

Supporting information for

**Phase-vanishing Method with Acetylene Evolution and Its Utilization
in Several Organic Syntheses**

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Table of Contents

I. General Remarks	S2
II. Materials	S2
III. General Procedure for the Preparation of Azides 3a-h	S2
IV. Typical Procedure for Sonogashira coupling by the Phase-vanishing Method.....	S3
V. Typical Procedure for Copper-catalyzed Azide-Alkyne Cycloaddition (CuAAC) by the Phase-vanishing Method	S3
VI. Typical Procedure for Aldehyde-Alkyne-Amine Coupling (A^3 -coupling) by the Phase-vanishing Method	S4
VII. Spectroscopic Data.....	S5
VIII. References	S10
IX. NMR Spectra.....	S11

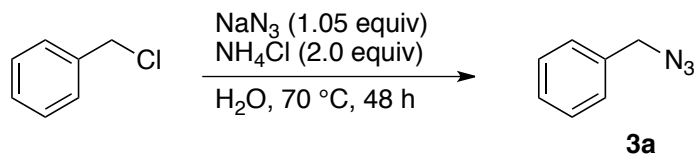
I. General Remarks

Melting points were obtained with Yanako micro melting point apparatus and are not corrected. Products were purified by column chromatography on silica gel (Kanto Chemical Co., Inc., Silica Gel 60N (spherical, neutral), 63-210 mm). ^1H NMR spectra were recorded with a JEOL-ECP-500 (500 MHz) and a JEOL-ECS-400 (400 MHz) spectrometer in CDCl_3 . Chemical shifts were reported in parts per million (δ) referenced to the solvent peak at 7.26 ppm. ^{13}C NMR spectra were recorded with a JEOL-ECP-500 (126 MHz) and a JEOL-ECS-400 (100 MHz) spectrometer in CDCl_3 and referenced to the solvent peak at 77.0 ppm. Coupling constants, J , were reported in Hertz (Hz), and splitting patterns were designated as s (singlet), d (doublet), t (triplet), q (quartet), quint (quintet) and m (multiplet). IR spectra were obtained on a JASCO FT/IR-4100 spectrometer; absorptions were reported in reciprocal centimeters. Conventional mass spectra were recorded with a SHIMADZU GCMS-QP2010Plus spectrometer and high-resolution mass spectra were recorded with a JEOL MS-700 spectrometer.

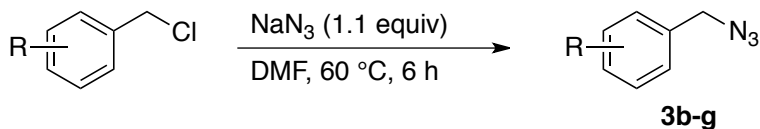
II. Materials

Benzyl azides (**3a-3h**) were prepared according to procedures (Methods A–C) in literature. The other compounds were purchased from commercially available sources and used without further purification.

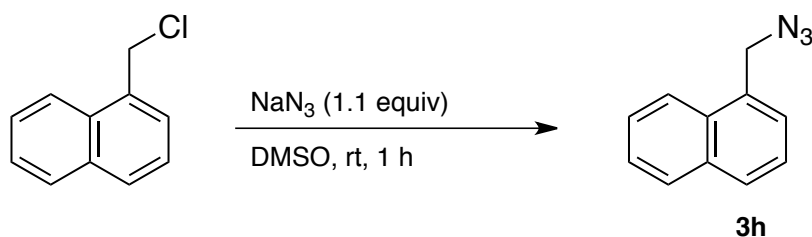
III. General Procedures for the Preparation of Azides 3a–h



Method A^{S1}: Benzyl chloride (10 mmol) was suspended in water at a concentration of 1.5 M. To the suspension, sodium azide (10.6 mmol) and ammonium chloride (20 mmol) were added, and the mixture was heated at $70\text{ }^\circ\text{C}$ for 48 h with vigorous stirring. After cooling, the mixture was extracted with diethyl ether, and collected organic layer was dried over sodium sulfate. After filtration, evaporation of the solvent yielded pure benzyl azide.

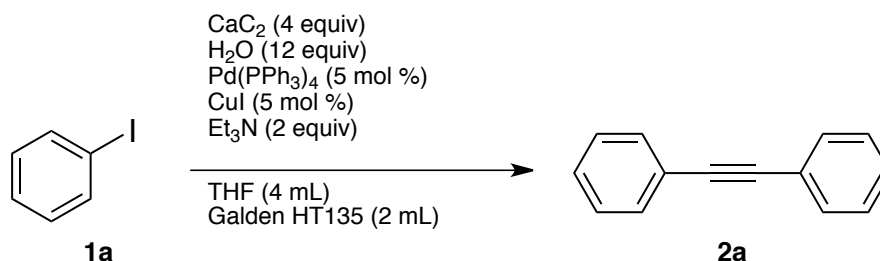


Method B^{S2}: A mixture of substituted benzyl chloride (10 mmol) and sodium azide (11 mmol) in DMF (5 mL) was gradually heated to $60\text{ }^\circ\text{C}$ and stirred for 6 h. After cooling, diethyl ether and water were added to the reaction mixture. The organic layer was separated and the aqueous layer was extracted with diethyl ether. The combined organic layers were washed with water and brine, and dried over sodium sulfate. After filtration, evaporation of the solvent yielded the analytically pure substituted azidomethylbenzenes.



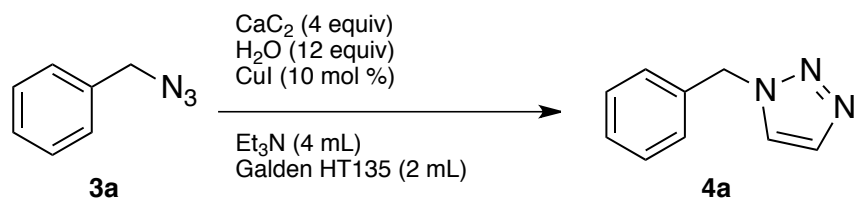
Method C^{S3}. To a solution of sodium azide (11 mmol) in DMSO (5 mL), 1-(chloromethyl)naphthalene (10 mmol) was added, and the mixture was stirred at room temperature for 1 h. After checking that the starting material (chloromethylnaphthalene) was consumed, the reaction was quenched with water and allowed to cool to room temperature with stirring. The mixture was then extracted with ethyl acetate. The collected organic layer was washed with water and brine, and dried over sodium sulfate. After filtration, the solvent was evaporated and the residue was purified by column chromatography on silica gel to afford 1-(azidomethyl)naphthalene.

IV. Typical Procedure for Sonogashira Coupling by the Phase-vanishing Method (Table 1, entry 1)



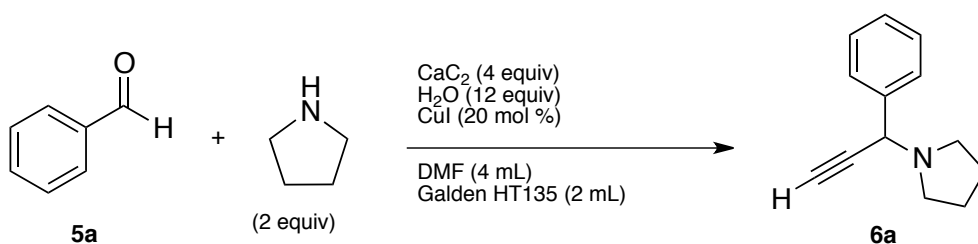
Calcium carbide (512 mg, 8.0 mmol) and a magnetic stirring bar (oval, 10 mm × 5 mm) were placed in a Pyrex test tube (15 mm ϕ × 130 mm), to which Galden HT135 (2 mL) was added slowly using a syringe. Subsequently, water (420 mg, 23.3 mmol), a solution of phenyl iodide (408 mg, 2.0 mmol) in THF (4 mL), Pd(PPh₃)₄ (115 mg, 0.10 mmol), CuI (24 mg, 0.13 mmol), and triethylamine (409 mg, 4.0 mmol) were added slowly in order, forming four layers. A rubber septum was fitted to the test tube, and a needle equipped with a balloon, which acted as a reservoir of acetylene gas during the reaction, was then pricked into the septum. The air in the test tube was removed by a syringe until the balloon was completely flattened. The test tube was heated in an oil bath at 55 °C for 20 h with slow stirring, taking care not to mix the layers, then allowed to cool to ambient temperature. After removal of the Galden HT135 by glass pipette, the organic layer and inorganic salts were moved to a separatory funnel using diethyl ether and water. The organic layer was separated, and the aqueous layer was extracted with diethyl ether. The combined organic layers were then dried over sodium sulfate, filtered, and concentrated. Purification by column chromatography on silica gel with hexane gave diphenylacetylene as white crystalline solid (165 mg, 94%).

V. Typical Procedure for Copper-catalyzed Azide-Alkyne Cycloaddition (CuAAC) by the Phase-vanishing Method (Table 2, entry 1)



Calcium carbide (259 mg, 4.0 mmol) and a magnetic stirring bar (oval, 10 mm \times 5 mm) were placed at bottom of a Pyrex test tube (15 mm ϕ \times 130 mm), to which Galden HT135 (2 mL) was added slowly using a syringe. Subsequently, water (300 mg, 16.6 mmol), a solution of benzyl azide (132 mg, 1.0 mmol) in triethylamine (4 mL), and CuI (20 mg, 0.11 mmol) were added slowly in order, forming four layers. A rubber septum was fitted to the test tube, and a needle equipped with a balloon, which acted as a reservoir of acetylene gas during the reaction, was then pricked into the septum. The air in the test tube was removed by a syringe until the balloon was completely flattened. The test tube was heated in an oil bath at 55 $^\circ\text{C}$ for 20 h with slow stirring, taking care not to mix the layers, and then allowed to cool to ambient temperature. The organic layer was taken up with a glass pipette. Ethyl acetate was placed on the residual Galden HT135 layer, and then decanted. The collected organic layer was mixed with hydrochloric acid (2 M). After checking that the pH was less than 7 with pH indicator paper, the layers were separated, and the aqueous layer was extracted with ethyl acetate. The combined organic layers were dried over sodium sulfate, filtered, and concentrated. Purification by column chromatography on silica gel with hexane–ethyl acetate (1:1) gave 1-benzyl-1H-1,2,3-triazole as pale yellow crystalline solid (137 mg, 85%).

VI. Typical Procedure for Aldehyde-Alkyne-Amine Coupling (A^3 -coupling) by the Phase-vanishing Method (Table 3, entry 1)

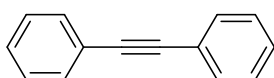


Calcium carbide (259 mg, 4.0 mmol) and a magnetic stirring bar (oval, 10 mm \times 5 mm) were placed in a Pyrex test tube (15 mm ϕ \times 130 mm), to which Galden HT135 (2 mL) was added slowly using a syringe. Subsequently, water (250 mg, 13.9 mmol), a solution of benzaldehyde (**1a**) (110 mg, 1.0 mmol) in DMF (4 mL), pyrrolidine (147 mg, 2.1 mmol), and CuI (40 mg, 0.21 mmol) were added slowly in order, forming four layers. A rubber septum was fitted to the test tube, and a needle equipped with a balloon, which acted as a reservoir of acetylene gas during the reaction, was then pricked into the septum. The air in the test tube was removed by a syringe until the balloon was completely flattened. The test tube was heated in an oil bath at

55 °C for 20 h with slow stirring, taking care not to mix the layers, and then allowed to cool to ambient temperature. After removal of the Galden HT135 by glass pipette, the organic layer and inorganic salts were moved to a separatory funnel using diethyl ether and water. The organic layer was separated, and the aqueous layer was extracted with diethyl ether. The combined organic layers were then dried over sodium sulfate, filtered, and concentrated. Purification by a column chromatography on silica gel with hexane-ethyl acetate (4/1) gave 1-(1-phenylprop-2-ynyl)pyrrolidine (**6a**) as reddish yellow oil (144 mg, 75%).

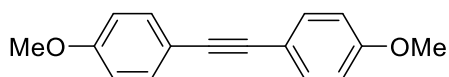
VII. Spectroscopic Data

Diphenyl acetylene (**2a**)^{S4}:



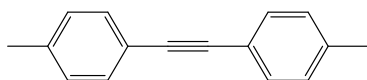
White crystalline solid (165 mg); ¹H NMR (400 MHz, CDCl₃): δ 7.56–7.53 (m, 4H), 7.38–7.33 (m, 6H); ¹³C NMR (126 MHz, CDCl₃): δ 131.50, 128.23, 128.14, 123.16, 89.26.

1,2-Bis(4-methoxyphenyl)acetylene (**2b**)^{S4}:



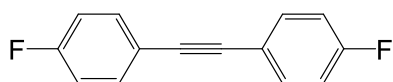
Yellow crystalline solid (186 mg); ¹H NMR (400 MHz, CDCl₃): δ 7.45 (d, *J* = 8.0 Hz, 4H), 6.87 (d, *J* = 8.0 Hz, 4H), 3.83 (s, 6H); ¹³C NMR (100 MHz, CDCl₃): δ 159.32, 132.81, 115.65, 113.91, 87.90, 55.21.

Bis(p-tolyl)acetylene (**2c**)^{S4}:



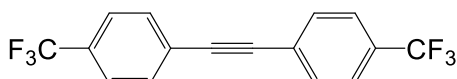
Pale yellow crystalline solid (196 mg); ¹H NMR (500 MHz, CDCl₃): δ 7.41 (d, *J* = 8.3 Hz, 4H), 7.15 (d, *J* = 8.3 Hz, 4H), 2.36 (s, 6H); ¹³C NMR (126 MHz, CDCl₃): δ 138.15, 131.41, 129.07, 120.37, 88.85, 21.48.

1,2-Bis(4-fluorophenyl)acetylene (**2d**)^{S4}:



Pale yellow crystalline solid (208 mg); ¹H NMR (500 MHz, CDCl₃): δ 7.50 (dt, *J* = 8.7, 3.2 Hz, 4H), 7.05 (t, *J* = 8.7 Hz, 4H); ¹³C NMR (126 MHz, CDCl₃): δ 162.33 (d, *J*_{C-F} = 251 Hz), 133.23 (d, *J*_{C-F} = 8.6 Hz), 119.00, 115.53 (d, *J*_{C-F} = 22.1 Hz), 87.84.

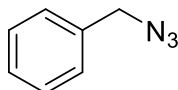
1,2-Bis(4-(trifluoromethyl)phenyl)acetylene (**2e**)^{S4}:



Greenish pale yellow crystalline solid (302 mg); ¹H NMR

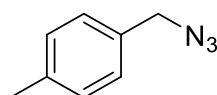
(500 MHz, CDCl₃): δ 7.67–7.62 (m, 8H); ¹³C NMR (100 MHz, CDCl₃): δ 131.96, 130.46 (q, $J_{\text{C-F}} = 32.8$ Hz), 126.36, 125.37 (broad d, $J_{\text{C-F}} = 2.9$ Hz), 123.84 (q, $J_{\text{C-F}} = 271.6$ Hz), 90.11.

Benzyl azide (3a)^{S5}:



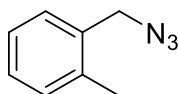
Pale yellow oil (1.15 g); ¹H NMR (400 MHz, CDCl₃): δ 7.42–7.26 (m, 5H), 4.34 (s, 2H); ¹³C NMR (100 MHz, CDCl₃): δ 135.23, 128.69, 128.12, 128.05, 54.56.

1-(Azidomethyl)-4-methylbenzene (3b)^{S2}:



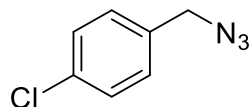
Pale yellow oil (646 mg); ¹H NMR (400 MHz, CDCl₃): δ 7.22–7.20 (m, 4H), 4.29 (s, 2H), 2.36 (s, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 138.15, 132.26, 129.50, 128.27, 54.56, 21.18.

1-(Azidomethyl)-2-methylbenzene (3c)^{S6}:



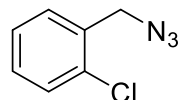
Pale yellow oil (617 mg); ¹H NMR (400 MHz, CDCl₃): δ 7.27–7.21 (m, 4H), 4.35 (s, 2H), 2.37 (s, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 136.77, 133.37, 130.64, 129.31, 128.62, 126.20, 53.03, 18.96.

1-(Azidomethyl)-4-chlorobenzene (3d)^{S2}:



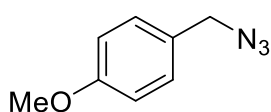
Pale yellow oil (799 mg); ¹H NMR (400 MHz, CDCl₃): δ 7.38–7.34 (m, 2H), 7.27–7.24 (m, 2H), 4.32 (s, 2H); ¹³C NMR (100 MHz, CDCl₃): δ 134.21, 133.84, 129.49, 129.02, 54.02.

1-(Azidomethyl)-4-chlorobenzene (3e)^{S2}:



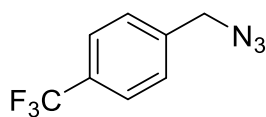
Pale yellow liquid (801 mg); ¹H NMR (400 MHz, CDCl₃): δ 7.44–7.39 (m, 2H), 7.32–7.26 (m, 2H), 4.50 (s, 2H); ¹³C NMR (100 MHz, CDCl₃): δ 133.77, 133.28, 130.00, 129.78, 129.64, 127.15, 52.26.

1-(Azidomethyl)-4-methoxybenzene (3f)^{S2}:



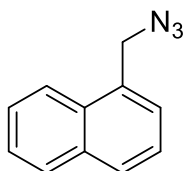
Pale red oil (791 mg); ¹H NMR (400 MHz, CDCl₃): δ 7.25 (d, $J = 8.0$ Hz, 2H), 6.91 (d, $J = 8.0$ Hz, 2H), 4.27 (s, 2H), 3.82 (s, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 159.61, 129.74, 127.37, 114.16, 55.24, 54.35.

1-(Azidomethyl)-4-trifluoromethylbenzene (3g)^{S5}:



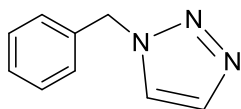
Yellow oil (976 mg); ^1H NMR (400 MHz, CDCl_3): δ 7.65 (d, J = 8.0 Hz, 2H), 7.45 (d, J = 8.0 Hz, 2H), 4.43 (s, 2H); ^{13}C NMR (100 MHz, CDCl_3): δ 139.41, 130.43 (q, $J_{\text{C-F}}$ = 32.4 Hz), 128.24, 125.78 (q, $J_{\text{C-F}}$ = 3.8 Hz), 123.95 (q, $J_{\text{C-F}}$ = 269 Hz), 54.04.

1-(Azidomethyl)naphthalene (3h)^{S3}:



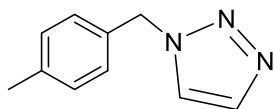
Pale yellow oil (1.75 g); ^1H NMR (400 MHz, CDCl_3): δ 8.03 (d, J = 8.0 Hz, 1H), 7.91–7.87 (m, 2H), 7.59–7.46 (m, 4H), 4.77 (s, 2H); ^{13}C NMR (100 MHz, CDCl_3): δ 133.73, 131.18, 130.81, 129.25, 128.66, 127.08, 126.57, 126.00, 125.06, 123.33, 52.76.

1-Benzyl-1H-1,2,3-triazole (4a)^{S7}:



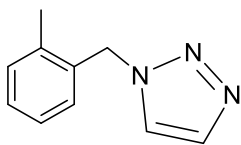
Pale yellow crystalline solid (137 mg); ^1H NMR (400 MHz, CDCl_3): δ 7.71 (s, 1H), 7.47 (s, 1H), 7.47–7.35 (m, 3H), 7.28–7.25 (m, 2H), 5.57 (s, 2H); ^{13}C NMR (100 MHz, CDCl_3): δ 134.66, 134.23, 129.09, 128.71, 127.98, 123.26, 53.94.

1-(4-Methylbenzyl)-1H-1,2,3-triazole (4b)^{S7}:



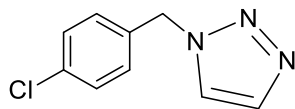
White crystalline solid (101 mg); ^1H NMR (400 MHz, CDCl_3): δ 7.69 (s, 1H), 7.47 (s, 1H), 7.17 (m, 4H), 5.52 (s, 2H), 2.35 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3): δ 138.15, 133.72, 131.46, 129.35, 127.68, 123.12, 53.29, 20.77.

1-(2-Methylbenzyl)-1H-1,2,3-triazole (4c)^{S7}:



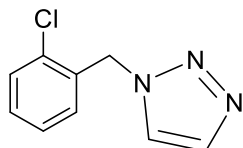
Light brown crystalline solid (100 mg); ^1H NMR (400 MHz, CDCl_3): δ 7.69 (s, 1H), 7.35 (s, 1H), 7.30–7.13 (m, 4H), 5.58 (s, 2H), 2.78 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3): δ 136.49, 133.63, 132.38, 130.64, 128.93, 128.72, 126.30, 123.08, 51.74, 18.61.

1-(4-Chlorobenzyl)-1H-1,2,3-triazole (4d)^{S7}:



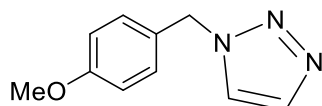
Light brown crystalline solid (170 mg); ^1H NMR (400 MHz, CDCl_3): δ 7.73 (s, 1H), 7.48 (s, 1H), 7.37–7.34 (m, 2H), 7.22–7.19 (m, 2H), 5.54 (s, 2H); ^{13}C NMR (100 MHz, CDCl_3): δ 134.31, 134.02, 133.11, 129.08, 128.95, 123.33, 52.84.

1-(2-Chlorobenzyl)-1H-1,2,3-triazole (4e)^{S7}:



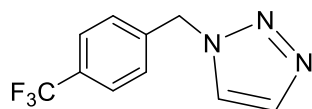
Light brown crystalline solid (162 mg); ¹H NMR (400 MHz, CDCl₃): δ 7.73 (s, 1H), 7.59 (s, 1H), 7.45–7.43 (m, 1H), 7.32–7.27 (m, 2H), 7.18–7.15 (m, 1H), 5.71 (s, 2H); ¹³C NMR (100 MHz, CDCl₃): δ 133.82, 133.14, 132.31, 129.97×2, 129.62, 127.34, 123.67, 50.97.

1-(4-Methoxybenzyl)-1H-1,2,3-triazole (4f)^{S7}:



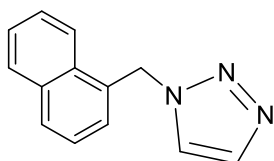
Light brown crystalline solid (157 mg); ¹H NMR (400 MHz, CDCl₃): δ 7.69 (s, 1H), 7.43 (s, 1H), 7.23 (d, *J* = 8.4 Hz, 2H), 6.91–6.89 (m, 2H), 5.50 (s, 2H), 3.81 (s, 3H); ¹³C NMR (100 MHz, CDCl₃): δ 159.0, 133.73, 129.27, 126.46, 123.00, 114.08, 54.99, 53.08.

1-(4-Trifluoromethylbenzyl)-1H-1,2,3-triazole (4g):



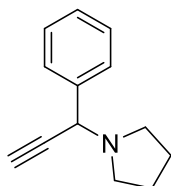
Light brown crystalline solid (168 mg); mp 52.5–53.0 °C; ¹H NMR (400 MHz, CDCl₃): δ 7.76 (s, 1H), 7.64 (d, *J* = 8.0 Hz, 2H), 7.52 (s, 1H), 7.36 (d, *J* = 8.0 Hz, 2H), 5.64 (s, 2H); ¹³C NMR (100 MHz, CDCl₃): δ 138.68, 134.16, 130.51 (q, *J*_{C-F} = 32.5 Hz), 127.91, 125.76 (d, *J*_{C-F} = 4.0 Hz), 123.65, 123.60 (q, *J*_{C-F} = 270.2 Hz), 52.93; IR (KBr): 2965, 2927, 2857, 2812, 2076, 1444, 1123 883, 747 cm⁻¹; GC-MS (EI): 227 (5), 198 (47), 172 (19), 159 (100), 130 (58), 109 (62), 40 (21 %); HRMS (EI) Calcd. For C₁₀H₈N₃F₃: 227.0670, Found: 227.0672.

1-(1-Naphthylmethyl)-1H-1,2,3-triazole (4h)^{S8}:



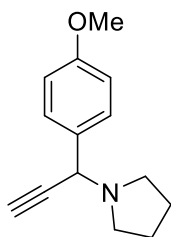
Off-white crystalline solid (185 mg); ¹H NMR (400 MHz, CDCl₃): δ 7.97–7.89 (m, 3H), 7.65 (s, 1H), 7.54–7.34 (m, 4H), 7.26 (s, 1H), 6.03 (s, 2H); ¹³C NMR (100 MHz, CDCl₃): δ 133.76, 133.59, 130.82, 129.66, 128.65, 127.44, 126.93, 126.10, 125.11, 123.20, 122.6, 51.72.

1-(1-phenylprop-2-ynyl)pyrrolidine (6a)^{S9}:



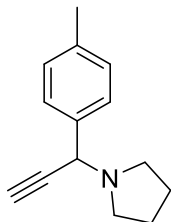
Reddish yellow oil (144 mg); ¹H NMR (400 MHz, CDCl₃): δ 7.54 (d, *J* = 3.6 Hz, 2H), 7.36–7.28 (m, 3H), 4.68 (d, *J* = 1.2 Hz, 1H), 2.61 (t, *J* = 6.8 Hz, 4 H), 2.48 (d, *J* = 1.2 Hz, 1H), 1.79–1.76 (m, 4H); ¹³C NMR (100 MHz, CDCl₃): δ 138.95, 128.22, 128.08, 127.58, 80.76, 74.52, 58.33, 50.01, 23.38.

1-(1-(4-Methoxyphenyl)prop-2-ynyl)pyrrolidine (6b)^{S10}:



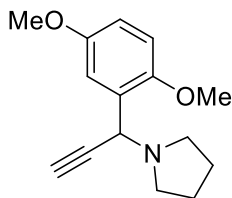
Reddish yellow oil (89 mg); ¹H NMR (400 MHz, CDCl₃): δ 7.45 (d, *J* = 8.8 Hz, 2H), 6.87 (d, *J* = 8.4 Hz), 4.61 (d, *J* = 2.0 Hz, 1H), 3.81 (s, 3H), 2.61-2.57 (m, 4H), 2.46 (d, *J* = 2.4 Hz, 1H), 1.77-1.75 (m, 4H); ¹³C NMR (100 MHz, CDCl₃): δ 158.93, 131.14, 129.11, 113.45, 81.03, 74.23, 57.64, 55.14, 49.91, 23.30.

1-(1-(4-Methylphenyl)prop-2-ynyl)pyrrolidine (6d):



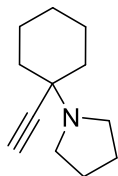
Reddish yellow oil (93 mg); ¹H NMR (400 MHz, CDCl₃): δ 7.42 (d, *J* = 8.0 Hz, 2H), 7.15 (d, *J* = 8.0 Hz, 2H), 4.63 (d, *J* = 2.0 Hz, 1H), 2.60 (t, *J* = 7.2 Hz, 4H), 2.46 (d, *J* = 2.4 Hz, 1H), 2.34 (s, 3H), 1.78–1.75 (m, 4H); ¹³C NMR (100 MHz, CDCl₃): δ 137.20, 136.01, 128.88, 127.96, 81.03, 74.24, 58.07, 50.02, 23.35, 21.07; IR (neat): 3297, 2966, 2875, 2813, 1511, 1267 cm⁻¹; GC-MS (EI): 199 (29), 129 (100), 108 (79), 70 (40 %); HRMS (EI) Calcd. For C₁₄H₁₇N: 199.1361, Found: 199.1357.

1-(1-(2,5-Dimethylphenyl)prop-2-ynyl)pyrrolidine (6e):



Reddish yellow oil (95 mg); ¹H NMR (400 MHz, CDCl₃): δ 7.20 (d, *J* = 2.4 Hz, 1H), 6.81–6.79 (m, 2H), 5.07 (d, *J* = 2.0 Hz, 1H), 3.81–3.79 (d, *J* = 1.2 Hz, 3H×2), 2.67 (s, 4H), 2.37 (d, *J* = 2.4 Hz, 1H), 1.78–1.77 (m, 4H); ¹³C NMR (100 MHz, CDCl₃): δ 153.32, 150.77, 128.32, 115.32, 113.31, 112.06, 81.80, 72.96, 56.37, 55.63, 51.09, 50.32, 23.26; IR (neat): 3286, 2962, 2833, 1499, 1464, 1279, 1246, 1216 cm⁻¹; GC-MS (EI): 245 (23), 216 (21), 175 (100), 161 (49), 132 (44), 108 (99), 91 (43), 70 (94 %); HRMS (EI) Calcd. For C₁₅H₁₉NO₂: 245.1416, Found: 245.1412.

1-(1-ethynyl-cyclohexyl)-pyrrolidine (6f):

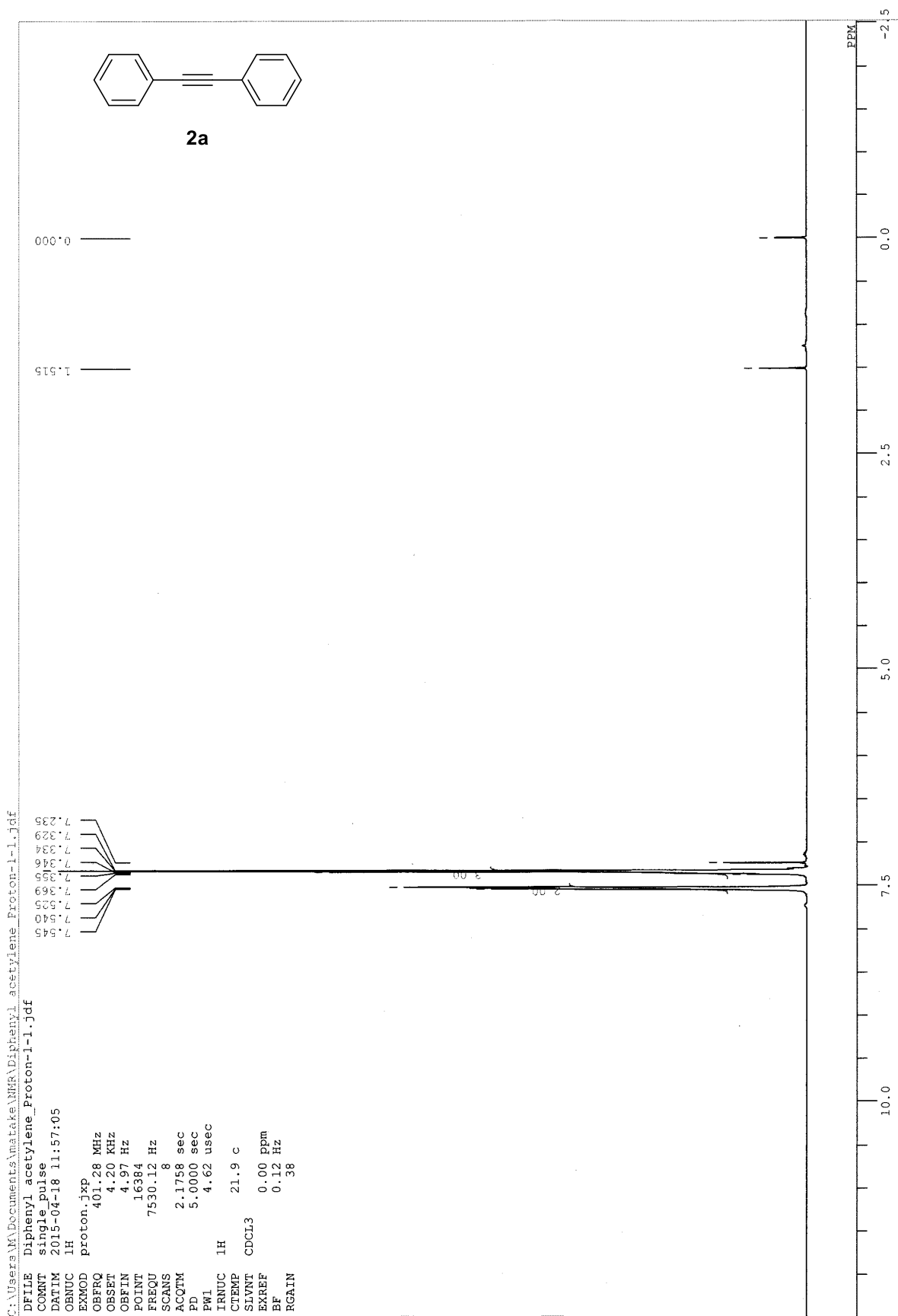


Light brown crystalline solid (55 mg); mp 62.5–63.5 °C; ¹H NMR (400 MHz, CDCl₃): δ 2.73–2.70 (m, 4H), 2.28 (s, 1H), 1.94 (d, *J* = 12.4 Hz, 2H), 1.78 (quint, *J* = 3.6 Hz, 4 H), 1.67–1.46 (m, 8H); ¹³C NMR (100 MHz, CDCl₃): δ 84.48, 72.98, 58.63, 46.78, 37.71, 25.55, 23.35, 22.74; IR (KBr): 3127, 2420, 1623, 1424, 1325, 1217, 1167, 1136, 1107, 1065, 830, 798 cm⁻¹; GC-MS (EI): 177 (11), 162 (18), 148 (21), 134 (100), 120 (21), 79 (10), 70 (26), 65 (10), 41 (10 %); HRMS (EI) Calcd. For C₁₂H₁₉N: 177.1517, Found: 177.1520.

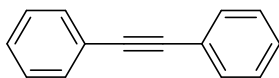
VIII. References

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IX. NMR Spectra

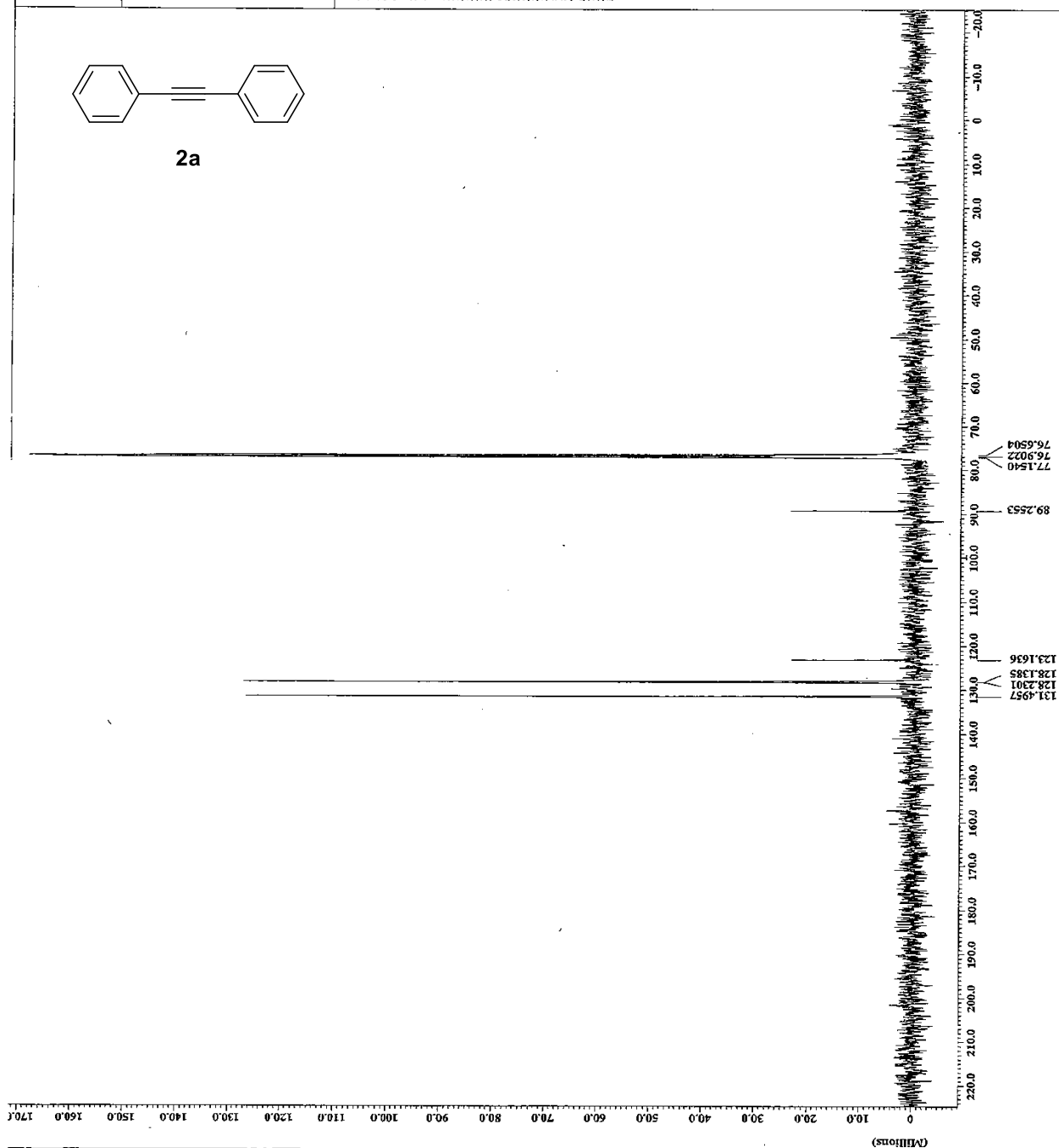


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Acquisition : 77.000 [ppm]



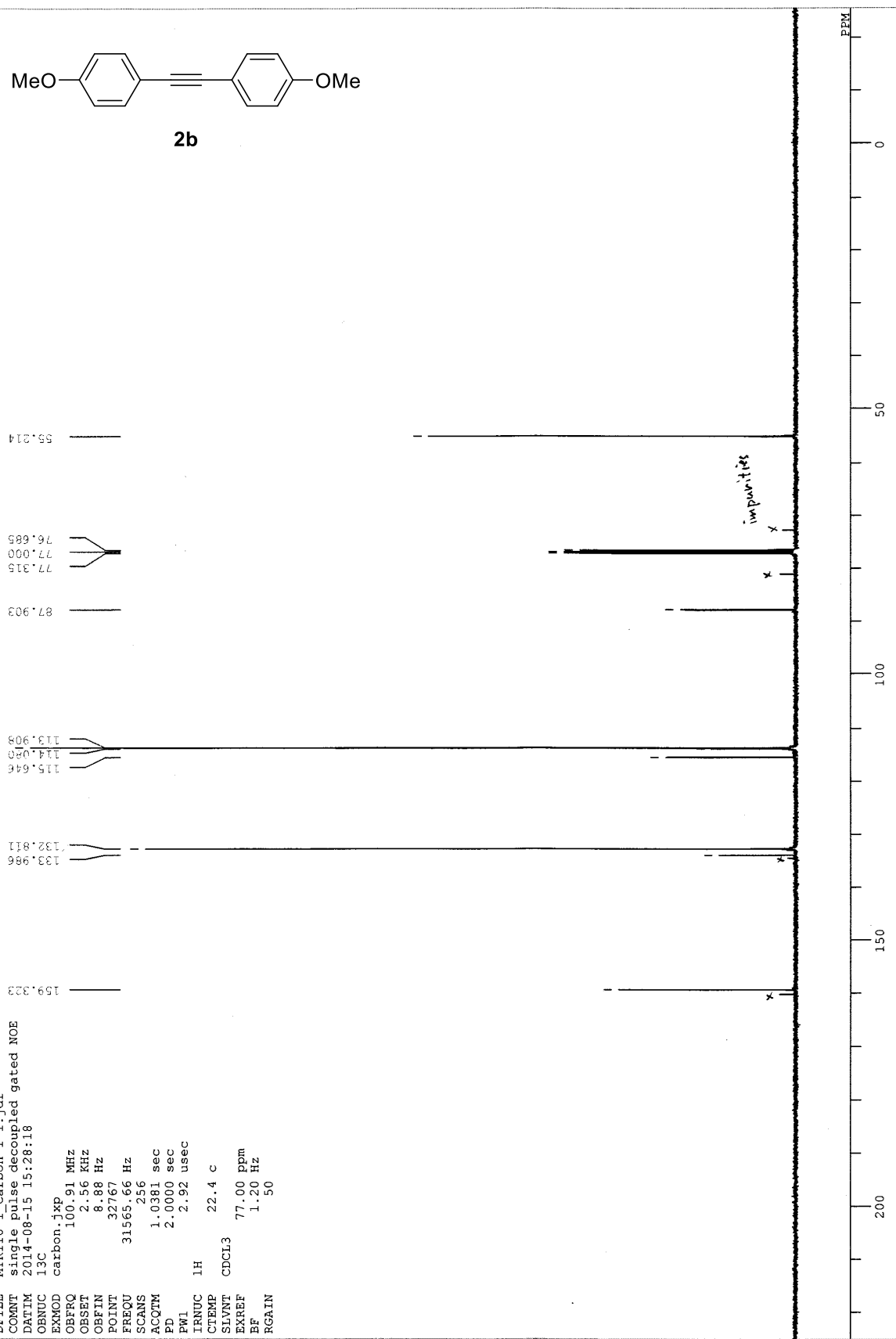
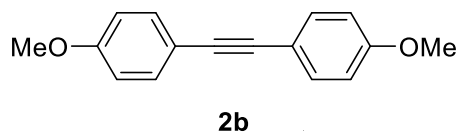
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Sample ID : diphenylacetylene
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Revision Date : 15-MAR-2010 10:11:13
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Data Units : [ppm]
Scans : 256
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X-Offset : 100 [ppm]
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X-Group : 15.7473579 [Hz]
Solvent : CDCl3
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Spin Mod : 30 [Hz]
Recvr Gain : 11.7473579 [V]
Field Strength : 125.7606221 [MHz]
Filter Mod : 15.7473579 [Hz]
Filter Width : 15.7473579 [Hz]



C:\Users\A\Documents\NMR\MTK110-1 Carbon-1-1.jdf

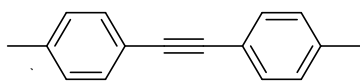
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 POINT 32767
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 ACQTM 1.0381 sec
 PD 2.0000 sec
 PW1 2.92 usec
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 SLVNT CDCL3
 EXREF 77.00 ppm
 BF 1.20 Hz
 RGAIN 50





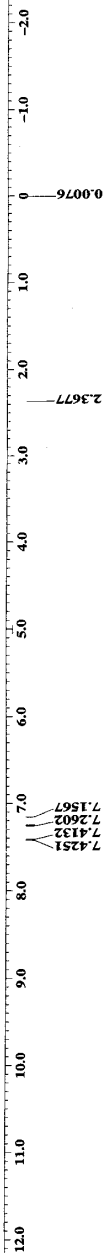
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reference = 7.255 [ppm] : 7.26 [ppm]

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Content = Single Pulse Experiment
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Revision Date = 17-MAR-2010 19:37:39
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Spec Type = DELTA_NMR
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Dim Size = 16384
Dim Units = [ppm]
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Split = 1
Temp_set = 21.4 [°C]
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Filter_mode = BUTTERWORTH
Filter_width = 3.75119936 [kHz]



2c

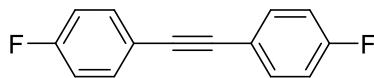
x in purities



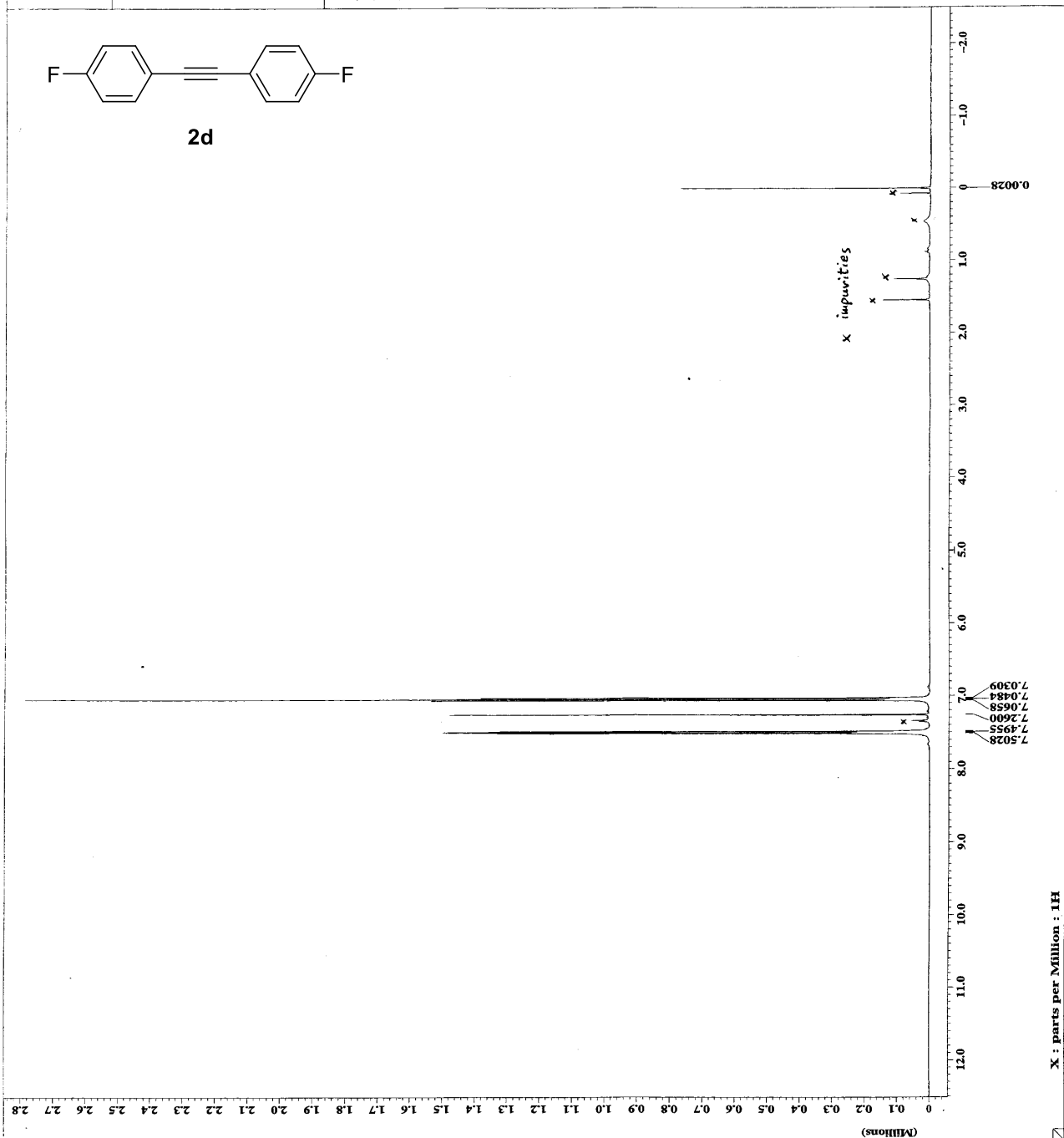
X : parts per Million : 1H

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machinphase
Pyr
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reference : 7.26 [ppm] : 7.26 [ppm]

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Sample ID :
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Scales :
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2d

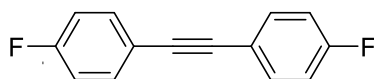


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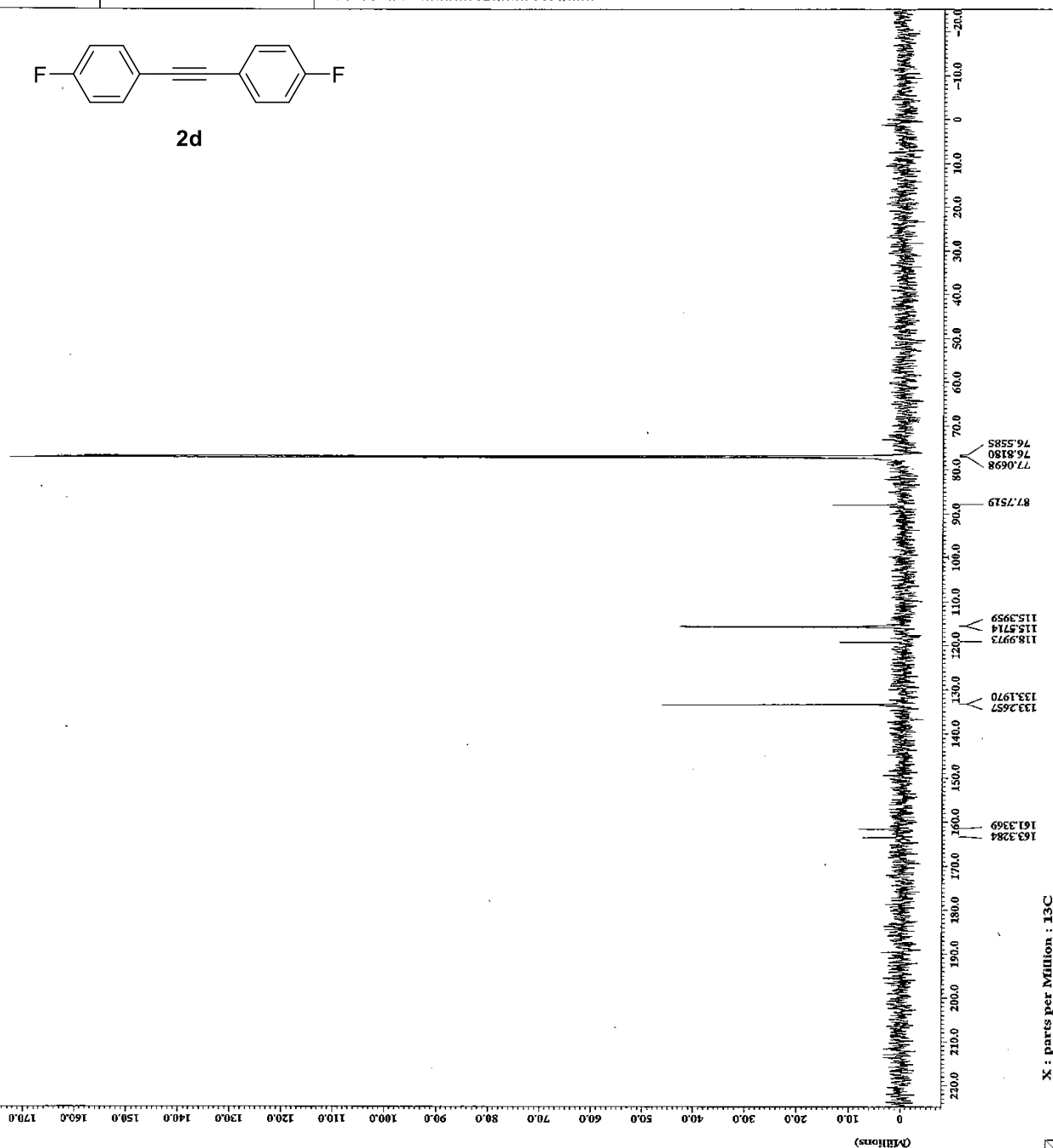


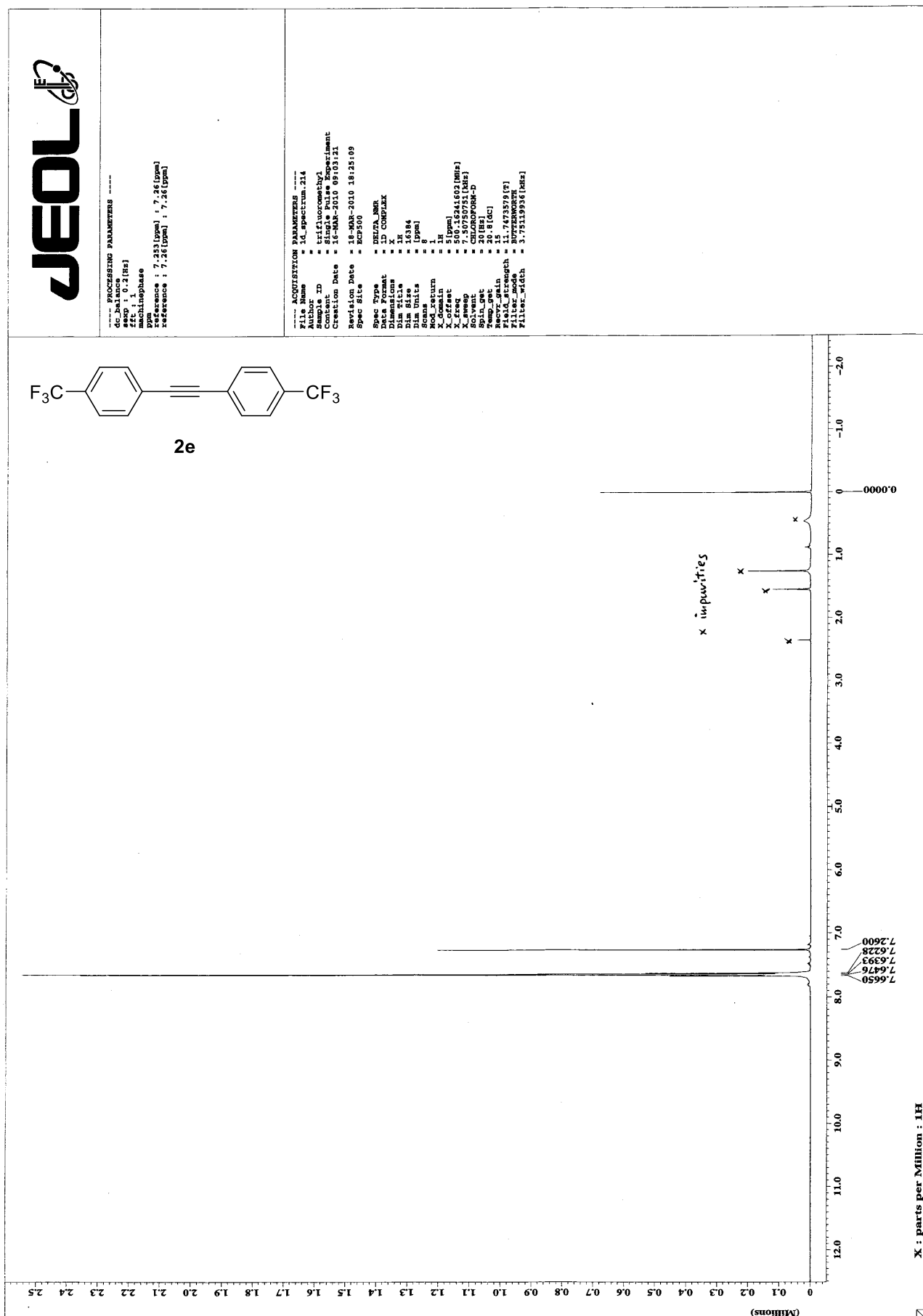
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machineshape
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2d



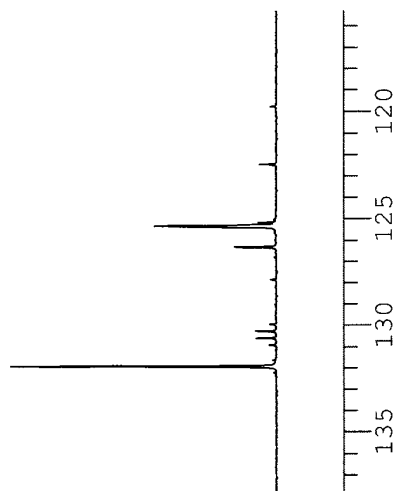
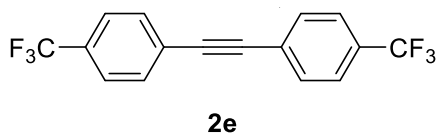


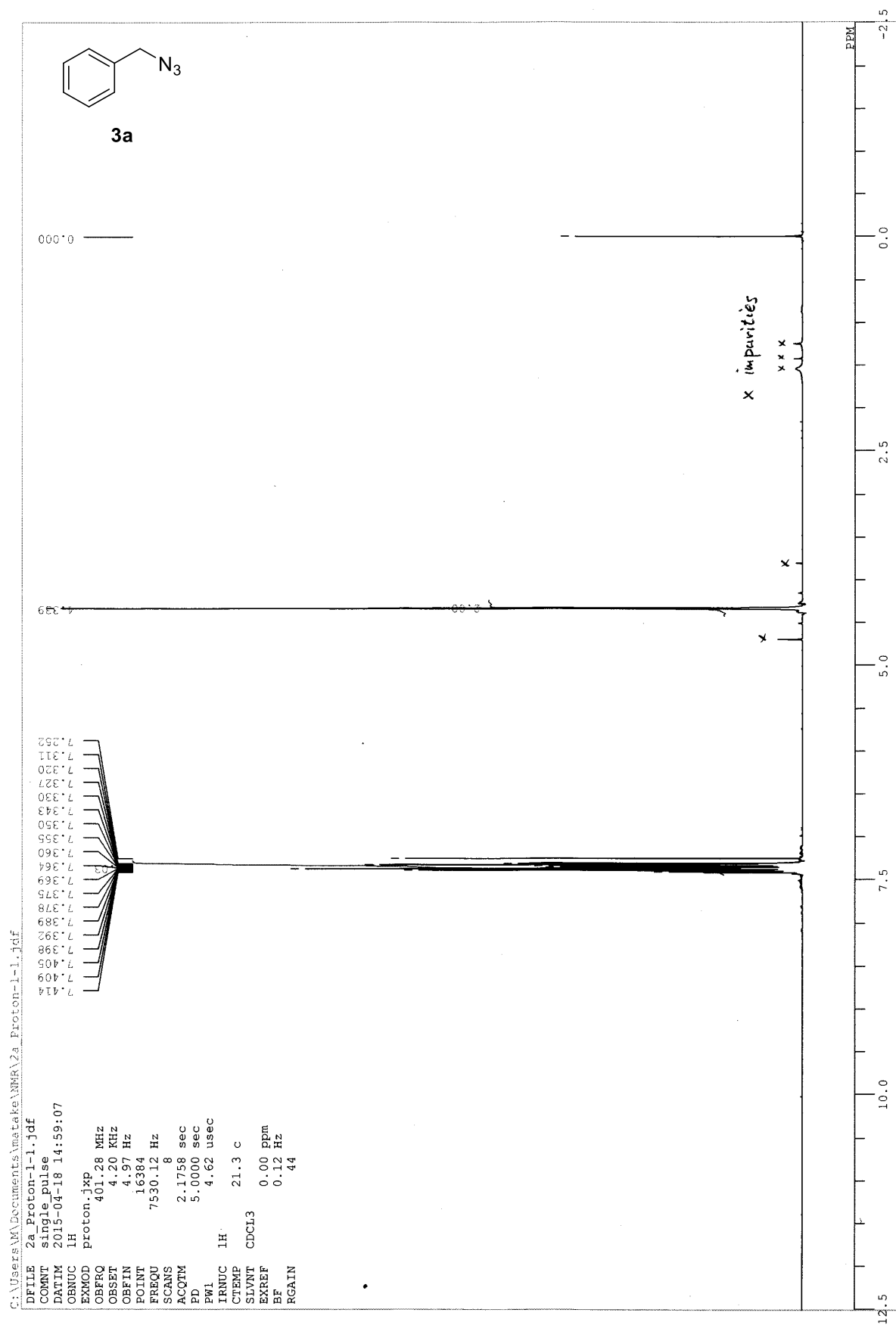
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 OBSET 8.88 Hz
 OBFIN 32767
 POINT 31565.66 Hz
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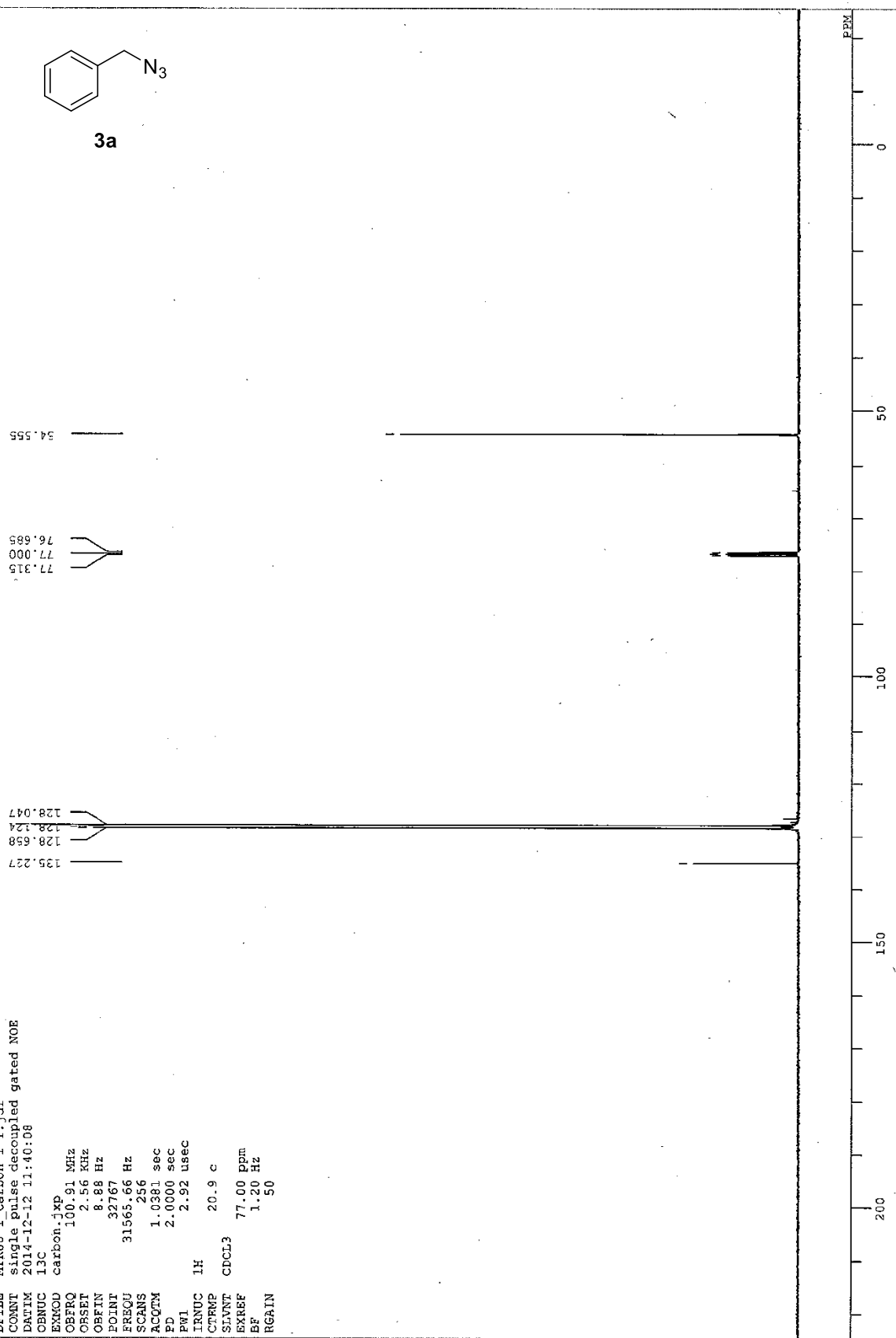
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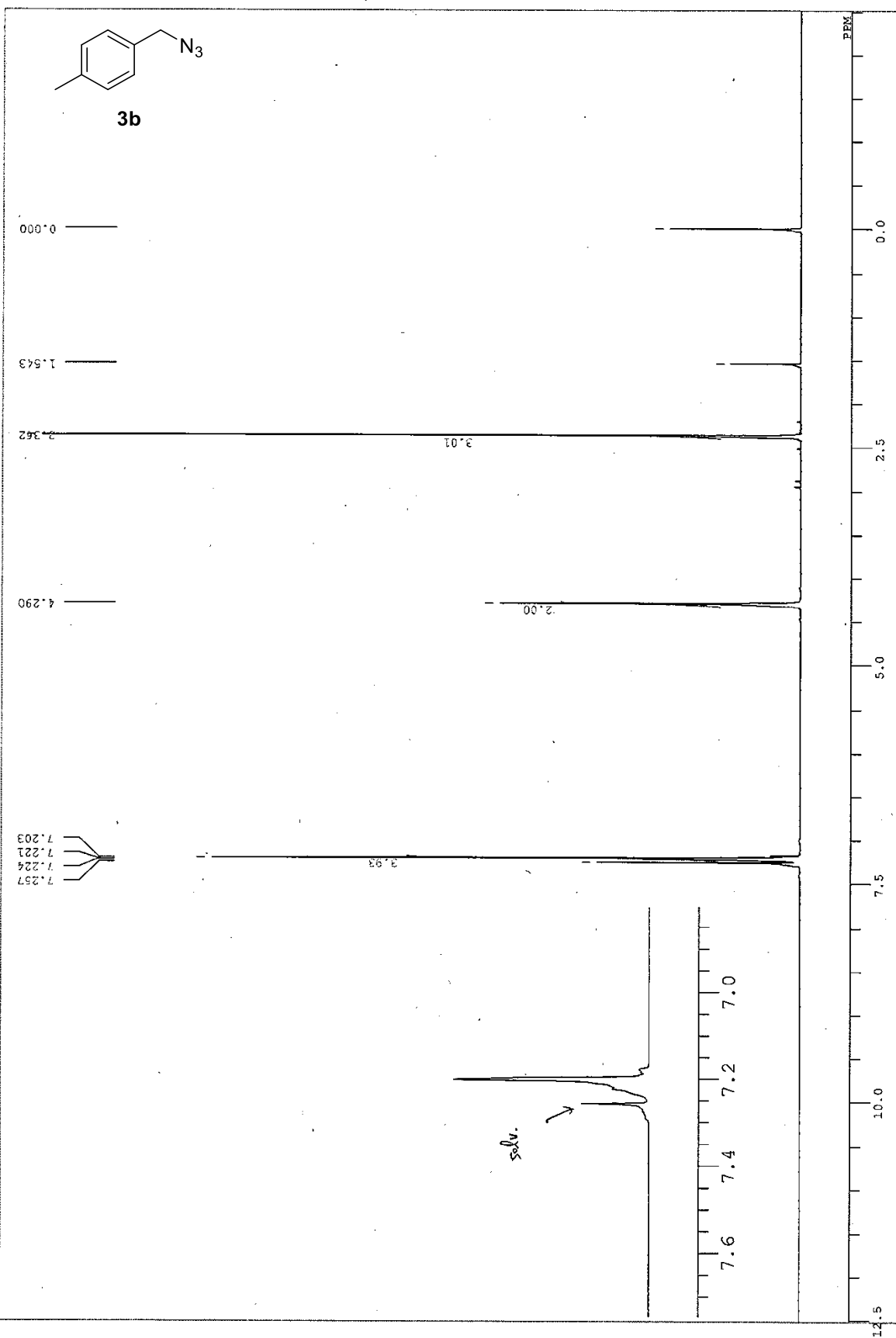


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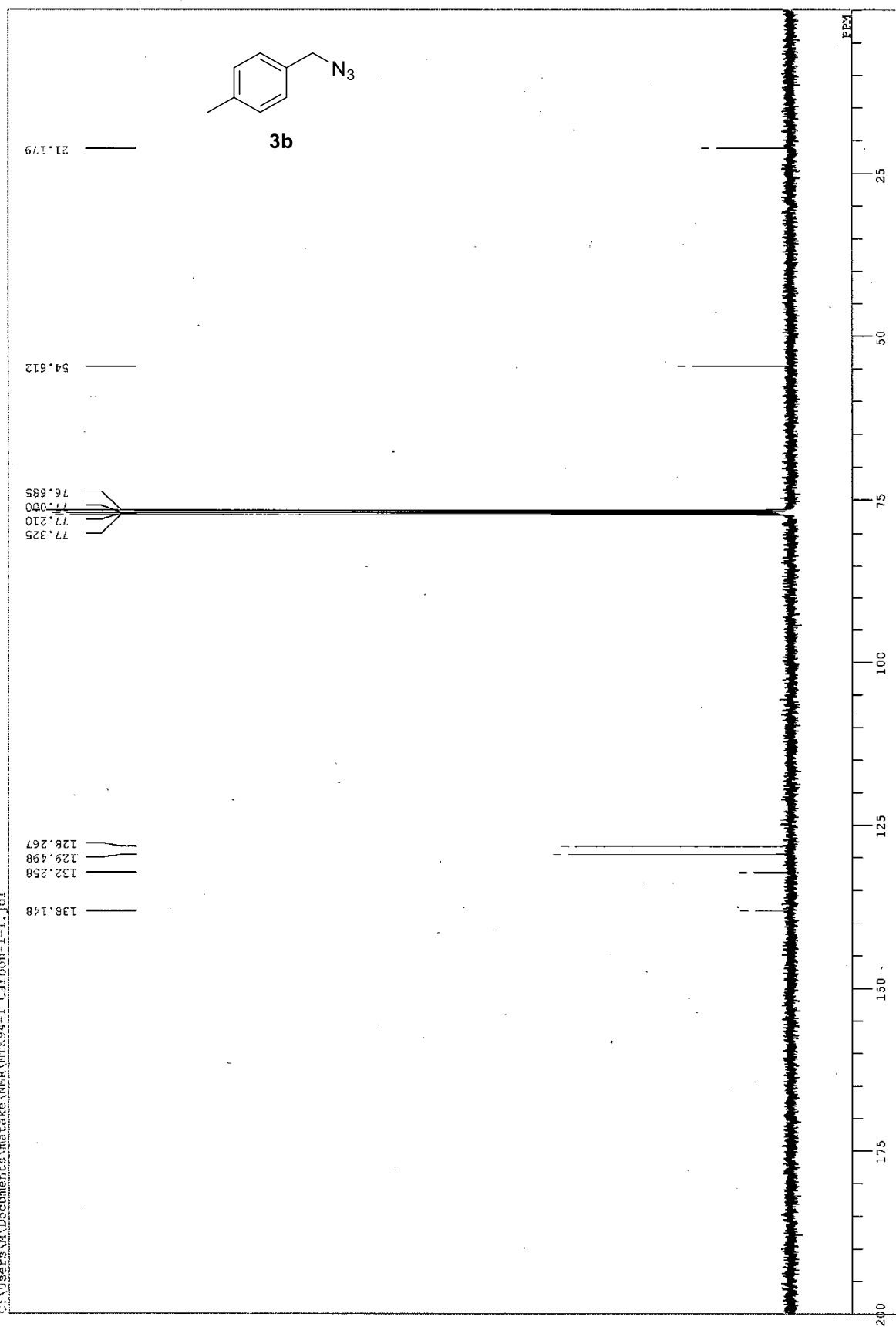
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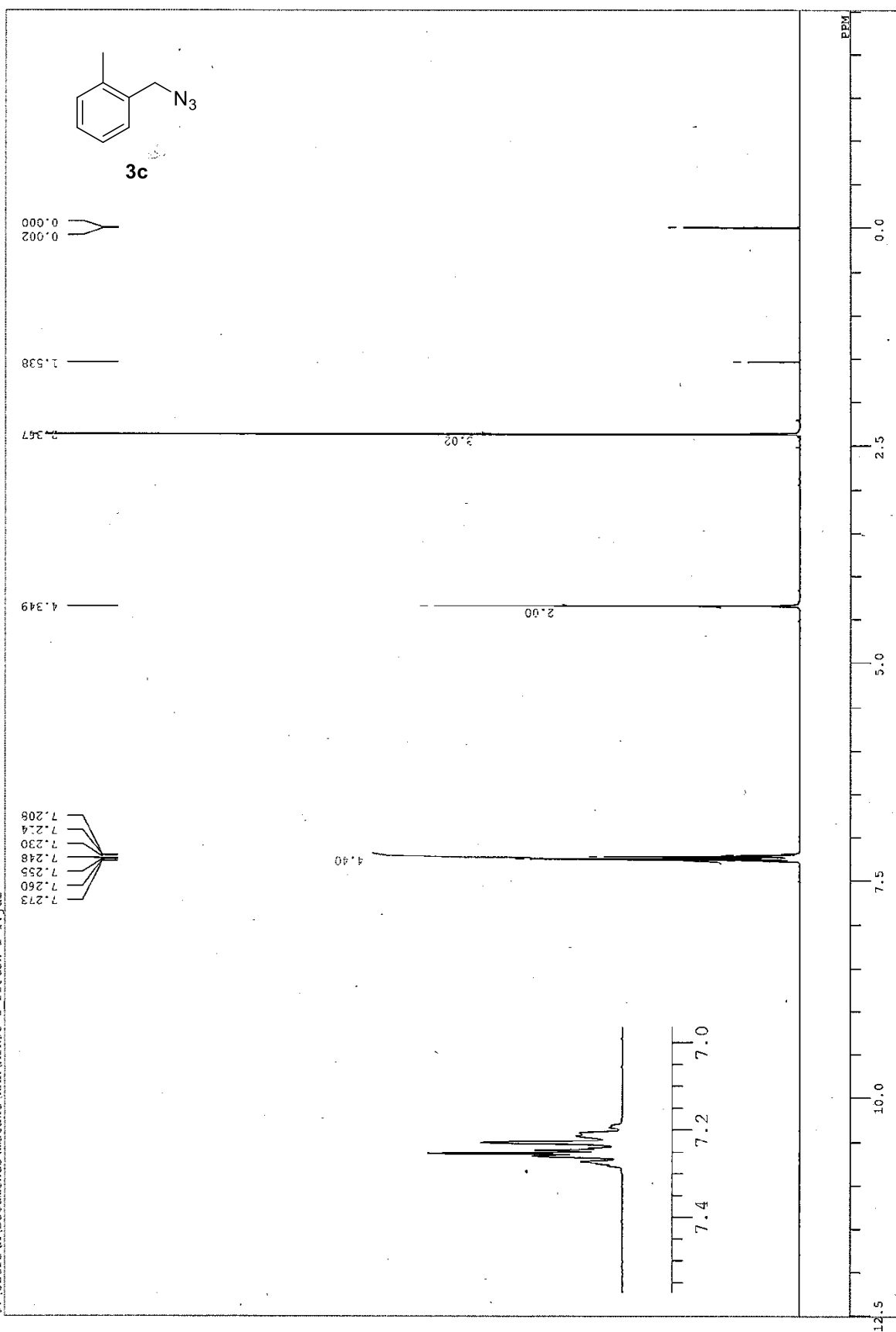
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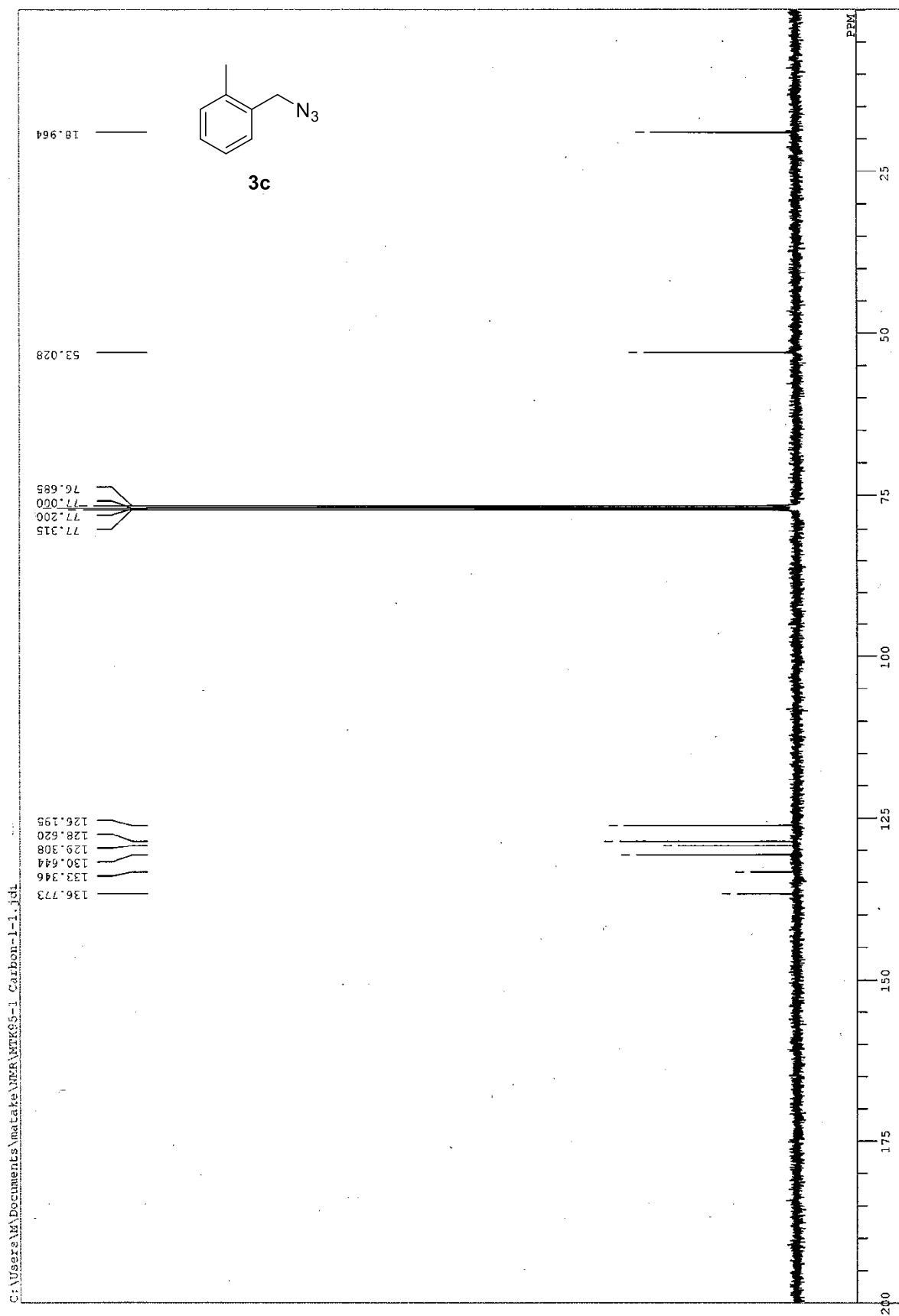


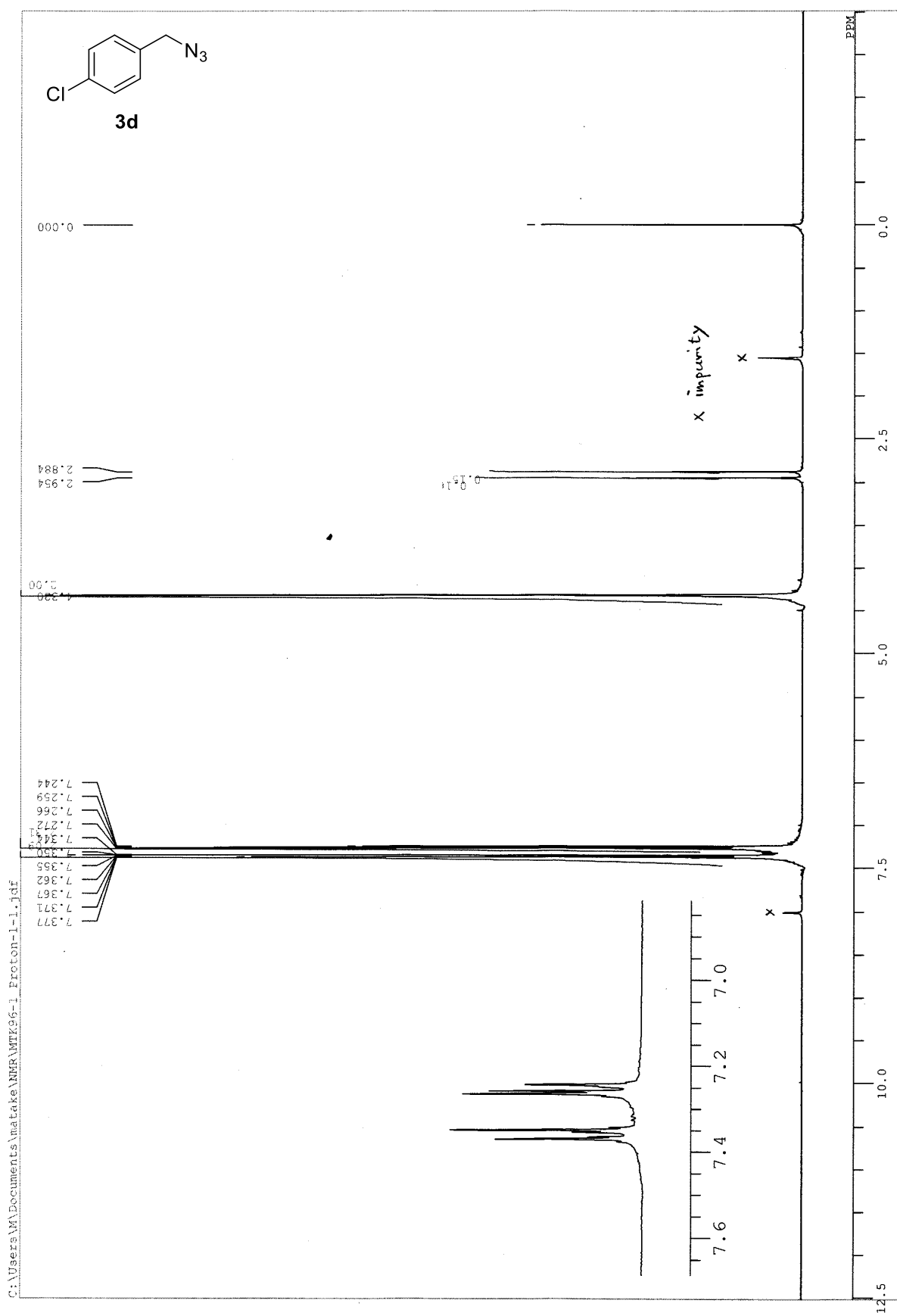
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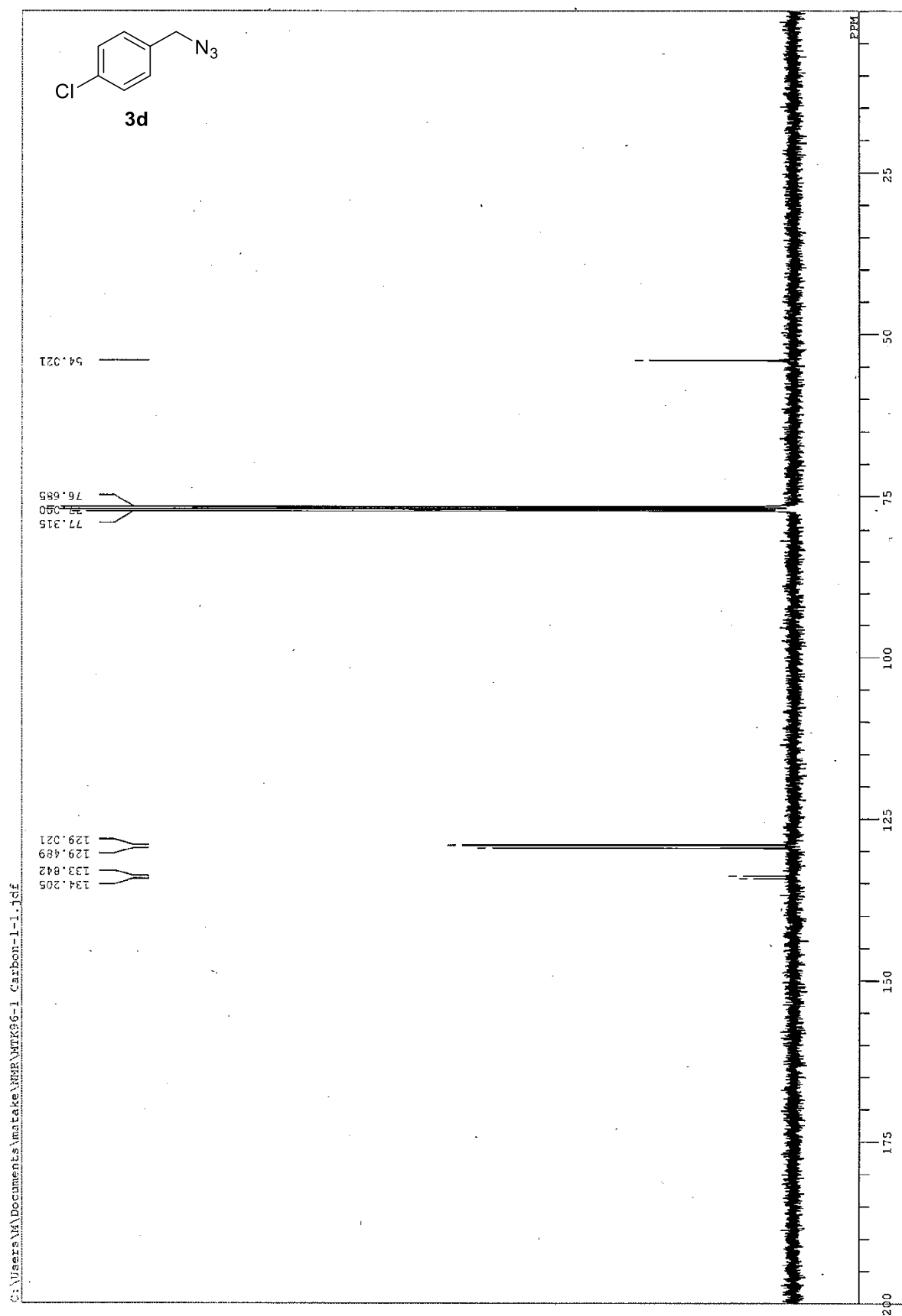


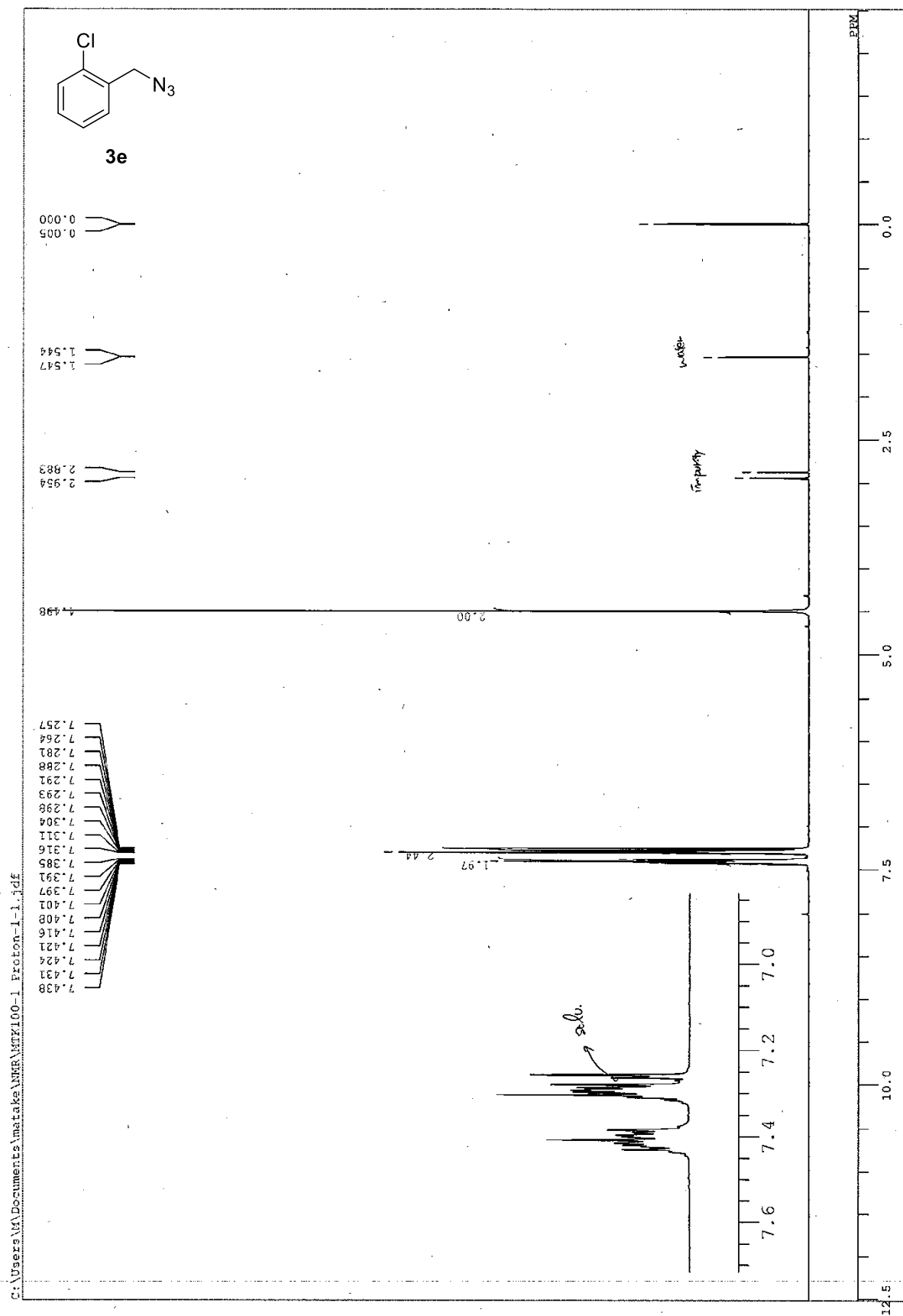
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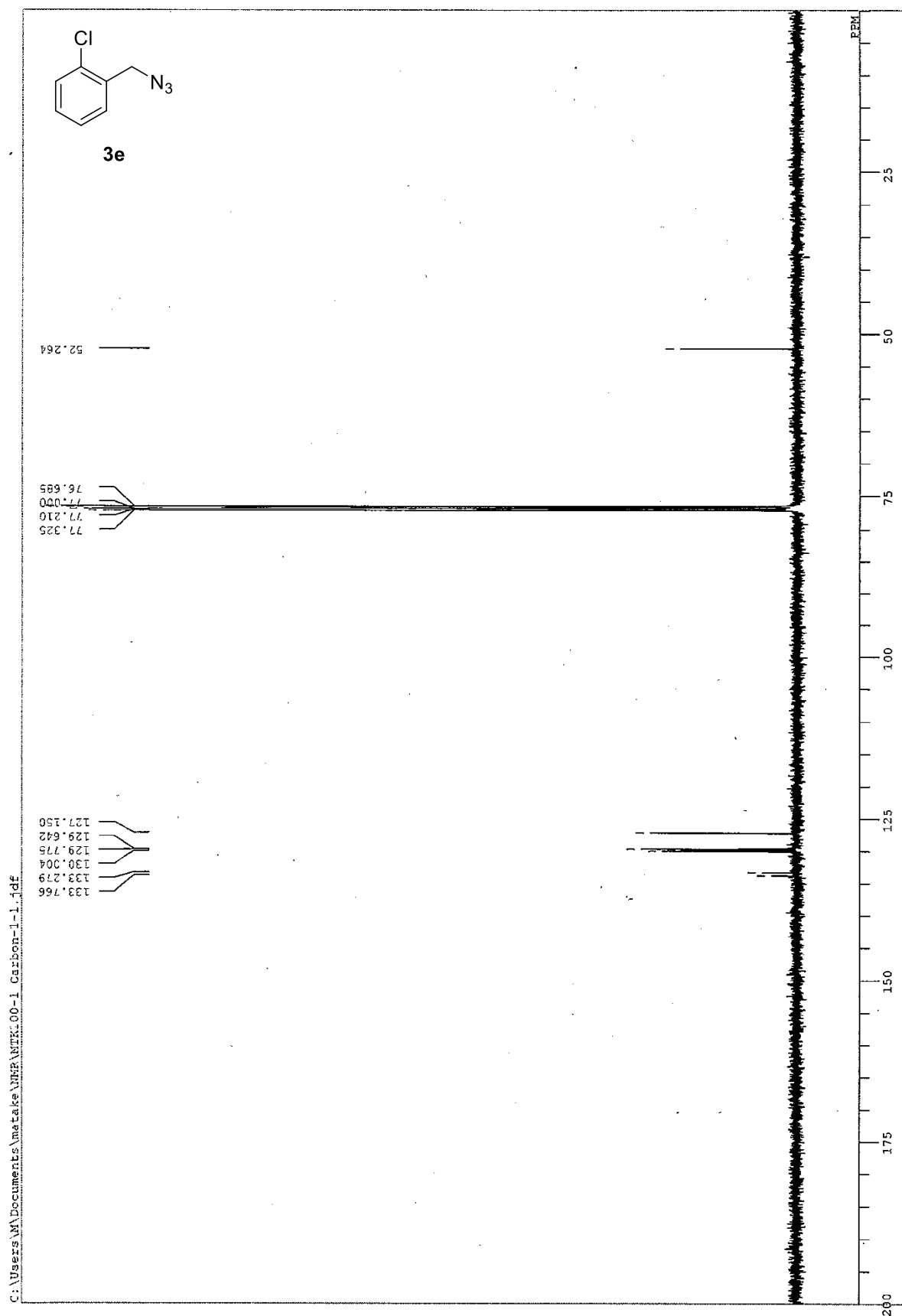




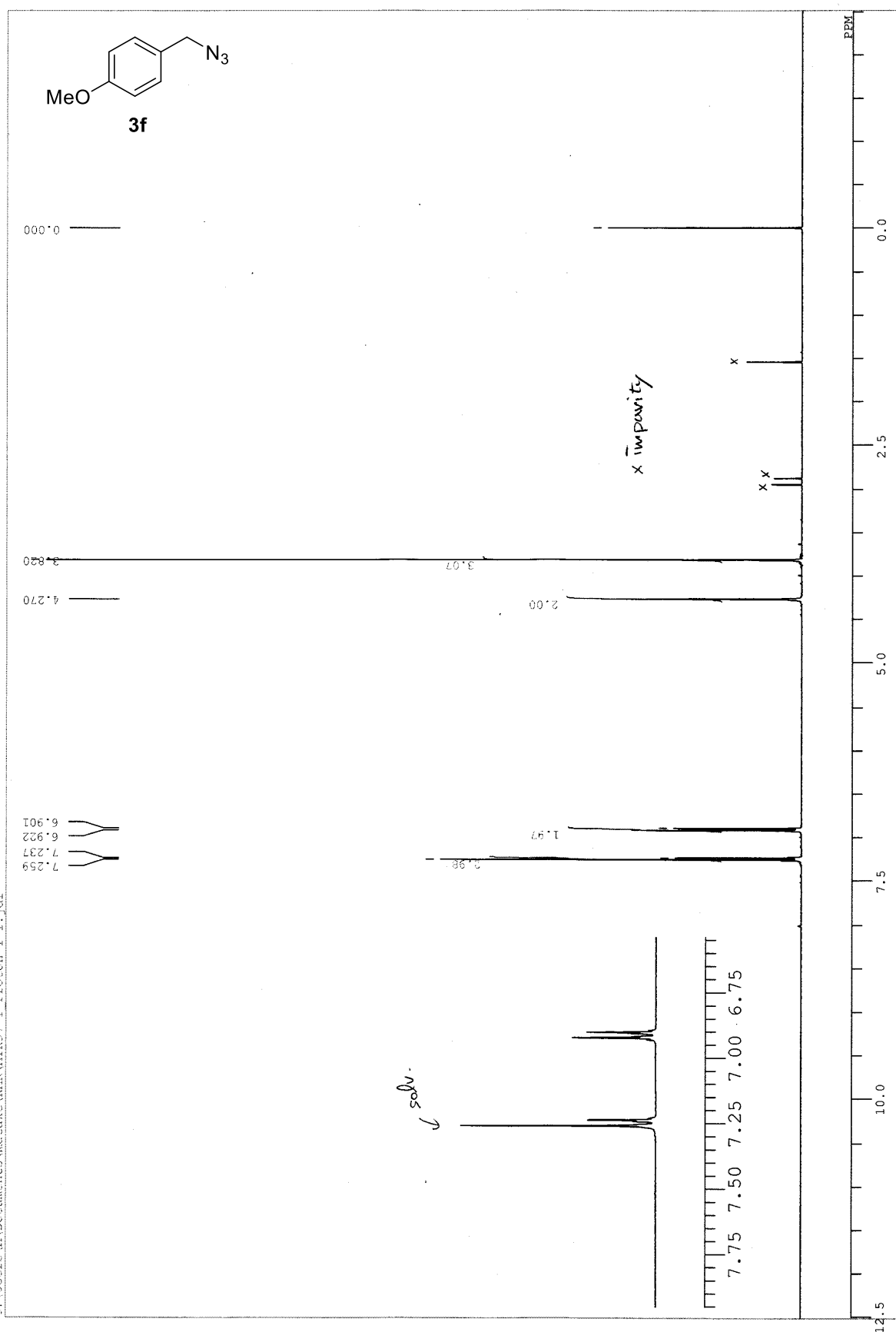




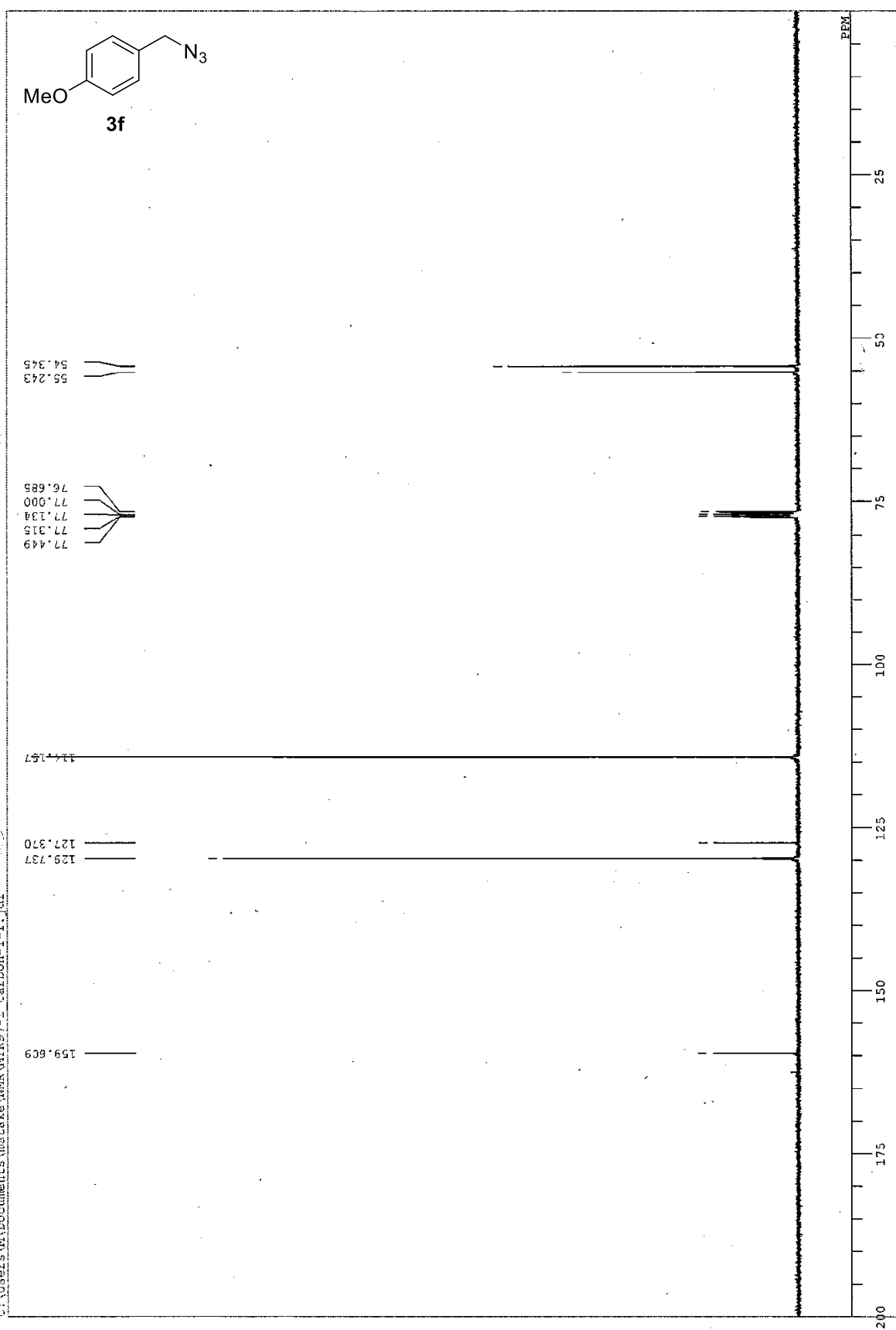


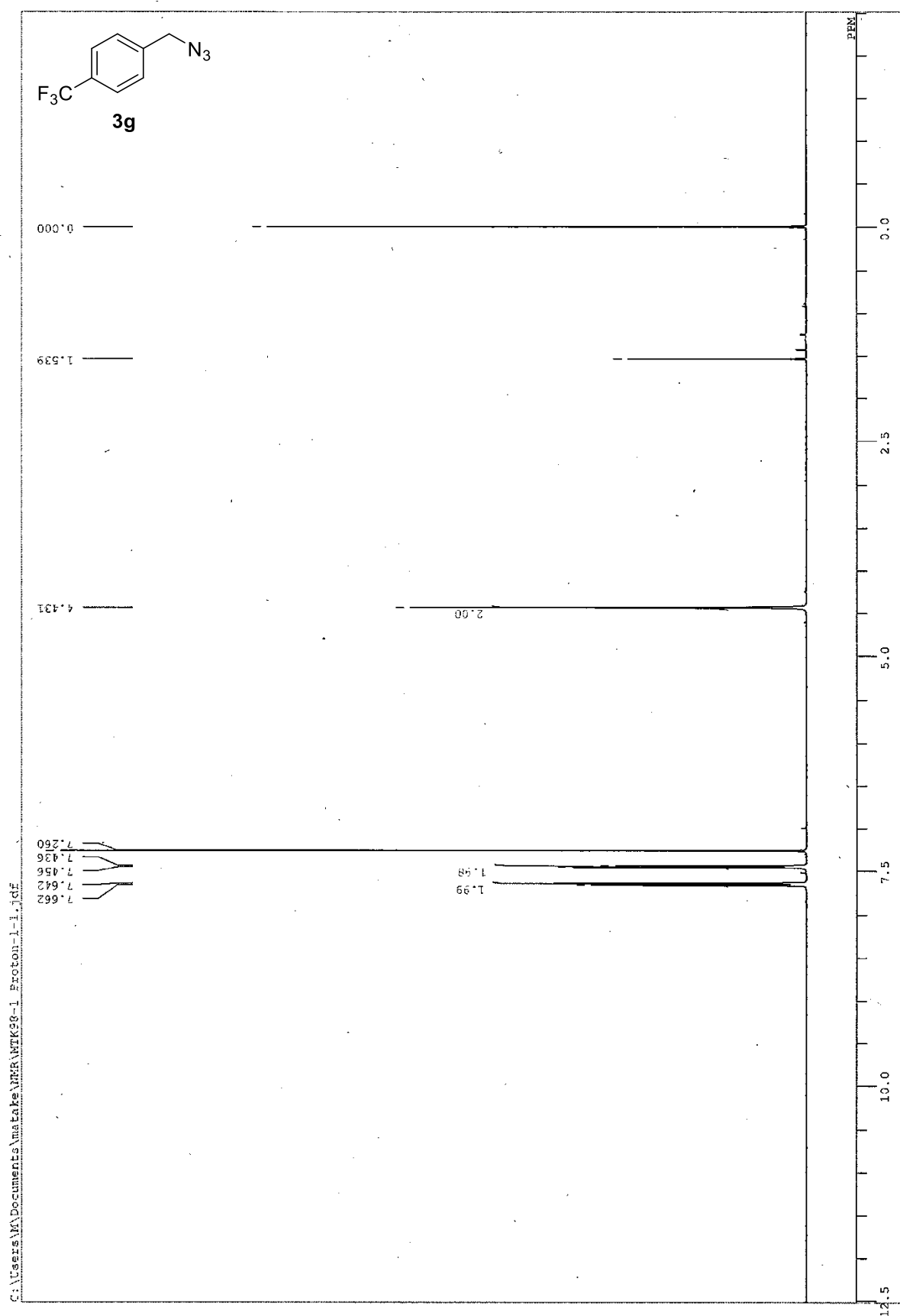


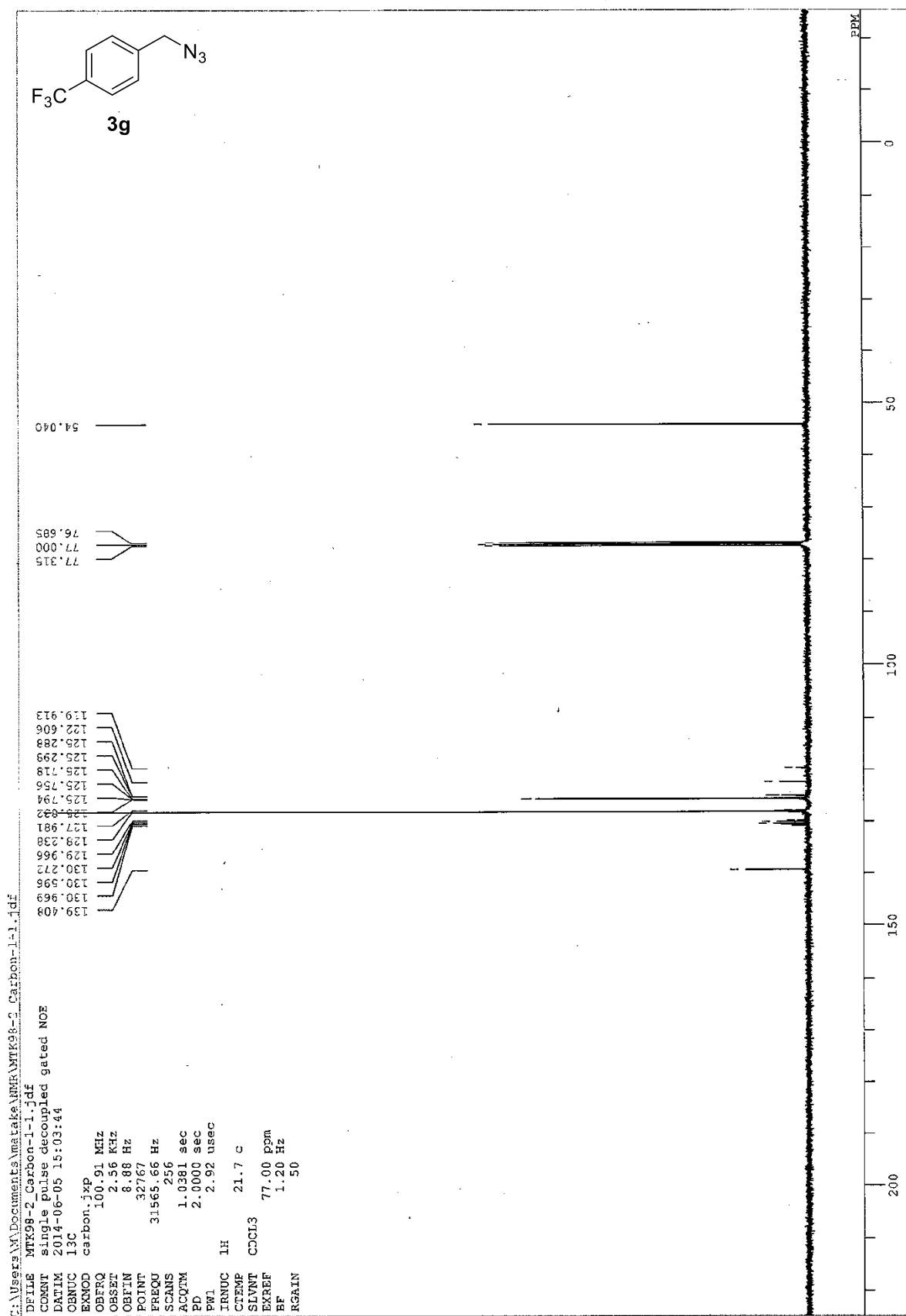
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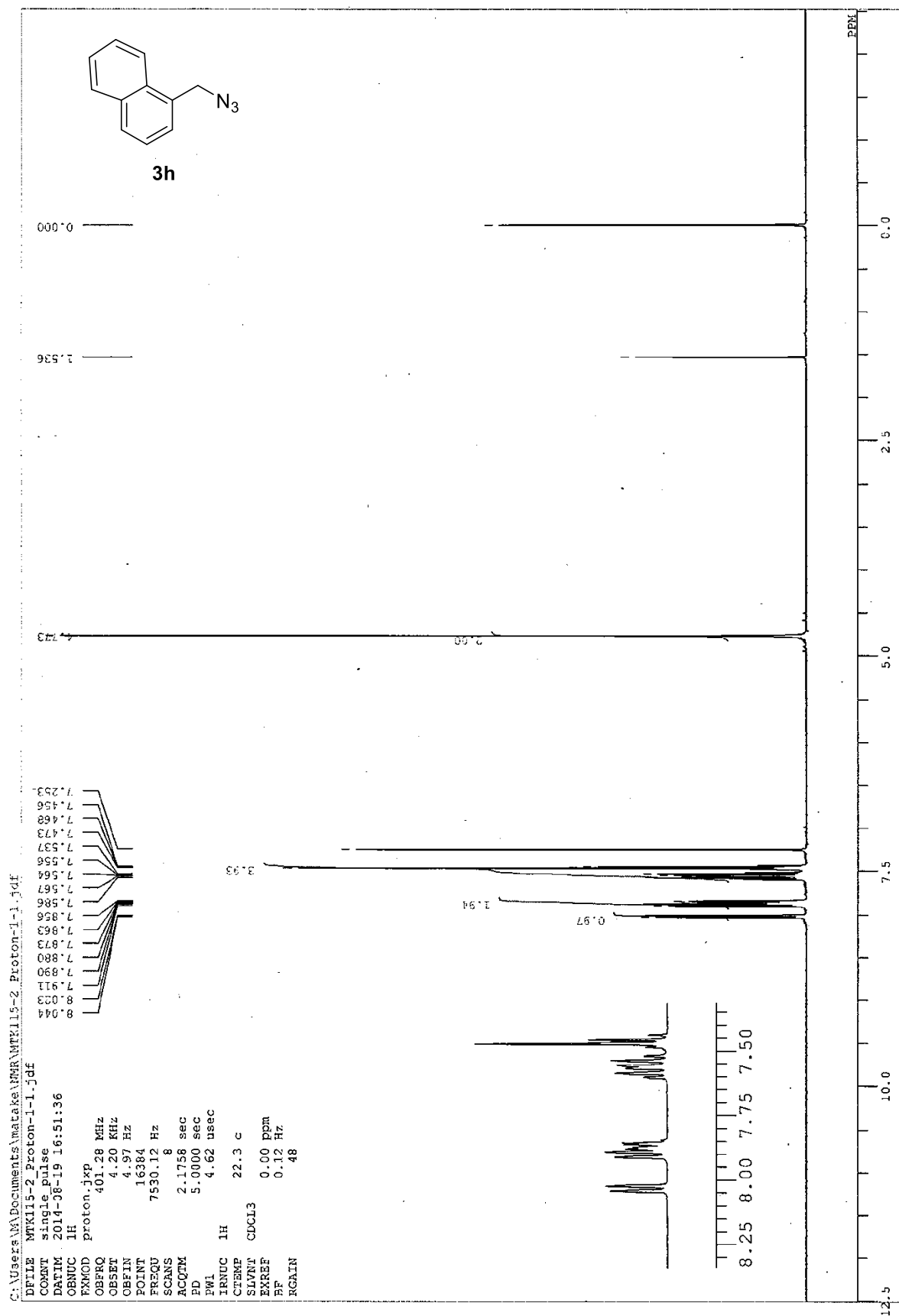


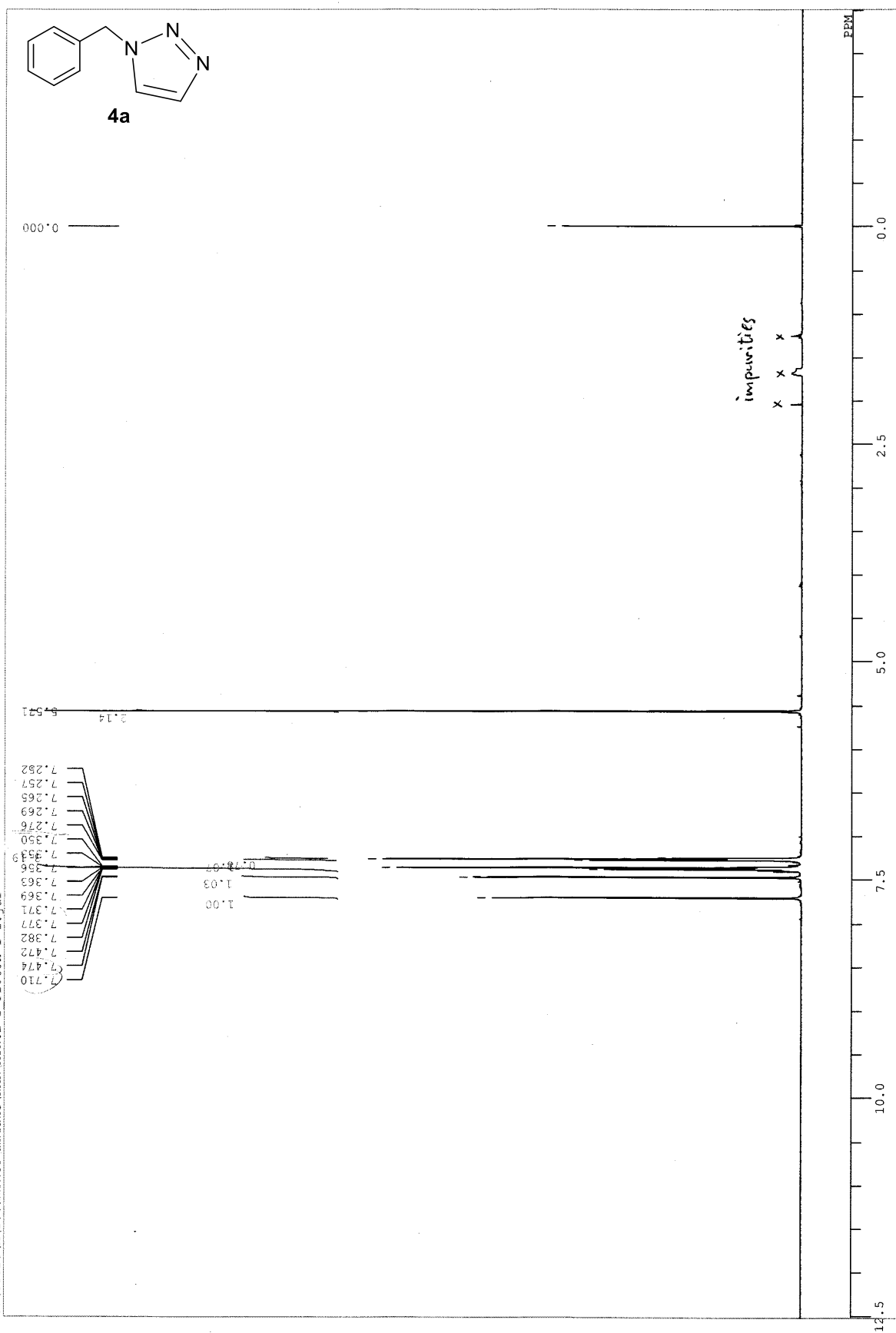
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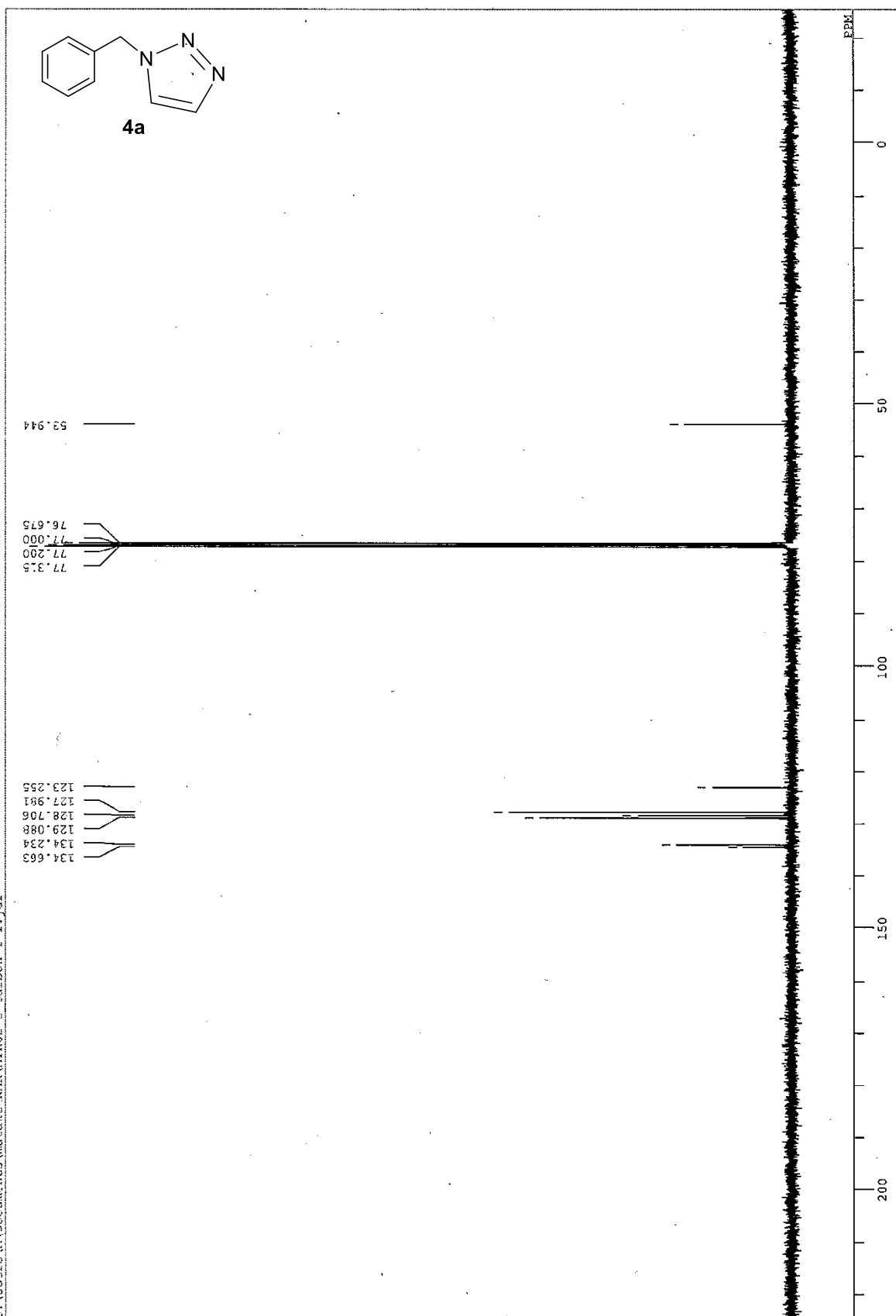


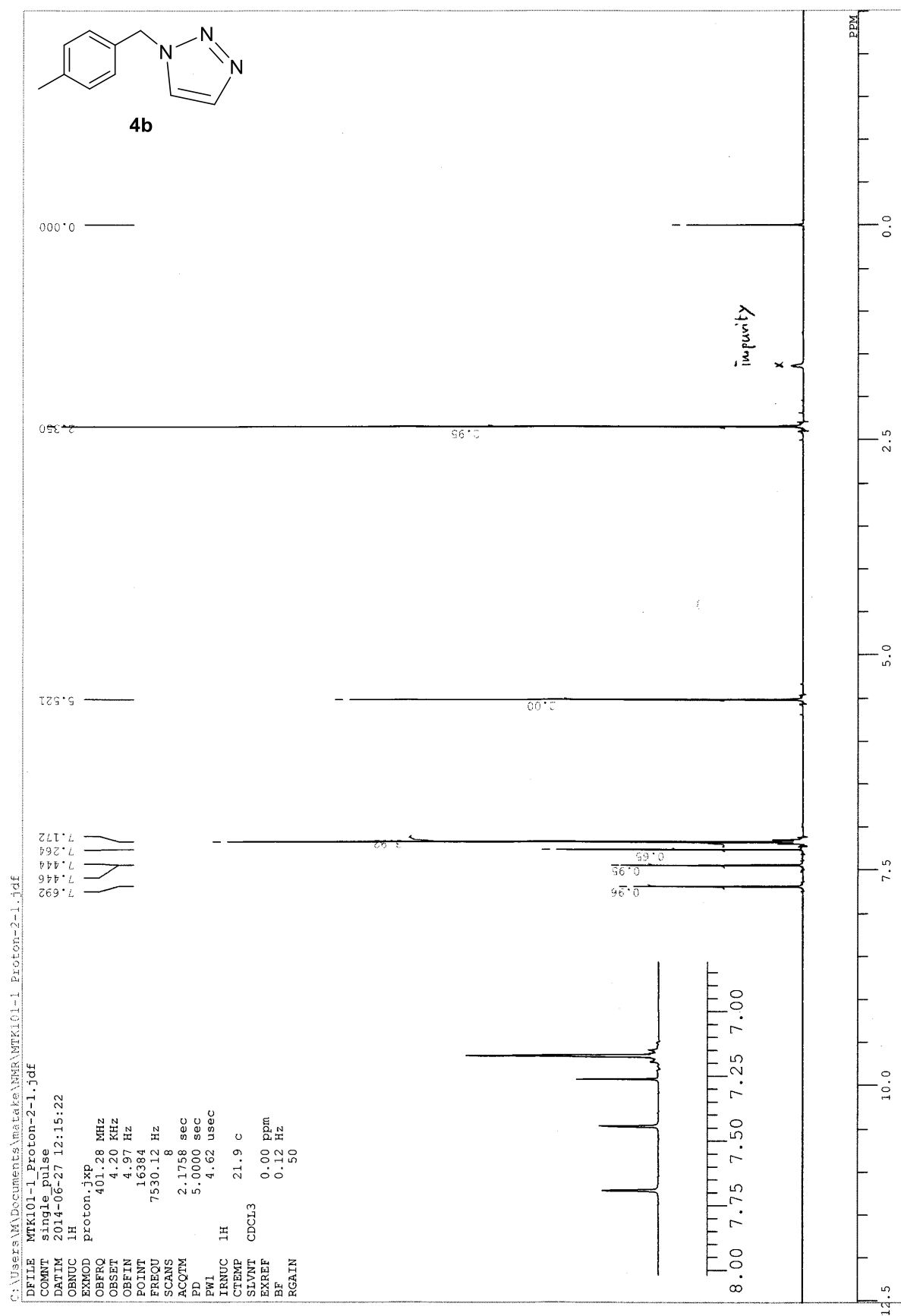






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IRNUC 1H

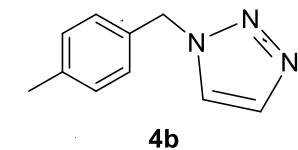
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SLVNT CDCL3

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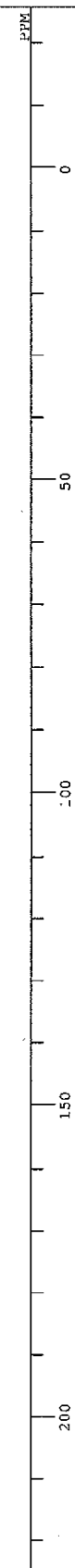


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53.285

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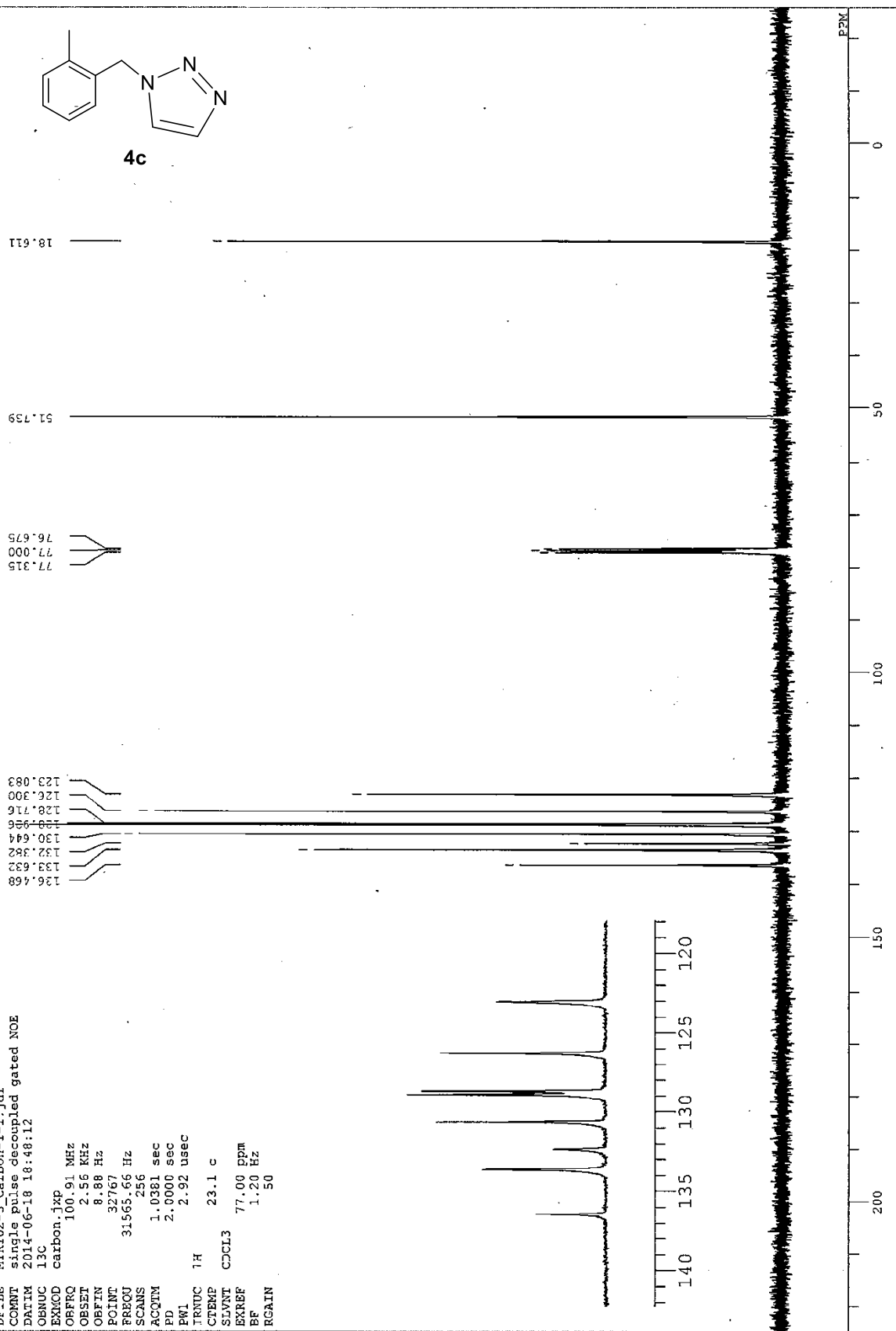
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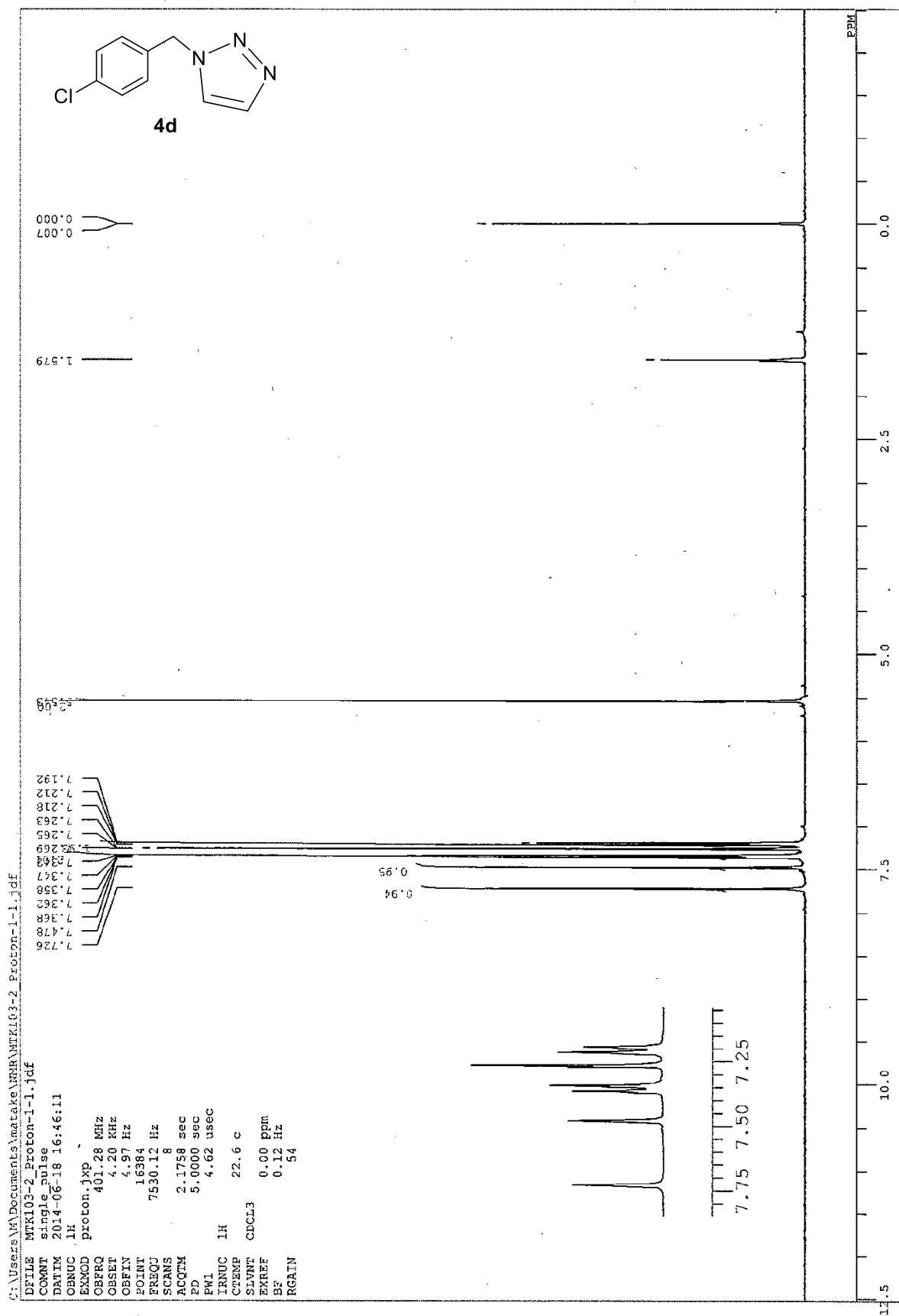


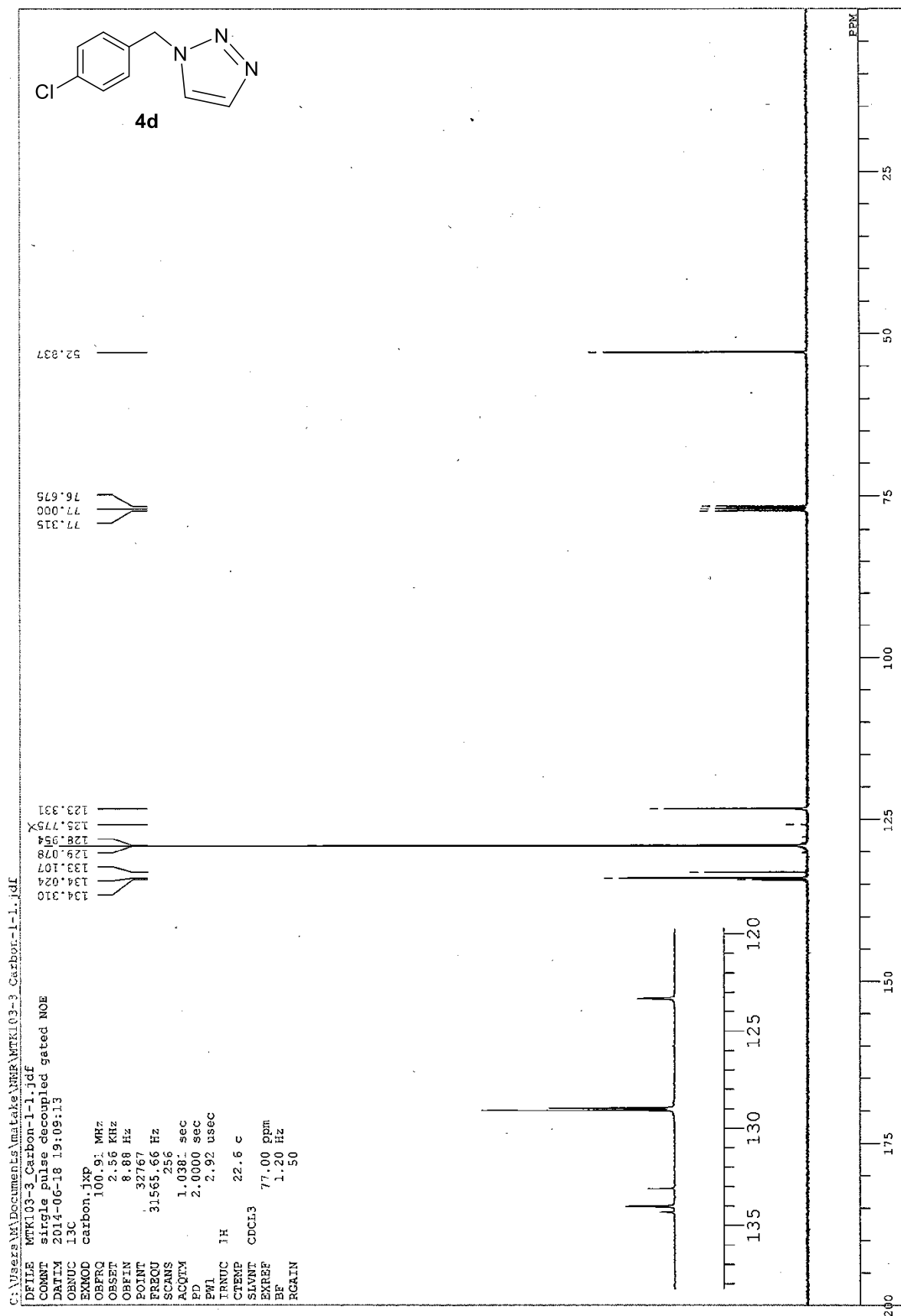
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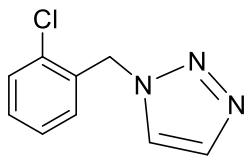
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BF 1.20 Hz
RGAIN 50

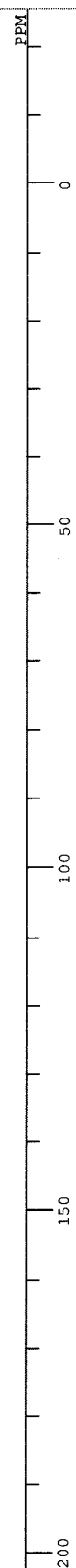
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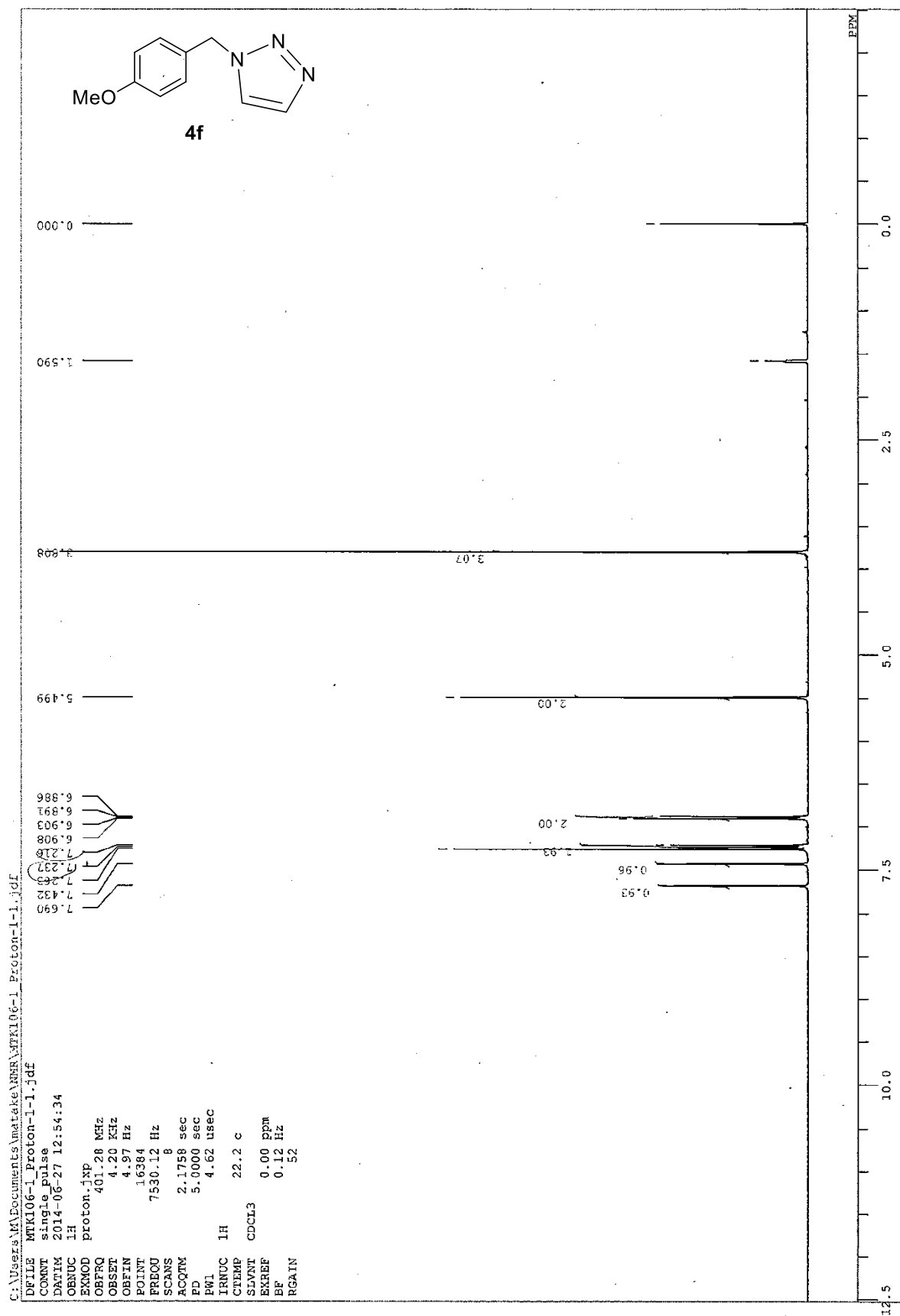
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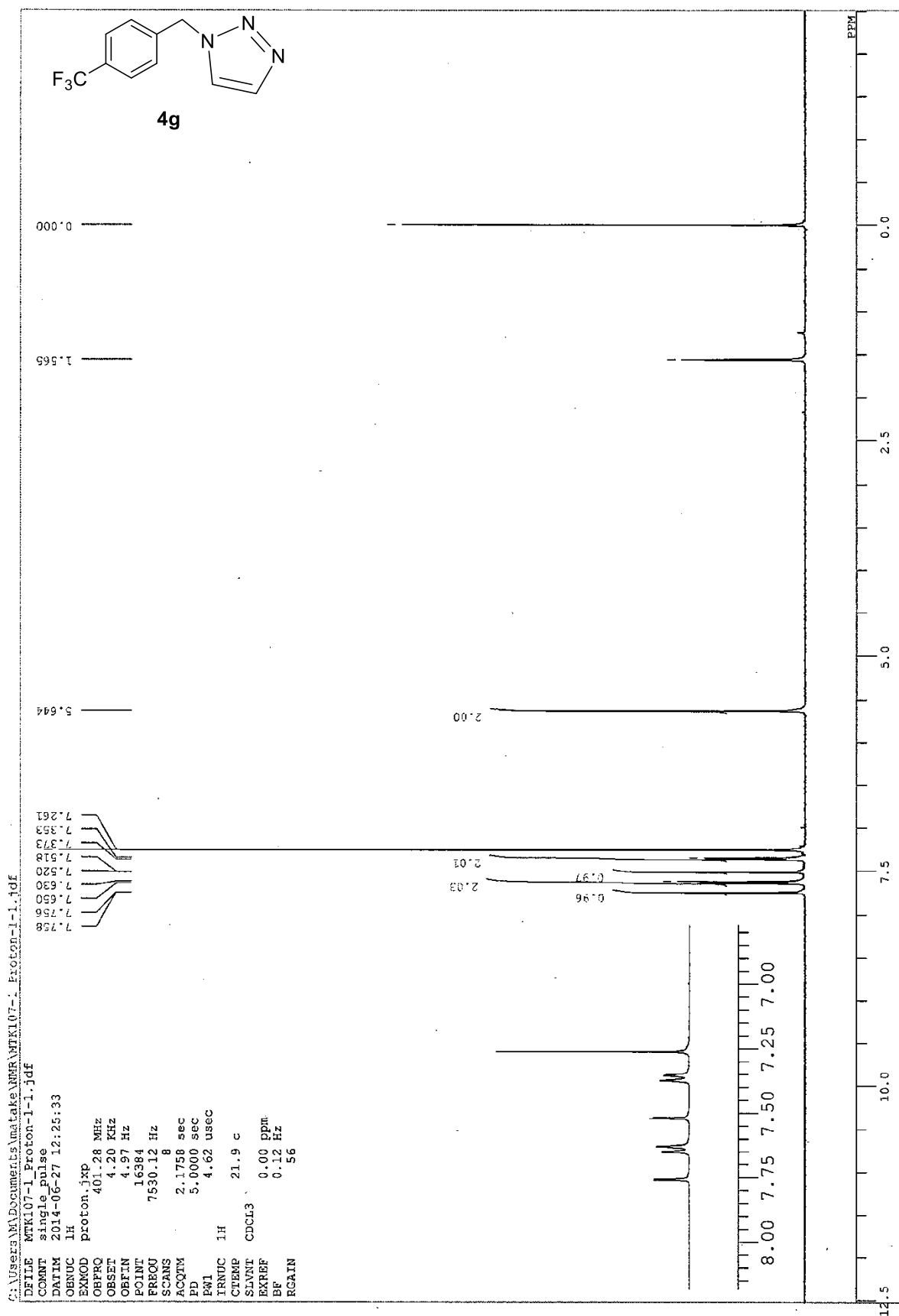
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133.823



4e

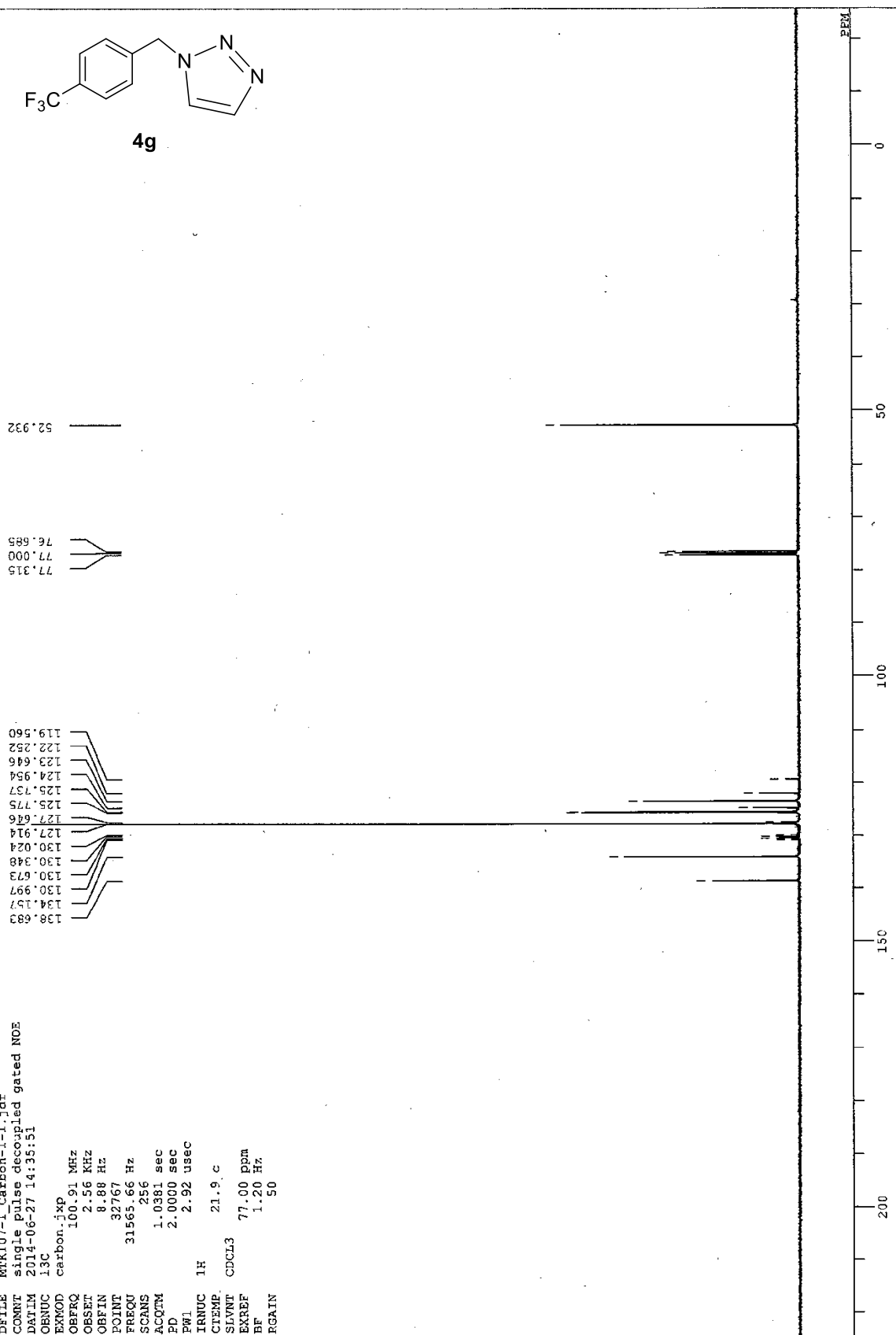






C:\Users\matate\Documents\matate\NMR\MTK107-1_Carbon-1-1.jdf

DFILE MTK107-1_Carbon-1-1.jdf
 COMNT single pulse decoupled gated NOE
 DATIM 2014-06-27 14:35:51
 OBSNUC 13C
 EXMOD carbon.jxp
 OBPRQ 100.91 MHz
 OBSET 2.56 KHz
 OBFIN 8.88 Hz
 POINT 32767
 FREQ 31565.66 Hz
 SCANS 256
 ACQTM 1.0381 sec
 PD 2.0000 sec
 PW1 2.92 usec
 IRNUC 1H
 CTEMP 21.9 c
 SLVNT CDCL3
 EXREF 77.00 ppm
 BF 1.20 Hz
 FGAIN 50



C:\Users\A\Documents\NMR\MTK118-2_Carbon-1-2_1.fdf

DFILE MTK118-2_Carbon-1-2_1.fdf

COMNT single pulse decoupled gated NOE

DATIM 2014-08-25 15:49:10

CBNUC 13C

EXMOD carbon-1xp

CBFREQ 100.61 MHz

OBSET 2.56 KHz

OBFIN 8.88 Hz

POINT 40960

FREQ 39457.07 Hz

SCANS 256

ACQIM 1.0381 sec

PD 2.0000 sec

FW 2.92 usec

IRNUC 1H

CTEMP 22.7 c

SLVNT CDCL3

EXREF 77.00 ppm

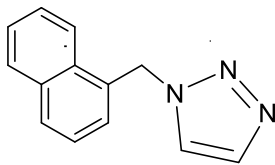
RF 0.12 Hz

RGAIN 50

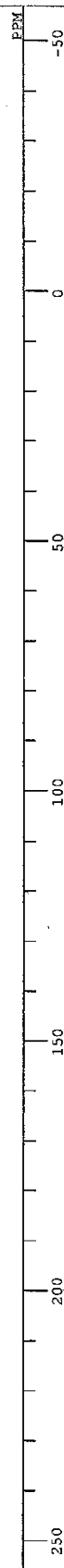
122.558
123.198
125.107
126.100
127.930
127.943
128.649
129.661
129.661
130.816
133.594
133.756

76.673
77.000
77.313

51.720

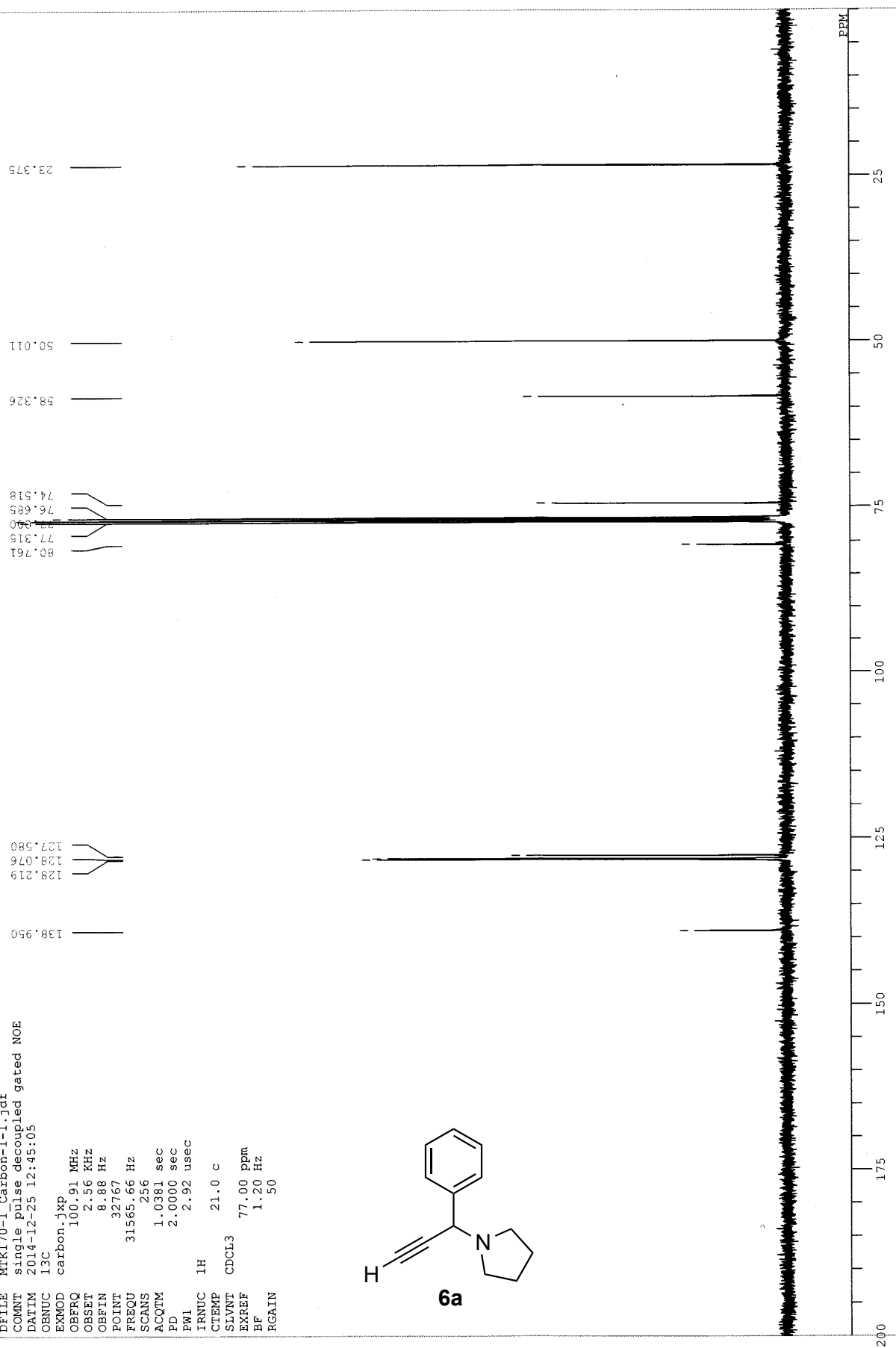
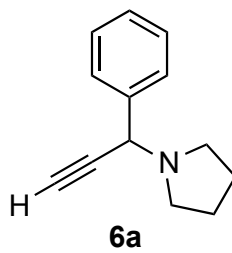


4h



C:\Users\A\Documents\anatake\NMR\MTKI70-1 Carbon-1-1.jdf

FILE MTKI70-1 Carbon-1-1.jdf
 COMNT single pulse decoupled gated NOE
 DATIM 2014-12-25 12:45:05
 OBNUC 13C
 EXMOD carbon.jxp
 OBFRQ 100.91 MHz
 OBSET 2.56 KHz
 OBFIN 8.88 Hz
 POINT 32767
 FREQU 31565.66 Hz
 SCANS 256
 ACQTM 1.0381 sec
 PD 2.0000 sec
 PW1 2.92 usec
 IRNUC 1H
 CTEMP 21.0 c
 SLVNT CDCL3
 EXREF 77.00 ppm
 BF 1.20 Hz
 RGAIN 50



C:\Users\N\Documents\data\NMR\MTK196-1H Proton-1-1.jdf

DELF MTK196-1H Proton-1-1.jdf

COMNT single_pulse

DATIM 2015-02-12 00:02:05

ORNUC 1H

EXMOD proton.jxp

OBFRQ 401.28 MHz

OBSET 4.20 KHz

OBFIN 4.97 Hz

POINT 16384

FREQ 7530.12 Hz

SCANS 8

ACQTM 2.1758 sec

PD 5.0000 sec

PW1 4.62 usec

IRNUC 1H

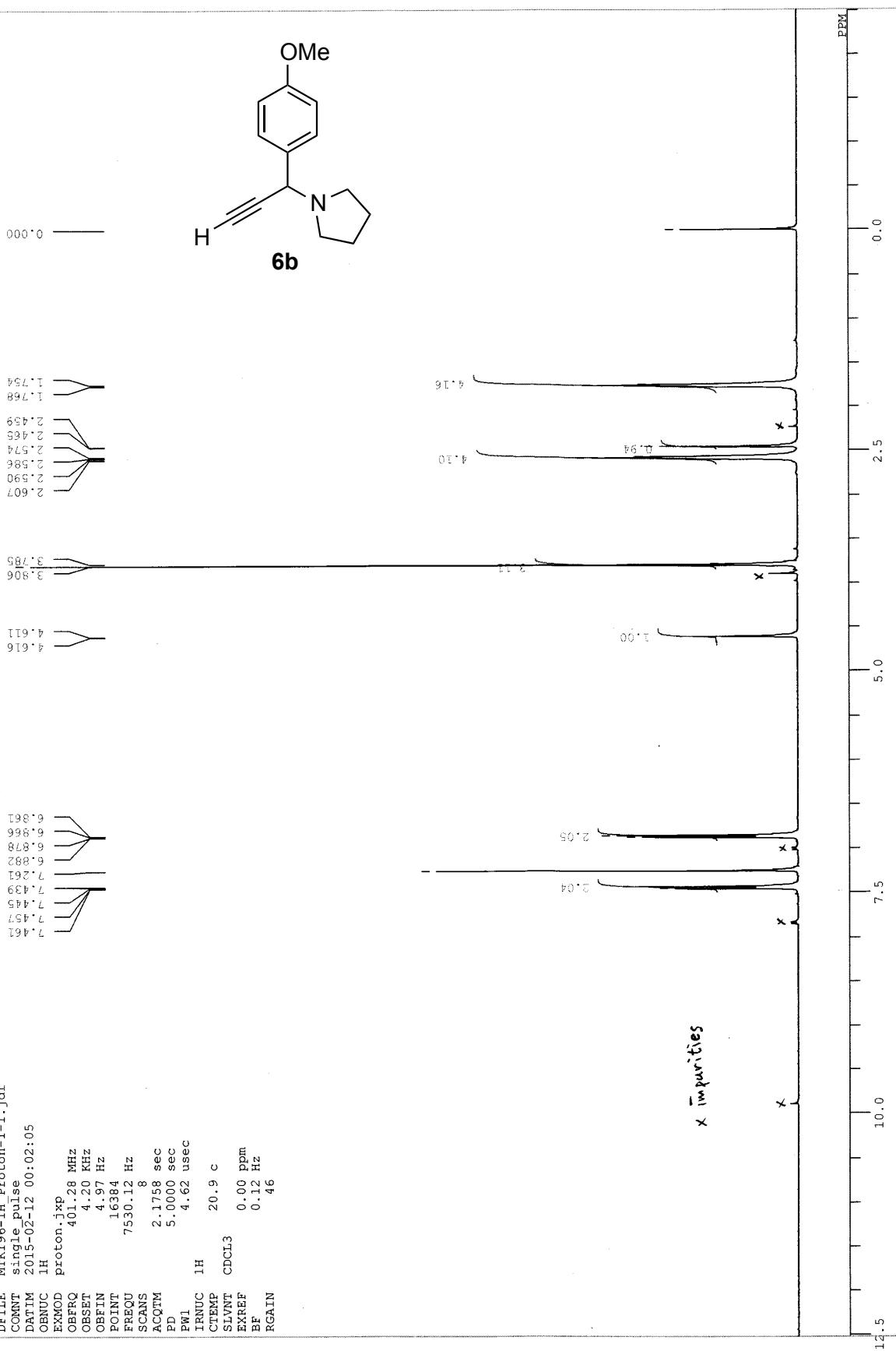
CTEMP 20.9 C

SLVNT CDCL3

EXREF 0.00 ppm

BF 0.12 Hz

RGAIN 46



C:\Users\A\Documents\matak\NMR\MTK196-1C_Carbon-1-1.fdf

DFILE MTK196-1C_Carbon-1-1.fdf

COMNT single pulse decoupled gated NOE

DATEM 2015-02-12 00:11:43

OBNUC 13C

EXMOD carbon-1xp

OBFRQ 100.91 MHz

OBSET 2.56 kHz

OBFTN 8.88 Hz

POINT 32767

FREQU 31565.66 Hz

SCANS 256

ACQTM 1.0381 sec

PD 2.0000 sec

PW1 2.92 usec

IRNUC 1H

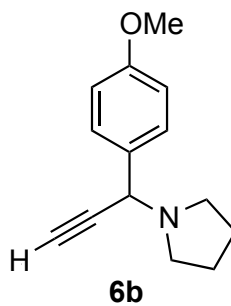
CTEMP 20.9 c

SLVNT CDCL3

EXREF 77.00 ppm

BF 1.20 Hz

RGAIN 50



81.029
77.315
77.000
76.685
74.231

57.639
55.138
49.906

131.141
129.107
113.450

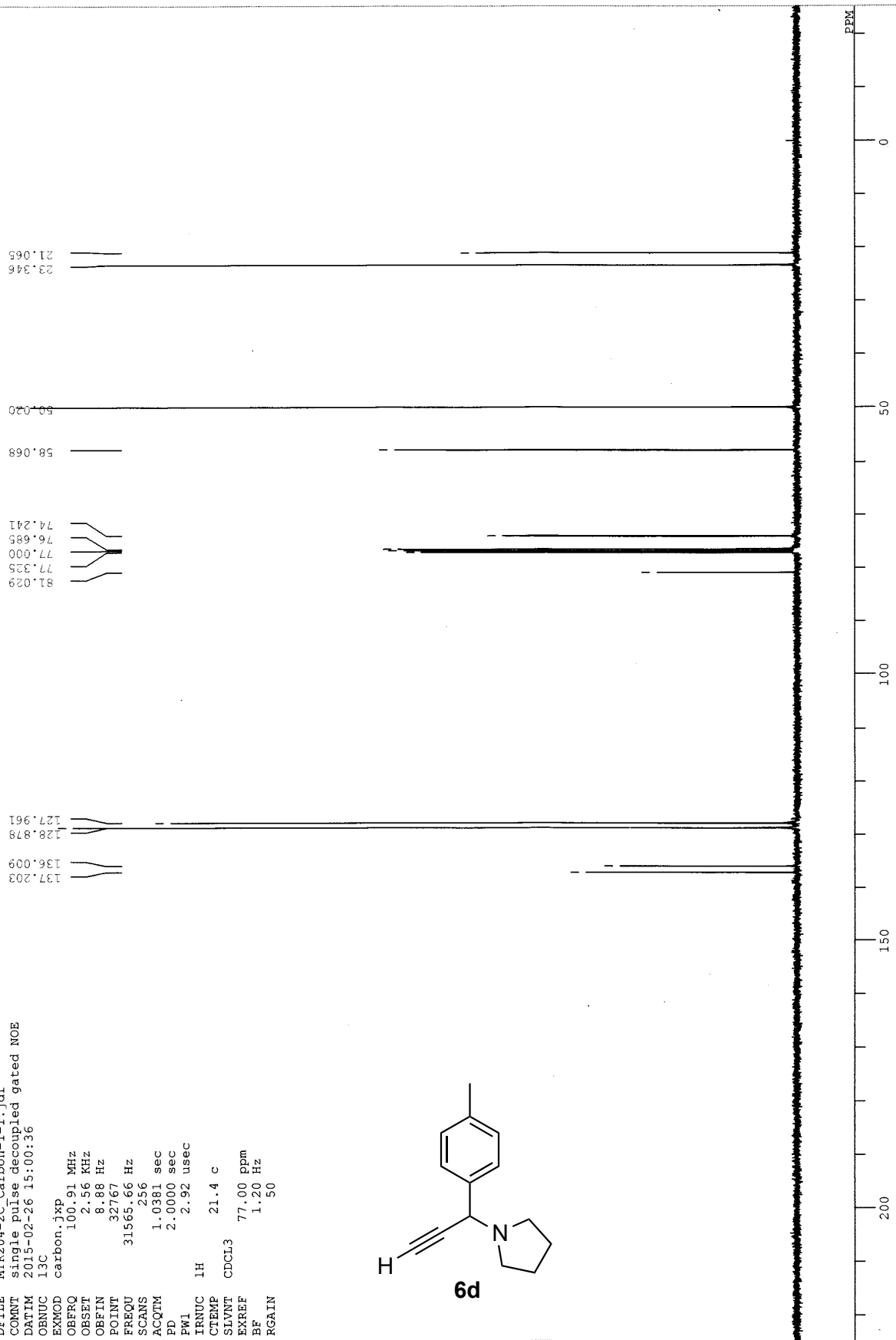
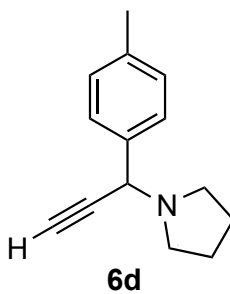
159.932

23.299

C:\Users\MI\Documents\data\NMR\MTK204-2C Carbon-1-1.jdf

DfFILE MTK204-2C Carbon-1-1.jdf
 COMNT single pulse decoupled gated NOE
 DATIM 2015-02-26 15:00:36

OBNUC 13C
 EXMOD carbon.jxp
 OBFRQ 100.91 MHz
 OBSET 2.56 KHz
 OBFIN 8.88 Hz
 POINT 32767
 FREQU 31565.66 Hz
 SCANS 256
 ACQTM 1.0381 sec
 PD 2.0000 sec
 PW1 2.92 usec
 IRNUC 1H
 CTMP 21.4 c
 SLVNT CDCL3
 EXREF 77.00 ppm
 BF 1.20 Hz
 RGAIN 50



C:\Users\MTK205-2H\Documents\NMR\MTK205-2H Proton-1-1.jdf

DPFILE MTK205-2H Proton-1-1.jdf

COMNT single_pulse

DATIM 2015-02-26 15:20:26

OBNUC 1H

EXMOD proton.jxp

OBFRQ 401.28 MHz

OBSET 4.20 KHz

POINT 16384

FREQ 7530.12 Hz

SCANS 8

ACQTM 2.1758 sec

PD 5.0000 sec

PW1 4.62 usec

IRNUC 1H

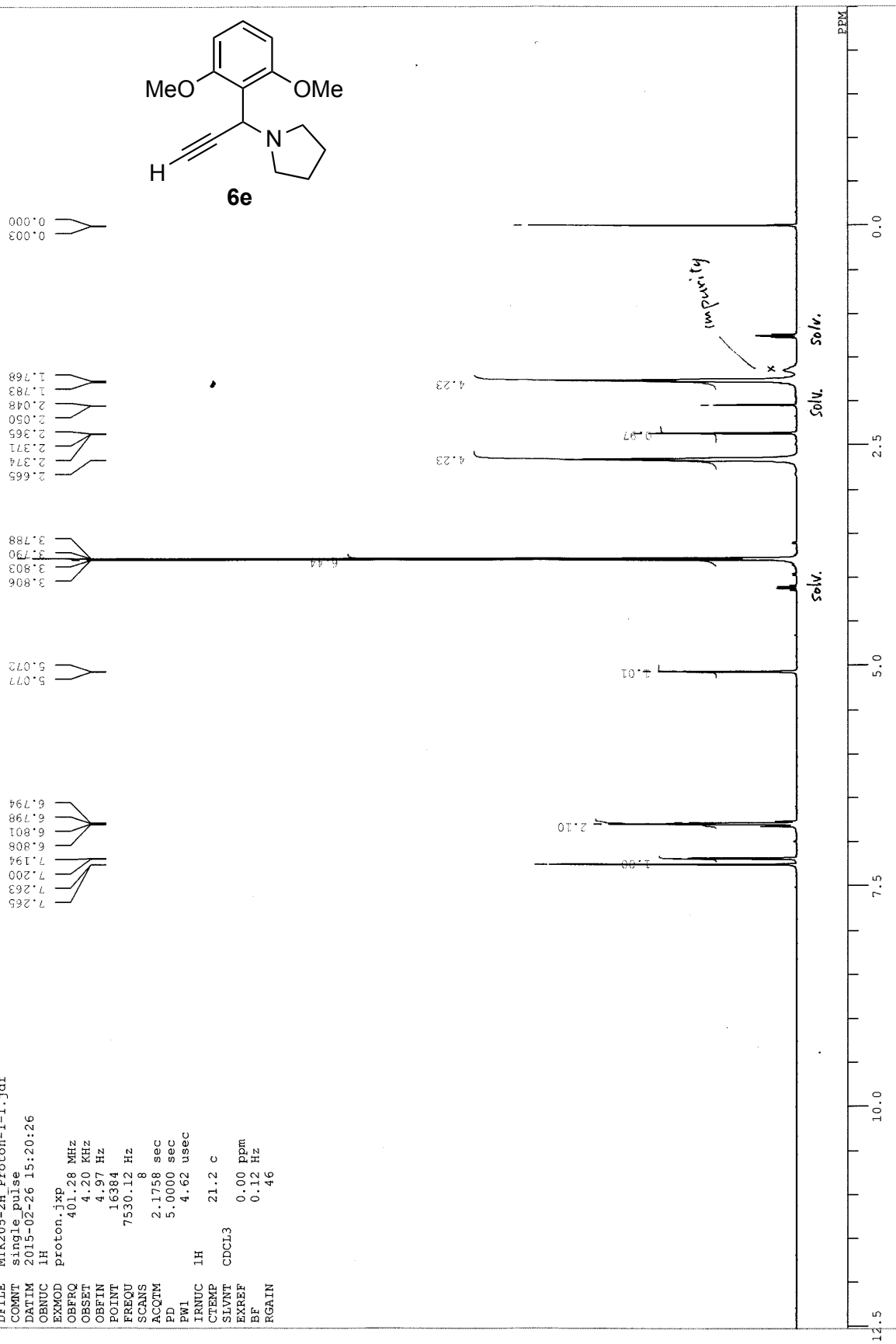
CTEMP 21.2 C

SLVNT CDCL3

EXREF 0.00 ppm

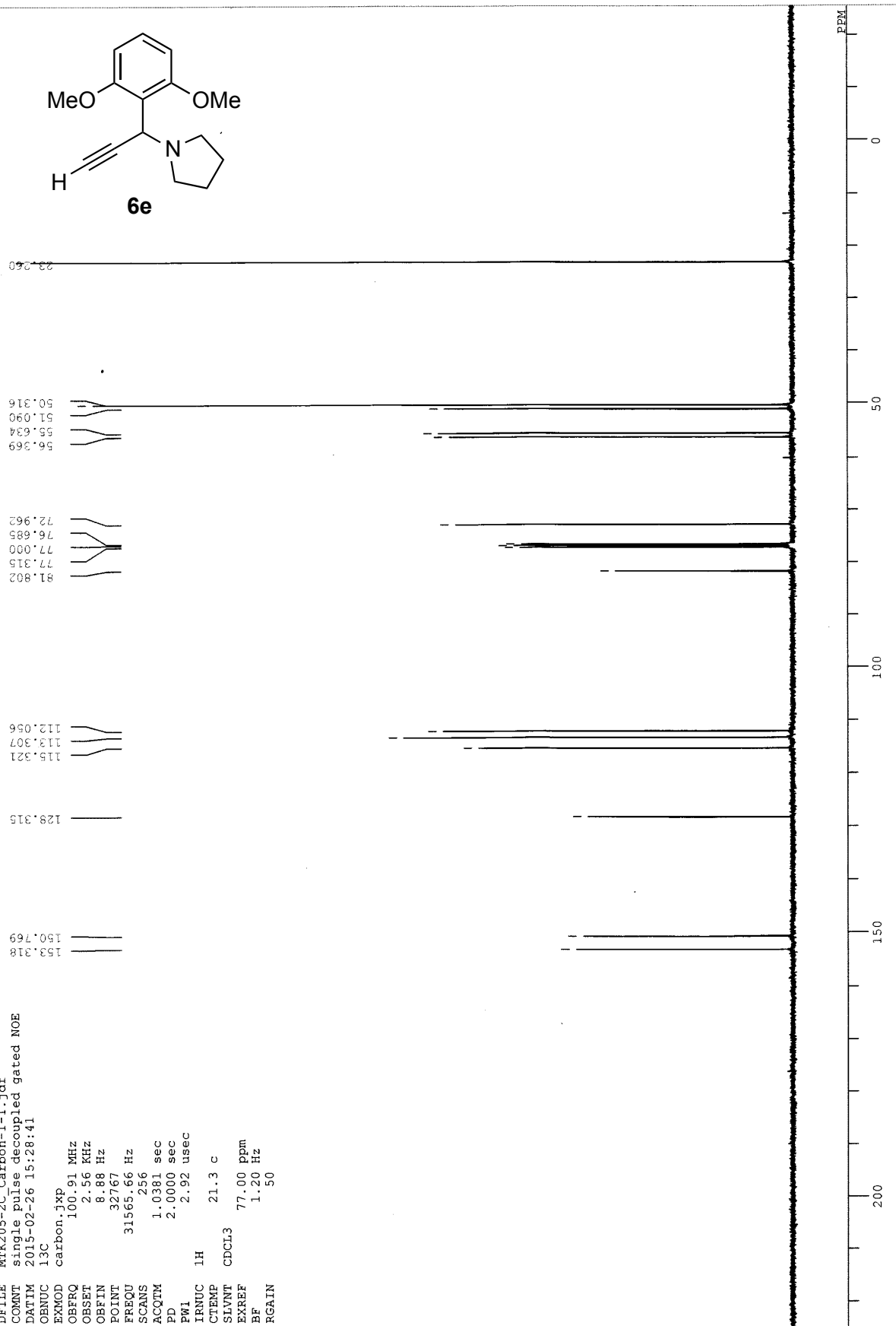
BF 0.12 Hz

RGAIN 46



C:\Users\A\Documents\matatek\NMR\MTK205-2C Carbon-1-1.jdf

DFIL MTK205-2C Carbon-1-1.jdf
 COMNT single pulse decoupled gated NOE
 DATIM 2015-02-26 15:28:41
 OBNUC 13C
 EXMOD carbon.jpg
 OBFRQ 100.91 MHz
 OBSET 2.56 KHz
 OBFIN 8.88 Hz
 POINT 32767
 FREQU 31565.66 Hz
 SCANS 256
 ACQTM 1.0381 sec
 PD 2.0000 sec
 PW1 2.92 usec
 IRNUC 1H
 CTMP 21.3 c
 SLVNT CDCL3
 EXREF 77.00 ppm
 BF 1.20 Hz
 RGAIN 50



C:\Users\A\Documents\NMR\MTK199-1H Proton-1-1.jdf

DELE MTK199-1H Proton-1-1.jdf

COMNT single pulse

DATIM 2015-02-12 01:25:11

ORNUC 1H

EXMOD proton.jxp

OBFRQ 401.28 MHz

OBSET 4.20 KHz

OBFIN 4.97 Hz

POINT 16384

FREQU 7530.12 Hz

SCANS 8

ACQTM 2.1758 sec

PD 5.0000 sec

PW1 4.62 usec

IRNUC 1H

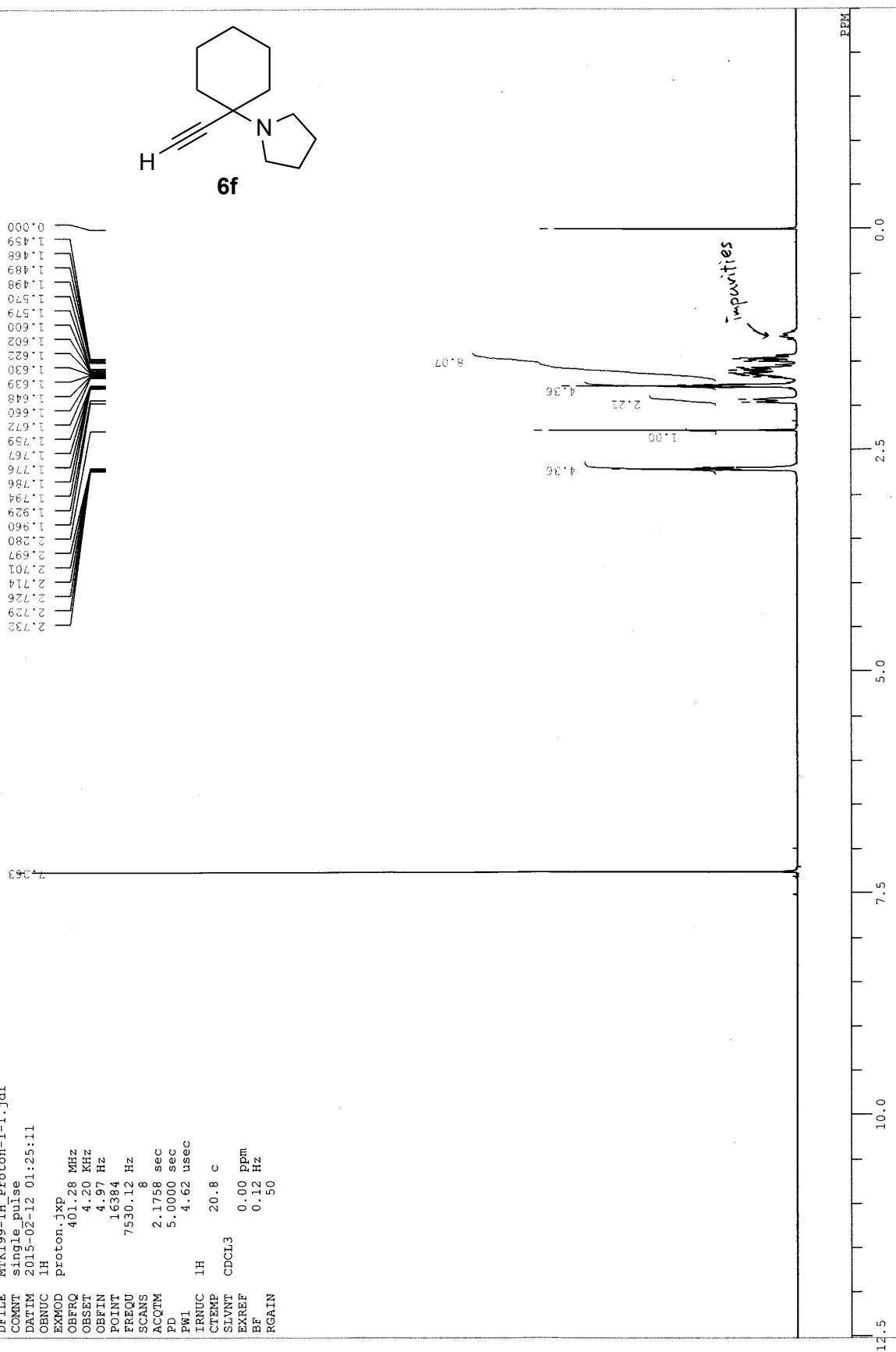
CTEMP 20.8 c

SLVNT CDCL3

EXREF 0.00 ppm

BF 0.12 Hz

RGAIN 50



C:\Users\N\Documents\data\NMR\MYK199-1C Carbon-1-1.jdf

DFILE MYK199-1C Carbon-1-1.jdf
 COMNT single pulse decoupled gated NOE
 DATIM 2015-02-12 01:35:42
 OBNUC 13C
 EXMOD carbon.jxp
 OBFRQ 100.91 MHz
 OBSFQ 2.56 KHz
 OBSFQ 8.88 Hz
 OBSFQ 32767
 POINT 31565.66 Hz
 FREQU 256
 SCANS 1.0381 sec
 ACQTM 2.0000 sec
 PD 2.92 usec
 PW1 1H
 IRNUC 20.9 c
 CTMP CDCL3
 SLVNT 77.00 ppm
 EXREF 1.20 Hz
 Bf 50
 RGAIN

