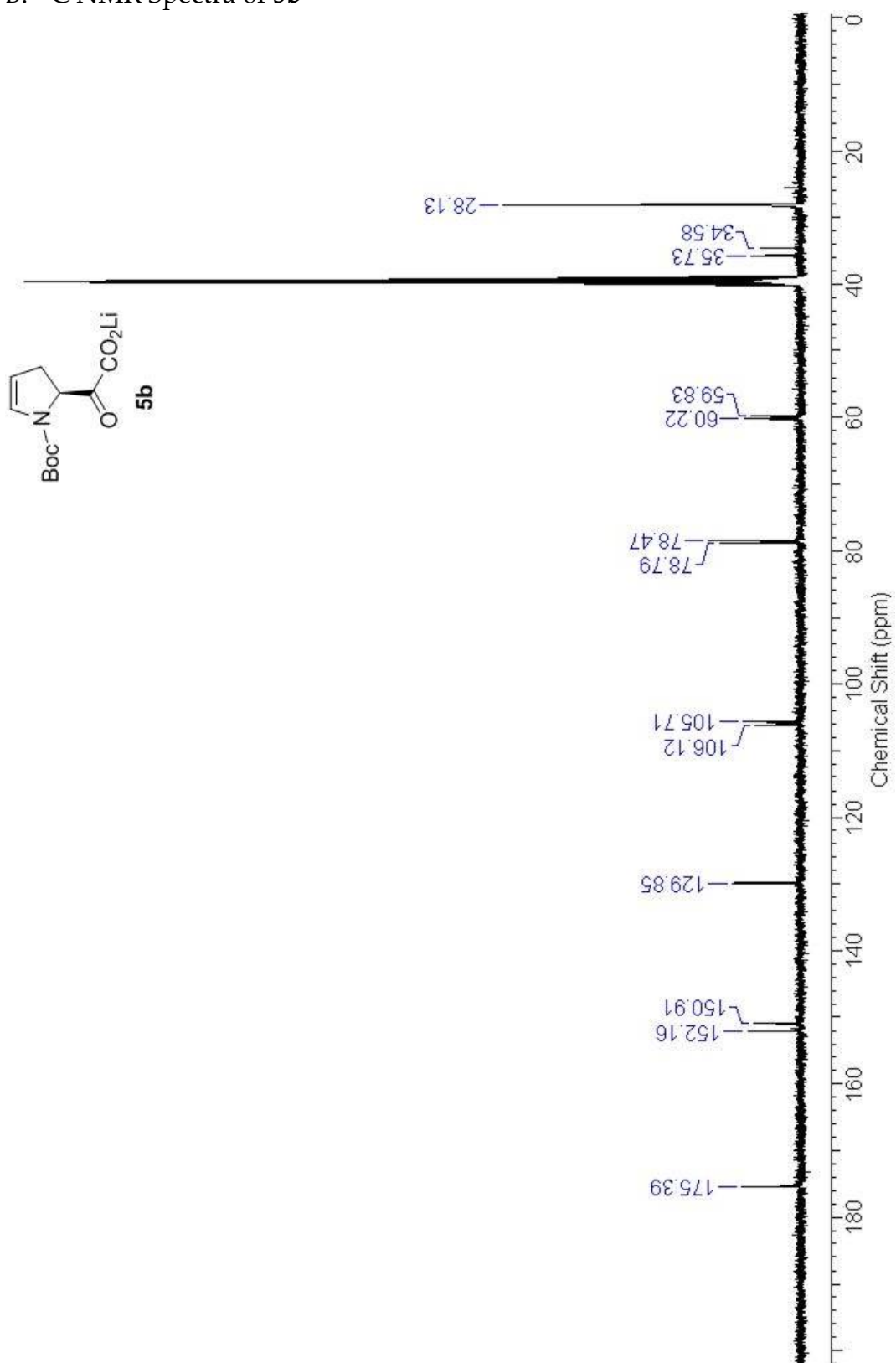
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I. Spectroscopic Data for the Characterization of Compounds 5b and 1a.

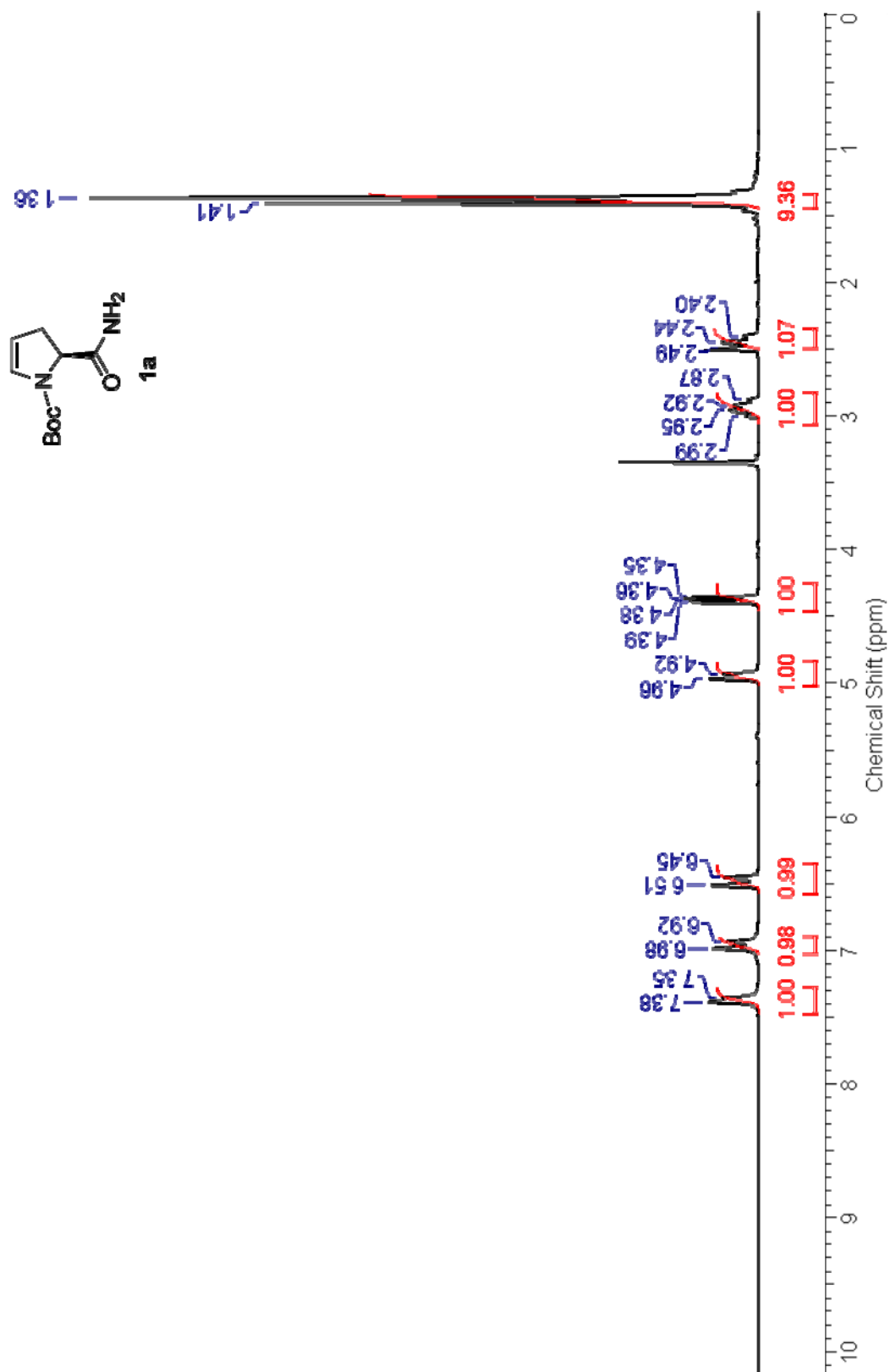
A. ^1H NMR Spectra of 5b



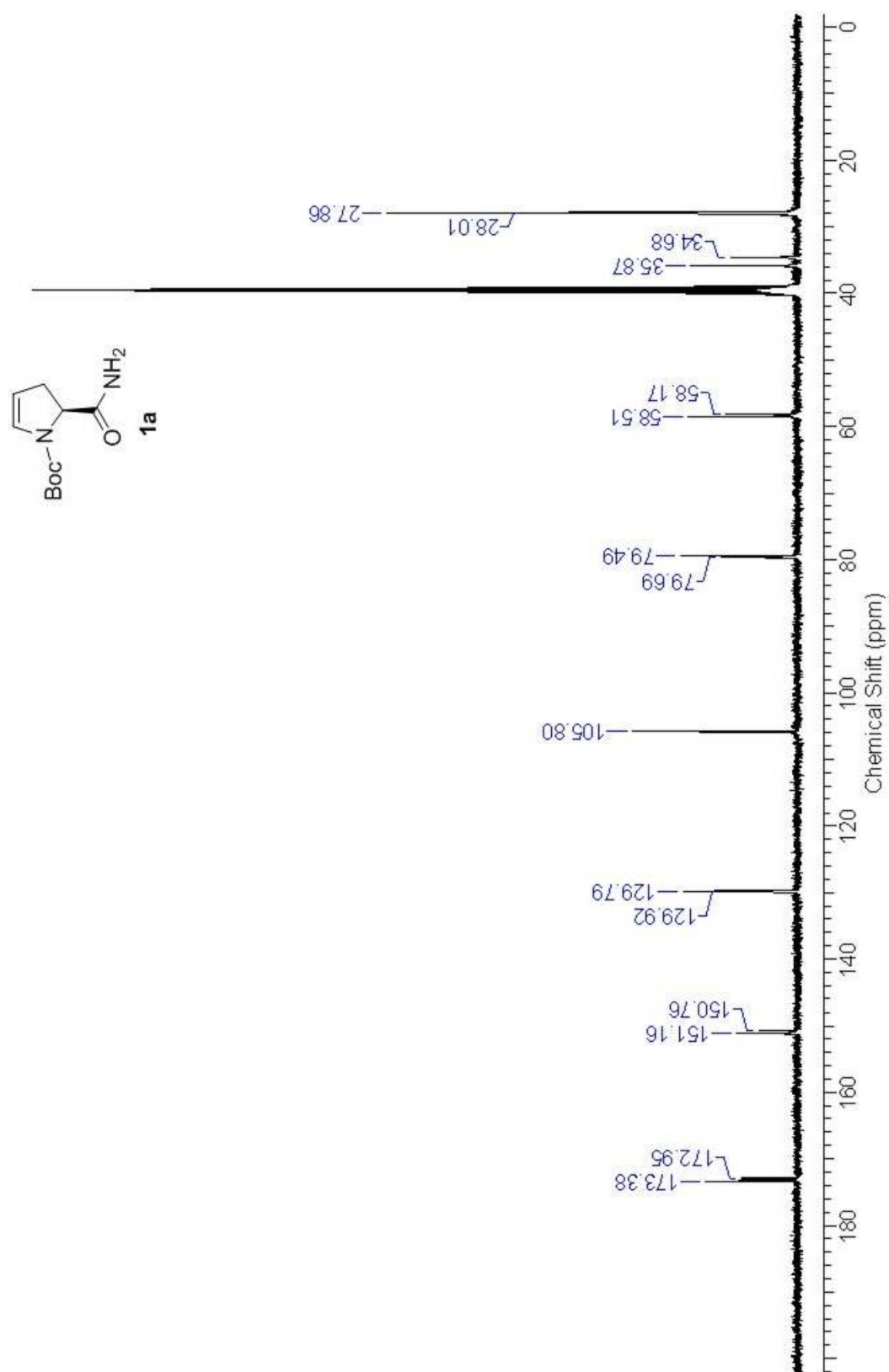
B. ^{13}C NMR Spectra of **5b**



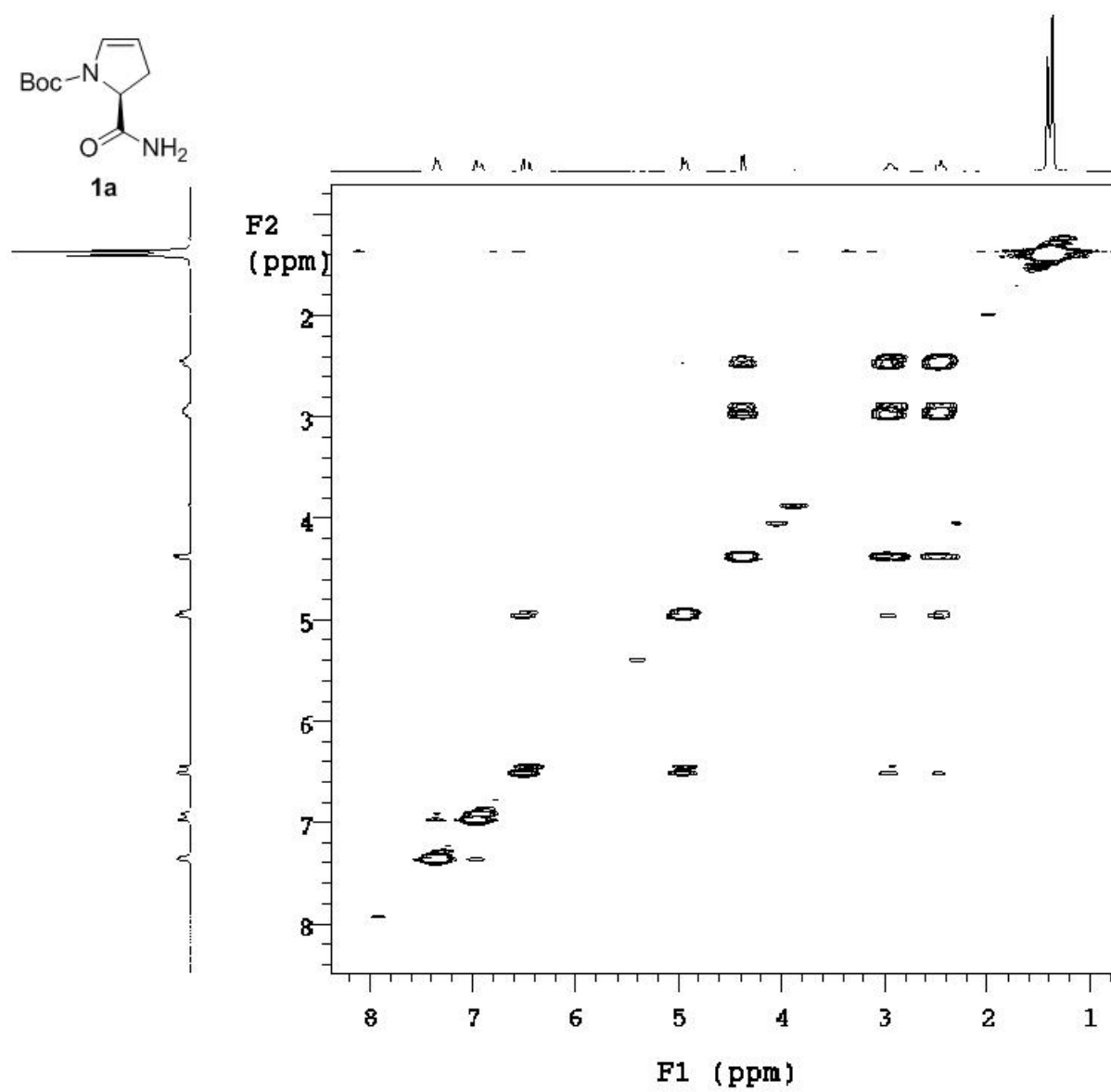
C. ^1H NMR Spectra of **1a**



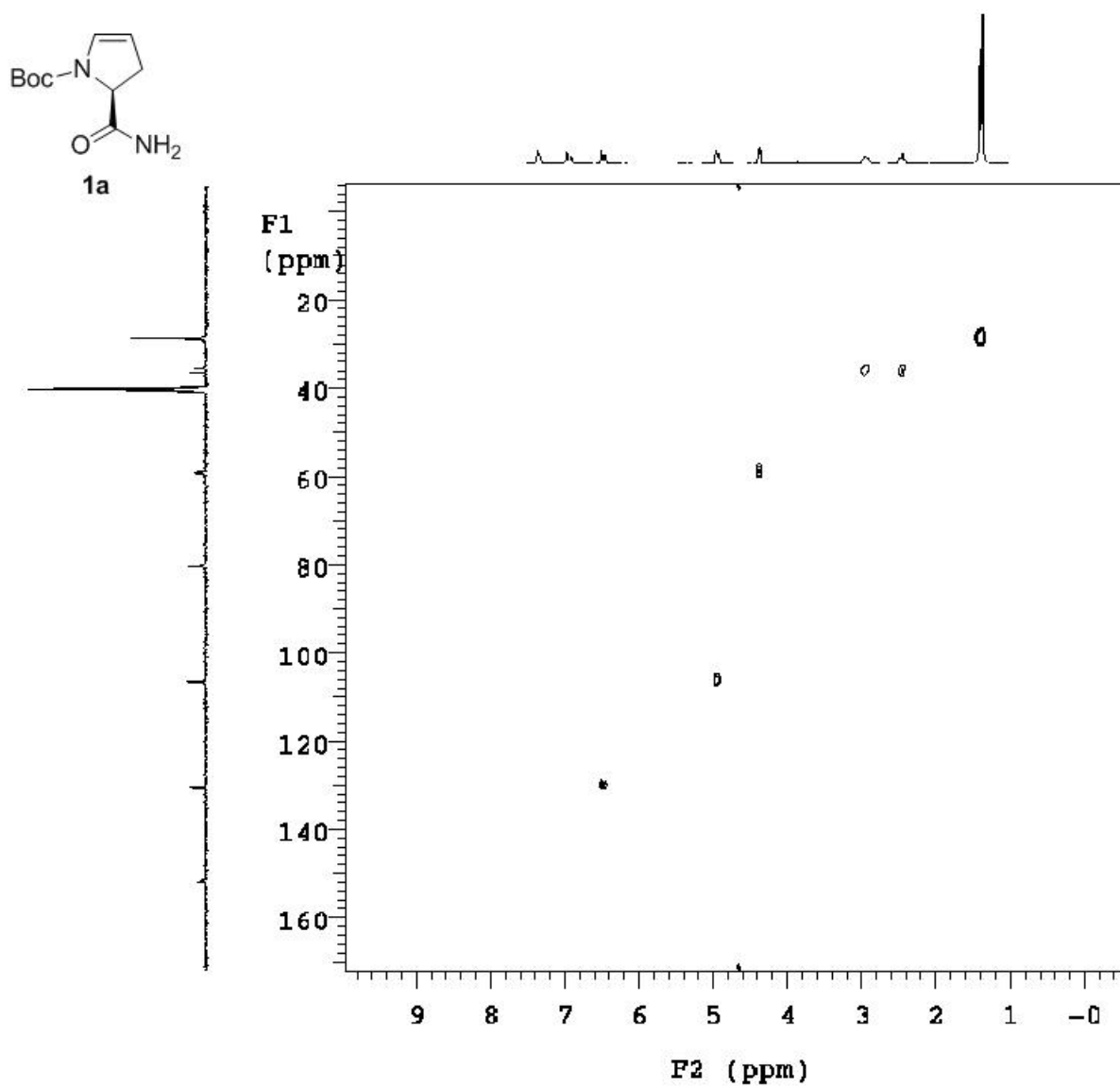
D. ^{13}C NMR Spectra of **1a**



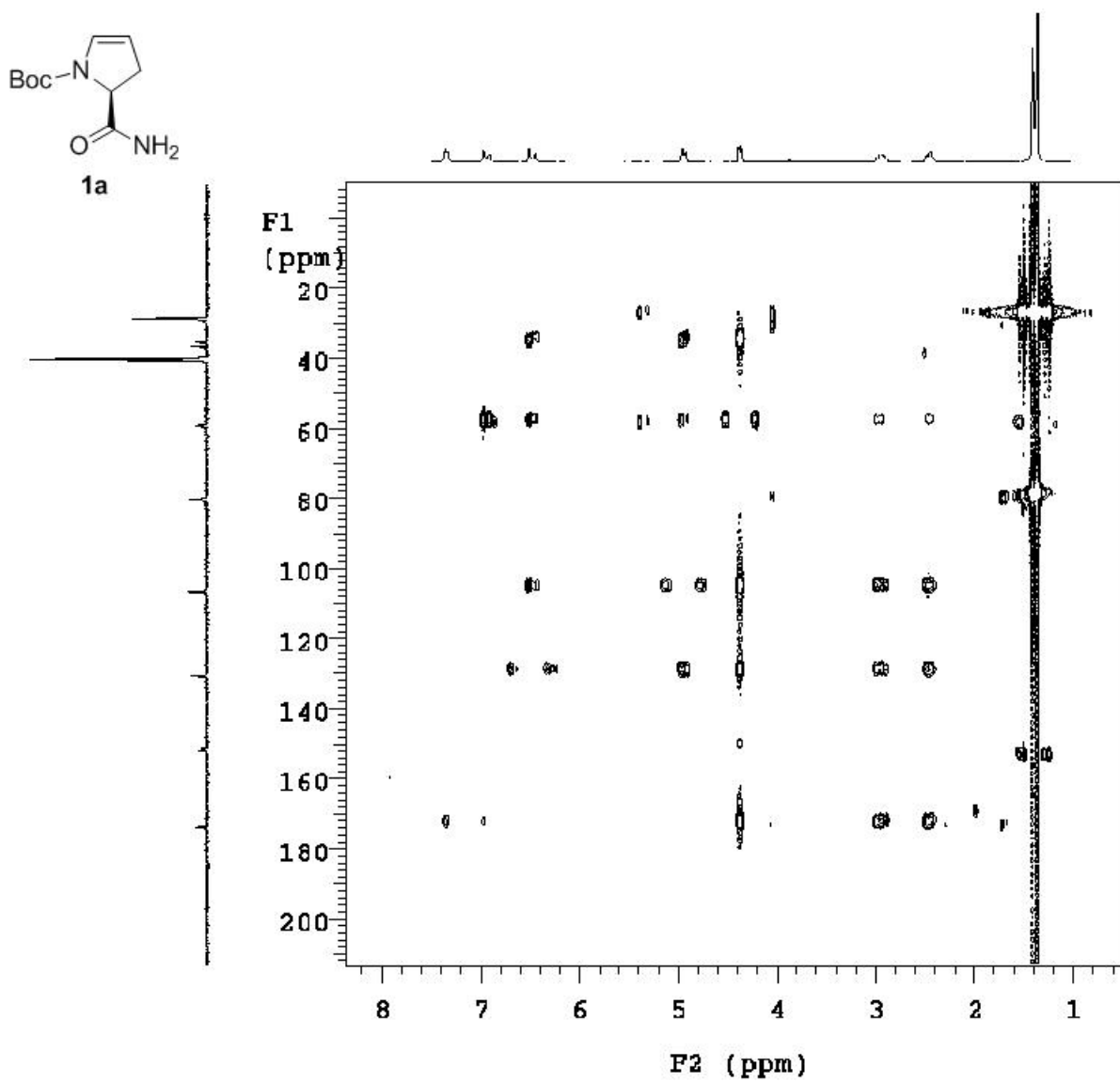
E. ^1H - ^1H COSY NMR Spectra of **1a**



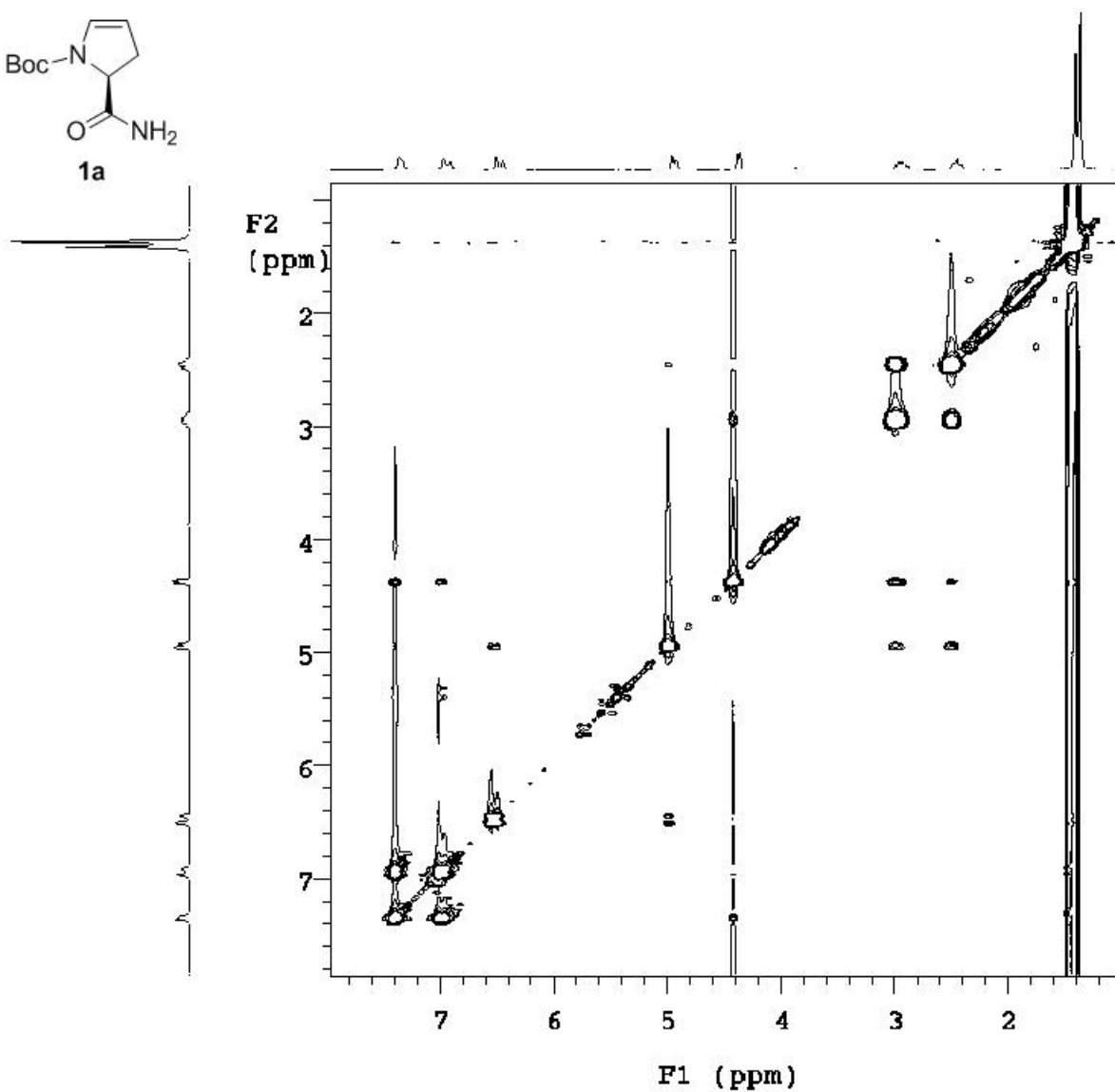
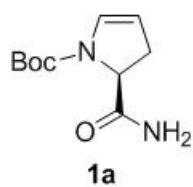
F. ^1H - ^{13}C HSQC NMR Spectra of **1a**



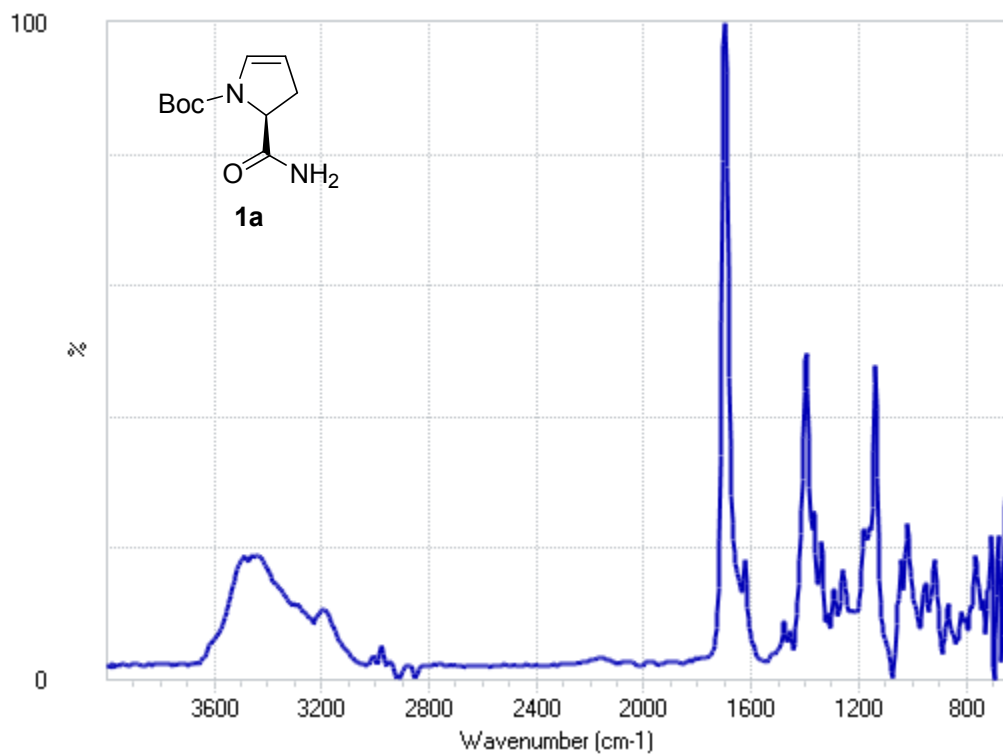
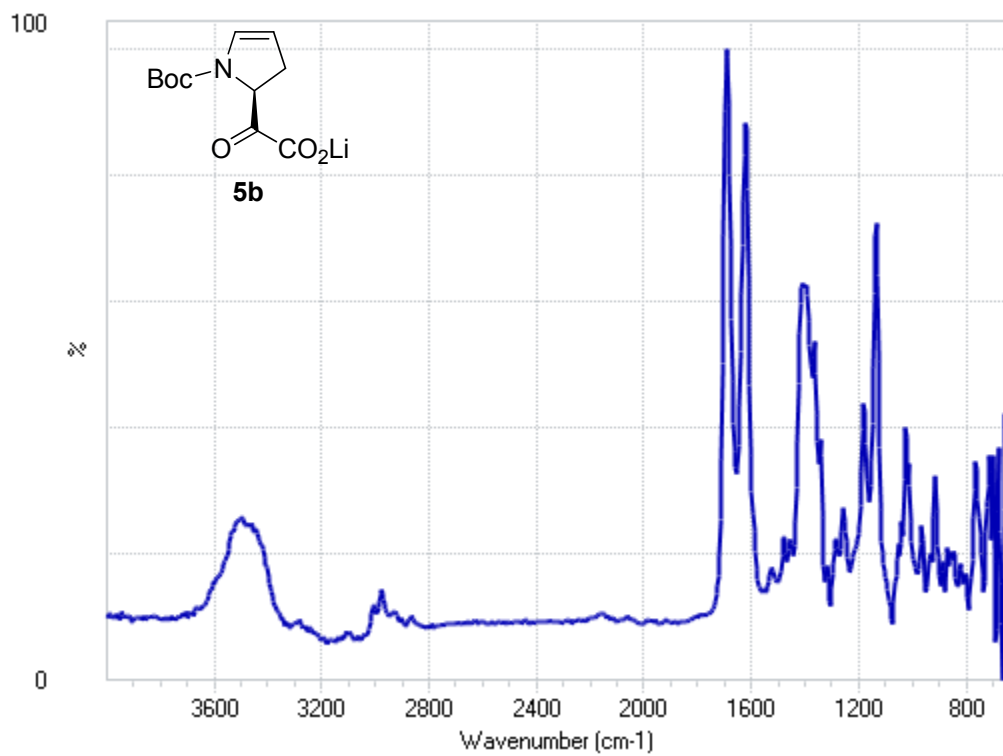
G. ^1H - ^{13}C HMBC NMR Spectra of **1a**



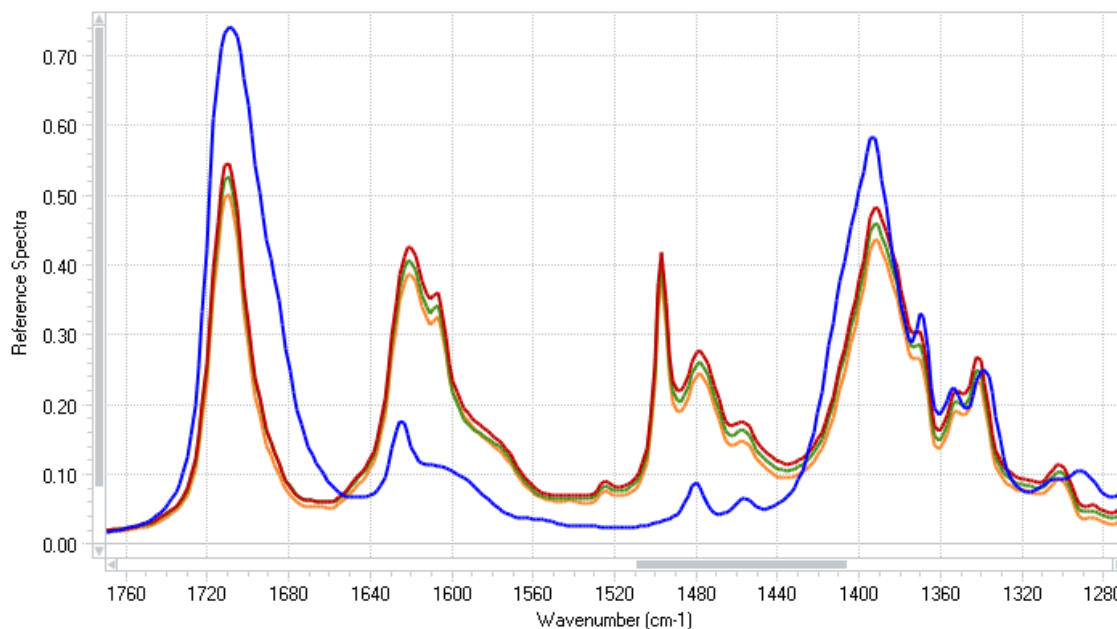
H. NOESY NMR Spectra of **1a**



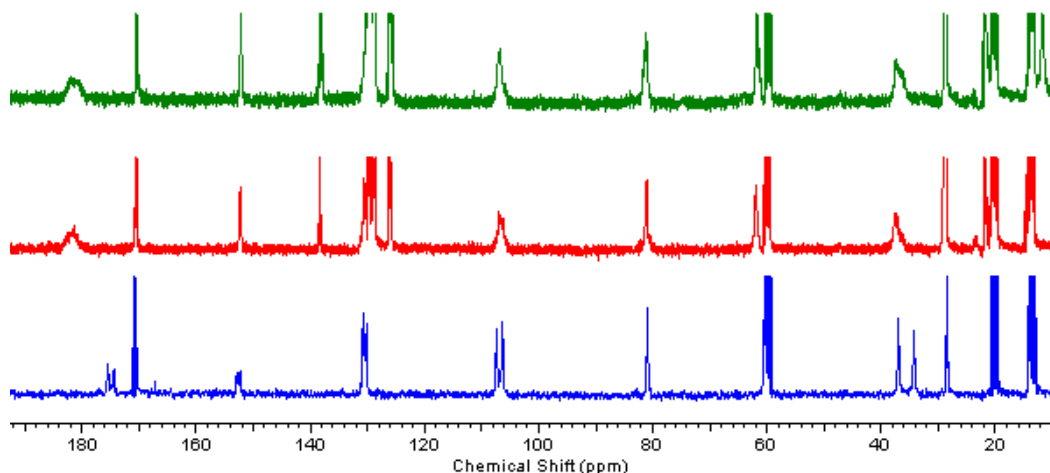
I. IR Spectra of **5b** and **1a**



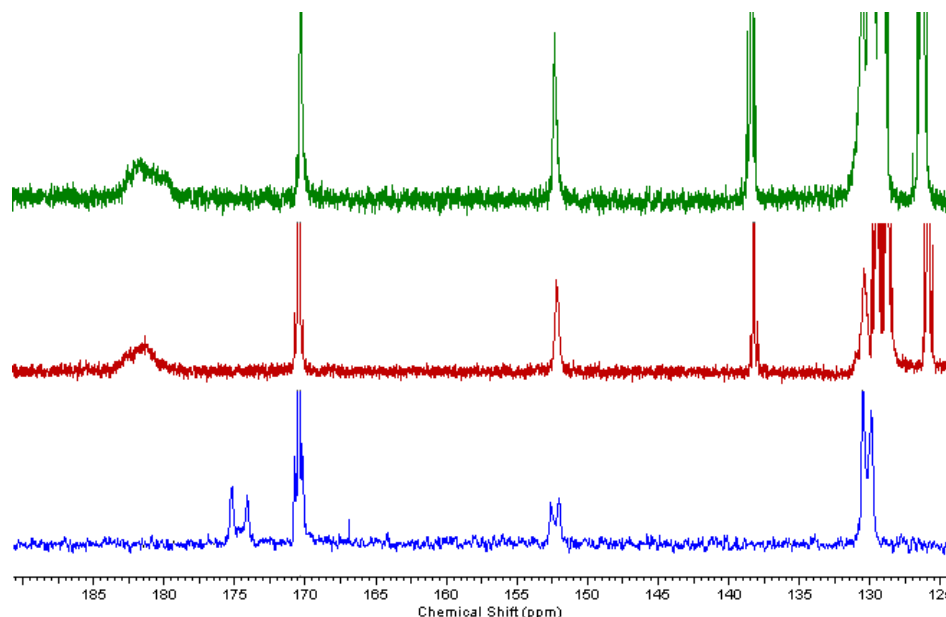
II. IR and ^{13}C NMR Spectra of the Reaction of **1a** with Et_2Zn



A. Overlay of IR spectra showing diagnostic bands¹ for dihydropyrrole **1a** (1710 and 1685 cm^{-1} , blue line) and **7a** (1710, 1620 and 1580 cm^{-1} , red line) upon treatment of **1a** (0.47 M in AcOEt) with 1 equiv Et_2Zn (1.1 M in toluene) at 0 °C. The IR frequencies were insensitive to the charge of a second equiv of Et_2Zn (green line) or the addition of 1 equiv $\text{Zn}(\text{CH}_2\text{I}_2)_2$ (orange line).

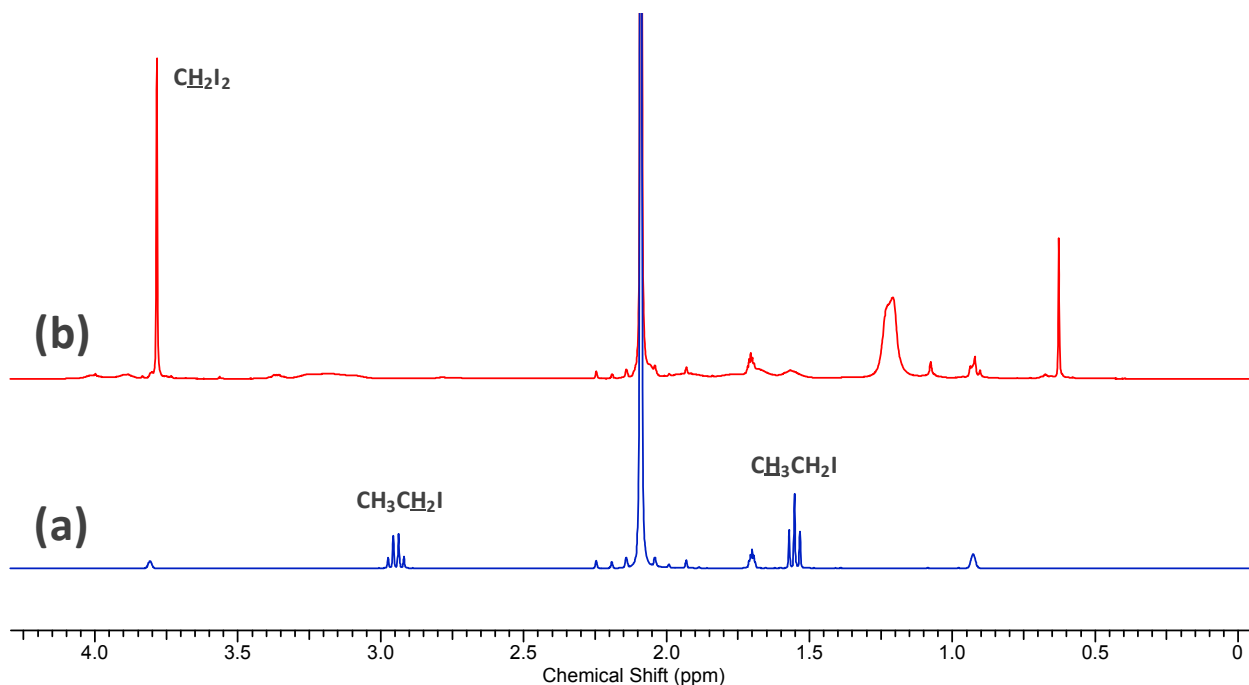


B. Representative titration of dihydropyrrole **1a** (0.41 M in $\text{AcOEt}-d_8$) with Et_2Zn (2.0 equiv, 1.1 M in toluene) at 25.0 °C as monitored by ^{13}C NMR. Blue spectrum: mixture before the addition of Et_2Zn ; red spectrum: mixture after the addition of 1.0 equiv Et_2Zn ; (c) green spectrum: mixture after the addition of 2.0 equiv Et_2Zn .

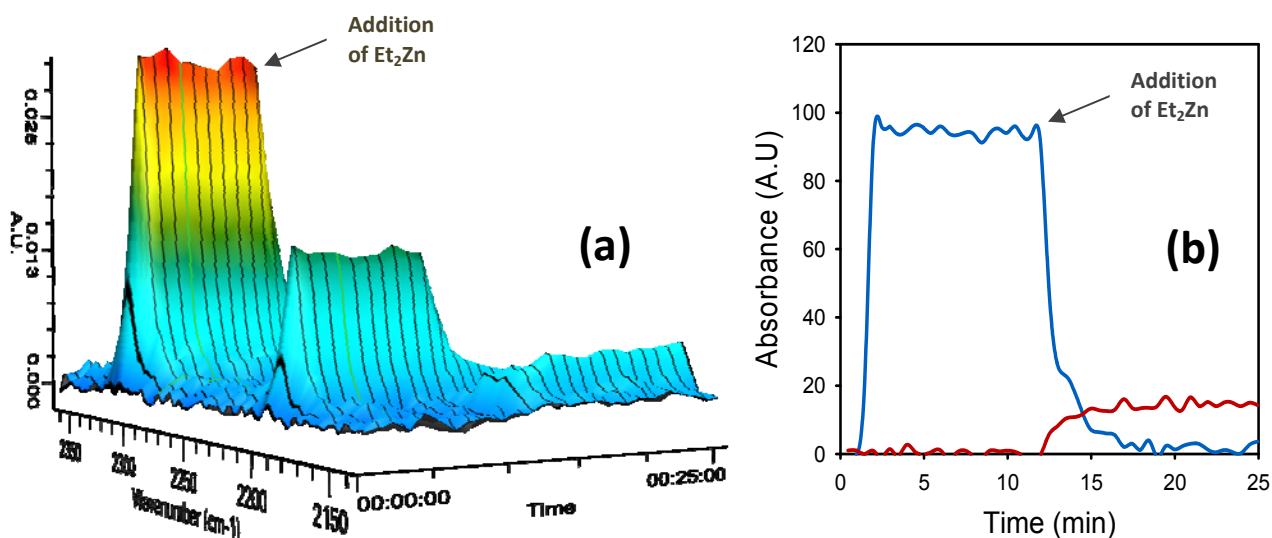


C. Expansion of ^{13}C NMR spectra in Section II.B between 125 and 190 ppm. Blue spectrum: mixture before the addition of Et_2Zn ; red spectrum: mixture after the addition of 1.0 equiv Et_2Zn ; (c) green spectrum: mixture after the addition of 2.0 equiv Et_2Zn .

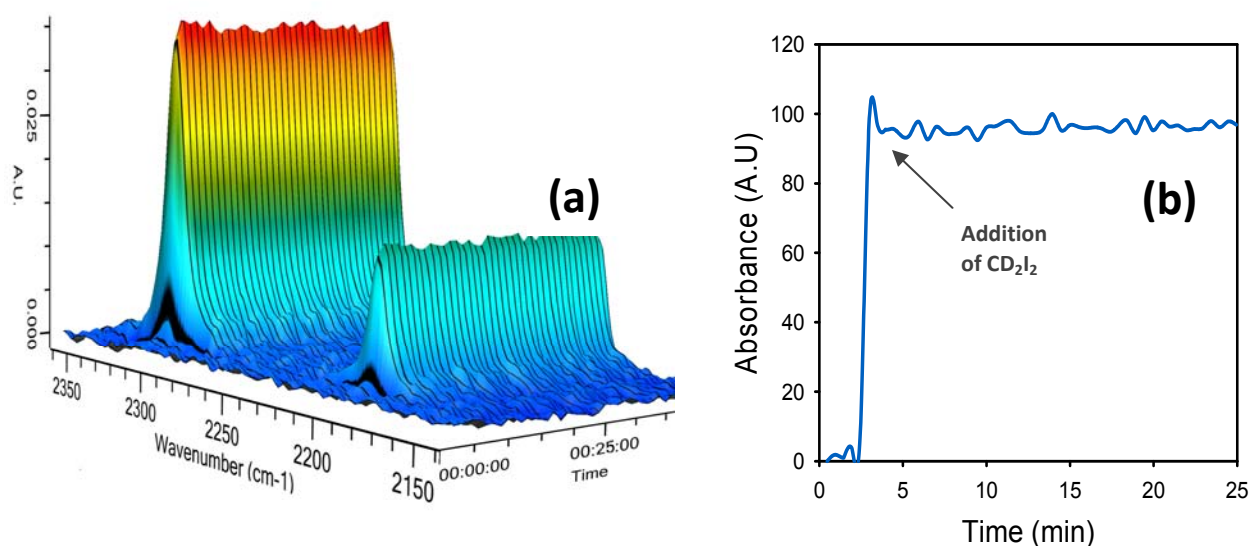
III. ^1H NMR and IR Spectra of the Reaction of **7b** with $\text{CH}_2\text{I}_2/\text{CD}_2\text{I}_2$



A. ^1H spectra of reaction mixtures containing (a) Et_2Zn (0.12 M) and CH_2I_2 , and (b) zinc amidate **7b** (0.12 M) and CH_2I_2 (0.24 M) in $\text{AcOEt-}d_8/\text{toluene}$ at $25.0\text{ }^\circ\text{C}$. The spectra are referenced to the resonance corresponding to the methyl group of toluene at 2.08 ppm.

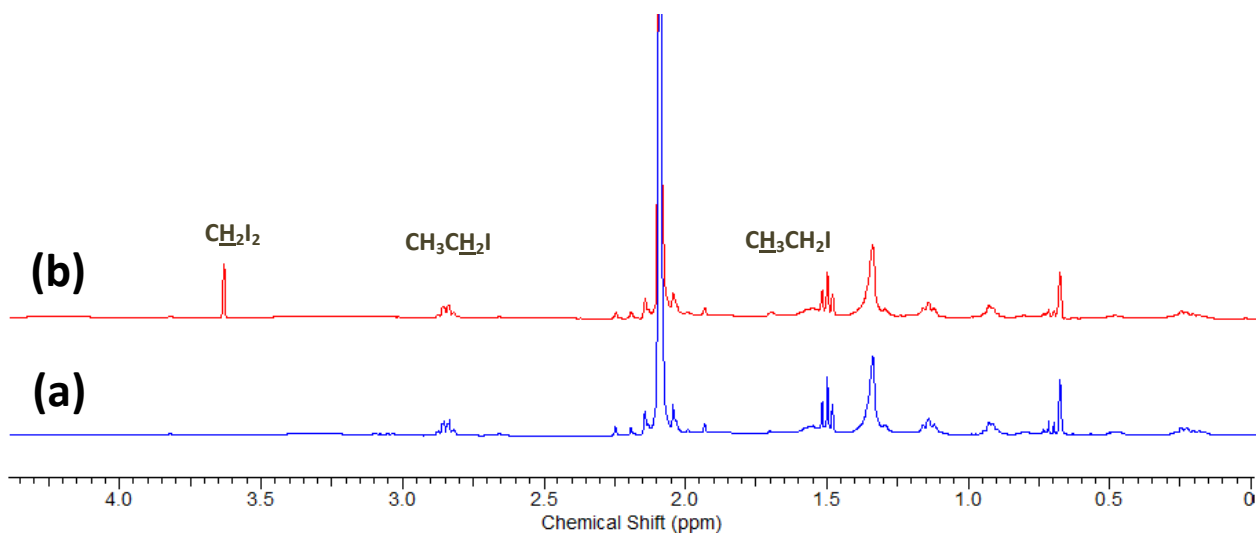


B. (a) In situ IR spectra of a reaction mixture containing Et₂Zn (0.32 M) and CD₂I₂ (0.64 M) in AcOEt/toluene solution at 0 °C. (b) Disappearance of CD₂I₂ (2310 cm⁻¹, blue line) and concomitant formation of Zn(CD₂I)₂ (2160 cm⁻¹, red line) in the same reaction.

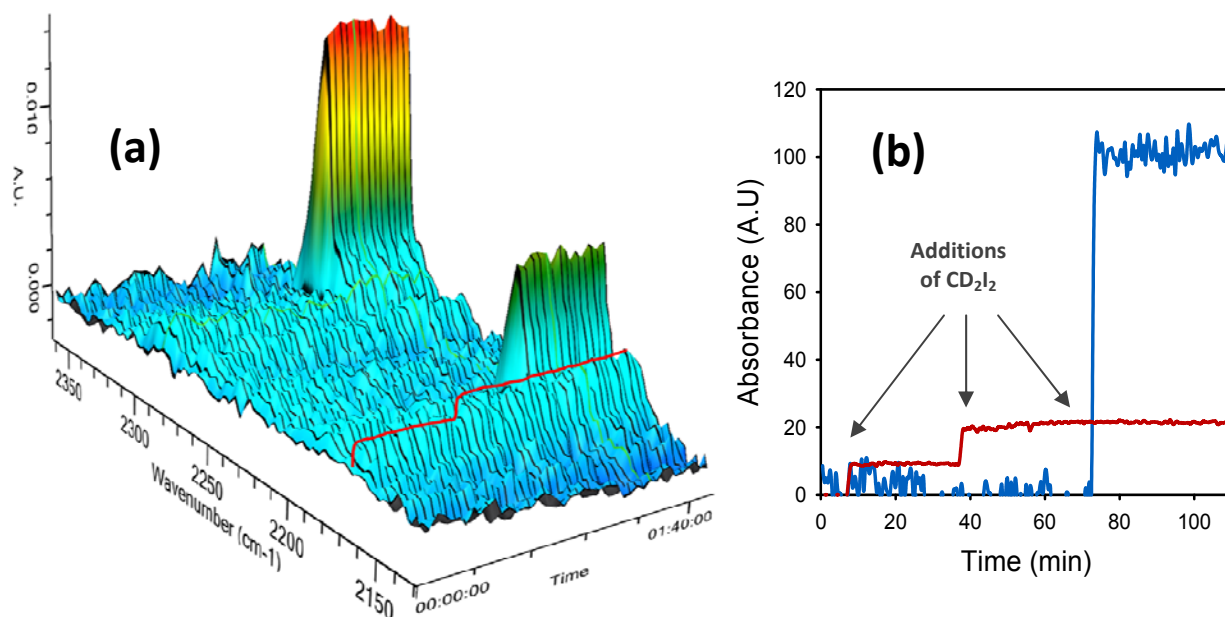


C. (a) In situ IR spectra of a reaction mixture containing amidate **7b** (0.32 M) and CD₂I₂ (0.64 M) in AcOEt/toluene solution at 25.0 °C. (b) Profile of the CD₂I₂ absorbance at 2310 cm⁻¹ (blue line) in the same reaction.

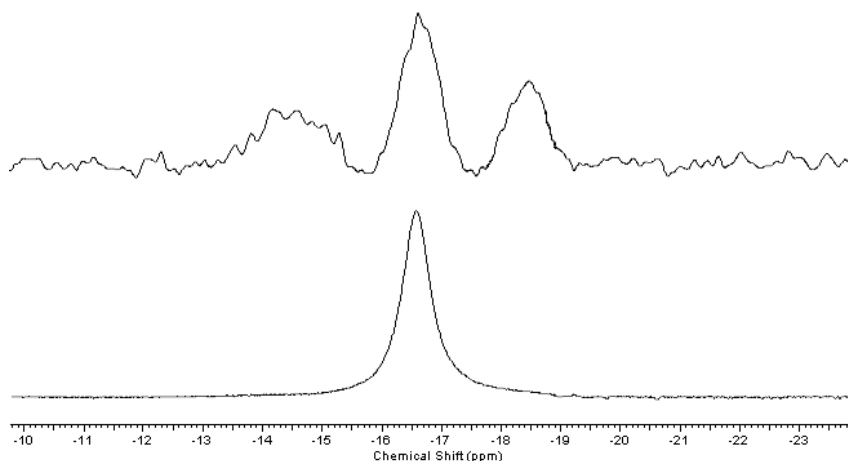
IV. NMR and IR Spectra of the Reaction of **7b** and Et₂Zn with CH₂I₂/CD₂I₂



A. ¹H spectra of reaction mixtures containing (a) zinc amidate **7b** (0.12 M), Et₂Zn (0.12 M) and CH₂I₂ (0.24 M), and (b) zinc amidate **7b** (0.12 M), Et₂Zn (0.12 M) and CH₂I₂ (0.36 M) in AcOEt-*d*₈/toluene at 25.0 °C. The spectra are referenced to the resonance corresponding to the methyl group of toluene at 2.08 ppm.

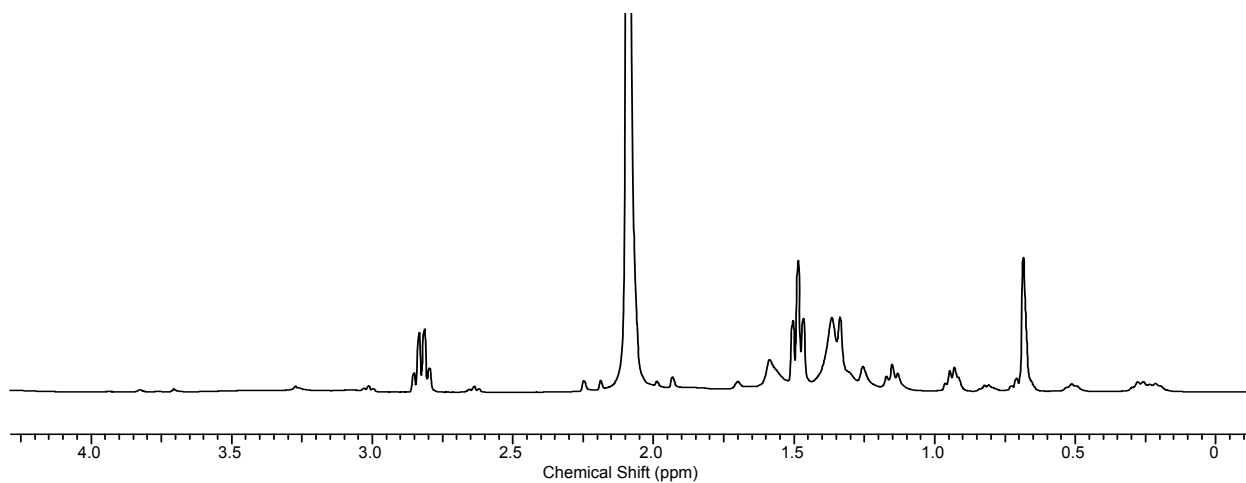


B. (a) In situ IR spectra of a reaction mixture containing amidate **7b** (0.32 M) and Et₂Zn (0.32 M) that was titrated with 3 equiv of CD₂I₂ (0.96 M) at intervals of 1 equiv of CD₂I₂ in AcOEt/toluene solution at 25.0 °C. (b) Profile of the CD₂I₂ (2310 cm⁻¹, blue line) and Zn(CD₂I)₂ (2160 cm⁻¹, red line) absorbances in the same reaction.

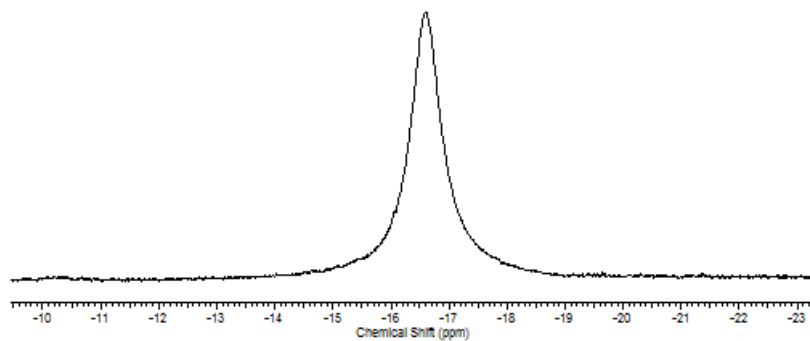


C. ^{13}C spectra of reaction mixtures containing (a) zinc amidate **7b** (0.12 M), Et_2Zn (0.12 M) and $^{13}\text{CH}_2\text{I}_2$ (0.12 M), and (b) zinc amidate **7b** (0.12 M), Et_2Zn (0.12 M) and $^{13}\text{CH}_2\text{I}_2$ (0.24 M) in $\text{AcOEt-}d_8$ /toluene at 25.0 °C. The spectra are referenced to the resonance corresponding to the methyl group of toluene at 20.4 ppm.

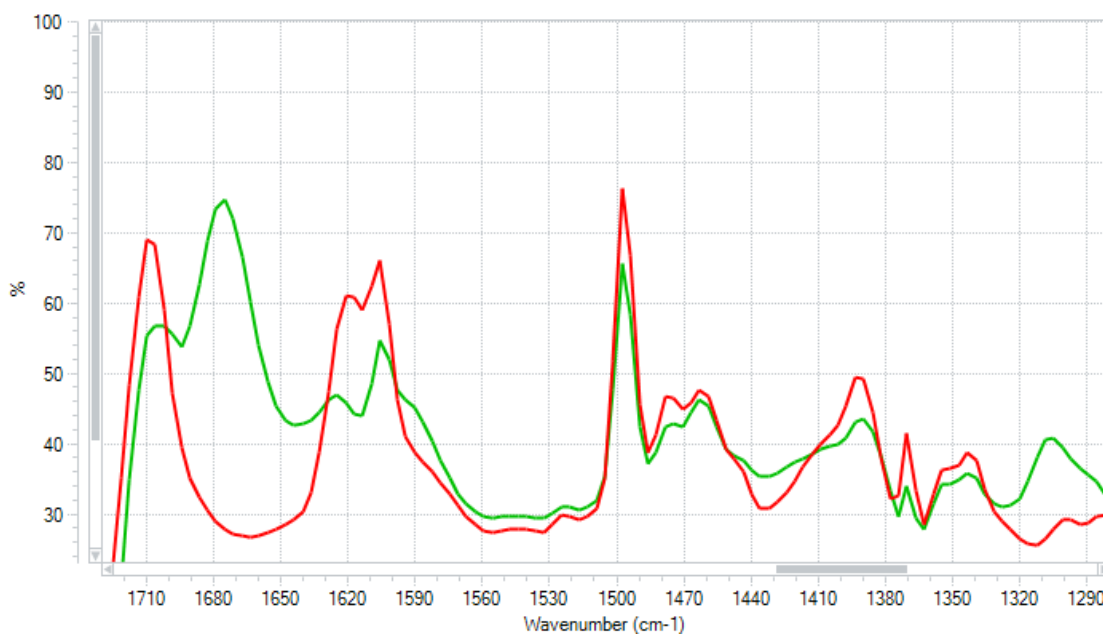
V. IR and NMR Spectra of the Reaction of **7b**/ $\text{Zn}(\text{CH}_2\text{I})_2$ with TFA



A. ^1H spectrum of reaction mixtures containing zinc amidate **7b** (0.12 M), $\text{Zn}(\text{CH}_2\text{I})_2$ (0.12 M) and TFA (0.02 M). The spectrum is referenced to the resonance corresponding to the methyl group of toluene at 2.08 ppm.

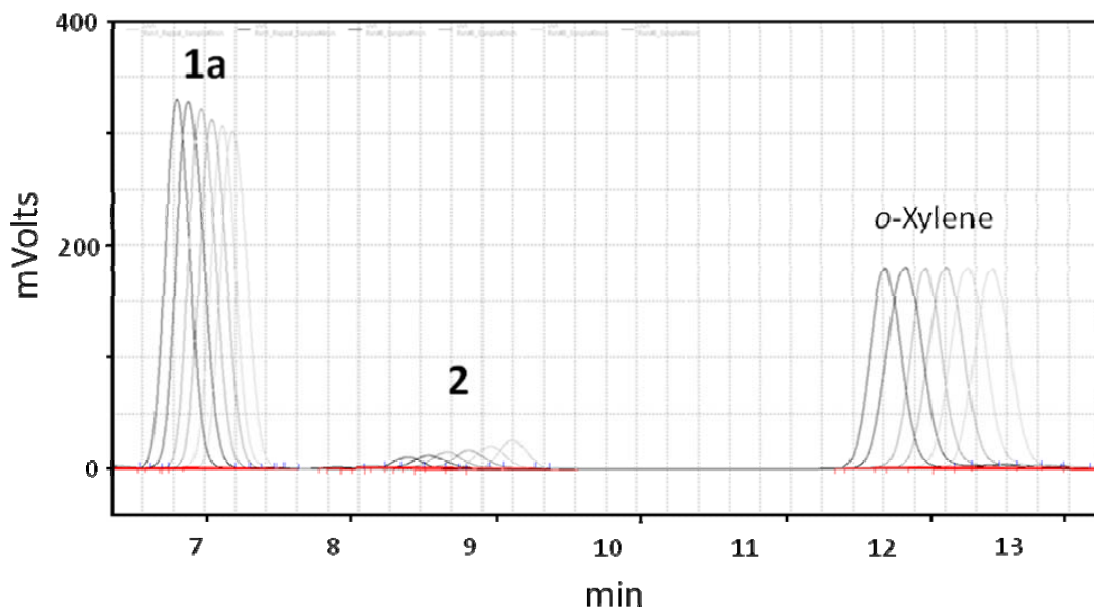


B. ^{13}C spectra of a reaction mixtures containing zinc amidate **7b** (0.12 M), $\text{Zn}(^{13}\text{CH}_2\text{I})_2$ (0.12 M) and TFA (0.02 M). The spectrum is referenced to the resonance corresponding to the methyl group of toluene at 20.4 ppm.



C. Overlay of IR spectra showing changes on the diagnostic bands¹ for amidate **7b** (1710, 1620 and 1580 cm⁻¹) upon treatment of an equimolar mixture of **7b** and $\text{Zn}(\text{CD}_2\text{I})_2$ (0.47 M in AcOEt , red line) with TFA (0.09 M) at 0 °C (green line).

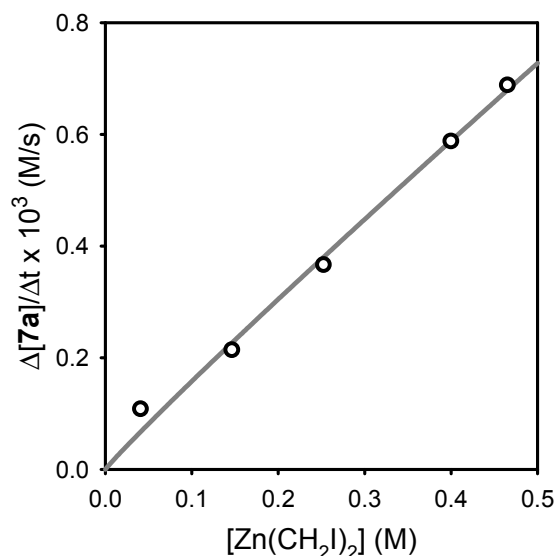
VI. Kinetic Studies



Representative overlay of HPLC traces with UV detection at 205 nm for the cyclopropanation of dihydropyrrole **1a** at 1 min intervals. $[1a] = 0.048$ M; $[Zn(CH_2I)_2] = 0.253$ M; $[AcOEt] = 5.230$ M. Analogous plots normalized to *o*-xylene internal standard were used to calculate the initial rates within the first 0-10% of substrate conversion using a linear fit.

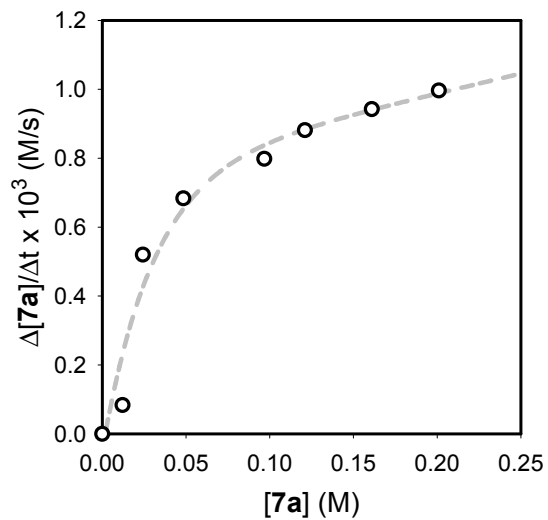
Plot of k_{obsd} versus $[Zn(CH_2I)_2]$ for the cyclopropanation of **7a** (0.048 M) by $Zn(CH_2I)_2$ at 0 °C ($[AcOEt] = 5.2$ M in toluene). The curve depicts an unweighted least-squares fit to $k_{obsd} = k[Zn(CH_2I)_2]^n$ ($k = 0.0014 \pm 0.0001$; $n = 0.95 \pm 0.08$).

$[Zn(CH_2I)_2]$ (M)	$\Delta[7a]/\Delta t$ (M/s)
0.041	$0.00011 \pm 2E-5$
0.147	$0.00021 \pm 2E-5$
0.253	$0.00037 \pm 4E-5$
0.400	$0.00059 \pm 7E-5$
0.465	$0.00068 \pm 6E-5$



Plot of k_{obsd} versus [7a] for the cyclopropanation of 7a by $\text{Zn}(\text{CH}_2\text{I})_2$ (0.465 M) at 0 °C ([AcOEt] = 5.2 M in toluene).

[7a] (M)	$\Delta[7a]/\Delta t$ (M/s)
0.012	$0.00008 \pm 1\text{E-}5$
0.024	$0.00052 \pm 7\text{E-}5$
0.048	$0.00068 \pm 6\text{E-}5$
0.097	$0.00080 \pm 9\text{E-}5$
0.121	$0.00088 \pm 9\text{E-}5$
0.161	$0.0009 \pm 1\text{E-}4$
0.201	$0.0010 \pm 1\text{E-}4$



VII. B3LYP and PCM Calculations

A. Computational Methods.

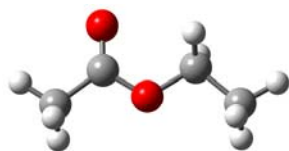
All calculations were executed using Gaussian 09, Revision B.01.² A series of geometries were systematically tested for reactants, intermediates, and transition structures at the B3LYP/6-311+G(2d,2p) level for all the atoms except iodine, which was described by Glukhovtsev's modification of the 6-311G basis set.³ Then, the lowest energy forms were submitted to single point B3LYP/6-311+G(2d,2p) calculations incorporating Tomasi's Polarized Continuum Model (PCM) corrections⁴ for AcOEt⁵ as the bulk solvent. The calculation of solution-phase energies using gas-phase geometries has been reported previously.⁶ All reactants and intermediates containing Zn atoms were reoptimized after the explicit addition of a molecule of AcOEt as coordinating solvent, and the resulting structures were only considered when their energies were lower than those of the unsolvated analogues. The calculated energies (ΔG , 298.15 K, 1.0 atm) result from the sum of thermal corrections to Gibbs free energy (TCGFE) as obtained from the frequency analysis at the B3LYP/6-311+G(2d,2p) level, and single point PCM corrections. Plausible structures for monomeric zinc amidate **7a** and its complexes with Et₂Zn, Zn(CH₂I)₂, and IZnCH₂I are presented in Figure 2. Calculated structures for transition states **TS1** and **TS2** are shown in Figures 5 and 6. Energy values are given relative to zinc amidate **10** in kcal·mol⁻¹. In the absence of structural data on the aggregation and solvation states of zinc amidate **7a** and plausible intermediates under the reaction conditions, these energies represent reactivity trends rather than precise values. Frequency calculations for all stationary points were carried out to describe them either as minima ($i = 0$) or as first-order transition states ($i = 1$). For all transition structures, visualization of the imaginary frequencies corresponded to the expected normal mode for the elementary step under investigation. Intrinsic reaction coordinate calculations (IRC) were performed from the transition states in forward and reverse directions to confirm the lowest energy reaction pathways that connect the corresponding minima.

B. Calculated B3LYP/6-311+G(2d,2p) Energies

Structure ^a	TCGFE (Hartree)	B3LYP/PCM (Hartree)	ΔG (kcal·mol ⁻¹)	Imaginary ν (cm ⁻¹)
AcOEt	0.085623	-307.820080	---	---
Et₂Zn	0.095996	-1937.848463	---	---
Zn(CH₂I)₂	0.013464	-15697.070394	---	---
IZnCH₂I	-0.008393	-15657.814317	---	---
9a (n = 0)	0.259140	-2584.171319	1.1	---
9b (n = 1)	0.363952	-2891.994255	11.4	---
9c (n = 2)	0.469066	-3199.812806	24.5	---
10 (n = 0)	0.260308	-2584.174263	0.0	---
11 (n = 0)	0.371042	-4522.001204	22.7	---
12 (n = 0)	0.294878	-18281.246947	11.8	---
13 (n = 0)	0.269480	-18242.000708	3.4	---
14 (n = 0)	0.370959	-4522.024458	8.1	---
15 (n = 0)	0.293860	-18281.254540	6.4	---
16 (n = 0)	0.272002	-18242.005233	2.1	---
TS1a-syn	0.295945	-18281.212266	34.2	-323
TS1b-syn	0.297744	-18281.220151	30.4	-375
TS1c-anti	0.291755	-18281.216925	31.6	-315
TS2a-syn	0.272710	-18241.961639	30.0	-306
TS2b-syn	0.274393	-18241.965834	28.4	-354
TS2c-anti	0.269907	-18241.958816	31.8	-299

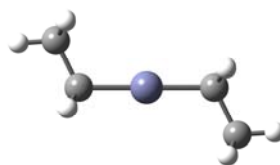
^a Multiple unconstrained searches for stationary points corresponding to structures **12** and **13** converged into related geometries lacking the C=O—Zn interaction.

C. Calculated Coordinates



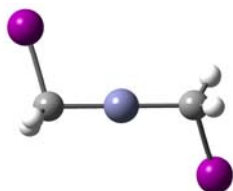
AcOEt

C	-1.039351	0.145032	-0.000149
O	-1.203552	1.339854	0.000030
O	0.176570	-0.440906	-0.000120
C	1.316983	0.452581	-0.000011
H	1.256907	1.091738	0.879968
H	1.257064	1.091742	-0.879998
C	-2.138571	-0.884997	0.000057
H	-3.101652	-0.385483	-0.000321
H	-2.050670	-1.524713	-0.877718
H	-2.051012	-1.523919	0.878452
C	2.567519	-0.398726	0.000097
H	3.448044	0.244408	0.000173
H	2.608775	-1.034350	0.884008
H	2.608928	-1.034348	-0.883809



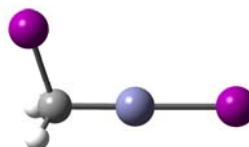
Et₂Zn

Zn	-0.000001	0.000015	-0.000057
C	-1.853155	-0.679573	-0.000055
H	-1.968365	-1.330980	-0.871142
H	-1.968276	-1.331209	0.870873
C	-2.956617	0.393138	0.000143
H	-2.896447	1.040128	-0.876930
H	-3.955916	-0.051336	0.000133
H	-2.896360	1.039890	0.877386
C	1.853168	0.679560	-0.000053
H	1.968301	1.331191	0.870877
H	1.968391	1.330968	-0.871138
C	2.956609	-0.393173	0.000143
H	2.896427	-1.040159	-0.876932
H	3.955919	0.051280	0.000134
H	2.896339	-1.039926	0.877384



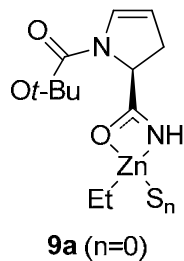
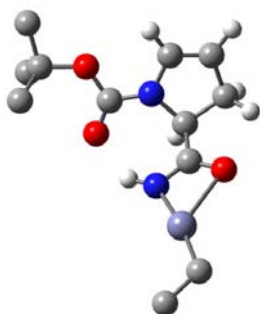
Zn(CH₂I)₂

Zn	0.000036	0.000177	0.435018
C	-1.554316	-1.176107	0.424161
H	-1.531644	-1.976431	-0.306965
H	-1.847230	-1.575058	1.388892
C	1.554450	1.176389	0.423742
I	3.255972	-0.098048	-0.191503
I	-3.256014	0.097905	-0.191529
H	1.531865	1.976505	-0.307615
H	1.847347	1.575576	1.388386

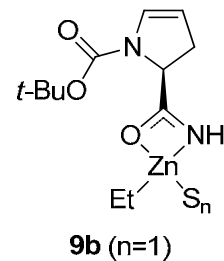
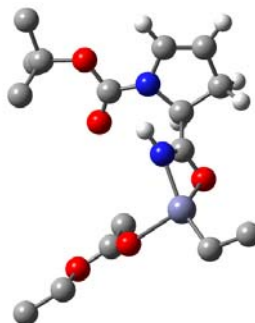


IZnCH₂I

Zn	0.000000	0.791320	0.000000
C	-1.903837	0.361188	0.000000
H	-2.430862	0.692698	0.887284
H	-2.430862	0.692698	-0.887284
I	-2.091197	-1.833449	0.000000
I	2.398457	1.318503	0.000000



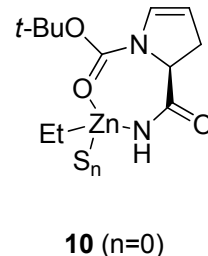
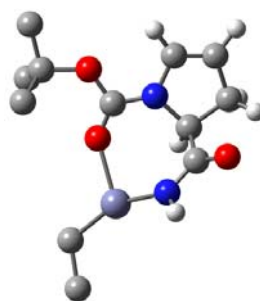
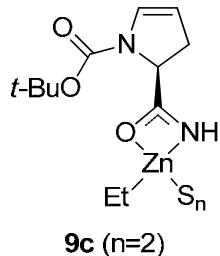
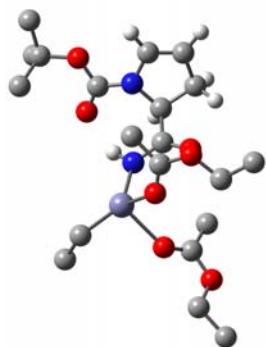
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C	1.820608	2.148351	-0.930540



C	-1.182344	-1.904332	1.097073
C	-1.442322	-3.444160	1.129283
C	-2.584055	-3.628421	0.160494
C	-3.015599	-2.454446	-0.289679

H	0.340756	1.611413	1.785492
H	-0.783626	3.585775	0.015804
H	0.502882	3.900803	1.176994
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H	1.440392	4.204597	-1.364878
H	2.591932	1.856979	-1.620252
C	-1.167035	1.096024	0.368654
N	-1.320410	-0.214036	0.531021
H	-0.533361	-0.734232	0.896193
O	-2.139132	1.771057	-0.040574
C	1.933329	0.035013	0.363098
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O	3.045965	-0.190032	-0.360477
C	3.861871	-1.407427	-0.159897
C	3.031536	-2.654067	-0.461668
H	3.682136	-3.529112	-0.456599
H	2.249243	-2.798497	0.277373
H	2.576869	-2.577517	-1.449539
C	4.967660	-1.231573	-1.198357
H	5.657208	-2.073909	-1.152652
H	4.549085	-1.182837	-2.203050
H	5.528567	-0.316478	-1.011803
C	4.444534	-1.419643	1.252531
H	4.987786	-0.494563	1.446141
H	3.666122	-1.535592	2.000605
H	5.146171	-2.249193	1.345990
Zn	-3.153999	-0.408776	-0.001059
C	-4.988475	-0.908341	-0.478425
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H	-5.145242	-0.584074	-1.509910
C	-5.357354	-2.392067	-0.324637
H	-6.396687	-2.578106	-0.609307
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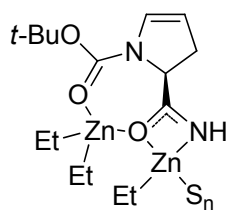
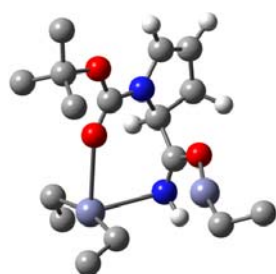
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N	-2.266972	-1.383797	0.241020
H	-2.995651	-4.584166	-0.118582
H	-3.817470	-2.241184	-0.973540
C	0.218397	-1.592903	0.581749
N	0.476914	-1.218725	-0.653805
H	-0.303794	-1.104809	-1.281502
O	1.180477	-1.740953	1.385766
C	-2.617681	-0.061687	0.268497
O	-1.997247	0.775538	0.898731
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C	-5.464169	1.279791	-1.566776
H	-5.981693	2.219049	-1.759061
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H	-5.424814	1.331675	1.182212
H	-3.906473	2.220456	1.384254
H	-5.282744	2.981863	0.573858
Zn	2.459981	-1.129117	-0.465047
C	4.302244	-1.667582	-0.955293
H	4.618321	-1.048637	-1.799459
H	4.947473	-1.379179	-0.121033
C	4.511898	-3.153125	-1.287877
H	5.555247	-3.378106	-1.531981
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H	3.910187	-3.466425	-2.143484
C	2.153687	1.868135	0.848072
O	2.609844	1.087331	0.026889
O	2.401757	3.177832	0.777767
C	3.223754	3.635399	-0.330045
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H	4.179945	3.117001	-0.284806
C	1.267844	1.502479	1.995380
H	0.235097	1.477176	1.638584
H	1.512804	0.502351	2.342346
H	1.344587	2.234954	2.793989
C	3.377454	5.133510	-0.194400
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C	-5.549793	-1.123212	0.266997
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H	-5.515052	-2.818686	1.590764
H	-3.990969	-2.567264	0.730753
H	-4.436533	-1.511768	2.081497
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H	-6.510371	-1.122990	-1.671924
H	-5.226695	-2.332597	-1.509762
H	-6.788604	-2.580533	-0.716480
Zn	1.113940	-1.285498	0.667757
C	1.409152	-2.868541	1.841746
H	2.293988	-2.651006	2.446843
H	0.579890	-2.951143	2.550522
C	1.597628	-4.211799	1.117400
H	1.798383	-5.036999	1.809432
H	0.712144	-4.489811	0.541419
H	2.432878	-4.175646	0.414509
C	3.586042	-0.728218	-1.486100
O	3.105692	-1.197553	-0.469264
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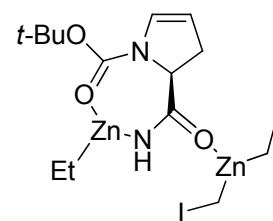
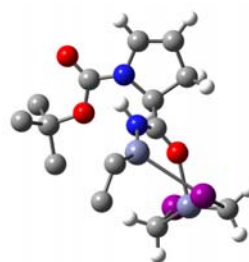
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H	0.579920	1.630548	-1.400652
H	-0.010810	4.288485	-0.457119
H	-0.455033	3.664457	-2.052643
N	-1.085343	1.319314	-0.154216
H	-2.679738	4.179633	-0.571308
H	-3.145445	1.748893	0.308451
C	1.104422	2.294519	0.605598
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H	2.540430	1.462132	1.591933
O	1.132889	3.347177	1.236017
C	-1.094879	-0.026990	-0.160710
O	-0.094337	-0.719983	-0.405027
O	-2.301400	-0.531426	0.109050
C	-2.561658	-1.992503	0.177851
C	-1.728856	-2.623925	1.291046
H	-2.049696	-3.655953	1.434503
H	-0.670497	-2.624713	1.049353
H	-1.879603	-2.090759	2.229584
C	-4.047498	-2.032727	0.526472
H	-4.380217	-3.067017	0.604733
H	-4.234229	-1.537606	1.478599
H	-4.637818	-1.538177	-0.243894
C	-2.307719	-2.631254	-1.185703
H	-2.878269	-2.116484	-1.958721
H	-1.254475	-2.607995	-1.448155
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H	5.555440	-0.841436	0.248743
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H	3.458960	0.309723	-3.360519
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C	7.172573	-1.246174	-1.145520
H	7.826577	-1.696953	-0.398765
H	7.327418	-1.763329	-2.091735
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C	1.156997	1.515399	2.150204
O	1.814005	0.574403	1.737171
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H	-0.915423	1.188703	1.950938
H	-0.228081	0.544505	3.439806
H	-0.503546	2.302447	3.269044
O	1.592876	2.768652	2.106211
C	2.831864	3.001854	1.378450
H	3.626883	2.441963	1.868672
H	2.693226	2.612036	0.372712
C	3.089795	4.491449	1.391428
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H	2.277284	5.030176	0.905844



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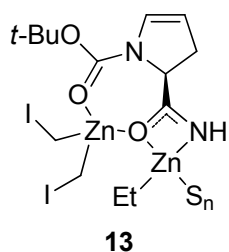
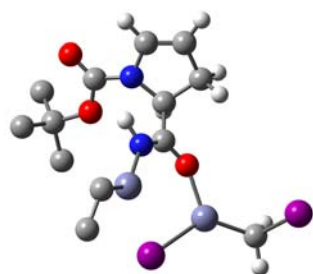
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C	-3.076637	-1.651378	-1.717690
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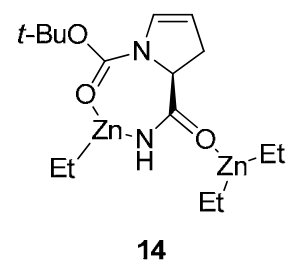
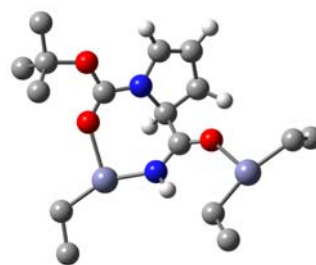
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H	-2.235990	0.039386	1.378797
H	-0.339013	-0.468076	3.503092
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N	-2.349803	-2.062232	1.594818
H	-1.760178	-2.327303	4.835525
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C	-0.300841	-0.710734	0.989071
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O	0.249439	0.413141	0.978719
C	-3.197924	-2.396016	0.558813
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C	-3.930624	-1.596549	-1.653132
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H	-1.335370	-2.791958	-2.308518
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C	-3.824530	-0.366876	-2.071035
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H	-3.165634	0.495769	-2.051774
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Zn	0.262313	2.474421	-0.570041
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C	-0.634226	4.937759	0.854701
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H	2.147128	2.743116	-3.247487
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H	0.970949	1.487745	-3.581174
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H	4.642723	-1.117049	-1.184990
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H	-4.080149	-2.878943	-3.370732
Zn	1.913518	-2.128042	-0.405756
C	3.071677	-3.325112	-1.470628
H	3.947056	-3.557973	-0.860215
C	3.511426	-2.773855	-2.835832
H	4.166212	-3.477021	-3.357458
H	2.659957	-2.583207	-3.491105
H	4.059264	-1.835562	-2.739314
Zn	1.112106	1.983757	-0.066755
C	-0.229798	2.658954	-1.354791
H	0.105240	3.416007	-2.055472
H	-0.788156	1.899585	-1.892473
C	2.917317	1.987886	0.777508
H	2.932099	1.970131	1.862834
H	2.535568	-4.267829	-1.601627
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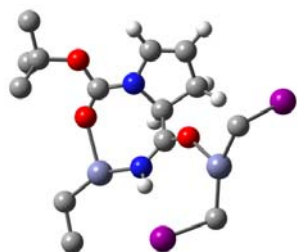


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H	1.88265	-2.54075	-0.60301
H	0.17049	-3.002	1.69844
H	1.0153	-4.2343	0.79093
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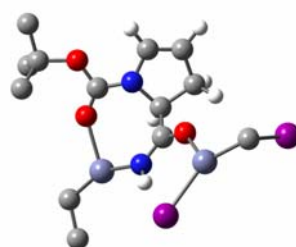
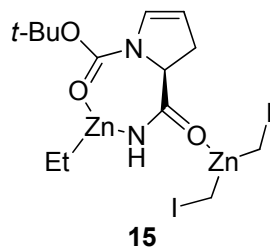
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H	0.727544	-0.448033	-2.208392
H	-1.497421	-2.277323	-2.082405
H	-0.127186	-2.428016	-3.187702
N	1.200794	-1.626399	-0.531264
H	-0.170094	-4.495962	-1.418614
H	1.594907	-3.575525	0.298261

C	0.87784	-0.96201	0.3733
N	1.16848	0.105	1.07294
H	2.03774	-0.03699	1.57584
O	-0.15171	-1.12624	-0.35109
C	4.19371	-1.40969	0.36835
O	5.30126	-1.36058	0.85987
O	3.82579	-0.9433	-0.83904
C	4.79825	-0.26049	-1.72605
C	5.9232	-1.2216	-2.10555
H	6.55657	-0.75178	-2.85851
H	6.53528	-1.47658	-1.2456
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C	3.9448	0.08373	-2.94404
H	4.55187	0.6018	-3.68569
H	3.54132	-0.81945	-3.40053
H	3.11485	0.73149	-2.66475
C	5.3151	1.00781	-1.04978
H	4.48329	1.64745	-0.75412
H	5.91347	0.77582	-0.17362
H	5.93243	1.56462	-1.75518
Zn	0.43172	1.92035	1.3274
C	0.69894	3.40713	2.6151
H	0.19019	3.11942	3.53876
C	0.21601	4.78763	2.15371
H	0.3802	5.54942	2.92106
H	0.73637	5.12061	1.25425
H	-0.85008	4.78754	1.92292
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H	1.76337	3.44011	2.86117
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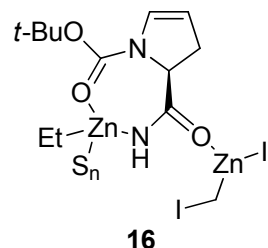
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H	4.181069	-2.704799	2.741381
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H	5.478014	-1.464009	-0.542857
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H	6.250176	-0.397132	0.629734
Zn	1.176454	1.967812	-0.663902
C	2.297869	3.565462	-0.932449
H	2.919448	3.672263	-0.039331
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H	2.274238	5.713705	-1.366510
H	0.969391	4.824617	-2.136650
H	0.897981	5.171072	-0.419270
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H	-4.225950	-2.176571	-1.646343
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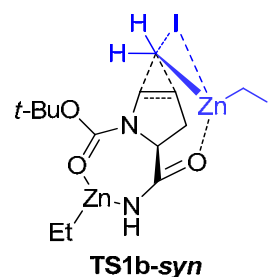
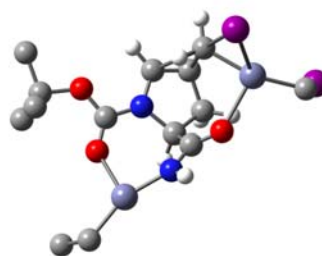
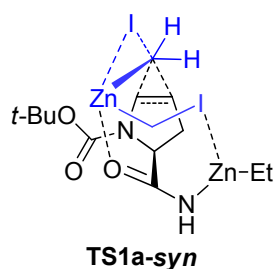
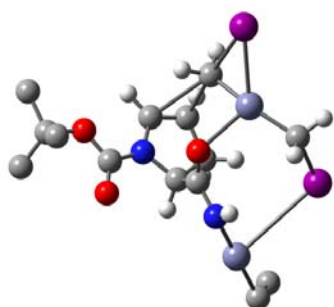


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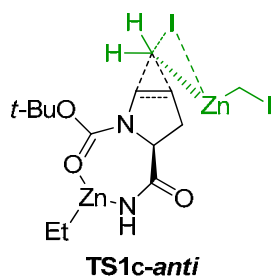
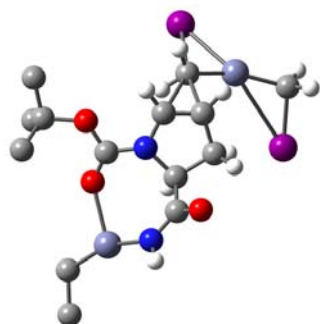
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N	-0.931713	-1.734532	-1.390333
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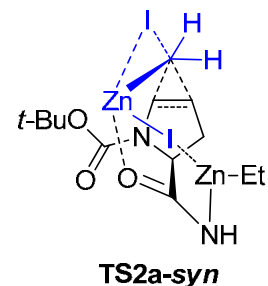
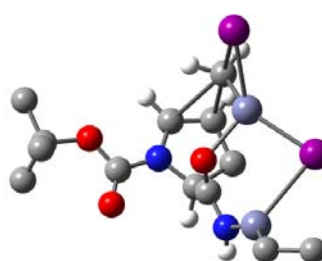
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O	2.647923	2.305810	0.762194
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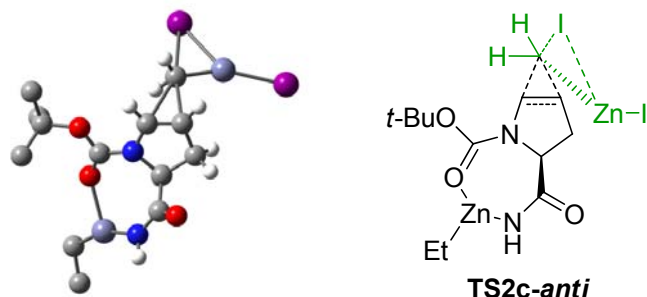
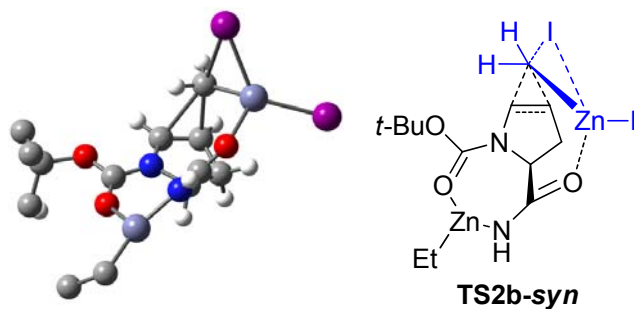
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C	0.85140000	-0.64073600	2.85892300
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H	1.503689	2.235520	1.501858

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C	-1.325990	1.507455	0.775907
C	4.945906	-2.813406	-1.679976
O	-0.763113	-1.093725	-0.733125
C	2.890933	0.498699	0.820308
O	3.286677	1.715194	1.163918
C	4.644900	2.253614	0.834958
O	3.589567	-0.331638	0.227433
H	-1.677851	2.345460	1.372290
H	-0.474194	1.802993	0.173020
H	-1.108542	-1.475881	1.699999
H	0.904801	-1.791570	-2.223797
H	1.156956	2.068160	2.242792
H	-0.967794	0.804105	3.182463
H	1.660142	-1.862193	1.231544
H	0.099053	-1.602709	2.962827
H	4.814060	-3.890096	-1.551535
H	5.089141	-2.663299	-2.752804
I	-2.901350	2.230699	-1.487095
C	5.708738	1.419081	1.540721
H	5.762217	0.411486	1.139643
H	6.679802	1.895301	1.405043
H	5.506788	1.367839	2.610417
C	4.572330	3.660283	1.419937
H	3.778223	4.236678	0.947392
H	4.387452	3.624621	2.492715
H	5.517081	4.175449	1.251993
C	4.827156	2.300984	-0.678720
H	5.743765	2.844615	-0.907330

N	1.734578	-0.080070	-1.230060
C	0.702652	0.303844	-2.086014
C	-0.294491	-0.602144	-2.096956
C	0.087569	-1.804488	-1.272560
C	1.500328	-1.468404	-0.736843
C	2.574828	-2.443005	-1.308038
N	3.723135	-2.523324	-0.610043
Zn	4.416547	-1.482377	0.833071
C	5.490851	-1.050711	2.422961
O	2.318518	-3.074934	-2.327491
C	2.662709	0.758148	-0.684416
O	2.613600	1.973157	-1.217610
C	3.522285	3.080011	-0.786697
O	3.434985	0.408838	0.210175
H	0.158106	-2.688285	-1.905564
H	4.330117	-3.206819	-1.048149
H	0.786265	1.226704	-2.631220
H	1.543402	-1.457962	0.350224
H	-0.622596	-2.012567	-0.473219
H	6.212459	-0.285686	2.124312
H	4.819241	-0.557826	3.130749
C	3.267837	3.408391	0.681311
H	3.581488	2.599893	1.334754
H	3.830002	4.303776	0.946801
H	2.211039	3.614024	0.850044
C	3.072714	4.224366	-1.689768
H	3.210763	3.966979	-2.739133
H	2.022701	4.460279	-1.522579
H	3.662392	5.114369	-1.474332
C	4.971797	2.697995	-1.070366
H	5.606404	3.568072	-0.901678
H	5.311529	1.896677	-0.421347
H	5.091267	2.391209	-2.109247
C	6.215531	-2.219647	3.107799
H	6.781140	-1.888026	3.983082
H	5.518335	-2.986353	3.448959
H	6.923347	-2.708773	2.436613

H	4.905987	1.306789	-1.107923
H	3.997850	2.828921	-1.148583
C	6.186300	-2.337265	-0.910552
H	6.089786	-2.507563	0.162754
H	7.087721	-2.860996	-1.239891
H	6.365802	-1.270300	-1.049665
I	-4.163131	-1.688538	0.815996

H	-1.116228	-0.588778	-2.791653
C	-1.494366	0.951909	-0.687663
Zn	-3.074339	-0.068923	-0.077814
I	-3.190065	2.432649	0.953793
H	-0.660438	1.139850	-0.023194
H	-1.572037	1.703752	-1.462762
I	-4.483844	-2.098986	0.084782

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