

Supporting Information

High-Temperature Ferroelectricity and Photoluminescence in a Hybrid Organic-Inorganic Compound: (3-Pyrrolinium)MnCl₃

Heng-Yun Ye ^{†,‡}, QiongHua Zhou, ^{‡,§}, XiangHong Niu[‡], Wei-Qiang Liao[†], Da-Wei Fu[†], Yi Zhang[†], Yu-Meng You^{†,*}, Jinlan Wang^{‡,*}, Zhong-Ning Chen[§], Ren-Gen Xiong^{†,*}

[†]Ordered Matter Science Research Center, Southeast University, Nanjing 211189, P. R. China.

[‡]Department of Physics, Southeast University, Nanjing 211189, P. R. China.

[§]Fujian Institute of Research on the Structure of Matter, The Chinese Academy of Sciences, Fuzhou 350002, P. R. China.

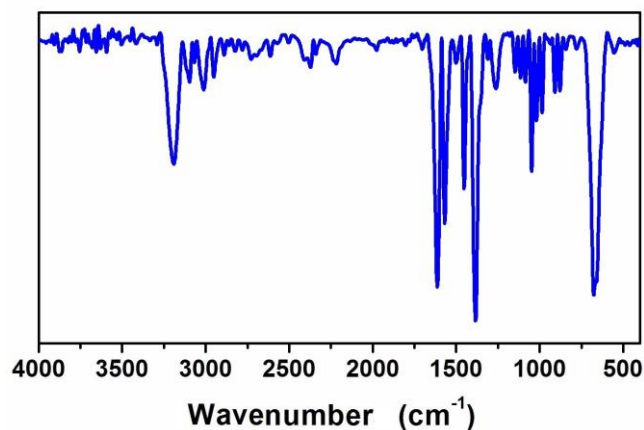


Figure S1. IR spectrum for (3-pyrrolinium)MnCl₃ (**1**).

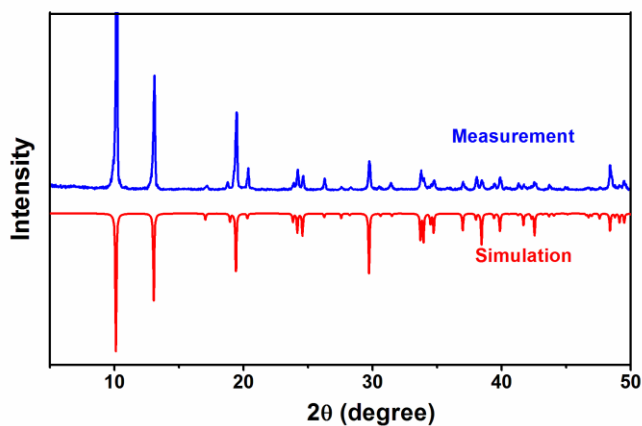


Figure S2. Pattern of the powder X-ray diffraction (PXRD) of (3-pyrrolinium)MnCl₃ (**1**), verifying the purity of the bulk phase.

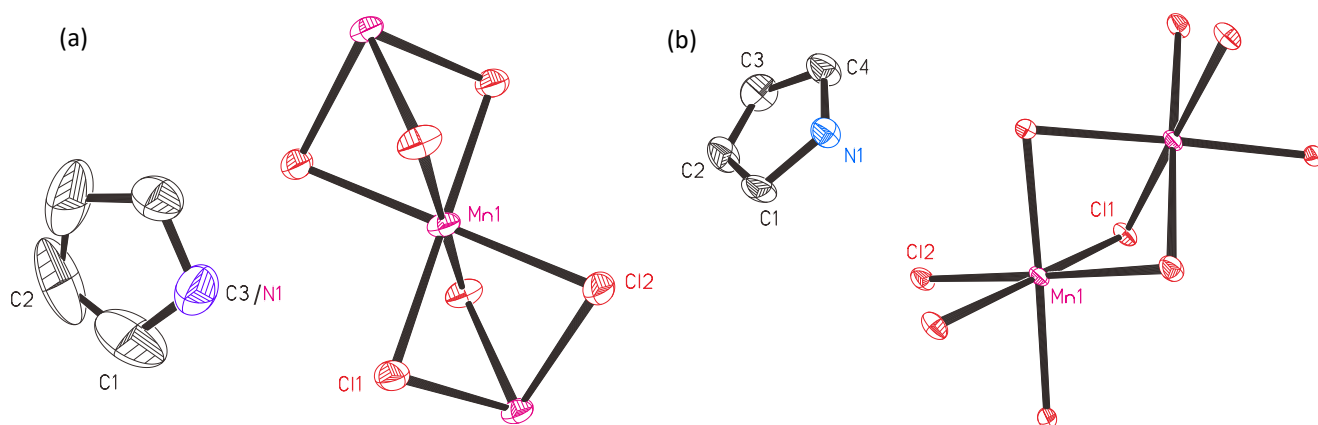


Figure S3. Ellipsoid drawings of symmetric units of (3-pyrrolinium) MnCl_3 (**1**) in (a) the HTP at 413 K and in (b) the LTP at 273 K. In (a), the site of N1/C3 is occupied equally by C and N atoms. Atoms not labeled are generated by symmetry operator. Displacement ellipsoids were drawn at 30% probability level.

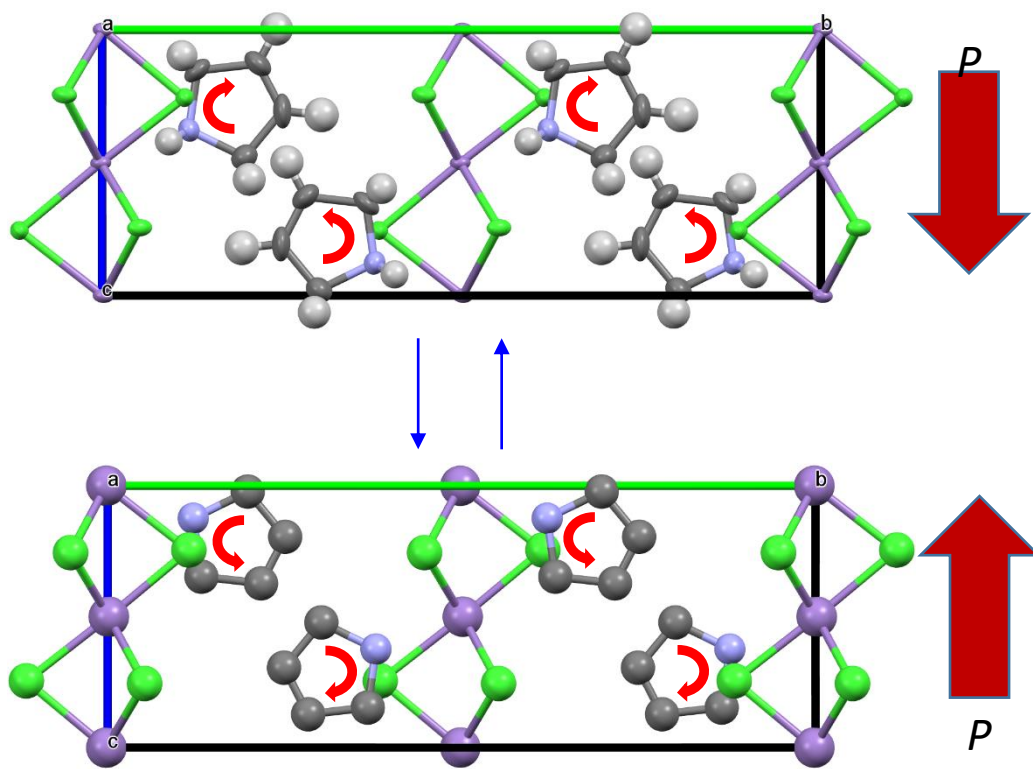


Figure S4. Schematic drawing of the realization of polarization reversal in (3-pyrrolinium) MnCl_3 (**1**).

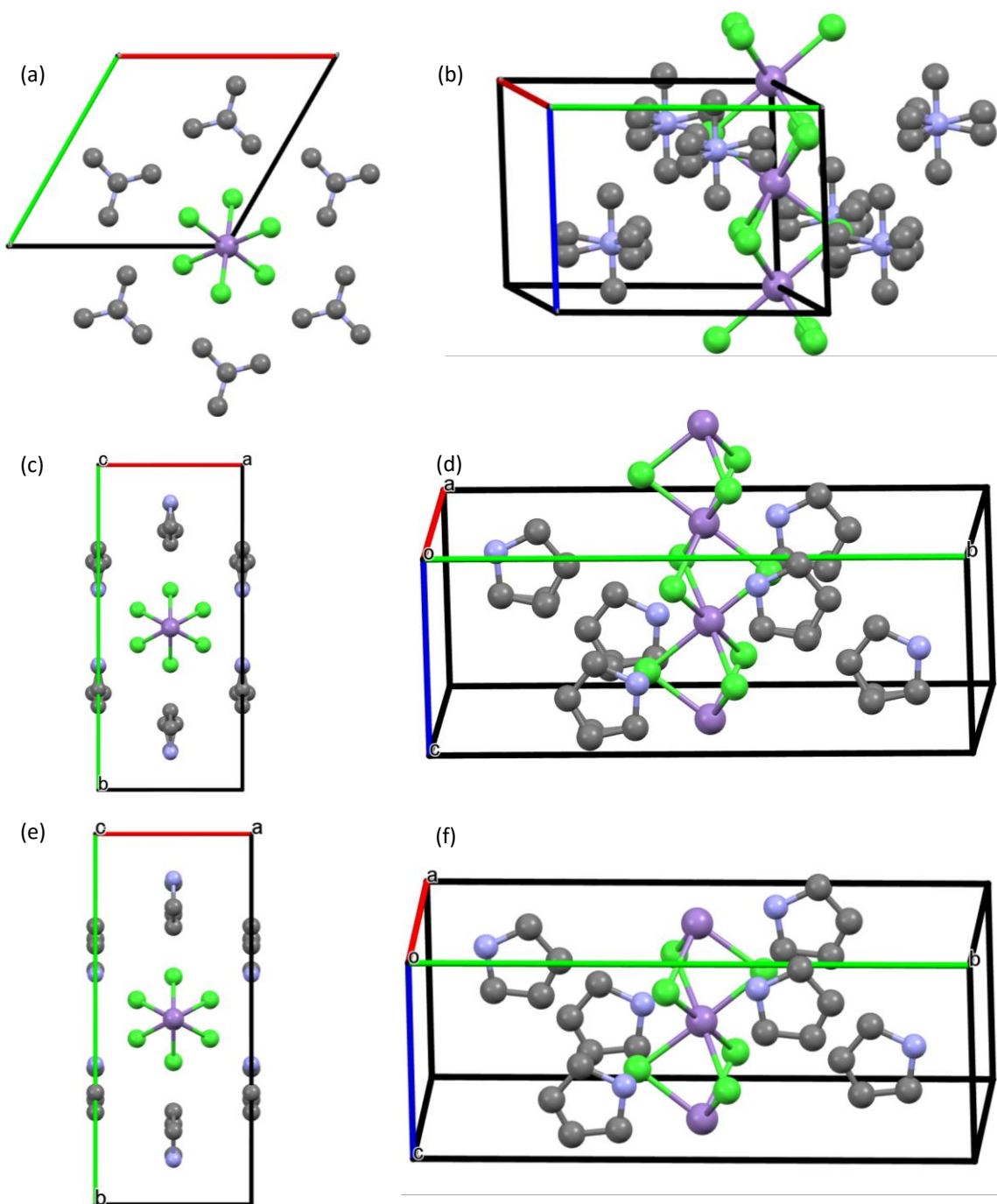


Figure S5. Comparison of crystal structures of (tetramethylammonium)MnCl₃ (a,b), (pyrrolidinium)MnCl₃ (c,d) and (3-pyrrolinium)MnCl₃ (1) (e, f). H atoms were omitted for clarity.

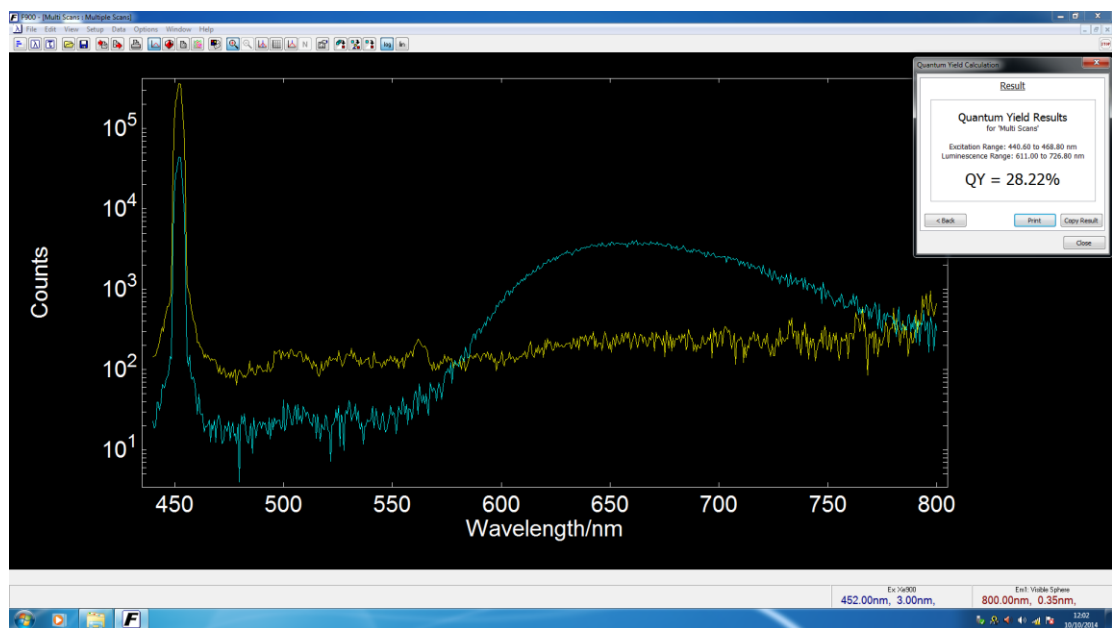


Figure S6. The quantum yield measurement of (3-pyrrolinium)MnCl₃ (**1**), giving a quantum yield of 28.22%.

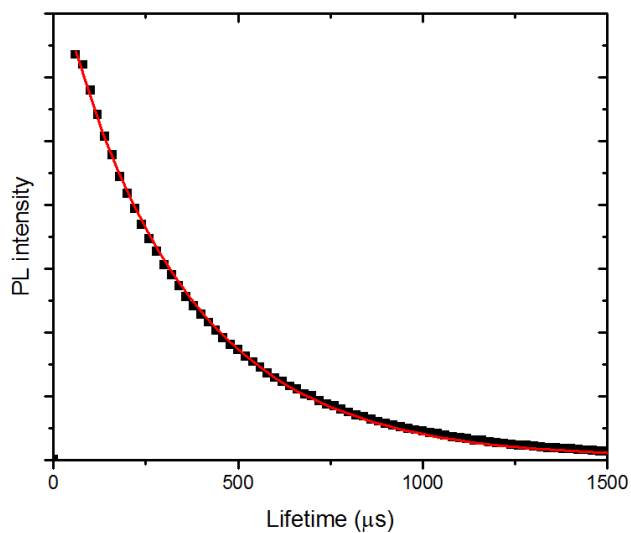


Figure S7. The lifetime measurement of (3-pyrrolinium)MnCl₃ (**1**). The fitted lifetime is 333.6 μs.

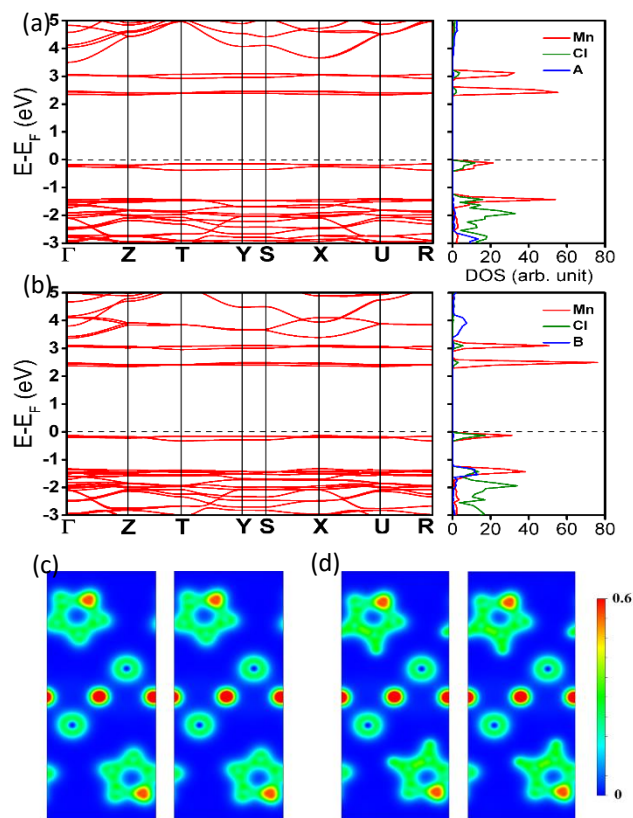


Figure S8. The band structure and corresponding density of states (DOS) of (a) AMnCl_3 and (b) BMnCl_3 . Highest occupied band and lowest unoccupied band of (c) AMnCl_3 and (d) BMnCl_3 . The unit of the scale is e/bohr^3 . A = (pyrrolidinium), B = (3-pyrrolinium).

Table S1. Fitted Curie constants at different frequencies.

Frequency	C_{para} (K)	C_{ferro} (K)	$C_{\text{para}}/C_{\text{ferro}}$	T_0 (K)
1 MHz	4086	1614	2.53	374.2
100 kHz	4953	2000	2.48	374.6
10 kHz	5906	2418	2.44	375.3
5 kHz	6134	2578	2.38	375.5
1 kHz	6706	3163	2.12	376.1
500 Hz	7085	3374	2.10	376.3

Table S2. Comparison between ground state of (pyrrolidinium)MnCl₃ and (3-pyrrolinium)MnCl₃.

Property	(pyrrolidinium)MnCl ₃	(3-pyrrolinium)MnCl ₃ .
Cation	pyrrolidinium	3-pyrrolinium
a / Å	7.65	7.36
b / Å	17.25	17.49
c / Å	6.48	6.47
Magnetic order	AFM	AFM
Magnetic moment / μ_B per Mn atom	4.4	4.4
P_s / $\mu C \cdot cm^{-2}$	8.04	8.04
Charge transfer /electron per cation	0.81	0.82

*The cell constants are fixed to experimental value during optimization.