

Supporting Information

Surface Morphology of Cu Adsorption on Different Terminations of the Hägg Iron Carbide (χ -Fe₅C₂) Phase

Xinxin Tian,^{a,b,c} Tao Wang,^d Yong Yang,^{a,b} Yong-Wang Li,^{a,b} Jianguo Wang,^a Haijun Jiao^{*,a,d}

a) State Key Laboratory of Coal Conversion, Institute of Coal Chemistry, Chinese Academy of Sciences, Taiyuan, 030001, China; b) National Energy Center for Coal to Liquids, Synfuels CHINA Co., Ltd, Huairou District, Beijing, 101400, China; c) University of Chinese Academy of Sciences, Beijing, 100049, China; d) Leibniz-Institut für Katalyse eV. an der Universität Rostock, Albert-Einstein-Strasse 29a, 18059 Rostock, Germany. E-mail address: haijun.jiao@catalysis.de

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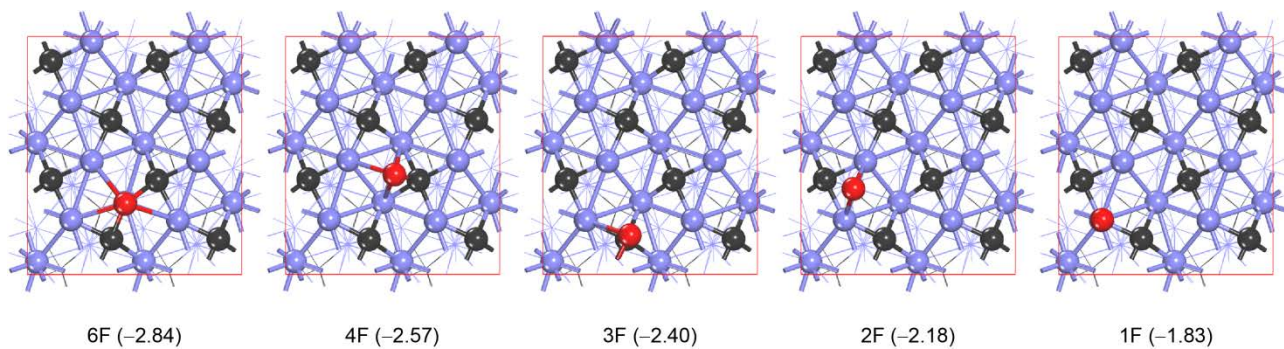
Table S1: Benchmark of slab thickness and relaxation thickness of single Cu atom adsorption on $\text{Fe}_5\text{C}_2(100)$, $\text{Fe}_5\text{C}_2(111)$, $\text{Fe}_5\text{C}_2(510)$, $\text{Fe}_5\text{C}_2(010)$ and $\text{Fe}_5\text{C}_2(001)$ surfaces. (L represents total layers, R represents relaxed layers.)

Test	$\text{Fe}_5\text{C}_2(100)$		$\text{Fe}_5\text{C}_2(111)$		$\text{Fe}_5\text{C}_2(510)$	
		$E_{\text{ads}}(\text{eV})$		$E_{\text{ads}}(\text{eV})$		$E_{\text{ads}}(\text{eV})$
Slab thickness	p(1×1)-4.60Å-7LR6	-2.82	p(2×1)-4.49Å-15LR12	-2.84	p(1×1)-5.18Å-8LR6	-3.27
	p(1×1)-5.72Å-8LR6	-2.82	p(2×1)-5.62Å-18LR12	-2.82	p(1×1)-5.63Å-9LR6	-3.26
	p(1×1)-7.06Å-10LR6	-2.86	p(2×1)-8.74Å-27LR12	-2.83	p(1×1)-7.22Å-11LR6	-3.30
	p(1×1)-9.02Å-12LR6	-2.80			p(1×1)-7.67Å-12LR6	-3.29
	p(1×1)-10.33Å-14LR6	-2.80			p(1×1)-9.71Å-15LR6	-3.29
Relaxation thickness	p(2×2)-5.72Å-8LR6	-2.84	p(2×1)-4.49Å-15LR11	-2.91	p(1×1)-5.63Å-9LR5	-3.26
	p(2×2)-5.72Å-8LR7	-2.82	p(2×1)-4.49Å-15LR12	-2.84	p(1×1)-5.63Å-9LR6	-3.26
	p(2×2)-5.72Å-8LR8	-2.85	p(2×1)-4.49Å-15LR13	-2.80	p(1×1)-5.63Å-9LR7	-3.26
			p(2×1)-4.49Å-15LR14	-2.79	p(1×1)-5.63Å-9LR8	-3.28

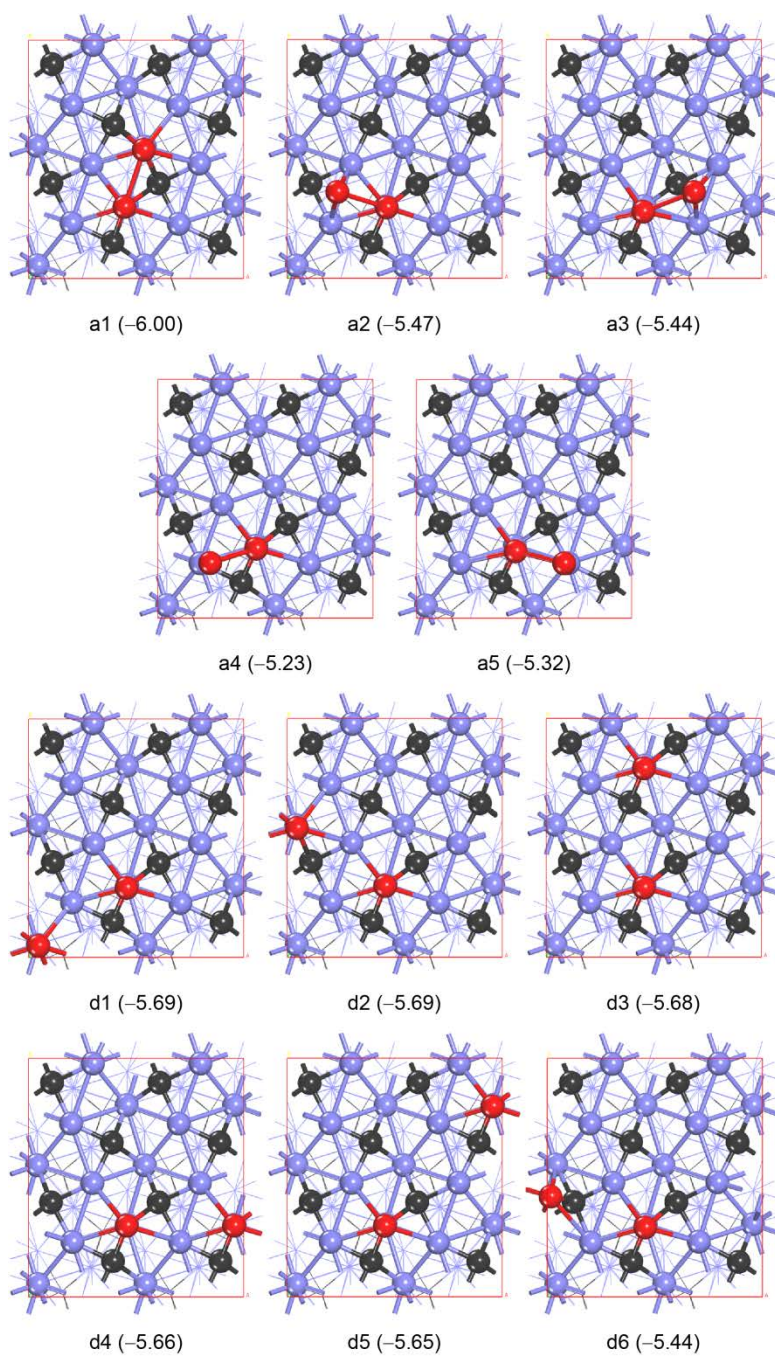
Test	$\text{Fe}_5\text{C}_2(010)$		$\text{Fe}_5\text{C}_2(001)$	
		$E_{\text{ads}}(\text{eV})$		$E_{\text{ads}}(\text{eV})$
Slab thickness	p(1×1)-4.04-8LR8	-3.31	p(1×1)-4.15Å-10LR8	-2.96
	p(1×1)-4.50-9LR8	-3.28	p(1×1)-6.07Å-14LR8	-2.93
	p(1×1)-5.07-10LR8	-3.28	p(1×1)-6.62Å-15LR8	-2.95
	p(1×1)-5.59-11LR8	-3.27	p(1×1)-9.09Å-20LR8	-2.93
	p(1×1)-6.74-13LR8	-3.26		
Relaxation thickness	p(1×1)-5.59-11LR6	-3.29	p(1×1)-6.07Å-14LR8	-2.93
	p(1×1)-5.59-11LR7	-3.29	p(1×1)-6.07Å-14LR9	-2.93
	p(1×1)-5.59-11LR8	-3.27	p(1×1)-6.07Å-14LR10	-2.92
	p(1×1)-5.59-11LR9	-3.25	p(1×1)-6.07Å-14LR11	-2.92
	p(1×1)-5.59-11LR10	-3.25	p(1×1)-6.07Å-14LR12	-2.92

Figure S1. Various optimized Cu₁₋₁₃, Cu₁₆ configurations on the $p(2\times 2)$ Fe₅C₂(100) surface (adsorption energy in eV)

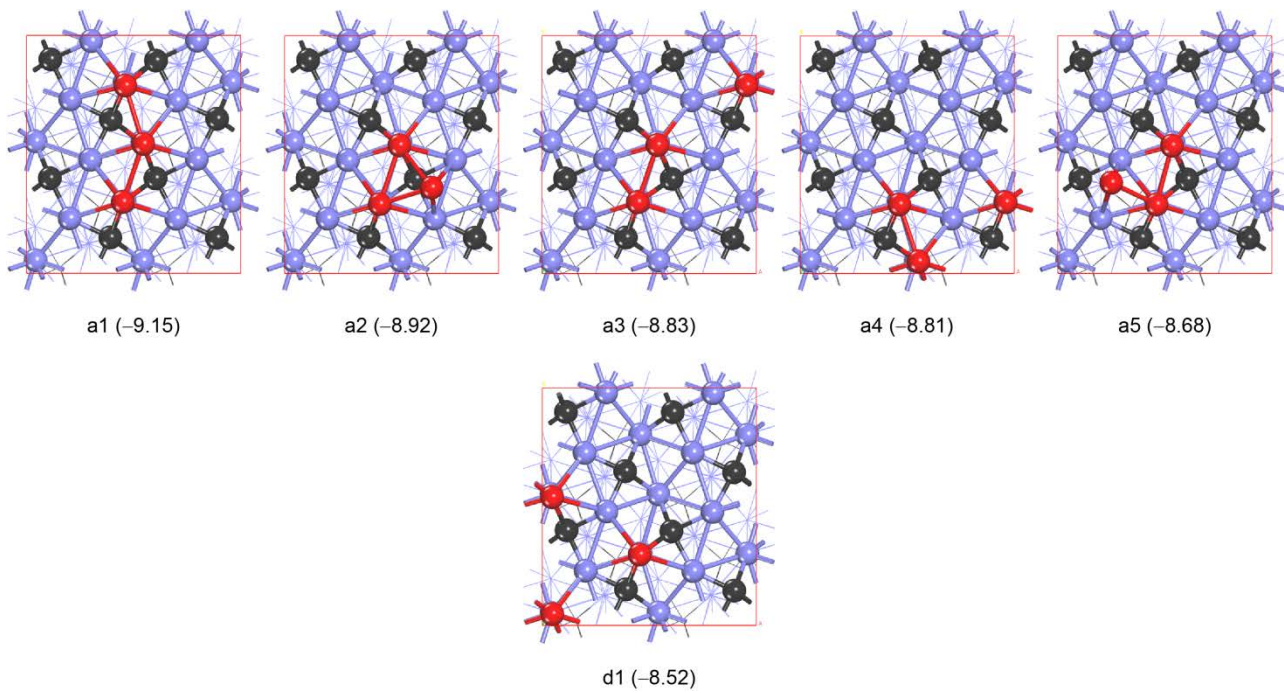
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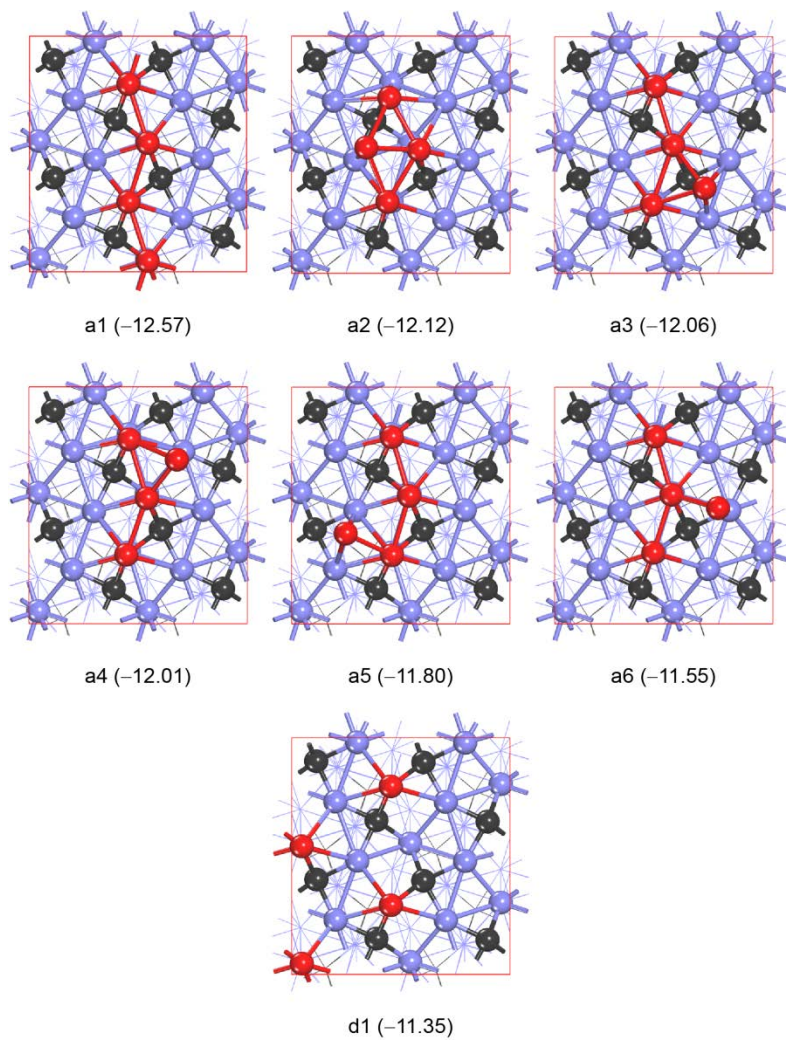
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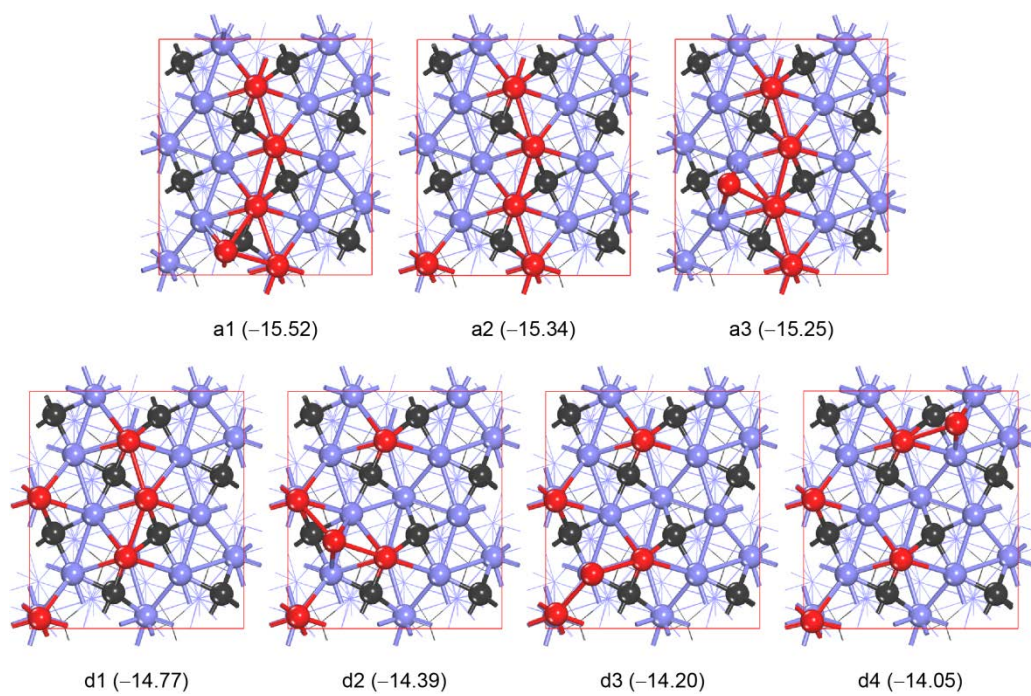
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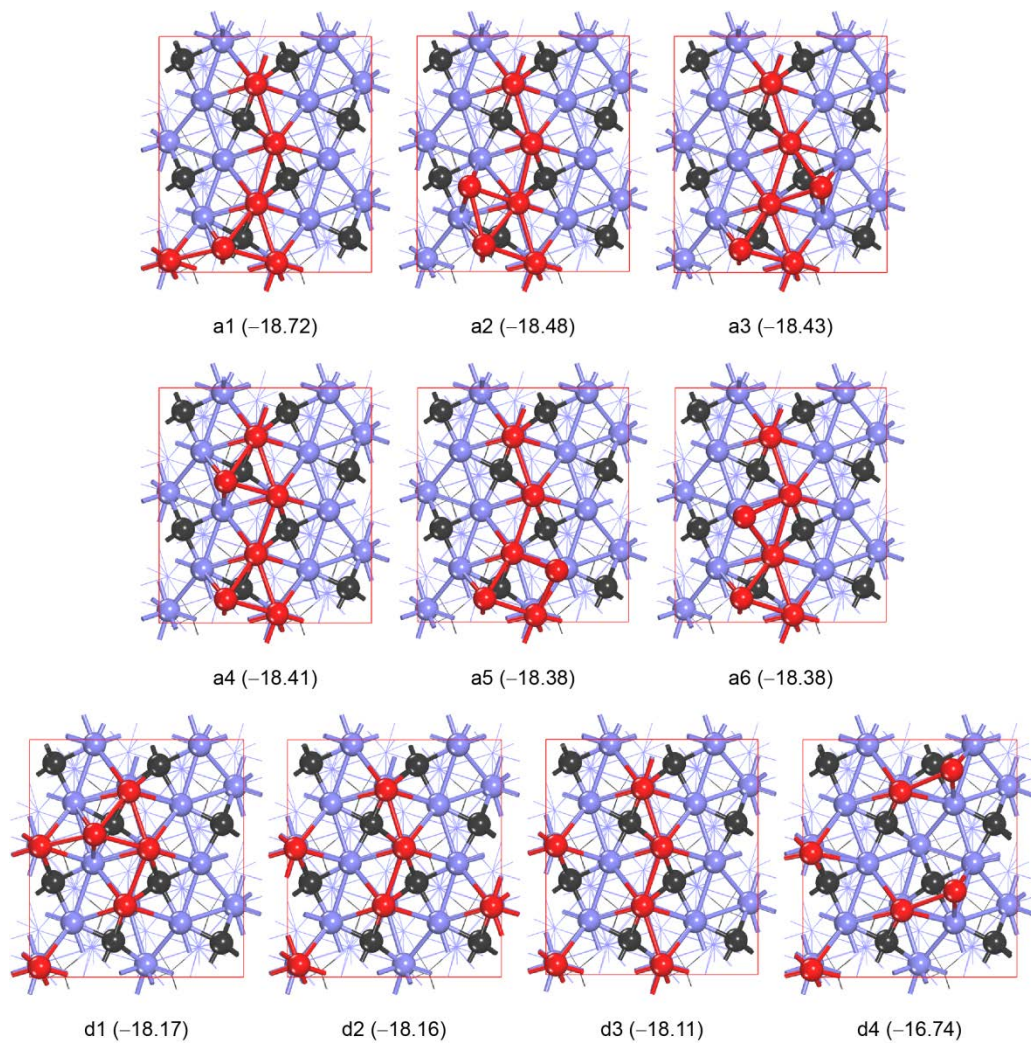
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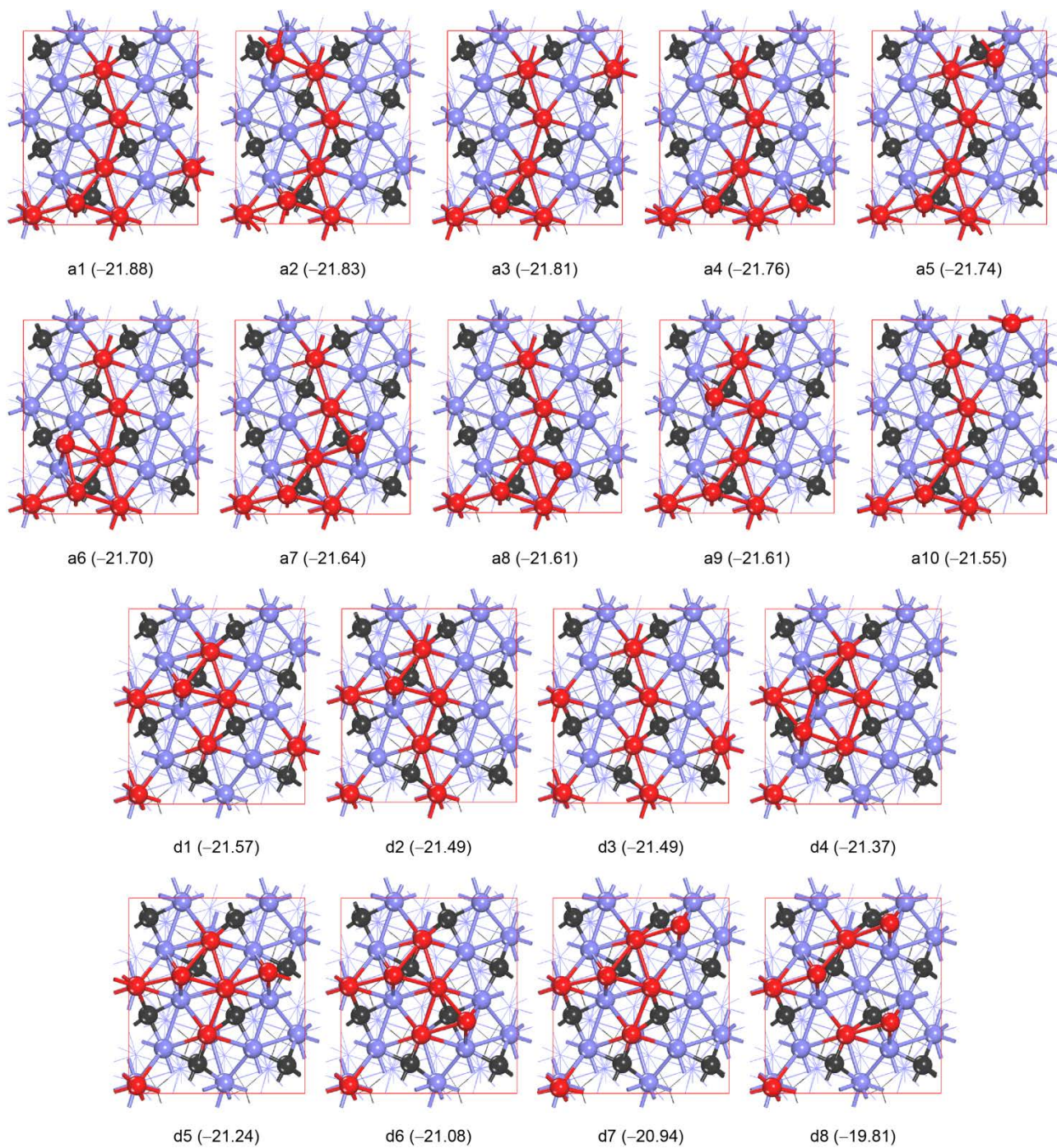


Cu5

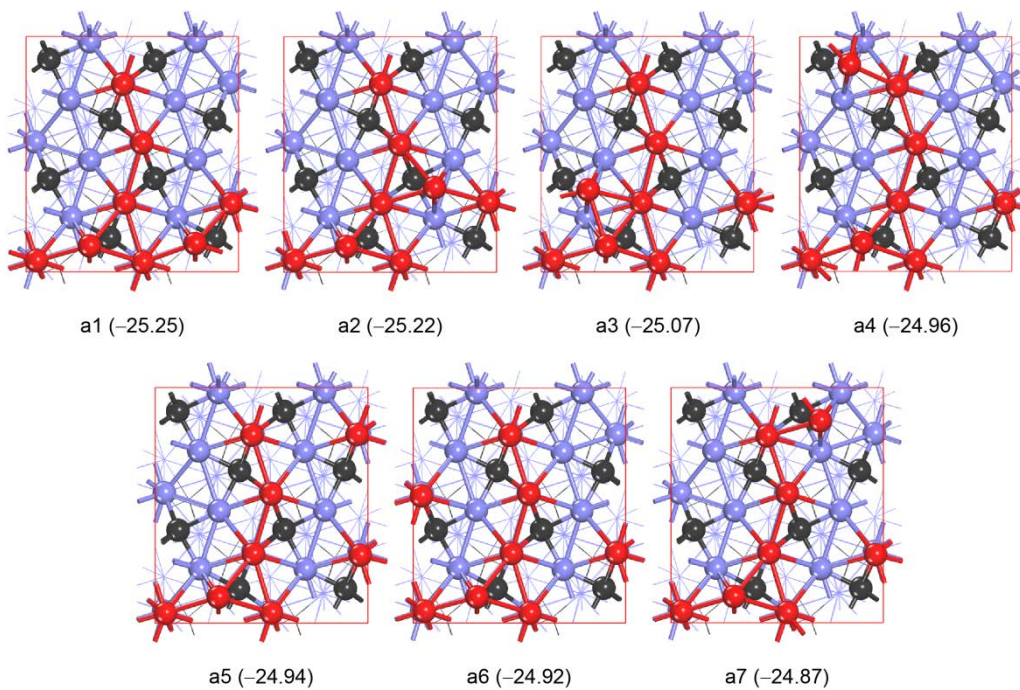


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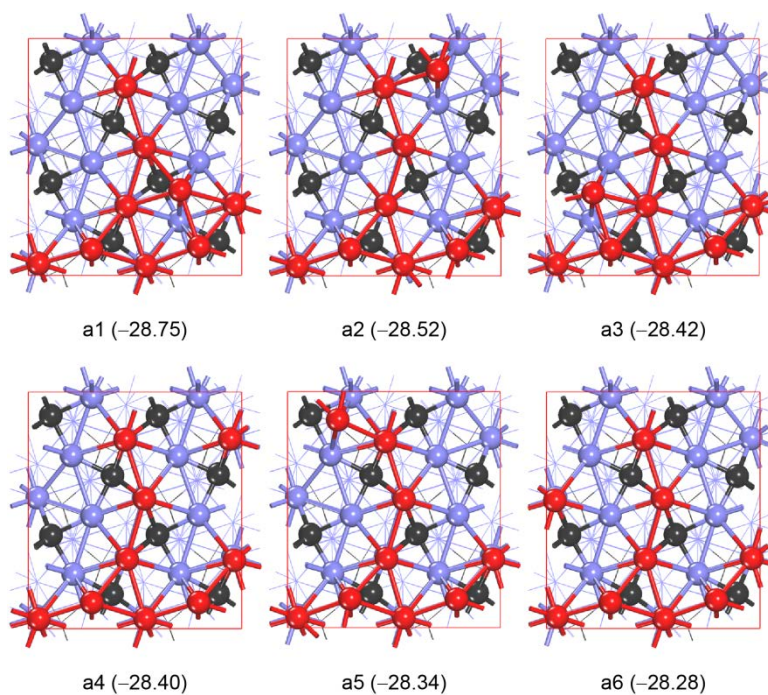




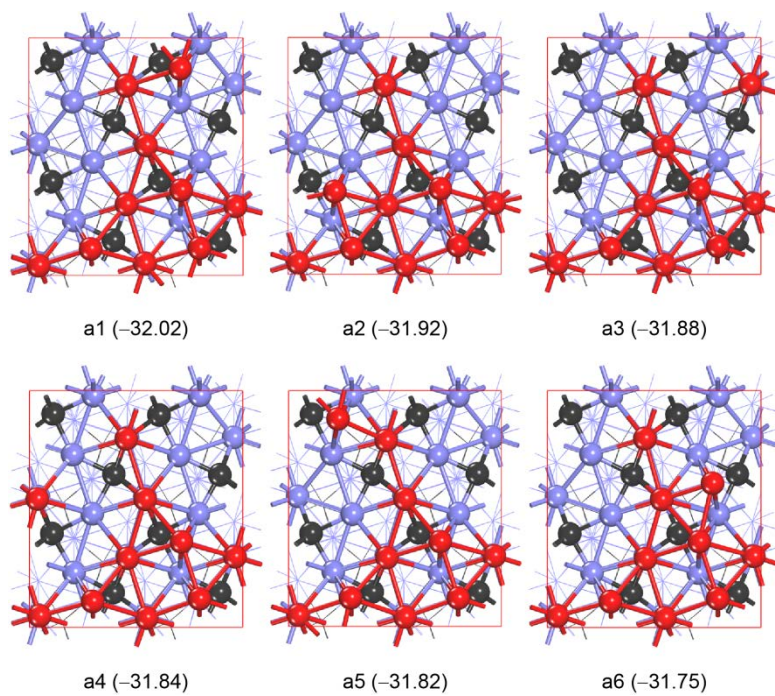
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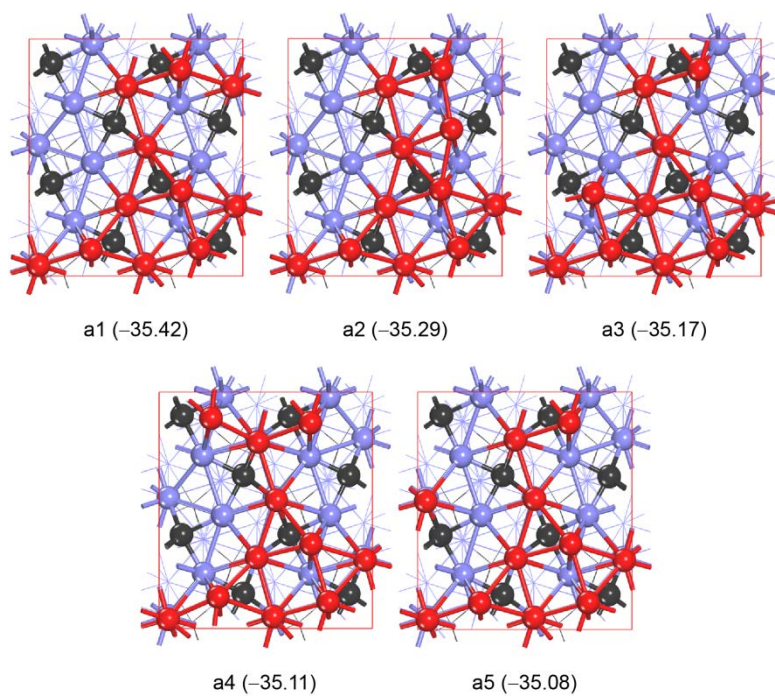
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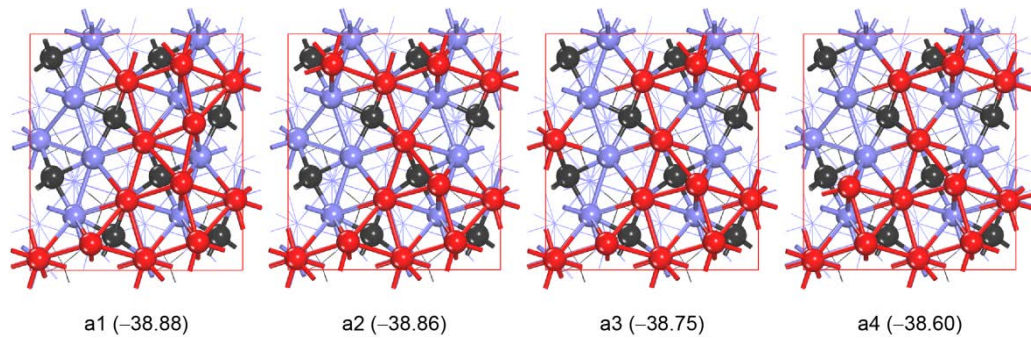
Cu10



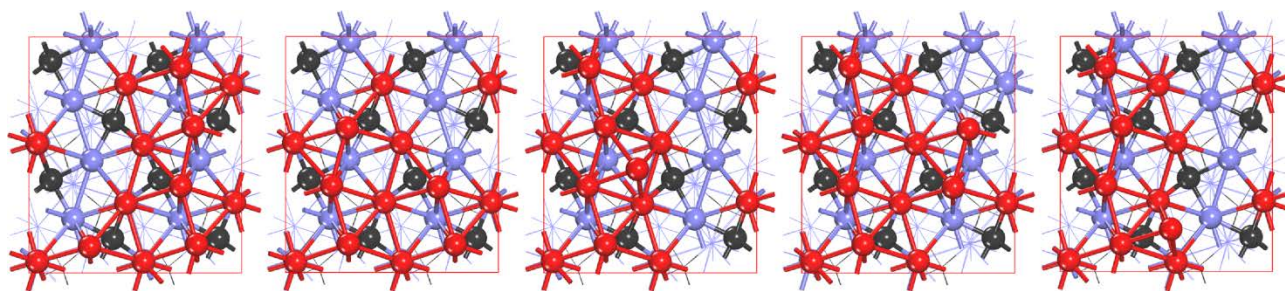
Cu11



Cu12



Cu13



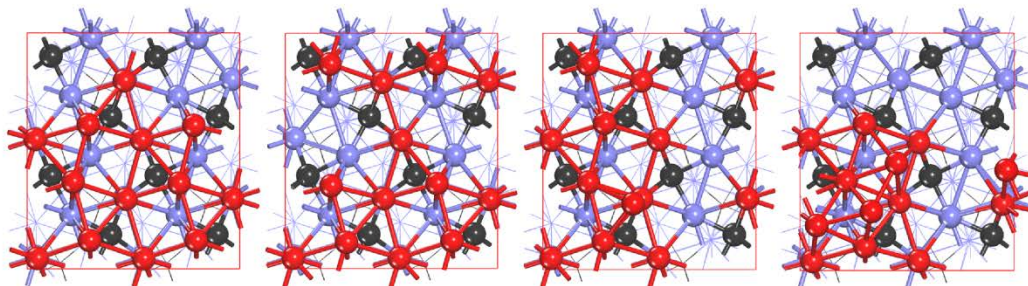
a1 (-42.55)

a2 (-42.28)

a3 (-42.19)

a4 (-42.09)

a5 (-42.08)



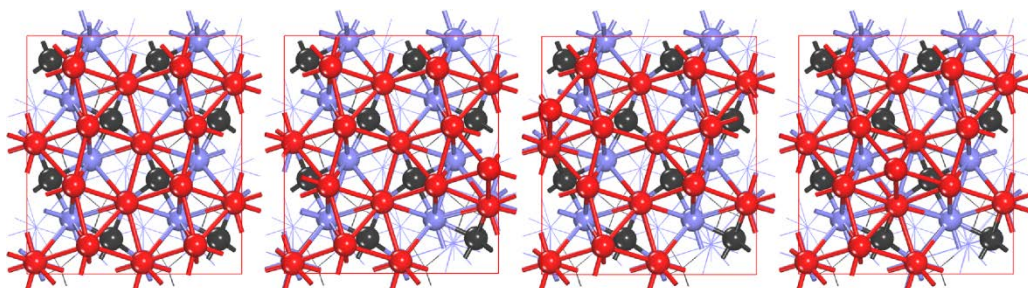
a6 (-42.07)

a7 (-42.06)

a8 (-41.95)

a9 (-40.28)

Cu16

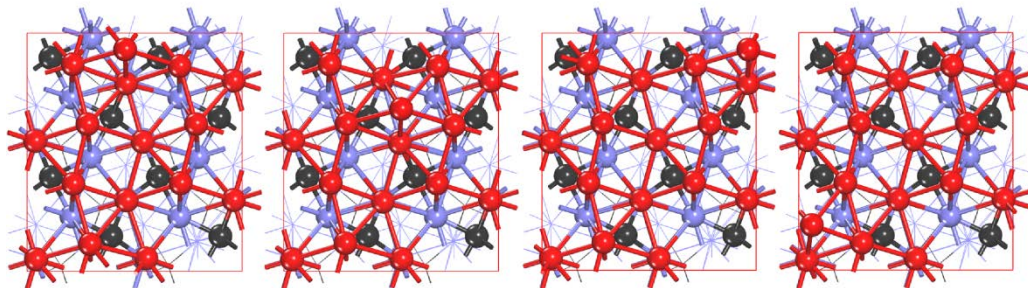


a1 (-53.23)

a2 (-52.75)

a3 (-52.74)

a4 (-52.73)

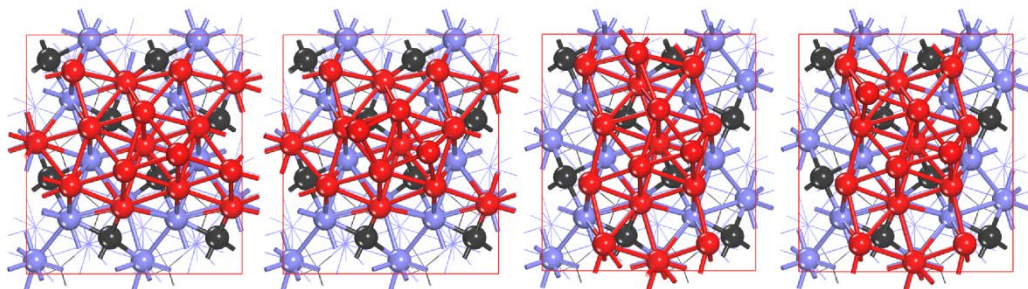


a5 (-52.73)

a6 (-52.71)

a7 (-52.53)

a8 (-52.47)



a9 (-51.71)

a10 (-50.81)

a11 (-50.05)

a12 (-49.37)

Figure S2. Diffusion pathways of single Cu atom on the $\text{Fe}_5\text{C}_2(100)$ surface

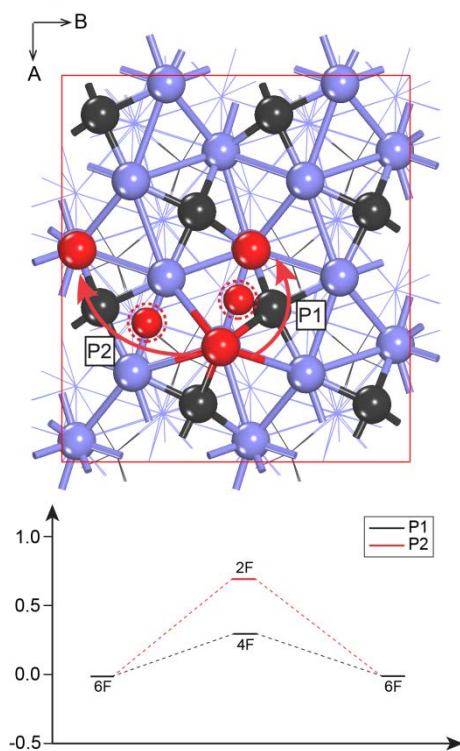
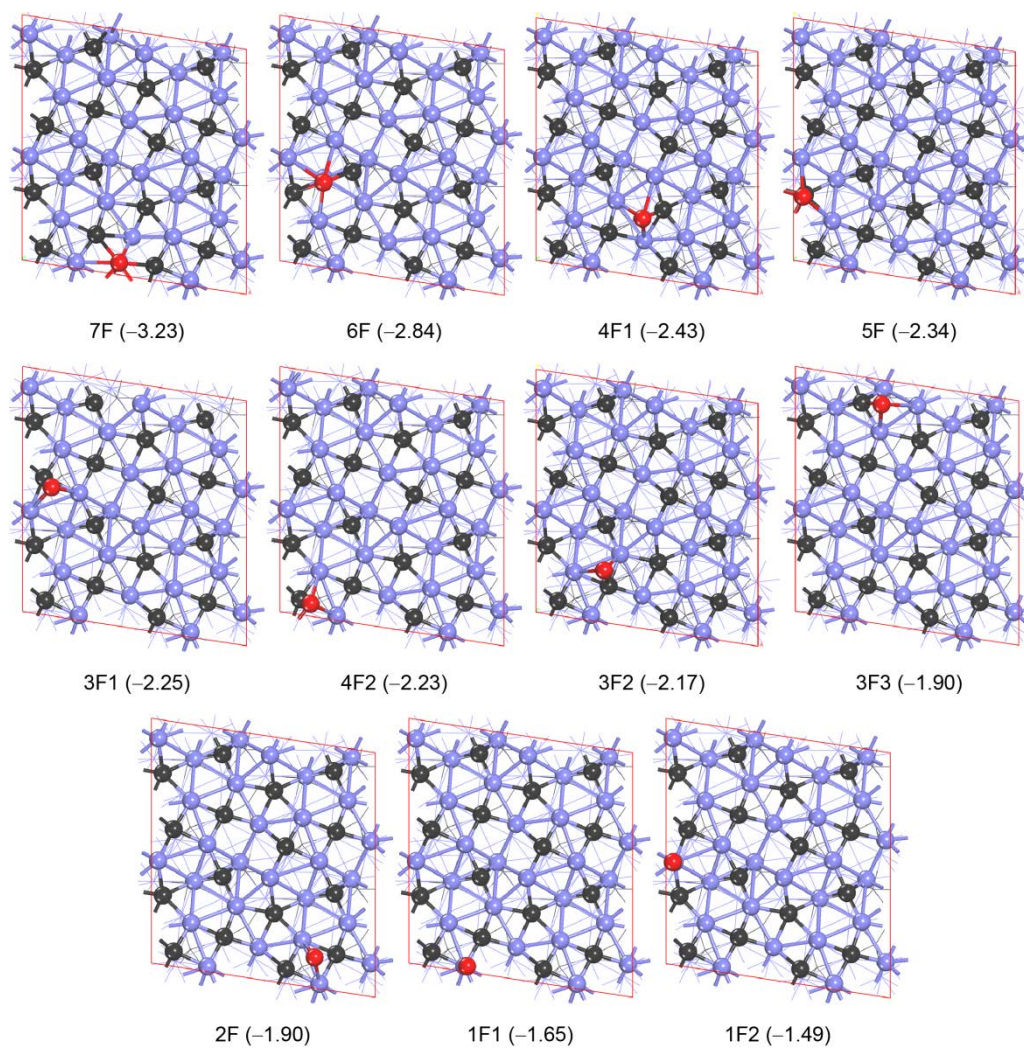
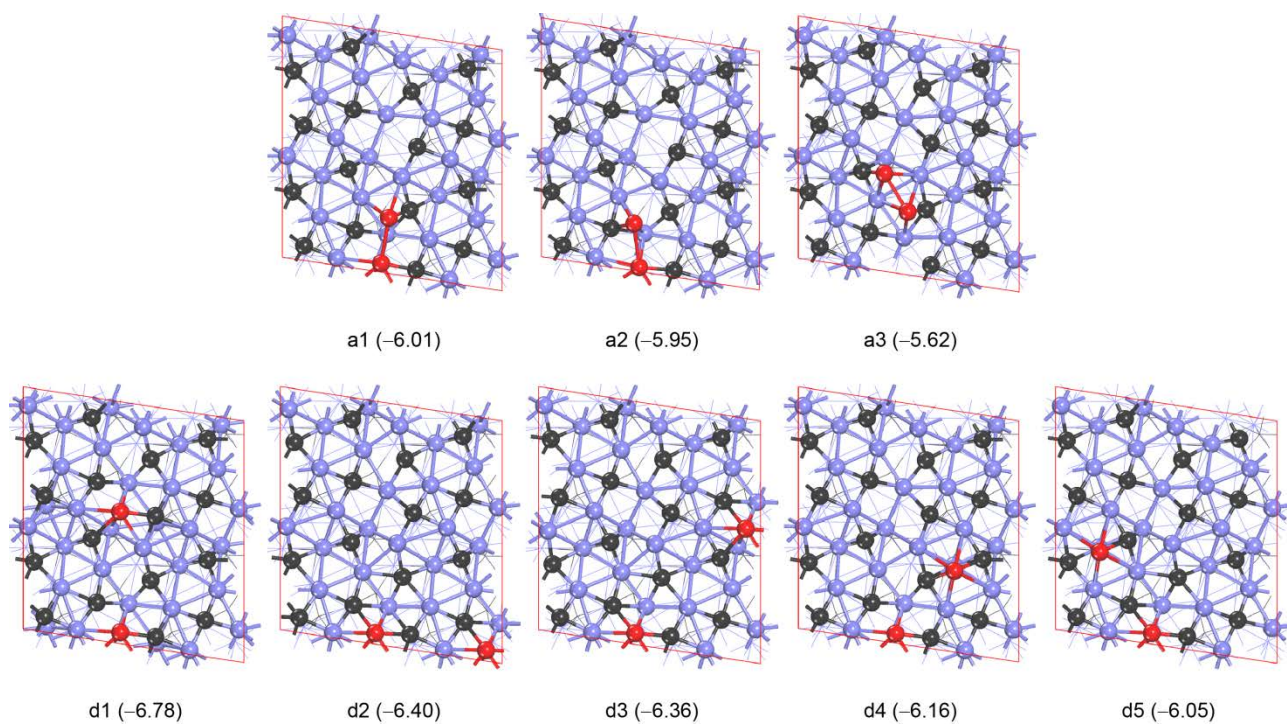


Figure S3. Various optimized Cu_{1-14} , Cu_{16} configurations on the $p(2\times 2)$ $\text{Fe}_5\text{C}_2(111)$ surface (adsorption energy in eV)

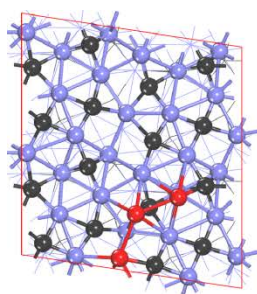
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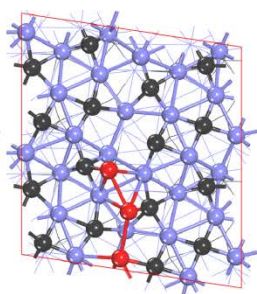
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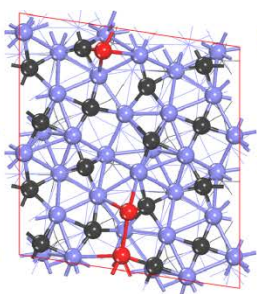
Cu3



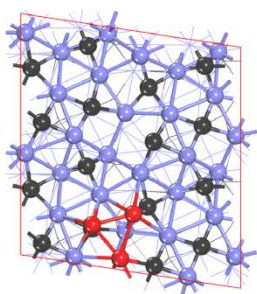
a1 (-9.36)



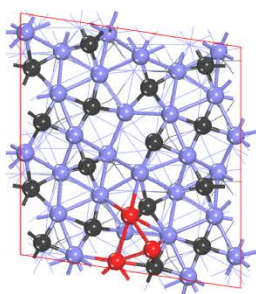
a2 (-9.23)



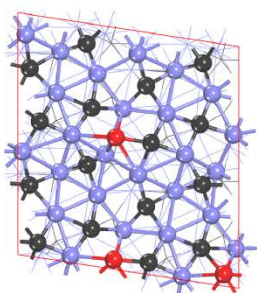
a3 (-8.96)



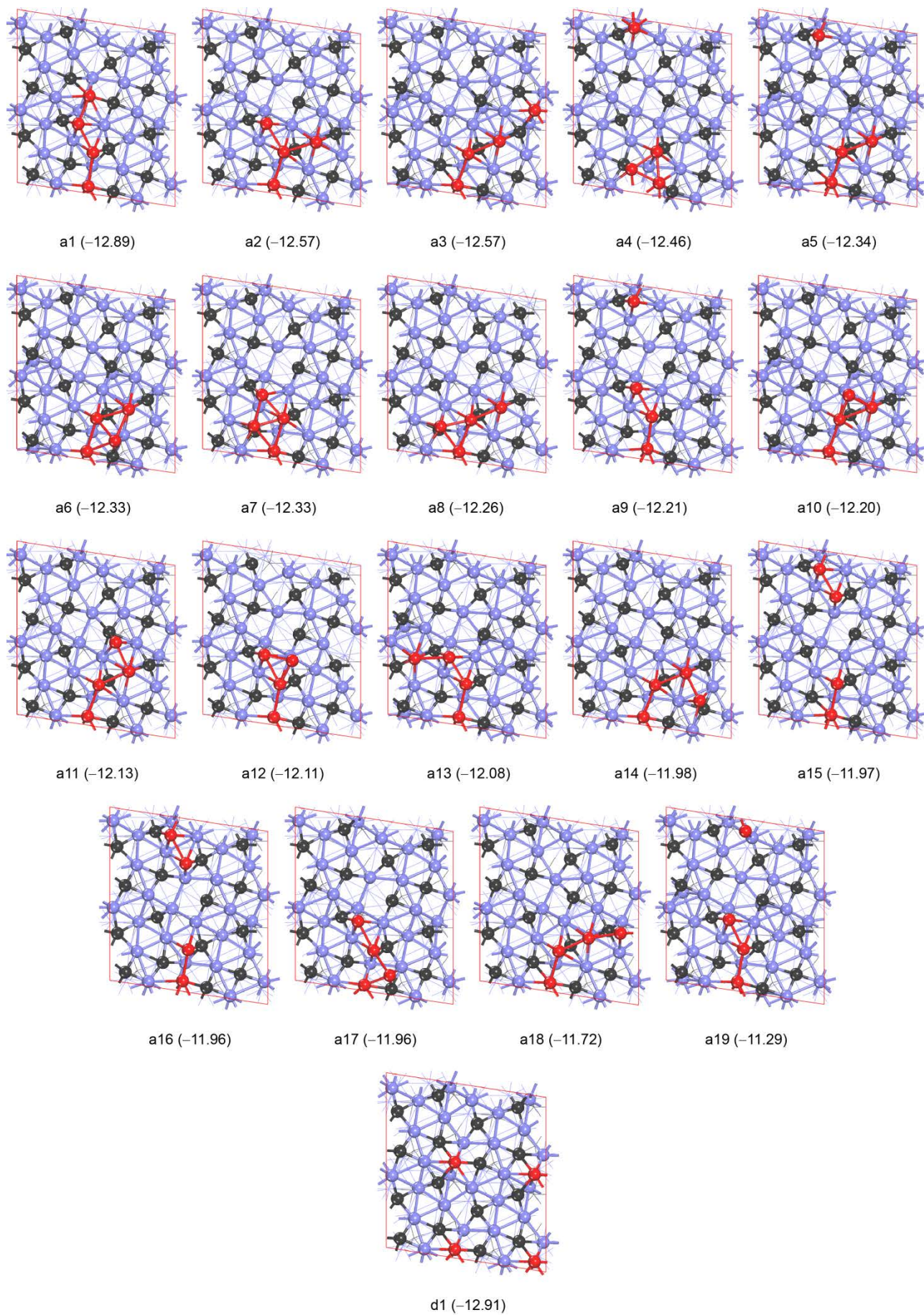
a4 (-8.87)

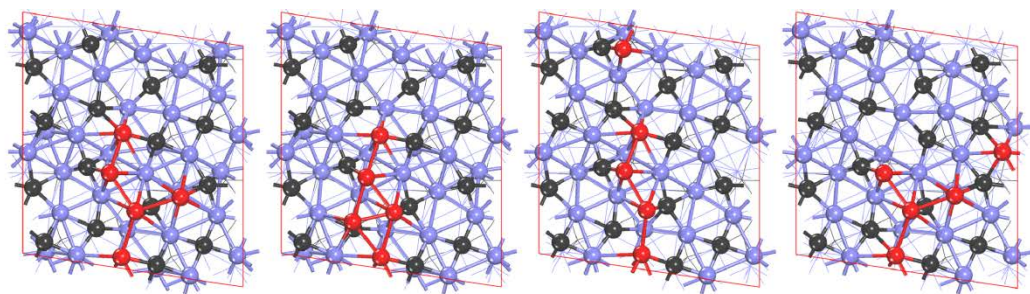


a5 (-8.78)



d1 (-9.77)



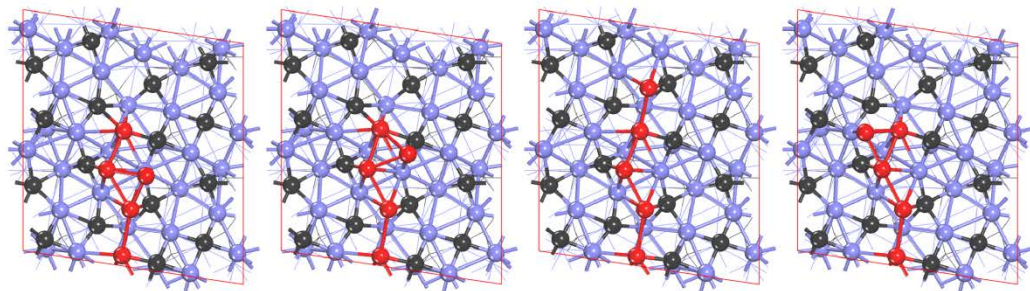


a1 (-16.18)

a2 (-15.95)

a3 (-15.80)

a4 (-15.72)

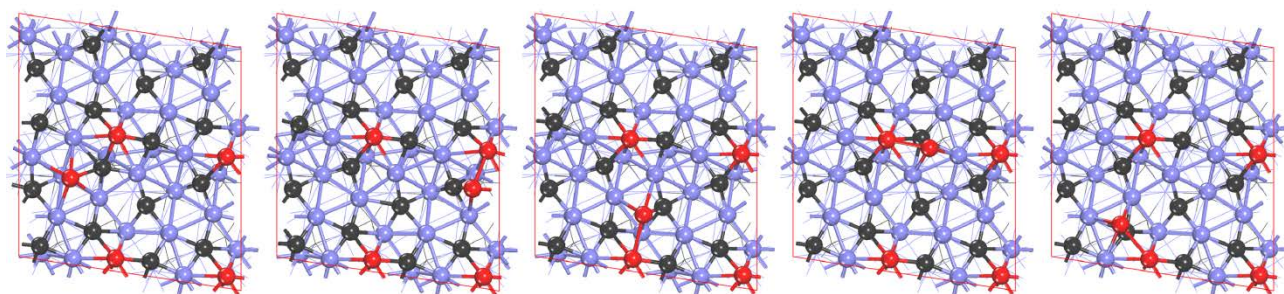


a5 (-15.71)

a6 (-15.71)

a7 (-15.57)

a8 (-15.51)



d1 (-16.09)

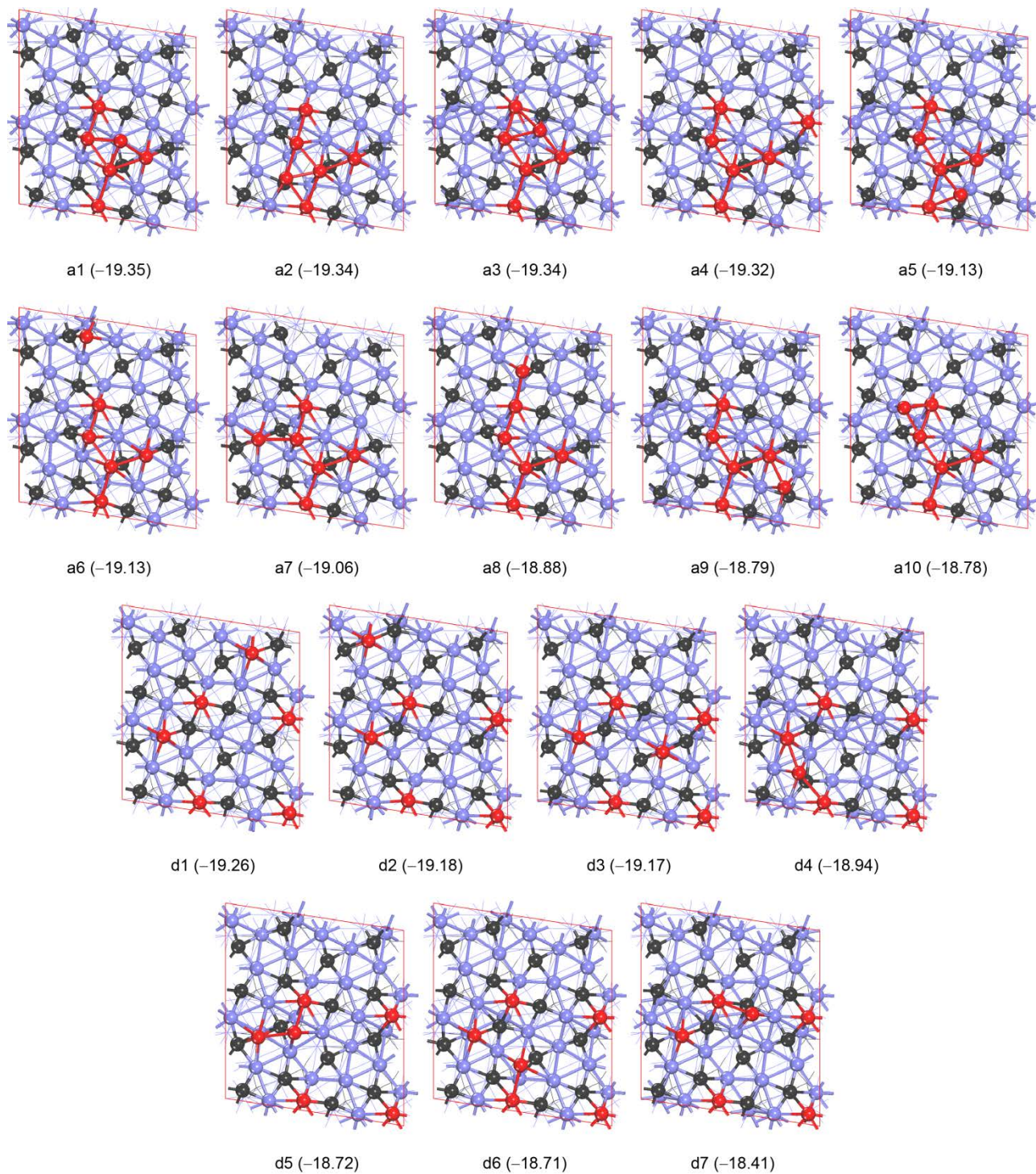
d2 (-15.78)

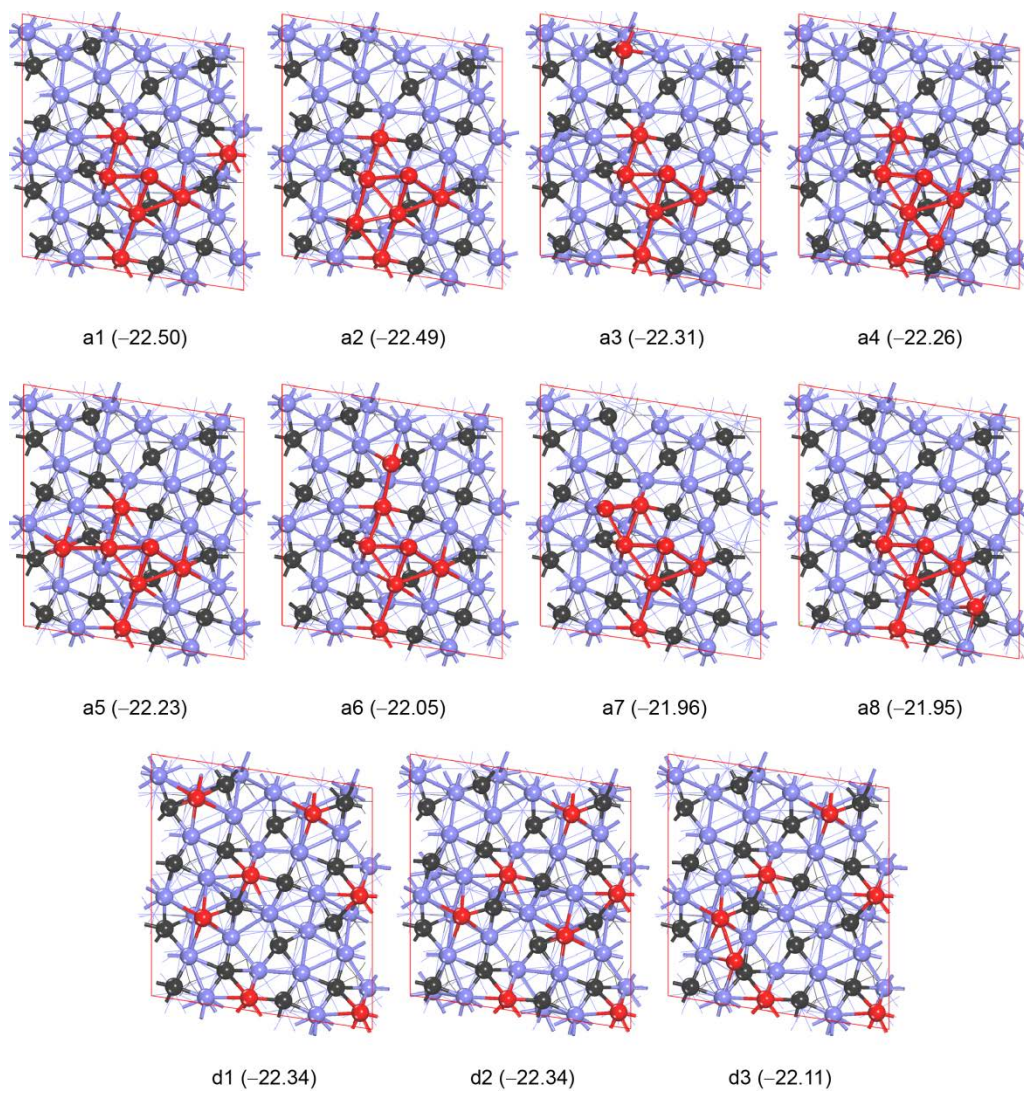
d3 (-15.59)

d4 (-15.36)

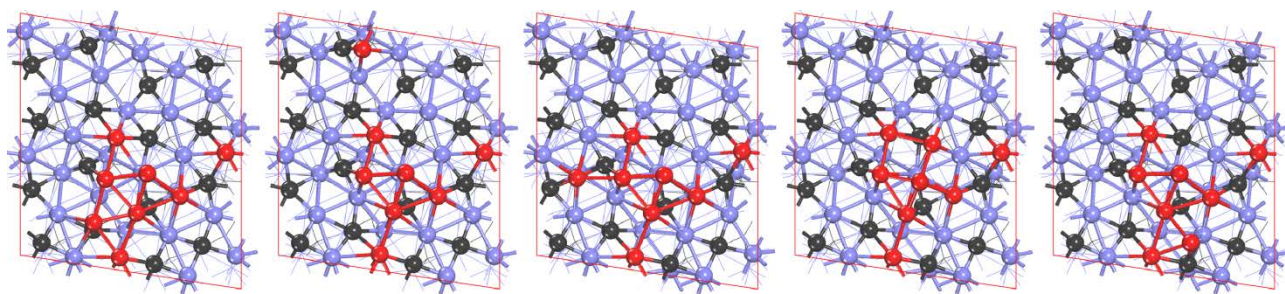
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Cu6





Cu8



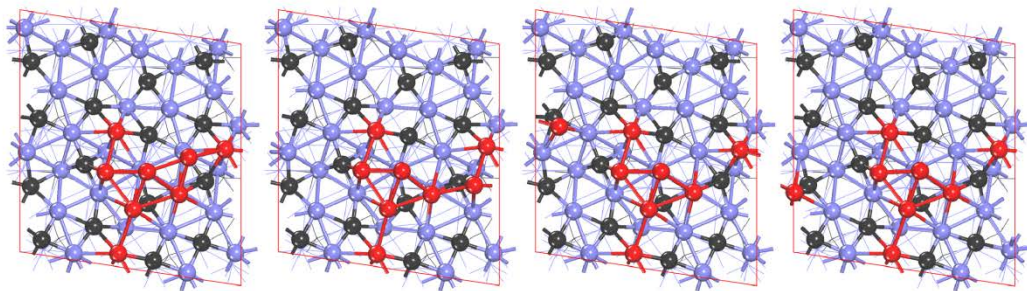
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a2 (-25.47)

a3 (-25.46)

a4 (-25.44)

a5 (-25.42)

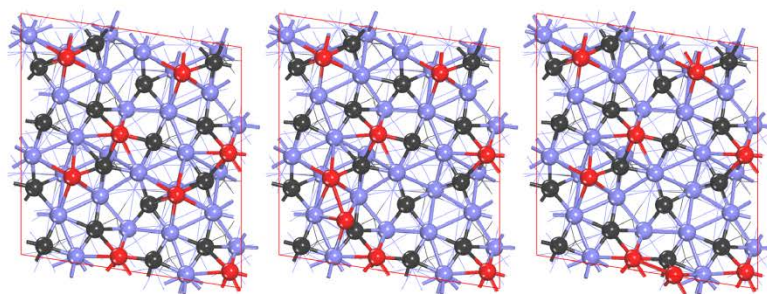


a6 (-25.19)

a7 (-25.17)

a8 (-24.92)

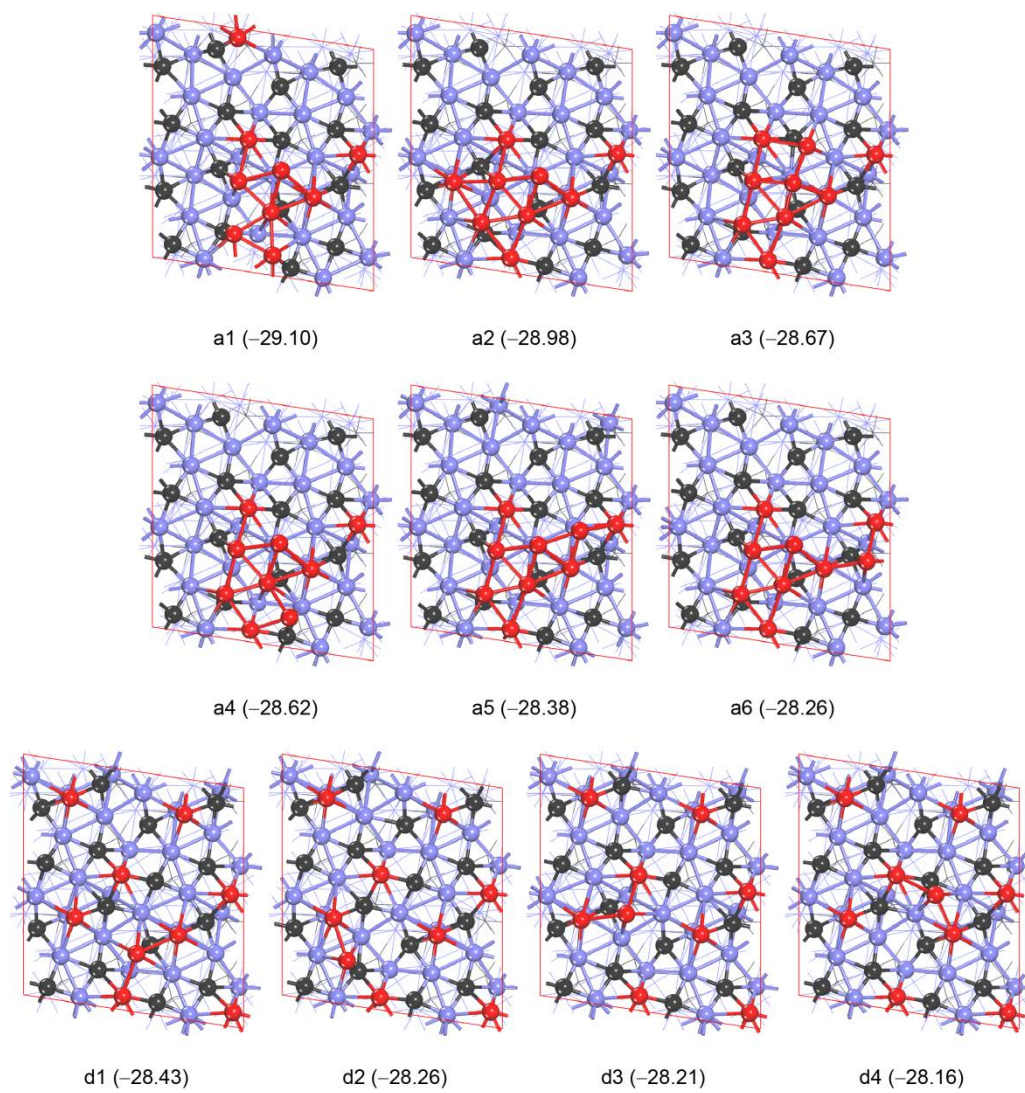
a9 (-24.80)



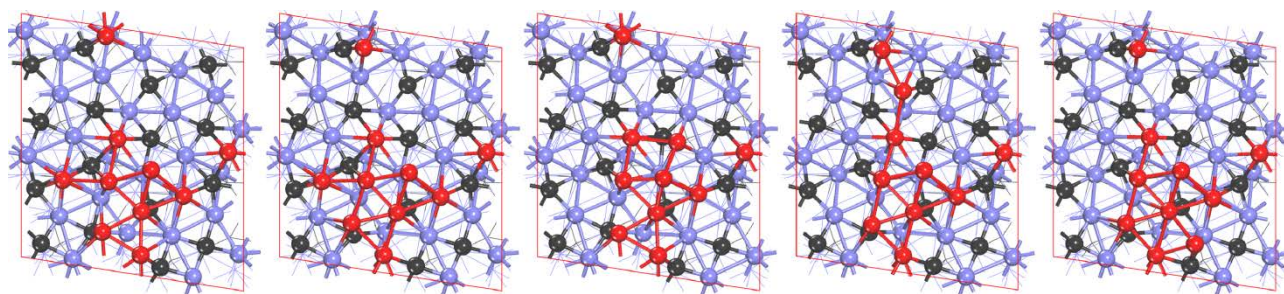
d1 (-25.40)

d2 (-25.19)

d3 (-25.13)



Cu10



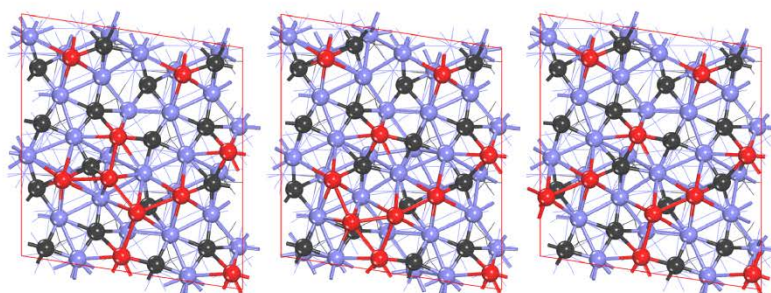
a1 (-32.20)

a2 (-32.05)

a3 (-31.96)

a4 (-31.95)

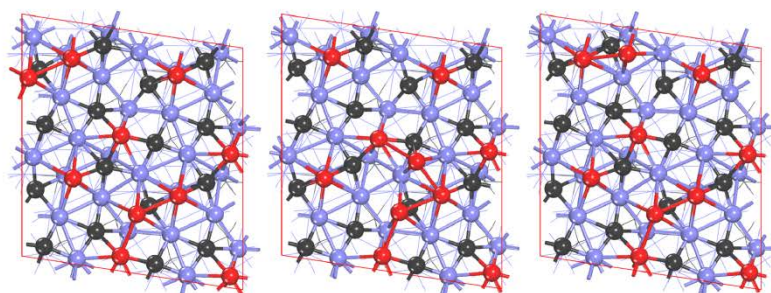
a5 (-31.66)



d1 (-31.77)

d2 (-31.58)

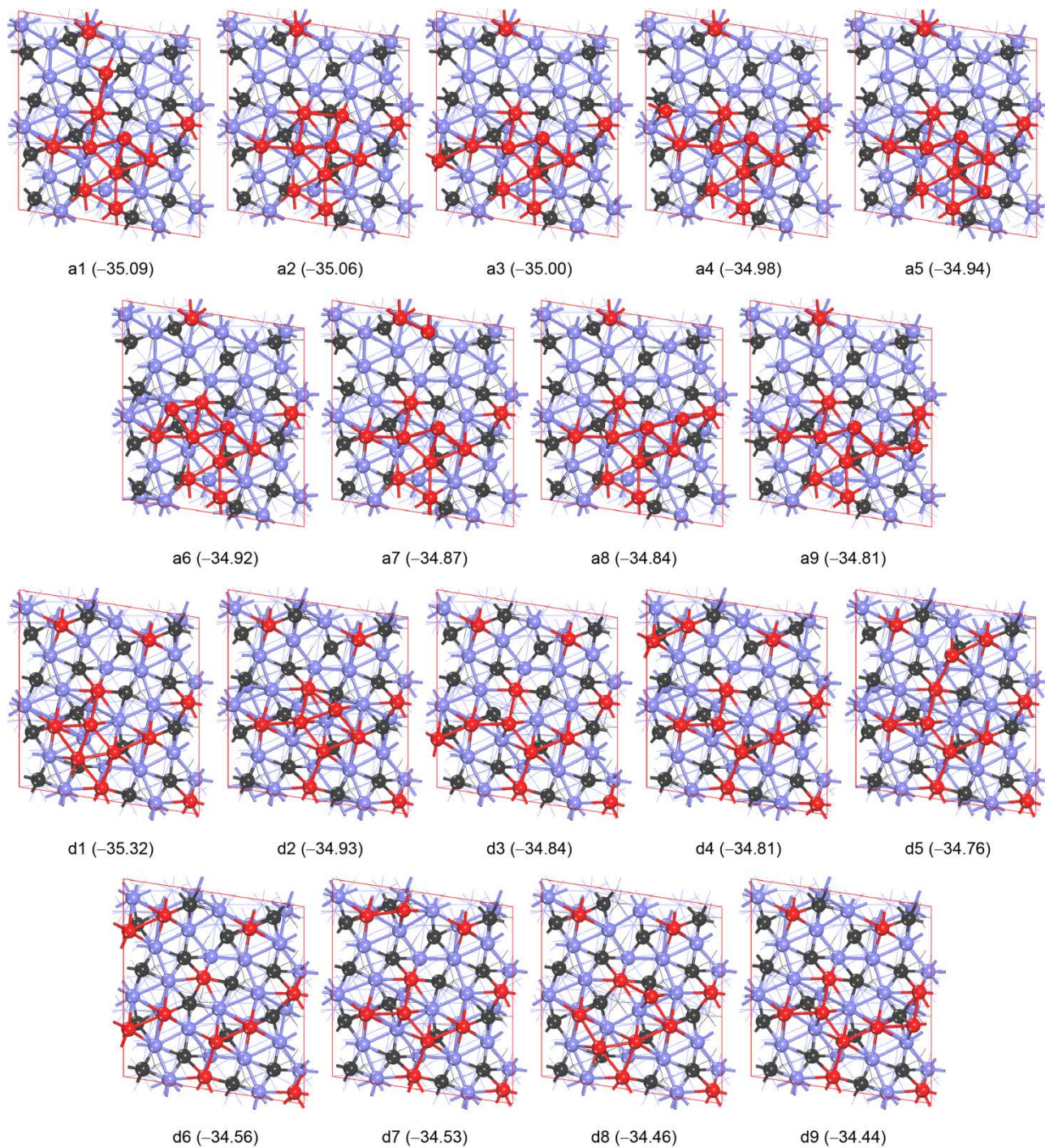
d3 (-31.56)

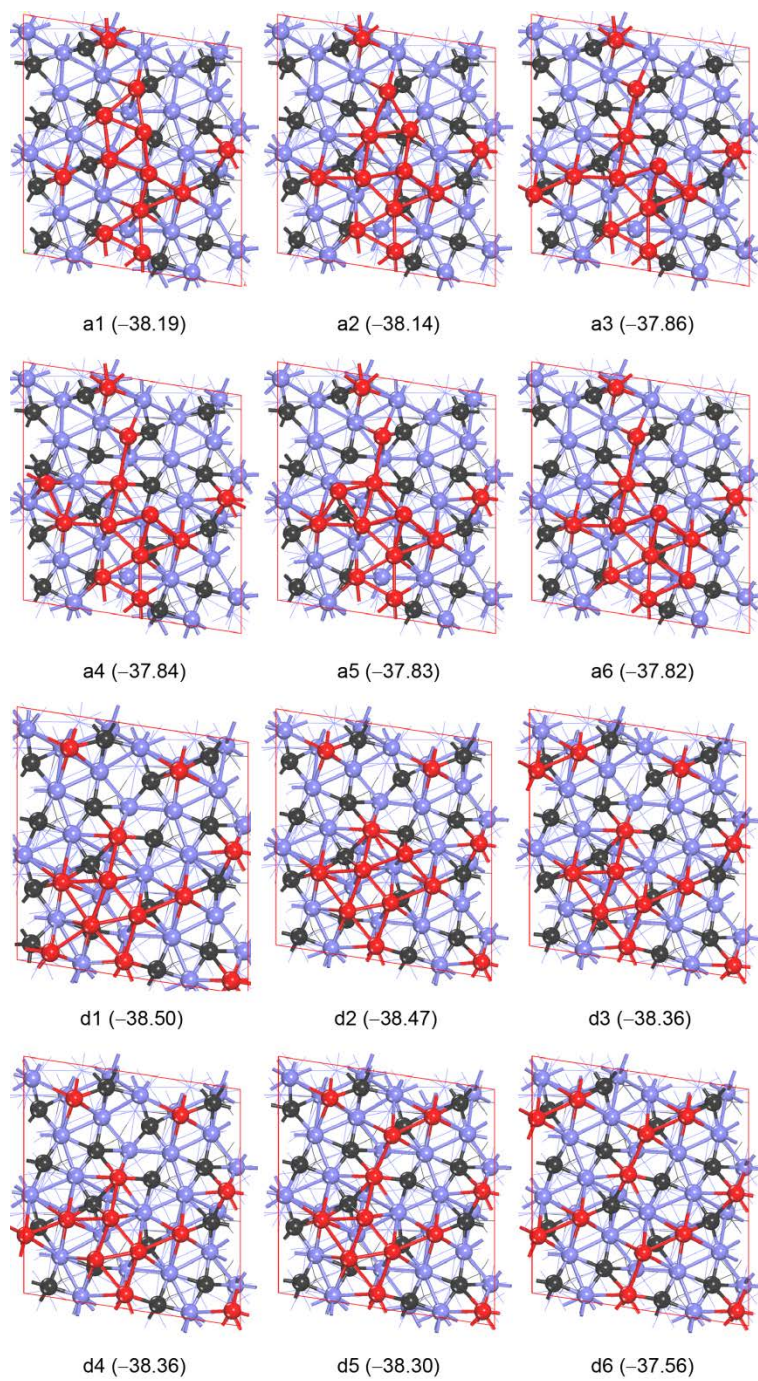


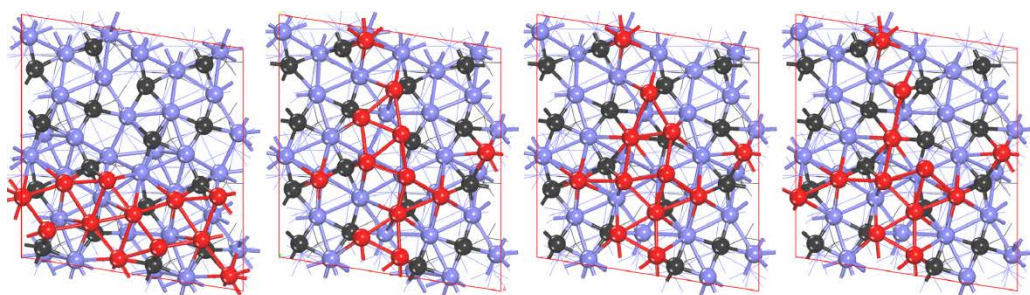
d4 (-31.45)

d5 (-31.41)

d6 (-31.21)





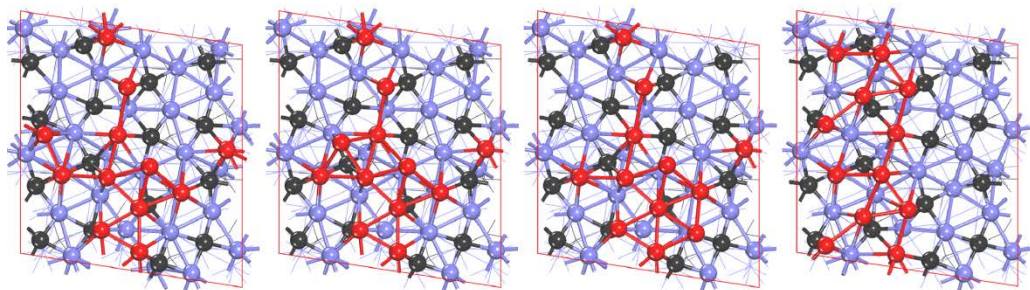


a1 (-38.22)

a2 (-38.19)

a3 (-38.14)

a4 (-37.86)

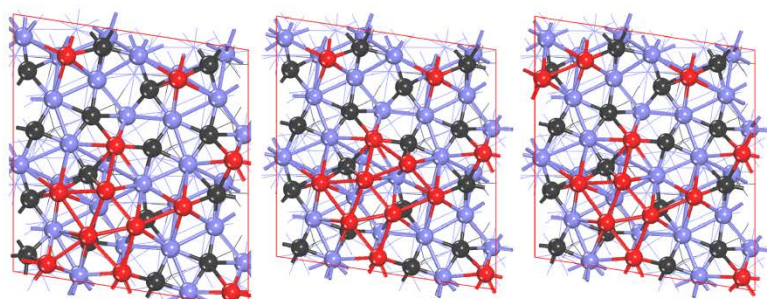


a5 (-37.84)

a6 (-37.83)

a7 (-37.82)

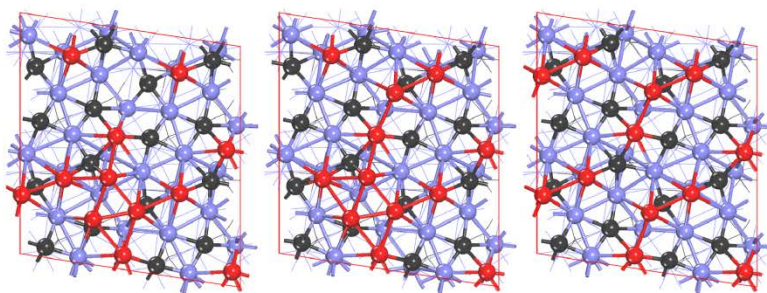
a8 (-37.48)



d1 (-38.50)

d2 (-38.47)

d3 (-38.36)

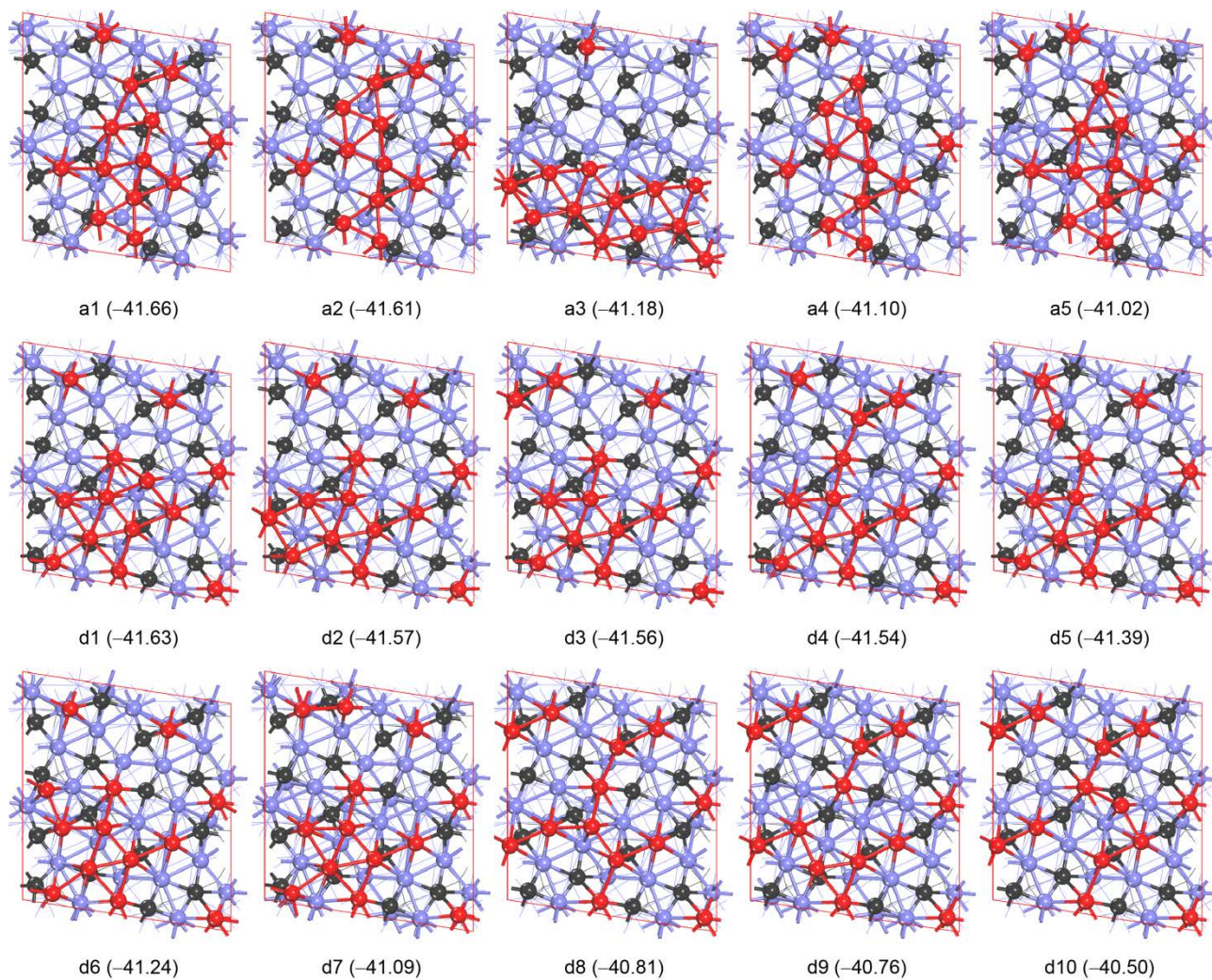


d4 (-38.36)

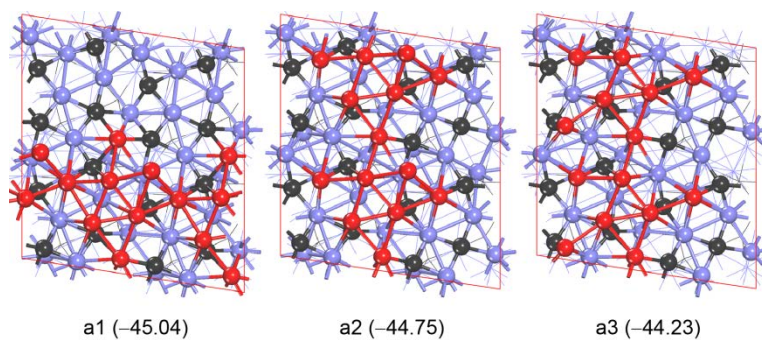
d5 (-38.30)

d6 (-37.56)

Cu13



Cu14



Cu16-*p*(2×1)-1ML

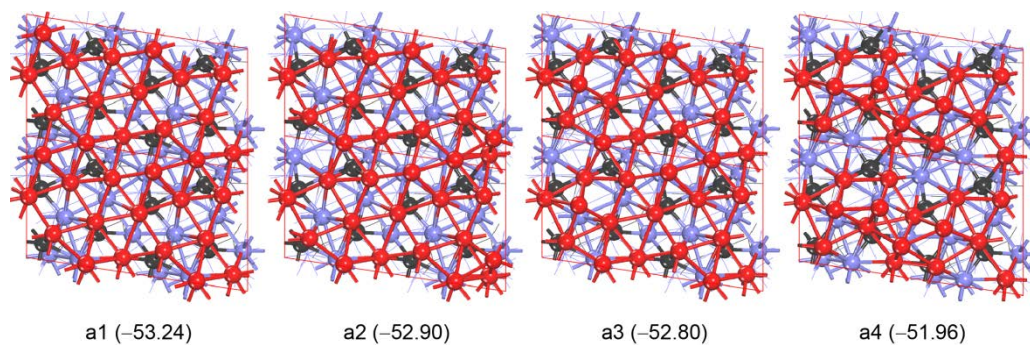


Figure S4. Diffusion pathways of single Cu atom on the $\text{Fe}_5\text{C}_2(111)$ surface

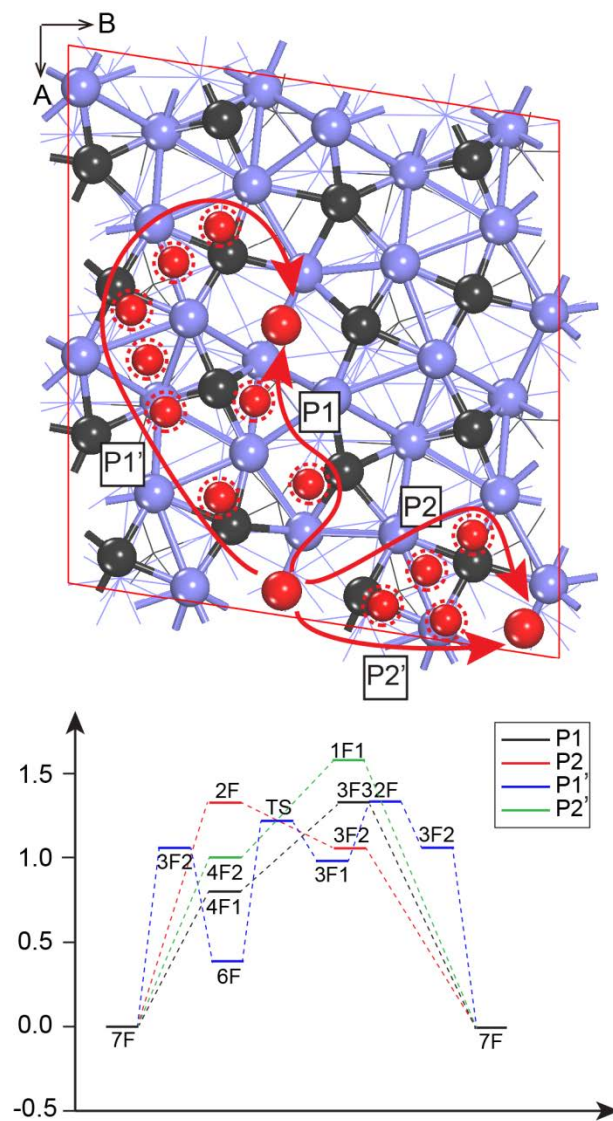
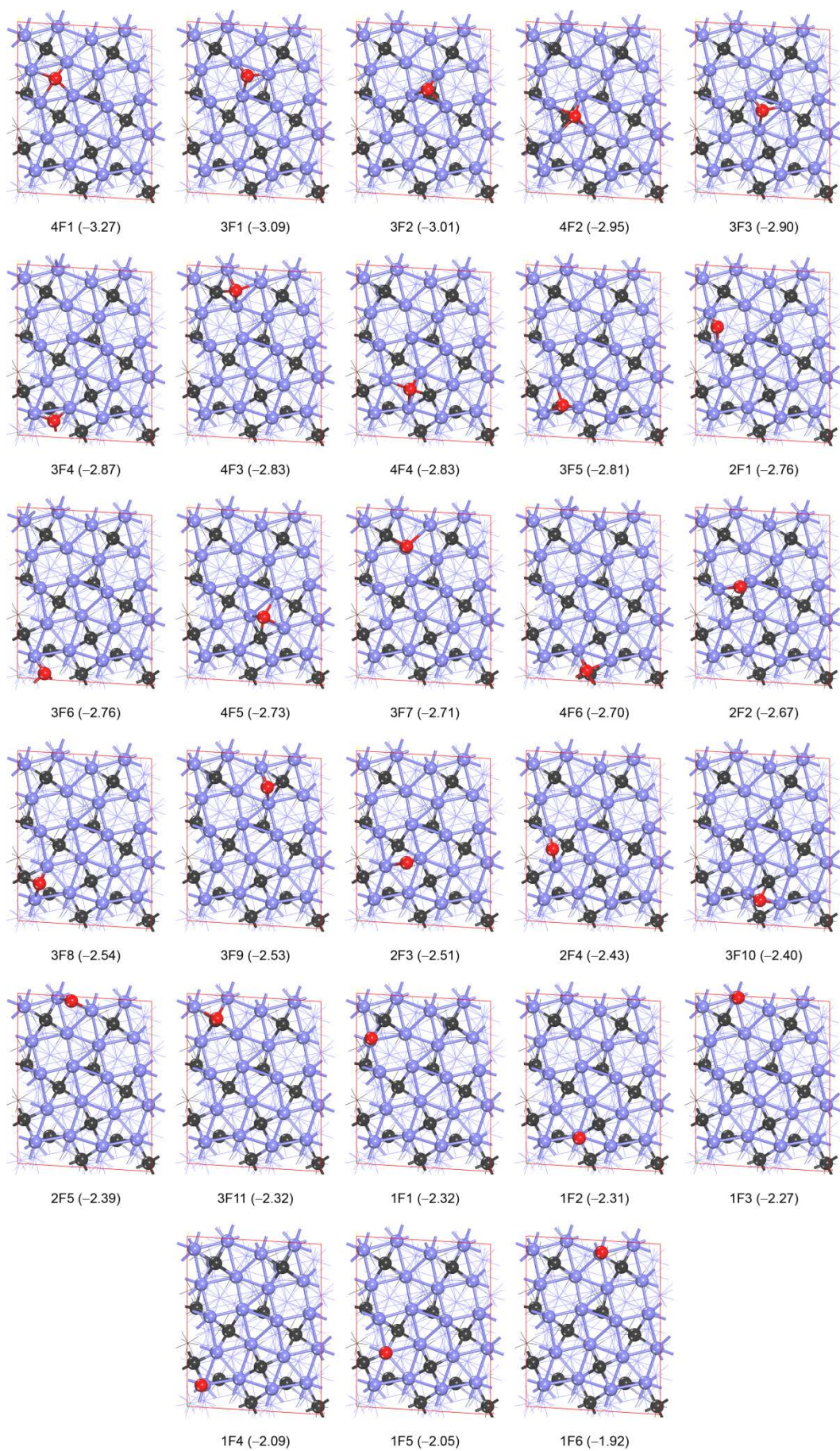
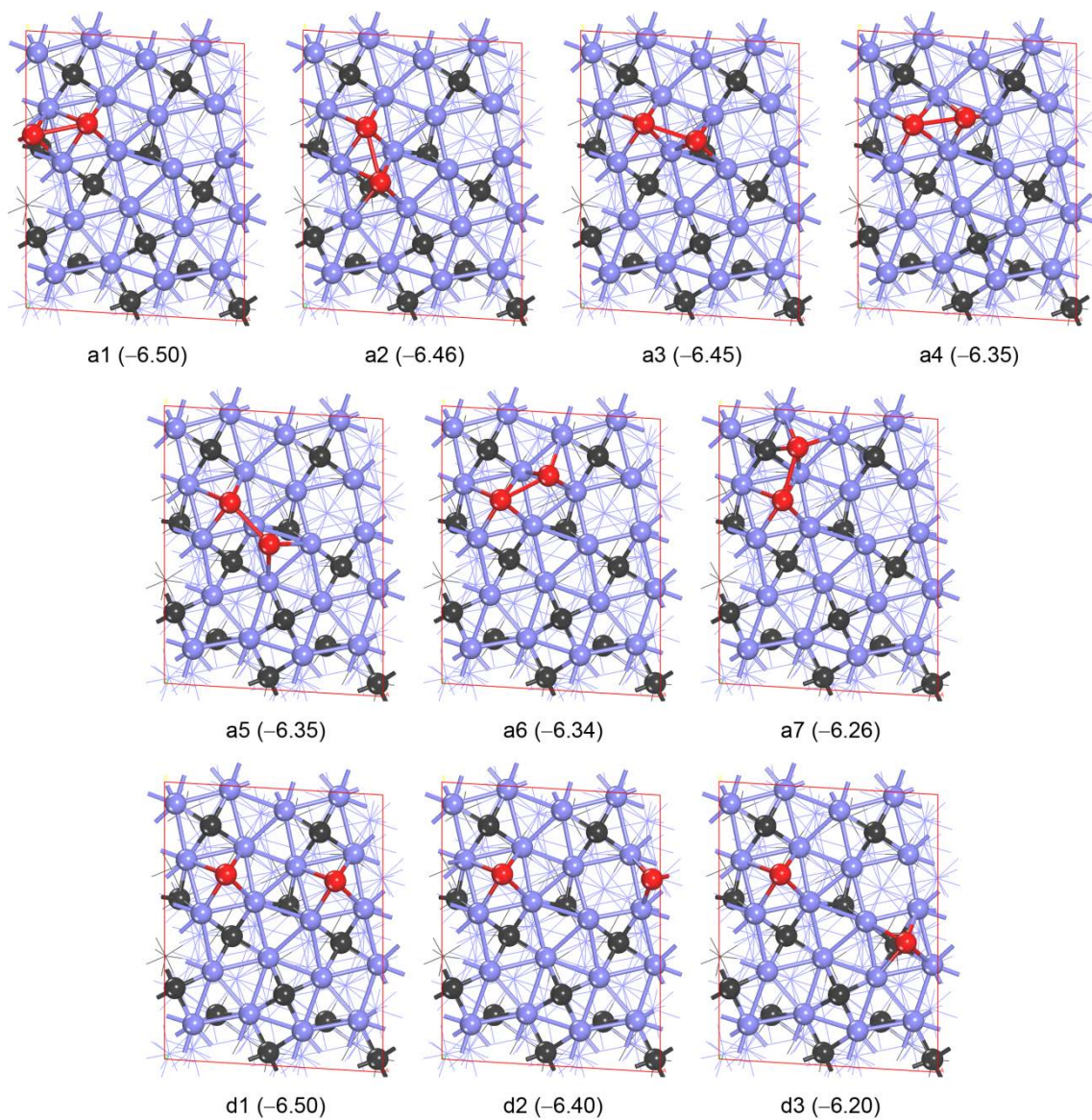
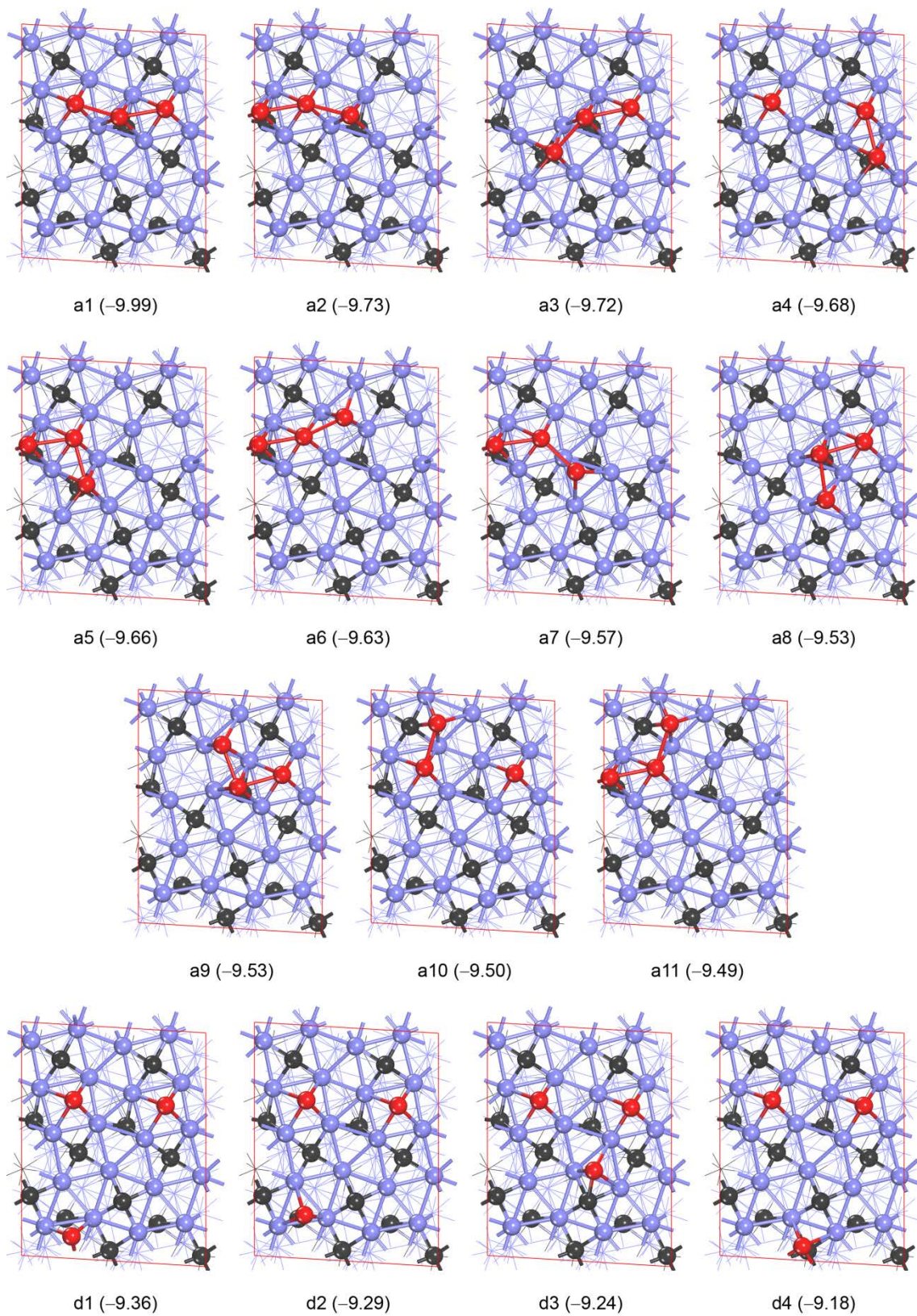


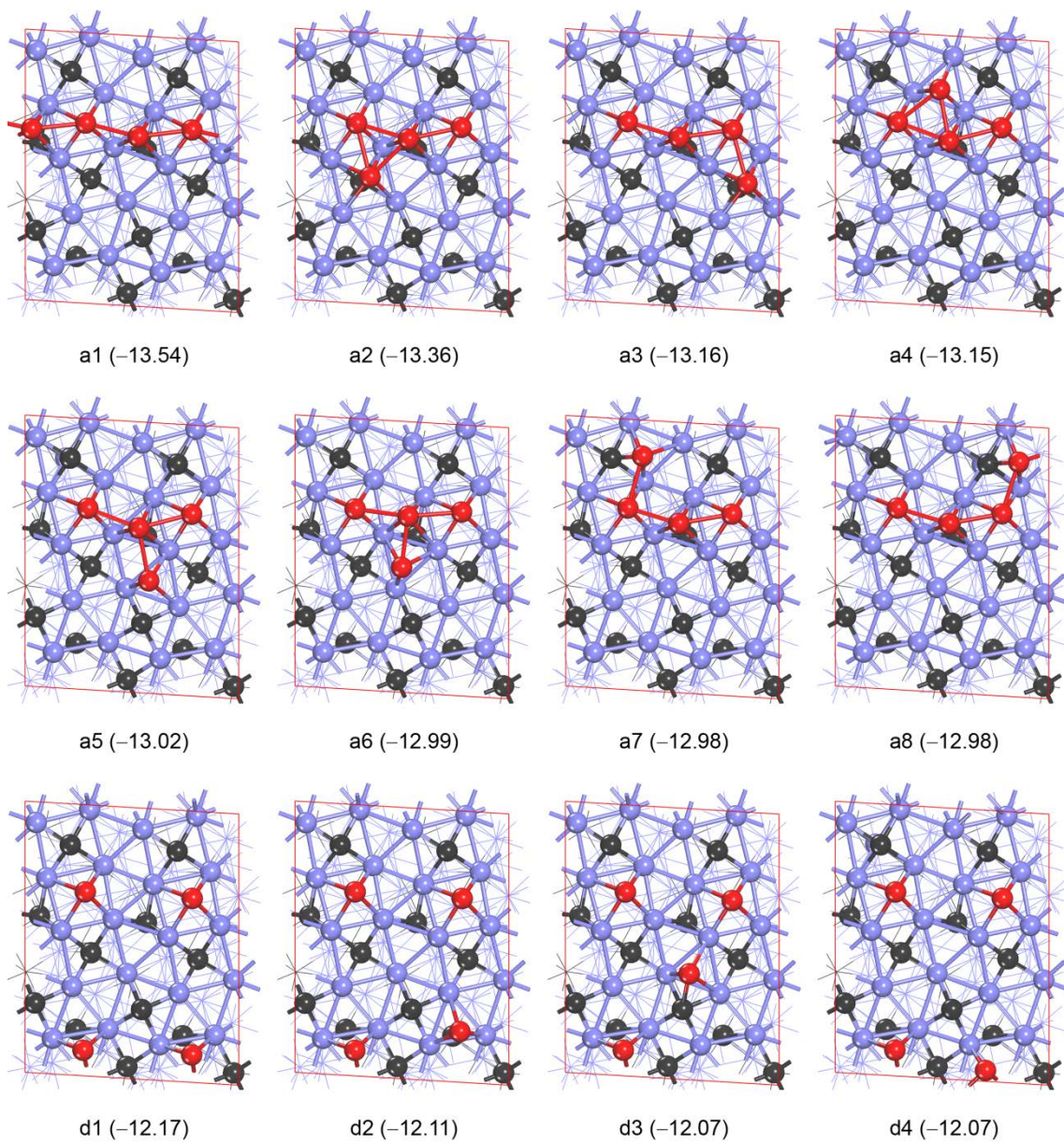
Figure S5. Various optimized Cu₁₋₁₇, Cu₂₀ configurations on the *p*(2×1) Fe₅C₂(510) surface (adsorption energy in eV)

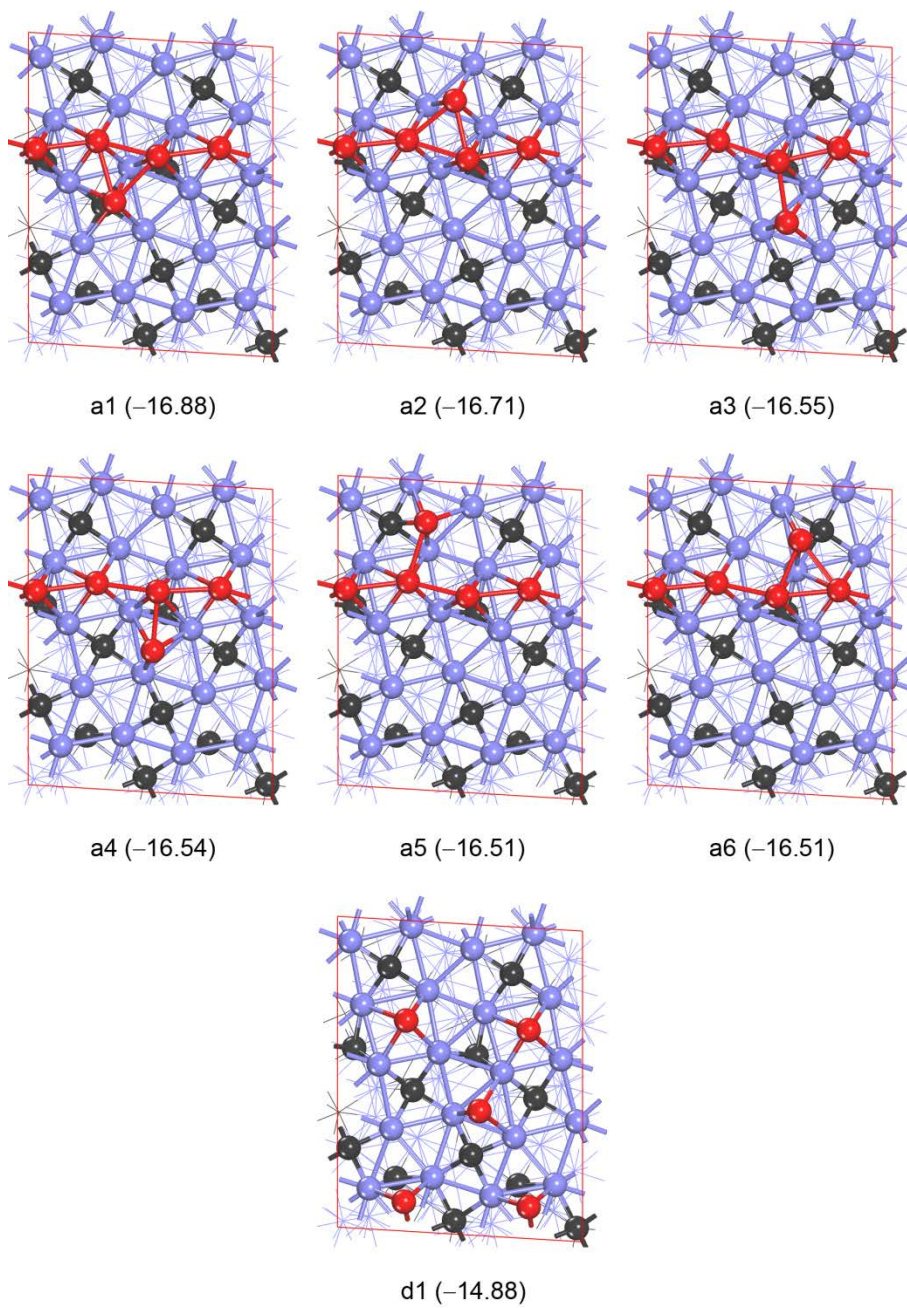
Cu1

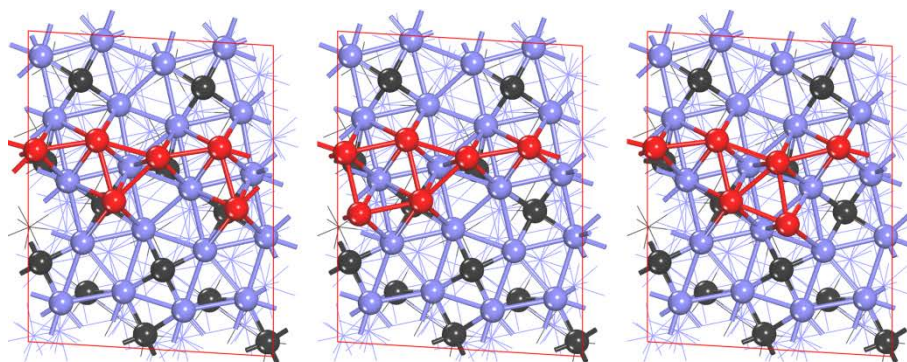








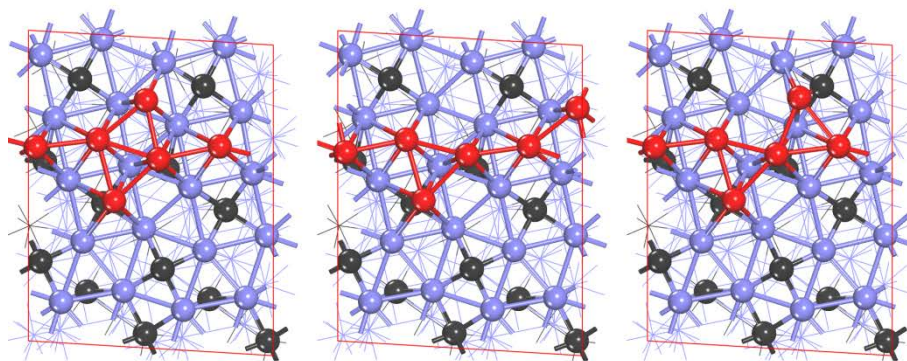




a1 (-20.25)

a2 (-20.15)

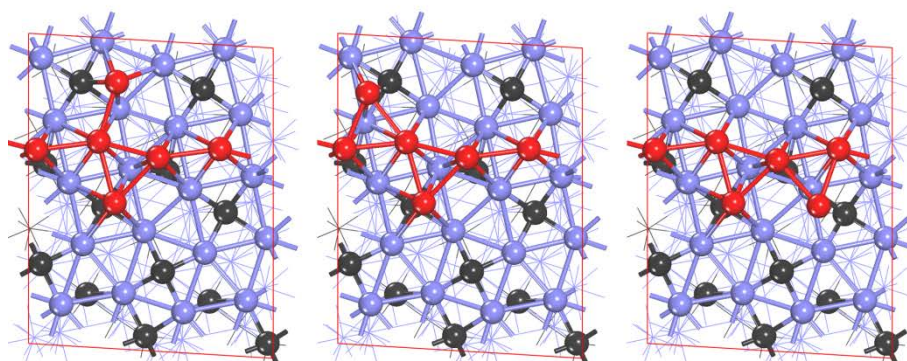
a3 (-20.14)



a4 (-20.10)

a5 (-20.07)

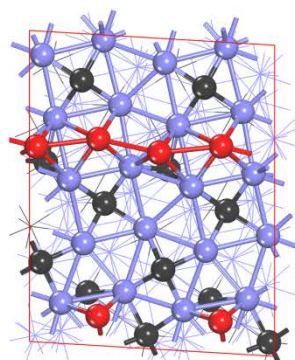
a6 (-19.90)



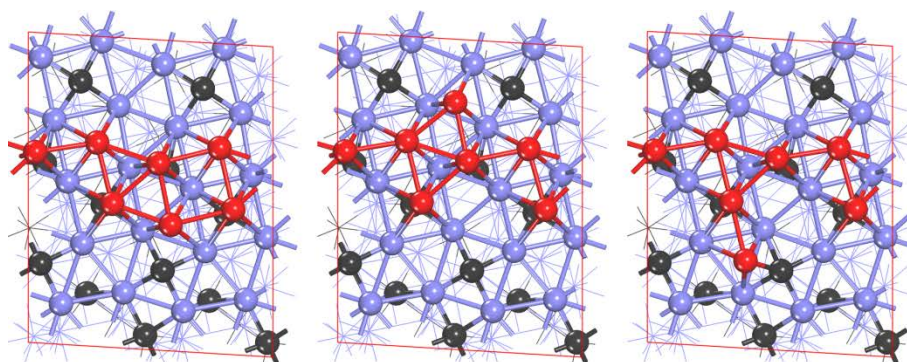
a7 (-19.89)

a8 (-19.87)

a9 (-19.81)



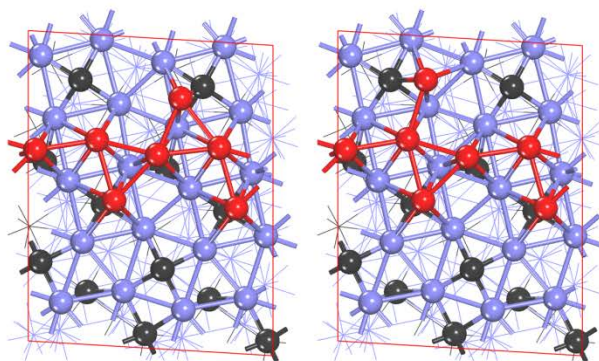
d1 (-19.19)



a1 (-23.78)

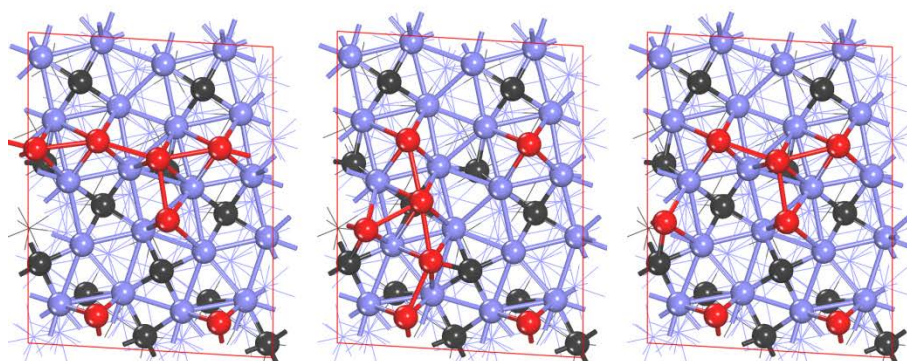
a2 (-23.47)

a3 (-23.43)



a4 (-23.27)

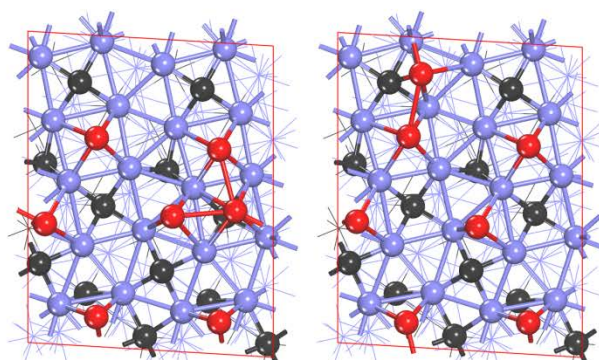
a5 (-23.24)



d1 (-22.19)

d2 (-21.74)

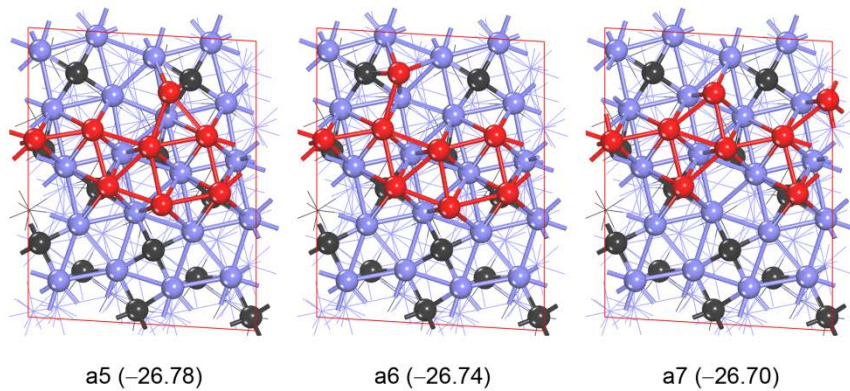
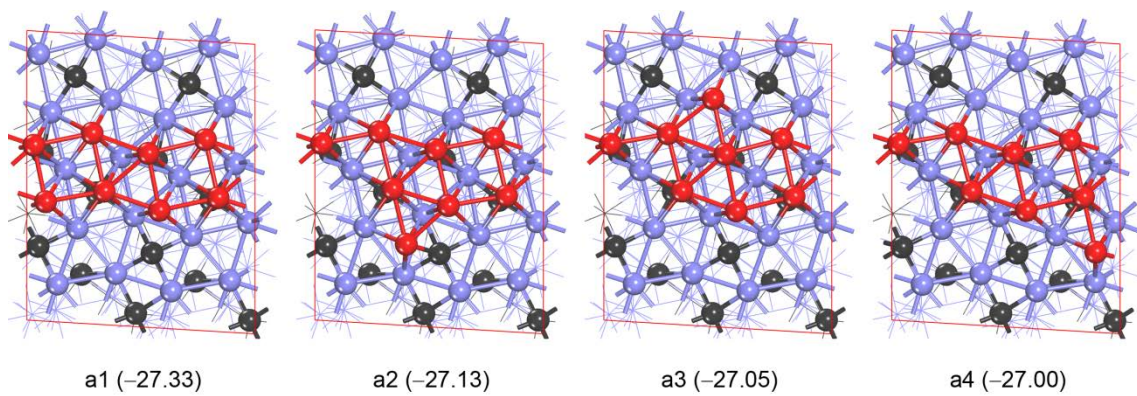
d3 (-21.39)



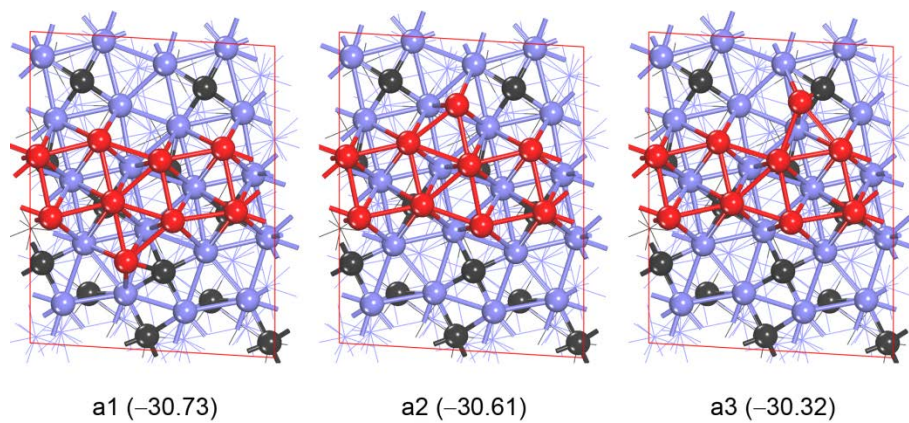
d4 (-21.38)

d5 (-20.85)

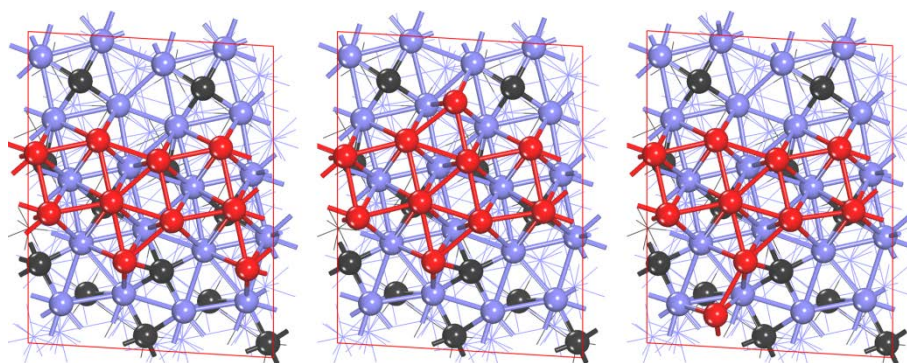
Cu8



Cu9



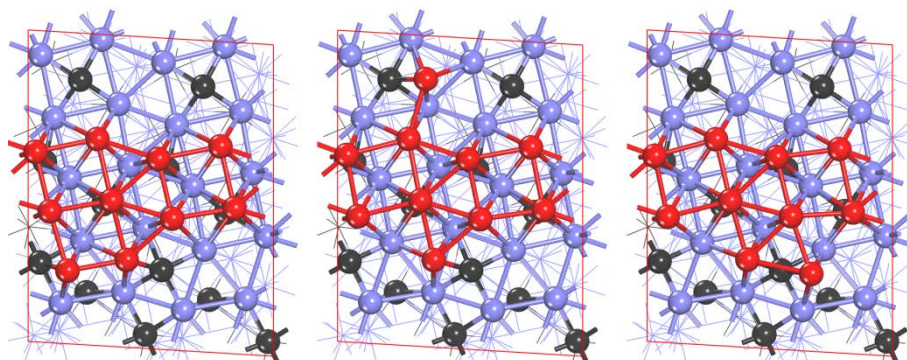
Cu10



a1 (-34.11)

a2 (-33.97)

a3 (-33.84)

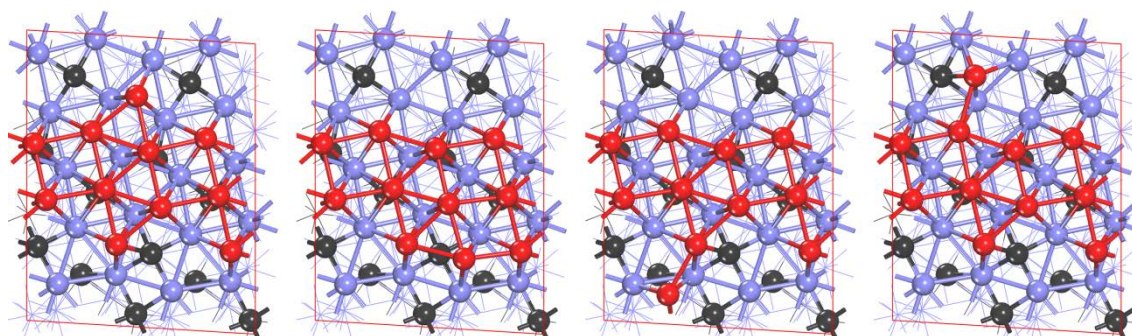


a4 (-33.78)

a5 (-33.70)

a6 (-33.70)

Cu11



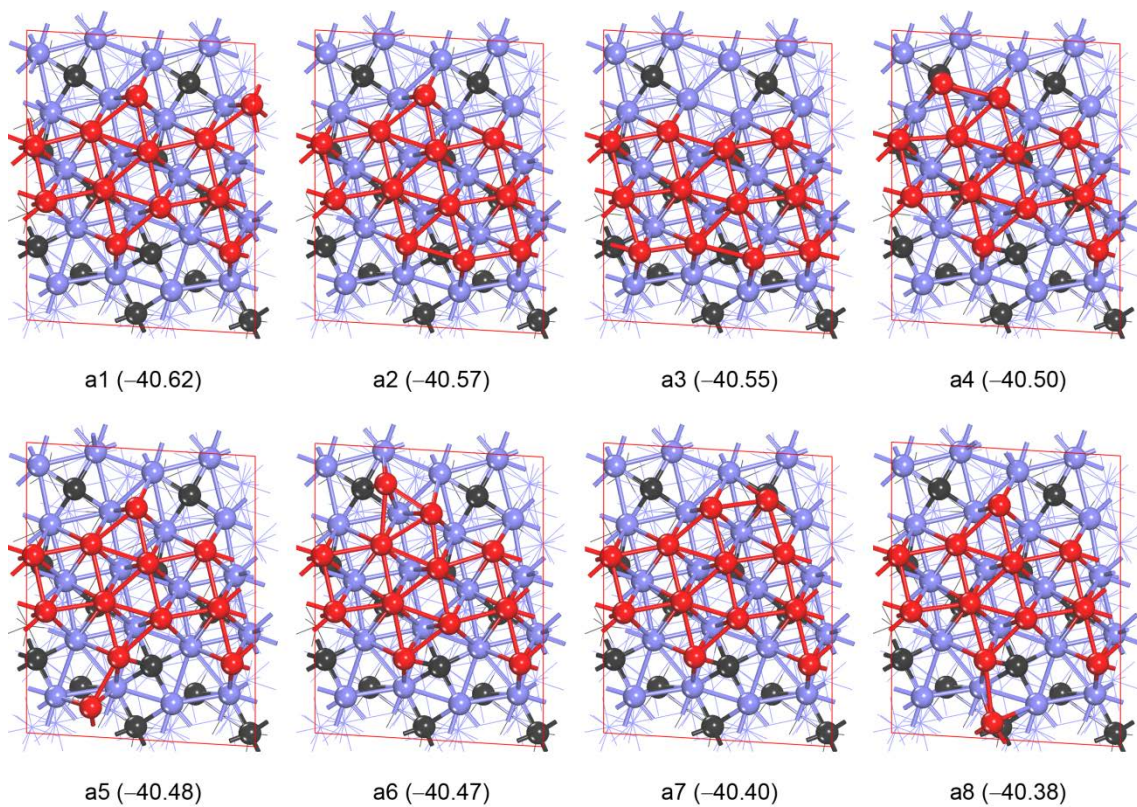
a1 (-37.35)

a2 (-37.33)

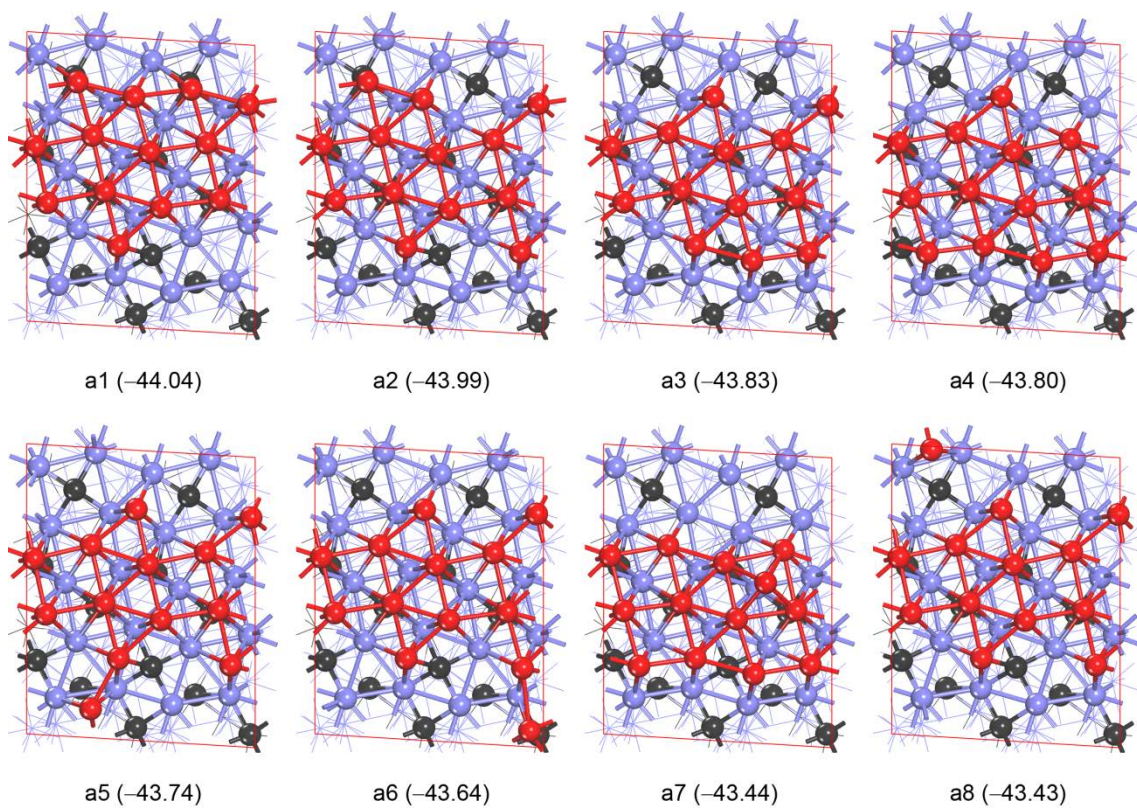
a3 (-37.23)

a4 (-37.08)

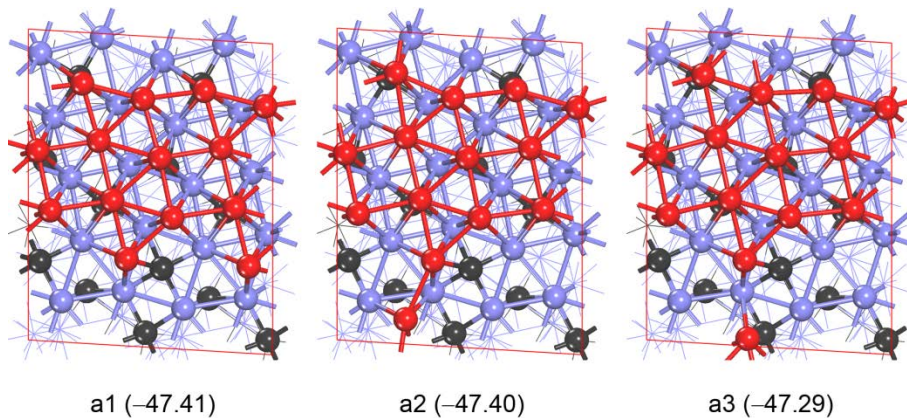
Cu12



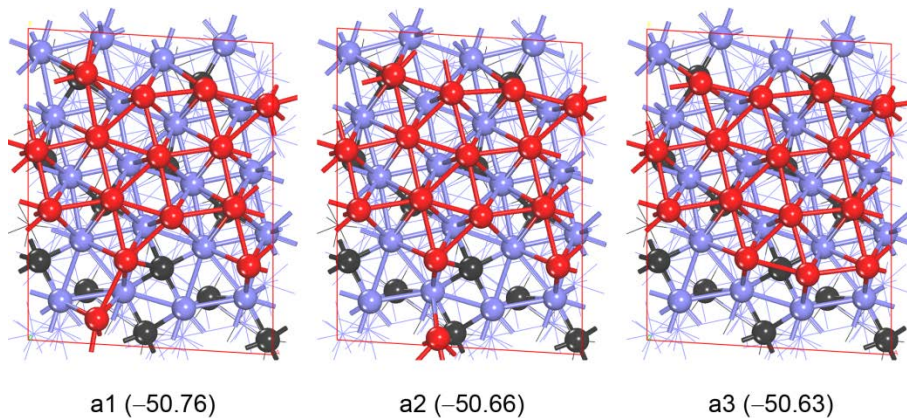
Cu13



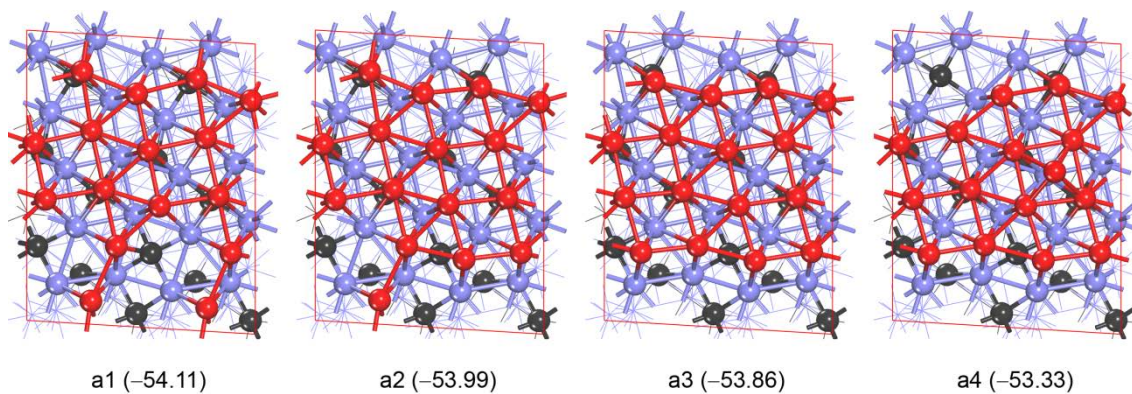
Cu14



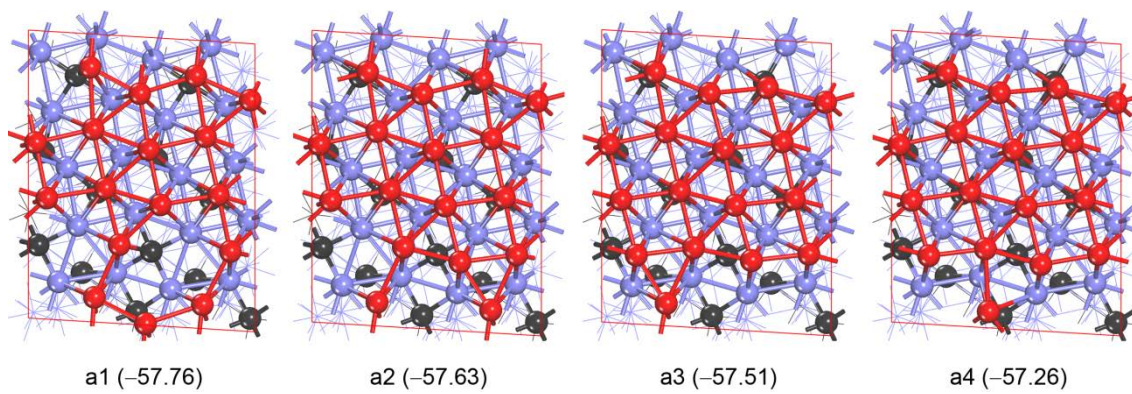
Cu15



Cu16



Cu17



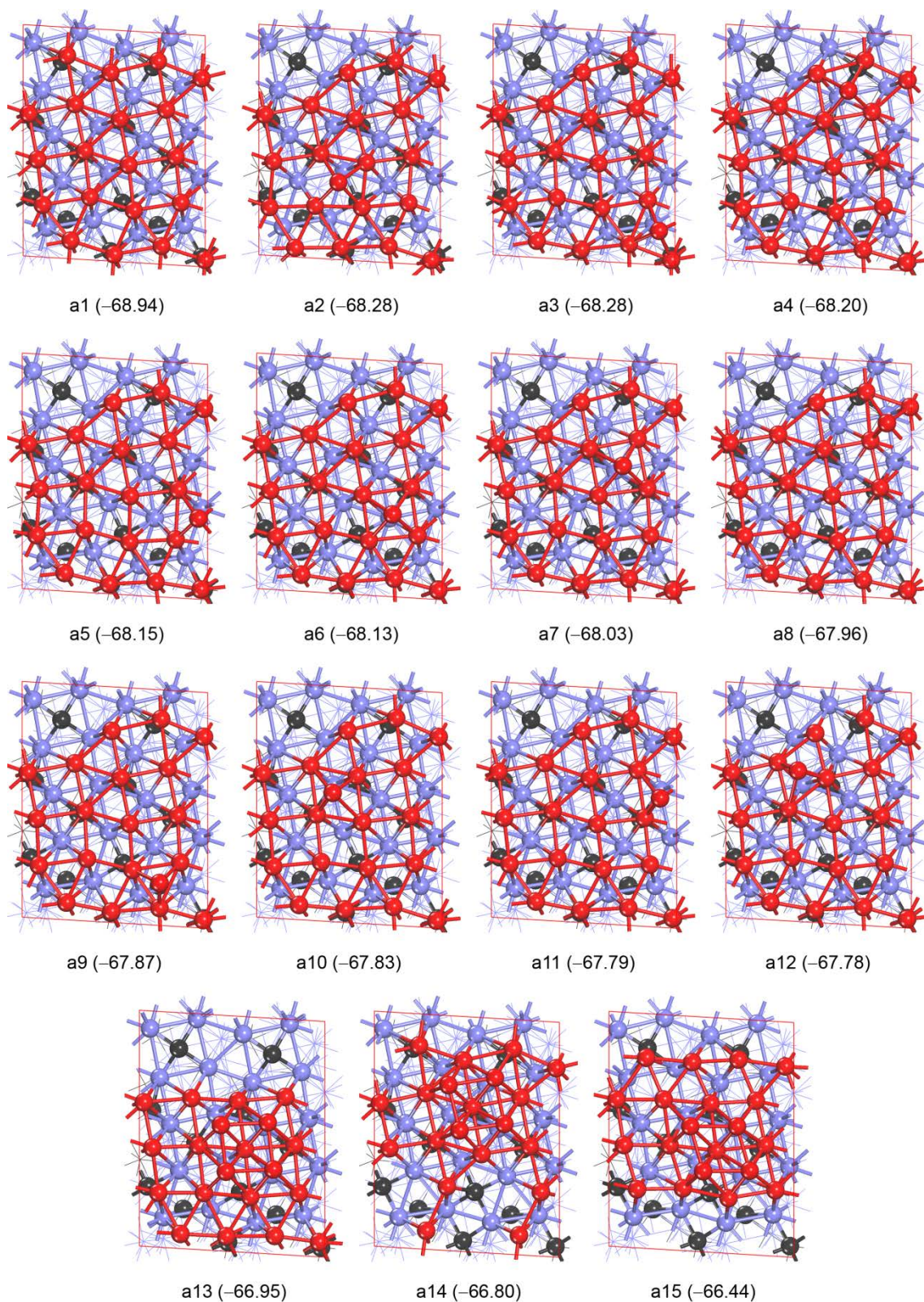


Figure S6. Diffusion pathways of single Cu atom on the $\text{Fe}_5\text{C}_2(510)$ surface

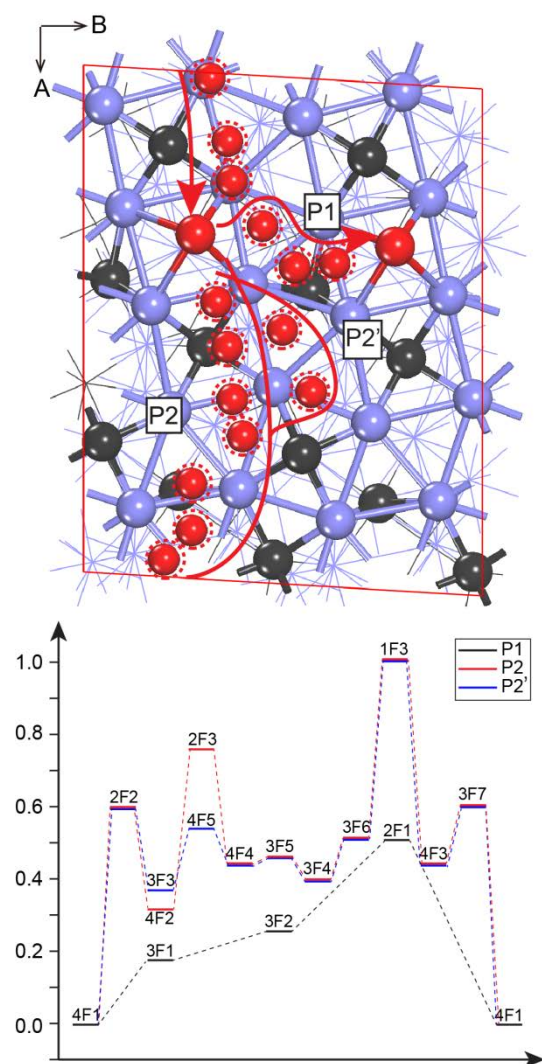
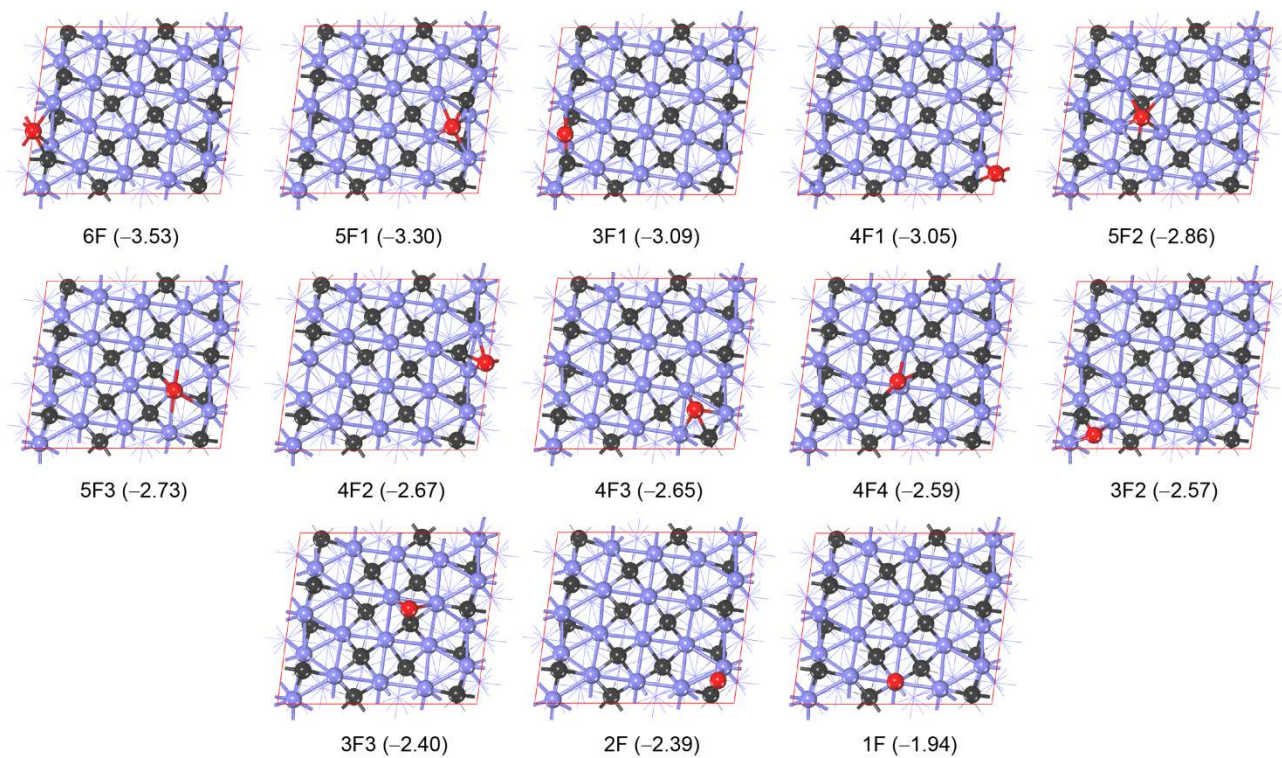
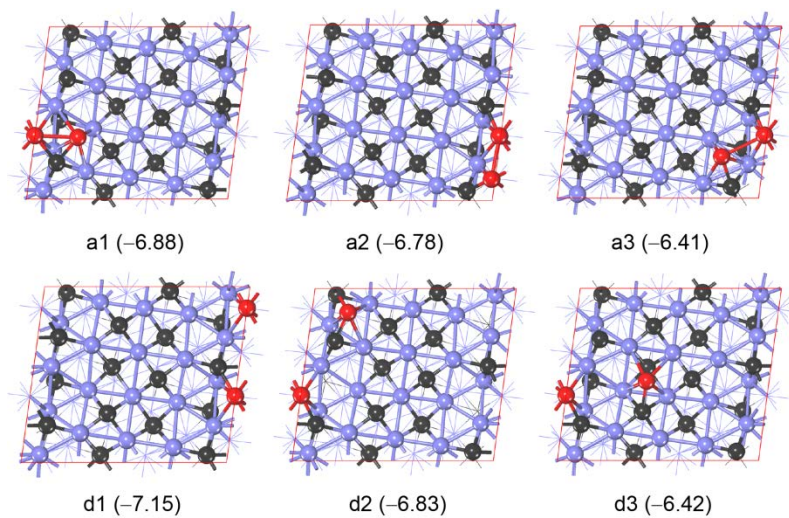


Figure S7. Various optimized Cu_{1-7} , Cu_{12} , Cu_{14} , Cu_{16} , Cu_{18} , Cu_{19} , Cu_{20} configurations on the $p(2\times 1)$ $\text{Fe}_5\text{C}_2(010)$ surface (adsorption energy in eV)

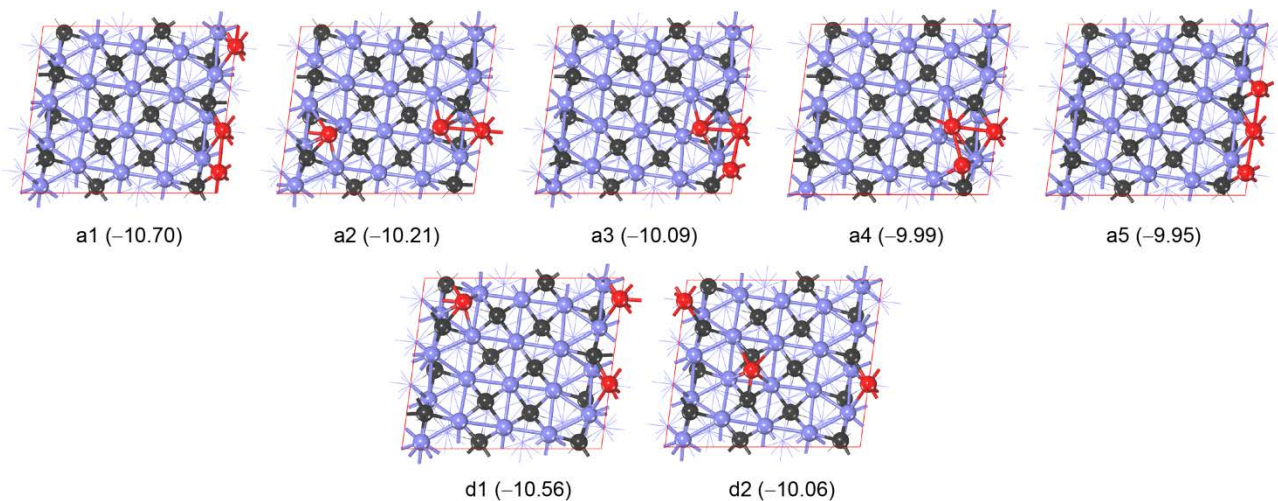
Cu1



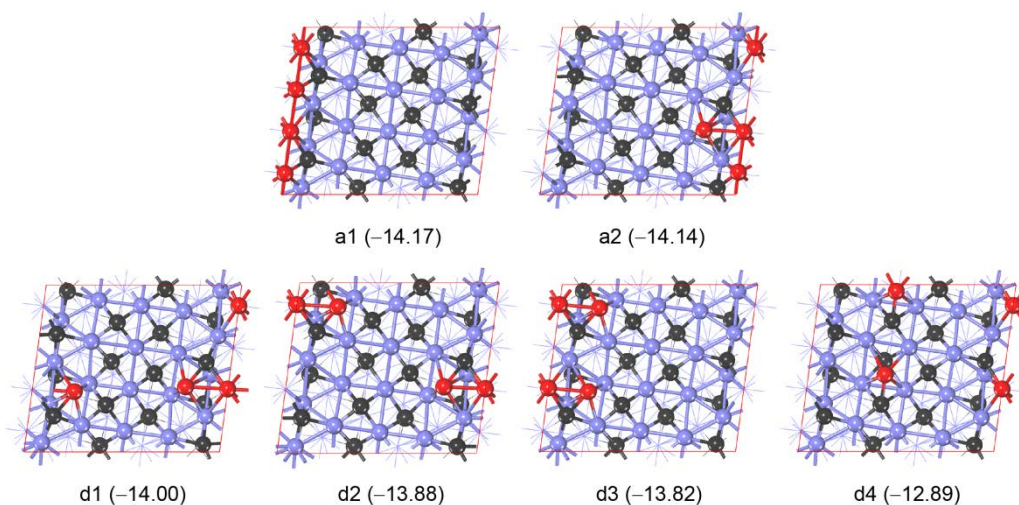
Cu2



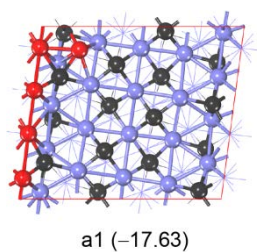
Cu3



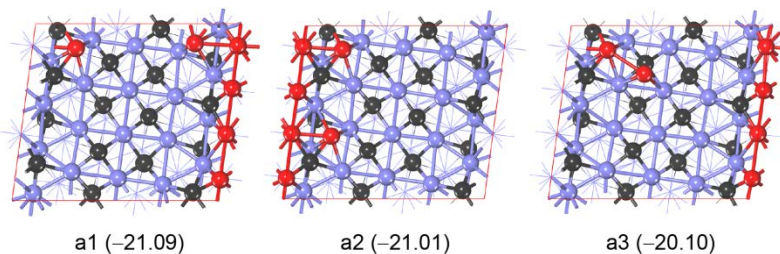
Cu4



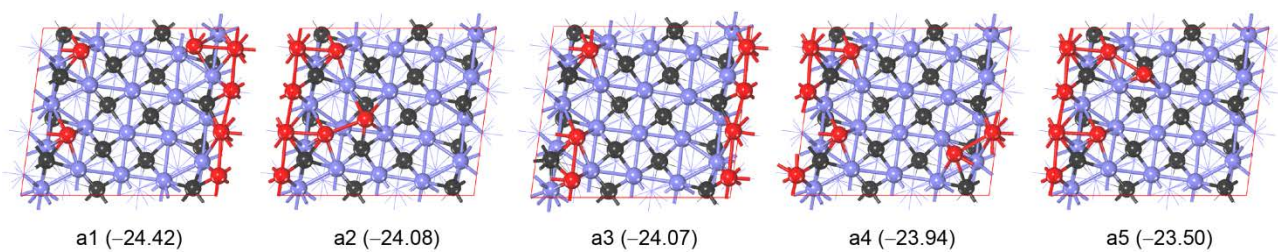
Cu5



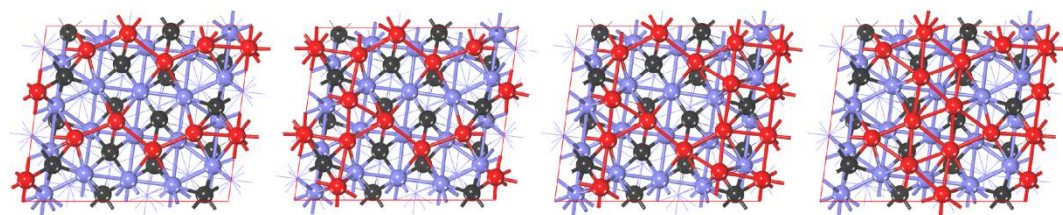
Cu6



Cu7



Cu12-19

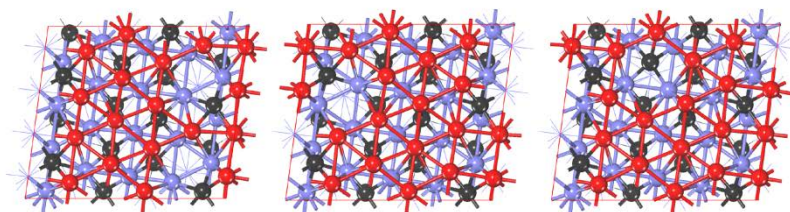


Cu12 (-40.01)

Cu14 (-47.20)

Cu16 (-54.20)

Cu16 (-54.15)

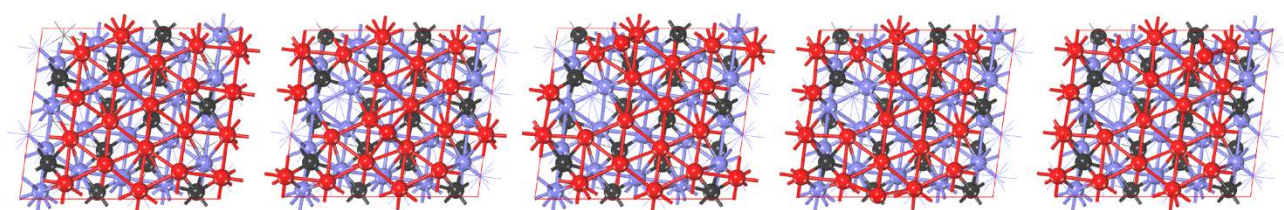


Cu18 (-61.83)

Cu19 (-65.64)

Cu19 (-65.46)

Cu20



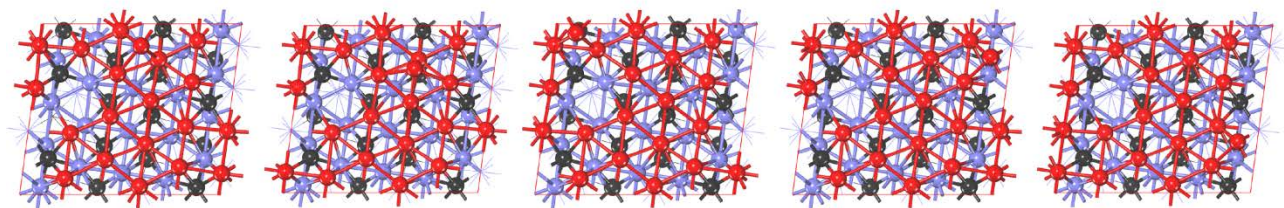
1 (-69.43)

2 (-68.73)

3 (-68.71)

4 (-68.71)

5 (-68.70)



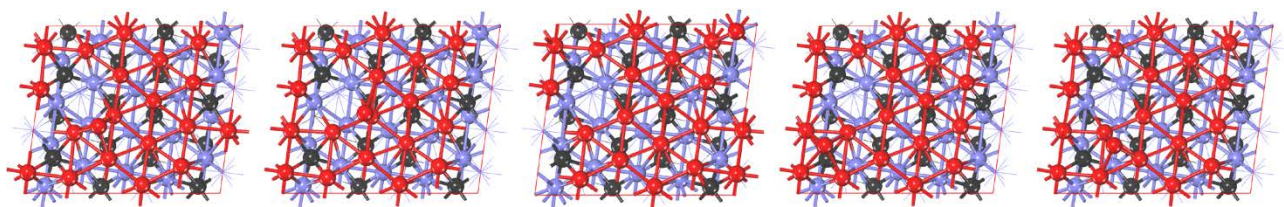
6 (-68.69)

7 (-68.68)

8 (-68.54)

9 (-68.53)

10 (-68.51)



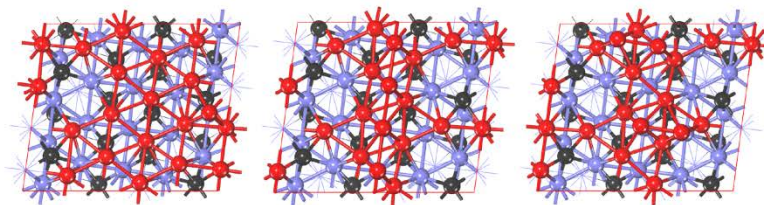
11 (-68.46)

12 (-68.36)

13 (-68.33)

14 (-68.32)

15 (-68.31)



16 (-68.30)

17 (-66.61)

18 (-66.11)

Figure S8. Diffusion pathways of single Cu atom on the $\text{Fe}_5\text{C}_2(010)$ surface

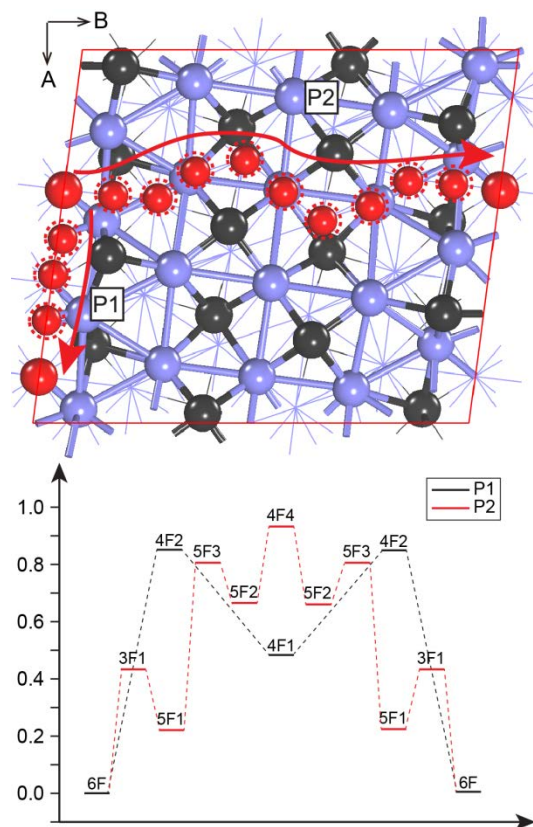
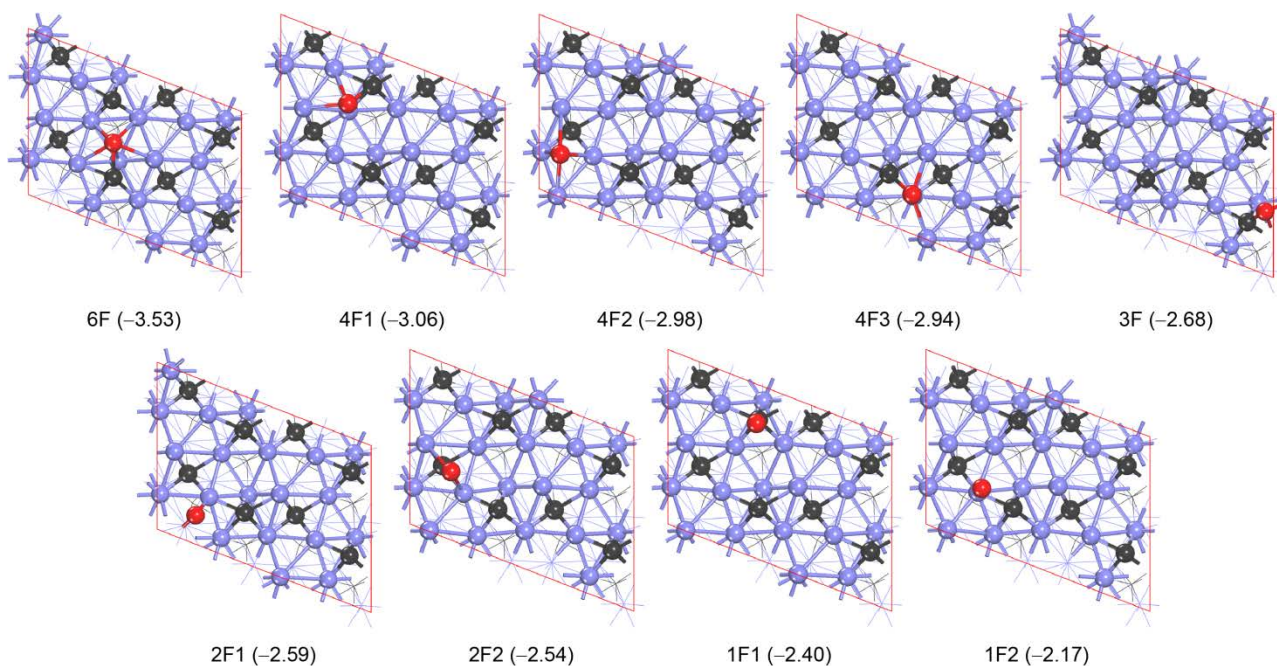
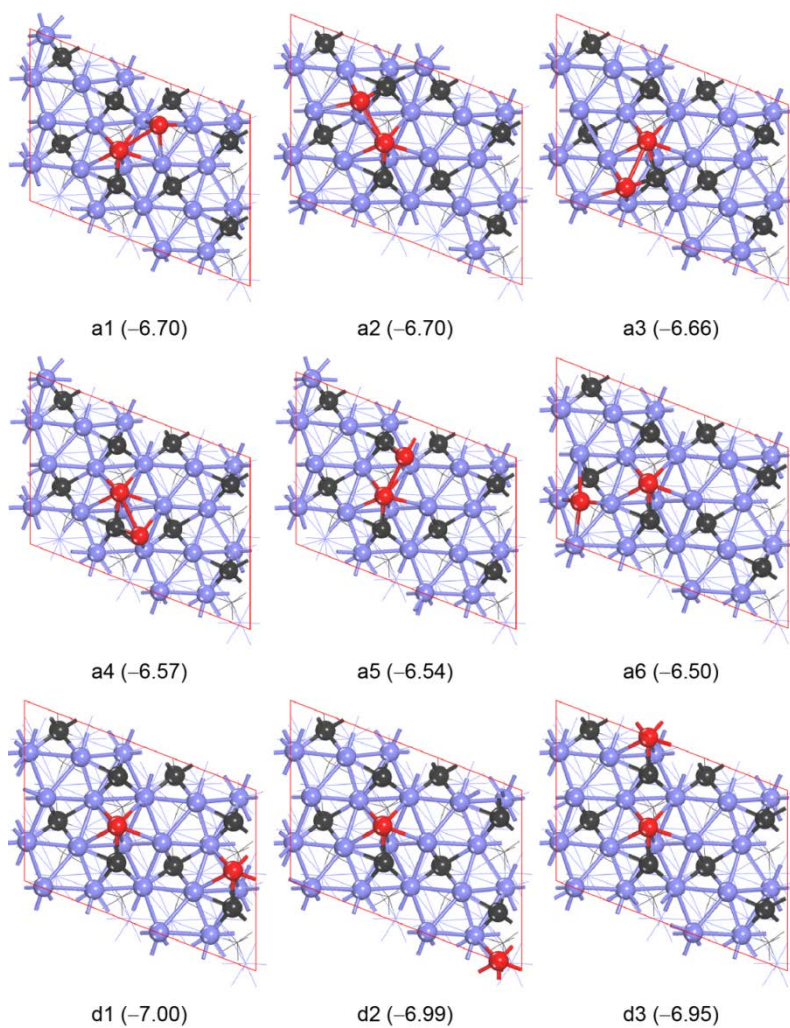


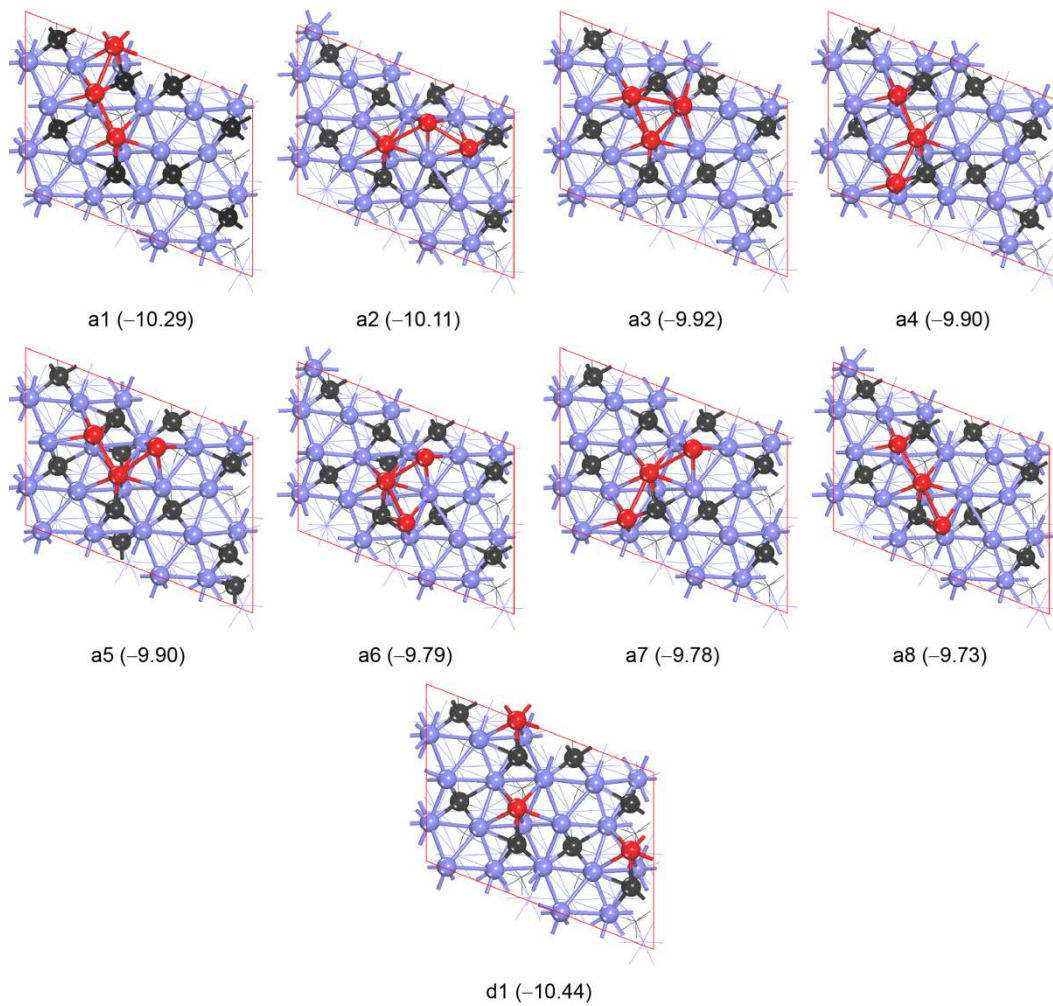
Figure S9. Various optimized Cu_{1-7} , Cu_{13} , Cu_{14} , Cu_{16} , Cu_{19} , Cu_{20} configurations on the $p(2\times 1)$ $\text{Fe}_5\text{C}_2(001)$ surface (adsorption energy in eV)

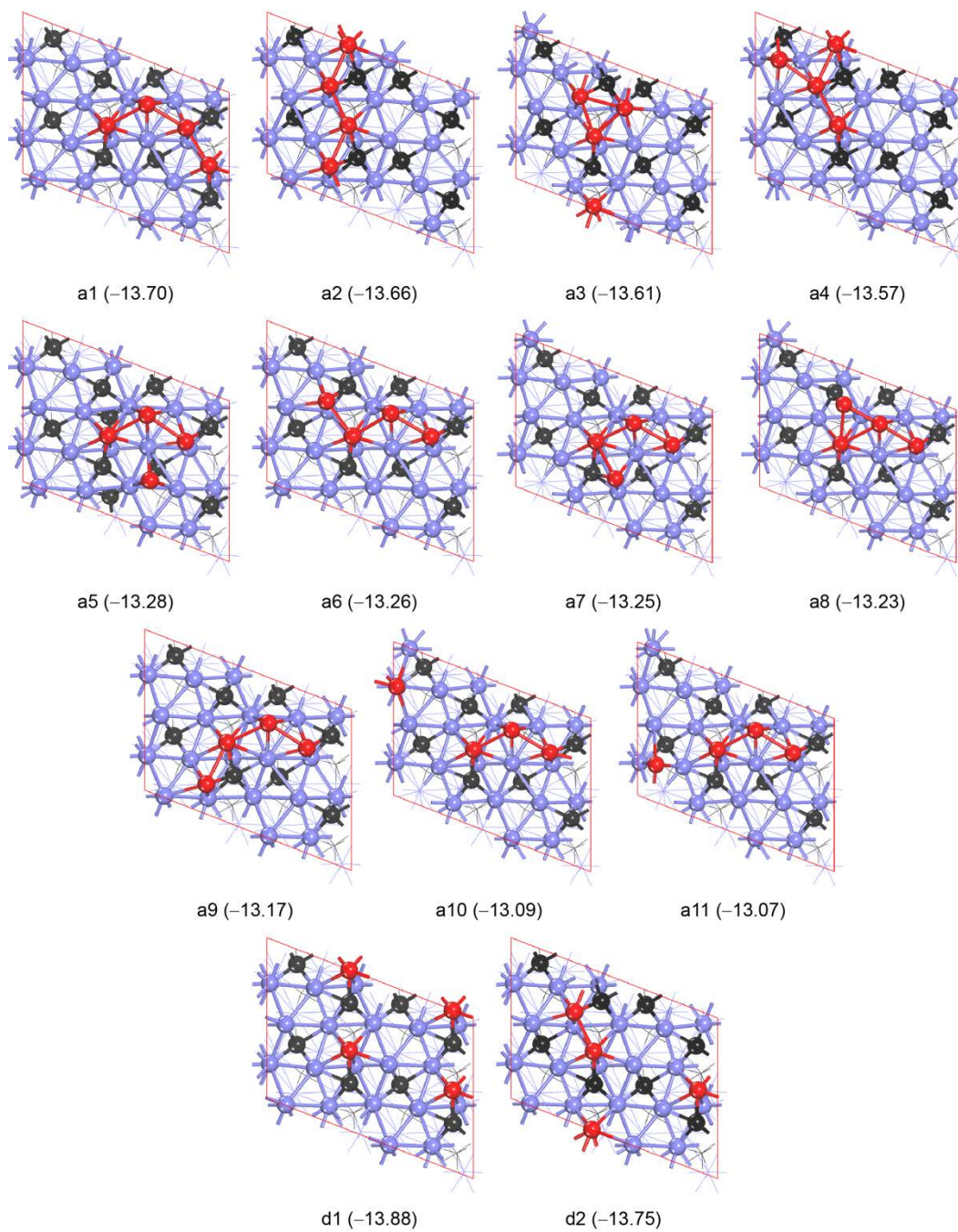
Cu1

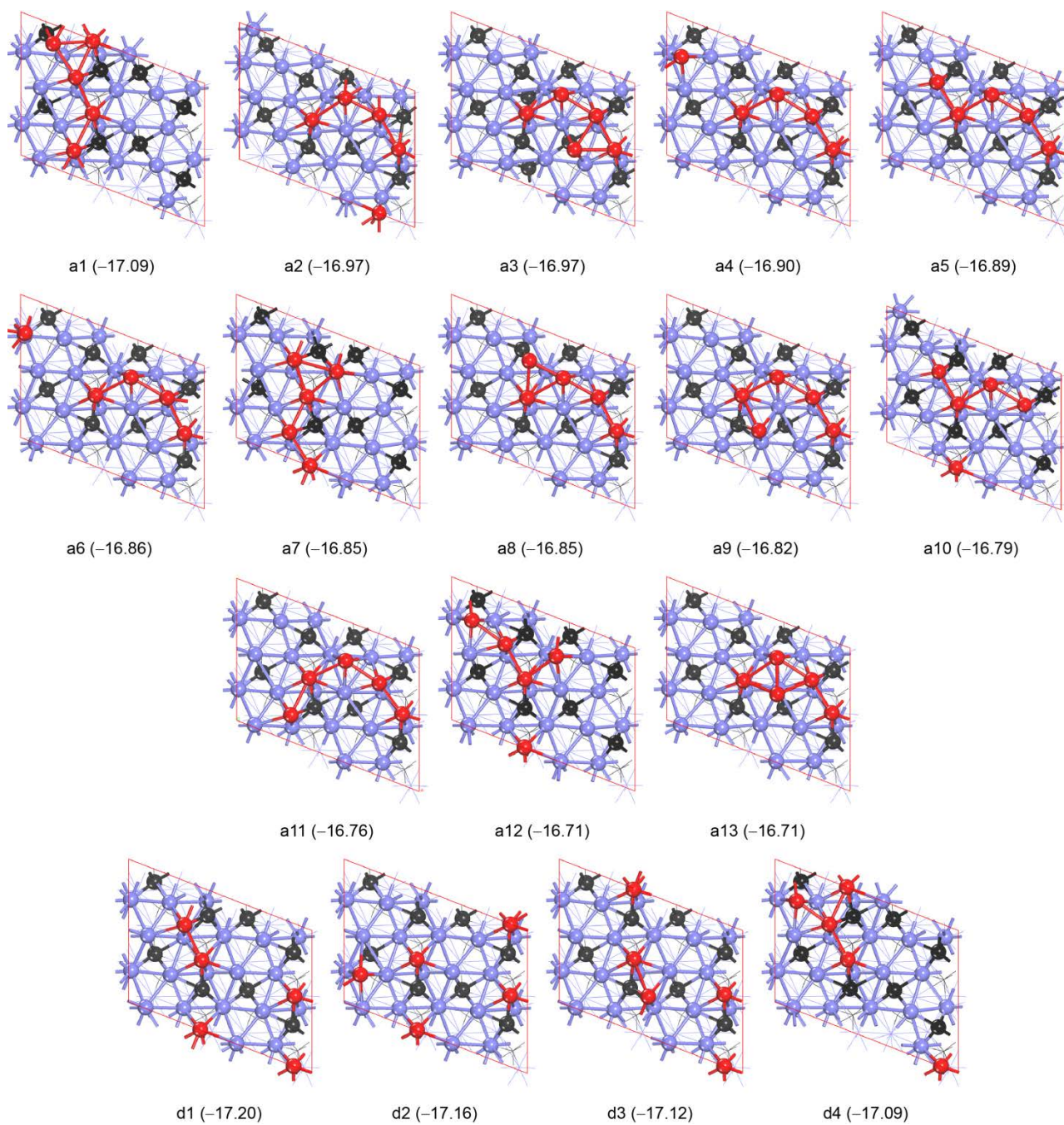


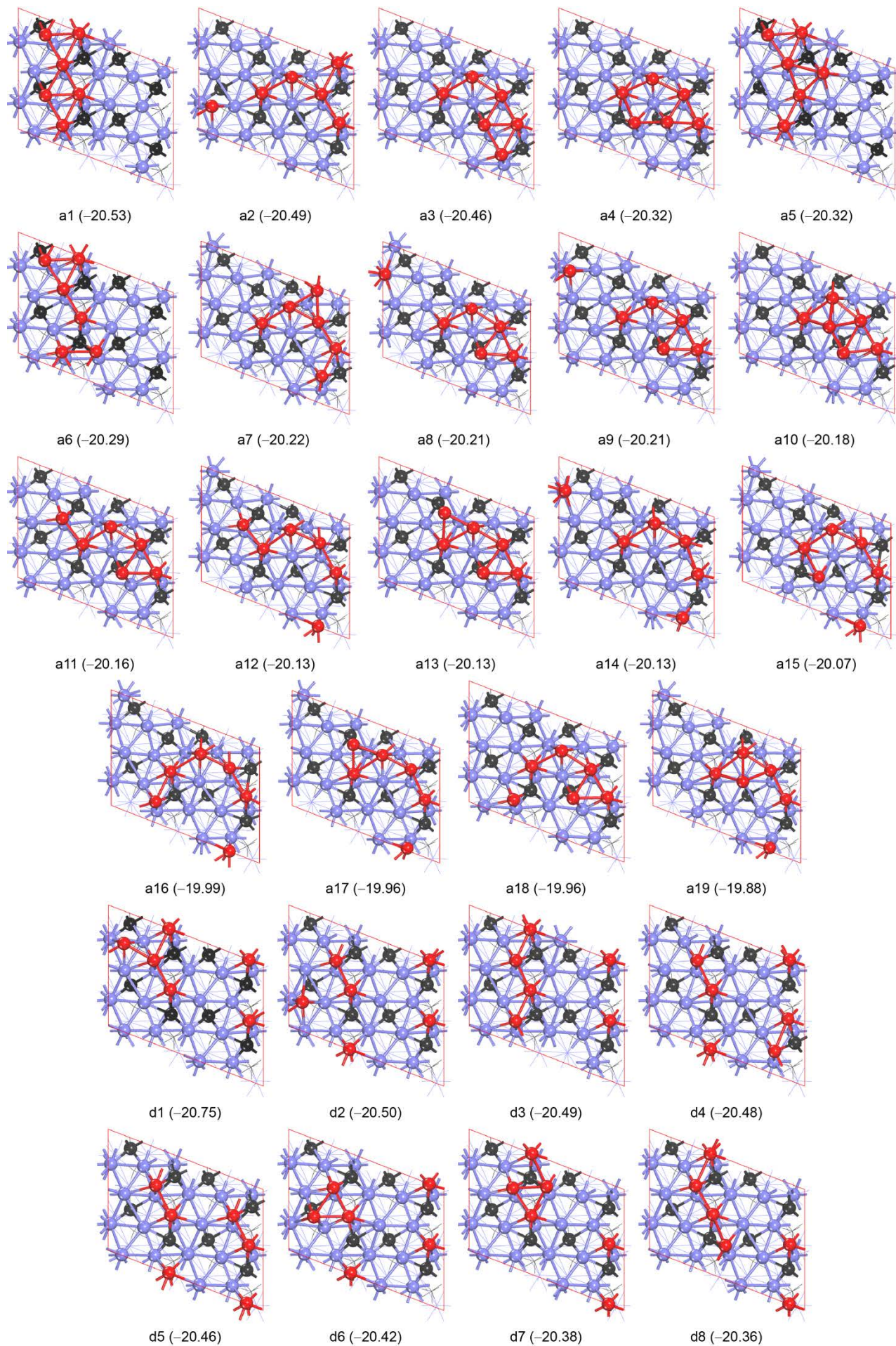
Cu2



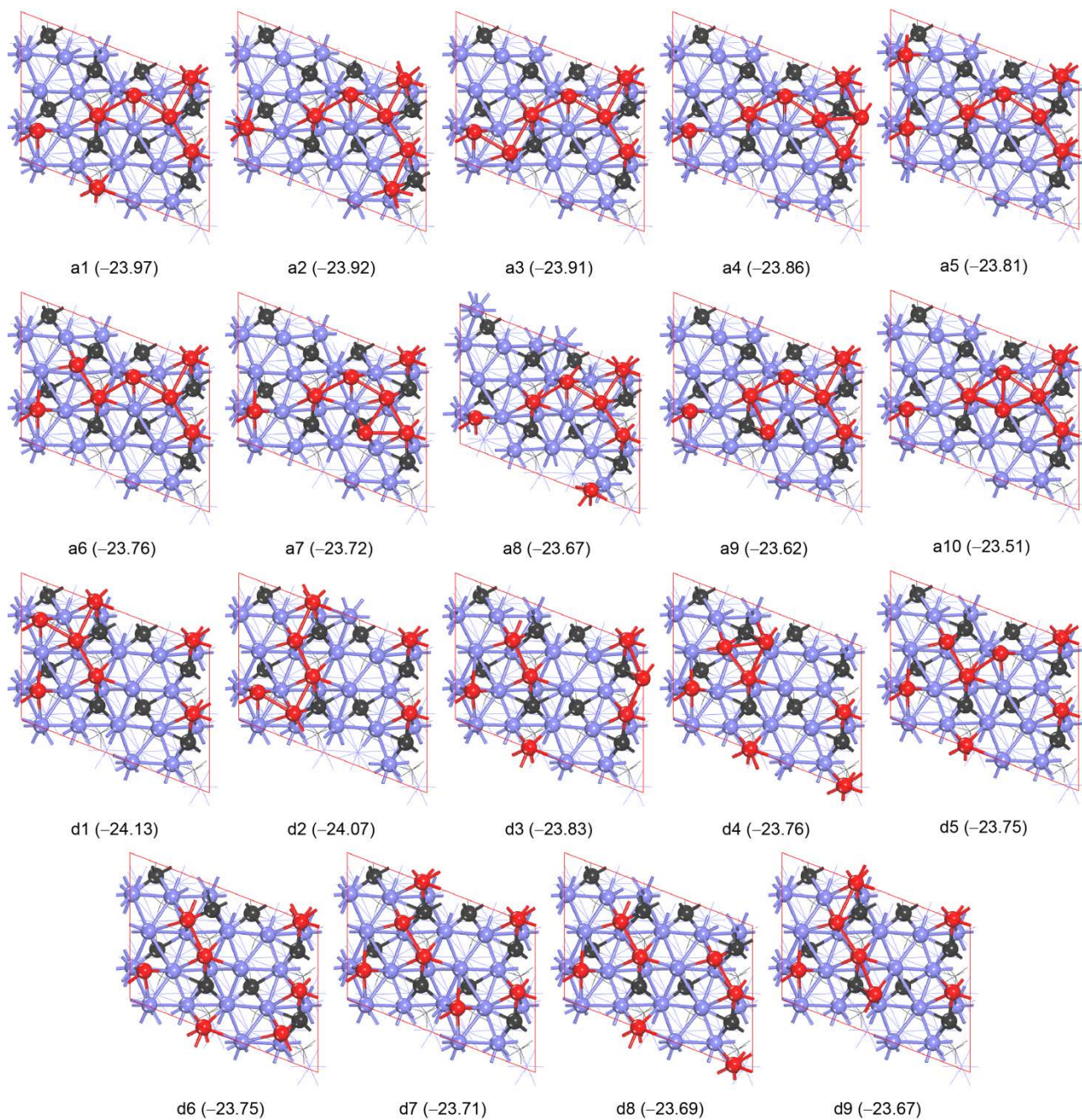


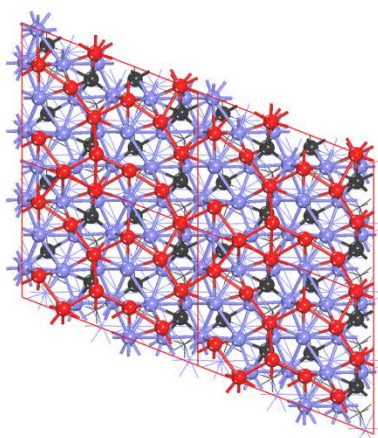




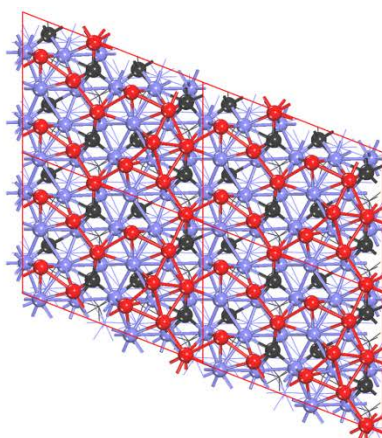


Cu7

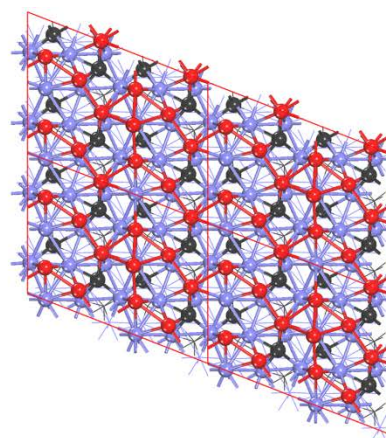




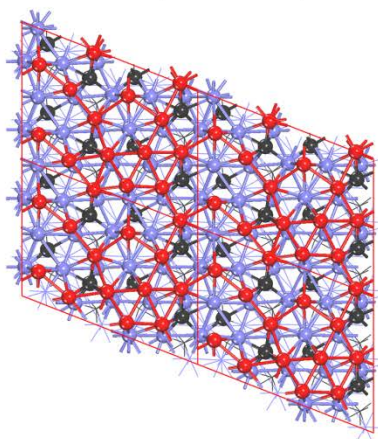
Cu13-1 (-44.87 / -3.45)



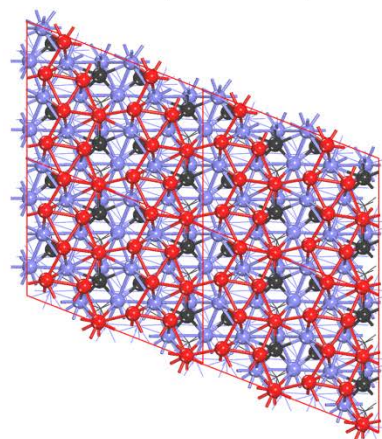
Cu13-2 (-44.83 / -3.45)



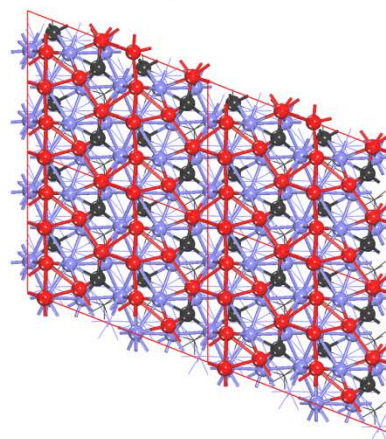
Cu13-3 (-44.82 / -3.45)



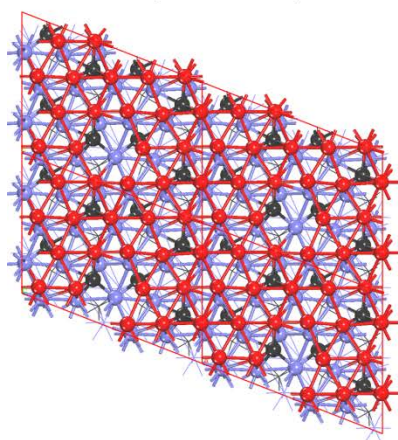
Cu14 (-48.71 / -3.48)



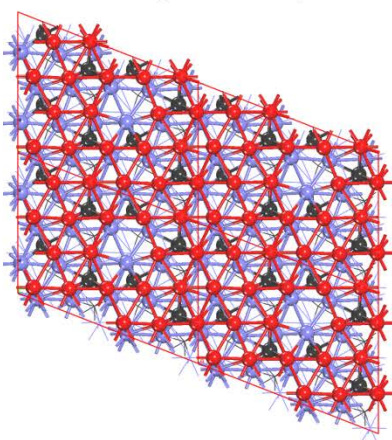
Cu16-1 (-54.99 / -3.44)



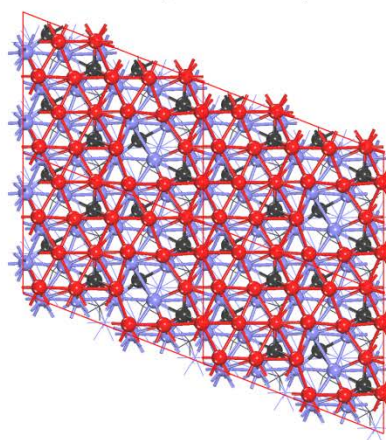
Cu16-2 (-54.98 / -3.44)



Cu19-1 (-67.29 / -3.54)



Cu19-2 (-67.22 / -3.54)



Cu19-3 (-67.16 / -3.53)

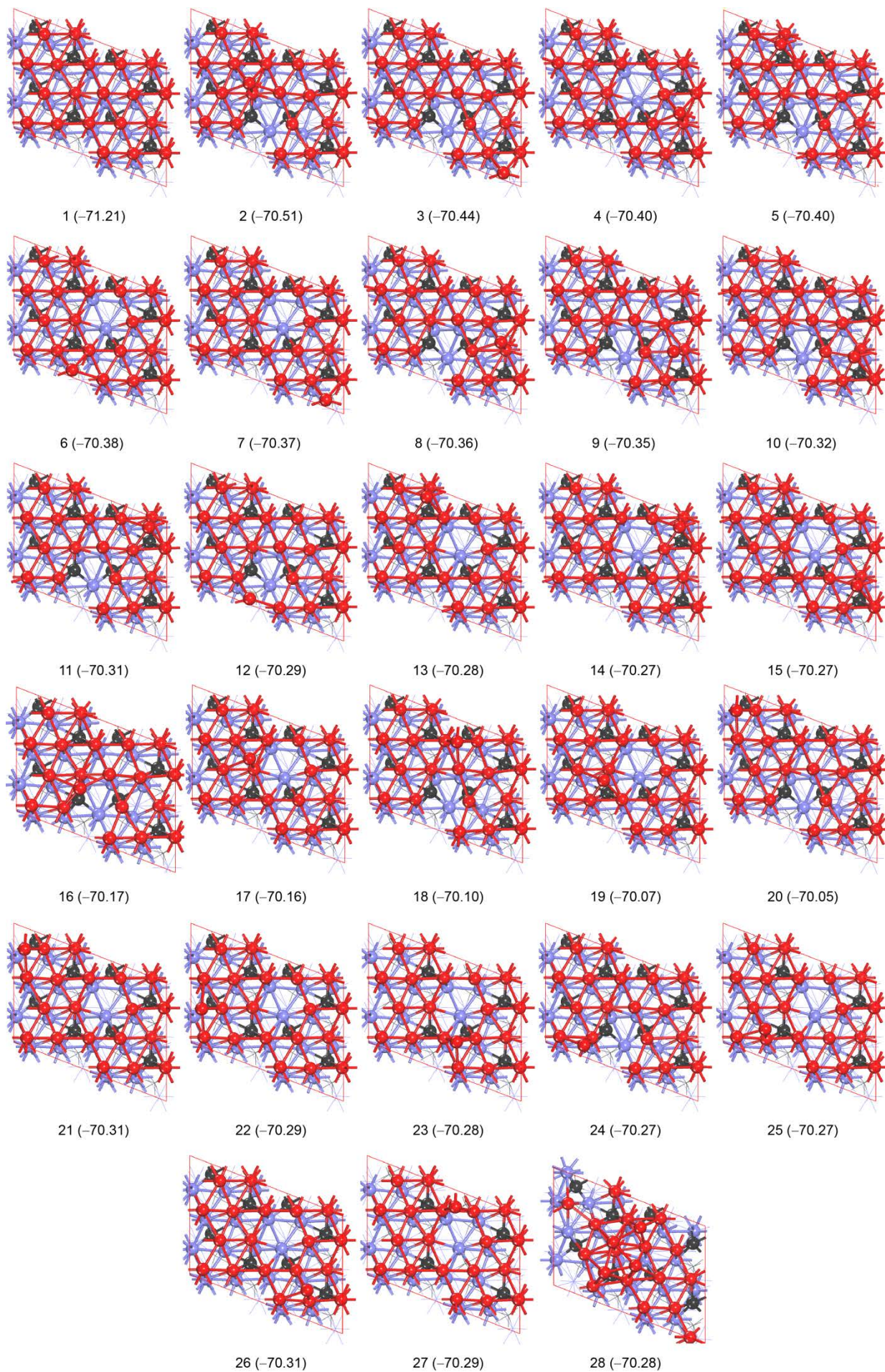


Figure S10. Diffusion pathways of single Cu atom on the $\text{Fe}_5\text{C}_2(001)$ surface

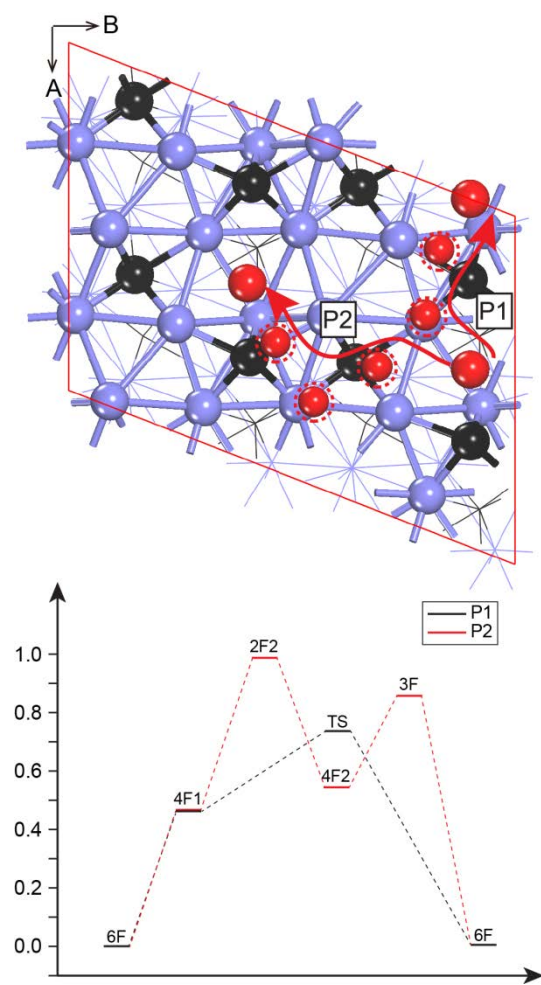


Figure S11. The top and side views of the Cu(111) and Cu_n (1 ML) on the five Fe₅C₂ surfaces.

