Structure, stability and spectroscopic properties of H-bonded complexes of HOSO and CH₃SO with H₂O

Antonija Lesar* and Simona Tušar

Department of Physical and Organic Chemistry, Institute Jožef Stefan, Jamova c. 39, SI-1000

Ljubljana, Slovenia

E-mail: antonija.lesar@ijs.si

^{*}To whom correspondence should be addressed

Supporting Information

- **Table SI-1:** Cartesian coordinate in Å for all structures studied in this work.
- **Table SI-2:** B3LYP/6-311++G(2df,2pd) harmonic and anharmonic (cm⁻¹) frequencies together with IR intensities (I, in km mol⁻¹) of H₂O molecule, H₂O dimer, and HOSO radical and CH₃SO radical.
- **Table SI-3:** B3LYP/6-311++G(2df,2pd) harmonic and anharmonic (cm⁻¹) frequencies together with IR intensities (I, in km mol⁻¹) for HOSO-water complexes.
- **Table SI-4:** B3LYP/6-311++G(2df,2pd) harmonic and anharmonic (cm⁻¹) frequencies together with IR intensities (I, in km mol⁻¹) for CH₃SO-water complexes.
- Figure SI-1 B3LYP/6-311++G(2df,2pd) computed IR spectrum of HOSO radical, CH₃SO radical, H₂O molecule and H₂O dimer.
- Figure SI-2 B3LYP/6-311++G(2df,2pd) computed IR spectrum of HOSO-water complexes.

Figure SI-3 B3LYP/6-311++G(2df,2pd) computed IR spectrum of CH₃SO-water complexes.

- **Figure SI-4** TD-B3LYP/6-311++G(2df,2pd) computed electronic spectrum of HOSO-water complexes.
- **Figure SI-5** TD-B3LYP/6-311++G(2df,2pd) computed electronic spectrum of CH₃SO-water complexes.

	a		• •	0 11				
Table SI-1.	Cartesian	coordinate	in A	for all	structures	studied in	1 this	work.

$\mathbf{H}_2\mathbf{O}$						
	B3LYP	/6-311++G(2	2df,2pd)	CCSI	D/aug-cc-pV	VDZ
0	0.000000	0.000000	0.116800	0.000000	0.000000	0.118740
Н	0.000000	0.763077	-0.467198	0.000000	0.760299	-0.474961
Н	0.000000	-0.763077	-0.467198	0.000000	-0.760299	-0.474961

$\mathbf{H}_2\mathbf{O}\cdot\cdot\cdot\mathbf{H}_2\mathbf{O}$

B3LYP/6-311++G(2df,2pd)			CCS	CCSD/aug-cc-pVDZ			
0	1.392028	-0.001638	0.108819	1.403409	-0.001708	0.110427	
Н	1.757241	-0.759690	-0.356887	1.772313	-0.756356	-0.364451	
Н	1.751633	0.771800	-0.335571	1.767468	0.768810	-0.342202	
Н	-0.563990	-0.002020	0.035241-	0.576984	-0.002103	0.038025	
0	-1.520021	0.001848	-0.120604-	1.533273	0.001915	-0.121712	
Н	-1.920937	-0.011776	0.751498-	1.923885	-0.012008	0.758908	

HOSO

B3LYP/6-311++G(2df,2pd)

0	1.332422	0.258730	-0.052524	-1.356396	0.274634	-0.000096
Η	1.261945	1.157305	0.304210	-1.170871	1.229652	0.000576
S	-0.144042	-0.478340	0.018631	0.145104	-0.502072	0.000035
0	-1.202081	0.553287	-0.022764	1.212546	0.575804	-0.000046

CH_3SO

B3LYP/6-311++G(2df,2pd) 0 1.245682 0.607973 0.000015 S 0.226446 -0.501169 0.000008 С $-1.404128 \quad 0.300882$ -0.000030 Η -2.160317 -0.483992-0.001712 -1.500815 0.918097 -0.891509 Η Н -1.502681 0.915529 0.893153

CCSD/aug-cc-pVDZ

CCSD/aug-cc-pVDZ

1.259622	0.628078	-0.000006
0.218014	-0.523920	-0.000010
-1.402817	0.320911	-0.000002
-2.186249	-0.455515	0.000091
-1.481282	0.944148	-0.904076
-1.480759	0.943995	0.904210

Table SI-1. Continued.

HOSO···	H_2O					
	B3LYP/6-3	311++G(2df	(,2pd)	CCS	D/aug-cc-p	VDZ
0	0.137320	1.279248	0.040486	0.157778	1.314265	0.097500
Н	-0.802151	0.952862	0.001119	-0.772935	0.988426	0.021679
S	1.162750	0.021075	-0.022785	1.177415	0.006449	-0.066868
0	0.391607	-1.252949	0.041943	0.365673	-1.274979	0.102298
Н	-1.661340	-0.875534	-0.018458	-1.676466	-0.857044	0.012269
0	-2.187445	-0.065218	-0.105651	-2.207604	-0.057154	-0.135055
Н	-2.872366	-0.103183	0.567680	-2.916021	-0.091631	0.517994

syn-HOSO···2H₂O

B3LYP/6-311++G(2df,2pd)

0	0.957866	1.212929	0.336319	0.943901	1.193784	0.508599
Н	-0.022598	1.329865	0.114276	0.008206	1.329539	0.193895
S	1.578452	-0.178822	-0.164075	1.593052	-0.152553	-0.196078
0	0.626589	-1.301633	0.086862	0.649609	-1.338512	-0.000027
Н	-1.971451	0.573071	-0.124984	-2.003584	0.608167	-0.171006
0	-1.582110	1.470735	-0.240798	-1.590099	1.479522	-0.341818
Н	-2.085962	2.069641	0.316685	-2.135746	2.122510	0.124745
Н	-2.587541	-1.713195	-0.496023	-2.605542	-1.780330	-0.367331
0	-2.172138	-1.134018	0.147976	-2.191764	-1.143109	0.225382
Н	-1.229343	-1.382339	0.172383	-1.245342	-1.372516	0.219861

anti-HOSO $\cdot \cdot \cdot 2H_2O$

B3LYP/6-311++G(2df,2pd)

0	-0.133090	-0.489412	-0.681298
Н	0.378980	0.368556	-0.607343
S	-1.698179	-0.304201	-0.255287
0	-1.787005	0.434687	1.026298
0	1.570344	1.515546	-0.335036
Н	1.361158	2.125714	0.378661
Н	2.204980	0.871174	0.029531
Н	1.769854	-1.311323	0.236825
0	2.619969	-0.924414	0.485565
Н	3.294152	-1.478167	0.082683

CCSD/aug-cc-pVDZ

CCSD/aug-cc-pVDZ

-0.176420	-0.675568	-0.905852
0.325139	0.187384	-0.915602
-1.589740	-0.410638	-0.045701
-1.344055	0.726903	0.935688
1.278172	1.549788	-0.496128
0.728321	1.898935	0.220357
1.965068	1.033833	-0.040834
1.683795	-1.286879	0.167012
2.437099	-0.828483	0.566291
3.175157	-1.444180	0.500281

Table SI-1. Continued.

cyc-CH₃SO···H₂O

	B3LYP/6-311++G(2df,2pd)			CCS	CCSD/aug-cc-pVDZ		
0	0.049119	-1.212009	0.183938	-0.113694	-1.269655	0.000244	
S	1.209325	-0.290486	-0.116090	-1.259227	-0.215926	-0.000157	
С	0.595488	1.402607	0.090602	-0.422122	1.406329	0.000154	
Н	1.389779	2.074830	-0.233088	-1.213374	2.174214	-0.000386	
Н	0.359042	1.576435	1.139260	0.205152	1.491511	-0.899700	
Н	-0.299695	1.536026	-0.514334	0.204192	1.491766	0.900645	
Н	-1.765543	-0.463978	0.089857	1.700913	-0.518718	0.000115	
0	-2.467241	0.189980	-0.049650	2.432901	0.119596	-0.000123	
Н	-3.260732	-0.314941	-0.242173	3.229834	-0.421461	-0.000049	

$acyc\text{-}\mathbf{CH}_3\mathbf{SO}\cdots\mathbf{H}_2\mathbf{O}$

B3LYP/6-311++G(2df,2pd)

0	-1.677871	0.348414	0.232391
S	-0.423140	-0.364666	-0.197962
С	0.920886	0.853572	-0.093185
Н	1.842104	0.357839	-0.399299
Н	1.007451	1.206614	0.933357
Н	0.703692	1.687312	-0.759238
0	1.028544	-0.457932	0.167981
Н	1.921197	-0.207204	0.542554
Н	0.965108	-1.455188	0.206145

CH_3SO-2H_2O

B3LYP/6-311++G(2df,2pd)								
0	0.801317	-1.267698	-0.000591					
S	1.712162	-0.095440	-0.289977					
С	1.003473	1.342269	0.546208					
Н	1.643408	2.192306	0.308517					
Н	1.002966	1.164941	1.621149					
Н	-0.013377	1.513143	0.188226					
Н	-1.045590	-1.418977	0.272563					
0	-2.015783	-1.355701	0.318236					
Н	-2.348630	-2.057135	-0.247330					
0	-2.208094	1.357681	-0.392691					
Н	-2.979734	1.790506	-0.020415					
Н	-2.293997	0.414386	-0.159957					

H_2O			HOSO			 CH ₃ SO				
Н	А	Ι	Exp ^{<i>a,b</i>}	Н	А	Ι	Н	А	Ι	Exp ^c
3818	3647	8	3660.6	3744	3555	101	3131	2981	2	2995 vw
1630	1575	72	1595.6	1168	1157	111	3130	2979	4	
3921	3736	64	3750.7	1067	1044	14	3037	2917	3	2919 vw
				753	740	194	1460	1419	8	1417 w
				395	367	30	1446	1404	10	1405 w
				109	405	110	1323	1299	1	1289 w
							1061	1047	47	1068 vs
							947	921	11	927 m
							887	869	1	868 vw
$H_2O \cdots H_2O$						662	646	14	670 w	
							332	333	7	341 w
H	А	Ι	Exp ^a				137	93	0	n.o.
3911	3715	87	3730							
3892	3691	89								
3811	3625	14								
3705	3547	330	$\sim \! 3530$							
1650	1591	37								
1630	1577	94								
638	499	121								
367	253	64								
190	78	206								
160	9	45								
158	87	82								
135	-25	135								

Table SI-2. B3LYP/6-311++G(2df,2pd) harmonic and anharmonic (cm⁻¹) frequencies together with IR intensities (I, in km mol⁻¹) of H₂O molecule, H₂O dimer, and HOSO radical and CH₃SO radical.

^a Forney, D.; Jacox, M. E.; Thompson, W. E. J. Mol. Spectrosc. **1993**, **157**, 479-493.

^b Tsuge, M.; Tsuji, K.; Kawai, A.; Shibuya, K. J. Phys. Chem. A 2007, 111, 3540-3547.

^c Reisenauer, H. P.; Romański, J.; Mlostoń, G.; Schreiner, P. R. ChemComm 2013, 49, 9467-9469.

$HOSO \cdot \cdot \cdot H_2O$			syn-H	syn-HOSO· · · $2H_2O$				anti-HOSO···2H ₂ O		
Н	А	Ι	Н	А	Ι	Н	А	Ι		
3785	3579	109	3881	3698	105	389	1 3700	135		
3682	3492	152	38/3	3694	8/	386	4 3664	96		
3233	2999	961	3599	3403	641	3/5	/ 3590	122		
1637	1586	80	3398	3173	696	360	3 3445	379		
1316	1237	/0	2911	2610	16/3	310	2 2850	1407		
1114	1096	82	1668	160	22	164	3 1601	23		
816	802	169	1640	1600	/8	163	2 1586	105		
736	640	143	1374	1445	65	134	9 1275	96		
574	493	153	1114	1098	131	113	9 1120	87		
451	413	7	946	829	135	90	0 838	99		
433	409	67	859	844	157	80	1 786	180		
387	333	128	811	741	65	67	3 585	193		
221	199	20	659	613	133	46	7 390	89		
158	130	10	514	464	79	44	1 342	161		
45	189	10	431	416	30	40	2 377	43		
			381	360	25	34	0 193	90		
			319	312	132	25	9 213	45		
			283	259	28	24	5 211	73		
			251	229	131	20	2 172	10		
			232	206	25	18	3 -18	126		
			178	164	4	11	8 77	4		
			110	106	7	6	7 19	5		
			79	38	4	4	0 49	6		
			41	31	2	2	4 10	2		

Table SI-3. B3LYP/6-311++G(2df,2pd) harmonic and anharmonic (cm⁻¹) frequencies together with IR intensities (I, in km mol⁻¹) for HOSO-water complexes.

cyc-CH ₃ SO···H ₂ O			acyc-	CH₃SO·	$\cdot \cdot H_2O$	CH ₃	$CH_3SO \cdots 2H_2O$			
Н	А	Ι	Н	А	Ι	Н	А	Ι		
3891	3713	101	3915	3733	73	3887	3682	66		
3692	3532	281	3814	3646	13	3882	3685	109		
3138	2986	1	3143	3000	0	3633	3456	582		
3129	2981	3	3126	2978	3	3569	3411	410		
3037	2918	3	3038	2923	6	3124	2975	3		
1646	1599	73	1634	1593	79	3112	2967	28		
1458	1414	13	1460	1413	9	3016	2935	34		
1443	1396	11	1456	1420	5	1668	1610	31		
1333	1289	1	1320	1288	2	1648	1599	77		
1055	1044	61	1055	1043	54	1470	1440	10		
958	929	11	957	934	11	1447	1405	9		
893	869	2	891	875	1	1343	1310	1		
667	655	10	663	663	14	1061	1044	73		
527	444	99	334	360	9	965	940	12		
368	350	54	173	178	23	905	877	2		
337	330	85	141	185	103	744	606	102		
157	143	14	128	77	164	672	635	10		
145	79	26	90	90	27	636	579	145		
121	130	97	43	14	10	394	338	114		
84	51	18	29	-102	1	370	332	33		
41	-17	3	20	-370	6	334	314	25		
						227	134	85		
						207	163	3		
						186	156	7		
						172	148	20		
						163	-127	116		
						122	93	11		
						77	64	1		
						51	23	4		
						32	14	5		
								_		

Table SI-4. B3LYP/6-311++G(2df,2pd) harmonic and anharmonic (cm⁻¹) frequencies together with IR intensities (I, in km mol⁻¹) for CH₃SO-water complexes.



Figure SI-1: B3LYP/6-311++G(2df,2pd) computed IR spectrum of HOSO radical, CH₃SO radical, H₂O molecule and H₂O dimer.



Figure SI-2: B3LYP/6-311++G(2df,2pd) computed IR spectrum of HOSO-water complexes.



Figure SI-3: B3LYP/6-311++G(2df,2pd) computed IR spectrum of CH₃SO-water complexes.



Figure SI-4: TD-B3LYP/6-311++G(2df,2pd) computed electronic spectrum of HOSO-water complexes.



Figure SI-5:TD-B3LYP/6-311++G(2df,2pd) computed electronic spectrum of CH₃SO-water complexes.