

Supporting Information

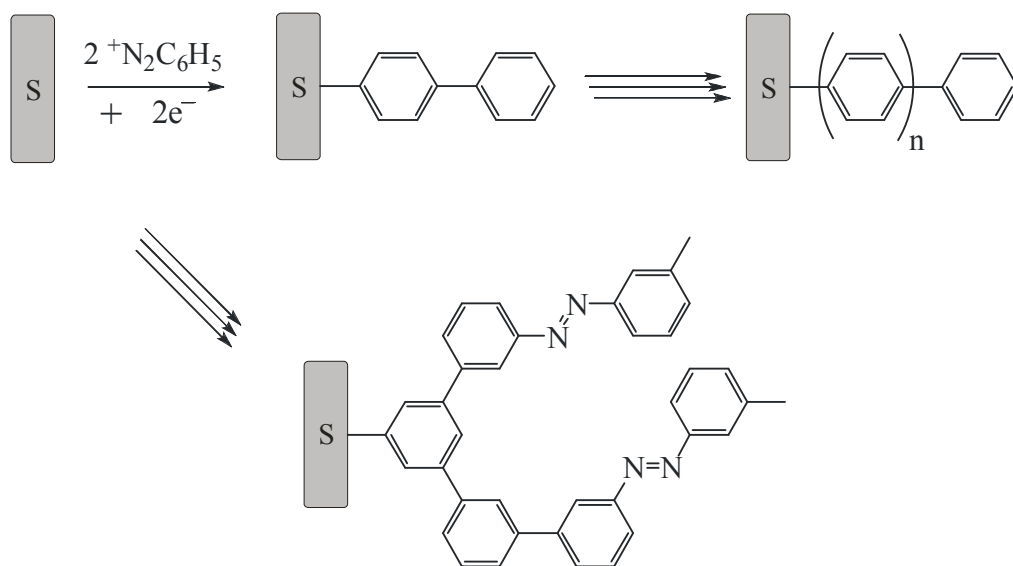
Electron Transfer Initiated Formation of Covalently Bound Organic Layers on Silicon Surfaces

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- I. Multilayer formation on a surface from aryl diazonium salts**
- II. Theoretical calculations**
- III. Additional IRRAS data for NBS/Fc Si(111) treated by succinimidyl anion and NBS/Fc Si(111) treated by pentachlorophenolate.**
- IV. Line profile details**
- V. Comparison of the XPS Si 2p signal for a modified Si(111) surface and an oxidized Si(111) surface**

I. Multilayer formation on a surface from aryl diazonium salts



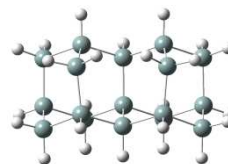
Scheme S1. Multilayer formation on a surface from an aryl diazonium salt.¹

II. Theoretical calculations

Theoretical calculations were performed using the Gaussian 09 software package.¹ Optimized structures for the Si₁₇H₂₄ cluster, meant to simulate a hydrogen terminated Si(111) surface, and the Si₁₇H₂₂ NBS cluster were calculated using the B3LYP method² with the 6-311G+(d,p) basis set. The method, basis set and silicon cluster size were chosen based on the results of a recently published paper.³ We checked that the obtained confirmations were real minima by running frequency calculations for the B3LYP calculations. No imaginary frequencies were observed. During the two cluster calculations, subsurface atoms were not held in a fixed position. Further calculations were not undertaken in the present study beyond the attachment of the succidimyl group and bromine atom to the Si(111) surface due to the uncertainty of the experimental surface structure. These calculations do show however, that existence of a succidimyl group and bromine atom on neighbouring surface sites on the Si(111) surface is theoretically possible.

Si₁₇H₂₄ Cluster

Total Energy: -4936.40670565 a.u.

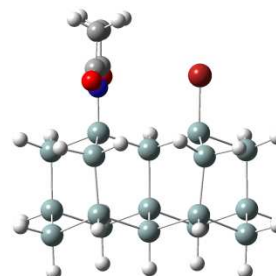


Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
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3	14	0	-1.900961	0.000000	-0.011823
4	1	0	-3.127586	0.000000	0.841913
5	14	0	-1.930216	-1.993291	-1.297858
6	1	0	-3.148844	-2.008310	-2.160567
7	14	0	0.000000	2.210714	-2.646836
8	1	0	0.000000	1.215704	-3.752016
9	1	0	0.000000	3.570073	-3.259273
10	14	0	0.000000	-2.210714	-2.646836
11	1	0	0.000000	-1.215704	-3.752016
12	1	0	0.000000	-3.570073	-3.259273
13	14	0	-1.957941	-3.869273	0.146029
14	1	0	-3.152080	-3.822546	1.036197
15	1	0	-2.042207	-5.116435	-0.665835
16	14	0	-1.957941	3.869273	0.146029
17	1	0	-2.042207	5.116435	-0.665835
18	1	0	-3.152080	3.822546	1.036197

19	14	0	0.000000	3.884634	1.473323
20	1	0	0.000000	5.090532	2.353836
21	14	0	1.957941	3.869273	0.146029
22	1	0	2.042207	5.116435	-0.665835
23	1	0	3.152080	3.822546	1.036197
24	14	0	0.000000	1.924334	2.788331
25	1	0	1.203528	1.890391	3.667217
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27	14	0	1.930216	1.993291	-1.297858
28	14	0	0.000000	0.000000	1.410114
29	14	0	0.000000	-1.924334	2.788331
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32	14	0	0.000000	-3.884634	1.473323
33	1	0	0.000000	-5.090532	2.353836
34	14	0	1.930216	-1.993291	-1.297858
35	14	0	1.900961	0.000000	-0.011823
36	1	0	3.127586	0.000000	0.841913
37	14	0	1.957941	-3.869273	0.146029
38	1	0	2.042207	-5.116435	-0.665835
39	1	0	3.152080	-3.822546	1.036197
40	1	0	3.148844	2.008310	-2.160567
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Si₁₇H₂₂ NBS Cluster

Total Energy: -7869.60185476 a.u.

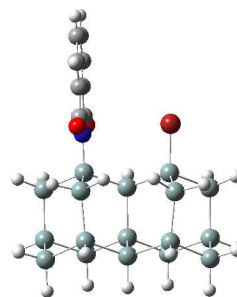


Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
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3	1	0	0.431788	-0.703997	-2.504743
4	14	0	-0.878847	-2.174379	-0.049697
5	14	0	1.838675	1.048581	2.196981
6	1	0	1.380519	-0.066511	3.056199
7	1	0	3.083075	1.651300	2.739547
8	14	0	-1.541502	-1.776647	2.183730
9	1	0	-0.365230	-1.668318	3.082669
10	1	0	-2.365061	-2.937204	2.624311
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14	14	0	3.086200	2.259203	-1.213126

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16	1	0	3.393815	1.918906	-2.626120
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18	1	0	2.121223	5.220759	-1.863237
19	14	0	1.005181	4.639709	1.080434
20	1	0	2.221393	5.140114	1.781686
21	1	0	-0.003439	5.736572	1.071868
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23	1	0	-1.440638	4.454282	-2.237518
24	1	0	-0.162297	2.963431	-3.633059
25	14	0	0.169473	2.737929	2.209877
26	14	0	-1.388528	1.492498	-1.096072
27	14	0	-3.373781	0.862142	-2.222703
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36	1	0	-5.565917	-1.496103	1.771347
37	1	0	-5.803143	0.791365	1.048671
38	1	0	-0.130845	3.102560	3.627237
39	1	0	-3.201267	0.517255	3.638816
40	35	0	0.570011	-3.964836	-0.075739
41	6	0	3.772951	-1.599902	-1.289169
42	6	0	3.988716	-1.556413	1.022680
43	6	0	4.722569	-2.742928	-0.961205
44	6	0	4.868858	-2.710742	0.563123
45	1	0	4.284817	-3.669368	-1.338277
46	1	0	5.659726	-2.585369	-1.498629
47	1	0	4.522985	-3.621599	1.055817
48	1	0	5.889420	-2.526727	0.904910
49	7	0	3.375336	-0.976511	-0.095691
50	8	0	3.402016	-1.254791	-2.386191
51	8	0	3.835121	-1.181657	2.162688

Si₁₇H₂₂ NBP Cluster

Total Energy: -8022.06022721 a.u.



Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
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32	14	0	-3.218241	-1.057314	2.179583
33	14	0	-3.052149	1.057206	1.118739
34	1	0	-4.421430	1.655933	1.113329
35	14	0	-4.732229	-2.439004	0.996157
36	1	0	-4.867346	-3.744742	1.701697
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41	6	0	3.668823	0.327399	-1.216470

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46	6	0	7.314271	-0.922456	0.513957
47	1	0	6.302846	-0.350790	2.349189
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52	35	0	1.763443	-3.262349	0.030217
53	8	0	3.207183	0.409689	-2.331130
54	8	0	3.504898	0.741743	2.237199
55	7	0	2.989053	0.705294	-0.035505

References

- (1) Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Mennucci, B.; Petersson, G. A.; Nakatsuji, H.; Caricato, M.; Li, X.; Hratchian, H. P.; Izmaylov, A. F.; Bloino, J.; Zheng, G.; Sonnenberg, J. L.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Montgomery Jr., J. A.; Peralta, J. E.; Ogliaro, F.; Bearpark, M.; Heyd, J. J.; Brothers, E.; Kudin, K. N.; Staroverov, V. N.; Keith, T.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Rega, N.; Millam, J. M.; Klene, M.; Knox, J. E.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Zakrzewski, V. G.; Voth, G. A.; Salvador, P.; Dannenberg, J. J.; Dapprich, S.; Daniels, A. D.; Farkas, O.; Foresman, J. B.; Ortiz, J. V.; Cioslowski, J.; Fox, D. J. *Gaussian 09*, Revision D.01, Gaussian, Inc., Wallingford CT, **2013**.
- (2) Becke, A. D. A new mixing of Hartree-Fock and local density-functional theories. *J. Chem. Phys.* **1993**, *98*, 1372-1377.
- (3) Tian, F.; Cui, Y.; Teplyakov, A. V. Nitroxidation of H-Terminated Si(111) Surfaces with Nitrobenzene and Nitrosobenzene. *J. Phys. Chem. C* **2014**, *118*, 502-512.

III. IRRAS data

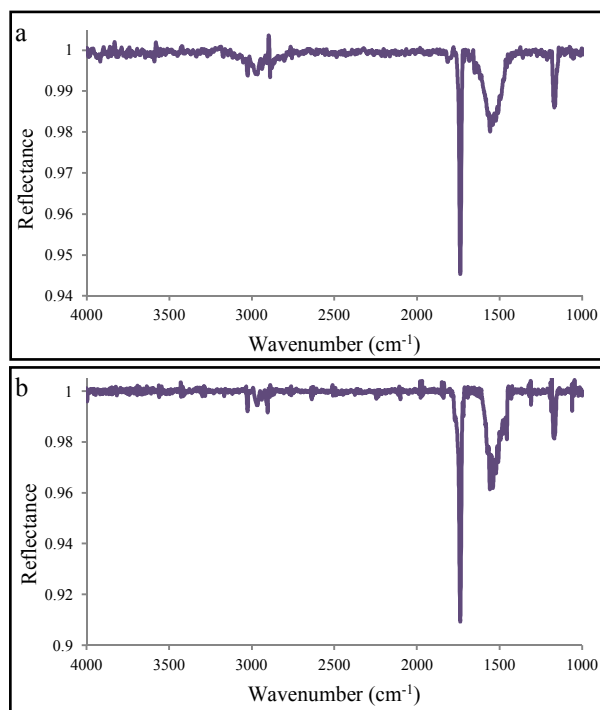


Figure S1. IRRAS data for Si(111) modified with (a) NBS/Fc followed by treatment with a succinimidyl anion.

IV. Line profile details

- The line profile shown in Figure 3d (in the manuscript) is taken (on purpose) in the perpendicular direction of the descending steps (in Figure 3c) to minimize the effect of the steps in deducing an average thickness. So the 10 steps observed in Figure 3c should not be expected in the line profile (Figure 3d).
- The size of the image in Figure 3c is only $0.8 \times 0.8 \mu\text{m}^2$ while that of the image in Figure 3b is $3 \times 3 \mu\text{m}^2$. When the whole scratched area is scanned the resolution is lost.
- The line profile in Figure 3d does indeed show silicon steps as can be seen in the line profile below (Figure S2. This is the same line profile reported in Figure 3d that has just been expended).

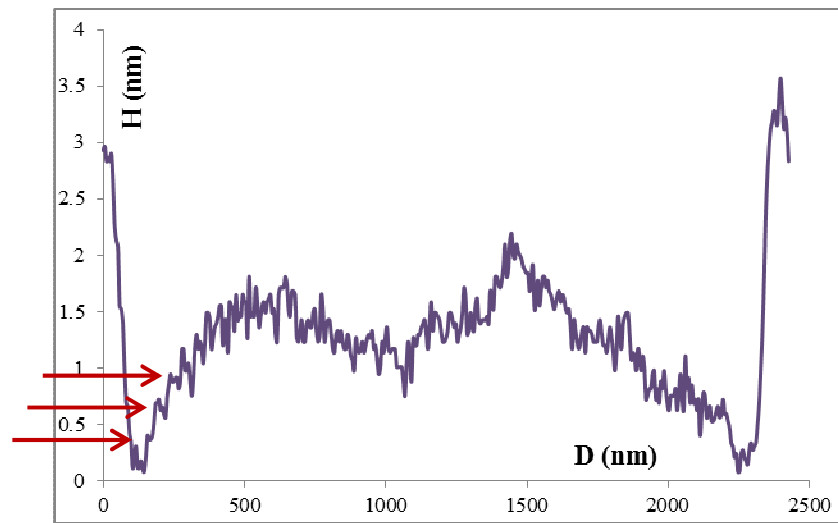


Figure S2. Representative line profile illustrating the approximate depth of the multilayers.

V. Comparison of the XPS Si 2p signal for a modified Si(111) surface and an oxidized Si(111) surface

The Si signal shows certainly the attachment of other elements in addition to O. The Figure below shows the Si 2p signal for an oxidized Si surface (a) and a Si surface modified using the bromosuccinimide / ferrocene mixture (b and c). The modified surface does show the signal observed for the oxidized surface but does also show other signals. The assignments are done in accordance with those reported in the literature to the best of our ability. Br, N and O (from the carbonyl groups of the succinimidyl) can all be attached to Si in the present work, given their potential involvement in radical reactions.

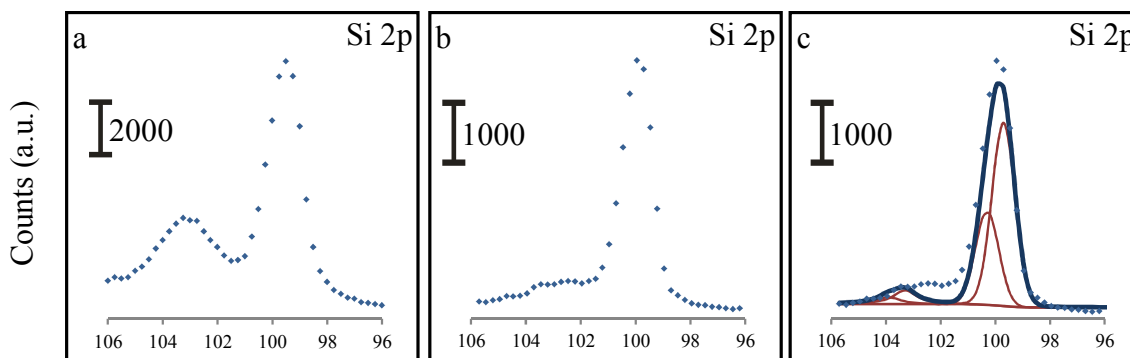


Figure S3. XPS spectra of (a) Si(111) with an oxide layer, (b) Si(111)-H modified with NBS/Fc and (c) Si(111)-H modified with NBS/Fc with peak fitting for the bulk silicon signal and silicon oxide signal. These spectra show differences in signal intensity due to surface modification and (c) suggests that more than more species are present on the surface in addition to silicon oxide.