

Supporting Information for the article titled:

Size Resolved High Temperature Oxidation Kinetics of Nano-Sized Titanium and Zirconium Particles

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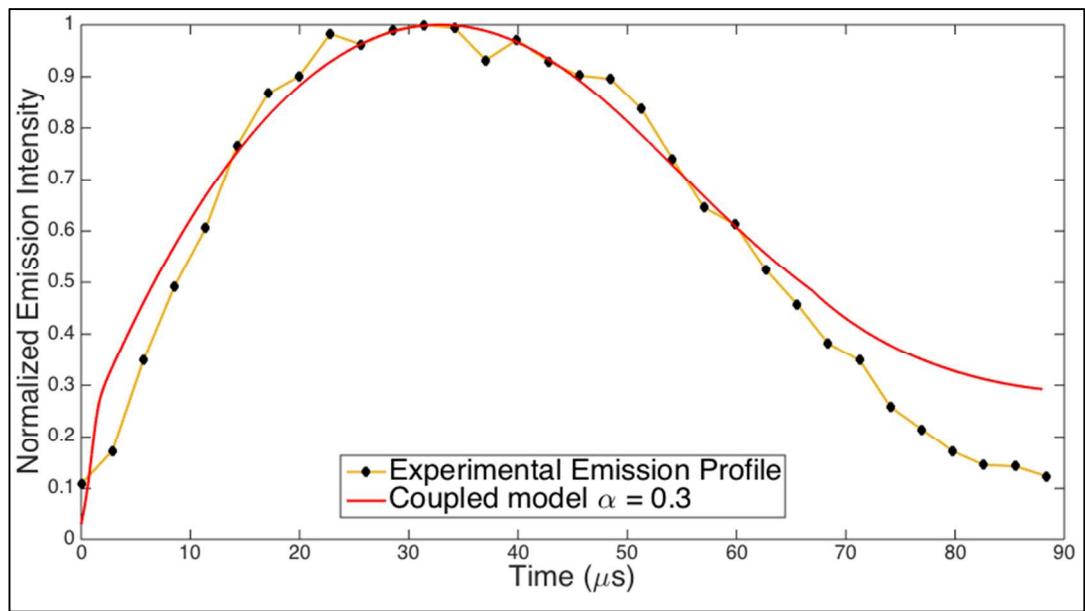


Figure S1. Observed emission profile vs. simulated emission profile for the coupled model fit for 90 nm particle using $\alpha = 0.3$.

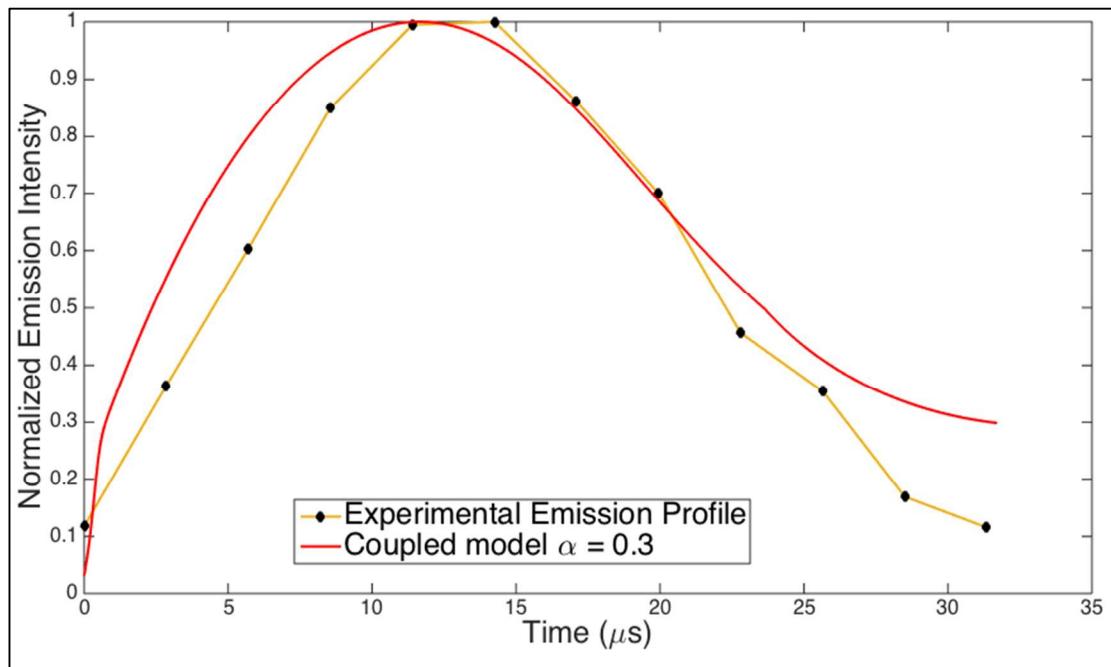


Figure S2. Observed emission profile vs. simulated emission profile for the coupled model fit for 28 nm particle using $\alpha = 0.3$.

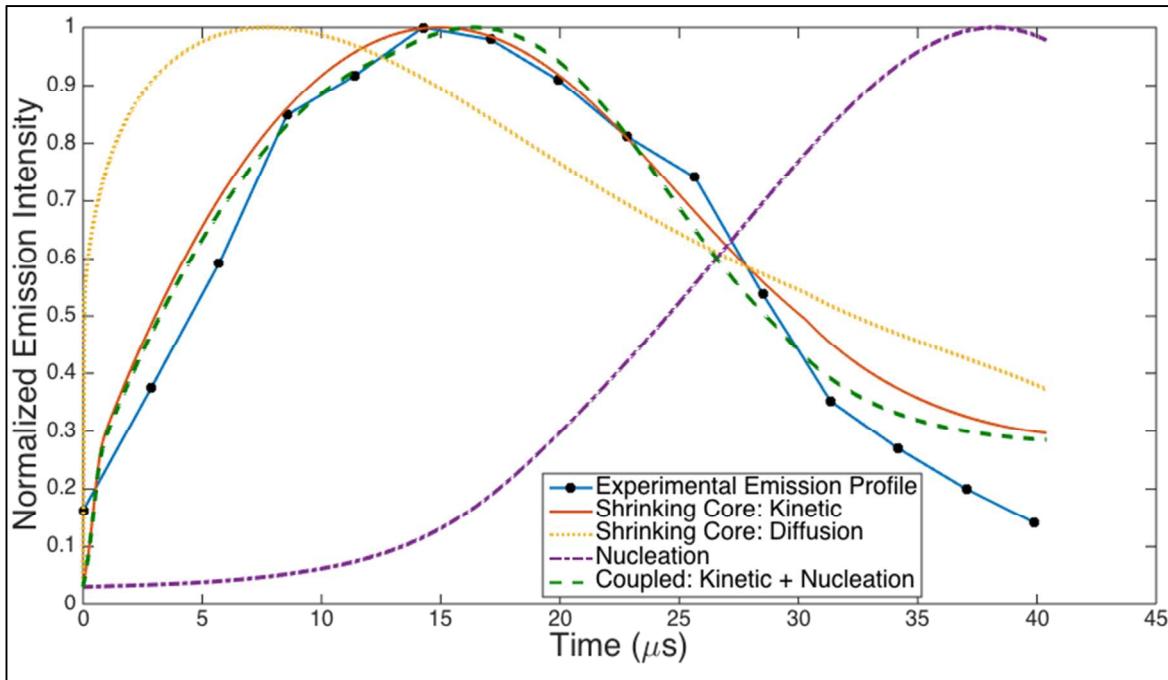


Figure S3. Comparison of different models using $\alpha = 0.3$ with the experimental emission profile for a 40 nm particle.

Variable Definitions for Model:

X_{Ti} : Volume fraction of Titanium/ Zirconium core left

$R_{K,D,N}$: Reaction rate for Kinetic limited, Diffusion limited and Nucleation models [s^{-1}]

τ : Burn time [s]

T : Particle Temperature [K]

P_{STiO_2} : Saturation Vapor pressure of Titania [N/m^2]

P_{SZrO_2} : Saturation Vapor pressure of Zirconia [N/m^2]

α : Thermal accommodation Coefficient

d_p : Particle diameter

m_g : Air molecular weight (4.8×10^{-26} Kg)

K : Boltzmann Constant ($1.38 \times 10^{-23} \text{ m}^2 \text{ Kg s}^{-2} \text{ K}^{-1}$)

P_g : Gas Pressure (1 atm)

γ : Adiabatic expansion factor (=1.3 at 1500 K)

T_g : Gas Temperature (1750 K)

σ : Stefan Boltzman constant ($5.67 \times 10^{-8} \text{ W/m}^2 \text{ K}^4$)

a_p : Surface area of particle (m^2)

N_{av} : Avogadro's constant

Equations used in the Model:

Kinetic Shrinking Core:

$$\frac{dR_K}{dt} = \frac{(-3) * X_{Ti}^{2/3}}{\tau}$$

Diffusion Shrinking Core:

$$\frac{dR_D}{dt} = \frac{1}{2\tau(1 - X_{Ti}^{-\frac{1}{3}})}$$

Nucleation/ Avrami-Erofeev (A4):

$$\frac{dR_N}{dt} = \frac{(-4) * X_{Ti} * (-\log(X_{Ti}))^{0.75}}{\tau}$$

Saturation Vapor Pressure for TiO₂:

$$\log_{10} P_{STiO_2} = 16.2 - 30361/T - 0.000492 * T$$

Saturation Vapor Pressure for ZrO₂:

$$\log_{10} P_{Szro_2} = 3 * 10^{-9} * T^3 - 3 * 10^{-5} * T^2 + 0.074 * T - 74.126$$

Conduction Heat Transfer:

$$Q_C = \alpha * \pi * \left[\frac{d_p}{2} \right]^2 * \frac{P_g}{2} * \left[\frac{8KT}{\pi m_g} \right]^{0.5} * \frac{(\gamma + 1)}{(\gamma - 1)} * \frac{(T - T_g)}{T_g}$$

Radiation Heat Transfer:

$$Q_R = \varepsilon_{avg} * \sigma * \pi d_p^2 * (T^4 - T_g^4)$$

Evaporative Heat Transfer:

$$Q_V = \frac{1 * P_S a_p}{\sqrt{2 \pi K T_p}} * \frac{\Delta H_{volatilization}}{N_{av}}$$

Heat generation:

$$H_{gen} = \frac{dT_{i_moles}}{dt} * \Delta H_{rxn}$$

Heat transfer equation:

$$C\Delta T = H_{gen} - \Delta t * \{Q_C + Q_R + Q_V\}$$