#### **Supporting Information**

Pyrene-Based Porous Organic Polymers as Efficient Catalytic Support for the Synthesis of Biodiesels at Room Temperature

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**Figure S1.** Wide angle powder XRD pattern of pyrene based organic polymers and corresponding sulfonated polymers.



Figure S2.  $N_2$  adsorption-desorption isotherms (A) and pore size distributions (B) of PPOP-2 and SPPOP-2, respectively.



**Figure S3.**  $N_2$  adsorption-desorption isotherms (A) and pore size distributions (B) of PPOP-3 and SPPOP-3, respectively.

Polymer	Pore diameter (nm)	BET surface area (m <sup>2</sup> g <sup>-1</sup> )	Total pore volume (ccg <sup>-1</sup> )
PPOP-2	1.54	570	0.3595
PPOP-3	1.52	508	0.3186
SPPOP-2	1.47	244	0.1751
SPPOP-3	1.48	205	0.1539

 Table 1S.Physical properties of PPOP-2,3 and corresponding sulfonated polymers



Figure S4. TEM images of SPPOP-3 polymer



Figure S5. FE SEM images of PPOP-2 (A), SPPOP-2 (B), PPOP-3 (C) and SPPOP-3 (D).



Figure S6. UV-visible absorption and photoluminescence spectra of PPOP-1.

UV-Vis absorption and photoluminescence spectra of PPOP-1 polymer are shown in FigureS2. UV-Vis absorption spectra showed broad band at 360 nm due to  $\pi \rightarrow \pi^*$  transition of the  $\pi$ conjugated pyrene rings. PPOP-1 exhibits strong photoluminescence properties and showed
emission with wavelength maxima at *ca*. 453 nm.



Figure S7. TG-DTA profiles of SPPOP-2 (A) and SPPOP-3 (B).



Figure S8 NH<sub>3</sub>-TPD profile for SPPOP-2



Figure S9 NH<sub>3</sub>-TPD profile for SPPOP-3



Figure S10. FT IR spectra of PPOP-2,3 and SPPOP-2,3





S13





## <sup>13</sup>C NMR of Methyl myristate



<sup>1</sup>H NMR of Methyl palmitate



# <sup>13</sup>C NMR of Methyl palmitate



# <sup>1</sup>H NMR of Methyl stearate



## <sup>13</sup>C NMR of Methyl stearate







# <sup>1</sup>H NMR of Dimethyl sebacate



# <sup>13</sup>C NMR of Dimethyl sebacate



# <sup>1</sup>H NMR of Dimethyl glutarate



## <sup>13</sup>C NMR of Dimethyl glutarate



# <sup>1</sup>H NMR of Dimethyl adipate



## <sup>13</sup>C NMR of Dimethyl adipate







<sup>1</sup>H NMR spectrum of transesterification of soybean oil at 60°C over SPPOP-1



 $^{13}\text{C}$  NMR spectrum of transesterification of soybean oil at 60°C over SPPOP-1



 $^1\text{H}$  NMR spectrum of transesterification of soybean oil at 25  $^{\rm o}\text{C}$  over SPPOP-2



<sup>1</sup>H NMR spectrum of transesterification of soybean oil at 60°C over SPPOP-2



## <sup>1</sup>H NMR spectrum of transesterification of soybean oil at 25°C over SPPOP-3



<sup>1</sup>H NMR spectrum of transesterification of soybean oil at 60°C over SPPOP-3



<sup>13</sup>C NMR spectrum of transesterification of soybean oil at 60°C over SPPOP-3



## <sup>1</sup>H NMR spectrum of transesterification of olive oil at 25°C over SPPOP-1



## <sup>1</sup>H NMR spectrum of transesterification of olive oil at 60°C over SPPOP-1



#### <sup>13</sup>C NMR spectrum of transesterification of olive oil at 60°C over SPPOP-1



## $^1\text{H}$ NMR spectrum of transesterification of olive oil at 25°C over SPPOP-2



## $^1\text{H}$ NMR spectrum of transesterification of olive oil at 60°C over SPPOP-2



#### <sup>13</sup>C NMR spectrum of transesterification of olive oil at 60°C over SPPOP-2



## <sup>1</sup>H NMR spectrum of transesterification of olive oil at 25°C over SPPOP-3



## <sup>13</sup>C NMR spectrum of transesterification of olive oil at 25°C over SPPOP-3

180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 ppm



## <sup>1</sup>H NMR spectrum of transesterification of olive oil at 60°C over SPPOP-3



## <sup>1</sup>H NMR spectrum of esterification of lauric acid at 25°C over amberlite IR-120(H) resin



<sup>1</sup>H NMR spectrum of transesterification of soybean oil at 25°C over amberlite IR-120(H) resin



<sup>1</sup>H NMR spectrum of transesterification of soybean oil at 60°C over amberlite IR-120(H) resin