Supporting Information

Nickel(0)-Catalyzed Enantio- and Diastereoselective Synthesis of Benzoxasiloles: Ligand-Controlled Switching from Inter- to Intramolecular Aryl-Transfer Process

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[1] General

All manipulations were conducted under a nitrogen atmosphere by using standard Schlenk or dry box techniques unless otherwise noted. ¹H and ¹³C spectra were recorded on Bruker AVANCE III 400 spectrometers at 25 °C. The chemical shifts in ¹H-NMR and ¹³C{¹H}-NMR spectra are reported in parts per million (ppm) and are referenced to the residual solvent signal as the internal standard: CDCl₃ $\delta = 7.26$ (¹H) and $\delta = 77.0$ (¹³C) ppm; C₆D₆: $\delta = 7.16$ (¹H) and $\delta = 128.1$ (¹³C) ppm. Splitting patterns are denoted as "s" for singlet; "d" for doublet; "t" for triplet; "q" for quartet; "sext" for sextet; "sept" for septet; "m" for multiplet, "br" for broad; "dt" for doublet of triplets; "td" for triplet of doublets, and "app" for apparent. Mass spectra were obtained by using a Shimadzu GCMS-QP 2010 instrument with an ionization voltage of 70 eV. Medium-pressure column chromatography was carried out on a Biotage Flash Purification System Isolera, equipped with a 250 nm UV detector. Analytical gas chromatography (GC) was carried out on a Shimadzu GC-2014 gas chromatograph, equipped with a flame ionization detector. High resolution mass spectrometry (HRMS) was performed at Instrumental Analysis Center, Faculty of Engineering, Osaka University. Enantioselectivities were recorded by means of either JASCO-Supercritical Fluid chromatography (SFC) equipped with PU-2080-CO₂ plus CO₂ delivery pump and MD-2018 plus as a photodiode array detector or HPLC of Shimadzu or Lacrome manufacturer using chiral columns of Diacel Chiralpak (I and O series). Optical rotations were measured in JASCO-DIP 1000 polarimeter with a path length of 1 dm using the sodium D line, 589 nm. THF, toluene, and benzene- d_6 were distilled from sodium benzophenone ketyl, and other solvents were distilled and degassed prior to use. All commercially available reagents were distilled over CaH₂ under reduced pressure prior to use. Ni(cod)₂ was recrystallized from toluene prior to use. Ni(acac)₂ was dried under reduced pressure (0.3 mmHg) at >100 °C prior to use. All synthesized starting materials were purified either by distillation over CaH₂ or recrystallized prior to use for catalytic reactions. Chiral N-heterocyclic carbene (NHC) salts were synthesized according to the reported procedures. Optimized ligand L5·HBF₄ was prepared according to Grubbs protocol. S1

[2] Preparation of Starting Materials: All starting materials (Figure S1) employed for catalytic reactions were prepared according to experimental procedure reported in our previous communication, S2 except for 1n and 1o. The experimental procedures for all new compounds (1g, 1i,

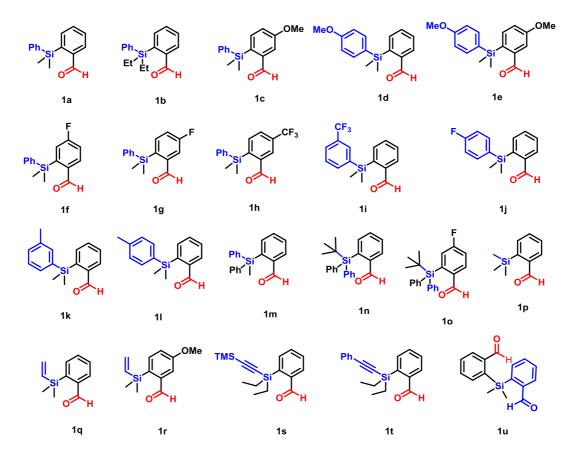


Figure S1: Substrates employed for catalytic asymmetric reactions

1j, 1t, and 1u) were described herein.

Synthesis of 2-dimethyl(4-methoxyphenyl)silyl-5-methoxybenzaldehyde (1e):

To the solution of 1-bromo-4-methoxybenzaldehyde dimethylacetal (5.9 g, 23.0 mmol) in THF (40 mL) was added ⁿBuLi (16 mL, 25.5 mmol, 1.6 M in hexane) dropwise at -78 °C and the mixture was stirred further for 0.5 h. This solution was transferred slowly to a flask containing a

solution of dichlorodimethylsilane (3.3 g, 25.5 mmol) in THF (40 mL) at -78 °C using cannula. The resulting reaction mixture was stirred further for 2 h at -78 °C and then allowed to warm to room temperature with stirring for overnight. The reaction mixture was brought to -78 °C again and a solution of 4-methoxyphenyllithium (prepared by the reaction of 1-bromo-4-methoxybenzene (5.2 g, 27.8 mmol) with ⁿBuLi (19 mL, 30.5 mmol, 1.6 M in hexane) in THF (40 mL) at -78 °C) was added slowly using cannula over 30 min and allowed to stir at -78 °C for additional 2 h followed by warming to room temperature with overnight stirring. 1M HCl (aq.) was added and the organic layer was extracted with ether, dried over anhydrous Na₂SO₄, and concentrated. It was purified by silica gel column chromatography (with 3% EtOAc:hexane) gave 2-dimethyl(4-methoxyphenyl)silyl-5-methoxybenzaldehyde (1e) as colorless oil in 63% overall yield (4.4 g, 14.6 mmol). ¹H NMR (400 MHz, CDCl₃): δ 0.59 (s, 6H, Si(CH₃)₂), 3.81 (s, 3H, OCH₃), 3.87 (s, 3H, OC H_3), 6.89 (d, J = 8.4 Hz, 2H, Ar-H), 7.11 (dd, J = 2.8, 8.4 Hz, 1H, Ar-H), 7.41 (d, J = 8.4Hz, 2H, Ar-H), 7.47 (d, J = 2.4 Hz, 1H, Ar-H), 7.55 (d, J = 8.4 Hz, 1H, Ar-H), 10.07 (s, 1H, CHO). $^{13}C{^1H}$ NMR (100 MHz, CDCl₃): δ –0.7, 55.0, 55.3, 113.7, 114.9, 119.3, 129.7, 132.7, 135.4, 137.9, 142.8, 160.4, 160.9, 192.7. **HRMS** (EI): m/z Calcd for $C_{17}H_{20}O_3Si$: (M⁺) 300.1182, found 300.1181.

Synthesis of 2-(dimethyl(phenyl)silyl)-5-fluorobenzaldehyde (1g):

To the solution of 1-bromo-4-fluorobenzaldehyde dimethylacetal (5.9 g, 24.0 mmol) in THF (50 mL) was added ⁿBuLi (18.2 mL, 29.2 mmol, 1.6 M in hexane) dropwise at -78 °C and the mixture was stirred further for 0.5 h. To this reaction mixture was added chlorodimethylphenylsilane (5.4 g, 31.6 mmol) and stirred for 2 h at -78 °C and then allowed to warm to room temperature for overnight. HCl aq. (1M) was added to the mixture and the organic layer was extracted with ether, dried over anhydrous Na₂SO₄, concentrated and purified by silica gel column chromatography (with 5% EtOAc:hexane) to get **1g** as a colorless liquid (5.0 g, 78% yield). ¹H NMR (400 MHz, CDCl₃): δ 0.63 (s, 6H, Si(CH₃)₂), 7.24–7.29 (m, 2H, Ar-H), 7.32–7.40 (m, 3H, Ar-H), 7.46–7.49 (m, 2H, Ar-H), 7.60–7.64 (m, 2H, Ar-H), 10.03 (s, 1H, CHO). ¹³C{¹H} NMR (100 MHz, CDCl₃): δ -1.1, 117.5 (d, J_{CF} = 20.5 Hz), 120.0 (d, J_{CF} = 20.5 Hz), 128.0, 129.3, 133.9, 136.6 (d, J_{CF} = 3.8 Hz), 138.2,

138.7 (d, $J_{\text{CF}} = 7.1 \text{ Hz}$), 143.4 (d, $J_{\text{CF}} = 5.0 \text{ Hz}$), 164.0 (d, $J_{\text{CF}} = 255.0 \text{ Hz}$), 191.5. **HRMS** (EI): m/z Calcd for $C_{15}H_{15}FOSi$: (M⁺) 258.0876, found 258.0875.

Synthesis of 2-(dimethyl(3-(trifluoromethyl)phenyl)silyl)benzaldehyde (1i):

To the solution of 2-bromobenzaldehyde diethyl acetal (2.0 g, 7.7 mmol) in THF (20 mL) was added ⁿBuLi (5.3 mL, 8.5 mmol, 1.6 M in hexane) dropwise at -78 °C and the mixture was stirred further for 0.5 h. This solution was transferred slowly to a flask containing a solution of dichlorodimethylsilane (1.1 g, 8.5 mmol) in THF (20 mL) at -78 °C using cannula. The resulting reaction mixture was stirred further for 2 h at -78 °C and then allowed to warm to room temperature with stirring for overnight. The reaction mixture was brought to -78 °C again and a solution of 3-trifluoromethylphenyllithium (prepared by the reaction of 3-Bromobenzotrifluoride (2.0 g, 9.3 mmol) with nBuLi (6.4 mL, 10.2 mmol, 1.6 M in hexane) in THF (20 mL) at -78 °C) was added slowly using cannula over 0.5 h and allowed to stir at -78 °C for additional 2 h followed by warming to room temperature with overnight stirring. 1M HCl (aq.) was added and the organic layer was extracted with ether, dried over anhydrous Na₂SO₄, and concentrated. It was purified by silica gel column chromatography (with 5% EtOAc:hexane) to get 1i as a colorless liquid (1.9 g, 79% yield). 1 H NMR (400 MHz, CDCl₃): δ 0.73 (s, 6H, Si(CH₃)₂), 7.51 (t, 1H, J = 7.5 Hz, Ar-H), 7.63-7.70 (m, 3H, Ar-H), 7.70-7.74 (m, 1H, Ar-H), 7.77 (d, 1H, J = 7.2 Hz, Ar-H), 7.84 (s, 1H, Ar-H), 7.93–7.80 (m, 1H, Ar-H), 10.03 (s, 1H, CHO). ${}^{13}C\{{}^{1}H\}$ NMR (100 MHz, CDCl₃): δ -1.5, 124.4 (q, $J_{CF} = 271.0 \text{ Hz}$), 125.5 (q, $J_{CF} = 3.7 \text{ Hz}$), 127.9, 129.8 (q, $J_{CF} = 31.7 \text{ Hz}$), 130.1, 130.2 (d, $J_{\text{CF}} = 3.7 \text{ Hz}$), 133.2, 133.6, 136.9, 137.3, 139.0, 140.5, 141.0, 192.9. **HRMS** (EI): m/z Calcd for $C_{16}H_{15}F_3OSi: (M+H^+) 309.0922$, found 309.0921.

Synthesis of 2-((4-fluorophenyl)dimethylsilyl)benzaldehyde (1j):

To the solution of 2-bromobenzaldehyde diethyl acetal (2.0 g, 7.7 mmol) in THF (20 mL) was added ⁿBuLi (5.3 mL, 8.5 mmol, 1.6 M in hexane) dropwise at -78 °C and the mixture was stirred further for 0.5 h. This solution was transferred slowly to a flask containing a solution of dichlorodimethylsilane (1.1 g, 8.5 mmol) in THF (20 mL) at -78 °C using cannula. The resulting reaction mixture was stirred further for 2 h at -78 °C and then allowed to warm to room temperature with stirring for overnight. The reaction mixture was brought to -78 °C again and a solution of (4-fluorophenyl)lithium (prepared by the reaction of 1-bromo-4-fluorobenzene (1.6 g, 9.3 mmol) with nBuLi (6.4 mL, 10.2 mmol, 1.6 M in hexane) in THF (20 mL) at -78 °C) was added slowly using cannula over 0.5 h and allowed to stir at -78 °C for additional 2 h followed by warming to room temperature with overnight stirring. 1M HCl (aq.) was added and the organic layer was extracted with ether, dried over anhydrous Na₂SO₄, and concentrated. It was purified by silica gel column chromatography (with 3% EtOAc:hexane) to get 1j as a colorless liquid (1.4 g, 68% yield). **H NMR** (400 MHz, CDCl₃): δ 0.62 (s, 6H, Si(CH₃)₂), 7.04 (t, J = 8.6 Hz, 2H, Ar-H), 7.48 (t, J =7.6 Hz, 2H, Ar-H), 7.50–7.65 (m, 3H, Ar-H), 7.89–7.93 (m, 1H, Ar-H), 10.01 (s, 1H, CHO). 13 C{ 1 H} NMR (100 MHz, CDCl₃): δ –1.2, 114.9 (d, J_{CF} = 19.6 Hz), 129.8, 132.6, 133.1, 134.3 (d, $J_{\text{CF}} = 3.8 \text{ Hz}$), 135.8 (d, $J_{\text{CF}} = 7.3 \text{ Hz}$), 136.7, 140.2, 141.1, 163.5 (d, $J_{\text{CF}} = 245.8 \text{ Hz}$), 191.0. **HRMS** (CI): m/z Calcd for $C_{15}H_{16}FOSi$: $(M+H^+)$ 259.0943, found 259.0946.

Synthesis of 2-(tert-butyldiphenylsilyl)benzaldehyde (1n):

To a stirring solution of (2-bromophenyl)methanol, 1n-(i) (2.8 g, 14.7 mmol) and imidazole (3.0 g, 44.1 mmol) in DMF (20 mL) was added TBDPSCl (6.39 mL, 25.0 mmol) at room temperature followed by stirring further for 24 h. Aqueous NaHCO₃ was added and extracted well with hexane twice. The organic extracts was dried over Na₂SO₄, concentrated and purified through silica gel column chromatography (with 5% EtOAc:hexane) to get pure protected benzylalcohol **1n-(ii)** (5.5 g, 89% yield). ¹**H NMR** (400 MHz, CDCl₃): δ 1.11 (s, 9H, SiC(C H_3)₃), 4.78 (s, 2H, OCH_2), 7.14 (td, J = 1.4, 7.4 Hz, 1H, Ar-H), 7.35–7.46 (m, 7H, Ar-H), 7.49 (dd, J = 1.1, 8.0 Hz, 1H, Ar-H), 7.71 (dd, J = 1.4, 8.0 Hz, 4H, Ar-H), 7.75 (d, J = 7.5 Hz, 1H, Ar-H). The spectral data was identified to that previously reported. S3 TBDPS-protected benzylalcohol, **1n-(ii)** (9 g, 21.2 mmol) was dissolved in dry THF (60 mL) was cooled to -78 °C and ⁿBuLi was added drop wise for 15 min. After 30 min it was allowed to warm up slowly to room temperature. Reaction mixture was quenched with NH₄Cl (aqueous) and extracted with ether (3 x 100 mL). Combined organic extracts were dried over Na₂SO₄ and concentrated to get isomerized product (O-Si to C-Si) 1n-(iii) quantitatively as a very thick liquid as confirmed by ¹H NMR analysis of residue. It was forwarded for next step without purification. ¹H NMR (400 MHz, CDCl₃): δ 1.17 (s, 9H, SiC(CH₃)₃), 4.11 (d, J $= 6.6 \text{ Hz}, 2H, OCH_2$, 7.33-7.43 (m, 7H, Ar-H), 7.47-7.57 (m, 2H, Ar-H), 7.61-7.67 (m, 4H, Ar-H),7.97 (dd, J = 1.0, 7.5 Hz, 1H, Ar-H). The *ortho*-TBDPS-benzylalcohol **1n-(iii)** (7.0 g, 20.2 mmol) was dissolved in DCM (60 mL) and PCC (8.7g, 40.4 mmol) was added in a small portion-wise under N₂ flow and allowed to stirred for 1-2 h at room temperature. After completion of reaction, DCM was evaporated completely then re-diluted with ether. The black sticky material was filtered off over celite and washed three times with ether. The ether solution was evaporated to get white solid (quantitative conversion), which was recrystalized with toluene to get pure compound 1n (6.5 g, 93% yield). H NMR (400 MHz, CDCl₃): δ 1.20 (s, 9H, SiC(CH₃)₂), 7.32–7.42 (m, 6H, Ar-H), 7.55–7.64 (m, 6H, Ar-H), 7.99 (dd, J = 1.8, 6.5 Hz, 1H, Ar-H), 8.07 (dd, J = 1.8, 6.5 Hz, 1H, Ar-H), 9.86 (s, 1H, CHO). 13 C{ 1 H} NMR (100 MHz, CDCl₃): δ 18.9, 28.7, 128.0, 128.4, 129.5, 129.9, 132.5, 135.1, 135.9, 137.7, 138.8, 141.9, 193.0. **HRMS** (EI): m/z Calcd for C₂₃H₂₄OSi: (M⁺) 344.1596, found 344.1591.

Synthesis of 2-(tert-butyldiphenylsilyl)-4-fluorobenzaldehyde (10):

Compound 10 was synthesized following the similar experimental procedure as employed for **1n** using (2-bromo-4-fluorophenyl)methanol (**10-(i)**, 6.9 g, 25.0 mmol) as a starting material. **10-(ii)**: ¹**H NMR** (400 MHz, CDCl₃): δ 1.12 (s, 9H, SiC(C H_3)₃), 4.74 (s, 2H, OC H_2), 7.08 (td, J = 2.4, 8.3 Hz, 1H, Ar-H), 7.22-7.27 (m, 1H, Ar-H), 7.35-7.48 (m, 6H, Ar-H), 7.64-7.74 (m, 1H, Ar-H)5H, Ar-H). ${}^{13}C\{{}^{1}H\}$ NMR (100 MHz, CDCl₃): δ 19.4, 26.9, 64.8, 114.4 (d, J_{CF} = 20.6 Hz), 119.4 (d, $J_{\text{CF}} = 24.6 \text{ Hz}$), 121.0 (d, $J_{\text{CF}} = 9.3 \text{ Hz}$), 127.9, 128.7 (d, $J_{\text{CF}} = 8.0 \text{ Hz}$), 129.9, 133.1, 135.5, 135.8 (d, $J_{\text{CF}} = 3.1 \text{ Hz}$), 161.4 (d, $J_{\text{CF}} = 247.8 \text{ Hz}$). **HRMS** (EI): m/z Calcd for $C_{23}H_{24}BrFOSiNa$: (M⁺+Na) 465.0661, found 465.0666. **1o-(iii)**: ¹**H NMR** (400 MHz, CDCl₃): δ 1.17 (s, 9H, SiC(CH₃)₃), 4.08 (d, J = 6.1 Hz, 2H, OC H_2), 7.18 (td, J = 2.9, 8.3 Hz, 1H, Ar-H), 7.35–7.45 (m, 6H, Ar-H), 7.53 (dd, J =5.5, 8.3 Hz, 1H, Ar-H), 7.61-7.67 (m, 5H, Ar-H). It was used for next oxidation-step without purification. **1o** (4.4 g, 95% yield): ¹**H NMR** (400 MHz, CDCl₃): δ 1.18 (s, 9H, SiC(CH₃)₃), 7.19 (td, J = 2.5, 8.3 Hz, 1H, Ar-H), 7.30-7.41 (m, 6H, Ar-H), 7.53-7.58 (m, 4H, Ar-H), 7.61 (dd, J = 2.5, 9.4)Hz, 1H, Ar-H), 8.09 (dd, J = 5.6, 8.7 Hz, 1H, Ar-H), 9.79 (s, 1H, CHO). ${}^{13}C({}^{1}H)$ NMR (100 MHz, $CDCl_3$): δ 19.0, 28.7, 117.1 (d, $J_{CF} = 21.1 \text{ Hz}$), 124.5 (d, $J_{CF} = 21.1 \text{ Hz}$), 128.2, 129.7, 131.3 (d, $J_{CF} = 21.1 \text{ Hz}$) 8.6 Hz), 134.3, 135.9, 138.1 (d, $J_{CF} = 3.3$ Hz), 142.9 (d, $J_{CF} = 5.7$ Hz), 165.0 (d, $J_{CF} = 257.9$ Hz), 191.2. **HRMS** (CI): *m/z* Calcd for C₂₃H₂₄FOSi: (M⁺+H) 363.1574, found 363.1578.

Synthesis of 2-(diethyl(phenylethynyl)silyl)benzaldehyde (1t):

To a solution of 2-bromobenzaldehyde diethyl acetal (7.5 g, 29 mmol) in THF (50 mL) was added dropwise n BuLi (20 mL, 32 mmol) at -78 °C, and the mixture was stirred for 0.5 h. This solution was slowly added to a solution of dichlorodimethylsilane (5 g, 32 mmol) in THF (100 mL), and stirred for 2 h at -78 °C. The resulting reaction mixture was allowed to warm to room temperature for overnight. To reaction mixture at -78 °C was added a solution of lithium phenylacetylide (prepared by the reaction of phenylacetylene (3.6 g, 35 mmol) with n BuLi (24 mL, 39 mmol) in THF (20 mL) at -78 °C) slowly for a period of 30 min. and further stirred for 2 h. After warming to room temperature for overnight, HCl aq. (1M) was added and stirred for 3 h. The organic layer was extracted with ether, dried over anhydrous Na₂SO₄, concentrated and purified by silica gel column chromatography (eluted with 3% EtOAc in hexane) to get **1t** as a colorless liquid (6.0 g, 73% yield). 1 H NMR (400 MHz, CDCl₃): δ 1.02-1.10 (m, 10H, Si(CH₃CH₂)₂), 7.33-7.36 (m, 3H, Ar*H*), 7.55-7.62 (m, 3H, Ar*H*), 7.65 (td, J = 1.5, 7.3 Hz, 1H, Ar-*H*), 7.89 (dd, J = 1.5, 7.3 Hz, 1H, Ar-*H*), 8.29 (d, J = 7.2 Hz, 1H, Ar-*H*), 10.16 (s, 1H, C*H*O). 13 C(1 H) NMR (100 MHz, CDCl₃): δ 6.0, 8.1, 92.4, 108.5, 123.4, 128.3, 128.6, 129.9, 132.1, 133.1, 133.4, 137.0, 138.51, 141.3, 192.7. HRMS (CI): m/z Calcd for C₁₉H₂₀OSi: (M+H⁺) 292.1283, found 292.1279.

Synthesis of 2,2'-(dimethylsilanediyl)dibenzaldehyde (1u):

A stirring solution of 2-bromobenzaldehyde diethyl acetal (5.0 g, 19.3 mmol) in THF (60 mL) was added "BuLi (13.2 mL, 1.6 M in THF, 21.2 mmol) drop wise at -78 °C under N₂. After 1 h, dichlorodimethylsilane (1.2 g, 9.65 mmol) was introduced drop wise to the reaction mixture at the same temperature and allowed it to stir for 1 h followed by warming to room temperature with overnight stirring. After completion of the reaction, 1 M HCl was added and stirred for 1 h. Biphasic solution was separated and aqueous layer was washed with ethyl acetate three times. The combined organic layers was dried over Na₂SO₄, concentrated, and purified by recrystallization with toluene to get required compound **1u** as a white solid (2.3 g, 89% yield). **H NMR** (400 MHz, CDCl₃): δ 0.66 (s, 6H, Si(CH₃)₃), 7.55 (td, J = 1.3, 7.4 Hz, 2H, Ar-H), 7.60 (td, J = 1.3, 7.4 Hz, 2H, Ar-H), 7.81 (dd, J = 1.1, 7.4 Hz, 2H, Ar-H), 7.87 (dd, J = 1.1, 7.4 Hz, 2H, Ar-H), 9.92 (s, 2H, CHO). **13C{¹H} NMR**

(100 MHz, CDCl₃): δ 0.0, 129.5, 132.1, 133.3, 136.2, 140.5, 141.6, 193.1. **<u>HRMS</u>** (EI): m/z Calcd for $C_{16}H_{16}O_2Si$: (M^+) 268.0920, found 268.0919.

[3] Optimization for catalytic enantioselective synthesis

1a. Survey of chiral NHCs (Scheme 2)

Scheme 2. Survey of chiral NHC Ligands

General method: A screw cap vial was charged with NHC-salt $\mathbf{Ln}\cdot\mathbf{X}$ (0.022 mmol) and NaO'Bu (0.02 mmol) in toluene or C_6D_6 (1 mL). After 10 min of stirring, Ni(cod)₂ (0.02 mmol) was added. To the stirring solution was added 2-dimethylphenylsilylbenzaldehyde (1a) and allowed to stir (or transferred to J. Young NMR tube) at room temperature (except in case of $\mathbf{L10}\cdot\mathbf{HBF_4}$ and $\mathbf{L11}\cdot\mathbf{HBF_4}$). The progress of reaction was monitored by either GC or ¹H NMR. The resulting reaction mixture was filtered through small pad of silica and concentrated. Formation of 2a was confirmed by spectroscopic data, reported previously for racemic compound. ^{S2} ¹H NMR (400 MHz, CDCl₃): δ 0.46 (s, 3H, SiCH₃), 0.54 (s, 3H, SiCH₃), 6.18 (s, 1H, CHOSi), 7.01–7.07 (m, 1H, Ar-H), 7.27–7.37 (m, 7H, Ar-H), 7.60–7.65 (m, 1H, Ar-H).

Note: Product 2a was sensitive toward silica gel column chromatography as well as HPLC conditions. S4 However, it could be possible to filter quickly through short silica gel column or

recycle GPC, if required. Enantioselectivity for **2a** was measured after converting it (after 100% conversion) into stable product, **3a** by Tamao-Fleming oxidation (see **Scheme S2**, *vide infra*).

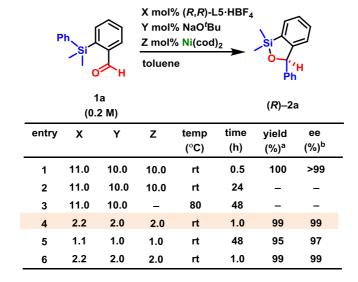
General procedure for Tamao-Fleming oxidation (TFO): To the solution of crude benzoxasilole 2 (for 0.2 mmol) in THF/MeOH (1:1; 1 mL) were added KHCO₃ (40 mg, 0.4 mmol) and H₂O₂ (35% aq, 0.19 mL, 2.0 mmol) at 0 °C. The reaction mixture was stirred to complete the reaction for overnight. The reaction was quenched carefully with aqueous NaHSO₃ solution at 0 °C and extracted well with ethyl acetate (3 x 10 mL), dried over Na₂SO₄, concentrated and purified by silica gel column chromatography (isolated with 20% EtOAc:Hexane) to get 3a as a white solid. (R)-2-(hydroxy(phenyl)methyl)phenol (3a): 1 H NMR (400 MHz, CDCl₃): δ 3.04 (br s, 1H, CHOH), 6.00 (s, 1H, CHOH), 6.82 (td, J = 1.0, 7.4 Hz, Ar-H, 1H), 6.84-6.92 (m, Ar-H, 2H), 7.19 (td, J = 2.0, 7.8 Hz, Ar-H, 1H), 7.30–7.40 (m, 5H, Ar-H), 7.92 (br s, 1H, Ar-OH). ${}^{13}C\{{}^{1}H\}$ NMR (100 MHz, CDCl₃): 8 76.9, 117.2, 119.9, 126.6, 126.8, 128.19, 128.23, 128.7, 129.3, 141.8, 155.4. **HRMS** (ESI) m/z calcd for $C_{13}H_{12}O_2$: (M^+) 200.0837, found 200.0836. The spectral data was identified to that previously reported. S5 Chiral separation: The enantioselectivity was measured by using either chiral HPLC (Chiralpak OD-H column. (10% isopropanol in hexane, Flow rate = 0.5 ml/min, t_R = 22.9 min (S-enantiomer) and 34.7 min (R-enantiomer)) or SFC using Chiralpak ID (Back pressure = 15 MPa, Flow (CO₂) = 2.0 mL/min, Flow (isopropanol) = 0.3 mL/min, 25 °C, λ = 250 nm). Retention time: $t_R = 3.7$ min (S-enantiomer) and 4.0 min (R-enantiomer). The absolute stereochemistry of 2a was assigned by analogy to 2h (determined to be (R) by X-ray with Flack parameter = 0.02(3); when (R,R)-L5 was employed, *vide infra*).

1b. Survey of chiral phosphine ligands (Scheme S1): Phosphine (mono- as well as bidentate) ligands were failed to give any conversion to required products. The reaction procedure was followed as such; To the stirring solution of phosphine ligand and $Ni(cod)_2$ in C_6D_6 (1 mL) was added **1a** and transferred to J. Young NMR tube. The reaction was monitored by ¹H NMR at room temperature for 24 h followed by heating at 80 °C for 18 h.

Scheme S1. Screening of phosphine ligands

1c. Further optimization with L5·HBF₄ (Table 1)

Table 1. Further Optimization with L5·HBF₄



General Procedure: A screw cap vial was charged with (R,R)-**L5·HBF**₄ (1–11 mol%) and NaO^tBu (1–10 mol%) in toluene (1 mL). After 10 min of stirring, Ni(cod)₂ (1–11 mol%) was added. 2-Dimethylphenylsilylbenzaldehyde (1a; 48.0 mg, 0.2 mmol) was added to the stirring solution and allowed to stir at room temperature. The progress of reaction was monitored and yields of 2a were

calculated by GC using *n*-pentadecane as an internal standard. The enantioselectivity was measured for corresponding TFO product, **3a** (see **Scheme S2**, *vide infra*).

Entry 1: Reaction was performed with (R,R)- L5·HBF₄ (12.0 mg, 0.022 mmol, 11 mol%), NaO^tBu (1.9 mg, 0.02 mmol, 10 mol%), and Ni(cod)₂ (5.5 mg, 0.02 mmol, 10 mol%) in 1 mL toluene. reaction was completed in 30 min at room temperature (100% GC yield, 99% ee).

Entry 2: Reaction was performed under identical reaction conditions except Ni(acac)₂ was used as a Ni(II) source in place of Ni(cod)₂ and stirred for 24 h. No reaction took place.

Entry 3: Reaction was performed in the absence of $Ni(cod)_2$ in C_6D_6 . No reaction took place even at 80 °C for 24 h.

Entry 4: Reaction was conducted with 2 mol% catalyst loading $\{(R,R)\text{-L5·HBF}_4 (2.4 \text{ mg}, 0.0044 \text{ mmol}), \text{NaO'Bu} (0.38 \text{ mg}, 0.004 \text{ mmol}), \text{ and Ni(cod)}_2 (1.1 \text{ mg}, 0.004 \text{ mmol})\}$. Reaction was completed in 1 h with 99% GC yield and 99% ee.

Entry 5: Reaction was conducted with 1 mol% catalyst loading $\{(R,R)\text{-L5}\text{-HBF}_4\ (1.2 \text{ mg},\ 0.0022 \text{ mmol}),\ \text{NaO}^t\text{Bu}\ (0.2 \text{ mg},\ 0.002 \text{ mmol}),\ \text{and}\ \text{Ni}(\text{cod})_2\ (0.6 \text{ mg},\ 0.004 \text{ mmol})\}$. Reaction was relatively slow and required longer time and result is denoted as conversion of starting material (95% conversion in 2 days). Ee was measured from aliquot taken from reaction mixture of 95% conversion.

Entry 6: Reaction was conducted under identical reaction condition as entry 4 using with (S,S)-L5·HBF₄, resulted (S)-2a with consistent efficiency.

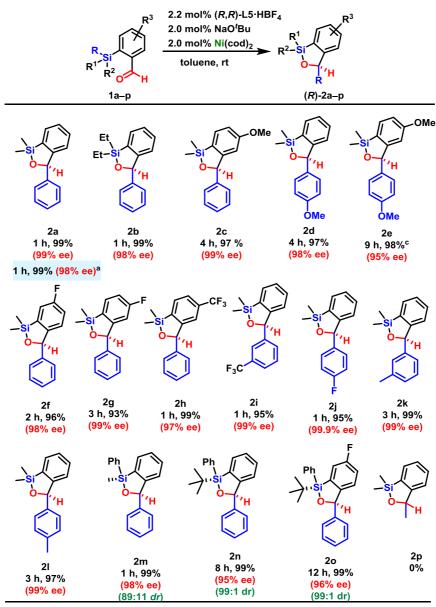
Optimized conditions employed for study of substrate scope: entry 4

[4] Substrate scope

Scope of 3-aryl-2,1-benzoxasilole (Table 2)

General procedures: A reaction tube was charged with (*R*,*R*)-**L5·HBF**₄ (2.2 mol%) and NaO'Bu (2.0 mol%) in toluene (1–10 mL; 0.2 M). After 5 to 10 min of stirring, Ni(cod)₂ (2.0 mol%) was added. *ortho*-Silylbenzaldehyde **1** (0.20–2.0 mmol) was added after 10 min and allowed to stir at room temperature and progress of the reaction was monitored either by GC or ¹H NMR. The products were isolated by short silica gel column chromatography (Wakogel[®] C-300, 5.0 g, eluted with ether). Further purification was carried out either by recrystallization or recycle GPC, if necessary.

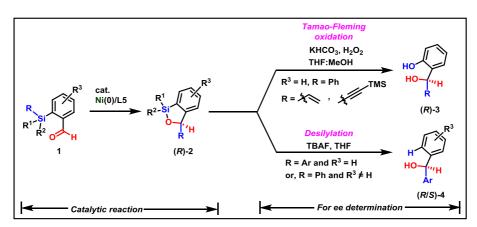
Table 2. Catalytic Enantio- and Diastereoselective Synthesis of 3-Arylbenzoxasiloles



^a Reaction was conducted at 5 mmol scale

Note: Since, 3-arylbenzoxasiloles 2 are sensitive to HPLC; it was converted it into either 3 (Tamao-Fleming oxidation) or 4 (desilylation) product to ascertain the enantioselectivity of corresponding benzoxasiloles, as shown below in Scheme S2. However, enantioselectivities measured after both transformations were consistent, as examined for 2h (vide infra).

Scheme S2. Nickel-catalyzed synthesis of benzoxasilole 2 and further transformations to 3 and 4



Reaction of 1a giving (*R*)-2a (Table 1, entry 3; 99% yield, 99% ee): The general procedure was followed with 2-(dimethyl(phenyl)silyl)benzaldehyde (1a) (48.0 mg, 0.20 mmol) for 1.0 h at room temperature. Reaction mixture was purified by short silica gel column chromatography giving (*R*)-1,1-dimethyl-3-phenyl-1,3-dihydrobenzo[c][1,2]oxasilole (2a) in 99% yield (47.5 mg, 0.198 mmol). $[\alpha]_D^{23} = (-) 143.3^\circ$ (c = 0.2, in CHCl₃). The absolute stereochemistry of 2a was assigned by analogy to 2h and enantioselectivity was evaluated after converting it into corresponding oxidation product 3a. It was synthesized from 2a as mentioned in general procedure for TFO in 81% yield. Chiral separation: The enantioselectivity was determined by SFC using Chiralpak ID (Back pressure = 15 MPa, Flow (CO₂) = 2.0 mL/min, Flow (isopropanol) = 0.3 mL/min, 25 °C, λ = 250 nm). Retention time: t_R = 3.7 min (minor *S*-enantiomer) and 4.0 min (major *R*-enantiomer).

Gram scale synthesis of 2a: Reaction procedure was followed as the optimized conditions (Table 1, entry 4) using 2-dimethylphenylsilylbenzaldehyde (**1a**; 1.2 g, 5.0 mmol), (*R*,*R*)-**L5·HBF**₄ (54.0 mg, 0.10 mmol), NaO'Bu (9.6 mg, 0.10 mmol) and Ni(cod)₂ (27.2 mg, 0.10 mmol) in toluene (25 mL). The reaction was completed in 1 h with consistent result. Enantioselectivity was measured after converting it into **3a**.

Reaction of 1b giving (*R*)-2b (99% yield, 98% ee): The general procedure was followed with 2-(diethyl(phenyl)silyl)benzaldehyde (1b) (54.0 mg, 0.2 mmol) and reaction was completed within 1.0 h at room temperature. Purification by short silica gel column chromatography gave (*R*)-1,1-diethyl-3-phenyl-1,3-dihydrobenzo[c][1,2]oxasilole (2b) as a pale yellow liquid in 99% yield (53.0 mg, 0.198 mmol). Spectroscopic data of 2b was identified to that previously reported for racemic compound. See 2b: (99% yield, pale yellow liquid). HNMR (400 MHz, CDCl₃): δ 0.85–1.13 (m, 10H, 2 x SiC H_2 C H_3), 6.16 (s, 1H, CHOSi), 7.02 (d, J = 6.6 Hz, 1H, Ar-H), 7.26–7.36 (m, 7H, Ar-H), 7.61 (d, J = 6.6 Hz, 1H, Ar-H). [α]_D²³ = (-) 109.1° (c = 0.1, in CHCl₃). The absolute stereochemistry of 2b was assigned by analogy to 2h and enantioselectivity was evaluated after converting it into 3a (80% yield, 98% ee). Chiral separation The enantioselectivity of 3a was determined by SFC using chiralpak ID (Back pressure = 15 MPa, Flow (CO₂) = 2.0 mL/min, Flow (isopropanol) = 0.3 mL/min, 25 °C, λ = 250 nm). Retention time: t_R = 4.1 min (minor *S*-enantiomer) and 4.4 min (major *R*-enantiomer).

Reaction of 1c giving (*R*)-2c (97% yield, 99% ee): The general procedure was followed with 2-(dimethyl(phenyl)silyl)-5-methoxybenzaldehyde (1c) (54.0 mg, 0.2 mmol) and reaction was conducted at room temperature for 4 h. Purification by short silica gel column chromatography gave (*R*)-5-methoxy-1,1-dimethyl-3-phenyl-1,3-dihydrobenzo[c][1,2]oxasilole (2c) as a white solid in 97% yield (52.0 mg, 0.192 mmol). Spectroscopic data of 2c was identified to that previously reported for racemic compound. See 1H NMR (400 MHz, CDCl₃): δ 0.40 (s, 3H, SiCH₃), 0.48 (s, 3H, SiCH₃), 3.68 (s, 3H, OCH₃), 6.09 (s, 1H, CHOSi), 6.49 (d, J = 2.0 Hz, 1H, Ar-H), 6.86 (dd, J = 2.4, 8.0 Hz, 1H, Ar-H), 7.21–7.33 (m, 5H, Ar-H), 7.52 (d, J = 8.0 Hz, 1H, Ar-H). [α]_D 23 = (–) 61.3° (c = 0.1, in CHCl₃). The absolute stereochemistry of 2c was assigned by analogy to 2h and

enantioselectivity was evaluated after converting it into desilylation product 4c (75% yield, 99% ee).

Experimental procedure for desilylation: For example, (R)-(3-methoxyphenyl)(phenyl)methanol (**4c**): To a reaction tube charged with **2c** (53.0 mg, 0.19 mmol) in 0.5 mL THF was added TBAF (1 M, 0.2 mL) at room temperature and stirred for 5 h to complete the reaction. The reaction mixture was quenched by aqueous NH₄Cl and stirred for 30 min. The biphasic mixture was separated and extracted with ethyl acetate (3 x 10 mL), dried over Na₂SO₄, concentrated and purified by silica gel column chromatography (with 20% EtOAc:Hexane). Spectroscopic data of **4c** was identified to that previously reported. See 1 MNR (400 MHz, CDCl₃): δ 2.14 (br s, CHOH, 1H), 3.68 (s, OCH₃, 3H), 5.71 (s, CHOH, 1H), 6.67–6.73 (m, Ar-H, 1H), 6.82–6.87 (m, Ar-H, 2H), 7.12–7.18 (m, Ar-H, 2H), 7.20–7.29 (m, Ar-H, 4H). Chiral separation: The enantioselectivity was determined by SFC using Chiralpak IA (Back pressure = 15 MPa, Flow (CO₂) = 2.0 mL/min, Flow (IPA) = 0.3 mL/min, 25 °C, λ = 250 nm). Retention time: t_R = 7.0 min (minor S-enantiomer) and 7.3 min (major R-enantiomer).

Reaction of 1d giving (*R*)-2d (97% yield, 98% ee): The general procedure was followed with 2-dimethyl(4-methoxyphenyl)silylbenzaldehyde (1d) (54.0 mg, 0.2 mmol) and reaction was conducted at room temperature for 4 h. Purification by short silica gel column chromatography gave (*R*)-3-(4-methoxyphenyl)-1,1-dimethyl-1,3-dihydrobenzo[c][1,2]oxasilole (2d) as a pale yellow solid in 97% yield (52.5 mg, 0.19 mmol). Spectroscopic data of 2d was identified to that previously reported for racemic compound. S2 1H NMR (400 MHz, CDCl₃): δ 0.47 (d, J = 2.4 Hz, 3H, SiCH₃), 0.54 (d, J = 2.4 Hz, 3H, SiCH₃), 3.81 (s, 3H, OCH₃), 6.16 (s, 1H, CHOSi), 6.89 (dt, J = 2.4, 8.7 Hz, 2H, Ar-H), 7.01–7.05 (m, 1H, Ar-H), 7.21 (dt, J = 2.4, 8.7 Hz, 2H, Ar-H), 7.30–7.38 (m, 2H, Ar-H), 7.61–7.67 (m, 1H, Ar-H). [α]_D²³ = (-) 115.0 (c = 0.1, in CHCl₃). The absolute stereochemistry of 2d was assigned by analogy to 2h and enantioselectivity was evaluated after converting it into desilylation product 4d (75% yield, 98% ee). Spectroscopic data of 4d was identified to that previously reported. S7 1H NMR (400 MHz, CDCl₃): δ 2.15 (br s, 1H, CHOH), 3.79 (s, 3H, OCH₃), 5.81 (s, 1H, CHOH), 6.86 (d, J = 8.6 Hz, 2H, Ar-H), 7.23–7.40 (m, 7H, Ar-H). Chiral separation:

The enantioselectivity for **4d** was determined by SFC using Chiralpak ID (Back pressure = 15 MPa, Flow (CO₂) = 2.0 mL/min, Flow (IPA) = 0.3 mL/min, 25 °C, λ = 250 nm). Retention time: t_R = 5.8 min (minor *R*-enantiomer) and 6.0 min (major *S*-enantiomer).

Reaction of 1e giving (R)-2e (98% yield, 95% ee): The general procedure was followed with 5-methoxy-2-((4-methoxyphenyl)dimethylsilyl)benzaldehyde (1e) (120.0 mg, 0.4 mmol) using (R, R)-L5 (10 mg, 0.018 mmol), NaO'Bu (1.6 mg, 0.016 mmol), Ni(cod)₂ (4.3 mg, 0.016 mmol) and reaction was conducted at room temperature for 9 h. Purification by short silica gel column chromatography gave (R)-5-methoxy-3-(4-methoxyphenyl)-1,1-dimethyl-1,3-dihydrobenzo[c][1,2] -oxasilole (2e) as pale yellow thick liquid in 98% yield (117 mg, 0.39 mmol). The spectroscopic data was identified to that reported data for racemic compound. S2 1H NMR (400 MHz, CDCl₃): δ 0.43 (s, 3H, SiCH₃), 0.50 (s, 3H, SiCH₃), 3.73 (s, 3H, OCH₃), 3.80 (s, 3H, OCH₃), 6.10 (s, 1H, CHOSi), 6.52 (s, 1H, Ar-H), 6.89 (dd, J = 2.4, 8.0 Hz, 1H, Ar-H), 7.20 (d, J = 8.8 Hz, 3H, Ar-H), 7.52 (d, J = 8.3Hz, 1H, Ar-H). $[\alpha]_D^{23} = (-) 27.4^\circ$ (c = 0.1, in CHCl₃). The absolute stereochemistry of **2e** was assigned by analogy to 2h and enantioselectivity was evaluated after converting it into desilylation product **4e** (50% yield, 95% ee). 1 **H NMR** (400 MHz, CDCl₃): δ 2.16 (d, J = 3.2 Hz, 1H, CHOH), 3.76 (s, 6H, 2 x OCH₃), 5.75 (d, J = 3.2 Hz, 1H, CHOH), 6.77 (dd, J = 1.9, 8.0 Hz, 1H Ar-H), 6.83(d, J = 8.0 Hz, 2H, Ar-H), 6.91 (d, J = 8.0 Hz, 1H, Ar-H), 6.92 (s, 1H, Ar-H), 7.19-7.30 (m, 3H, 4H)Ar-H). ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 55.2, 55.24, 75.7, 111.9, 112.8, 113.8, 118.7, 127.9, 129.4, 136.0, 145.7, 159.0, 159.7. **HRMS** (ESI) m/z calcd for $C_{15}H_{16}O_3$ 244.1099: (M⁺) found 244.1097. Chiral separation: The enantioselectivity for 4e was determined by SFC using Chiralpak ID (Back pressure = 15MPa, Flow (CO₂) = 2.0 mL/min, Flow (IPA) = 0.3 mL/min, 25 °C, λ = 250 nm). Retention time: $t_R = 7.0 \text{ min (minor } S\text{-enantiomer)}$ and 7.4 min (major R-enantiomer).

Reaction of 1f giving (*R*)-2**f** (96% yield, 98% ee): The general procedure was followed with 2-dimethylphenylsilyl-4-fluorobenzaldehyde (1**f**) (51.0 mg, 0.2 mmol) and reaction was conducted at room temperature for 2 h. Purification by short silica gel column chromatography gave (*R*)-6-fluoro-1,1-dimethyl-3-phenyl-1,3-dihydrobenzo[c][1,2]oxasilole (2**f**) as pale yellow oil in 96% yield (49.0 mg, 0.19 mmol). The spectroscopic data was identified to that previously reported for racemic compound. ^{S2} 1H NMR (400 MHz, CDCl₃): δ 0.44 (s, 3H, SiCH₃), 0.52 (s, 3H, SiCH₃), 6.12 (s, 1H, CHOSi), 6.96–6.99 (m, 2H, Ar-H), 7.23–7.35 (m, 6H, Ar-H). [α]_D²³ = (–) 110.9 (c = 0.1, in CHCl₃). The absolute stereochemistry of 2**f** was assigned by analogy to 2**h** and enantioselectivity was evaluated after converting it into desilylation product 4**f** (70% yield, 98% ee). The spectroscopic data of 4**f** was identified to that previously reported. ^{S7} 1H NMR (400 MHz, CDCl₃): δ 2.20 (br s, 1H, CHOH), 5.71 (d, J = 2.6 Hz, 1H, CHOH), two set of double at 6.90 and 6.93 (d, J = 8.0 Hz, 2H Ar-H), 7.15–7.28 (m, 7H, Ar-H). Chiral separation: The enantioselectivity for 4**f** was determined by SFC using Chiralpak ID (Back pressure = 15MPa, Flow (CO₂) = 4.0 mL/min, Flow (CH₂Cl₂) = 0.6 mL/min, 25 °C, λ = 250 nm). Retention time: t_R = 3.4 min (major R-enantiomer) and 3.9 min (minor S-enantiomer).

Reaction of 1g giving (*R*)-2g (93% yield, 99% ee): The general procedure was followed with 2-(dimethyl(phenyl)silyl)-5-fluorobenzaldehyde (1g) (129 mg, 0.5 mmol) and reaction was conducted at room temperature for 3 h. Purification by short silica gel column chromatography gave (*R*)-5-fluoro-1,1-dimethyl-3-phenyl-1,3-dihydrobenzo[c][1,2]oxasilole (2g) as pale yellow oil in 93% yield (120 mg, 0.47 mmol). 1 H NMR (400 MHz, CDCl₃): δ 0.44 (s, 3H, SiCH₃), 0.52 (s, 3H, SiCH₃), 6.11 (s, 1H, CHOH), 6.69 (d, J = 9.9 Hz, 1H, Ar-H), 7.00 (td, J = 2.3, 7.6 Hz, 1H, Ar-H),

7.24–7.36 (m, 5H, Ar-H), 7.55 (t, J = 6.9 Hz, 1H, Ar-H). $^{13}\text{C}\{^1\text{H}\}$ NMR (100 MHz, CDCl₃): δ 0.6, 1.3, 83.6, 110.7 (d, $J_{\text{CF}} = 21.0$ Hz), 115.0 (d, $J_{\text{CF}} = 21.0$ Hz), 127.0, 128.0, 128.6, 130.4, 132.3 (d, $J_{\text{CF}} = 9.0$ Hz), 143.1, 155.3 (d, $J_{\text{CF}} = 7.3$ Hz), 164.4 (d, $J_{\text{CF}} = 248.0$ Hz). $^{14}\text{HRMS}$ (EI): m/z Calcd for $C_{15}H_{15}FOSi$: (M^+) 258.0876, found 258.0871. [α] $_0^{23} = (-)$ 131.7 (c = 0.1, in CHCl₃). The absolute stereochemistry of **2g** was assigned by analogy to **2h** and enantioselectivity of **2g** was evaluated after converting it into desilylation product **4g** (76% yield, 99% ee). ^{1}H NMR (400 MHz, CDCl₃): δ 2.26 (d, J = 3.1 Hz, 1H, CHOH), 5.83 (J = 3.1 Hz, 1H, CHOH), 6.95 (t, J = 8.0 Hz, 1H, Ar-H), 7.14 (t, J = 8.0 Hz, 2H, Ar-H), 7.29-7.37 (m, 6H, Ar-H). Spectroscopic data of **4g** was identified to that previously reported. Section Chiral separation: The enantioselectivity was determined by SFC using Chiral pak IC (Back pressure = 15MPa, Flow (CO₂) = 4.0 mL/min, Flow (isopropanol) = 0.2 mL/min, 25 °C, $\lambda = 250$ nm). Retention time: $t_R = 3.6$ min (major R-enantiomer) and 3.9 min (minor S-enantiomer).

Reaction of 1h giving (*R*)-2h (99% yield, 97% ee): The general procedure was followed with 2-dimethylphenylsilyl-5-trifluoromethylbenzaldehyde (1h) (61.0 mg, 0.2 mmol) and reaction was conducted at room temperature for 2 h. Purification by silica gel column chromatography gave (*R*)-1,1-dimethyl-3-phenyl-5-(trifluoromethyl)-1,3-dihydrobenzo[c][1,2]oxasilole (2h) as a pale yellow solid in 99% yield (60.0 mg, 0.19 mmol). Spectroscopic data of 2h was identified to that previously reported. S2 HNMR (400 MHz, CDCl₃): δ 0.48 (s, 3H, SiCH₃), 0.56 (s, 3H, SiCH₃), 6.19 (s, 1H, CHOSi), 7.27–7.40 (m, 6H, Ar-H), 7.57 (d, J = 7.2 Hz, 1H, Ar-H), 7.73 (d, J = 7.2 Hz, 1H, Ar-H). [α]_D²³ = (-) 103.8 (c = 0.1, in CHCl₃). The absolute stereochemistry of 2h was assigned by X-ray analysis (Flack parameter 0.02(3)) and enantioselectivity was evaluated after converting it into corresponding oxidation product 3h as well as desilylation product 4h prepared as described above. *Enantioselectivities were consistent in both products* 3h and 4h (97% ee).

(R)-2-(hydroxy(phenyl)methyl)-4-(trifluoromethyl)phenol (3h, 77% yield, 97% ee): ¹H NMR (400

MHz, CDCl₃): δ 3.12 (br s, 1H, CHO*H*)), 6.02 (s, 1H, C*H*OH), 6.95 (d, J = 8.5 Hz, 1H, Ar-*H*), 7.12 (d, J = 1.6 Hz, 1H, Ar-*H*), 7.33–7.41 (m, 5H, Ar-*H*), 7.44 (dd, J = 1.6, 8.5 Hz, 1H, Ar-*H*), 8.52 (br s, 1H, Ar-O*H*). ${}^{13}\text{C}\{{}^{1}\text{H}\}$ NMR (100 MHz, CDCl₃): δ 76.9, 117.6, 122.2 (q, J_{CF} = 33.1 Hz), 125.4 (q, J_{CF} = 3.7 Hz), 125.6 (q, J_{CF} = 271.0 Hz), 126.5 (q, J_{CF} = 3.7 Hz), 126.6, 126.8, 128.7, 129.0, 141.0, 158.3. HRMS (FAB) m/z calcd for $C_{14}H_{10}F_{3}O_{2}$: (M⁺-H) 267.0632, found 267.0632. Chiral separation: The enantioselectivity of 3h was determined by HPLC using Chiralpak OD-H (IPA/Hexane 1:9; flow 0.5 mL/min, λ = 254 nm); retention time: t_{R} = 16.2 min (minor *S*-enantiomer) and 19.9 min (major *R*-enantiomer).

(*R*)-phenyl(3-(trifluoromethyl)phenyl)methanol (**4h**, 58% yield, 97% ee): Spectroscopic data of **4h** was identified to that previously reported for racemic compound. So $\frac{1}{2}$ H NMR (400 MHz, CDCl₃): δ 2.27 (d, J = 3.5 Hz, 1H, CHOH), 5.89 (d, J = 3.5 Hz, 1H, CHOH), 7.25–7.37 (m, 5H, Ar-H), 7.44 (dd, J = 7.9, 7.9 Hz, 1H, Ar-H), 7.53 (dd, J = 8.2, 8.2 Hz, 2H, Ar-H), 7.70 (br s, 1H, Ar-H). Chiral separation: The enantioselectivity of **4h** was determined by HPLC using Chiralpak OB-H (IPA/Hexane 1:9; flow 1.0 mL/min, δ = 254 nm); retention time: δ = 9.2 min (major *R*-enantiomer) and 21.1 min (minor *S*-enantiomer).

Reaction of 1i giving (*R*)-2i (95% yield, 99% ee): The general procedure was followed with 2-(dimethyl(3-(trifluoromethyl)phenyl)silyl)benzaldehyde (1i) (154.0 mg, 0.5 mmol) and reaction was conducted at room temperature for 1 h. Purification by short silica gel column chromatography gave (*R*)-1,1-dimethyl-3-(3-(trifluoromethyl)phenyl)-1,3-dihydrobenzo[c][1,2]oxasilole (2i) as colorless liquid in 95% yield (147.0 mg, 0.48 mmol). ¹H NMR (400 MHz, CDCl₃): δ 0.47 (d, J = 2.3 Hz, 3H, SiCH₃), 0.55 (d, J = 2.3 Hz, 3H, SiCH₃), 6.21 (1H, CHOSi), 7.02 (br s, 1H, Ar-H), 7.32–7.36 (m, 2H, Ar-H), 7.46 (br s, 2H, Ar-H), 7.52–7.68 (m, 2H, Ar-H), 7.60–7.65 (m, 1H, Ar-H). ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 0.4, 1.1, 83.3, 123.5, 123.8 (q, $J_{CF} = 3.7$ Hz), 124.0 (d, $J_{CF} = 272.0$ Hz), 124.6 (q, $J_{CF} = 3.7$ Hz), 127.5, 129.0, 130.0, 130.3, 130.8 (d, $J_{CF} = 32.0$ Hz), 130.9, 135.1, 144.8, 151.5 (d, $J_{CF} = 9.0$ Hz). HRMS (EI): m/z Calcd for C₁₆H₁₅F₃OSi: (M⁺) 308.0844, found 308.0841. [α]_D²³ = (-) 116.3° (c = 0.1, in CHCl₃). The absolute stereochemistry of 2i was assigned by analogy to 2h and enantioselectivity was evaluated after converting it into desilylation product 4h.

(73% yield, 99% ee). <u>Chiral separation</u>: The enantioselectivity was determined by means of SFC using Chiralpak IB (Back pressure = 15 MPa, Flow (CO₂) = 5.0 mL/min, Flow (isopropanol) = 0.2 mL/min, 25 °C, λ = 230 nm). Retention time: t_R = 3.4 min (major *S*-enantiomer) and 3.8 min (minor *R*-enantiomer).

Reaction of 1j giving (*R*)-2**j** (95% yield, 99.9% ee): The general procedure was followed with 2-((4-fluorophenyl)dimethylsilyl)benzaldehyde (1**j**) (129.0 mg, 0.5 mmol) and reaction was conducted at room temperature for 1 h. Purification by short silica gel column chromatography gave (*R*)-3-(4-fluorophenyl)-1,1-dimethyl-1,3-dihydrobenzo[c][1,2]oxasilole (2**j**) as pale yellow oil in 98% yield (127.0 mg, 0.47 mmol). HNMR (400 MHz, CDCl₃): δ 0.45 (s, 3H, SiC*H*₃), 0.52 (s, 3H, SiC*H*₃), 6.15 (s, 1H, C*H*OSi), 6.96–7.06 (m, 3H, Ar-*H*), 7.20–7.28 (m, 2H, Ar-*H*), 7.29–7.36 (m, 2H, Ar-*H*), 7.59–7.65 (m, 1H, Ar-*H*). HC{1H} NMR (100 MHz, CDCl₃): δ 0.4, 1.2, 83.3, 115.3 (d, J_{CF} = 22.8 Hz), 123.7, 127.2, 128.8 (d, J_{CF} = 7.9 Hz), 129.8, 130.7, 135.1, 139.7 (d, J_{CF} = 3.0 Hz), 152.2, 162.3 (d, J_{CF} = 248.0 Hz). HRMS (EI): m/z Calcd for C₁₅H₁₅FOSi: (M⁺) 258.0876, found 258.0878. [α]_D²³ = (-) 112.2 (c = 0.16, in CHCl₃). The absolute stereochemistry of 2**j** was assigned by analogy to 2**h** and enantioselectivity was evaluated after converting it into desilylation product 4**f** (70% yield, 99.9% ee). Chiral separation: The enantioselectivity for 4**f** was determined by SFC using Chiralpak ID (Back pressure = 15MPa, Flow (CO₂) = 3.5 mL/min, Flow (CH₂Cl₂) = 0.6 mL/min, 25 °C, λ = 250 nm). Retention time: t_R = 2.8 min (minor *R*-enantiomer) and 3.1 min (major *S*-enantiomer).

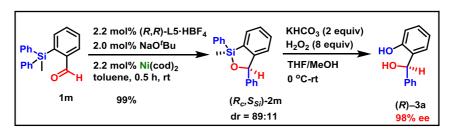
Reaction of 1k giving (*R*)-2k (99% yield, 99% ee): The general procedure was followed with 2-dimethyl(3-methylphenyl)silylbenzaldehyde (1k) (51.0 mg, 0.2 mmol) and reaction was conducted

at room temperature for 3 h. Purification by short silica gel column chromatography gave (R)-1,1-dimethyl-3-(m-tolyl)-1,3-dihydrobenzo[c][1,2]oxasilole ($2\mathbf{k}$) as pale yellow oil in 99% yield (50.0 mg, 0.2 mmol). 1 H NMR (400 MHz, CDCl₃): δ 0.44 (s, 3H, SiCH₃), 0.53 (s, 3H, SiCH₃), 2.33 (s, 3H, Ar-CH₃), 6.13 (s, 1H, CHOSi), 7.00–7.12 (m, 4H, Ar-H), 7.22 (t, J = 7.8 Hz, 1H, Ar-H), 7.28–7.35 (m, 2H, Ar-H), 7.58–7.64 (m, 1H, Ar-H). [α]_D²³ = (–) 133.3° (c = 0.1, in CHCl₃). Spectroscopic data of $2\mathbf{k}$ was identified to that previously reported for racemic compound. Should after converting it into desilylation product $4\mathbf{k}$. (74% yield, 96% ee). 1 H NMR (400 MHz, CDCl₃): δ 2.24 (br s, 1H, CHOH), 2.33 (s, 3H, Ar-CH₃), 5.79 (s, 1H, CHOH), 7.07 (d, J = 7.3 Hz, 1H, Ar-H), 7.14-7.28 (m, 4H, Ar-H), 7.30-7.40 (m, 4H, Ar-H). Spectroscopic data of $4\mathbf{k}$ was identified to that previously reported. 57 Chiral separation: The enantioselectivity was determined by means of SFC using Chiralpak IA (Back pressure = 15 MPa, Flow (CO₂) = 2.0 mL/min, Flow (isopropanol) = 0.3 mL/min, 25 °C, λ = 250 nm). Retention time: t_R = 5.6 min (major S-enantiomer) and 5.9 min (minor R-enantiomer).

Reaction of 11 giving (*R*)-21 (97% yield, 99% ee): The general procedure was followed with 2-dimethyl(4-methylphenyl)silylbenzaldehyde (11) (50.5 mg, 0.2 mmol) and reaction was conducted at room temperature for 3 h. Purification by silica gel column chromatography gave (*R*)-1,1-dimethyl-3-(*p*-tolyl)-1,3-dihydrobenzo[c][1,2]oxasilole (21) as pale yellow oil in 97% yield (49.0 mg, 0.19 mmol). Spectroscopic data of 21 was identified to that previously reported. ⁵² 1 H NMR (400 MHz, CDCl₃): δ 0.44 (s, 3H, SiCH₃), 0.51 (s, 3H, SiCH₃), 2.33 (s, 3H, Ar-CH₃), 6.14 (, 1H, CHO), 6.99–7.05 (m, 1H, Ar-H), 7.11–7.18 (m, 4H, Ar-H), 7.28–7.34 (m, 2H, Ar-H), 7.58–7.63 (m, 1H, Ar-H). [α]_D²³ = (–) 121.7 (c = 0.1, in CHCl₃). The absolute stereochemistry of 21 was assigned by analogy to 2h and enantioselectivity of 2l was evaluated after converting it into desilylation product 41 (75% yield, 99% ee). 1 H NMR (400 MHz, CDCl₃): δ 2.21 (s, 1H, CHOH), 2.30 (s, 3H, CH₃), 5.77 (s, 1H, CHOH), 7.11 (d, J = 8.0 Hz, 2H, Ar-H), 7.19-7.25 (m, 3H, Ar-H), 7.26-7.37 (m, 4H, Ar-H). Spectroscopic data of 41 was identified to that previously reported. ⁵⁷

<u>Chiral separation</u>: The enantioselectivity was determined by SFC using Chiralpak IB (Back pressure = 15MPa, Flow (CO₂) = 3.0 mL/min, Flow (isopropanol) = 0.3 mL/min, 25 °C, λ = 250 nm). Retention time: t_R = 3.9 min (major *S*-enantiomer) and 4.2 min (minor *R*-enantiomer).

Diastereoselective synthesis of 3-arylbenzoxasiloles (2m-o)

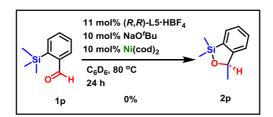


Reaction of 1m giving ($R_{\rm C}$, $S_{\rm Si}$)-2m (99% yield, 89:11 dr, 98% ee): The general procedure for catalytic reaction was slightly modified. A reaction tube charged with (R,R)-L5·HBF₄ (12.0 mg, 0.022 mmol), NaO'Bu (1.9 mg, 0.02 mmol) and Ni(cod)₂ (6.0 mg, 0.022 mmol) in toluene (5.0 mL) was stirred for 15 min. 2-Diphenylmethylsilylbenzaldehyde (1m) (300.0 mg, 1.0 mmol) was added and stirred at room temperature for 0.5 h. Purification by short silica gel column chromatography gave 1-methyl-1,3-diphenyl-1,3-dihydrobenzo[c][1,2]oxasilole (2m) as pale yellow oil in 99% yield (296.0 mg, 0.2 mmol) as a mixture of diastereomers (89:11 dr). 1 H NMR (400 MHz, C₆D₆): δ 0.43 (s, 3H, SiCH₃, minor), 0.48 (s, 3H, SiCH₃, major), 6.03 (s, 1H, CHOSi, major), 6.05 (s, 1H, CHOSi, major), 6.72–7.76 (m, 1H each for both diastereomers, Ar-H), 6.83–6.92 (m, 3H each for both diastereomers, Ar-H), 6.92–7.00 (m, 5H each for both diastereomers, Ar-H), 7.07 (d, J = 7.0Hz, 2H, Ar-H, minor), 7.09 (d, J = 7.0 Hz, 2H, Ar-H, major), 7.23–7.29 (m, 1H each for both diastereoisomers, Ar-H), 7.38–7.46 (m, 2H each for both diastereomers, Ar-H). ¹³C{¹H} NMR (100 MHz, CDCl₃): $\delta -0.7_{\text{minor}}$, -0.5_{major} , 85.1_{minor} , 85.3_{major} , 124.9, 128.1, 128.2, 128.5, 128.7, 128.9, 129.2, 129.3, 130.8, 131.1, 132.1, 132.2, 134.6, 134.9, 135.2, 137.5, 144.9, 145.1, 154.5. Several carbons of phenyl groups are overlapped. $[\alpha]_D^{23} = (-) 176.3$ (c = 0.1, in CHCl₃). The absolute stereochemistries of **2m** was assigned to be (R_c, S_{Si}) by analogy to **2n** (determined by X-ray analysis, vide infra). The enantioselectivity of 2m with respect to carbon-stereogenic centre was evaluated after converting it into 3a using TFO conditions as employed previously (98% ee at carbon) for 2a and 2b. Chiral separation: The enantioselectivity was determined by SFC using Chiralpak ID (Back pressure = 15 MPa, Flow (CO₂) = 2.0 mL/min, Flow (isopropanol) = 0.3 mL/min, 25 $^{\circ}$ C, λ = 250 nm). Retention time: $t_R = 4.0 \text{ min (minor } S\text{-enantiomer)}$ and 4.5 min (major R-enantiomer).

Reaction of 1n giving (R_C, S_{Si}) **-2n** (99% yield, 99:1 dr, 95% ee): A reaction tube was charged with (R,R)-L5·HBF₄ (24.0 mg, 0.044 mmol), NaO^tBu (3.8 mg, 0.04 mmol) and Ni(cod)₂ (12.0 mg, 0.044 mmol) in toluene (10.0 mL) was stirred for 15 min. 2-(tert-Butyldiphenylsilyl)benzaldehyde (1n) (720.0 mg, 2.0 mmol) was added and stirred at room temperature for 8 h. Purification by short silica gel column chromatography gave 1-(tert-butyl)-1,3-diphenyl-1,3-dihydrobenzo[c][1,2]oxasilole (2n) as a while solid in 99% yield (712.0 mg, 1.9 mmol) as a mixture of diastereomers (99:1 dr). It was recrystallized (in hexane, -20 °C) to a single diastereomer. ¹H NMR (400 MHz, CDCl₃): δ 1.14 (s, 9H, $C(CH_3)_3$, 6.30 (s, 1H, CHOSi), 7.06 (d, J = 6.0 Hz, 1H, Ar-H), 7.15–7.33 (m, 5H, Ar-H), 7.32–7.56 (m, 5H, Ar-H), 7.82 (d, J = 6.0 Hz, 2H, Ar-H), 7.88 (d, J = 6.0 Hz, 1H, Ar-H). 13 C{ 1 H} NMR (100 MHz, CDCl₃): δ 19.8, 25.7, 85.2, 123.9, 127.2, 127.3, 127.7, 127.8, 128.4, 129.8, 130.0, 131.7, 134.1, 134.3, 143.6, 153.5. **HRMS** (ESI): m/z Calcd for C₂₃H₂₄OSiNa: (M⁺+Na) 367.1494, found 367.1481. $[\alpha]_D^{23} = (-) 172.2^{\circ}$ (c = 0.1, in CHCl₃). The absolute stereochemistries ($S_{(Si)}, R_{(C)}$) of 2n was confirmed by X-ray analysis (Flack parameter 0.03(15). The enantioselectivity with respect to carbon centre was evaluated after converting it into 3a using TFO conditions as employed previously. Chiral separation: The enantioselectivity of 3a was determined by SFC using Chiralpak ID (Back pressure = 15 MPa, Flow (CO₂) = 2.0 mL/min, Flow (isopropanol) = 0.3 mL/min, 25 °C, λ = 250 nm). Retention time: t_R = 3.7 min (minor *S*-enantiomer) and 3.9 min (major *R*-enantiomer).

Reaction of 10 giving (R_C , S_{Si})-20 (99% yield, 99:1 dr, 96% ee): A reaction tube was charged with (R,R)- L5·HBF₄ (12.0 mg, 0.022 mmol), NaO^tBu (1.9 mg, 0.02 mmol) and Ni(cod)₂ (6.0 mg, 0.022 mmol) in toluene (5.0 mL) was stirred for 15 min. 2-(*tert*-Butyldiphenylsilyl)-5-fluorobenzaldehyde

(10) (362.0 mg, 1.0 mmol) was added and stirred at room temperature for 12 h. Purification by short silica gel column chromatography gave 1-(tert-butyl)-6-fluoro-1,3-diphenyl-1,3-dihydrobenzo[c]-[1,2]oxasilole (20) as a while solid in 99% yield (358.0 mg, 0.99 mmol) as a mixture of diastereomers (99:1 dr). It was recrystallized (in hexane, -20 °C) to single diastereomer. ¹H NMR (400 MHz, CDCl₃): δ 1.06 (s, 9H, C(CH₃)₃), 6.20 (s, 1H, CHOSi), 6.91–6.97 (m, 1H, Ar-H), 7.00 (td, J = 2.1, 8.7 Hz, 1H, Ar-H), 7.10-7.17 (m, 2H, Ar-H), 7.19-7.27 (m, 3H, Ar-H), 7.35-7.48 (m, 4H, Ar-H)Ar-H), 7.73 (dd, J = 1.7, 7.0 Hz, 2H, Ar-H). ${}^{13}C\{{}^{1}H\}$ NMR (100 MHz, CDCl₃): δ 19.8, 25.7, 84.7, $117.3 \text{ (d, } J_{\text{CF}} = 15.4 \text{ Hz), } 117.5 \text{ (d, } J_{\text{CF}} = 17.0 \text{ Hz), } 125.6 \text{ (d, } J_{\text{CF}} = 7.3 \text{ Hz), } 127.2, 127.9, 128.0, 128.5, \\ I = 12.4 \text{ Hz), } 12.2 \text{ (d, } J_{\text{CF}} = 17.0 \text{ Hz), } 12.2 \text{ (d, } J_{\text{CF}} =$ 130.1, 133.6, 134.0, 134.5 (d, $J_{CF} = 5.7 \text{ Hz}$), 143.3, 148.9 (d, $J_{CF} = 2.1 \text{ Hz}$), 162.2 (d, $J_{CF} = 246.4 \text{ Hz}$). **<u>HRMS</u>** (EI): m/z Calcd for $C_{23}H_{23}FOSi$: (M^+) 362.1502, found 362.1501. $[\alpha]_D^{23} = (-)$ 156.8° (c = -)0.1, in CHCl₃). The absolute stereochemistries of 20 was assigned by analogy to 2n and the enantioselectivity with respect to carbon centre was evaluated after converting it into desilylation product 4f using TBAF (83% yield, 96% ee). The spectroscopic data of 4f was identified to that previously reported as obtained from 2f. Chiral separation: The enantioselectivity of 4f, obtained from 20 was determined by SFC using Chiralpak ID (Back pressure = 15 MPa, Flow (CO₂) = 4.0 mL/min, Flow (CH₂Cl₂) = 0.6 mL/min, 25 °C, λ = 250). Retention time: t_R = 3.1 min (major *R*-enantiomer) and 3.9 min (minor *S*-enantiomer).



Reaction of 1p: A sample vial was charged with (R,R)-L5·HBF₄ (12.0 mg, 0.022 mmol) and NaO^tBu (1.9 mg, 0.02 mmol) and after 10 min of stirring Ni(cod)₂ (5.5 mg, 0.02 mmol) was added. To the stirring solution was added 2-trimethylsilylbenzaldehyde (**1p**) and transferred to J. Young NMR tube. The reaction was monitored by NMR. After 24 h (observed no reaction), it was hearted to 80 °C. No reaction took place and starting material was intact as such, recovered after filtration.

Scope of 3-vinyl- and alkynyl-2,1-benzoxasiloles (Table 3)

Table 3. Catalytic Enantioselective Synthesis of 3-Vinyl- and Alkynylbenzoxasiloles

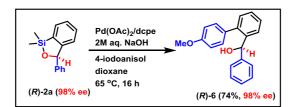
Reaction of 1q giving (R)-2q (80% yield, 76% ee): A screwed reaction tube was charged with (R,R)-L5·HBF₄ (54.3 mg, 0.1 mmol) and NaO'Bu (9.6 mg, 0.1 mmol) in THF (1.5 mL). After 10 min of stirring, Ni(cod)₂ (13.7 mg, 0.05 mmol) was added. The resulting reaction mixture was stirred at room temperature for 1 h followed by addition of a solution of 1q (190.0 mg, 1.0 mmol in 1 mL THF). The reaction mixture was stirred at 60 °C for 18 h. The reaction mixture was filtered through recycle small silica column. It was further purification by **GPC** (R)-1,1-dimethyl-3-vinyl-1,3-dihydrobenzo[c][1,2]oxasilole (2q) as volatile colorless oil in 80% yield (136.0 mg, 0.36 mmol). The spectroscopic data was identified to that reported data for racemic compound. S2 H NMR (400 MHz, CDCl₃): δ 0.40 (s, 3H, SiCH₃), 0.41 (s, 3H, SiCH₃), 5.18 (d, J =10.1 Hz, 1H, CH= CH_2), 5.40 (d, J = 16.8 Hz, 1H, CH= CH_2), 5.58 (d, J = 6.8 Hz, 1H, CHOSi), 5.91 $(ddd, J = 6.8, 10.1, 16.8 \text{ Hz}, 1H, CH=CH_2), 7.21 (d, J = 7.2 \text{ Hz}, 1H, Ar-H), 7.31 (t, J = 7.2 \text{ Hz}, 1H, Ar-H)$ Ar-H), 7.40 (dt, J = 1.3, 7.6 Hz, 1H, Ar-H), 7.56 (d, J = 7.2 Hz, 1H, Ar-H). $[\alpha]_D^{23} = (-) 15.1$ (c = 1.3) 0.09, in CHCl₃). The absolute stereochemistry of 2q was assigned by analogy to 2h and enantioselectivity was evaluated after converting it into oxidation product 3q. To the stirred solution of benzoxasilole **2q** (30 mg, 0.15 mmol) in 1 mL MeOH and THF (1:1) KHCO3 (30 mg) and H₂O₂ (35% aqueous solution, 0.12 mL) was added. The reaction flask was closed with stopper and stirred at room temperature for overnight gave vinyl alcohol **3q** (isolated with 15% EtOAc:Hexane; 10 mg, 66% yield, 76% ee). 1 H NMR (400 MHz, CDCl₃): δ 2.80 (br s, 1H, CHOH), 5.24–5.37 (m, 3H, CHOH and CH=CH₂), 6.14 (ddd, J = 16.9, 10.3, 6.3 Hz, 1H, CH=CH₂), 6.83–6.89 (m, 2H, Ar-H), 7.01 (dd, J = 7.5, 1.6 Hz, 1H, Ar-H), 7.19 (td, J = 8.1, 1.7 Hz, 1H, Ar-H), 7.74 (br s, 1H, Ar-OH). 13 C[1 H} NMR (100 MHz, CDCl₃): δ 76.3,116.4, 117.2, 120.0, 125.5, 127.6, 129.4, 138.0,155.5. HRMS (ESI) m/z calcd for C₉H₁₀O₂ 150.0681, found 150.0680. Chiral separation: The enantioselectivity for **3q** was determined by HPLC using Chiralpak OB-H; Isopropanol/Hexane = 3:97; Flow = 0.5 mL/min; T = 40 °C; λ = 250 nm. Retention time: t_R = 43.1 min (minor *S*-enantiomer) and 47.4 min (major *R*-enantiomer).

Reaction of 1r giving 2r (78% yield, 73% ee): The general procedure was followed with 2-dimethylvinylsilyl-5-methoxybenzaldehyde (1r) (220.0 mg, 1.0 mmol) and reaction was conducted at 60 °C for 24 h. The reaction mixture was filtered through small silica column. It was further purification by recycle GPC to get (*R*)-5-methoxy-1,1-dimethyl-3-vinyl-1,3-dihydrobenzo[c] -[1,2]oxasilole (2r) as colorless and volatile oil in 78% yield (170.0 mg, 0.77 mmol). The spectroscopic data was identified to that reported data for racemic compound. S2 1H NMR (400 MHz, CDCl₃): δ 0.44 (s, 3H, SiCH₃), 0.45 (s, 3H, SiCH₃), 3.82 (s, 3H, OCH₃), 5.22 (td, J = 1.2, 10.0Hz, 1H, CH=C H_2), 5.46 (td, J = 1.2, 17.0 Hz, 1H, CH=C H_2), 5.58 (d, J = 6.8 Hz, 1H, C H_2), 5.97 $(ddd, J = 6.8, 10.0, 17.0 \text{ Hz}, 1H, CH = CH_2), 6.77 (d, J = 2.4 \text{ Hz}, 1H, Ar-H), 6.93 (dd, J = 2.0, 8.0 \text{ Hz}, 1H, Ar-H)$ 1H, Ar-H), 7.51 (d, J = 8.0 Hz, 1H, Ar-H). $[\alpha]_D^{23} = (-) 9.7$ (c = 0.09, in CHCl₃). The absolute stereochemistry of 2r was assigned by analogy to 2h and enantioselectivity was evaluated after converting it into oxidation product **3r** (69 % yield, 73% ee). ¹H NMR (400 MHz, CDCl₃): δ 2.62 (br s, 1H, CHOH), 3.74 (s, 3H, OCH₃), 5.25-5.37 (m, 3H, CHOH and CH=CH₂), 6.09-6.18 (m, 1H, $CH=CH_2$), 6.59 (s, J=2.7 Hz, 1H, Ar-H), 6.75 (dd, J=8.7, 2.5 Hz, 1H, Ar-H), 6.81 (d, J=8.7 Hz, 1H, Ar-H), 7.20 (d, J = 2.5 Hz, 1H, Ar-OH). ${}^{13}C\{{}^{1}H\}$ NMR (100 MHz, CDCl₃): δ 55.9, 75.9, 113.2, 114.3, 116.5, 117.7, 126.5, 137.9, 149.2, 153.0. **HRMS** (ESI) m/z calcd for $C_{10}H_{13}O_3$ 181.0864: (M^+) found 181.0861. <u>Chiral separation</u>: The enantioselectivity for **3r** was determined by SFC using Chiralpak IB (Back pressure = 15 MPa, Flow (CO₂) = 4.0 mL/min; Flow (isopropanol) = 0.2 mL/min; T = 25 °C; λ = 230 nm). Retention time: t_R = 6.0 min (major *R*-enantiomer) and 6.9 min (minor *S*-enantiomer).

Reaction of 1s giving 2s (76% yield, 93% ee): A screwed reaction tube was charged with (R,R)-**L5·HBF**₄ (27.3 mg, 0.05 mmol) and NaO^tBu (4.8 mg, 0.05 mmol) in THF (1.5 mL). After 10 min of stirring, Ni(cod)₂ (6.8 mg, 0.025 mmol) was added. The resulting reaction mixture was stirred at room temperature for h followed addition of a solution of 2-diethyl(trimethylsilylethynyl)silylbenzaldehyde (1s) (144.0 mg, 0.5 mmol) in THF (1.0 mL) was added the solution. The reaction mixture was stirred at 60 °C for 8 h. The reaction mixture was filtered through small silica column. It was further purified by Kugelrohr distillation (0.4 mmHg, 100 °C) to get (R)-1,1-diethyl-3-((trimethylsilyl)ethynyl)-1,3-dihydrobenzo[c][1,2]oxasilole (2s) as colorless oil in 76% (109.4 mg, 0.38 mmol). The spectroscopic data was identified to that reported data for racemic compound. Start NMR (400 MHz, CDCl₃): δ 0.18 (s, 9H, Si(CH₃)₃), 0.77–1.02 (m, 10H, Si(CH₂CH₃)₂), 5.91 (s, 1H, CHO), 7.31–7.36 (m, 1H, Ar-H), 7.43–7.46 (m, 2H, Ar-H), 7.54 (dd, J = 0.8, 7.3 Hz 1H, Ar-H). $[\alpha]_D^{23} = (+) 8.9$ (c = 0.2, in CHCl₃). The absolute stereochemistry of 2s was assigned by analogy to 2h and enantioselectivity was evaluated after converting it into oxidation product 3s according to the previously reported procedure (68 % yield). The spectroscopic data was identified to that reported data for racemic compound. S2 1H NMR (400 MHz, CDCl₃): δ 0.22 (s, 9H, TMS), 2.59 (br s, 1H, OH), 5.66 (s, 1H, ArCH), 6.87–6.91 (m, 2H, Ar-H), 7.10 (br s, 1H, OH), 7.22-7.26 (m, 1H, Ar-H), 7.36 (dd, J = 1.4, 7.8 Hz, 1H, Ar-H). Chiral separation: The enantioselectivity for 3s was determined by SFC using Chiralpak IA (Back pressure = 15 MPa, Flow (CO₂) = 4.0 mL/min; Flow (isopropanol) = 0.3 mL/min; T = 25 °C; λ = 220 nm). Retention time: $t_R = 3.1 \text{ min (minor } S\text{-enantiomer)}$ and 3.2 min (major R-enantiomer).

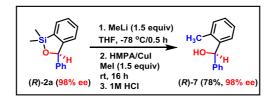
[5] Synthetic utilities of 3-arylbenzoxasiloles

Reaction of (*R*)-2a giving (*R*)-5 (61% yield, 98% ee): To the stirring solution of (*R*)-2a (86 mg, 0.4 mmol, 98% ee) in THF (2 mL) was added Ag(I)F (208 mg, 1.6 mmol) followed by *N*-iodosuccinamide (360 mg, 1.6 mmol). The reaction flask was covered with aluminum foil and allowed to stir at room temperature for overnight under N₂ atmosphere. After completion of reaction, it was filtered through celite and further purified by silica gel column chromatography to yield yellow oil product 6 (189.0 mg, 61% yield, isolated with 10% EtOAc:Hexane). Spectral data was identified to that previously reported. ^{S10} H NMR (400 MHz, CDCl₃): δ 2.36–2.50 (m, 1H, CHO*H*), 6.06 (d, J = 3.2 Hz, 1H, CHOH), 6.99 (td, J = 1.4, 7.6 Hz, 1H, Ar-*H*), 7.26-7.44 (m, 6H, Ar-*H*), 7.52 (dd, J = 1.0, 7.7 Hz, 1H, Ar-*H*), 7.83 (dd, J = 1.0, 7.7 Hz, 1H, Ar-*H*). [α]_D²³ = (+) 53.9° (c = 0.18, in CHCl₃) (Lit. ^{S11} [α]_D = (+) 68.2° (c = 1.5, in acetone for >95% ee, *R*-analogue). Chiral separation: The enantioselectivity was determined by SFC using Chiralpak IB (Back pressure = 15 MPa, Flow (CO₂) = 2.0 mL/min, Flow (IPA) = 0.3 mL/min, 25 °C, λ = 250 nm). Retention time: t_R = 5.8 min (major *R*-enantiomer) and 6.4 min (minor *S*-enantiomer). The absolute stereochemistry of 5 was assigned by analogy to previous report. ^{S10}



Reaction of (*R*)-2a giving (*R*)-6 (74% yield, 98% ee): To a stirring solution of 2a (96 mg, 0.4 mmol, 98% ee) and 4-iodoanisol (140 mg, 0.48 mmol) in dioxane (2 mL) was added a freshly prepared solution of Pd(OAc)₂ (9 mg, 8 mol%) and 1,2-bis(dicyclohexylphosphino)ethane (18 mg, 16 mol%) in dioxane (2 mL). After stirring the reaction mixture for 30 min at room temperature, aqueous NaOH (2M, 2.0 mmol) was added. The biphasic solution was stirred further for 30 min followed by heated at 65 °C. After completion of reaction, it was cooled to rt and then diluted with ethyl acetate, filtered through silica and concentrated. It was dried over Na₂SO₄ and purified by silica gel column chromatography (with 15% EtOAc:hexane, 74% yield). ¹H NMR (400 MHz, CDCl₃): δ 2.18 (br s,

1H, CHO*H*), 3.77 (s, 3H, OC*H*₃), 5.88 (s, 1H, C*H*OH), 6.84 (d, J = 8.8 Hz, 2H, Ar-*H*), 7.10–7.15 (m, 4H, Ar-*H*), 7.16–7.23 (m, 4H, Ar-*H*), 7.25 (td, J = 1.6, 7.5 Hz, 1H, Ar-*H*), 7.30 (td, J = 1.6, 7.5 Hz, 1H, Ar-*H*), 7.47 (dd, J = 1.6, 7.5 Hz, 1H, Ar-*H*). ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 55.4, 72.5, 113.5, 126.7, 127.2, 127.3, 127.4, 127.7, 128.3, 130.3, 130.5, 133.2, 141.1, 141.3, 144.0, 158.9. HRMS (EI): m/z Calcd for C₂₀H₁₈O₂: (M⁺) 290.1307, found 290.1308. [α]_D²³ = (+) 160.8° (c = 0.2, in CHCl₃). Chiral separation: The enantioselectivity was determined by SFC using Chiralpak IB (Back pressure = 15 MPa, Flow (CO₂) = 2.0 mL/min, Flow (IPA) = 0.3 mL/min, 25 °C, λ = 250 nm). Retention time: t_R = 5.6 min (major R-enantiomer) and 6.1 min (minor S-enantiomer).



Reaction of (R)-2a giving (R)-7 (78% yield, 98% ee): To a solution of (R)-2a (240 mg, 1.0 mmol, 98% ee) in dry THF (3 mL) was added MeLi (1.5 mmol, 1.05 M in Et₂O) at -78 °C. After stirring for 30 min, the solution was transferred through cannula slowly to a pre-mixed homogeneous solution of HMPA (2.5 mL) and copper iodide (228 mg, 1.2 mmol) at room temperature, which turns the color of solution black. After 30 min, MeI (2.0 mmol) was added drop wise and the resulting solution was stirred at room temperature for 16 h, then diluted with reaction mixture was diluted with ether (10 mL) and quenched with aqueous 1M HCl (10 mL), and allowed to stir for 30 min. The reaction mixture was extracted with ethyl acetate (3 x 15 mL). The combined organic layers were washed with brine, dried with Na₂SO₄, and concentrated in vacuo. The crude product was purified by silica gel chromatography (with 10% EtOAc:hexane) to yield 7 as a white solid (155.0 mg, 78%). The spectral data was identified to that previously reported. S7 ¹H NMR (400 MHz, CDCl₃): δ 2.12 (br s, 1H, CHOH), 2.18 (s, 3H, Ar-CH₃), 5.93 (d, J = 2.4 Hz, 1H, CHOH), 7.08 (d, J = 7.5 Hz, 1H, Ar-H), 7.11-7.30 (m, 7H, Ar-H), 7.45 (d, J = 7.5 Hz, 1H, Ar-H). $[\alpha]_D^{23} = (-) 4.6^{\circ}$ (c = 0.1, in CHCl₃) (Lit. S⁷ [α]_D²³ = (+) 3.0° (c = 0.65, in CHCl₃ for 86% ee, S-analogue). Chiral separation: The enantioselectivity was determined by SFC using Chiralpak IB (Back pressure = 15 MPa, Flow $(CO_2) = 2.0$ mL/min, Flow (IPA) = 0.3 mL/min, 25 °C, $\lambda = 250$ nm). Retention time: $t_R = 4.2$ min (minor S-enantiomer) and 4.4 min (major R-enantiomer). The absolute stereochemistry of 7 was assigned by analogy to previous report.^{S7}

Synthesis of (*R*)-orphenadrine from (*R*)-7: To a suspension of (*R*)-2a (105 mg, 0.53 mmol, 98% ee) and hydrochloride salt of 2-(*N*,*N*-dimethylamino)ethylchloride (152.7 mg, 1.06 mmol) in DMSO (1.5 mL), a grinded KOH (excess, 297 mg, 5.3 mmol) was added. The reaction mixture was stirred at room temperature. After 16 h, it was diluted with ether and aqueous NaOH (1M, 10 mL) was added and stirred vigorously for 30 min. The biphasic mixture was extracted well and aqueous phase was extracted with ether (3 x 10 mL). The combined organic layer was dried over Na₂SO₄, concentrated and purified by short silica gel column chromatography (with 20% EtOAc:hexane:1% TEA; 76% yield). The spectral data was identified to previously reported. S12,13 H NMR (400 MHz, CDCl₃): δ 2.18 (br s, 9H, Ar-CH₃, N(CH₃)₂), 2.51 (t, J = 6.1 Hz, 2H, NCH₂), 3.45–3.53 (m, 2H, OCH₂), 5.46 (s, 1H, CHOCH₂), 7.04–7.20 (m, 5H, Ar-H), 7.21–7.27 (m, 3H, Ar-H), 7.36 (d, J = 7.5 Hz, 1H, Ar-H). $\frac{13}{1000}$ Cf Hg NMR (100 MHz, CDCl₃): δ 19.4, 46.0, 59.0, 67.6, 81.3, 125.9, 127.1, 127.3, 127.4, 127.6, 128.2, 130.5, 135.9, 139.8, 141.1. HRMS (CI): m/z Calcd for C₁₈H₂₄NO: (M⁺+H) 270.1852, found 270.1856. [α]_D²³ = (+) 3.4° (c = 0.1, in CHCl₃) {(Lit. S13 [α]_D²⁷ = (+) 11.0° (c = 3.61, in THF for 81% ee, *R*-analogue)}.

One pot synthesis of (S)-neobenodine from 11 (75% yield): In a globe box, a screw cap vial was charged with NHC-salt L5·HBF₄ (2.4 mg, 2.2 mol%) and NaO^tBu (0.4 mg, 2 mol%) in toluene (1 mL). After 10 min of stirring, Ni(cod)₂ (1.1 mg, 2 mol%) was added. After 10 min, 11 (51 mg, 0.2 mmol) was added and allowed it stir till completion of reaction, monitored by GC. Dry DMSO (1 mL) was added to the reaction mixture. To the reaction solution, freshly grinded KOH (112 mg, 2.0 mmol) and hydrochloride salt of 2-(N,N-dimethylamino)ethylchloride (57 mg, 0.4 mmol) were added. The reaction mixture was stirred at room temperature for 12 h. GC analysis shows a sole

product of required product. It was diluted with ether and aqueous NaOH (1M, 5 mL) was added and stirred vigorously for 30 min. The biphasic mixture was extracted well and aqueous phase was extracted with ether (3 x 10 mL). The combined organic layer was dried over Na₂SO₄, concentrated and purified by silica gel column chromatography (with 20% EtOAc:hexane:1% TEA; 45.0 mg, overall yield 75%). ¹H NMR (400 MHz, CDCl₃): δ 2.26 (s, 6H, N(CH₃)₂), 2.30 (s, 3H, Ar-CH₃), 2.59 (t, J = 6.1 Hz, 2H, NCH₂), 3.55 (t, J = 6.1 Hz, 2H, OCH₂), 5.32 (s, 1H, CHOCH₂), 7.10 (d, J = 8.0 Hz, 2H, Ar-H), 7.20–7.36 (m, 7H, Ar-H). ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 21.1, 46.0, 59.0, 67.5, 83.8, 126.9, 127.0, 127.2, 128.3, 129.0, 137.0, 139.3, 142.5. HRMS (FAB): m/z Calcd for C₁₈H₂₄NO: (M⁺+H) 270.1852, found 270.1853. [α]_D²³ = (–) 9.4° (c = 0.1, in CHCl₃).

Synthesis of (S)-3-Phenylisobenzofuran-1(3H) one (8) from 1u (65% yield, 87% ee): In a globe box, a screw cap vial was charged with NHC-salt L5·HBF₄ (44 mg, 4.0 mol%) and NaO'Bu (7.7 mg, 4.0 mol%) in THF (20 mL). After 10 min of stirring, Ni(cod)₂ (11 mg, 2.0 mol%) was added and stirred at room temperature for additional 60 min. To a stirring solution, 1u (536 mg, 2 mmol) was added and stirred for 30 min followed by heating at 60 °C till completion of reaction (ca. 16 h), monitored by GC. The reaction mixture was filtered though short silica column. 2u: ¹H NMR (400 MHz, CDCl₃): δ 0.47 (s, 3H, SiCH₃), 0.55 (s, 3H, SiCH₃), 6.91–6.99 (m, 1H, Ar-H), 6.15 (s, 1H, CHOSi), 7.27-7.36 (m, 3H, Ar-H), 7.46 (t, J = 7.4 Hz, 1H, Ar-H), 7.53 (t, J = 7.4 Hz, 1H, Ar-H), 7.59–7.67 (m, 1H, Ar-H), 7.90 (d, J = 7.4 Hz, 1H, Ar-H), 10.39 (s, 1H, Ar-CHO). ${}^{13}C\{{}^{1}H\}$ NMR (100 MHz, CDCl₃): δ 0.5, 0.9, 78.4, 123.7, 127.3, 128.0, 128.3, 129.9, 130.9, 131.8, 133.6, 134.1, 135.2, 151.8, 192.8. **HRMS** (EI): m/z Calcd for $C_{16}H_{16}O_2Si$: (M⁺) 268.0920, found 268.0919. $[\alpha]_D^{23}$ = (-) 148.8° (c = 0.1, in CHCl₃). It was forwarded for without further purification. To a solution of 2u (110 mg, 0.4 mmol) in THF was added TBAF (1M, 1.0 mL, 1.0 mmol) at 0 °C allowed to stir at room temperature for 5 h. After the completion of reaction, it was quenched with aqueous NH₄Cl solution and stirred vigorously for further 30 min. The reaction mixture was extracted and aqueous phase was washed with ethyl acetate (3 x 15 min). Combined organic layer was dried over Na₂SO₄, concentrated and used for next oxidation step. The compound was dissolved in dry CH₂Cl₂ and then

PCC (172 mg, 0.8 mmol) was added. After 1 h, CH_2Cl_2 was evaporated completely and ether was added to the black residue. It was filtered through celite, washed the residue (3 x 10 mL) and filtered, concentrated to get desired product with quantitative conversion. It was further purified by silica gel column chromatography (with 15% ethyl acetate:hexane, white solid, 65% over all yield). The spectroscopic data was identified to previously reported. S13 1H NMR (400 MHz, CDCl₃): δ 6.3 (s, 1H, OCHPh), 7.14–7.20 (m, 2H, Ar-H), 7.20–7.31 (m, 4H, Ar-H), 7.44 (t, J = 7.5 Hz, 1H, Ar-H), 7.54 (t, J = 7.5 Hz, 1H, Ar-H), 7.85 (d, J = 7.5 Hz, 1H, Ar-H). [α] $_D^{23}$ = (+) 32.8° (c = 0.1, in CHCl₃) (Lit. S14 [α] $_D^{27}$ = (+) 40.3° (c = 1.0, in CH₂Cl₂ for 84% ee, S-analogue). Chiral separation: The enantioselectivity was determined by SFC using Chiralpak IA (Back pressure = 15 MPa, Flow (CO₂) = 2.4 mL/min, Flow (CH₂Cl₂) = 0.4 mL/min, 40 °C, λ = 250 nm). Retention time: t_R = 7.0 min (minor R-enantiomer) and 7.3 min (major S-enantiomer). The absolute stereochemistry of 8 was assigned by analogy to literature report. S14

[6] Isolation of $(\eta^2: \eta^2$ -dialdehyde)Ni complex (2u')

To the stirring solution of Ni(cod)₂ (22.0 mg, 0.08 mmol) and IPr (31.0 mg, 0.08 mmol) in C₆D₆ (1 mL) was added 2,2'-(dimethylsilanediyl)dibenzaldehyde (21.5, 0.08 mmol). After 5 min the volatile materials were evaporated under high vacuum for a long time. A yellow crystalline solid **2u'** was obtained quantitatively and the complex was analyzed by 1 H, 13 C and X-ray analyses. 1 H NMR (400 MHz, C₆D₆): δ 0.37 (s, 3H, SiCH₃), 0.96 (d, J = 6.8 Hz, 6H, IPr-CH₃), 1.08 (d, J = 6.8 Hz, 6H, IPr-CH₃), 1.21 (d, J = 6.8 Hz, 6H, IPr-CH₃), 1.68 (d, J = 6.8 Hz, 6H, IPr-CH₃), 3.08 (sept, J = 6.8 Hz, 2H, IPr-CHCH₃), 3.14 (sept, J = 6.8 Hz, 2H, IPr-CHCH₃), 4.97 (s, 2H, CHO), 6.04 (d, J = 7.6 Hz, 2H, Ar-H), 6.66 (s, 2H, IPr-NCHCHN), 6.96 (t, J = 7.2 Hz, 2H, Ar-H), 7.14 (t, J = 7.6 Hz, 2H, Ar-H), 7.31 (d, J = 7.6 Hz, 2H, Ar-H), 7.36 (d, J = 7.6 Hz, 2H, Ar-H), 7.40 (t, J = 7.6 Hz, 2H, Ar-H). 13 C{ 1 H} NMR (100 MHz, CDCl₃): δ 0.4, 23.4, 23.6, 24.8, 26.0, 28.7, 29.4, 117.1, 121.6, 123.9, 124.1, 124.2, 126.5, 129.4, 129.6, 133.5, 133.8, 137.2, 146.5, 147.3, 149.3, 193.7.

[7] Crossover experiments

(a) Crossover experiments with 1c and 1d (Scheme S3): To a stirring solution of NHC salt L5·HBF₄ (4.8 mg, 0.0088 mmol), NaO^tBu (0.8 mg, 0.008 mmol), and Ni(cod)₂ (2.2 mg, 0.008 mmol) in toluene (1 mL) was added a pre-dissolved solution of 1c (54.0 mg, 0.2 mmol) and 1d (54.0, 0.2 mmol) in toluene (1 mL) at room temperature. The reaction mixture was stirred at room temperature. The reaction was monitored by GC using *n*-pentadecane as an internal standard for 4 h. The corresponding benzoxasiloles 2c and 2d were obtained in 80% yield each, with no formation of crossover products (entry 1). The influence of salts and base on mechanism switching was studied. In all of these cases, crossover products were obtained (entries 2–5).

Crossover Products 1d entry Conversions L = NHC/Sal 100% $L = (R.R)-L5-HBF_A$ L = IPr 46% 100% 100% 52% 60% 42% 76% L = IPr·HCI 15% 50% 25% 27% 58% 87% L = IPr/NaBF. 27% 54% 30% 32% L = SIPr·HBF 38% 44% 100% 100% 64% 56%

Scheme S3. Crossover experiments with 1c and 1d^a

Entry 1: Under the optimized enantioselective reaction conditions using L5·HBF₄.

Entry 2: ref. S2.

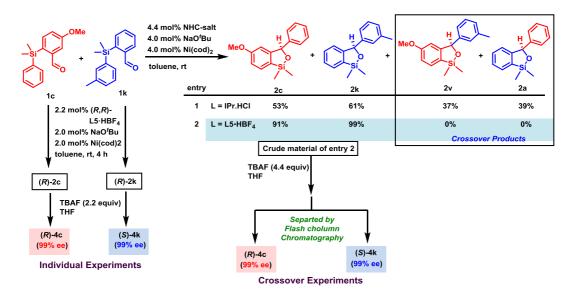
Entry 3: IPr HCl salt was used for *in situ* generation of IPr (free carbene) in the presence of KO^tBu and yields were obtained after 42% and 76% conversion of 1c and 1d, respectively.

Entry 4: NaBF₄ was used as an additive in Ni(cod)₂/IPr catalytic system. Yields were mentioned at 58% and 87% conversion of **1c** and **1d**, respectively.

Entry 5: Commercially available SIPr·HBF₄ was employed as a precursor of SIPr in the presence of NaO'Bu. Yields were mentioned after quantitative conversion of 1c and 1d.

(b) Crossover experiments with 1c and 1k (Scheme S4): The crossover experiments was examined with **1c** (54.0 mg, 0.2 mmol) and **1k** (51.0 mg, 0.2 mmol) under the identical reaction conditions

^aYields were determined by GC using *n*-pentadecane as an internal standard.

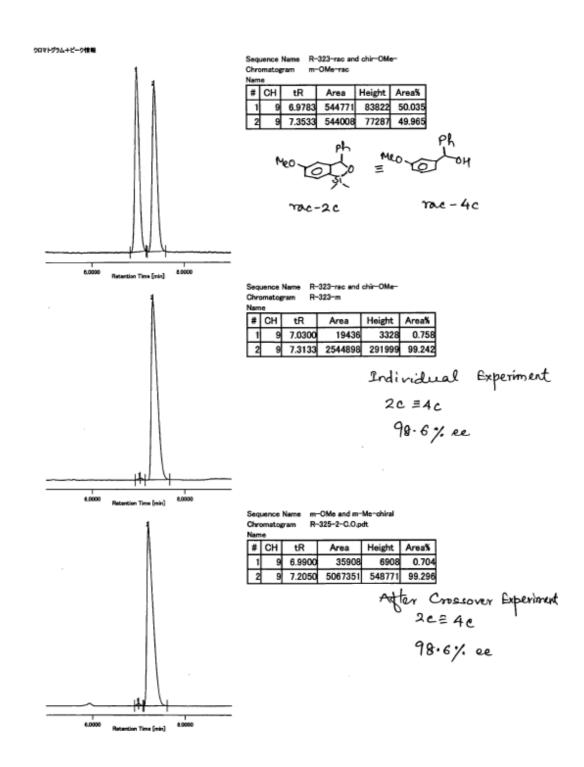


Scheme S4: Crossover experiments with 1c and 1k^a

^aYields were determined by GC using *n*-pentadecane as an internal standard.

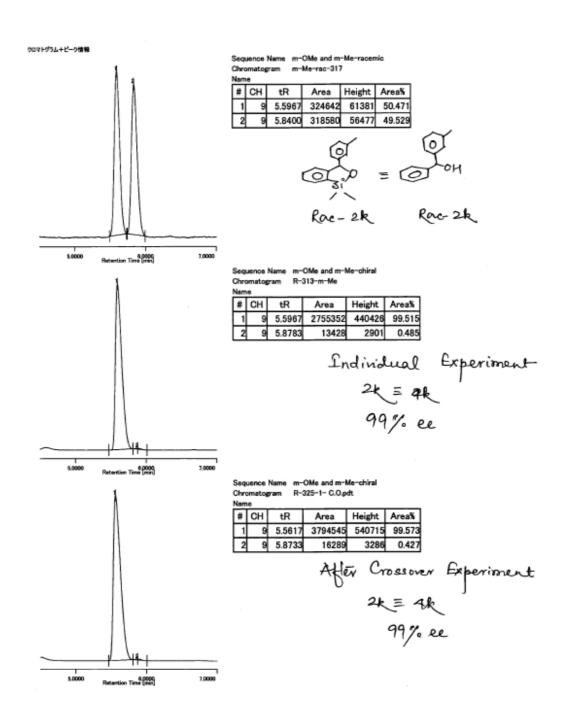
using NHC salt **L5·HBF**₄. The corresponding benzoxasiloles **2c** and **2k** were obtained in 91 and 99% yields, respectively, with no crossover products (entry 2). After completion of reaction, it was passed through the small pad of silica and volatile material was evaporated. The crude material was subjected to desilylation using TBAF (0.9 mL, 0.88 mmol) in THF (2 mL). The desilylated products **4c** (31 mg, 72% yield) and **4k** (29 mg, 73% yield) were obtained after silica column chromatography (isolated with 20 and 10% EtOAc:hexane, respectively). The enantioselectivities measured for desilylation products (**4c** and **4k**) correspond to benzoxasiloles (**2c** and **2k**, respectively), which were remain consistent with the results of those of individual experiments (see below the chiral SFC charts). Entry 1 showed the results of crossover experiments, when IPr·HCl was employed.

Comparative Chiral SFC chart of 4c correspond to 2c



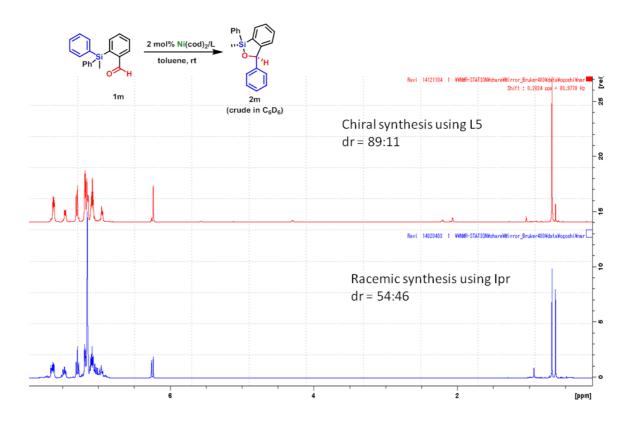
Top SFC (for racemic sample), center SFC (for individual experiment's product) and bottom SFC (for crossover experiment's product).

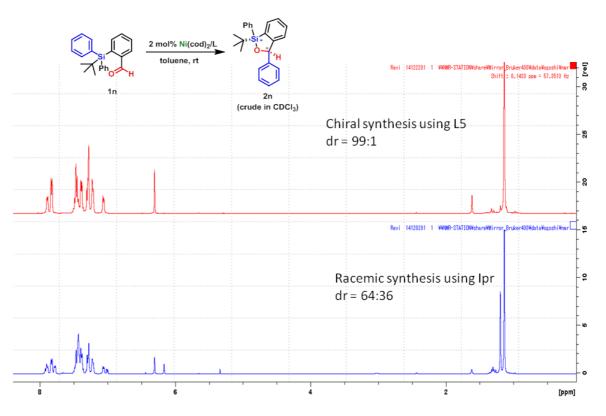
Comparative Chiral SFC chart of 4k correspond to 2k

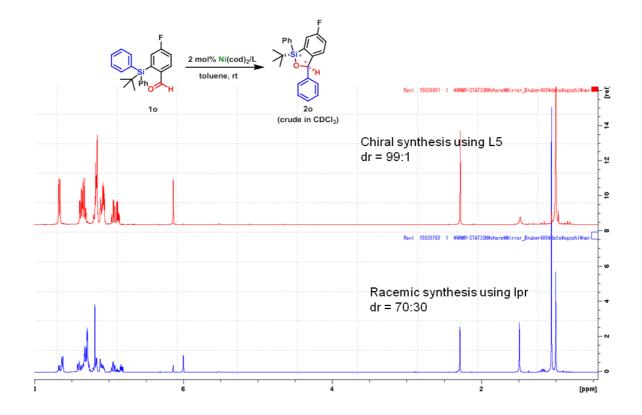


Top SFC (for racemic sample), center SFC (for individual experiment's product) and bottom SFC (for crossover experiment's product).

[8] Comparative ¹H NMR spectra for diastereoselectivity (IPr verses L5) in case of 2m-o







[9] Computational Details:

All calculations were performed with the Gaussian 09 package^{S15} of programs with the hybrid B3LYP functional. S16 For geometry optimizations, nickel was represented by the effective core potentials (ECPs) of Hay and Wadt with a double- ζ basis set (LANL2DZ)^{S17} augmented with an f-type polarization function. S18 The 6-311G(d,p) basis set was used for C, H, and N, whereas the 6-311G+(d) basis set was used for Si and O. The geometry optimizations were performed without any symmetry constraint followed by analytical frequency calculations to confirm that a minimum had been reached. The Gibbs energies were calculated at 298 K and 1 atm from the harmonic approximation for frequencies. Potential energies were also obtained by performing single point calculations using the above optimized geometries with the following basis sets: the LANL2DZ effective core potential for the inner electrons and the LANL2TZ^{S19} basis set, augmented by an f polarization function, for the outer ones for Ni, the 6-311+G(2d,p) basis set for all the other atoms. Solvent effects (toluene, $\varepsilon = 2.3741$) were introduced through the single-point calculation at the optimized gas-phase geometries for all the minima by means of the polatizable continuum model (IEFPCM) implemented in Gaussian 09. S20 Relative Gibbs energies were obtained by adding the Gibbs energy corrections derived from the analytical frequency calculations. These calculations involve a certain margin error.

Coordinates, E and G at 298 K in a.u. for all calculated extrema:

 $Ni(NHC^*)$ (NHC* = (R,R)-L5)

E = -1556.579862, G = -1556.649748

```
Ν
   -1.04907600
                 -0.52881600
                                0.01674300
C
   -2.14893700
                  -1.31576900
                               -0.46580500
C
   -3.15875400
                  -3.38867800
                                -1.14113800
C
    -4.37753600
                  -2.74970800
                               -1.34023300
Н
    -3.08538200
                  -4.45257700
                               -1.33871900
Н
    -5.23633100
                  -3.31764400
                               -1.68060100
C
    2.00254900
                  1.26731300
                               -0.47193600
C
    3.00742300
                  3.36168500
                               -1.08527700
C
    4.21399300
                  2.72417800
                               -1.35189000
Н
    2.93653200
                  4.43391700
                               -1.23212900
Н
    5.06625200
                  3.30111600
                               -1.69354000
C
    0.58641000
                  0.39324600
                                1.45340100
Н
     0.31182400
                  1.39195200
                                1.81030300
C
    -0.68148800
                  -0.51623400
                                1.45242100
Н
    -0.39768800
                  -1.52888000
                                1.75918300
С
    1.70864000
                  -0.12122400
                                2.33351100
С
    2.33969100
                 -1.34326300
                                2.07187500
С
     2.11314500
                  0.61669500
                                3.44768400
С
    3.34823000
                  -1.81410800
                                2.90640900
Н
    2.04645200
                  -1.92481200
                                1.20468100
C
    3.12188500
                  0.14644300
                                4.28832900
Η
    1.63441000
                  1.56668700
                                3.66080700
C
    3.74222600
                  -1.07011200
                                4.01943400
Н
    3.82970600
                  -2.76093100
                                2.68850000
Н
    3.42285800
                  0.73297000
                                5.14906000
Н
     4.52891100
                  -1.43704100
                                4.66887300
С
   -1.77779100
                  -0.03969200
                                2.38496400
C
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                  -0.82251400
                                3.47997400
С
    -2.41731200
                  1.19048700
                                2.19094800
C
   -3.13297800
                  -0.38847800
                                4.36782000
Η
   -1.66301800
                  -1.77926800
                                3.64062700
C
   -3.40149500
                  1.62538500
                                3.07276500
Н
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                  1.80666100
                                1.33939500
C
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                                4.16605200
Η
   -3.40815700
                  -1.00965600
                                5.21272300
Н
   -3.89015900
                  2.57908200
                                2.90712500
Η
   -4.52991600
                  1.17536500
                                4.85229600
С
    1.87856400
                  2.65844900
                                -0.64326300
C
    -2.02115900
                  -2.69692300
                                -0.70347900
```

```
C
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                  -0.01311500
                                -0.80219600
C
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                                -0.41170500
Н
    -0.15865900
                   2.67588500
                                -0.03998900
С
     0.70775800
                   4.51039400
                                 0.64508500
Н
     1.09043400
                   4.11664000
                                 1.59056500
Н
     1.39059600
                   5.29757100
                                 0.31293300
Н
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                   4.97590000
                                 0.83852500
C
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                   3.95065100
                                -1.73458300
Н
     0.67800200
                                -2.17992300
                   4.68887400
Н
    -0.13873500
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                                -2.45052700
Н
    -0.96214700
                   4.43512900
                                -1.56628700
C
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                                -0.54729600
                  -3.43742100
Н
     0.03300800
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                                -0.16061300
C
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                  -4.60069200
                                 0.45596700
Η
    -1.15697400
                  -4.25845600
                                 1.43075200
Н
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                  -5.37685800
                                 0.10461300
Н
     0.18085600
                  -5.06602900
                                 0.59774300
C
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                  -3.91644200
                                -1.91206700
Η
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                  -4.63493400
                                -2.37560400
Н
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                  -3.06649400
                                -2.58903800
Н
     0.80286600
                  -4.40311000
                                -1.79584200
Ni
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                   0.02804100
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C
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                  -1.38102000
                                -1.10532900
Н
    -5.42975000
                                -1.26191400
                  -0.86842700
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                                -0.66782100
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                                -0.48531700
С
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                                -0.74193700
Н
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                  -0.44521300
                                -0.60966800
С
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                                -1.18267100
Н
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                                -1.39234400
                   0.83413700
```

o-dimethylphenylsilylbenzaldehyde (1a)

E = -945.9365012, G = -945.9844812

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Si
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                 -1.09307300
0
    -4.35933400
                  -1.33513100
                               -0.0000300
С
    -3.15165800
                  -1.13886100
                               -0.0000300
C
    -2.51658500
                  0.19622500
                               -0.0000300
С
                  -0.31302900
                                0.0000100
    1.92066300
С
    2.58537700
                  -0.02023600
                               -1.20082400
С
    -0.61852000
                  1.66745500
                               -0.0000200
С
    -1.11006000
                  0.35323200
                               -0.0000200
С
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                  0.82808100
                                0.00000200
С
    -2.85155300
                  2.59646900
                               -0.00000400
```

```
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                               -1.20459900
C
                  0.54397400
    3.86006300
                                1.20460300
C
    -3.36861200
                  1.31033400
                               -0.0000400
C
    -1.46854200
                  2.77243000
                               -0.0000300
C
    2.58537800
                 -0.02024100
                                1.20082700
C
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                 -2.16377600
                                1.56722900
Н
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                 -2.00427000
                               -0.0000300
Н
    -4.43774000
                  1.13464800
                               -0.00000400
Н
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                  3.45372200
                               -0.0000500
Н
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                               -0.00000400
Н
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                               -0.0000200
Н
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                                2.47006700
Η
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                                1.56021500
                                1.62989100
Н
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Η
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                 -0.23540800
                                2.15102000
Η
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                  0.75908000
                                2.14638000
Н
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Н
    4.35311000
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Η
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                               -2.47006500
Н
    0.82583000
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Н
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                 -2.73630300
                               -1.56021100
```

$(\eta^2$ -1a)Ni(NHC*) (Figure S2)

E = -2502.569344, G = -2502.660506

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Н
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                  -2.10017900
C
     1.00796800
                  -1.27278400
                                -2.69077900
C
     2.30943300
                  -0.79594900
                                -2.91658600
Н
     2.97915800
                  -0.68460300
                                -2.07138000
C
     2.74141800
                  -0.44326800
                                -4.18781400
Н
     3.74880500
                                -4.33461100
                  -0.06922000
C
     1.87158900
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                                -5.26772100
Н
     2.18850700
                  -0.30087800
                                -6.26813200
С
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                  -1.09113400
                                -5.05712100
Н
    -0.04768400
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                                -5.91917800
C
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                                -3.78531500
0
    -0.62274300
                  -1.79323500
                                -0.95566100
Si
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                  -2.35539900
                                -3.70367600
C
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                  -2.79992800
                                -5.51683100
Н
    -1.26394100
                  -3.41586600
                                -5.98358200
```

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                                -5.50500700
Н
    -2.20585900
                  -1.92432900
                                -6.14838600
C
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                  -4.03100900
                                -2.80021000
Н
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                                -1.73207200
Н
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                                -2.96234800
Н
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                                -3.21221500
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                                -3.86560500
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                                -1.58616100
Н
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                                -3.50843800
Η
    -2.98506300
                                -4.76613900
                   0.07550200
C
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                   0.12520300
                                -2.36978900
Η
    -5.47547400
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                                -0.69270000
Η
    -5.00382000
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                                -4.12341500
Н
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                                -2.09595000
Ni
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                                -0.17160400
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Ν
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                                 1.38939300
Ν
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                                 0.86981000
C
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                                 1.35929300
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                                 1.98776300
С
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                   1.02752300
                                 0.42443100
C
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                   2.86759000
                                 1.84488600
С
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                   1.87753700
                                 2.52808100
С
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                                 0.13767300
Н
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                                 3.04435400
С
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                                 1.31854700
C
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                   1.63053700
                                -0.41839800
C
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                                 0.87858300
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                                 1.40652500
C
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                   2.46876100
                                 3.16193500
C
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                   1.89026200
                                 2.42174100
C
    -1.63270000
                   1.54339900
                                 3.86396300
Η
    -1.59372500
                   2.53097200
                                -0.74669300
C
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                                 0.05770700
C
     0.34150800
                   4.81499700
                                -0.07348300
C
     0.06842400
                   5.79227000
                                 2.10943100
C
     5.21831300
                   0.88062700
                                -0.74943000
C
     3.91207800
                   3.03066700
                                -0.99716800
С
     4.43936900
                  -1.01046200
                                 0.51834900
Н
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                  -0.72382600
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С
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                                 3.74122400
С
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С
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С	-1.96721500	0.10357000	4.29919000
Н	-4.06315300	2.48828100	-0.89698200
С	0.11554600	6.05823200	-0.65725500
Н	0.52273400	3.95318300	-0.70554900
С	-0.15641500	7.03895100	1.52675600
Н	0.04444500	5.69174200	3.18966600
С	5.40515500	-0.41872900	-0.29320100
Н	5.96978600	1.32378000	-1.39329200
Н	2.94007200	3.40799500	-0.67700100
С	4.99278300	3.99821300	-0.47793400
С	3.88792400	3.01532800	-2.53701700
Н	4.56576500	-2.02533700	0.87685200
С	4.26343600	3.00921900	4.97615600
Н	3.98855200	4.22192800	3.22373400
С	2.95026500	0.99197500	5.07354900
Н	1.65665800	0.62310800	3.40393800
Н	-5.40828000	2.11222700	1.16671600
Н	-1.75148100	3.57813400	4.66673200
Н	-3.07457900	2.53102700	5.18339500
Н	-1.46805200	2.32785800	5.88822200
Н	-3.03804900	-0.02050900	4.48281100
Н	-1.67356100	-0.61658400	3.53153200
Н	-1.43863400	-0.14609700	5.22395200
С	-0.13256700	7.17472700	0.14142000
Н	0.12765200	6.15487100	-1.73699300
Н	-0.35583700	7.89942800	2.15535100
Н	6.29612300	-0.96857800	-0.57451800
Н	5.01827500	4.02840100	0.61435400
Н	5.98724700	3.69866600	-0.82089700
Н	4.80547000	5.01190100	-0.84330600
Н	4.84389900	2.68197900	-2.95075500
Н	3.10779000	2.35105100	-2.91384800
Н	3.69568000	4.02228300	-2.91893000
С	3.89273900	1.84616900	5.64567700
Н	5.00022000	3.67619200	5.40926900
Н	2.66222900	0.08066400	5.58533700
Н	-0.31188200	8.14144200	-0.31503600
Н	4.33811900	1.60236800	6.60338400

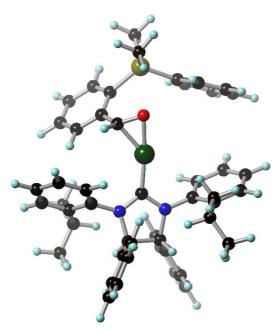


Figure S2 Optimized structures of $(\eta^2$ -1a)Ni(NHC*)

 $(\eta^2:\eta^2$ -1a)Ni(NHC*) (Figure S3)

where NHC* = (R,R)L5

E = -2502.568039, G = -2502.657091

```
С
    1.29503500
                 -0.05654300
                               -2.76167200
Н
    2.28978800
                  0.40153900
                               -2.73056400
С
    0.32739900
                  0.55387200
                               -3.69444600
С
    0.62141000
                  1.80508200
                               -4.25614200
Н
    1.55498400
                  2.29361000
                               -3.99141400
С
    -0.24757400
                  2.41707700
                               -5.15257400
Н
    0.00073600
                  3.38260700
                               -5.58022500
С
    -1.43033400
                  1.77282400
                               -5.50567400
   -2.11589500
Н
                  2.22982000
                               -6.21101100
С
   -1.72706200
                  0.52623900
                               -4.95011600
Н
   -2.65407000
                  0.04597400
                               -5.24108000
С
    -0.87547100
                  -0.11252800
                               -4.03942100
0
    1.18907000
                 -1.34762200
                               -2.51217900
Si
   -1.39850800
                 -1.83279300
                               -3.33719000
С
    -3.21800600
                 -2.09800700
                               -3.90902300
Н
   -3.26889900
                 -2.21852100
                               -4.99490800
Н
   -3.60229500
                 -3.01731900
                               -3.45826700
Н
    -3.88879500
                 -1.28256400
                               -3.62854900
С
    -0.46568000
                 -3.32217500
                               -4.06942800
Н
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                 -3.31766800
                               -3.77449500
```

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                                -3.73698300
Н
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                                -5.16117000
C
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                                -1.40839900
C
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                  -2.55184500
                                -0.60794000
                  -1.44999300
C
    -2.66202900
                                -0.75791500
С
    -0.71549700
                  -2.71176700
                                 0.77150600
Н
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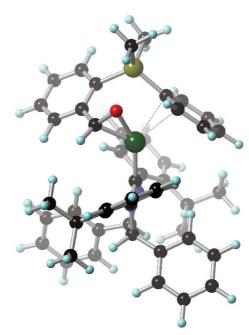


Figure S3 Optimized structures of $(\eta^2:\eta^2-1a)$ Ni(NHC*)

Ni(IPr)

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$(\eta^2$ -1a)Ni(IPr) (Figure S4)

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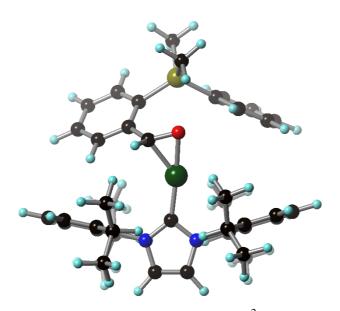


Figure S4 Optimized structures of $(\eta^2$ -1a)Ni(IPr)

Table S1. Selected Geometries for $(\eta^2-1\mathbf{a})\text{Ni}(\text{NHC}^*)$, $(\eta^2:\eta^2-1\mathbf{a})\text{Ni}(\text{NHC}^*)$, and $(\eta^2-1\mathbf{a})\text{Ni}(\text{IPr})$

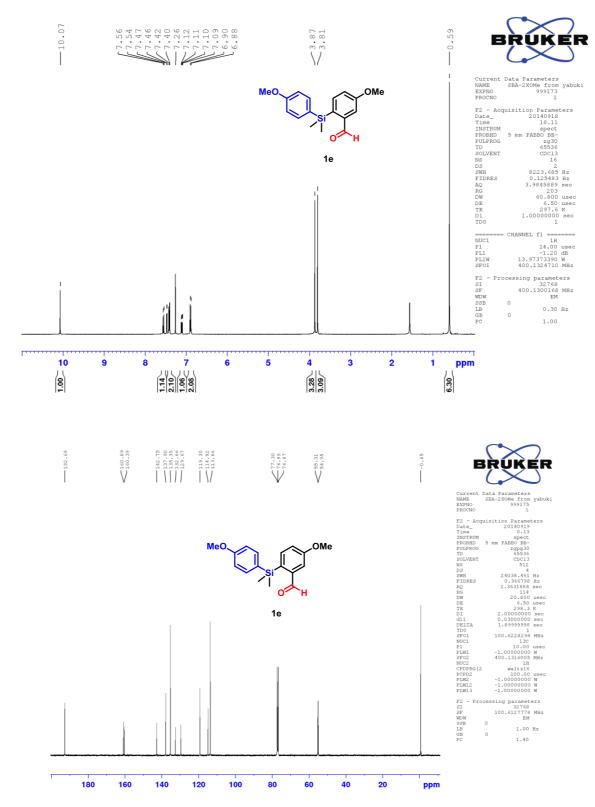
	$(n^2$ -1a)Ni(NHC*)	$(n^2: n^2-1a)$ Ni(NHC*)	$(n^2$ -1a)Ni(IPr)
bond distances / \mathring{A}			
Ni-O	1.8875	1.9217	1.8862
Ni–C	1.9485	1.9721	1.9542
C-O	1.3193	1.3192	1.3186
$Ni-C_{NHC}$	1.8962	1.9083	1.9011
Ni-C _{inso}	(4.3276)	2.6285	(4.1982)
Ni- Cortho	(4.3593)	2.4543	(3.9997)
Si···O	2.9635	2.7589	2.9513
Ni···Si	4.4197	3.4270	4.3609
bond angles / $^{\circ}$			
C _{NHC} -Ni-O	172.14	152.05	170.93
C_{NHC} -Ni-C	134.25	116.07	135.51
torsion angles / °			
N_{NHC} - C_{NHC} - Ni - C	-7.23	-79.70	3.86

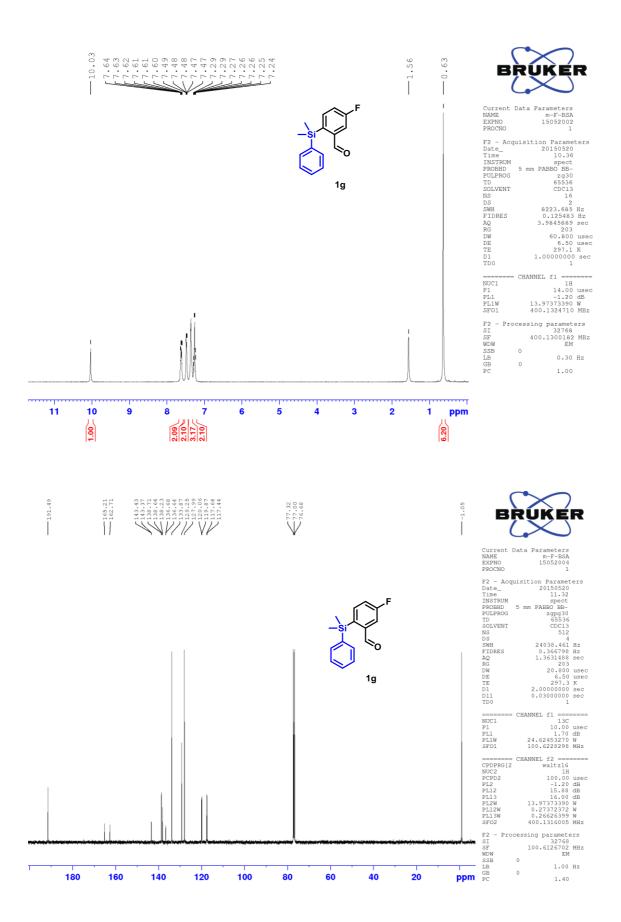
[10] References for the Supporting Information

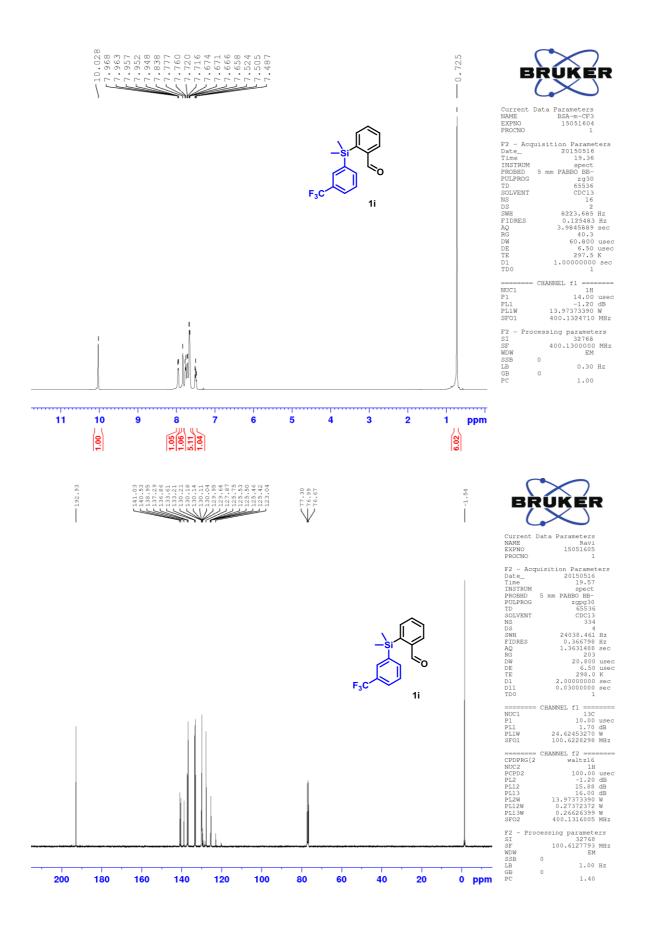
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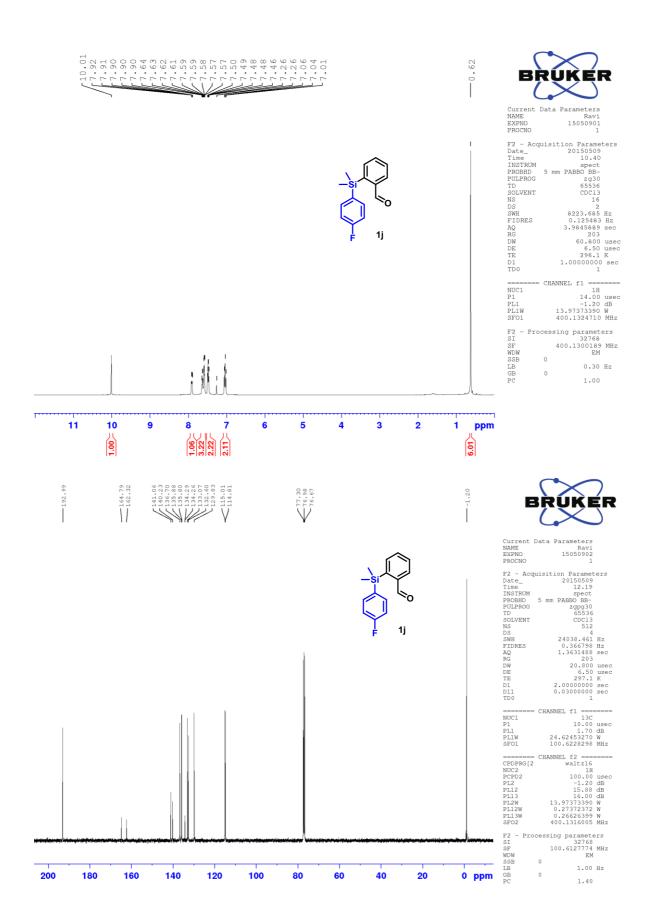
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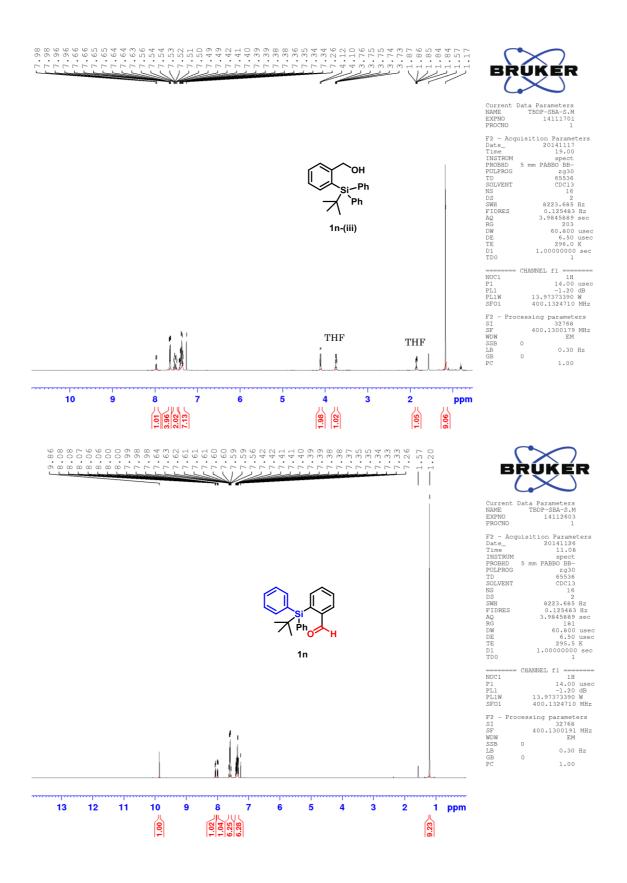
[11] ¹H, ¹³C NMR spectra and SFC/HPLC chart

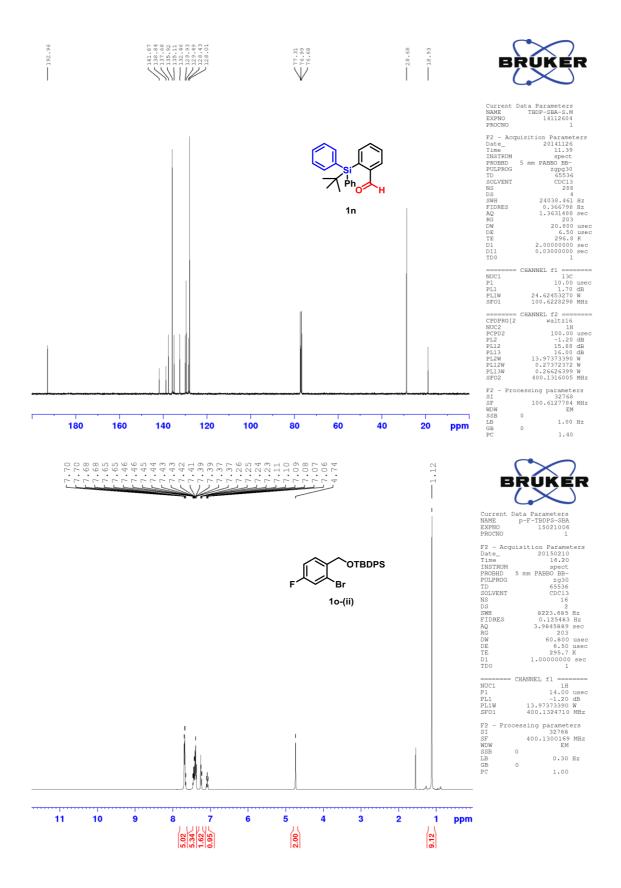


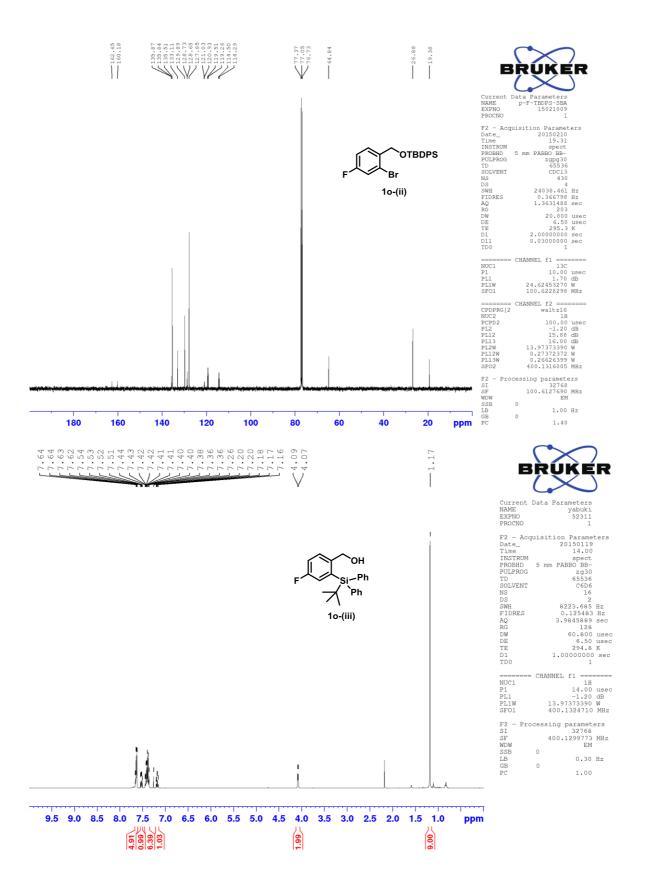


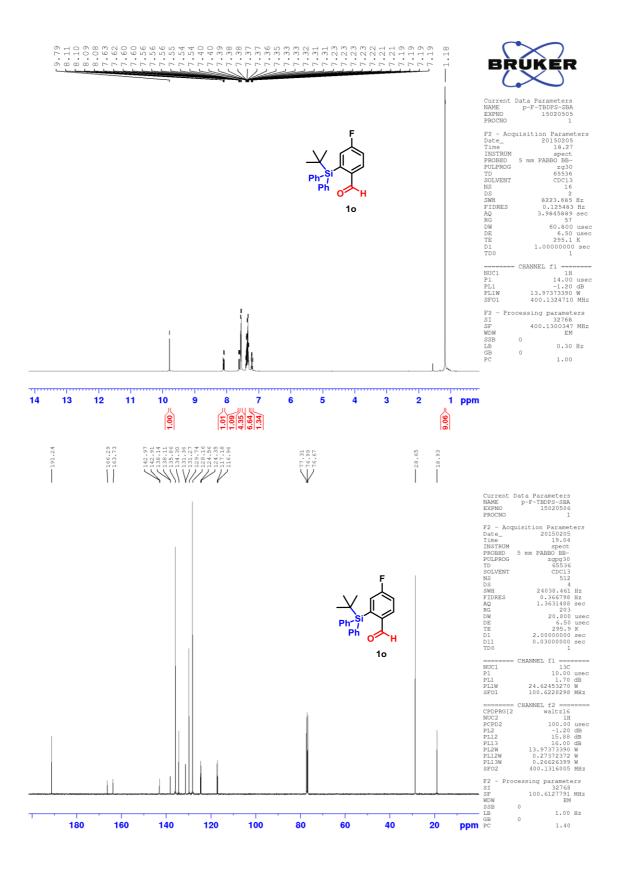


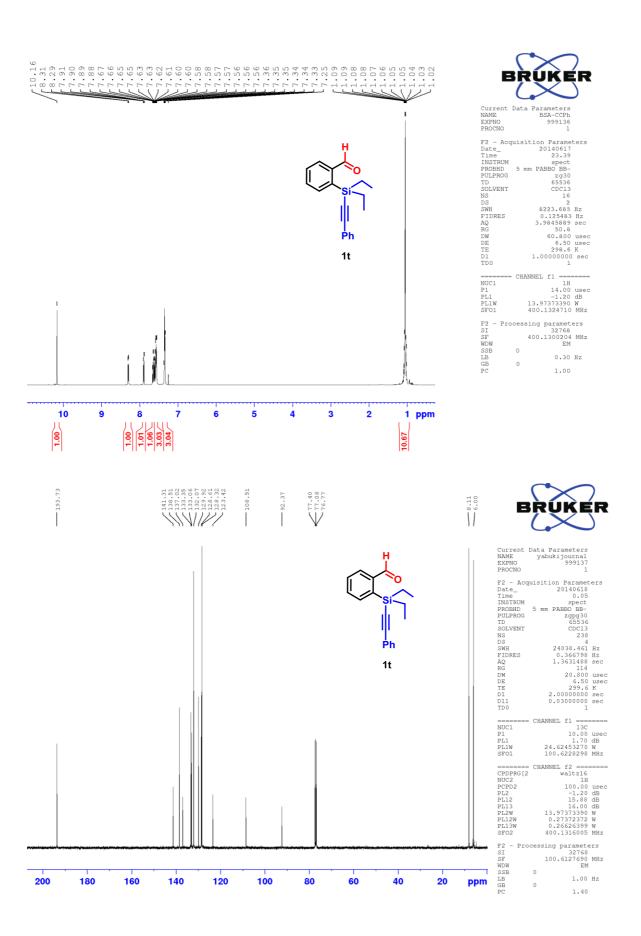


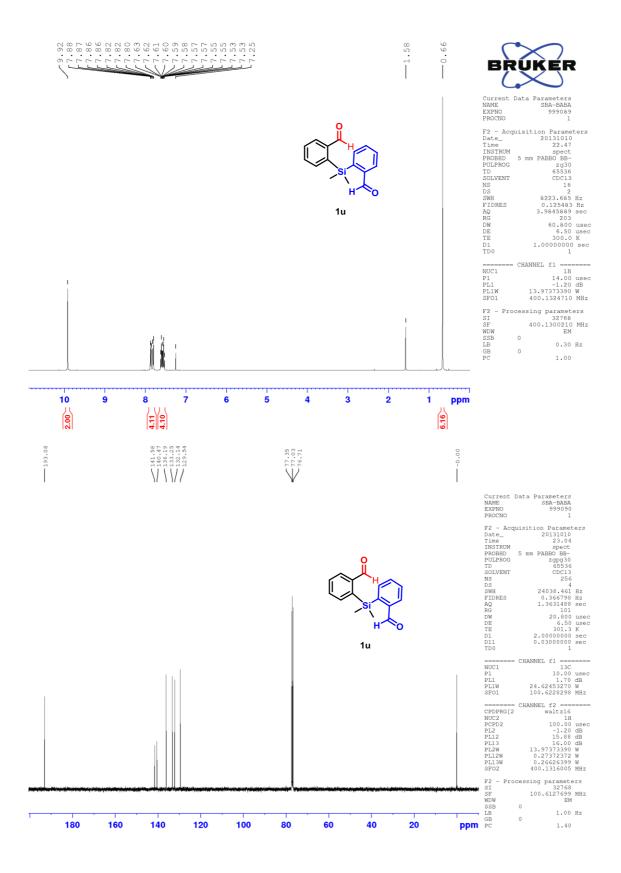


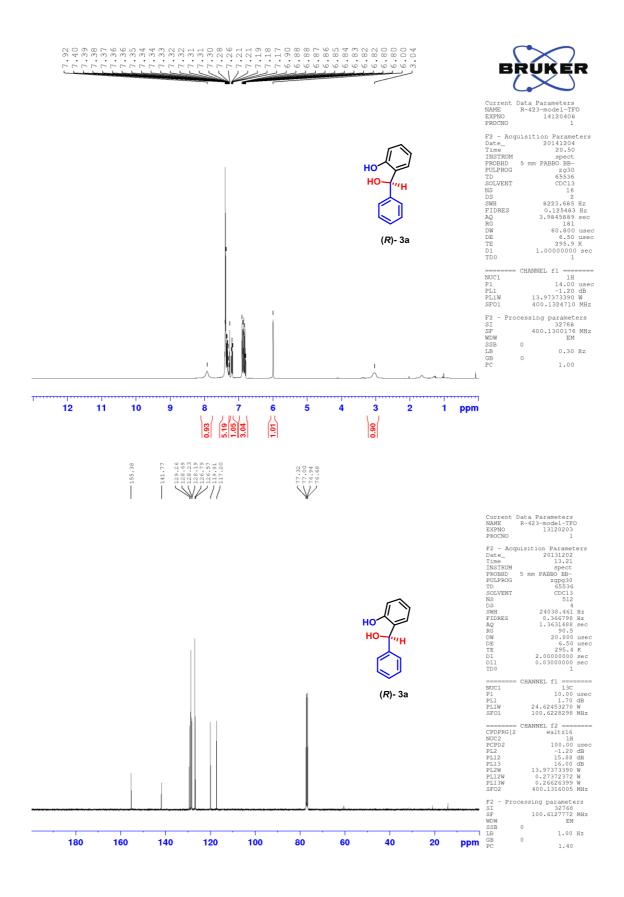


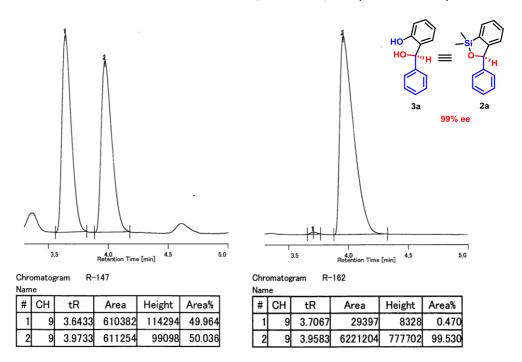


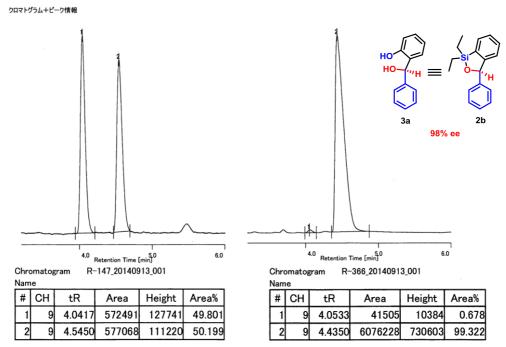




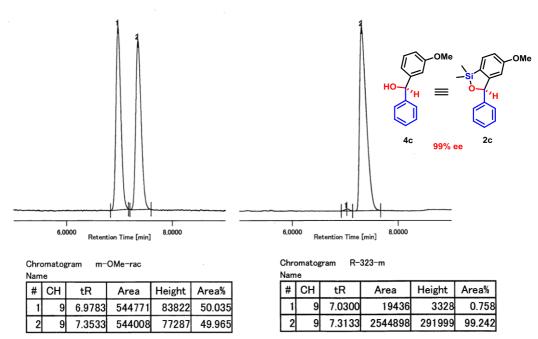




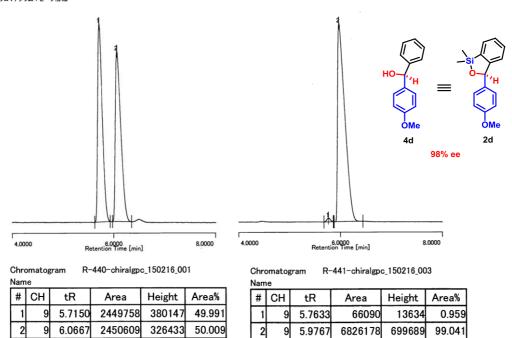


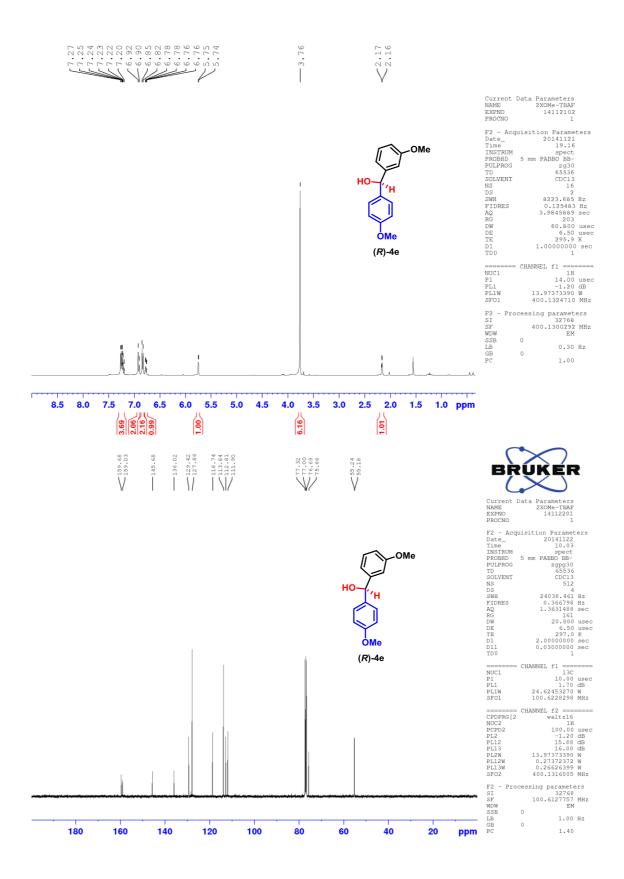


<u>Chiral Separation</u>: SFC; Chiralpak IA; Flow (CO₂) = 2.0 mL/min; Flow (isopropanol) = 0.3 mL/min; T = 25 °C; λ = 250 nm; Back pressure = 15 Mpa

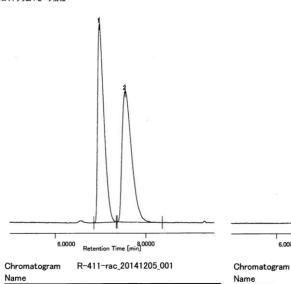


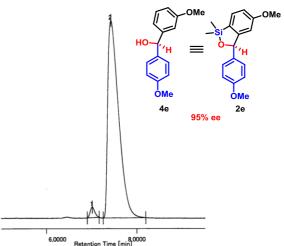
Chiral Separation:SFC; Chiralpak ID; Flow (CO_2) = 2.0 mL/min; Flow (isopropanol) = 0.3 mL/min;T = 25 °C; λ = 250 nm; Back pressure = 15 Mpa





<u>Chiral Separation</u>: SFC; Chiralpak ID; Flow (CO₂) = 2.0 mL/min; Flow (isopropanol) = 0.3 mL/min; T = 25 °C; λ = 250 nm; Back pressure = 15 Mpa 1ロマトグラム+ピーク情報





		Naı
Height	Area%	#

50.722

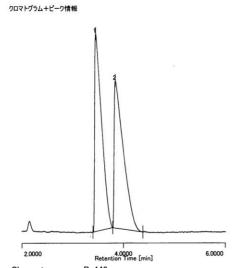
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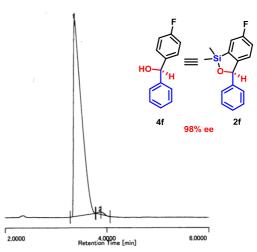
R-411-chiral_20141205_002

N	am	е
_		

#	СН	tR	Area	Height	Area%
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2	9	7.4350	4033877	243167	97.708

<u>Chiral Separation</u>: SFC; Chiralpak ID; Flow (CO₂) = 4.0 mL/min; Flow (CH₂Cl₂) = 0.6 mL/min; T = 25 °C; λ = 250 nm; Back pressure = 15 Mpa





Chromatogram R-442

Name

CH

tR

6.9817

7.5500

Area

1848336

1795685

186846

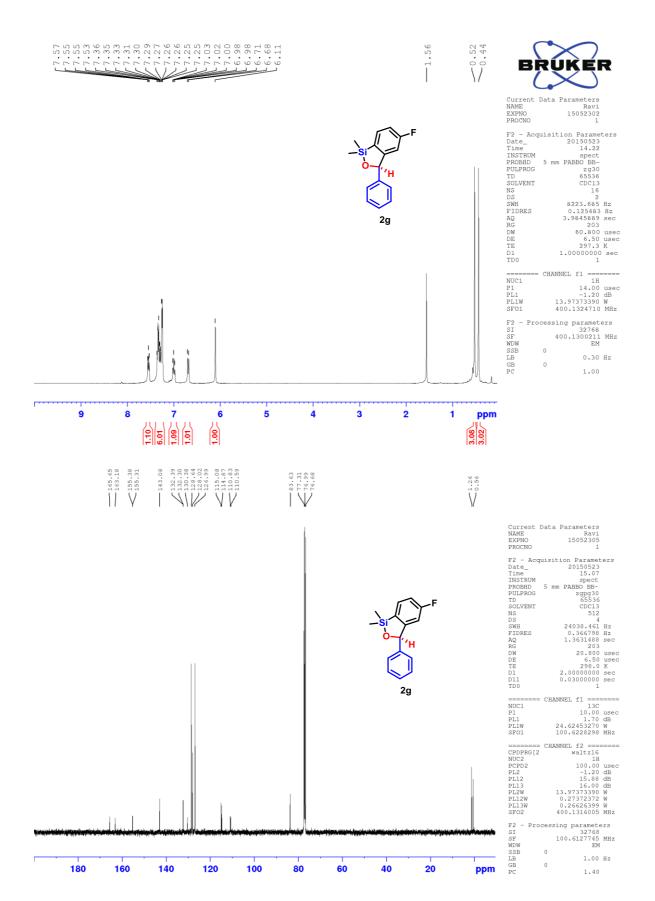
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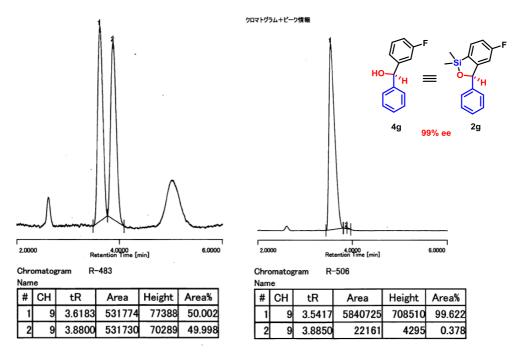
	#	СН	tR	Area	Height	Area%
I	1	9	3.4617	1566649	166133	50.379
	2	9	3.8483	1543080	123901	49.621

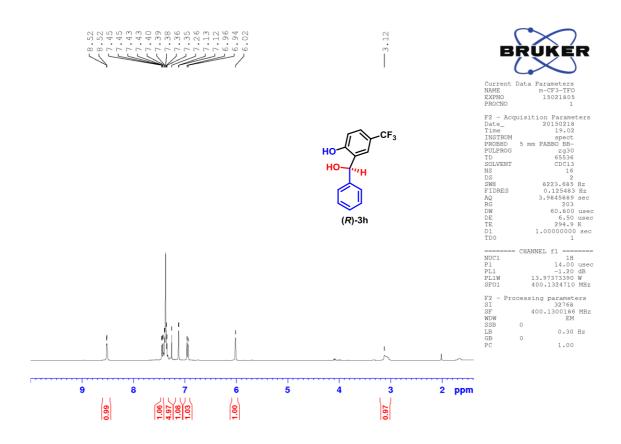
Chromatogram R-387

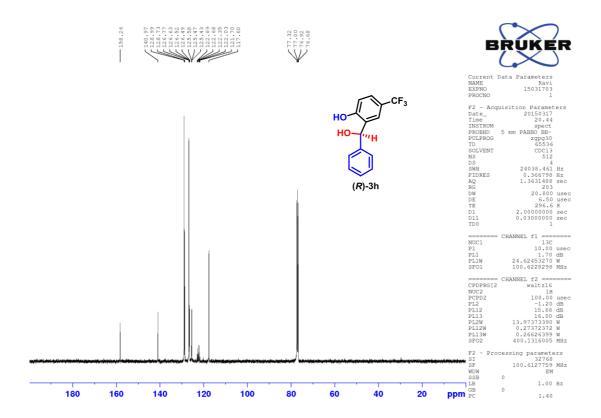
Name

#	СН	tR	Area	Height	Area%
	1 9	3.3733	5257878	490293	99.132
Г	2 9	3.8833	46057	5806	0.868

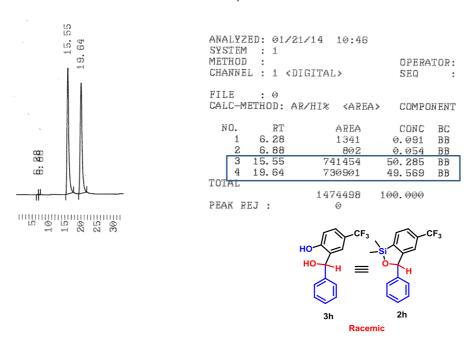




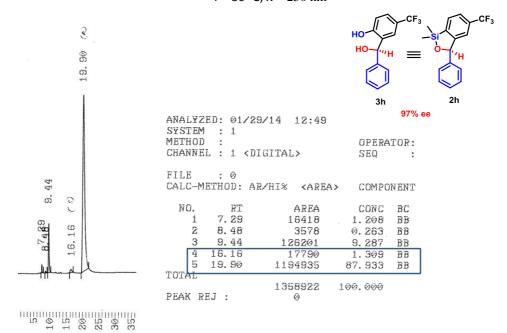




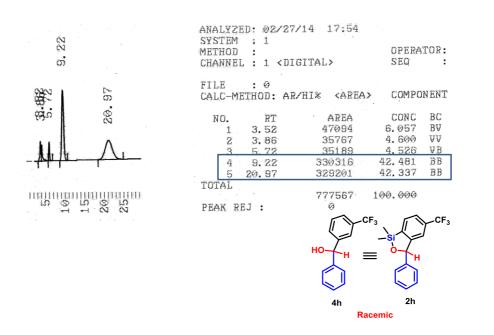
<u>Chiral Separation</u>: HPLC; Chiralpak OD-H; Isopropanol/Hexane = 1:9; Flow = 0.5 mL/min; T = 30 °C; λ = 250 nm



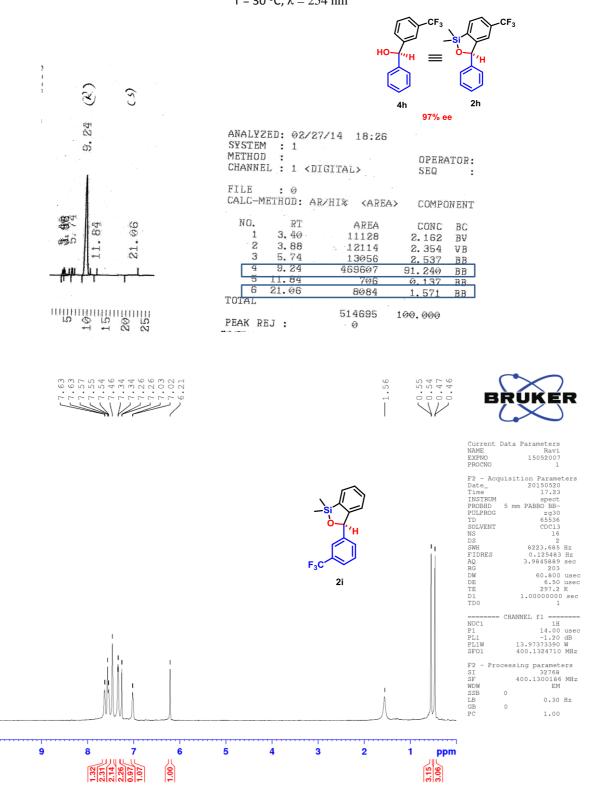
<u>Chiral Separation</u>: HPLC; Chiralpak OD-H; Isopropanol/Hexane = 1:9; Flow = 0.5 mL/min; T = 30 °C; $\lambda = 250$ nm

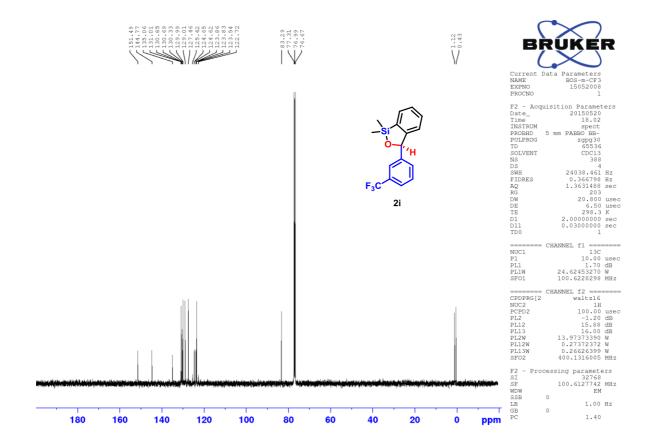


<u>Chiral Separation</u>: HPLC; Chiralpak OB-H; Isopropanol/Hexane = 1:9; Flow = 1.0 mL/min; T = 30 °C; λ = 250 nm

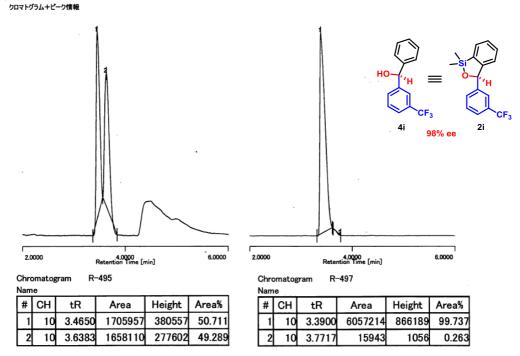


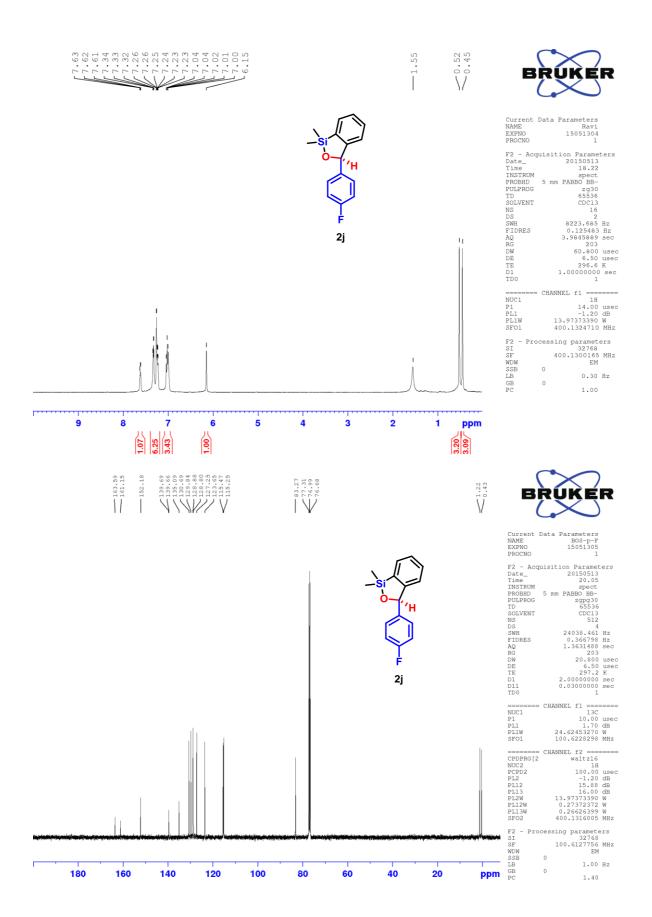
<u>Chiral Separation</u>: HPLC; Chiralpak OB-H; Isopropanol/Hexane = 1:9; Flow = 1.0 mL/min; T = 30 °C; $\lambda = 254$ nm

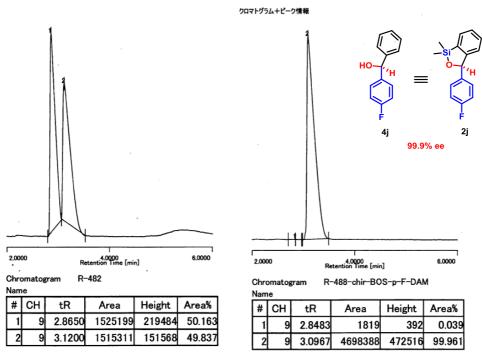


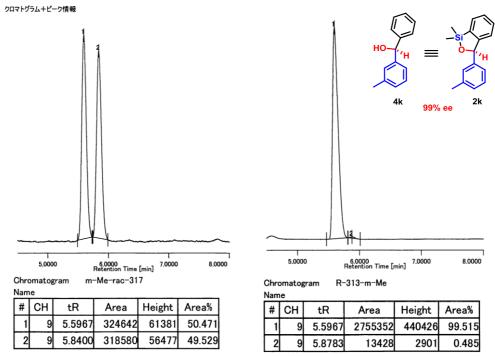


<u>Chiral Separation</u>: SFC; Chiralpak IB; Flow (CO₂) = 5.0 mL/min; Flow (isopropanol) = 0.2 mL/min; T = 25 °C; λ = 230 nm; Back pressure = 15 Mpa









 $\underline{\textbf{Chiral Separation}}; SFC; Chiralpak \ IB; \ Flow \ (CO_2) = 3.0 \ \text{mL/min}; \ Flow \ (Isopropanol)$ =0.3 mL/min;T = 25 °C; $\lambda = 250$ nm; Back pressure = 15 Мра 99% ee 3.0000 6,0000 6.0000 4.0000 Retention Time [min] 5,0000 4.0000 5.0000 Retention Time [min] R-347-4_20141001_001 Chromatogram R-348-1_20141001_001 Chromatogram Name Name

CH

9

tR

3.8567

4.1800

Area

1113148

5134

Height

221329

1674

Area%

99.541

0.459

CH

tR

3.8600

4.1117

Area

801408

802202

Height

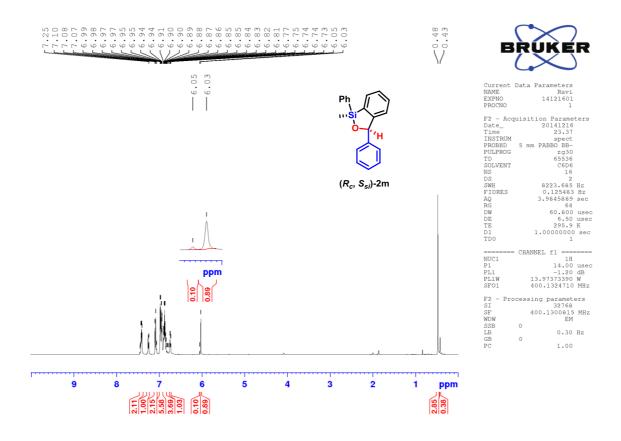
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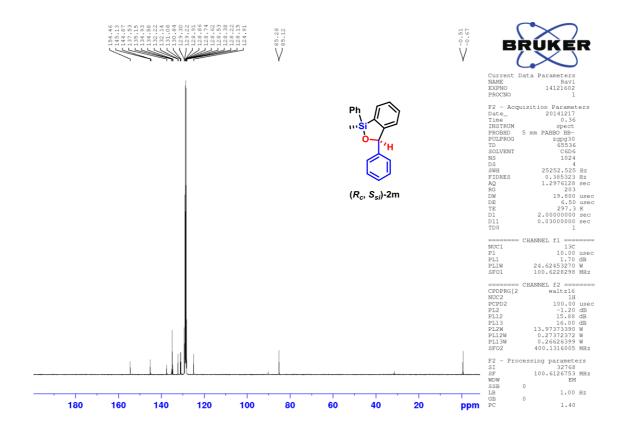
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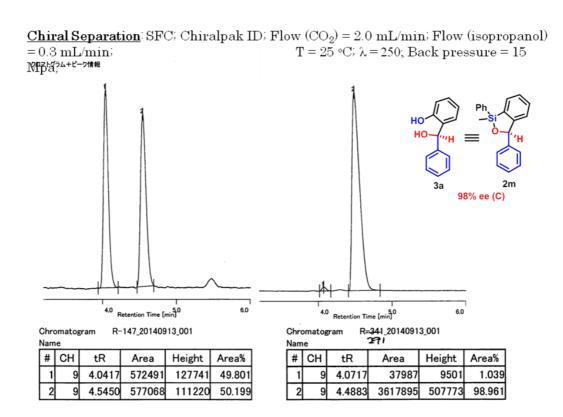
Area%

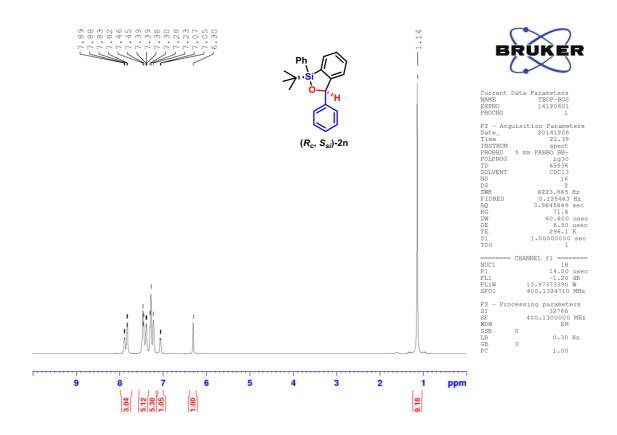
49.975

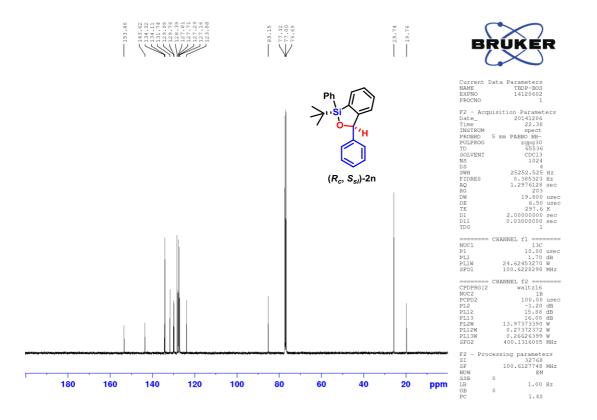
50.025

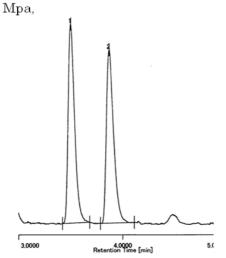


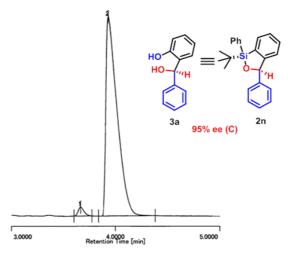










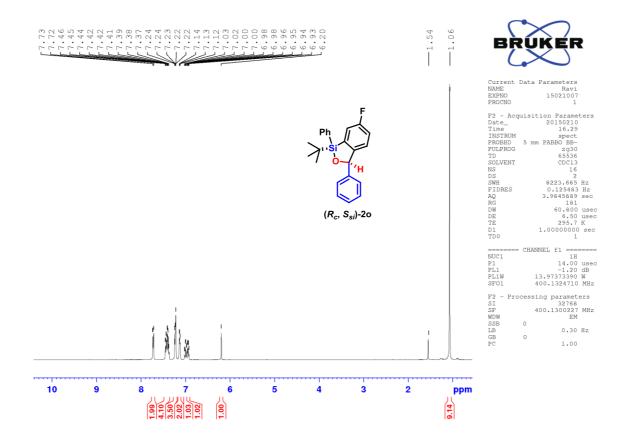


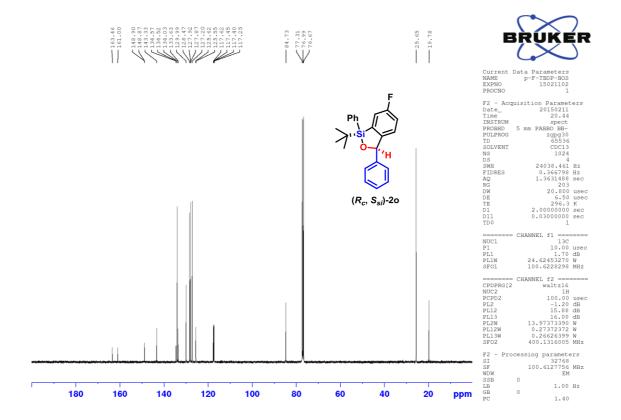
Chromatogram R-147_20140926_001

Name							
#	СН	tR	Area	Height	Area%		
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2	9	3.8700	284717	58297	49.959		

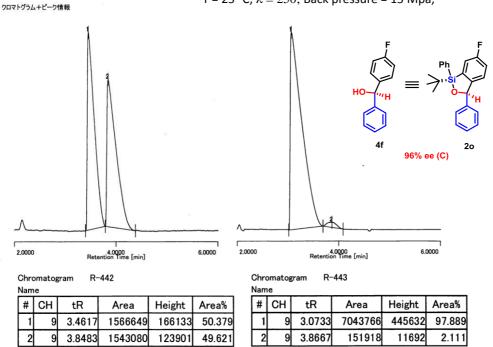
Chromatogram R-426_20141205_001 Name

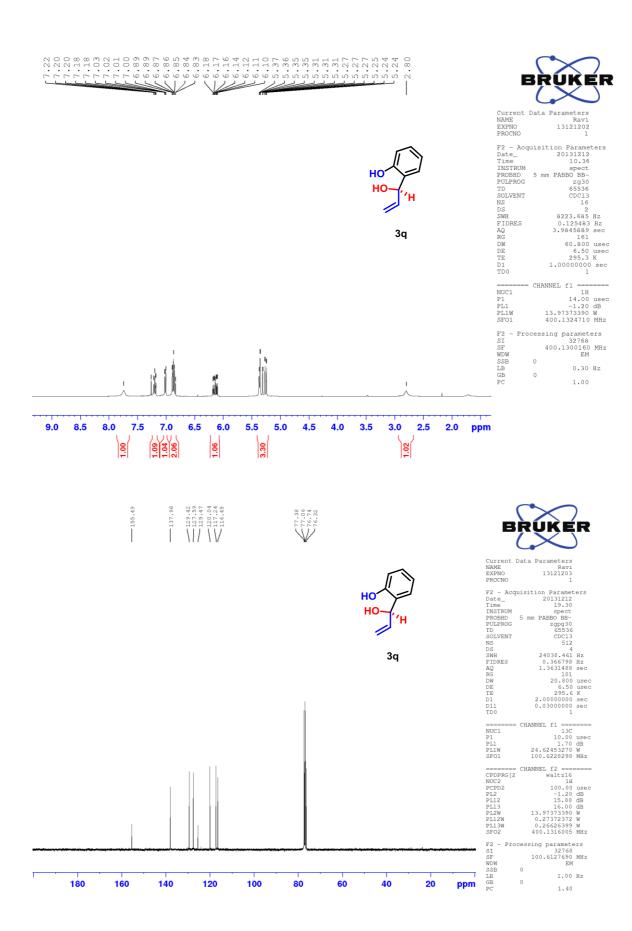
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2	9	3.9333	5209633	700713	97.752

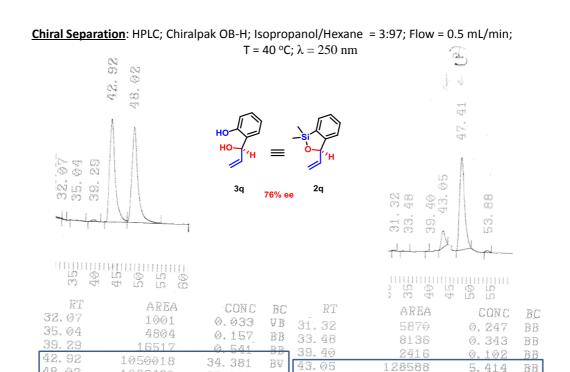




<u>Chiral Separation</u>: SFC; Chiralpak ID; Flow (CO₂) = 4.0 mL/min; Flow (CH₂Cl₂) = 0.6 mL/min; T = 25 °C; $\lambda = 250$; Back pressure = 15 Mpa,







48.02

62.96

80.42

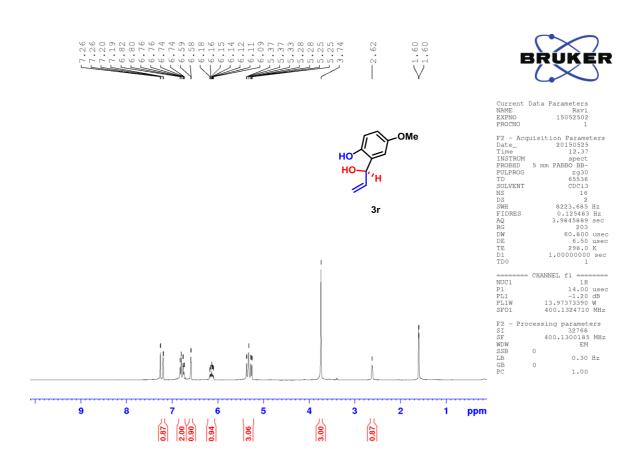
1069490

3740

35.019

1.071

0.122



47.41

53. 88

BB

BB

5.414

0.631

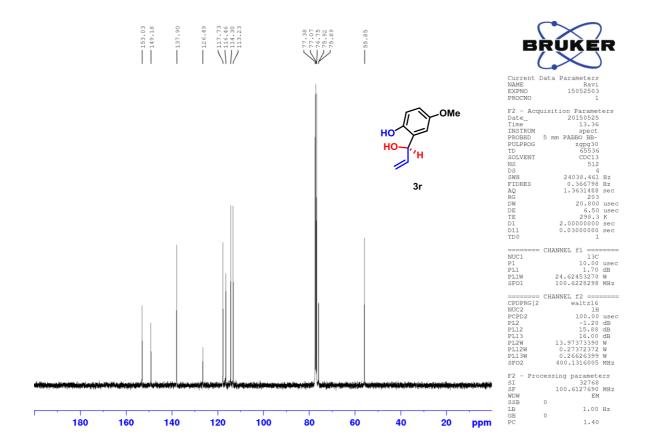
49.060

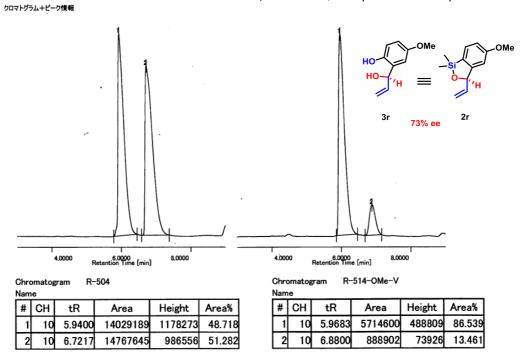
951404

14984

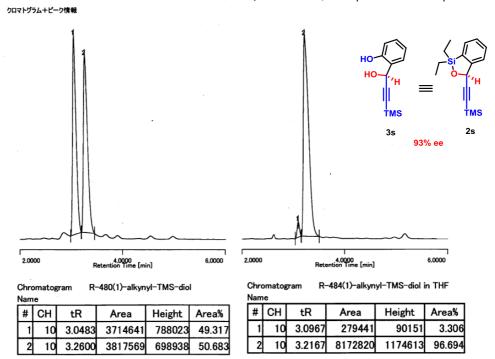
BB

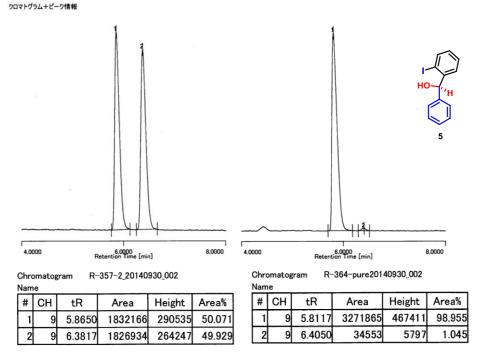
BB

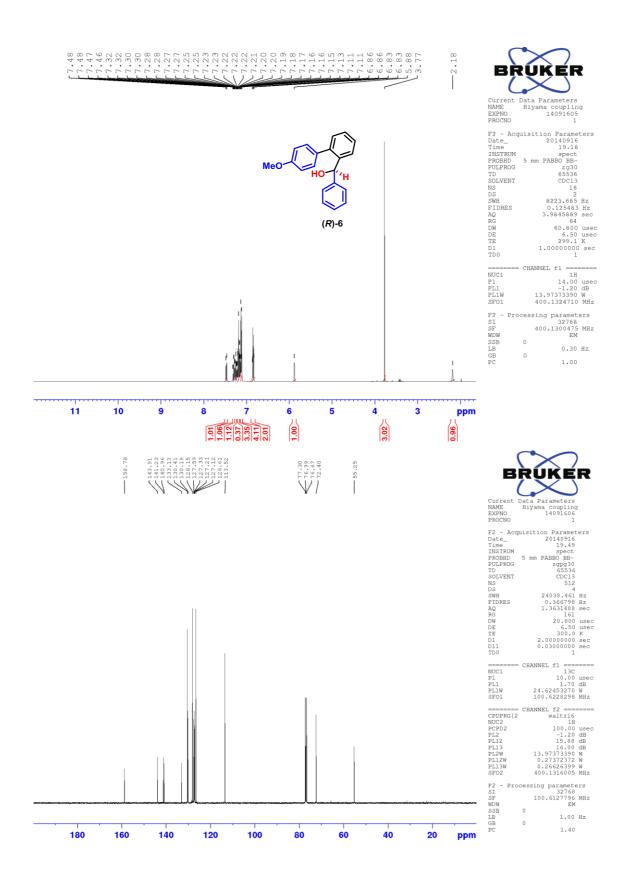


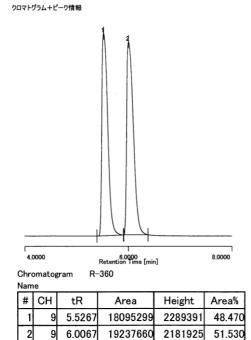


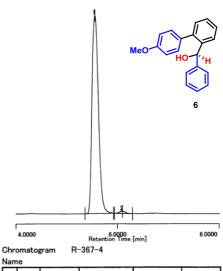
<u>Chiral Separation</u>: SFC; Chiralpak IA; Flow (CO₂) = 4.0 mL/min; Flow (isopropanol) = 0.3 mL/min; T = 25 °C; $\lambda = 220$ nm; Back pressure = 15 Mpa





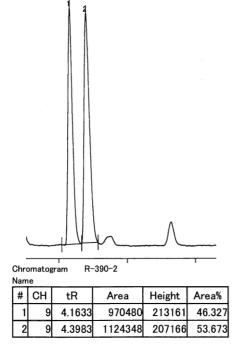


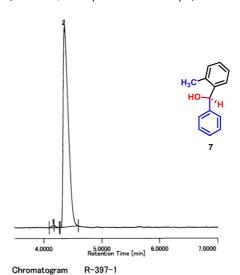




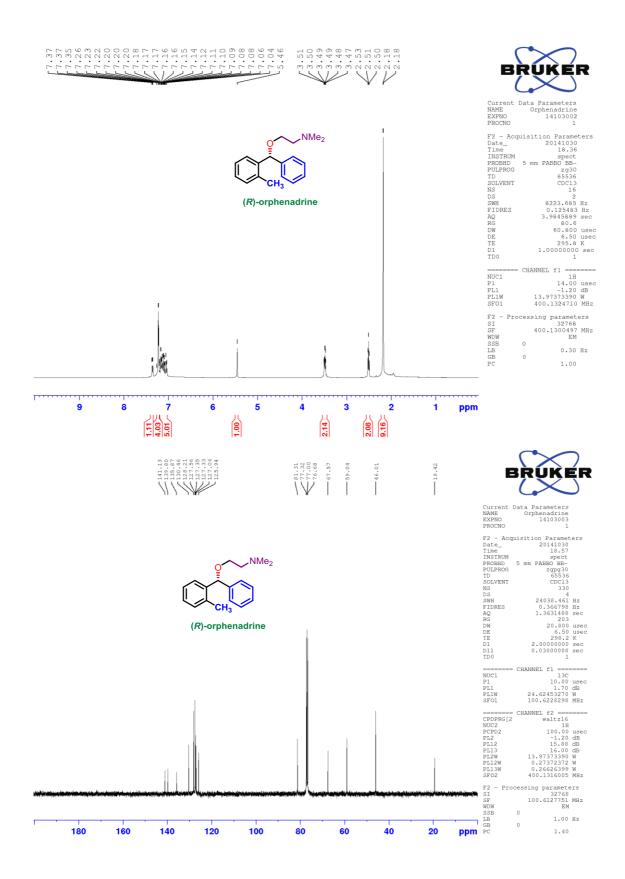
Nam	Name							
#	СН	tR	Area	Height	Area%			
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2	9	6.0967	119839	14440	1.010			

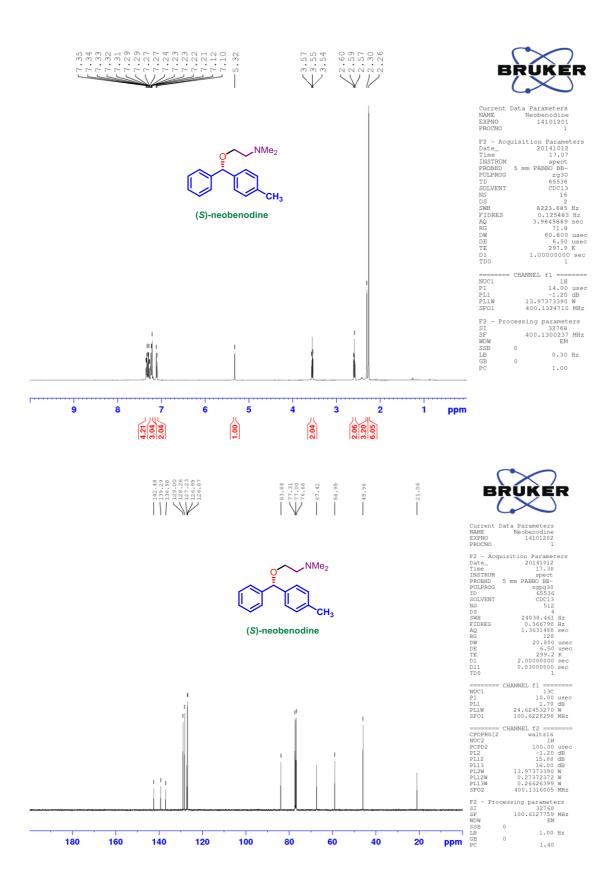
<u>Chiral Separation</u>: SFC; Chiralpak IB; Flow (CO₂) = 2.0 mL/min; Flow (Isopropanol) = 0.3 mL/min; T = 25 °C; $\lambda = 250$; Back pressure = 15 Mpa,

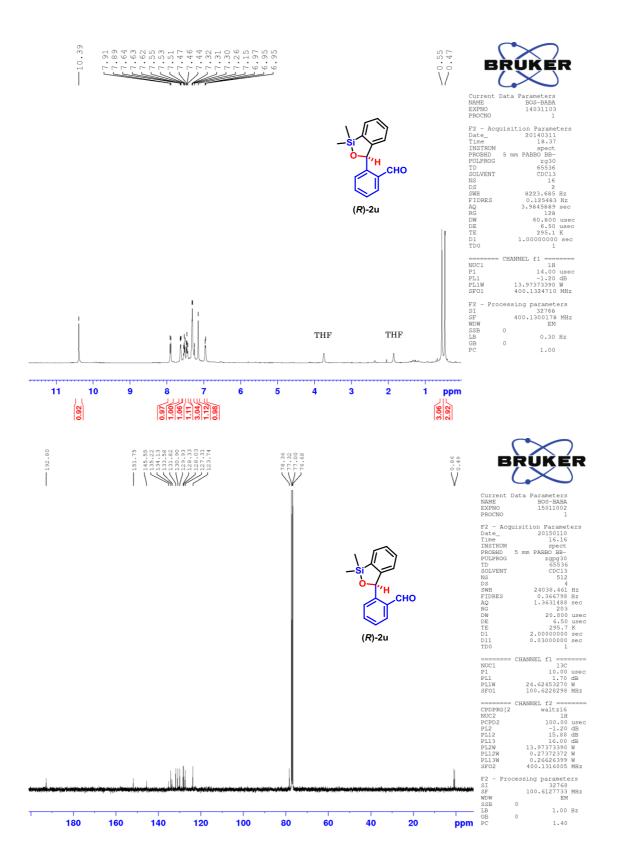




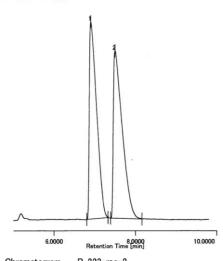
N	am	e				
F	#	СН	tR	Area	Height	Area%
Γ	1	9	4.1800	11939	2844	0.882
Γ	2	9	4.3683	1,341724	219554	99.118

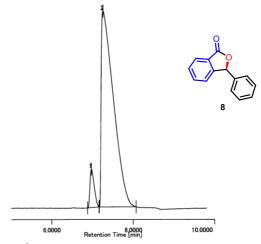






<u>Chiral Separation</u>: SFC; Chiralpak IA; Flow $(CO_2) = 2.4 \text{ mL/min}$; Flow $(CH_2Cl_2) = 0.4 \text{ mL/min}$; T = 40° C; $\lambda = 250$; Back pressure = 15 Mpa, クロマトグラム+ピーク情報





Chromatogram R-223-rac-2

14ame							
#	СН	tR	Area	Height	Area%		
1	10	6.9067	7128018	573902	49.207		
2	10	7.5017	7357845	486547	50.793		

R-223-chir-2 Chromatogram

Name

#	СН	tR	Area	Height	Area%
1	10	6.9717	1222845	171222	6.338
2	10	7.2650	18070834	870158	93.662

