## Gum arabic and $Fe^{2+}$ synergistically improve the heat and acid stability of norbixin at pH 3.0-5.0

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## **Supporting information**

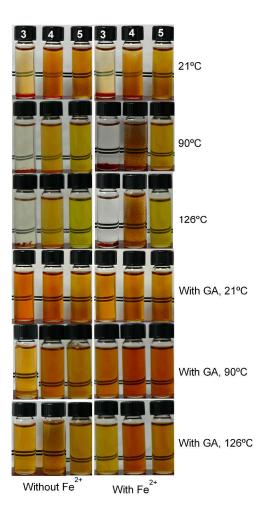


Figure S1. Photographs of dispersions with norbixin, gum arabic and  $Fe^{2+}$  at pH 3.0, 4.0, and 5.0, before and after heating at 90 or 126 °C for 30 min.

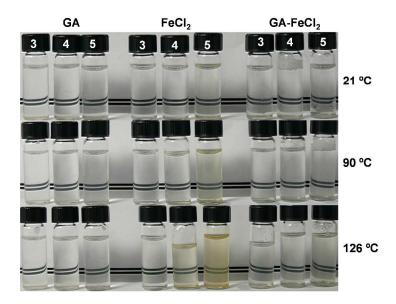


Figure S2. Photographs of 2% w/v GA, 0.02% w/v FeCl<sub>2</sub>, and the mixture with 2% w/v GA and 0.02% w/v FeCl<sub>2</sub> at pH 3.0, 4.0 and 5.0, before and after heating at 90 and 126 °C for 30 min. Supplemental.

Table S1. Changes of absorbance at 460 nm ( $A_{460}$ ) of norbixin and norbixin-gum arabic (GA) dispersions with and without Fe<sup>2+</sup>at pH 3.0-5.0, before and after heating at 90 or 126°C for 30 min.<sup> $\zeta$ </sup>

Heating temperature	рН	$A_{460}$	
		Norbixin	Norbixin-GA
None	3.0	N/D*	0.152±0.041 <sup>ab</sup>
	4.0	N/D	$0.161\pm0.032^{abc}$
	5.0	$0.218\pm0.046^{cd}$	$0.215 \pm 0.001^{cd}$
90°C	3.0	N/D	N/D
	4.0	N/D	$0.128\pm0.02^{cd}$
	5.0	$0.107\pm0.018^{a}$	$0.222 \pm 0.029^d$
126°C	3.0	N/D	N/D
	4.0	N/D	N/D
	5.0	$0.106\pm0.059^{a}$	$0.232 \pm 0.012^d$
None, with Fe <sup>2+</sup>	3.0	N/D	$0.147 \pm 0.001^{ab}$
	4.0	N/D	$0.135 \pm 0.022^{cd}$
	5.0	$0.226 \pm 0.018^d$	$0.243 \pm 0.002^d$
90°C, with Fe <sup>2+</sup>	3.0	N/D	$0.132 \pm 0.008^{ab}$
	4.0	N/D	$0.228 \pm 0.007^d$
	5.0	$0.139\pm0.005^{ab}$	$0.251 \pm 0.038^d$
126°C, with Fe <sup>2+</sup>	3.0	N/D	$0.121\pm0.050^a$
	4.0	N/D	$0.189 \pm 0.032^{bcd}$
	5.0	$0.144\pm0.002^{a}$	$0.235 \pm 0.076^d$

 $<sup>^{\</sup>zeta}$ Numbers are mean  $\pm$  standard deviation (n = 3). Different superscript letters represent significant differences in the mean (p < 0.05).

<sup>\*</sup>N/D: not determined due to precipitation.