Supplementary Information

Shape Control of Zn(II) Metal-Organic Frameworks by Modulation Synthesis and Their Morphology-Dependent Catalytic Performance

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Figure S1. Chemical structure of H₂oba, 4-bpdb, 4-bpdh and 4-bpmb.

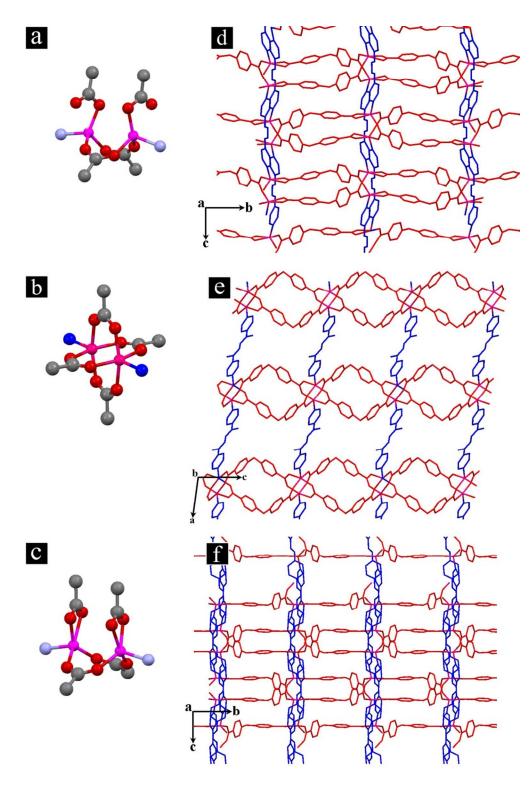


Figure S2. Representation of binuclear Zn_2 units in (a) TMU-4, (b) TMU-5 and (c) TMU-6. Layers of Zn(II)-oba (in red) pillared by N-donor ligand (in blue), (d) 4-bpdb in TMU-4, (e) 4-bpdh in TMU-5 and (f) 4-bpmb in TMU-6.

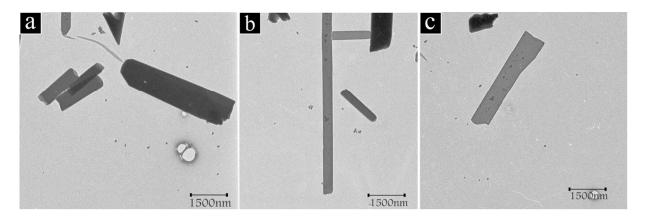


Figure S3. (a-c) TEM images of TMU-5 which shown in Figure 1d; r = 2, T = 100 °C, 24h; r represents the ratio between acetic acid and oba. The concentration of oba is c = 0.025 M.

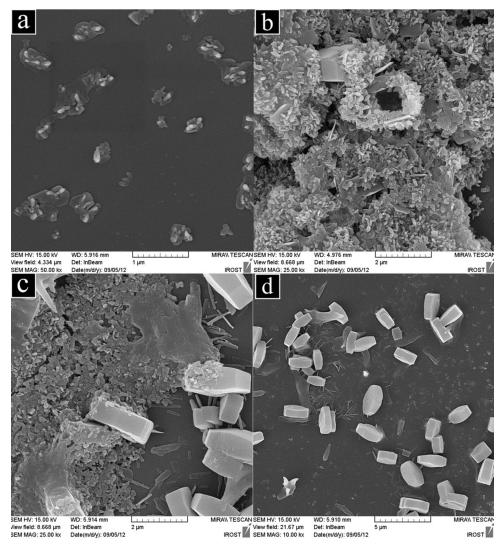


Figure S4. Time dependent FE-SEM images of TMU-5 micro-plates that was obtained under conditions of r = 30, T = 100 °C, after (a) 2 h (b) 15 h, (c) 20 h and (d) 24 h.

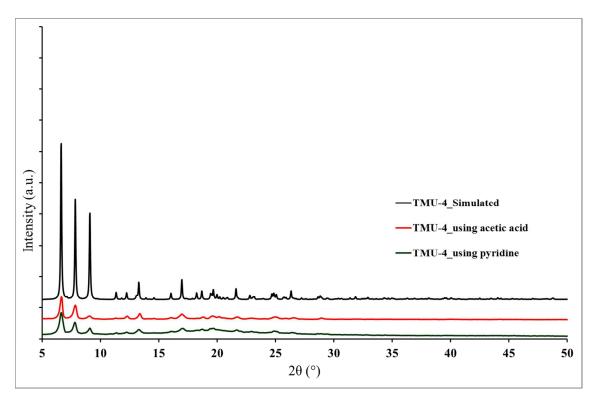


Figure S5. Powder X-ray diffraction patterns of TMU-4. While using acetic acid conditions are r = 10, T = RT, 24h and when using pyridine conditions are r = 10, T = RT, 2 h.

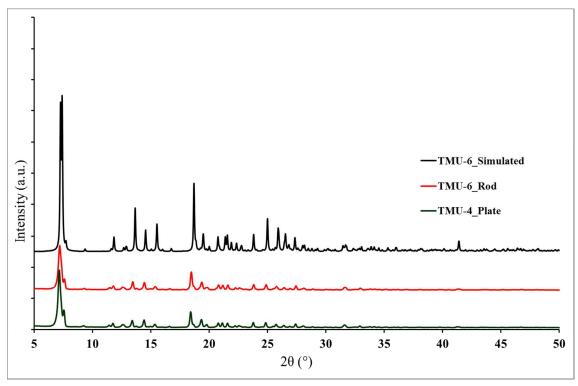


Figure S6. Powder X-ray diffraction patterns of TMU-6. For rod-shaped conditions are r = 1, T = RT, 24h and for plate-shaped conditions is r = 1, T = RT, 24h.

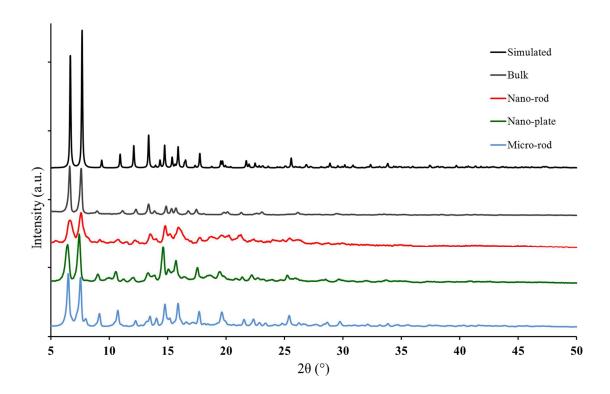


Figure S7. Powder X-ray diffraction patterns of TMU-5's simulated, bulk, nano-rod, nano-plate and micro-rod after 4 successive runs.

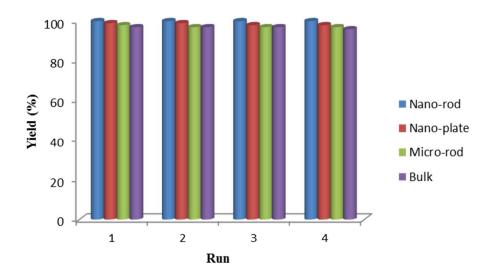


Figure S8. Recyclability of TMU-5 morphologies after 4 successive runs.

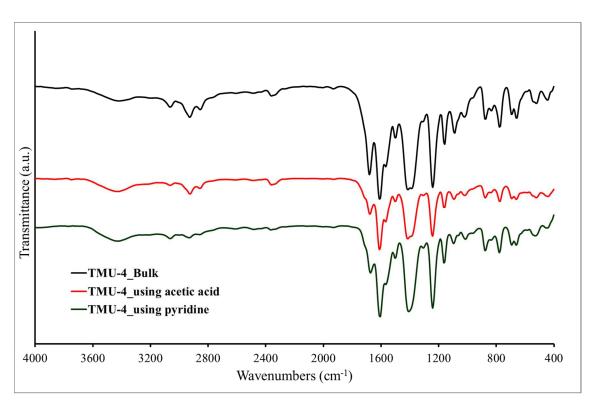


Figure S9. IR spectra of TMU-4 bulk, upon using acetic acide (r = 10, T = RT, 24h) and upon using pyridine (r = 10, T = RT, 2 h).

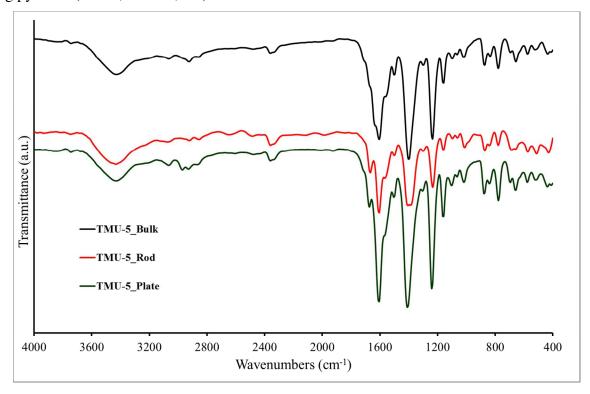


Figure S10. IR spectra of TMU-5 bulk, rod-shaped (r = 10, T = 100 °C, 24h) and plate-shaped morphology (r = 10, T = RT, 24h).

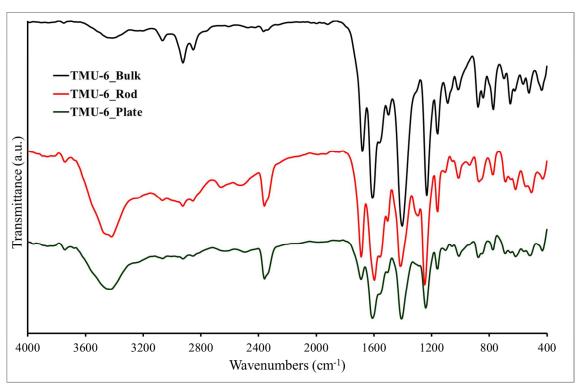


Figure S11. IR spectra of TMU-6 bulk, rod-shaped (r = 1, T = RT, 24h) and plate-shaped morphology (r = 1, T = RT, 24h).