

**Supporting information for:**

**“Corroles that “Click”: Modular Synthesis of Azido- and Propargyl- Functionalized Metallocorrole Complexes and Convergent Synthesis of a bis-Corrole Scaffold”**

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### A. Cyclic voltammetry of compounds 7, 9, 14 and 15

Cyclic voltammetric measurements were performed with a Gamry Reference 600 potentiostat, using either 0.1 M tetraethylammonium tetrafluoroborate (TEABF<sub>4</sub>) in acetonitrile or 0.1 M tetra-*n*-butylammonium tetrafluoroborate (TBABF<sub>4</sub>) in CH<sub>2</sub>Cl<sub>2</sub>. A three-electrode cell was used, with a glassy carbon working electrode (3 mm diameter), a platinum mesh counter electrode, and a Ag/Ag<sup>+</sup> reference electrode. The reference electrode consisted of a silver wire in a solution of 0.01 M AgNO<sub>3</sub> and 0.1 M TEABF<sub>4</sub> in acetonitrile, separated from the bulk solution by a Vycor frit. Electrolyte solutions were deoxygenated with nitrogen for at least 10 minutes before any compounds were added and were kept under a blanket of nitrogen during analysis. Between measurements, the working electrode was polished with 0.05 µm alumina and rinsed with distilled water and acetonitrile.

All potentials are referenced to Ag/Ag<sup>+</sup> (0.01 M Ag<sup>+</sup>). Potentials reproduced from other publications have been converted from SCE to Ag/Ag<sup>+</sup> by subtracting 298 mV.<sup>1</sup> All scans shown were taken at 100 mV/s, except for **15** (the Cu-Fe complex), which was not sufficiently soluble in acetonitrile (although solubility was better in acetonitrile than in CH<sub>2</sub>Cl<sub>2</sub>) and had to be measured at 500 mV/s to observe well-resolved peaks. Currents are reported as current densities, calculated using an electrode area of 0.071 cm<sup>2</sup>. Arrows on the voltammograms (in text and SI) indicate the direction of the first potential sweep.

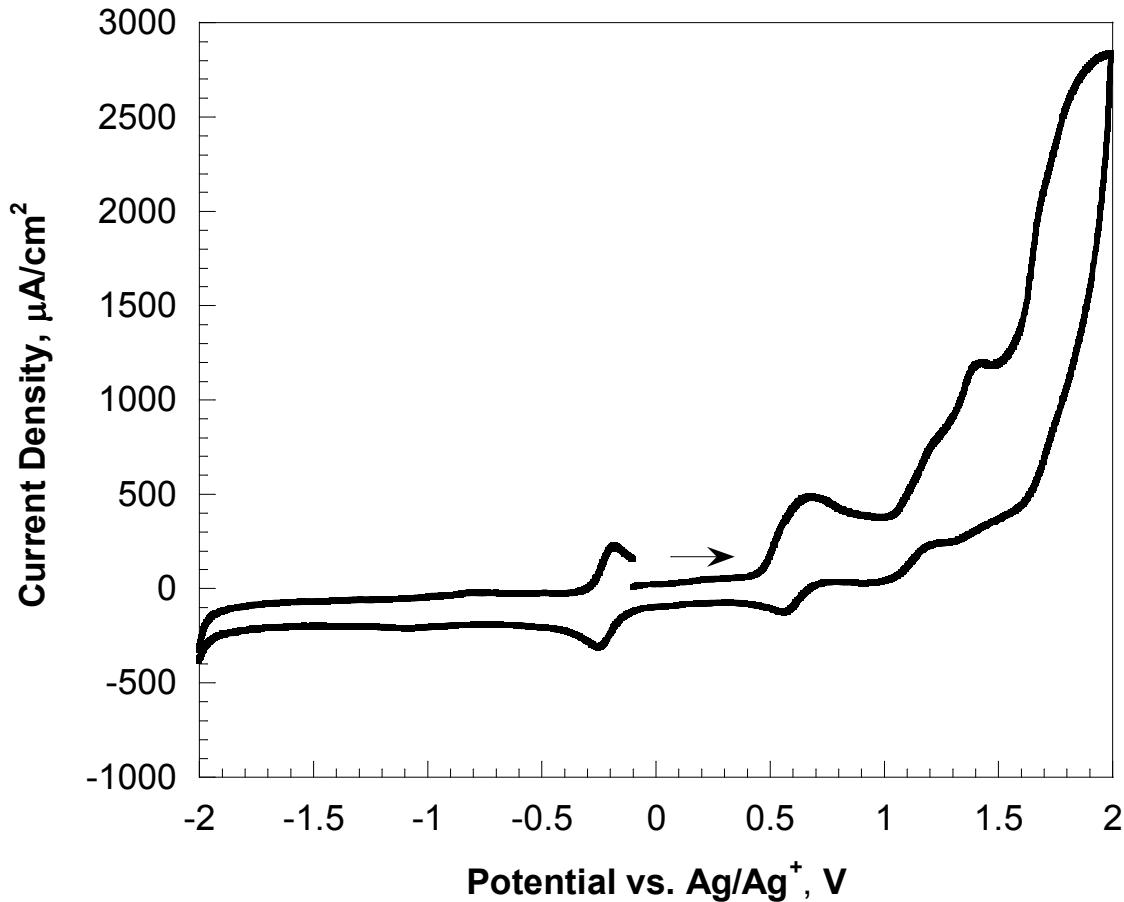


Figure S1. Cyclic voltammogram of 1 mM  $(\text{C}_6\text{F}_5)_2(\text{p-OMePh})\text{corroleCu}$  (**7**) in acetonitrile, 100 mV/s.

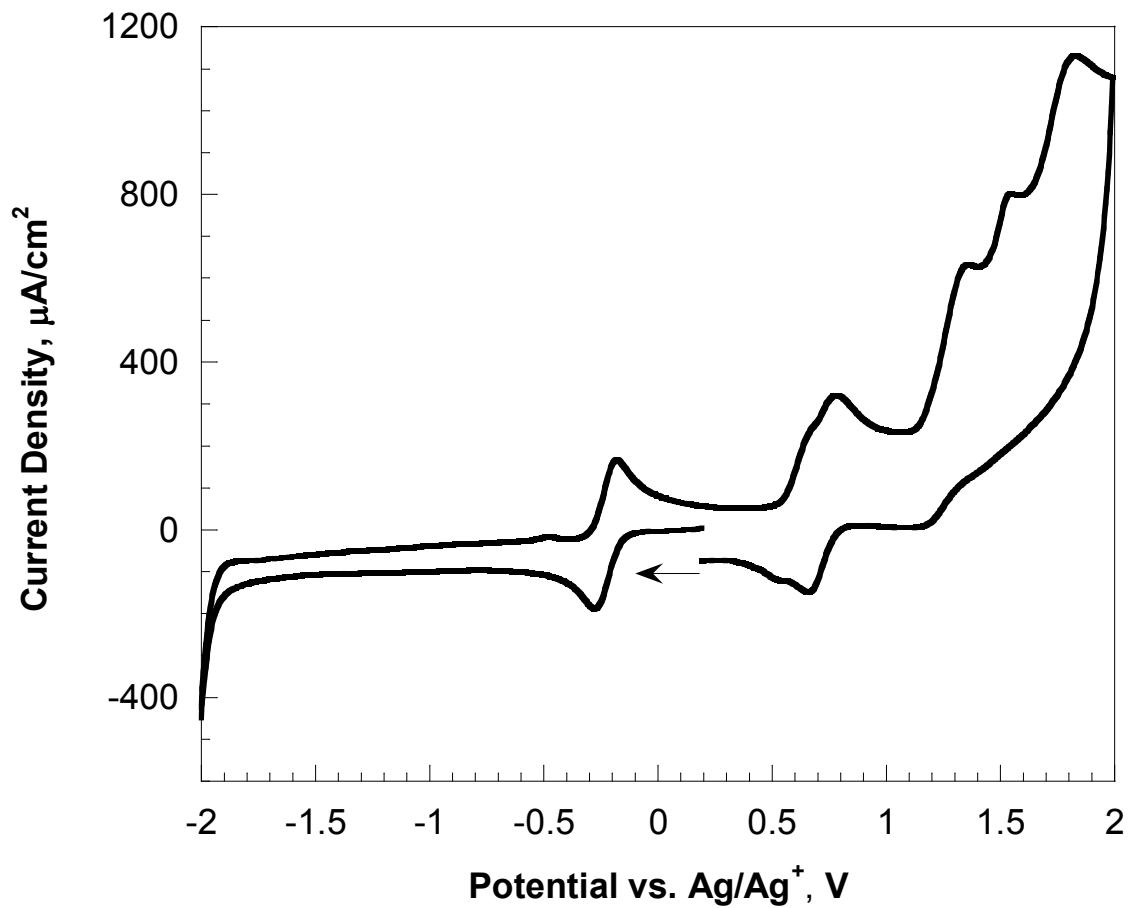


Figure S2. Cyclic voltammogram of 1 mM  $(\text{C}_6\text{F}_5)_2(\text{p-OMePh})\text{corroleCu}$  (**7**) in  $\text{CH}_2\text{Cl}_2$ , 100 mV/s.

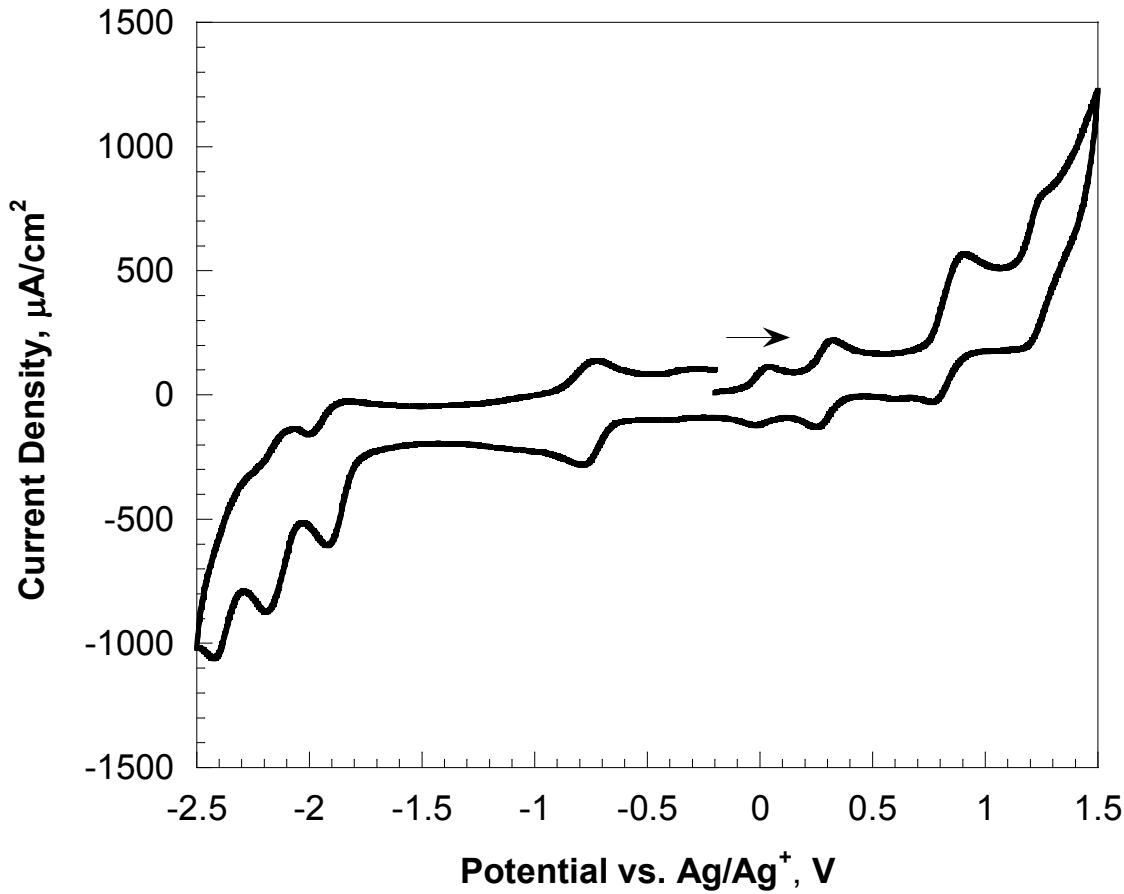


Figure S3. Cyclic voltammogram of 1 mM  $((\text{C}_6\text{F}_5)_2(p\text{-OMePh})\text{corroleFe}(\text{Et}_2\text{O})_2)$  (**9**) in acetonitrile, 100 mV/s.

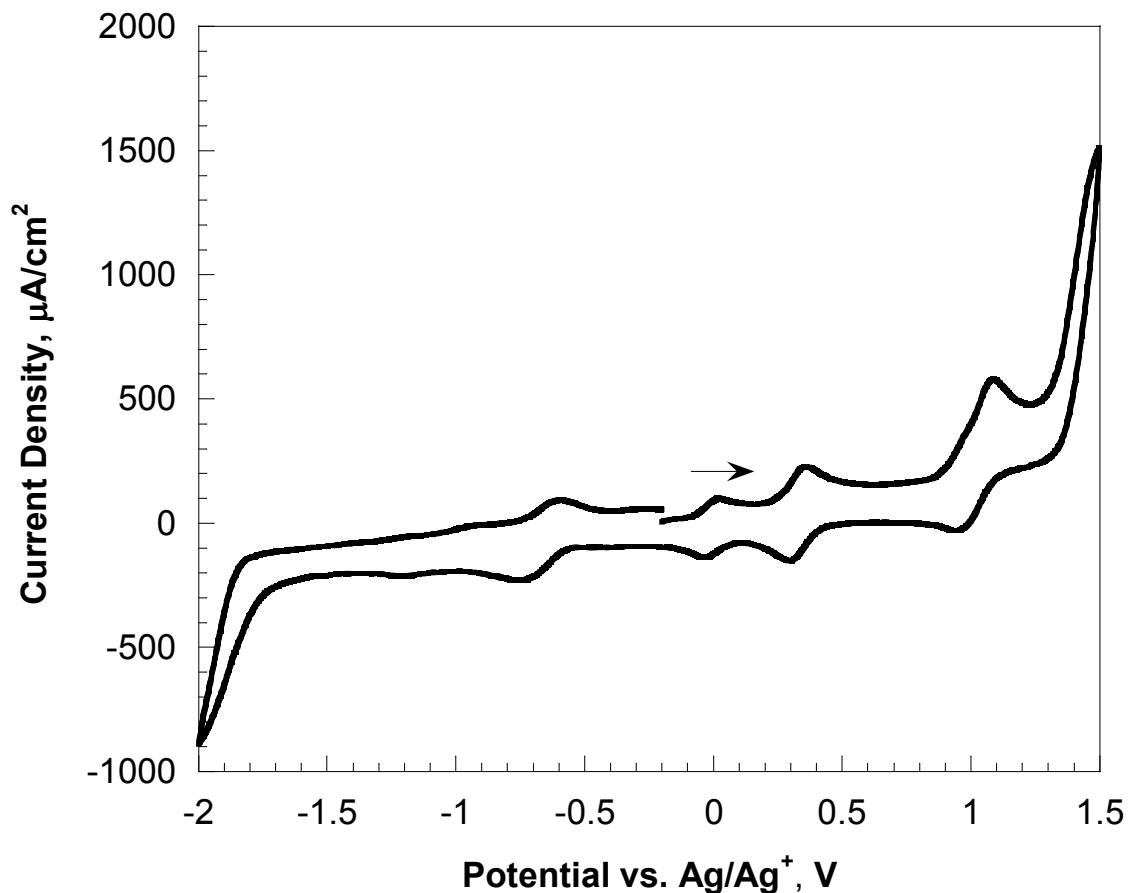


Figure S4. Cyclic voltammogram of 1 mM  $((\text{C}_6\text{F}_5)_2(p\text{-OMePh})\text{corroleFe}(\text{Et}_2\text{O})_2)$  (**9**) in  $\text{CH}_2\text{Cl}_2$ , 100 mV/s.

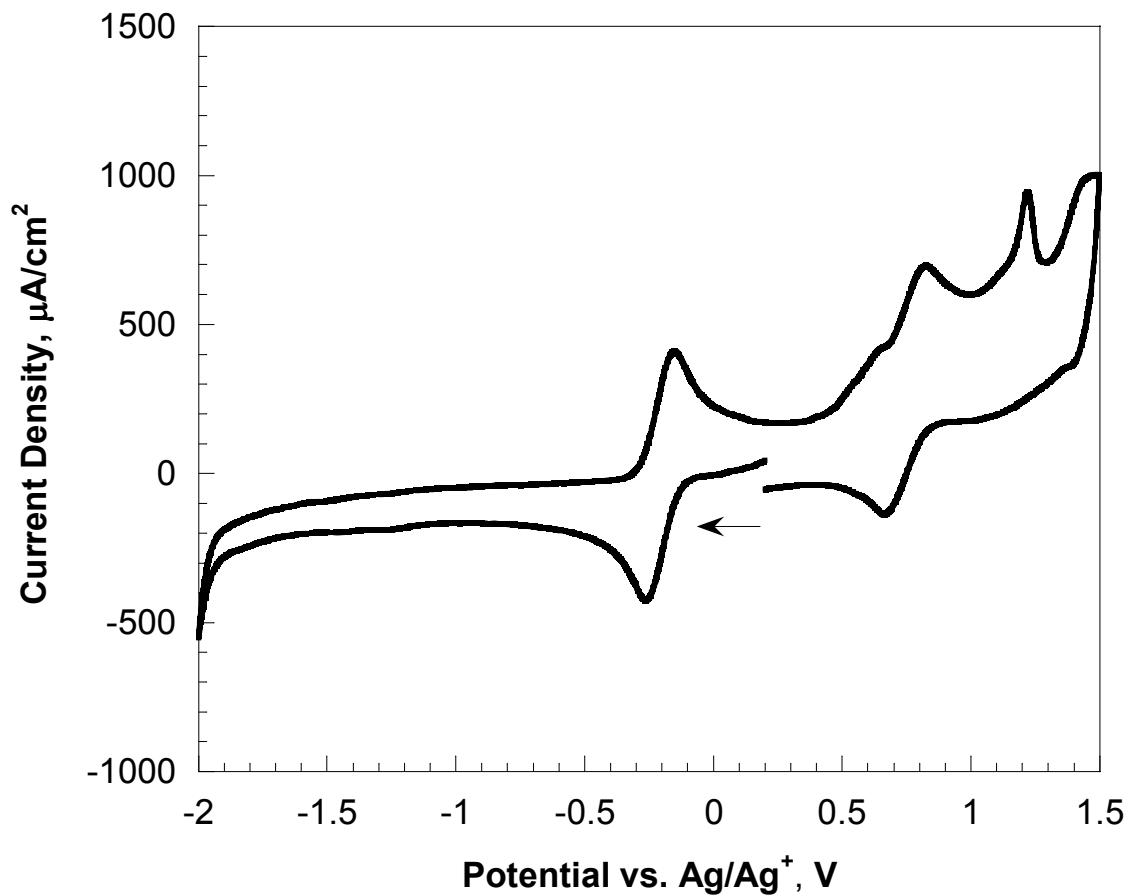


Figure S5. Cyclic voltammogram of 1 mM Cu(corrrole((C<sub>6</sub>F<sub>5</sub>)<sub>2</sub>(Ph))-*p*-O-(CH<sub>2</sub>)(C<sub>2</sub>HN<sub>3</sub>)(CH<sub>2</sub>)-*m*-(Ph(C<sub>6</sub>F<sub>5</sub>)<sub>2</sub>)corrle)Cu (**14**) corrle in CH<sub>2</sub>Cl<sub>2</sub>, 100 mV/s.

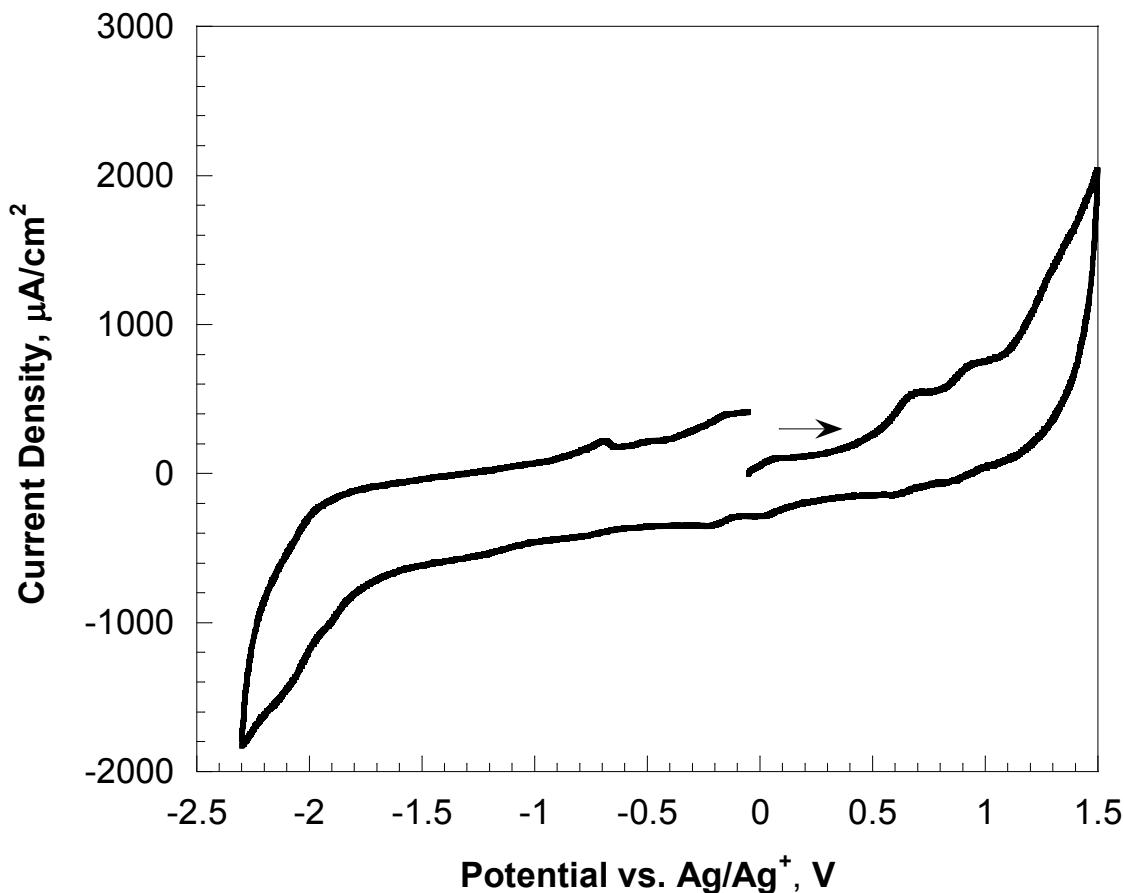


Figure S6. Cyclic voltammogram of <1 mM  $\text{Fe}(\text{CH}_3\text{CN})_2(\text{corrole}((\text{C}_6\text{F}_5)_2\text{Ph}))\text{-}p\text{-O}-(\text{CH}_2)(\text{C}_2\text{HN}_3)(\text{CH}_2)\text{-}m\text{-}(\text{Ph}(\text{C}_6\text{F}_5)_2\text{corrole})\text{Cu}$  (**15**) in acetonitrile, 500 mV/s.

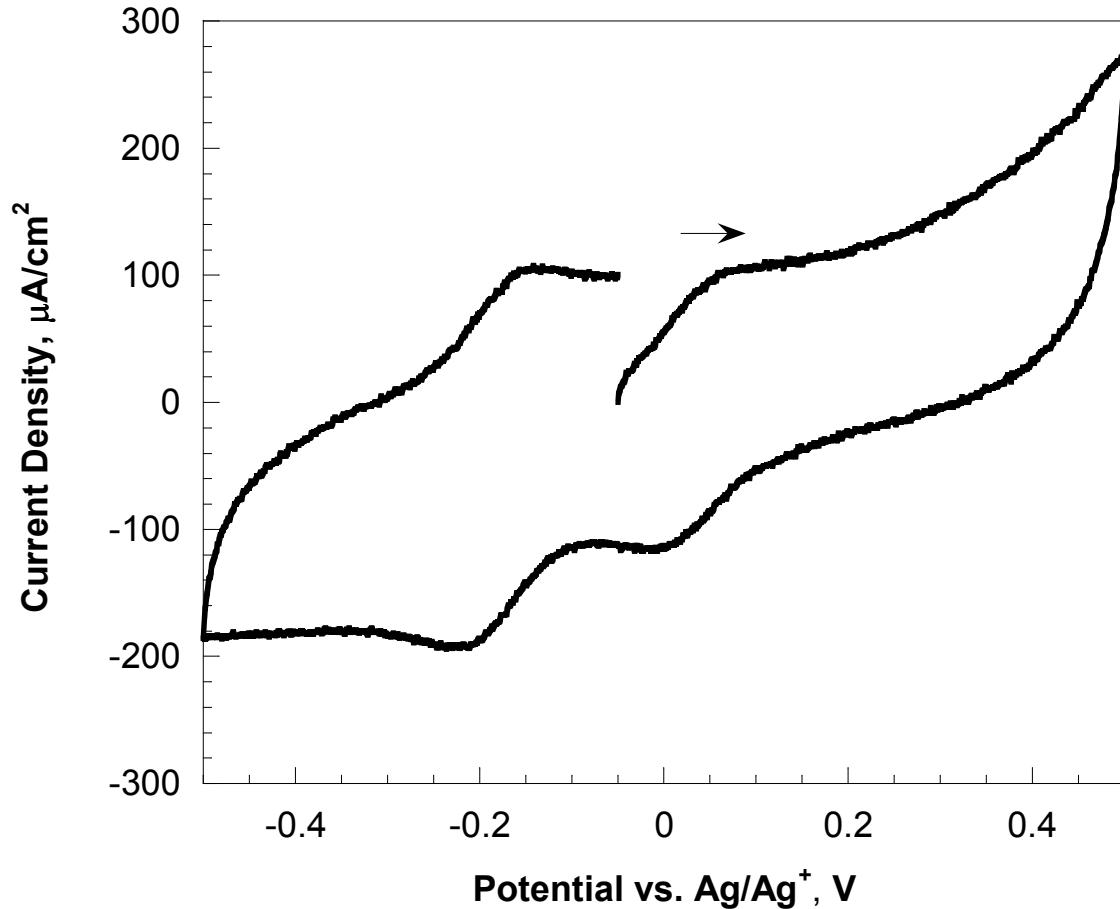


Figure S7. Cyclic voltammogram of <1 mM  $\text{Fe}(\text{CH}_3\text{CN})_2(\text{corrole}((\text{C}_6\text{F}_5)_2\text{Ph}))\text{-}p\text{-O}(\text{CH}_2)(\text{C}_2\text{HN}_3)(\text{CH}_2)\text{-}m\text{-}(\text{Ph}(\text{C}_6\text{F}_5)_2\text{corrole})\text{Cu}$  (**15**) in acetonitrile, focus on Cu and Fe peaks, 500 mV/s.

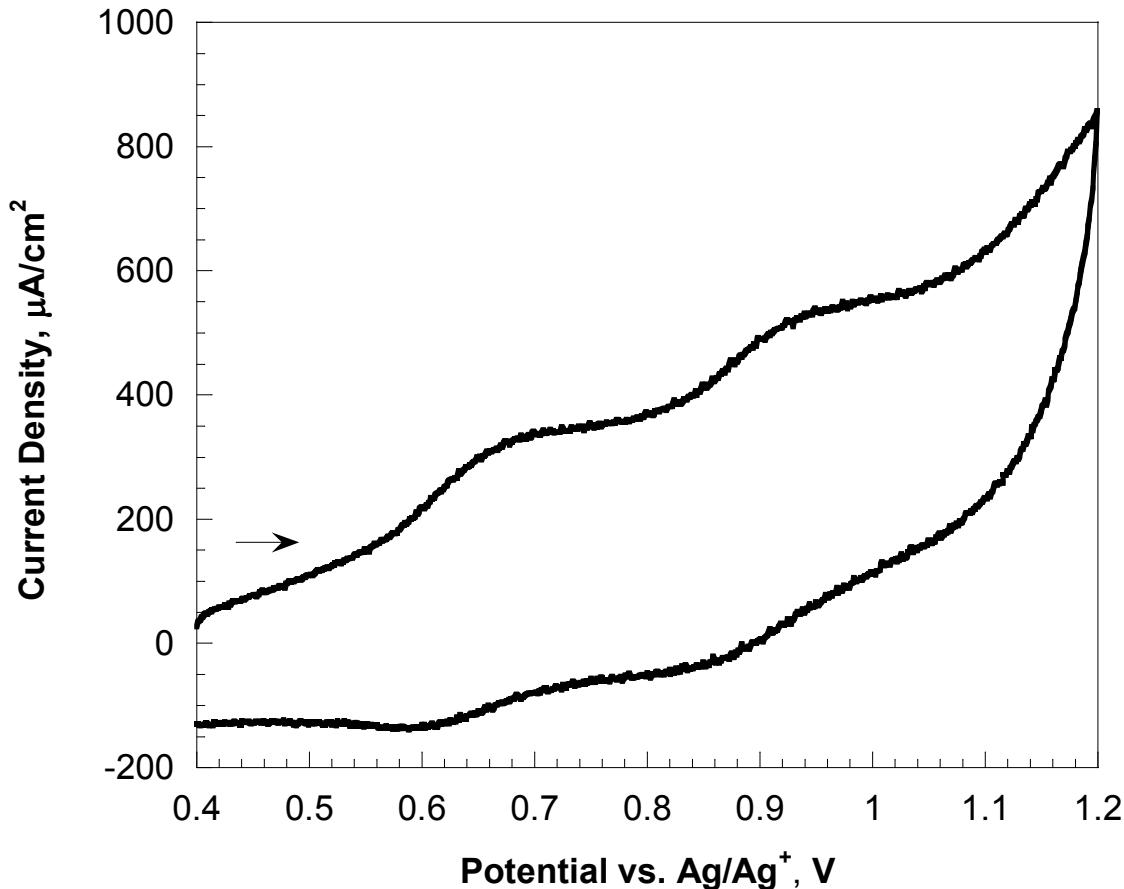


Figure S8. Cyclic voltammogram of <1 mM  $\text{Fe}(\text{CH}_3\text{CN})_2(\text{corrole}((\text{C}_6\text{F}_5)_2(\text{Ph}))\text{-}p\text{-O}\text{-}(\text{CH}_2)(\text{C}_2\text{H}_\text{N}_3)\text{(CH}_2\text{)}\text{-}m\text{-(Ph(C}_6\text{F}_5)_2\text{corrole})\text{Cu (15)}$  in acetonitrile, focus on ligand oxidation, 500 mV/s.

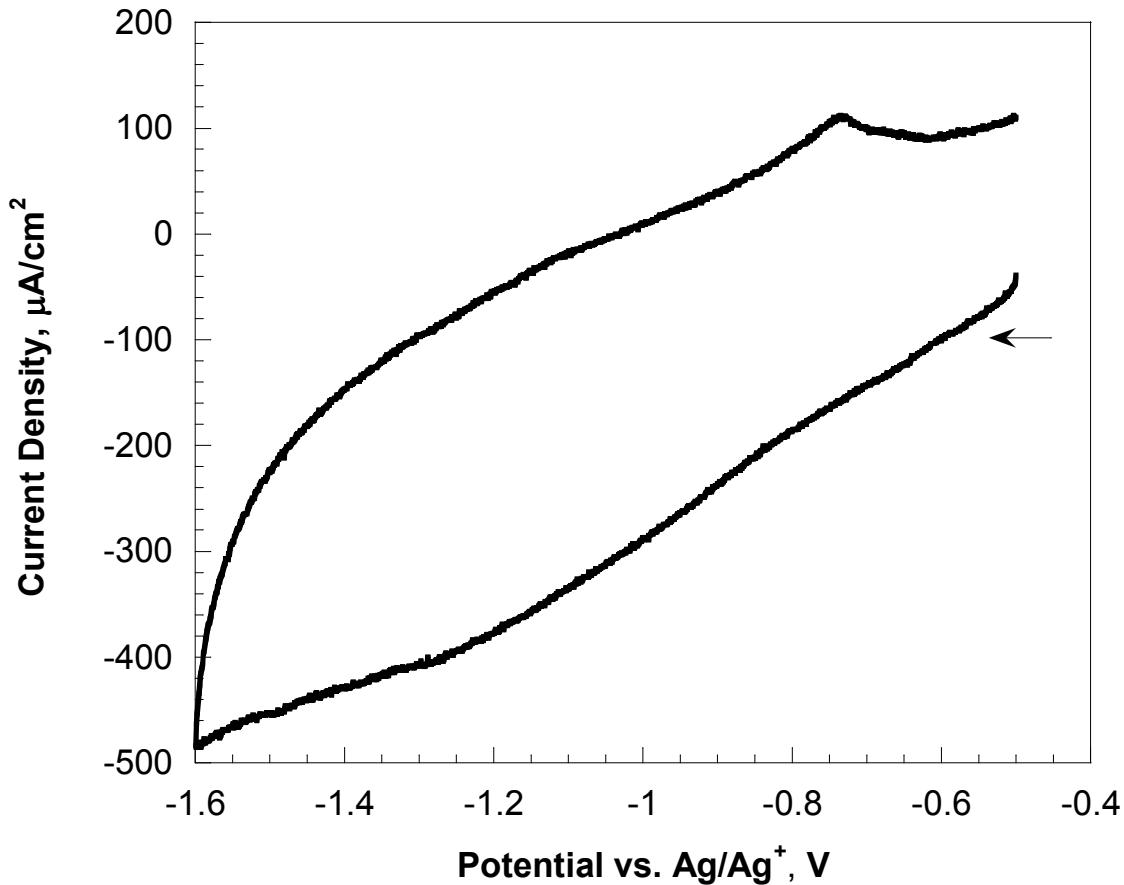


Figure S9. Cyclic voltammogram of <1 mM  $\text{Fe}(\text{CH}_3\text{CN})_2(\text{corrole}((\text{C}_6\text{F}_5)_2\text{Ph}))\text{-}p\text{-O}\text{-}(\text{CH}_2)(\text{C}_2\text{H}\text{N}_3)(\text{CH}_2)\text{-}m\text{-}(\text{Ph}(\text{C}_6\text{F}_5)_2\text{corrole})\text{Cu}$  (**15**) in acetonitrile, focus on oxidative adsorption peak, 500 mV/s.

## B. NMR Spectroscopy of compounds 1-3, 5-15

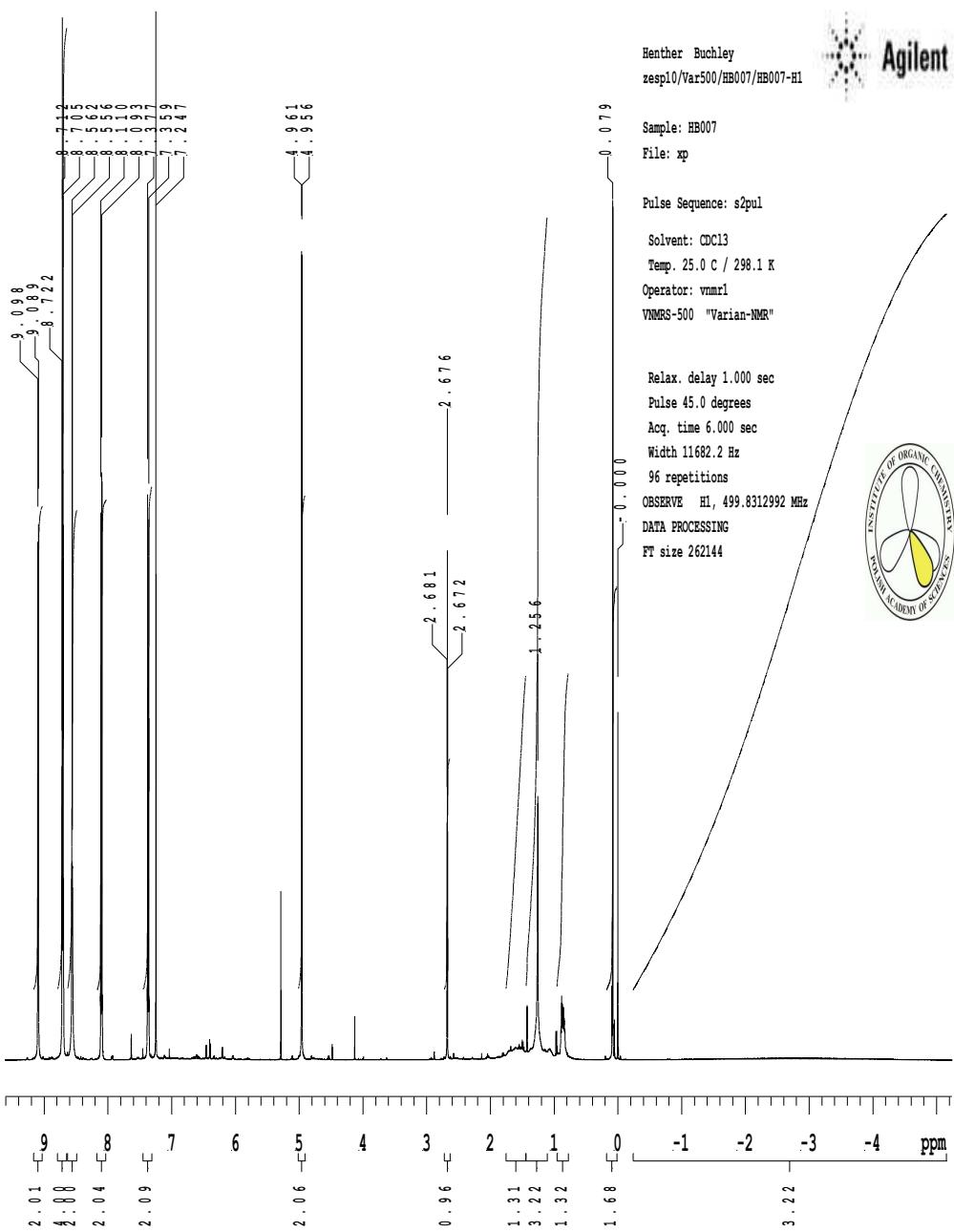


Figure S10.  $^1\text{H}$  NMR spectrum of **1** in  $\text{CDCl}_3$ .

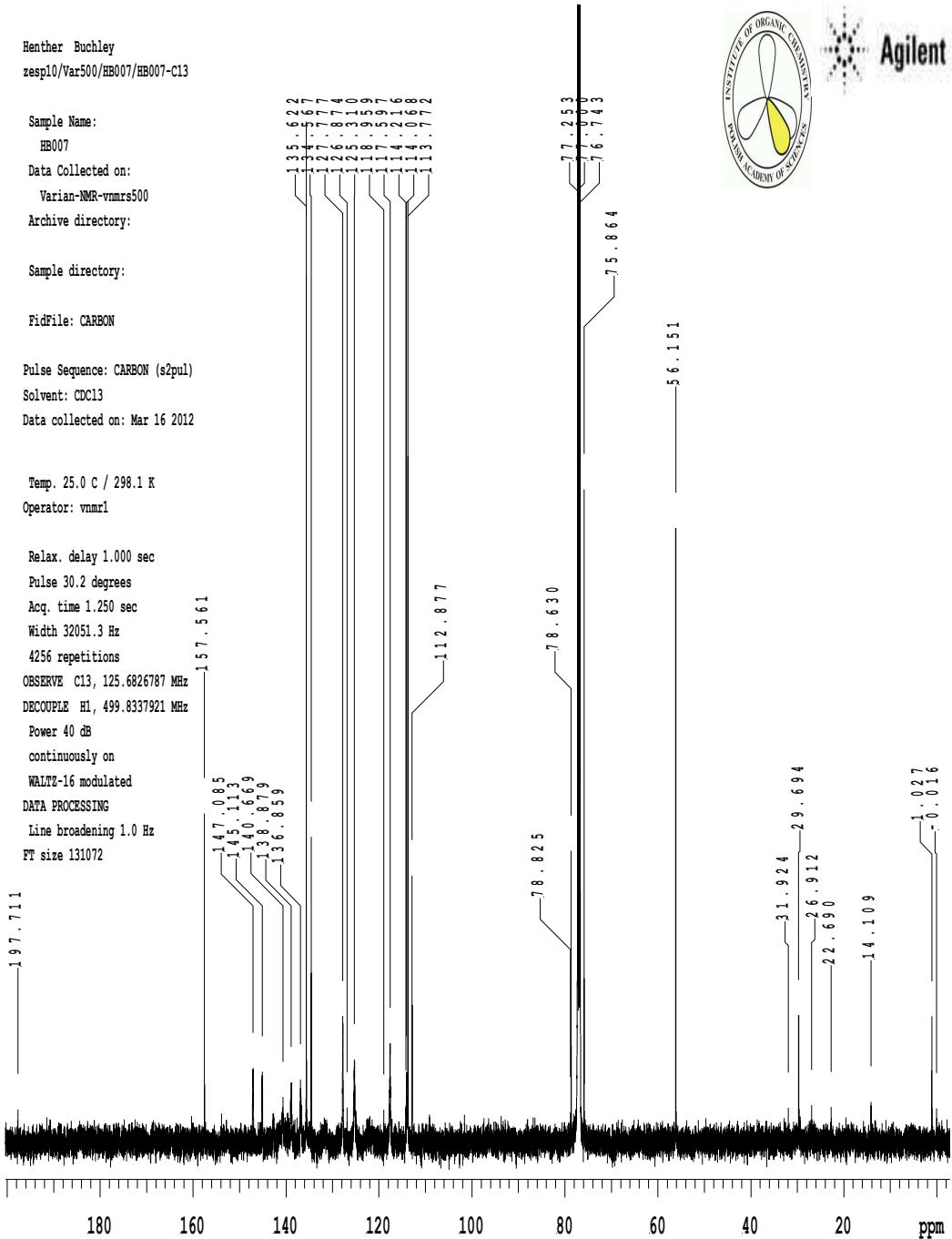


Figure S11.  $^{13}\text{C}$  NMR spectrum of **1** in  $\text{CDCl}_3$ .

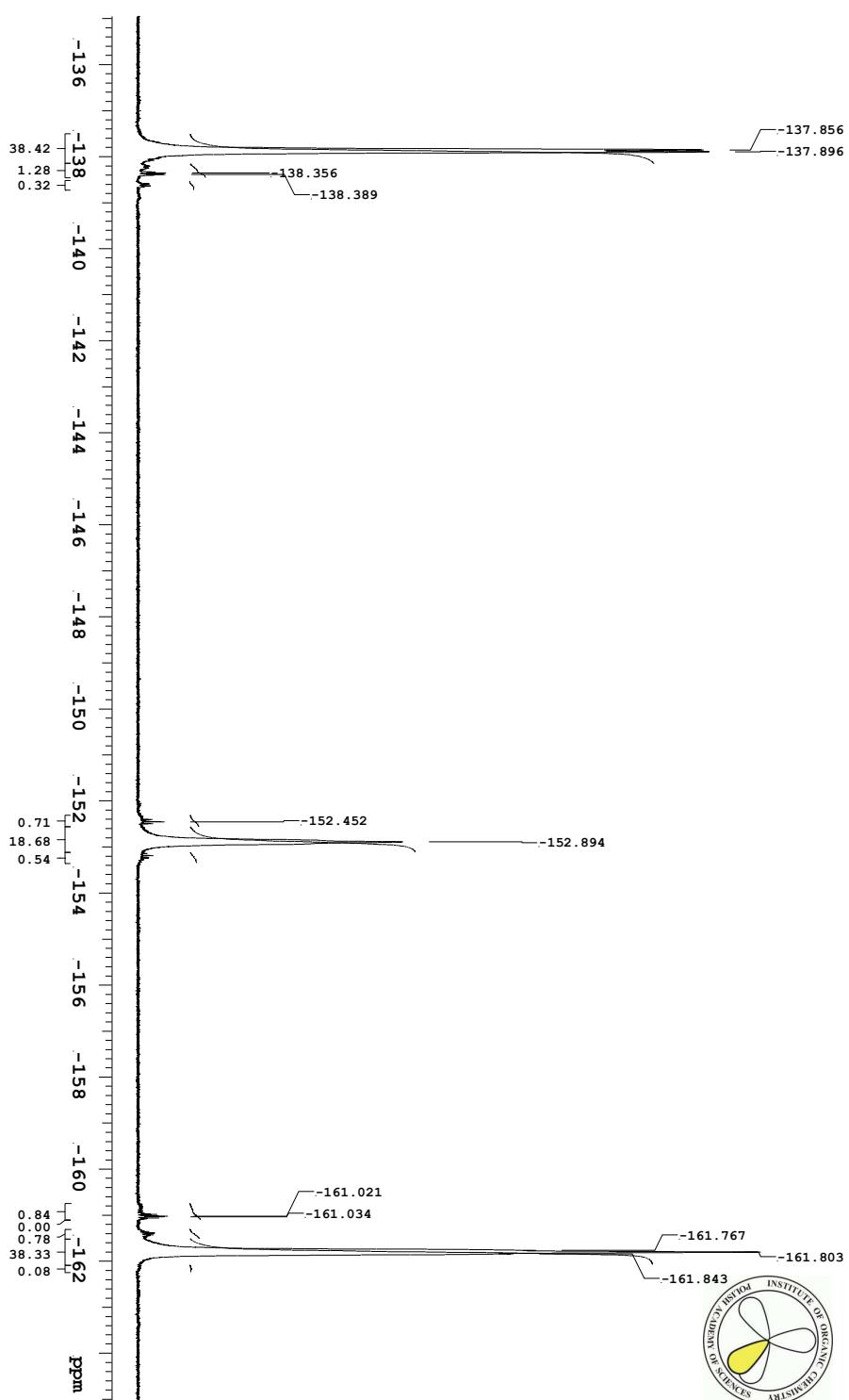


Figure S12.  ${}^{19}\text{F}$  NMR spectrum of **1** in  $\text{CDCl}_3$ .

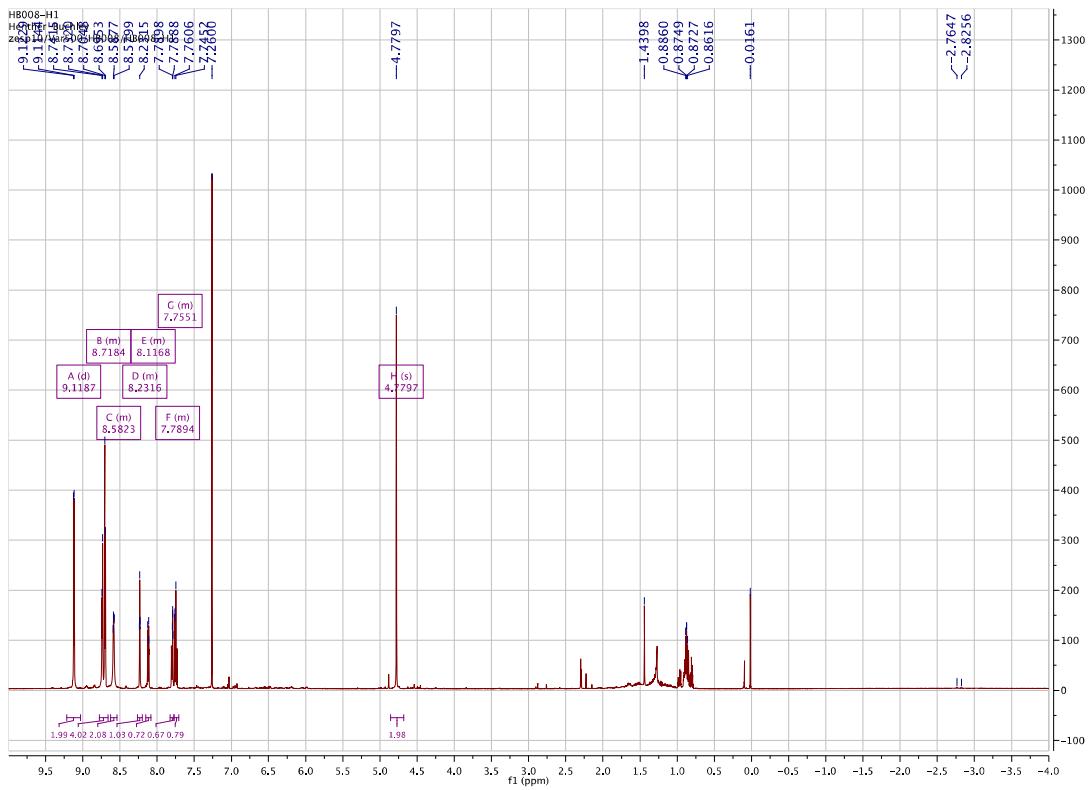


Figure S13.  $^1\text{H}$  NMR spectrum of **2** in  $\text{CDCl}_3$ .

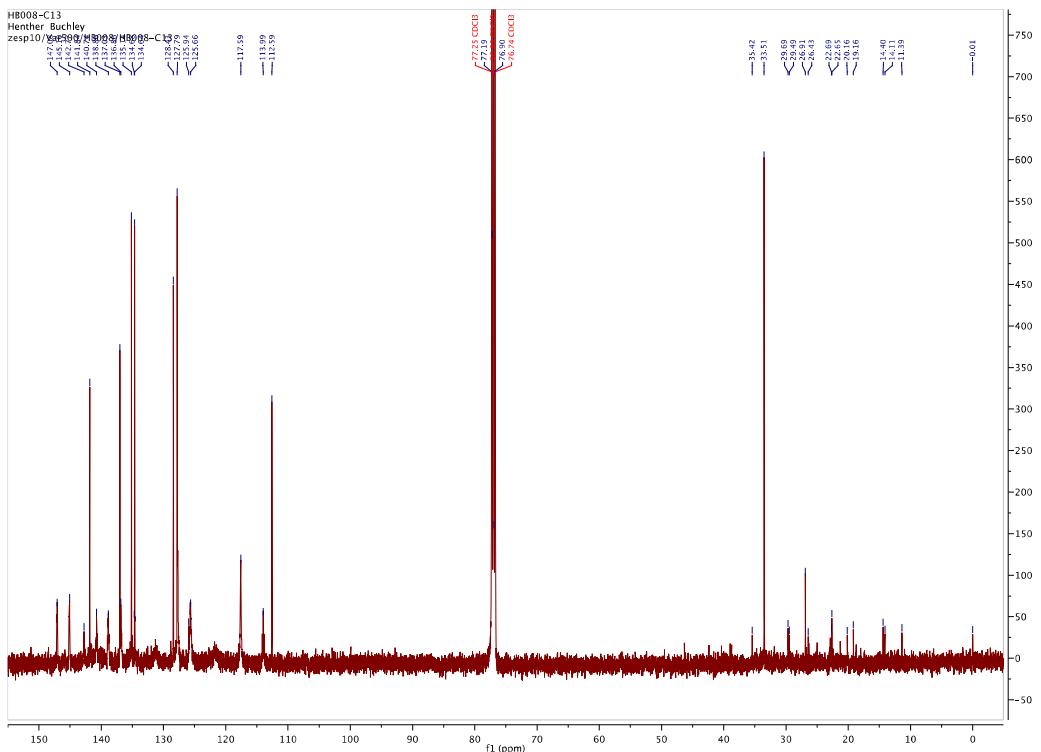


Figure S14.  $^{13}\text{C}$  NMR spectrum of **2** in  $\text{CDCl}_3$ .

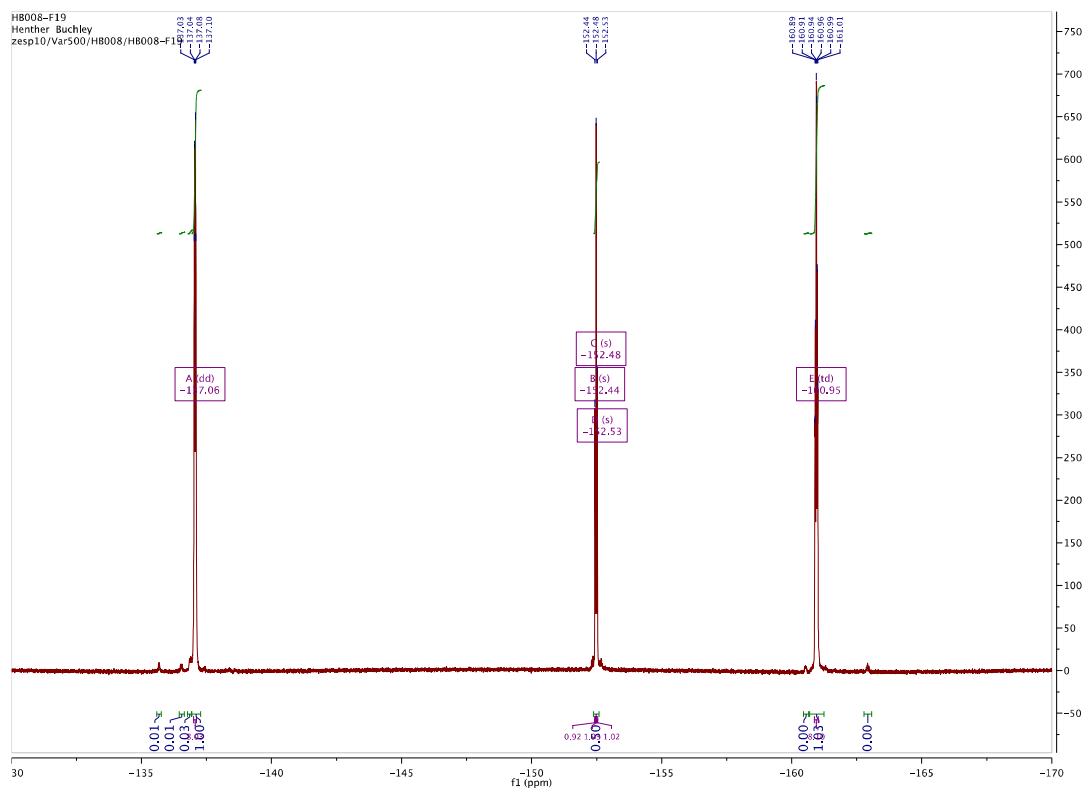


Figure S15.  $^{19}\text{F}$  NMR spectrum of **2** in  $\text{CDCl}_3$ .

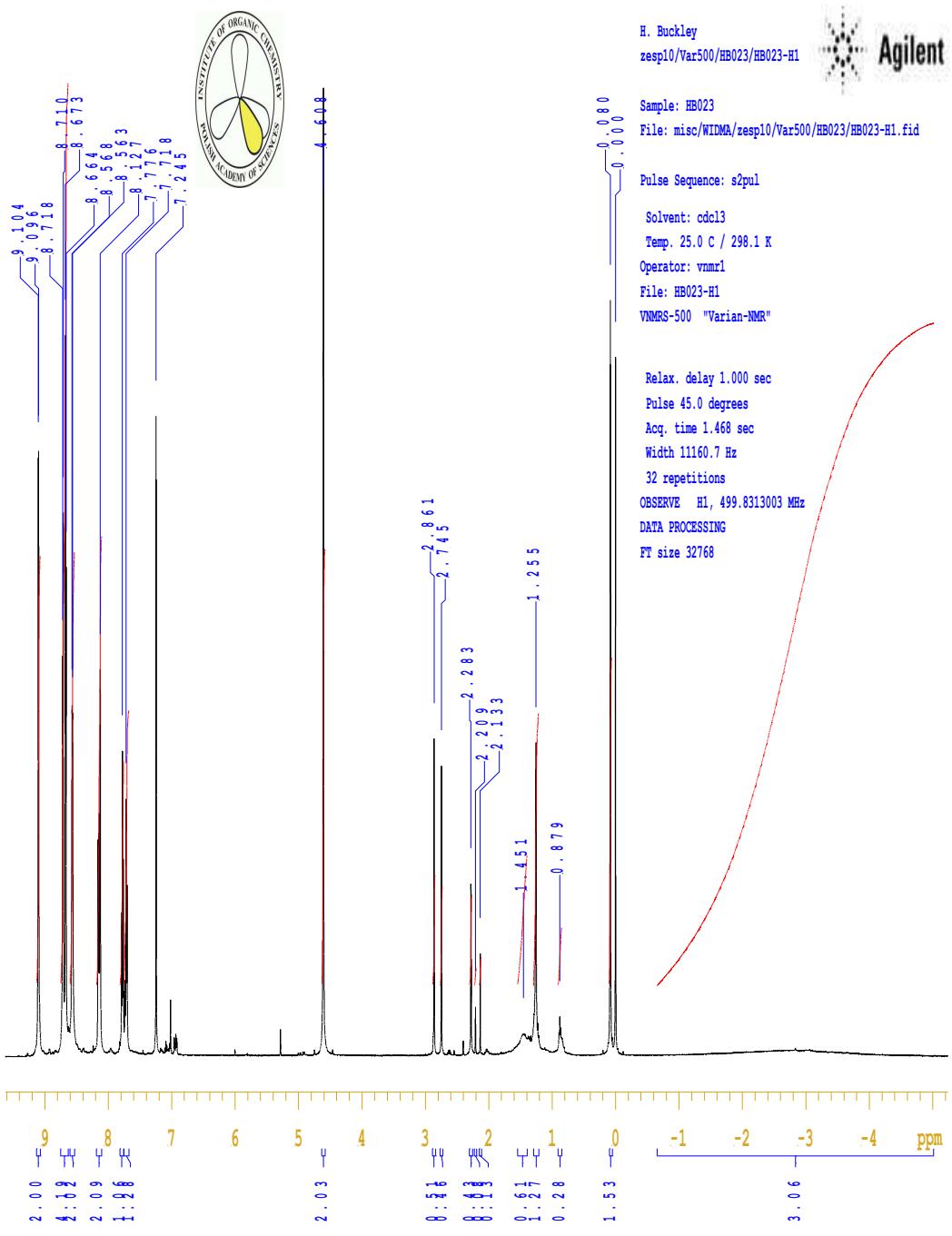


Figure S16.  $^1\text{H}$  NMR spectrum of **3** in  $\text{CDCl}_3$ .

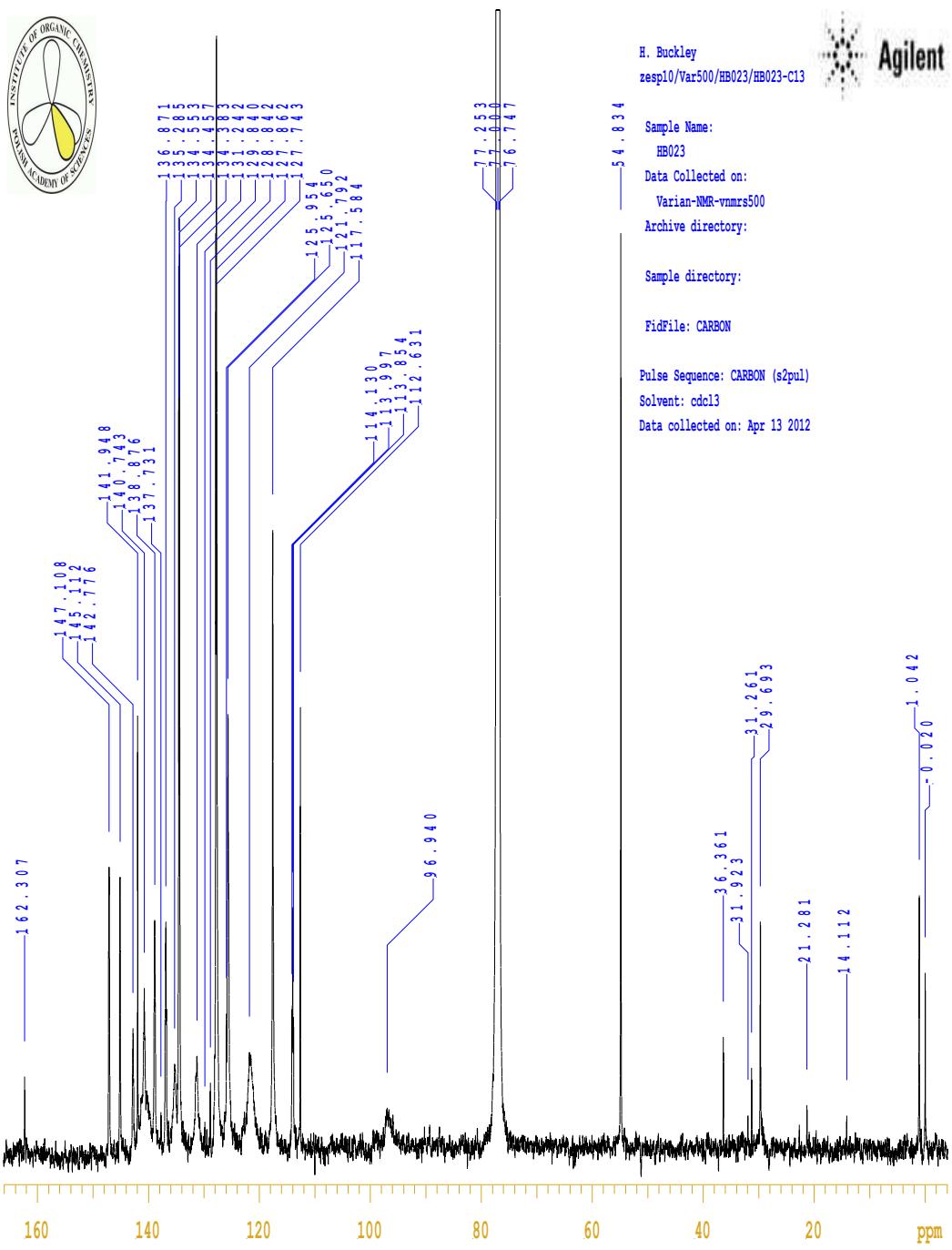


Figure S17.  $^{13}\text{C}$  NMR spectrum of **3** in  $\text{CDCl}_3$ .



H. Buckley  
zesp10/Var500/HB023/HB023-F19  
zesp10/Var500/HB023/HB023-F19



Sample Name:

302-148-19-01

Data Collected on:

Varian-NMR-vnmrs500

Archive directory:

Sample directory:

Fidfile: FLUORINE

Pulse Sequence: FLUORINE (s2pul)

Solvent: dmso

Data collected on: Apr 13 2012

Temp. 25.0 C / 298.1 K

Operator: vnmrl

Relax. delay 1.000 sec

Pulse 30.0 degrees

Acq. time 0.524 sec

Width 15625.0 Hz

128 repetitions

OBSERVE F19, 470.3135458 MHz

DATA PROCESSING

Line broadening 0.5 Hz

FT size 16384

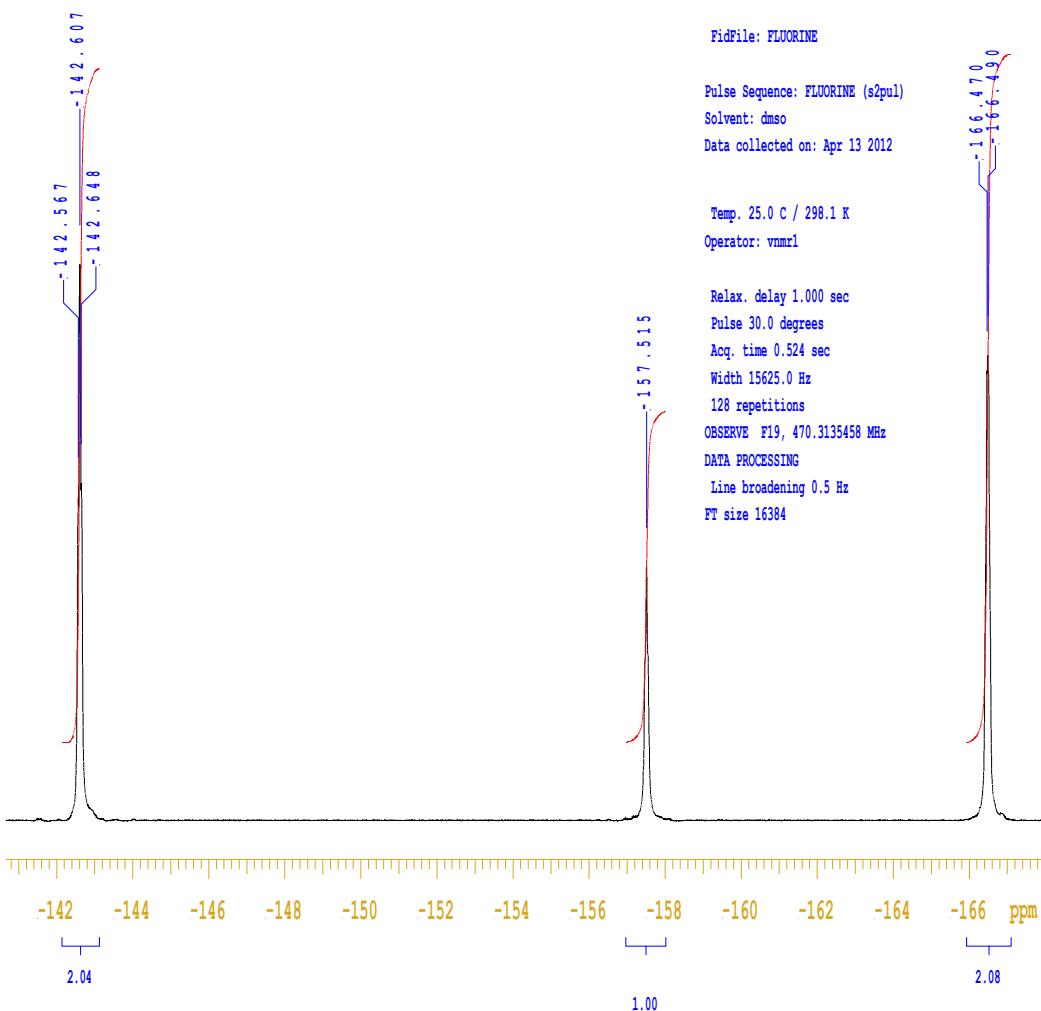


Figure S18. <sup>19</sup>F NMR spectrum of **3** in  $\text{CDCl}_3$ .

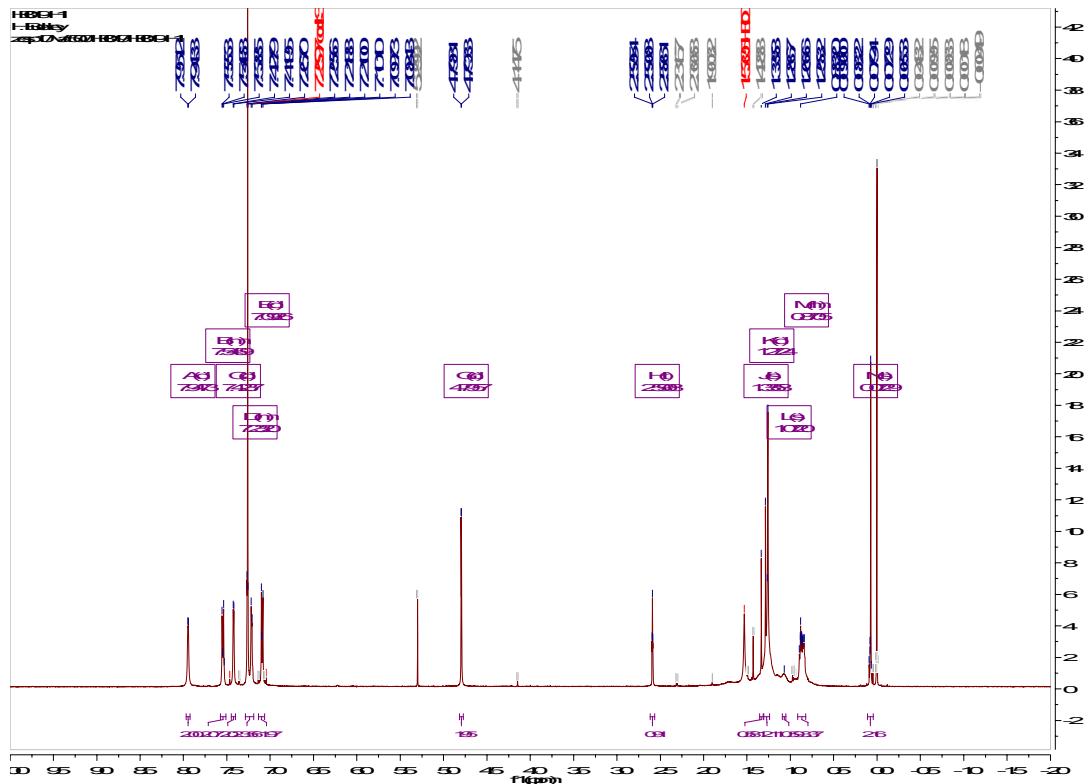


Figure S19.  $^1\text{H}$  NMR spectrum of **5** in  $\text{CDCl}_3$ .

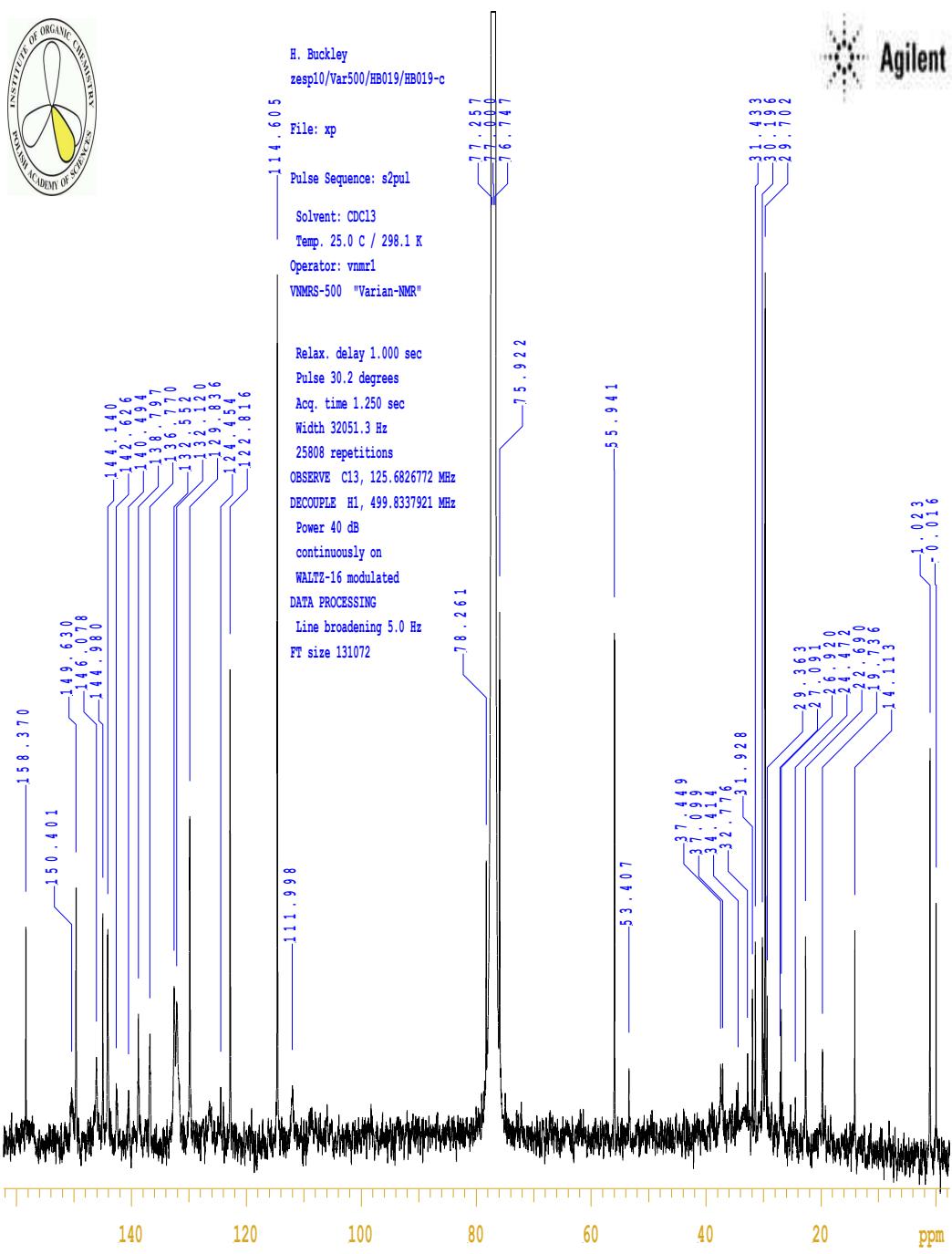


Figure S20. <sup>13</sup>C NMR spectrum of **5** in CDCl<sub>3</sub>.

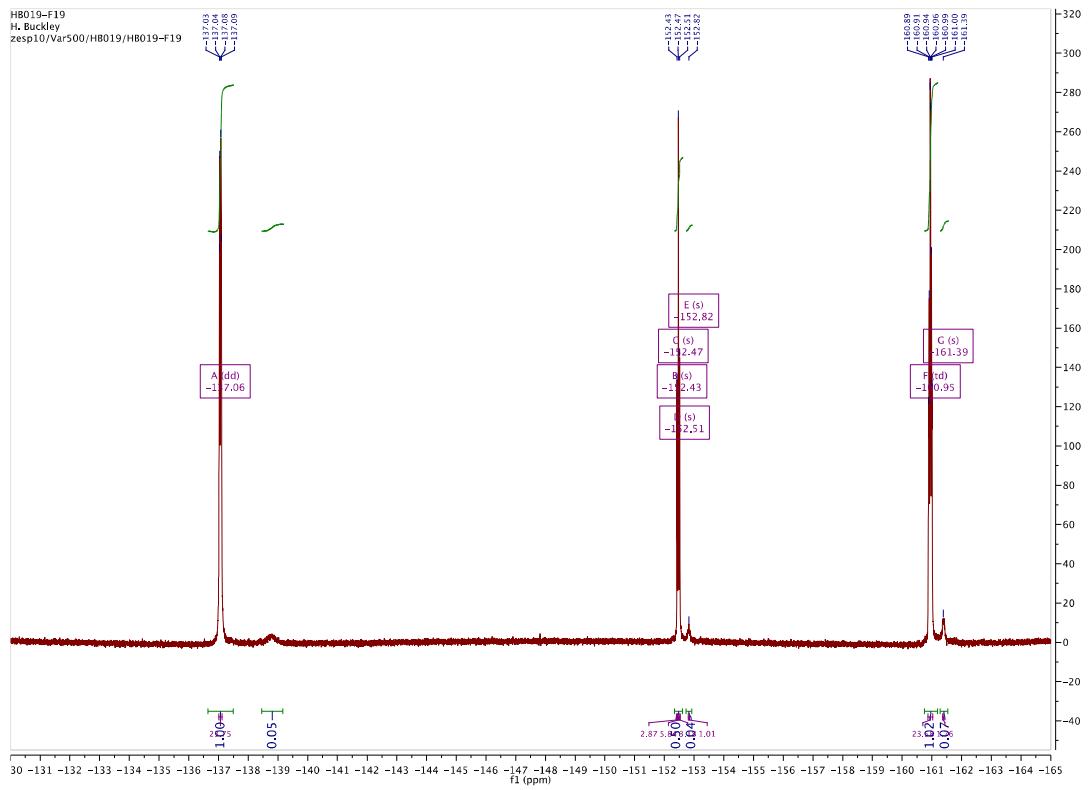


Figure S21.  $^{19}\text{F}$  NMR spectrum of **5** in  $\text{CDCl}_3$ .

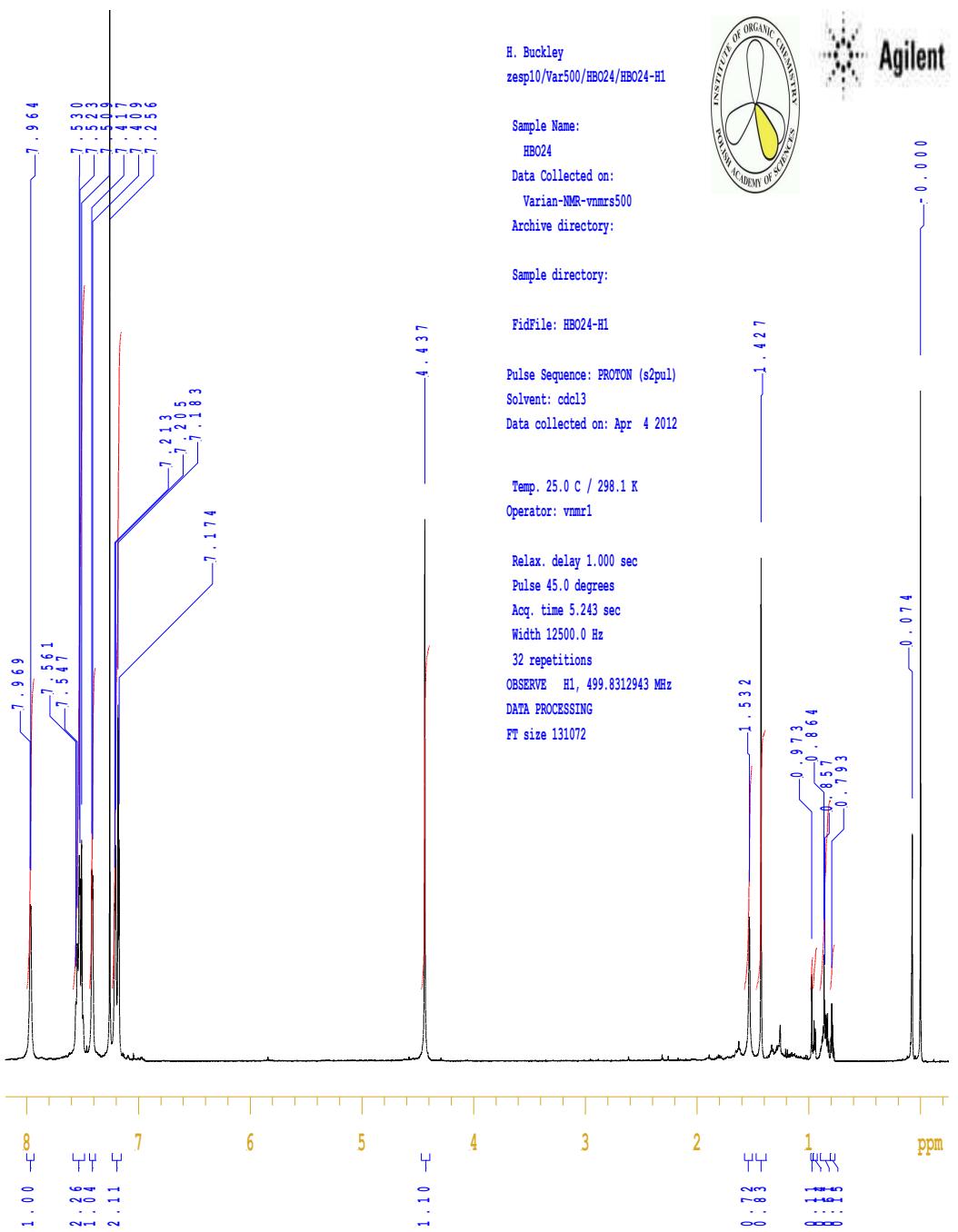


Figure S22.  $^1\text{H}$  NMR spectrum of **6** in  $\text{CDCl}_3$ .

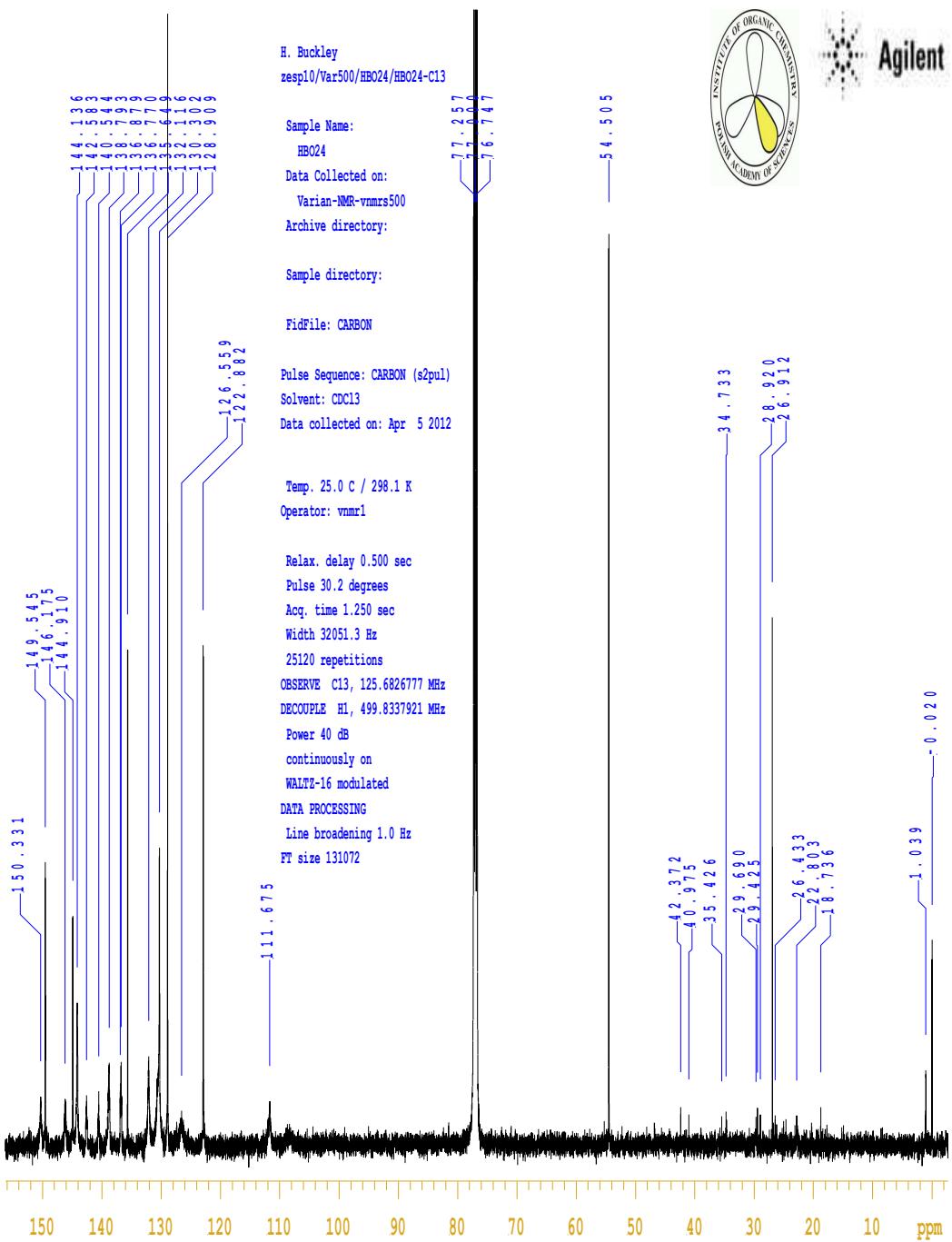


Figure S23.  $^{13}\text{C}$  NMR spectrum of **6** in  $\text{CDCl}_3$ .

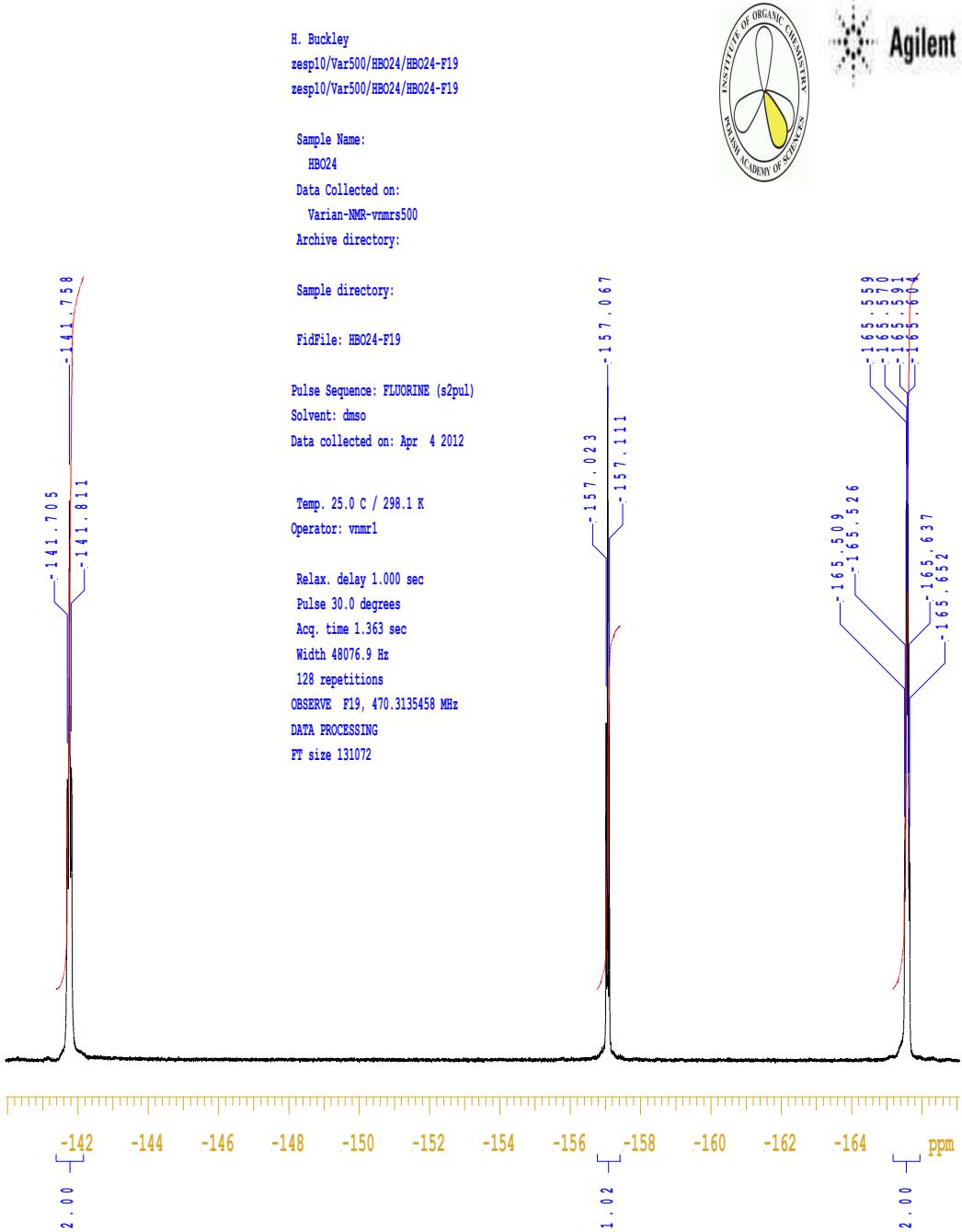


Figure S24.  $^{19}\text{F}$  NMR spectrum of **6** in  $\text{CDCl}_3$ .

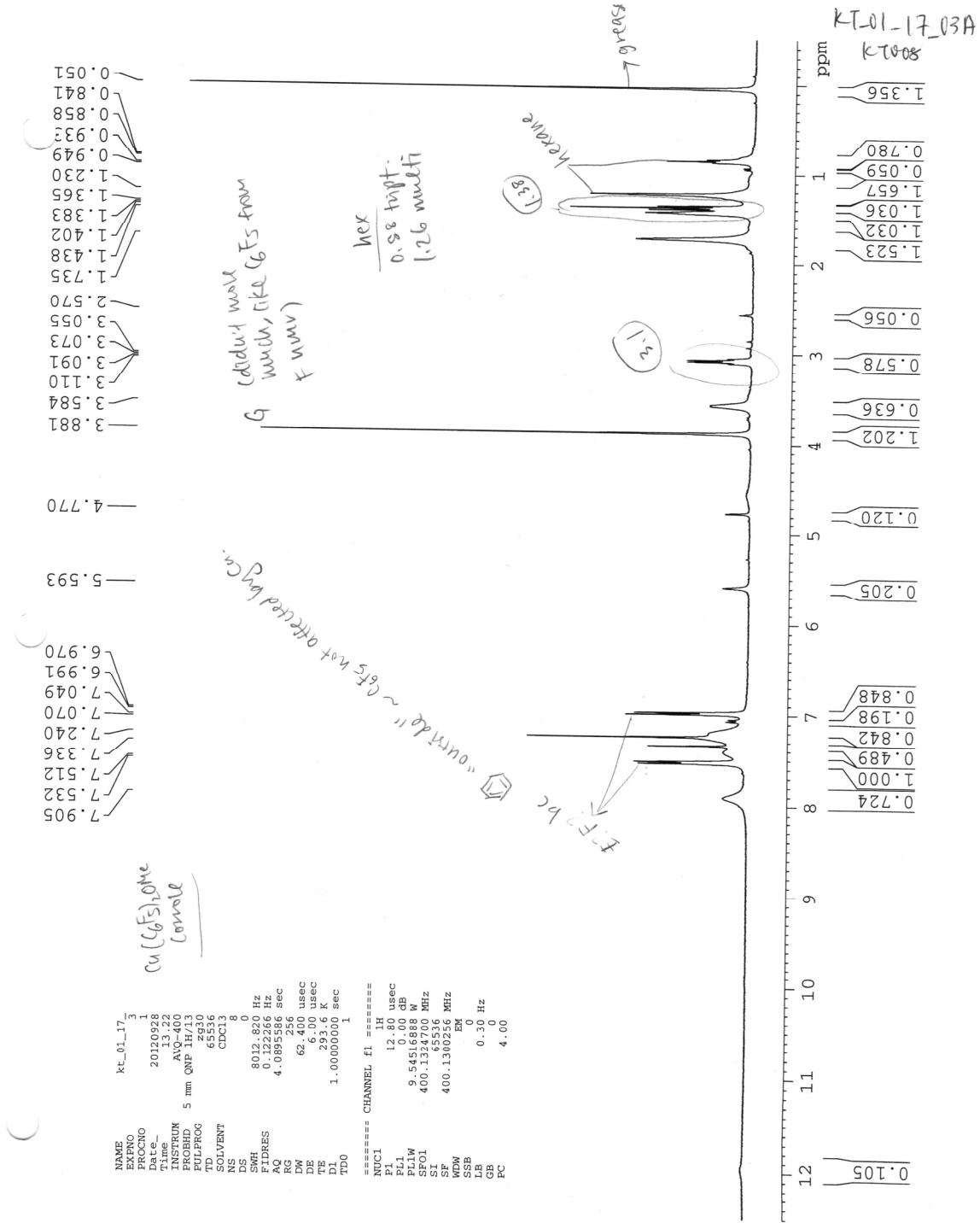


Figure S25.  $^1\text{H}$  NMR spectrum of **7** in  $\text{CDCl}_3$ .

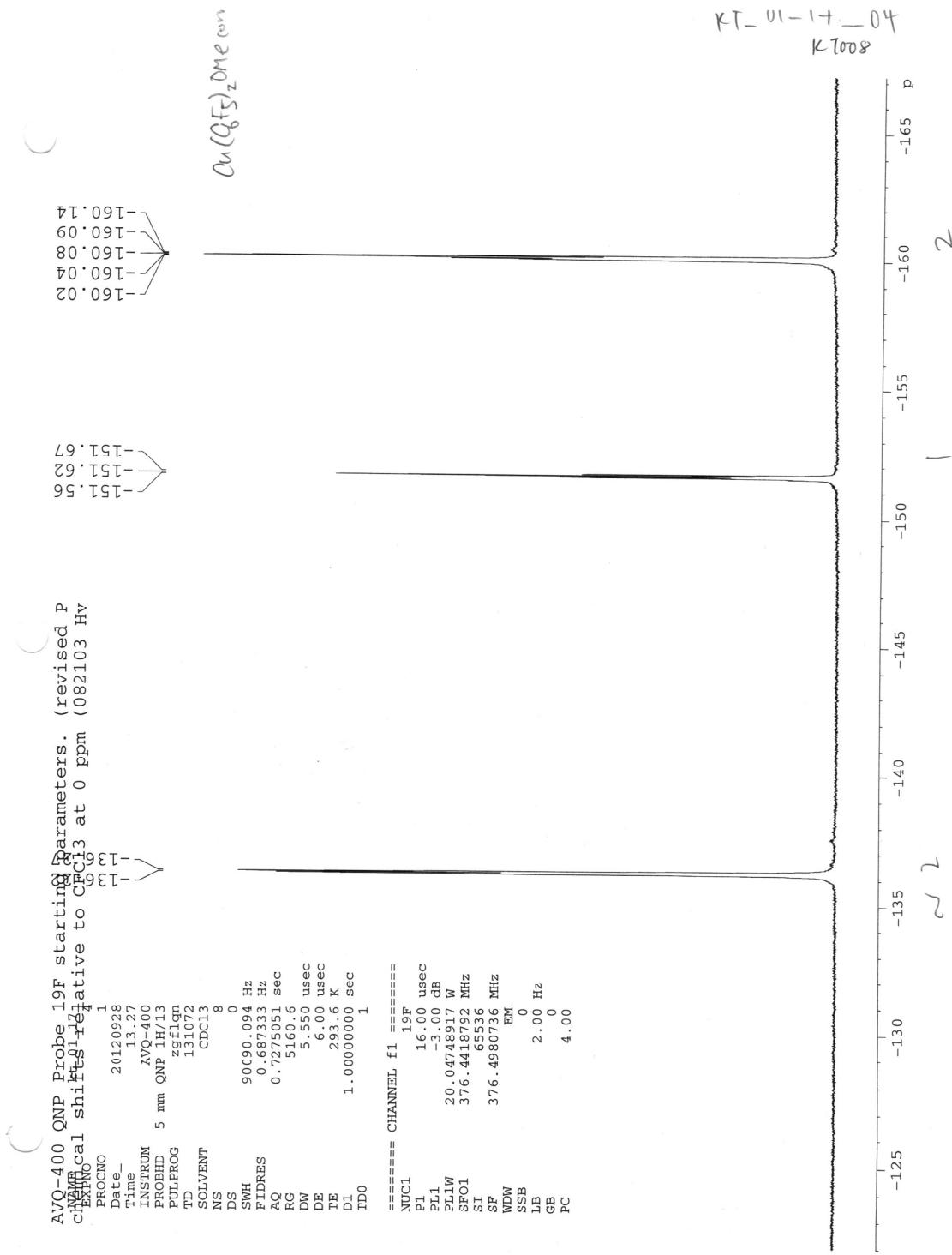


Figure S26.  $^{19}\text{F}$  NMR spectrum of **7** in  $\text{CDCl}_3$ .

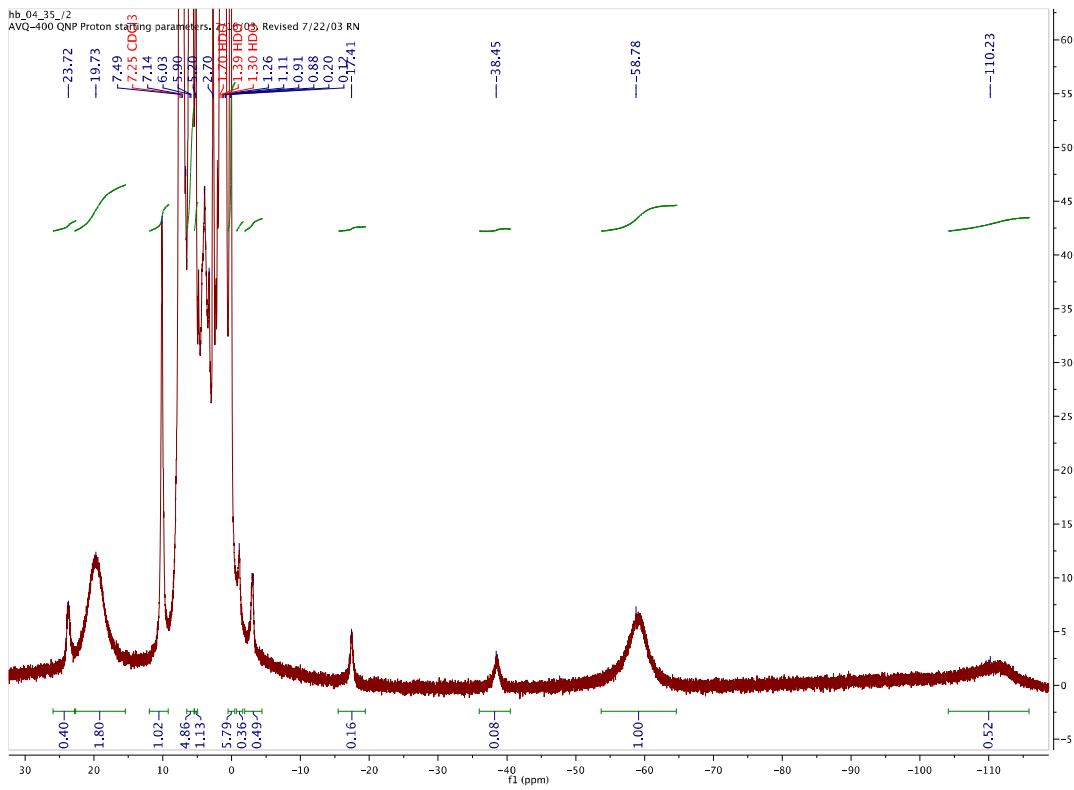


Figure S27.  $^1\text{H}$  NMR spectrum of **8** in  $\text{C}_6\text{D}_6$  (broad window).

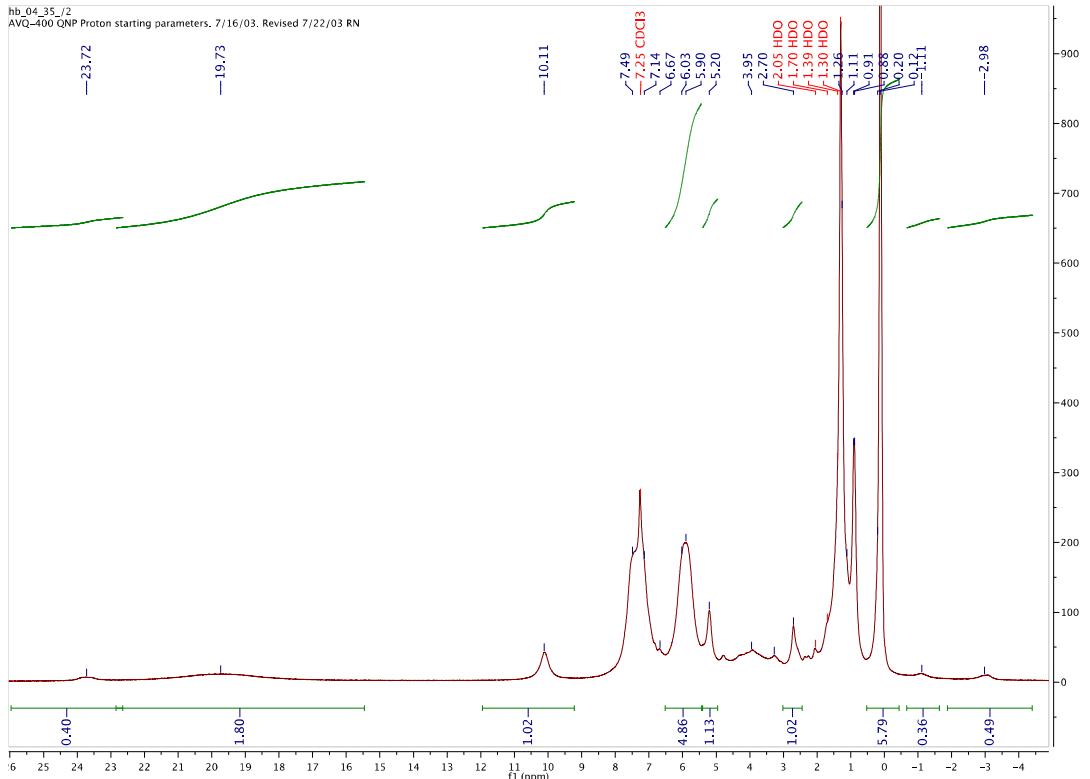


Figure S28.  $^1\text{H}$  NMR spectrum of **8** in  $\text{C}_6\text{D}_6$  (narrow window).

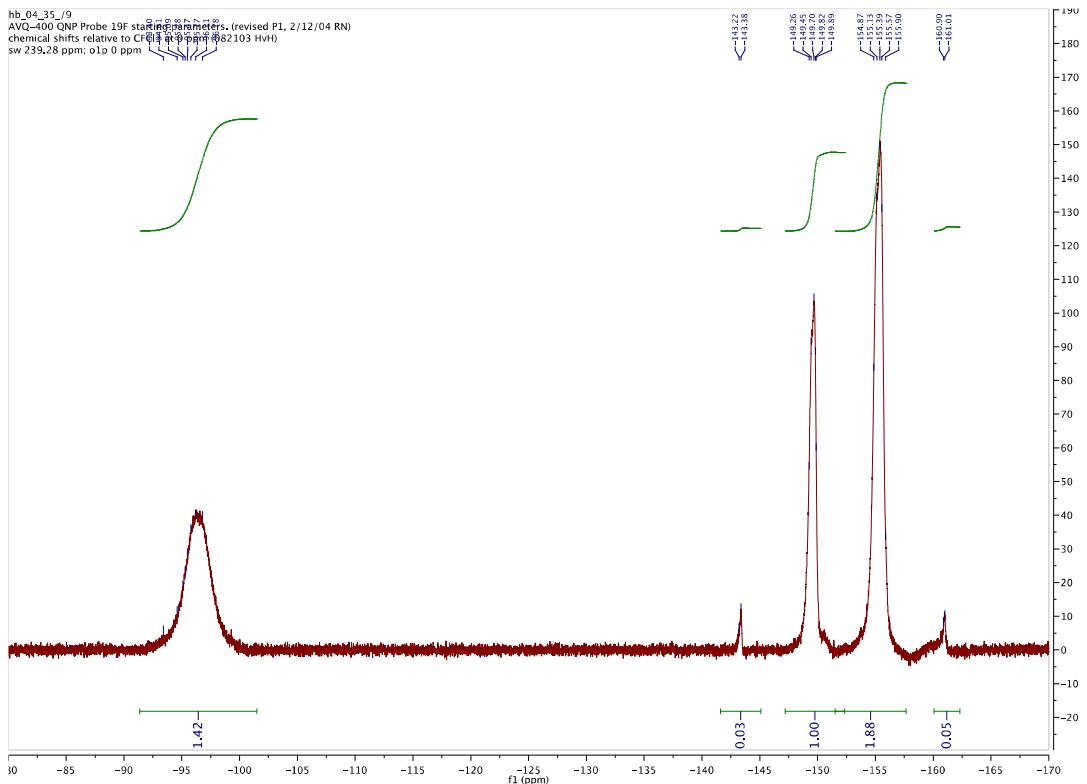


Figure S29.  $^{19}\text{F}$  NMR spectrum of **8** in  $\text{C}_6\text{D}_6$ .

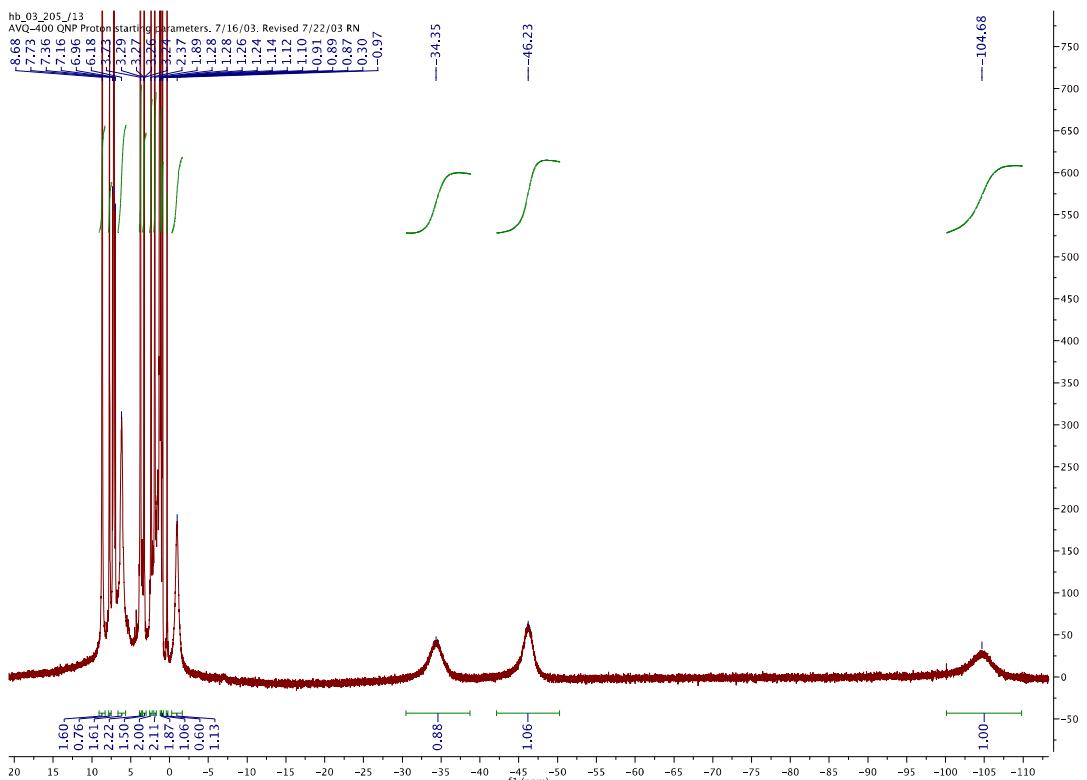


Figure S30.  $^1\text{H}$  NMR spectrum of **9** in  $\text{C}_6\text{D}_6$  (broad window)

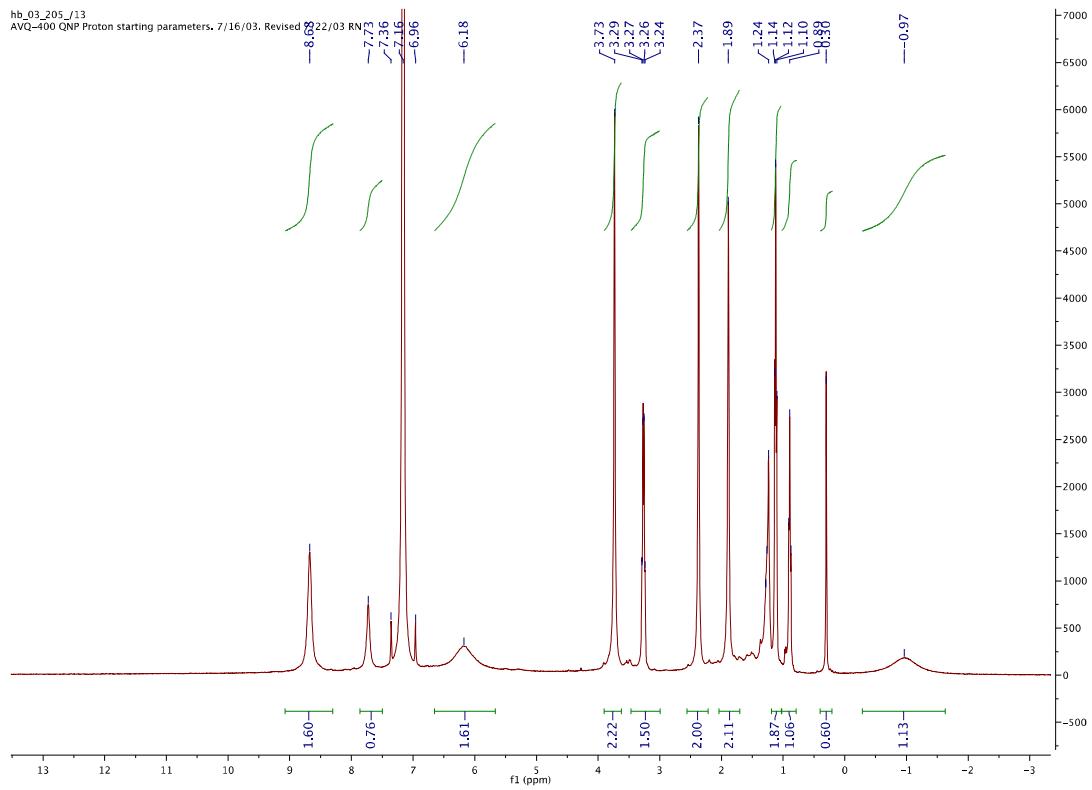


Figure S31. <sup>1</sup>H NMR spectrum of **9** in C<sub>6</sub>D<sub>6</sub> (narrow window).

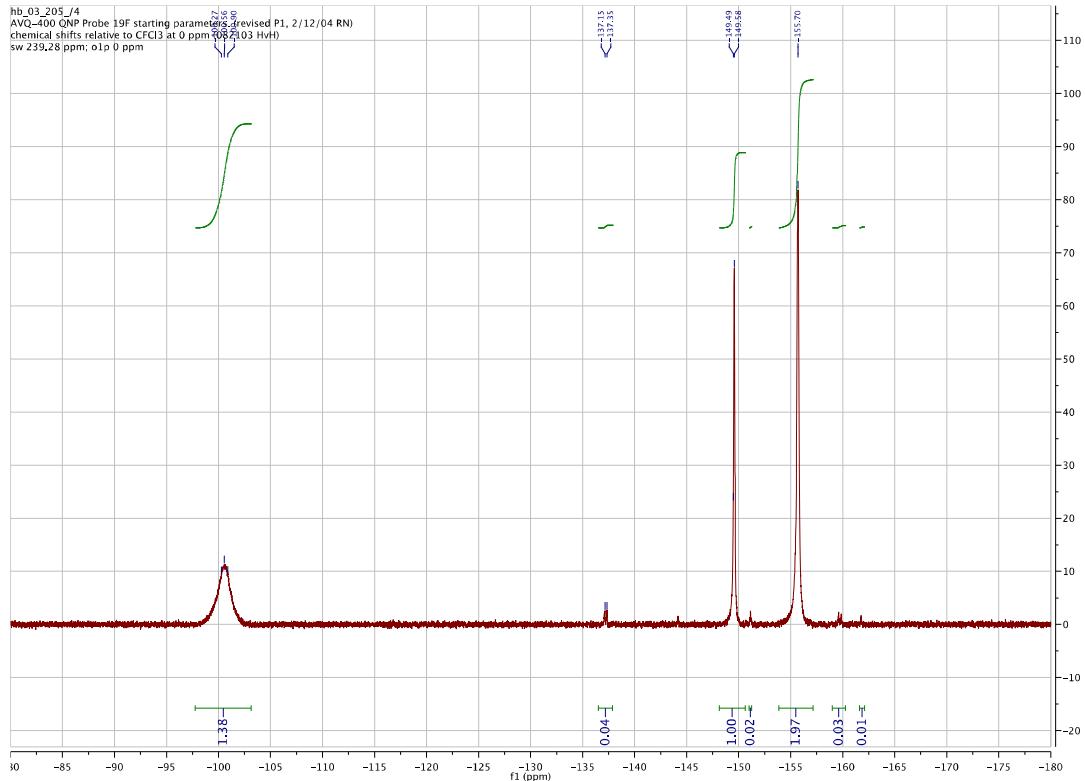


Figure S32. <sup>19</sup>F NMR spectrum of **9** in C<sub>6</sub>D<sub>6</sub>.

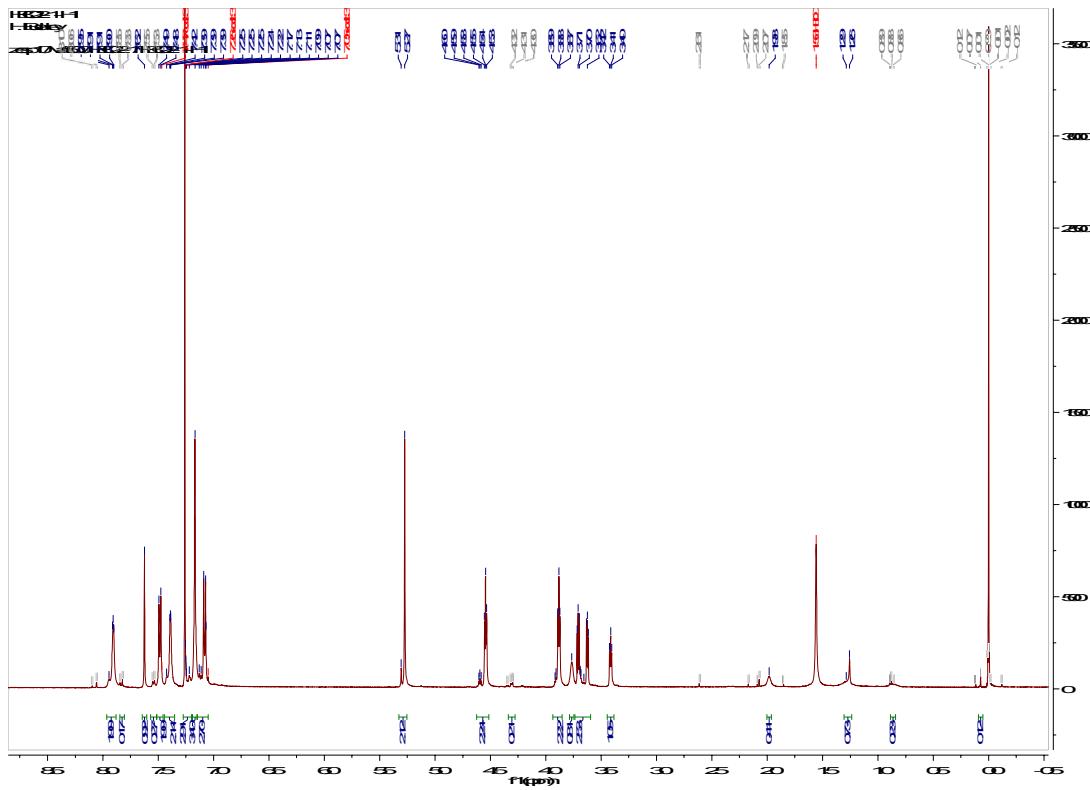


Figure S33.  $^1\text{H}$  NMR spectrum of **10** in  $\text{CDCl}_3$ .

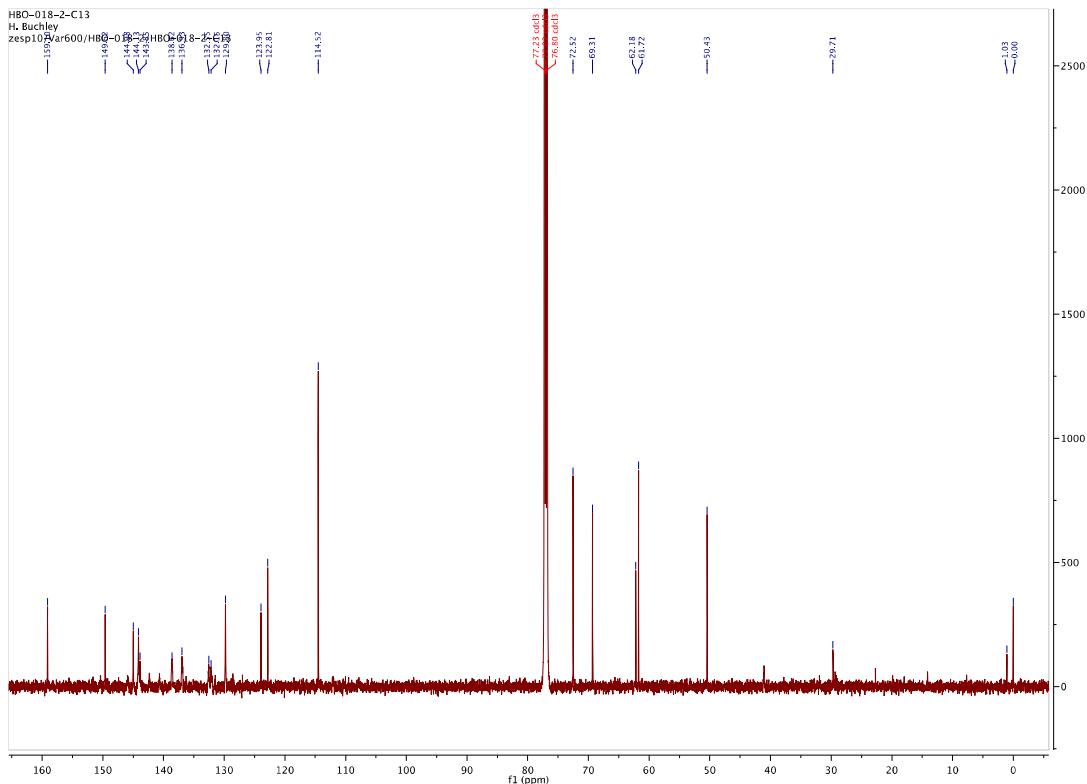


Figure S34.  $^{13}\text{C}$  NMR spectrum of **10** in  $\text{CDCl}_3$ .

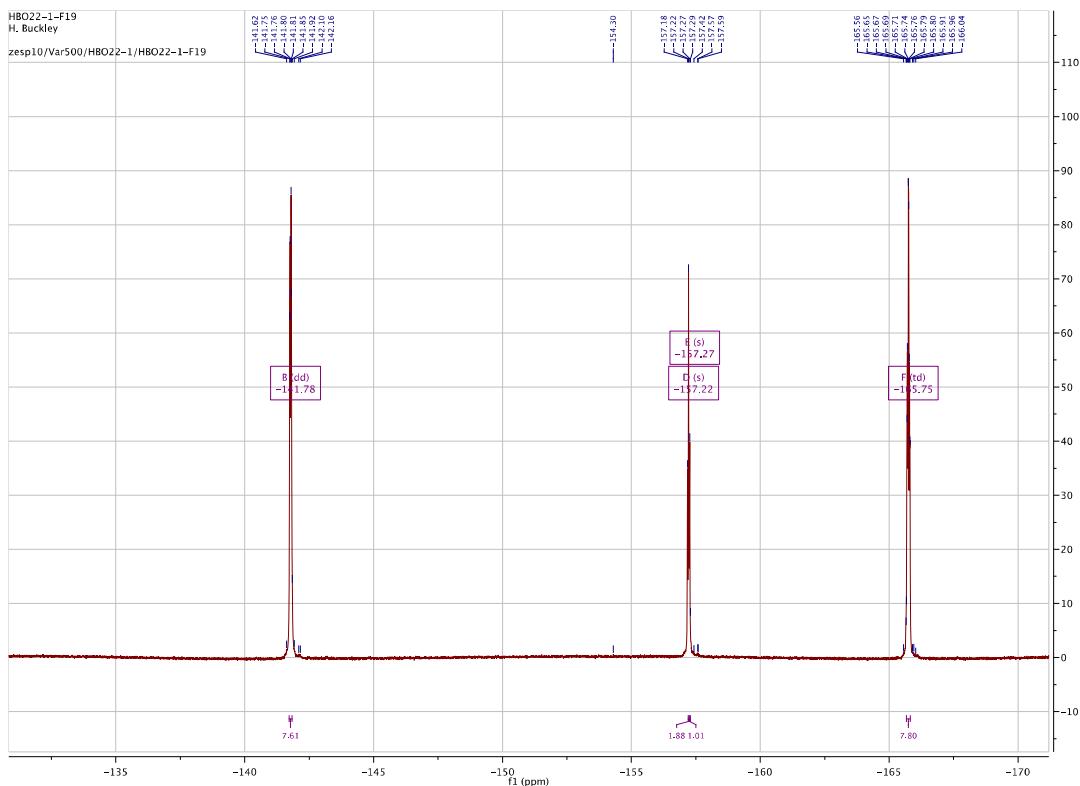


Figure S35.  $^{19}\text{F}$  NMR spectrum of **10** in  $\text{CDCl}_3$ .

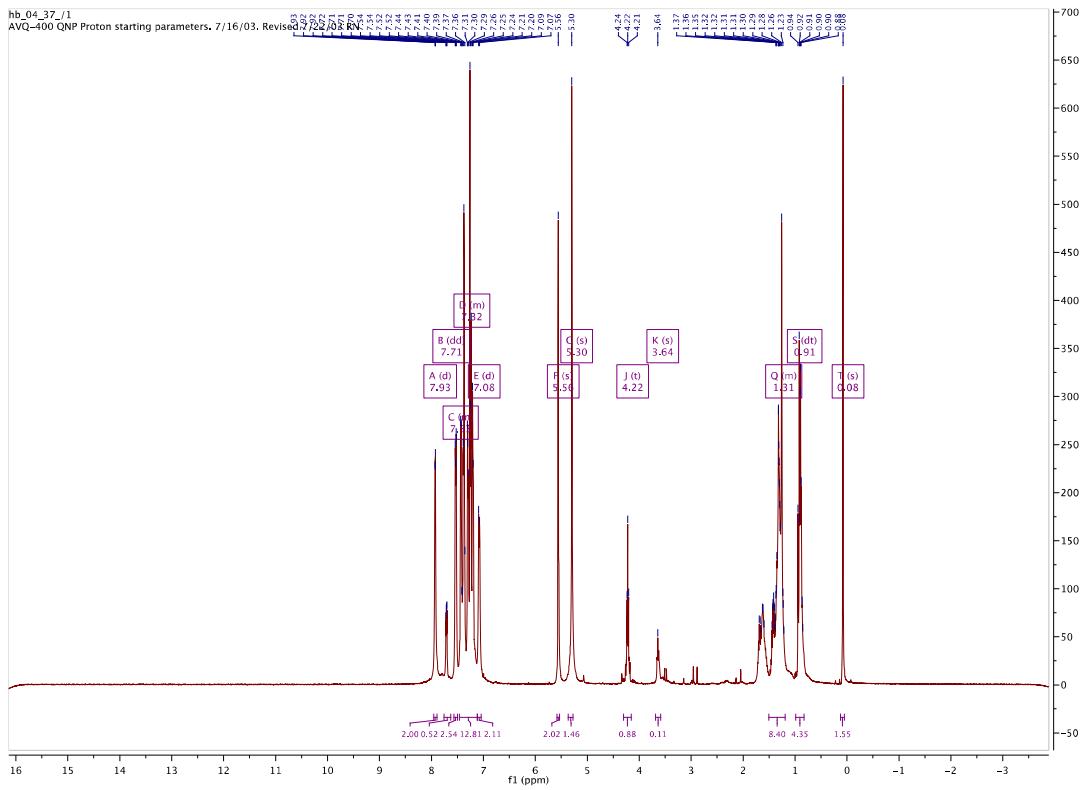


Figure S36.  $^1\text{H}$  NMR spectrum of **11** in  $\text{CDCl}_3$ .

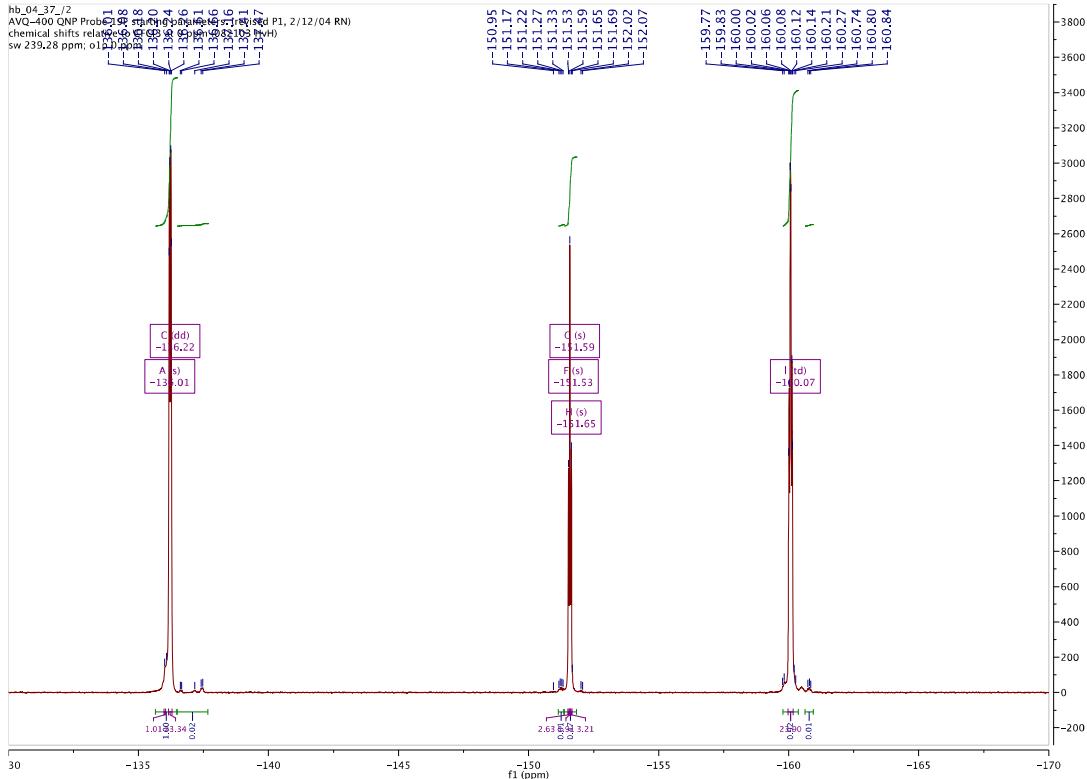


Figure S37.  $^{19}\text{F}$  NMR spectrum of **11** in  $\text{CDCl}_3$ .

*NO FIGURE. DATA IS MISSING. WE APOLOGIZE FOR THE INCONVENIENCE*

Figure S38.  $^1\text{H}$  NMR spectrum of **12**.

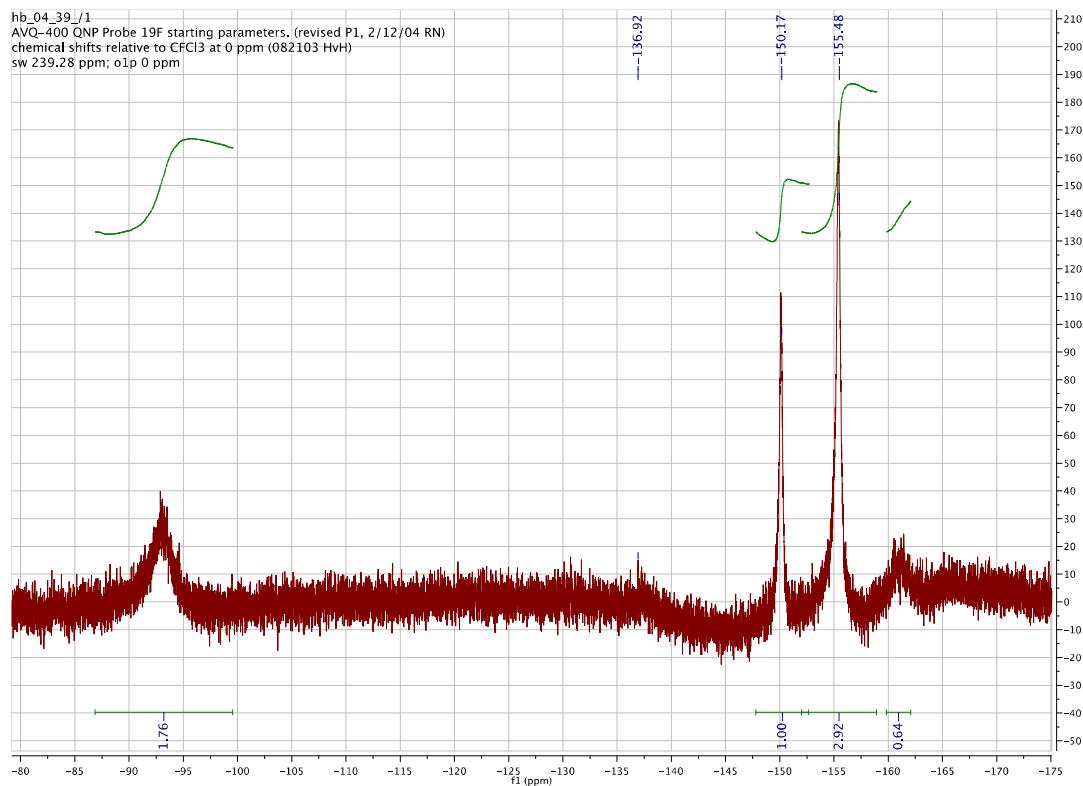


Figure S39.  $^{19}\text{F}$  NMR spectrum of **12** in  $\text{C}_6\text{D}_6$ .

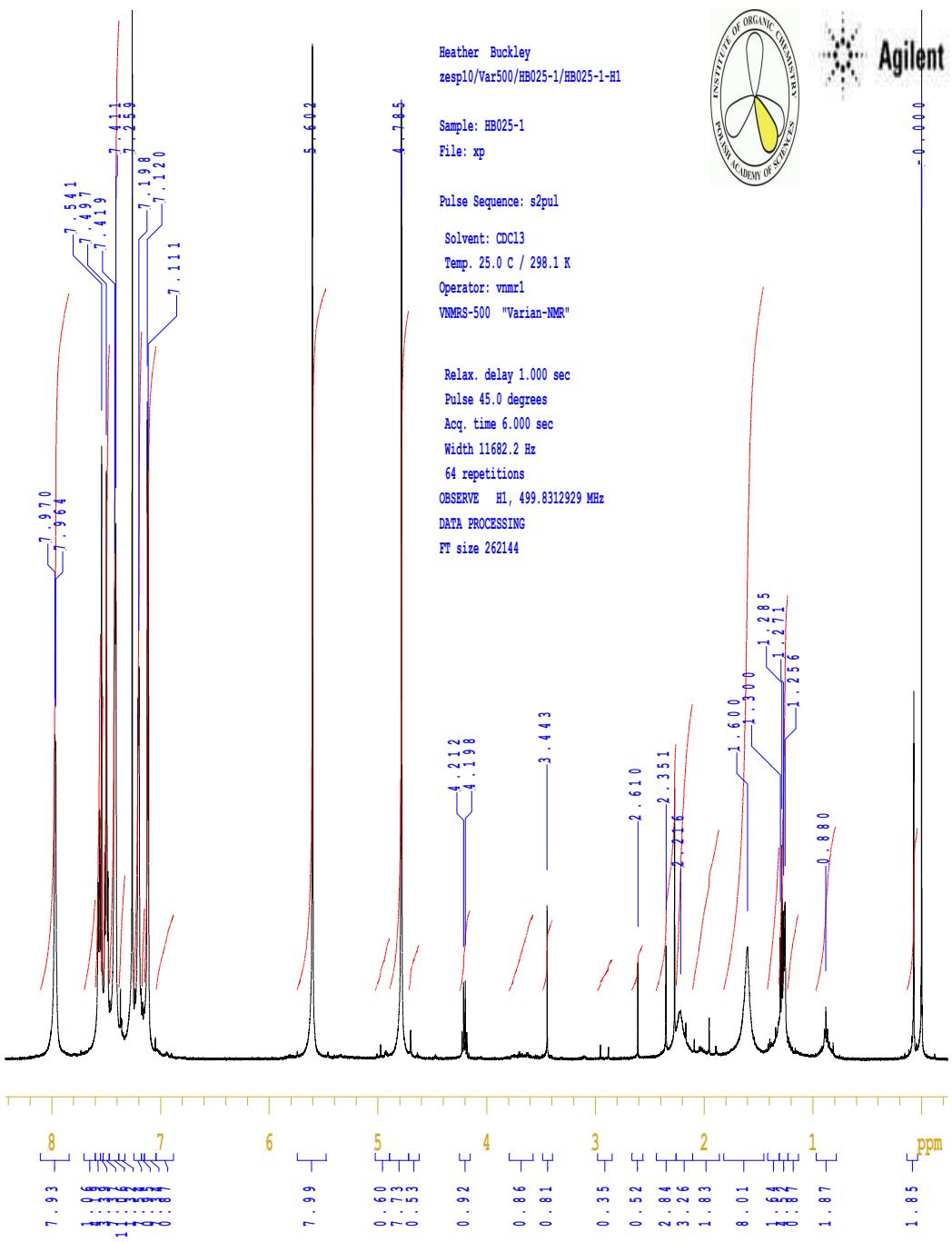


Figure S40. <sup>1</sup>H NMR spectrum of **13** in CDCl<sub>3</sub>.

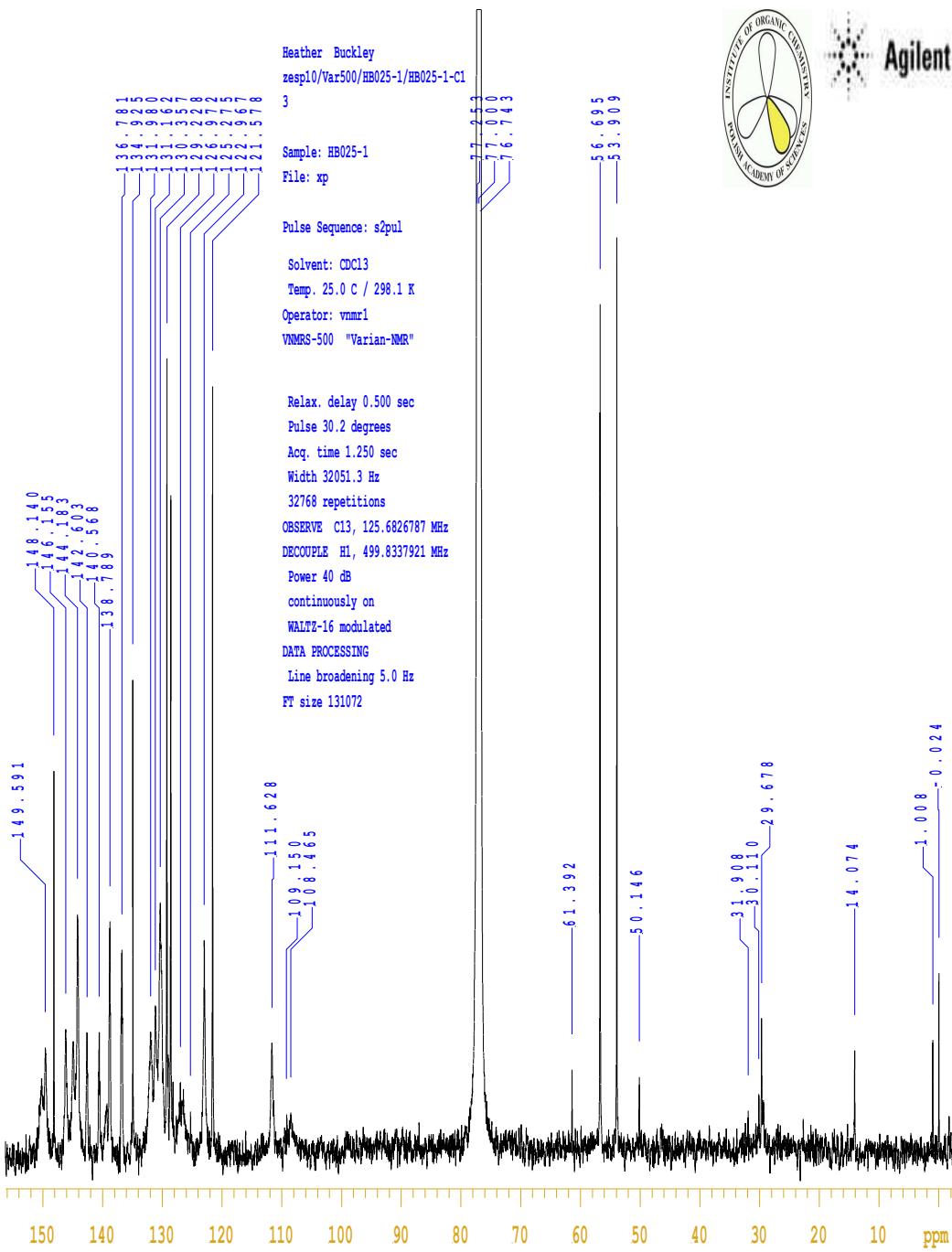


Figure S41. <sup>13</sup>C NMR spectrum of **13** in CDCl<sub>3</sub>.

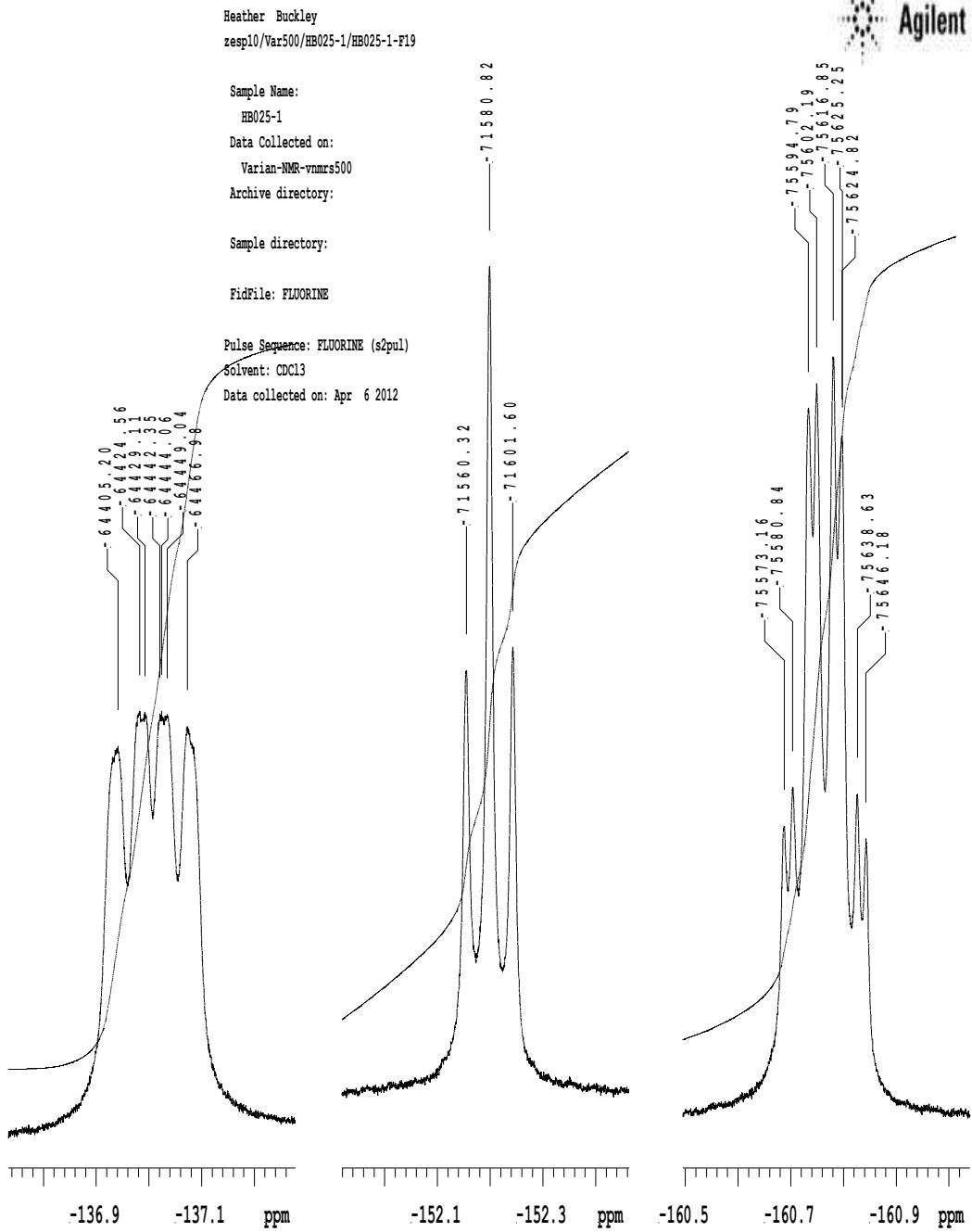


Figure S42. <sup>19</sup>F NMR spectrum of **13** in CDCl<sub>3</sub>.

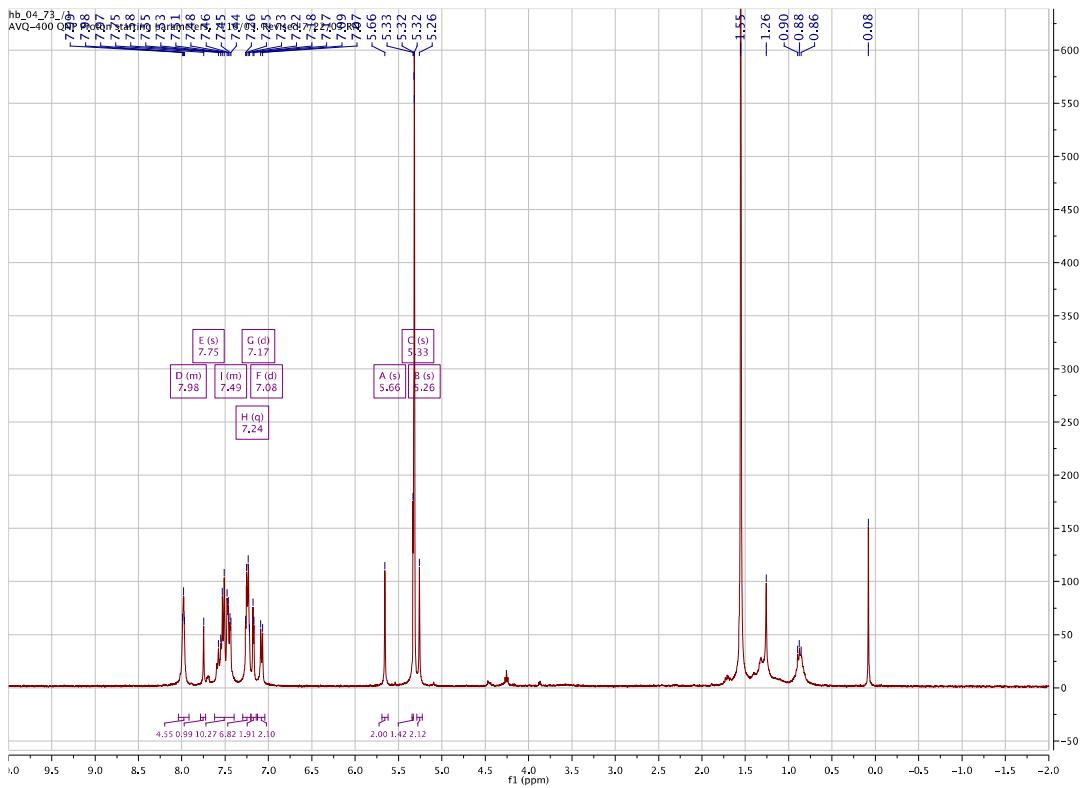


Figure S43.  $^1\text{H}$  NMR spectrum of **14** in  $\text{CD}_2\text{Cl}_2$ .

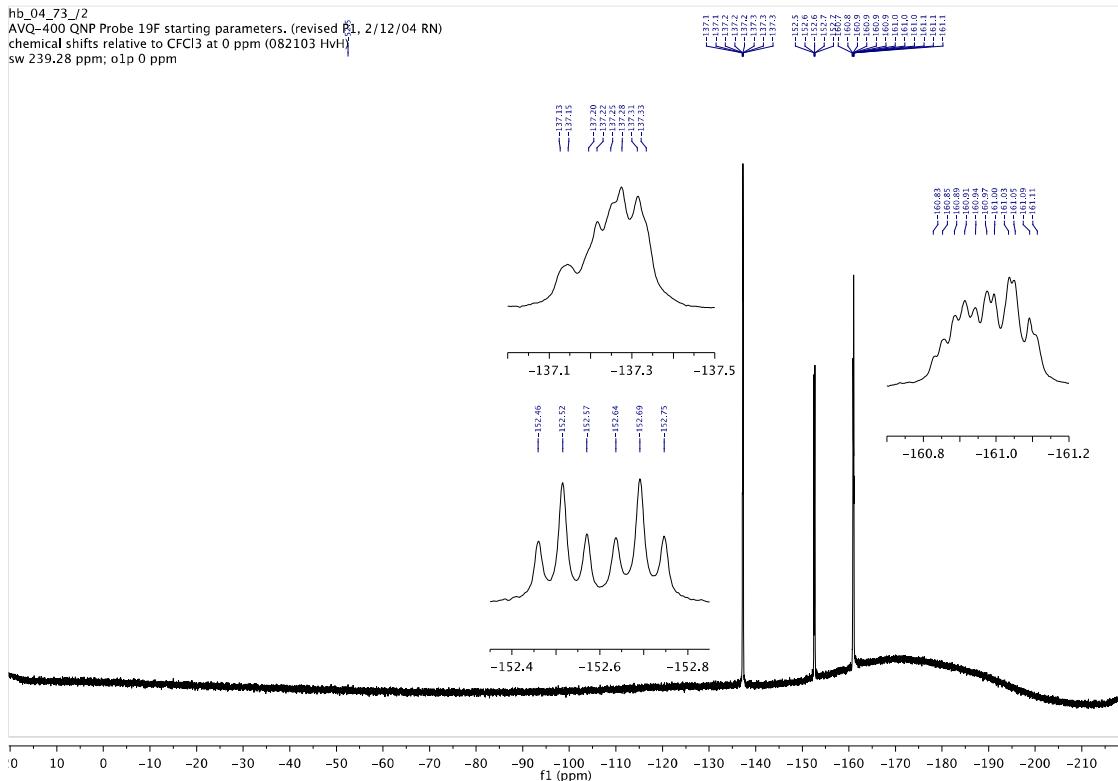


Figure S44.  $^{19}\text{F}$  NMR spectrum of **14** in  $\text{CD}_2\text{Cl}_2$ .

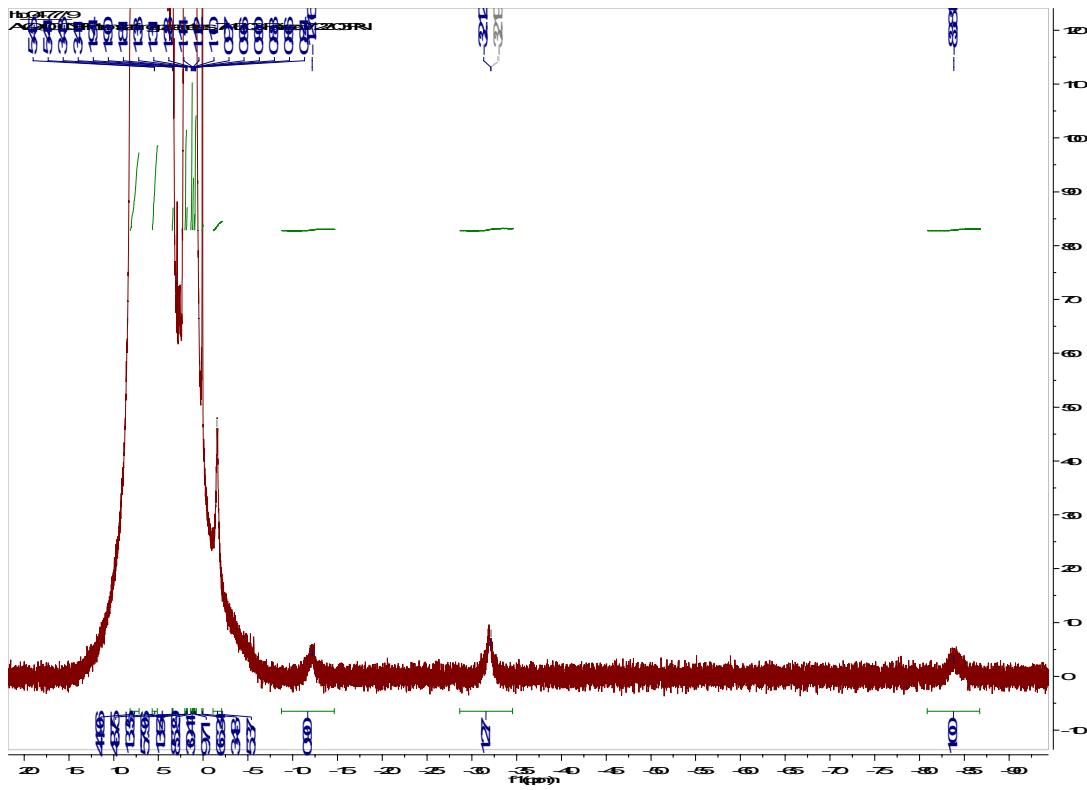


Figure S45. <sup>1</sup>H NMR spectrum of **15** in  $\text{CD}_3\text{CN}$  (broad window).

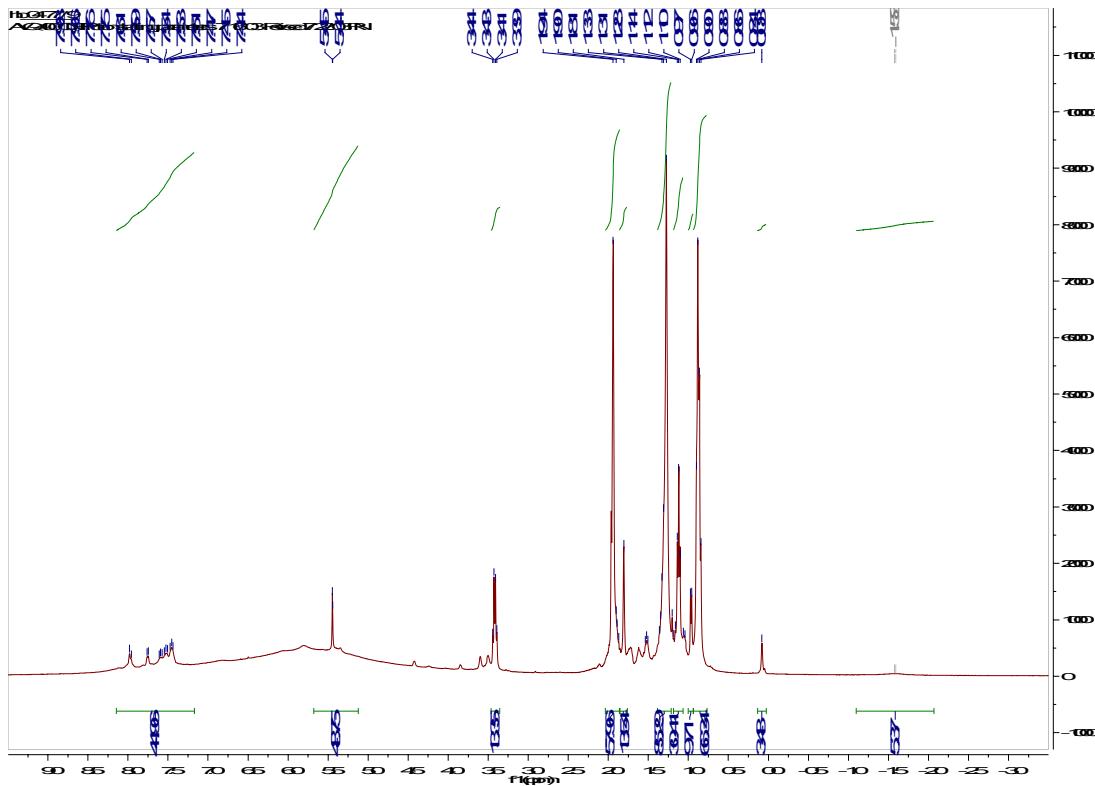


Figure S46. <sup>1</sup>H NMR spectrum of **15** in  $\text{CD}_3\text{CN}$  (narrow window).

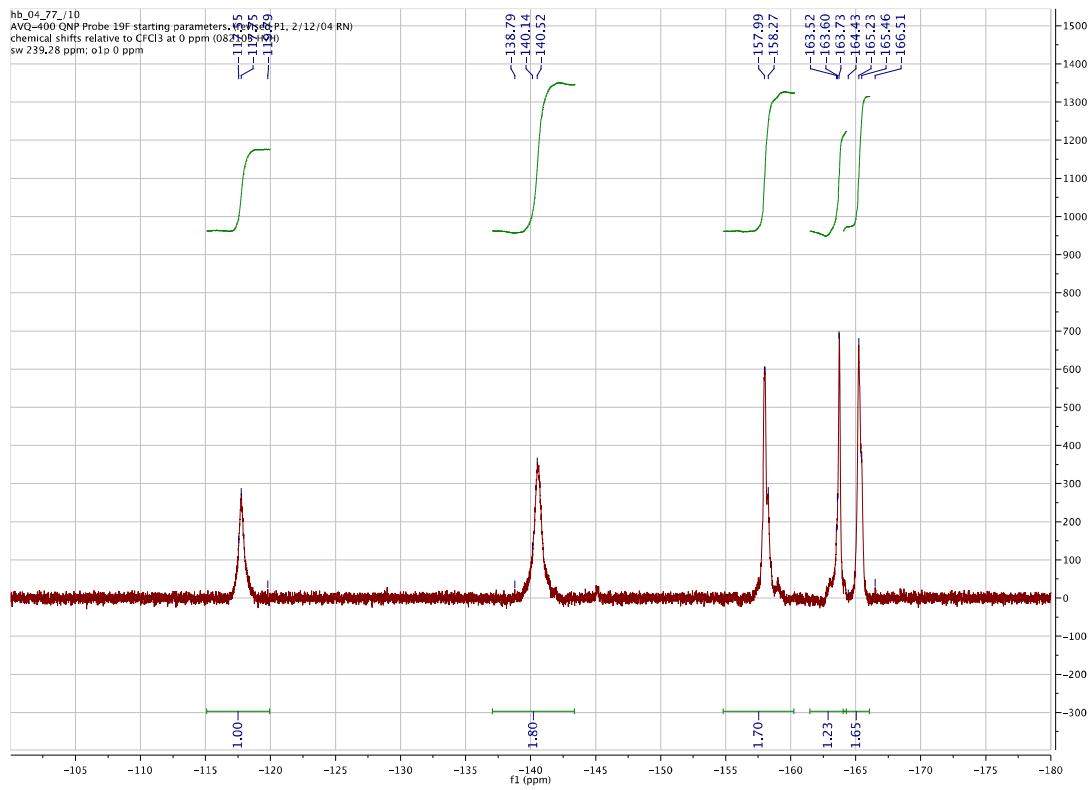


Figure S47.  $^{19}\text{F}$  NMR spectrum of **15** in  $\text{CD}_3\text{CN}$ .

## C. References

- (1) Pavlishchuk, V. V.; Addison, A. W. *Inorg. Chim. Acta* **2000**, *298*, 97–102.