## **Supporting Information**

## Nitrogen-Doped Graphene/Pt Counter Electrodes for Dye-Sensitized Solar Cells

1-An Lin,<sup>a,b</sup> Chuan-Pei Lee,<sup>b</sup> Shu-Te Ho,<sup>a</sup> Tzu-Chiao Wei,<sup>b</sup> Yu-Wen Chi,<sup>c</sup> K. P. Huang,<sup>c</sup> and Jr-Hau He<sup>a,</sup>\*

nputer, Electrical and Mathematical Sciences and Engineering (CEMSE) Division, King Abdullah University of Science &

nology (KAUST), Thuwal 23955-6900, Saudi Arabia

itute of Photonics and Optoelectronics, & Department of Electrical Engineering, National Taiwan University, Taipei, Taiwan, R

chanical and Systems Research Laboratories, Industrial Technology Research Institute, Hsinchu, Taiwan, ROC

## surements and instruments:

The electronic structures of NGRs were confirmed by PHI Quantera XPS. Morphological study of NGR has been performed w L JSM-6500 field emission scanning electron microscopy and a JEOL JEM-3000F high-resolution transmission electroscopy. The excitation source of Raman spectroscopy is a 532 nm laser beam (2.33 eV) with a power below 1 mW to a --induced heating. Optical reflectance spectroscopy was measured for wavelength region between 400 and 1100 nm und vis-NIR spectrometer (JASCO ARN-733). IPCE measurement has also been carried out with a spectral response system (anology Co., Ltd. R3011) for wavelength range from 400 to 850 nm. The Surface of the DSSC was covered by a mask we --illuminated area of 0.16 cm<sup>2</sup> and then illuminated under 1 sun intensity (AM1.5G). Incident light intensity (100 mW cm<sup>-2</sup>) rated with a standard Si Cell (Hong-Ming tech. Co., Ltd.). Photocurrent-voltage curves of the DSSC's were obtained with a dicce meter (2400, Keithley).

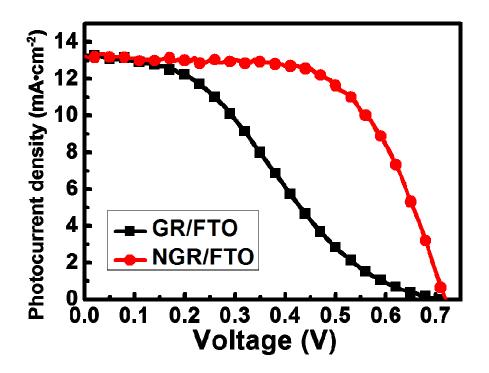
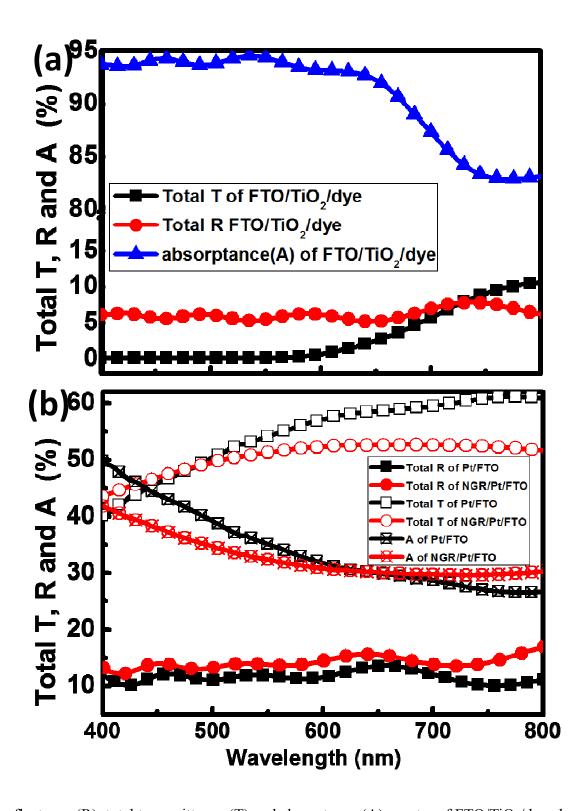


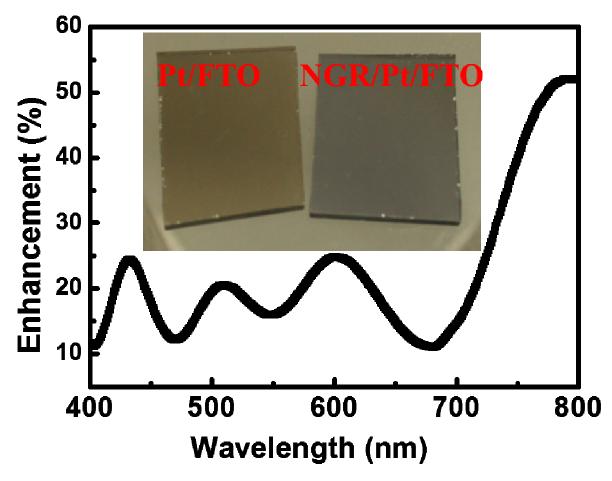
Figure S1 The *J*-*V* curves of DSSCs with CEs using GR and NGR.

**e S1** Photovoltaic parameters of DSSCs with CEs using GR and NGR alone, measured under 100 mW·cm<sup>-2</sup> of AM 1.5G nination.

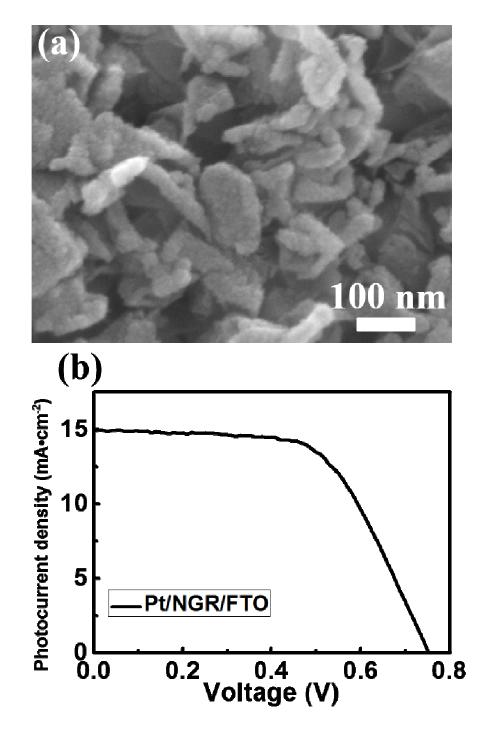
CEs	$V_{\rm OC}({\rm mV})$	$J_{\rm SC}({\rm mA}{\cdot}{\rm cm}^{-2})$	FF	η (%)
GR/FTO	710	13.24	0.31	2.94
NGR/FTO	718	13.26	0.61	5.84



**Tre S2** (a) Total reflectance (R), total transmittance (T) and absorptance (A) spectra of FTO/TiO<sub>2</sub>/dye electrodes. (b) Total R, to A spectra of Pt/FTO and NGR/Pt/FTO CEs.



re S3. The enhancement of total R spectrum. The inset is the photoimage of Pt/FTO and NGR/Pt/FTO CEs.

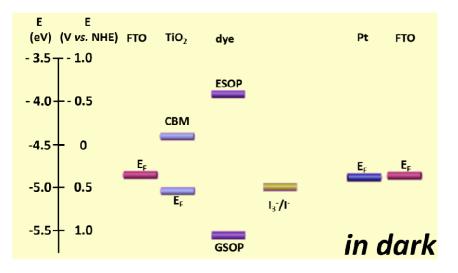


re S4. (a) The SEM image of Pt/NGR/FTO and (b) the *J-V* curve of the DSSC with Pt/NGR/FTO CEs.

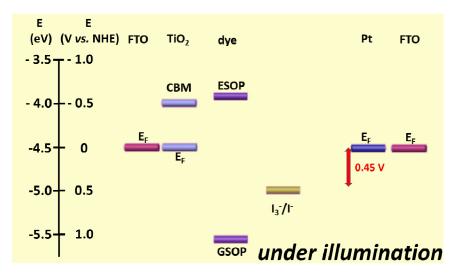
In the case of the DSSC with CE using a reverse bi-layer structure, Pt/NGR/FTO, shows a  $\eta$  of 6.81%, and its  $J_{SC}$ ,  $V_{OC}$ , and es are 14.92 mA·cm<sup>-2</sup>, 745 mV, and 0.61, respectively

According to reported literatures, the conduction band minimum (CBM, -4.4 eV)<sup>S1</sup> and Fermi level ( $E_F$ , -5.1 eV)<sup>S2</sup> of TiO<sub>2</sub> figround/excited state oxidation potentials (GSOP/ESOP, -0.6/1.12 V *vs.* NHE)of N719 dye<sup>S3</sup>, the redox potential of I<sub>3</sub><sup>-</sup>/ $\Gamma$  (0.5 V E)<sup>S1</sup>, and the  $E_F$  of FTO (-4.9 eV)<sup>S2</sup> and Pt (-4.95 eV)<sup>S4</sup> in the dark without  $E_F$  alignment have been schematically shown in Fi In Figure S5, the VBM of TiO<sub>2</sub> is not shown because it is much deeper than GSOP of dye and thus it does not participate in p ting.<sup>S5</sup> Obviously, the  $E_F$  of Pt (0.45 V *vs.*NHE) is slightly higher than the redox potential of I<sub>3</sub><sup>-</sup>/ $\Gamma$  (0.5 V *vs.* NHE)<sup>S1</sup> under lition before  $E_F$  alignment.

While the DSSC is illuminated by solar light, the band diagram would be changed as shown in Figure S6. Under illumination A of TiO<sub>2</sub> film (-4.0 eV)<sup>S6</sup>, the GSOP/ESOP (-0.6/1.12 V, *vs.* NHE)<sup>S3</sup> of N719 dye, the redox potential of I<sub>3</sub>/T (0.5 V *vs.* NH e schematically shown in Figure S6. Note that, under illumination, the E<sub>F</sub> alignment of FTO, TiO<sub>2</sub> and Pt is achieved du ocarrier injection, and thus the E<sub>F</sub> is called quasi-Fermi level (quasi-E<sub>F</sub>).<sup>S7-S9</sup> Zhang *et al.* have been reported the quasi-E<sub>F</sub> of measuring the potential difference between the Ti electrode on the top of TiO<sub>2</sub> and the redox potential of I<sub>3</sub><sup>-</sup>/T u nination.<sup>S8,S9</sup> The minimum of quasi-E<sub>F</sub> is higher than the redox potential of I<sub>3</sub><sup>-</sup>/T at short-circuit (without applying voltage) by he quasi-E<sub>F</sub> will be raised by applying voltage to measure the *J-V* curves of DSSCs.<sup>S8,S9</sup> While the measured photocurrent is er open-circuit condition), the potential difference of quasi-E<sub>F</sub> and redox potential of I<sub>3</sub><sup>-</sup>/T is open-circuit voltage.<sup>S8,S9</sup> Thus, u nination, the E<sub>F</sub> of Pt is much higher than the redox potential of I<sub>3</sub><sup>-</sup>/T.

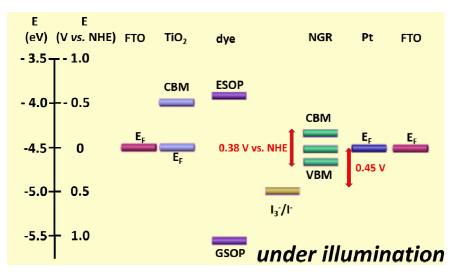


re S5. Schematics of the energy diagram of a DSSC with Pt/FTO CE in the dark before E<sub>F</sub> alignment.



re S6. Schematics of the energy diagram of a DSSC with Pt/FTO CEs under light illumination.

Song *et al.* have reported that the work function of graphene can be determined by the work function of the contact material trates.<sup>S10</sup> In addition, Rani *et al.* have been reported that the bandgap of NGR is *ca.* 0.38 eV.<sup>S11</sup> Thus, we can estimate that the VIGR should be located between the redox potential of  $I_3^-/I^-$  and the  $E_F$  of Pt/FTO CEs. The energy diagram of a DSSC R/Pt/FTO CE under light illumination is shown in Figure S7. Therefore, in our work, the mechanism of cascade-hole transvere electrolyte and NGR/Pt/FTO CEs is confirmed.



re S7. Schematics of the energy diagram of a DSSC with NGR/Pt/FTO CE under light illumination.

## erence

- Dai, F. R.; Wu, W. J.; Wang, Q. W.; Tian, H.; Wong, W. Y. Heteroleptic ruthenium complexes containing uncom disubstituted-2,2'-bipyridine chromophores for dye-sensitized solar cells. *DaltonTrans.* **2011**, 40, 2314–2323
- Wei, X.; Xie, T.; Xu, D.; Zhao, Q.; Pang, S.; Wang, D. A study of the dynamic properties of photo-induced charge carrie porous TiO<sub>2</sub>/conductive substrate interfaces by the transient photovoltage technique. *Nanotechnology* **19**, 2008, 275707-6.
- Daeneke, T.; Kwon, T. H.; Holmes, A. B.; Duffy, N. W.; Bach, U.; Spiccia, L. High-efficiency dye-sensitized solar cells beene-based electrolytes. *Nat. Chem.* **2011**, 3, 211-215.
- Sim, k.; Sung, S. J.; Jo, H. J.; Jeon, D. H.; Kim, D. H.; Kang, J. K. Electrochemical investigation of high-perform sensitized solar cells based on molybdenum for preparation of counter electrode. *Int. J. Electrochem. Sci.* **2013**, 8, 8272-8281.
- Hagfeldt, A.; Boschloo, G.; Sun, L.; Kloo, L.; Pettersson, H. Dye-sensitized solar cells. Chem. Rev., 2010, 110, 6595-6663.
- De Angelis, F.; Fantacci, S.; Selloni, A. Alignment of the dye's molecular levels with the TiO2 band edges in dye-sensitized so : a DFT–TDDFT study. *Nanotechnology* **2008**, 19, 424002-7.
- Grätzel, M. Photoelectrochemical cells. Nature 2001, 414, 338-344.
- Zhang, S.; Yanagida, M.; Yang, X.; Han, L. Effect of 4-tert-butylpyridine on the quasi-fermi level of dye-sensitized TiO<sub>2</sub> for *Phys. Express* **2011**, 4, 042301-3.
- Zhang, S.; Yang, X; Zhang, K; Chen, H.; Yanagida, M.; Han, L. Effects of 4-tert-butylpyridine on the quasi-Fermi levels of TiC s in the presence of different cations in dye-sensitized solar cells. Phys. *Chem. Chem. Phys.* **2011**, 13, 19310-19313.
- )) Song, S. M.; Park, J. K.; Sul, O. J.; Jin, Cho, B. J. Determination of work function of graphene under a metal electrode an in contact resistance. *Nano lett.* **2012**, 12, 3887-3892.

) Rani, P.; Jindal, V. K. Designing band gap of graphene by B and N dopant atoms. *Rsc Adv.* 2013, 3, 802-812.