

Optimal Synthesis of Refinery Property-Based Water Networks with Electrocoagulation Treatment Systems

Supporting Information

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Pages: S1-S30

Tables: Table S1-Table S11

S1. MATHEMATICAL MODEL

This sections shows a thorough explanation about proposed mathematical model. A considerable amount of detail was omitted in the manuscript to avoid unnecessary details. The mathematical formulation for this work includes the mass balances and property tracking to model the splitters and mixers considering in the superstructure shown in **Figure 2**. The formulation also includes the relationships to model the treatment units, the cost functions, and the process an environmental constraint. The following is a detailed description of the model equations.

Mass balances

The total flowrate inlet to the treatment system (m^{In}) is obtained from the different process sources (h_i) as follows:

$$m^{In} = \sum_i h_i \quad (1)$$

This stream enters to all treatment units; however, during the electrocoagulation process there is a water lost. Thus, the stream leaving the electrocoagulation is related to the inlet stream through an efficiency factor (β):

$$m^{Out} = m^{In} \beta \quad (2)$$

Additionally, the outlet water from the treatment system (m^{Out}) is equal to the water sent to the process sinks (g_j) plus the water sent to the reverse osmosis unit (g_j^{RO}) as follows:

$$m^{Out} = \sum_j g_j + g_{Waste} + m^{Out-RO} \quad (3)$$

Fresh water (F_r) can be distributed to the different process sinks ($f_{r,j}$), except the waste discharge to the environment:

$$F_r = \sum_j f_{r,j}, \quad \forall r \quad (4)$$

The process sources (W_i) can be segregated and recycled to the process units $w_{i,j}$, discharged to the environment ($w_{i,waste}$), or sent to the treatment system (h_i):

$$W_i = \sum_j w_{i,j} + w_{i,Waste} + h_i, \quad \forall i \quad (5)$$

The required water for the process units (G_j^{Sink}) is equal to the fresh water ($f_{r,j}$), water directly recycled from the process streams ($w_{i,j}$), treated water leaving the electrocoagulation system (g_j) and the treated water processed in the reverse osmosis unit (g_j^{RO}):

$$G_j^{Sink} = \sum_r f_{r,j} + \sum_i w_{i,j} + g_j + g_j^{RO}, \quad \forall j \quad (6)$$

The wastewater discharged to the environment ($Waste$) is equal to the water sent from the process sources ($w_{i,j}$), plus the treated water (g_{Waste}) and the water from the reverse osmosis unit (g_{Waste}^{RO}) as follows:

$$Waste = \sum_i w_{i,Waste} + g_{Waste} + g_{Waste}^{RO} \quad (7)$$

Property tracking

The property operator for the stream entering to the treatment unit ($\psi_p^{InMix}(P)$) is obtained by the weighted sum of the property operators of the different process sources ($\psi_{p,i}^{Source}(P)$):

$$\psi_p^{InMix}(P) m^{In} = \sum_i \psi_{p,i}^{Source}(P) h_i, \quad \forall p \quad (8)$$

It should be noted that the property operators ($\psi_{p,i}$) depend on each property and these can be determined experimentally.⁶

The property operator for the sinks ($\psi_{p,j}^{Sink}(P)$) must be equal to the sum of the property operators for fresh sources ($\psi_{p,r}^{Fresh}(P)$), direct recycled sources ($\psi_{p,i}^{Source}(P)$), treated streams ($\psi_p^{OutSplit}(P)$) and treated water from the reverse osmosis unit ($\psi_p^{OutSplit}(P)$):

$$G_j \psi_{p,j}^{Sink}(P) = \sum_r f_{r,j} \psi_{p,r}^{Fresh}(P) + \sum_i w_{i,j} \psi_{p,i}^{Source}(P) + g_j \psi_p^{OutSplit}(P) + g_j^{RO} \psi_p^{RO}(P), \quad \forall j, j \neq waste, \forall p \quad (9)$$

Similarly for the wastewater stream, notice that fresh water cannot be sent to the waste discharge:

$$G_{\text{waste}} \psi_{p,\text{waste}}^{\text{waste}}(P) = \sum_i w_{i,\text{waste}} \psi_{p,i}^{\text{Source}}(P) + g_{\text{waste}} \psi_p^{\text{OutSplit}}(P) + g_{\text{waste}}^{\text{RO}} \psi_p^{\text{RO}}(P), \quad \forall p \quad (10)$$

There are specific constraints for the process sinks and for the wastewater discharged to the environment. These constraints are given in terms of lower $\psi_{p,j}^{\text{Sink-Min}}(P)$ and upper $\psi_{p,j}^{\text{Sink-Max}}(P)$ limits for the property operators for the streams entering the process sinks as follows:

$$\psi_{p,j}^{\text{Sink-Min}}(P) \leq \psi_{p,j}^{\text{Sink}}(P) \leq \psi_{p,j}^{\text{Sink-Max}}(P), \quad \forall P, \forall j \quad (11)$$

$$\psi_{p,\text{waste}}^{\text{Sink-Min}}(P) \leq \psi_{p,\text{waste}}^{\text{Sink}}(P) \leq \psi_{p,\text{waste}}^{\text{Sink-Max}}(P), \quad \forall P \quad (12)$$

Treatment units

First, there is needed to adjust the properties for the inlet stream to desired values, which have been identified through experimental studies.

pH adjustment process

For modelling the treatment system, first it is necessary to consider a pH adjustment unit to set the pH of the mixed stream to a desired value. Therefore, if the pH is greater than the desired one, it is required to add an acid and if the mixed stream has a pH lower than the desired one, it is necessary to add a base. This is modeled through the following disjunction:

$$\left[\begin{array}{c} Y^{pH-a} \\ 10^{-pH^{\text{In-Mix}}} \leq 10^{-pH^{\text{desired}}} \\ \text{Acid} = \frac{(10^{-pH^{\text{desired}}} - 10^{-pH^{\text{In-Mix}}}) m^{\text{In}}}{\text{Conc}^{\text{Acid}} \rho^{\text{In-Mix}}} \\ \text{Cost}^{pH} = \text{UC}^{\text{Acid}} \text{Acid} \end{array} \right] \vee \left[\begin{array}{c} Y^{pH-b} \\ 10^{-pH^{\text{In-Mix}}} \geq 10^{-pH^{\text{desired}}} \\ \text{Base} = \frac{(10^{-pH^{\text{In-Mix}}} - 10^{-pH^{\text{desired}}}) m^{\text{In}}}{\text{Conc}^{\text{Base}} \rho^{\text{In-Mix}}} \\ \text{Cost}^{pH} = \text{UC}^{\text{Base}} \text{Base} \end{array} \right] \vee \left[\begin{array}{c} Y^{pH-c} \\ 10^{-pH^{\text{In-Mix}}} = 10^{-pH^{\text{desired}}} \\ \text{Acid} = 0 \\ \text{Base} = 0 \\ \text{Cost}^{pH} = 0 \end{array} \right]$$

In the previous relationship, the optimization variable is $10^{-pH^{\text{In-Mix}}}$, which corresponds to the property operator for the pH. The previous disjunction is reformulated as follows:

First, only one option can be selected, for modelling this, the binary variables y^{pH-b} and y^{pH-a} are introduced for the different segments as follows:

$$y^{pH-a} + y^{pH-b} + y^{pH-c} = 1 \quad (13)$$

Then, if the pH is greater than the desired value the binary variable y^{pH-a} is activated:

$$10^{-pH^{In_Mix}} \leq 10^{-pH_{desired}} + M^{pH} (1 - y^{pH-a}) \quad (14)$$

If the pH is lower than the desired value, the binary variable y^{pH-b} is activated:

$$10^{-pH^{In_Mix}} \geq 10^{-pH_{desired}} - M^{pH} (1 - y^{pH-b}) \quad (15)$$

When the binary variable y^{pH-a} is activated, then the acid required and the cost for the pH treatment are calculated through the following relationships:

$$Acid \leq \frac{(10^{-pH_{desired}} - 10^{-pH^{In_Mix}}) m^{In}}{Conc^{Acid} \rho^{In_Mix}} + M^{Acid} (1 - y^{pH-a}) \quad (16)$$

$$Acid \geq \frac{(10^{-pH_{desired}} - 10^{-pH^{In_Mix}}) m^{In}}{Conc^{Acid} \rho^{In_Mix}} - M^{Acid} (1 - y^{pH-a}) \quad (17)$$

$$Cost^{pH} \geq UC^{Acid} Acid - M^{Cost^{pH}} (1 - y^{pH-a}) \quad (18)$$

On the other hand, if the binary variable y^{pH-a} is zero, the previous relationships are relaxed through the big-M parameter. Similarly, when the binary variable y^{pH-b} is activated, the corresponding base required is calculated as follows:

$$Base \leq \frac{(10^{-pH^{In_Mix}} - 10^{-pH_{desired}}) m^{In}}{Conc^{Base} \rho^{In_Mix}} + M^{Base} (1 - y^{pH-b}) \quad (19)$$

$$Base \geq \frac{(10^{-pH^{In_Mix}} - 10^{-pH_{desired}}) m^{In}}{Conc^{Base} \rho^{In_Mix}} - M^{Base} (1 - y^{pH-b}) \quad (20)$$

Then, the cost for the pH is calculated depending on the amount of base used:

$$Cost^{pH} \geq UC^{Base} Base - M^{Cost^{pH}} (1 - y^{pH-b}) \quad (21)$$

When the pH is equal to the desired value for the electrocoagulation process, the binary variable y^{pH-c} is activated and it is not require to add base or acid to the inlet stream and the cost for pH treatment will be zero.

$$10^{-pH^{In_Mix}} \leq 10^{-pH_{desired}} + M^{pH} (1 - y^{pH-c}) \quad (22)$$

$$10^{-pH_{In_Mix}} \geq 10^{-pH_{desired}} - M^{pH} (1 - y^{pH-c}) \quad (23)$$

$$Base \leq M^{Base} (1 - y^{pH-c}) \quad (24)$$

$$Base \geq -M^{Base} (1 - y^{pH-c}) \quad (25)$$

$$Acid \leq M^{Acid} (1 - y^{pH-c}) \quad (26)$$

$$Acid \geq -M^{Acid} (1 - y^{pH-c}) \quad (27)$$

$$Cost^{pH} \leq M^{Cost^{pH}} (1 - y^{pH-c}) \quad (28)$$

$$Cost^{pH} \geq -M^{Cost^{pH}} (1 - y^{pH-c}) \quad (29)$$

Otherwise, these equations are also relaxed through the big-M parameter.

Sodium chloride concentration adjustment.

The concentration of NaCl is an important property for the electrocoagulation treatment unit.⁴⁶ Thus, if the stream entering to the treatment unit has a concentration lower than the desired value, it is necessary to adjust it through the addition of NaCl. On the other hand, when the NaCl concentration of the inlet to the treatment stream is greater than the desired value, it is necessary to dilute this stream by adding fresh water to adjust this concentration. This is modeled through the following disjunction:

$$\left[\begin{array}{c} Y^{NaCl-a} \\ C_{NaCl}^{In_Mix} \leq C_{NaCl}^{desired} \\ m^{NaCl} = \left(C_{NaCl}^{desired} - C_{NaCl}^{In_Mix} \right) \frac{m^{In}}{\rho^{In_Mix}} \\ Cost^{NaCl} = UC^{NaCl} m^{NaCl} \end{array} \right] \vee \left[\begin{array}{c} Y^{NaCl-b} \\ C_{NaCl}^{In_Mix} \geq C_{NaCl}^{desired} \\ m^{Fresh} = m^{In} \rho^{In_Mix} \left(\frac{C_{NaCl}^{In_Mix}}{C_{NaCl}^{desired}} - 1 \right) \\ Cost^{H_2O} = UC^{H_2O} m^{Fresh} \end{array} \right] \vee \left[\begin{array}{c} Y^{NaCl-c} \\ C_{NaCl}^{In_Mix} = C_{NaCl}^{desired} \\ m^{Fresh} = 0 \\ m^{NaCl} = 0 \\ Cost^{H_2O} = 0 \\ Cost^{NaCl} \end{array} \right]$$

This disjunction is also reformulated in terms of algebraic equations, which include three Boolean variables associated with binary variables. Then, when the stream at the inlet of this unit has a NaCl concentration lower than the target value, the binary variable y^{NaCl-a} is set as one; otherwise y^{NaCl-b} is equal to one for the case when the concentration is greater than the mentioned value and

y^{NaCl-c} results one when the concentration is equal to the desired value. Then, only one of these options must be selected and this is modeled as follows:

$$y^{NaCl-a} + y^{NaCl-b} + y^{NaCl-c} = 1 \quad (30)$$

To activate these binary variables, the following relationships are used:

$$C^{NaCl} \leq C_{NaCl}^{desired} + M^{C^{NaCl}} (1 - y^{NaCl-a}) \quad (31)$$

$$C^{NaCl} \geq C_{NaCl}^{desired} - M^{C^{NaCl}} (1 - y^{NaCl-b}) \quad (32)$$

$$C^{NaCl} \leq C_{NaCl}^{desired} + M^{C^{NaCl}} (1 - y^{NaCl-c}) \quad (33)$$

$$C^{NaCl} \geq C_{NaCl}^{desired} - M^{C^{NaCl}} (1 - y^{NaCl-c}) \quad (34)$$

When the concentration of NaCl is lower than the desired value, the binary variable y^{NaCl-a} is equal to one and the corresponding NaCl that is required is calculated as follows:

$$m^{NaCl} \leq \left(C_{NaCl}^{desired} - C_{NaCl}^{In-Mix} \right) \frac{m^{In}}{\rho^{In-Mix}} + M^{NaCl} (1 - y^{NaCl-a}) \quad (35)$$

$$m^{NaCl} \geq \left(C_{NaCl}^{desired} - C_{NaCl}^{In-Mix} \right) \frac{m^{In}}{\rho^{In-Mix}} - M^{NaCl} (1 - y^{NaCl-a}) \quad (36)$$

If the binary variable y^{NaCl-a} is equal to zero, no additional NaCl is required.

Thus, the cost associated with the previous operation is determined as follows:

$$Cost^{NaCl} \leq UC^{NaCl} m^{NaCl} + M^{Cost^{NaCl}} (1 - y^{NaCl-a}) \quad (37)$$

$$Cost^{NaCl} \geq UC^{NaCl} m^{NaCl} - M^{Cost^{NaCl}} (1 - y^{NaCl-a}) \quad (38)$$

Then, for the case when the NaCl concentration is greater than the desired value, the binary variable y^{NaCl-b} is set as one and the corresponding water required to adjust the NaCl concentration is computed as follows:

$$m^{Fresh} \leq m^{In} \rho^{In-Mix} \left(\frac{C_{NaCl}^{In-Mix}}{C_{NaCl}^{desired}} - 1 \right) + M^{Fresh} (1 - y^{NaCl-b}) \quad (39)$$

$$m^{Fresh} \geq m^{In} \rho^{In_Mix} \left(\frac{C_{NaCl}^{In_Mix}}{C_{NaCl}^{desired}} - 1 \right) - M^{Fresh} (1 - y^{NaCl_b}) \quad (40)$$

If the binary variable is zero, no dilution is required.

Through the following relationships, the cost for the case when it is necessary to add fresh water is evaluated:

$$Cost^{H_2O} \leq UC^{H_2O} m^{Fresh} + M^{Cost^{H_2O}} (1 - y^{NaCl_b}) \quad (41)$$

$$Cost^{H_2O} \geq UC^{H_2O} m^{Fresh} - M^{Cost^{H_2O}} (1 - y^{NaCl_b}) \quad (42)$$

Finally, for the last case when the concentration of NaCl is equal to the desired value, the binary variable y^{NaCl_c} is one. Hence, according to the disjunction, no action is required and the cost is equal to zero for this treatment, which is represented as follows:

$$m^{NaCl} \leq M^{NaCl} (1 - y^{NaCl_c}) \quad (43)$$

$$m^{NaCl} \geq -M^{NaCl} (1 - y^{NaCl_c}) \quad (44)$$

$$m^{Fresh} \leq M^{Fresh} (1 - y^{NaCl_c}) \quad (45)$$

$$m^{Fresh} \geq -M^{Fresh} (1 - y^{NaCl_c}) \quad (46)$$

$$Cost^{NaCl} \leq UC^{NaCl} m^{NaCl} + M^{Cost^{NaCl}} (1 - y^{NaCl_c}) \quad (47)$$

$$Cost^{NaCl} \geq UC^{NaCl} m^{NaCl} - M^{Cost^{NaCl}} (1 - y^{NaCl_c}) \quad (48)$$

$$Cost^{H_2O} \leq UC^{H_2O} m^{Fresh} + M^{Cost^{H_2O}} (1 - y^{NaCl_c}) \quad (49)$$

$$Cost^{H_2O} \geq UC^{H_2O} m^{Fresh} - M^{Cost^{H_2O}} (1 - y^{NaCl_c}) \quad (50)$$

Temperature adjustment

The inlet temperature for the electrocoagulation system has to be adjusted to the desired value. First, if the inlet temperature is greater than this value, then a cooling utility is used to quench it. On the

other hand, if the inlet temperature is lower than the desired value, steam is employed to raise its temperature. This is modeled through the following disjunction:

$$\left[\begin{array}{c} Y^{EXC_a} \\ T^{In_Mix} \geq T^{desired} \\ Q^{cw} = m^{In} C_p^{In_Mix} (T^{In_Mix} - T^{desired}) \\ Cost^{cw} = UC^{cw} Q^{cw} \\ A^{EXC} = \frac{Q^{cw}}{ULMTD} \\ LMTD = \frac{(T^{In_Mix} - T^{Out_cw}) - (T^{desired} - T^{In_cw})}{\ln\left(\frac{T^{In_Mix} - T^{Out_cw}}{T^{desired} - T^{In_cw}}\right)} \\ Cost^{Cool} = A^{Cool} + B^{Cool} (A^{EXC})^{C^{Cool}} \end{array} \right] \vee \left[\begin{array}{c} Y^{EXC_b} \\ T^{In_Mix} \leq T^{desired} \\ Q^{steam} = m^{In} C_p^{In_Mix} (T^{desired} - T^{In_Mix}) \\ Cost^{steam} = UC^{steam} Q^{steam} \\ A^{EXC2} = \frac{Q^{steam}}{ULMTD} \\ LMTD = \frac{(T^{In_steam} - T^{desired}) - (T^{Out_steam} - T^{In_mix})}{\ln\left(\frac{T^{In_steam} - T^{desired}}{T^{Out_steam} - T^{In_mix}}\right)} \\ Cost^{Heat} = A^{Heat} + B^{Heat} (A^{EXC2})^{C^{Heat}} \end{array} \right] \vee \left[\begin{array}{c} Y^{EXC_c} \\ T^{In_Mix} = T^{desired} \\ Q^{steam} = 0 \\ Q^{cw} = 0 \\ Cost^{steam} = 0 \\ Cost^{cw} = 0 \\ A^{EXC} = 0 \\ A^{EXC2} = 0 \\ LMTD = 0 \\ Cost^{Heat} = 0 \\ Cost^{Cool} = 0 \end{array} \right]$$

Similar to the previous disjunctions, each Boolean variable is related to a binary variable. The binary variable y^{EXC_a} is employed to model the case when the inlet temperature is greater than $T^{desired}$, y^{EXC_b} is used to model the case when the inlet temperature is lower than the desired value, and y^{EXC_c} is used to model the case when the inlet temperature is equal to the optimal value. Then, the following relationship states that only one option can be selected:

$$y^{EXC_a} + y^{EXC_b} + y^{EXC_c} = 1 \quad (51)$$

To activate the binary variable y^{EXC_a} , the following relationship is used:

$$T^{In_Mix} \geq T^{desired} - M^{T^{In}} (1 - y^{EXC_a}) \quad (52)$$

To activate the binary variable y^{EXC_b} , the following relationship is employed:

$$T^{In_Mix} \leq T^{desired} + M^{T^{In}} (1 - y^{EXC_b}) \quad (53)$$

Whereas to activate the binary variable y^{EXC_c} , the following relationships are used:

$$T^{In_Mix} \leq T^{desired} + M^{T^{In}} (1 - y^{EXC_c}) \quad (54)$$

$$T^{In_Mix} \geq T^{desired} - M^{T^{In}} (1 - y^{EXC_c}) \quad (55)$$

For the case when the binary variable y^{EXC_a} is set as one, the cooling load (Q^{cw}) is calculated as follows:

$$Q^{cw} \leq m^{In} Cp^{In_Mix} (T^{In_Mix} - T^{desired}) + M^{Q^{cw}} (1 - y^{EXC_a}) \quad (56)$$

$$Q^{cw} \geq m^{In} Cp^{In_Mix} (T^{In_Mix} - T^{desired}) - M^{Q^{cw}} (1 - y^{EXC_a}) \quad (57)$$

In order to determine the capital and operating costs associated with the case when cooling is required, the following relationships are required. The operating cost of cooling is calculated as follows:

$$Cost^{cw} \leq UC^{cw} Q^{cw} + M^{cost^{cw}} (1 - y^{EXC_a}) \quad (58)$$

$$Cost^{cw} \geq UC^{cw} Q^{cw} - M^{cost^{cw}} (1 - y^{EXC_a}) \quad (59)$$

For the capital cost, it is necessary to determine the size of the heat exchanger unit (A^{EXC}). This is done through the following relationships:

$$A^{EXC} \leq \frac{Q^{cw}}{U LMTD} + M^A (1 - y^{EXC_a}) \quad (60)$$

$$A^{EXC} \geq \frac{Q^{cw}}{U LMTD} - M^A (1 - y^{EXC_a}) \quad (61)$$

As can be seen from the previous relationships, it is also necessary to determine the log-mean temperature difference ($LMTD$). The following relationship is employed for this purpose:

$$LMTD \leq \frac{(T^{In_Mix} - T^{Out_cw}) - (T^{desired} - T^{In_cw})}{\ln \left(\frac{T^{In_Mix} - T^{Out_cw}}{T^{desired} - T^{In_cw}} \right)} + M^{LMTD} (1 - y^{EXC_a}) \quad (62)$$

$$LMTD \geq \frac{(T^{In_Mix} - T^{Out_cw}) - (T^{desired} - T^{In_cw})}{\ln \left(\frac{T^{In_Mix} - T^{Out_cw}}{T^{desired} - T^{In_cw}} \right)} - M^{LMTD} (1 - y^{EXC_a}) \quad (63)$$

Finally, the capital cost for the cooler is calculated as follows:

$$Cost^{Cool} \leq A^{Cool} + B^{Cool} (A^{EXC})^{C^{Cool}} + M^{Cost^{Cool}} (1 - y^{EXC-a}) \quad (64)$$

$$Cost^{Cool} \geq A^{Cool} + B^{Cool} (A^{EXC})^{C^{Cool}} - M^{Cost^{Cool}} (1 - y^{EXC-a}) \quad (65)$$

If the binary variable y^{EXC-b} is equal to one, it means that the inlet temperature is lower than the optimal value. Consequently, a heater is required to achieve the desired temperature. Then, the heating load, operating cost, area, log-mean temperature difference and the capital cost are calculated as follows:

$$Q^{steam} \leq m^{In} Cp^{In-Mix} (T^{desired} - T^{In-Mix}) + M^{Q^{steam}} (1 - y^{EXC-b}) \quad (66)$$

$$Q^{steam} \geq m^{In} Cp^{In-Mix} (T^{desired} - T^{In-Mix}) - M^{Q^{steam}} (1 - y^{EXC-b}) \quad (67)$$

$$Cost^{steam} \leq UC^{steam} Q^{steam} + M^{Cost^{steam}} (1 - y^{EXC-b}) \quad (68)$$

$$Cost^{steam} \geq UC^{steam} Q^{steam} - M^{Cost^{steam}} (1 - y^{EXC-b}) \quad (69)$$

$$A^{EXC2} \leq \frac{Q^{steam}}{U2 LMTD} + M^{A2} (1 - y^{EXC-b}) \quad (70)$$

$$A^{EXC2} \geq \frac{Q^{steam}}{U2 LMTD} - M^{A2} (1 - y^{EXC-b}) \quad (71)$$

$$LMTD \leq \frac{(T^{In-steam} - T^{desired}) - (T^{Out-steam} - T^{In-mix})}{\ln\left(\frac{T^{In-steam} - T^{desired}}{T^{Out-steam} - T^{In-mix}}\right)} + M^{LMTD} (1 - y^{EXC-b}) \quad (72)$$

$$LMTD \geq \frac{(T^{In-steam} - T^{desired}) - (T^{Out-steam} - T^{In-mix})}{\ln\left(\frac{T^{In-steam} - T^{desired}}{T^{Out-steam} - T^{In-mix}}\right)} - M^{LMTD} (1 - y^{EXC-b}) \quad (73)$$

The capital cost is calculated using the following relationships:

$$Cost^{Heat} \leq A^{Heat} + B^{Heat} (A^{EXC2})^{C^{Heat}} + M^{Cost^{Heat}} (1 - y^{EXC-b}) \quad (74)$$

$$Cost^{Heat} \geq A^{Heat} + B^{Heat} (A^{EXC2})^{C^{Heat}} - M^{Cost^{Heat}} (1 - y^{EXC-b}) \quad (75)$$

When y^{EXC-c} is equal to one, it means that the inlet temperature to treatment is equal to the optimal temperature. Therefore, it is not necessary to modify the temperature and the costs are set to zero:

$$Q^{cw} \leq M^{Q^{cw}} (1 - y^{EXC-c}) \quad (76)$$

$$Q^{cw} \geq -M^{Q^{cw}} (1 - y^{EXC-c}) \quad (77)$$

$$Q^{steam} \leq M^{Q^{steam}} (1 - y^{EXC-c}) \quad (78)$$

$$Q^{steam} \geq -M^{Q^{steam}} (1 - y^{EXC-c}) \quad (79)$$

$$Cost^{cw} \leq M^{cost^{cw}} (1 - y^{EXC-c}) \quad (80)$$

$$Cost^{cw} \geq -M^{cost^{cw}} (1 - y^{EXC-c}) \quad (81)$$

$$Cost^{steam} \leq M^{cost^{steam}} (1 - y^{EXC-c}) \quad (82)$$

$$Cost^{steam} \geq -M^{cost^{steam}} (1 - y^{EXC-c}) \quad (83)$$

$$A^{EXC} \leq M^A (1 - y^{EXC-c}) \quad (84)$$

$$A^{EXC} \geq -M^A (1 - y^{EXC-c}) \quad (85)$$

$$A^{EXC2} \leq M^{A2} (1 - y^{EXC-c}) \quad (86)$$

$$A^{EXC2} \geq -M^{A2} (1 - y^{EXC-c}) \quad (87)$$

$$LMTD \leq M^{LMTD} (1 - y^{EXC-c}) \quad (88)$$

$$LMTD \geq -M^{LMTD} (1 - y^{EXC-c}) \quad (89)$$

$$Cost^{Cool} \leq M^{Cost^{cool}} (1 - y^{EXC-c}) \quad (90)$$

$$Cost^{Cool} \geq -M^{Cost^{cool}} (1 - y^{EXC-c}) \quad (91)$$

$$Cost^{Heat} \leq M^{Cost^{Heat}} (1 - y^{EXC-c}) \quad (92)$$

$$Cost^{Heat} \geq -M^{Cost^{Heat}} (1 - y^{EXC-c}) \quad (93)$$

Reverse osmosis

According to the proposed design for the recycle and reuse network, a reverse osmosis unit is placed at the exit of the electrocoagulation process with the purpose of reducing the NaCl concentration to meet the constraints for the process sinks. In this regard, the existence for this unit is modeled through the following disjunction.

$$\left[\begin{array}{c} Y^{RO-a} \\ m^{Out-RO} \geq m^{OutRO-Min} \\ C_{NaCl}^{RO} = 0 \\ \sum_j g_j^{RO} + g_{Waste}^{RO} = \alpha m^{Out-RO} \\ Cost^{OP-RO} = UC^{RO}(m^{Out-RO}) \\ Cost^{RO} = A^{RO} + B^{RO}(m^{Out-RO})^{C^{RO}} \end{array} \right] \vee \left[\begin{array}{c} Y^{RO-b} \\ m^{Out-RO} = 0 \\ C_{NaCl}^{RO} = C_{NaCl}^{desired} \\ \sum_j g_j^{RO} + g_{Waste}^{RO} = 0 \\ Cost^{OP-RO} = 0 \\ Cost^{RO} = 0 \end{array} \right]$$

In the previous disjunction, Y^{RO-a} represents the Boolean variable associated with the existence of the RO unit; while Y^{RO-b} is the Boolean variable that is set as true when the unit is not required. Then, associated with each Boolean variable there is a binary variable which helps to obtain the following algebraic relationships. Firstly, only one of these two options can be selected:

$$y^{RO-a} + y^{RO-b} = 1 \quad (94)$$

The next equations are useful to state that if the RO unit is installed, then the outlet stream is free of NaCl; otherwise, if the reverse osmosis is not needed, then the concentration for the stream is fixed in $C_{NaCl}^{desired}$.

$$C_{NaCl}^{RO} \leq M_{NaCl}^{RO} (1 - y^{RO-a}) \quad (95)$$

$$C_{NaCl}^{RO} \geq -M_{NaCl}^{RO} (1 - y^{RO-a}) \quad (96)$$

$$C_{NaCl}^{RO} \leq C_{NaCl}^{desired} + M_{NaCl}^{RO} (1 - y^{RO-b}) \quad (97)$$

$$C_{NaCl}^{RO} \geq C_{NaCl}^{desired} - M_{NaCl}^{RO} (1 - y^{RO-b}) \quad (98)$$

Furthermore, if the RO unit is required, then the inlet flowrate (m^{Out-RO}) must be higher than a minimum value to ensure an adequate operation ($m^{OutRO-Min}$); otherwise, if the RO unit is not installed, the inlet flowrate must not exist. This is modeled through the next expressions:

$$m^{Out-RO} \geq m^{OutRO-Min} - M^{OutRO-Max} (1 - y^{RO-a}) \quad (99)$$

$$m^{Out-RO} \leq M^{OutRO-Max} (1 - y^{RO-b}) \quad (100)$$

$$m^{Out-RO} \geq -M^{OutRO-Max} (1 - y^{RO-b}) \quad (101)$$

According to the proposed disjunction, the next aspect to be modeled is the flowrate leaving the RO unit. In this sense, the reverse osmosis process involves a loss of water, which is considered through the factor α . Thus, the remaining water is split and sent to the process sinks (g_j^{RO}) and to the waste (g_{Waste}^{RO}):

$$\sum_j g_j^{RO} + g_{Waste}^{RO} \leq \alpha m^{Out-RO} + M^{OutRO-Max} (1 - y^{RO-a}) \quad (102)$$

$$\sum_j g_j^{RO} + g_{Waste}^{RO} \geq \alpha m^{Out-RO} - M^{OutRO-Max} (1 - y^{RO-a}) \quad (103)$$

On the other hand, if the RO unit is not required, then these streams do not exist, which is modeled as follows:

$$\sum_j g_j^{RO} + g_{Waste}^{RO} \leq M^{OutRO-Max} (1 - y^{RO-b}) \quad (104)$$

$$\sum_j g_j^{RO} + g_{Waste}^{RO} \geq -M^{OutRO-Max} (1 - y^{RO-b}) \quad (105)$$

The next relationships determine the operating cost for the RO unit:

$$Cost^{OP-RO} \leq UC^{RO} (m^{Out-RO}) + M^{OPROCost} (1 - y^{RO-a}) \quad (106)$$

$$Cost^{OP-RO} \geq UC^{RO} (m^{Out-RO}) - M^{OPROCost} (1 - y^{RO-a}) \quad (107)$$

$$Cost^{OP-RO} \leq M^{OPROCOST} (1 - y^{RO-b}) \quad (108)$$

$$Cost^{OP-RO} \geq -M^{OPROCOST} (1 - y^{RO-b}) \quad (109)$$

The capital cost is calculated considering fixed and variable charges as follows:

$$Cost^{RO} \leq A^{RO} + B^{RO} (m^{Out-RO})^{C^{RO}} + M^{RO} (1 - y^{RO-a}) \quad (110)$$

$$Cost^{RO} \geq A^{RO} + B^{RO} (m^{Out-RO})^{C^{RO}} - M^{RO} (1 - y^{RO-a}) \quad (111)$$

$$Cost^{RO} \leq M^{RO} (1 - y^{RO-b}) \quad (112)$$

$$Cost^{RO} \geq -M^{RO} (1 - y^{RO-b}) \quad (113)$$

It should be noted that the operating and capital costs depend of the treated flowrate.

Electrocoagulation

For the purpose of properly modelling the electrocoagulation system, the experimental data by Abdelwahab et al.¹ were used. First, the optimal desired for the wastewater treatment system were matched with the experimentally found results of Abdelwahab et al.¹ (in this case the best conditions were the following: NaCl concentration: 2 kg/m³, pH: 7, time: 2h, c.d: 19.3 mA/cm², and temperature: 293 K). The energy required and aluminum consumption are the key factors used to determine the electrocoagulation cost in optimization. By analyzing the experimental data of Abdelwahab et al.¹, these two aspects can be correlated with the initial phenol concentration (**Figure 3**). Thus, the energy and aluminum consumption are determined directly through the initial phenol concentration.

As can be seen from **Figure 3**, the initial phenol concentration is a key factor for modeling the electrocoagulation unit and in the optimization formulation this is modeled through the following disjunction:

$$\begin{bmatrix} Y^{C^{In_Ph_a}} \\ C^{In_Ph} \leq C^{In_Ph_Min} \\ m^{In} = 0 \\ E.C. = 0 \\ Al.C. = 0 \\ Cost^{OP_Treatm} = 0 \end{bmatrix} \vee \begin{bmatrix} Y^{C^{In_Ph_b}} \\ C^{In_Ph} \geq C^{In_Ph_Max} \\ m^{In} = 0 \\ E.C. = 0 \\ Al.C. = 0 \\ Cost^{OP_Treatm} = 0 \end{bmatrix} \vee \begin{bmatrix} Y^{C^{In_Ph_c}} \\ C^{In_Ph} \geq C^{In_Ph_Min} \\ C^{In_Ph} \leq C^{In_Ph_Max} \\ m^{In} \geq 0 \\ E.C. = f_1(C^{In_Ph}) \\ Al.C. = f_2(C^{In_Ph}) \\ Cost^{OP_Treatm} = UC^{KWh} E.C. + UC^{Al.C.} Al.C. \end{bmatrix}$$

In the previous disjunction, three options are considered. The first one (associated with the Boolean variable $Y^{C^{In_Ph_a}}$) represents the case when the initial phenol concentration is lower than the minimum allowed ($C^{In_Ph_Min}$), the second option (associated with the Boolean variable $Y^{C^{In_Ph_b}}$) represents the case when the initial phenol concentration is greater than the maximum allowed ($C^{In_Ph_Max}$). In these two options, the electrocoagulation unit is not used and the associated operating cost must be zero. In the third scenario (associated with the Boolean variable $Y^{C^{In_Ph_c}}$), the electrocoagulation unit is required, and the corresponding energy ($E.C.$) and aluminum ($Al.C.$) consumed are calculated to determine the associated operating cost ($Cost^{OP_Treatm}$). Again, the disjunction is reformulated as a set of algebraic constraints as follows. First, only one option must be selected:

$$y^{C^{In_Ph_a}} + y^{C^{In_Ph_b}} + y^{C^{In_Ph_c}} = 1 \quad (114)$$

The binary variables are activated through the following relationships:

$$C^{In_Ph} \leq C^{In_Ph_Min} + M^{Ph} (1 - y^{C^{In_Ph_a}}) \quad (115)$$

$$C^{In_Ph} \geq C^{In_Ph_Max} - M^{Ph} (1 - y^{C^{In_Ph_b}}) \quad (116)$$

$$C^{In_Ph} \geq C^{In_Ph_Min} - M^{Ph} (1 - y^{C^{In_Ph_c}}) \quad (117)$$

$$C^{In_Ph} \leq C^{In_Ph_Max} + M^{Ph} (1 - y^{C^{In_Ph_c}}) \quad (118)$$

The upper limits for the variables involved in the first section of the disjunction are set as follows:

$$m^{In} \leq M^{m^{In}} \left(1 - y^{C^{In_Ph} - a}\right) \quad (119)$$

$$m^{In} \geq -M^{m^{In}} \left(1 - y^{C^{In_Ph} - a}\right) \quad (120)$$

$$E.C. \leq M^{E.C.} \left(1 - y^{C^{In_Ph} - a}\right) \quad (121)$$

$$E.C. \geq -M^{E.C.} \left(1 - y^{C^{In_Ph} - a}\right) \quad (122)$$

$$Al.C. \leq M^{Al.C.} \left(1 - y^{C^{In_Ph} - a}\right) \quad (123)$$

$$Al.C. \geq -M^{Al.C.} \left(1 - y^{C^{In_Ph} - a}\right) \quad (124)$$

$$Cost^{OP_Treatm} \leq M^{Cost^{OP_Treatm}} \left(1 - y^{C^{In_Ph} - a}\right) \quad (125)$$

$$Cost^{OP_Treatm} \geq -M^{Cost^{OP_Treatm}} \left(1 - y^{C^{In_Ph} - a}\right) \quad (126)$$

The upper limits for the second section of the disjunction are stated as follows:

$$m^{In} \leq M^{m^{In}} \left(1 - y^{C^{In_Ph} - b}\right) \quad (127)$$

$$m^{In} \geq -M^{m^{In}} \left(1 - y^{C^{In_Ph} - b}\right) \quad (128)$$

$$E.C. \leq M^{E.C.} \left(1 - y^{C^{In_Ph} - b}\right) \quad (129)$$

$$E.C. \geq -M^{E.C.} \left(1 - y^{C^{In_Ph} - b}\right) \quad (130)$$

$$Al.C. \leq M^{Al.C.} \left(1 - y^{C^{In_Ph} - b}\right) \quad (131)$$

$$Al.C. \geq -M^{Al.C.} \left(1 - y^{C^{In_Ph} - b}\right) \quad (132)$$

$$Cost^{OP_Treatm} \leq M^{Cost^{OP_Treatm}} \left(1 - y^{C^{In_Ph} - b}\right) \quad (133)$$

$$Cost^{OP_Treatm} \geq -M^{Cost^{OP_Treatm}} \left(1 - y^{C^{In_Ph} - b}\right) \quad (134)$$

The relationships for the last section are stated as follows:

$$E.C. \leq -1.1917 \times 10^{-7} (C^{In-Ph})^3 + 6.5373 \times 10^{-5} (C^{In-Ph})^2 - 1.1315 \times 10^{-2} (C^{In-Ph}) + 0.7793 + M^{Al.C.} (1 - y^{C^{In-Ph}-c}) \quad (135)$$

$$E.C. \geq -1.1917 \times 10^{-7} (C^{In-Ph})^3 + 6.5373 \times 10^{-5} (C^{In-Ph})^2 - 1.1315 \times 10^{-2} (C^{In-Ph}) + 0.7793 - M^{Al.C.} (1 - y^{C^{In-Ph}-c}) \quad (136)$$

$$Al.C. \leq 2.8521 \times 10^{-8} (C^{In-Ph})^3 - 4.2077 \times 10^{-6} (C^{In-Ph})^2 + 8.6930 \times 10^{-4} (C^{In-Ph}) + 0.1676 + M^{E.C.} (1 - y^{C^{In-Ph}-c}) \quad (137)$$

$$Al.C. \geq 2.8521 \times 10^{-8} (C^{In-Ph})^3 - 4.2077 \times 10^{-6} (C^{In-Ph})^2 + 8.6930 \times 10^{-4} (C^{In-Ph}) + 0.1676 - M^{E.C.} (1 - y^{C^{In-Ph}-c}) \quad (138)$$

$$Cost^{OP-Treatm} \leq UC^{KWh} E.C. + UC^{Al.C.} Al.C. + M^{Cost^{OP-Treatm}} (1 - y^{C^{In-Ph}-c}) \quad (139)$$

$$Cost^{OP-Treatm} \geq UC^{KWh} E.C. + UC^{Al.C.} Al.C. - M^{Cost^{OP-Treatm}} (1 - y^{C^{In-Ph}-c}) \quad (140)$$

It should be noted that the removal efficiency for the electrocoagulation system, γ , is determined experimentally and depends on the following functionality:

$$\gamma = f(pH, I, C^{In-Ph}, C^{NaCl}, T) \quad (141)$$

Notice that the desired conditions determined experimentally for the electrocoagulation system were considered; however, the initial phenol concentration is the only optimization variable in the formulation. This is modeled employing the correlation shown in **Figure 4**.

$$\gamma = \frac{-0.3005 C^{In-Ph} + 108.74}{100} \quad (142)$$

The outlet phenol concentration from the electrocoagulation unit is determined as follows:

$$P_{C^{Ph}}^{Out} = P_{C^{Ph}}^{In} (1 - \gamma) \quad (143)$$

Finally, the capital cost for the entire electrocoagulation unit ($Cost^{Treatm}$) depends on the flowrate treated as follows:

$$Cost^{Treatm} \leq M^{Cost^{Treatm}} (1 - y^{C^{In-Ph}-a}) \quad (144)$$

$$Cost^{Treatm} \geq -M^{Cost^{Treatm}} (1 - y^{C^{In-Ph}-a}) \quad (145)$$

$$Cost^{Treatm} \leq M^{Cost^{Treatm}} \left(1 - y^{C^{In_Ph} - b}\right) \quad (146)$$

$$Cost^{Treatm} \geq -M^{Cost^{Treatm}} \left(1 - y^{C^{In_Ph} - b}\right) \quad (147)$$

$$Cost^{Treatm} \leq A^{Treatm} + B^{Treatm} (m^{In})^{C^{Treatm}} + M^{Cost^{Treatm}} \left(1 - y^{C^{In_Ph} - c}\right) \quad (148)$$

$$Cost^{Treatm} \geq A^{Treatm} + B^{Treatm} (m^{In})^{C^{Treatm}} - M^{Cost^{Treatm}} \left(1 - y^{C^{In_Ph} - c}\right) \quad (149)$$

Objective function

The total annualized cost is the sum of the total operating costs including the fresh water cost ($F_r UC^{Fresh}$) and the costs associated with operate each unit required plus the annualized capital costs for the units involved in the treatment.

$$TAC = H_Y \left[F_r UC^{Fresh} + \left(Cost^{pH} + Cost^{NaCl} + Cost^{H_2O} + Cost^{cw} + Cost^{steam} + Cost^{OP_RO} + Cost^{OP_Treatm} \right) \right] + k_F \left(Cost^{Cool} + Cost^{Heat} + Cost^{Treatm} + Cost^{RO} \right) \quad (150)$$

In the previous equation, H_Y represents the operating hours (8000 h/year) and k_F corresponds to the annualization factor for the depreciation of capital cost. It should be noted that the corresponding optimization formulation is a mixed-integer nonlinear programming (MINLP) problem, which was coded in the GAMS software.²

S2. NOMENCLATURE

Variables

<i>Acid</i>	Acid added in treatment
A^{EXC}	Cooler's area
A^{EXC2}	Heater's area
<i>Al.C.</i>	Aluminum consumed
<i>Base</i>	Base added in treatment
$C^{In_Mix}_{NaCl}$	NaCl concentration inlet to MIX-1
C^{NaCl}	NaCl concentration
Cp^{In_Mix}	Heat capacity

C^{In_Ph}	Phenol concentration
$Cost^{OP_Tratm}$	Operating treatment cost
$Cost^{Treatm}$	Treatment cost
$Cost^{steam}$	Steam cost
$Cost^{cw}$	Cooling cost
$Cost^{Heat}$	Heater cost
$Cost^{NaCl}$	NaCl cost
$Cost^{H_2O}$	Fresh water cost
$Cost^{pH}$	pH adjust cost
$Cost^{cool}$	Cooler cost
$Cost^{OP_RO}$	Operating cost for reverse osmosis
$Cost^{RO}$	Reverse osmosis unit cost
$E.C.$	Energy consumed
F_r	Total flowrate of fresh sources r
$f_{r,j}$	Segregated flowrate from fresh source r to sink j
g_{Waste}	Segregated flowrate from treatment to waste
g_{Waste}^{RO}	Segregated flowrate from RO to waste
g_j	Segregated flowrate from treatment to sink j
g_j^{RO}	Segregated flowrate from RO to sink j
h_i	Segregated flowrate from sources to treatment
$LMTD$	Logarithmic mean temperature difference
m^{In}	Total flowrate from sources to treatment
m^{Out}	Total flowrate from treatment to sinks

m^{NaCl}	Total flowrate of NaCl added in treatment
m^{Fresh}	Total flowrate of fresh water added in treatment
m^{Out_RO}	Total flowrate from RO to sinks
m^{OutRO_Min}	Minimum total flowrate from RO to sinks
m^{OutRO_Max}	Maximum total flowrate from RO to sinks
pH^{In_Mix}	pH inlet to MIX-1
Q^{cw}	Heat removed
Q^{steam}	Heat added
T^{In_Mix}	Temperature inlet to MIX-1
$w_{i,j}$	Segregated flowrate from sources to sinks
$w_{i,waste}$	Segregated flowrate from sources to waste
$Waste$	Total flowrate for the waste stream discharged to the environment
TAC	Total annual cost

Parameters

A	Constant for the cost estimation
B	Constant for the cost estimation
C	Constant for the cost estimation
$Conc^{Acid}$	Acid concentration
$Conc^{Base}$	Base concentration
$C_{NaCl}^{desired}$	Desired NaCl concentration for electrocoagulation process
$C^{In_Ph_Max}$	Maximum concentration of phenol
$C^{In_Ph_Min}$	Minimum concentration of phenol
G_j^{Sink}	Total flowrate inlet to sink j

H_Y	Time of operation of the plant in h/year
k_F	Factor used to annualize the inversion
$pH_{desired}$	Desired pH for electrocoagulation process
U	Global heat transfer coefficient
UC^{Acid}	Unit cost for acid in treatment
$UC^{Al.C.}$	Unit cost for aluminum anode
UC^{Base}	Unit cost for base in treatment
UC^{cw}	Unit cost for coolant in treatment
UC^{Fresh}	Unit cost for fresh water
UC^{H_2O}	Unit cost for fresh water in treatment
UC^{KWh}	Unit cost for electricity
UC^{NaCl}	Unit cost for sodium chloride
UC^{RO}	Unit cost for treating wastewater in reverse osmosis
UC^{steam}	Unit cost for the steam
$T^{desired}$	Desired temperature for the electrocoagulation process
W_i	Total flowrate for the process source i

Greek symbols

α	Efficiency factor for the reverse osmosis process
β	Efficiency factor for the electrocoagulation process
γ	Efficiency factor for phenol removal in electrocoagulation process
ρ^{In_Mix}	Inlet treatment density
$\psi_p^{InMix}(P)$	Properties for the streams inlet to the treatment unit
$\psi_{p,i}^{Source}(P)$	Properties of the streams from the different process sources

$\psi_{p,j}^{Sink}(P)$	Property operator for the sinks
$\psi_{p,j}^{Sink_Max}(P)$	Maximum for the property operator for the sinks
$\psi_{p,j}^{Sink_Min}(P)$	Minimum for the property operator for the sinks
$\psi_{p,waste}^{Sink_Max}(P)$	Maximum for the property operator for the waste
$\psi_{p,waste}^{Sink_Min}(P)$	Minimum for the property operator for the waste
$\psi_{p,r}^{Fresh}(P)$	Property operator for fresh sources
$\psi_p^{OutSplit}(P)$	Property operator for treated wastewater
$\psi_p^{RO}(P)$	Property operator for reverse osmosis outlet stream

Big M Parameters

M^{pH}	pH adjustment
M^{Base}	Mass of base
$M^{cost^{pH}}$	pH adjustment cost
M^{Acid}	Mass of acid
$M^{C^{NaCl}}$	NaCl concentration
M^{NaCl}	Mass of NaCl
$M^{cost^{NaCl}}$	NaCl concentration adjustment cost
M^{Fresh}	Mass of fresh water
$M^{cost^{H_2O}}$	NaCl concentration adjustment cost
$M^{T^{In}}$	Inlet temperature adjustment
$M^{Q^{cw}}$	Heat removal
$M^{cost^{cw}}$	Heat removal cost
M^A	Cooler area

M^{LMTD}	Logarithmic mean temperature
$M^{cost^{cool}}$	Cooler cost
$M^{Q^{steam}}$	Adding heat
$M^{cost^{steam}}$	Heating cost
M^{A^2}	Heater area
$M^{cost^{Heat}}$	Heater cost
M_{NaCl}^{RO}	NaCl concentration in reverse osmosis
$M^{cost^{OutRO_Max}}$	Inlet flowrate to reverse osmosis
$M^{OPROCost}$	Operating cost of the reverse osmosis unit
M^{RO}	Capital cost of reverse osmosis
M^{Ph}	Phenol concentration
$M^{m^{In}}$	Inlet flow to treatment
$M^{E.C.}$	Energy consumption
$M^{Al.C.}$	Aluminum consumption
M^{costOP_Treatm}	Operating cost of treatment
$M^{cost^{Treatm}}$	Capital cost of treatment

Boolean Variables

Y^{pH_a}	Logic variable for case a for pH adjustment
Y^{pH_b}	Logic variable for case b for pH adjustment
Y^{pH_c}	Logic variable for case c for pH adjustment
Y^{NaCl_a}	Logic variable for case a for NaCl concentration adjustment
Y^{NaCl_b}	Logic variable for case b for NaCl concentration adjustment

Y^{NaCl}_{-c}	Logic variable for case b for NaCl concentration adjustment
Y^{EXC}_{-a}	Logic variable for case a for temperature adjustment
Y^{EXC}_{-b}	Logic variable for case b for temperature adjustment
Y^{EXC}_{-c}	Logic variable for case c for temperature adjustment
Y^{RO}_{-a}	Logic variable for case a for reverse osmosis process
Y^{RO}_{-b}	Logic variable for case b for reverse osmosis process
$Y^{In_Ph}_{-a}$	Logic variable for case a for electrocoagulation process
$Y^{In_Ph}_{-b}$	Logic variable for case b for electrocoagulation process
$Y^{In_Ph}_{-c}$	Logic variable for case c for electrocoagulation process

Subscripts and superscripts

Max	Maximum
Min	Minimum
In	Inlet
Out	Outlet
Cool	Cooler
Heat	Heater
RO	Reverse osmosis
Treat	Treatment
desired	Desired value

Indices

i	Process source
j	Sink
p	Property

r Fresh source

Waste Waste stream

S3. RESULTS AND PARAMETERS

This section presents complementary tables that provide the entire results and the parameters used for solving the presented case studies.

Table S1. Flowrate and property constrains for the sinks for case study 1.

G_j	Flow (m ³ /h)	Phenol (mg/l)	pH		NaCl (kg/m ³)	Temperature (K)	
		Max	Min	Max	Max	Min	Max
1	60	5	6.7	8.2	2.0	300.00	305.00
2	160	10	6.5	8.5	3.0	300.00	305.00
3	80	3	6	8	1.0	295.00	300.00
4	70	13	7	7.8	0.9	295.00	298.00
Waste	-	1	6	8	2.0	293.00	303.00

Table S2. Segregated flowrates from sources, fresh water, treatment unit, and RO to the sinks and waste for case study 1.

G_j	Fresh source (m ³ /h)	Treatment unit (m ³ /h)	Sources (m ³ /h)								Total (m ³ /h)
			W_1	W_2	W_3	W_4	W_5	W_6	W_7	W_8	
1	0	23.70	0	0	30.60	0	0	0	0	5.72	60.0
2	0	0	0	0	46.09	3.15	39.43	0	50.47	20.86	160.0
3	25.00	25.00	0	0	30.00	0	0	0	0	0	80.0
4	26.01	25.95	6.31	3.15	8.59	0	0	0	0	0	70.0
Waste	-	209.21	0	21.19	0	0	0	0	0	0	230.4

Table S3. Sink and waste properties for case study 1.

G_j	Flow (m ³ /h)	Phenol (mg/l)	pH	NaCl (kg/m ³)	Temperature (K)
1	60	5	7.75	1.0	303.00
2	160	10	7.80	2.0	312.82
3	80	0.756	6.98	1.0	300.00
4	70	3.214	7.08	0	297.00
Waste	230.4	1	7.97	2.0	294.56

Table S4. Flowrate from sources to treatment for case study 1.

h_i	Flow (m ³ /h)	Phenol (mg/l)	pH	NaCl (kg/m ³)	Temperature (K)
3	21.271	1.2	8.0	1.0	310.00
6	11.040	8.5	6.8	3.0	298.00
8	257.330	42	7.0	2.0	307.00
Total	289.640	37.73	7.22	2.0	306.88

Table S5. Flowrate in the electrocoagulation treatment system for case study 1.

	Flow (m ³ /h)	Phenol (mg/l)	pH	NaCl (kg/m ³)	Temperature (K)
m^{In}	289.64	37.73	7.22	2.0	306.88
m^{Out}	283.85	0.98	8	2.0	293.00

Table S6. Flowrate and properties constrains for the sinks and waste for case study 2.

G_j	Flow (m ³ /h)	Phenol (mg/l)	pH		NaCl (kg/m ³)	Temperature (K)	
		Max	Min	Max	Max	Min	Max
Desalter	27.50	4.5	6.7	7.5	2.0	293.00	305.00
Sweetening	45.00	5	6.3	7.8	1.0	288.00	307.00
Quenching	50.00	7	7	8	3.0	290.00	298.00
Waste	-	1	6	8	2.0	293.00	303.00

Table S7. Segregated flowrates from sources, fresh water, treatment unit and RO to sinks for case study 2.

G_j	Fresh source (m ³ /h)	Treatment unit (m ³ /h)	Sources (m ³ /h)						Total (m ³ /h)
			W_1	W_2	W_3	W_4	W_5	W_6	
Desalter	23.54	0	0	0	0	0	0.96	3.00	27.50
Sweetening	27.09	0	0	0	0	16.07	1.84	0	45.00
Quenching	4.67	14.76	3.00	13.63	0	13.93	0	0	50.00
Waste	-	0	0	0	0	0	0	0	0

Table S8. Sink properties for case study 2.

G_j	Flow (m ³ /h)	Phenol (mg/l)	pH	NaCl (kg/m ³)	Temperature (K)
Desalter	27.50	4.5	7.13	0.1	297.97
Sweetening	45.00	5	6.44	0.3	297.67
Quenching	50.00	7	7.61	0.8	297.20
Waste	0	-	-	-	-

Table S9. Flowrates from sources to treatment for case study 2.

h_i	Flow (m ³ /h)	Phenol (mg/l)	pH	NaCl (kg/m ³)	Temperature (K)
Hydroskimmer	8.87	15	7.6	0.4	298.00
Condensate flare	6.00	50	8.2	3.2	303.00
Hydrocracker flare	0.20	35	7.8	3.0	308.00
Total	15.07	29.20	7.79	2.0	300.12

Table S10. Flowrates in the electrocoagulation treatment system for case study 2.

	Flowrate (m ³ /h)	Phenol (mg/l)	pH	NaCl (kg/m ³)	Temperature (K)
m^{In}	15.06	29.20	7.22	2.0	300.12
m^{Out}	14.76	0.01	8	2.0	293.00

Table S11. Parameters used in the case studies

Concept	Value
Hours of operation per year, H_Y (hr/year)	8000
Factor used to annualize the inversion, k_F (year ⁻¹)	0.24
Unitary price of aluminum (\$/kg)	2
Unitary price of cooling (\$/10 ⁶ kJ)	4
Unitary price of steam (\$/10 ³ kJ)	7.8
Unitary price of fresh water (\$/m ³)	2.48
Unitary price of electricity (\$/kWh)	0.0676
Unitary price of NaCl (\$/kg)	0.2
Unitary price of reverse osmosis (\$/m ³)	8.086
Global heat transfer coefficient, U (kJ/(h m ² K))	6750
Global heat transfer coefficient, U_2 (kJ/(h m ² K))	8100