

Revised on-2014-012209

Ether-substituted group 4 metallocene complexes: cytostatic effects and applications in ethylene polymerization §

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Supplementary Material

Table S1. Crystallographic data of 8 and 4a.

Compound	8	4a
Empirical formula	C ₂₂ H ₃₄ Cl ₂ O ₄ Zr	C ₁₄ H ₁₅ ClO ₂ Ti
Molecular weight (g/mol)	524.61	298.61
Temperature (K)	100(2)	293(2)
Wavelength (Å)	0.71069	0.71073
Crystal system	Monoclinic	Monoclinic
Space group	C2/c	P2 ₁ /n
a (pm)	2557.7(5)	650.00(2)
b (pm)	666.2(2)	1486.00(5)
c (pm)	1400.1(3)	1397.90(5)
α (°)	90.00	90
β (°)	94.620(4)	101.328(4)
γ (°)	90.00	90
Volume (Å³)	2377.7(8)	1.32393(8)
Z	4	4
Calculated density (mg/m³)	1.465	1.498
Absorption coefficient μ (mm⁻¹)	0.712	0.838

F(000)	1088	616
Crystal size (mm)	$0.20 \times 0.14 \times 0.12$	$0.08 \times 0.06 \times 0.03$
θ range ($^{\circ}$)	2.92 a 20.89	2.02 a 26.37
hkl ranges	-31 $\leq h \leq$ 31 -5 $\leq k \leq$ 8 -15 $\leq l \leq$ 17	-8 $\leq h \leq$ 8 -18 $\leq k \leq$ 18 -17 $\leq l \leq$ 17
Collected reflections	6807	74651
Independent reflections	2424 [$R_{(int)} = 0.0519$]	2711 [$R_{(int)} = 0.1265$]
Completeness	99.5 % ($\theta = 20.89^{\circ}$)	100 % ($\theta = 26.37^{\circ}$)
Maximum and minimum transition	1.000 and 0.820	1 and 0.97588
Refinement method	Least squares on F^2	Least squares on F^2
Data / Restraints / Parameters	2424 / 0 / 136	2711 / 0 / 178
Goodness-of-fit on F^2	1.052	1.009
Final R indices [$I > 2\sigma(I)$]	$R_1 = 0.0531$ $wR_2 = 0.1307$	$R_1 = 0.0547$ $wR_2 = 0.1343$
Final R indices (all data)	$R_1 = 0.0683$ $wR_2 = 0.1409$	$R_1 = 0.1011$ $wR_2 = 0.1618$
Largest diffraction peak and hole ($e \cdot \text{\AA}^3$)	0.677 and -0.912	0.845 and -0.207

Table S2. Data obtained by X-ray diffraction for materials KIT-6 and K4–K7.

Material	(<i>hkl</i>)	2θ (°)	d _{hkl} (Å)	a ₀ (Å)
KIT-6	(211)	0.99	89	103.10
	(220)	1.65	-	
	(320)	1.91	46	
K4	(211)	1.01	87	100.71
	(220)	1.64	-	
	(320)	1.99	44	
K5	(211)	0.99	89	103.20
	(220)	1.62	-	
	(320)	1.98	-	
K6	(211)	0.99	89	102.47
	(220)	1.66	-	
	(320)	1.94	-	
K7	(211)	0.99	89	102.79
	(220)	1.64	-	
	(320)	1.94	-	

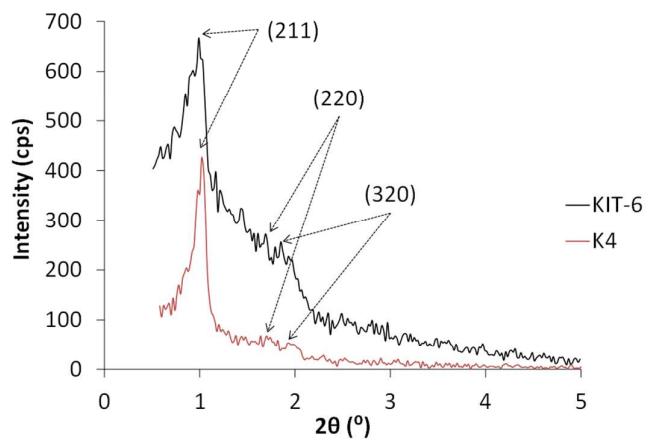


Figure S1. X-ray diffractograms of KIT-6 and K4.

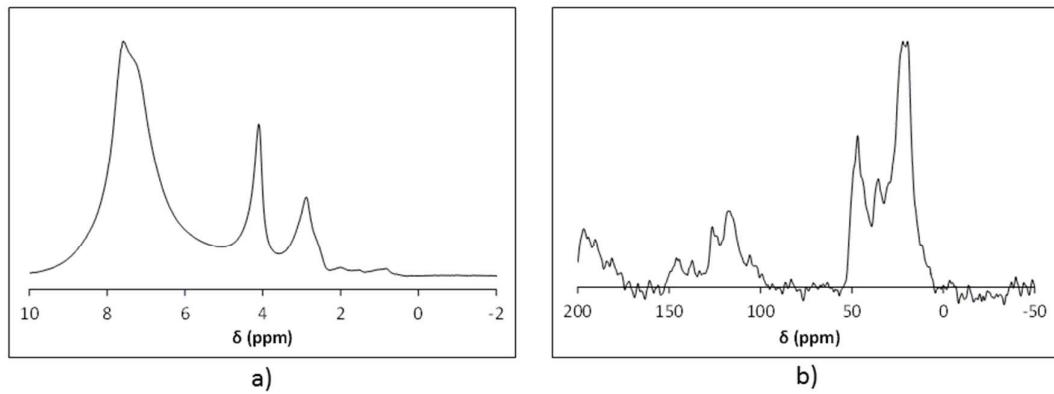


Figure S2. ^1H MAS NMR spectrum of K5 (a) and ^{13}C CP MAS NMR spectrum of K4 (b).

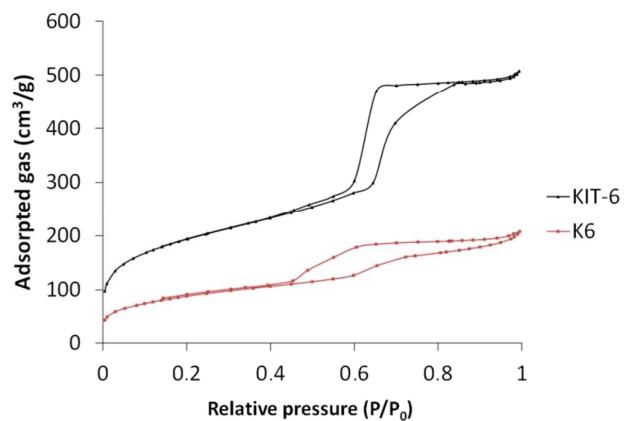


Figure S3. Adsorption-desorption isotherms of KIT-6 and K6.