

Cadmium Stabilization Efficiency and Leachability by CdAl_4O_7 Monoclinic Structure

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Supporting Information

Including 2 tables and 8 figures over 10 pages

Table S1. Quantitative phase analysis results and the calculated transformation ratios (TR) of the authentic CdO : CdAl₄O₇ : glass (amorphous) : CaF₂ determined via the Rietveld refinement method. The refinement was evaluated by the factors derived from the fitness criteria provided in Supporting Information Table S1.

Table S2. Weight variation of CdAl₄O₇ powder with or without calcination at 1025 °C for 3 h.

Figure S1. X-ray diffraction (XRD) patterns of the alumina powder (AlOOH, boemite; ICDD PDF no. 74-1895), γ -Al₂O₃ obtained by heating the alumina powder at 650°C for 3 h and the cadmium oxide powder (CdO, ICDD PDF no. 75-0594).

Figure S2. Schematic diagram of the cadmium stabilization reaction pathway with thermal treatment.

Figure S3. XRD patterns of the products obtained from the reaction of CdO + γ -Al₂O₃ with Cd/Al molar ratios of 2/1 and 3/4 heated at 1000 °C for 24 h. The standard patterns were derived from the International Centre for Diffraction Data (ICDD) database, including CdO (ICDD PDF no. 75-0594), CdAl₄O₇ (ICDD PDF no. 22-1061) and α -Al₂O₃ (ICDD PDF no. 74-1081).

Figure S4. A comparison of the observed (red cross symbol) and calculated XRD patterns (blue line) between $2\theta = 28.75^\circ$ and 30.00° in Figure 2 shows the unreported Bragg diffraction position (400) of the CdAl₄O₇ phase.

Figure S5. X-ray photoelectron spectroscopy (XPS) spectra for Cd 3d of (a) CdO and (b) CdAl₄O₇.

Figure S6. Particle size distributions of (a) CdO and (b) as-prepared CdAl₄O₇ powders for CPLT.

Figure S7. Decrease of the (a) Cd 3d signal and increase of the (b) Al 2p signal on the surface of the CdAl₄O₇ sample leached for 120 min and observed via X-ray photoelectron spectroscopy (XPS). The leaching process comprised the constant pH leaching test (CPLT) at pH 4.0 with nitric acid solution serving as the leaching fluid.

Figure S8. XRD patterns of the un-leached and leached (pH 4.0 for 120 min) CdAl₄O₇ samples. The comparison shows that both diffraction peaks were nearly identical and that no new crystalline phase was found after the CPLT process.

Table S1. Quantitative phase analysis results and the calculated transformation ratios (TR) of the authentic CdO : CdAl₄O₇ : glass (amorphous) : CaF₂ determined via the Rietveld refinement method. The refinement was evaluated by the factors derived from the fitness criteria provided in Supporting Information Table S1.

Known		Refinement Results				
Composition (wt. %)	TR (%)	Composition (wt %)	TR (%)	GOF	R _p (%)	R _{wp} (%)
20.00 : 80.00 : 0.00 : 0.00	60.72	21.14 : 79.86 : -- : --	59.34	1.87	4.81	6.30
40.00 : 60.00 : 0.00 : 0.00	36.69	41.34 : 58.36 : -- : --	35.41	1.87	4.94	6.43
60.00 : 20.00 : 0.00 : 0.00	20.48	62.13 : 37.87 : -- : --	19.06	1.84	6.92	6.32
80.00 : 20.00 : 0.00 : 0.00	8.81	82.31 : 17.69 : -- : --	7.67	1.97	6.85	5.35
24.00 : 36.00 : 20.00 : 20.00	36.69	20.35 : 29.34 : 30.32 : 20.00	35.78	1.62	4.35	5.69
16.00 : 24.00 : 40.00 : 20.00	36.69	13.31 : 19.56 : 47.13 : 20.00	36.22	1.65	4.54	5.96
36.00 : 24.00 : 20.00 : 20.00	20.48	31.75 : 19.38 : 28.87 : 20.00	19.08	1.75	4.76	6.25
24.00 : 16.00 : 40.00 : 20.00	20.48	20.41 : 12.93 : 46.66 : 20.00	19.67	1.71	4.71	6.14

"R-pattern", R_p,
$$R_p = \frac{\sum |Y_{o,m} - Y_{c,m}|}{\sum Y_{o,m}}$$

"R-weight pattern", R_{wp},
$$R_{wp} = \sqrt{\frac{\sum w_m (Y_{o,m} - Y_{c,m})^2}{\sum w_m Y_{o,m}^2}}$$

"R-expected", R_{exp},
$$R_{exp} = \sqrt{\frac{\sum M - P}{\sum w_m Y_{o,m}^2}}$$

"Goodness of fit", GOF,
$$GOF = \chi^2 = \frac{R_{wp}}{R_{exp}} = \sqrt{\frac{\sum w_m (Y_{o,m} - Y_{c,m})^2}{M - P}}$$

Table S2. Weight variation of CdAl_4O_7 powder with or without calcination at 1025 °C for 3 h.

Uncalcined CdAl_4O_7 , g	Calcined CdAl_4O_7 , g
1.0005	1.0003
1.0002	1.0001
1.0003	1.0004

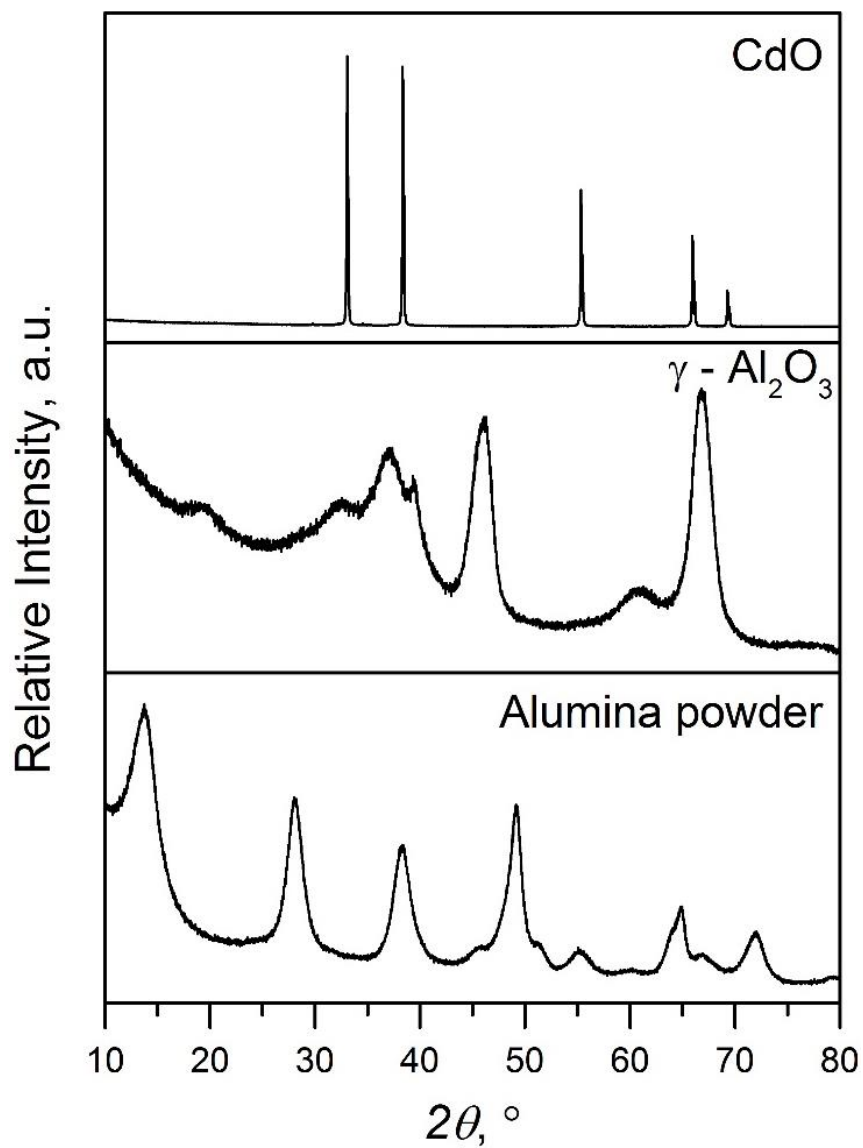


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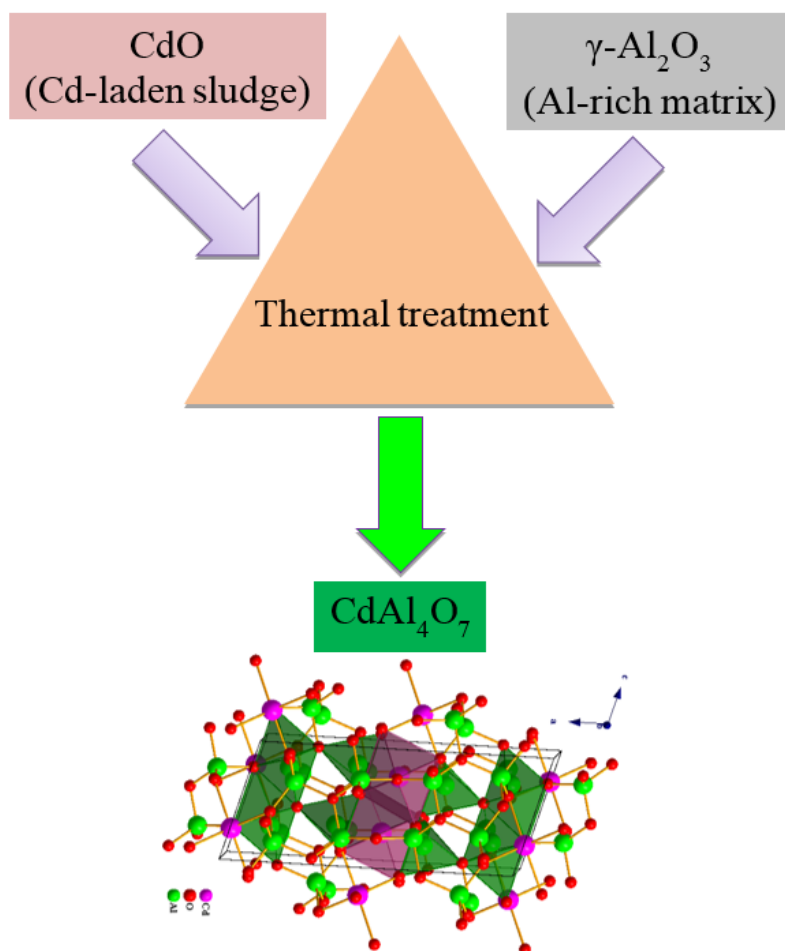


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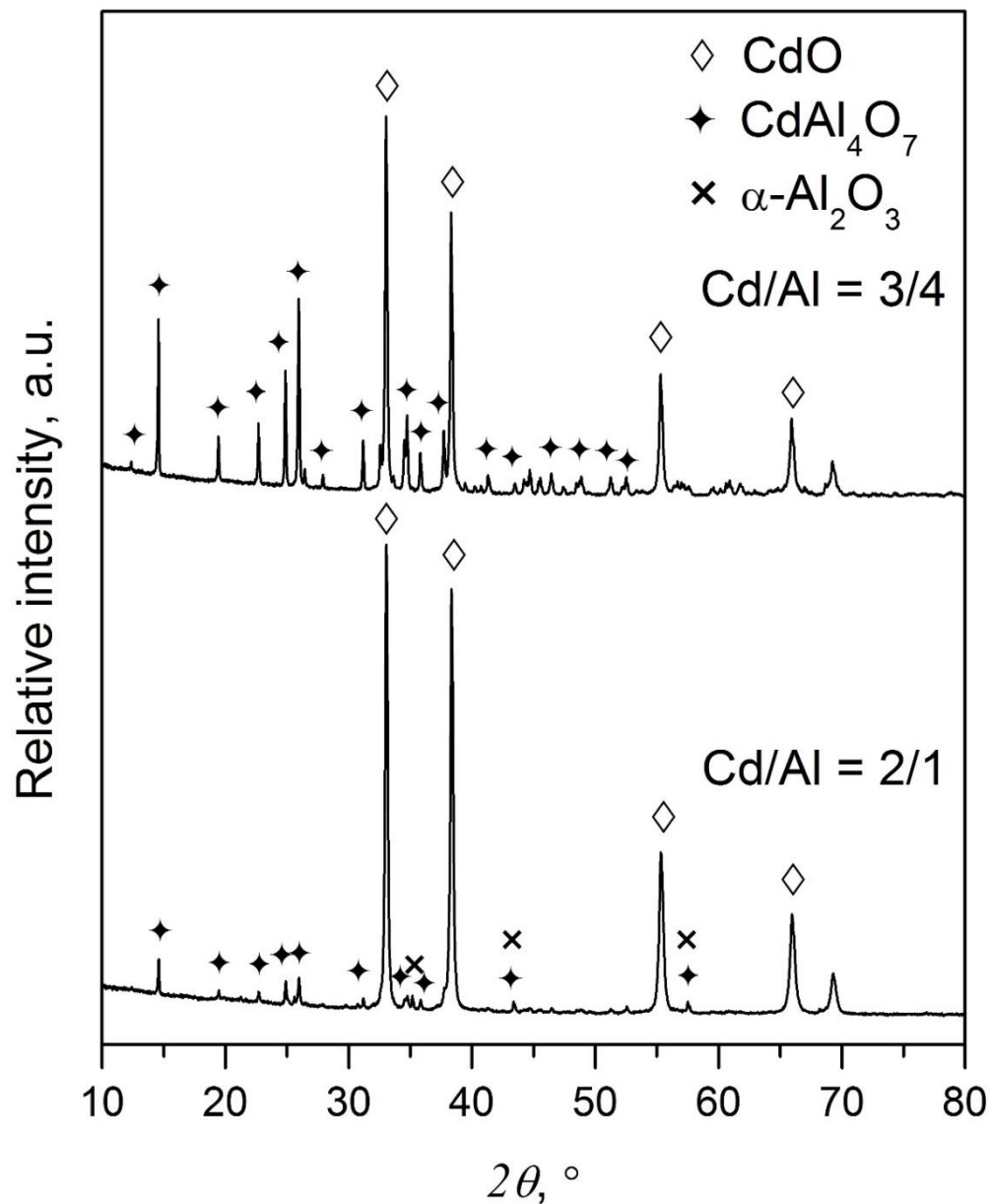


Figure S3. XRD patterns of the products obtained from the reaction of $\text{CdO} + \gamma\text{-Al}_2\text{O}_3$ with Cd/Al molar ratios of 2/1 and 3/4 heated at 1000 °C for 24 h. The standard patterns were derived from the International Centre for Diffraction Data (ICDD) database, including CdO (ICDD PDF no. 75-0594), CdAl_4O_7 (ICDD PDF no. 22-1061) and $\alpha\text{-Al}_2\text{O}_3$ (ICDD PDF no. 74-1081).

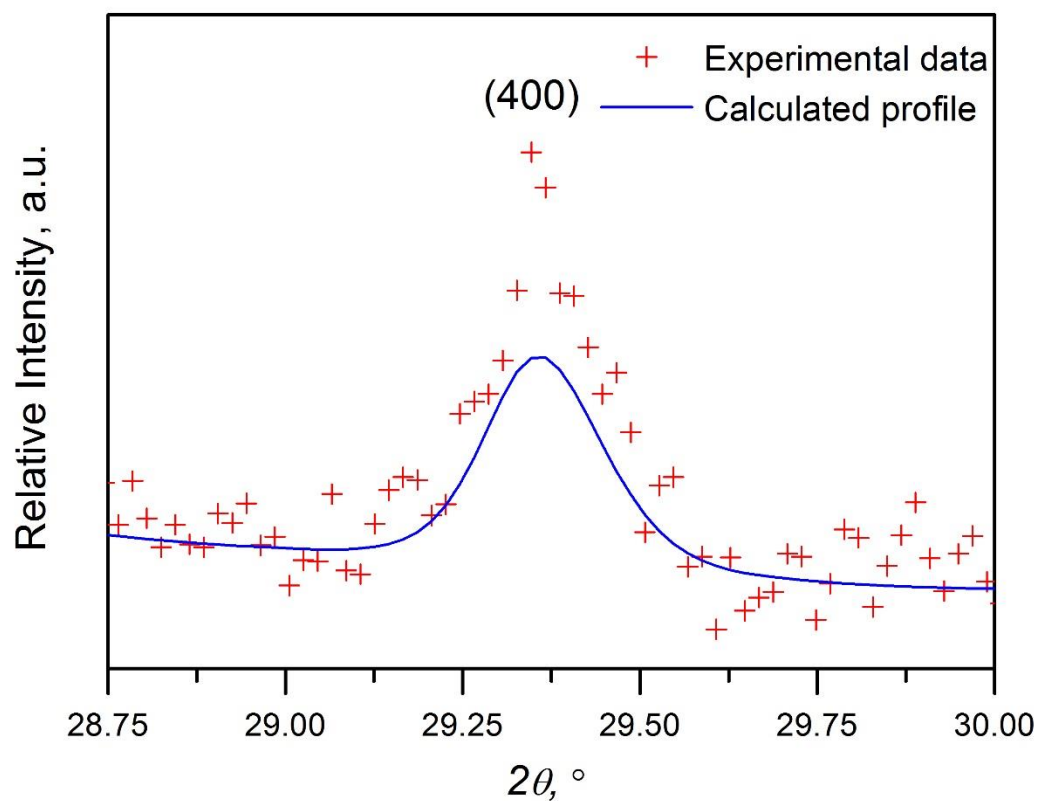


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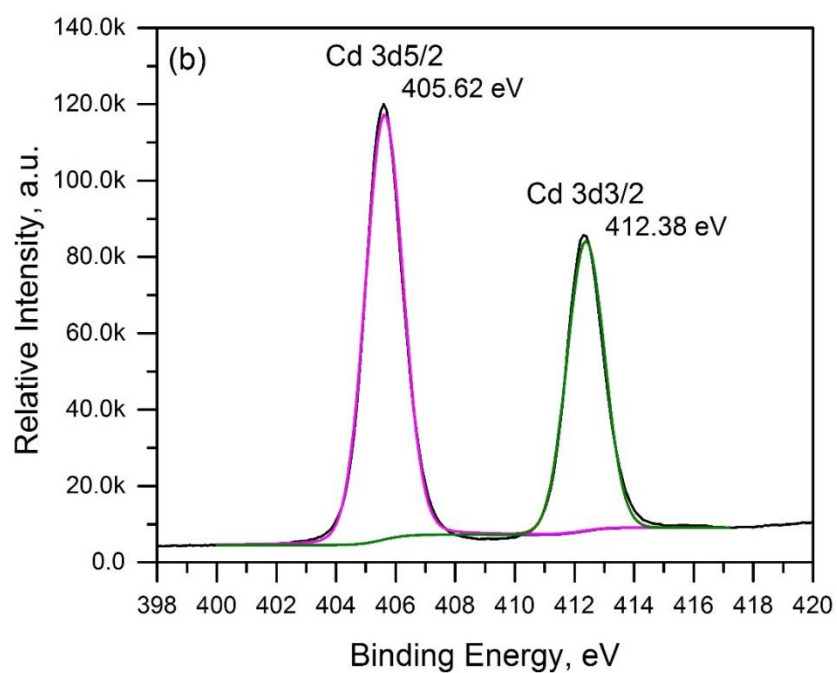
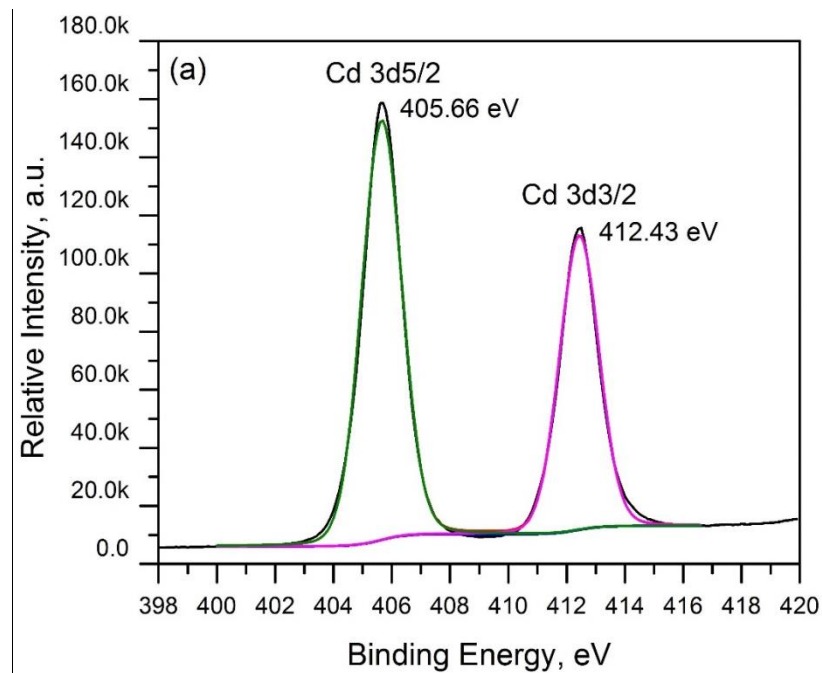


Figure S5. X-ray photoelectron spectroscopy (XPS) spectra for Cd 3d of (a) CdO and (b) CdAl₄O₇.

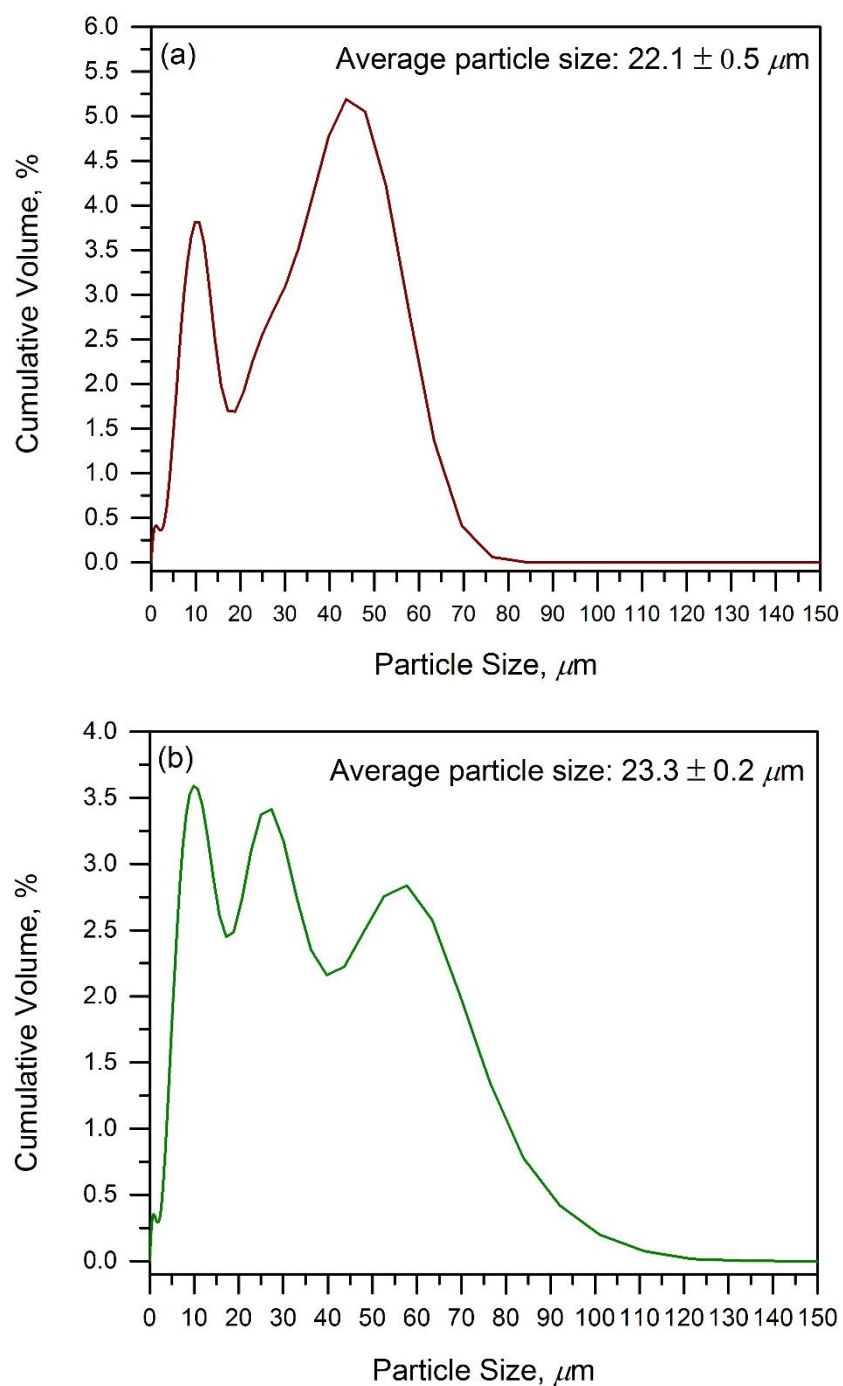


Figure S6. Particle size distributions of (a) CdO and (b) as-prepared CdAl_4O_7 powders for CPLT.

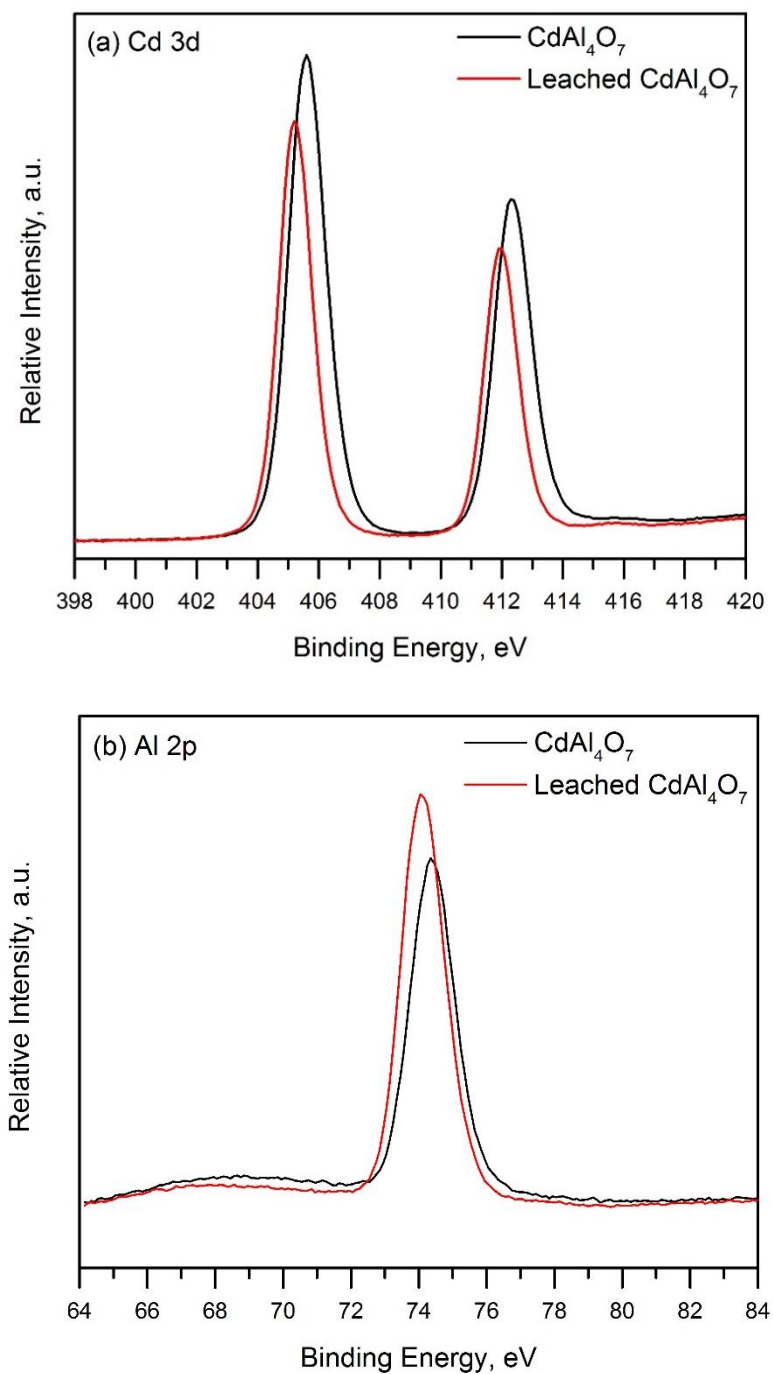


Figure S7. Decrease of the (a) Cd 3d signal and increase of the (b) Al 2p signal on the surface of the CdAl_4O_7 sample leached for 120 min and observed via X-ray photoelectron spectroscopy

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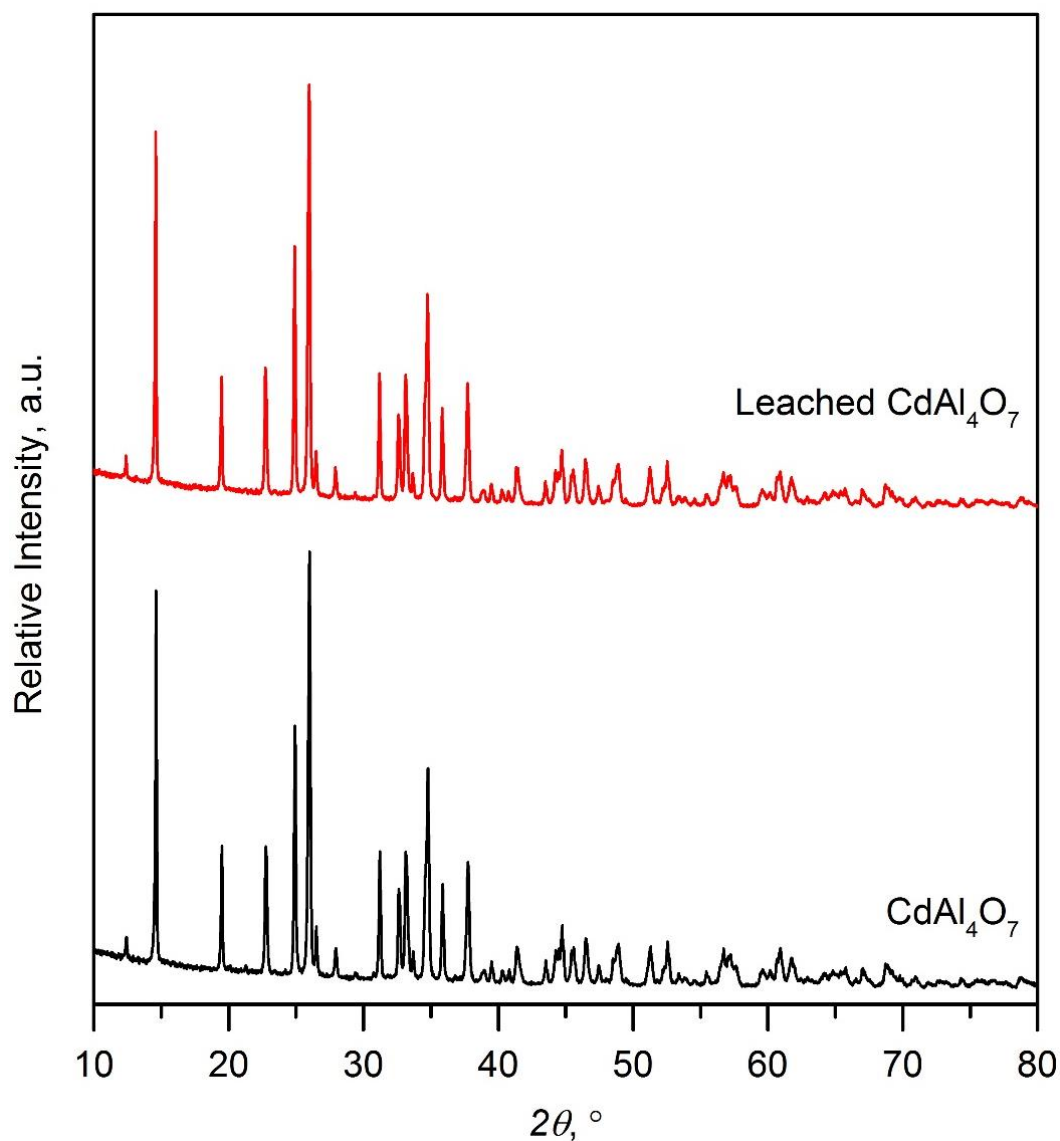


Figure S8. XRD patterns of the un-leached and leached (pH 4.0 for 120 min) CdAl_4O_7 samples. The comparison shows that both diffraction peaks were nearly identical and that no new crystalline phase was found after the CPLT process.