

Table S1. Anthocyanins half-life values.

symbol	half-life (days)	
	C3G	P3G
C	44.6 ± 1.3 <b>a</b>	60.8 ± 2.3 <b>a</b>
Ste200	40.6 ± 1.6 <b>a</b>	60.9 ± 0.6 <b>a</b>
Ste125	43.1 ± 0.8 <b>a</b>	59.6 ± 1.7 <b>a</b>
Ste50	43.5 ± 2.3 <b>a</b>	60.6 ± 2.5 <b>a</b>
RebA200	42.3 ± 1.5 <b>a</b>	58.1 ± 0.6 <b>a</b>
RebA125	42.3 ± 1.3 <b>a</b>	58.7 ± 0.2 <b>a</b>
RebA50	44.1 ± 0.1 <b>a</b>	59.0 ± 3.1 <b>a</b>
Suc	59.4 ± 2.0 <b>b</b>	81.2 ± 5.0 <b>b</b>
GF	53.1 ± 3.0 <b>c</b>	70.2 ± 1.8 <b>c</b>
Mix	68.3 ± 1.1 <b>d</b>	92.2 ± 2.7 <b>d</b>

Sample symbols as in Table 1. Statistical significance of data was determined by Tukey test ( $\alpha=0.05$ ); statistics are separate for each column and are presented in table as lowercase letters.

Table S2. Ascorbic and dehydroascorbic acid degradation rates.

symbol	rate of the reaction ( $s^{-1}$ ) $\times 10^6$	
	$k_1$ (AA)	$k_2$ (DHAA)
pH4.0/C	$11.66 \pm 0.10$ a	$14.27 \pm 0.06$ a
pH4.0/Ste200	$10.77 \pm 0.05$ b	$8.55 \pm 0.12$ b
pH4.0/Ste125	$10.98 \pm 0.12$ b	$9.67 \pm 0.23$ c
pH4.0/Ste50	$11.19 \pm 0.04$ c	$10.47 \pm 0.24$ d
pH4.0/Suc	$9.68 \pm 0.08$ d	$10.75 \pm 0.16$ b
pH3.6/C	$5.71 \pm 0.05$ e	$7.30 \pm 0.15$ e
pH3.6/Ste200	$4.80 \pm 0.02$ f	$4.36 \pm 0.06$ f
pH3.6/Ste125	$5.04 \pm 0.05$ g	$5.14 \pm 0.39$ f
pH3.6/Ste50	$5.28 \pm 0.02$ h	$5.25 \pm 0.03$ g
pH3.6/Suc	$4.79 \pm 0.07$ h	$5.64 \pm 0.11$ f
pH3.2/C	$4.43 \pm 0.03$ i	$5.43 \pm 0.13$ f
pH3.2/Ste200	$3.44 \pm 0.04$ j	$3.16 \pm 0.01$ gh
pH3.2/Ste125	$3.76 \pm 0.06$ jk	$3.64 \pm 0.01$ hi
pH3.2/Ste50	$3.90 \pm 0.02$ l	$3.81 \pm 0.00$ i
pH3.2/Suc	$3.65 \pm 0.03$ k	$4.05 \pm 0.12$ gh

Sample symbols as in Table 2. Statistical significance of data was determined by Tukey test ( $\alpha=0.05$ ); statistics are separate for each column and are presented in table as lowercase letters.