Supporting Information

Highly Efficient and Reversible Absorption of SO₂ by Aqueous Triethylenetetramine Tetralactate Solutions

Jianguo Qian, ^a Shuhang Ren, ^a Shidong Tian, ^a Yucui Hou, ^b Chenxing Wang ^a and Weize Wu^{*,a}

^aState Key Laboratory of Chemical Resource Engineering, Beijing University of Chemical Technology, Beijing 100029, China;

^bDepartment of Chemistry, Taiyuan Normal University, Taiyuan 030031, China

^{*}Corresponding author; Email: wzwu@mail.buct.edu.cn, Tel./Fax: +86 10 64427603.

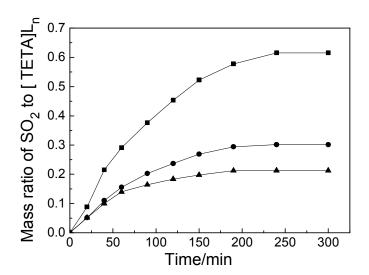


Figure S1. The absorption of 3 % SO_2 in nitrogen based on mass ratio by three kinds of aqueous [TETA] L_n solutions with 50 wt% H_2O under atmospheric pressure at 50 °C: \blacksquare , [TETA] L_2 ; \bullet , [TETA] L_3 ; \blacktriangle , [TETA] L_4 .

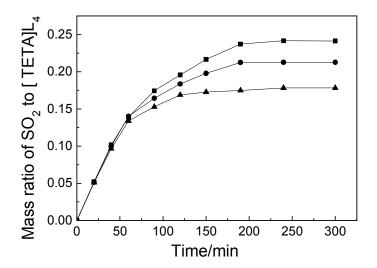


Figure S2. The absorption of 3 % SO_2 in nitrogen based on mass ratio by aqueous [TETA]L₄ solution with 50 wt% H₂O under atmospheric pressure at different temperatures: \blacksquare , 40 °C; \bullet , 50 °C; \blacktriangle , 60 °C.

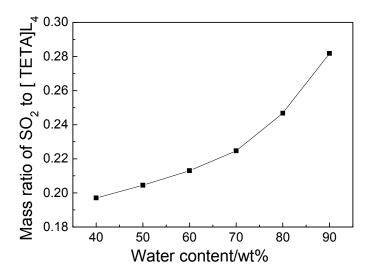


Figure S3. The absorption capacity of 3 % SO_2 in nitrogen based on mass ratio by aqueous [TETA]L₄ solution with different water contents at 50 °C under atmospheric pressure.

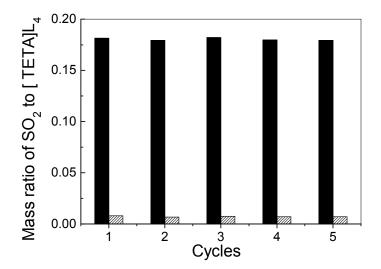


Figure S4. The absorption and desorption of 3 % SO_2 in nitrogen based on mass ratio by aqueous [TETA]L₄ solution with 30 wt% water at 60 °C under atmospheric pressure: \blacksquare , absorption; \square , desorption.