

Supporting Information

Luminescence Properties of 1,8-naphthalimide Derivatives in Solution, in Their Crystals and in Co- Crystals: Towards Room Temperature Phosphorescence from Organic Materials

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Absorption and luminescence data in solution

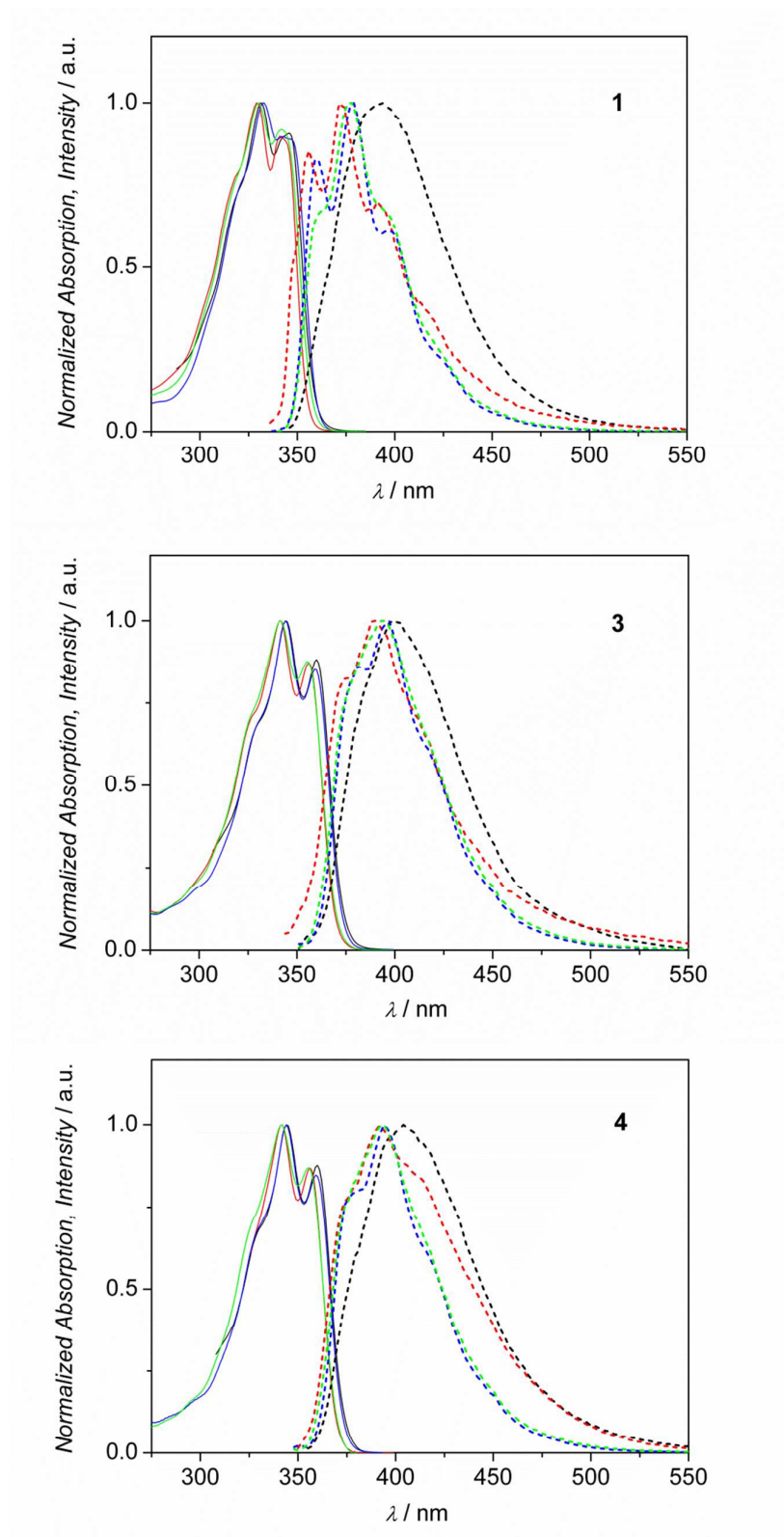


Figure S1. Absorption (full lines) and emission (dotted lines) spectra of **1**, **3**, and **4** in TOL (black), THF (red), DCM (blue) and ACN (green). Excitation at 330 nm for **1** and at 340 nm for **3** and **4**.

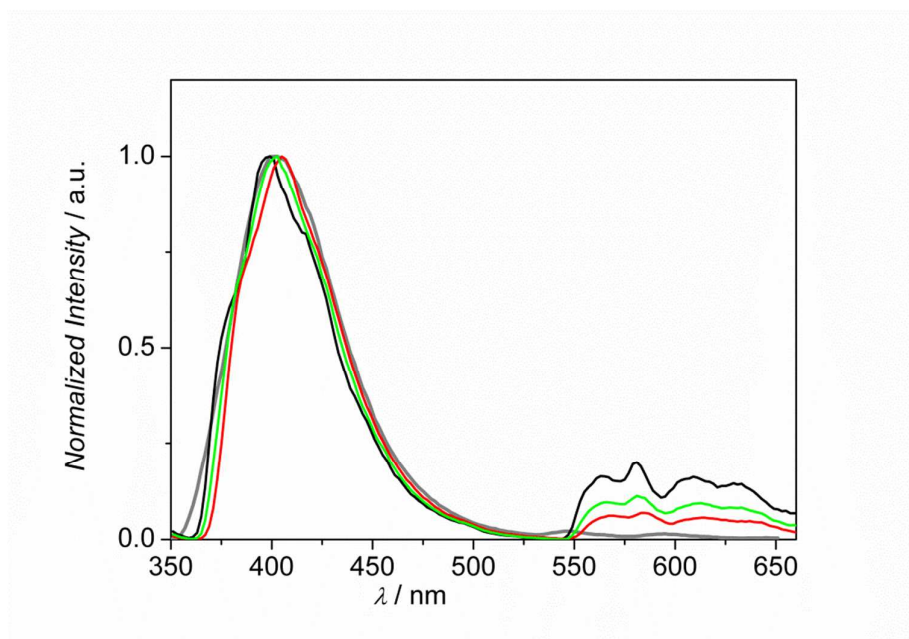


Figure S2. Luminescence spectra of **1** (gray), **2** (black), **3** (red) and **4** (green) in TOL at 77 K. Excitation at 330 nm. The spectra are normalized on the maximum of fluorescence.

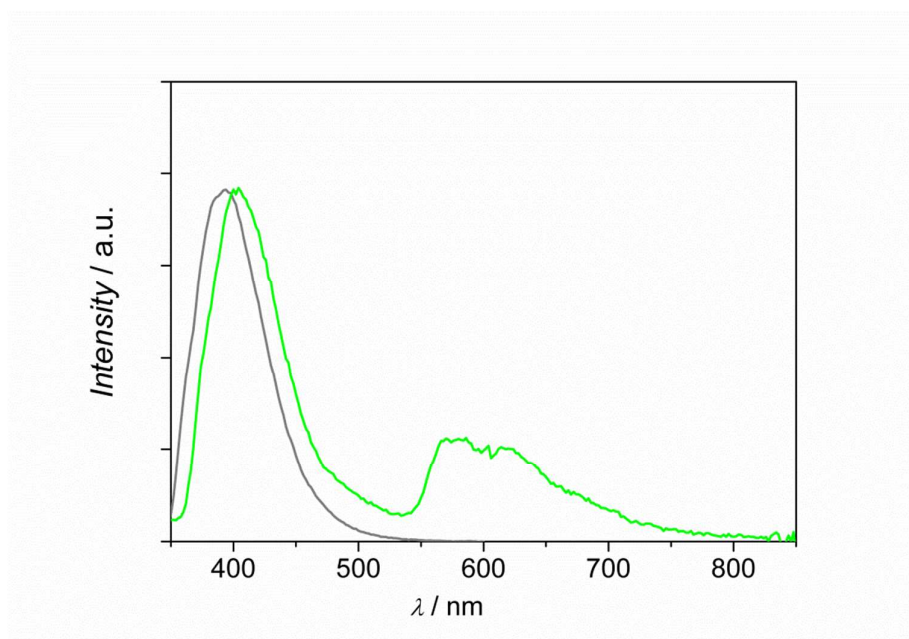


Figure S3. Luminescence spectra of **1** (grey) and **4** (green) in air-free TOL at room temperature. Excitation at 310 nm. The spectra are normalized on the maximum of fluorescence intensity.

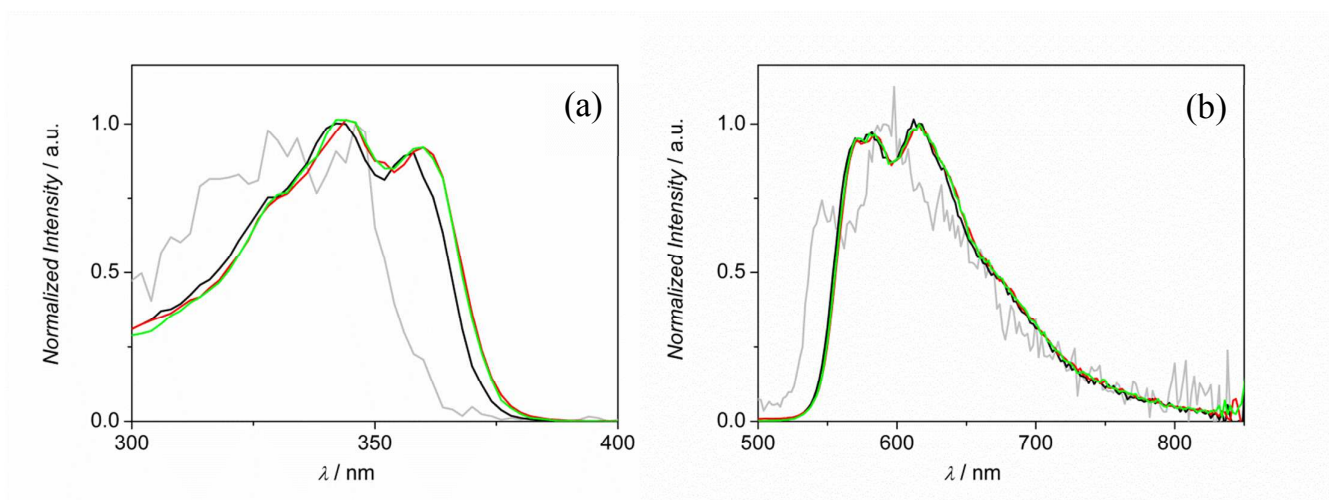


Figure S4. Excitation (a) and phosphorescence (b) spectra of **1** (gray), **2** (black), **3** (red) and **4** (green) in air-free TOL at room temperature, registered in gated mode. Gating parameters: delay: 0.1 ms, gate: 0.2-0.5 ms. In (a) $\lambda_{\text{em}} = 600 \text{ nm}$; in (b) $\lambda_{\text{exc}} = 340 \text{ nm}$.



Figure S5. Picture of an air-free TOL solution of **4** upon UV irradiation (UVGL-55 lamp). $A_{340} = 0.080$.

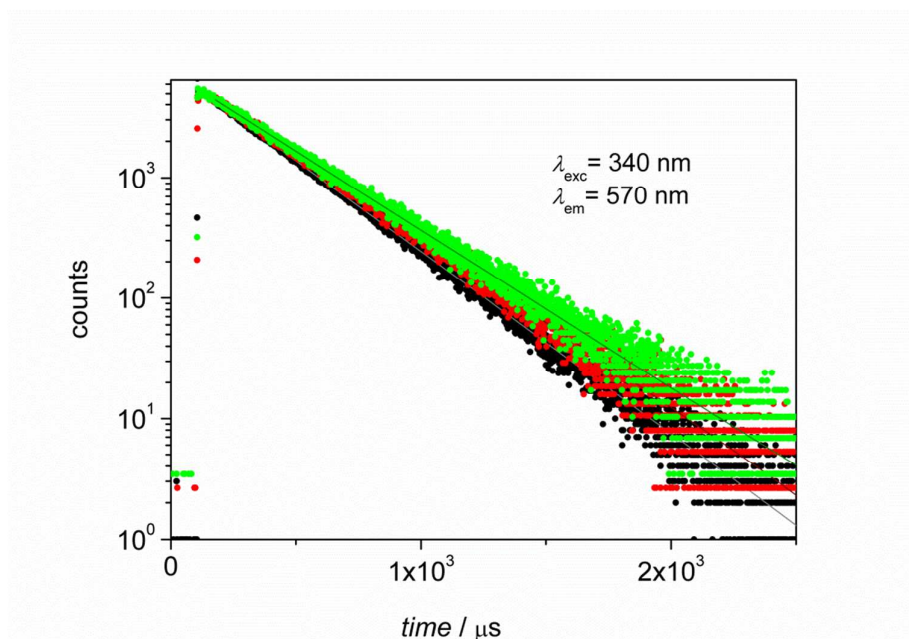


Figure S6. Phosphorescence decays, on a semi-log scale, of **2** (black), **3** (red) and **4** (green) in air-free TOL at room temperature and mono-exponential fittings.

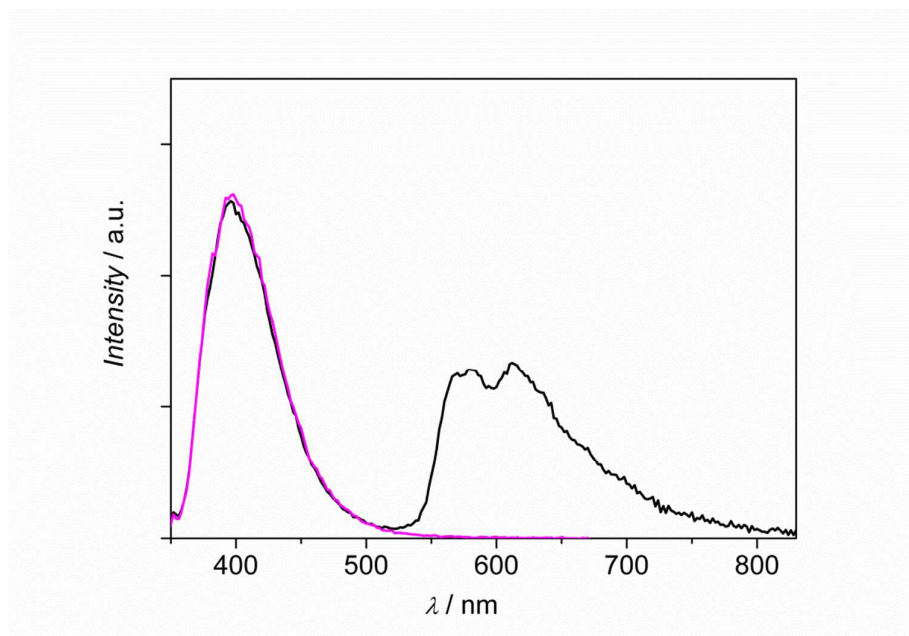


Figure S7. Luminescence spectra from optically matched TOL solutions of **2** in air-free solution (black) and in air-equilibrated solution (pink). Excitation at 340 nm, $A_{340} = 0.080$.

IR Spectra

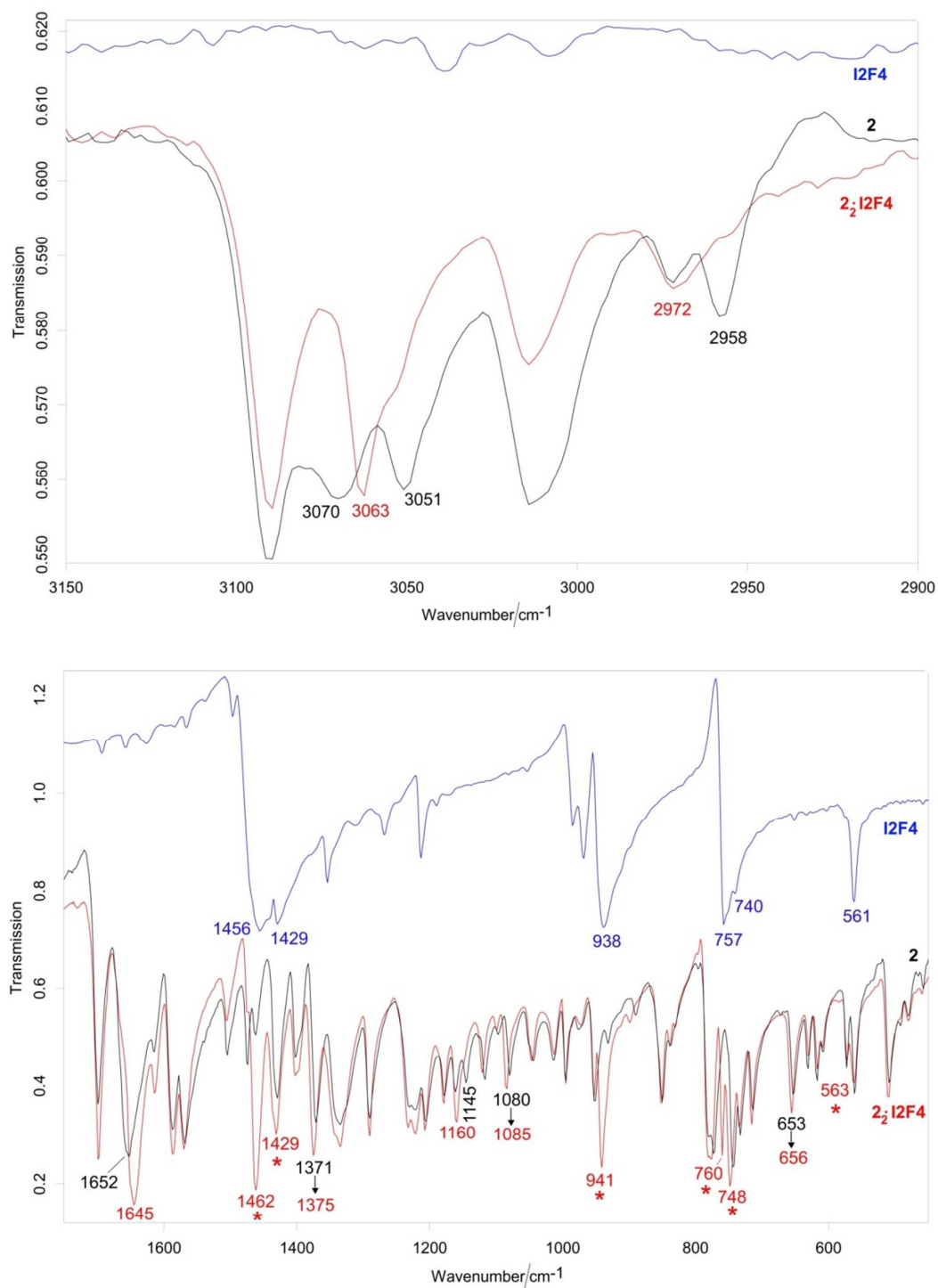


Figure S8. 3150-2900 cm^{-1} (top) and 1750-450 cm^{-1} (bottom) IR spectra of **2** (black), **I2F4** (blue) and **2₂·I2F4** (red). The bands of **I2F4** are indicated with an asterisk.

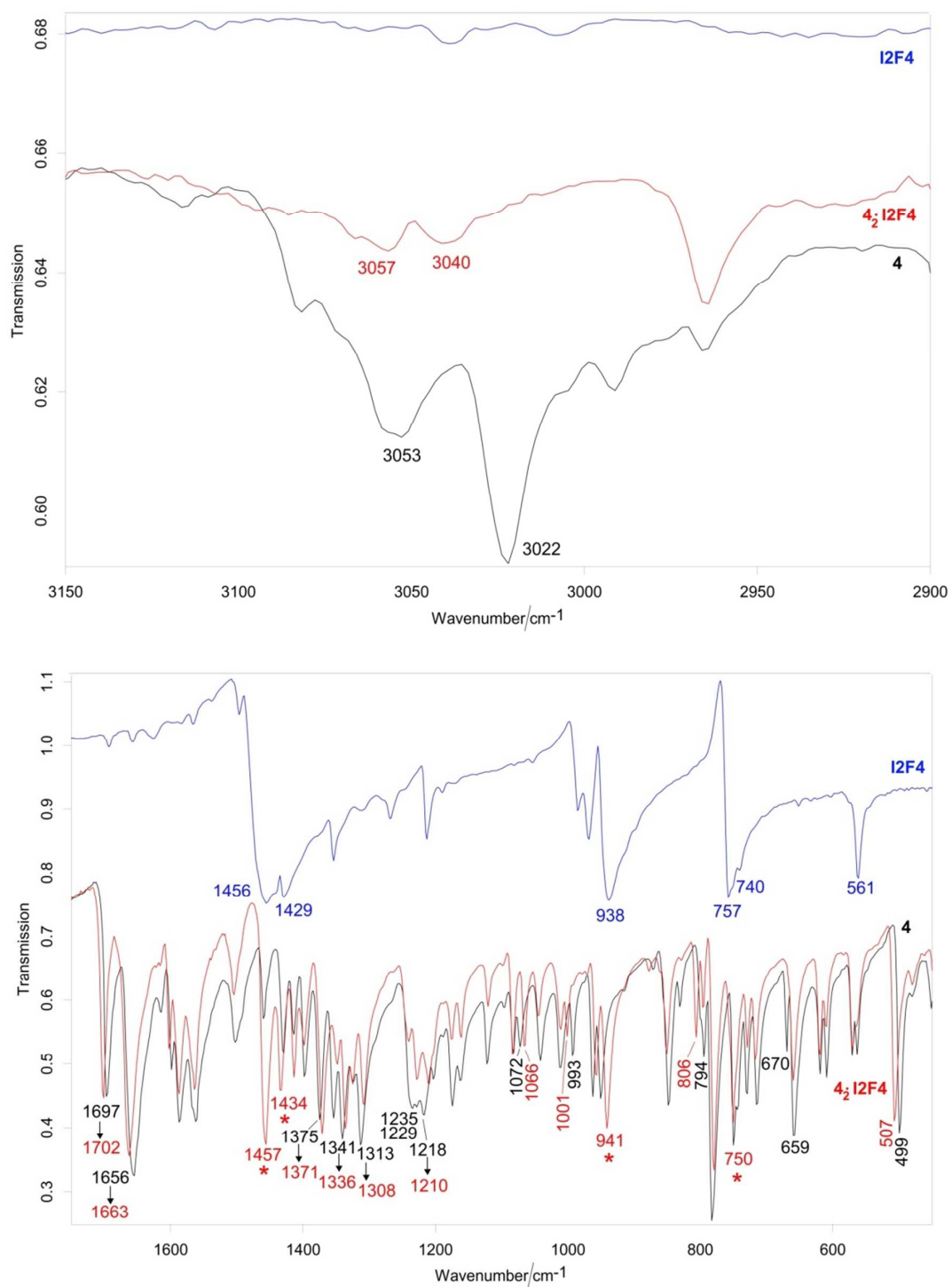


Figure S9. 3150-2900 cm⁻¹ (top) and 1750-450 cm⁻¹ (bottom) IR spectra of **4** (black), **12F4** (blue) and **4₂·12F4** (red). The bands of 12F4 are indicated with an asterisk.

Absorption and luminescence data in the solid state

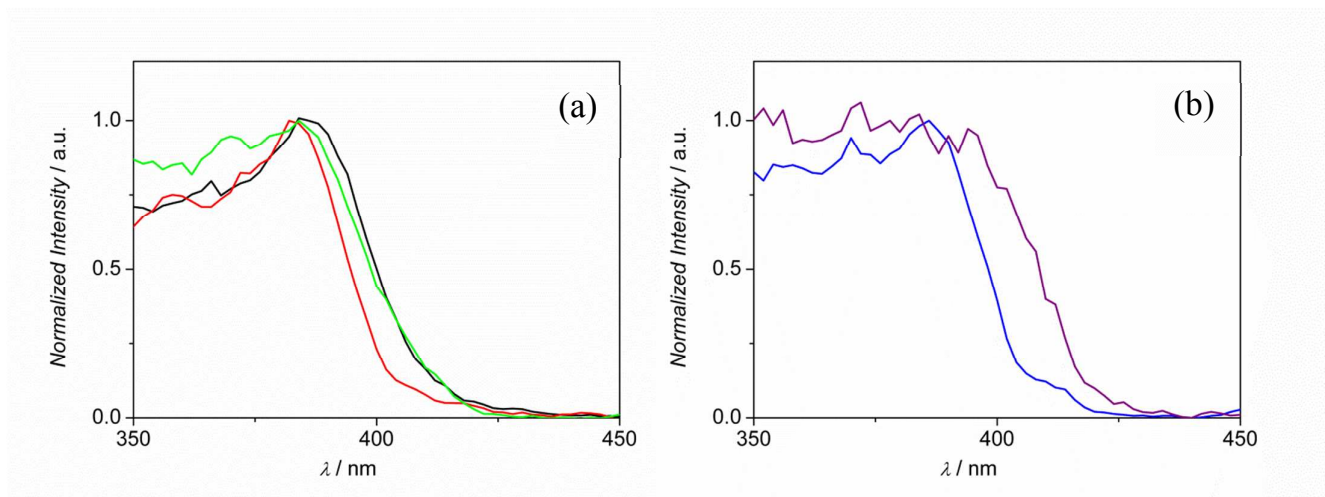


Figure S10. Excitation spectra collected on phosphorescence maxima of (a) **2** (black), **3** (red), and **4** (green) and (b) **2₂·I₂F₄** (blue) and **4₂·I₂F₄** (purple) in the solid state. Gating parameters: delay: 0.1 ms, gate: 2-3 ms.

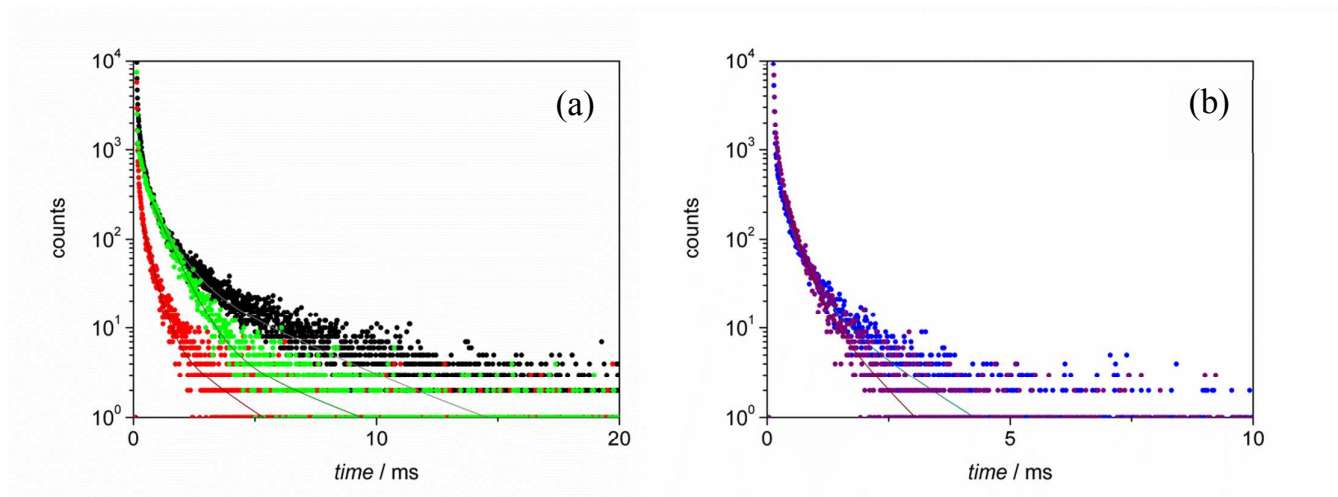


Figure S11. Phosphorescence decays, on a semi-log scale, of (a) **2** (black), **3** (red), and **4** (green) and (b) **2₂·I₂F₄** (blue) and **4₂·I₂F₄** (purple) in the solid state at room temperature, and bi-exponential fittings.

X-ray data

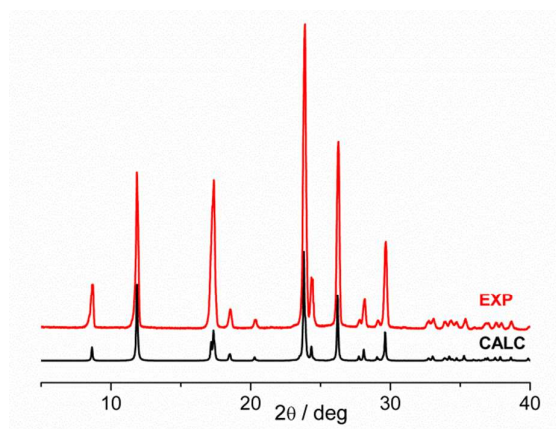


Figure S12. Comparison of calculated (on the basis of single crystal data) and measured X-ray powder patterns for compound **1**.

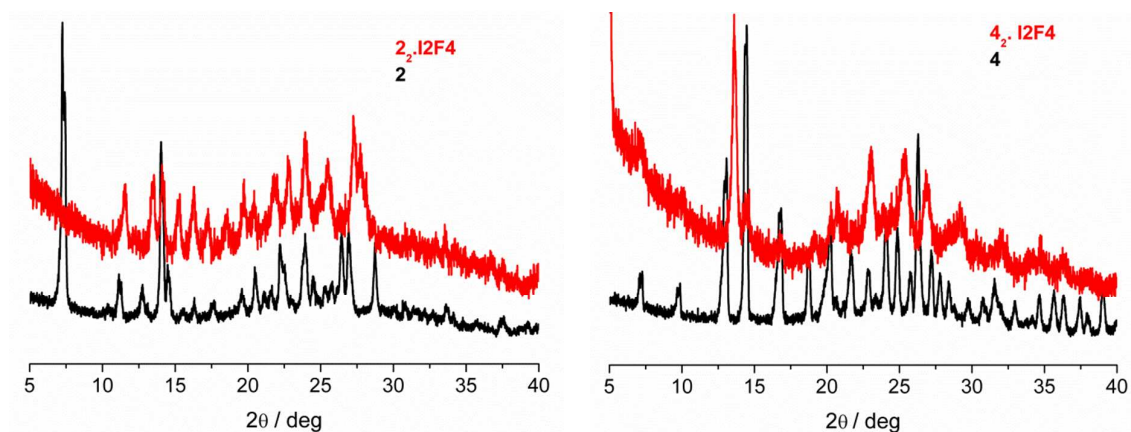


Figure S13. Comparison of experimental X-ray powder patterns for compound **2** and **4** and the relative co-crystals with the **12F4** co-former, namely $2_2 \cdot 12F4$ and $4_2 \cdot 12F4$, left and right respectively.

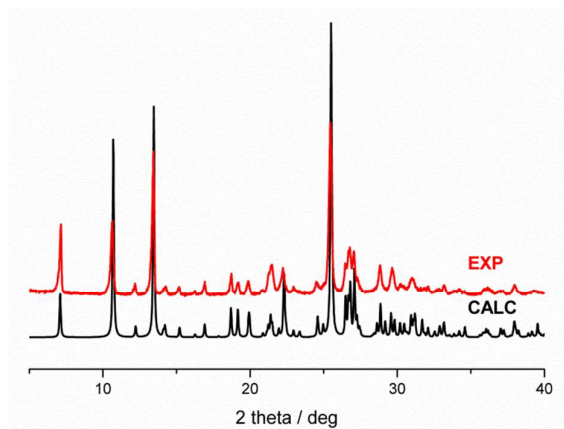


Figure S14. Comparison of calculated (on the basis of single crystal data) and measured X-ray powder patterns for compound **3**.

Table S1. Crystallographic data and details of measurements for compound **3**.

	3
Formula	C ₁₈ H ₁₁ BrN ₂ O ₂
fw	367.20
Cryst. System	Triclinic
Space group	P-1
Z	2
a (Å)	7.0727(8)
b (Å)	8.796 (1)
c (Å)	12.650 (1)
α (deg)	92.803(9)
β (deg)	98.280(9)
γ (deg)	109.299(10)
V (Å ³)	731.06(1)
D _{calc} (Mg/m ³)	1.668
μ (mm ⁻¹)	2.823
Measd reflns	4785
Indep reflns	2819
R ₁ [on F ₀ ² , I>2 σ (I)]	0.1040
wR ₂ (all data)	0.3016