

Electronic Supplementary Information

Photodegradation of Imidacloprid insecticide by Ag deposited titanate nanotubes - A study of intermediates and their reaction pathways

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Scheme 1: Procedure of sample preparation for GC-MS analysis

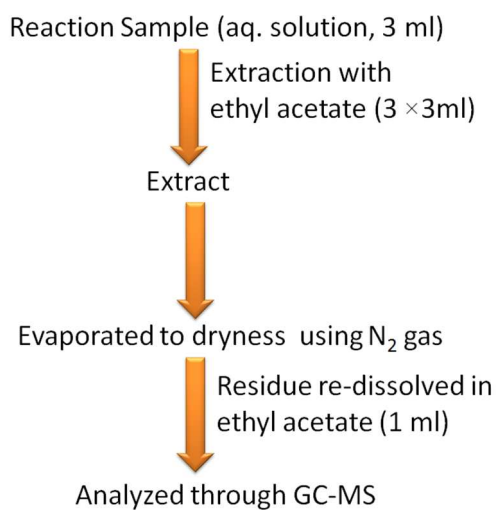


Figure S1: Kinetics for photooxidation of imidacloprid

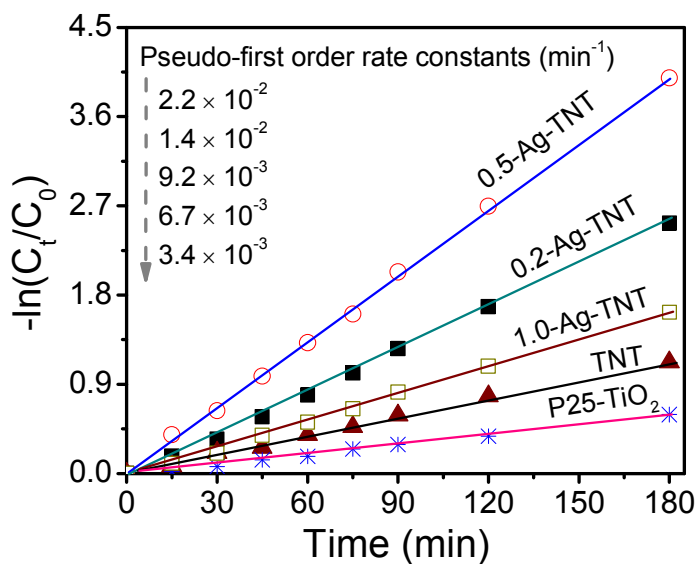


Table 1: Half-life of imidacloprid in presence of studied photocatalysts

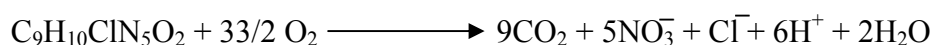
S. No	Catalyst	Rate Constant (k, min ⁻¹)	Half life = 0.693/k (min)
1	P25-TiO ₂	3.4×10^{-3}	204
2	TNT	6.7×10^{-3}	103
3	0.2-Ag-TNT	1.4×10^{-2}	50
4	0.5-Ag-TNT	2.2×10^{-2}	32
5	1.0-Ag-TNT	9.2×10^{-3}	75

Table 2: Calculation for mass balance of CO₂ formation by photooxidation of imidacloprid using bare titania catalysts

Initial concentration of Imidacloprid applied = 50 ppm or 0.2 mM (exact = 0.198 mM)

Amount of Imidacloprid in 5 ml = 1.0 (μmol)

Complete mineralization proceeds according as given below:



Means, 9 molecule of CO₂ produced by =1 molecules of imidacloprid

Therefore, highest amount of CO₂ could be produced = 1.0×9

$$= 9.0 (\mu \text{ mol})$$

S. No	Catalyst	Amount of CO ₂ produced (ppm) (A)	Amount of CO ₂ produced in 5ml $\frac{A}{1000 \times 1000 \times 44} \times 5 (\mu \text{ mol})$ (B)	Percentage of CO ₂ formed $\frac{(B)}{4.5} \times 100$
1	P25-TiO ₂	5.8	0.66	7.3
2	TNT	12.3	1.4	15.5
3	0.2-Ag-TNT	22.4	2.5	28.3
4	0.5-Ag-TNT	29.6	3.5	37.4
5	1.0-Ag-TNT	18.2	2.0	23.0

Table 3: Rate of CO₂ formation during photooxidation of imidacloprid by studied photocatalysts

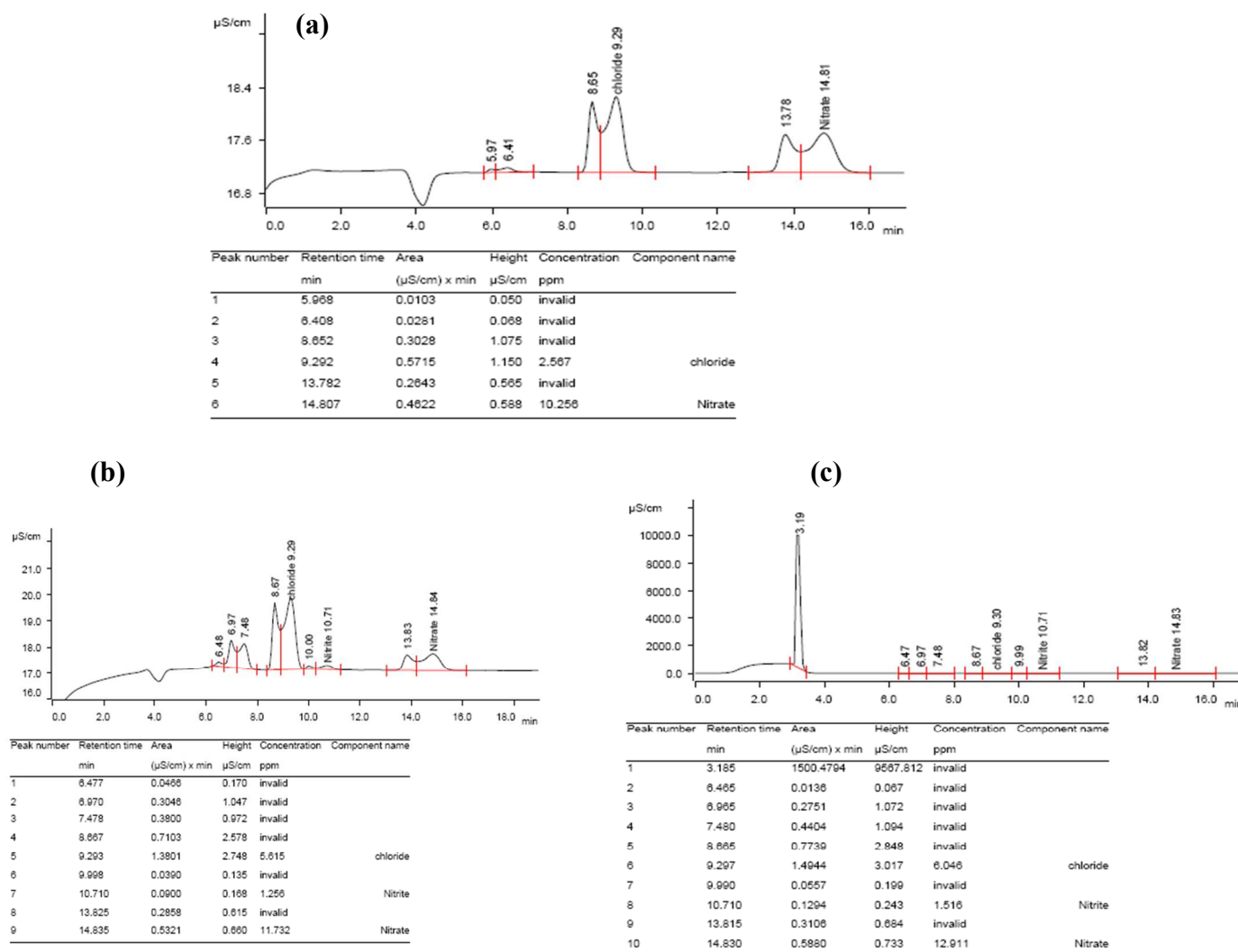
Balanced chemical equation expected after complete mineralization of imidacloprid

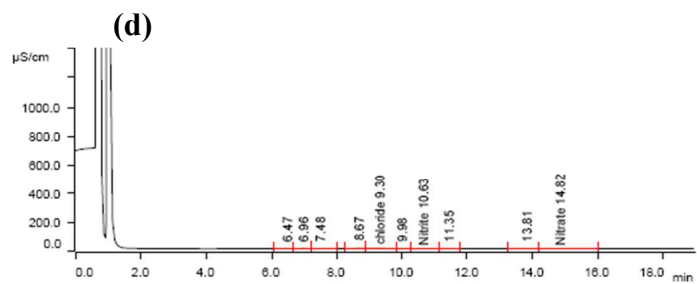


Rate of CO₂ formation (**k**₁) = (**1/9**) rate of degradation of imidacloprid (**k**)

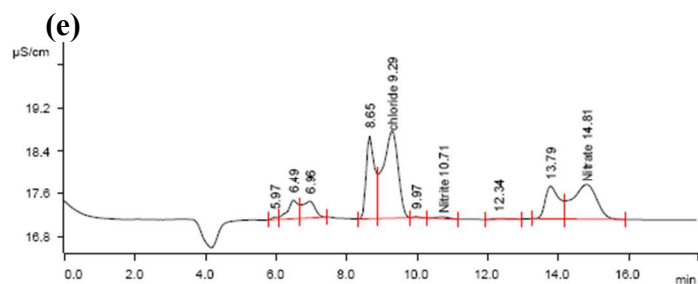
S. No	Catalyst	Rate Constant (k, min ⁻¹)	Rate of CO ₂ formation (k ₁ , min ⁻¹)
1	P25-TiO ₂	3.4×10^{-3}	3.7×10^{-4}
2	TNT	6.7×10^{-3}	7.4×10^{-4}
3	0.2-Ag-TNT	1.4×10^{-2}	1.0×10^{-3}
4	0.5-Ag-TNT	2.2×10^{-2}	1.6×10^{-3}
5	1.0-Ag-TNT	9.2×10^{-3}	2.4×10^{-3}

Figure S2: Ion Chromatograms for (a) Blank water, photooxidation of imidacloprid using 0.5 wt% Ag-photodeposited sodium titanate nanotubes after (b) 60 min, (c) 120 min, (d) 180 min and (e) after 180 min of photoreaction in presence of P25-TiO₂ catalyst.





Peak number	Retention time min	Area ($\mu\text{S/cm}$) x min	Height $\mu\text{S/cm}$	Concentration ppm	Component name
1	6.467	0.1616	0.671	invalid	
2	6.963	0.6256	1.967	invalid	
3	7.482	0.5121	1.385	invalid	
4	8.668	0.8945	3.213	invalid	
5	9.300	1.6779	3.413	6.737	chloride
6	9.977	0.0134	0.067	invalid	
7	10.627	0.0489	0.112	0.985	Nitrite
8	11.345	0.0103	0.028	invalid	
9	13.807	0.4339	0.963	invalid	
10	14.822	0.8277	1.041	17.972	Nitrate



Peak number	Retention time min	Area ($\mu\text{S/cm}$) x min	Height $\mu\text{S/cm}$	Concentration ppm	Component name
1	5.972	0.0083	0.047	invalid	
2	6.492	0.1185	0.344	invalid	
3	6.960	0.1290	0.308	invalid	
4	8.652	0.4322	1.549	invalid	
5	9.292	0.8041	1.641	3.444	chloride
6	9.968	0.0045	0.022	invalid	
7	10.705	0.0132	0.031	0.749	Nitrite
8	12.343	0.0072	0.014	invalid	
9	13.785	0.2798	0.609	invalid	
10	14.808	0.5030	0.643	11.116	Nitrate