

Development of a New Dual Polarity and Viscosity Probe Based on the Foldamer Concept

Delia Miguel,^a Sara P. Morcillo,^a Ana Martín-Lasanta,^a Noelia Fuentes,^b Lara Martínez-Fernández,^c Inés Corral,^c Maria J. Ruedas-Rama,^d Diego J. Cárdenas,^b Luis Álvarez de Cienfuegos,^{*a} Angel Orte,^{*d} and Juan M. Cuerva^{*a}

^a Organic Chemistry Department, Science Faculty, Universidad de Granada, C. U. Fuentenueva, 18071 Granada (Spain). E-mail: luc@ugr.es, jmcuerva@ugr.es.

^b Organic Chemistry Department, Universidad Autónoma de Madrid (UAM), Ciudad Universitaria de Cantoblanco, 28049 Madrid (Spain).

^c Chemistry Department, UAM, 28049 Madrid (Spain).

^d Physical Chemistry Department, Pharmacy Faculty, Universidad de Granada, C. U. Cartuja, 18071 Granada (Spain). E-mail: angelort@ugr.es.

Index

General Information.....	S2
Synthesis and characterization data of the precursor II and the compound PyomoPy	S3
Copies of ¹ H-NMR and ¹³ C-NMR of compounds II and PyomoPy	S4
Absorption and Steady State Fluorescence.....	S8
Computational Details	S10
Time-resolved fluorescence of PyomoPy in different solvents	S30
Time-resolved fluorescence of PyomoPy as a function of the media viscosity.....	S34
References	S36

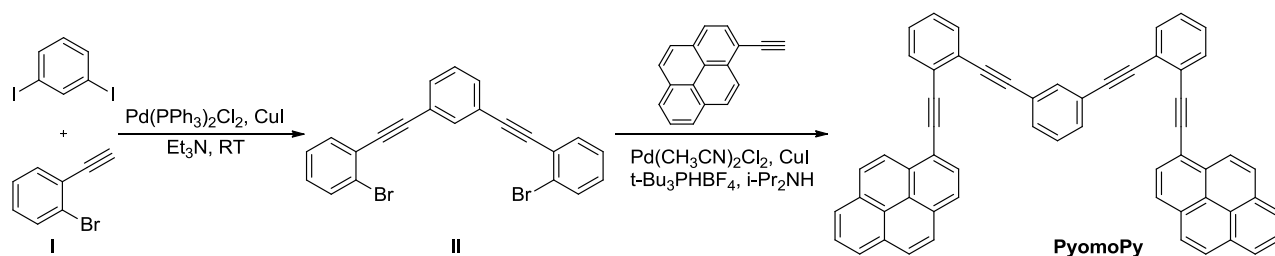
General Information

Sonogashira couplings were carried out in dry glassware under positive pressure of argon. All commercially available reagents and solvents used in extraction and purification, (triethylamine, diisopropylamine, dichlorometane, hexane and ethyl acetate) were obtained from standard chemical suppliers and used without further purification. Compound **I** was also commercially available. TLC was performed on aluminium-backed plates coated with silica gel 60 (230-240 mesh) with F₂₅₄ indicator. The spots were visualized with UV light (254 nm) and/or staining with Ce/Mo reagent or phosphomolybdic acid solution and subsequent heating.

NMR Spectra were measured at room temperature. ¹H NMR spectra were recorded at 300 or 600 MHz. Chemical shifts are reported in ppm using residual solvent peak as reference (CHCl₃: δ 7.26). Data are reported as follows: chemical shift, multiplicity (s: singlet, d: doublet, m: multiplet, dd: doublet of doublets), coupling constant (*J* in Hz) and integration. ¹³C-NMR spectra were recorded at 75.4 or 151 MHz using broadband proton decoupling and chemical shifts are reported in ppm using residual solvent peaks as reference (CDCl₃: δ 77.16). Carbon multiplicities were assigned by DEPT techniques.

High resolution mass spectra (HRMS) were recorded on a Micromass AutoSpec using EI at 70eV.

Synthesis of **PyomoPy**:



Compound II: 1,3-diiodobenzene (231 mg, 0.7 mmol), Pd(PPh₃)₂Cl₂ (48mg, 0.07 mmol) and CuI (13mg, 0.07 mmol) were carefully deoxygenated and suspended in Et₃N (5 mL). After 10 minutes, a solution of the commercial alkyne **I** (300 mg, 1.7 mmol) in 2 mL of Et₃N was added dropwise over 10 minutes and the reaction mixture was stirred overnight at room temperature under argon atmosphere. The mixture was then diluted with CH₂Cl₂ (3 x 15 mL), washed with saturated NH₄Cl (3 x 15 mL), dried over anhydrous Na₂SO₄ and the solvent was removed. The residue was purified by flash chromatography (CH₂Cl₂:Hexane, 1:4) to give the corresponding coupling product **II** (250 mg, 82%) as a pale brown solid.

¹H-NMR (300 MHz, CDCl₃): δ = 7.70 (s, 1H), 7.52 (d, *J* = 8.0 Hz, 2H), 7.49-7.43 (m, 4H), 7.26 (t, *J* = 7.8 Hz, 1H), 7.19 (t, *J* = 7.5 Hz, 2H), 7.12 – 7.05 (m, 2H).

¹³C-NMR (75 MHz, CDCl₃): δ = 134.7 (CH), 133.4 (CH), 132.6 (CH), 131.9 (CH), 129.7 (CH), 128.7 (CH), 127.2 (CH), 125.8 (C), 125.2 (C), 123.5 (C), 93.1 (C), 88.8 (C).

HRMS (EI, 70 eV) *m/z* calcd. for C₂₂H₁₂Br₂ [M]⁺: 433.9306; found: 433.9317, [M+2]⁺: 435.9285; found: 435.9287.

PyomoPy: Compound **I** (93 mg, 0.2 mmol), Pd(CH₃CN)₂Cl₂ (5 mg, 0.02 mmol), CuI (4 mg, 0.02 mmol) and *t*Bu₃PHBF₄ (12 mg, 0.04 mmol) were carefully deoxygenated and suspended in a mixture of dioxane (3 mL) and *i*Pr₂NH (3 mL). After 10 minutes, a solution of 1-ethynylpyrene (116 mg, 0.5 mmol) in 2 mL of *i*Pr₂NH was added dropwise over 10 minutes and the reaction mixture was stirred under argon atmosphere at 80°C for 18h. The mixture was then diluted with CH₂Cl₂, washed with saturated NH₄Cl (3 x 15 mL), dried over anhydrous Na₂SO₄ and the solvent was removed. The residue was purified by flash chromatography (CH₂Cl₂:Hexane, 1:4) to give the corresponding coupling product **PyomoPy** (110 mg, 76%) as a brown solid.

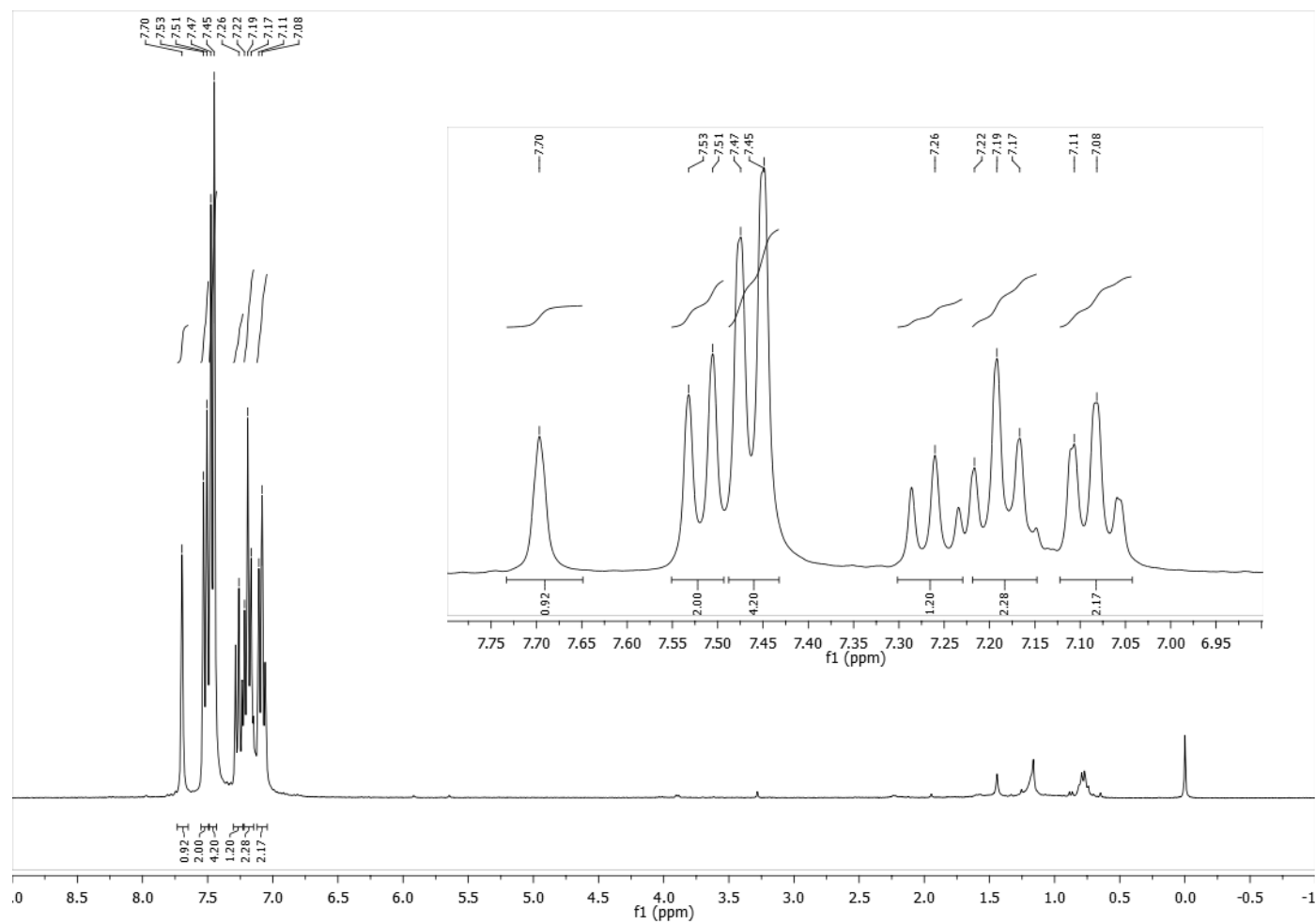
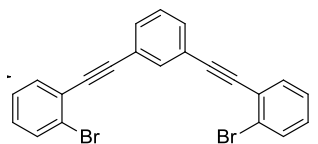
¹H-NMR (600 MHz, CDCl₃): δ = 8.37 (d, *J* = 8.9 Hz, 2H), 8.21 (s, 1H), 8.04 (d, *J* = 7.5 Hz, 2H), 7.96 (d, *J* = 2.8 Hz, 2H), 7.94 (d, *J* = 4.1 Hz, 2H), 7.81 (d, *J* = 9.0 Hz, 2H), 7.78 (d, *J* = 8.8 Hz, 2H), 7.77 – 7.70 (m, 6H), 7.67 (d, *J* = 7.7 Hz, 2H), 7.63 (d, *J* = 7.6 Hz, 2H), 7.56 (d, *J* = 7.6 Hz, 2H), 7.42 – 7.38 (m, 3H), 7.36 (dd, *J* = 14.4, 6.9 Hz, 2H).

¹³C-NMR (151 MHz, CDCl₃): δ = 135.8 (CH), 132.4 (CH), 132.2 (CH), 132.1 (CH), 131.6 (C), 131.2 (C), 131.1 (C), 129.4 (CH), 128.77 (CH), 128.75 (CH), 128.4 (CH), 128.03 (CH), 127.96 (CH), 127.2 (CH), 126.5 (C), 125.95 (CH), 125.94 (CH), 125.4 (CH), 125.29 (CH), 125.27 (C), 124.18 (CH), 124.15 (C), 124.10 (C), 123.9 (C), 117.2 (C), 93.7 (C), 93.2 (C), 92.8 (C), 89.8 (C).

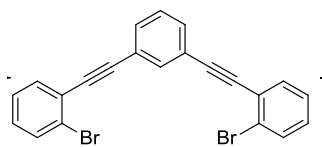
HRMS (ESI) *m/z* calcd. for C₅₈H₃₁ [M⁺+1]: 727.2426; found: 727.2405.

Copies of ^1H -NMR and ^{13}C -NMR of compounds **II** and **PyomoPy**

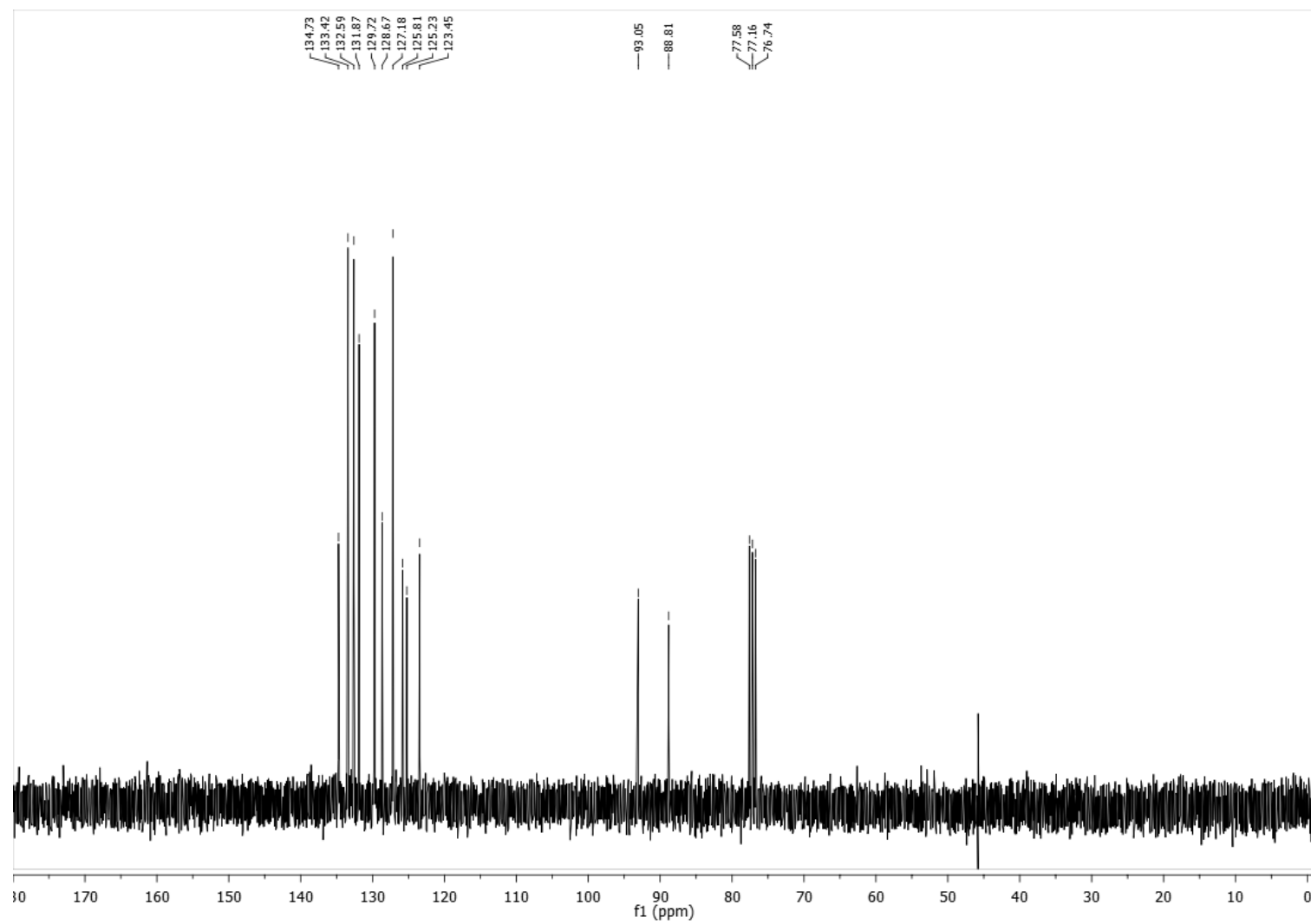
^1H -NMR (CDCl_3 , 300 MHz)



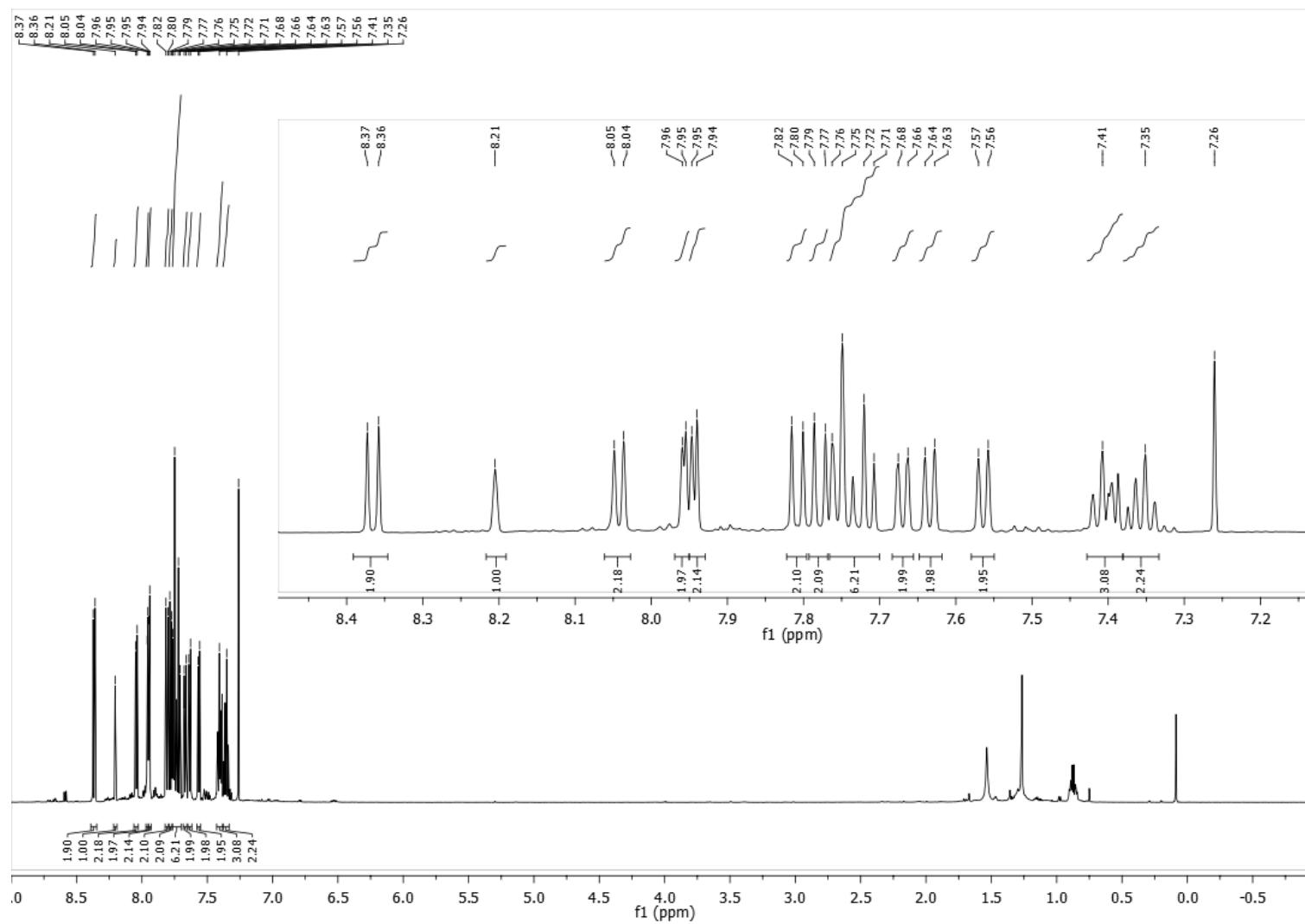
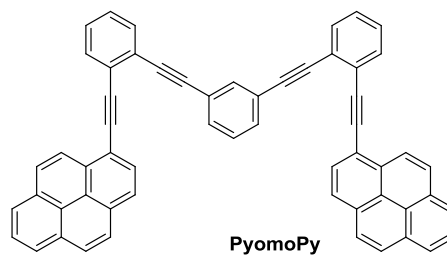
^{13}C -NMR (CDCl_3 , 75 MHz)



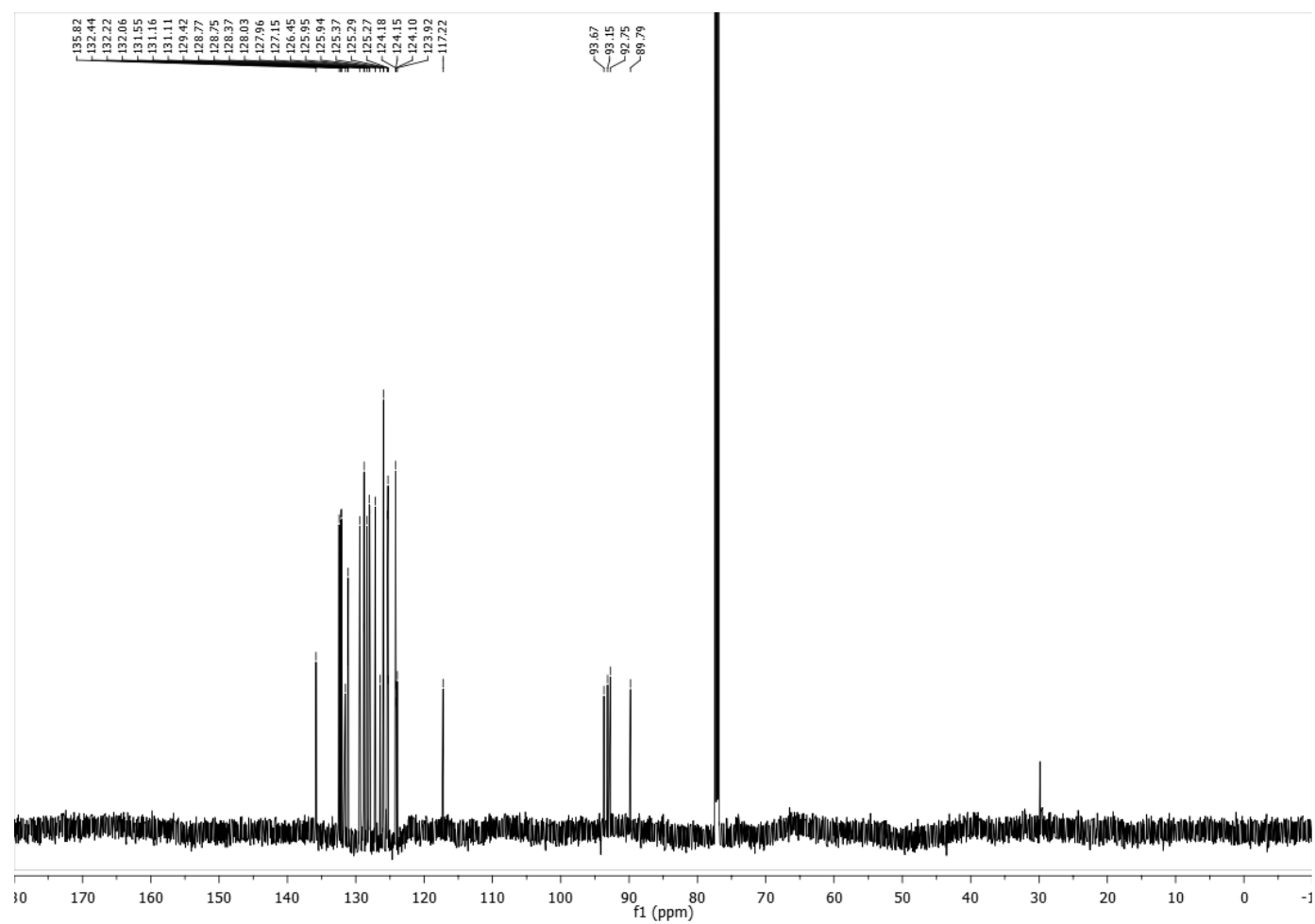
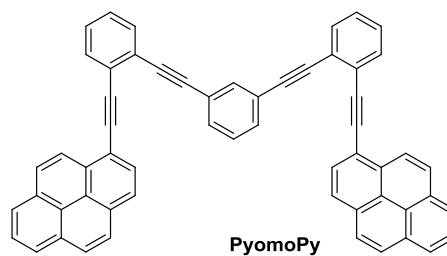
II



$^1\text{H-NMR}$ (CDCl_3 , 600 MHz)



^{13}C -NMR (CDCl_3 , 151 MHz)



Absorption and Steady State Fluorescence

Absorption spectra were recorded at 25 °C with a Perkin-Elmer Lambda 650 UV/vis spectrophotometer equipped with a Peltier temperature-controlled cell holder. Steady-state fluorescence emission spectra were collected on a JASCO FP-6500 spectrofluorometer equipped with a 450 W xenon lamp for excitation, with temperature controller ETC-273T at 25 °C. All experiments were performed using 5x10-mm quartz cuvettes.

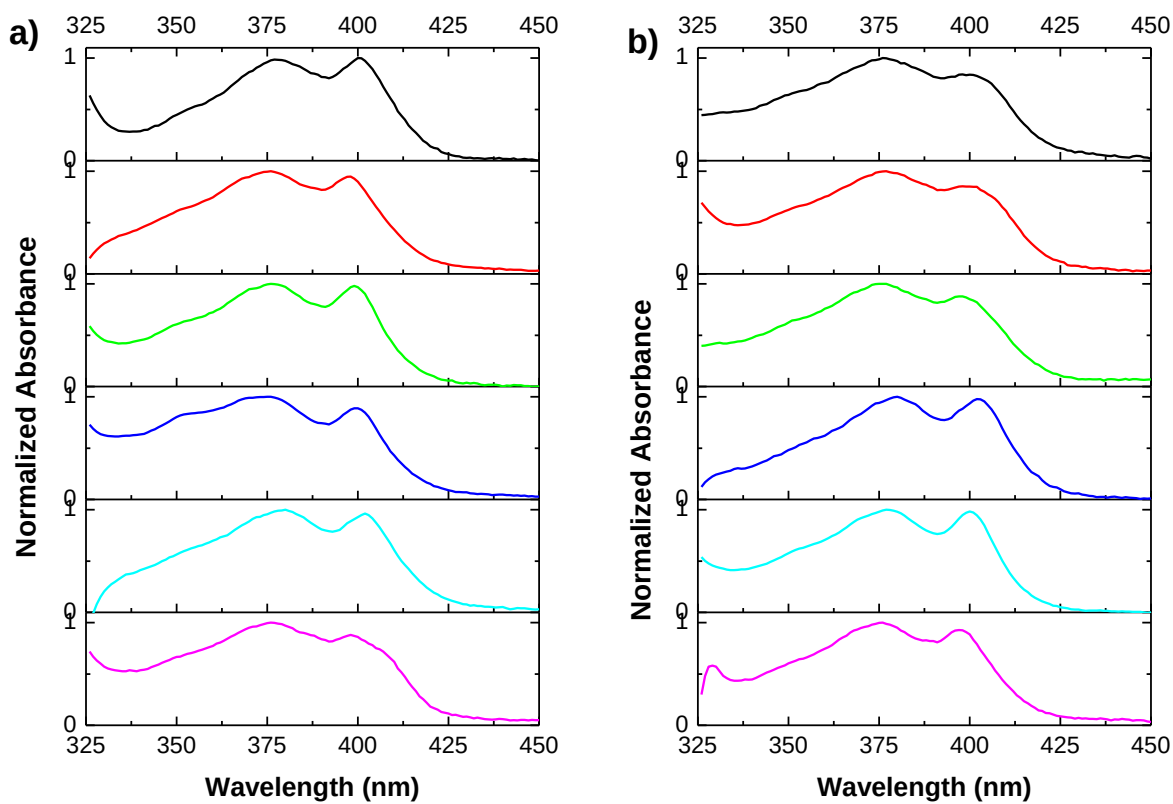


Figure S1. Normalized absorption spectra of **PyomoPy** in different solvents (2×10^{-6} M): (a) n-hexane (magenta), toluene (cyan), 1,4-dioxane (blue), THF (green), ethyl-acetate (red), CH_2Cl_2 (black). (b) acetone (magenta), DMF (cyan), DMSO (blue), acetonitrile (green), ethanol (red), methanol (black).

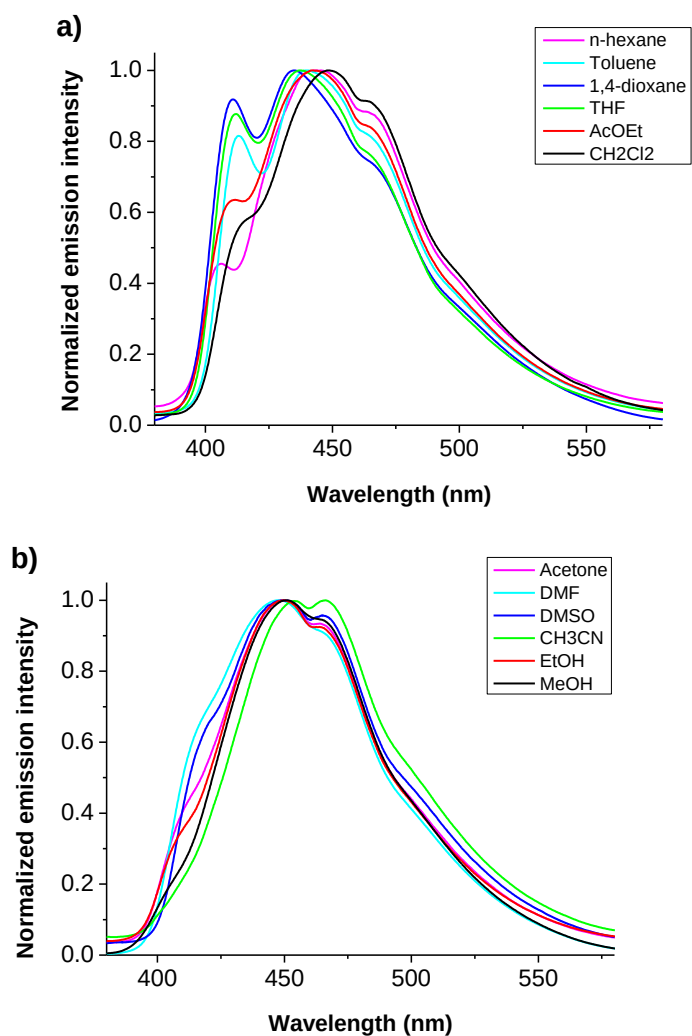


Figure S2. Normalized emission spectra of **PyomoPy** in different solvents (2×10^{-6} M): (a) *n*-hexane (magenta), toluene (cyan), 1,4-dioxane (blue), THF (green), ethyl acetate (red), CH_2Cl_2 (black). (b) acetone (magenta), DMF (cyan), DMSO (blue), acetonitrile (green), ethanol (red), methanol (black). $\lambda_{\text{ex}} = 375$ nm.

Computational Details

The emission spectra were theoretically modeled in the frame of Time Dependent Density Functional Theory (TD-DFT). To this aim, the equilibrium geometries of the S_1 states were optimized using the long-range corrected version of the hybrid functional B3LYP, at TD-CAMB3LYP/6-31G* level of theory.¹ The default grid values as implemented in Gaussian09 had been used during the DFT optimizations. The importance of dispersion for the modeling of the structure of these compounds was evaluated by reoptimizing the first singlet excited states of the species presenting the largest overlap between the pirene macrocycles after including Empirical Dispersion Correction (D3).² The comparison of the geometries calculated with and without including dispersion corrections revealed very small differences between the two optimized geometries.

Final vertical emission energies on TD-CAMB3LYP/6-31G* geometries were calculated with the more flexible triple- ξ quality basis set 6-311+G**.³ The effect of including the dispersion correction in the emission energies was also found to be very small, amounting at most to ca. 3kcal/mol. As expected, due to the important charge transfer character of the transitions the performance of standard pure and hybrid functionals, such as BLYP⁴ and B3LYP⁵, was found to be very poor, leading to important underestimation of the emission energies that amounted up to 1 eV. Finally, bulk n-hexane and methanol solvent effects were estimated by using Polarizable Continuum Model (PCM).⁶ All the calculations have been performed with Gaussian09 program suite.⁷

Table S1. Relative stabilities of the different conformers of PyomoPy taking the most stable one A (VI) as a reference.

	ΔE (kcal/mol)
D (I)	1.37
C (II)	0.67
B (III)	1.09
A (IV)	2.86
A (V)	0.62
A (VI)	0
A (VII)	1.51
A (VIII)	0.20
A (IX)	2.73

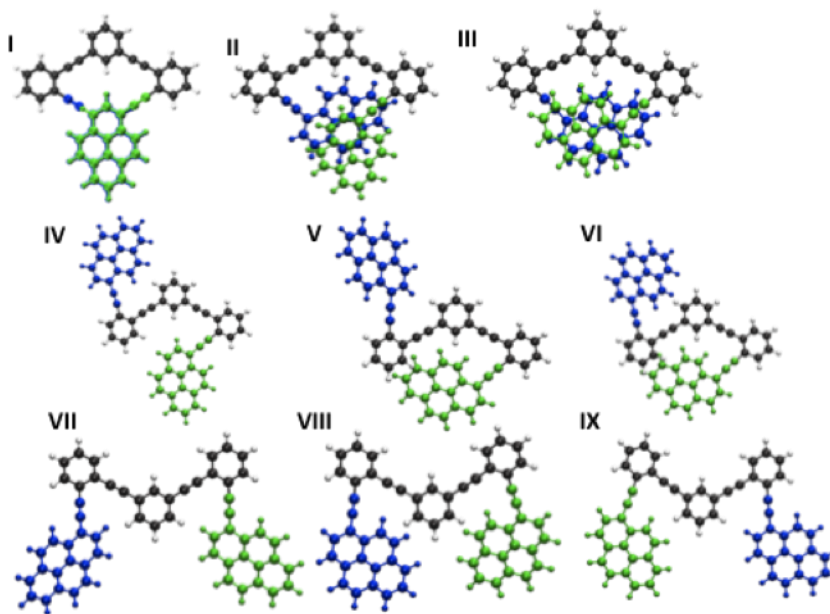


Table S2. TD-CAM-B3LYP/6-311+G** calculated emission energies (E_{emiss}) and oscillator strenghts (f) at the S_1 minimum for the different conformers of PyomoPy (conformer labeling as referred in Table S1).

	Gas Phase		Hexane		Methanol	
	E_{emiss} (nm)	f	E_{emiss} (nm)	f	E_{emiss} (nm)	f
D (I)	479.21	0.0725	483.98	0.1009	482.62	0.0976
C (II)	435.93	0.5337	445.82	0.7252	442.93	0.7065
B (III)	445.05	0.3643	452.74	0.5209	449.29	0.5015
A (IV)	433.06	0.9010	445.52	1.0530	442.77	1.0389
A (V)	435.43	0.5488	445.71	0.7544	443.04	0.7372
A (VI)	435.71	0.5255	445.83	0.7278	443.17	0.7102
A (VII)	425.86	1.1404	439.94	1.3212	437.35	1.3046
A (VIII)	430.59	0.6661	442.66	0.8521	440.12	0.8345
A (IX)	432.79	0.7076	445.14	0.8890	442.52	0.8717

Table S3. Cartesian coordinates (in Angstroms) of the optimized S₁ minimum for the different conformers of PyomoPy.

Atom	X	Y	Z
D (I)			
C	-6.411239	0.515181	-1.683331
C	-5.723993	-0.637372	-2.035029
C	-4.314585	-0.656936	-2.064754
C	-3.600544	0.522777	-1.712814
C	-4.313463	1.694820	-1.341284
C	-5.724072	1.664547	-1.326501
C	-3.580996	-1.820501	-2.415619
C	-2.178033	0.519431	-1.703910
C	-1.476039	-0.668078	-2.047550
C	-2.209809	-1.821171	-2.409781
C	-0.056861	-0.662276	-2.003909
H	0.479698	-1.570668	-2.259251
C	0.634892	0.468139	-1.647742
C	-0.034037	1.653888	-1.286907
C	-1.469056	1.690305	-1.322485
C	-2.204154	2.847523	-0.974290
C	-3.574871	2.850976	-0.979506
H	-4.117064	3.750653	-0.702950
H	-1.656806	3.739585	-0.689938
H	-4.127118	-2.717977	-2.691207
H	-7.496818	0.511025	-1.670199
H	-6.267327	-1.536639	-2.310076
H	-6.265853	2.563921	-1.048729
H	-1.663229	-2.721382	-2.675425
H	1.719380	0.460080	-1.624405
C	-6.411374	-0.515047	1.682989
C	-5.724217	-1.664463	1.326286
C	-4.313615	-1.694784	1.341174
C	-3.600685	-0.522748	1.712706
C	-4.314709	0.657005	2.064533
C	-5.724115	0.637498	2.034680
C	-3.575029	-2.850982	0.979500
C	-2.178176	-0.519452	1.703923
C	-1.469210	-1.690371	1.322623
C	-2.204315	-2.847574	0.974390
C	-0.034196	-1.654000	1.287175
C	0.634759	-0.468256	1.647976
C	-0.056985	0.662208	2.004024
C	-1.476159	0.668041	2.047559
C	-2.209922	1.821178	2.409689
C	-3.581106	1.820558	2.415417
H	-4.127225	2.718061	2.690923
H	-1.663324	2.721375	2.675339
H	-4.117229	-3.750650	0.702931
H	-7.496951	-0.510855	1.669758
H	-6.266016	-2.563822	1.048501
H	-6.267436	1.536799	2.309638
H	-1.656972	-3.739668	0.690127
H	1.719248	-0.460241	1.624706
H	0.479583	1.570596	2.259351

C	0.713173	-2.786064	0.910594
C	1.373650	-3.754907	0.588040
C	2.140682	-4.888921	0.229956
C	3.526943	-4.778133	-0.049589
C	1.534187	-6.154292	0.153783
C	4.251496	-5.930655	-0.378532
C	2.268502	-7.278112	-0.177103
H	0.474028	-6.234407	0.368870
C	3.634163	-7.168431	-0.442135
H	5.311265	-5.833738	-0.588402
H	1.779611	-8.245885	-0.226381
H	4.213646	-8.048939	-0.699863
C	4.185592	-3.516695	-0.015364
C	4.766812	-2.456187	-0.009509
C	5.471291	-1.214728	-0.004869
C	4.778755	0.000081	-0.000016
C	6.873291	-1.204076	-0.006804
C	5.471451	1.214803	0.004628
H	3.694483	0.000170	0.000180
C	7.562413	-0.000092	-0.000465
H	7.408935	-2.146991	-0.011125
C	6.873447	1.203979	0.006104
H	8.647717	-0.000165	-0.000634
H	7.409197	2.146836	0.010274
C	4.767075	2.456319	0.009485
C	4.185706	3.516743	0.015489
C	3.527002	4.778156	0.049822
C	2.140737	4.888878	-0.229676
C	4.251513	5.930710	0.378749
C	1.534169	6.154213	-0.153416
C	3.634115	7.168447	0.442441
H	5.311293	5.833836	0.588584
C	2.268433	7.278062	0.177480
H	0.473997	6.234266	-0.368461
H	4.213563	8.048971	0.700194
H	1.779492	8.245806	0.226810
C	1.373719	3.754869	-0.587826
C	0.713303	2.785977	-0.910357

Atom	X	Y	Z
C (II)			
C	5.946741	5.242237	-2.242403
C	4.694545	5.728732	-1.891214
C	3.655672	4.855076	-1.560560
C	3.893695	3.455767	-1.586865
C	5.175777	2.966601	-1.946684
C	6.186759	3.875359	-2.270472
C	2.348097	5.326215	-1.193103
C	2.846643	2.544498	-1.252610
C	1.566794	3.043482	-0.894590
C	1.355616	4.464220	-0.876603
C	0.551782	2.138655	-0.569579
H	-0.428753	2.512280	-0.289077
C	0.784279	0.776694	-0.595666
C	2.039588	0.263607	-0.944190
C	3.089049	1.146843	-1.278503
C	4.396260	0.681029	-1.648254
C	5.386595	1.545693	-1.965142
H	6.369295	1.175656	-2.243839
H	4.568535	-0.389428	-1.668978
H	2.174151	6.398419	-1.176128
H	6.742641	5.935212	-2.496936
H	4.511191	6.799327	-1.871101
H	7.167696	3.498706	-2.546258
H	0.371216	4.831395	-0.601057
H	-0.005821	0.083832	-0.330963
C	1.773980	-3.566368	3.091623
C	0.646724	-3.373755	2.310496
C	-0.059499	-2.151902	2.356437
C	0.403635	-1.113918	3.209284
C	1.559446	-1.324158	4.013950
C	2.226939	-2.560917	3.936961
C	-1.219244	-1.922177	1.575414
C	-0.280553	0.130893	3.253685
C	-1.422680	0.337384	2.437678
C	-1.879813	-0.715623	1.620521
C	-2.085847	1.629267	2.466024
C	-1.597478	2.629428	3.363132
C	-0.510743	2.404631	4.151493
C	0.190901	1.159899	4.117274
C	1.331104	0.923512	4.911198
C	1.999729	-0.275199	4.864031
H	2.880225	-0.435197	5.479215
H	1.681597	1.715344	5.567239
H	-1.579923	-2.717541	0.931050
H	2.309917	-4.508813	3.039277
H	0.295680	-4.157669	1.646529
H	3.110338	-2.719074	4.548997
H	-2.768691	-0.557270	1.017245
H	-2.111404	3.584020	3.391210
H	-0.151787	3.182064	4.819554
C	-3.153430	1.907458	1.640130
C	-4.106217	2.161301	0.898068
C	-5.202677	2.498288	0.117992
C	-5.722451	1.625163	-0.898948

C	-5.859670	3.740510	0.339811
C	-6.875882	2.018610	-1.602285
C	-6.975919	4.093660	-0.374227
H	-5.458237	4.400329	1.101473
C	-7.496184	3.223608	-1.352268
H	-7.264295	1.349504	-2.362781
H	-7.462000	5.044806	-0.181807
H	-8.381117	3.504118	-1.914000
C	-5.104565	0.401896	-1.225702
C	-4.613774	-0.654460	-1.575271
C	-4.085666	-1.901448	-2.001082
C	-2.764677	-2.263905	-1.703537
C	-4.889183	-2.797771	-2.727924
C	-2.257130	-3.503046	-2.100766
H	-2.129236	-1.574342	-1.161256
C	-4.382685	-4.024057	-3.127734
H	-5.909922	-2.518326	-2.964500
C	-3.078504	-4.386334	-2.816026
H	-5.012279	-4.709699	-3.685905
H	-2.683187	-5.348306	-3.122839
C	-0.916594	-3.876400	-1.773281
C	0.212580	-4.210002	-1.498276
C	1.528167	-4.666307	-1.185526
C	2.578748	-3.761425	-0.909909
C	1.785741	-6.042699	-1.152388
C	3.849681	-4.268978	-0.611957
C	3.049952	-6.525005	-0.854248
H	0.972641	-6.727760	-1.366974
C	4.085374	-5.634134	-0.583090
H	4.650239	-3.568623	-0.400141
H	3.228831	-7.595195	-0.834334
H	5.078393	-6.004837	-0.349746
C	2.371279	-2.351179	-0.934477
C	2.234871	-1.149541	-0.945369

Atom	X	Y	Z
B (III)			
C	4.407197	-1.208991	-2.287102
C	4.022437	-2.543585	-2.309052
C	2.666533	-2.908464	-2.204269
C	1.685771	-1.884274	-2.076470
C	2.087181	-0.521638	-2.062133
C	3.459799	-0.205966	-2.164257
C	2.241189	-4.262744	-2.216812
C	0.311218	-2.223438	-1.953633
C	-0.078815	-3.593104	-1.976272
C	0.909760	-4.587010	-2.111627
C	-1.466269	-3.914951	-1.860414
H	-1.758486	-4.960643	-1.885755
C	-2.409536	-2.945891	-1.706576
C	-2.052586	-1.563417	-1.649742
C	-0.657046	-1.198424	-1.804935
C	-0.237128	0.146758	-1.811960
C	1.091407	0.477865	-1.934797
H	1.396010	1.519999	-1.924969
H	-0.987336	0.924720	-1.711368
H	2.988356	-5.044702	-2.316785
H	5.458754	-0.949657	-2.359721
H	4.770199	-3.325637	-2.406785
H	3.761891	0.836672	-2.138623
H	0.601632	-5.628785	-2.128700
H	-3.456679	-3.206175	-1.600586
C	-4.441073	-1.066879	2.307757
C	-3.462970	-0.084634	2.213677
C	-2.112356	-0.430899	2.112523
C	-1.747560	-1.801198	2.094808
C	-2.752162	-2.799727	2.194069
C	-4.091753	-2.409274	2.301638
C	-1.075314	0.556656	2.015023
C	-0.376530	-2.175339	1.964056
C	0.621720	-1.174004	1.844396
C	0.225953	0.204305	1.887419
C	1.969541	-1.566476	1.679122
C	2.299234	-2.930086	1.666451
C	1.327216	-3.900142	1.798097
C	-0.021572	-3.549761	1.940551
C	-1.055566	-4.535091	2.055718
C	-2.357355	-4.178492	2.174588
H	-3.130874	-4.936875	2.258240
H	-0.769815	-5.583166	2.042151
H	-1.360750	1.604482	2.036031
H	-5.485386	-0.779477	2.379551
H	-3.741457	0.965005	2.211658
H	-4.859872	-3.174058	2.377030
H	1.000062	0.960663	1.810192
H	3.338322	-3.211751	1.535767
H	1.602944	-4.950438	1.779438
C	3.003643	-0.608054	1.490123
C	3.903866	0.173459	1.280276
C	5.014172	1.033338	1.045143
C	4.843383	2.385188	0.669341

C	6.314336	0.528051	1.184461
C	5.975351	3.182819	0.456369
C	7.421081	1.331420	0.965778
H	6.436642	-0.510416	1.472152
C	7.251913	2.665386	0.602096
H	5.832059	4.219124	0.170114
H	8.418163	0.918589	1.081532
H	8.115033	3.300602	0.431376
C	3.549474	2.960840	0.492354
C	2.490909	3.519172	0.317739
C	1.266850	4.231043	0.118339
C	0.060562	3.548869	-0.058191
C	1.275940	5.633421	0.098614
C	-1.136252	4.252763	-0.250399
H	0.048087	2.465541	-0.038775
C	0.091556	6.333379	-0.088778
H	2.214212	6.159729	0.234325
C	-1.106325	5.658208	-0.260642
H	0.104028	7.418660	-0.100482
H	-2.031824	6.204083	-0.407129
C	-2.370524	3.569726	-0.424865
C	-3.453568	3.037741	-0.570163
C	-4.747402	2.487247	-0.693704
C	-4.960010	1.093879	-0.961888
C	-5.866216	3.321395	-0.518970
C	-6.296095	0.611641	-0.999848
C	-7.151525	2.824908	-0.578088
H	-5.694175	4.374662	-0.323563
C	-7.364052	1.454136	-0.814571
H	-6.451174	-0.444707	-1.191416
H	-7.996586	3.490389	-0.435423
H	-8.375200	1.061573	-0.855351
C	-3.911324	0.211423	-1.193047
C	-3.028101	-0.616080	-1.426341

Atom	X	Y	Z
A (IV)			
C	-3.253865	-0.463039	0.000000
C	-4.154878	0.613283	0.000000
C	-3.680051	1.917252	0.000000
C	-2.317511	2.171821	0.000000
C	-1.397783	1.109969	0.000000
C	-1.883045	-0.206638	0.000000
H	-5.220402	0.412385	0.000000
H	-4.382866	2.744857	0.000000
H	-1.945520	3.188953	0.000000
H	-1.183219	-1.034087	0.000000
C	0.000000	1.350538	0.000000
C	1.204825	1.518576	0.000000
C	-3.743838	-1.805140	0.000000
C	-4.178610	-2.932606	0.000000
C	2.610485	1.592524	0.000000
C	3.323560	2.840832	0.000000
C	3.349853	0.394477	0.000000
C	4.746874	2.803411	0.000000
C	4.727056	0.394747	0.000000
H	2.800204	-0.540985	0.000000
C	5.430562	1.615508	0.000000
H	5.279617	3.748169	0.000000
H	5.269714	-0.545008	0.000000
H	6.515800	1.615845	0.000000
C	-4.758040	-4.235778	0.000000
C	-6.152520	-4.363938	0.000000
C	-3.960104	-5.402302	0.000000
C	-6.754693	-5.611546	0.000000
H	-6.755561	-3.462529	0.000000
C	-4.587816	-6.654535	0.000000
C	-5.968923	-6.761145	0.000000
H	-7.837164	-5.688472	0.000000
H	-3.969248	-7.545471	0.000000
H	-6.434024	-7.741669	0.000000
C	2.691198	4.074892	0.000000
C	2.228183	5.219245	0.000000
C	-2.536734	-5.330554	0.000000
C	-1.327287	-5.314375	0.000000
C	1.753100	6.512330	0.000000
C	0.333008	6.810074	0.000000
C	2.681065	7.600371	0.000000
C	-0.087183	8.165529	0.000000
C	-0.640155	5.791584	0.000000
C	2.258153	8.893962	0.000000
H	3.739895	7.366080	0.000000
C	-1.474443	8.475530	0.000000
C	0.867694	9.222006	0.000000
C	-1.982001	6.088534	0.000000
H	-0.303495	4.759241	0.000000
H	2.981147	9.704346	0.000000
C	-2.438432	7.430578	0.000000
C	-1.904161	9.833554	0.000000
C	0.416584	10.558022	0.000000
H	-2.719523	5.291130	0.000000

C	-3.810908	7.761389	0.000000
C	-3.282623	10.112257	0.000000
C	-0.921929	10.860845	0.000000
H	1.154150	11.355686	0.000000
C	-4.218550	9.084405	0.000000
H	-4.544741	6.960543	0.000000
H	-3.609858	11.147966	0.000000
H	-1.249092	11.896367	0.000000
H	-5.277737	9.322338	0.000000
C	0.097471	-5.273392	0.000000
C	0.860439	-6.462326	0.000000
C	0.748747	-4.033309	0.000000
C	2.276484	-6.374015	0.000000
C	0.252307	-7.763213	0.000000
C	2.127682	-3.951892	0.000000
H	0.147203	-3.130866	0.000000
C	3.065623	-7.563808	0.000000
C	2.915509	-5.106850	0.000000
C	1.001267	-8.889175	0.000000
H	-0.830450	-7.818979	0.000000
H	2.613175	-2.980148	0.000000
C	2.436669	-8.835117	0.000000
C	4.482627	-7.479210	0.000000
C	4.350561	-5.051961	0.000000
H	0.523786	-9.864958	0.000000
C	3.226999	-9.987497	0.000000
C	5.235236	-8.656284	0.000000
C	5.096701	-6.179609	0.000000
H	4.826271	-4.075302	0.000000
C	4.612143	-9.897053	0.000000
H	2.743425	-10.960202	0.000000
H	6.319541	-8.590400	0.000000
H	6.181523	-6.123453	0.000000
H	5.211942	-10.801785	0.000000

Atom	X	Y	Z
A (V)			
C	0.096693	-1.477392	-0.734202
C	-0.908745	-2.436603	-0.543274
C	-0.566324	-3.765960	-0.329828
C	0.762653	-4.156745	-0.296575
C	1.785846	-3.209371	-0.481507
C	1.436467	-1.870609	-0.707938
H	-1.948525	-2.129272	-0.571005
H	-1.347827	-4.505637	-0.187440
H	1.029151	-5.193917	-0.125605
H	2.214183	-1.134614	-0.872965
C	3.145684	-3.611718	-0.440617
C	4.298196	-3.999008	-0.403029
C	-0.252721	-0.109419	-0.952921
C	-0.582596	1.039304	-1.135008
C	5.601652	-4.531186	-0.394457
C	6.769713	-3.699451	-0.287577
C	5.772534	-5.923120	-0.515266
C	8.048003	-4.321971	-0.344530
C	7.025754	-6.494260	-0.554153
H	4.885038	-6.542646	-0.588690
C	8.173596	-5.681124	-0.473443
H	8.926064	-3.689173	-0.273114
H	7.127786	-7.569951	-0.653364
H	9.160213	-6.131882	-0.509570
C	-0.982138	2.387841	-1.369909
C	-0.048841	3.331752	-1.813630
C	-2.324387	2.785868	-1.165282
C	-0.428080	4.642553	-2.059670
H	0.979160	3.019800	-1.960251
C	-2.688034	4.111883	-1.423712
C	-1.750745	5.032220	-1.867593
H	0.309906	5.360340	-2.402680
H	-3.718754	4.409905	-1.264890
H	-2.052042	6.056647	-2.061657
C	6.693657	-2.326206	-0.110709
C	6.699323	-1.105750	0.071572
C	-3.291633	1.852707	-0.689598
C	-4.097838	1.054740	-0.270413
C	6.721060	0.256383	0.275351
C	5.501052	1.013336	0.489280
C	7.968703	0.954140	0.274113
C	5.583827	2.419206	0.663158
C	4.237930	0.390707	0.535304
C	8.027239	2.302921	0.448050
H	8.874611	0.377646	0.122356
C	4.398169	3.177434	0.862720
C	6.844515	3.081024	0.642307
C	3.088613	1.120973	0.728083
H	4.187227	-0.687499	0.418326
H	8.986095	2.812895	0.438445
C	3.133888	2.528154	0.893417
C	4.471621	4.589616	1.032308
C	6.890596	4.479257	0.815793
H	2.123838	0.624073	0.755776

C	1.967034	3.300083	1.085753
C	3.280676	5.315858	1.221048
C	5.746617	5.215268	1.003055
H	7.857757	4.973902	0.798626
C	2.048603	4.673981	1.244461
H	1.003656	2.800003	1.104643
H	3.337545	6.393059	1.349338
H	5.803786	6.291968	1.133016
H	1.141411	5.252913	1.386162
C	-5.002885	0.083138	0.250009
C	-6.391012	0.168242	0.008840
C	-4.495398	-0.973829	1.017662
C	-7.247547	-0.823821	0.552518
C	-6.969078	1.225227	-0.773022
C	-5.332980	-1.938180	1.545290
H	-3.426640	-1.019389	1.198242
C	-8.654417	-0.755993	0.319589
C	-6.713284	-1.885773	1.327845
C	-8.301924	1.287141	-0.992824
H	-6.303947	1.975369	-1.186304
H	-4.919871	-2.747440	2.140324
C	-9.197461	0.300406	-0.455709
C	-9.514838	-1.745822	0.862650
C	-7.608030	-2.872656	1.865010
H	-8.722378	2.093216	-1.587454
C	-10.577025	0.348104	-0.672353
C	-10.888417	-1.660210	0.622086
C	-8.940429	-2.806563	1.644366
H	-7.185837	-3.679309	2.457446
C	-11.412492	-0.623232	-0.137968
H	-10.991199	1.157594	-1.266655
H	-11.546195	-2.418022	1.038108
H	-9.605784	-3.559685	2.056813
H	-12.482126	-0.571865	-0.315486

Atom	X	Y	Z
A (VI)			
C	0.439464	1.142275	-1.330188
C	1.521188	2.032895	-1.384404
C	1.299166	3.399932	-1.266891
C	0.017201	3.896256	-1.095290
C	-1.081559	3.019301	-1.038958
C	-0.854785	1.641523	-1.164062
H	2.525155	1.646847	-1.516130
H	2.140166	4.084737	-1.312364
H	-0.154536	4.963025	-1.002793
H	-1.693593	0.956192	-1.140259
C	-2.393136	3.530737	-0.864755
C	-3.501773	4.012189	-0.726543
C	0.650512	-0.266903	-1.447008
C	0.823234	-1.459656	-1.550026
C	-4.753186	4.647463	-0.616085
C	-5.959050	3.921828	-0.321002
C	-4.835598	6.036901	-0.826665
C	-7.187562	4.639247	-0.291933
C	-6.040709	6.702670	-0.776730
H	-3.919504	6.576469	-1.042290
C	-7.228233	5.991792	-0.512228
H	-8.095591	4.085495	-0.078947
H	-6.075261	7.773572	-0.947994
H	-8.177577	6.516701	-0.478901
C	0.952049	-2.870928	-1.717018
C	-0.184625	-3.628536	-2.026674
C	2.199566	-3.522627	-1.584449
C	-0.097866	-4.998898	-2.213272
H	-1.138498	-3.120907	-2.116472
C	2.267108	-4.907474	-1.781472
C	1.132621	-5.639440	-2.094071
H	-0.991135	-5.567075	-2.451120
H	3.228378	-5.399303	-1.680158
H	1.207397	-6.712027	-2.242021
C	-5.963009	2.562805	-0.045873
C	-6.036191	1.362503	0.230952
C	3.382379	-2.803971	-1.243411
C	4.406747	-2.237542	-0.938760
C	-6.133066	0.024168	0.542161
C	-4.953670	-0.805315	0.709247
C	-7.420619	-0.575365	0.703596
C	-5.116757	-2.184463	1.000683
C	-3.651716	-0.278042	0.598389
C	-7.555777	-1.900097	0.985989
H	-8.294705	0.055575	0.585589
C	-3.972727	-3.012868	1.160469
C	-6.416658	-2.749152	1.138730
C	-2.542486	-1.076449	0.752037
H	-3.537914	0.781729	0.391616
H	-8.544321	-2.335736	1.098187
C	-2.669077	-2.460067	1.034314
C	-4.127658	-4.399152	1.447934
C	-6.543001	-4.123710	1.424966
H	-1.547487	-0.652234	0.658657

C	-1.545717	-3.300993	1.192288
C	-2.977290	-5.196629	1.596462
C	-5.440055	-4.927886	1.573820
H	-7.539495	-4.544139	1.527959
C	-1.706930	-4.649037	1.468064
H	-0.552334	-2.875202	1.090230
H	-3.096305	-6.253969	1.815363
H	-5.559083	-5.984844	1.793178
H	-0.832807	-5.282374	1.581606
C	5.618824	-1.576385	-0.583743
C	5.598920	-0.309143	0.038980
C	6.845715	-2.195057	-0.857162
C	6.828671	0.316528	0.369207
C	4.372454	0.369452	0.351898
C	8.041295	-1.583088	-0.532592
H	6.843306	-3.168838	-1.334697
C	6.829705	1.599458	0.995587
C	8.060489	-0.326111	0.079964
C	4.376189	1.585467	0.945636
H	3.437706	-0.120953	0.101273
H	8.981482	-2.079152	-0.755158
C	5.603283	2.248473	1.290142
C	8.057915	2.229954	1.325456
C	9.287555	0.335418	0.425780
H	3.438160	2.084716	1.171487
C	5.626132	3.503958	1.903338
C	8.035522	3.485911	1.937157
C	9.287053	1.551217	1.017761
H	10.223047	-0.167659	0.198467
C	6.831081	4.114997	2.222510
H	4.685941	3.998601	2.130096
H	8.975775	3.967974	2.189389
H	10.222535	2.040391	1.273849
H	6.831797	5.090392	2.698865

Atom	X	Y	Z
A (VII)			
C	-1.241857	-2.379434	0.234512
C	-1.257072	-0.978440	0.155522
C	-0.098998	-0.288795	-0.177470
C	1.080536	-0.972021	-0.431785
C	1.116701	-2.374900	-0.352660
C	-0.055082	-3.068086	-0.020296
H	-2.179923	-0.446368	0.359780
H	-0.117611	0.794666	-0.240659
H	1.982568	-0.435256	-0.704983
H	-0.037516	-4.149898	0.042038
C	2.324448	-3.076244	-0.609207
C	3.366832	-3.656207	-0.837463
C	-2.434906	-3.087161	0.576139
C	-3.462217	-3.651996	0.868403
C	4.580693	-4.310161	-1.130094
C	5.827412	-3.592581	-1.094663
C	4.589535	-5.667318	-1.488475
C	7.013780	-4.281416	-1.464206
C	5.762721	-6.310824	-1.827076
H	3.645498	-6.201492	-1.505613
C	6.981360	-5.606760	-1.820179
H	7.951286	-3.735826	-1.446702
H	5.743592	-7.359178	-2.106225
H	7.900636	-6.115031	-2.093265
C	-4.676943	-4.307140	1.226027
C	-4.653812	-5.617903	1.714940
C	-5.917290	-3.638058	1.103063
C	-5.826061	-6.259725	2.082180
H	-3.698180	-6.122749	1.805247
C	-7.089686	-4.301785	1.481640
C	-7.046550	-5.599190	1.967113
H	-5.788649	-7.276160	2.460530
H	-8.037053	-3.782840	1.383590
H	-7.966555	-6.097945	2.254528
C	5.878784	-2.268788	-0.684154
C	5.904036	-1.101695	-0.284253
C	-5.974336	-2.309358	0.590652
C	-6.006746	-1.186694	0.141940
C	5.896300	0.194835	0.180695
C	7.090241	1.018699	0.165288
C	4.684538	0.749268	0.702212
C	7.018705	2.344458	0.663810
C	8.319374	0.542208	-0.333978
C	4.636570	2.024465	1.176113
H	3.803860	0.116743	0.724602
C	8.178707	3.165830	0.656821
C	5.792422	2.863417	1.173702
C	9.442936	1.333052	-0.342371
H	8.363938	-0.472235	-0.716129
H	3.708197	2.424071	1.573666
C	9.407942	2.661114	0.150209
C	8.114541	4.498317	1.157667
C	5.756126	4.185081	1.663645
H	10.378985	0.943196	-0.731494

C	10.548043	3.496183	0.153943
C	9.275757	5.289540	1.139981
C	6.874234	4.980773	1.658945
H	4.816921	4.570141	2.050981
C	10.474830	4.787871	0.642698
H	11.485087	3.107152	-0.233770
H	9.226347	6.305162	1.521943
H	6.825135	5.996219	2.040822
H	11.359431	5.417185	0.638949
C	-5.990189	0.126415	-0.412922
C	-7.050903	1.031451	-0.191877
C	-4.894584	0.518758	-1.194200
C	-6.984233	2.326286	-0.768350
C	-8.197681	0.687942	0.601220
C	-4.833836	1.780596	-1.754058
H	-4.091521	-0.191940	-1.358295
C	-8.046032	3.256546	-0.555539
C	-5.865939	2.703510	-1.556645
C	-9.202317	1.570778	0.801793
H	-8.235738	-0.303637	1.038498
H	-3.977455	2.063743	-2.359132
C	-9.167116	2.889119	0.232047
C	-7.983399	4.552731	-1.130990
C	-5.830290	4.021123	-2.126831
H	-10.062473	1.294590	1.405097
C	-10.195383	3.814679	0.428611
C	-9.033483	5.447388	-0.909505
C	-6.837020	4.901161	-1.924894
H	-4.969101	4.296007	-2.729107
C	-10.127436	5.080466	-0.137286
H	-11.053105	3.531795	1.032207
H	-8.984951	6.439057	-1.350306
H	-6.797287	5.894651	-2.362400
H	-10.934610	5.787899	0.024976

Atom	X	Y	Z
A (VIII)			
C	-1.264843	1.974626	-0.504057
C	-1.227313	0.573272	-0.560472
C	-0.065256	-0.103315	-0.211133
C	1.066460	0.591903	0.187371
C	1.051540	1.996545	0.240451
C	-0.125619	2.675720	-0.105496
H	-2.111223	0.028427	-0.871019
H	-0.043954	-1.188259	-0.248472
H	1.969062	0.059479	0.464890
H	-0.149772	3.758359	-0.062542
C	2.200127	2.725013	0.646117
C	3.165587	3.369708	1.008602
C	-2.454976	2.684771	-0.855328
C	-3.457795	3.289878	-1.154244
C	4.226415	4.169175	1.476806
C	5.563505	3.657933	1.613593
C	3.969155	5.499315	1.856165
C	6.557654	4.509806	2.169918
C	4.960718	6.303517	2.375302
H	2.959066	5.878089	1.741947
C	6.265205	5.797676	2.539275
H	7.561965	4.115862	2.282404
H	4.733523	7.324080	2.665402
H	7.042844	6.430577	2.954727
C	-4.595780	4.059953	-1.539028
C	-4.418883	5.389092	-1.942648
C	-5.897779	3.510566	-1.532178
C	-5.497340	6.163746	-2.337780
H	-3.415833	5.801521	-1.943949
C	-6.974946	4.308696	-1.937820
C	-6.779686	5.620652	-2.337156
H	-5.338404	7.191330	-2.648361
H	-7.970559	3.878754	-1.929574
H	-7.628392	6.222089	-2.646573
C	5.914272	2.381934	1.198988
C	6.289759	1.269426	0.819135
C	-6.136561	2.169054	-1.114431
C	-6.385097	1.040702	-0.756628
C	6.711024	0.030800	0.388283
C	5.839236	-0.839530	-0.380025
C	8.031434	-0.421148	0.697285
C	6.314779	-2.115975	-0.776717
C	4.538071	-0.447841	-0.752797
C	8.469131	-1.649752	0.307433
H	8.677524	0.240947	1.263198
C	5.473808	-2.981657	-1.527909
C	7.631197	-2.538526	-0.434353
C	3.722310	-1.282596	-1.479767
H	4.187627	0.536848	-0.459049
H	9.473823	-1.976139	0.559550
C	4.161912	-2.567205	-1.886769
C	5.942811	-4.266431	-1.926019
C	8.071262	-3.814919	-0.840235
H	2.722700	-0.957219	-1.753181

C	3.344095	-3.443103	-2.633777
C	5.089816	-5.103566	-2.667835
C	7.259342	-4.656329	-1.559663
H	9.076638	-4.127953	-0.572680
C	3.808245	-4.690784	-3.013924
H	2.343705	-3.121424	-2.908582
H	5.447853	-6.083240	-2.970877
H	7.617273	-5.636152	-1.861691
H	3.167335	-5.352662	-3.588068
C	-6.686389	-0.289729	-0.343078
C	-5.769491	-1.035829	0.429551
C	-7.910493	-0.863365	-0.711312
C	-6.106945	-2.359983	0.811535
C	-4.504182	-0.497885	0.844591
C	-8.235983	-2.152643	-0.335997
H	-8.603867	-0.276030	-1.303408
C	-5.192054	-3.132130	1.589315
C	-7.350668	-2.923253	0.424241
C	-3.639178	-1.231769	1.582407
H	-4.259434	0.518206	0.553565
H	-9.189603	-2.579021	-0.633380
C	-3.947687	-2.576544	1.982984
C	-5.524515	-4.458231	1.971437
C	-7.660339	-4.266942	0.826001
H	-2.685486	-0.808482	1.884614
C	-3.066232	-3.349017	2.744121
C	-4.613810	-5.196100	2.731846
C	-6.792099	-4.997735	1.561336
H	-8.615788	-4.685491	0.522862
C	-3.397646	-4.645494	3.113545
H	-2.114500	-2.921287	3.046285
H	-4.868882	-6.210848	3.023977
H	-7.039032	-6.013011	1.858453
H	-2.701940	-5.232260	3.705182

Atom	X	Y	Z
A (IX)			
C	2.184696	-0.411432	0.000000
C	0.785914	-0.520121	0.000000
C	0.000000	0.623672	0.000000
C	0.583578	1.881001	0.000000
C	1.982044	2.013682	0.000000
C	2.772076	0.854048	0.000000
H	0.331762	-1.504632	0.000000
H	-1.081922	0.532307	0.000000
H	-0.028678	2.774201	0.000000
H	3.852025	0.943932	0.000000
C	2.599570	3.290396	0.000000
C	3.169220	4.365370	0.000000
C	3.003735	-1.581823	0.000000
C	3.699029	-2.570135	0.000000
C	3.932965	5.547718	0.000000
C	3.338456	6.856739	0.000000
C	5.337365	5.444469	0.000000
C	4.201988	7.988571	0.000000
C	6.143025	6.561672	0.000000
H	5.775791	4.452292	0.000000
C	5.565287	7.846743	0.000000
H	3.748138	8.973665	0.000000
H	7.222395	6.450875	0.000000
H	6.200600	8.726631	0.000000
C	4.573304	-3.696513	0.000000
C	5.958236	-3.489152	0.000000
C	4.075538	-5.019570	0.000000
C	6.839647	-4.557822	0.000000
H	6.329430	-2.470145	0.000000
C	4.983083	-6.086629	0.000000
C	6.349825	-5.861442	0.000000
H	7.909364	-4.375234	0.000000
H	4.594322	-7.099178	0.000000
H	7.034930	-6.703098	0.000000
C	1.967508	7.066214	0.000000
C	0.768416	7.359321	0.000000
C	2.675909	-5.287610	0.000000
C	1.496464	-5.556616	0.000000
C	-0.560312	7.723943	0.000000
C	-1.626318	6.739828	0.000000
C	-0.908544	9.110779	0.000000
C	-2.974405	7.183361	0.000000
C	-1.360893	5.356237	0.000000
C	-2.207831	9.516372	0.000000
H	-0.103736	9.837641	0.000000
C	-4.031484	6.232953	0.000000
C	-3.281421	8.573943	0.000000
C	-2.381010	4.435546	0.000000
H	-0.324713	5.031479	0.000000
H	-2.448743	10.575338	0.000000
C	-3.739038	4.841698	0.000000
C	-5.387153	6.670030	0.000000
C	-4.631674	8.980943	0.000000
H	-2.159699	3.372230	0.000000

C	-4.804159	3.915242	0.000000
C	-6.413456	5.708512	0.000000
C	-5.654449	8.066029	0.000000
H	-4.853922	10.044382	0.000000
C	-6.118486	4.350036	0.000000
H	-4.576891	2.853062	0.000000
H	-7.447012	6.042476	0.000000
H	-6.687791	8.400049	0.000000
H	-6.926427	3.624951	0.000000
C	0.102074	-5.851191	0.000000
C	-0.361257	-7.185746	0.000000
C	-0.822362	-4.798627	0.000000
C	-1.758731	-7.431479	0.000000
C	0.534822	-8.307733	0.000000
C	-2.182216	-5.041499	0.000000
H	-0.447999	-3.780760	0.000000
C	-2.246919	-8.773142	0.000000
C	-2.677094	-6.349400	0.000000
C	0.070603	-9.577867	0.000000
H	1.600467	-8.108084	0.000000
H	-2.881405	-4.210379	0.000000
C	-1.337497	-9.861620	0.000000
C	-3.644353	-9.023007	0.000000
C	-4.085107	-6.632307	0.000000
H	0.763470	-10.414576	0.000000
C	-1.835720	-11.167155	0.000000
C	-4.100068	-10.343712	0.000000
C	-4.545911	-7.903608	0.000000
H	-4.776517	-5.794495	0.000000
C	-3.203489	-11.403860	0.000000
H	-1.137564	-11.999376	0.000000
H	-5.169626	-10.533702	0.000000
H	-5.613668	-8.103356	0.000000
H	-3.574500	-12.423977	0.000000

Time-resolved fluorescence of PyomoPy in different solvents

Time-resolved fluorescence decay traces were collected in single photon timing (SPT) mode on a FluoTime 200 fluorometer (PicoQuant, GmbH). The excitation source was a 405-nm pulsed diode laser (EPL405, Edinburgh Instruments, Ltd.) using a 5 MHz excitation frequency. The full width at half maximum (fwhm) of the laser pulses was around 90 ps. The fluorescence emission was collected at a 90° geometry, focused at the detector after crossing through a polarizer (set at the magic angle), 2-mm slits, and a 2-nm bandwidth monochromator. SPT was achieved by a TimeHarp200 board, set at 36 ps/channel. Fluorescence decay traces were collected for the necessary time to reach 20,000 counts at the peak channel. Time-resolved emission spectroscopy (TRES) of **PyomoPy** dissolved in different solvents was performed by collecting 51 fluorescence decay traces in the 420 - 570 nm emission range ($\Delta\lambda_{em} = 3$ nm) during a fixed amount of time to maintain the overall intensity information.

The fluorescence decay traces were fitted to a four-exponential function, by using a Levenberg-Marquard algorithm-based nonlinear least-squares error minimization deconvolution method iterative reconvolution methods (FluoFit 4.4 package, Picoquant GmbH). For each sample, the decay traces collected at different emission wavelengths were fitted globally with the decay times linked as shared parameters, whereas the pre-exponential factors were local adjustable parameters. The quality of fittings was assessed by the value of the reduced chi-squared, χ^2 , parameter and random distributions of the weighted residuals and the autocorrelation functions. Figure S3 shows four decay traces of PyomoPy dissolved in ethanol (Fig. S3a) and n-hexane (Fig. S3b) at different emission wavelengths. The insets show the weighted residuals for the 4-exponential function fit.

For the TRES analysis and the estimation of the species-associated emission spectra (SAEMS), the fitting procedure described above was performed, by fitting globally the 51 decay traces. The SAEMS of each species i at any given emission wavelength (SAEMS _{i} (λ_{em})) is given by the fluorescence intensity emitted by the species i ($A_{i,\lambda_{em}} \times \tau_i$), normalized by the total intensity and corrected for the different detection sensitivity using the total intensity of the steady-state spectrum ($I_{ss,\lambda_{em}}$):

$$SAEMS_i(\lambda_{em}) = \frac{A_{i,\lambda_{em}} \times \tau_i}{\sum_i A_{i,\lambda_{em}} \times \tau_i} \cdot I_{ss,\lambda_{em}} \quad (\text{eq. S1})$$

The approximate contribution of each species can be assessed as the area under the SAEMS. This estimation assumes equal excitation rate for all the species, as the initial amount of each form in the excited state (after the pulsed excitation) is unknown.

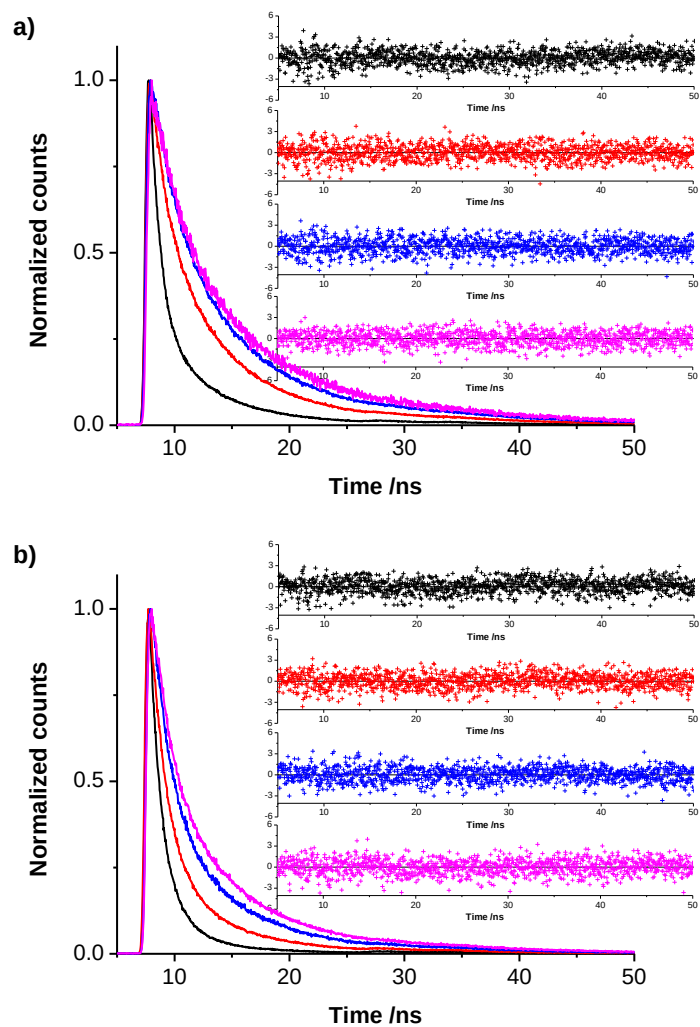
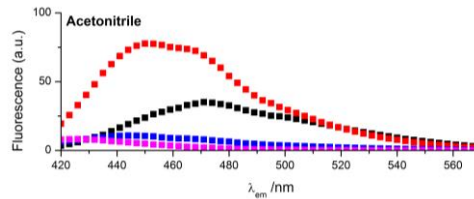
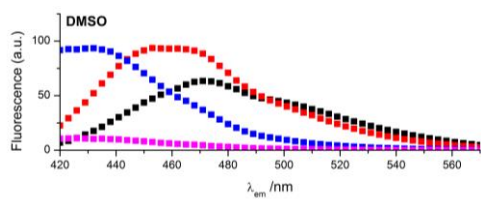
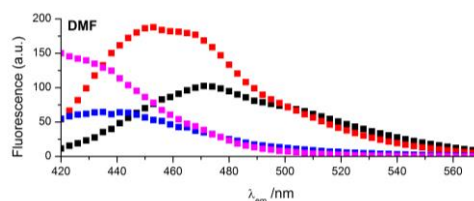
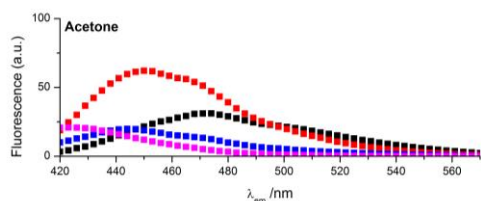
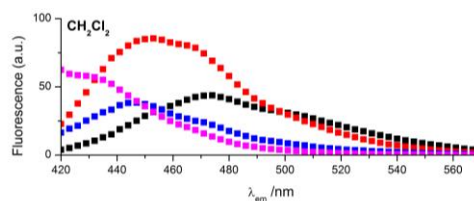
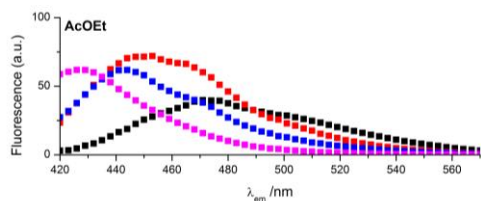
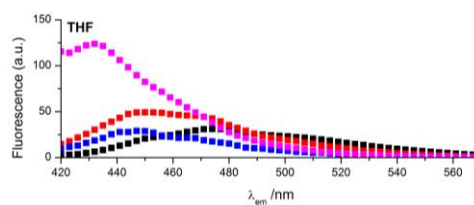
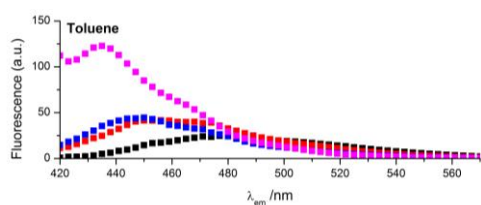


Figure S3. Fluorescence decay traces of **PyomoPy** dissolved in (a) ethanol and (b) n-hexane, collected at 420 (black), 460 (red), 500 (blue), and 560 nm (magenta) (extracted from the full TRES experiments). The insets show the corresponding weighted residuals from the 4-exponential function fits.

Table S3. Fluorescence decay times of **PyomoPy** in different solvents recovered from global analyses of 51 decay traces in the emission range 420 – 570 nm.^a

Solvent	τ_1 /ns	τ_2 /ns	τ_3 /ns	τ_4 /ns
n-hexane	0.54	1.55	5.58	9.00
Toluene	0.72	1.94	5.44	9.17
Tetrahydrofuran	0.83	1.62	4.72	9.94
AcOEt	0.63	1.84	4.95	8.97
Dichloromethane	0.59	1.88	5.73	12.12
Acetone	0.40	1.53	5.31	9.66
Dimethylformamide	0.85	1.57	7.36	13.40
Dimethylsulfoxide	0.19	1.86	9.03	16.89
Acetonitrile	0.37	1.46	6.51	10.46
Ethanol	0.63	2.11	5.19	10.49
Methanol	0.32	1.53	5.28	10.33
1,4-dioxane	1.00	2.77	8.68	14.71

^a The standard deviation from the fits were always <0.27 ns for each decay time, except for 1,4-dioxane, in which the standard deviation in τ_4 was 1.34 ns.



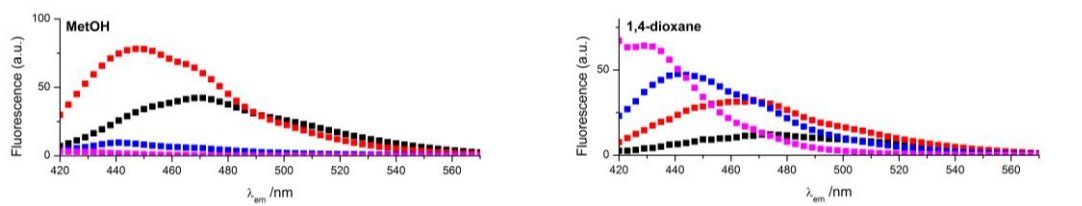


Figure S4. TRES deconvolution of the SAEMS associated to the shortest (τ_1 , magenta), short intermediate (τ_2 , blue), long intermediate (τ_3 , red), and the longest (τ_4 , black) decay times of PyomoPy dissolved in the different solvents tested in this work. The corresponding SAEMS on n-hexane and ethanol are displayed in Figure 2 of the main text.

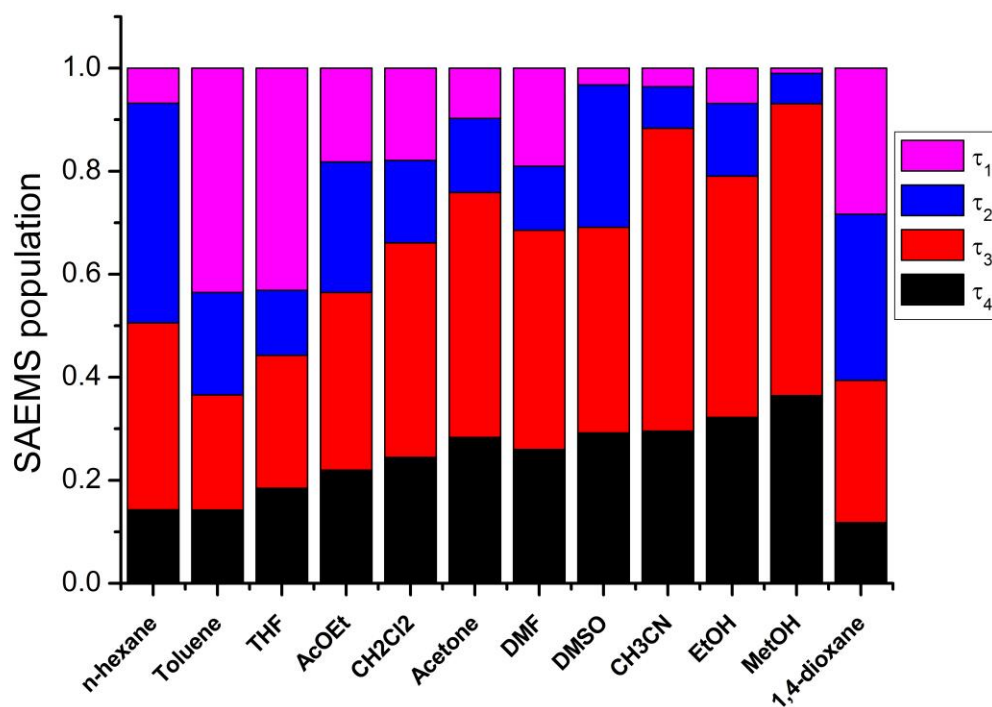


Figure S5. Relative population of the SAEMS associated to the shortest (τ_1 , magenta), short intermediate (τ_2 , blue), long intermediate (τ_3 , red), and the longest (τ_4 , black) decay times of PyomoPy dissolved in the different solvents tested in this work.

Time-resolved fluorescence of PyomoPy as a function of the media viscosity

The viscosity values of the pure solvents were taken from CRC Handbook of Chemistry and Physics.⁸ Ethanol : glycerine and methanol : glycerine mixtures were prepared by mixing different proportions of each solvent within the ratio 19:1 to 1:2 for methanol : glycerine and 9:1 to 3:7 for ethanol : glycerine mixtures. The viscosity of these mixtures was measured with a falling ball viscometer (Viscoba, Fungilab). In this system the time a ball (of known size and density) requires to traverse a definite distance within the fluid is measured, and Stokes' law is used to calculate the viscosity of that fluid. Measurements were made in triplicate. The temperature of the liquid mixtures was controlled at 25 °C during the viscosity measurements.

Time-resolved fluorescence decay traces were collected and analyzed using the same instrumentation and procedures described above. Fluorescence decay traces of **PyomoPy** dissolved in ethanol:glycerine mixtures were collected in the 420-580 nm emission range ($\Delta\lambda_{em} = 40$ nm, 5 traces). For methanol:glycerine mixtures, the same range was used but with $\Delta\lambda_{em} = 20$ nm (9 traces).

Table S4. Fluorescence decay times of PyomoPy in EtOH:glycerine mixtures (between 9:1 and 3:7) of different viscosities, recovered from global analyses of 5 decay traces in the emission range 420 – 580 nm.^a

Viscosity /mPa·s	τ_1 /ns	τ_2 /ns	τ_3 /ns	τ_4 /ns
1.667	0.71	3.58	8.21	14.18
2.969	0.93	3.86	8.94	15.77
5.616	1.07	3.87	9.00	16.58
11.677	0.18	1.73	6.55	15.77
23.665	0.20	1.77	6.72	15.96
42.652	0.26	2.22	7.77	17.09
75.646	0.29	2.38	8.35	17.50

^a The standard deviation from the fits were always <1.37 ns for τ_4 .

Table S5. Fluorescence decay times of PyomoPy in MetOH:glycerine mixtures (between 19:1 and 1:2) of different viscosities, recovered from global analyses of 9 decay traces in the emission range 420 – 580 nm.^a

Viscosity /mPa·s	τ_1 /ns	τ_2 /ns	τ_3 /ns	τ_4 /ns
0.751	--	1.58	6.19	11.89
0.961	--	1.57	6.57	12.76
1.244	--	1.57	6.82	13.37
1.663	--	1.54	7.14	14.38
2.089	--	1.56	7.44	15.15
2.684	--	1.57	7.57	15.50
4.067	0.65	2.04	7.61	16.11
7.940	0.40	1.86	7.82	16.91
12.660	0.35	1.99	8.10	17.19
17.250	0.42	2.17	8.32	17.34
44.320	0.38	1.79	8.13	17.10

^a The standard deviation from the fits were always <0.30 ns for τ_4 .

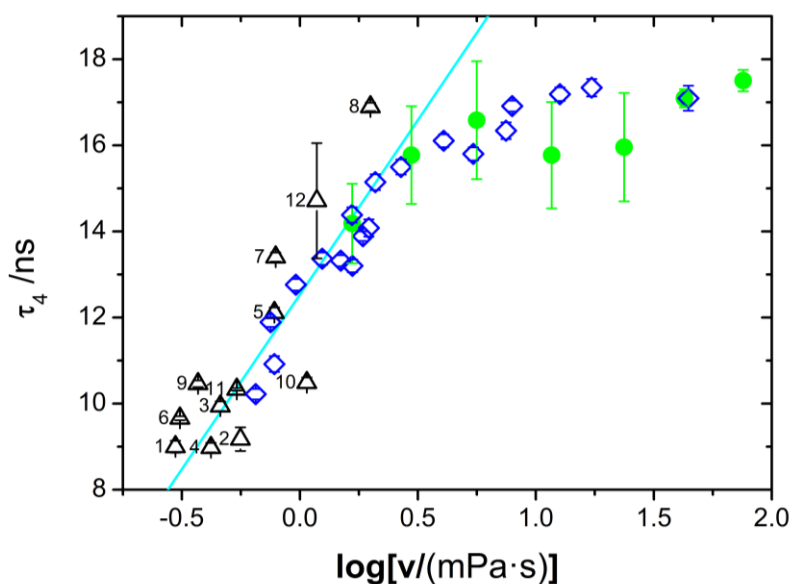


Figure S6. Longest decay time (τ_4) of **PyomoPy** dissolved in pure solvents (black triangles), MeOH:glycerine mixtures (blue diamonds), or EtOH:glycerine mixtures (green circles) as a function of $\log(\text{viscosity})$. Pure solvents: 1. n-hexane, 2. Toluene, 3. THF, 4. AcOEt, 5. CH_2Cl_2 , 6. Acetone, 7. Dimethylformamide, 8. Dimethylsulfoxide, 9. CH_3CN , 10. Ethanol, 11. Methanol, 12. 1,4- Dioxane. MeOH:glycerine mixtures between 19:1 and 1:2. EtOH:glycerine mixtures between 9:1 and 3:7. The blue line is the linear fit of the datapoints in the linear region as detailed in the main text.

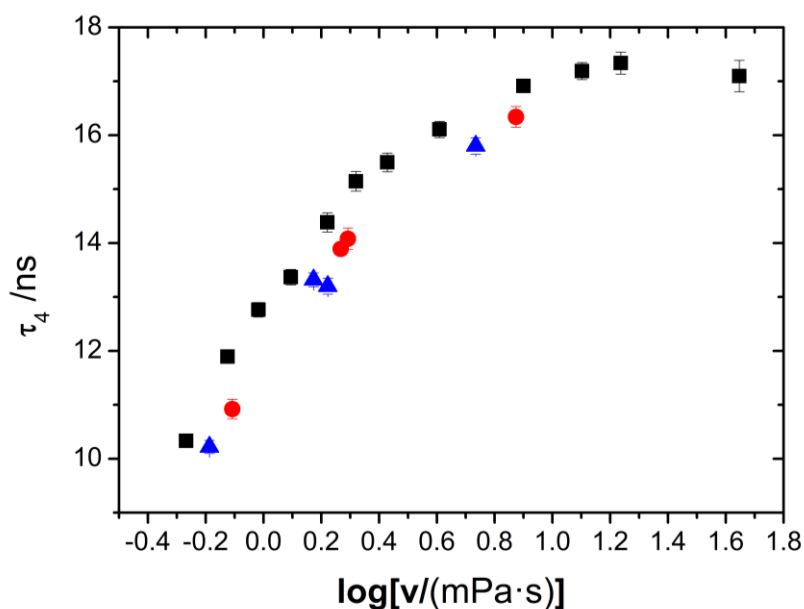


Figure S7. Longest decay time (τ_4) of **PyomoPy** dissolved in MeOH:glycerine mixtures at 20 °C (black squares), 30 °C (red circles), and 40 °C (blue triangles), as a function of $\log(\text{viscosity})$. At higher temperatures, the decay time is slightly smaller due to the enhancement of non-radiative deactivation processes.

References

- [1] Yanai, D.P. Tew, N.C. Handy *Chem. Phys. Lett.* 2004, **39**, 51–57; P. C. Hariharan, J. A. Pople *Theor. Chim. Acta* 1973, **28**, 213.
- [2] S. Grimme, J. Antony, S. Ehrlich, H. Krieg *J. Chem. Phys.* 2010, **132**, 154104.
- [3] R. Krishnan, J.S. Binkley, R. Seeger and J.A. Pople *J. Chem. Phys.* 1980, **72**, 650.
- [4] A.D. Becke, *Phys. Rev. A* 1988, **38**, 3098–3100; C.T. Lee, W.T. Yang, R.G. Parr *Phys. Rev. B* 1988, **37**, 785–789.
- [5] A.D. Becke *J. Chem. Phys.* 1993, **98**, 5648–5652.
- [6] J. Tomasi, B. Mennucci, R. *Chem. Rev.* 2005, **105**, 2999–3094.
- [7] Gaussian 09, Revision A.1, M. J. Frisch, et al., Gaussian, Inc., Wallingford CT, 2009.
- [8] "Viscosity of Liquids", in CRC Handbook of Chemistry and Physics, Internet Version 2005, David R. Lide, ed., <<http://www.hbcpnetbase.com>>, CRC Press, Boca Raton, FL, 2005