

Supporting Information

Constructing High-Efficiency MoO₃/Polyimide Hybrid Photocatalyst Based on Strong Interfacial Interaction

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Table S1. The amounts of the reactants for synthesis of the samples

Samples*	Reactants (g)			
	MA	PMDA	(NH ₄) ₆ Mo ₇ O ₂₄ ·4H ₂ O	CS(NH ₂) ₂
PI	0.388	0.669	—	—
BMO	—	—	1.227	1.431
0.5BMO/PI	0.386	0.666	0.006	0.007
1.0BMO/PI	0.384	0.662	0.012	0.014
3.0BMO/PI	0.376	0.649	0.036	0.042
5.0BMO/PI	0.368	0.635	0.060	0.070
MoO ₃	—	—	1.277	—
3.0MoO ₃ /PI	0.376	0.649	0.036	—
3.0S/PI	0.376	0.649	—	0.042

*All samples were synthesized by one-pot method, heating at the rate of 7 °C min⁻¹ up to 325 °C and keeping at this temperature for 4 h prior to cooling. The amounts of the reactants were calculated according to the synthesis of 1 g sample.

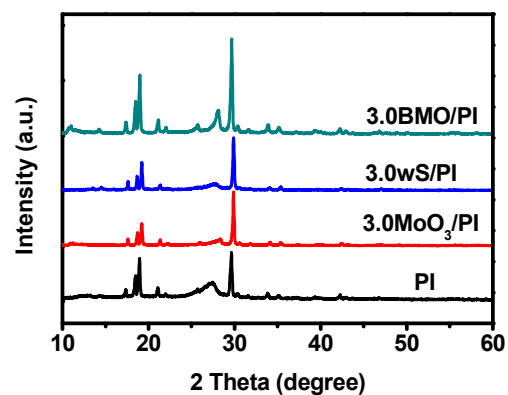


Figure S1. Powder XRD patterns: 3.0BMO/PI, 3.0S/PI, 3.0MoO₃/PI and PI.

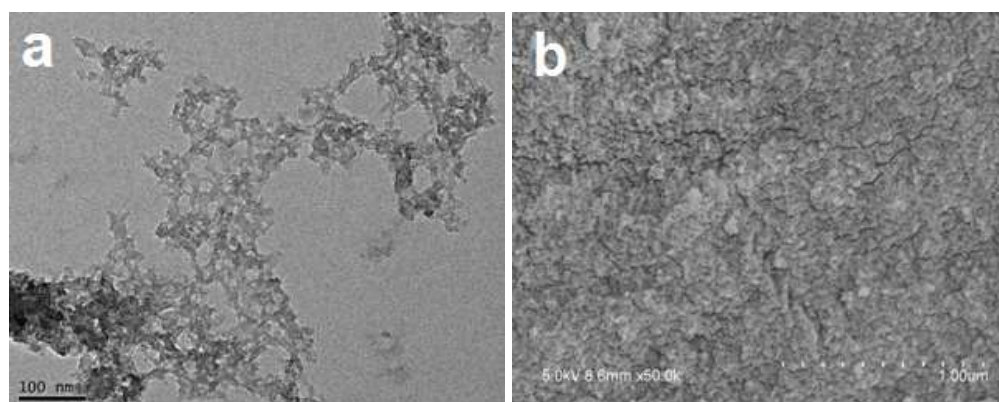


Figure S2. (a) TEM and (b) SEM images of PI.

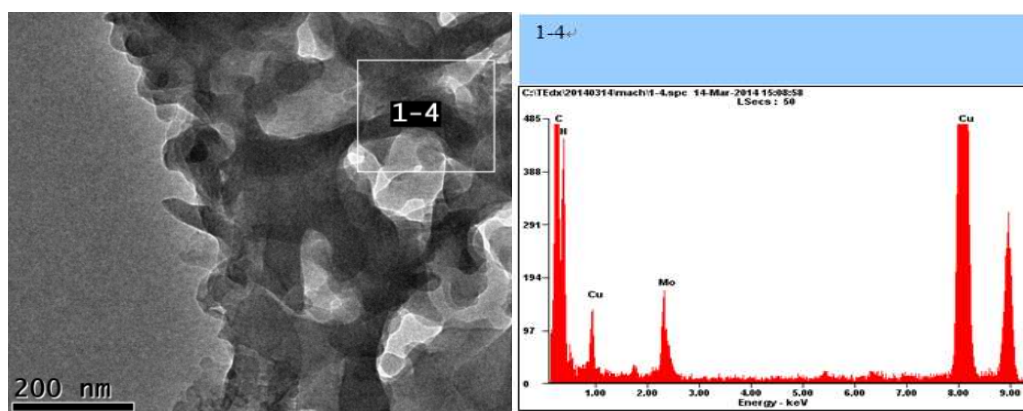


Figure S3. EDS analysis of 3.0BMO/PI composite photocatalyst.

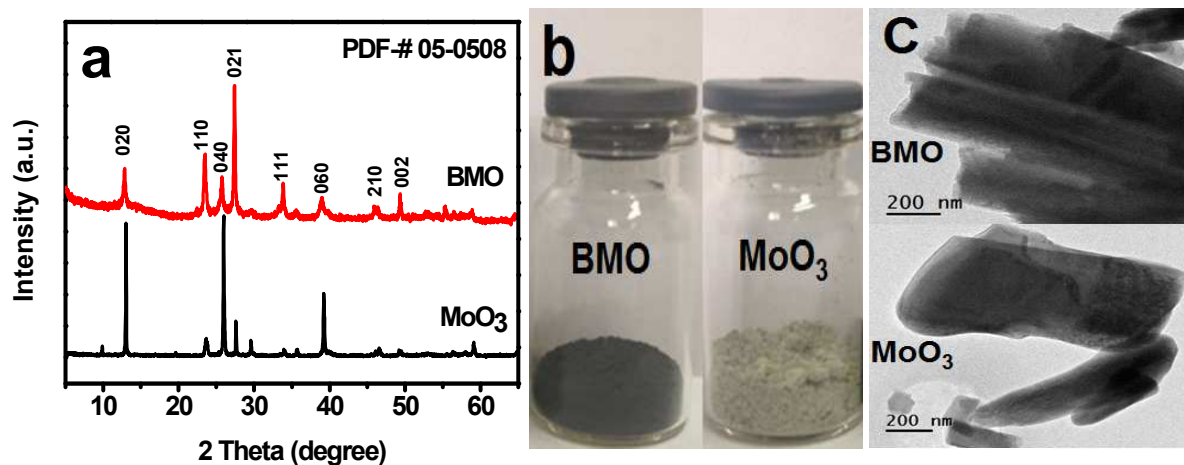


Figure S4. (a) XRD patterns, (b) Photographs and (c) TEM images of BMO and MoO₃.

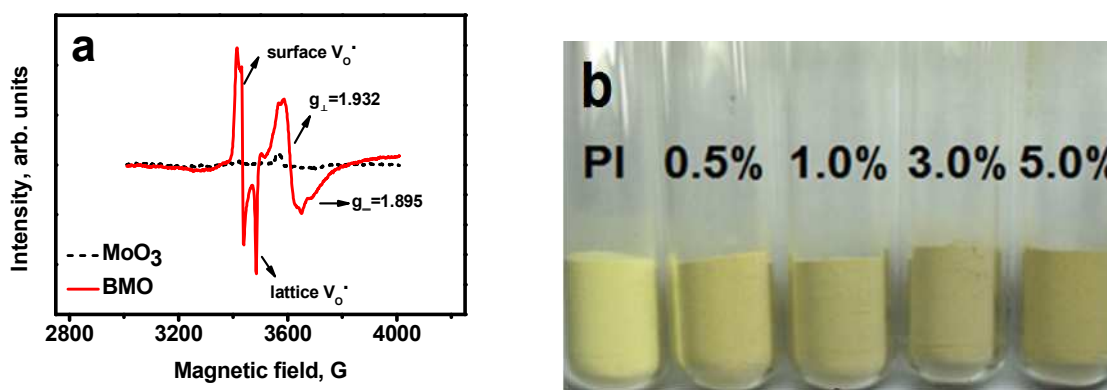


Figure S5. (a) EPR spectra of pristine MoO₃ and BMO samples at room temperature; (b) Colors of PI and (0.5-5.0)BMO/PI.

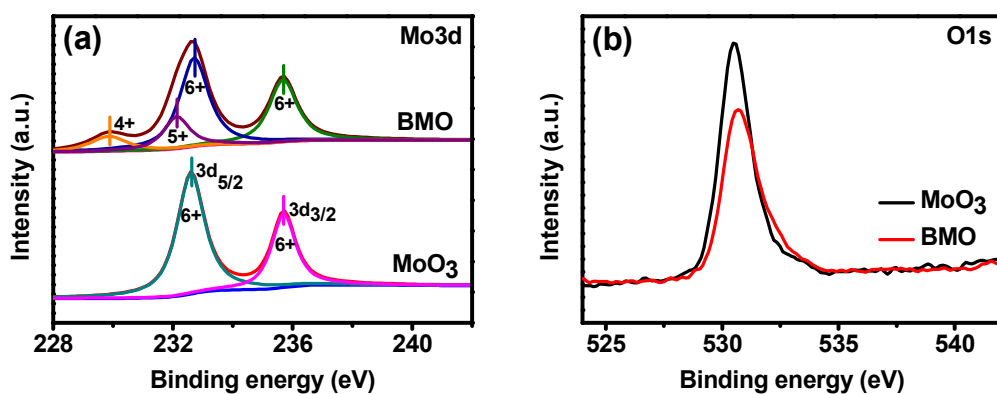


Figure S6. (a) Mo 3d XPS of MoO₃ and BMO; (b) O 1s XPS core level spectra of the pristine MoO₃ and BMO powders.

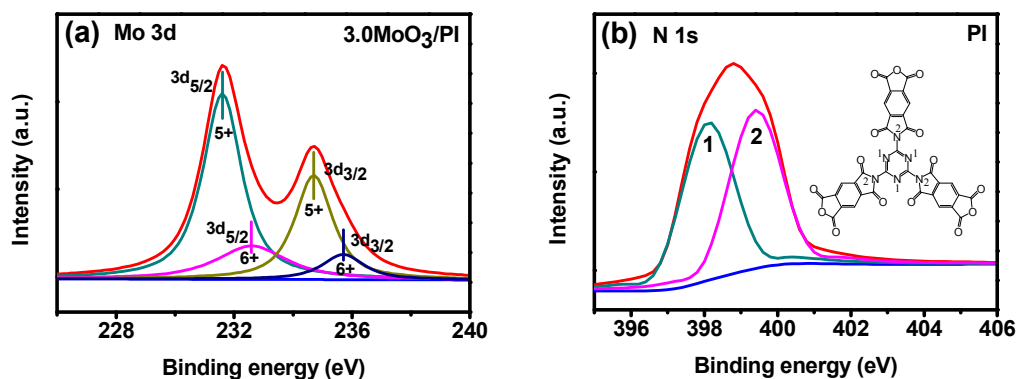


Figure S7. (a) Mo 3d XPS spectra of 3.0MoO₃/PI composite; (b) N 1s XPS spectra of PI.

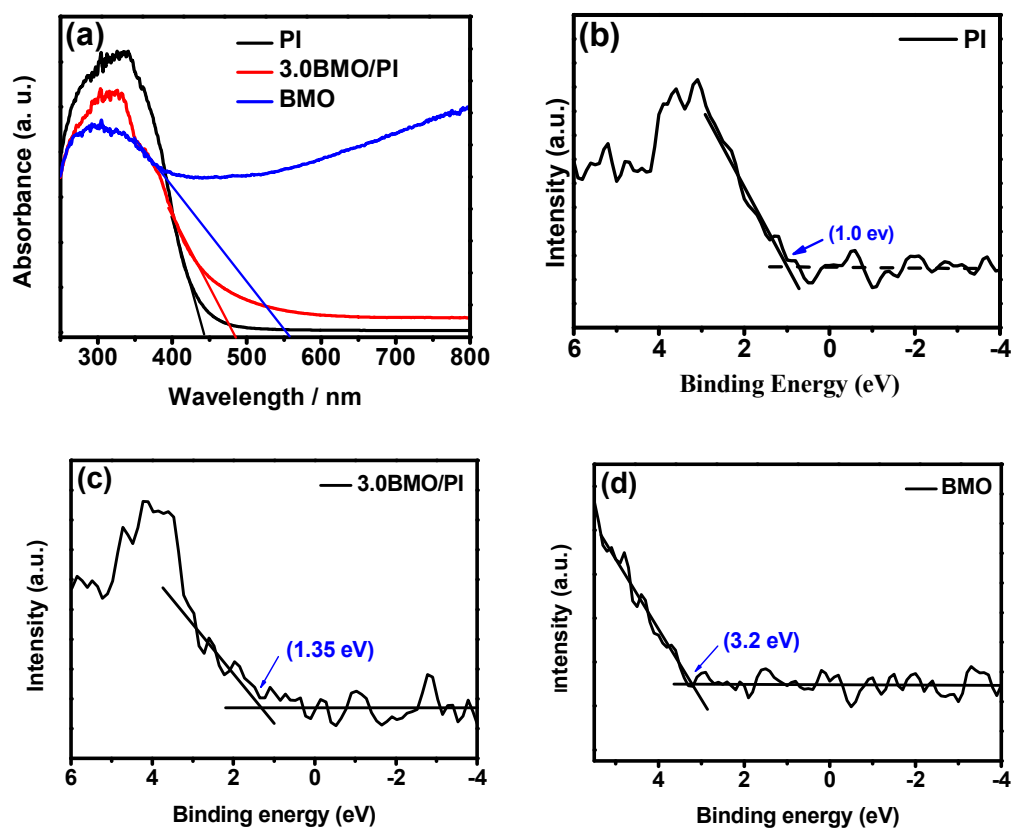


Figure S8. UV-vis absorption spectra (a), VB XPS spectra (b-d) of PI, 3.0BMO/PI and BMO.

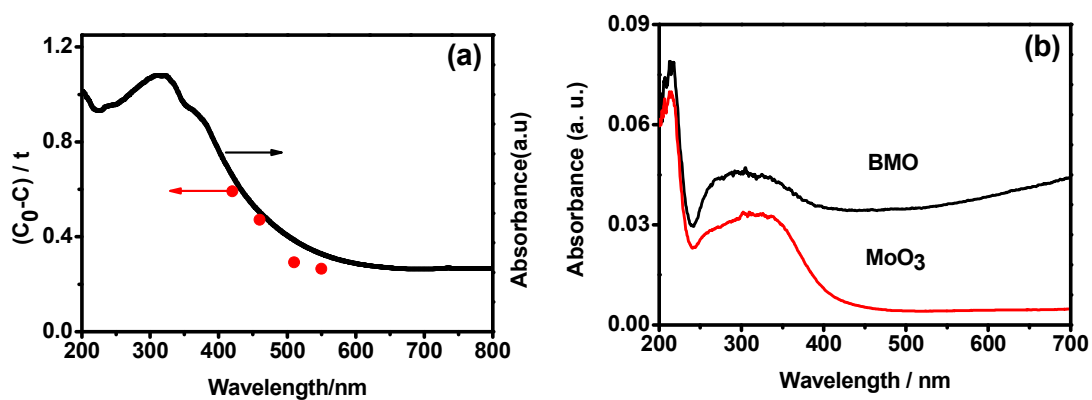


Figure S9. (a) Dependence of degradation activity on wavelength by 3.0BMO/PI; (b) UV-vis absorption spectra of MoO₃ and BMO.

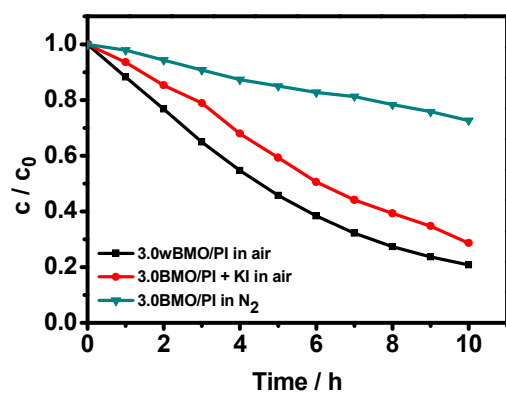


Figure S10. Effects of KI and purging-N₂ as h⁺ and O₂^{•-} scavengers on the degradation of MO in the presence of 3.0BMO/PI under visible light irradiation.

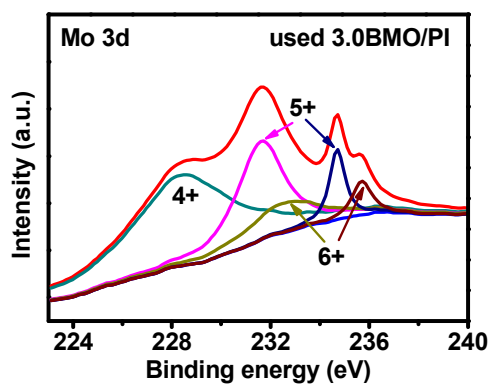


Figure S11. Mo 3d XPS spectra of used 3.0BMO/PI composite.