## Supporting Information Rapid Synthesis of C-TiO<sub>2</sub>: Tuning the Shape from Spherical to Rice Grain Morphology for Visible Light Photocatalytic Applications

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S.No	Time in min	Morphology	Crystalline phase
CT-13 CT-14	10 18	spherical spherical and oval	Anatase Anatase
CT-15	40	spherical and oval	Anatase

## Table S1: Results from varying reaction time\*\*

<sup>\*\*</sup>Conditions: Volume: 80% ; Temperature: 150 °C; Concentrations: Same as in C1 (see Table 1), power: 400 W

S.No	Volume	Morphology	Crystalline phase
CT-16	50%	spherical	Anatase
CT-17	40%	distorted spherical	Anatase
CT-18	30%	no specific morphology	Anatase

## Table S2: Results from varying the volume $\frac{\#}{2}$

<sup>#</sup>Conditions: Temperature: 150 °C; Time: 30 min; Concentrations: Same as in C1; power: 400 W

Table S3: Results from varying the temperature	Table S3:	<b>Results from</b>	varying the	temperature
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S.No	Temp in (° C)	Morphology	Crystalline phase
CT-19	70	no specific morphology	Anatase
CT-20	100	no specific morphology	Anatase

<sup>\$</sup>Conditions: Time: 30 min; Concentrations: Same as in C1; power: 400 W, Volume: 80%

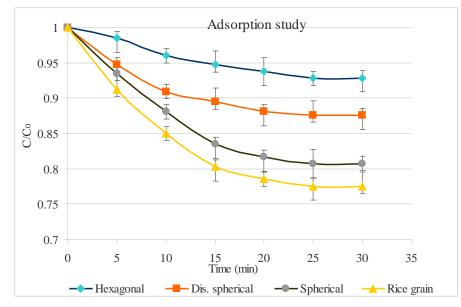


Figure S1. Removal of carbamazepine using different shapes of C-TiO $_2$  catalysts in dark.

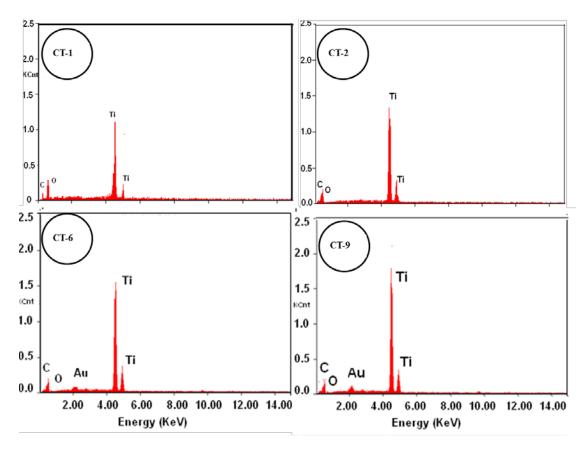


Figure S2. EDS pattern for four different morphologies of C-TiO<sub>2</sub>.

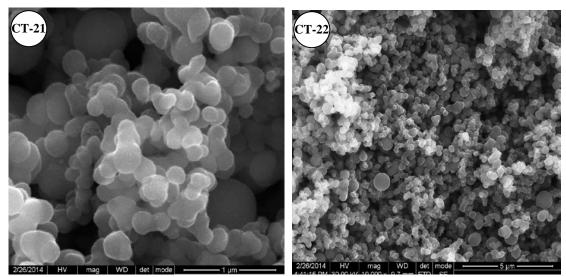


Figure S3. HRSEM images for C-TiO<sub>2</sub> at higher concentration of the ingredients.

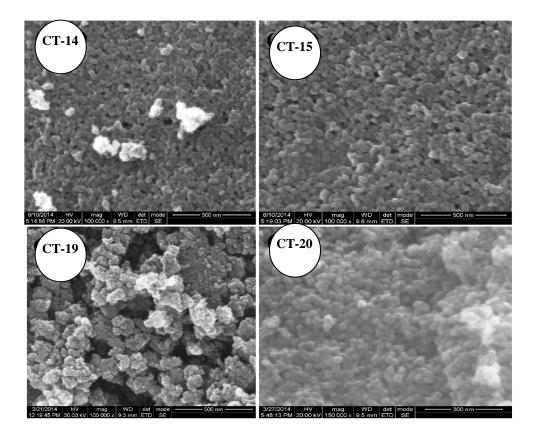


Figure S4. HRSEM images for C-TiO<sub>2</sub> of CT-14 and CT-15 (see Tables S1 and S3) showcasing time and temperature dependence of synthesis.

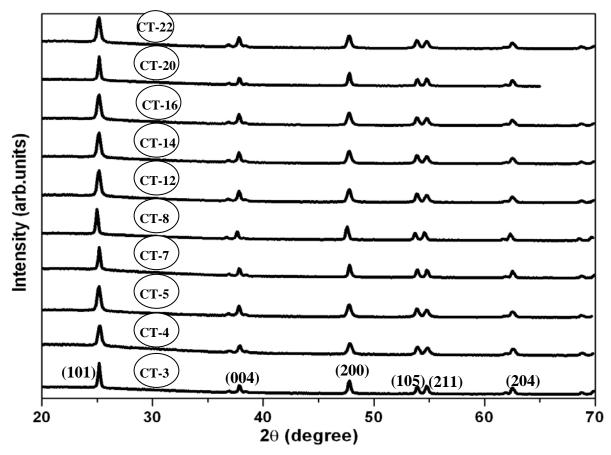


Figure S5. XRD profile for C-TiO<sub>2</sub> at various preparative conditions.

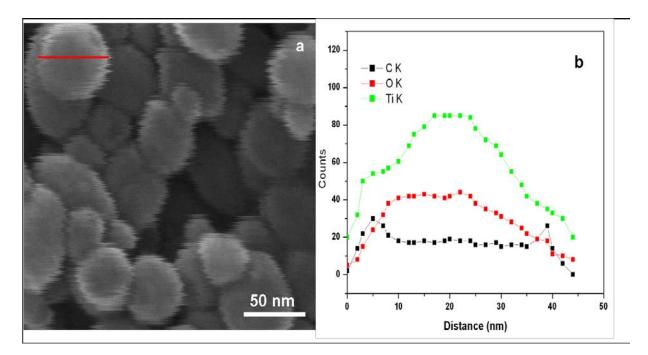


Figure S6. (a) HRSEM image of spherical shaped C-TiO<sub>2</sub> and (b) corresponding EDS line scan data for a single particle as marked by a red line in (a).

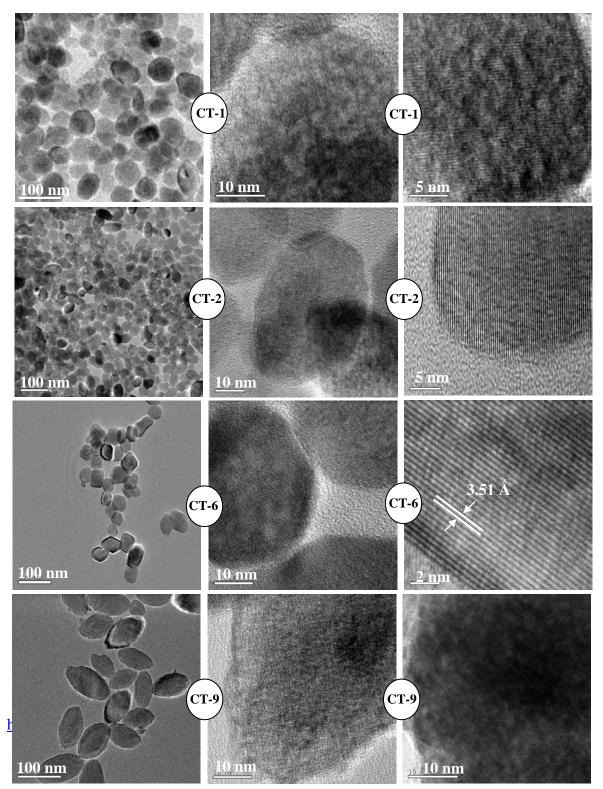


Figure S7. HRTEM images for the four different morphologies of C-TiO<sub>2</sub> nanocrystals.

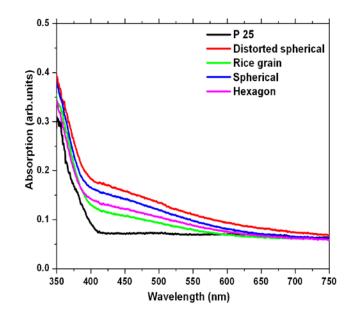


Figure S8. UV-vis diffuse reflectance profiles for four different morphologies of C- $TiO_2$  in comparison with P25 Degussa (bulk  $TiO_2$ ).

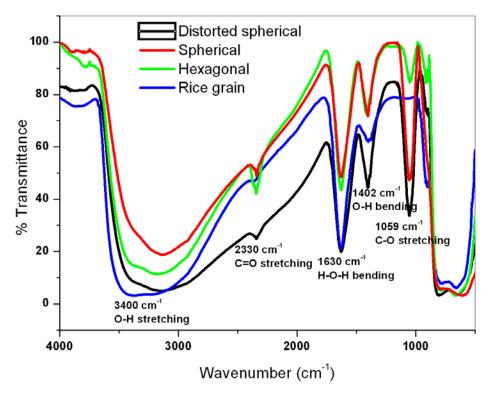


Figure S9. FT-IR spectra for C-TiO<sub>2</sub> of four different morphologies.

Table S4.	BET st	irface area	ı for th	e different	t morphol	logies of	C-TiO <sub>2</sub>
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Sample code	Surface area,	Pore volume,	Pore diameter,
	m²/g	cc/g	Å
Spherical (CT-1)	62	0.24	180
Distorted spherical (CT-2)	61	0.19	224
Hexagonal (CT-6)	33	0.18	190
Rice grain (CT-9)	229	0.17	36

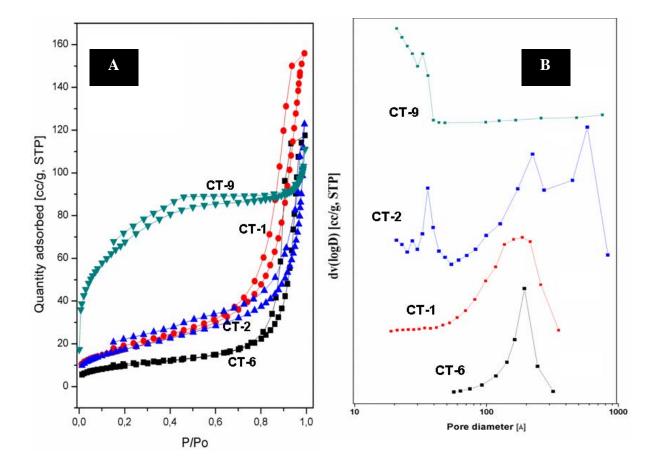


Figure S10.  $N_2$ -adsorption-desorption isotherm liner plot (A) and BJH desorption average pore diameter (B) for four different morphologies of C-TiO<sub>2</sub>.

Table S5. Rate of degradation of Carbamazepine using different shaped C-TiO<sub>2</sub>

Pollutant	Catalyst shape	$\mathbf{R}^2$	K min <sup>-1</sup>
	Rice grain (CT-9)	0.98	0.094
Carbamazepine	Spherical (CT-1)	0.99	0.059
(50µg/L)	Distorted spherical (CT-2)	0.99	0.044
	Hexagonal (CT-6)	0.97	0.020

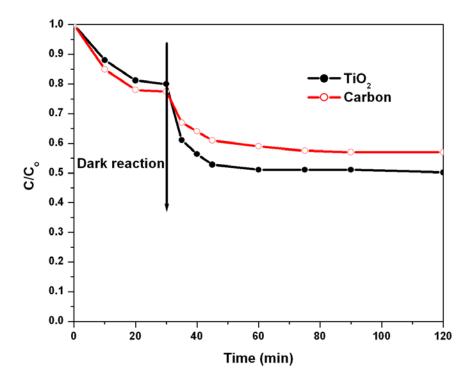


Figure S11. Photocatalytic activity for carbamazepine by  $TiO_2$  and carbon prepared by CT-1 method.