

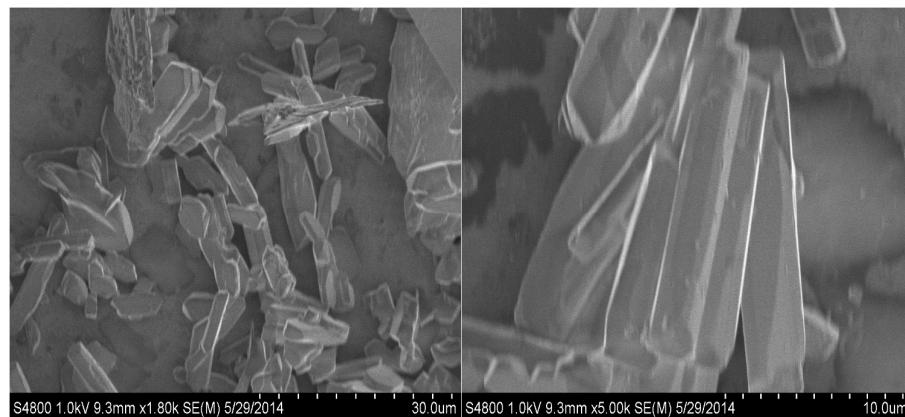
# Supporting Information

## **Eu(III)-functionalized MIL-124 as fluorescent probe for highly selectively sensing ions and organic small molecules especially for Fe(III) and Fe(II)**

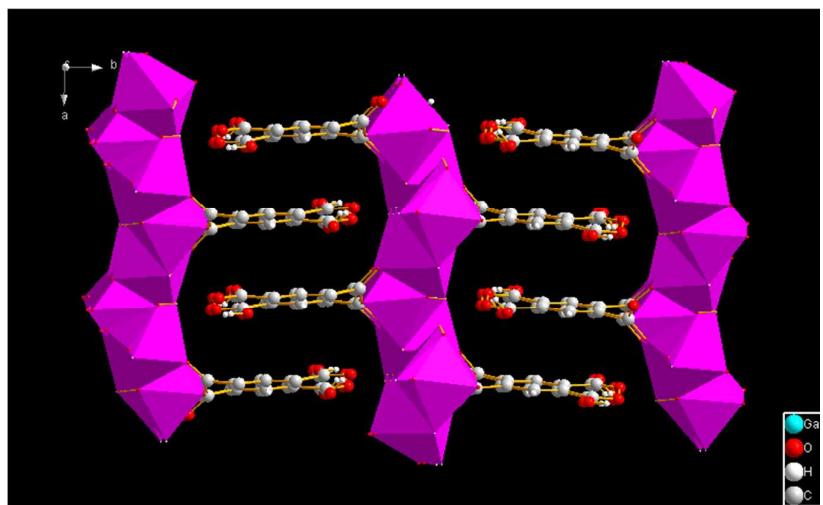
Xiao-Yu Xu, Bing Yan\*

Department of Chemistry, Tongji University, Shanghai 200092, China

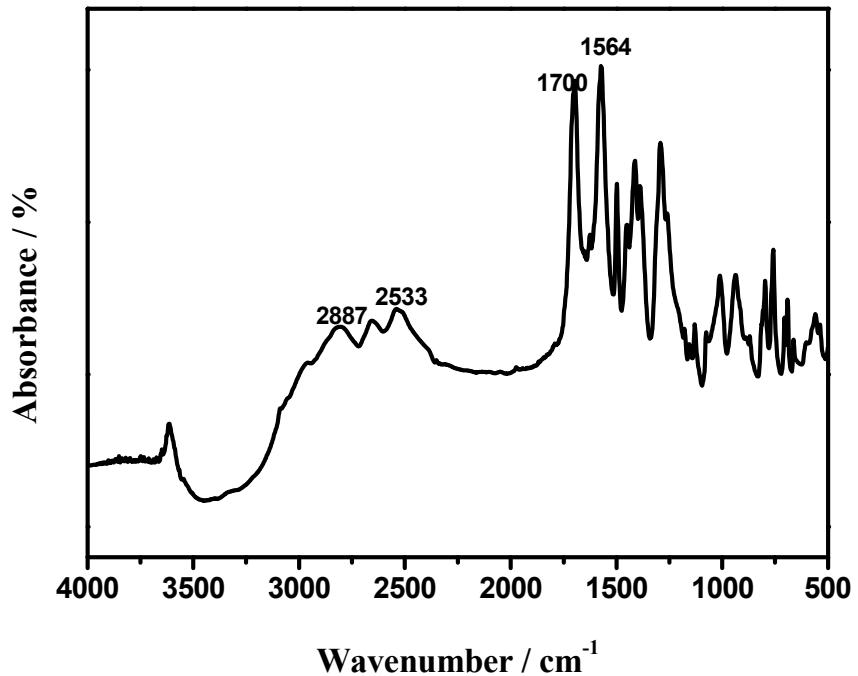
Corresponding author: Prof. Dr. Bing Yan, Email: [byan@tongji.edu.cn](mailto:byan@tongji.edu.cn)



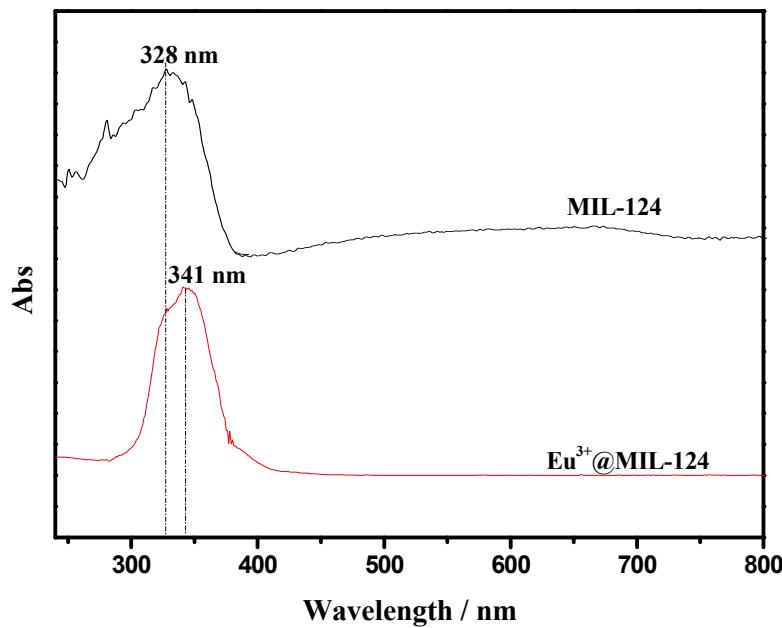
**Figure S1** SEM image of MIL-124.



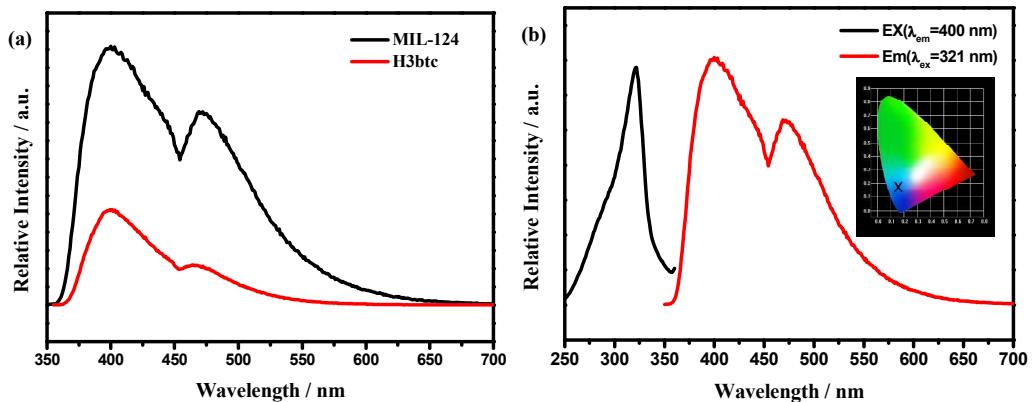
**Figure S2** Representation of the structure of MIL-124 viewed along c axis.



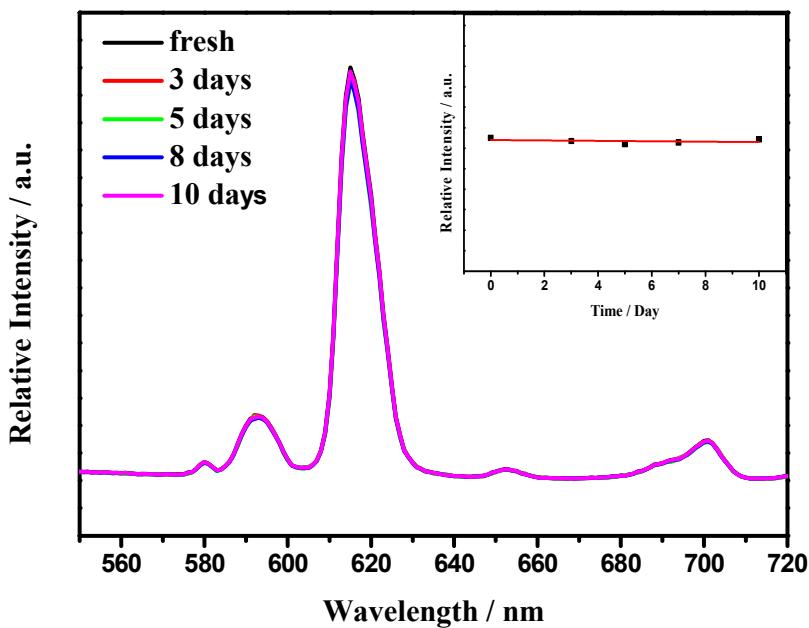
**Figure S3** FTIR spectra of MIL-124.



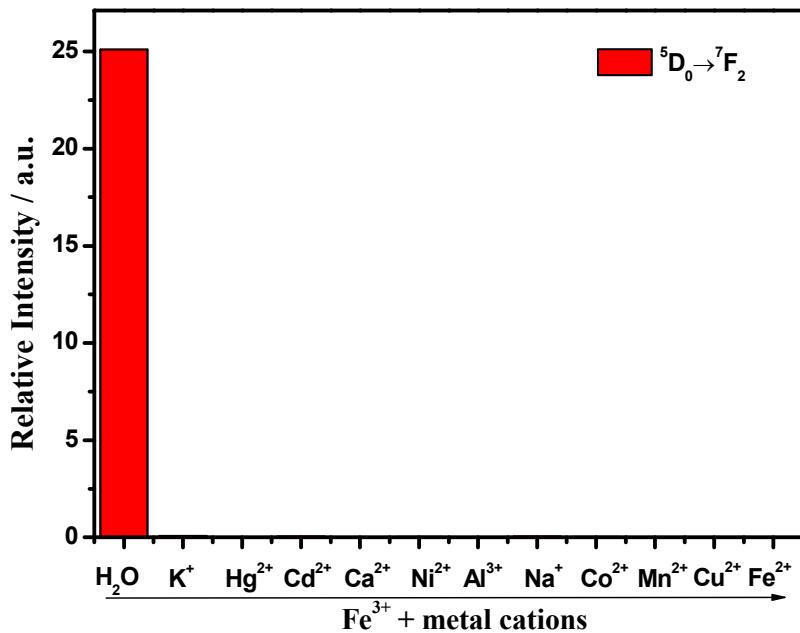
**Figure S4** UV–vis absorption spectra of suspended MIL-121 and  $\text{Eu}^{3+}$ @MIL-121 in  $\text{H}_2\text{O}$  solution.



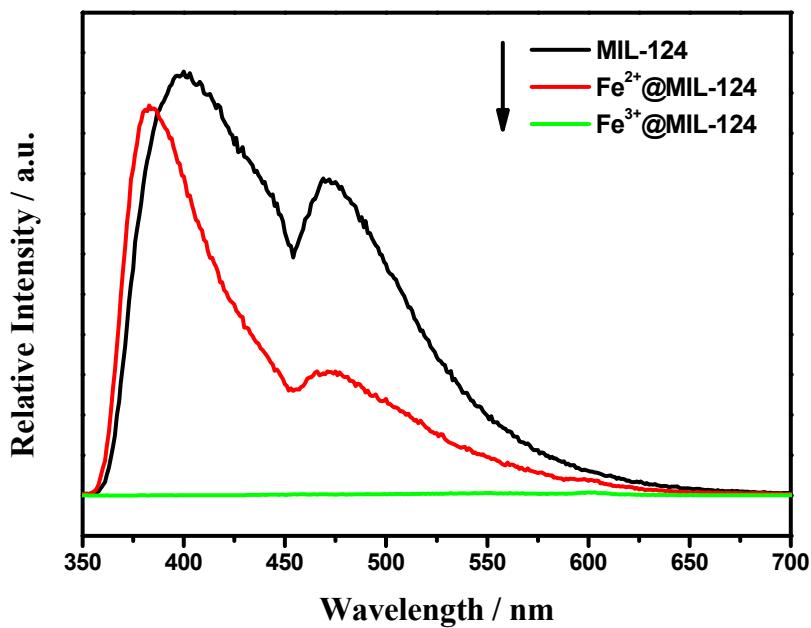
**Figure S5** The fluorescence emission spectra of  $\text{H}_3\text{btc}$  and MIL-124 (a) and the excitation and emission spectra of MIL-124 (b).



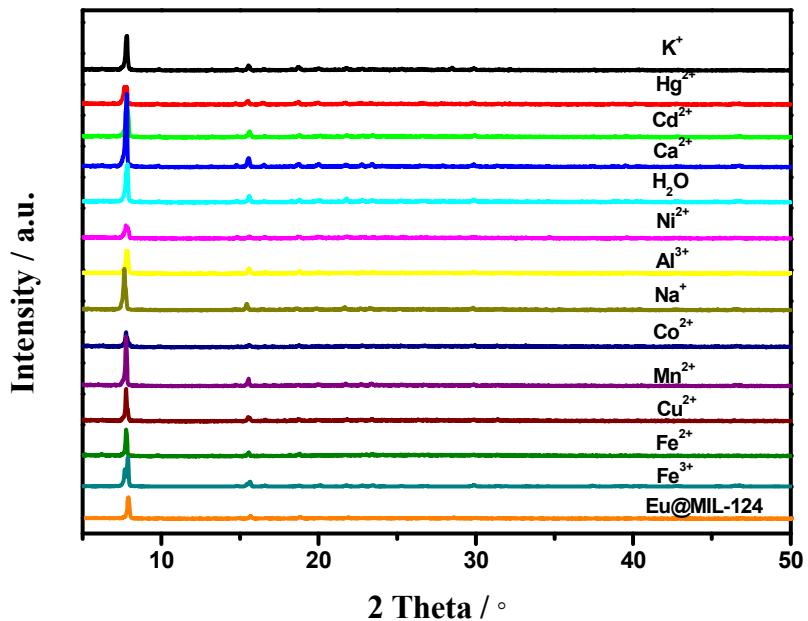
**Figure S6** Day-to-day fluorescence stability of  $\text{Eu}^{3+}$ @MIL-124 under excitation at 298 nm.



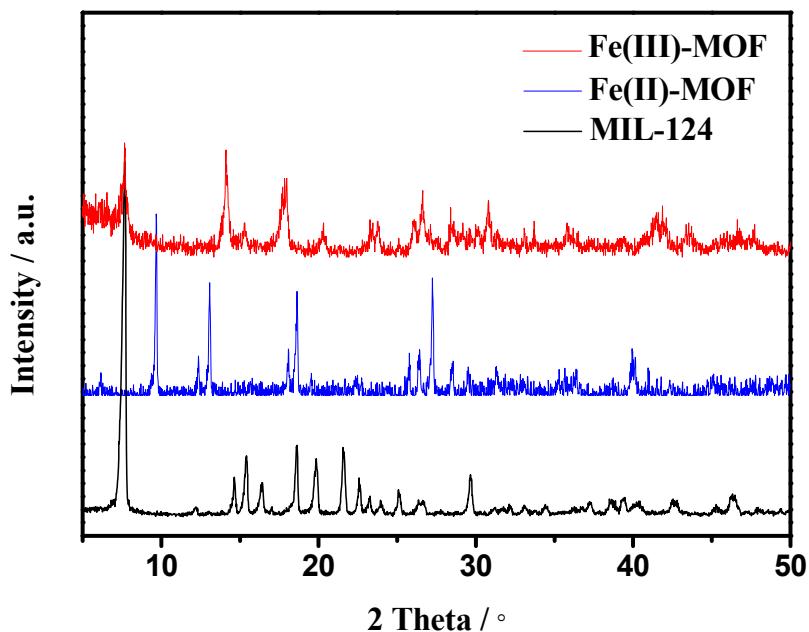
**Figure S7** Luminescence intensity of  $\text{Eu}^{3+}$ @MIL-124 upon the addition of  $\text{Fe}^{3+}$  in the presence of background of metal cations in aqueous solution ( $\lambda_{\text{ex}} = 298$  nm;  $\lambda_{\text{em}} = 615$  nm).



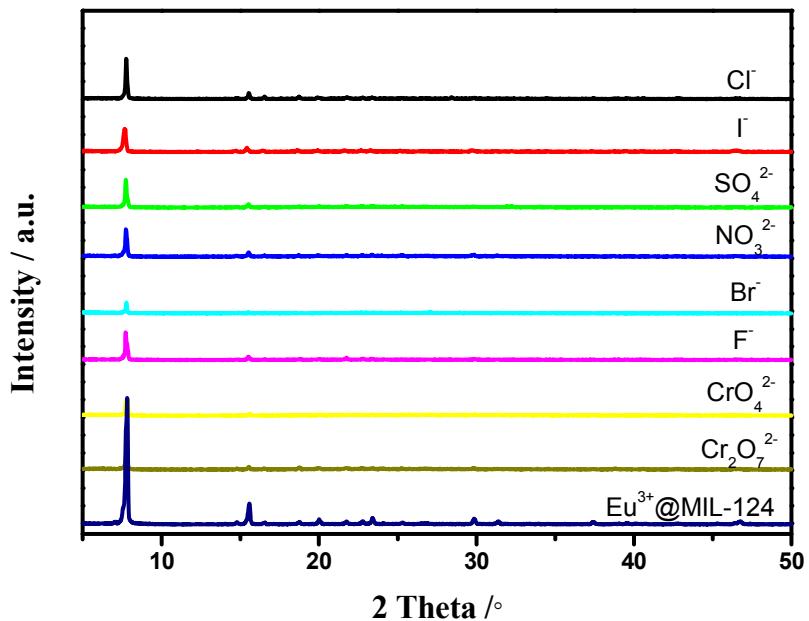
**Figure S8** Responses of the fluorescence of pure MOF towards aqueous solution of  $\text{Fe}^{3+}$  and  $\text{Fe}^{2+}$ .



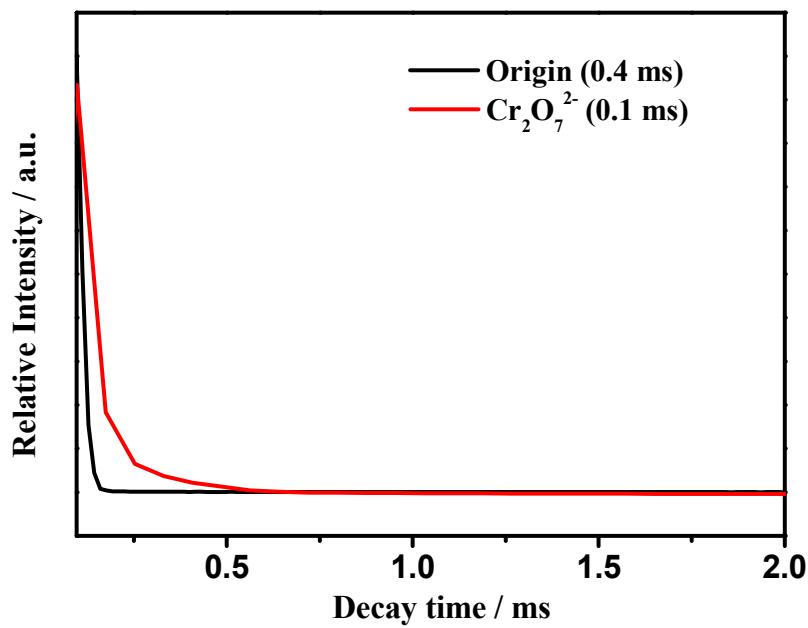
**Figure S9** PXRD patterns of the  $\text{Eu}^{3+}$ @MIL-124 after immersing in aqueous solution with various metal ions.



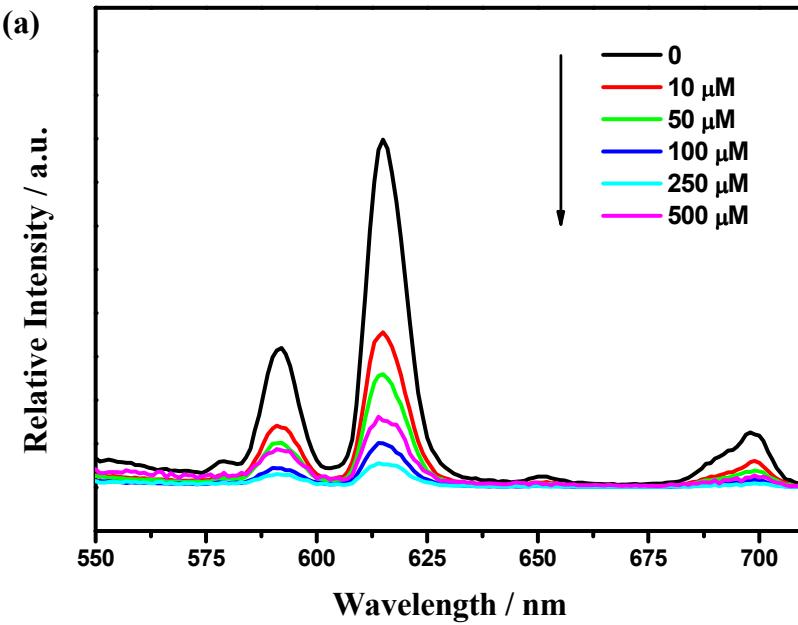
**Figure S10**PXRD patterns of synthesized Fe(III)-MOF and Fe(II)-MOF.

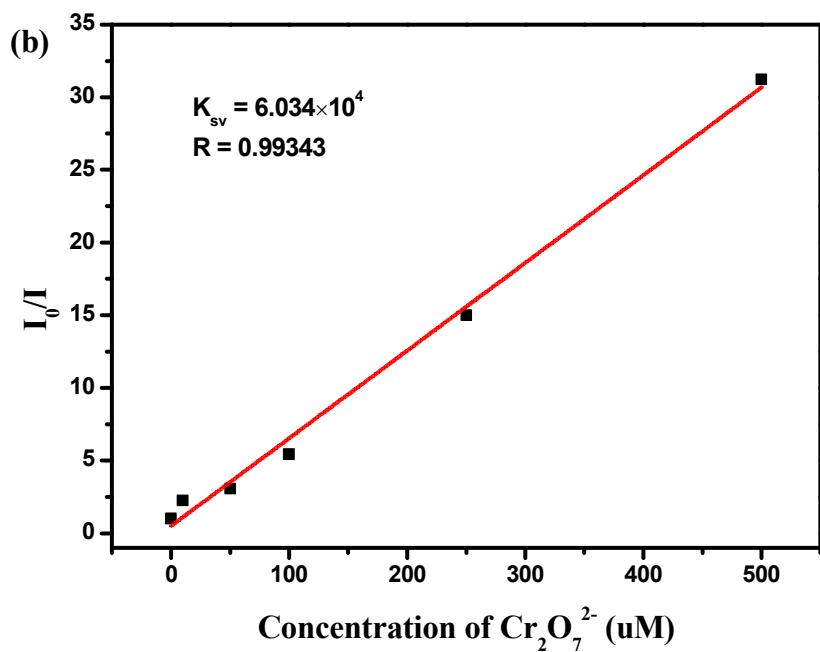


**Figure S11** PXRD patterns of the Eu<sup>3+</sup>@MIL-124 after immersing in aqueous solution with various anions.

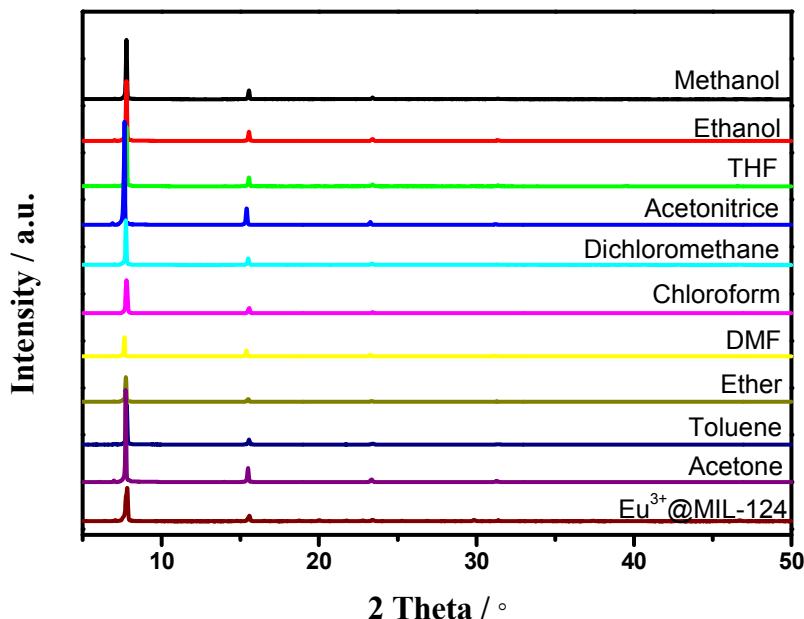


**Figure S12** Fluorescence lifetime of  $\text{Eu}^{3+}$ @MIL-124 in the absence (red) and presence (black) of  $\text{Cr}_2\text{O}_7^{2-}$  in aqueous solution.

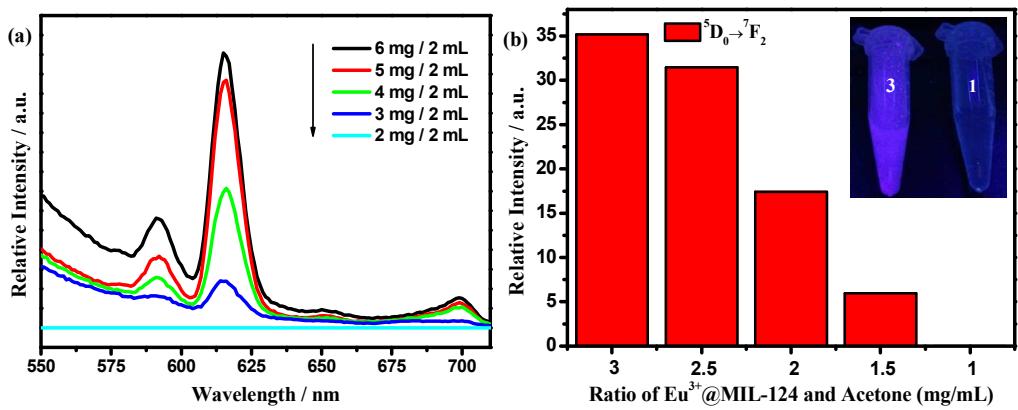




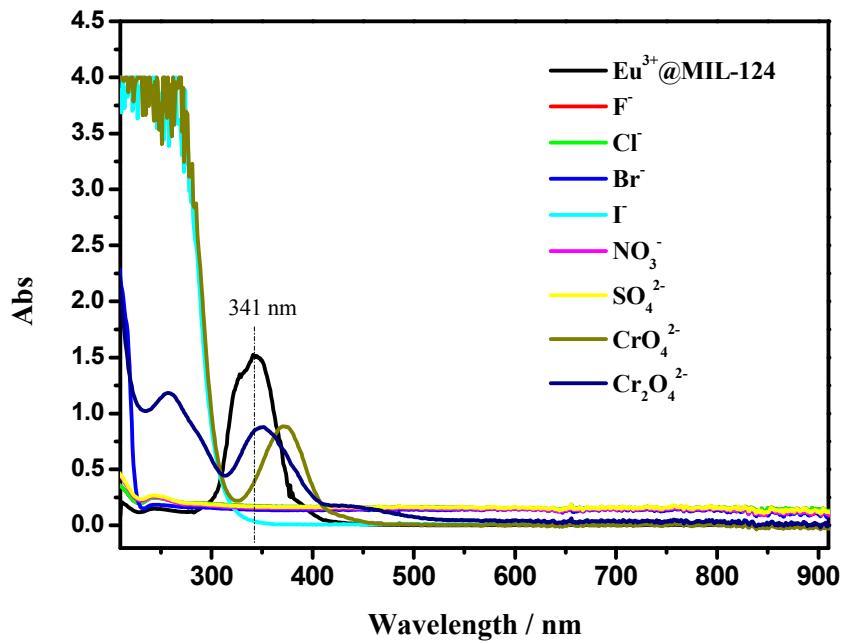
**Figure S13** Emission spectra (a) and Ksv curve (b) of  $\text{Eu}^{3+}$ @MIL-124 in aqueous solutions in the presence of various concentrations of  $\text{Cr}_2\text{O}_7^{2-}$  under excitation at 298 nm.



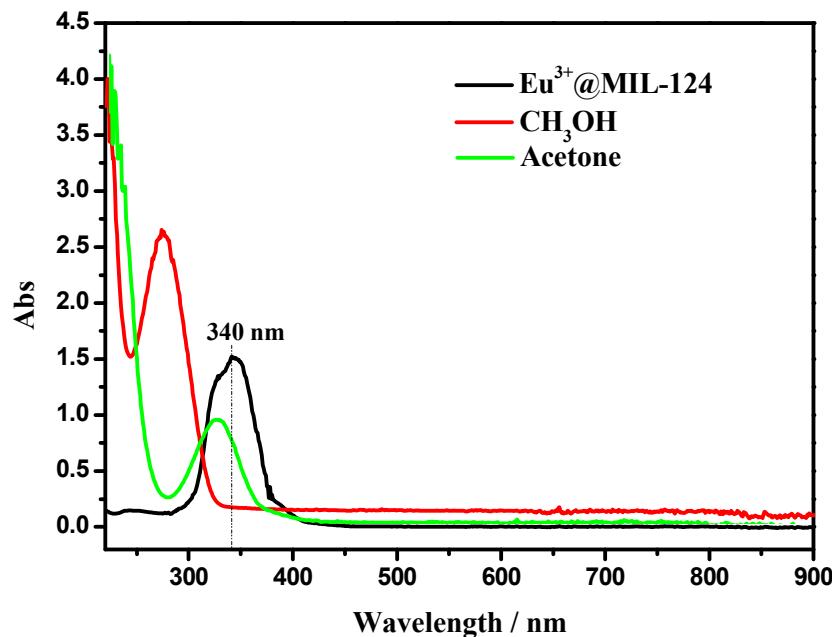
**Figure S14** PXRD patterns of the  $\text{Eu}^{3+}$ @MIL-124 after immersing in different organic small molecules.



**Figure S15** PL spectra (a) and luminescence intensity of the  $^5D_0 \rightarrow ^7F_2$  of Eu<sup>3+</sup>@MIL-124(b) in the presence of various ratios of Eu<sup>3+</sup>@MIL-124 and acetone under excitation at 298 nm.



**Figure S16** Liquid UV-vis absorption of different anions.



**Figure S17** UV-visabsorption spectra of  $\text{Eu}^{3+}$ @MIL-124 and target solvents ( $\text{CH}_3\text{OH}$  and acetone).

Table S1 The detailed ICP studies of MIL-124,  $\text{Eu}^{3+}$ @MIL-124 and target metal ions ( $\text{K}^+$ ,  $\text{Co}^{2+}$ ,  $\text{Fe}^{2+}$  and  $\text{Fe}^{3+}$ )

Samples	$\text{Ga}^{3+}$ (ppm)	$\text{Eu}^{3+}$ (ppm)	$\text{M}^{\text{n}+}$ (ppm)	$\text{Ga}^{3+}:\text{Eu}^{3+}:\text{M}^{\text{n}+}$
MIL-124	35.02	--	--	--
$\text{Eu}^{3+}$ @MIL-124 (1a)	33.24	0.56	--	--
$\text{K}^+$ @1a	32.45	0.60	0.23	141:3:1
$\text{Co}^{2+}$ @1a	30.72	0.23	0.84	134:1:4
$\text{Fe}^{2+}$ @1a	31.56	0.07	0.58	451:1:8
$\text{Fe}^{3+}$ @1a	13.41	0.04	3.84	335:1:96