- 1 Supporting Information for
- 2 Modeling the effect of external carbon source addition under different electron acceptor
- 3 conditions in biological nutrient removal activated sludge systems
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- 16 This Supporting Information contains 18-page document, seven tables, two figures, including
- this cover page. This material is available free of charge via the Internet at <a href="http://pubs.acs.org">http://pubs.acs.org</a>.

#### Model limitations

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- Even though, the developed model is complex, there are still a few important issues that may be incorporated in further expansions or modifications:
- 21 (1) The competition between PAOs and GAOs was ignored due to a negligible activity of 22 the GAOs (confirmed experimentally). However, in the EBPR systems with a substantial 23 competition between PAOs and GAOs, the metabolism of GAOs cannot be ignored.
- 24 (2) When abundant in municipal wastewater, propionate may be incorporated in a model 25 as a separate state variable in order to describe better the response of PAOs to different 26 carbon sources (acetate and propionate) and the competition between PAOs and GAOs.
- 27 (3) Since the nitrite effect on PAOs has not ultimately been clarified, the actual inhibitory 28 effect of nitrite or FNA in BNR systems requires further investigation.
- 29 (4) Two-step denitrification is considered. The growing concern of greenhouse gas 30 (GHG) emissions from WWTPs may require further extensions by including four steps  $(NO_3 \rightarrow NO_2 \rightarrow NO \rightarrow N_2O \rightarrow N_2)$ .

#### Organization of the modeling study procedure

The modelling study followed the consecutive steps presented in SI Figure S1. First (step 1), the original ASM2d was calibrated/validated under dynamic conditions with results of both batch tests with the settled wastewater (without the addition of external carbon) and 96-hour measurement campaign in the full-scale MUCT bioreactor. Next, sensitivity analysis using a one-variable-at-a-time approach was performed to establish number of parameters that indeed influence the model outputs (step 2). Considering the most influential parameters, the expanded model was calibrated (step 3) based on experimental data of the selected two-phase batch tests carried out under anaerobic/aerobic and anaerobic/anoxic

conditions with acetate (SA) and ethanol (SA,1) as two different types of external carbon sources. For model calibration, a special GPS-X utility called "Optimizer" was used. Parameters were estimated based on the Nelder-Mead simplex method with the maximum likelihood as an objective function. The 95% confidence intervals for the estimated parameters were calculated from the parameter estimation error covariance matrix. Furthermore, a correlation matrix was developed in step 4 for each calibration step to evaluate the degree of correlation between pairs of the adjusted parameters and determine if a change in the value of one parameter could be compensated by a change in the value of another parameter (to avoid potential overparametrization). The details on the computational methods can be found in the GPS-X "Technical Reference" (Hydromantis, Canada ). The expanded model was validated based on results of the remaining one- and two-phase batch experiments (step 5). In order to evaluate the goodness of fit between observed data and model predictions, the absolute and relative standard errors as well as correlation coefficients of determination (R<sup>2</sup>) were determined for each measured variable. Finally (step 6), the long-term acclimation experiment with addition of the S<sub>A,1</sub> type of external carbon source (fusel oil) to the anoxic zone of a bench-scale pilot system was simulated with the expanded ASM2d. The biomass composition for setting the initial conditions was determined based on steadystate and dynamic simulations of the full-scale activated sludge system at the studied plant. This procedure was described by Swinarski et al.<sup>28</sup> Specifically, the original ASM2d model was first calibrated/validated under steady-state and dynamic simulations in the full-scale plant to determine each soluble and particulate fraction in the biomass. Then the initial biomass concentrations in each batch experiment were calculated based on the ratio of MLVSS concentration in the batch reactor and the MLVSS concentration in the sampling point at the full-scale plant.

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#### Parameter estimation, confidence intervals and correlation matrix

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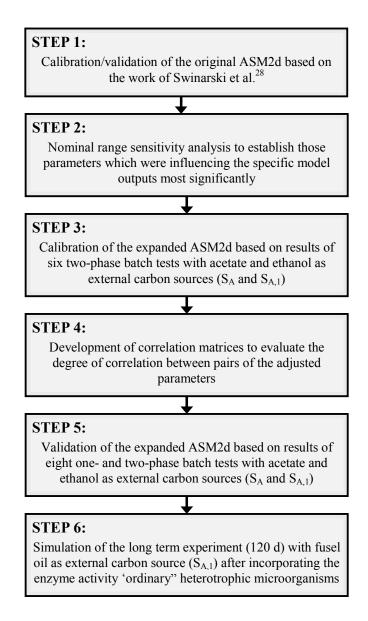
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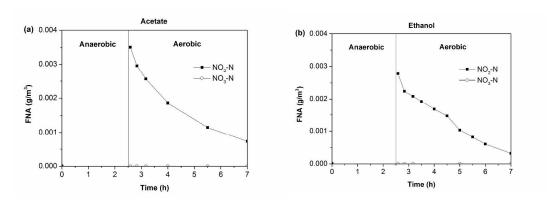
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The nominal range sensitivity analysis using a one-variable-at-a-time approach was performed to establish number of parameters that indeed influence the model outputs (Table S3 in SI).<sup>14</sup> Based on those results, 13 sensitive parameters were identified in the expanded ASM2d model, excluding the parameters which were directly adopted from the study of Swinarski et al.<sup>28</sup> Parameter estimation was performed based on data derived from the selected batch experiments, including two-phase tests with the different external carbon sources (acetate, ethanol) and electron acceptors (DO, NO<sub>2</sub>-N, NO<sub>3</sub>-N). Results of parameter estimation are shown in SI Table S5 along with uncertainty of the parameter estimates, quantified as the 95% confidence intervals. Several new parameters revealed relatively large confidence intervals, which implied rather poor parameter estimation in those cases e.g. K<sub>IOPHA</sub> =  $3.60\pm1.135$  or  $K_{IPHA} = 0.02\pm0.138$ . Therefore, a correlation matrix between the new estimated parameters was developed and analyzed in order to explain the large confidence intervals. Correlations between the parameters estimated in step 3.1 and steps 3.2-3.4 are shown in SI Table S6 and Table S7, respectively. Strong correlations (values close to 1) were found between the parameters K<sub>IPHA</sub> and K<sub>MAX1</sub> (0.98), and the parameters  $Y_{\rm H1}$  and  $\mu_{\rm H1}$  (0.96) in step 3.1 and 3.2, respectively. The strong correlation implied that a change in the value of one parameter could be compensated by a change in the value of another parameter. This could make it difficult to find unique estimates for those new parameters. Nevertheless, all optimization runs using different initial parameter estimates converged to the same optimal parameter set. The low standards errors listed in SI Table S4 and high coefficients of determination  $R^2$  (the average value of approximately 0.92) confirmed the goodness of fit with respect to the estimated parameters for each batch experiment.



- 92 Figure S1. Calibration/validation procedure of the expanded ASM2d for predicting the
- 93 effects of external carbon sources in BNR activated sludge systems.



**Figure S2.** FNA concentrations measured in two-phase anaerobic-anoxic experiments with different external carbon sources and different electron acceptors: (a) acetate, NO<sub>3</sub>-N and NO<sub>2</sub>-N, (b) ethanol, NO<sub>3</sub>-N and NO<sub>2</sub>-N.

**Table S1.** Stoichiometric matrix for the expanded ASM2d (only new and modified processes)

	Component	$S_o$	$S_F$	$S_A$	$S_{A1}$	$S_{NH4}$	$S_{NO2}$	$S_{NO3}$	$S_{PO4}$	$S_I$	$S_{ALK}$		
	Growth of heterotrophic organisms $(X_H)$												
1	Aerobic growth of $X_H$ on $S_{A,1}$	$1-\frac{1}{Y_{\rm HI}}$			$-\frac{1}{Y_{_{\rm H1}}}$	$\nu_{\rm l,NH4}$			$V_{1,PO4}$				
	Anoxic growth on $S_F$ , denitrification ( $S_{NO2}$ )		$-\frac{1}{Y_H}$			V <sub>2,NH4</sub>	$-\frac{1-Y_H}{1.71Y_H}$		$v_{2,PO4}$				
	Anoxic growth on $S_F$ , denitrification ( $S_{NO3}$ )		$-\frac{1}{Y_H}$			$V_{3,\mathrm{NH4}}$		$-\frac{1-Y_{\rm H}}{2.86Y_{\rm H}}$	$v_{3,PO4}$				
4	Anoxic growth on $S_A$ , denitrification ( $S_{NO2}$ )			$-\frac{1}{Y_{_{\rm H}}}$		$v_{_{4,NH4}}$	$-\frac{1-Y_H}{1.71Y_H}$		$V_{4,PO4}$				
	Anoxic growth on $S_A$ , denitrification ( $S_{NO3}$ )			$-\frac{1}{Y_H}$		$v_{5,NH4}$		$-\frac{1-Y_H}{2.86Y_H}$	$v_{5,PO4}$				
	Anoxic growth on $S_{A1}$ , denitrification ( $S_{NO2}$ )				$-\frac{1}{Y_{H1}}$	$v_{\scriptscriptstyle 6,NH4}$	$-\frac{1-Y_{H1}}{1.71Y_{H1}}$		$\nu_{\scriptscriptstyle 6,PO4}$				
7	Anoxic growth on $S_{A1}$ , denitrification $(S_{NO3})$				$-\frac{1}{Y_{_{H1}}}$	$v_{7,NH4}$		$-\frac{1-Y_{H1}}{2.86Y_{H1}}$	$\nu_{7,PO4}$				
Act	Activity of Phosphorus Accumulating Organisms (PAO) (X <sub>PAO</sub> )												
8	Storage of X <sub>PHA</sub> by PAOs			-1					$Y_{PO4}$				

	Component	$S_o$	$S_F$	$S_A$	$S_{A1}$	$S_{NH4}$	$S_{NO2}$	$S_{NO3}$	$S_{PO4}$	$S_I$	$\begin{array}{c} 10 \\ S_{ALK} \end{array}$
9	Aerobic storage of $X_{pp}$ by PAOs on $X_{PHA}$	$-Y_{PHA}$							-1		
	Anoxic storage of $X_{pp}$ , denitrification ( $S_{NO2}$ )						$ u_{10,NO2}$		-1		
11	Anoxic storage of $X_{pp}$ , denitrification ( $S_{NO3}$ )							$v_{_{11,NO3}}$	-1		
12	Aerobic storage of $X_{pp}$ by PAO on $S_A$	$-Y_{SA}$		$-Y_{SA}$					-1		
13	Anoxic storage of $X_{pp}$ on $S_A$ , denitrification $(S_{NO2})$			$-Y_{SA}$			$V_{13,NO2}$		-1		
14	Anoxic storage of $X_{pp}$ on $S_A$ , denitrification $(S_{NO3})$			$-Y_{SA}$				$V_{14,NO3}$	-1		
15	Aerobic storage of $X_{pp}$ by PAO on $S_{A,1}$	$-Y_{SAI}$			$-Y_{SAI}$				-1		
16	Anoxic storage of $X_{pp}$ on $S_{A,1}$ , denitrification $(S_{NO2})$				$-Y_{SAI}$		$ u_{16,NO2} $		-1		
	Anoxic storage of $X_{pp}$ on $S_{A,1}$ , denitrification $(S_{NO3})$				$-Y_{SA1}$			V <sub>17,NO3</sub>	-1		
18	Aerobic growth of X <sub>PAO</sub> on X <sub>PHA</sub>	$\nu_{\rm 18,O}$				V <sub>18,NH4</sub>			−i <sub>P,BM</sub>		
19	Anoxic growth of $X_{PAO,}$ denitrification $(S_{NO2})$					$ u_{19,NH4}$	$ u_{19,NO2}$		-i <sub>Р,ВМ</sub>		
20	Anoxic growth of $X_{PAO}$ , denitrification $(S_{NO3})$					$ u_{20,NH4} $		$V_{20,NO3}$	–i <sub>Р,ВМ</sub>		
21	Aerobic growth of $X_{PAO}$ on $S_A$	$\nu_{\rm 21,O}$		$-\frac{1}{Y_{PAO,SA}}$		$ u_{21,NH4}$			− <b>i</b> <sub>P,BN</sub>		
	Anoxic growth of $X_{PAO}$ on $S_A$ , denitrification $(S_{NO2})$			$-\frac{1}{Y_{PAO,SA}}$		$V_{22,NH4}$	$V_{22,NO2}$		-i <sub>P,BM</sub>		
23	Anoxic growth of $X_{PAO}$ on $S_{A}$ , denitrification $(S_{NO3})$			$-\frac{1}{Y_{PAO,i}}$		$V_{23,NH4}$		$V_{23NO3}$	- I , , , , , , , , , , , , , , , , , ,		
24	Aerobic growth of $X_{PAO}$ on $S_{A,1}$	$V_{24,O}$			$-\frac{1}{Y_{PAO,S}}$	V <sub>24,NH4</sub>			-i <sub>Р,ВМ</sub>		
25	Anoxic growth of $X_{PAO}$ on $S_{A,1}$ denitrification $(S_{NO2})$				$-\frac{1}{Y_{PAO,S}}$	V <sub>25,NH4</sub>	$V_{25,NO2}$		–i <sub>Р,ВМ</sub>		
26	Anoxic growth of $X_{PAO}$ on $S_{A,1}$ denitrification $(S_{NO3})$				$-\frac{1}{Y_{PAO,S}}$	V <sub>26,NH4</sub>		$V_{26NO3}$	-i <sub>P,ВМ</sub>		

Process		$S_o$	$S_F$	$S_A$	$S_{A1}$	$S_{NH4}$	$S_{NO2}$	$S_{NO3}$	$S_{PO4}$	$S_I$	$\begin{array}{c} 10 \\ S_{ALK} \end{array}$
	Hydrolysis processes										
	Anoxic hydrolysis of $X_S$ with nitrite $(S_{NO2})$		$1-f_{SI}$			V <sub>27, NH4</sub>			$v_{27,PO4}$	$f_{\scriptscriptstyle SI}$	
28 Anoxic hydrolysis of $X_S$ with nitrate $(S_{NO3})$			$1-f_{SI}$			$\nu_{_{28,NH4}}$			ν <sub>28,PO4</sub>	$f_{SI}$	
	Activity of t	he denitr	rificatio	n enzyi	me for t	the exteri	nal carbo	n source (S	$S_{A,1}$		
29	Enzyme synthesis										
30 Enzyme decay											

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**Table S1.** (continued) Stoichiometric matrix for the expanded ASM2d (only new and modified processes)

	Process	$X_S$	12 X <sub>H</sub>	$X_{PAO}$	14 X <sub>PP</sub>	15 X <sub>PHA</sub>	16 E <sub>sat</sub>							
	Growth of heterotrophic organisms ( $X_H$ )													
1	Aerobic growth of $X_H$ on $S_{A,1}$		1											
2	Anoxic growth on $S_F$ , denitrification ( $S_{NO2}$ )		1											
3	Anoxic growth on $S_F$ , denitrification $(S_{NO3})$		1											
4	Anoxic growth on $S_A$ , denitrification ( $S_{NO2}$ )		1											
5	Anoxic growth on $S_A$ , denitrification ( $S_{NO3}$ )		1											
6	Anoxic growth on $S_{A1}$ , denitrification ( $S_{NO2}$ )		1											
7	Anoxic growth on $S_{A1}$ , denitrification ( $S_{NO3}$ )		1											
	Activity of Phosphorus Accumula	ating Org	ganisms (	(PAO) (X	PAO)									
8	Storage of $X_{PHA}$ by PAOs				$-Y_{PO4}$	1								
9	Aerobic storage of $X_{pp}$ by PAOs on $X_{PHA}$				1	$-Y_{PHA}$								
10	Anoxic storage of $X_{pp}$ , denitrification ( $S_{NO2}$ )				1	$-Y_{PHA}$								

_	Component	11	12	13	14	15	16
	Process	$X_{\scriptscriptstyle S}$	$X_H$	$X_{PAO}$	$X_{PP}$	$X_{PHA}$	E <sub>sat</sub>
11	Anoxic storage of $X_{pp}$ , denitrification ( $S_{NO3}$ )				1	$-Y_{PHA}$	
12	Aerobic storage of X <sub>pp</sub> by PAO on S <sub>A</sub>				1		
13	Anoxic storage of $X_{pp}$ on $S_A$ , denitrification $(S_{NO2})$				1		
14	Anoxic storage of $X_{pp}$ on $S_A$ , denitrification $(S_{NO3})$				1		
15	Aerobic storage of $X_{pp}$ by PAO on $S_{A,1}$				1		
16	Anoxic storage of $X_{pp}$ on $S_{A,1}$ , denitrification $(S_{NO2})$				1		
17	Anoxic storage of $X_{pp}$ on $S_{A,1}$ , denitrification $(S_{NO3})$				1		
18	Aerobic growth of $X_{PAO}$ on $X_{PHA}$			1		$-\frac{1}{Y_{PAO}}$	
19	Anoxic growth of $X_{PAO}$ , denitrification ( $S_{NO2}$ )			1		$-\frac{1}{Y_{PAO}}$	
20	Anoxic growth of $X_{PAO}$ , denitrification ( $S_{NO3}$ )			1		$-\frac{1}{Y_{PAO}}$	
21	Aerobic growth of $X_{PAO}$ on $S_A$			1			
22	Anoxic growth of $X_{PAO}$ on $S_{A}$ , denitrification $(S_{NO2})$			1			
23	Anoxic growth of $X_{PAO}$ on $S_{A}$ , denitrification $(S_{NO3})$			1			
24	Aerobic growth of $X_{PAO}$ on $S_{A,1}$			1			
25	Anoxic growth of $X_{PAO}$ on $S_{A,1}$ , denitrification $(S_{NO2})$			1			
26	Anoxic growth of $X_{PAO}$ on $S_{A,1}$ , denitrification $(S_{NO3})$			1			
	Hydrolysis p	processes	S				
27	Anoxic hydrolysis of $X_S$ with nitrite ( $S_{NO2}$ )	-1					
28	Anoxic hydrolysis of $X_S$ with nitrate $(S_{NO3})$	-1					
	Activity of the denitrification enzyme f	for the ex	ternal ca	rbon sou	rce (S <sub>A,1</sub>	)	
29	Enzyme synthesis						1
30	Enzyme decay						-1

**Table S2.** Process rates for the expanded ASM2d (only new and modified processes)

	Process	Process rate, $\rho_i$
		Growth of heterotrophic organisms $(X_H)$
1	Aerobic growth of $X_H$ on $S_{A,1}$	$\mu_{H1} \cdot \frac{S_{O,H}}{K_{O,H} + S_O} \cdot \frac{S_{A1}}{K_{SA1,H} + S_{A1}} \cdot \frac{S_{NH4,H}}{K_{NH4,H} + S_{NH4}} \cdot \frac{S_{PO4}}{K_{PO4,H} + S_{PO4}} \cdot \frac{S_{ALK}}{K_{ALK,H} + S_{ALK}} \cdot X_H$
2	Anoxic growth on $S_F$ , denitrification $(S_{NO2})$	$ \frac{\mu_{H} \cdot \eta_{NO2,H} \cdot \frac{K_{O,H}}{K_{O,H} + S_{O}} \cdot \frac{S_{F}}{K_{SF,H} + S_{F}} \cdot \frac{S_{F}}{S_{F} + S_{A}} \cdot \frac{S_{NO2}}{K_{NO2,H} + S_{NO2}} \cdot \frac{S_{NH4,H}}{K_{NH4,H} + S_{NH4}} }{\frac{S_{PO4}}{K_{NO2,H} + S_{NO2}} \cdot \frac{S_{NH4,H}}{K_{NH4,H} + S_{NH4}} } $
3	$(S_{NO3})$	$\cdot \frac{S_{PO4}}{K_{PO4,H} + S_{PO4}} \cdot \frac{S_{ALK}}{K_{ALK,H} + S_{ALK}} \cdot X_H \cdot E_{Sat}$
4	S <sub>A</sub> , demunication	N N
5	Anoxic growth on $S_A$ , denitrification $(S_{NO3})$	$\begin{split} & \frac{S_{PO4}}{K_{PO4,H} + S_{PO4}} \cdot \frac{S_{AlK}}{K_{ALK,H} + S_{ALK}} \cdot X_H \cdot E_{Sat} \\ & \mu_H \cdot \eta_{NO3,H} \cdot \frac{K_{O,H}}{K_{O,H} + S_O} \cdot \frac{S_A}{K_{SA,H} + S_A} \cdot \frac{S_A}{S_F + S_A} \cdot \frac{S_{NO3}}{K_{NO3,H} + S_{NO3}} \cdot \frac{S_{NH4,H}}{K_{NH4,H} + S_{NH4}} \\ & \cdot \frac{S_{PO4}}{K_{PO4,H} + S_{PO4}} \cdot \frac{S_{ALK}}{K_{ALK,H} + S_{ALK}} \cdot X_H \cdot E_{Sat} \end{split}$
6	Anoxic growth on $S_{A1}$ , denitrification $(S_{NO2})$	$\frac{S_{PO4}}{K_{PO4,H} + S_{PO4}} \cdot \frac{S_{ALK}}{K_{ALK,H} + S_{ALK}} \cdot X_H \cdot E_{Sat}$
7	Anoxic growth on $S_{A1}$ , denitrification $(S_{NO3})$	$ \frac{\mu_{H1} \cdot \eta_{NO3,H1} \cdot \frac{K_{O,H}}{K_{O,H} + S_O} \cdot \frac{S_{A1}}{K_{SA1,H} + S_{A1}} \cdot \frac{S_{NO3}}{K_{NO3,H} + S_{NO3}} \cdot \frac{S_{NH4,H}}{K_{NH4,H} + S_{NH4}} \cdot \frac{S_{PO4}}{K_{PO4,H} + S_{PO4}} }{\frac{S_{ALK}}{K_{ALK,H} + S_{ALK}}} \cdot X_H \cdot E_{Sat} $
	Activ	vity of Phosphorus Accumulating Organisms (PAO) (X <sub>PAO</sub> )
8	$X_{PAO}$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$
9	Aerobic storage of $X_{pp}$ by PAO on	$q_{PP} \cdot \frac{S_O}{K_{O,PAO} + S_O} \cdot \frac{S_{PO4}}{K_{PO4,PAO} + S_{PO4}} \cdot \frac{S_{ALK}}{K_{ALK,PAO} + S_{ALK}} \cdot \frac{X_{PHA} / X_{PAO}}{K_{PHA} + X_{PHA} / X_{PAO}}$
10	$(S_{NO2})$	$ \frac{\cdot \frac{K_{MAX} - X_{PP} + X_{PAO}}{K_{IPP} + K_{MAX} - X_{PP} / X_{PAO}} \cdot X_{PAO} }{K_{PAO} \cdot \frac{K_{O,PAO}}{K_{O,PAO}} \cdot \frac{S_{NO2}}{K_{NO2,PAO}} \cdot \frac{S_{PO4}}{K_{PO4,PAO} + S_{PO4}} \cdot \frac{S_{ALK}}{K_{ALK,PAO} + S_{ALK}} } \cdot \frac{X_{PHA} / X_{PAO}}{K_{PHA} + X_{PHA} / X_{PAO}} \cdot \frac{K_{MAX} - X_{PP} / X_{PAO}}{K_{IPP} + K_{MAX} - X_{PP} / X_{PAO}} \cdot X_{PAO} $
11	Anoxic storage of $X_{pp}$ on $X_{PHA}$ , denitrification $(S_{NO3})$	$q_{PP} \cdot \eta_{NO3,PAO} \cdot \frac{K_{O,PAO}}{K_{O,PAO}} \cdot \frac{S_{NO3}}{K_{NO3,PAO} + S_{NO3}} \cdot \frac{S_{PO4}}{K_{PO4,PAO} + S_{PO4}} \cdot \frac{S_{ALK}}{K_{ALK,PAO} + S_{ALK}} \cdot \frac{X_{PHA}/X_{PAO}}{K_{PHA} + X_{PHA}/X_{PAO}} \cdot \frac{K_{MAX} - X_{PP}/X_{PAO}}{K_{MPP} + K_{MAX} - X_{PP}/X_{PAO}} \cdot X_{PAO}$

	Process	Process rate, $\rho_i$
		$S_O$ $S_A$ $S_{PO4}$ $S_{ALK}$
12	Aerobic storage of $X_{pp}$ by PAO on $S_A$	$K_{MAY} - X_{RR} / X_{RAQ}$
		$K_{IPP} + K_{MAX} - X_{PP} / X_{PAO}$
13	Anoxic storage of $X_{pp}$ on $S_A$ , denitrification $(S_{NO2})$	$ \frac{\cdot \frac{MAA}{K_{IPP} + K_{MAX} - X_{PP} / X_{PAO}}{K_{IPP} + K_{MAX} - X_{PP} / X_{PAO}} \cdot X_{PAO} }{ q_{PPSA} \cdot \eta_{NO2,PAOSA} \cdot \frac{K_{O,PAO}}{K_{O,PAO} + S_{O}} \cdot \frac{S_{A}}{K_{SA,PAO} + S_{A}} \cdot \frac{S_{NO2}}{K_{NO2,PAO} + S_{NO2}} \cdot \frac{S_{PO4}}{K_{PO4,PAO} + S_{PO4}} } \cdot \frac{S_{ALK}}{K_{ALK,PAO} + S_{ALK}} \cdot \frac{K_{MAX} - X_{PP} / X_{PAO}}{K_{IPP} + K_{MAX} - X_{PP} / X_{PAO}} \cdot X_{PAO} $
	Anoxic storage of	***
14	$X_{pp}$ on $S_A$ , denitrification $(S_{NO3})$	$ \frac{S_{ALK}}{K_{ALK,PAO} + S_{ALK}} \cdot \frac{K_{O,PAO} + S_{O}}{K_{IPP} + K_{MAX} - X_{PP} / X_{PAO}} \cdot X_{PAO} \cdot X_{PAO} $
15		$\cdot \frac{K_{\scriptscriptstyle MAX} - X_{\scriptscriptstyle PP} / X_{\scriptscriptstyle PAO}}{K_{\scriptscriptstyle IPP} + K_{\scriptscriptstyle MAX} - X_{\scriptscriptstyle PP} / X_{\scriptscriptstyle PAO}} \cdot X_{\scriptscriptstyle PAO}$
16	denitrification	$q_{PPSA1} \cdot \eta_{NO2,PAOSA1} \cdot \frac{K_{O,PAO}}{K_{O,PAO} + S_{O}} \cdot \frac{S_{A1}}{K_{SA1,PAO} + S_{A1}} \cdot \frac{S_{NO2}}{K_{NO2,PAO} + S_{NO2}} \cdot \frac{S_{PO4}}{K_{PO4,PAO} + S_{PO4}} \cdot \frac{S_{ALK}}{K_{PO4,PAO} + S_{ALK}} \cdot \frac{K_{MAX} - X_{PP} / X_{PAO}}{K_{ALK,PAO} + S_{ALK}} \cdot X_{PAO}$
17	Anoxic storage of $X_{pp}$ on $S_{A1}$ , denitrification $(S_{NO3})$	$ q_{PPSA1} \cdot \eta_{NO3,PAOSA1} \cdot \frac{K_{O,PAO}}{K_{O,PAO} + S_o} \cdot \frac{S_{A1}}{K_{SA1,PAO} + S_{A1}} \cdot \frac{S_{NO3}}{K_{NO3,PAO} + S_{NO3}} \cdot \frac{S_{PO4}}{K_{PO4,PAO} + S_{PO4}} \cdot \frac{S_{ALK}}{K_{ALK,PAO} + S_{ALK}} \cdot \frac{K_{MAX} - X_{PP} / X_{PAO}}{K_{IPP} + K_{MAX} - X_{PP} / X_{PAO}} \cdot X_{PAO} $
18	Aerobic growth of $X_{PAO}$ on $X_{PHA}$	$\cdot \frac{X_{\scriptscriptstyle PHA}/X_{\scriptscriptstyle PAO}}{K_{\scriptscriptstyle PHA}+X_{\scriptscriptstyle PHA}/X_{\scriptscriptstyle PAO}} \cdot X_{\scriptscriptstyle PAO}$
19	Anoxic growth of $X_{PAO}$ on $X_{PHA}$ , denitrification $(S_{NO2})$	$ \frac{\mu_{PAO} \cdot \eta_{NO2,PAO} \cdot \frac{K_{O,PAO}}{K_{O,PAO} + S_O} \cdot \frac{S_{NO2}}{K_{NO2,PAO} + S_{NO2}} \cdot \frac{S_{NH4}}{K_{NH4,PAO} + S_{NH4}} \cdot \frac{S_{PO4}}{K_{PO4,PAO} + S_{PO4}} }{\frac{S_{ALK}}{K_{ALK,PAO} + S_{ALK}} \cdot \frac{X_{PHA} / X_{PAO}}{K_{PHA} + X_{PHA} / X_{PAO}} \cdot X_{PAO} } \cdot X_{PAO} $
20	Anoxic growth of $X_{PAO}$ on $X_{PHA}$ , denitrification $(S_{NO3})$	$\frac{\mu_{PAO} \cdot \eta_{NO3,PAO}}{K_{O,PAO} \cdot K_{O,PAO} + S_O} \cdot \frac{S_{NO3}}{K_{NO3,PAO} + S_{NO3}} \cdot \frac{S_{NH4}}{K_{NH4,PAO} + S_{NH4}} \cdot \frac{S_{PO4}}{K_{PO4,PAO} + S_{PO4}} \cdot \frac{S_{ALK}}{K_{ALK,PAO} + S_{ALK}} \cdot \frac{X_{PHA} / X_{PAO}}{K_{PHA} / X_{PAO}} \cdot X_{PAO}$
21	Aerobic growth of $X_{PAO}$ on $S_A$	$\mu_{PAOSA} \cdot \frac{S_{O}}{K_{O,PAO} + S_{O}} \cdot \frac{S_{A}}{K_{SA,PAO} + S_{A}} \cdot \frac{S_{NH4}}{K_{NH4,PAO} + S_{NH4}} \cdot \frac{S_{PO4}}{K_{PO4,PAO} + S_{PO4}} \cdot \frac{S_{ALK}}{K_{ALK,PAO} + S_{ALK}} \cdot X_{PAO}$
22	Anoxic growth of $X_{PAO}$ on $S_{A}$ , denitrification $(S_{NO2})$	$ \frac{\mu_{PAOSA} \cdot \eta_{NO2, PAOSA} \cdot \frac{K_{O, PAO}}{K_{O, PAO} + S_{O}} \cdot \frac{S_{A}}{K_{SA, PAO} + S_{A}} \cdot \frac{S_{NO2}}{K_{NO2, PAO} + S_{NO2}} \cdot \frac{S_{NH4}}{K_{NH4, PAO} + S_{NH4}} } \cdot \frac{S_{PO4}}{K_{PO4, PAO} + S_{PO4}} \cdot \frac{S_{ALK}}{K_{ALK, PAO} + S_{ALK}} \cdot X_{PAO} $

	Process	Process rate, $\rho_i$					
23	Anoxic growth of $X_{PAO}$ on $S_{A,}$ denitrification $(S_{NO3})$	$\frac{\mu_{PAOSA} \cdot \eta_{NO3,PAOSA}}{K_{O,PAO} + S_{O}} \cdot \frac{K_{O,PAO}}{K_{SA,PAO} + S_{A}} \cdot \frac{S_{NO3}}{K_{NO3,PAO} + S_{NO3}} \cdot \frac{S_{NH4}}{K_{NH4,PAO} + S_{NH4}} \cdot \frac{S_{PO4}}{K_{PO4,PAO} + S_{PO4}} \cdot \frac{S_{ALK}}{K_{ALK,PAO} + S_{ALK}} \cdot X_{PAO}$					
24	Aerobic growth of $X_{PAO}$ on $S_{A,1}$	$\mu_{PAOSA1} \cdot \frac{S_O}{K_{O,PAO} + S_O} \cdot \frac{S_{A1}}{K_{SA1,PAO} + S_{A1}} \cdot \frac{S_{NH4}}{K_{NH4,PAO} + S_{NH4}} \cdot \frac{S_{PO4}}{K_{PO4,PAO} + S_{PO4}} \cdot \frac{S_{ALK}}{K_{ALK,PAO} + S_{ALK}} \cdot X_{PAO}$					
25		$ \frac{\mu_{\scriptscriptstyle PAOSA1} \cdot \eta_{\scriptscriptstyle NO2,PAOSA1} \cdot \frac{K_{\scriptscriptstyle O,PAO}}{K_{\scriptscriptstyle O,PAO} + S_{\scriptscriptstyle O}} \cdot \frac{S_{\scriptscriptstyle A1}}{K_{\scriptscriptstyle SA1,PAO} + S_{\scriptscriptstyle A1}} \cdot \frac{S_{\scriptscriptstyle NO2}}{K_{\scriptscriptstyle NO2,PAO} + S_{\scriptscriptstyle NO2}} \cdot \frac{S_{\scriptscriptstyle NH4}}{K_{\scriptscriptstyle NH4,PAO} + S_{\scriptscriptstyle NH4}} }{\frac{S_{\scriptscriptstyle PO4}}{K_{\scriptscriptstyle PO4,PAO} + S_{\scriptscriptstyle PO4}} \cdot \frac{S_{\scriptscriptstyle ALK}}{K_{\scriptscriptstyle ALK,PAO} + S_{\scriptscriptstyle ALK}} \cdot X_{\scriptscriptstyle PAO}} $					
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$							
Hydi	olysis processes						
27	(SNO2)	$k_{\rm h} \cdot \eta_{\rm NO2,HYD} \cdot \frac{K_{\rm O,H}}{K_{\rm O,H} + S_{\rm O}} \cdot \frac{S_{\rm NO2}}{K_{\rm NO2,HYD} + S_{\rm NO2}} \cdot \frac{K_{\rm S} / K_{\rm H}}{K_{\rm X} + X_{\rm S} / X_{\rm H}} \cdot X_{\rm H}$					
28	Anoxic hydrolysis of $X_S$ with nitrate $(S_{NO3})$	$k_{h} \cdot \eta_{\text{NO3,HYD}} \cdot \frac{K_{\text{O,H}}}{K_{\text{O,H}} + S_{\text{O}}} \cdot \frac{S_{\text{NO3}}}{K_{\text{NO3,HYD}} + S_{\text{NO3}}} \cdot \frac{X_{\text{S}}/X_{\text{H}}}{K_{\text{X}} + X_{\text{S}}/X_{\text{H}}} \cdot X_{\text{H}}$					
	Activity o	f the denitrification enzyme for the external carbon source ( $S_{A,l}$ )					
29		$\begin{split} k_{S,ENZ} \cdot & (\frac{S_{NO3}}{K_{NO3,ENZ} + S_{NO3}} \cdot \frac{S_{NO3}}{S_{NO2} + S_{NO3}} + \frac{S_{NO2}}{K_{NO2,ENZ} + S_{NO2}} \cdot \frac{S_{NO2}}{S_{NO2} + S_{NO3}}) \\ \cdot & \frac{K_{O,ENZ}}{K_{O,ENZ} + S_{O}} \cdot & (l - E_{Sat}) \end{split}$					
30	Enzyme decay	$\mathbf{k}_{\mathrm{D,ENZ}} \cdot \mathbf{E}_{\mathrm{Sat}}$					

## Table S3. List of sensitivity coefficients calculated for the adjusted parameters during the

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Symbol	Anaerob	Anaerobic				Aerobic		
Symbol	COD	PO <sub>4</sub> -P	COD	PO <sub>4</sub> -P	NO <sub>3</sub> -N	NO <sub>2</sub> -N	COD	PO <sub>4</sub> -P
Original kir	netic paramet	ers in ASM	2d					
Heterotroph	ic organisms	$(X_H)$ :						
$\mu_{\mathrm{H}}$	-0.046	-0.029	-0.371	-0.160	-1.062	-0.466	-0.304	0.407
	Accumulating	Organisms (	$X_{PAO}$ ):			•	•	
$q_{PHA}$	-0.498	0.510	-0.092	0.217	0.503	0.004	-0.045	0.125
$q_{PP}$	-0.001	0.004	-0.061	-0.336	-0.714	0.000	0.091	-1.464
K <sub>IPP</sub>	0.003	-0.026	0.017	0.280	0.220	0.002	0.016	0.539
K <sub>PHA</sub>	-0.003	-0.014	0.007	0.049	0.175	-0.004	-0.006	0.259
	f particulate	substrate (X	s):	•		•	•	•
K <sub>h</sub>	0.082	0.061	0.217	0.213	-0.882	-0.080	0.103	0.039

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Symbol	Anaerob	oic	Anoxic				Aerobic	
Symbol	COD	PO <sub>4</sub> -P	COD	PO <sub>4</sub> -P	NO <sub>3</sub> -N	NO <sub>2</sub> -N	COD	PO <sub>4</sub> -P
η <sub>fe</sub>	0.270	0.220	0.008	0.078	-0.033	-0.005	0.005	0.062
$K_X$	0.142	0.237	-0.062	-0.075	0.380	0.037	-0.034	-0.005
New kinetic p			ASM2d	•			•	
Heterotrophic					T.			
$\mu_{\mathrm{H}1}$	-0.025	0.016	-0.061	0.018	0.123	-0.118	-0.520	1.072
$\eta_{ m NO2,H}$	-0.001	0.000	-0.332	0.035	0.094	-0.462	0.000	0.001
$\eta_{ m NO3,H}$	-0.042	-0.030	-0.371	-0.158	-1.064	-0.002	0.012	-0.024
$\eta_{ m NO2,H1}$	-0.001	0.000	-0.053	0.009	0.000	-0.193	0.012	-0.015
$\eta_{ m NO3,H1}$	-0.001	0.000	-0.061	0.018	-0.111	0.008	0.011	-0.014
$K_{SA,H}$	0.001	0.001	0.014	0.020	0.080	0.025	0.012	-0.036
K <sub>SA1,H</sub>	-0.001	0.000	0.008	-0.002	0.012	0.013	0.012	-0.010
K <sub>NO2,H</sub>	0.001	0.000	0.000	0.001	-0.002	0.013	-0.011	0.014
K <sub>NO3,H</sub>	0.020	0.015	0.008	0.040	0.604	0.012	0.010	-0.010
Phosphorus A	ccumulating	Organisms (.	$X_{PAO}$ ):			•		
q <sub>PPSA</sub>	-0.001	0.008	-0.016	-0.058	-0.140	-0.001	-0.008	-0.263
q <sub>PPSA1</sub>	0.000	0.000	-0.006	-0.122	-0.019	0.008	0.013	-0.851
$\mu_{PAOSA}$	0.009	-0.001	-0.018	-0.068	-0.188	-0.001	-0.074	-0.286
μ <sub>PAOSA1</sub>	-0.013	0.008	-0.009	-0.135	-0.093	-0.093	-1.156	-0.434
η <sub>NO2,PAO</sub>	0.001	0.001	0.000	-0.227	0.000	0.109	-0.011	0.014
$\eta_{NO3,PAO}$	0.005	0.009	-0.044	-0.133	-0.481	0.000	0.001	0.002
η <sub>NO2,PAOSA</sub>	0.000	0.001	-0.090	-0.192	0.000	0.202	0.000	0.000
η <sub>NO3,PAOSA</sub>	-0.005	0.021	-0.053	-0.134	-0.463	0.000	-0.003	-0.006
η <sub>NO2,PAOSA1</sub>	0.000	0.001	-0.105	-0.198	0.000	0.113	0.000	0.000
η <sub>NO3,PAOSA1</sub>	-0.022	0.045	-0.092	-0.149	-0.144	0.008	-0.031	0.025
K <sub>SA,PAO</sub>	0.044	-0.035	0.027	-0.020	0.026	0.037	0.007	0.031
K <sub>SA1,PAO</sub>	0.002	-0.001	0.001	0.004	0.005	0.000	0.033	0.136
K <sub>NO2,PAO</sub>	-0.001	0.001	-0.124	0.191	0.000	0.000	0.002	-0.001
K <sub>NO3,PAO</sub>	0.001	-0.015	0.007	0.024	0.270	0.000	0.000	0.002
K <sub>O,PAO</sub>	-0.001	0.002	0.000	0.000	0.000	0.000	-0.022	0.373
K <sub>IPHA</sub>	0.000	0.000	-0.012	0.000	-0,013	0.000	0.000	0.610
K <sub>IOPHA</sub>	0.000	0.000	0.000	0.000	0.000	0.000	0.000	-0.525
K <sub>max1</sub>	0.000	0.000	0.006	0.049	0.053	0.000	-0,629	1.073
Hydrolysis of	particulate :	substrate (X		ı	l.		1 - )	
$\eta_{ ext{NO2,HYD}}$	0.001	0.001	0.192	0.018	-0.003	-0.075	0.000	0.000
$\eta_{NO3,HYD}$	0.018	0.015	0.029	0.093	-0.843	0.008	0.012	-0.005
K <sub>NO2,HYD</sub>	-0.001	-0.001	-0.005	-0.002	0.003	0.001	-0.012	0.016
K <sub>NO3,HYD</sub>	-0.014	-0.009	-0.012	-0.011	0.204	0.000	0.002	-0.008
Denitrification								
k <sub>s,enz</sub>	-	-	-0.032	-0.004	-0.082	-0.054	-	-
k <sub>D,ENZ</sub>	-	-	0.000	0.000	0.002	0.000	-	-
K <sub>NO2,ENZ</sub>	-	-	0.000	0.000	0.000	0.000	-	-
K <sub>NO3,ENZ</sub>	-	-	0.000	0.000	0.000	0.000	-	-
K <sub>O,ENZ</sub>	-	-	0.000	0.000	0.000	0.000	-	-
Original stoic			in ASM2d					
Heterotrophic	c organisms	$(X_H)$ :						
$\mathbf{Y}_{\mathbf{H}}$	-0.529	-0.411	0.728	-0.663	1.910	3.974	0.076	-0.993
Phosphorus A	1ccumulatin	g Organism	$s(X_{PAO})$ :				·	
$Y_{PO4}$	0.008	0.963	-0.009	0.820	-0.266	0.008	-0.016	1.895
New stoichion	metric para	meters in ex	panded AS	SM2d				
Heterotrophic								
$Y_{H1}$	0.001	0.000	0.065	-0.051	0.684	0.560	-0.002	0.002
Phosphorus A								
Y <sub>PAOSA</sub>	0.024	-0.101	0.046	-0.073	0.669	-0.001	0.011	-0.258

Symbol	Anaerobic		Anoxic		Aerobic			
	COD	PO <sub>4</sub> -P	COD	PO <sub>4</sub> -P	NO <sub>3</sub> -N	NO <sub>2</sub> -N	COD	PO <sub>4</sub> -P
$Y_{PAOSA1}$	0.002	-0.015	0.079	-0.060	0.878	-0.010	0.730	-0.198
Y <sub>SA1</sub>	0.000	0.000	-0.008	0.008	-0.085	0.000	-0.023	0.000

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**Table S4.** Calibration/validation procedure for the expanded ASM2d based on results of the batch tests, list of the adjusted parameters, and evaluation of simulation results against experimental data

Batch	Electron	External	Target Variable	Standar	d Error	$\mathbb{R}^2$	Adjusted					
Test	Acceptor	Carbon	Target Variable	Relative	Absolute	K	parameters					
			Mode	l calibratio	n							
	Γ	I		Step 3.1								
			OUR	0.054	0.130	0.838	$ Y_{\mathrm{H}}$					
2	DO	naatata	COD	0.003	0.172	0.906						
phase	БО	acetate	PO <sub>4</sub> -P	0.108	0.132	0.967	$K_{\text{IPP}}, K_{\text{PHA}}, \\ K_{\text{IO,PHA}}, \\ K_{\text{IPHA}}, K_{\text{max1}}$					
	l	II.	S	Step 3.2	l		min man					
			OUR	0.153	0.163	0.925	Y <sub>H1</sub>					
2 phase	DO	ethanol	COD	0.147	0.202	0.934	$\mu_{\mathrm{H1},}$					
phase			PO <sub>4</sub> -P	0.199	0.243	0.987	$q_{\mathrm{PPSA1}}$					
	T	T		Step 3.3								
2	NO <sub>3</sub> -N	no external	NO <sub>3</sub> -N	0.007	0.022	0.993	$\eta_{\text{NO3,H}}$					
phase	NO <sub>2</sub> -N	carbon	NO <sub>2</sub> -N	0.015	0.032	0.965	$\eta_{\text{NO2},\text{H}}$					
Step 3.4												
2	NO <sub>3</sub> -N	ethanol	NO <sub>3</sub> -N	0.035	0.088	0.997	$\eta_{NO3,H1}$					
phase	NO <sub>2</sub> -N	Ctilatioi	NO <sub>2</sub> -N	0.012	0.049	0.967	$\eta_{ m NO2,H1}$					
		M	odel validation wi	th other tes	sts							
			Step 5									
1		acetate	OUR	0.104	0.146	0.925						
1 phase	DO		COD	0.056	0.086	0.972						
pinase			PO <sub>4</sub> -P	0.004	0.038	0.995						
1			OUR	0.0029	0.116	0.936						
1 phase	DO	ethanol	COD	0.0156	0.036	0.984						
phase			PO <sub>4</sub> -P	0.120	0.315	0.978						
2	NO N	no	COD	0.025	0.096	0.952						
phase	NO <sub>3</sub> -N	external carbon	$PO_4$ -P	0.142	0.155	0.972						
2	NO N	no	COD	0.026	0.113	0.958						
phase	NO <sub>2</sub> -N	external carbon	PO <sub>4</sub> -P	0.222	0.255	0.897						
			COD	0.011	0.092	0.848						
1	NO N	nantata	NO <sub>2</sub> -N	0.047	0.085	0.888						
phase	NO <sub>3</sub> -N	acetate	NO <sub>3</sub> -N	0.178	0.304	0.792						
			PO <sub>4</sub> -P	0.228	0.245	0.810						

	]			0.4.4	0.055							
	acetate	COD	0.142	0.142	0.965							
NO <sub>2</sub> -N		NO <sub>3</sub> -N	0.087	0.090	0.988							
		PO <sub>4</sub> -P	0.089	0.013	0.942							
NO N	athonal	COD	0.036	0.091	0.966							
NO <sub>3</sub> -N	ethanoi	PO <sub>4</sub> -P	0.007	0.082	0.952							
NO N	athonal	COD	0.0125	0.109	0.972							
1102-11	etilalioi	PO <sub>4</sub> -P	0.035	0.079	0.957							
		COD	0.071	0.144	0.895							
NO <sub>3</sub> -N	acetate	NO <sub>3</sub> -N	0.279	0.279	0.883							
		PO <sub>4</sub> -P	0.061	0.092	0.922							
		COD	0.052	0.006	0.960							
$NO_2$ -N	acetate	NO <sub>2</sub> -N	0.024	0.097	0.892							
		PO <sub>4</sub> -P	0.091	0.136	0.862							
		COD	0.063	0.068	0.862							
NO <sub>3</sub> -N	ethanol	NO <sub>3</sub> -N	0.062	0.064	0.884							
		PO <sub>4</sub> -P	0.011	0.083	0.956							
		COD	0.091	0.112	0.873							
NO <sub>2</sub> -N	ethanol	NO <sub>2</sub> -N	0.056	0.098	0.948							
									PO <sub>4</sub> -P	0.029	0.064	0.943
	NO <sub>3</sub> -N NO <sub>2</sub> -N NO <sub>3</sub> -N NO <sub>3</sub> -N	NO <sub>3</sub> -N ethanol  NO <sub>2</sub> -N ethanol  NO <sub>3</sub> -N acetate  NO <sub>2</sub> -N acetate  NO <sub>3</sub> -N ethanol		$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$							

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### Table S5. Kinetic and stoichiometric parameters adjusted in the ASM2d and new parameters

# in the expanded ASM2d

				Calibrated value							
Symbol	Definition	Unit	Defaulted value	Swiniarski et al. (2012)	This study	95% confidence intervals					
	Kinetic parameters in the original ASM2d										
	Heterotrophic organisms $(X_H)$ :										
$\mu_{\mathrm{H}}$	Maximum growth rate of $X_H$ on $S_A$ and $S_F$	1/d	6.00								
	Phospho	rus Accumul	lating Organisn	$ns(X_{PAO})$ :							
$q_{\mathrm{PHA}}$	Rate constant for storage of $X_{PHA}$	1/d	3.00	6.00	6.00						
$q_{PP}$	Rate constant for storage of $X_{PP}$	1/d	1.50	4.50	4.50						
K <sub>IPP</sub>	Inhibition coefficient for X <sub>PP</sub> storage	g P/g COD	0.02	0.10	0.13	±0.114					
$K_{PHA}$	$K_{PHA}$ Saturation coefficient for $X_{PHA}$		0.01	0.15	0.10	±0.029					
		lysis of parti	culate substra	te (X <sub>S</sub> ):							
$k_h$	Hydrolysis rate constant	1/d	3.00	2.50	2.50						
$\eta_{\text{fe}}$	Anaerobic hydrolysis reduction factor	-	0.40	0.10	0.10						
K <sub>X</sub>	K <sub>X</sub> Saturation coefficient for particulate COD		0.10	0.20	0.20						
	New kinetic	parameter	s in the expan	ded ASM2d							
	Heterotrophic organisms $(X_H)$ :										
$\mu_{\mathrm{H}1}$	Maximum growth rate of $X_H$ on $S_{A,1}$	1/d		0.40	0.26	±0.042					

	Calibrated value									
Symbol	Definition	Unit	Defaulted value	Swiniarski et al. (2012)	This study	95% confidence intervals				
$\eta_{\text{NO2,H}}$	Reduction factor for denitrification ( $S_{NO2}$ )	-			0.32	±0.054				
$\eta_{\text{NO3,H}}$	Reduction factor for denitrification ( $S_{NO3}$ )	-	0.80	0.80	0.35	±0.138				
$\eta_{NO2,H1}$	Reduction factor on $S_{A,1}$ for denitrification $(S_{NO2})$	-			0.06	±0.028				
$\eta_{NO3,H1}$	Reduction factor on $S_{A,1}$ for denitrification $(S_{NO3})$	-		0.80	0.21	±0.029				
K <sub>SA1,H</sub>	Saturation coefficient for growth of $X_H$ on $S_{A,1}$	COD/m <sup>3</sup>		4.00	4.00					
K <sub>NO2,H</sub>	Saturation coefficient for growth of X <sub>H</sub> (S <sub>NO2</sub> )	$g\ N/m^3$			0.50					
		rus Accumul	lating Organism	ns (X <sub>PAO</sub> ):						
q <sub>PPSA</sub>	Rate constant for storage of $X_{PP}$ on $S_A$	1/d		1.00	1.00					
<b>Q</b> PPSA1	Rate constant for storage of $X_{PP}$ on $S_{A,1}$ Maximum growth rate of	1/d		4.50	5.00	±1.115				
$\mu_{PAOSA}$	$X_{PAO}$ on $S_A$	1/d		1.00	1.00					
μ <sub>PAOSA1</sub>	Maximum growth rate of $X_{PAO}$ on $S_{A,1}$	1/d		1.00	1.00					
η <sub>NO2,PAOSA</sub>	Reduction factor for anoxic activity on $S_A$ ( $S_{NO2}$ )	-			0.00					
$\eta_{NO3,PAOSA}$	Reduction factor for anoxic activity on $S_A$ ( $S_{NO3}$ )	-		0.60	0.60					
$\eta_{NO2,PAOSA1}$	Reduction factor for anoxic activity on $S_{A,1}$ ( $S_{NO2}$ )	-			0.00					
$\eta_{NO3,PAOSA1}$	Reduction factor for anoxic activity on $S_{A,1}$ ( $S_{NO3}$ )	-		0.60	0.60					
K <sub>SA,PAO</sub>	Saturation coefficient for storage of $X_{PP}$ and $X_{PAO}$ growth on $S_A$	COD/m³		4.00	4.00					
K <sub>SA1,PAO</sub>	Saturation coefficient for storage of $X_{PP}$ and $X_{PAO}$ growth on $S_{A,1}$	COD/m <sup>3</sup>		4.00	4.00					
K <sub>NO2,PAO</sub>	Saturation coefficient for storage of $X_{PP}$ and $X_{PAO}$ growth $(S_{NO2})$	g N/m³			3.00					
K <sub>O,PAO</sub>	Saturation coefficient for oxygen for storage of $X_{pp}$ and $X_{PAO}$ growth	$g O_2/m^3$		0.20	0.20					
K <sub>IOPHA</sub>	Inhibition coefficient for $X_{PHA}$	$g O_2/m^3$			3.60	±1.135				
K <sub>MAX1</sub>	Maximum ratio of X <sub>PHA</sub> /X <sub>PAO</sub>	g COD/g COD			0.34	±0.126				
K <sub>IPHA</sub>	X <sub>PHA</sub> storage inhibition coefficient	g COD/g COD			0.02	±0.138				
	Hydrolysis of particulate substrate $(X_S)$ :									
$\eta_{\text{NO2,HYD}}$	Anoxic hydrolysis reduction factor (S <sub>NO2</sub> )	-			0.6					
K <sub>NO2,HYD</sub>	Saturation coefficient for hydrolysis (S <sub>NO2</sub> )	g N/m <sup>3</sup>			0.5					
	Countle agia mate	Denitrifica	tion enzymes							
k <sub>S,ENZ</sub>	Synthesis rate constant for denitrification enzymes	1/d			30.00					

				Ca	Calibrated value		
Symbol	Definition	Unit	Defaulted value	Swiniarski et al. (2012)	This study	95% confidence intervals	
$k_{D,ENZ}$	Decay rate constant for denitrification enzymes	1/d			4.00		
K <sub>NO2,ENZ</sub>	Nitrite half saturation coefficient for denitrification enzymes	g N/m³			0.50		
K <sub>NO3,ENZ</sub>	Nitrate half saturation coefficient for denitrification enzymes	g N/m <sup>3</sup>			0.50		
K <sub>O,ENZ</sub>	Oxygen half saturation coefficient for denitrification enzymes	g O <sub>2</sub> /m <sup>3</sup>			0.10		
	Stoichiomet	ric paramet	ers in the orig	ginal ASM2d			
	Не		organisms (X	́н):			
$Y_{H}$	Yield coefficient for S <sub>A</sub>	g COD/g COD	0.63	0.67	0.82	±0.016	
		us Accumul	ating Organis	$ms(X_{PAO})$ :			
Y <sub>PO4</sub>	$X_{PP}$ requirement per $X_{PHA}$ stored ( $S_{PO4}$ release)	g P/g COD	0.40	0.34	0.40		
	New stoichiome	tric param	eters in the ex	panded ASM2d			
		eterotrophic	organisms (X	( <sub>H</sub> ):			
$Y_{H1}$	Yield coefficient of $X_H$ for $S_{A,1}$	g COD/g COD		0.70	0.43	±0.049	
	Phosphor	rus Accumul	lating Organisn	$ns(X_{PAO})$ :			
Y <sub>PAOSA</sub>	$ \begin{array}{c} \text{Yield coefficient of $X_{PAO}$ for} \\ S_A \end{array} $	g COD/g COD		0.625	0.625		
Y <sub>PAOSA1</sub>	$S_A$ Yield coefficient of $X_{PAO}$ for $S_{A,1}$	g COD/g COD		0.625	0.625		
Y <sub>SA1</sub>	$S_{A,1}$ requirement for storage of $X_{PP}$	g COD/g P		0.20	0.20		

# 116 **Table S6.** Approximate correlation matrix of parameter estimates using experimental data for

# step 3.1 of model calibration

	Parameter	Unit			Correlation matrix				
	r ar ameter	Omt	$Y_{H}$	K <sub>MAX1</sub>	$K_{IPP}$	$K_{IPHA}$	$K_{PHA}$	$K_{IOPHA}$	
	$Y_{H}$	g COD/ g COD	1.00						
	$K_{MAX1}$	g P/ g COD	0.05	1.00					
3.1	$K_{IPP}$	g P/g COD	-0.70	0.11	1.00				
Step	$K_{IPHA}$	g P/g COD	0.12	0.98	0.08	1.00			
	$K_{PHA}$	g COD/ g COD	-0.58	-0.10	0.13	-0.16	1.00		
	$K_{IOPHA}$	$g O_2/m^3$	0.34	-0.55	-0.55	-0.54	-0.07	1.00	

Table S7. Approximate correlation matrix of parameter estimates using experimental data for
 steps 3.2-3.4 of model calibration

			Correlation matrix								
	Parameter	Unit	$Y_{\rm H1}$	$\mu_{\rm H1}$	$q_{PP,SA1} \qquad \eta_{NO3,H}$		$\eta_{\text{NO2,H}}$	$\eta_{\text{NO3,H1}}$	$\eta_{\text{NO2,H1}}$		
3.2	$Y_{\rm H1}$	g COD/ g COD	1.00								
Step 3.	$\mu_{\rm H1}$	1/d	0.96	1.00							
S	qpp,sa1	1/d	-0.02	0.01	1.00						
Step 3.3	$\eta_{\text{NO3,H}}$	-				1					
St.	$\eta_{\text{NO2,H}}$	-				-0.51	1				
Step 3.4	$\eta_{\text{NO3,H1}}$	-						1			
St.	$\eta_{\rm NO2,H1}$	-						-0.62	1		