

Supporting Information:

Light-induced Efficient Molecular Oxygen Activation on Cu(II)

Grafted TiO₂/graphene Photocatalyst for Phenol Degradation

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(9 Figures, 1 Table, 6 Pages)

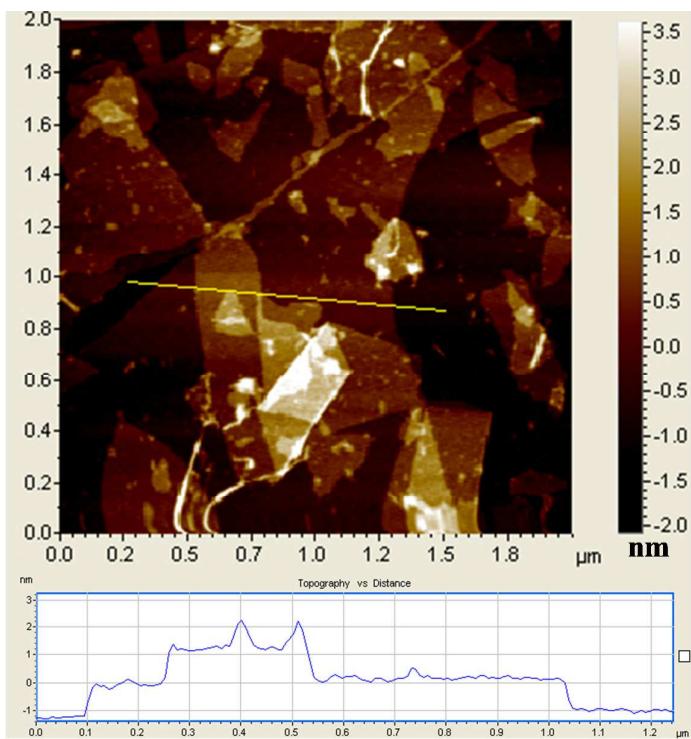


Figure S1. Atomic force microscopy images of graphene oxide sheets.

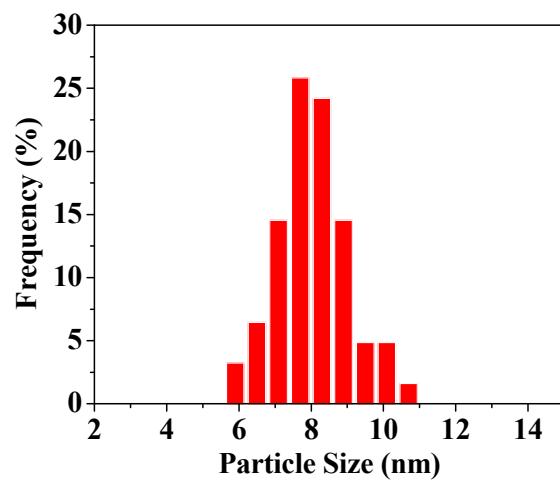


Figure S2. The statistical data of the TiO_2 NPs size in the $\text{TiO}_2/\text{RGO}/\text{Cu(II)}$ composites.

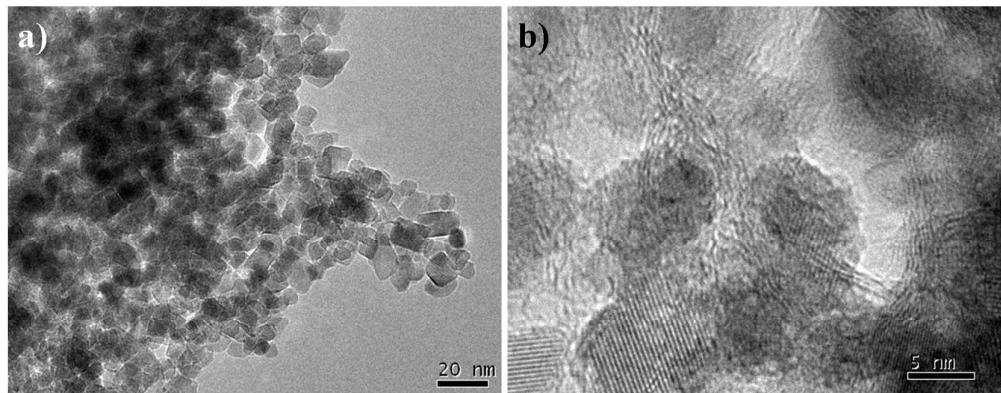


Figure S3. (a) TEM and (b) High-resolution TEM images of the TiO₂/RGO composites.

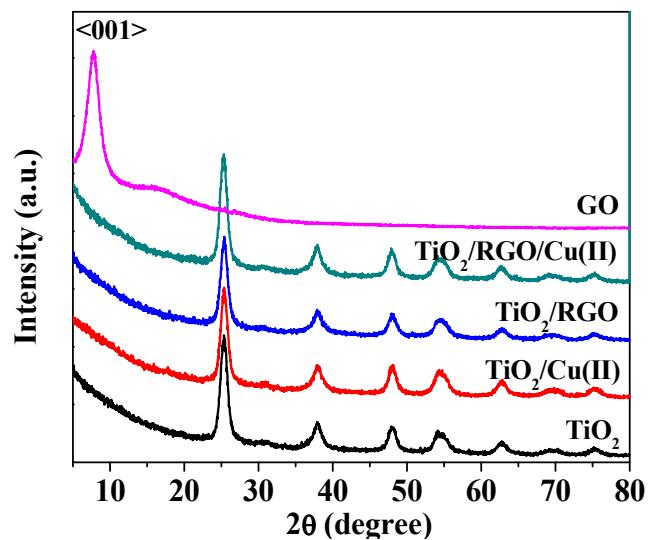


Figure S4. XRD patterns of GO, TiO₂, TiO₂/Cu(II), TiO₂/RGO and TiO₂/RGO/Cu(II) composites.

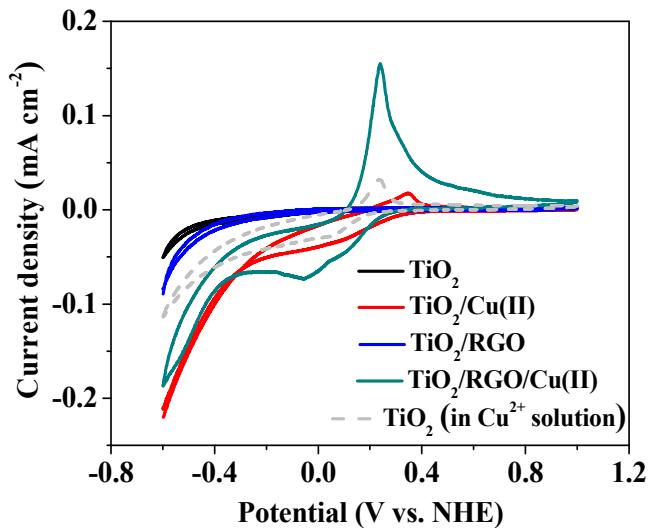


Figure S5. Cyclic voltammograms of the prepared TiO_2 , $\text{TiO}_2/\text{Cu(II)}$, TiO_2/RGO and $\text{TiO}_2/\text{RGO/Cu(II)}$ electrodes in NaClO_4 electrolyte solution. The gray dashed line represents the cyclic voltammograms of the bare TiO_2 electrode in 0.2 mM CuCl_2 solution.

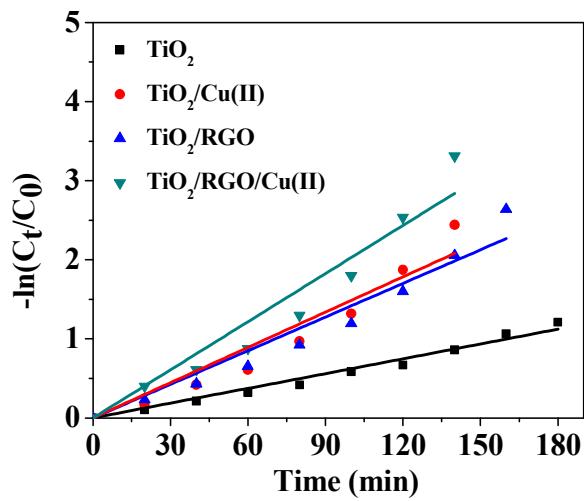


Figure S6. The calculation of phenol degradation rate constants for the TiO_2 , $\text{TiO}_2/\text{Cu(II)}$, TiO_2/RGO and $\text{TiO}_2/\text{RGO/Cu(II)}$ composites with R square of 0.99, 0.98, 0.98 and 0.97, respectively.

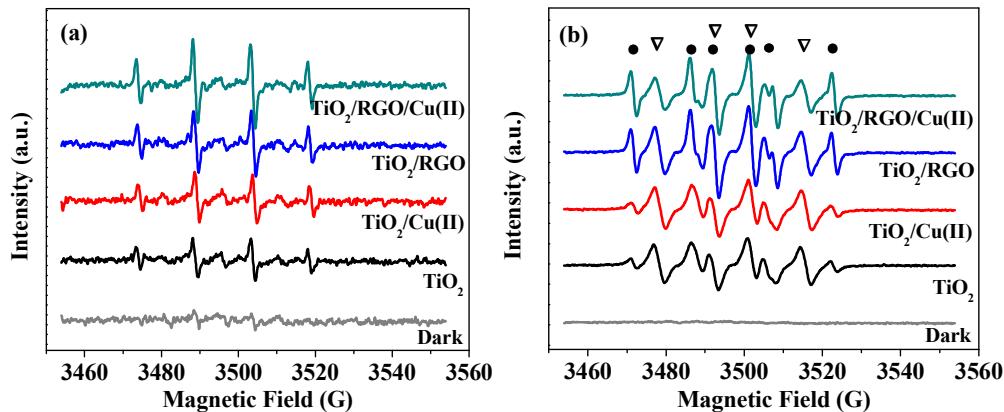


Figure S7. Electron spin resonance spectra of radical adducts trapped by DMPO in TiO₂, TiO₂/Cu(II), TiO₂/RGO and TiO₂/RGO/Cu(II) dispersions after 10 min UV irradiation: (a) DMPO-·OH formed in aqueous dispersions and (b) DMPO-·O₂⁻ formed in methanol suspensions (DMPO-·O₂⁻ and methyl radical were marked with ▽ and ● symbols, respectively).

Table S1. BET surface area of TiO₂, TiO₂/Cu(II), TiO₂/RGO and TiO₂/RGO/Cu(II).

Sample	TiO ₂	TiO ₂ /Cu(II)	TiO ₂ /RGO	TiO ₂ /RGO/Cu(II)
BET surface area m ² /g	152.3	145.2	155.7	156.8

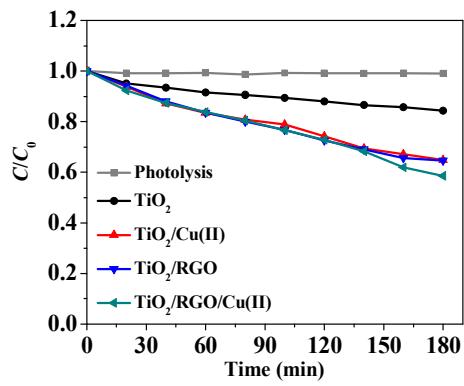


Figure S8. Phenol degradation curves under visible light irradiation ($\lambda > 400$ nm) over TiO_2 , $\text{TiO}_2/\text{Cu(II)}$, TiO_2/RGO and $\text{TiO}_2/\text{RGO/Cu(II)}$ composites.

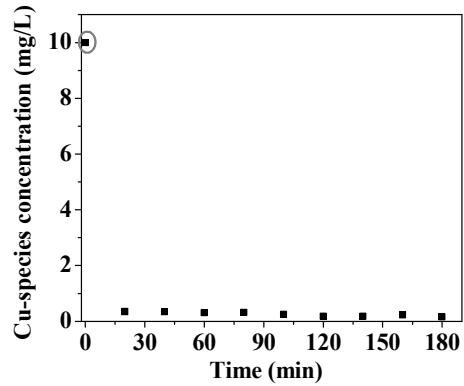


Figure S9. Change of Cu-species concentration in the $\text{TiO}_2/\text{RGO/Cu(II)}$ composites suspension during light irradiation. The Cu-species concentration at $t = 0$ min was calculated to be 10 mg/L by assuming that Cu(II) in the composites was completely dissolved.