## Supporting Information

## Design, synthesis and biological evaluation of novel matrix metalloproteinase inhibitors as potent antihemorrhagic agents: from hit identification to an optimized lead.

Josune Orbe, ${ }^{1}$ Juan A. Sánchez, ${ }^{2}$ Obdulia Rabal, ${ }^{2}$ José A. Rodríguez, ${ }^{1}$ Agustina Salicio, ${ }^{1}$ Ana Ugarte, ${ }^{2}$ Miriam Belzunce, ${ }^{1}$ Musheng Xu, ${ }^{4}$ Wei Wu, ${ }^{4}$ Haizhong Tan, ${ }^{4}$ Hongyu Ma, ${ }^{4}$ José A. Páramo, ${ }^{1,3, *}$ and Julen Oyarzabal ${ }^{2,{ }^{*}}$

## Table of Contents

S1. Protocol for Prep-HPLC purification method................................................ S2
S2. Protocol for SFC purification method.............................................................. S2
S3. Details for Optical Rotation.......................................................................... S2
S4. Table S1. Selectivity profile $10 \mu \mathrm{M}$ against different MMPs............................ S 3
S5. Synthesis of benzyl 4-chlorocarbonyl benzoate (Int. 1)................................... S3

S7. HPLC traces for final compounds.......................................................................... 5
S8. Synthesis of compound 23b...................................................................... S20
S9. X-ray crystallographic data of compound 23b.............................................. 21

## Protocol for Prep-HPLC purification method:

The HPLC measurement was performed using Gilson 281 from 233 pump (binary), an autosampler, and a UV detector. The fractions were detected by LC-MS. The MS detector was configured with an electrospray ionization source. The source temperature was maintained at $300-350{ }^{\circ} \mathrm{C}$. Reverse phase HPLC was carried out on Luna C18 $(100 \times 30 \mathrm{~mm} ; 4$ or 5 um$)$. Solvent A: water with $0.075 \%$ trifluoroacetic acid; Solvent B: acetonitrile with $0.075 \%$ trifluoroacetic acid. Gradient: At room temperature, $20 \%$ of B to $40 \%$ of B within 6 min at $25 \mathrm{~mL} / \mathrm{min}$; then $40 \%$ B at $25 \mathrm{~mL} / \mathrm{min}$ over 2 min , UV detector.

## Protocol for SFC purification method:

## Analytical separation method:

Analytical separations were performed using a Thar analytical SFC with a ChiralPak AD-H column ( $250 \times 4.6 \mathrm{~mm}$ ). Solvent A: $\mathrm{CO}_{2}$; Solvent B: methanol with $0.05 \%$ DEA. Mobile phase: $50 \%$ of A and $50 \%$ of B at $2.0 \mathrm{~mL} / \mathrm{min}$. Back pressure: 100 bar; and column temperature: $35^{\circ} \mathrm{C}$. UV detector at 220 nm .

## Preparative separation method:

Preparative separation was performed using a Mg II preparative SFC with a ChiralPak AD-H column ( $250 \times 30 \mathrm{~mm}$ ). Solvent A: $\mathrm{CO}_{2}$; Solvent B: ethanol with $0.1 \%$ $\mathrm{NH}_{3} \cdot \mathrm{H}_{2} \mathrm{O}$. Mobile phase $50 \%$ of A and $50 \%$ of B at $40 \mathrm{~mL} / \mathrm{min}$. Back pressure: 100 bar; and column temperature: $38^{\circ} \mathrm{C}$. UV detector at 220 nm .

Samples were prepared by dissolving in methanol to $\sim 12 \mathrm{mg} / \mathrm{ml}$ and 2.0 mL were used per injection. After separation, the fractions were dried off via rotary evaporation (bath temperature $40^{\circ} \mathrm{C}$ ) to get the desired isomers. Purity was tested via LCMS.

## Details for Optical Rotation of compounds 5 and 22:

Optical rotation was measured with an AUTOPOL V polarimeter at room temperature and with a wavelength of 589 nm .
Compound 5
$\mathrm{C}=0.9762 \mathrm{~g} / 100 \mathrm{ml}$ diluted with methanol
$\mathrm{L}=1 \mathrm{dm}$
$\alpha_{1}=+0.163 ; \alpha_{2}=+0.162 ; \alpha_{3}=+0.163 ; \bar{\alpha}=+0.163 ; \operatorname{RSD}=0.35 \% ;$
$[\alpha]_{D}^{20}=+16.66^{\circ} \pm 0.06^{\circ}$
Compound 22
$\mathrm{C}=0.9668 \mathrm{~g} / 100 \mathrm{ml}$ diluted with methanol
$\mathrm{L}=1 \mathrm{dm}$
$\alpha_{1}=-0.166 ; \alpha_{2}=-0.165 ; \alpha_{3}=-0.165 ; \bar{\alpha}=-0.165 ; \mathrm{RSD}=0.35 \% ;$
$[\alpha]_{D}^{2 \mathrm{o}}=-17.10^{\circ} \pm 0.06^{\circ}$

## Table S1. Selectivity profile, \% inhibition, at $10 \mu \mathrm{M}$ against different MMPs

| Cpd | MMP1 | MMP2 | MMP7 | MMP8 | MMP9 | MMP12 | MMP13 | MMP14 | MMP20 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 13 | 90.37 | 101.14 | 59.67 | 101.34 | 99.92 | 104.89 | 100 | 100.2 | N/D |
| 16 | 85.11 | 124.81 | 56.48 | 102.31 | 101.56 | 101.57 | N/D | 100.37 | N/D |
| 19d | 86.1 | 100.65 | 65.59 | 99.82 | 102.15 | 99.67 | N/D | 92.48 | N/D |
| 19p | 9 | 100.69 | 26.55 | 96.52 | 98.07 | 100.03 | N/D | 55.78 | N/D |
| 20a | N/D | 100 | N/D | N/D | 100 | N/D | 100 | N/D | N/D |

## Synthesis of benzyl 4-chlorocarbonyl benzoate (Int. 1)



To a solution of commercially available terepthalic acid ( $10 \mathrm{~g}, 0.06 \mathrm{~mol}$ ) in DCM (100 mL ) was added $\mathrm{Et}_{3} \mathrm{~N}(12.12 \mathrm{~g}, 2$ eq). Then bromide ( $9.27 \mathrm{~g}, 0.9 \mathrm{eq})$ in DCM ( 20 mL ) was added dropwise at room temperature. The resulting mixture was stirred at room temperature for 3 h until TLC (PE:AE=1:1) detected the most starting material was consumed, then concentrated and the mixture was extracted with EtOAc, washed with brine, dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated to give the crude compound which was chromatographed on silica gel to give pure compound Int.1-A ( $2.2 \mathrm{~g}, 14 \%$ ) as a white solid. MS $m / z 257[\mathrm{M}+\mathrm{H}]^{+}$calc. for $\mathrm{C}_{15} \mathrm{H}_{12} \mathrm{O}_{4}$.

To the solution of Int.1-A ( 800 mg .3 .13 mmol ) in DCM ( 20 mL ) and DMF (cat.) was added oxalyl dichloride ( $788 \mathrm{mg}, 2 \mathrm{eq}$ ) at $0-5^{\circ} \mathrm{C}$ and the reaction was stirred at RT for 2 h. Then the mixture was concentrated to give the crude product Int. $1(860 \mathrm{mg})$ which was used for the next step directly. MS $m / z 275[\mathrm{M}+\mathrm{H}]^{+}$calc. for $\mathrm{C}_{15} \mathrm{H}_{11} \mathrm{ClO}_{3}$.

## HPLC purities of final compounds

HPLC-analysis was performed using a Shimadzu LC-20AB or LC-20AD with a LunaC18(2), 5um, $2.0 * 50 \mathrm{~mm}$ column at $40^{\circ} \mathrm{C}$ and UV detection. Three different methods were used.

## Method 1

Solvent A: water with $0.056 \%$ trifluoroacetic acid; Solvent B: acetonitrile with $0.056 \%$ trifluoroacetic acid. Gradient: After 0.1 minutes at the initial condition of $90 \% \mathrm{~A}$ and $10 \% \mathrm{~B}$, solvent B was increased to $80 \%$ over 4 minutes, maintained at $80 \%$ for 0.9 minutes, then a linear gradient to initial conditions was applied for 0.02 minutes and maintained for 0.58 minutes to re-equilibrate the column, giving a cycle time of 5.50 minutes. Flow rate was $0.8 \mathrm{~mL} / \mathrm{min}$ from 0.01 to 4.90 minutes, increased to $1.2 \mathrm{~mL} / \mathrm{min}$ in 0.03 minutes and maintained until the end of the run.

## Method 2

Solvent A: water with $0.056 \%$ trifluoroacetic acid; Solvent B: acetonitrile with $0.056 \%$ trifluoroacetic acid. Gradient: After 0.4 minutes at the initial condition of $100 \% \mathrm{~A}$, solvent B was increased to $60 \%$ over 4 minutes, maintained at $60 \%$ for 0.8 minutes, then a linear gradient to initial conditions was applied for 0.02 minutes and maintained for 0.68 minutes to re-equilibrate the column, giving a cycle time of 5.90 minutes. Flow rate was $0.8 \mathrm{~mL} / \mathrm{min}$ from 0.01 to 5.21 minutes, increased to $1.2 \mathrm{~mL} / \mathrm{min}$ in 0.02 minutes and maintained until the end of the run.

## Method 3

Solvent A: water with $0.037 \%$ trifluoroacetic acid; Solvent B: acetonitrile with $0.019 \%$ trifluoroacetic acid. Gradient: After 0.1 minutes at the initial condition of $100 \% \mathrm{~A}$, solvent B was increased to $60 \%$ over 4 minutes, maintained at $60 \%$ for 2 minutes, then a linear gradient to initial conditions was applied for 0.01 minutes and maintained for 1.79 minutes to re-equilibrate the column, giving a cycle time of 7.80 min . Flow rate was $3 \mathrm{~mL} / \mathrm{min}$.

## Method 4

Solvent A: water with $0.056 \%$ trifluoroacetic acid; Solvent B: acetonitrile with $0.019 \%$ trifluoroacetic acid. Gradient: After 0.1 minutes at the initial condition of $90 \% \mathrm{~A}$ and $10 \% \mathrm{~B}$, solvent B was increased to $80 \%$ over 4 minutes, maintained at $80 \%$ for 2 minutes, then a linear gradient to initial conditions was applied for 0.01 minutes and maintained for 0.68 minutes to re-equilibrate the column, giving a cycle time of 7.80 minutes. Flow rate was $3 \mathrm{~mL} / \mathrm{min}$.

## HPLC purities of final compounds

| Compound | Method | Rt | Purity (\%) |
| :---: | :---: | :---: | :---: |
| $\mathbf{1 1}$ | 3 | 2.38 | 99.53 |
| $\mathbf{1 2}$ | 1 | 2.56 | 98.38 |
| $\mathbf{1 3}$ | 4 | 2.70 | 99.36 |
| $\mathbf{1 4}$ | 1 | 2.18 | 96.97 |
| $\mathbf{1 5}$ | 1 | 2.31 | 97.88 |
| $\mathbf{1 6}$ | 3 | 3.90 | 98.96 |
| $\mathbf{1 7}$ | 1 | 3.30 | 97.98 |
| $\mathbf{1 8}$ | 1 | 2.58 | 95.33 |
| $\mathbf{1 9 a}$ | 3 | 3.53 | 97.68 |
| $\mathbf{1 9 b}$ | 4 | 3.30 | 99.74 |
| $\mathbf{1 9 c}$ | 1 | 2.37 | 99.38 |
| $\mathbf{1 9 d}$ | 3 | 3.04 | 98.23 |
| $\mathbf{1 9 e}$ | 3 | 3.28 | 95.46 |
| $\mathbf{1 9 f}$ | 3 | 2.58 | 95.31 |
| $\mathbf{1 9 g}$ | 3 | 3.23 | 98.50 |
| $\mathbf{1 9 h}$ | 3 | 3.44 | 98.01 |
| $\mathbf{1 9 i}$ | 3 | 3.31 | 98.35 |


| $\mathbf{1 9 j}$ | 2 | 2.82 | 96.13 |
| :---: | :---: | :---: | :---: |
| $\mathbf{1 9 k}$ | 2 | 3.41 | 99.63 |
| $\mathbf{1 9 1}$ | 1 | 2.69 | 97.88 |
| $\mathbf{1 9 m}$ | 3 | 3.50 | 98.88 |
| $\mathbf{1 9 n}$ | 1 | 1.98 | 97.99 |
| $\mathbf{1 9 0}$ | 3 | 2.50 | 98.14 |
| $\mathbf{1 9 p}$ | 1 | 3.51 | 100 |
| $\mathbf{2 0 a}$ | 4 | 2.66 | 95.45 |
| $\mathbf{2 0 b}$ | 4 | 2.96 | 96.71 |
| $\mathbf{2 0 c}$ | 4 | 3.21 | 99.09 |
| $\mathbf{2 1 a}$ | 4 | 2.71 | 99.59 |
| $\mathbf{2 1 b}$ | 2 | 2.79 | 96.16 |
| $\mathbf{2 2}$ | 1 | 1.61 | 99.46 |

## HPLC traces for final compounds

## Compound 11



| $0 \mathrm{~nm}, 8 \mathrm{~nm}$ |  |  |  |
| :---: | :---: | :---: | :---: |
| Retention Time | Height | Area | Area Percent |
| 2.28 | 2151 | 4456 | 0.10 |
| 2.38 | 1387223 | 4644288 | 99.53 |
| 3.35 | 1530 | 5109 | 0.11 |
| 5.35 | 1497 | 4427 | 0.09 |
| 5.65 | 2029 | 7734 | 0.17 |

## Compound 12



## Compound 13



## Compound 14



| 1: $\mathbf{2 2 0} \mathbf{n m , ~ 8 ~ n m ~}$ <br> Retention Time | Height | 18071 | Area |
| :---: | :---: | :---: | :---: |

## Compound 15



## Compound 16



| 1: $220 \mathrm{~nm}, 8 \mathrm{~nm}$ Retention Time | Height | Area | Area Percent |
| :---: | :---: | :---: | :---: |
| 3.80 | 8198 | 21976 | 0.49 |
| 3.90 | 1397215 | 4438032 | 98.96 |
| 5.21 | 2179 | 6477 | 0.14 |
| 5.46 | 4427 | 18278 | 0.41 |

## Compound 17



## Compound 18



1: $220 \mathrm{~nm}, 8 \mathrm{~nm}$

| Retention Time | Height | Area | Area Percent |
| :---: | :---: | :---: | :---: |
| 0.33 | 54806 | 184568 | 3.17 |
| 2.18 | 9637 | 37987 | 0.65 |
| 2.58 | 1423241 | 5553713 | 95.33 |
| 2.71 | 7322 | 18007 | 0.31 |
| 2.89 | 8881 | 31711 | 0.54 |

## Compound 19a



## Compound 19b



## Compound 19c



## Compound 19d



| 1: $\mathbf{2 2 0} \mathbf{~ n m , ~} \mathbf{8} \mathbf{~ n m}$ <br> Retention Time | Height | Area | Area Percent |
| :---: | :---: | :---: | :---: |
| 0.73 | 2682 | 9227 | 0.26 |
| 2.01 | 17306 | 42971 | 1.21 |
| 3.04 | 1225413 | 3494283 | 98.23 |
| 3.77 | 1350 | 3700 | 0.10 |
| 4.43 | 2773 | 7185 | 0.20 |

## Compound 19e



| 1: $\mathbf{2 2 0} \mathbf{n m , ~} \mathbf{8} \mathbf{n m}$ <br> Retention Time | Height |
| :---: | :---: | :---: | :---: |

## Compound 19f



## Compound 19g



## Compound 19h



| 1: 220 nm, $\mathbf{8} \mathbf{n m}$ |  |  |  |
| :---: | :---: | :---: | :---: |
| Retention Time | Height | 4281 | Area |

## Compound 19i



## Compound 19j



1: $220 \mathrm{~nm}, 8 \mathrm{~nm}$

| Retention Time | Height | Area | Area Percent |
| :---: | :---: | :---: | :---: |
| 1.89 | 19658 | 59812 | 1.17 |
| 2.77 | 72141 | 94405 | 1.84 |
| 2.82 | 1310362 | 4923237 | 96.13 |
| 2.93 | 14962 | 44026 | 0.86 |

## Compound 19k



## Compound 191



## Compound 19m



| 1: $220 \mathrm{~nm}, \mathbf{8} \mathbf{n m}$ |  |  |  |
| :---: | :---: | :---: | :---: |
| Retention Time | Height | 999780 | Area |

## Compound 19n



## Compound 190



| 1: $\mathbf{2 2 0} \mathbf{n m , 8} \mathbf{~ n m}$ |  |  |  |
| :---: | :---: | :---: | :---: |
| Retention Time | Height | 4362 | Area |

## Compound 19p



## Compound 20a



## 1: $220 \mathrm{~nm}, 8 \mathrm{~nm}$

| Retention Time | Height | Area | Area Percent |
| :---: | :---: | :---: | :---: |
| 2.55 | 3225 | 8529 | 0.18 |
| 2.66 | 1145745 | 4416883 | 95.45 |
| 2.92 | 4770 | 17847 | 0.39 |
| 4.07 | 3830 | 14497 | 0.31 |
| 4.18 | 1748 | 13377 | 0.29 |
| 4.44 | 1624 | 10097 | 0.22 |
| 4.76 | 1976 | 13714 | 0.30 |
| 5.52 | 7047 | 132430 | 2.86 |

## Compound 20b



| 1: 220 nm, 8 nm |  |  |  |
| :---: | :---: | :---: | :---: |
| Retention Time | Height | Area | Area Percent |
| 2.49 | 3172 | 6850 | 0.18 |
| 2.79 | 4329 | 11072 | 0.30 |
| 2.90 | 5273 | 11743 | 0.32 |
| 2.96 | 1118856 | 3589961 | 96.71 |
| 3.13 | 5052 | 47269 | 1.27 |
| 4.14 | 2265 | 7259 | 0.20 |
| 4.26 | 5092 | 12388 | 0.33 |
| 4.48 | 3652 | 9455 | 0.25 |
| 4.96 | 6331 | 16001 | 0.43 |

## Compound 20c



## Compound 21a



## Compound 21b



| 3: $\mathbf{2 5 4} \mathbf{n m , ~ 8 ~ n m ~}$ <br> Retention Time | Height | Area | Area Percent |
| :---: | :---: | :---: | :---: |
| 2.10 | 13371 | 57640 | 0.86 |
| 2.66 | 19474 | 45756 | 0.68 |
| 2.79 | 1498949 | 6471428 | 96.16 |
| 2.88 | 42065 | 116223 | 1.73 |
| 3.03 | 4091 | 7877 | 0.12 |
| 3.09 | 9621 | 19399 | 0.29 |
| 3.18 | 4351 | 11663 | 0.17 |

## Compound 22



| 1: 220 nım, 8 nin <br> Retention Time | Height | Area | Area Percent |
| :---: | :---: | :---: | :---: |
| 1.61 | 2322641 | 5618005 | 99.46 |
| 1.76 | 14640 | 30406 | 0.54 |

## Synthesis of (3S)-3-(4-fluorophenyl)sulfonyl-8-azaspiro[4.5]decane-3-carboxylic acid (23b)



A racemic mixture of compound $\mathbf{2 6}^{1}$ was first separated by SFC. Then, enantiomerically pure $26 \mathrm{a}(60 \mathrm{mg}, 0.14 \mathrm{mmol})$ was dissolved in $\mathrm{HCl} / \mathrm{EtOAc}(2 \mathrm{~N}, 10 \mathrm{~mL})$ and the mixture was stirred at RT for 1 h . Then, concentrated to give the crude product which was purified by prep-HPLC to obtained pure compound $\mathbf{2 3 b}(30 \mathrm{mg}, 59 \%)$ as a yellow oil. MS $m / z 342[\mathrm{M}+\mathrm{H}]^{+}$calc. for $\mathrm{C}_{16} \mathrm{H}_{2} 2 \mathrm{FNO} 4 \mathrm{~S}$. ${ }^{1} \mathrm{H}$ NMR (MeOD, 400 MHz ): $\delta$ 7.97-7.94 (m, 2H), 7.38-7.34 (m, 2H), 3.21-3.15 (m, 4H), 2.53-2.50 (m, 2H), 2.45$2.38(\mathrm{~m}, 2 \mathrm{H}), 1.88-1.84(\mathrm{~m}, 3 \mathrm{H}), 1.70-1.67(\mathrm{~m}, 3 \mathrm{H})$.

## Protocol for compound 23b crystallization

We confirm the structure of compound 23b by X-ray single crystal diffraction analysis. Compound 23b was dissolved in a three-phase solvent system $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} / \mathrm{THF}$ ( $10: 1: 1$ ) and slow evaporation at room temperature for three days led to creation of crystals which were utilized for X-ray diffraction analysis on a rigaku saturn diffractometer using graphic-monochromated Mo K $\alpha$ radiation ( $\lambda=0.71073 \AA$ ). The structures were solved using the SHELXS program and refined with SHELXL.

## X-ray crystallographic data of compound 23b

| Crystal size | $0.20 \times 0.18 \times 0.12 \mathrm{~mm}^{3}$ |
| :--- | :--- |
| Radiation type | $\mathrm{Mo} \mathrm{K} \alpha(? \lambda=0.71073 \AA)$ |
| Space group | Orthorhombic $/ \mathrm{P} 2(1) 2(1) 2(1)$ |
| Cell size | $\mathrm{a}=7.2894(15) \AA$ |
|  | $\mathrm{b}=11.834(2) \AA$ |
|  | $\mathrm{c}=21.401(4) \AA$ |
|  | $\alpha=90.00^{\circ}$ |
|  | $\beta=90.00^{\circ}$ |
|  | $\gamma=90.00^{\circ}$ |
| Cell volume | $V=1846.1(6) \AA^{3}$ |
| Cell formula units | $Z=4$ |
| Crystal density | $D_{c}=1.359 \mathrm{Mg} / \mathrm{m}^{3}$ |
| Crystal $F(000)$ | 792.0 |
| Absorpt coefficient mu | $\mu(\mathrm{Mo} \mathrm{K} \alpha)=0.348 \mathrm{~mm}^{-1}$ |
| Limiting indices | $-7 \leq h \leq 9$ |
|  | $-15 \leq k \leq 15$ |
|  | $-24 \leq l \leq 28$ |
| Cell measurement temperature | $T=293(2) \mathrm{K}$. |

## Details for compound 23b coordinates

```
Table 1. Crystal data and structure refinement for 23b
Empirical formula C16 H21 Cl F N O4 S
Formula weight 377.85
Temperature 293(2) K
Wavelength 0.71073 A
Crystal system, space group Orthorhombic, P2(1)2(1)2(1)
Unit cell dimensions a = 7.2894(15)A alpha = 90 deg.
    b = 11.834(2)A beta = 90 deg.
    c = 21.401(4) A gamma = 90 deg.
Volume
    1846.1(6) A^3
Z, Calculated density
    4, 1.359 Mg/m^3
Absorption coefficient
    0.348 mm^-1
F(000)
    7 9 2
Crystal size 0.20 x 0.18 x 0.12 mm
Theta range for data collection 1.90 to 27.96 deg.
Limiting indices -7<=h<=9, -15<= k<=15, -24<=l<=28
```

| Reflections collected / unique | $17535 / 4398$ [R(int) $=0.0461]$ |
| :---: | :---: |
| Completeness to theta $=27.96$ | 98.8 \% |
| Absorption correction | Semi-empirical from equivalents |
| Max. and min. transmission | 0.9594 and 0.9336 |
| Refinement method | Full-matrix least-squares on $\mathrm{F}^{\wedge} 2$ |
| Data / restraints / parameters | $4398 / 3 / 225$ |
| Goodness-of-fit on $\mathrm{F}^{\wedge} 2$ | 1.050 |
| Final R indices [I>2sigma(I)] | $\mathrm{R} 1=0.0509, \mathrm{wR} 2=0.1117$ |
| R indices (all data) | $\mathrm{R} 1=0.0645, \mathrm{wR} 2=0.1198$ |
| Absolute structure parameter | -0.02 (8) |
| Extinction coefficient | 0.025 (2) |
| Largest diff. peak and hole | 0.273 and -0.237 e. $\mathrm{A}^{\wedge}-3$ |

Table 2. Atomic coordinates ( x 10^4) and equivalent isotropic displacement parameters ( $A^{\wedge} 2 \times 10^{\wedge} 3$ ) for $23 b(e q)$ is defined as one third of the trace of the orthogonalized Uij tensor.

|  | X | Y | z | $\mathrm{U}(\mathrm{eq})$ |
| :---: | :---: | :---: | :---: | :---: |
| S (1) | 9763 (1) | 6512 (1) | 1995 (1) | 47 (1) |
| F (1) | 4549 (3) | 2780 (2) | 1722 (1) | 107 (1) |
| O(1) | 12893 (3) | 6339 (2) | 668 (1) | 78 (1) |
| O (2) | 10158 (3) | 5511 (2) | 639 (1) | 67 (1) |
| O(3) | 11425 (3) | 6031 (2) | 2248 (1) | 58 (1) |
| O (4) | 8827 (3) | 7373 (2) | 2350 (1) | 65 (1) |
| N(1) | 11794 (3) | 11312 (2) | 374 (1) | 58 (1) |
| C (1) | 11310 (4) | 6296 (2) | 827 (1) | 48 (1) |
| C (2) | 10394 (4) | 7167 (2) | 1248 (1) | 43 (1) |
| C (3) | 8678 (4) | 7717 (2) | 937 (1) | 51 (1) |
| C (4) | 8854 (4) | 8979 (2) | 1064 (1) | 50 (1) |
| C (5) | 10943 (4) | 9203 (2) | 1051 (1) | 42 (1) |
| C (6) | 11693 (4) | 8163 (2) | 1415 (1) | 49 (1) |
| C (7) | 11526 (4) | 10293 (2) | 1377 (1) | 55 (1) |
| C (8) | 10994 (5) | 11352 (2) | 1014 (2) | 60 (1) |
| C (9) | 11698(4) | 9226 (2) | 377 (1) | 47 (1) |
| C (10) | 11171 (4) | 10288(2) | 26 (1) | 52 (1) |
| C (11) | 8177 (4) | 5398 (2) | 1875 (1) | 45 (1) |
| C (12) | 8823 (5) | 4301 (2) | 1803(1) | 53 (1) |
| C (13) | 7578 (5) | 3406 (3) | 1749 (2) | 60 (1) |
| C (14) | 5758 (5) | 3654 (3) | 1773 (2) | 66 (1) |
| C (15) | 5086 (5) | 4719 (3) | 1857 (2) | 78 (1) |
| C(16) | 6319 (4) | 5619 (3) | 1911 (2) | 64 (1) |
| Cl(1) | 6149 (1) | 1346 (1) | 262 (1) | 55 (1) |

Table 3. Bond lengths [A] and angles [deg] for 23b

| S (1)-O(4) | 1.443 (2) |
| :---: | :---: |
| S (1)-O(3) | 1.444 (2) |
| S (1) - C (11) | 1.772 (3) |
| S (1)-C(2) | 1.836 (3) |
| F(1)-C(14) | 1.364 (4) |
| $\mathrm{O}(1)-\mathrm{C}(1)$ | 1.205 (3) |
| $\mathrm{O}(2)-\mathrm{C}(1)$ | 1.316 (3) |
| $\mathrm{O}(2)-\mathrm{H}(2)$ | 0.8200 |
| $\mathrm{N}(1)-\mathrm{C}(8)$ | 1.490 (4) |
| $\mathrm{N}(1)-\mathrm{C}(10)$ | 1.493 (4) |
| $\mathrm{N}(1)-\mathrm{H}(1 \mathrm{~A})$ | 0.916 (10) |
| $\mathrm{N}(1)-\mathrm{H}(1 \mathrm{~B})$ | 0.915 (10) |
| $\mathrm{C}(1)-\mathrm{C}(2)$ | 1.522 (4) |
| C (2) - C (6) | 1.553 (4) |
| C (2)-C(3) | 1.558 (4) |
| $\mathrm{C}(3)-\mathrm{C}(4)$ | 1.523 (4) |
| $\mathrm{C}(3)-\mathrm{H}(3 \mathrm{~A})$ | 0.9700 |
| C (3) - H (3B) | 0.9700 |
| C (4) - C (5) | 1.546 (4) |
| $\mathrm{C}(4)-\mathrm{H}(4 \mathrm{~A})$ | 0.9700 |
| $\mathrm{C}(4)-\mathrm{H}(4 \mathrm{~B})$ | 0.9700 |
| C (5) - C ( 7 ) | 1.528 (4) |
| $\mathrm{C}(5)-\mathrm{C}(9)$ | 1.542 (4) |
| C (5)-C (6) | 1.556 (4) |
| $\mathrm{C}(6)-\mathrm{H}(6 \mathrm{~A})$ | 0.9700 |
| $\mathrm{C}(6)-\mathrm{H}(6 \mathrm{~B})$ | 0.9700 |
| $\mathrm{C}(7)-\mathrm{C}(8)$ | 1.524 (4) |
| $\mathrm{C}(7)-\mathrm{H}(7 \mathrm{~A})$ | 0.9700 |
| $\mathrm{C}(7)-\mathrm{H}(7 \mathrm{~B})$ | 0.9700 |
| $\mathrm{C}(8)-\mathrm{H}(8 \mathrm{~A})$ | 0.9700 |
| C (8) - H (8B) | 0.9700 |
| C (9)-C (10) | 1.515 (4) |
| C (9) - H (9A) | 0.9700 |
| C (9) - H (9B) | 0.9700 |
| C (10) - H (10A) | 0.9700 |
| $\mathrm{C}(10)-\mathrm{H}(10 \mathrm{~B})$ | 0.9700 |
| C (11) - C (16) | 1.382 (4) |
| $\mathrm{C}(11)-\mathrm{C}(12)$ | 1.390 (4) |
| C (12) - C (13) | 1.399 (4) |
| C (12) - H (12) | 0.9300 |
| C (13) - C (14) | 1.360 (5) |
| C (13) - $\mathrm{H}(13)$ | 0.9300 |
| C (14)-C (15) | 1.364 (5) |
| C (15) - $\mathrm{C}(16)$ | $1.398(5)$ |
| C (15) - H (15) | 0.9300 |
| C (16) - $\mathrm{H}(16$ ) | 0.9300 |
| O(4)-S(1)-O(3) | 118.50(14) |
| $\mathrm{O}(4)-\mathrm{S}(1)-\mathrm{C}(11)$ | 107.06(13) |
| $\mathrm{O}(3)-\mathrm{S}(1)-\mathrm{C}(11)$ | 107.96(13) |
| O(4)-S (1)-C(2) | 106.22 (12) |
| $\mathrm{O}(3)-\mathrm{S}(1)-\mathrm{C}(2)$ | 106.49(12) |
| $\mathrm{C}(11)-\mathrm{S}(1)-\mathrm{C}(2)$ | 110.55(12) |
| $\mathrm{C}(1)-\mathrm{O}(2)-\mathrm{H}(2)$ | 109.5 |
| $\mathrm{C}(8)-\mathrm{N}(1)-\mathrm{C}(10)$ | 111.4(2) |
| $\mathrm{C}(8)-\mathrm{N}(1)-\mathrm{H}(1 \mathrm{~A})$ | 111.3(19) |
| $\mathrm{C}(10)-\mathrm{N}(1)-\mathrm{H}(1 \mathrm{~A})$ | 108 (2) |
| $\mathrm{C}(8)-\mathrm{N}(1)-\mathrm{H}(1 \mathrm{~B})$ | 109.3(19) |


| $\mathrm{C}(10)-\mathrm{N}(1)-\mathrm{H}(1 \mathrm{~B})$ | 108(2) |
| :---: | :---: |
| $\mathrm{H}(1 \mathrm{~A})-\mathrm{N}(1)-\mathrm{H}(1 \mathrm{~B})$ | 109.3(15) |
| $\mathrm{O}(1)-\mathrm{C}(1)-\mathrm{O}(2)$ | 123.7(3) |
| $\mathrm{O}(1)-\mathrm{C}(1)-\mathrm{C}(2)$ | 124.0(3) |
| $\mathrm{O}(2)-\mathrm{C}(1)-\mathrm{C}(2)$ | 112.3(2) |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(6)$ | 112.5 (2) |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | 112.5 (2) |
| $\mathrm{C}(6)-\mathrm{C}(2)-\mathrm{C}(3)$ | 105.7 (2) |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{S}(1)$ | 109.77(18) |
| $\mathrm{C}(6)-\mathrm{C}(2)-\mathrm{S}(1)$ | 105.79(17) |
| $\mathrm{C}(3)-\mathrm{C}(2)-\mathrm{S}(1)$ | 110.29(19) |
| C (4)-C (3)-C (2) | 105.4(2) |
| $\mathrm{C}(4)-\mathrm{C}(3)-\mathrm{H}(3 \mathrm{~A})$ | 110.7 |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{H}(3 \mathrm{~A})$ | 110.7 |
| $\mathrm{C}(4)-\mathrm{C}(3)-\mathrm{H}(3 \mathrm{~B})$ | 110.7 |
| C (2) - C (3) - H (3B) | 110.7 |
| H (3A) - C (3) - $\mathrm{H}(3 \mathrm{~B})$ | 108.8 |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | 104.3(2) |
| C (3) - C (4)-H(4A) | 110.9 |
| C (5) - C (4)-H (4A) | 110.9 |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{H}(4 \mathrm{~B})$ | 110.9 |
| C (5) - C (4) - H ( 4B) | 110.9 |
| H (4A) - C (4) - H (4B) | 108.9 |
| $\mathrm{C}(7)-\mathrm{C}(5)-\mathrm{C}(9)$ | 108.2(2) |
| C (7)-C(5)-C(4) | 114.2 (2) |
| C (9)-C (5)-C (4) | 111.8(2) |
| $C(7)-C(5)-C(6)$ | 110.0(2) |
| $C(9)-C(5)-C(6)$ | 110.9(2) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | 101.6(2) |
| $\mathrm{C}(2)-\mathrm{C}(6)-\mathrm{C}(5)$ | 105.7(2) |
| $\mathrm{C}(2)-\mathrm{C}(6)-\mathrm{H}(6 \mathrm{~A})$ | 110.6 |
| $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{H}(6 \mathrm{~A})$ | 110.6 |
| $\mathrm{C}(2)-\mathrm{C}(6)-\mathrm{H}(6 \mathrm{~B})$ | 110.6 |
| $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{H}(6 \mathrm{~B})$ | 110.6 |
| H (6A) -C (6) -H (6B) | 108.7 |
| C (8) - C (7)-C (5) | 113.0 (2) |
| $\mathrm{C}(8)-\mathrm{C}(7)-\mathrm{H}(7 \mathrm{~A})$ | 109.0 |
| $\mathrm{C}(5)-\mathrm{C}(7)-\mathrm{H}(7 \mathrm{~A})$ | 109.0 |
| $\mathrm{C}(8)-\mathrm{C}(7)-\mathrm{H}(7 \mathrm{~B})$ | 109.0 |
| $\mathrm{C}(5)-\mathrm{C}(7)-\mathrm{H}(7 \mathrm{~B})$ | 109.0 |
| H (7A) - C ( 7 ) - $\mathrm{H}(7 \mathrm{~B})$ | 107.8 |
| $\mathrm{N}(1)-\mathrm{C}(8)-\mathrm{C}(7)$ | 110.0 (2) |
| $\mathrm{N}(1)-\mathrm{C}(8)-\mathrm{H}(8 \mathrm{~A})$ | 109.7 |
| $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{H}(8 \mathrm{~A})$ | 109.7 |
| $\mathrm{N}(1)-\mathrm{C}(8)-\mathrm{H}(8 \mathrm{~B})$ | 109.7 |
| $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{H}(8 \mathrm{~B})$ | 109.7 |
| H (8A) -C (8) -H (8B) | 108.2 |
| C (10) -C (9)-C (5) | 112.9(2) |
| $\mathrm{C}(10)-\mathrm{C}(9)-\mathrm{H}(9 \mathrm{~A})$ | 109.0 |
| C (5) - C (9)-H(9A) | 109.0 |
| C (10) - C (9)-H(9B) | 109.0 |
| C (5) - C (9)-H(9B) | 109.0 |
| H (9A) - C (9) - $\mathrm{H}(9 \mathrm{~B})$ | 107.8 |
| $\mathrm{N}(1)-\mathrm{C}(10)-\mathrm{C}(9)$ | 110.4(2) |
| $\mathrm{N}(1)-\mathrm{C}(10)-\mathrm{H}(10 \mathrm{~A})$ | 109.6 |
| $\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{H}(10 \mathrm{~A})$ | 109.6 |
| $\mathrm{N}(1)-\mathrm{C}(10)-\mathrm{H}(10 \mathrm{~B})$ | 109.6 |
| $\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{H}(10 \mathrm{~B})$ | 109.6 |
| H (10A) - C (10)-H (10B) | 108.1 |
| $\mathrm{C}(16)-\mathrm{C}(11)-\mathrm{C}(12)$ | 121.0(3) |
| C (16) -C (11)-S (1) | 119.4(2) |

```
C(12)-C(11)-S(1) 119.3(2)
C(11)-C(12)-C(13) 119.8(3)
C(11)-C(12)-H(12) 120.1
C(13)-C(12)-H(12) 120.1
C(14)-C(13)-C(12) 117.8(3)
C(14)-C(13)-H(13) 121.1
C(12)-C(13)-H(13) 121.1
C(13)-C(14)-F(1) 117.6(4)
C(13)-C(14)-C(15) 123.7(3)
F(1)-C(14)-C(15) 118.7(3)
C(14)-C(15)-C(16) 118.9(3)
C(14)-C(15)-H(15) 120.5
C(16)-C (15)-H(15) 120.5
C(11)-C(16)-C(15) 118.8(3)
C(11)-C(16)-H(16) 120.6
C(15)-C (16)-H(16) 120.6
```


## REFERENCES

1. Orbe, J.; Rodriguez, J.A.; Sanchez, J.A.; Salicio, A.; Belzunce, M., Ugarte, A.; Chang, H.C.Y.; Rabal, O.; Oyarzabal, J.; Paramo, J.A. Discovery of a potent and safe pre-clinical candidate, CM-352, for the prevention and treatment of hemorrhage. J. Med. Chem., 2015, XX, XXXX
