

'Choose-a-size' approach in dynamic combinatorial chemistry: a single substrate dynamic combinatorial library of oligomacrocycles that adapts to the size and shape of carboxylates

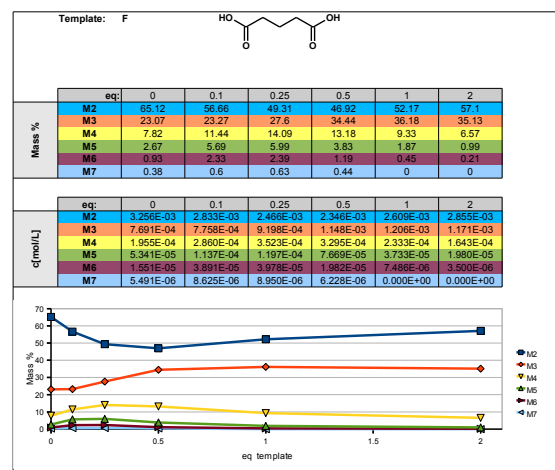
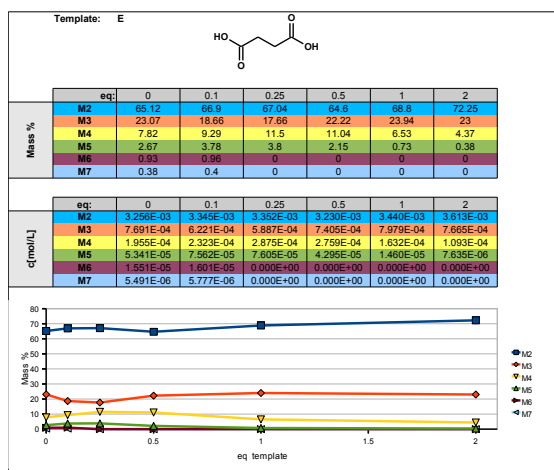
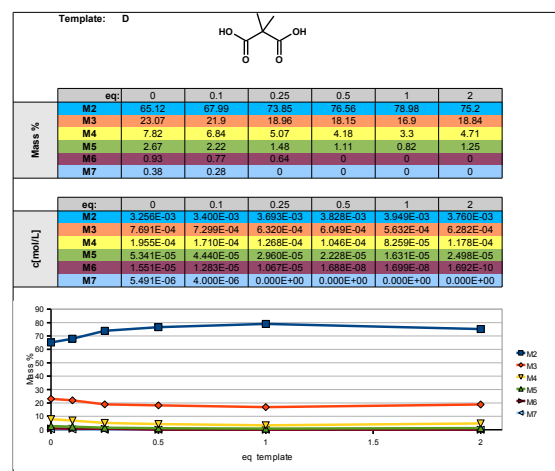
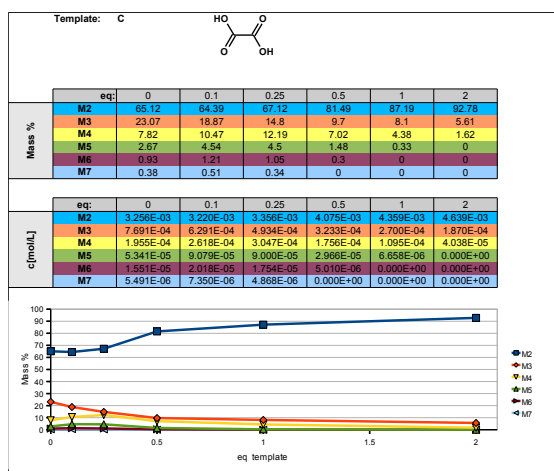
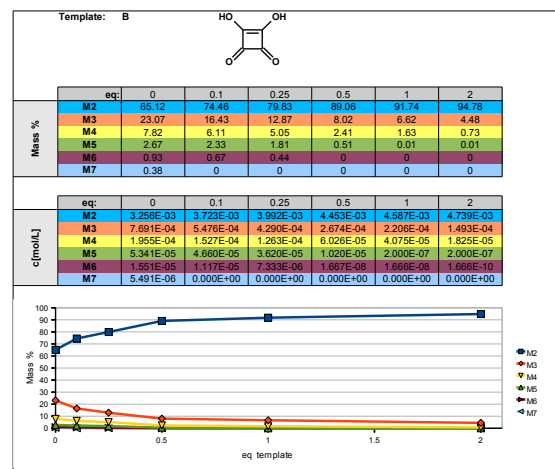
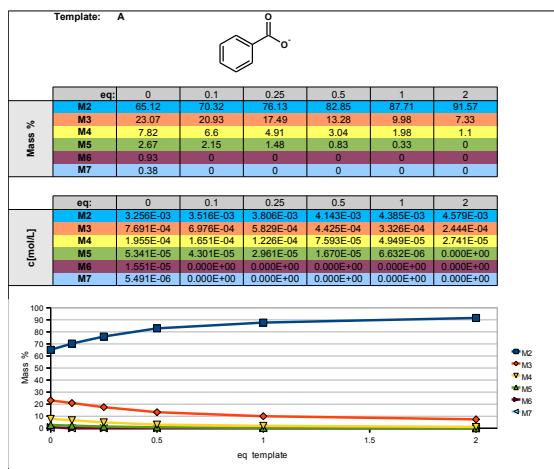
Supporting Information

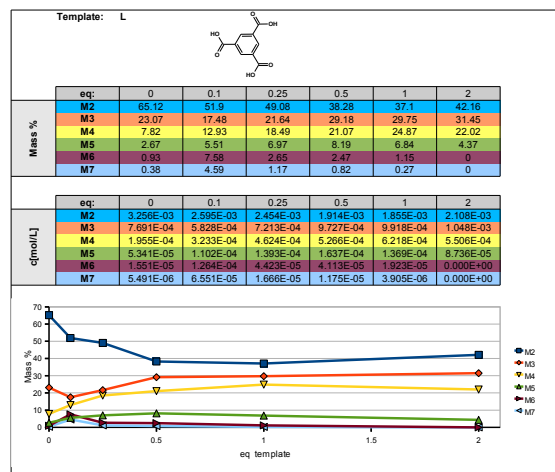
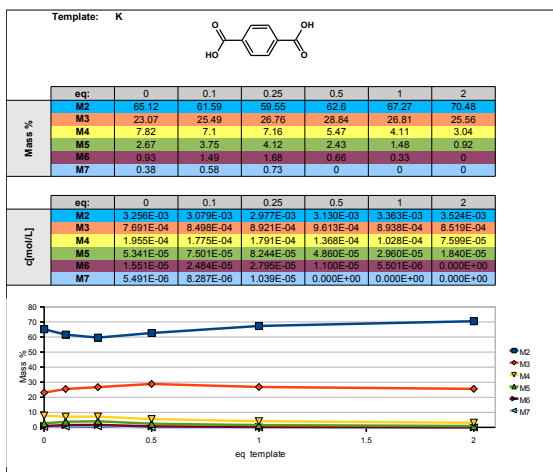
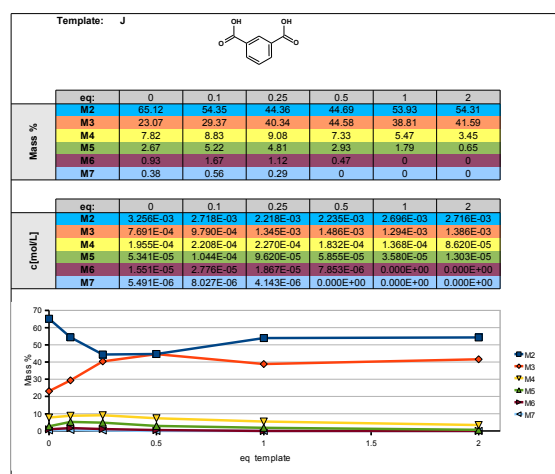
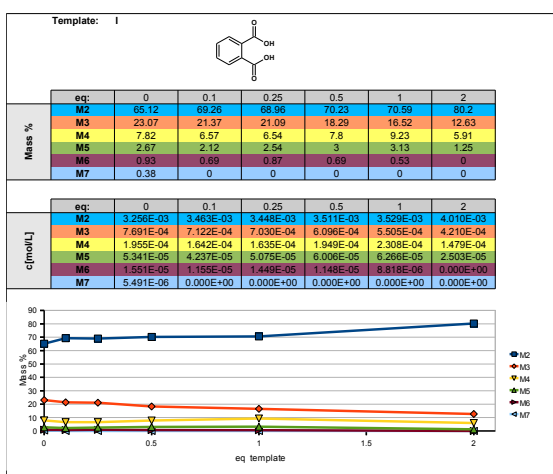
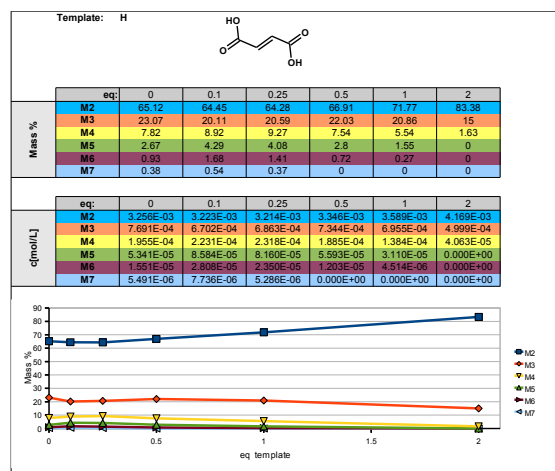
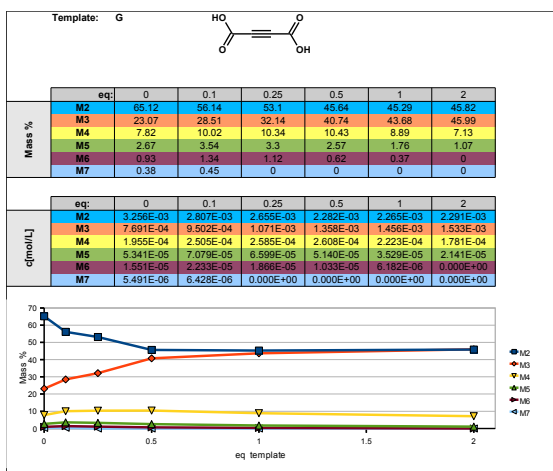
Filip Ulatowski, Agnieszka Sadowska-Kuzioła, Janusz Jurczak*

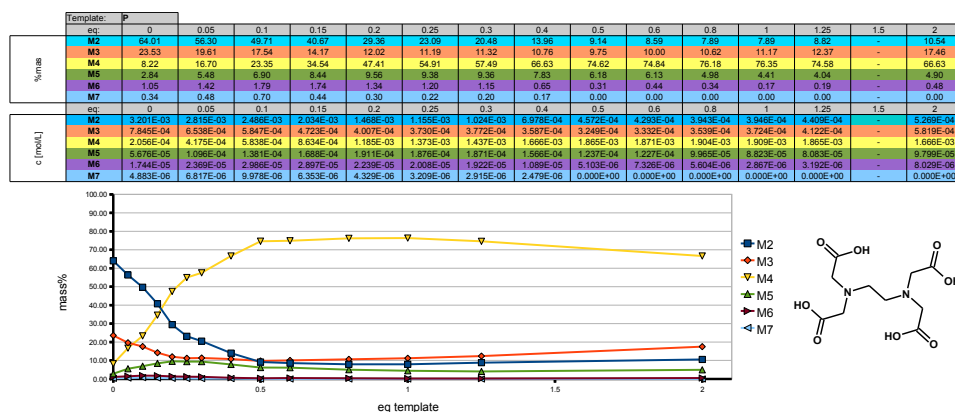
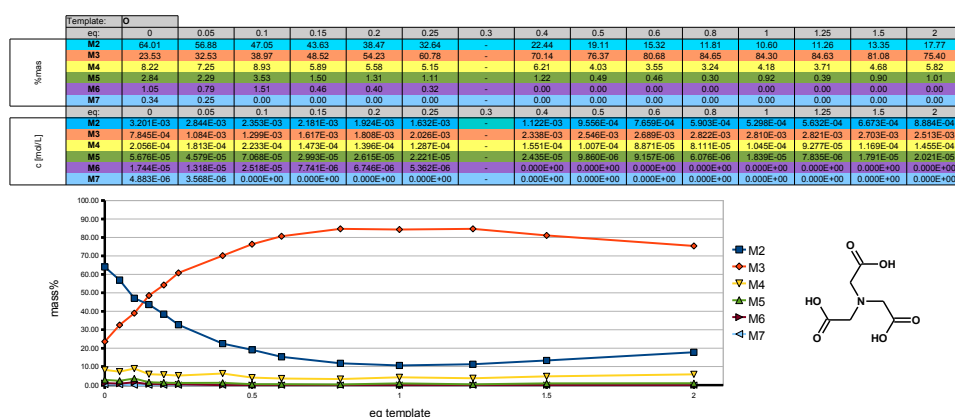
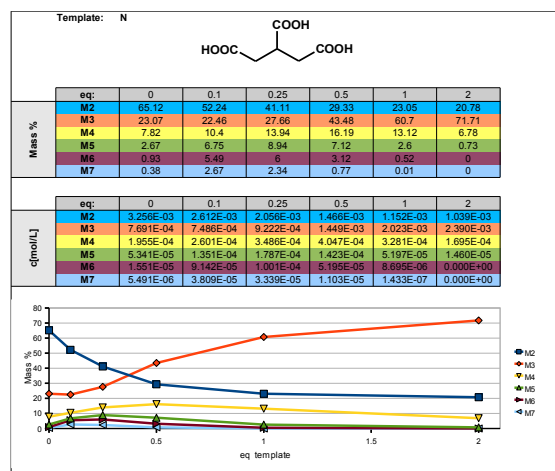
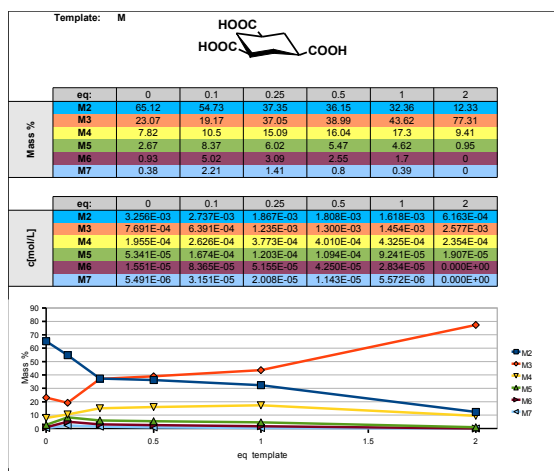
Contents

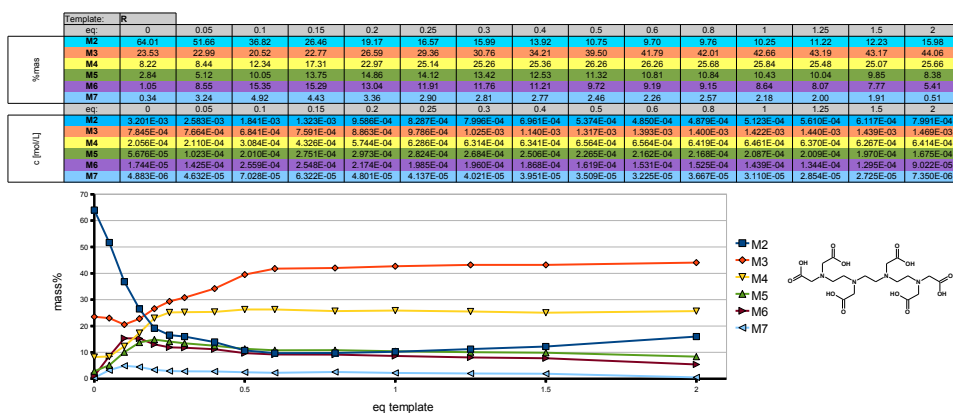
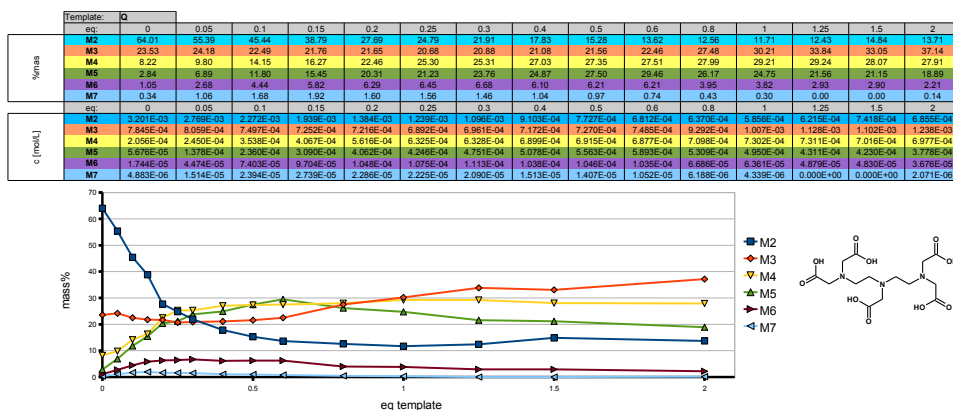
S1 Compositions of templated libraries	S2
S2 Reaction of a library with malonate	S5
S3 HPLC-HRMS analysis	S6
S4 Preparative HPLC	S8
S5 DCLfit analyses	S8
S5.1 Analysis of untemplated library	S8
S5.2 Analyses of templated libraries	S9
S6 Competitive ¹H NMR titration	S12
S7 NMR spectra of new compounds	S14

S1 Compositions of templated libraries



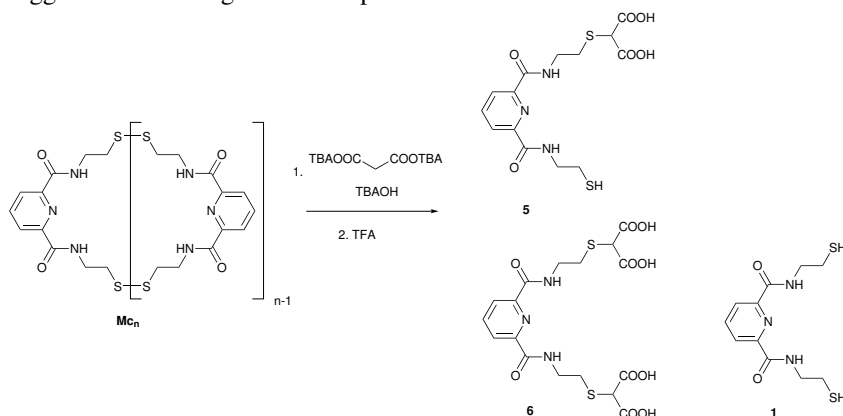






S2 Reaction of a library with malonate

Templation of a library Mc_n with an excess of malonate diTBA salt resulted in disappearance of the macrocycles. We suggest the following reaction to proceed:



In the reaction mixture we identified compounds **1** and **5** by the means of MS.

S3 HPLC-HRMS analysis

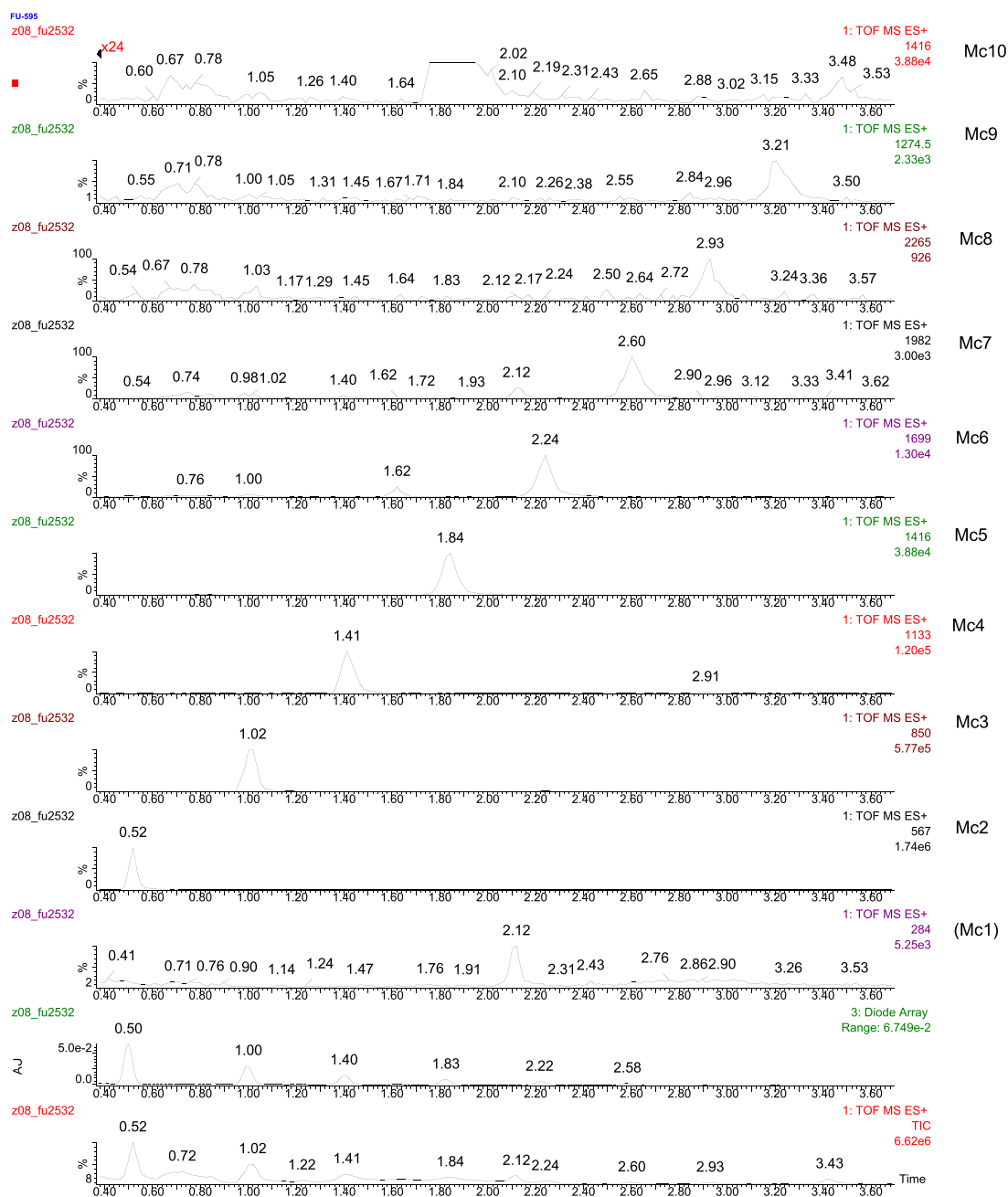


Figure S1: LC-MS analysis of the library. Total ion current, UV (220 nm), and extracted ion chromatograms for **Mc₁-Mc₁₀**

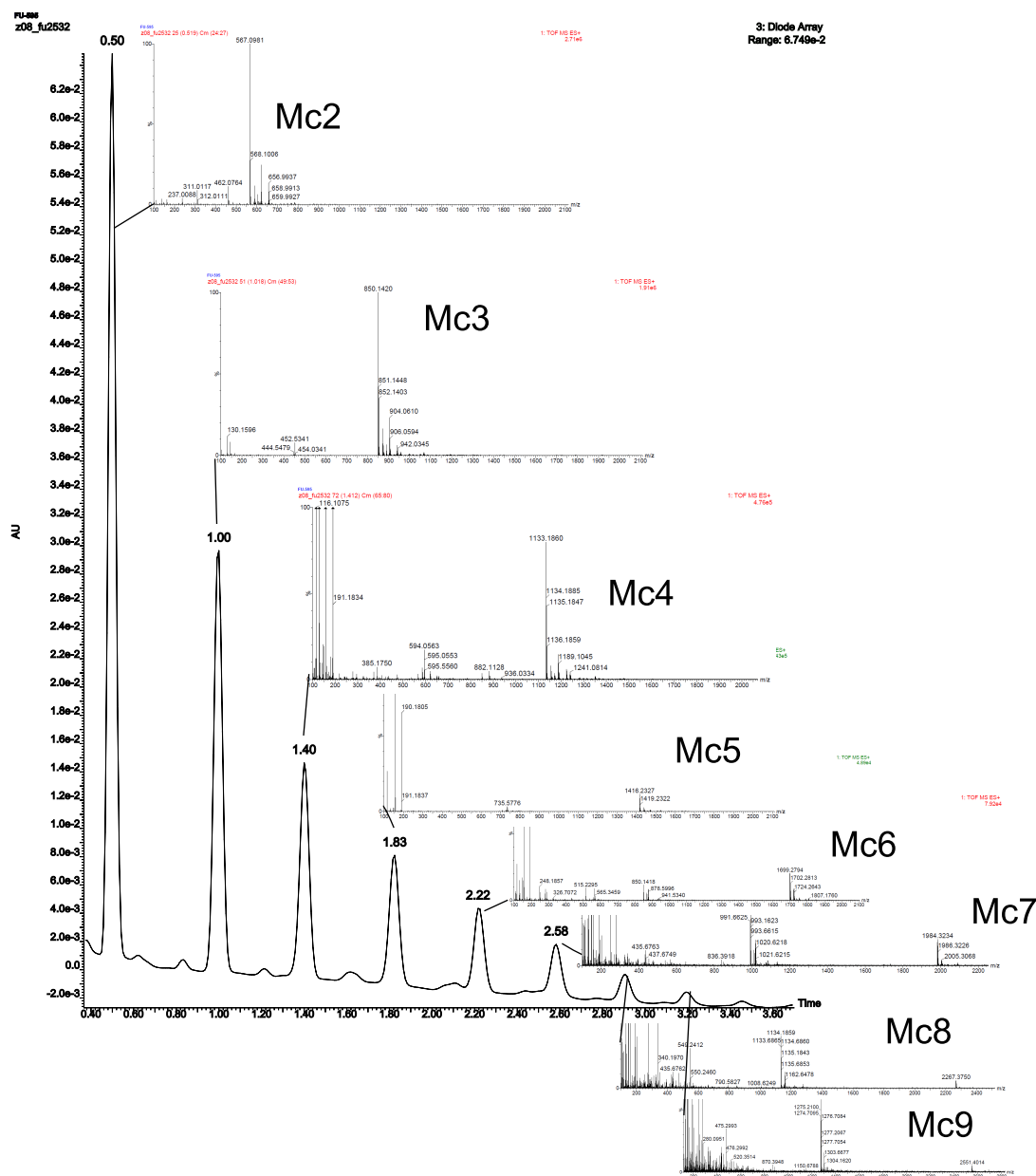


Figure S2: UV trace of LCMS analysis and MS spectra for each peak

HRMS results for oligomers:

Mc₂	Calc. for C ₂₂ H ₂₇ N ₆ O ₄ S ₄ (M + H ⁺):	567.0971
	found:	567.0981
Mc₃	Calc. for C ₃₃ H ₄₀ N ₉ O ₆ S ₆ (M + H ⁺):	850.1420
	found:	850.1420
Mc₄	Calc. for C ₄₄ H ₅₃ N ₁₂ O ₈ S ₈ (M + H ⁺):	1133.1869
	found:	1133.1860
Mc₅	Calc. for C ₅₅ H ₆₆ N ₁₅ O ₁₀ S ₁₀ (M + H ⁺):	1416.2318
	found:	1416.2327
Mc₆	Calc. for C ₆₆ H ₇₉ N ₁₈ O ₁₂ S ₁₂ (M + H ⁺):	1699.276
	found:	1699.279
Mc₇	Calc. for C ₇₇ H ₉₂ N ₂₁ O ₁₄ S ₁₄ (M + H ⁺):	1982.321
	found:	1982.327

S4 Preparative HPLC

The libraries were prepared as described above (typically 5 mL, 0.02 mmol, 28 mg) and were templated with appropriate template at its optimal concentration:

Template	isolated yields [%]			
	Mc ₂	Mc ₃	Mc ₄	Mc ₅
A, 2 eq	70	0	0	0
O, 1 eq	15	60	5	0
P, 0.75 eq	25	20	40	4
Q, 0.6 eq	10	18	22	25

Preparative isolation of oligomeric macrocycles was performed with HPLC column (Poligrorep C18 12um i.d. 20 mm l 300 mm) with gradient elution:

time [min]	eluent composition [%]	
	DMF	H ₂ O
0	60	40
5	60	40
20	73	27
40	80	20

UV detector was set to $\lambda = 285$ nm.

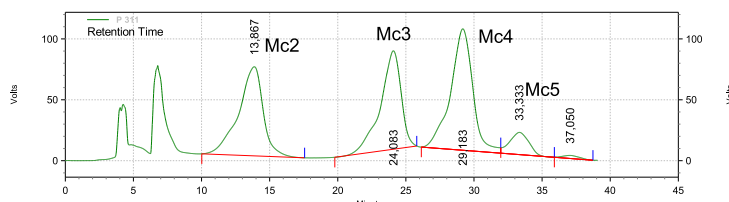


Figure S3: An example of preparative resolution of library members. DCC templated with EDTA (P).

S5 DCLfit analyses

S5.1 Analysis of untemplated library

Since monomeric macrocycle does not form upon oxidation of **1** it is treated as virtual molecule which is used to calculate formation constants of higher oligomers. The first formation constant K_2 corresponds to the reaction:



$$K_2 = \frac{[\text{Mc}_2]}{[\text{Mc}_1]^2} \quad (2)$$

This formation constant is arbitrary set to $\log(K_2) = 1$, and this defines the concentration of virtual **Mc**₁. Other macrocycles are in equilibrium according to following equations:



$$K_n = \frac{[\text{Mc}_n]}{[\text{Mc}_{n-1}] \cdot [\text{Mc}_1]} \quad (4)$$

The association constants K_3 - K_6 were fitted for a series of libraries with various total concentrations. The fitted values are as follows:

$$\log(K_2) \stackrel{\text{def}}{=} 1.00$$

$$\log(K_3) = 1.13$$

$$\log(K_4) = 1.19$$

$$\log(K_5) = 1.21$$

$$\log(K_6) = 1.20$$

K_3 is slightly lower than following values which indicates that $\mathbf{Mc}_2$ is more stable. K_4 - K_6 being nearly equal indicate simple polymerisation model for higher oligomers with a single formation constant K_n . The values obtained in this set of experiments were used in analyses of templated libraries.

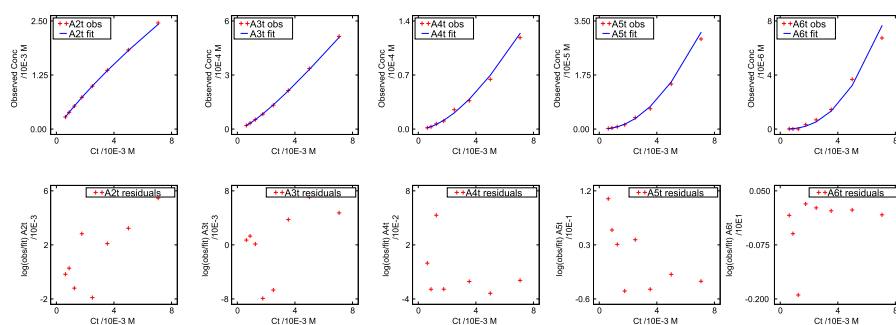


Figure S4: Concentrations of components of untemplated libraries at various total concentrations fitted with DCLfit software

S5.2 Analyses of templated libraries

Association constants fitted in the experiments were defined as follows:

$$K(\mathbf{Mc}_n - \mathbf{T}_m) = \frac{[\mathbf{Mc}_n \mathbf{T}_m]}{[\mathbf{Mc}_n \mathbf{T}_{m-1}] \cdot [\mathbf{T}]} \quad (5)$$

$$\mathbf{Mc}_n \mathbf{T}_{m-1} + \mathbf{T} \rightleftharpoons \mathbf{Mc}_n \mathbf{T}_m \quad (6)$$

Template A Applied model includes following species (**T** stands for template) : $\mathbf{Mc}_2$; $\mathbf{Mc}_3$; $\mathbf{Mc}_4$; $\mathbf{Mc}_5$; $\mathbf{Mc}_2\text{-T}$; $\mathbf{Mc}_3\text{-T}$; $\mathbf{Mc}_3\text{-T}_2$; $\mathbf{Mc}_4\text{-T}$; $\mathbf{Mc}_4\text{-T}_2$; $\mathbf{Mc}_5\text{-T}$; $\mathbf{Mc}_5\text{-T}_2$.

Values obtained:

$$\log(K(\mathbf{M}_2\mathbf{T})) = 2.78$$

$$\log(K(\mathbf{M}_3\mathbf{T})) = 2.55$$

$$\log(K(\mathbf{M}_4\mathbf{T})) = 2.67$$

$$\log(K(\mathbf{M}_5\mathbf{T})) = 2.85$$

$$\log(K(\mathbf{M}_3\mathbf{T}_2)) = 0.85$$

$$\log(K(\mathbf{M}_4\mathbf{T}_2)) = 1.02$$

$$\log(K(\mathbf{M}_5\mathbf{T}_2)) = 1.61$$

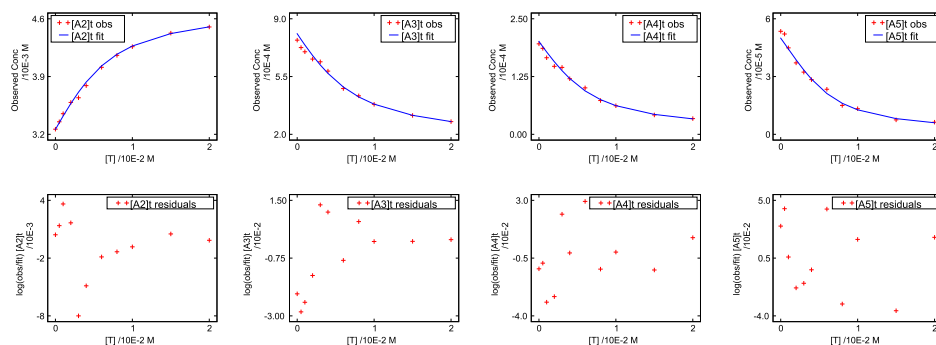


Figure S5: Templatation effects of anion A fitted with DCLfit

Template O Applied model includes following species (**T** stands for template) : **Mc₂**; **Mc₃**; **Mc₄**; **Mc₅**; **Mc₂-T**; **Mc₃-T**; **Mc₃-T₂**; **Mc₄-T**; **Mc₄-T₂**; **Mc₅-T**; **Mc₅-T₂**.
Values obtained:

$$\log(K(Mc_2T)) = 1.99$$

$$\log(K(Mc_3T)) = 4.06$$

$$\log(K(Mc_4T)) = 3.50$$

$$\log(K(Mc_5T)) = 3.58$$

$$\log(K(Mc_3T_2)) = 0.00$$

$$\log(K(Mc_4T_2)) = 1.93$$

$$\log(K(Mc_5T_2)) = 2.00$$

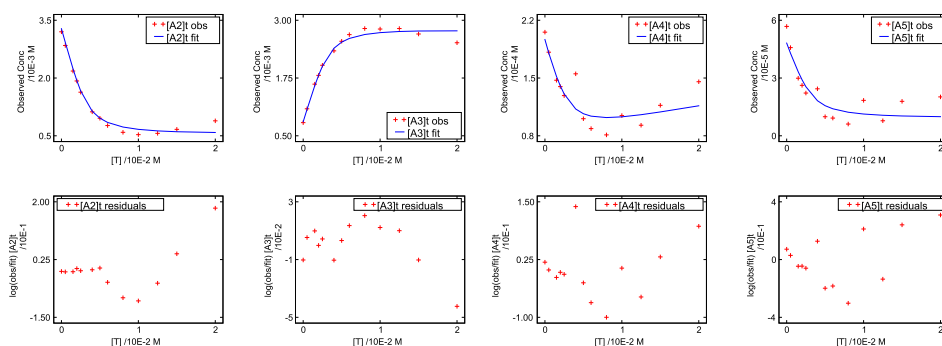


Figure S6: Templatation effects of anion **O** fitted with DCLfit

Template P Applied model includes following species (**T** stands for template) : **Mc₂**; **Mc₃**; **Mc₄**; **Mc₅**; **Mc₂-T**; **Mc₃-T**; **Mc₃-T₂**; **Mc₄-T**; **Mc₄-T₂**; **Mc₅-T**; **Mc₅-T₂**.
Values obtained:

$$\log(K(Mc_2T)) = 1.64$$

$$\log(K(Mc_3T)) = 3.33$$

$$\log(K(Mc_4T)) = 5.10$$

$$\log(K(Mc_5T)) = 4.97$$

$$\log(K(Mc_3T_2)) = 0.00$$

$$\log(K(Mc_4T_2)) = 1.86$$

$$\log(K(Mc_5T_2)) = 0.00$$

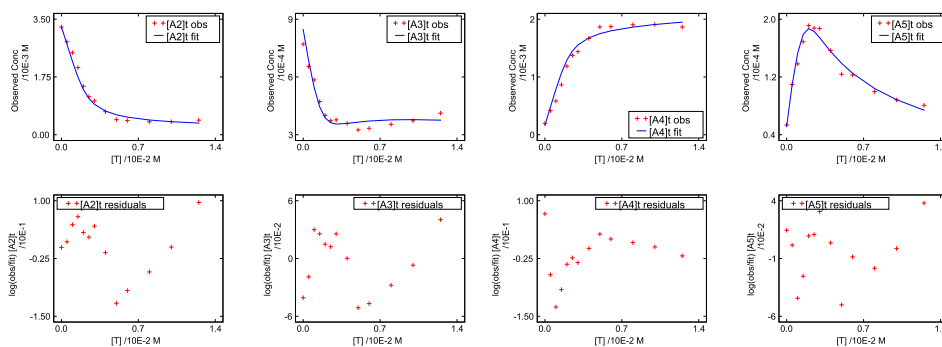


Figure S7: Templatation effects of anion **P** fitted with DCLfit

Template Q Applied model includes following species (**T** stands for template) : **Mc₂**; **Mc₃**; **Mc₄**; **Mc₅**; **Mc₂-T**; **Mc₃-T**; **Mc₃-T₂**; **Mc₄-T**; **Mc₄-T₂**; **Mc₅-T**; **Mc₅-T₂**; **Mc₆-T**; **Mc₆-T₂**.
Values obtained:

$$\log (K (Mc_2T)) = 1.85$$

$$\log (K (Mc_3T)) = 3.38$$

$$\log (K (Mc_4T)) = 4.24$$

$$\log (K (Mc_5T)) = 4.84$$

$$\log (K (Mc_6T)) = 6.21$$

$$\log (K (Mc_3T_2)) = 1.43$$

$$\log (K (Mc_4T_2)) = 1.33$$

$$\log (K (Mc_5T_2)) = 2.04$$

$$\log (K (Mc_6T_2)) = 1.84$$

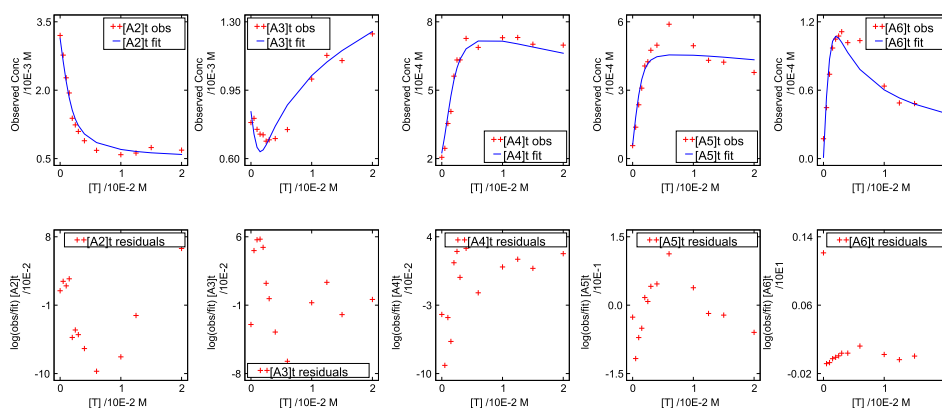


Figure S8: Templatation effects of anion **Q** fitted with DCLfit

Template R Applied model includes following species (**T** stands for template) : **Mc₂**; **Mc₃**; **Mc₄**; **Mc₅**; **Mc₂-T**; **Mc₃-T**; **Mc₃-T₂**; **Mc₄-T**; **Mc₄-T₂**; **Mc₅-T**; **Mc₅-T₂**; **Mc₆-T**; **Mc₆-T₂**.
Values obtained:

$$\log (K (Mc_2T)) = 2.49$$

$$\log (K (Mc_3T)) = 4.30$$

$$\log (K (Mc_4T)) = 5.05$$

$$\log (K (Mc_5T)) = 5.58$$

$$\log (K (Mc_6T)) = 7.51$$

$$\log (K (Mc_3T_2)) = 1.21$$

$$\log (K (Mc_4T_2)) = 1.51$$

$$\log (K (Mc_5T_2)) = 2.60$$

$$\log (K (Mc_6T_2)) = 3.13$$

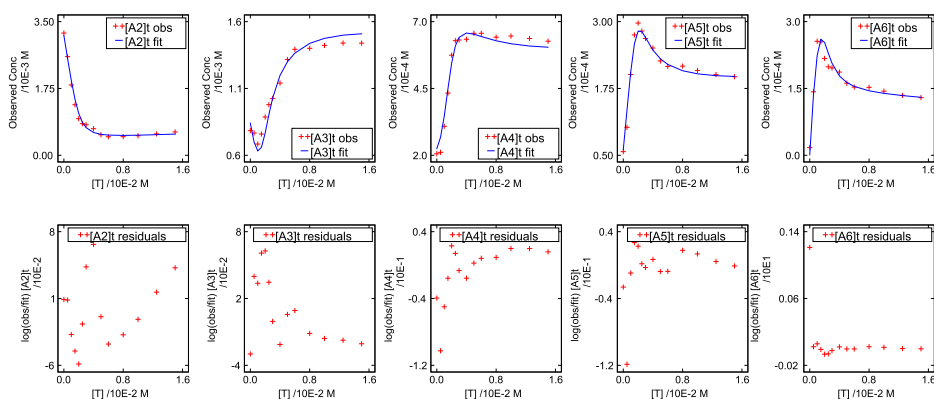


Figure S9: Templatation effects of anion **R** fitted with DCLfit

S6 Competitive ^1H NMR titration

Separate stock solutions of pure **Mc**₂–**Mc**₅ were prepared in DMSO-*d*₆, *c*=10 mM. In the competitive experiments 50 μL of these solutions were mixed in an NMR tube and diluted with 0.3 mL to obtain a final concentration of 1 mM of each macrocycle. The solutions were titrated with solutions of appropriate TBA salt (**O**, **P**, and **Q**) in DMSO-*d*₆ (*c*=33 mM) added in portions of 5 μL (1/3 eq of each macrocycle). Additional titrations of pure **Mc**₃ and **Mc**₄ hosts were performed for unambiguous signals assignment.

The titrations are presented herein:

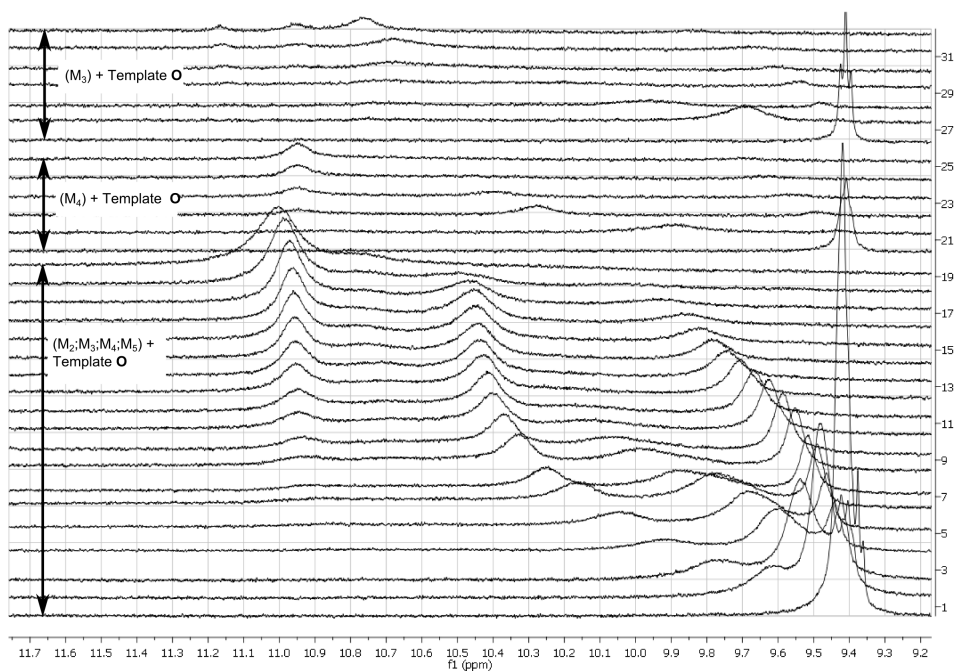


Figure S10: Competitive titration with template **O** (NTA anion)

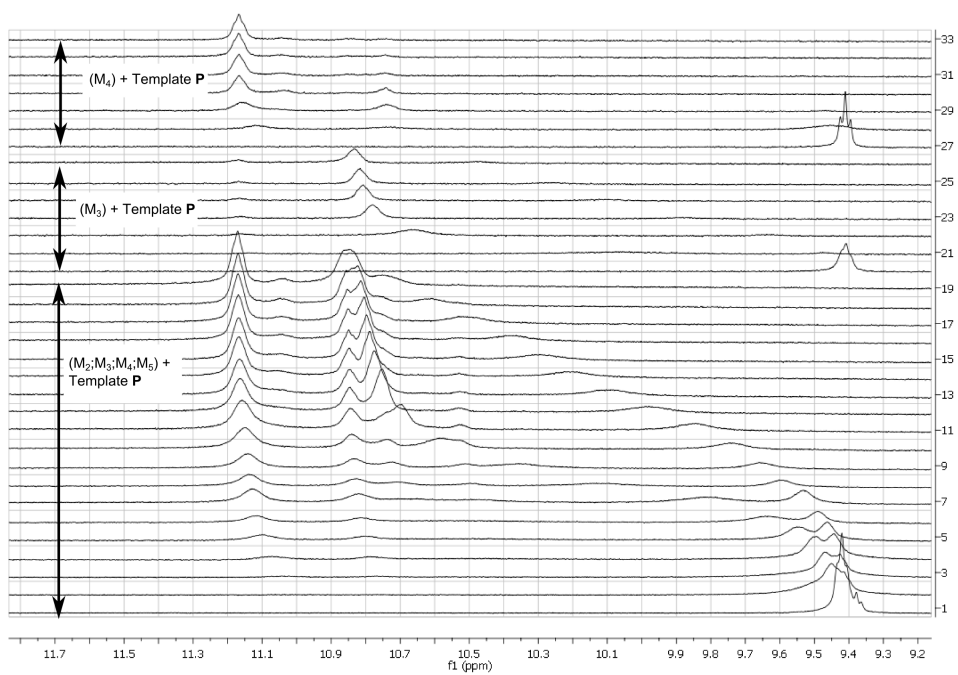


Figure S11: Competitive titration with template **P** (EDTA anion)

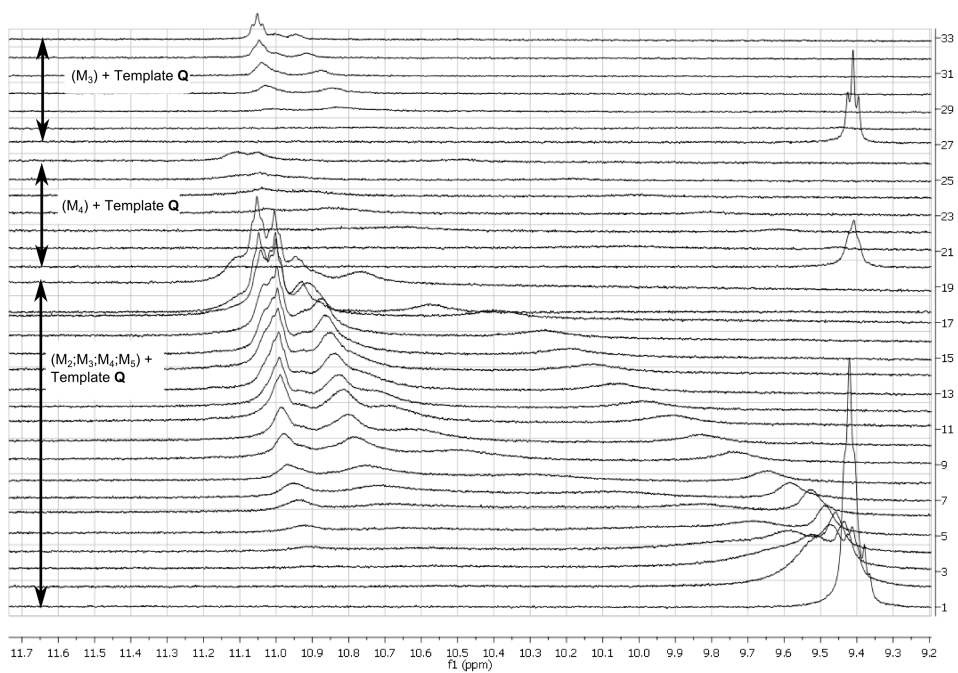


Figure S12: Competitive titration with template **Q** (DTPA anion)

S7 NMR spectra of new compounds

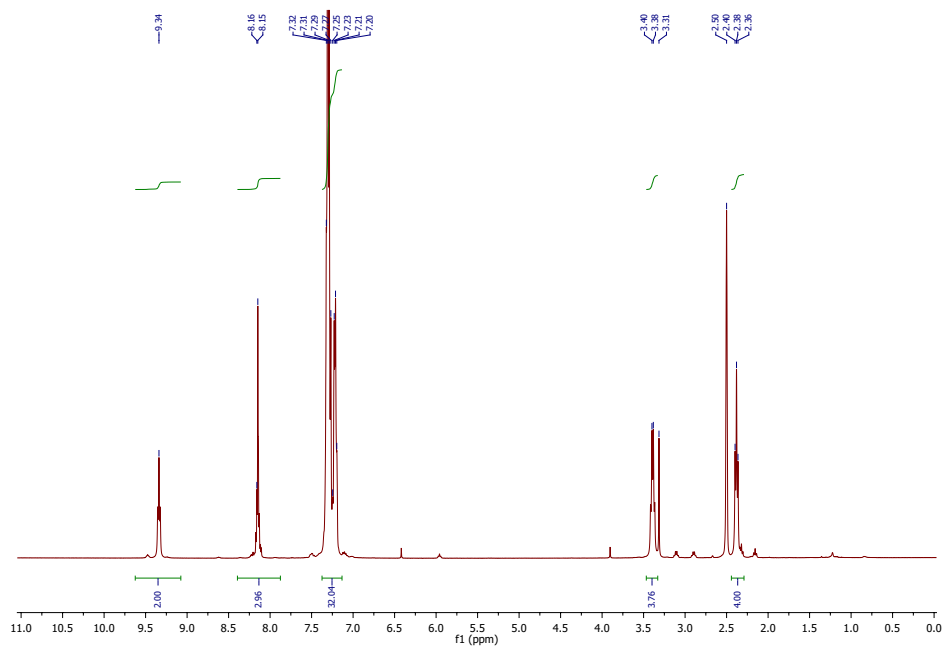


Figure S13: ¹H NMR spectrum of **4c** in CDCl₃

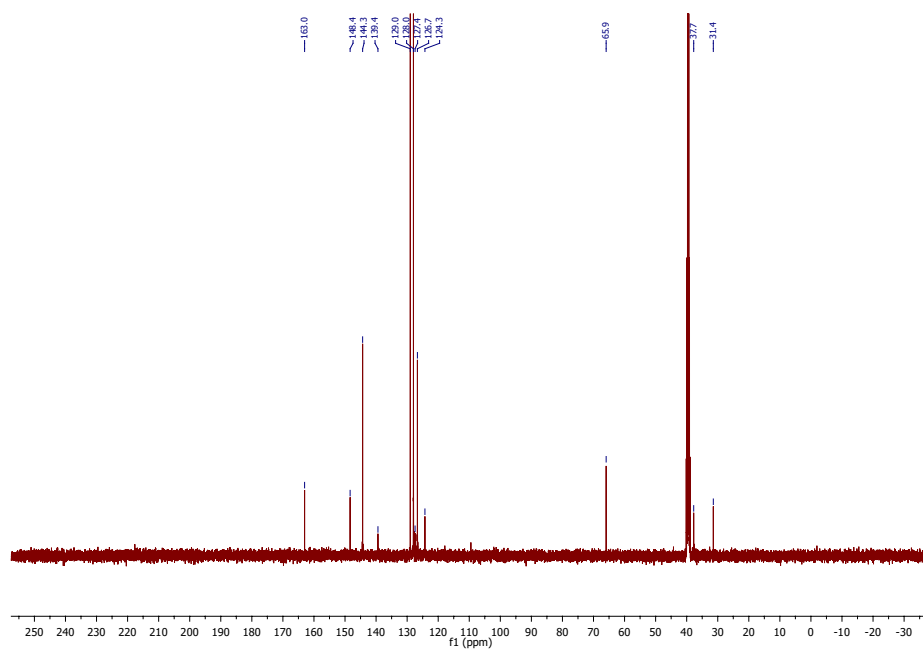


Figure S14: ¹³C NMR spectrum of **4c** in CDCl₃

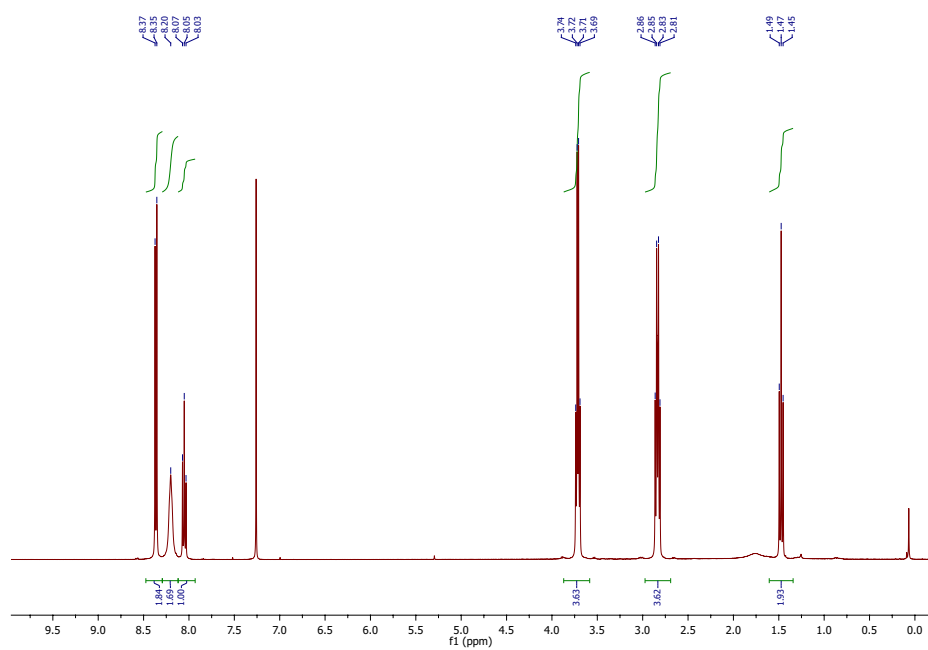


Figure S15: ¹H NMR spectrum of **1** in DMSO-d₆

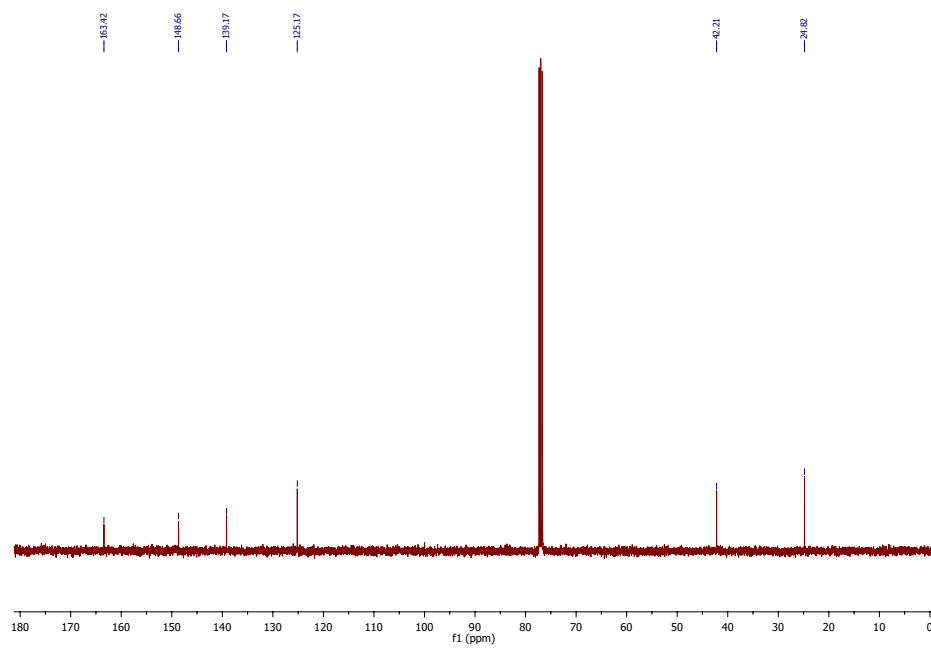


Figure S16: ¹³C NMR spectrum of **1** in DMSO-d₆