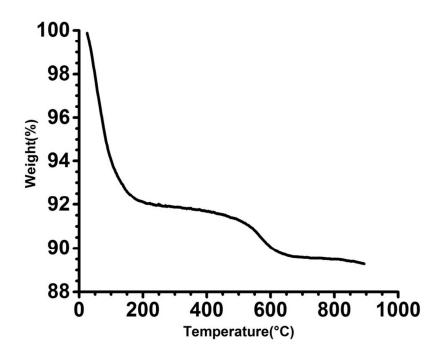
## **Supporting Information**

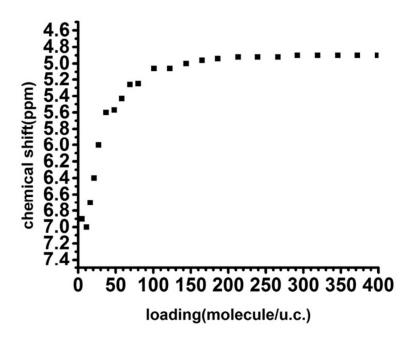
## Water Interactions in Zeolite Catalysts and Their Hydrophobically-Modified Analogues

Kuizhi Chen, Jarred Kelsey, Lu Zhang, Daniel Resasco, and Jeffery L. White\*

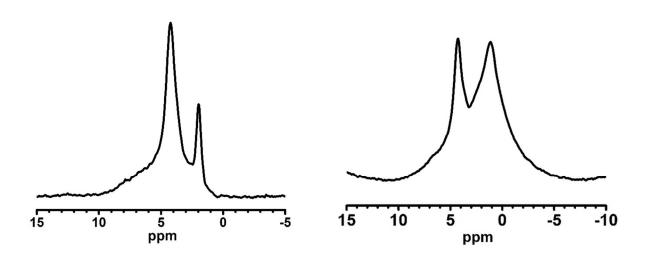
Department of Chemistry, Oklahoma State University, Stillwater, OK 74078



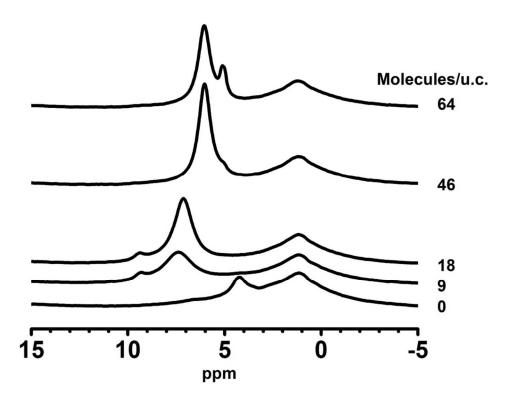
**Figure S1.** Thermal gravimetric analysis (TGA) curves for ETS-HZSM-5, with the experiment done in air. The 8% mass loss at 100-150 C is from water, and the ca. 3% mass loss near 500 C is from decomposition of the surface ETS groups.



**Figure S2.** Complete water loading/chemical shift titration curve for water loaded in Si/Al = 15 H-ZSM-5 up to 400 water molecules per unit cell.



**Figure S3.** Representative <sup>1</sup>H MAS NMR spectra for calcined and dehydrated (**left**) H-ZSM-5 and (**right**) ETS-H-ZSM-5.



**Figure S4.**  $^{1}$ H MAS NMR spectra for  $D_{2}O$  in ETS-HZSM-5 as a function of its loading per zeolitic unit cell. Peak assignments are as in the text for Figures 8 and 9.