# Synthesis and Physicochemical Properties of Double-Chain Cationic Surfactants 

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Supporting Information $\mathrm{C}:{ }^{1} \mathrm{H}$ NMR Spectra and ${ }^{13} \mathrm{C}$ NMR Spectra of 12-0-12
${ }^{1} \mathrm{H}$ NMR Spectra and ${ }^{13} \mathrm{C}$ NMR Spectra of 16-0-16


Figure S4a. ${ }^{1} \mathrm{H}$ NMR Spectra of 12-0-12


Figure S4b. ${ }^{1}$ H NMR Spectra of 12-0-12


Figure S4c. ${ }^{13} \mathrm{C}$ NMR Spectra of 12-0-12


Figure S5a. ${ }^{1} \mathrm{H}$ NMR Spectra of 16-0-16

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H1, 299.9477120 MHz
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Figure S5b. ${ }^{13}$ C NMR Spectra of 16-0-16

The purities of double-chain cationic surfactants $\mathrm{m}-0-\mathrm{m}(\mathrm{m}=10,12$ and 16) are estimated to be $\geq 99.5 \mathrm{~mol} \%$ from their ${ }^{1} \mathrm{H}$ NMR spectra. Because, there are no integral relating to any foreign proton.

On the other hand, the purity of the gemini surfactant 12-2-12 are estimated to be $97 \mathrm{~mol} \%$ from ${ }^{1} \mathrm{H}$ NMR spectra. The one impurity is the starting material $\mathrm{N}, \mathrm{N}, \mathrm{N}$ ', N 'tetramethylethylenediamine and the peaks come from its hydrogen are observed in 2.3-2.5 ppm . The purity of the gemini surfactant 12-2-12 was determined by the integral ratio of this proton to total integral of all hydrogen atoms. The purity of the gemini surfactant 16-2-16 are estimated to be $98,5 \mathrm{~mol} \%$ from ${ }^{1} \mathrm{H}$ NMR spectra. The one impurity is the same of 12-2-12 ( $\mathrm{N}, \mathrm{N}, \mathrm{N}^{\prime}, \mathrm{N}$ '-tetramethylethylenediamine) and the peaks come from its hydrogen are observed in 2.3-2.5 ppm. The purity of the gemini surfactant 16-2-16 was determined by the integral ratio of this proton to total integral of all hydrogen atoms.

