Chemoselective Intramolecular Functionalization of Methyl Groups in non-Constrained Molecules Promoted by *N*-iodosulfonamides.

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General Information and Instrumentation. Melting points were determined with a hot-stage apparatus and are uncorrected. Optical rotations were measured at the sodium line at ambient temperature in CHCl₃ solutions. IR spectra were measured as thin films on a NaCl plate. NMR spectra were determined at 400 MHz for ¹H and 100.6 MHz for ¹³C in CDCl₃ unless otherwise stated, in the presence of TMS as internal standard. Mass spectra were determined at ESI+ unless otherwise stated. TLC was performed by using Merck silica gel coated aluminium 60 F₂₅₄ plates; detection with UV or dipping into a solution of KMnO₄ (1.5 g in 400 mL H₂O, 5 g NaHCO₃) or a solution of nihidrin in EtOH (0.2%) and further heating until development of color. Merck silica gel 60 PF (0.063 - 0.2 mm) was used for column chromatography. Flash chromatography refers to a column chromatography accelerated by air pressure. Ascentis®Si columns (25 cm x 10 mm, 5µm or 15 cm x 4.6 mm, 3µm) were used for HPLC purifications. Commercially available reagents and solvents were analytical grade or were purified by standard procedures prior to use. All reactions involving air- or moisture-sensitive materials were carried out under argon atmosphere. The relative emission of light sources were measured with spectrophotometer [avantes AVS-USB2000 with grating #3(350-850)].UV-visible spectra were measured with spectrophotometer (Varian CARY-1E). The specifications for the light source were the following: [A] for tungsten lamps Philips PAR 38EC, Flood 30°, 80 W; [A+f] for tungsten lamps Philips PAR 38EC, Flood 30°, 80 W, filtered by 1 cm pathlength of aqueous solution of K₂CrO₄ (0.27 g/L) and Na₂CO₃ (1 g/L) into pyrex glasses (λ >445 nm); [B] for green LED lamps Lexman GU10, 0.9W; [C] for white LED lamps Lexman GU10, 4W, 285 Lumens.

General procedures:

Functionalization of methyl group conducted by slow addition of PhI(OAc)₂; General procedure for SHAT (GP1): To a solution of *N*-sulfonamidate (1 mmol) in DCE (10 mL), in a sealing tube equipped with a stirring bar, were added iodine (5 mmol, *saturated solution*) and PhI(OAc)₂ (0.25 mmol) and sealed off. The reaction mixture was stirred and irradiated with two tungsten filament lamps at 15-20 cm allowing to reach 65-75 °C, and portions of PhI(OAc)₂ (0.25 mmol) were added each 15 min until the reaction was completed, monitored by TLC. The reaction mixture was further stirred for 1 hour and then poured into an aqueous solution of Na₂S₂O₃ (20%) and extracted with DCM. The organic phase was dried with Na₂SO₄ and concentrated. The residue was analyzed by ¹H NMR and then purified by flash chromatography (hexanes/EtOAc). Changes in amounts of reagents and use of other additives are specified in text or tables. All reactions were performed with 100-200 mg of *N*-sulfonamidates.

Functionalization of methyl group conducted by slow addition of PhI(OAc)₂; General procedure for SHAT (GP1)_{CSA}: Same as GP1 but iodine (0.6 mmol) and camphorsulfonic acid CSA (1 mmol) were added instead of the large amount of iodine (*saturated solution*).

Functionalization of methyl group conducted by slow addition of $PhI(OAc)_2$; General procedure for SHAT (GP1)_{Zn(OTf)2}: Same as GP1 but iodine (0.5 mmol) and Zn(OTf)₂ (0.6 mmol) were added instead of the large amount of iodine (*saturated solution*).

Functionalization of methyl group conducted by slow addition of iodine; General procedure for MHAT (GP2): To a solution of *N*-sulfonamidate (1 mmol) in DCE (33 mL) in a round bottomed flask, equipped with a stirring bar and a septum stopper, were added PhI(OAc)₂ (4-6 mmol) and NaHCO₃ (100%) (in most cases). The reaction mixture was stirred and irradiated with two tungsten filament lamps at 35 cm at rt (refrigerating with a fan) while a solution of iodine (0.6-1.5 mmol) in DCE (0.15 M) was added dropwise within 3-6 h with a syringe pump until consumption of the starting material, monitored by TLC. The reaction mixture was further stirred for 1-18 hours depending on the ability to afford the lactone product. Longer time of further stirring is needed if higher amount of the carboxylic acid intermediate is detected to promote its cyclization. The reaction mixture was then poured into an aqueous solution of Na₂S₂O₃ (20%) and extracted with DCM. The organic phase was dried with Na₂SO₄ and concentrated. The residue was analyzed by ¹H NMR and then purified by flash chromatography (hexanes/EtOAc). Changes in amounts of reagents and use of other additives are stated in text or tables.

Functionalization of methyl group conducted by simultaneously all-mixed protocol; General procedure for SHAT or MHAT (GP3): To a solution of *N*-sulfonamidate (1 mmol) in DCE (0.1-0.03M) in a sealing tube equipped with a stirring bar were added iodine (0.6-1.5 mmol), PhI(OAc)₂ (1.25-5 mmol) and, in some cases, additives such NaHCO₃ (100%), Na₂CO₃ (100-200%) or CSA (0.5-1 mmol) and sealed off. The reaction mixture was stirred and irradiated with two tungsten filament lamps at 35 cm at rt (refrigerating with a fan) until the reaction was completed, monitored by TLC. The reaction mixture was then poured into an aqueous solution of Na₂S₂O₃ (20%) and extracted with DCM. The organic phase was dried with Na₂SO₄ and concentrated. The residue was analyzed by ¹H NMR and then purified by flash chromatography (hexanes/EtOAc). Changes in amounts of reagents and use of other additives are stated in text or tables.

Functionalization of methyl group conducted by simultaneously all-mixed protocol; General procedure for MHAT (GP3)_{Na2CO3}: To a solution of *N*-sulfonamidate (1 mmol) in DCE or CDCl₃ (0.03M) in a sealing tube equipped with a stirring bar were added PhI(OAc)₂ (6-10 mmol), Na₂CO₃ (100-200%) and stirred for 20-30 min prior to the addition of iodine (1-2 mmol). The reaction mixture was stirred and irradiated at rt (refrigerating with a fan) until the reaction was completed, monitored by TLC. The reaction mixture was then poured into an aqueous solution of Na₂S₂O₃ (20%) and extracted with DCM. The organic phase was dried with Na₂SO₄ and concentrated. The residue was analyzed by ¹H NMR and then purified by flash chromatography (hexanes/EtOAc). Changes in amounts of reagents and source of light are stated in text or tables.

Functionalization of methyl group conducted by previous formation of AcOI in CDCl₃ and monitored by NMR; General procedure for SHAT or MHAT (GP4): A solution of iodine, $PhI(OAc)_2$ and, in some cases, additives such NaHCO₃ or Na₂CO₃ (100-200%) in CDCl₃ (0.03 – 0.2 M) was stirred for 30-45 min in the dark in a sealed tube, until the $PhI(OAc)_2$ was consumed (monitored by ¹H NMR). *N*-sulfonamidate was then added to this reaction mixture, which was vigorously stirred for 1-2 minutes and transferred to a NMR-tube. The evolution of the reaction was monitored by ¹H NMR either in the dark (usually, without taking the

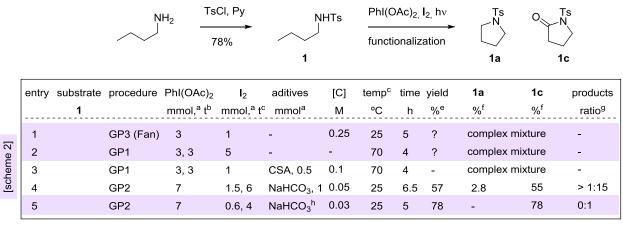
sample out of the NMR machine) or under irradiation (the NMR tube was rotated while it was irradiated outside of the NMR machine).

Note: When the reaction was performed under irradiation, we assumed that the processes promoted by radiation stopped during the measurement of the ¹H NMR (\approx 3 min of darkness), while the processes that could occur in absence of light continued ongoing, introducing an error that did not disturbed drastically the evolution of reactions.

General procedure for *N*-sulfonamidate products (GP5): $CISO_2Ar$ (1.1 mmol) was added to a solution of the amine (1 mmol) in pyridine (1 mL) at 0 °C and stirred at rt for 1-3 h. The reaction mixture was poured into a diluted aqueous solution of HCl (10%) and extracted twice with EtOAc. The organic extracts were washed with saturated aqueous solution of NaHCO₃ and brine, dried over anhydrous NaSO₄ and concentrated under vacuum. Column chromatography (hexanes–EtOAc) of the residue afforded the corresponding products.

Synthesis and Functionalizacion of Tosylamine 1:

Table S1.



^a millimoles per millimol of **1**; ^b Time of addition in hours: Portionwise (0.25 equiv / 15 min); ^c time of continous addition of I_2 disolved in DCE (0.15M) by syringe pump (in hours); ^d 25 °C (reaction mixture refrigerated by a fan) and 65-75 °C (without fan); ^e total isolated yield; ^f isolated yield; ^g calculated by integration in the ¹H NMR of the crude of reaction; ^h 100% w/w. CSA = Camphorsulfonic acid.

NHTs *N*-Butyl-4-methylbenzenesulfonamide (1):¹ According to GP5, *N*-tosylamine 1 was prepared from butan-1-amine in 78% as a crystalline solid: Mp 41.5-43.0 °C (ethanol/*n*hexane); ¹H NMR δ 0.85 (t, *J* = 7.4 Hz, 3H), 1.24-1.33 (m, 2H), 1.40-1.47 (m, 2H), 2.43 (s, 3H), 2.93 (br dd, *J* = 7.0, 12.6 Hz, 2H), 4.47 (br s, 1H), 7.31 (d, *J* = 8.0 Hz, 2H), 7.75 (d, *J* = 8.0 Hz, 2H); ¹³C NMR δ 13.9 (CH₃), 20.1 (CH₂), 21.9 (CH₃), 31.9 (CH₂), 43.3 (CH₂), 127.5 (2 x CH), 130.0 (2 x CH), 137.8 (C), 143.6 (C).

Ts **1-Tosylpyrrolidine (1a):**² Crystalline solid: ¹H NMR δ 1.73-1.77 (m, 4H), 2.43 (s, 3H), 3.22-3.25 (m, 4H), 7.31 (d, J = 8.2 Hz, 2H), 7.72 (d, J = 8.5 Hz, 2H).

O Ts 1-Tosylpyrrolidin-2-one (1c):³ Crystalline solid: ¹H NMR δ2.03-2.11 (m, 2H), 2.41-2.45 (m, 2H), 2.44 (s, 3H), 3.90 (t, *J* = 7.2 Hz, 2H), 7.34 (d, *J* = 8.5 Hz, 2H), 7.93 (d, *J* = 8.5 Hz, 2H); ¹³C NMR δ18.2 (CH₂), 21.6 (CH₃), 32.2 (CH₂), 47.2 (CH₂), 128.1 (2 x CH), 129.6 (2 x CH), 135.2 (C), 145.1 (C), 173.3 (C).

¹ Spectral data are in good agreement with those reported in the literature: (a) Rad, M. N. S.; Khlafi-Nezhad, A.; Asrari, Z.; Behrouz, S.; Amini, Z.; Behrouz, M. *Synthesis* **2009**, 3983; (b) Li, Y.; Zhao, Y.; Zhang, Z.;Xu, Y. *Tetrahedron Lett.* **2010**, *51*, 1434.

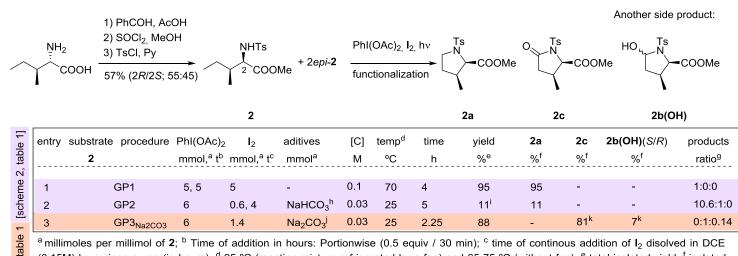
² spectral data are in good agreement with those reported in the literature: Kato, Y.; Yen, D. H.; Fukudome, Y.; Hata, T.; Urabe, H. *Org. Lett.* **2010**, *12*, 4137.

³ Spectral data are in good agreement with those reported in the literature: (a) Guo, J.; Harling, J. D.; Steel, P. G.; Woods, T. M. Org. Biomol. Chem. 2008, 6, 4053; (b) Boal, B. W.; Schammel, A. W.; Garg, N. K. Org. Lett. 2009, 11, 3458; (c) Schammel, A. W.; Boal, B. W.; Zu, L.; Mesganaw, T.; Garg N. Tetrahedron 2010, 66, 4687.

Synthesis and Functionalization of Tosylamine 2:

Table S2.

NHTs



^a millimoles per millimol of **2**; ^b Time of addition in hours: Portionwise (0.5 equiv / 30 min); ^c time of continous addition of **I**₂ disolved in DCE (0.15M) by syringe pump (in hours); ^d 25 °C (reaction mixture refrigerated by a fan) and 65-75 °C (without fan); ^e total isolated yield; ^f isolated yield; ^g [**2a/2c/2b(OH)**] calculated by integration in the ¹H NMR of the crude of reaction; ^h 100% w/w; ⁱ startingt material **2** was recovered (73%); ^j 200% w/w; ^k **2c** and **2b(OH)** were isolated together by column chromatography and independently isolated by further HPLC.

(2R,3S) Methyl 3-methyl-2-(4-methylphenylsulfonamido)pentanoate (2): A solution
CO₂Me of L-isoleucine (5 g, 37.9 mmol) in glacial acetic acid (25 mL) was treated with benzaldehyde (400 mg), and the mixture heated at 100 °C in a nitrogen atmosphere for 3

h. The mixture was cooled to 50 °C and filtered through Celite. The solution was concentrated under vacuum to a volume of 5 mL. 2-Propanol (16 mL) was then added and the mixture stirred for 1 h. The mixture was filtered and washed with more 2-propanol (2x4 mL). The solid was dried under vacuum to afford an epimeric mixture of L-isoleucine and D-alloisoleucine (4.3 g, 86%, ≈1:1, estimated by ¹H NMR).⁴ Part of this epimeric mixture (2.41 g, 20.6 mmol) was then dissolved in dry MeOH (80 mL) and treated at 0 °C with SOCl₂ (2.99 mL, 41.2 mmol). The mixture was refluxed for 4 h and concentrated to give the crude methyl ester as viscous oil. The product was dissolved in pyridine (70 mL), TsCl (7.85 g, 41.2 mmol) was added and the mixture was heated at 40 °C for 24 h. The mixture was concentrated, dissolved in EtOAc and washed with diluted aqueous solution of HCl (10%). The organic extract was washed with saturated aqueous solution of NaHCO₃ and brine, dried over anhydrous NaSO₄ and concentrated under vacuum. Column chromatography (hexanes-EtOAc, 75:25) of the residue afforded pure D-alloisoleucine derivative 2 (165 mg, 0.55 mmol, 2.7%) as a white solid collecting the initial fractions of the column and the epimeric mixture $(2/2epi-2, \approx 1:1)$ (3.95 g, 13.2 mmol, 64%) with equal R_f by TLC analysis which was successively subjected to column chromatography to obtain further small portions of pure 2: Crystalline solid; Mp 75.6-76.5 °C (hexane / Et₂O); $[\alpha]_D$ -20.9 (c 1.0); ¹H NMR δ 0.81 (d, J = 7.0 Hz, 3H), 0.90 (t, J = 7.4 Hz, 3H), 1.23 (dquin, $J = 7.4 \times 4$, 14.3 Hz, 1H), 1.46 (dquin, $J = 7.4 \times 4$, 14.3 Hz, 1H), 1.77 (tqd, J = 4.1, 7.1 x 3, 7.4 x 2 Hz, 1H), 2.41 (s, 3H), 3.44 (s, 3H), 3.90 (dd, J = 4.1, 10.2 Hz, 1H), 5.06 (d, J = 10.2 Hz, 1H), 7.28 (d, J = 10.2 Hz, 1H), 7. 8.1 Hz, 2H), 7.71 (d, J = 8.3 Hz, 2H); ¹³C NMR δ 11.5 (CH₃), 14.3 (CH₃), 21.6 (CH₃), 26.0 (CH₂), 38.1

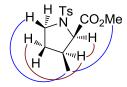
⁴ Cambiè, M.; D'Arrigo, P.; Fasoli, E.; Servi, S.; Tessaro, D.; Canevotti, F.; Coron, L. D. *Tetrahedron: Asymmetry* **2003**, *14*, 3189.

(CH), 52.2 (CH₃), 59.1 (CH), 127.4 (2 x CH), 129.6 (2 x CH), 136.7 (C), 143.6 (C), 172.1 (C); MS (ESI+) m/z (rel intensity) 322 (M⁺+ Na, 100); HRMS calcd for C₁₄H₂₁NO₄NaS, 322.1089, found 322.1088.

(2*R*,3*S*) Methyl 3-methyl-1-tosylpyrrolidine-2-carboxylate (2a): Colorless oil: $[\alpha]_D$ +53.2 (*c* 1.2); ¹H NMR δ 0.94 (d, *J* = 6.9 Hz, 3 H), 1.80 (dddd, *J* = 8.2, 10.1, 12.3, 12.3 Hz,

1 H), 1.92 (dddd, J = 2.2, 6.6, 6.6, 12.8 Hz, 1 H), 2.26 (ddquin, $J = 6.9 \times 4$, 8.5, 11.7 Hz, 1 H), 2.43 (s, 3 H), 3.24 (ddd, J = 6.8, 9.1, 10.0 Hz, 1 H), 3.60 (ddd, J = 1.9, 8.2, 8.2 Hz, 1 H), 3.68 (s, 3 H), 4.24 (d, J = 8.5 Hz, 1 H), 7.31 (d, J = 8.5 Hz, 2 H), 7.73 (d, J = 8.2 Hz, 2 H); ¹³C NMR δ 14.8 (CH₃), 21.9 (CH₃), 32.4 (CH₂), 37.9 (CH), 48.0 (CH₂), 52.1 (CH₃), 64.5 (CH), 127.8 (2 x CH), 130.0 (2 x CH), 135.9 (C), 143.8 (C), 171.5 (C); MS (ESI+) m/z (rel intensity) 320 (M⁺+ Na, 100); HRMS calcd for C₁₄H₁₉NO₄NaS, 320.0932, found 320.0937.

Main correlations observed by NOESY are shown in the figure:



Ts (2*R*,3*S*) Methyl 3-methyl-5-oxo-1-tosylpyrrolidine-2-carboxylate (2c): Colorless oil: $[\alpha]_D + 13.8 (c \ 0.5); {}^{1}H \ NMR \ \delta 1.05 (d, J = 6.9 \ Hz, 3 \ H), 2.30 (dd, J = 12.3, 17.0 \ Hz, 1 \ H), 2.44 (s, 3 \ H), 2.48 (dd, J = 8.0, 17.0 \ Hz, 1 \ H), 2.68-2.81 (m, 1 \ H), 3.76 (s, 3 \ H), 4.83$

(d, J = 8.6 Hz, 1 H), 7.33 (d, J = 8.2 Hz, 2 H), 7.92 (d, J = 8.3 Hz, 2 H); ¹³C NMR (125 MHz) δ 15.3 (CH₃), 22.1 (CH₃), 31.1 (CH), 38.6 (CH₂), 52.6 (CH₃), 64.1 (CH), 129.3 (2 x CH), 129.7 (2 x CH), 135.4 (C), 145.7 (C), 170.1 (C), 172.5 (C); MS (ESI+) m/z (rel intensity) 334 (M⁺+ Na, 100); HRMS calcd for C₁₄H₁₇NO₅NaS, 334.0725, found 334.0723.

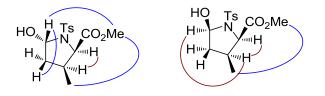
 $HO_{V} \xrightarrow{N} CO_{2}Me$ (2R,3S) Methyl 5-hydroxy-3-methyl-1-tosylpyrrolidine-2-carboxylate [2b(OH)]: $CO_{2}Me$ CO_{2}

(5S)-**2b(OH**): ¹H NMR (500 MHz) δ 0.95 (d, J = 6.9 Hz, 3 H), 1.90 (ddd, J = 5.0, 12.9 x 2 Hz, 1 H), 1.96 (dd, J = 6.3, 12.6 Hz, 1 H), 2.43 (s, 3 H), 2.93 (dddq, J = 6.9 x 4, 8.2, 13.6 Hz, 1 H), 3.55 (s, 3 H), 4.38 (d, J = 8.2 Hz, 1 H), 5.60 (d, J = 5.0 Hz, 1 H), 7.29 (d, J = 7.6 Hz, 2 H), 7.8 (d, J = 8.2 Hz, 2 H); ¹³C NMR δ 14.2 (CH₃), 21.9 (CH₃), 35.0 (CH), 40.6 (CH₂), 52.1 (CH₃), 64.4 (CH), 84.5 (CH), 127.8 (2 x CH), 129.9 (2 x CH), 137.2 (C), 144.2 (C), 170.8 (C).

(5R)- **2b**(**OH**: ¹H NMR (500 MHz) δ 0.95 (d, J = 6.9 Hz, 3 H), 1.65 (ddd, J = 5.4, 11.7, 13.2 Hz, 1 H), 2.32-2.40 (m, 1 H), 2.42 (s, 3 H), 2.49 (ddd, J = 6.3, 6.9, 13.9 Hz, 1 H), 3.62 (s, 3 H), 4.31 (d, J = 8.2 Hz, 1 H), 5.62 (d, J = 6.3 Hz, 1 H), 7.31 (d, J = 7.6 Hz, 4 H), 7.8 (d, J = 8.2 Hz, 2 H); ¹³C NMR δ 15.0 (CH₃), 21.9 (CH₃), 36.9 (CH), 42.1 (CH₂), 52.5 (CH₃), 64.4 (CH), 85.7 (CH), 127.8 (2 x CH), 129.9 (2 x CH), 137.2 (C),

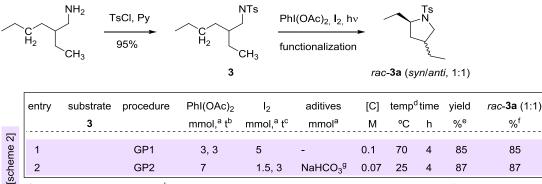
144.1 (C), 172.2 (C); MS (ESI+) m/z (rel intensity) 336 (M⁺+ Na, 100); HRMS calcd for C₁₄H₁₉NO₆NaS, 336.0882, found 336.0879.

Main correlations observed by NOESY are shown in the figure:



Synthesis and Functionalization of Tosylamine 3:

Table S3.



^a millimoles per millimol of **3**; ^b Time of addition in hours: Portionwise (0.25 equiv / 15 min); ^c time of continous addition of I₂ disolved in DCE (0.15M) by syringe pump (in hours); ^d 25 °C (reaction mixture refrigerated by a fan) and 65-75 °C (without fan); ^e total isolated yield; ^f isolated yield; ^g 100% w/w.

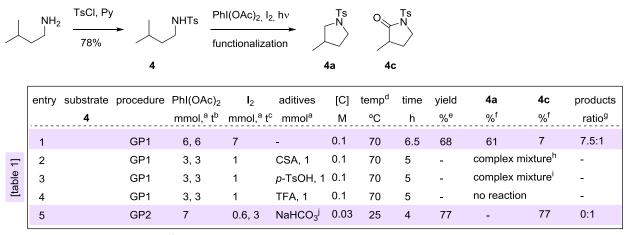
N-(2-Ethylhexyl)-4-methylbenzenesulfonamide (3): According to GP5, N-tosyl amine 3 NTs was prepared from 2-ethylhexan-1-amine (1.16 g, 9.0 mmol) in 95% as a colorless oil: ¹H NMR $\delta 0.78$ (t, J = 7.4 Hz, 3H), 0.83 (t, J = 7.0 Hz, 3H), 1.10-1.31 (m, 8H), 1.35 (sept, J = 6.1 Hz, 1H), 2.24 (s, 3H), 2.83 (d, J = 5.7 Hz, 2H), 4.87 (br s, 1H), 7.30 (d, J = 8.0 Hz, 2H), 7.77 (d, J = 5.7 Hz, 2H), 4.87 (br s, 1H), 7.30 (d, J = 8.0 Hz, 2H), 7.77 (d, J = 5.7 Hz, 2H), 4.87 (br s, 1H), 7.30 (d, J = 8.0 Hz, 2H), 7.77 (d, J = 5.7 Hz, 2H), 4.87 (br s, 1H), 7.30 (d, J = 8.0 Hz, 2H), 7.77 (d, J = 5.7 Hz, 2H), 4.87 (br s, 1H), 7.30 (d, J = 8.0 Hz, 2H), 7.77 (d, J = 5.7 Hz, 2H), 4.87 (br s, 1H), 7.30 (d, J = 8.0 Hz, 2H), 7.77 (d, J = 5.7 Hz, 2H), 7.77 (d, 8.0 Hz, 2H); ¹³C NMR δ11.0 (CH₃), 14.4 (CH₃), 21.9 (CH₃), 23.3 (CH₂), 24.2 (CH₂), 29.0 (CH₂), 31.0 (CH₂), 39.6 (CH), 46.2 (CH₂), 127.5 (2 x CH), 130.0 (2 x CH), 137.5 (C), 143.6 (C); MS (ESI⁺) m/z (rel intensity) 306 (M⁺+Na, 100); HRMS calcd for C₁₅H₂₅NO₂NaS, 306.1504, found 306.1505.

2,4-Diethyl-1-tosylpyrrolidine (3a): Crystalline solid; NMR showed a mixture of isomer *syn/anti* in 1:1 ratio. ¹H NMR (500Mz) δ 0.80 (t, J = 7.3 Hz, 3H), 0.82 (t, J = 7.3 Hz, 3H),

0.89 (t, J = 7.2 Hz, 3H), 0.90 (t, J = 7.2 Hz, 3H), 1.09-1.19 (m, 4H), 1.23-1.29 (m, 2H), 1.30-1.40 (m, 1H), 1.46-1.62 (m, 2H), 1.72 (ddd, J = 1.9, 6.2, 6.2 Hz, 1H), 1.80-1.88 (m, 1H), 1.97-2.15 (m, 3H), 2.43 (s, 6H), 2.63 (t, J = 9.5 Hz, 1H), 2.85 (br t, J = 11.0 Hz, 1H), 3.48-3.58 (m, 3H), 3.62 (br dd, J = 7.3, 11.3 Hz, 1H), 7.31 (d, J = 7.9 Hz, 4H), 7.72 (d, J = 8.0 Hz, 4H); ¹³C NMR δ 10.1 (CH₃), 10.9 (CH₃), 12.8 (CH₃), 12.9 (CH₃), 21.9 (2 x CH₃), 25.9 (CH₂), 26.3 (CH₂), 29.7 (CH₂), 29.8 (CH₂), 36.3 (CH₂), 38.2 (CH₂), 39.2 (CH), 40.3 (CH), 54.5 (CH₂), 55.0 (CH₂), 62.2 (CH), 62.6 (CH), 127.8 (2 x CH), 127.9 (2 x CH), 129.9

(2 x CH), 130.0 (2 x CH), 135.2 (C), 136.2 (C), 143.5 (2 x C); MS (ESI⁺) m/z (rel intensity) 304 (M⁺+Na, 100); HRMS calcd for C₁₅H₂₃NO₂NaS, 304.1347, found 304.1352.

Synthesis and Functionalization of Tosylamine 4: Table *S4*.



^a millimoles per millimol of **4**; ^b Time of addition in hours: Portionwise (0.25 equiv / 15 min); ^c time of continous addition of **I**₂ disolved in DCE (0.15M) by syringe pump (in hours); ^d 25 °C (reaction mixture refrigerated by a fan) and 65-75 °C (without fan); ^e total isolated yield; ^f isolated yield; ^g calculated by integration in the ¹H NMR of the crude of reaction; ^h aproximatly 60% of reaction was completed; ⁱ aproximatly 10% of reaction was completed; ^j 100% w/w. CSA = Camphorsulfonic acid; *p*-TsOH = *p*-toluenesulfonic acid; TFA = trifluoroacetic acid.

NHTs *N*-Isopentyl-4-methylbenzenesulfonamide (4): According to GP5, *N*-tosylamine 4 was prepared from 3-methylbutan-1-amine (704 mg, 8.1 mmol) in 57% as a pale yellow oil: ¹H NMR δ0.81 (d, J = 6.6 Hz, 6H), 1.33 (q, J = 7.0 Hz, 2H), 1.57 (non, J = 6.7 x 8 Hz, 1H), 2.43 (s, 3H), 2.92 (t, J = 7.2 Hz, 2H), 4.97 (br s, 1H), 7.31 (d, J = 8.0 Hz, 2H), 7.77 (d, J = 8.1 Hz, 2H); ¹³C NMR δ21.9 (CH₃), 22.6 (2 x CH₃), 25.8 (CH), 38.7 (CH₂), 41.9 (CH₂), 127.5 (2 x CH), 130.0 (2 x CH), 137.4 (C), 143.6 (C); MS (ESI⁺) *m*/*z* (rel intensity) 264 (M⁺+Na, 100); HRMS calcd for C₁₂H₁₉NO₂NaS, 264.1034, found 264.1027.

Ts 3-Methyl-1-tosylpyrrolidine (**4a**):⁵ Crystalline solid: Mp 69-71 °C (Et₂O/hexane); [literature = 67-68 °C (hexane/DCM)]; ¹H NMR (500 MHz) δ 0.92 (d, J = 6.6 Hz, 3H), 1.35 (dddd, J = 8.3, 8.3, 8.3, 12.3 Hz, 1H), 1.90 (dddd, J = 4.2, 7.3, 7.3, 12.3 Hz, 1H), 2.07-2.17 (m, 1H), 2.43 (s, 3H), 2.75 (dd, J = 7.9, 9.8 Hz, 1H), 3.22 (ddd, J = 7.3, 8.2, 9.8 Hz, 1H), 3.34 (ddd, J = 4.1, 8.2, 9.8 Hz, 1H), 3.42 (dd, J = 7.3, 9.8 Hz, 1H), 7.32 (d, J = 7.9 Hz, 2H), 7.71 (d, J = 8.0 Hz, 2H); ¹³C NMR (125.7 MHz) δ 17.7 (CH₃), 21.5 (CH₃), 33.3 (CH₂), 33.3 (CH), 47.6 (CH₂), 54.8 (CH₂), 127.6 (2 x CH), 129.6 (2 x CH), 134.2 (C), 143.2 (C); MS (ESI⁺) *m*/*z* (rel intensity) 262 (M⁺ + Na, 100); HRMS calcd for C₁₂H₁₇NO₂NaS, 262.0878, found 262.0876.

⁵ Spectral data are in good agreement with those reported in the literature: (a) Tang, J.; Shinokubo, H.; Oshima, K. *Tetrahedron*, **1999**,*55*, 1893; (b) Rai, K. M. L.; Hassner, A. *Heterocycles*, **1990**, *30*, 817.

3-Methyl-1-tosylpyrrolidin-2-one (4c):⁶ Crystalline solid. Mp 84-86 °C (hexane/EtOAc); ¹H NMR δ 1.13 (d, J = 7.1 Hz, 3H), 1.65-1.75 (m, 1H), 2.22-2.30 (m, 1H), 2.43 (s, 3H), 2.42-2.52 (m, 1H), 3.69 (ddd, J = 6.9, 9.7, 9.7 Hz, 1H), 3.94 (ddd, J = 2.4, 8.6, 9.9 Hz, 1H), 7.33 (d, J = 7.9 Hz, 2H), 7.91 (d, J = 8.2 Hz, 2H); ¹³C NMR δ 15.3 (CH₃), 22.0 (CH₃), 27.5 (CH₂), 38.6 (CH), 45.6 (CH₂), 128.4 (2 x CH), 130.0 (2 x CH), 135.6 (C), 145.5 (C), 176.2 (C); MS (ESI⁺) m/z (rel intensity) 276 (M⁺+Na, 100); HRMS calcd for C₁₂H₁₅NO₃NaS, 276.0670, found 276.0674.

Synthesis and Functionalization of Tosylamine 5:

Table S5.

	NH ₂	HCI —	CI, Et₃N ► 2%	NH 5	-	hI(OAc) _{2,} I _{2,} hv ////////////////////////////////////	-	Ts N (3S)-4a	0	Ts N 5c			
	entry	substrate	procedure	· /2	I ₂	aditives	[C]	temp ^d	time	yield	(3S)- 4a	5c	products
		5		mmol, ^a t ^b	mmol, ^a t	^c mmol ^a	_ <u>M</u> _	°C	h	% ^e	% ^f	% ^f	ratio ^g
	1		GP1	6, 6	7	-	0.1	70	6.5	65	65	-	1:0
e 1]	2		GP1	2, 2	0.6	<i>p</i> -TSOH, 1	0.1	70	3	-	no read	ction	-
[table	3		GP2	7	0.6, 4	NaHCO ₃ ^h	0.1	25	4.5	61	9	52	1:3
	4		GP2	7	0.6, 4	NaHCO ₃ ^h	0.03	25	4.5	78	-	78	0:1

^a millimoles per millimol of **5**; ^b Time of addition in hours: Portionwise (0.25 equiv / 15 min); ^c time of continous addition of I_2 disolved in DCE (0.15M) by syringe pump (in hours); ^d 25 °C (reaction mixture refrigerated by a fan) and 65-75 °C (without fan); ^e total isolated yield; ^f isolated yield; ^g calculated by integration in the ¹H NMR of the crude of reaction; ^h 100% w/w.

NHTs (*S*)-4-Methyl-*N*-(2-methylbutyl)benzenesulfonamide (5):⁷ TsCl (3.59 g, 18.86 mmol) was added to a solution of (*S*)-2-methylbutan-1-amine⁸ (1.16 g, 9.43 mmol) and Et₃N (7.9 mL, 56.58 mmol) in dry DCM (17 mL) at 0 °C and stirred at rt for 2 h. The reaction mixture was poured into a diluted aqueous solution of HCl (10%) and extracted twice with EtOAc. The organic extracts were washed with saturated aqueous solution of NaHCO₃ and brine, dried over anhydrous NaSO₄ and concentrated under vacuum. Column chromatography (hexanes–EtOAc, 95:5-90:10) of the residue afforded the corresponding sulfonamide **6** (1.41 g, 5.85 mmol, 62%) as a colorless oil: $[\alpha]_D$ +2.5 (*c* 0.7); ¹H NMR δ 0.82 (t, *J* = 7.4 Hz, 3H), 0.85 (d, *J* = 6.6 Hz, 3H), 1.10 (dquin, *J* = 7.5 x 4, 13.5 Hz, 1H), 1.35 (ddq, *J* = 5.3, 7.5 x 3, 13.5 Hz, 1H), 1.48 (oct, *J* = 6.7 x 7 Hz, 1H), 2.43 (s, 3H), 2.72 (ddd, *J* = 6.8, 6.8, 12.4 Hz, 1H),

⁶ Spectral data are in good agreement with those reported in the literature: (a) Padwa, A.; Kissel, W. S.; Eidell, C. K. *Can. J. Chem.* **2001**, *79*, 1681; (b) Boal, B. W.; Schammel, A. W.; Garg, N. K. *Org. Lett.* **2009**, *11*, 3458; (c) Schammel, A. W.; Boal, B. W.; Zu, L.; Mesganaw, T.; Garg N. *Tetrahedron* **2010**, *66*, 4687.

⁷ Spectral data are in good agreement with those reported in the literature for the racemic compund: Zhu, M.; Fujita, K.-I.; Yamaguchi, R. *Org. Lett.* **2010**, *12*, 1336.

⁸ This compound was obtained from L-isoleucine according the procedure reported in the literature: Hashimoto, M.; Eda, Y.; Osanai, Y.; Iwai, T.; Aoki, S. *Chem. Lett.* **1986**, 893.

2.84 (ddd, J = 6.3, 6.3, 12.3 Hz, 1H), 4.66 (t, J = 6.2 Hz, 1H), 7.31 (d, J = 8.0 Hz, 2H), 7.75 (d, J = 8.2 Hz, 2H); ¹³C NMR δ 11.4 (CH₃), 17.3 (CH₃), 21.8 (CH₃), 27.0 (CH₂), 35.1 (CH), 49.1 (CH₂), 127.5 (2 x CH), 130.0 (2 x CH), 137.6 (C), 143.6 (C); MS (ESI⁺) m/z (rel intensity) 264 (M⁺+ Na, 100); HRMS calcd for C₁₂H₁₉NO₂NaS, 264.1034, found 264.1032.

^{Ts} (*S*)-3-Methyl-1-tosylpyrrolidine [(3*S*)-4a]:⁹ Crystalline solid: Mp 87–91 °C (hexane/Et₂O); [α]_D -8.51 (*c* 1.6); ¹H NMR δ 0.91 (d, *J* = 6.9 Hz, 3H), 1.35 (dddd, *J* = 8.3, 8.3, 8.3, 12.3 Hz, 1H), 1.90 (dddd, *J* = 4.2, 7.3, 7.3, 12.3 Hz, 1H), 2.08-2.16 (m, 1H), 2.43 (s, 3H), 2.75 (dd, *J* = 7.8, 9.7 Hz, 1H), 3.22 (ddd, *J* = 7.1, 8.2, 9.8 Hz, 1H), 3.34 (ddd, *J* = 4.0, 8.3, 9.7 Hz, 1H), 3.42 (dd, *J* = 7.1, 9.5 Hz, 1H), 7.32 (d, *J* = 7.7 Hz, 2H), 7.71 (d, *J* = 8.1 Hz, 2H); ¹³C NMR δ 18.0 (CH₃), 21.9 (CH₃), 33.6 (CH₂), 33.7 (CH), 48.0 (CH₂), 55.1 (CH₂), 127.9 (2 x CH), 130.0 (2 x CH), 134.6 (C), 143.6 (C); MS (ESI⁺) *m/z* (rel intensity) 262 (M⁺+ Na, 100); HRMS calcd for C₁₂H₁₇NO₂NaS, 262.0878, found 262.0877.

 $\begin{array}{c} (S) - 4 - Methyl - 1 - tosylpyrrolidin - 2 - one (5c): ^{10} Crystalline solid: Mp 87 - 88 °C (hexane/Et_2O); \\ [\alpha]_D + 10.0 (c 0.9); ^1H NMR \delta 1.09 (d, J = 6.6 Hz, 3H), 2.07 (dd, J = 7.4, 16.7 Hz, 1H), 2.44 (s, 3H), 2.42 - 2.52 (m, 1H), 2.57 (dd, J = 7.9, 16.7 Hz, 1H), 3.42 (dd, J = 6.4, 9.8 Hz, 1H), 4.03 (dd, J = 7.3, 9.7 Hz, 1H), 7.34 (d, J = 8.7 Hz, 2H), 7.92 (d, J = 8.5 Hz, 2H); ^{13}C NMR \delta 18.1 (CH_3), 22.0 (CH_3), 27.2 (CH), 40.8 (CH_2), 54.4 (CH_2), 128.4 (2 x CH), 130.1 (2 x CH), 135.7 (C), 145.5 (C), 173.3 (C); MS (ESI⁺) <math>m/z$ (rel intensity) 276 (M⁺ + Na, 100); HRMS calcd for C₁₂H₁₅NO₃NaS, 276.0670, found 276.0676.

Synthesis and Functionalization of Tosylamine 6:

Table S6.

		TsCl, 68%	· •	NHTs		c) _{2,} I _{2,} h∨ → nalization		s }	O,	Ts N			
				6			6	а		6b			
	entry	substrate	procedure	PhI(OAc) ₂	I ₂	aditives	[C] 1	temp ^d	time	yield	6a	6b	products
		6		mmol, ^a t ^b	mmol, ^a t ^c	equiv	M	°C	h	% ^e	% ^f	% ^f	ratio ^g
e 1]	1		GP1	4.5, 4.5	7	-	0.1	70	5	65	65	-	1:0
[table	2		GP2	5	0.6, 3	NaHCO ₃ ^h	0.03	25	4	56	-	56	0:1

^a millimoles per millimol of **6**; ^b Time of addition in hours: Portionwise (0.25 equiv / 15 min); ^c time of continous addition of **I**₂ disolved in DCE (0.15M) by syringe pump (in hours); ^d 25 °C (reaction mixture refrigerated by a fan) and 65-75 °C (without fan); ^e total isolated yield; ^f isolated yield; ^g calculated by integration in the ¹H NMR of the crude of reaction; ^h 100% w/w.

⁹ Spectral data are in good agreement with those reported in the literature: reference 5a.

¹⁰ This ¹H NMR spectral data was in good agreement with that reported in the literature: Ozaki, S.; Matsushita, H.; Ohmori, H. J. Chem. Soc., Perkin Trans. 1 **1993**, 2339.

NHTs *rac*-4-Methyl-*N*-(pentan-2-yl)benzenesulfonamide (6):¹¹ According to GP5, *N*-tosylamine 6 was prepared from pentan-2-amine (1.85 g, 21.27 mmol) in 68% as a crystalline solid: ¹H NMR $\delta 0.79$ (t, J = 7.1 Hz, 3H), 1.01 (d, J = 6.6 Hz, 3H), 1.15-1.40 (m, 4H), 2.42 (s, 3H), 3.25-3.35 (m, 1H), 4.48 (br s, 1H), 7.29 (d, J = 7.9 Hz, 2H), 7.77 (d, J = 8.1 Hz, 2H); ¹³C NMR $\delta 14.1$ (CH₃), 19.1 (CH₂), 21.9 (CH₂), 22.0 (CH₃), 40.0 (CH₂), 50.1 (CH), 127.4 (2 x CH), 130.0 (2 x CH), 138.8 (C), 143.4 (C); MS (ESI⁺) m/z (rel intensity) 264 (M⁺+ Na, 100); HRMS calcd for C₁₂H₁₉NO₂NaS, 264.1034, found 264.1032.



rac-2-Methyl-1-tosylpyrrolidine (6a):¹² Crystalline solid: Mp 95–95.5 °C (Et₂O); ¹H NMR δ 1.31 (d, J = 6.4 Hz, 3H), 1.45-1.57 (m, 2H), 1.64-1.74 (m, 1H), 1.77-1.88 (m, 1H), 2.43 (s, 3H),

3.15 (ddd, J = 7.3, 7.3, 10.3 Hz, 1H), 3.44 (ddd, J = 4.9, 7.1, 10.0 Hz, 1H), 3.71 (ddq, J = 4.0, 6.6, 6.6, 6.6, 7.2 Hz, 1H), 7.31 (d, J = 8.0 Hz, 2H), 7.72 (d, J = 8.2 Hz, 2H); ¹³C NMR δ 21.9 (CH₃), 23.2 (CH₃), 24.3 (CH₂), 33.9 (CH₂), 49.4 (CH₂), 56.5 (CH), 127.9 (2 x CH), 130.0 (2 x CH), 135.5 (C), 143.5 (C); MS (ESI⁺) m/z (rel intensity) 262 (M⁺+ Na, 100); HRMS calcd for C₁₂H₁₇NO₂NaS, 262.0878, found 262.0877.

 $\begin{array}{c} \begin{array}{c} & \quad \textbf{rac-5-Methyl-1-tosylpyrrolidin-2-one (6b):}^{13} \mbox{ Crystalline solid: Mp 135-136 °C (n-hexane/$ Et_2O); $^{1}H \mbox{ NMR } \delta 1.46 (d, $J = 6.4 \mbox{ Hz}, 3H), 1.69-1.75 (m, 1H), 2.21-2.31 (m, 1H), 2.35 (ddd, $J = 2.5, 9.3, 17.3 \mbox{ Hz}, 1H), 2.43 (s, 3H), 2.56 (ddd, $J = 8.9, 10.5, 17.2 \mbox{ Hz}, 1H), 4.52 (ddq, $J = 2.1, 6.4, 6.4, 6.4, 8.2 \mbox{ Hz}, 1H), 7.32 (d, $J = 8.0 \mbox{ Hz}, 2H), 7.94 (d, $J = 8.5 \mbox{ Hz}, 2H); $^{13}C \mbox{ NMR } \delta 21.9 (CH_3), 22.0 (CH_3), 27.0 (CH_2), 30.9 (CH_2), 56.8 (CH), 128.7 (2 \ x \ CH), 129.9 (2 \ x \ CH), 136.6 (C), 145.3 (C), 173.7 (C); \mbox{ MS} (ESI^+) m/z (rel intensity) 276 (M^++ \ Na, 100); HRMS calcd for $C_{12}H_{15}NO_3NaS$, 276.0670, found 276.0676. \end{array}$

¹¹ Spectral data are in good agreement with those reported in the literature: Bossart, M.; Faessler, R.; Schoenberger, J.; Studer, A. *Eur. J. Org. Chem.* **2002**, 2742.

¹² Spectral data are in good agreement with those reported in the literature: (a) Andrés, J. M.; Herráiz-Sierra, I.; Pedrosa, R.; Pérez-Encabo, A. *Eur. J. Org. Chem.* **2000**, 1719; (b) Kato, Y.; Yen, D. H.; Fukudome, Y.; Hata, T.; Urabe, H. *Org. Lett.* **2010**, *12*, 4137.

¹³ Spectral data are in good agreement with those reported in the literature: Wang, J.; Hou, Y. J. Chem. Soc., *Perkin Trans. 1* **1998**, 1919.

Synthesis and Functionalization of Tosylamine 7:

Table S7.

	NH ₂ COOH 1) SOCI ₂ , Me 2) TsCI, Py 90%) TsCl, Py	NHTs			PhI(OAc) ₂ , I ₂ , hv			Ts N-COOMe 7a		O N COOMe 7c	
	entry	substrate 7	procedure	· /2	Ι ₂ mmol, ^a t ^c	aditives mmol ^a	[C] M	temp ^d °C	time h	yield % ^e	7a % ^f	7c % ^f	products ratio ^g
e 1]	1		GP1	4.5, 4.5	5	-	0.1	70	5	86 ^h	86	-	1:0
	2		GP1	3, 3	1	CSA, 10%	0.04	70	10	60	60	-	1:0
[table	3		GP2	7	1.2, 4	NaHCO ₃ ⁱ	0.03	25	20	82	9	73	1:9
	4		GP3	6	1.4	NaHCO ₃ ⁱ	0.03	25	20	83	-	83	0:1

^a millimoles per millimol of **7**; ^b Time of addition in hours: Portionwise (0.25 equiv / 15 min); ^c time of continous addition of I_2 disolved in DCE (0.15M) by syringe pump (in hours); ^d 25 °C (reaction mixture refrigerated by a fan) and 65-75 °C (without fan); ^e total isolated yield; ^f isolated yield; ^g calculated by integration in the ¹H NMR of the crude of reaction; ^h starting material **7** was recovered in 12%; ⁱ 100% w/w.

(S)-Methyl 2-(4-methylphenylsulfonamido)pentanoate (7): A solution of D-norvaline NHTs (1.17 g, 9.95 mmol) in MeOH (60 mL) was treated at 0 °C with SOCl₂ (1.44 mL, 19.9 CO₂Me mmol). The mixture was refluxed for 4 h and concentrated to give the crude methyl ester as viscous oil. The product was dissolved in pyridine (47 mL), TsCl (5.69 g, 29.85 mmol) was added and the mixture was heated at 40 °C for 24 h. The mixture was concentrated, dissolved in EtOAc and washed with diluted aqueous solution of HCl (10%). The organic extract was washed with saturated aqueous solution of NaHCO₃ and brine, dried over anhydrous NaSO₄ and concentrated under vacuum. Column chromatography (hexanes–EtOAc, 75:25) of the residue afforded the corresponding compound 7 (2.56 g, 8.98 mmol, 90%) as a crystalline solid: Mp 57.5-58.5 °C (hexane/AcOEt); $[\alpha]_{\rm D}$ -18.1 (c 1.3); ¹H NMR δ 0.87 (t, J = 7.3 Hz, 3H), 1.36 (sxt, $J = 7.5 \times 5 \text{ Hz}$, 2H), 1.55-1.73 (m, 2H), 2.41 (s, 3H), 3.48 (s, 3H), 3.91 (ddd, J = 5.3, 7.7, 9.3 Hz, 1H), 5.19 (d, J = 9.5 Hz, 1H), 7.29 (d, J = 7.9 Hz, 2H), 7.72 (d, J = 8.2 Hz, 2H); ¹³C NMR δ 13.7 (CH₃), 18.6 (CH₂), 21.9 (CH₃), 35.8 (CH₂), 52.7 (CH₃), 55.9 (CH), 127.7 (2 x CH), 130.0 (2 x CH), 137.3 (C), 144.0 (C), 172.7 (C); MS (ESI⁺) m/z (rel intensity) 308 (M⁺+ Na, 100); HRMS calcd for C₁₃H₁₉NO₄NaS, 308.0932, found 308.0934.

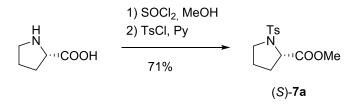
Ts (*R*)-Methyl 1-tosylpyrrolidine-2-carboxylate (7a):¹⁴ Crystalline solid: Mp 75.5-75.7 °C (hexane/Et₂O); $[\alpha]_D$ +82.7 (c 1.5); ¹H NMR δ 1.70-1.78 (m, 1H), 1.91-2.04 (m, 3H), 2.43 (s, 3H), 3.30 (ddd, J = 7.1, 7.1, 9.5 Hz, 1H), 3.49 (ddd, J = 4.7, 7.2, 9.8 Hz, 1H), 3.71 (s, 3H), 4.28 (dd, J = 4.6, 8.0 Hz, 1H), 7.32 (d, J = 8.2 Hz, 2H), 7.75 (d, J = 8.2 Hz, 2H); ¹³C NMR δ 21.5 (CH₃), 24.6 (CH₂), 30.9 (CH₂), 48.5 (CH₂), 52.3 (CH₃), 60.4 (CH), 127.5 (2 x CH), 129.7 (2 x CH), 135.2 (C), 143.7 (C), 172.6

¹⁴ Spectral data are consistent with previously published literature values of its enantiomer: Lalonde, R. L.; Sherry, B. D.; Kang, E. J.; Toste, F.D. *J. Am. Chem. Soc.* **2007**, *129*, 2452.

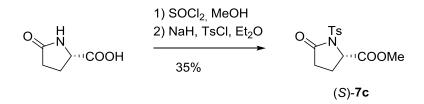
(C); MS (ESI⁺) m/z (rel intensity) 306 (M⁺+ Na, 100); HRMS calcd for C₁₃H₁₇NO₄NaS, 306.0776, found 306.0776.

 $\begin{array}{l} (\textbf{R}) \textbf{-Methyl 5-oxo-1-tosylpyrrolidine-2-carboxylate (7c): Crystalline solid: Mp 119.5-}\\ (\textbf{R}) \textbf{-CO_2Me} & 121.5 \ ^\circ\text{C} (\text{Et}_2\text{O}); \ [\alpha]_{\text{D}} + 44.1 \ (c \ 1.5); \ ^1\text{H} \text{ NMR} \ (500 \text{ MHz}) \ \delta 2.09-2.14 \ (m, \ 1\text{H}), \ 2.39-2.48}\\ (m, \ 2\text{H}), \ 2.44 \ (s, \ 3\text{H}), \ 2.51-2.57 \ (m, \ 1\text{H}), \ 3.78 \ (s, \ 3\text{H}), \ 4.89 \ (dd, \ J = 2.7, \ 8.7 \ \text{Hz}, \ 1\text{H}), \ 7.33 \ (d, \ J = 8.2 \ \text{Hz}, \ 2\text{H}), \ 7.96 \ (d, \ J = 8.5 \ \text{Hz}, \ 2\text{H}); \ ^{13}\text{C} \text{ NMR} \ (125.7 \ \text{MHz}) \ \delta 21.7 \ (\text{CH}_3), \ 23.3 \ (\text{CH}_2), \ 30.5 \ (\text{CH}_2), \ 52.9 \ (\text{CH}_3), \ 59.4 \ (\text{CH}), \ 129.1 \ (2 \ \text{x} \ \text{CH}), \ 129.4 \ (2 \ \text{x} \ \text{CH}), \ 135.0 \ (\text{C}), \ 145.4 \ (\text{C}), \ 171.3 \ (\text{C}), \ 172.7 \ (\text{C}); \ \text{MS} \ (\text{ESI+}) \ m/z \ (\text{rel intensity}) \ 320 \ (\text{M}^+ + \text{Na}, \ 100); \ \text{HRMS} \ \text{calcd for} \ \text{C}_{13}\text{H}_{15}\text{NO}_5\text{NaS}, \ 320.0569, \ \text{found} \ 320.0572. \end{array}$

Synthesis of the enantiomers of 7a and 7c [(S)-7a and (S)-7c]:



Ts (*S*)-Methyl 1-tosylpyrrolidine-2-carboxylate [(*S*)-7a]: A solution of L-proline (234 mg, N :::CO₂Me 2.03 mmol) in MeOH (12 mL) was treated at 0 °C with SOCl₂ (0.3 mL, 4.14 mmol). The mixture was refluxed for 4 h and concentrated to give the crude methyl ester as viscous oil. The product was dissolved in pyridine (8 mL), TsCl (450 mg, 2.36 mmol) was added and the mixture was heated at 40 °C for 24 h. The mixture was concentrated, dissolved in EtOAc and washed with diluted aqueous solution of HCl (10%). The organic extract was washed with saturated aqueous solution of NaHCO₃ and brine, dried over anhydrous NaSO₄ and concentrated under vacuum. Column chromatography (hexanes–EtOAc, 75:25) of the residue afforded the corresponding L-proline derivative (*S*)-7a (408 mg, 1.44 mmol, 71%) as a crystalline solid: Mp 75.0-75.5 °C (hexane/Et₂O); $[\alpha]_D$ -82.7 (*c* 1.5) [Literature: -73.3 (c 1.0);¹⁴ -93.3 (c 1.5)¹⁵]



 O_{N} (*S*)-Methyl 5-oxo-1-tosylpyrrolidine-2-carboxylate [(*S*)-7c]: 1st step: A solution of (*S*)-2-pyrrolidinone-5-carboxylic acid (195 mg, 1.51 mmol) in MeOH (30 mL) was treated at 0 °C with SOCl₂ (110 µL, 1.51 mmol). The mixture was stirred at rt for 40 min, then neutralized with amberjet 4400-OH resin, filtered and concentrated. The crude product was purified by silica gel column

¹⁵ Fujita, Y.; Gottlieb, A.; Peterkofsky, B.; Udenfriend, S.; Witkop, B. J. Am. Chem. Soc. 1964, 86, 4709.

chromatography (DCM/MeOH, 95:5) to afford the corresponding methyl ester (125 mg, 0.876 mmol, 58%) as viscous oil.

2nd step: To a suspension of NaH (60% in oil, 29 mg, 0.724 mmol) in dry ether (1 mL) were added at 0 °C the previously obtained methyl ester (94 mg, 0.658 mmol) in dry ether (5 mL) and TsCl (125 mg, 0.658 mmol) in dry CH₂Cl₂ (2 mL). The reaction mixture was worked up according to the usual manner. The crude product was purified by silica gel column chromatography (hexane/AcOEt, 75:25) to give (*S*)-**7a** (117 mg, 0.39 mmol, 60%) as a crystalline solid: Mp 121.5–123.2 °C (hexane/CH₂Cl₂); $[\alpha]_D$ -46.6 (*c* 1.6).

Synthesis and Functionalization of Tosylamine 8:

Table S8.

\downarrow	NH ₂	TsCl, Et ₃ N 91.8%			PhI(OAc) _{2,}		Ts	COOI	Me	Ts N COOMe		
			8	3			8a (syr	n/anti)	8c	(syn/anti)		
	entry substrat	e procedure	PhI(OAc) ₂	I ₂	aditives	[C]	temp ^d	time	yield	8a (syn/anti)	8c (syn/anti)	products
	8		mmol, ^a t ^b	mmol, ^a	t ^c mmol ^a	M	°C	h	% ^e	% ^f	% ^f	ratio ^g
	1	GP1	3, 3	5	-	0.1	70	4.5	89	81(1:1.4)	8(1:0)	3.3:5:1:0
=	2	GP2	7	1, 6	NaHCO ₃ ^h	0.03	25	7	72	-	72(1.1:1)	0:0:1.1:1
[table	3	GP1 _{CSA}	2.5, 2.5	0.6	CSA, 1	0.1	70	3	92	92(1.1:1)	-	1.1:1:0:0
<u> </u>	4	GP1 _{Zn(OTf)2}	3, 3	0.5	Zn(TfO) ₂ , 0.6	0.1	70	4	88	88(1:1)	-	1:1:0:0
	5	GP1	3, 3	3	Zn(TfO) ₂ , 0.6	0.1	70	4	85	85(1:1)	-	1:1:0:0
	6	GP1	3, 3	3	Zn(TfO) ₂ , 0.6	0.1	-20	4	0	n	o reaction	-
	7	GP3	3	0.5	Zn(TfO) ₂ , 1	0.2	30	4	85	85(1.2:1)	-	1.2:1:0:0
	8	GP1	3, 3	-	Zn(TfO) ₂ , 0.6	0.1	70	4	0	n	o reaction	-
	9	GP1	3, 3	0.5	-	0.1	70	4	nd	60% rea	action completed	0.8:3:2.3:1

^a millimoles per millimol of **9**; ^b Time of addition in hours: Portionwise (0.25 equiv / 15 min); ^c time of continous addition of **I**₂ disolved in DCE (0.15M) by syringe pump (in hours); ^d 25 °C (reaction mixture refrigerated by a fan) and 65-75 °C (without fan); ^e total isolated yield; ^f isolated yield; ^g (**9a**-*syn*/**9a**-*anti*/**9c**-*syn*/**9c**-*anti*/**9A**) calculated by integration in the ¹H NMR of the crude of reaction; ^h 100% w/w. nd = not determined.

NHTs (*S*) Methyl 4-methyl-2-(4-methylphenylsulfonamido)pentanoate (8):¹⁶ TsCl (4.24 g, CO₂Me 22.25 mmol) was added to a solution of L-leucine methyl ester hydrochloride (2.02 g, 11.13 mmol) and Et₃N (6 mL, 41.72 mmol) in dry DCM (50 mL) at 0 °C and stirred at rt for 3 h. The reaction mixture was poured into a diluted aqueous solution of HCl (10%) and extracted twice with EtOAc. The organic extracts were washed with saturated aqueous solution of NaHCO₃ and brine, dried over anhydrous NaSO₄ and concentrated under vacuum. Column chromatography (hexanes–EtOAc, 90:10) of the residue afforded the corresponding sulfonamide **8** (3.05 g, 10.21 mmol, 91.8%) as crystalline solid: Mp 49.4-50.2 °C (hexane/Et₂O); $[\alpha]_D$ +7.3 (*c* 1.5); ¹H NMR δ 0.88 (d, *J* = 6.6 Hz, 3H), 0.90 (d, *J* = 6.6 Hz, 3H), 1.48 (ddd, *J* = 0.8, 6.5, 7.8 Hz, 2H), 1.78 (non, *J* = 6.7 x 8 Hz, 1H), 2.42 (s, 3H), 3.44 (s, 3H), 3.93 (ddd, *J* = 6.2, 8.1, 10.1 Hz, 1H), 5.10 (br s, 1H), 7.29 (d, *J* = 8.5 Hz, 2H), 7.71 (d, *J* = 8.2 Hz, 2H); ¹³C NMR δ 21.8

¹⁶ Spectral data are in good agreement with those reported in the literature: Ordóñez, M.; Cruz-Cordero, R.; Fernández-Zertuche, M.; Muñoz-Hernández, M. A.; García-Barradas, O. *Tetrahedron: Asymmetry* **2004**, *15*, 3035.

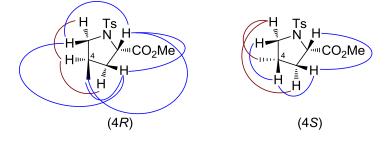
(CH₃), 21.9 (CH₃), 23.1 (CH₃), 24.7 (CH), 42.8 (CH₂), 52.6 (CH₃), 54.8 (CH), 127.7 (2 x CH), 129.9 (2 x CH), 137.2 (C), 144.0 (C), 173.1 (C); MS (ESI+) m/z (rel intensity) 322 (M⁺+ Na, 100); HRMS calcd for C₁₄H₂₁NO₄NaS, 322.1089, found 322.1101.

Ts (2S) Methyl 4-methyl-1-tosylpyrrolidine-2-carboxylate (8a). Colorless oil: NMR showed a mixture of isomers (4R/4S)-8a in 7.2:1 ratio. (4R)-8a: ¹H NMR δ 0.90 (d, J = 6.6 Hz, 3H), 1.64 (ddd, J = 9.6, 9.6, 12.6 Hz, 1H), 2.07 (ddd, J = 3.0, 6.3, 12.8 Hz, 1H),

2.43 (s, 3H), 2.42-2.48 (m, 1H), 2.81 (dd, J = 8.8, 8.8 Hz, 1H), 3.58 (dd, J = 7.1, 9.0 Hz, 1H), 3.71 (s, 3H), 4.34 (dd, J = 3.0, 9.0 Hz, 1H), 7.32 (d, J = 7.9 Hz, 2H), 7.76 (d, J = 8.2 Hz, 2H); ¹³C NMR δ 17.2 (CH₃), 21.9 (CH₃), 32.9 (CH), 38.9 (CH₂), 52.7 (CH₃), 55.3 (CH₂), 60.7 (CH), 127.9 (2 x CH), 130.0 (2 x CH), 135.7 (C), 143.9 (C), 173.0 (C). (4*S*)-**8a**: ¹H NMR δ 0.99 (d, J = 6.3 Hz, 3H), 1.59 (ddd, J = 8.8, 10.4, 12.6 Hz, 1H), 1.95 (ddq, J = 6.7 x 3, 9.9, 9.9 Hz, 1H), 2.34 (ddd, J = 7.3, 7.3, 12.6 Hz, 1H), 2.43 (s, 3H), 2.97 (dd, J = 9.8, 10.4 Hz, 1H), 3.54-3.62 (m, 1H), 3.75 (s, 3H), 4.25 (dd, J = 7.9, 8.8 Hz, 1H), 7.32 (d, J = 8.2Hz, 2H), 7.77 (d, J = 8.2 Hz, 2H); ¹³C NMR δ 17.0 (CH₃), 21.9 (CH₃), 33.9 (CH), 39.2 (CH₂), 52.7 (CH₃), 55.6 (CH₂), 61.4 (CH), 127.9 (2 x CH), 130.0 (2 x CH), 135.9 (C), 143.9 (C), 173.1 (C).

MS (ESI+) m/z (rel intensity) 320 (M⁺+ Na, 100); HRMS calcd for C₁₄H₁₉NO₄NaS, 320.0932, found 320.0935.

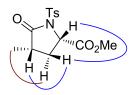
Main correlations observed by NOESY are shown in the figure:



 $\begin{array}{l} (2S,4S) \text{ Methyl 4-methyl-5-oxo-1-tosylpyrrolidine-2-carboxylate (syn-8c): Colorless} \\ (2S,4S) \text{ Methyl 4-methyl-5-conducted (s$

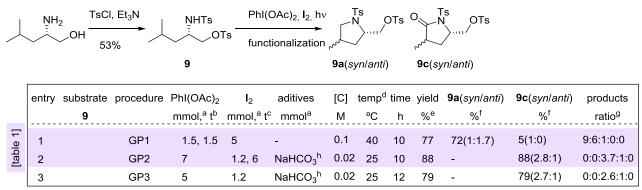
¹H NMR (500 MHz, C₆D₆) δ 0.69 (d, J = 7.3 Hz, 3H), 1.07 (ddd, J = 7.1, 8.3, 12.7 Hz, 1H), 1.61 (ddd, J = 8.7, 8.7, 12.7 Hz, 1H), 1.68-1.78 (m, 1H), 1.77 (s, 3H), 3.37 (s, 3H), 4.51 (dd, J = 7.1, 8.4 Hz, 1H), 6.78 (d, J = 8.2 Hz, 2H), 8.31 (d, J = 8.2 Hz, 2H); ¹³C NMR (125.7 MHz, C₆D₆) δ 15.2 (CH₃), 20.8 (CH₃), 30.5 (CH₂), 36.5 (CH), 51.8 (CH₃), 57.7 (CH), 128.9 (2 x CH), 129.5 (2 x CH), 136.0 (C), 144.5 (C), 171.7 (C), 174.6 (C); MS (ESI+) m/z (rel intensity) 334 (M⁺+ Na, 100); HRMS calcd for C₁₄H₁₇NO₅NaS, 334.0725, found 334.0725.

Main correlations observed by NOESY (in C_6D_6) are shown in the figure:



Synthesis and Functionalization of Tosylamine 9:

Table S9.



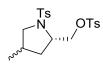
^a millimoles per millimol of **9**; ^b Time of addition in hours: Portionwise (0.25 equiv / 15 min); ^c time of continous addition of **I**₂ disolved in DCE (0.15M) by syringe pump (in hours); ^d 25 °C (reaction mixture refrigerated by a fan) and 65-75 °C (without fan); ^e total isolated yield; ^f isolated yield; ^g (**9a**-*syn*/**9a**-*anti*/**9c**-*syn*/**9c**-*anti*) calculated by integration in the ¹H NMR of the crude of reaction; ^h 100% w/w.

 (S)-4-Methyl-2-(4-methylphenylsulfonamido)pentyl 4-methylbenzenesulfonate (9):¹⁷

According to GP5, *N*-tosylamine **9** was prepared from L-leucinol (1 mmol) employing TsCl (2.2 mmol) in 53% as a crystalline solid: Mp 100.5-102.0 °C (hexane/Et₂O); $[\alpha]_D$ –47.6 (*c* 1.0); ¹H NMR δ 0.58 (d, *J* = 6.5 Hz, 3H), 0.75 (d, *J* = 6.6 Hz, 3H), 1.26 (dd, *J* = 7.0, 7.0 Hz, 2H), 1.36-1.46 (m, 1H), 2.41 (s, 3H), 2.44 (s, 3H), 3.36-3.46 (m, 1H), 3.83 (dd, *J* = 4.8, 9.9 Hz, 1H), 3.94 (dd, *J* = 3.5, 10.0 Hz, 1H), 5.01 (d, *J* = 8.4 Hz, 1H), 7.26 (d, *J* = 8.5 Hz, 2H), 7.33 (d, *J* = 8.5 Hz, 2H), 7.70 (d, *J* = 8.0 Hz, 2H), 7.72 (d, *J* = 8.2 Hz, 2H), ¹³C NMR δ 21.9 (2 x CH₃), 22.0 (CH₃), 23.1 (CH₃), 24.4 (CH), 41.2 (CH₂), 51.0 (CH), 72.0 (CH₂), 127.4 (2 x CH), 128.4 (2 x CH), 130.1 (2 x CH), 130.3 (2 x CH), 132.9 (C), 138.0 (C), 143.9

¹⁷ Spectral data are in good agreement with those reported in the literature: Craven, A. P.; Dyke, H. J.; Thomas, E. J. *Tetrahedron* **1989**, *45*, 2417.

(C), 145.5 (C); MS (ESI+) m/z (rel intensity) 448 (M⁺+ Na, 100); HRMS calcd for C₂₀H₂₇NO₅NaS₂, 448.1228, found 448.1224.



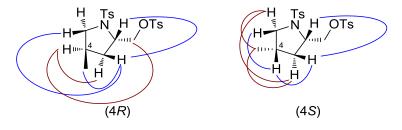
(2S)-4-Methyl-1-tosylpyrrolidin-2-yl)methyl 4-methylbenzenesulfonate (9a): Colorless oil. NMR showed a mixture of isomers (*anti/syn*)-9a in 1.7:1 ratio. *Anti*-9a: ¹H NMR $\delta 0.84$ (d, J = 6.6 Hz, 3H), 1.20 (ddd, J = 8.8, 11.0, 12.9 Hz, 1H), 1.98 (dd, J = 6.3, 13.2

Hz, 1H), 2.26-2.35 (m, 1H), 2.44 (s, 3H), 2.47 (s, 3H), 2.49 (dd, J = 9.5, 9.5 Hz, 1H), 3.49 (dd, J = 6.6, 9.1 Hz, 1H), 3.75-3.78 (m, 1H), 3.97 (dd, J = 8.2, 10.1 Hz, 1H), 4.23 (dd, J = 3.5, 9.8 Hz, 1H), 7.31 (d, J = 7.9 Hz, 2H), 7.37 (d, J = 7.9 Hz, 2H), 7.65 (d, J = 8.2 Hz, 2H) 7.82 (d, J = 8.2 Hz, 2H), ¹³C NMR δ 16.7 (CH₃), 21.6 (CH₃), 21.7 (CH₃), 31.4 (CH), 36.4 (CH₂), 55.9 (CH₂), 57.9 (CH), 71.5 (CH₂), 127.7 (2 x CH), 128.1 (2 x CH), 129.8 (2 x CH), 130.0 (2 x CH), 132.7 (C), 133.3 (C), 143.9 (C), 145.1 (C).

*Syn-9*a: ¹H NMR δ 0.90 (d, *J* = 6.6 Hz, 3H), 1.42 (ddd, *J* = 8.5, 11.0, 12.6 Hz, 1H), 1.48-1.58 (m, 1H), 2.10 (ddd, *J* = 6.5, 6.5, 12.5 Hz, 1H), 2.43 (s, 3H), 2.47 (s, 3H), 2.83 (dd, *J* = 10.4, 11.4 Hz, 1H), 3.47-3.49 (m, 1H), 3.70-3.76 (m, 1H), 4.04 (dd, *J* = 7.6, 9.8 Hz, 1H), 4.41 (dd, *J* = 3.8, 9.8 Hz, 1H), 7.31 (d, *J* = 7.9 Hz, 2H), 7.37 (d, *J* = 7.9 Hz, 2H), 7.65 (d, *J* = 8.2 Hz, 2H) 7.82 (d, *J* = 8.2 Hz, 2H); ¹³C NMR δ 16.4 (CH₃), 21.6 (CH₃), 21.7 (CH₃), 32.7 (CH), 37.8 (CH₂), 56.3 (CH₂), 58.6 (CH), 72.6 (CH₂), 127.6 (2 x CH), 128.1 (2 x CH), 129.9 (2 x CH), 130.0 (2 x CH), 132.8 (C), 134.4 (C), 143.8 (C), 145.0 (C).

MS (ESI+) m/z (rel intensity) 446 (M⁺+ Na, 100); HRMS calcd for C₂₀H₂₅NNaO₅S₂, 446.1072, found 446.1062.

Main correlations observed by NOESY are shown in the figure:



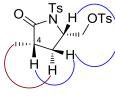
O TS OTS

[(2*S*,4*S*)-4-Methyl-5-oxo-1-tosylpyrrolidin-2-yl]methyl 4-methylbenzenesulfonate (*syn*-9c): Crystalline solid: Mp 172.5-174.0 °C (hexane/CH₂Cl₂); $[\alpha]_D$ -18.9 (*c* 3.0); ¹H NMR (500 MHz) δ 1.16 (d, *J* = 6.6 Hz, 3H), 1.61-1.68 (m, 2H), 2.44 (s, 3H), 2.40-2.50

(m, 1H), 2.47 (s, 3H), 4.36-4.42 (m, 3H), 7.30 (d, J = 8.2 Hz, 2H), 7.37 (d, J = 8.2 Hz, 2H), 7.76 (d, J = 8.2 Hz, 2H) 7.86 (d, J = 8.2 Hz, 2H); ¹H NMR (500 MHz, C₆D₆) δ 0.76 (d, J = 7.3 Hz, 3H), 1.10 (ddd, J = 6.3, 8.2, 13.2 Hz, 1H), 1.52 (ddd, J = 8.2, 10.2, 13.2 Hz, 1H), 1.64-1.74 (m, 1H), 1.80 (s, 3H), 1.83 (s, 3H), 3.92-4.00 (m, 1H), 4.23 (dd, J = 2.7, 10.3 Hz, 1H), 4.40 (dd, J = 5.4, 10.4 Hz, 1H), 6.72 (d, J = 8.2 Hz, 2H), 6.77 (d, J = 7.9 Hz, 2H), 7.70 (d, J = 8.2 Hz, 2H) 8.07 (d, J = 8.5 Hz, 2H); ¹³C NMR (125 MHz) δ 16.0 (CH₃), 21.7 (2 x CH₃), 29.2 (CH₂), 36.4 (CH), 55.9 (CH), 70.0 (CH₂), 128.0 (2 x CH), 128.4 (2 x CH), 129.6 (2 x CH))

CH), 130.1 (2 x CH), 132.3 (C), 135.1 (C), 145.4 (2 x C), 176.5 (C); MS (ESI+) m/z (rel intensity) 460 (M⁺+ Na, 100); HRMS calcd for $C_{20}H_{23}NNaO_6S_2$, 460.0865, found 460.0873.

Main correlations observed by NOESY (in C_6D_6) are shown in the figure:

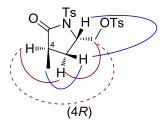






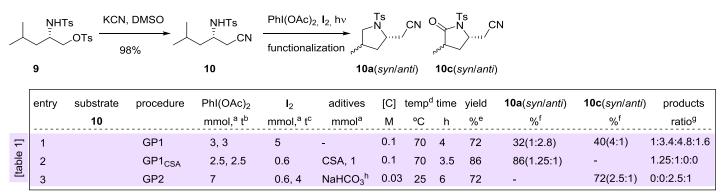
[(2*S*,4*R*)-4-Methyl-5-oxo-1-tosylpyrrolidin-2-yl]methyl 4-methylbenzenesulfonate (*anti*-9c): Crystalline solid: Mp 163.0-164.5 °C (hexane/CH₂Cl₂); $[\alpha]_D$ -9.6 (c 1.6); ¹H NMR δ 1.08 (d, J = 6.9 Hz, 3H), 1.85 (ddd, J = 9.0, 12.4, 12.4 Hz, 1H), 2.28 (dd, J = 8.8, 12.3 Hz, 1H), 2.44 (s, 3H), 2.47 (s, 3H), 2.71-2.75 (m, 1H), 4.26 (dd, J = 2.5, 10.4 Hz, 1H), 4.38 (dd, J = 2.5, 10.4 Hz, 1H), 4.5 4.4, 10.4 Hz, 1H), 4.41-4.48 (m, 1H), 7.30 (d, J = 8.5 Hz, 2H), 7.37 (d, J = 8.2 Hz, 2H), 7.74 (d, J = 8.2 Hz, 2H) 7.86 (d, J = 8.5 Hz, 2H); ¹³C NMR δ 15.2 (CH₃), 21.7 (CH₃), 22.0 (CH₃), 31.2 (CH₂), 36.1 (CH), 55.4 (CH), 70.2 (CH₂), 127.9 (2 x CH), 128.4 (2 x CH), 129.6 (2 x CH), 130.1 (2 x CH), 132.2 (C), 135.2 (C), 145.4 (C), 145.4 (C), 175.8 (C); MS (ESI+) m/z (rel intensity) 460 (M⁺+ Na, 100); HRMS calcd for C₂₀H₂₃NNaO₆S₂, 460.0865, found 460.0869.

Main correlations observed by NOESY are shown in the figure:



Synthesis and Functionalization of Tosylamine 10:

Table S10.



^a millimoles per millimol of **10**; ^b Time of addition in hours: Portionwise (0.25 equiv / 15 min); ^c time of continous addition of **I**₂ disolved in DCE (0.15M) by syringe pump (in hours); ^d 25 °C (reaction mixture refrigerated by a fan) and 65-75 °C (without fan); ^e total isolated yield; ^f isolated yield; ^g (10a-syn/10a-anti/10c-syn/10c-anti) calculated by integration in the ¹H NMR of the crude of reaction; ^h 100% w/w.

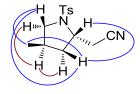


(S)-N-(1-Cyano-4-methylpentan-2-yl)-4-methylbenzenesulfonamide (10): KCN (217

^{CN} mg, 3.34 mmol) was added to a solution of **10** (710 mg, 1.67 mmol) in DMSO (4 mL) and stirred at rt for 5 h. The reaction mixture was poured into brine and extracted with EtOAc. The organic phase was dried with Na₂SO₄, filtrated and concentrated. The residue was purified by column chromatography (hexanes/EtOAc, 60:40) to afford the product **10** (435 mg, 1.64 mmol, 98%) as a crystalline solid: Mp 83.0-83.5 °C (hexane/Et₂O); $[\alpha]_D$ -75.5 (*c* 1.6); ¹H NMR δ 0.61 (d, *J* = 6.4 Hz, 3H), 0.82 (d, *J* = 6.4 Hz, 3H), 1.30-1.40 (m, 1H), 1.42-1.52 (m, 2H), 2.44 (s, 3H), 2.51 (dd, *J* = 3.5, 16.7 Hz, 1H), 2.63 (dd, *J* = 6.1, 16.7 Hz, 1H), 3.45-3.54 (m, 1H), 5.03 (d, *J* = 7.9 Hz, 1H), 7.33 (d, *J* = 8.0 Hz, 2H), 7.78 (d, *J* = 8.4 Hz, 2H); ¹³C NMR δ 21.6 (CH₃), 21.9 (CH₃), 23.1 (CH₃), 24.6 (CH), 25.7 (CH₂), 43.4 (CH₂), 48.7 (CH), 117.5 (C), 127.5 (2 x CH), 130.3 (2 x CH), 137.7 (C), 144.3 (C); MS (ESI+) *m/z* (rel intensity) 303 (M⁺+ Na, 100); HRMS calcd for C₁₄H₂₀N₂O₂NaS, 303.1143, found 303.1145.

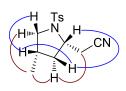
^{Ts} N CN N 2-[(2S,4R)-4-Methyl-1-tosylpyrrolidin-2-yl]acetonitrile (*anti*-10a): Colorless oil: $[\alpha]_D$ -72.7 (*c* 1.0); ¹H NMR δ 0.87 (d, *J* = 6.4 Hz, 3H), 1.47 (ddd, *J* = 8.6, 10.3, 13.1 Hz, 1H), 1.99 (ddd, *J* = 2.5, 6.2, 13.1 Hz, 1H), 2.45 (s, 3H), 2.45-2.52 (m, 1H), 2.61 (dd, *J* = 9.1, 9.1 Hz, 1H), 2.77 (dd, *J* = 8.1, 16.8 Hz, 1H), 2.88 (dd, *J* = 3.7, 16.7 Hz, 1H), 3.60 (dd, *J* = 6.4, 9.0 Hz, 1H), 3.85 (dddd, *J* = 3.1, 3.1, 8.4, 8.4, 1H), 7.36 (d, *J* = 8.5 Hz, 2H), 7.73 (d, *J* = 8.2 Hz, 2H); ¹³C NMR δ 17.1 (CH₃), 21.9 (CH₃), 25.8 (CH₂), 31.7 (CH), 39.3 (CH₂), 56.3 (CH), 56.5 (CH₂), 117.9 (C), 128.1 (2 x CH), 130.3 (2 x CH), 133.9 (C), 144.5 (C); MS (ESI+) *m*/z (rel intensity) 301 (M⁺+ Na, 100); HRMS calcd for C₁₄H₁₈N₂O₂NaS, 301.0987, found 301.0989.

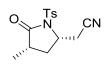
Main correlations observed by NOESY are shown in the figure:



^{Ts} CN ^{CN} 2-[(2*S*,4*S*)-4-Methyl-1-tosylpyrrolidin-2-yl]acetonitrile (*syn*-10a): Colorless oil: $[\alpha]_D$ - 88.5 (*c*, 1.0); ¹H NMR δ 0.96 (d, *J* = 6.4 Hz, 3H), 1.49 (ddd, *J* = 9.3, 11.6, 12.5 Hz, 1H), 1.59-1.69 (m, 1H), 2.25 (ddd, *J* = 6.1, 6.1, 12.2 Hz, 1H), 2.45 (s, 3H), 2.89 (dd, *J* = 6.6, 16.7 Hz, 1H), 2.95 (dd, *J* = 4.1, 16.8 Hz, 1H), 2.97 (d, *J* = 10.6 Hz, 1H), 3.56 (ddd, *J* = 1.3, 7.2, 11.4 Hz, 1H), 3.79 (dddd, *J* = 4.1, 6.8, 6.8, 9.2, 1H), 7.34 (d, *J* = 8.0 Hz, 2H), 7.73 (d, *J* = 8.2 Hz, 2H); ¹³C NMR δ 16.6 (CH₃), 21.9 (CH₃), 25.7 (CH₂), 33.2 (CH), 40.6 (CH₂), 56.6 (CH₂), 57.1 (CH), 117.6 (C), 127.9 (2 x CH), 130.3 (2 x CH), 135.0 (C), 144.4 (C); MS (ESI+) *m/z* (rel intensity) 301 (M⁺+ Na, 100); HRMS calcd for C₁₄H₁₈N₂O₂NaS, 301.0987, found 301.0989.

Main correlations observed by NOESY are shown in the figure:



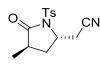


2-((2*S***,4***S***)-4-Methyl-5-oxo-1-tosylpyrrolidin-2-yl)acetonitrile (***syn***-10c): Crystalline solid: Mp 137.5-139.0 °C (hexane/Et₂O); [\alpha]_D -72.6 (***c* **1.0); ¹H NMR \delta 1.23 (d,** *J* **= 7.0 Hz, 3H), 1.64 (ddd,** *J* **= 8.0, 9.8, 13.0 Hz, 1H), 2.45 (s, 3H), 2.48-2.56 (m, 1H), 2.65 (ddd,** *J* **=**

7.5, 9.7, 13.0 Hz, 1H), 3.12 (dd, J = 3.7, 16.6 Hz, 1H), 3.18 (dd, J = 6.3, 16.6 Hz, 1H), 4.39 (dddd, J = 3.9, 6.4, 7.7, 7.7 Hz, 1H), 7.36 (d, J = 8.0 Hz, 2H), 7.95 (d, J = 8.3 Hz, 2H); ¹³C NMR δ 15.8 (CH₃), 22.1 (CH₃), 25.6 (CH₂), 33.0 (CH₂), 37.0 (CH), 54.0 (CH), 116.3 (C), 128.9 (2 x CH), 130.1 (2 x CH), 135.6 (C), 146.1 (C), 176.5 (C); MS (ESI+) m/z (rel intensity) 315 (M⁺+ Na, 100); HRMS calcd for C₁₄H₁₆N₂O₃NaS, 315.0779, found 315.0771.

Main correlations observed by NOESY are shown in the figure:





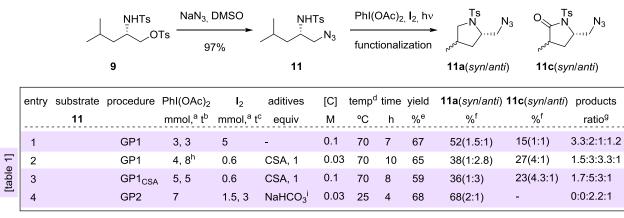
2-[(2S,4R)-4-Methyl-5-oxo-1-tosylpyrrolidin-2-yl]acetonitrile (*anti*-10c): Crystalline solid: Mp 123.5-124.0 °C (hexane/Et₂O); $[\alpha]_D$ -46.8 (*c* 0.4); ¹H NMR δ 1.14 (d, *J* = 6.9 Hz, 3H), 2.01 (ddd, *J* = 8.7, 11.6, 13.5 Hz, 1H), 2.33 (ddd, *J* = 1.3, 8.7, 13.4 Hz, 1H), 2.45 (s,

3H), 2.82-2.92 (m, 1H), 2.98 (d, J = 5.5 Hz, 2H), 4.52 (dddd, J = 1.3, 5.0, 6.1, 8.8 Hz, 1H), 7.36 (d, J = 8.8 Hz, 2H), 7.93 (d, J = 8.5 Hz, 2H); ¹³C NMR δ 15.3 (CH₃), 22.1 (CH₃), 24.2 (CH₂), 33.3 (CH₂), 35.9 (CH), 53.6 (CH), 116.8 (C), 128.7 (2 x CH), 130.2 (2 x CH), 135.4 (C), 146.2 (C), 175.4 (C); MS (ESI+) m/z (rel intensity) 315 (M⁺+ Na, 100); HRMS calcd for C₁₄H₁₆N₂O₃NaS, 315.0779, found 315.0771. Main correlations observed by NOESY are shown in the figure:



Synthesis and Functionalization of Tosylamine 11:

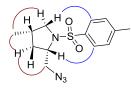




^a millimoles per millimol of **11**; ^b Time of addition in hours: Portionwise (0.25 equiv / 15 min); ^c time of continous addition of **I**₂ disolved in DCE (0.15M) by syringe pump (in hopurs); ^d 25 °C (reaction mixture refrigerated by a fan) and 65-75 °C (without fan); ^e total isolated yield; ^f isolated yield; ^g (**11a**-*syn*/**11a**-*anti*/**11c**-*syn*/**11c**-*anti*) calculated by integration in the ¹H NMR of the crude of reaction; ^h Portionwise (0.25 equiv / 30 min); ⁱ 100% w/w.

NHTs (*S*)-*N*-(1-Azido-4-methylpentan-2-yl)-4-methylbenzenesulfonamide (11): NaN₃ (0.59 g, 9.16 mmol) was added to a solution of **9** (973 mg, 2.29 mmol) in DMSO (13 mL) and stirred at rt for 24 h. The reaction mixture was poured into brine and extracted with EtOAc. The organic phase was dried with Na₂SO₄, filtrated and concentrated. The residue was purified by column chromatography (hexanes/EtOAc, 60:40) to afford the product **11** (658 mg, 2.22 mmol, 97%) as a crystalline solid: Mp 59.0-60.3 °C (hexane/Et₂O); $[\alpha]_D$ -45.4 (*c* 1.1); ¹H NMR δ 0.60 (d, *J* = 6.4 Hz, 3H), 0.73 (d, *J* = 6.6 Hz, 3H), 1.13-1.28 (m, 2H), 1.37-1.47 (m, 1H), 2.35 (s, 3H), 3.18 (dd, *J* = 3.9, 12.2 Hz, 1H), 3.22 (dd, *J* = 4.8, 12.5 Hz, 1H), 3.26-3.35 (m, 1H), 5.01 (br s, 1H), 7.24 (d, *J* = 8.0 Hz, 2H), 7.72 (d, *J* = 8.2 Hz, 2H); ¹³C NMR δ 20.5 (CH₃), 20.7 (CH₃), 21.8 (CH₃), 23.3 (CH), 40.9 (CH₂), 50.5 (CH), 54.3 (CH₂), 126.1 (2 x CH), 128.7 (2 x CH), 136.9 (C), 142.6 (C); MS (ESI+) *m/z* (rel intensity) 319 (M⁺+ Na, 100); HRMS calcd for C₁₃H₂₀N₄O₂NaS, 319.1205, found 319.1210.

Ts N_3 (2*S*,4*S*)-2-(Azidomethyl)-4-methyl-1-tosylpyrrolidine (*syn*-11a): *syn*-11a was isolated from a mixture (*syn/anti*)-11a (1.5:1 ratio) by HPLC (hexanes / EtOAc, 90:10) as a colorless oil: $[\alpha]_D$ -32.0 (*c* 0.2); ¹H NMR (500 MHz) δ 0.94 (d, *J* = 6.6 Hz, 3H), 1.43 (ddd, *J* = 8.8, 11.4, 12.6 Hz, 1H), 1.51-1.63 (m, 1H), 2.05 (ddd, *J* = 1.2, 7.2, 12.6 Hz, 1H), 2.44 (s, 3H), 2.88 (dd, *J* = 10.7, 11.3 Hz, 1H), 3.57 (dd, *J* = 3.2, 12.3 Hz, 1H), 3.58 (dd, *J* = 1.3, 11.3 Hz, 1H), 3.64 (dd, *J* = 6.3, 12.3 Hz, 1H), 3.70-3.75 (m, 1H), 7.32 (d, *J* = 8.0 Hz, 2H), 7.72 (d, *J* = 8.2 Hz, 2H); ¹³C NMR (125 MHz) δ 16.4 (CH₃), 21.6 (CH₃), 32.9 (CH), 38.2 (CH₂), 55.5 (CH₂), 56.3 (CH₂), 60.0 (CH), 127.5 (2 x CH), 129.9 (2 x CH), 134.9 (C), 143.7 (C); MS (ESI+) *m/z* (rel intensity) 317 (M⁺+ Na, 100); HRMS calcd for C₁₃H₁₈N₄O₂NaS, 317.1048, found 317.1041. Main correlations observed by NOESY are shown in the figure:





(*2S*,*4R*)-2-(Azidomethyl)-4-methyl-1-tosylpyrrolidine (*anti*-11a): [data obtained from the mixture (*anti/syn*)-11a; 1.0:1.1]: Colorless oil; ¹H NMR (500 MHz) δ0.86 (d, *J* = 6.6 Hz, 3H), 1.22-1.32 (m, 1H), 1.91 (ddd, *J* = 2.2, 6.3, 12.9 Hz, 1H), 2.32-2.41 (m, 1H), 2.45 (s,

3H), 2.58 (dd, J = 9.3, 9.3 Hz, 1H), 3.48 (dd, J = 7.9, 12.3 Hz, 1H), 3.55-3.60 (m, 2H), 3.68-3.73 (m, 1H), 7.34 (d, J = 8.0 Hz, 2H), 7.73 (d, J = 8.2 Hz, 2H); ¹³C NMR (125 MHz) δ 16.9 (CH₃), 21.6 (CH₃), 31.6

(CH), 37.3 (CH₂), 55.4 (CH₂), 56.1 (CH₂), 58.9 (CH), 127.7 (2 x CH), 129.8 (2 x CH), 133.7 (C), 143.8 (C). $\sim \sum_{N=1}^{N} \sum_{N_3}^{N_3} [\alpha]_{\rm D} -68.0 (c \ 0.5); {}^{1}{\rm H} \text{ NMR} (500 \text{ MHz}) \delta 1.12 (d, J = 7.3 \text{ Hz}, 3\text{H}), 1.86 (ddd, J = 8.8, 11.6, 1.6)$

13.0 Hz, 1H), 2.20 (ddd, J = 1.0, 9.0, 12.8 Hz, 1H), 2.44 (s, 3H), 2.79 (ddq, J = 7.1 x 3, 8.7, 11.6 Hz, 1H), 3.68 (dd, J = 2.8, 12.6 Hz, 2H), 3.84 (dd, J = 5.4, 12.9 Hz, 1H), 4.40 (dddd, J = 1.0, 2.8, 5.4, 8.8 Hz, 1H), 7.34 (d, J = 7.9 Hz, 2H), 7.95 (d, J = 8.4 Hz, 2H); ¹³C NMR δ 15.4 (CH₃), 21.7 (CH₃), 31.9 (CH₂), 36.2 (CH), 54.3 (CH), 56.1 (CH₂), 128.4 (2 x CH), 129.7 (2 x CH), 135.5 (C), 145.4 (C), 175.9 (C); MS (ESI+) m/z (rel intensity) 331 (M⁺+ Na, 100); HRMS calcd for C₁₃H₁₆N₄O₃NaS, 331.0841, found 331.0836.

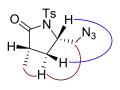
Main correlations observed by NOESY are shown in the figure:



(3*S*,5*S*)-5-(Azidomethyl)-3-methyl-1-tosylpyrrolidin-2-one (*syn*-11c): Crystalline solid: Mp 98-99 °C (hexane/Et₂O); $[\alpha]_D$ -8.4 (*c* 1.0); ¹H NMR (500 MHz) δ 1.21 (d, *J* = 6.9 Hz, 3H), 1.60 (ddd, *J* = 6.0, 7.3, 13.9 Hz, 1H), 2.43 (ddd, *J* = 8.2, 10.4, 13.0 Hz, 1H), 2.44 (s,

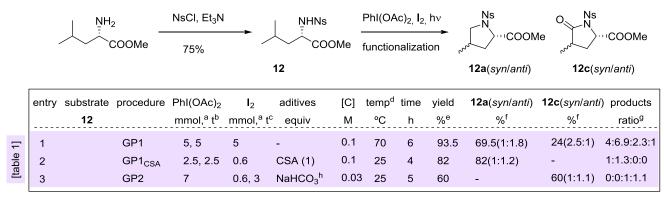
3H), 2.48 (ddq, J = 6.9 x 4, 10.1 Hz, 1H), 3.75 (dd, J = 2.5, 12.6 Hz, 1H), 3.94 (dd, J = 5.7, 12.6 Hz, 1H), 4.33 (dddd, J = 2.8, 5.7, 6.9, 8.2 Hz, 1H), 7.34 (d, J = 8.2 Hz, 2H), 7.94 (d, J = 8.2 Hz, 2H); ¹³C NMR δ 16.3 (CH₃), 22.1 (CH₃), 30.4 (CH₂), 36.9 (CH), 54.8 (CH₂), 57.1 (CH), 128.7 (2 x CH), 130.0 (2 x CH), 135.9 (C), 145.7 (C), 176.9 (C); MS (ESI+) m/z (rel intensity) 331 (M⁺+ Na, 100); HRMS calcd for C₁₃H₁₆N₄O₃NaS, 331.0841, found 331.0840.

Main correlations observed by NOESY are shown in the figure:



Synthesis and Functionalization of 4-nitrophenylsulfonamine 12:

Table S12.



^a millimoles per millimol of **12**; ^b Time of addition in hours: Portionwise (0.25 equiv / 15 min); ^c time of continous addition of **I**₂ disolved in DCE (0.15M) by syringe pump (in hours); ^d 25 °C (reaction mixture refrigerated by a fan) and 65-75 °C (without fan); ^e total isolated yield; ^f isolated yield; ^g (**12a**-*syn*/**12a**-*anti*/**12c**-*syn*/**12c**-*anti*) calculated by integration in the ¹H NMR of the crude of reaction; ^h 100% w/w.

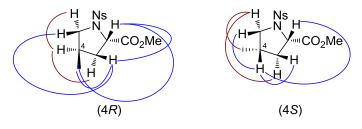
NHNs (*S*)-Methyl 4-methyl-2-(4-nitrophenylsulfonamido)pentanoate (12):¹⁸ Triethylamine CO₂Me (2 mL) was added to a suspension of L-leucine methyl ester hydrochloride (1.04 g, 5.74 mmol) in dry CH₂Cl₂ (7 mL) at room temperature and stirred for 30 min. The resulting mixture was cold at 0 °C and *p*-nitrobenzenesulfonyl chloride (1.91 g, 8.60 mmol) was added in one portion, allowed to reach rt and stirred for further 3-4 h. The reaction was poured into ice–water and extracted with CH₂Cl₂. The organic extracts were dried over Na₂SO₄ and concentrated under vacuum. Column chromatography (hexanes– EtOAc, 85:15) of the residue afforded compound **13** (1.42 g, 4.31 mmol, 75%) as a pale yellow solid: Mp 97.5-99.0 °C (*n*-hexane/Et₂O); $[\alpha]_D$ +27.4 (*c* 1.5); ¹H NMR δ 0.90 (d, *J* = 6.9 Hz, 3H), 0.92 (d, *J* = 7.0 Hz, 3H), 1.54 (t, *J* = 6.9 Hz, 2H), 1.70-1.82 (m, 1H), 3.51 (s, 3H), 4.03 (ddd, *J* = 6.8, 7.9, 9.8 Hz, 1H), 5.45 (br s, 1H), 8.05 (d, *J* = 8.8 Hz, 2H), 8.35 (d, *J* = 8.8 Hz, 2H); ¹³C NMR δ 21.7 (CH₃), 23.0 (CH₃), 24.8 (CH), 42.6 (CH₂), 52.9 (CH₃), 55.0 (CH), 124.6 (2 x CH), 128.9 (2 x CH), 146.2 (C), 150.5 (C), 172.7 (C); MS (ESI+) *m/z* (rel intensity) 353 (M⁺+ Na, 100); HRMS calcd for C₁₃H₁₈N₂O₆NaS, 353.0783, found 353.0783.

Ns (2S)-Methyl 4-methyl-1-((4-nitrophenyl)sulfonyl)pyrrolidine-2-carboxylate (12a): Colorless oil. NMR showed a mixture of isomer (*anti/syn*)-12a in 1.8:1.0 ratio. *anti*-12a: ¹H NMR (500 MHz) δ 0.99 (d, J = 6.9 Hz, 3H), 1.80 (ddd, J = 9.6, 9.6, 12.9 Hz, 1H), 2.14 (ddd, J = 2.8, 6.2, 13.0 Hz, 1H), 2.41-2.49 (m, 1H), 2.96 (dd, J = 8.8, 8.8 Hz, 1H), 3.57 (dd, J = 7.1, 9.0 Hz, 1H), 3.71 (s, 3H), 4.49 (dd, J = 2.8, 9.1 Hz, 1H), 8.08 (d, J = 8.8 Hz, 2H), 8.38 (d, J = 8.8 Hz, 2H); ¹³C NMR δ 17.2 (CH₃), 33.0 (CH), 38.9 (CH₂), 52.8 (CH₃), 55.1 (CH₂), 61.0 (CH), 124.5 (2 x CH), 129.1 (2 x CH), 145.0 (C), 150.5 (C), 172.5 (C).

¹⁸ Spectral data are in good agreement with those reported in the literature: (a) Reichwein, J. F.; Liskamp, R. M. J. *Eur. J. Org. Chem.* **2000**, 2335; (b) Gioia, M. L. D.; Leggio, A.; Pera, A. L.; Liguori, A.; Napoli, A.; Siciliano, C.; Sindona, G. *J. Org. Chem.* **2003**, 68, 7416.

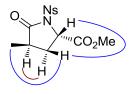
syn-12a: ¹H NMR (500 MHz) δ 1.03 (d, J = 6.6 Hz, 3H), 1.62 (ddd, J = 8.7, 10.2, 12.6 Hz, 1H), 2.14-2.24 (m, 1H), 2.40-2.47 (m 1H), 2.94 (dd, J = 9.8, 9.8 Hz, 1H), 3.68-3.75 (m, 1H), 3.74 (s, 3H), 4.43 (dd, J = 8.3, 8.3 Hz, 1H), 8.10 (d, J = 8.8 Hz, 2H), 8.38 (d, J = 8.8 Hz, 2H); ¹³C NMR δ 17.2 (CH₃), 34.2 (CH), 39.2 (CH₂), 52.8 (CH₃), 55.5 (CH₂), 61.5 (CH), 124.5 (2 x CH), 129.1 (2 x CH), 145.3 (C), 150.5 (C), 172.7 (C). MS (ESI+) m/z (rel intensity) 351 (M⁺+ Na, 100); HRMS calcd for C₁₃H₁₆N₂O₆NaS, 351.0627, found 351.0621.

Main correlations observed by NOESY are shown in the figure:



(m, 1H), 3.82 (s, 3H), 4.91 (d, J = 9.4 Hz, 1H), 8.31 (d, J = 8.8 Hz, 2H), 8.38 (d, J = 8.8 Hz, 2H); ¹³C NMR δ 14.8 (CH₃), 32.6 (CH₂), 36.5 (CH), 53.4 (CH₃), 57.9 (CH), 124.0 (2 x CH), 131.2 (2 x CH), 143.5 (C), 151.4 (C), 171.3 (C), 175.3 (C); MS (ESI+) m/z (rel intensity) 365 (M⁺+ Na, 100); HRMS calcd for C₁₃H₁₄N₂O₇NaS, 365.0419, found 365.0420.

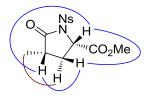
Main correlations observed by NOESY are shown in the figure:



Ns (2*S*,4*S*)-Methyl 4-methyl-1-((4-nitrophenyl)sulfonyl)-5-oxopyrrolidine-2carboxylate (*syn*-12c): Crystalline solid: Mp 146.0-146.5 °C (hexane/Et₂O); $[\alpha]_D$ -17.2 (*c* 0.8); ¹H NMR δ 1.17 (*d*, *J* = 6.9 Hz, 3H), 1.73 (ddd, *J* = 6.8, 6.8, 11.8 Hz, 1H), 2.65-

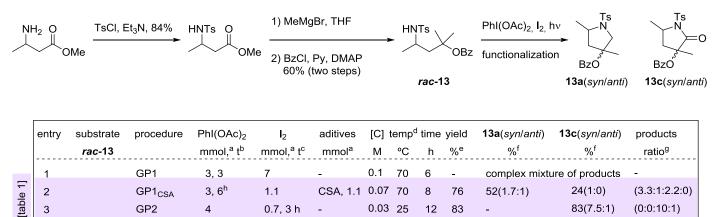
2.76 (m, 2H), 3.85 (s, 3H), 4.84 (dd, J = 7.5, 7.5 Hz, 1H), 8.35 (d, J = 8.8 Hz, 2H), 8.39 (d, J = 8.8 Hz, 2H); ¹³C NMR δ 16.0 (CH₃), 31.5 (CH₂), 37.4 (CH), 53.4 (CH₃), 58.3 (CH), 124.1 (2 x CH), 131.1 (2 x CH), 143.8 (C), 151.4 (C), 171.9 (C), 175.8 (C); MS (ESI+) m/z (rel intensity) 365 (M⁺+ Na, 100); HRMS calcd for C₁₃H₁₄N₂O₇NaS, 365.0419, found 365.0422.

Main correlations observed by NOESY are shown in the figure:



Synthesis and Functionalization of Tosylamine 13:





^a millimoles per millimol of **13**; ^b Time of addition in hours: Portionwise (0.25 equiv / 15 min); ^c time of continous addition of I₂ disolved in DCE (0.15M) by syringe pump (in hours); ^d 25 °C (reaction mixture refrigerated by a fan) and 65-75 °C (without fan); ^e total isolated yield; ^f isolated yield; ^g (13a-syn/13a-anti/13c-syn/13c-anti) calculated by integration in the ¹H NMR of the crude of reaction; ^h portionwise (0.5 equiv / 60 min).



Ethyl 3-(4-methylphenylsulfonamido)butanoate:¹⁹ TsCl (570 mg, 3 mmol) was added portionwise to a suspension of 3-amino-butyric acid ethyl ester (175 mg, 1.5 mmol) and OEt distilled Et₃N (0.56 mL, 4 mmol) in dry DCM (15 mL). The resulting mixture was stirred at room temperature for 4 h. 1 N Aqueous HCl was then added and the acidified solution (pH = 2) was extracted with DCM. The organic phase was carefully washed with a saturated aqueous NaHCO₃, brine and finally dried with Na₂SO₄ and the solvent was evaporated. The crude material was purified by column chromatography (hexane/EtOAc 70:30) to give the desired product as colorless oil (341 mg, 84% yield). ¹H NMR δ 1.15 (d, J = 6.9 Hz, 3H), 1.22 (dd, J = 7.2, 7.2 Hz, 3H), 2.41 (d, J = 6.9 Hz, 2H), 2.42 (s, 3H), 3.65-3.73 (m, 1H), 4.06 (dd, J = 4.5, 7.1 Hz, 1H), 4.09 (dd, J = 4.5, 7.1 Hz, 1H), 5.17 (d, J = 8.2 Hz, 1H), 7.30 (d, J = 4.5, 7.1 Hz, 1H), 5.17 (d, J = 8.2 Hz, 1H), 7.30 (d, J = 4.5, 7.1 Hz, 1H), 5.17 (d, J = 8.2 Hz, 1H), 7.30 (d, J = 4.5, 7.1 Hz, 1Hz, 1Hz, 1Hz), 7.30 (d, J = 4.5, 7.1 Hz, 1Hz), 7.30 (d, J = 4.5, 7.1 Hz, 1Hz), 7.30 (d, J = 4.5, 7.1 Hz), 7.30 (d, J = 4(d, J = 8.5 Hz, 2H), 7.76 (d, J = 8.5 Hz, 2H); ¹³C NMR δ 14.4 (CH₃), 21.4 (CH₃), 21.8 (CH₃), 41.5 (CH₂), 47.1 (CH), 61.0 (CH₂), 127.4 (2 x CH), 130.0 (2 x CH), 138.5 (C), 143.6 (C), 171.5 (C).

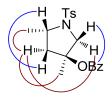
NHTs 2-Methyl-4-(4-methylphenylsulfonamido)pentan-2-yl benzoate (rac-13): A solution of the previously prepared N-tosyl derivative (100 mg, 0.29 mmol) in THF (1 mL) was added to a ÒB7 slurry of MeMgCl in THF (ca. 2 M, 0.72 mL, 5 equiv) at 0 °C. The resulting mixture was rac stirred for 4 h at 0 °C and poured into ice-cooled aqueous NH₄Cl. The phases were separated and the aqueous phase was extracted with EtOAc twice. The combined organic phase was dried over Na₂SO₄ and filtered. To the concentrated crude residue was added DCM (1 mL), BzCl (203 mg, 0.17 mL, 1.45 mmol) and DMAP (88 mg, 0.86 mmol) and stirred for 16 h at room temperature. The reaction was quenched by the addition of aqueous HCl solution (10 %) and diluted with EtOAc. The layers were separated and the aqueous phase was extracted with more EtOAc. The combined organic layer was washed with aqueous HCl

¹⁹ Spectral data are in good agreement with those reported in the literature: Jefford, C. W.; McNulty, J.; Lu, Z-H.; Wang, J. B. Helv. Chim. Acta 1996, 79, 1203.

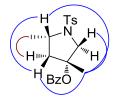
(10%), saturated aqueous NaHCO₃ solution and brine, dried over anhydrous Na₂SO₄, and the solvent was evaporated under vacuum. The residue was purified by column chromatography (hexane/EtOAc 70:30) to give the desired benzoyl derivative as colorless oil (60 mg, 60% after 2 steps). ¹H NMR δ 1.09 (d, *J* = 6.4, Hz, 3H), 1.50 (s, 3H), 1.55 (s, 3H), 2.01 (dd, *J* = 5.3, 14.8 Hz, 1H), 2.10 (dd, *J* = 7.4, 14.8 Hz, 1H), 2.36 (s, 3H), 3.57-3.67 (m, 1H), 5.32 (d, *J* = 7.7 Hz, 1H), 7.19 (d, *J* = 7.9 Hz, 2H), 7.42 (dd, *J* = 7.7, 7.7 Hz, 2H), 7.68 (d, *J* = 8.2 Hz, 2H), 7.95 (d, *J* = 7.7 Hz, 2H), 8.11 (d, *J* = 7.9 Hz, 1H); ¹³C NMR δ 21.8 (CH₃), 23.5 (CH₃), 26.7 (CH₃), 26.9 (CH₃), 47.4 (CH), 48.3 (CH₂), 82.8 (C), 127.4 (2 x CH), 128.7 (2 x CH), 129.8 (2 x CH), 130.0 (2 x CH), 131.9 (C), 133.1 (CH), 138.6 (C), 143.6 (C), 166.2 (C); MS (ESI+) *m/z* (rel intensity) 398 (M⁺+ Na, 100); HRMS calcd for C₂₀H₂₅NO₄NaS, 398.1402, found 398.1404.

Ts *rac-3,5-syn-Dimethyl-1-tosylpyrrolidin-3-yl benzoate* (*syn-13a*): Colorless oil; ¹H NMR δ 1.47 (d, J = 6.0 Hz, 3 H), 1.56 (s, 3 H), 1.66 (dd, J = 10.4, 13.6 Hz, 1 H), 1.89 (s, 3 H), 2.48 (ddd, $J = B_{ZO}$ 2.2, 6.3, 13.6 Hz, 1 H), 3.49 (d, J = 13.2 Hz, 1 H), 3.70 (ddq, $J = 6.2 \times 4$, 10.3 Hz, 1 H), 4.21 (dd, J = 2.1, 13.1 Hz, 1 H), 6.86 (d, J = 7.9 Hz, 2 H), 7.22-7.29 (m, 2 H), 7.44 (tt, J = 1.3, 7.3 Hz, 1 H), 7.49 - 7.57 (m, 4 H); ¹³C NMR δ 21.1 (CH₃), 21.3 (CH₃), 22.1 (CH₃), 48.0 (CH₂), 55.6 (CH), 58.1 (CH₂), 84.7 (C), 127.4 (2 x CH), 128.0 (2 x CH), 129.3 (2 x CH), 129.4 (2 x CH), 130.4 (C), 132.8 (CH), 134.2 (C), 143.1 (C), 165.2 (C); MS (EI+) m/z (rel intensity) 251 (M⁺- PhCOOH, 97), 236 [M⁺- (PhCOOH + Me), 17]; HRMS calcd for C₁₃H₁₇NO₂S, 251.0980, found 251.0986; IR 1714, 1599, 1453, 1337, 1295, 1241, 1158, 1097 cm⁻¹.

Main correlations observed by NOESY are shown in the figure:

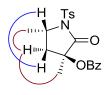


Ts rac-3,5-anti-Dimethyl-1-tosylpyrrolidin-3-yl benzoate (anti-13a): Colorless oil: ¹H NMR $(500 MHz) <math>\delta$ 1.36 (s, 3 H), 1.43 (d, J = 6.6 Hz, 3 H), 2.15 (dd, J = 8.8, 13.6 Hz, 1 H), 2.26 (dd, J = 5.5, 13.7 Hz, 1 H), 2.44 (s, 3 H), 3.50 (d, J = 11.3 Hz, 1 H), 3.79 (dquin, J = 6.2 x 4, 8.1 Hz, 1 H), 3.90 (d, J = 11.3 Hz, 1 H), 7.33 (d, J = 7.9 Hz, 2 H), 7.42 (d, J = 8.2 Hz, 1H), 7.44 (d, J = 7.6 Hz, 1H), 7.56 (tt, J = 1.3, 7.4 Hz, 1 H), 7.76 (d, J = 8.5 Hz, 2 H), 7.96 (d, J = 8.5 Hz, 2 H); ¹³C NMR (125 MHz) δ 21.6 (CH₃), 22.2 (CH₃), 22.8 (CH₃), 45.4 (CH₂), 55.0 (CH), 59.0 (CH₂), 84.0 (C), 127.6 (2 x CH), 128.4 (2 x CH), 129.6 (2 x CH), 129.7 (2 x CH), 130.6 (C), 133.1 (CH), 134.8 (C), 143.6 (C), 165.6 (C); MS (EI+) m/z (rel intensity) 251 (M⁺- PhCOOH, 100), 236 [M⁺- (PhCOOH + Me), 27]; HRMS calcd for C₁₃H₁₇NO₂S, 251.0980, found 251.0979. Main correlations observed by NOESY are shown in the figure:



Ts rac-3,5-syn-Dimethyl-2-oxo-1-tosylpyrrolidin-3-yl benzoate (*syn*-13c): Colorless oil; NMR showed a mixture of isomers (*syn/anti*)-13c in 7.5:1.0 ratio; ¹H NMR δ 1.63 (d, J = 6.6 Hz, 3H), 1.67 (s, 3H), 1.92 (dd, J = 2.8, 13.9 Hz, 1H), 2.45 (s, 3H), 2.88 (dd, J = 9.8, 13.9 Hz, 1H), 4.50-4.57 (m, 1H), 7.34 (d, J = 7.9 Hz, 2H), 7.38 (d, J = 8.5 Hz, 2H), 7.53 (dd, J = 7.4, 7.4 Hz, 1H), 7.82 (dd, J = 1.2, 8.5 Hz, 2H), 7.98 (d, J 0 8.5 Hz, 2H); ¹³C NMR δ 21.7 (CH₃), 23.9 (CH₃), 25.2 (CH₃), 39.6 (CH₂), 52.5 (CH), 79.5 (C), 128.3 (2 x CH), 128.9 (2 x CH), 129.3 (2 x CH), 129.7 (2 x CH), 133.4 (CH), 133.5 (C), 135.0 (C), 145.0 (C), 164.9 (C), 171.7 (C); MS (ESI+) *m/z* (rel intensity) 410 (M⁺+ Na, 100); HRMS calcd for C₂₀H₂₁NO₅NaS, 410.1038, found 410.1050.

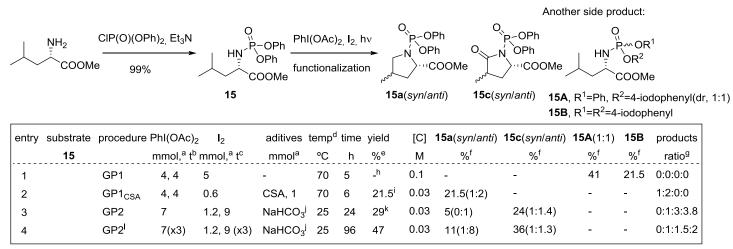
Main correlations observed by NOESY are shown in the figure:



Ts rac-3,5-anti-Dimethyl-2-oxo-1-tosylpyrrolidin-3-yl benzoate (anti-13c): colorless oil; NMRshowed a mixture of isomers (*syn/anti* $)-13c in 7.5:1.0 ratio; ¹H NMR <math>\delta$ 1.54 (s, 3H), 1.61 (d, J = 6.0 Hz, 3H), 2.31 (dd, J = 6.9, 13.2 Hz, 1H), 2.44 (s, 3H), 2.46 (dd, J = 7.9, 13.2 Hz, 1H), 4.33-4.37 (m, 1H), 7.33-7.99 (m, 9H).

Synthesis and Functionalization of (diphenoxyphosphoryl)amine 15:

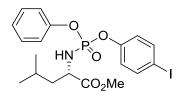
Table S14.



^a millimoles per millimol of **15**; ^b Time of addition in hours: Portionwise (0.25 equiv / 15 min); ^c time of continous addition of **I**₂ disolved in DCE (0.15M) by syringe pump (in hours); ^d 25 °C (reaction mixture refrigerated by a fan) and 65-75 °C (without fan); ^e total isolated yield; ^f isolated yield; ^g (**15a**-*syn*/**15a**-*anti*/**15c**-*anti*) calculated by integration in the ¹H NMR of the crude of reaction; ^h 10% of **15** was recovered; ⁱ addition of PhI(OAc)₂ and **I**₂ was equally repited every 3 continous days. reaction performed with 300 mg.

NHPO(OPh)₂ (S)-Methyl 2-[(diphenoxyphosphoryl)amino]-4-methylpentanoate (15)

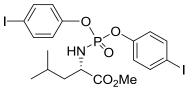
CO₂Me L-Leucine methylester hydrochloride (1.01 g, 5.58 mmol) was dissolved in dry CH₂Cl₂ (62 mL) and treated with TEA (3.1 mL, 22.32 mmol) and diphenylchlorophosphate (2.3 mL, 11.16 mmol). The mixture was stirred at rt for 1.5 h under nitrogen, poured into a dilute hydrochloric acid (5%) solution, and extracted with CHCl₃. The organic layer was washed with an aqueous saturated solution of NaHCO₃ and concentrated in vacuo. Column chromatography of the residue (hexanes–EtOAc, 85:15) afforded the title compound **15** (1.43 g, 3.79 mmol, 68%) as colorless oil: $[\alpha]_D$ -2.0 (*c* 1.2); ¹H NMR δ 0.86 (d, *J* = 6.6 Hz, 3H), 0.87 (d, *J* = 6.6 Hz, 3H), 1.46 (ddd, *J* = 6.3, 8.5, 13.7 Hz, 1H), 1.55 (ddd, *J* = 2.6, 5.7, 13.7 Hz, 1H), 1.64 (non, *J* = 6.6 x 8 Hz, 1H), 3.62 (s, 3H), 3.66 (dd, *J* = 11.0, 11.0 Hz, 1H), 4.03-4.10 (m, 1H), 7.13-7.34 (m, 10H); ¹³C NMR δ 22.3 (CH₃), 23.0 (CH₃), 24.7 (CH), 44.3 (CH₂), 52.6 (CH₃), 53.8 (CH), 120.6 (4 X CH), 125.4 (2 x CH), 130.1 (4 X CH), 151.1 (2 x C), 174.1 (C); MS (ESI+) *m/z* (rel intensity) 400 (M⁺+ Na, 100); HRMS calcd for C₁₉H₂₄NO₅NaP, 400.1290, found 400.1282.



(2*S*)-Methyl 2-[((4-iodophenoxy)(phenoxy)phosphoryl)amino]-4methylpentanoate (15A). NMR showed only one diastereoisomer: Pale brown syrup: $[\alpha]_D$ +1.3 (*c* 1.0); ¹H NMR δ 0.86 (d, *J* = 6.6 Hz, 1H), 0.87 (d, *J* = 6.6 Hz, 1H), 0.87 (d, *J* = 6.6 Hz, 1H), 0.88 (d, *J* = 6.9 Hz, 3H), 1.47 (ddd, *J* = 7.2,

7.2, 14.8 Hz, 1H), 1.51-1.58 (m, 1H), 1.60-1.69 (m, 1H), 3.59 (dd, J = 11.0, 11.0 Hz, 1H), 3.64 (s, 3H), 4.02-4.09 (m, 1H), 6.98 (d, J = 8.8 Hz, 1H), 7.01 (d, J = 8.8 Hz, 1H), 7.15-7.24 (m, 3H), 7.30-7.34 (m, 2H), 7.61 (d, J = 8.8 Hz, 1H), 7.62 (d, J = 8.8 Hz, 1H); ¹³C NMR δ 22.2 (CH₃), 23.0 (CH₃), 24.7 (CH), 44.2 (CH₂), 52.6 (CH₃), 53.8 (CH), 89.0 (C), 120.5 (2 x CH), 122.9 (2 x CH), 125.5 (CH), 130.1 (2 x CH), 139.0

(2 x CH), 151.0 (C), 151.1 (C), 174.1 (C); MS (ESI+) m/z (rel intensity) 526 (M⁺+ Na, 100); HRMS calcd for C₁₉H₂₃NO₅NaPI, 526.0253, found 526.0256.

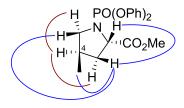


(S)-Methyl 2-((bis(4-iodophenoxy)phosphoryl)amino)-4-methyl pentanoate (15B): Pale brown syrup: $[\alpha]_D$ -0.3 (c 1.0); ¹H NMR δ 0.87 (d, J = 6.9 Hz, 3H), 0.88 (d, J = 6.9 Hz, 3H), 1.46 (ddd, J = 6.0, 8.4, 13.9 Hz, 1H), 1.51-1.58 (m, 1H), 1.60-1.69 (m, 1H), 3.60 (dd, J = 11.0, 11.0 Hz, 1H),

3.65 (s, 3H), 4.00-4.05 (m, 1H), 6.97 (d, J = 8.8 Hz, 2H), 7.00 (d, J = 8.8 Hz, 2H), 7.62 (d, J = 8.8 Hz, 1H), 7.63 (d, J = 8.8 Hz, 2H), 7.65 (d, J = 8.8 Hz, 1H); ¹³C NMR δ 22.2 (CH₃), 23.0 (CH₃), 24.7 (CH), 44.2 (CH₂), 52.7 (CH₃), 53.7 (CH), 89.2 (2 x C), 122.7 (4 x CH), 139.1 (4 x CH), 150.9 (2 x C), 174.0 (C); MS (ESI+) m/z (rel intensity) 652 (M⁺+ Na, 100); HRMS calcd for C₁₉H₂₂NO₅NaPI₂, 651.9223, found 651.9223.

PO(OPh)₂ (2*S*,4*R*)-Methyl 1-(diphenoxyphosphoryl)-4-methylpyrrolidine-2-carboxylate (*anti*-N----CO₂Me 15a): Colorless oil: $[\alpha]_D$ -0.3 (*c* 1.0); ¹H NMR (500 MHz) δ 0.97 (d, *J* = 6.6 Hz, 3H), 1.71 (ddd, *J* = 8.8, 9.8, 12.6 Hz, 1H), 2.08-2.13 (m, 1H), 2.40-2.46 (m, 1H), 2.97 (ddd, *J* = 2.8, 8.6, 8.6 Hz, 1H), 3.59 (s, 3H), 3.63 (dd, *J* = 8.6, 8.6 Hz, 1H), 4.42 (ddd, *J* = 3.2, 3.2, 8.7 Hz, 1H), 7.13-7.34 (m, 10H); ¹³C NMR δ 17.3 (CH₃), 32.9 (CH, ³*J*_{CP} = 9.5 Hz), 39.0 (CH₂, ³*J*_{CP} = 9.5 Hz), 52.1 (CH₃), 54.3 (CH₂, ²*J*_{CP} = 3.2 Hz), 60.9 (CH, ²*J*_{CP} = 6.4 Hz), 120.0 (2 x CH, ⁴*J*_{CP} = 5.3 Hz), 120.4 (2 x CH, ⁴*J*_{CP} = 4.2 Hz), 124.8 (2 x CH, ⁵*J*_{CP} = 7.4 Hz), 129.6 (4 x CH, ³*J*_{CP} = 10.6 Hz), 150.9 (2 x C, ²*J*_{CP} = 5.3 Hz), 173.6 (C); MS (ESI+) *m*/*z* (rel intensity) 398 (M⁺+ Na, 100); HRMS calcd for C₁₉H₂₂NO₅NaP, 398.1133, found 398.1122.

Main correlations observed by NOESY are shown in the figure:



PO(OPh)₂ (2*S*,4*R*)-Methyl 1-(diphenoxyphosphoryl)-4-methyl-5-oxopyrrolidine-2-carboxylate (*anti*-15c): Colorless oil: $[\alpha]_D$ -31.1 (*c* 2.0); ¹H NMR (500 MHz) δ 1.13 (d, *J* = 7.2 Hz, 3H), 1.81 (ddd, *J* = 9.1, 11.8, 13.1 Hz, 1H), 2.33 (dddd, *J* = 1.7, 1.7, 8.2, 13.0 Hz, 1H),

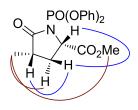
2.62-2.69 (m, 1H), 3.54 (s, 3H), 4.49 (ddd, J = 1.6, 4.6, 9.1Hz, 1H), 7.18-7.35 (m, 10H); ¹³C NMR δ 15.3 (CH₃), 33.8 (CH₂, ³ $J_{CP} = 7.8$ Hz), 36.5 (CH, ³ $J_{CP} = 8.5$ Hz), 52.9 (CH₃), 58.7 (CH, ² $J_{CP} = 5.0$ Hz), 121.2 (4 x CH, ⁴ $J_{CP} = 22.6$ Hz), 126.0 (2 x CH, ⁵ $J_{CP} = 24.0$ Hz), 130.0 (4 x CH, ³ $J_{CP} = 21.9$ Hz), 150.4 (2 x C), 171.8 (C), 179.3 (C); MS (ESI+) m/z (rel intensity) 412 (M⁺+ Na, 100); HRMS calcd for C₁₉H₂₀NO₆NaP, 412.0926 found 412.0930.

Main correlations observed by NOESY are shown in the figure:

PO(OPh)₂ н CO₂Me

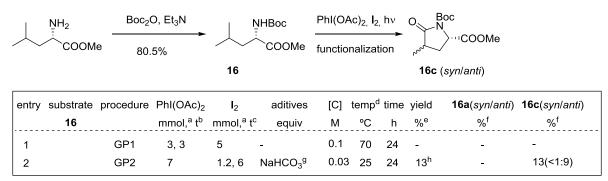
PO(OPh)₂ (2*S*,4*S*)-Methyl 1-(diphenoxyphosphoryl)-4-methyl-5-oxopyrrolidine-2-carboxylate N CO_2Me (*syn*-15c): colorless oil; [α]_D -25.1 (*c* 1.2); ¹H NMR (500 MHz) δ 1.25 (d, *J* = 6.6 Hz, 3H), 1.78 (dd, *J* = 6.2, 12.9 Hz, 1H), 2.48-2.60 (m, 2H), 3.55 (s, 3H), 4.43 (dd, *J* = 6.9, 8.8 Hz, 1H), 7.18-7.36 (m, 10H); ¹³C NMR δ 16.3 (CH₃), 32.7 (CH₂, ³*J*_{CP} = 8.5 Hz), 37.6 (CH, ³*J*_{CP} = 9.2 Hz), 52.9 (CH₃), 59.3 (CH, ²*J*_{CP} = 5.0 Hz), 121.2 (4 x CH, ⁴*J*_{CP} = 42.4 Hz), 126.0 (2 x CH, ⁵*J*_{CP} = 17.0 Hz), 130.0 (4 x CH, ³*J*_{CP} = 30.4 Hz), 150.4 (2 x C), 172.1 (C), 179.5 (C); MS (ESI+) *m*/*z* (rel intensity) 412 (M⁺+ Na, 100); HRMS calcd for C₁₉H₂₀NO₆NaP, 412.0926 found 412.0928.

Main correlations observed by NOESY are shown in the figure:



Synthesis and Functionalization of (tert-butoxycarbonyl)amine 16:

Table S15.



^a millimoles per millimol of **16**; ^a Time of addition in hours: Portionwise (0.25 equiv / 15 min); ^b time of continous addition of **I**₂ disolved in DCE (0.15M) by syringe pump (in hours); ^c 25 °C (reaction mixture refrigerated by a fan) and 65-75 °C (without fan); ^d total isolated yield; ^e isolated yield; ^f calculated by integration in the ¹H NMR of the crude of reaction; ^g 100% w/w; ^h large amount of products without the Boc group were observed in the ¹H NMR spectra after column chromatography.

NHBoc (S)-Methyl 2-[(*tert*-butoxycarbonyl)amino]-4-methylpentanoate (16):²⁰ (Boc)₂O (6.7 $\overline{CO_2Me}$ mmol, 1.5 g) was added to a solution of L-leucine methyl ester (5.9 mmol, 1 g) and triethylamine (11.2 mmol, 1.6 mL) in anhydrous THF (14 mL) at room temperature. The reaction mixture

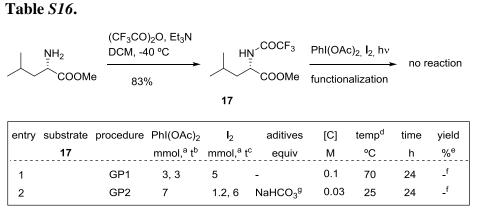
²⁰ Spectral data are in good agreement with those reported in the literature: Jahani, F.; Tajbakhsh, M.; Golchoubian, H.; Khaksar, S. *Tetrahedron Lett.* **2011**, *52*, 1260.

was stirred for 24 h at room temperature, quenched by the addition of aqueous HCl solution (10 %) and diluted with EtOAc. The layers were separated and the aqueous phase was extracted with EtOAc twice. The combined organic layer was washed with aqueous solution HCl (10%), saturated aqueous NaHCO₃ solution and brine, dried over anhydrous Na₂SO₄, and evaporated under vacuum. The residue was purified by column chromatography (hexane / EtOAc 70:30) to give the desired *N-tert*-butoxycarbonyl-L-leucine methyl ester as a crystalline solid (1.1 g, 80%). $[\alpha]_D$ -4.2 (*c* 1.5); ¹H NMR δ 0.94 (d, *J* = 6.6 Hz, 3H), 0.95 (d, *J* = 6.6 Hz, 3H), 1.44 (s, 9H), 1.44-1.77 (m, 3H), 3.73 (s, 3H), 4.28-4.36 (m, 1H), 4.91 (d like, *J* = 7.1 Hz, 1H); ¹³C NMR δ 22.3 (CH₃), 23.2 (CH₃), 25.2 (CH), 28.7 (3 x CH₃), 42.3 (CH₂), 52.5 (CH + CH₃), 80.3 (C), 155.8 (C), 174.2 (C); MS (ESI+) *m*/*z* (rel intensity) 268 (M⁺+ Na, 100); HRMS calcd for C₁₂H₂₃NO₄Na, 268.1525, found 268.1523.

Boc (2*S*,4*R*)-1-*tert*-Butyl 2-methyl 4-methyl-5-oxopyrrolidine-1,2-dicarboxylate (*anti*-16c):²¹ Colorless oil: $[\alpha]_D$ -9.6 (*c* 0.7); ¹H NMR δ 1.22 (d, *J* = 7.0 Hz, 3H), 1.50 (s, 9H), 1.93 (ddd, *J* = 9.6, 11.7, 13.2 Hz, 1H), 2.28 (ddd, *J* = 1.3, 8.6, 13.2 Hz, 1H), 2.62-2.74 (m, 1H), 3.78 (s, 3H), 4.57 (dd, *J* = 1.5, 9.5 Hz, 1H); ¹³C NMR δ 15.1 (CH₃), 27.9 (3 x CH₃), 30.5 (CH₂), 36.6 (CH), 52.5 (CH₃), 56.9 (CH), 83.5 (C), 149.6 (C), 171.8 (C), 175.5 (C). MS (ESI+) *m/z* (rel intensity)

Synthesis and Functionalization of Trifluoroacetylamine 17:

280 (M^+ + Na, 100); HRMS calcd for C₁₂H₁₉NO₅Na, 280.1161, found 280.1168.

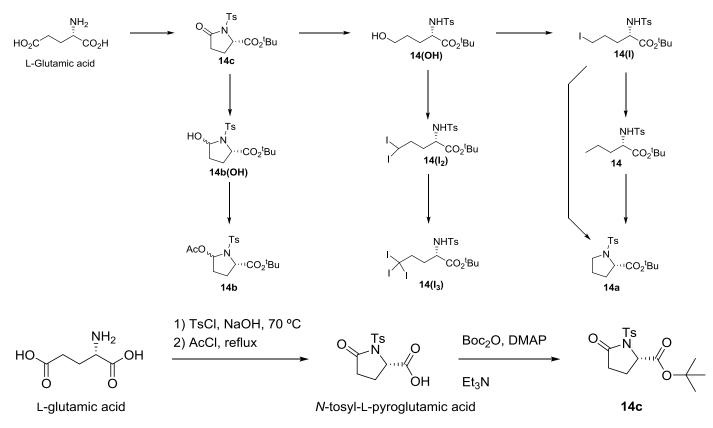


^a millimoles per millimol of **17**; ^b Time of addition in hours: Portionwise (0.25 equiv / 15 min); ^c time of continous addition of I₂ disolved in DCE (0.15M) by syringe pump (in hours); ^d 25 °C (reaction mixture refrigerated by a fan) and 65-75 °C (without fan); ^e isolated yield; ^f **17** was quantitatively recovered; ^g 100% w/w.

²¹ Spectral data are in good agreement with those reported in the literature for its enantiomer: Katoh, M.; Mizutani, H.; Honda, T. *Heterocycles* **2006**, *69*, 193.

MECHANISTIC STUDIES Description of products:

Scheme S1. Synthetic approaches



tert-Butyl (2S)-1-[(4-methylphenyl)sulfonyl]-2-pyrrolidinecarboxylate (14c):

Step 1: L-Glutamic acid (14.7 g, 0.1 mol) was placed in a 250-mL conical flask and NaOH solution (2N) was added slowly until all glutamic acid was dissolved and the mixture became distinctly alkaline. The reaction mixture was then stirred on the magnetic stirrer at 70°C using a hot water bath. TsCl (0.15 mol) was added in small portions with constant stirring and also, from time to time, small addition of NaOH (2 N) to keep the reaction mixture alkaline. The stirring was continued until a clear homogeneous solution resulted or the TLC showed that the reaction was completed. After cooled and filtered to separate non dissolved solid matter, the filtrate was acidified with concentrated HCl, saturated with sodium chloride and extracted with EtOAc. The organic phase was washed with brine solution and dried overnight over anhydrous sodium sulphate. The solvent was distilled off to get the *N*-tosyl-L-glutamic acid derivative (26.6 g, 0.093 mol, 93%).

Step 2:²² *N*-Tosyl-L-glutamic acid (0.01 mol) was placed in 100-mL round-bottomed flask, fitted with reflux condenser and calcium chloride guard tube. Acetyl chloride (0.025 mol) was added and refluxed for 2 h in a boiling water bath. After the reaction was completed (tested by TLC), the reaction mixture was cooled and then poured onto crushed ice with continuous stirring. The precipitated product was filtered, recrystallized from water, and directly used for the next step.

²² Steps 1-2: Srikanth, K.; Kumar, Ch. A.; Ghosh, B.; Jha, T. Bioorg. Med. Chem. 2002, 10, 2119.

Step 3: To the *N*-tosyl-L-pyroglutamic acid (7.2 g, 25.4 mmol) in CH₃CN (70 ml) was added DMAP (297 mg, 2.5 mmol), Et₃N (5 ml) and (Boc)₂O (8.1 g, 36.2 mmol) in CH₃CN (15 mL) and was stirred for 24 h at room temperature, before quenching by the addition of HCl (10 %) solution. After the reaction mixture was diluted with EtOAc, the layers were separated and the aqueous phase was extracted with EtOAc. The combined organic layers were washed with HCl (10%) solution, saturated aqueous NaHCO₃ solution and brine, dried over anhydrous Na₂SO₄, and the solvent was evaporated to give the crude product. The residue was purified by column chromatography (hexane /EtOAc, 70:30) to give **14c** (4.3 g, 12.7 mmol, 50%) as a crystalline solid: ¹H NMR δ 1.50 (s, 9H), 2.05 (dddd, *J* = 1.6, 4.1, 9.2, 11.1 Hz, 1H), 2.36–2.42 (m, 2H), 2.43 (s, 3H), 2.48–2.61 (m, 1H), 4.74 (dd, *J* = 2.0, 9.2 Hz, 1H), 7.32 (d, *J* = 8.2 Hz, 2H), 7.97 (d, *J* = 8.2 Hz, 2H); ¹³C NMR δ 21.7 (CH₃), 23.3 (CH₂), 27.9 (3 x CH₃), 30.5 (CH₂), 60.2 (CH), 83.0 (C), 129.1 (2 x CH), 129.2 (2 x CH), 135.2 (C), 145.2 (C), 169.9 (C), 171.0 (C).²³

(2S)-tert-Butyl 5-hydroxy-1-tosylpyrrolidine-2-carboxylate [14b(OH)]:



Diisobutylaluminium hydride 1.0 M in toluene (0.19 mL, 0.19 mmol) was added dropwise to a solution of lactam **14c** (43 mg, 0.13 mmol) in anhydrous THF (0.65 mL) at -78 °C under Ar atmosphere. The reaction was allowed to reach room temperature, and was stirred for 1 h. Then, it was cooled at -78 °C and MeOH (0.2 mL) was added, stirred for 10 min, poured into a saturated aqueous solution of K₂CO₃ and extracted with EtOAc. The organic phase was dried over sodium sulfate and concentrated under vacuum. The residue was purified by column chromatography (hexanes/EtOAc, 70:30) to yield **14b(OH)** (32 mg, 74 %) as a colorless oil. NMR showed a mixture of isomers in 1:1.7 ratio. ¹H NMR δ 1.38 (s, 9 H), 1.42 (s, 9 H), 1.87-2.22 (m, 6 H), 2.42 (s, 6 H), 2.39-2.45 (m, 1 H), 2.46-2.59 (m, 1 H), 4.21 (dd, *J* = 8.3, 5.5 Hz, 1 H), 4.40 (dd, *J* = 9.2, 0.9 Hz, 1 H), 5.55 (d, *J* = 5.0 Hz, 1 H), 5.57 (dt, *J* = 5.4, 2.8 Hz, 1 H), 7.29 (d, *J* = 8.5 Hz, 4 H), 7.77 (d, *J* = 8.2 Hz, 2 H), 7.79 (d, *J* = 8.2 Hz, 2 H); ¹³C NMR δ 21.47 (CH₃), 21.49 (CH₃), 27.8 (CH₃ x 6), 28.4 (CH₂), 29.0 (CH₂), 32.5 (CH₂), 34.0 (CH₂), 60.7 (CH), 61.4 (CH), 81.7 (C), 82.4 (C), 84.3 (CH), 85.2 (CH), 127.4 (2 x CH), 127.6 (2 x CH), 129.5 (2 x CH), 129.6 (2 x CH), 136.8 (C), 137.3 (C), 143.7 (2 x C), 170.6 (C), 173.3 (C); MS (ESI⁺) *m*/*z* (rel intensity) 364 (M⁺+ Na, 100); HRMS calcd for C₁₆H₂₃NO₅NaS, 364.1195, found 364.1190.

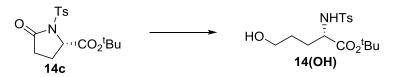
(2S)-tert-Butyl 5-acetoxy-1-tosylpyrrolidine-2-carboxylate (14b):



²³ spectral data are in good agreement with those reported in the literature: Yar, M.; Unthank, M. G.; McGarrigle, E. M.; Aggarwal V. K. *Chem. Asian J.* **2011**, *6*, 372 – 375.

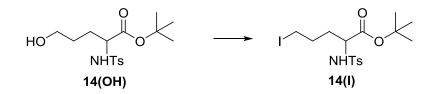
Acetic anhydride (0.05 mL) and pyridine (0.02 mL) were added to a solution of hemiaminal **14b(OH)** (32 mg, 0.094 mmol) in DCM (0.8 mL). The reaction was stirred for 17 h at room temperature and concentrated under vaccum to yield **14b** (35 mg, 97%) as colorless oil, unstable on silica gel and acidic conditions in general. NMR showed a mixture of isomers in 1:1.3 ratio. ¹H NMR δ 1.40 (s, 9 H), 1.40 (s, 9 H), 1.79 (s, 3 H), 1.82-2.32 (m, 8 H), 1.84 (s, 3 H), 2.35 (s, 3 H), 2.35 (s, 3 H), 4.20-4.26 (m, 2 H), 6.38 (t, *J* = 3.4 Hz, 1 H), 6.44 (d, *J* = 3.9 Hz, 1 H), 7.23 (d, *J* = 8.0 Hz, 4 H), 7.69 (d, *J* = 8.3 Hz, 2 H), 7.74 (d, *J* = 8.3 Hz, 2 H); ¹H NMR (500 MHz, C₆D₆) δ 1.26–1.36 (m, 2H), 1.39 (s, 9H), 1.40 (s, 9H), 1.44–1.64 (m, 4H), 1.55 (s, 3H), 1.59 (s, 3H), 1.81–1.95 (m, 1H), 1.85 (s, 6H), 2.00–2.08 (m, 1H), 4.28 (t, *J* = 8.0 Hz, 1H), 4.42 (d, *J* = 9.1 Hz, 1H), 6.72 (d, *J* = 5.4 Hz, 1H), 6.75 (d, *J* = 8.2 Hz, 4H), 6.81 (d, *J* = 4.7 Hz, 1H), 7.92 (d, *J* = 8.5 Hz, 2H), 7.98 (d, *J* = 8.2 Hz, 2H); ¹³C NMR (125.7 MHz, C₆D₆) δ 20.9 (CH₃), 21.1 (CH₃), 21.4 (2 x CH₃), 28.2 (3 x CH₃), 28.5 (CH₂), 28.7 (CH₂), 32.7 (CH₂), 33.4 (CH₂), 62.2 (CH), 63.2 (CH), 81.6 (C), 81.8 (C), 85.1 (CH), 85.3 (CH), 128.7 (2 x CH), 129.8 (2 x CH), 129.9 (4 x CH), 138.1 (C), 138.7 (C), 143.7 (2 x C), 169.4 (C), 169.7 (C), 171.0 (C), 171.2 (C); MS (ESI⁺) *m/z* (rel intensity) 406 (M⁺ + Na, 100); HRMS calcd for C₁₈H₂₅NO₆NaS, 406.1300, found 406.1304.

tert-Butyl (2S)-5-hydroxy-2-{[(4-methylphenyl)sulfonyl]amino}pentanoate [14(OH)]:



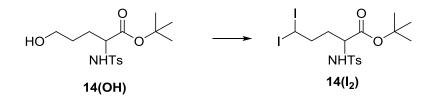
To a solution of **14c** (660 mg, 1.9 mmol) in anhydrous THF (47 mL) was added NaBH₄ (216 mg, 5.7 mmol), and the mixture was heated at 50-60°C. Then a solution of *tert*-butyl alcohol:THF (1:1, 10mL) was added dropwise. After the reaction was completed, the reaction mixture was cooled at room temperature and quenched by the slow addition of saturated citric acid solution (pH~4), diluted with EtOAc and washed with brine. After the layers were separated, the aqueous phase was extracted with EtOAc twice. The combined organic layer was dried over anhydrous Na₂SO₄, evaporated and purified by column chromatography (hexane/EtOAc 6:4) to give the desired product **14(OH)** (481 mg, 72%) as a crystalline solid: Mp. 68.9-70.1°C (hexane/EtOAc); [α] _D +20.0° (*c* 0.35); ¹H NMR δ 1.24 (s, 9H), 1.64–1.71 (m, 3H), 1.80–1.86 (m, 1H), 2.40 (s, 3H), 3.60–2.69 (m, 2H), 3.81 (dd, *J* = 4.4, 7.2 Hz, 1H), 5.50 (bs, 1H), 7.27 (d, *J* = 8.2 Hz, 2H), 7.72 (d, *J* = 8.2 Hz, 2H); ¹³C NMR δ 21.3 (CH₃), 27.9 (3 x CH₃), 28.0 (CH₂), 30.1 (CH₂), 56.0 (CH), 62.0 (CH₂), 82.4 (C), 127.3 (2 x CH), 129.6 (2 x CH), 136.9 (C), 143.2 (C), 170.8 (C); MS (ESI⁺) *m/z* (rel intensity) 366 (M⁺ + Na, 100); HRMS (ESI+) calcd for C₁₆H₂₅NO₅SNa, 366.1353 found 366.1353.

tert-Butyl (2S)-5-iodo-2-{[(4-methylphenyl)sulfonyl]amino}pentanoate [14(I)]:



I₂ (350 mg, 1.4 mmol) was added to a solution of PPh₃ (913 mg, 3.5 mmol) and imidazole (254 mg, 3.7 mmol) in anhydrous dichloromethane (5 ml) at 0°C and allowed to warm at room temperature. Then a solution of alcohol **14(OH)** (600 mg, 1.75 mmol) in anhydrous dichloromethane (3 mL) was added at 0°C and the reaction was stirred during 2h at this temperature. The reaction mixture was poured into a saturated aqueous solution of Na₂S₂O₃ and extracted with dichloromethane. The organic phase was dried over sodium sulfate and concentrated under vacuum. The residue was purified by column chromatography (hexanes/EtOAc, 70:30) to yield **14(I)** (760 mg, 1.7 mmol, 96 %) as a crystalline solid. Mp. 122.5-123.7°C (hexane-EtOAc); [α] _D+51° (*c* 0.21), ¹H NMR δ 1.26 (s, 9H), 1.64–1.73 (m, 1H), 180–1.97 (m, 3H), 2.40 (s, 3H), 3.12–3.23 (m, 2H), 3.76 (ddd, *J* = 4.2, 8.5, 8.5 Hz, 1H), 5.23 (d, *J* = 8.5 Hz, 1H), 7.29 (d, *J* = 8.2 Hz, 2H), 7.72 (d, *J* = 8.2 Hz, 2H). ¹³C NMR δ 5.38 (CH₂), 21.4 (CH₃), 27.6 (3 x CH₃), 28.7 (CH₂), 34.2 (CH₂), 55.2 (CH), 82.8 (C), 127.3 (2 x CH), 129.6 (2 x CH), 136.7 (C), 143.6 (C), 170.3 (C). MS (ESI⁺) *m/z* (rel intensity) 476 (M⁺+ Na, 100); HRMS (ESI+) calc for C₁₆H₂₄NO₄SINa, 476.0369 found 476.0363.

tert-Butyl (2S)-5,5-diiodo-2-{[(4-methylphenyl)sulfonyl]amino}pentanoate [14(I₂)]:

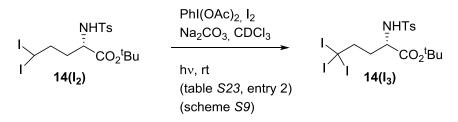


Dess-Martin periodinane (488 mg, 1.15 mmol) was added to a solution of alcohol **14(OH)** (329 mg, 0.96 mmol) in DCM (10 mL) and stirred at room temperature during 45 min. Then, the reaction was poured into a saturated aqueous solution of $Na_2S_2O_3$ and extracted with EtOAc. The organic phase was dried over sodium sulfate and concentrated under vacuum. The crude residue was used in the next step without further purification. Hydrazine monohydrate (0.9 mL) was added to a solution of the crude in dry DCM (1mL) and stirred at room temperature during 1 h. Then, water and DCM were added and the aqueous layer was extracted with CH_2Cl_2 . The combined organic extracts were dried over $MgSO_4$, filtered and concentrated under vacuum. To this crude pale yellow oil were added, at room temperature and under argon, TEA (0.25 mL), CH_2Cl_2 (1mL) and iodine in 3 portions (300 mg) and the reaction was stirred for 10 min. After hydrolysis with an aqueous solution of Na_2SO_3 , the aqueous layer was extracted with CH_2Cl_2 and the combined organic extracts **u** and concentrated under vacuum.²⁴ Purification by chromatography gave the diiodo derivative **14(I_2)** as a yellow solid (110 mg, 21% after 3 steps). Mp.

²⁴ Français, A.; Leyva-Pérez, A.; Etxebarria-Jardi, G.; Peña, J.; Ley, S. V. Chem. Eur. J. 2011, 17, 329.

117-119°C (hexane/EtOAc); $[\alpha]_{D}$ +24.1° (*c* 0.22); ¹H NMR δ 1.29 (s, 9H), 1.73 (dddd, J = 5.0, 8.2, 10.2, 12.9 Hz, 1H), 1.92 (dddd, J = 5.0, 5.3, 10.2, 13.9 Hz, 1H), 2.35–1.49 (m, 2H), 2.41 (s, 3H), 3.79 (ddd, J = 5.0, 8.2, 8.2 Hz, 1H), 5.13 (dd, J = 5.6, 6.7 Hz, 1H), 5.17 (d, J = 8.3 Hz, 1H), 7.30 (d, J = 8.4 Hz, 2H), 7.72 (d, J = 8.4 Hz, 2H); ¹³C NMR δ -29.0 (CH), 21.5 (CH₃), 27.7 (3 x CH₃), 35.4 (CH₂), 43.3 (CH₂), 54.5 (CH), 83.2 (C), 127.4 (2 x CH), 129.8 (2 x CH), 136.6 (C), 143.8 (C), 170.0 (C); MS (ESI⁺) *m/z* (rel intensity) 602 (M⁺+ Na, 100); HMRS (ESI +) *m/z* calc for C₁₆H₂₃NO₄SNa, 601.9335 found 601.9341.

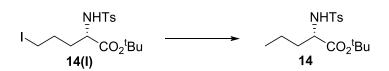
tert-Butyl (2S)-5,5,5-triiodo-2-{[(4-methylphenyl)sulfonyl]amino}pentanoate [14(I₃)]:



Following conditions described in table *S23*, entry 2, compound **14(I₃)** was detected as major product in the crude reaction by ¹H NMR (87.2% conversion)(scheme *S9*). **14(I₃)** was relatively stable when reaction mixture was kept at -22 °C in the dark.

Description of the compound **14(I₃)** from a crude reaction. ¹H NMR (500 MHz, -20 °C) δ 1.27 (s, 9H), 1.81–1.90 (m, 1H), 1.90–2.00 (m, 1H), 3.10 (ddd, J = 3.5, 14.8, 15.1 Hz, 1H), 3.14 (ddd, J = 4.7, 14.8, 15.1 Hz, 1H), 3.87 (ddd, J = 4.4, 7.9, 8.2 Hz, 1H), 5.53 (d, J = 8.2 Hz, 1H), 7.30 (d, J = 8.0 Hz, 2H), 7.75 (d, J = 8.0 Hz, 2H); ¹³C NMR (125 MHz, -20 °C) δ –108.8 (C), 21.5 (CH₃), 27.5 (3 x CH₃), 38.1 (CH₂), 52.8 (CH), 60.4 (CH₂), 83.3 (C), 127.2 (2 x CH), 129.8 (2 x CH), 135.6 (C), 143.9 (C), 169.8 (C); MS (ESI⁺) m/z (rel intensity) 728 (M⁺+ Na, 100); HMRS (ESI+) m/z calc for C₁₆H₂₂I₃SNaNO₄, 727.8302, found 727.8303.

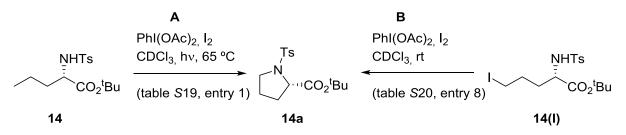
tert-Butyl (2S)-2-{[(4-methylphenyl)sulfonyl]amino}pentanoate (14):²⁵



Bu₃SnH (0.9 mL, 3.3 mmol) and AIBN (54 mg, 0.33 mmol) were added to a solution of **14(I)** (760 mg, 1.65 mmol) in benzene (80 mL) and was refluxed during 1 h. Then, the reaction mixture was concentrated under vacuum and the residue was purified by column chromatography (hexanes and gradient elution to 30 % EtOAC) to give the derivate **14** (512 mg, 1.57 mmol, 94 %) as a crystalline solid. ¹H NMR δ 0.90 (dd, J = 7.4, 7.4 Hz, 3H), 1.24 (s, 9H), 1.35-1.45 (m, 2H), 1.52-1.68 (m, 2H), 2.40 (s, 3H), 3.76 (ddd, J = 5.0, 7.7, 9.2 Hz, 1H), 5.10 (d, J = 9.2 Hz, 1H), 7.27 (d, J = 8.2 Hz, 2H), 7.72 (d, J = 8.2 Hz, 2H); ¹³C NMR δ 13.5 (CH₃), 18.2 (CH₂), 21.4 (CH₃), 27.7 (3 x CH₃), 35.7 (CH₂), 56.0 (CH), 82.2 (C), 127.4 (2 x CH), 129.6 (2 x CH), 137.1 (C), 143.4 (C), 171.0 (C).

²⁵ spectral data are in good agreement with those reported in the literature: Baldwin, J. E.; Spivey, A. C.; Schofield, C. J.; Sweeney, J. B. *Tetrahedrom* **1993**, *49*, 6309–6330.

(S)-tert-Butyl 1-tosylpyrrolidine-2-carboxylate (14a):²⁶



According to GP1 (table *S*19, entry 1), pyrrolidine **14a** was prepared from **14** (100 mg, 0.306 mmol) in 71% following approach **A** or from **14(I)** (100 mg, 0.22 mmol) according to GP3 (table *S*20, entry 8) in quantitative yield following approach **B**, as a crystalline solid compound. ¹H NMR δ 1.45 (s, 9H), 1.72–1.79 (m, 1H), 1.91–2.04 (m, 3H), 2.42 (s, 3H), 3.32 (ddd, *J* = 7.0, 7.0, 9.2 Hz, 1H), 3.44–3.49 (m, 1H), 4.19 (dd, *J* = 3.5, 8.0 Hz, 1H), 7.30 (d, *J* = 8.1 Hz, 2H), 7.76 (d, *J* = 8.1 Hz, 2H); ¹³C NMR δ 21.5 (CH₃), 24.5 (CH₂), 27.9 (3 x CH₃), 30.9 (CH₂), 48.3 (CH₂), 61.2 (CH), 81.5 (C), 127.5 (2 x CH), 129.5 (2 x CH), 136.0 (C), 143.3 (C), 171.3 (C).

²⁶ spectral data are in good agreement with those reported in the literature: Foschi, F.; Landini, D.; Lupi, V.; Mihali, V.; Penso, M.; Pilati, T.; Tagliabue, A. *Chem. Eur. J.* **2010**, *16*, 10667 – 10670.

MECHANISTIC STUDIES (NMR monitored reactions):

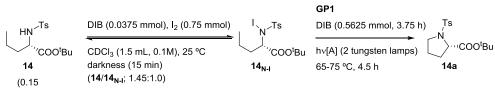
NMR monitored experiments with 14.

Table S17. Monitored experiments with 14:

	HN ^{_Ts}	D ^t Bu	CONDITIONS] CDCI ₃ (1 mL)	* (HN ^{_T}		⊡COO ^t Bu			s OO ^t Bu		:OO ^t Bu	
	14				14(I)		14a		14(l ₂)		14c		
	entry	substrate	e procedure ^a l	PhI(OAc) ₂	I 2	aditives	[C]	temp	time	conv.	14a	14c	notes ^e
		14 mmo		mmol ^b	mmol	^b w/w	М	°C	h	% ^c	% ^d	% ^d	
	1	0.15	GP1	4	5	-	0.1	65	4.5	73	73	-	hν [A]
3]	2	0.15	GP3 _{Na2CO3}	6	1.4	Na ₂ CO _{3,} 100%	0.03	27	6	95 ^f	5	84.3	hν [A]
[table 3	3	0.05	GP3 _{Na2CO3}	6	1.4	Na ₂ CO _{3,} 100%	0.03	27	6	100 ^g	4	88	hv [A+f]
	4	0.05	GP3	2	2	-	0.05	27	0.41	0 ^h	-	-	dark
	5	0.05	GP3	2	2	Na ₂ CO _{3,} 100%	0.05	27	3.6	0 ⁱ	-	-	dark

^a GP1: PhI(OAc)₂ added portionwise (0.25 mmol per mmol of **14**, every 15 min), GP3: all reagents added in one portion; ^b millimols per millimol of **14**; ^c global convertion of starting material. Detection of **14_{N-I}**, was not considered as converted product since this afforded starting material after quenching; ^d convertions; ^e [A] = tungsten lamps (80W), [A+f] = light filtered by 1 cm pathlength of aqueous solution of K₂CrO₄ (0.27 g/L) and Na₂CO₃ (1 g/L) into pyrex glasses (λ >445 nm); ^f besides **14**, **14a** and **14c**, **14(I)** (5.7%) was detected by ¹H NMR; ^g besides **14a** and **14c**, **14(I)** (8%) was detected by ¹H NMR; ^h **14_{N-I}** was detected by ¹H NMR (**14/14_{N-I}**, 1:1.52) after 20 min; ⁱ **14_{N-I}** was detected by ¹H NMR (**14/14_{N-I}**, 1:0.31) after 3.6 h.

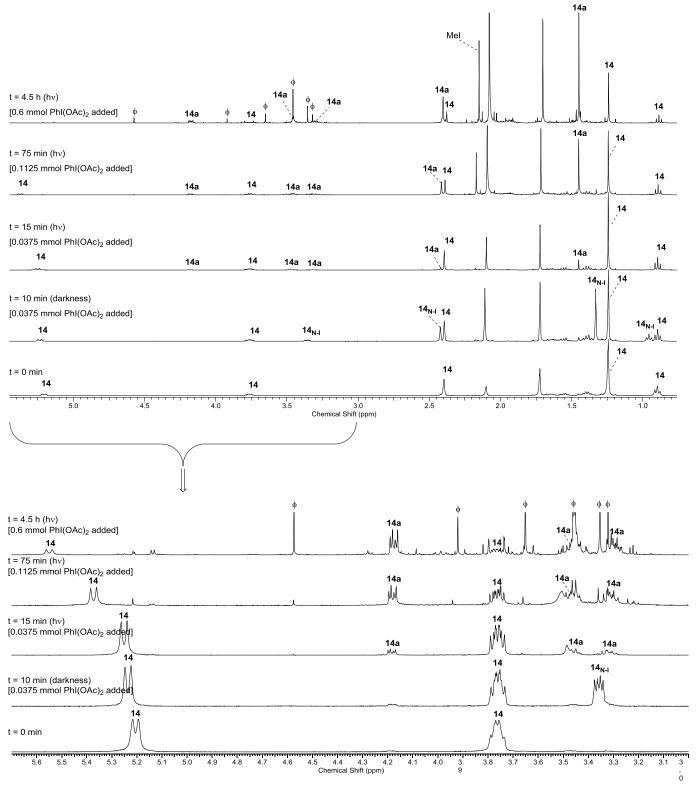
Scheme S2: Table S17 (entry 6):



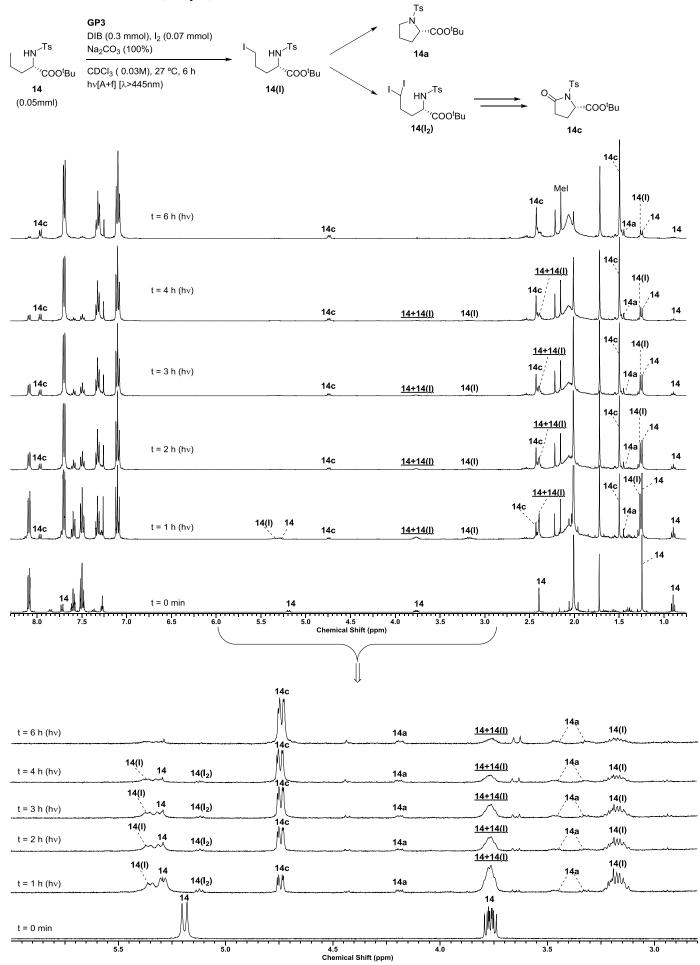


GP1: PhI(OAc)₂ added portionwise (0.25 equiv/15 min)

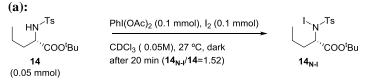
(ϕ) signals from side products that appear along the time when the reaction is performed with the system PhI(OAc)₂/I₂ in CDCl₃ as solvent and tungsten lamps as light source (see scheme *S5*). These signals are independent from our substrates.

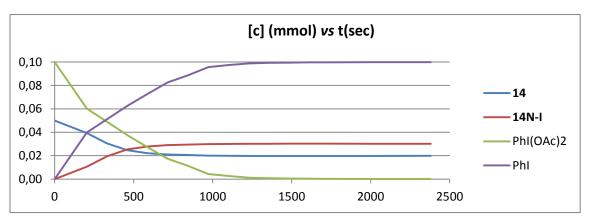


Scheme S3: Table S17 (entry 8):

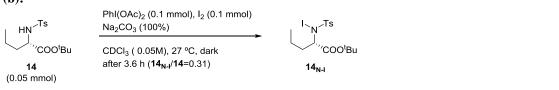


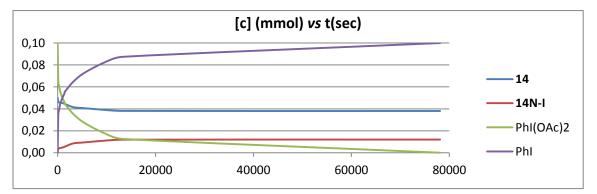
Scheme *S4***:** (a) Formation of 14_{N-I} in absence of Na₂CO₃. (b) Formation of 14_{N-I} in presence of Na₂CO₃. (c) Consumption of PhI(OAc)₂ in the presence of I₂: (A) in absence of Na₂CO₃, (B) in presence of Na₂CO₃.

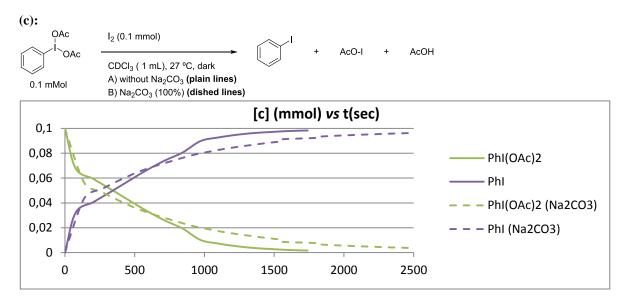






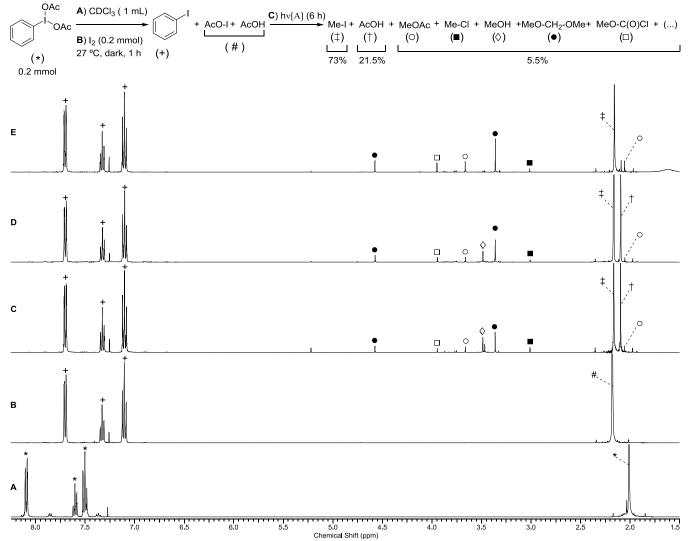




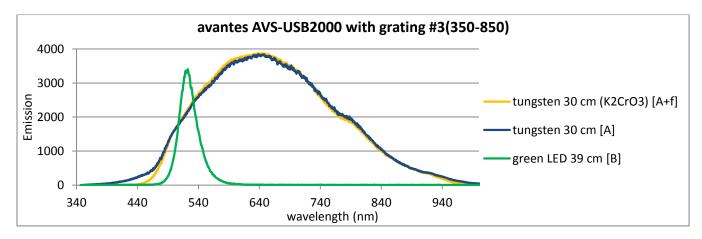


Determination of products from photolysis of PhI(OAc)₂ with I₂:

Scheme *S5*: Formation of side products derived from photolysis of AcOI and I_2 in CDCl₃ under irradiation with tungsten lamps (hv[A]).



^a A: PhI(OAc)₂ in CDCl₃; B: I₂ was added and stirred in the darknes for 1 h; C: Reaction mixture was irradiated with 2 tunsgten lamps at 20 cm for 6 h while flask was refrigerated with a fun; D: Reaction mixture was further irradiated for 6 h; E: Reaction mixture was quenched with 20% aqueous solution of Na₂S₂O₃. MeOH and AcOH were extracted into the water phase.



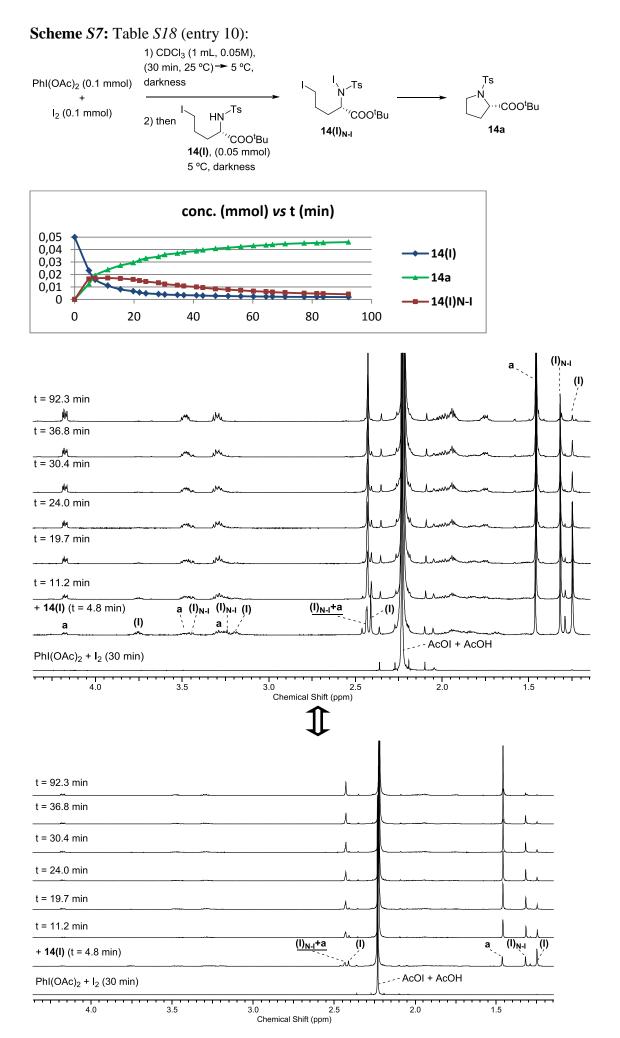
Scheme S6: Relative emission spectrum of different light sources in the region 340-950 nm

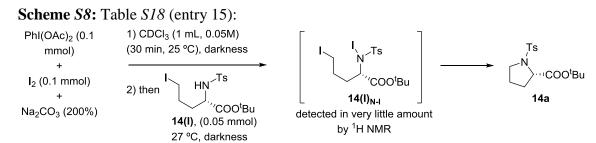
NMR monitored experiments with 14(I).

Table S18. Monitored experiments with 14(I)

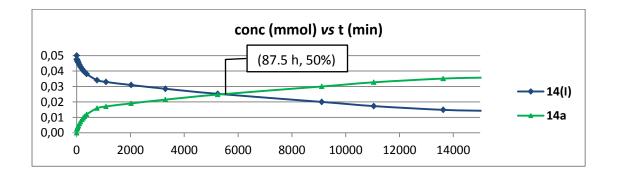
]	HN ^{_Ts} [cond		[conditions]	conditions]		Ts N ⊂ O COO ^t Bu			1		∠Ts		
COO ^t Bu C			CDCl ₃ (1 mL)		,	COO ^t Bu			.,	(COOH			
	14(I) entry substrate		NMR monitoring		1	14a	14c			14(I,	CO ₂ H)		
ſ			procedure ^a P	PhI(OAc) ₂	I_2	aditives	[C] temp ^c		time c	onv.	14a	14c	notes ^f
	14	(I) mmol		mmol ^b	mmol ^b	mmol ^b or w/w	М	°C	h	% ^d	% ^e	% ^e	
	1	0.1	-	-	-	-	0.1	60	10	0	no reaction		dark or hv [A]
	2	0.1	GP3	-	5	-	0.1	27	14	50 ^g	no reaction		dark or hv [A]
	3	0.1	GP3	-	1	-	0.05	27	48	0	no reaction		dark or hv [B]
	4	0.1	GP3	-	-	AcOH, 2	0.1	60	16	0	no reaction		dark or hv [A]
	5	0.1	GP3	-	-	CSA, 1	0.1	27	12	15 ^g	no reaction		dark
	6	0.1	GP3	-	-	Zn(OTf) _{2,} 1	0.1	27	12	0	no reaction		dark
	7	0.052	GP3	2	-	-	0.052	27	10	0	no reaction		dark or $h\nu$ [A]
	8	0.05	GP3	2	1	-	0.05	27	<5 min	100	100	-	dark
	9	0.05	GP4	2	2	-	0.05	27	<3 min	100	100	-	dark
I	10	0.05	GP4	2	2	-	0.05	5	1	100	100	-	dark
	11	0.05	GP3	2	2	-	0.05	5	3.5	100	100	-	dark
	12	0.05	GP3	2	0.2	-	0.05	5	10	100	100	-	dark
	13	0.05	GP3	2	2	Na ₂ CO _{3,} 200%	0.05	5	12	0	-	-	dark
	14	0.05	GP3	2	0.2	Na ₂ CO _{3,} 200%	0.05	27	24	86	86	-	dark, 11.1 h / 50
	15	0.05	GP4	2	2	Na ₂ CO _{3,} 200%	0.05	27	257	71.6	71.6	-	dark, 87.5 h / 50
ſ	16	0.03	GP3	4	1	NaHCO _{3,} 100%	0.05 ^k	25	16	95	49.7	45.3	hν [A]
	17	0.03	GP3 _{Na2CO3}	6	1.3	Na ₂ CO _{3,} 200%	0.03	24	10	100 ¹	12	82	hν [A]
	18	0.03	GP3 _{Na2CO3}	6	1.3	Na ₂ CO _{3,} 200%	0.03	24	7.5	100 ^m	8	87	hv [B]

^a GP3: all reagents added in one portion, GP4: Phl(OAc)₂ and I₂ were stirred in the solvent at room temperature untill the Phl(OAc)₂ desappeared (15-45 min), previous to the addition of **14(I)**; ^b millimols per millimol of **14(I)**; ^c temperature controled by the NMR apparatus; ^d global conversion of starting material; ^e conversions; ^f [A] = tungsten lamps (80W), [B] = green LED light; ^g deprotected compound **14(I, COOH)** was observed; ^h equivalents of reagents added every 12 h; ⁱ 200% w/w; ^j DIB + I₂ previouly mixed; ^k DCE as solvent; ^I 6% of another not identified product seems to be formed.





Note: $14(I)_{N-I}$ was detected by ¹H NMR during the whole time in very little amount (<0.002 mmol) indicating that the equilibrium $14(I) \leftrightarrow 14(I)_{N-I}$ is displaced toward 14(I) in these conditions.

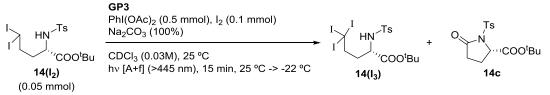


NMR monitored experiments with 14(I₂)

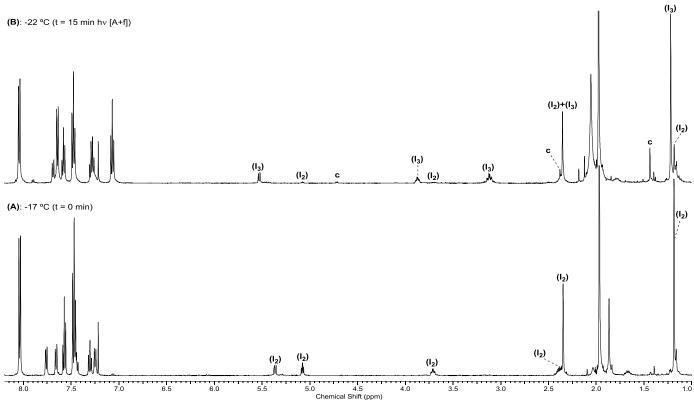
Та	Table S19. Monitored experiments with $14(I_2)$													
>		_Ts	[conditions]	<u> </u>	N ^{Ts}	A	،co	Ts ∽N			I HN Ts	0	Ts N	
											Bu			
	14(l ₂	.)	NMR monitorir	ימ 1	14(I ₂) _{N-I}			14b			14(I ₃)		14c	
	entr	y substra	te procedure ^a	Phl(OAc) ₂	I 2	aditives	[C]	temp	time	conv.	14b	14c	14(I ₃)	notes ^e
		14(l ₂) m	mol	mmol ^b	mmol ^t	^o mmol ^b	М	°C	h	% ^c	% ^d	% ^d	% ^d	
	1	0.05	GP3	3.5	1	Na ₂ CO _{3,} 100%	0.03	27	2	87 ^f	-	78.3	-	hν [A]
	2	0.05	GP3	10	2	Na ₂ CO _{3,} 100%	0.03	25	0.25	95.3	-	8.1	87.2 ^g	hν [A+f]
ole 31	3	0.05	GP3	3	1	-	0.03	5-35	4	71.6 ^h	11.2	8.8	40.8	dark + h∨ [A+f] ⁱ
[table	. 4	0.05	GP3	3	1	Na ₂ CO _{3,} 100%	0.03	27	2.4	58 ^j	-	34.0	13.4	dark + hv [A+f] ⁱ
	5	0.05	GP4	4	1	-	0.03	27	3	100	>95	-	-	dark
	6	0.05	GP4	4	1	Na ₂ CO ₃ , 200%	0.03	27	3	12 ^k	7	-	-	dark

^a GP1: Phl(OAc)₂ added portionwise (0.25 equiv/15 min), GP3: all reagents added in one portion, GP4: Phl(OAc)₂ and I₂ were stirred in the CDCI₃ for 24 h, previous to the addition of **14(I₂)**; ^b millimols per millimol of **14(I₂)**; ^c global convertion of starting material; ^d convertions; ^e [A] = tungsten lamps (80W), [A+f] = tungsten lamps (80W) filtered by 1 cm pathlength of aqueous solution of [K₂CrO₄ (0.27 g/L) and Na₂CO₃ (1 g/L)] into pyrex glasses (λ >445 nm); ^f **14a** (8.7%) was also detected by ¹H NMR; ^g **14(I₃)** was relatively stable when reaction mixture was keept at -22 °C; ^h **14a** (10.8%) was also detected by ¹H NMR; ⁱ procedure described in schemes *S10* and *S11*, with more detail; ^j **14a** (10.3%) was also detected by ¹H NMR; ^k besides **14(I₂)** (66%) and **14b**, **14(I₂)_{N-I}** (22%) and **14a** (5%) were detected by ¹H NMR.

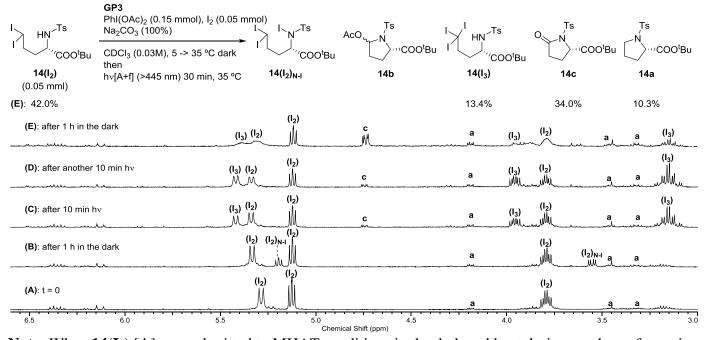
Scheme S9: Table S19 (entry 2)



Procedure: $PhI(OAc)_2$ and Na_2CO_3 were added to a solution of $14(I_2)$ in CDCl₃ at rt, and the mixture was submitted to a ¹H NMR at – 17 °C (A). Then, the mixture was allowed to reach 25 °C, iodine was added and the reaction was irradiated with filtered light [A+f] for 15 min. Finally, the crude reaction was submitted to a ¹H NMR at -22 °C (B).

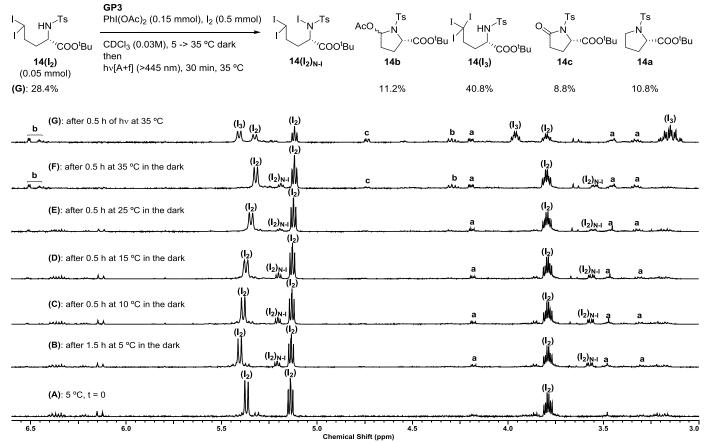


Scheme S10: Table S19 (entry 4)



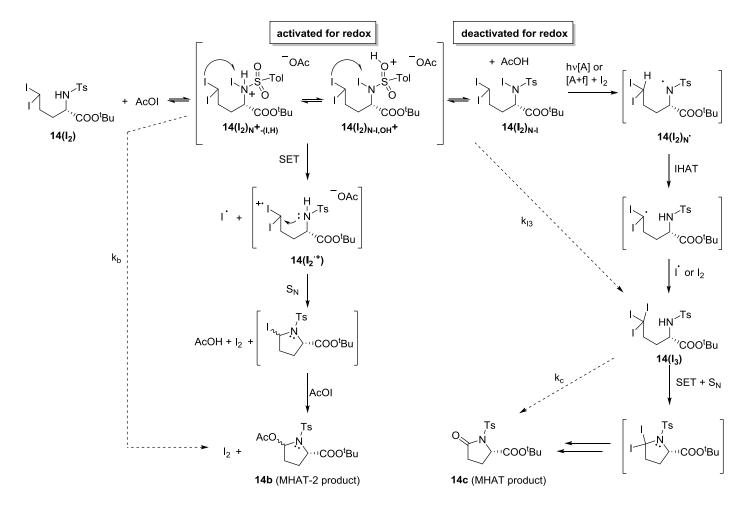
Note: When $14(I_2)$ [A] was submitted to MHAT conditions in the dark and kept during one hour, formation of $14(I_2)_{N-I}$ was observed, which did not lead to cyclic compound 14b [B]. [C and D] show how irradiation is necessary to further functionalize the iodinated moiety, however, more interestingly is the evolution between $[D \rightarrow E]$ indicating that the conversion of $14(I_3)$ toward 14c did not need either light or aqueous quenching.

Scheme S11: Table S19 (entry 3)



Note: When $14(I_2)$ [A] was submitted to standard conditions in the dark at 5 °C and kept during 1.5 hours, formation of $14(I_2)_{N-I}$ was observed, which did not lead to cyclic compound 14b [B]. Then, temperature was increased to 10, 15 and 25 °C keeping the reaction in the dark for half an hour at each temperature, not observing considerable formation of cyclic compound 14b [C, D and E]. Temperatures above 30 °C were

necessary to observe considerable formation of 14b [F]. Irradiation of reaction mixture [F] with filtered light led to a complete consumption of intermediate $14(I_2)_{N-I}$ affording $14(I_3)$ as major product. As a conclusion, it seems to be evident that, under irradiation, the IHAT reaction with $14(I_2)$ toward $14(I_3)$ is much faster than its cyclization toward 14b (Scheme *S12*: $k_{I3} >> k_b$).



Scheme S12: Proposed mechanism for 14(I₂) towards 14b and 14c.

Note 1: $k_{I3} >> k_b$ under irradiation [A] or ([A+f] + I₂), with or without Na₂CO₃ (schemes *S9*, *S10* and *S11*) Note 2: $k_c > k_a$ and k_b in MHAT conditions with Na₂CO₃ in the darkness: Compare scheme *S8* and (table *S19*, entry 6) with the scheme *S10*.

Note 3: Conversion of $14(I_3)$ into 14c did not need light (scheme *S10*).

Note 4: Conversion of $14(I_3)$ into 14c did not need previous aqueous quenching (scheme S10).

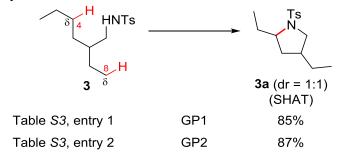
Note 5: Conversion of $14(I_3)$ into 14c required the presence of AcOI, either in absence or presence of Na₂CO₃. Conversion of $14(I_3)$ into 14c was dramatically slowed down when reactions were carried out in deficiency of PhI(OAc)₂, and, moreover, side products derived from $14(I_3)$ were formed.

DFT CALCULATIONS

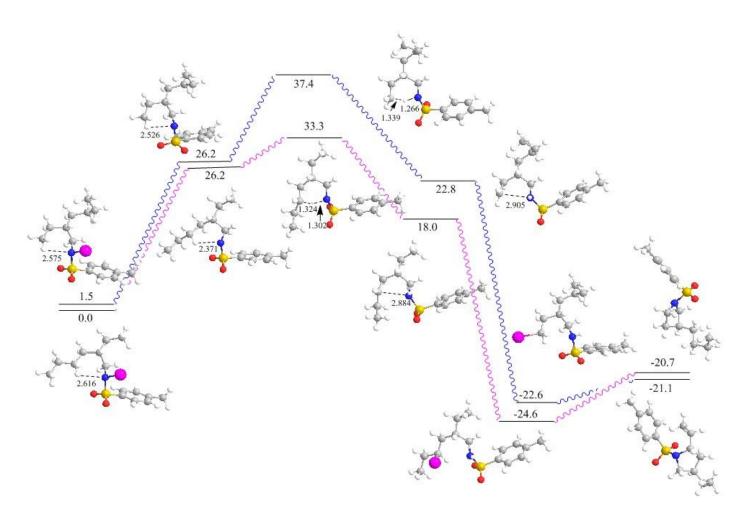
Density Functional Theory (DFT) calculations have been performed using two different density functionals, namely, M06-L (1,2) and M06-2X (2,3). The first one is a meta GGA whereas the second one is a hybrid meta functional. Geometries have been optimized using M06-L in conjunction with the cc-pVDZ basis set (4,5,6). Diffuse functions have been added to iodine in order to correctly describe its more diffuse electron density since we have found that iodine is involved in halogen- π interactions with the benzene ring. As a matter of fact, our calculations show that conformers that feature halogen- π bonding have enhanced stability. Only diffuse functions of low angular momentum have been added (minimally augmented basis or maug-cc-pVDZ) following the procedure described by Truhlar et al. (7). Thermochemical data based on the rigid rotor-harmonic oscillator-ideal approximations have been obtained at this level of theory. In order to improve the quality of the electronic energies we have carried out M06-2X/maug-cc-pVTZ (4,5,6) calculations on the M06-L geometries. The chosen level of theory for this work is therefore M06-2X/maug-cc-pVTZ/M06-L/maug-cc-pVDZ. All DFT calculations have been carried out with the G09 program package (8). We have additionally model bulk solvent effects through the Polarizable Continuum Model (PCM) as implemented in G09. In particular, we have used the polarizable conductor calculation model (CPCM) (9) using the united atom topological model (radii=uaks). The chosen solvent was diethylether.

- 1. Zhao, Y. and Truhlar, D. G. J. Chem. Phys. 2006, 125, 194101.
- 2. Zhao, Y. and Truhlar, D. G. Acc. Chem. Res. 2008, 41, 157
- 3. Zhao, Y. and Truhlar, D. G. Theor. Chem. Acc. 2008, 120, 215
- (a) Dunning, Jr., T.H. J. Chem. Phys. 1989, 90, 1007; (b) Woon, D.E. and Dunning, Jr., T.H. J. Chem. Phys. 1993, 98, 1358
- 5. Peterson, K.A.; Shepler, B.C.; Figgen, D. and Stoll, H. J. Phys. Chem. A 2006, 110, 13877
- 6. (a) Feller, D., J. Comp. Chem., 1996, 17(13), 1571; (b) Schuchardt, K.L., Didier, B.T., Elsethagen, T., Sun, L., Gurumoorthi, V., Chase, J., Li, J., and Windus, T.L. J. Chem. Inf. Model., 2007, 47(3), 1045
- 7. Papajak, E.; Zheng, J.; Xu, X.; Leverentz, H. R.; Truhlar, D. G. J. Chem. Theory Comput., 2011, 7, 3027
- Gaussian 09, Revision B.01, Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Mennucci, B.; Petersson, G. A.; Nakatsuji, H.; Caricato, M.; Li, X.; Hratchian, H. P.; Izmaylov, A. F.; Bloino, J.; Zheng, G.; Sonnenberg, J. L.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Montgomery, J. A., Jr.; Peralta, J. E.; Ogliaro, F.; Bearpark, M.; Heyd, J. J.; Brothers, E.; Kudin, K. N.; Staroverov, V. N.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Rega, N.; Millam, M. J.; Klene, M.; Knox, J. E.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Zakrzewski, V. G.; Voth, G. A.; Salvador, P.; Dannenberg, J. J.; Dapprich, S.; Daniels, A. D.; Farkas, Ö.; Foresman, J. B.; Ortiz, J. V.; Cioslowski, J.; Fox, D. J. Gaussian, Inc., Wallingford CT, 2009.
- (a) Barone, V.; Cossi, M. J. Phys. Chem. A 1998, 102, 1995; (b) Cossi, M.; Rega, N.; Scalmani, G. and Barone, V. J. Comp. Chem. 2003, 24, 669

Scheme S13: Regioselectivity observed with 3.

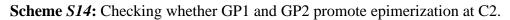


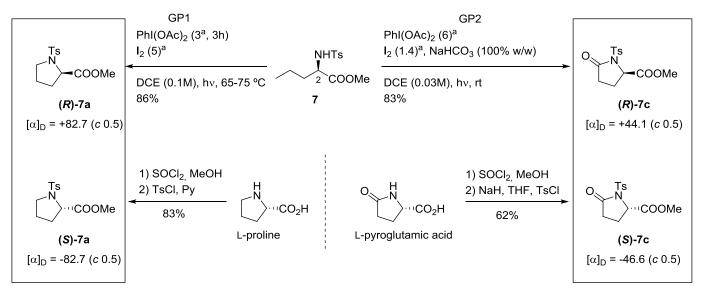
Note: The step of hydrogen atom transfer from C–H bond to Nitrogen centered radical is determinant for the regioselectivity observed within the functionalization of **3**. $BDE_{C4-H} = 93.9 \text{ kcal/mol}$; $BDE_{C8-H} = 97.5 \text{ kcal/mol}$



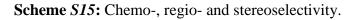
Schematic representation of the Gibbs Free Energy Surface (FES) (at 298.15 K; kcal/mol) showing the equilibrium structures involved in the functionalization of **3**. Solvent effects have been taken into account as described above. Notice the larger stability (4.8 kcal/mol) of the methylene radical over the methyl radical in agreement with the observed regioselectivity for this functionalization.

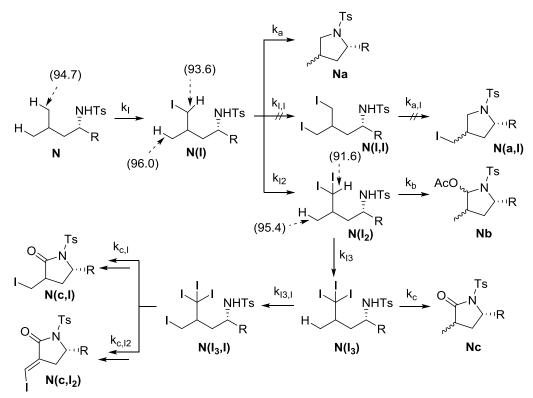
ADDITIONAL SCHEMES





^a millimoles per millimol of **7**.



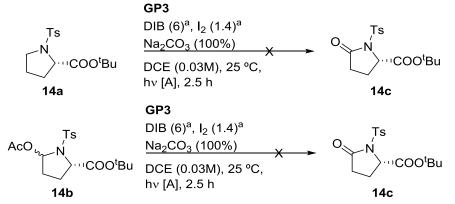


^a Values in parentheses indicate ab initio C–H BDE (kcal mol⁻¹) for derivatives of **8** (N = 8) at the M06-2X/cc-pVTZ level.

Experimental data support the following conclusions:

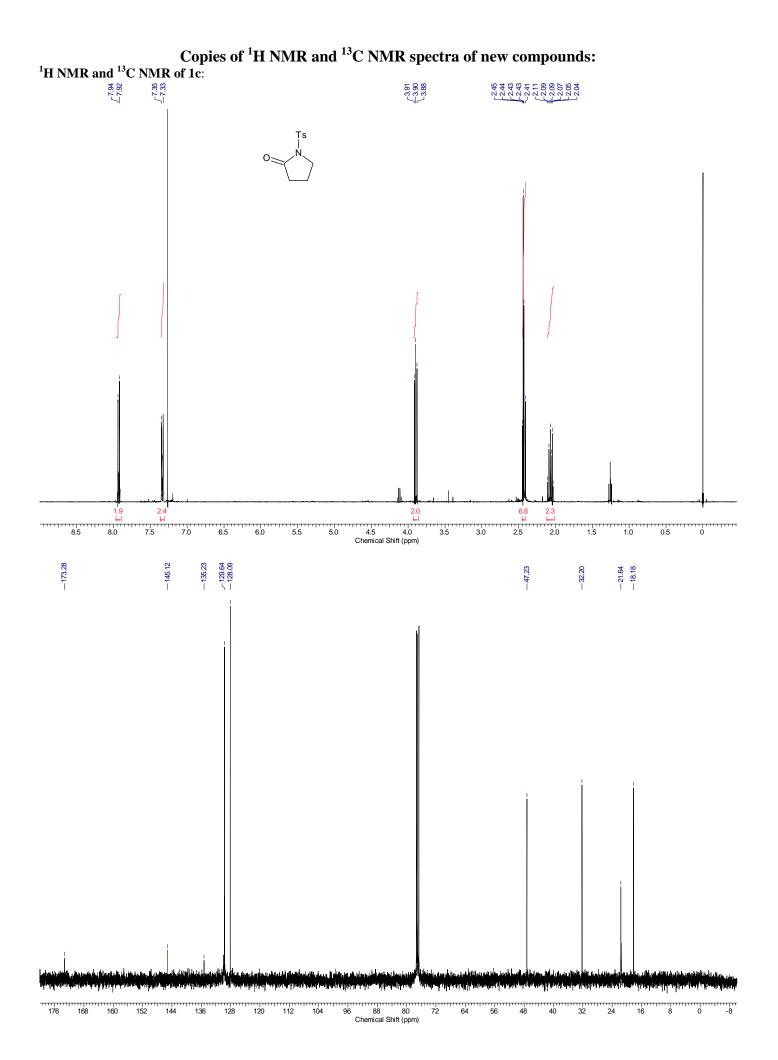
- $k_{I3} > k_{I2} > k_I$
- k_{I2} and $k_a > k_{I,I}$
- $k_{I3} > k_b$ (under irradiation)
- $k_c > k_{I3,I}$

Scheme *S16*: Determination whether **14a** and **14b** are intermediates toward **14c**.

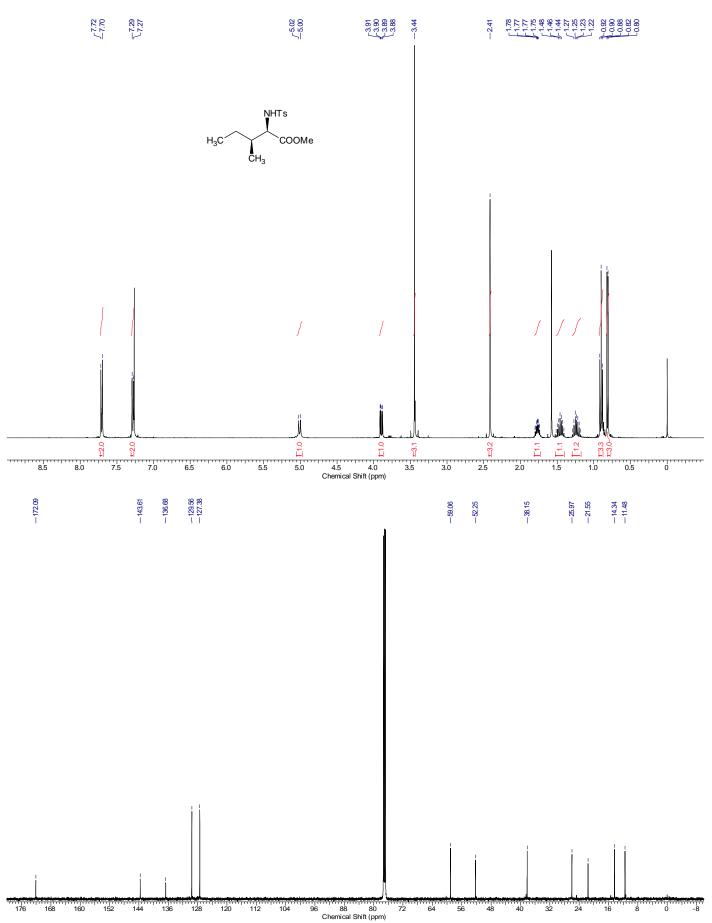


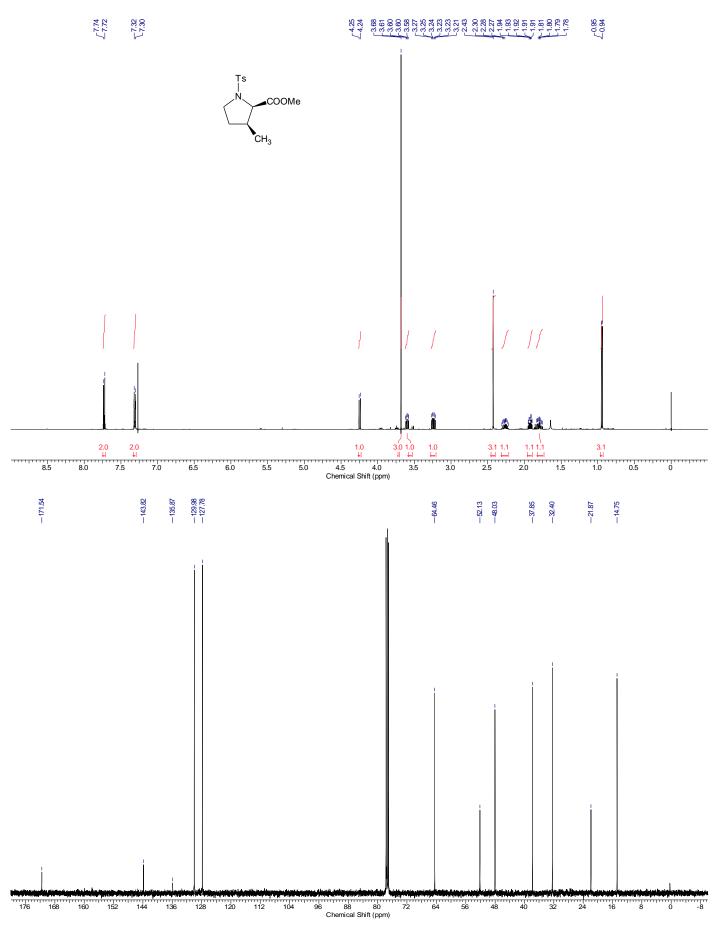
^a millimoles per millimol of 14a and 14b, respectively.

Note: starting material (14a and 14b) was completely recovered after 2.5 h of irradiation.

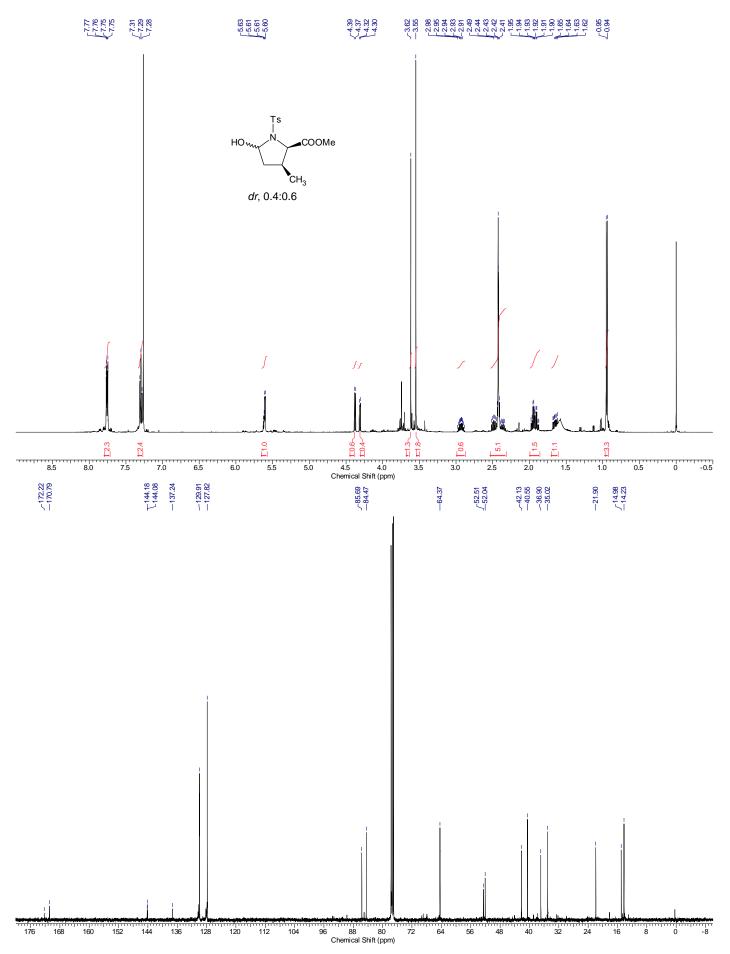


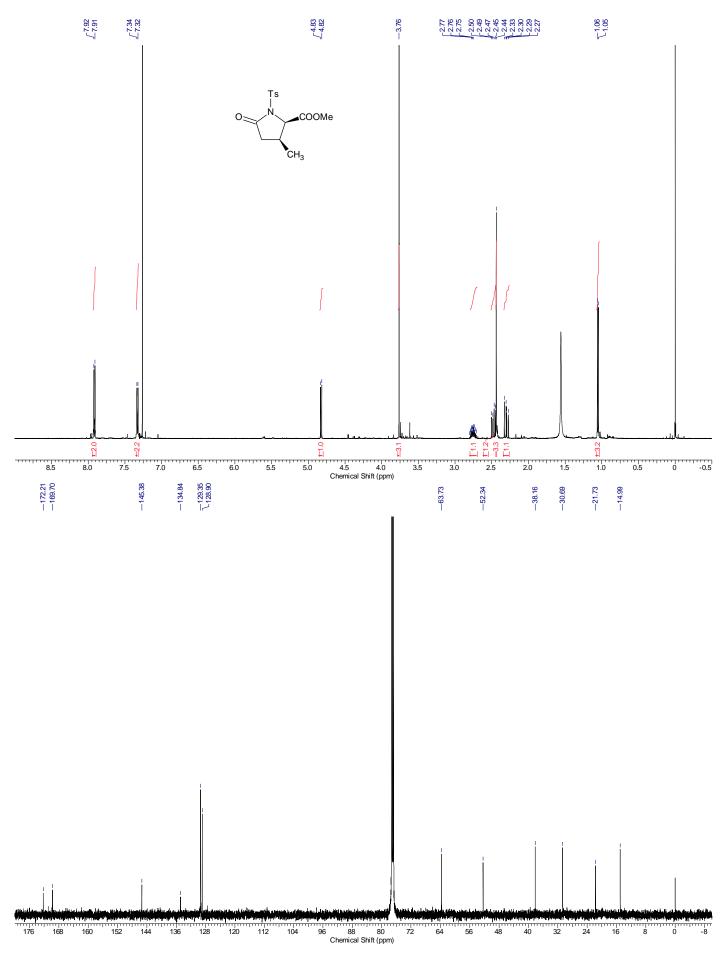
¹H NMR and ¹³C NMR of 2:

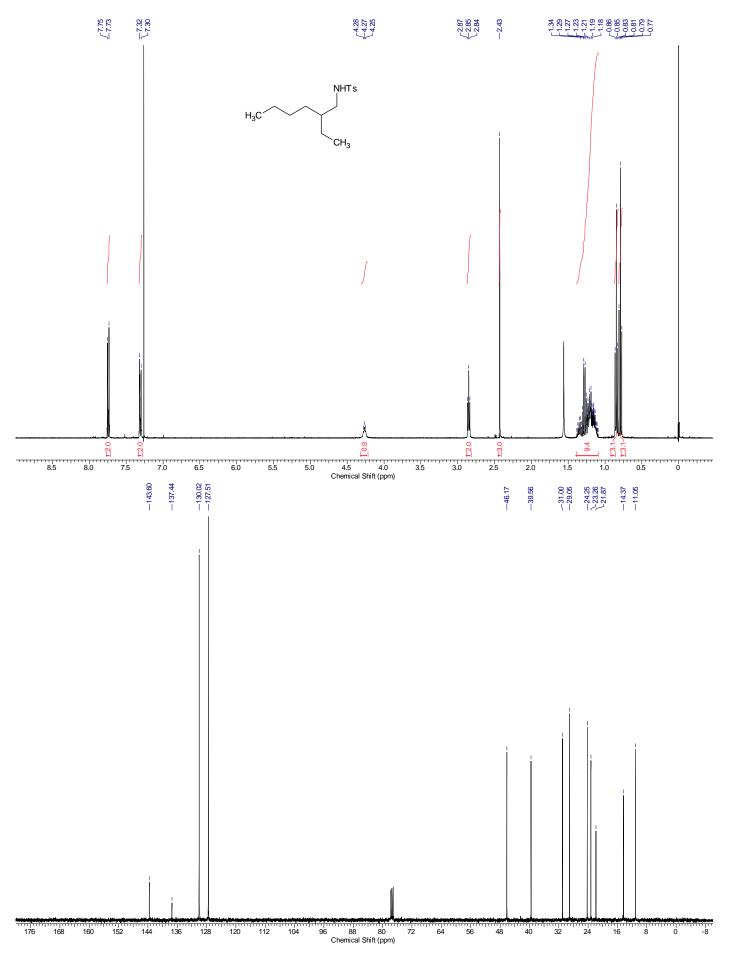


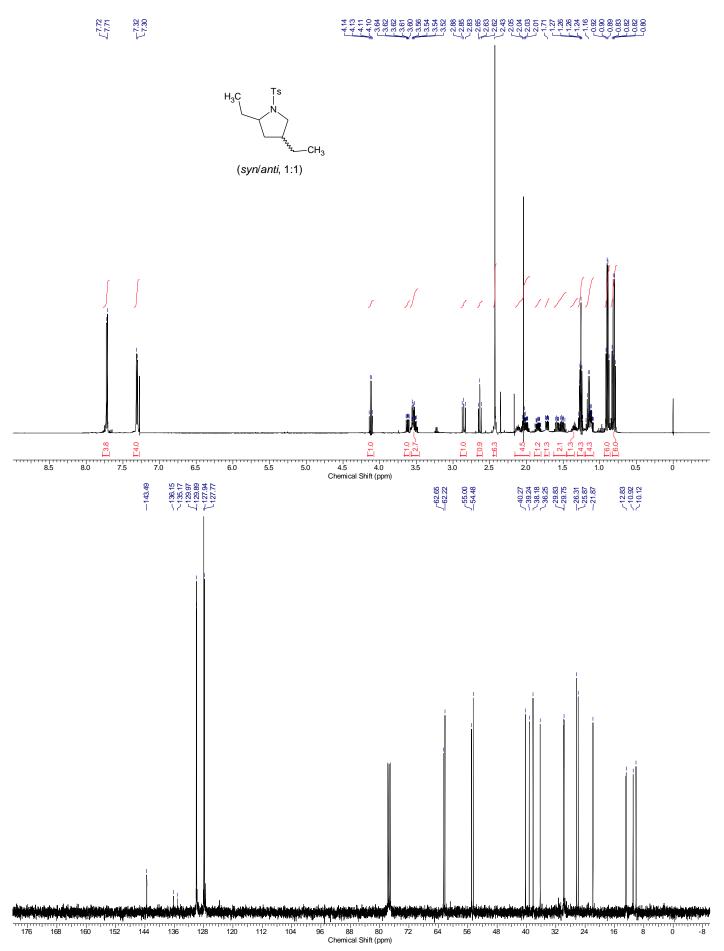


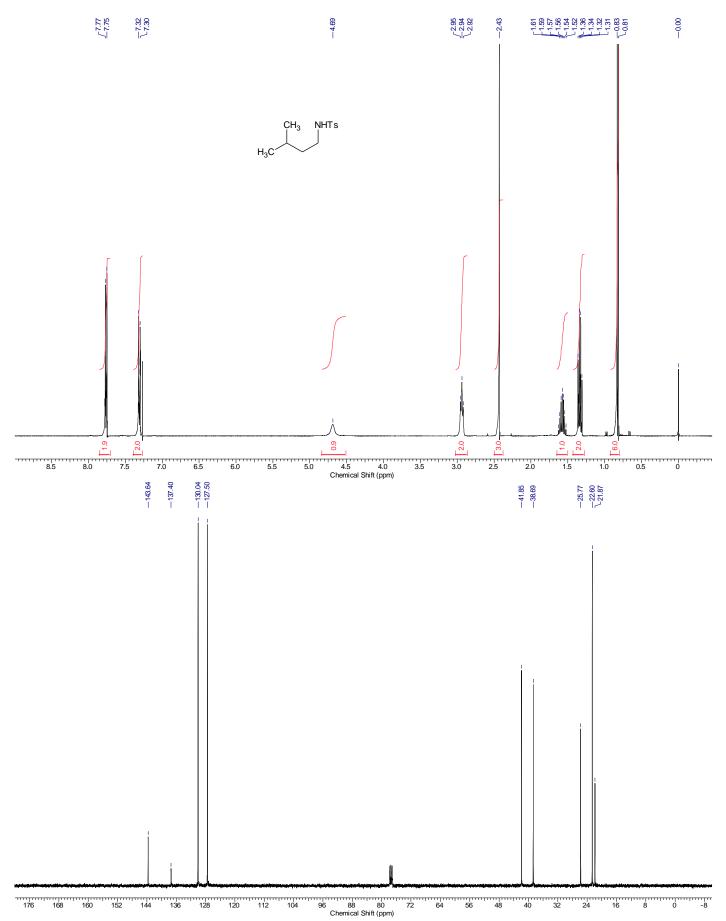
¹H NMR and ¹³C NMR of 2b:











S62

