

Supporting Information

for

Tube-graft-Sheet Nano-Objects Created by A Stepwise Self-Assembly of Polymer-Polyoxometalate Hybrids

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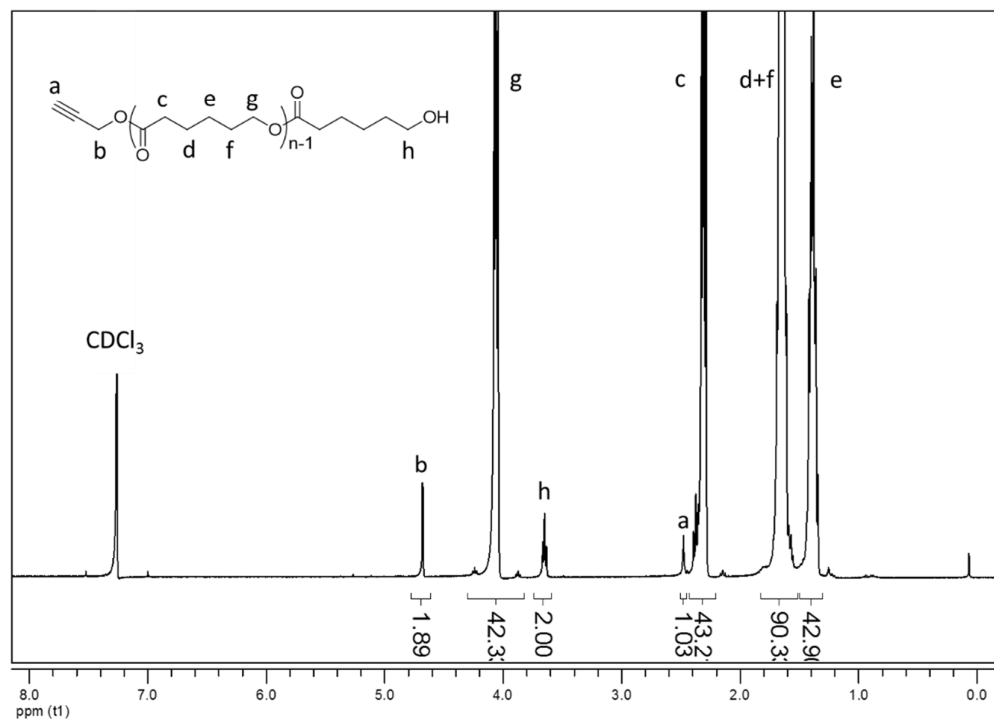


Figure S1. ^1H NMR spectrum of PCL₁₉-OH.

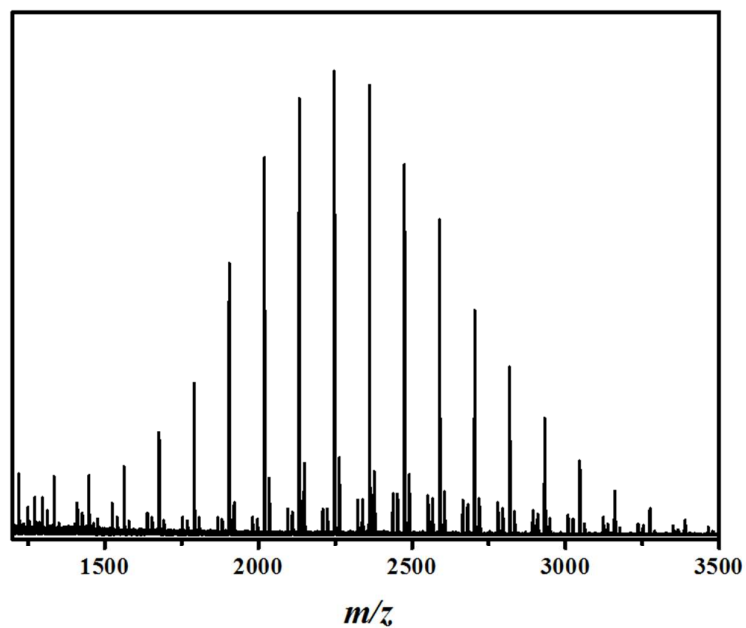


Figure S2. MALDI-TOF mass spectra of PCL₁₉. The calculated monoisotopic mass for the 19-mer of PCL₁₉ is 2247.0 Da ($[M_{19} \cdot Na]^+$) and the observed m/z equals 2247.3.

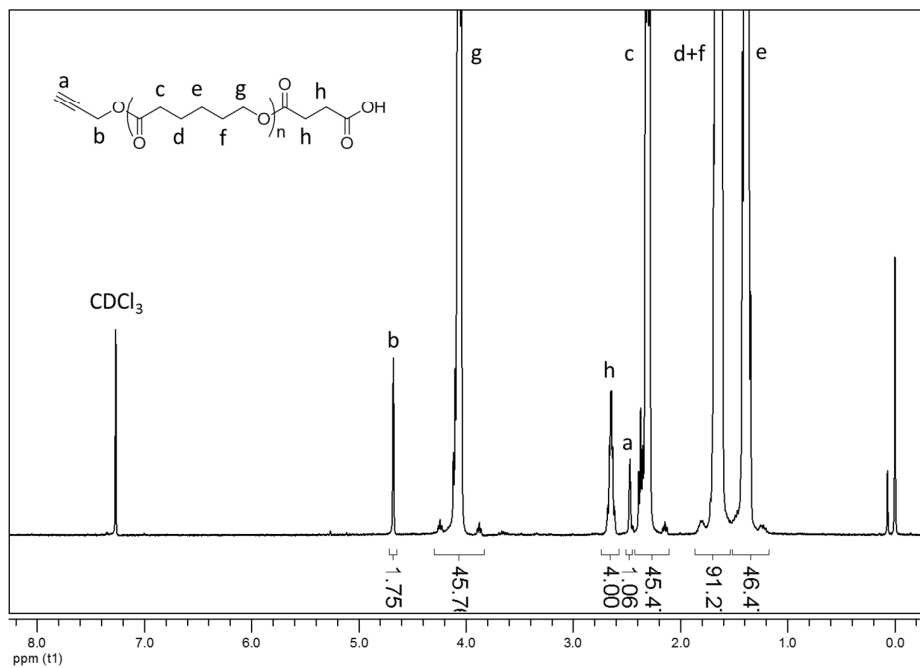


Figure S3. ¹H NMR spectrum of PCL₁₉-COOH.

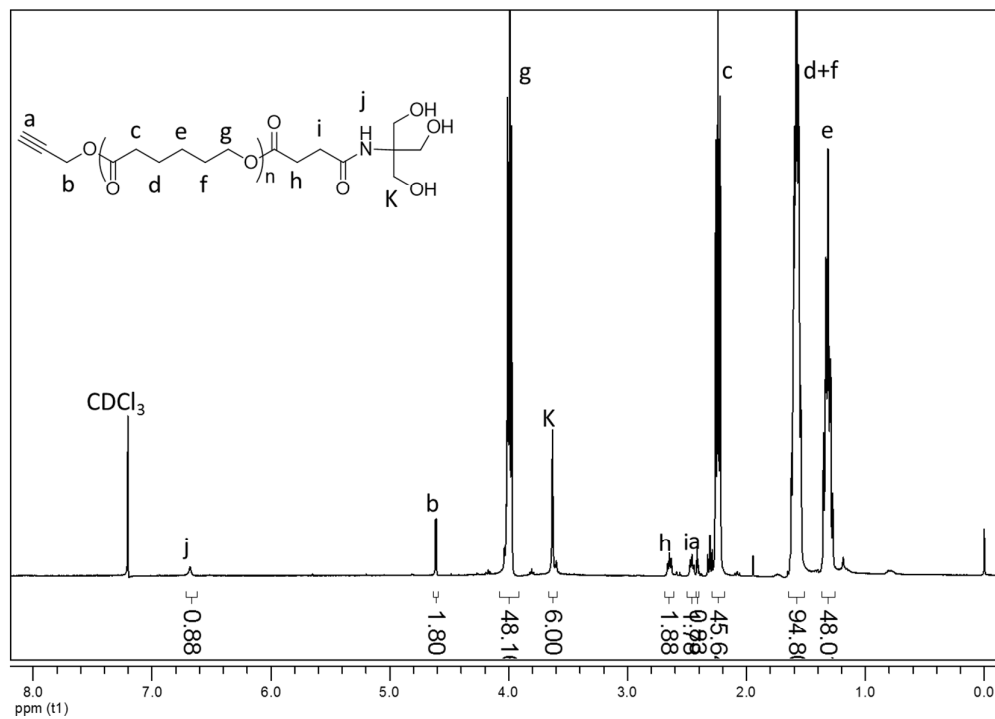


Figure S4. ¹H NMR spectrum of PCL₁₉-Tris.

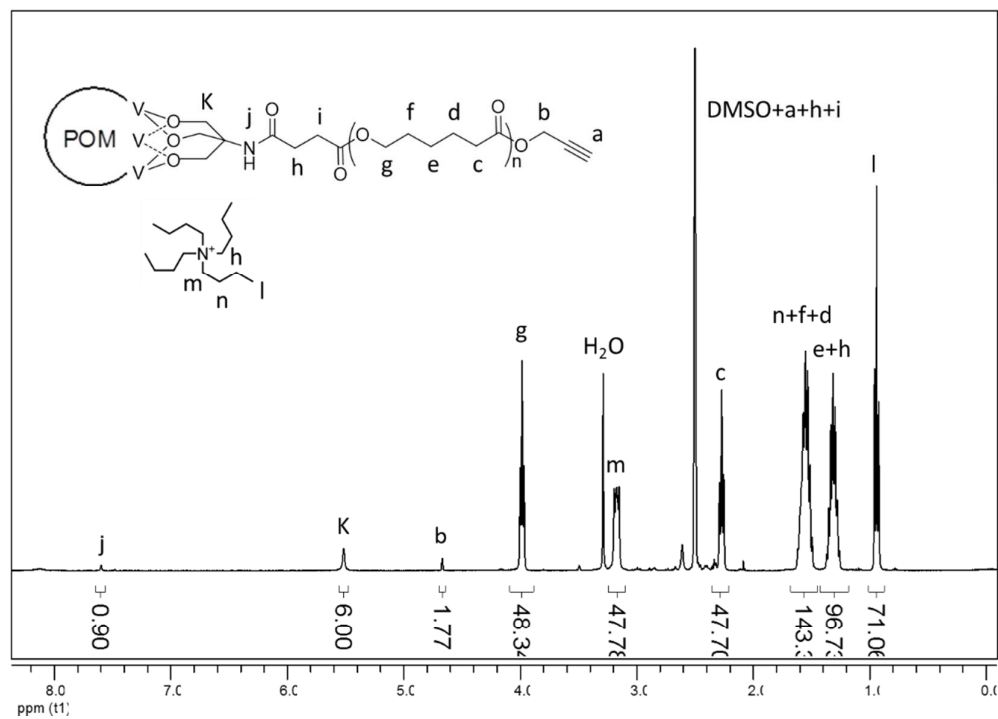


Figure S5. ¹H NMR spectrum of PCL₁₉-POM.

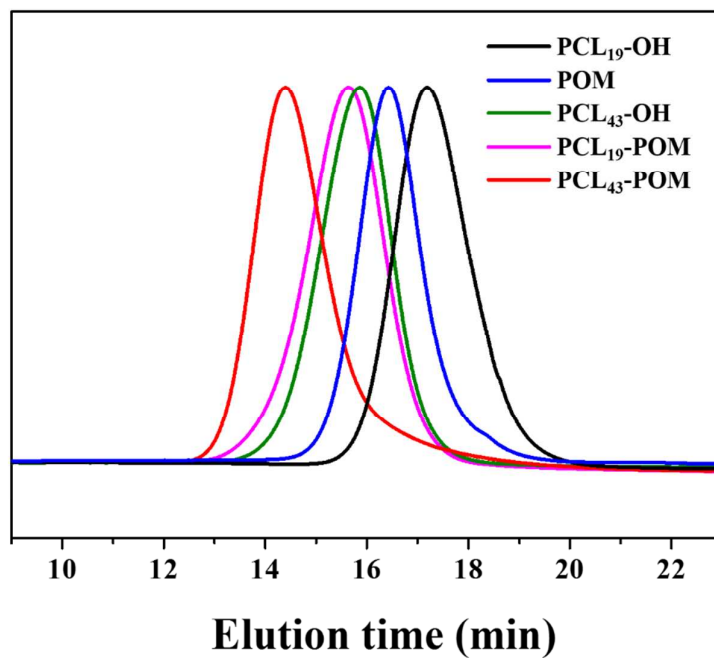


Figure S6. SEC chromatograms of the POM, PCL₁₉-OH, PCL₄₃-OH, PCL₁₉-POM and PCL₄₃-POM measured in DMF.

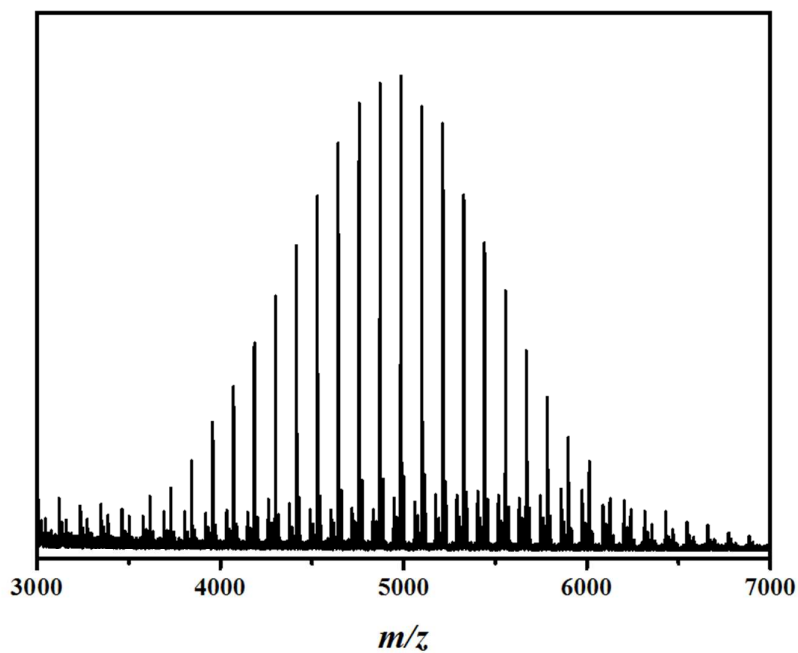


Figure S7. MALDI-TOF mass spectra of PCL₄₃. The calculated monoisotopic mass for the 43-mer of PCL₄₃ is 4985.4 Da ($[M_{43} \cdot Na]^+$) and the observed m/z equals 4985.7.

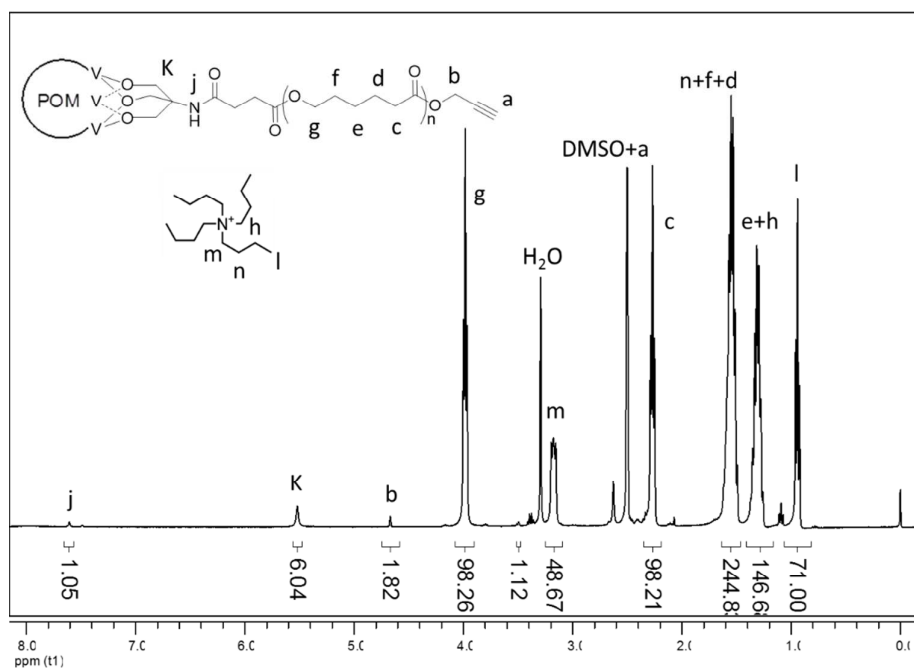
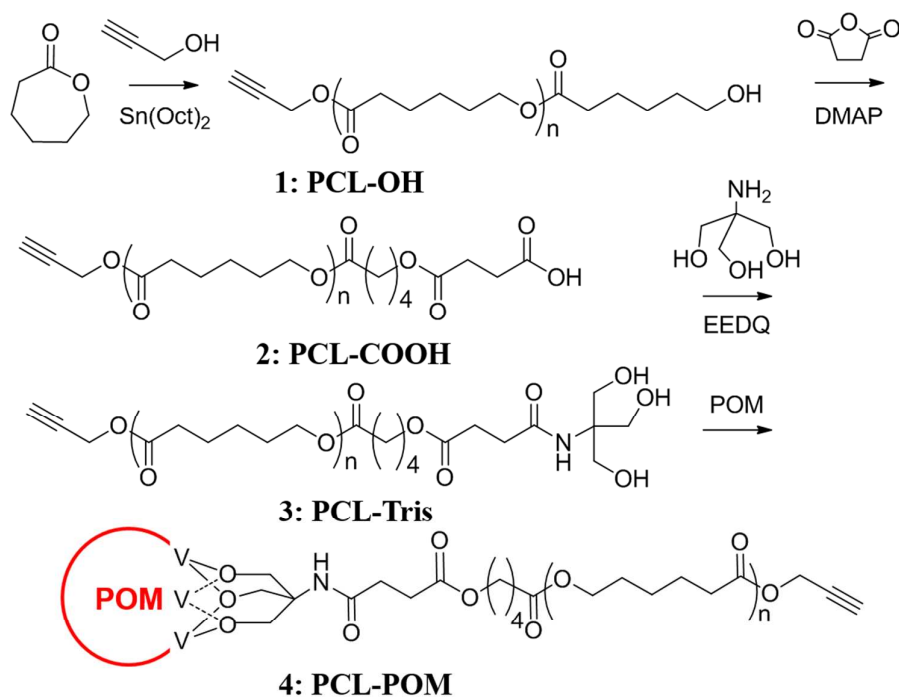


Figure S8. ^1H NMR spectrum of $\text{PCL}_{43}\text{-POM}$.

Table S1. Summary of molecular weights and molecular weight distributions ($\text{PDI} = \bar{M}_w / \bar{M}_n$) of POM-PCL hybrid.

Sample	\bar{M}_n kg/mol	\bar{M}_w kg/mol	\bar{M}_z kg/mol	PDI
POM	4.9	5.6	---	---
$\text{PCL}_{19}\text{-OH}$	2.0	2.3	3.5	1.15
$\text{PCL}_{19}\text{-POM}$	7.5	8.6	10.1	1.14
$\text{PCL}_{43}\text{-OH}$	5.1	6.1	6.5	1.19
$\text{PCL}_{43}\text{-POM}$	11.0	12.5	14.5	1.14



Scheme S1. Synthetic route of the PCL-POM hybrids.

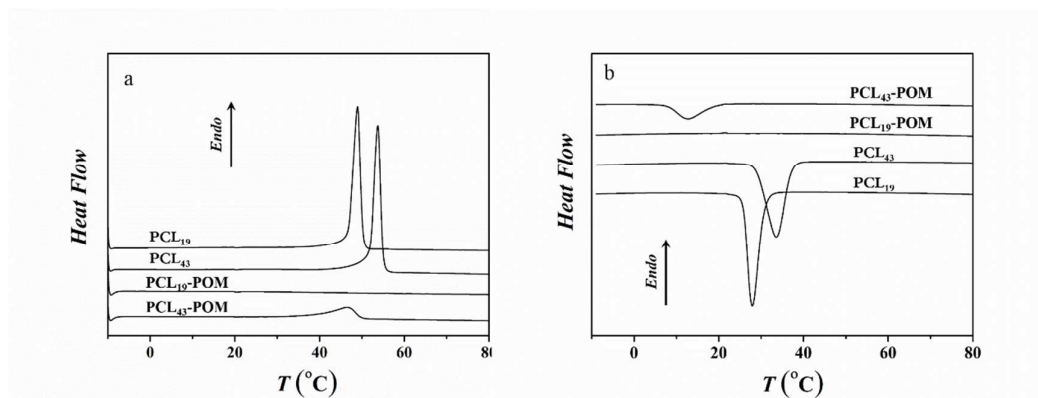


Figure S9. DSC thermographs of (a) second heating flow and (b) cooling flow for PCL₁₉, PCL₄₃, PCL₁₉-POM, and PCL₄₃-POM.

Table S2. DSC data of PCL₁₉, PCL₄₃, PCL₁₉-POM, and PCL₄₃-POM.

Sample	T_m (°C)	T_c (°C)	ΔH (J/g)	f_{PCL}^a
PCL ₁₉	49.0	27.9	93.5	0.69
PCL ₄₃	53.7	33.6	92.8	0.69
PCL ₁₉ -POM	---	---	---	---
PCL ₄₃ -POM	46.5	12.7	27.3	0.42

^a f_{PCL} is the degree of crystallinity of PCL. $f_{\text{PCL}} = \Delta H_c / \Delta H_{100}$, $H_c = \Delta H_{\text{DSC}} / \text{PCL}\%$, ΔH_{100} is the heat of crystallization of 100% crystalline PCL.

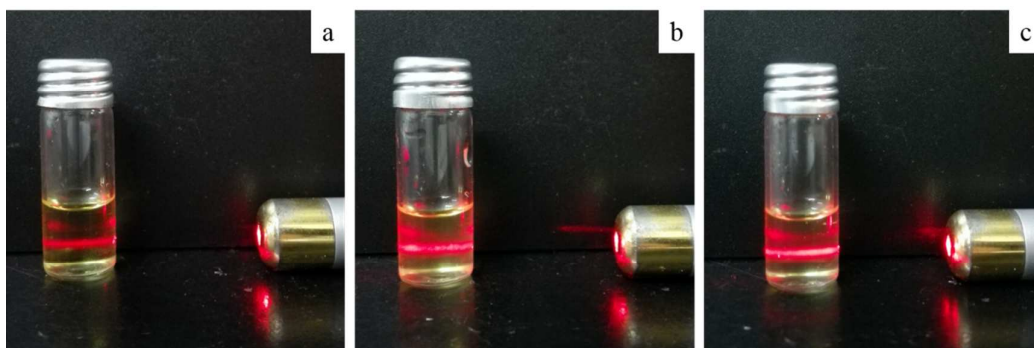


Fig. S10. Tyndall scattering of the solutions. (a) Solution of PCL₄₃-POM at $r = 5:1$ v/v of DMF/water. (b) Solution of PCL₁₉-POM at $r = 2:1$ v/v. (b) Solution prepared following a procedure: Dropping the DMF solution of PCL₁₉-POM into a suspension of the PCL₄₃-POM nanosheets at $r = 5:1$ v/v, and then immediately adding water into the suspension to change the ratio to $r = 2:1$ v/v.

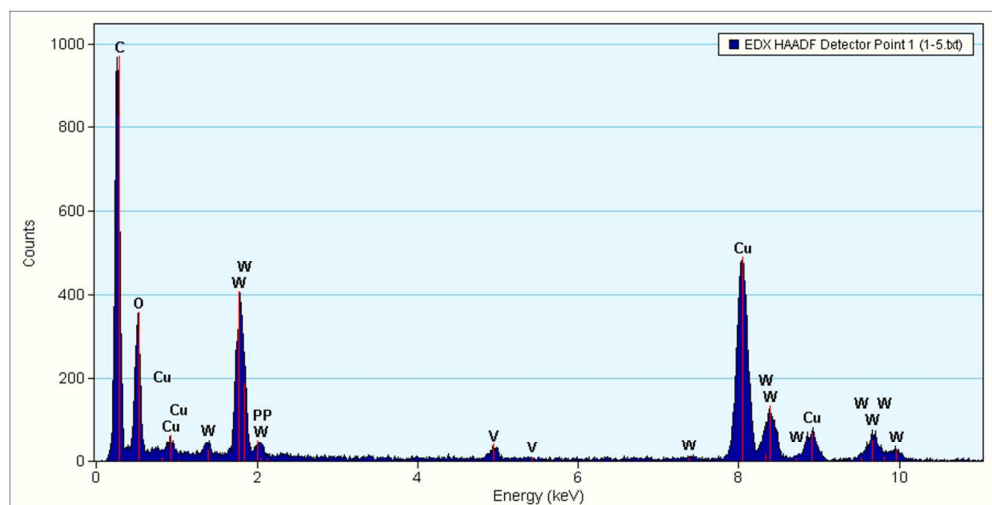


Figure S11. EDX spectrum taken on the nanosheets.

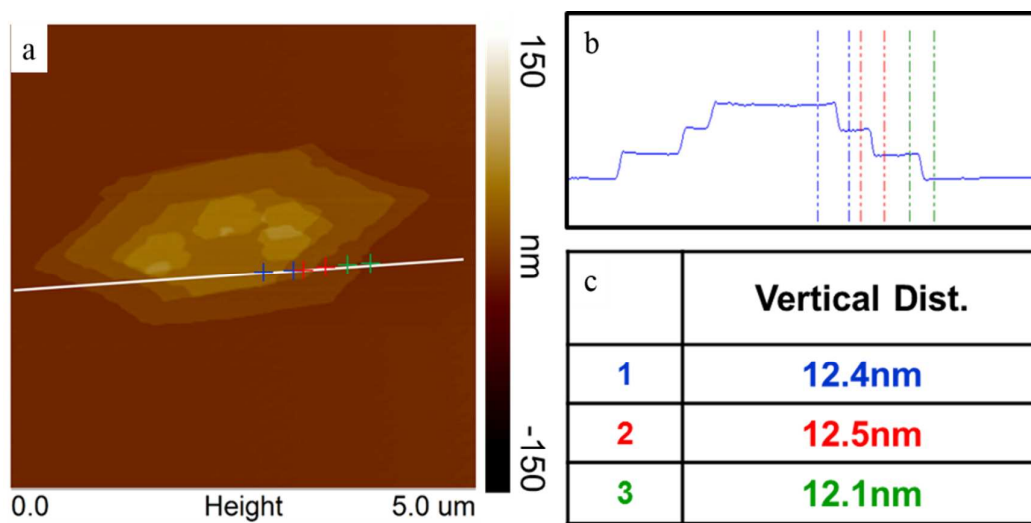


Figure S12. (a) AFM height images showing the PCL₄₃-POM nanosheets. (b) Height profiles of the nanosheets. (c) Measured vertical distance (or height) of the nanosheets.

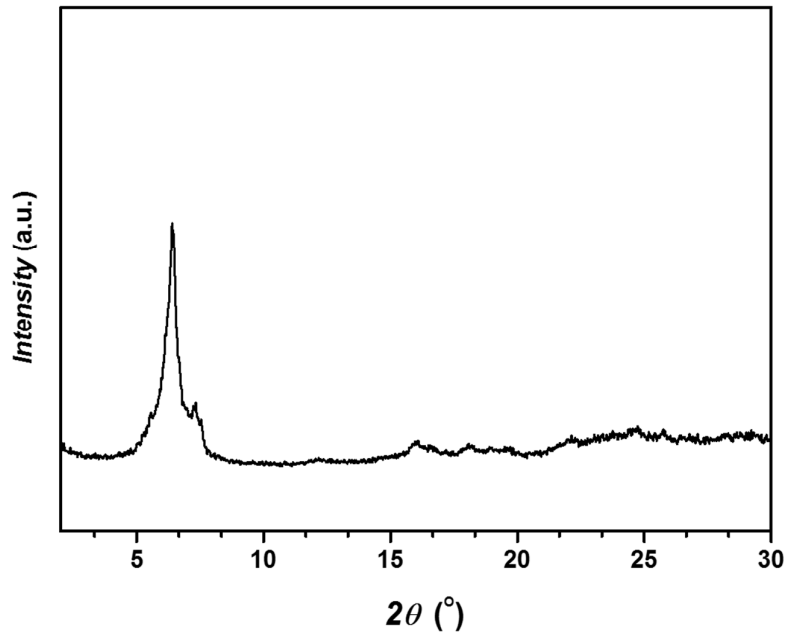


Figure S13. XRD pattern of the crystalline sample of the pure POM cluster.

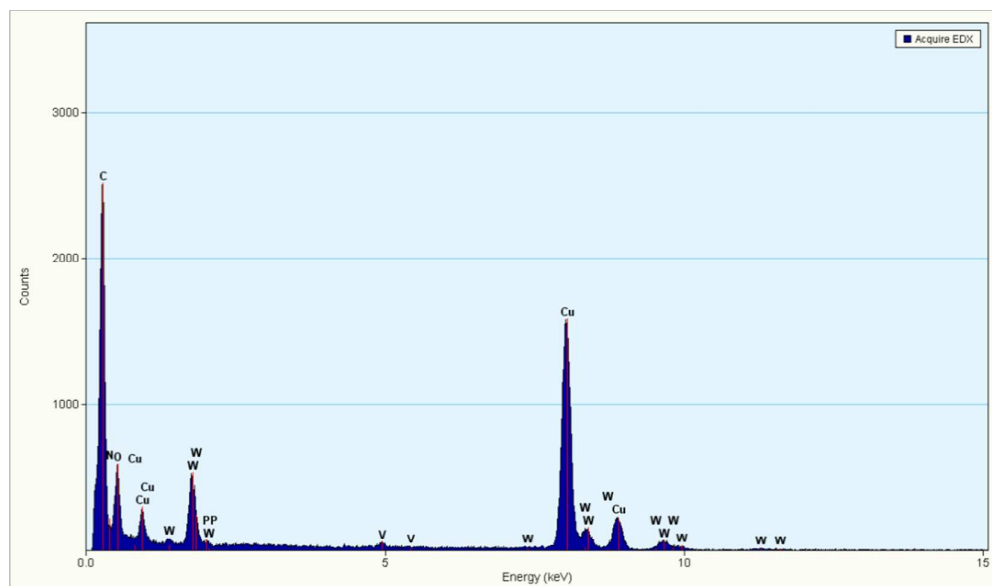


Figure S14. EDX spectrum taken on the nanotubes.

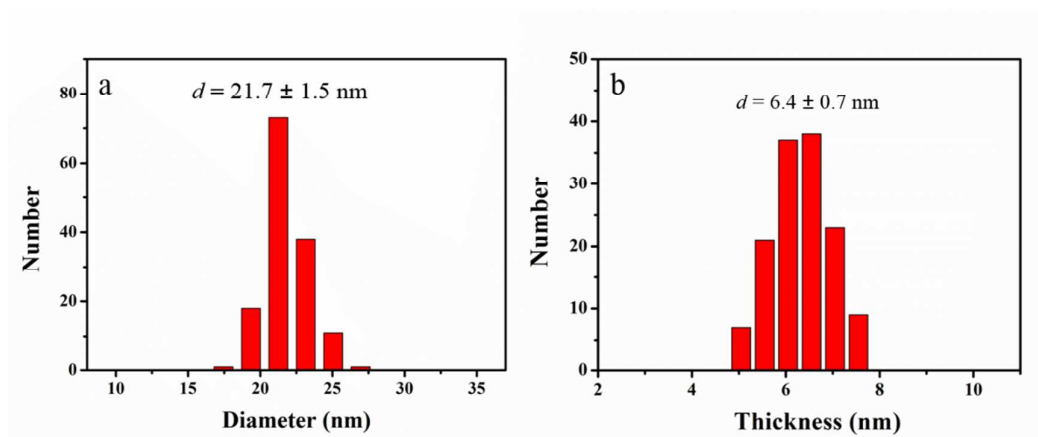


Figure S15. (a) The outside diameter distribution of nanotubes. (b) The outside diameter distribution of nanotubes.

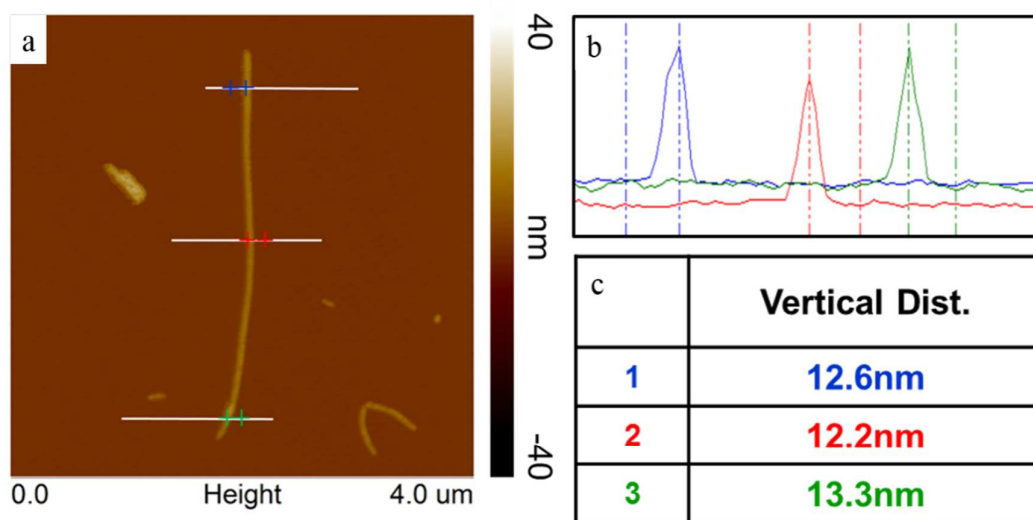


Figure S16. (a) AFM height images showing the PCL₁₉-POM nanotubes. (b) Height profiles of the nanotubes. (c) Measured vertical distance (or height) of the nanotubes.

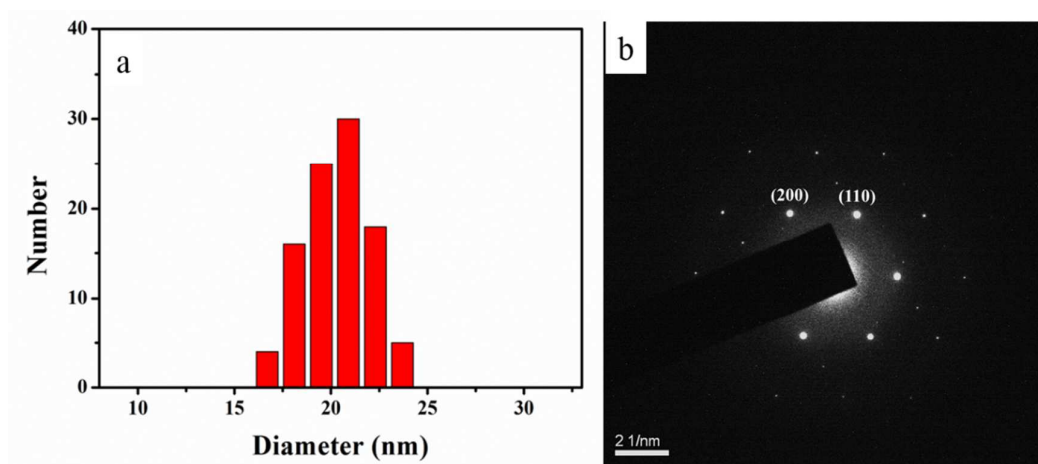


Figure S17. (a) The outside diameter distribution of grown nanotubes. (b) The ED pattern of the darker zone.

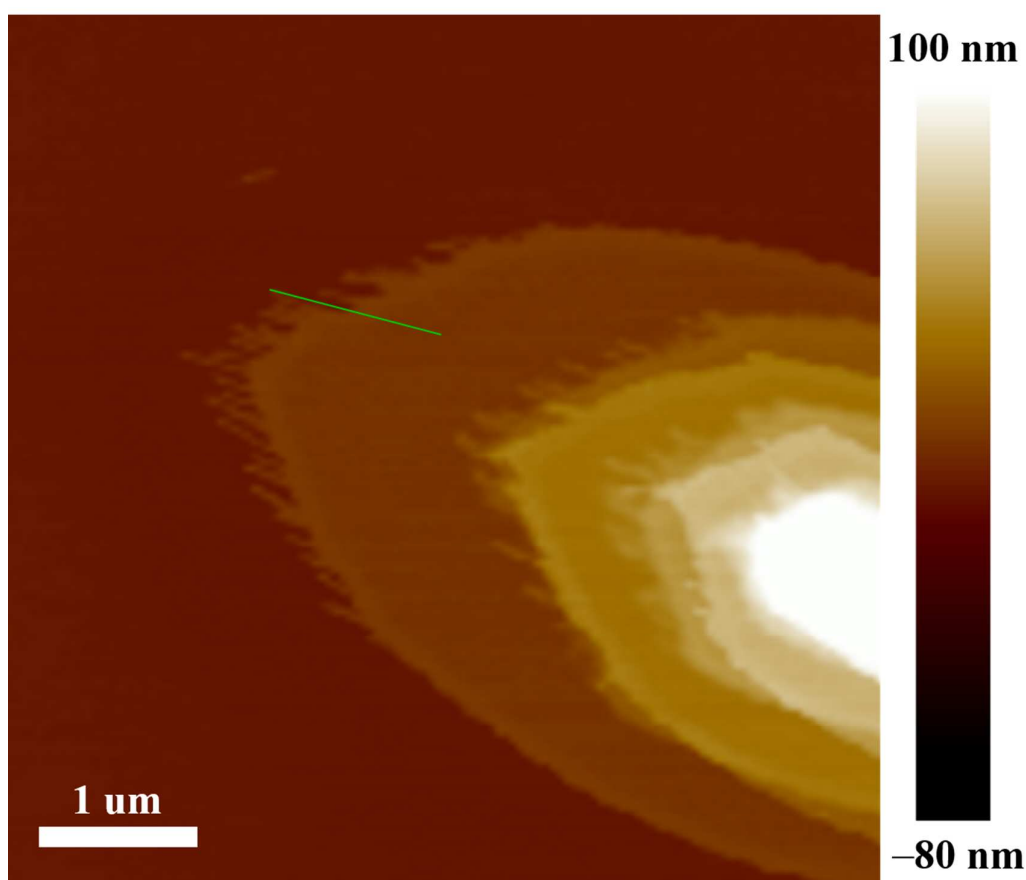


Figure S18. AFM height image showing the complex tube-graft-sheet nano-objects. The height profile in Figure 4c corresponds to the green line.