

# Supporting Information

## 2-D perovskite activation with an organic luminophore

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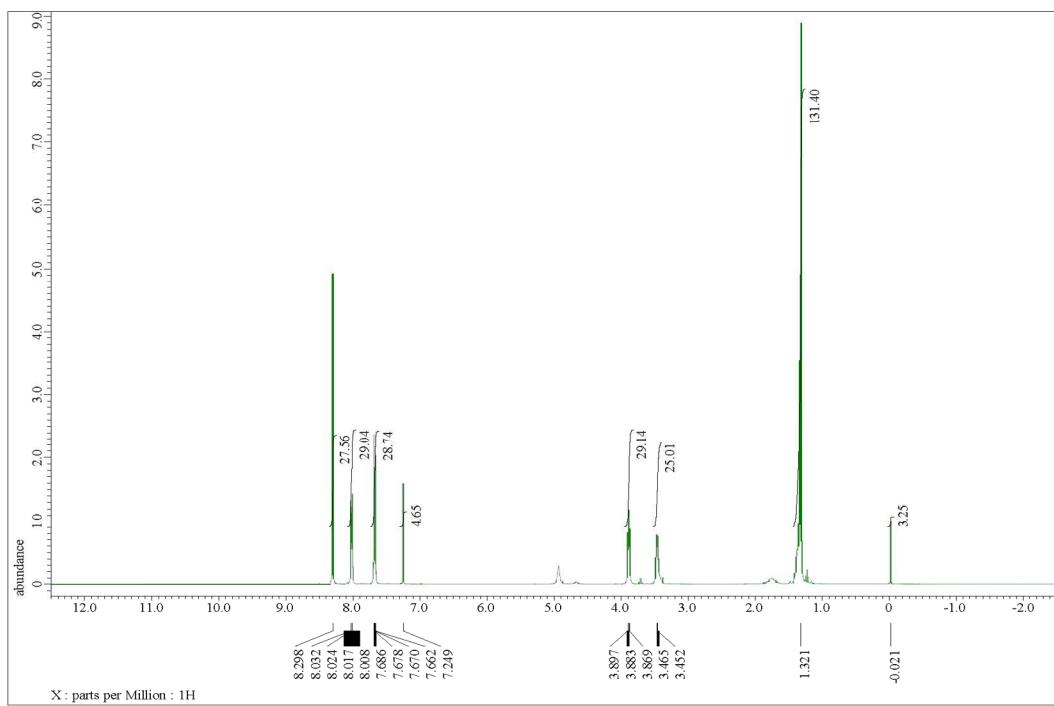
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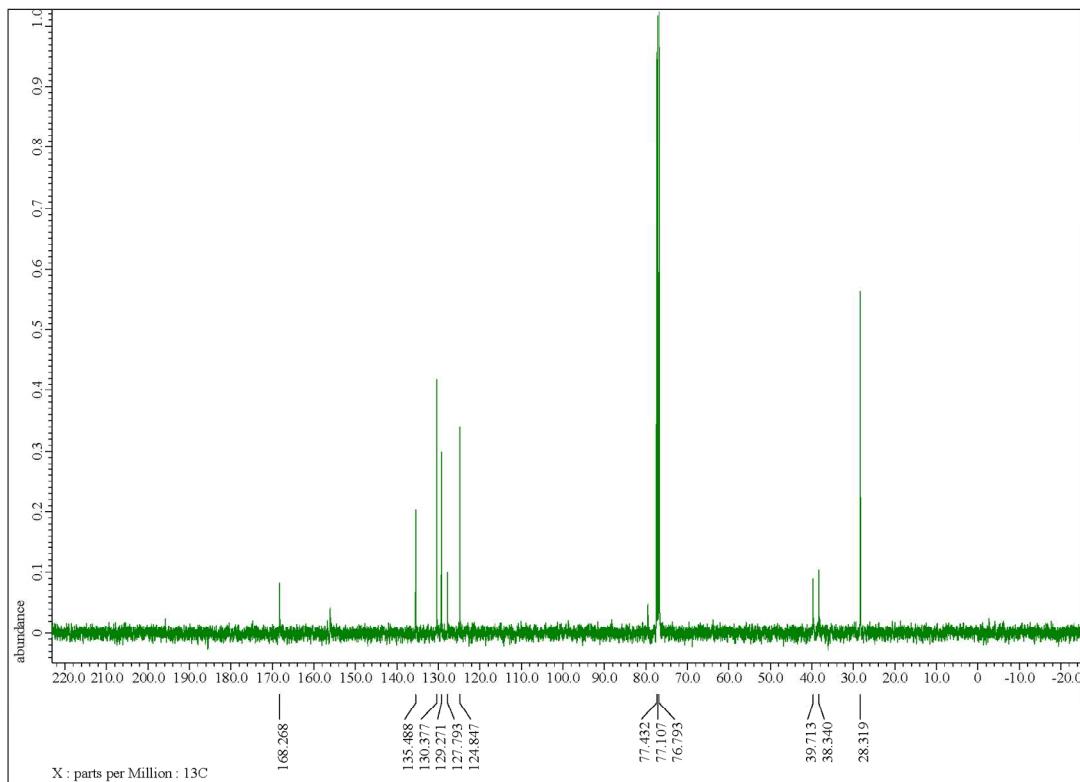
• **Nuclear Magnetic Resonance measurement of NAA<sub>B</sub>:**

The structure and the purity of the fluorophore containing ammonium salt have been checked by NMR measurement. The <sup>1</sup>H (figure S1) and <sup>13</sup>C spectra (figure S2) were recorded at 20 °C in a JOEL ESC-400MHz spectrometer with a JOEL autotune 5mm NMR broadband probe. Chemical shifts were given in ppm relative to chloroform CDCl<sub>3</sub> 7,26ppm (H) and 77,16ppm(C):

<sup>1</sup>H NMR(400MHz, CDCl<sub>3</sub>, $\delta$ ): 7.68 (dd , J=3.2Hz and 6.4Hz, 2H ; Ar H), 3.46(d, J=5.2 Hz,2H ; CH<sub>2</sub>N), 3.897(t, J= 6.8 and 5.6, 2H; CH<sub>2</sub>N), 1.32(s, 3H;CH<sub>3</sub> ),8.032(dd, J=3.2 and 6.2,2H; Ar H), 8.298( s, 2H, Ar H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>,  $\delta$ ):168.26(C=O), 135.48, 127.79(C<sub>Ar</sub> ), 130.37, 129.27,124. 84(C<sub>Ar</sub>-H), 77.43(C<sub>4</sub>-(CH<sub>3</sub>)<sub>3</sub> ), 39.71,38.34 (CH<sub>2</sub> N), 28.31(CH<sub>3</sub> ).

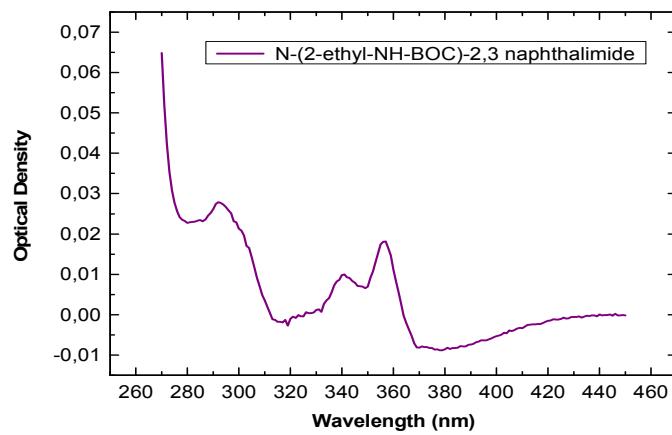


**Figure S1:**  $^1\text{H}$  NMR N-(2-ethyl-NH-Boc)-2,3 naphthalimide ( $\text{CDCl}_3$ )



**Figure S2:**  $^{13}\text{C}$  NMR N-(2-ethyl-NH-Boc)-2,3 naphthalimide ( $\text{CDCl}_3$ )

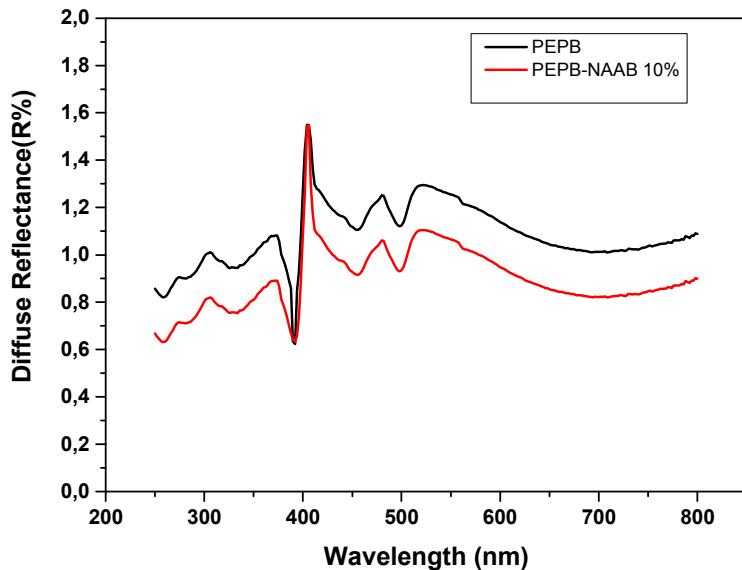
- **Absorption spectrum of a solution of pure NAAB in DMF**



**Figure S3:** Absorption spectrum of a solution of pure NAAB in DMF

- **Diffuse reflectance measurement of PEPB and PEPB-NAAB10%**

Diffuse reflectance measurements are performed with a conventional Perkin Elmer 950 UV/Vis spectrometer, using a 100 mm integrating sphere.



**Figure S4:** Diffuse reflectance measurement of the samples PEPB and PEPB-NAAB10%

- **Calculated values of the brilliance improvement from measured values of the optical density enhancement**

Table S1 reports the optical density of PEPB and PEPB-NAAB10%: respectively  $A_{PEPB}$  and  $A_{PEPB-NAAB10\%}$ , measured from figure 3 at three wavelengths: 250 nm, 340 nm, 360 nm

**Table S1:**

Wavelength (nm)	250 nm	340 nm	360 nm
$A_{PEPB}$	$0,658 \pm 0,005$	$0,237 \pm 0,005$	$0,178 \pm 0,005$
$A_{PEPB-NAAB10\%}$	$0,797 \pm 0,005$	$0,386 \pm 0,005$	$0,297 \pm 0,005$
$I_{abs}(PEPB-NAAB10\%)$	$1,08 \pm 0,01$	$1,4 \pm 0,01$	$1,47 \pm 0,01$
$I_{abs}(PEPB)$			

According to the definition of the optical density  $A$ , if  $I_0$  is the intensity of the incident beam, the intensity of the transmitted beam is  $I_t = I_0 \cdot 10^{-A}$ . Assuming that the reflectance diffusion is negligible (which is reasonable, see figure S4), the intensity which has been absorbed by the sample is  $I_{abs} = I_0 - I_t = I_0 \cdot (1 - 10^{-A})$ . Assuming a complete energy transfer, the calculated expected brilliance improvement is then:

$$\frac{I_{abs}(PEPB - NAAB10\%)}{I_{abs}(PEPB)} = \frac{1 - 10^{-A_{PEPB-NAAB10\%}}}{1 - 10^{-A_{PEPB}}} \quad (1)$$