## Supporting Information

# Chalcogen and Pnicogen Bonds in Complexes of Neutral Icosahedral and Bicapped SquareAntiprismatic Heteroboranes 

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Table S1. The structural and energetic characteristics of $\mathbf{C l}^{-} \mathbf{S B}_{11}$ and $\mathbf{S B}_{\mathbf{9}}$ complexes. The B12-chalcogen-chalcogen-bond acceptor angle ( $\alpha$ ) in degrees. The chalcogen...chalcogen-bond acceptor distance (d) in $\AA$. The interaction energy ( $\Delta \mathrm{E}$ ) and its decomposition into electrostatic $\left(\mathrm{E}_{1}{ }^{\mathrm{Pol}}\right.$ ), exchange-repulsion $\left(\mathrm{E}_{1}{ }^{\mathrm{Ex}}\right)$, dispersion $\left(E_{2}{ }^{\mathrm{D}}\right.$ ) and induction ( $\mathrm{E}^{\text {Ind }}$ ) terms; energy in kcal $\mathrm{mol}^{-1}$.

| Complex | $\alpha$ | d | DFD-D3 | DFT-SAPT |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | $\Delta \mathrm{E}$ | $\mathrm{E}_{1}{ }^{\text {Pol }}$ | $\mathrm{E}_{1}{ }^{\text {Ex }}$ | $\mathrm{E}_{2}{ }^{\text {D }}$ | $\mathrm{E}^{\operatorname{lnd}[a}$ | $\Delta \mathrm{E}$ |
| Cl-SB ${ }_{11}$ |  |  |  |  |  |  |  |  |
| Cl-SB ${ }_{11} \ldots$ BEN | 180 | 3.11 | $-6.2$ | -5.4 | 9.2 | -8.4 | -2.1 | -6.7 |
|  | 165 | 3.13 | -6.3 | -5.4 | 9.3 | -8.6 | -2.0 | -6.7 |
| Cl-SB ${ }_{11} \ldots$ TMA | 180 | 3.24 | -4.2 | -4.0 | 5.6 | $-4.8$ | -1.1 | $-4.3$ |
|  | 140 | 3.03 | $-6.3$ | -7.5 | 11.5 | -7.8 | -2.2 | -6.1 |
| $\mathrm{Cl}^{\text {-SB }} 11 \ldots$ DME | 180 | 3.18 | -3.2 | -2.8 | 3.7 | -3.4 | -0.7 | -3.2 |
|  | 135 | 3.07 | -4.9 | -4.5 | 5.8 | -5.2 | -1.1 | -5.0 |
| $\begin{aligned} & \mathrm{Cl}- \\ & \mathrm{SB}_{11} \ldots \mathrm{DMK} \end{aligned}$ | 180 | 3.21 | -3.7 | -3.4 | 3.7 | -3.2 | -0.9 | -3.9 |
|  | 135 | 3.04 | -5.4 | -5.8 | 7.1 | $-5.5$ | -1.6 | -5.9 |
| Cl-SB ${ }_{11} \ldots \mathrm{FA}$ | 180 | 3.15 | -3.7 | -3.8 | 3.8 | -2.8 | -1.0 | -3.8 |
|  | 130 | 3.01 | -5.9 | -6.6 | 7.6 | -5.3 | -1.8 | -6.1 |
| SB9 |  |  |  |  |  |  |  |  |
| SB9...BEN | 180 | 3.13 | -5.0 | -4.8 | 8.4 | -7.2 | -1.8 | -5.5 |
|  | 150 | 3.19 | -5.3 | -4.6 | 8.2 | -7.8 | -1.6 | -5.8 |
| SB ${ }_{9} \ldots$.. ${ }^{\text {TMA }}$ | 180 | 3.19 | -3.6 | -4.2 | 6.0 | -4.4 | -1.1 | -3.7 |
|  | 125 | 2.98 | -6.1 | -8.3 | 13.1 | -8.4 | -2.4 | -6.0 |
| SB9...DME | 180 | 3.23 | -2.8 | -2.2 | 3.3 | -3.4 | -0.6 | -2.8 |
|  | 125 | 3.08 | $-4.5$ | -4.3 | 6.0 | -5.3 | -1.1 | -4.6 |
| SB9...DMK | 180 | 3.20 | -3.1 | -3.2 | 4.2 | -3.3 | -0.9 | -3.3 |
|  | 120 | 3.11 | -5.7 | -5.2 | 6.6 | -5.6 | -1.4 | -5.6 |
| $\mathbf{S B}_{9} \ldots$..FA | 180 | 3.17 | -3.2 | -3.3 | 3.4 | -2.4 | -1.0 | -3.4 |
|  | 115 | 3.14 | -5.3 | -3.3 | 3.4 | -2.4 | -1.1 | -3.6 |

$[\mathrm{a}] \mathrm{E}^{\mathrm{Ind}}=\mathrm{E}_{2}^{\mathrm{Ind}}+\mathrm{E}_{2}^{\mathrm{Ex}-\mathrm{Ind}}+\delta \mathrm{HF}$

Figure S1. Interaction energy ( $\Delta \mathrm{E}$ ) plotted against B12-S-chalcogen bond acceptor angle. Structures of complexes at an optimal angle (right) and at 180 degrees (left) are shown. Energy in $\mathrm{kcal} \mathrm{mol}^{-1}$, angle in degrees and Color coding as follows: black, C; light-pink, B; yellow, S; white, H.


Figure S2. Interaction energy ( $\Delta \mathrm{E}$ ) plotted against the centre of B9 and B12 -the centre of P1 and P2 -pnictogenbond acceptor angle. Energy in $\mathrm{kcal} \mathrm{mol}^{-1}$, angle in degrees


