

SUPPORTING INFORMATION

Generation of recommendable values for the surface tension of water using a non-parametric regression

Jaroslav Pátek*, Monika Součková, Jaroslav Klomfar

Institute of Thermomechanics of the Czech Academy of Sciences, v.v.i., Dolejškova 1402/5,
CZ 182 00 Prague 8, Czech Republic

*Corresponding author. +420 266053153. Fax: +420 286584695. E-mail: patek@it.cas.cz (J. Pátek).

Table S1. Literature sources of experimental data on the gas-liquid surface tension-temperature relation for water with year of publication, method, number of data in the data set, and the quantitative estimate of the data accuracy $u(\gamma)$ given by their author. The literature sources are ordered according to year of publication

author(s)	year	temp. range T/K	methods	no. data	$u(\gamma)$ $mN \cdot m^{-1}$
Buliginsky [1]	1868	288.15	capillary rise	1	—
Rodenbeck [2]	1880	290.65	capillary rise	1	—
Volkmann [3]	1882	288.65–292.15	capillary rise	2	—
Traube [4]	1885	288.15	capillary rise	1	—
Ramsay and Shields [5]	1893	273.15–413.15	capillary rise	15	—
Humphreys and Mohler [6]	1895	264.85–291.65	capillary rise	41	—
Volkmann [7]	1895	273.15–313.15	capillary rise	15	—
Sentis [8]	1897	286.65–298.25	capillary rise	2	—
Sohet [9]	1898	289.15–423.65	capillary rise	6	—
Grabowsky [10]	1904	283.15–303.15	capillary rise	2	—
Hartley et al. [11]	1908	273.15–298.15	capillary rise	2	—
Grunmach [12]	1909	292.25	capillary waves	1	—
Drucker and Moles [13]	1910	291.15	capillary rise	1	—
Magini [14]	1911	291.15	capillary rise	1	—
Morgan and McAfee [15]	1911	273.15–350.15	drop weight	20	—
Grunmach [16]	1912	293.15	capillary waves	1	—
Morgan and Neidle [17]	1913	303.15	drop weight	1	—
Ferguson [18]	1914	288.95–289.15	maximum bubble pressure	3	—
Richards and Coombs [19]	1915	293.15	capillary rise	1	—
Sentis [20]	1915	273.15–353.15	capillary rise	4	—
Morgan and Davis [21]	1916	273.15–323.15	drop weight	3	—
Morgan and Egloff [22]	1916	273.15–338.15	drop weight	10	—
Harkins et al. [23]	1917	283.15–313.15	drop weight	18	—
Jaeger [24]	1917	273.55–373.05	capillary rise	11	—
Morgan and Scarlett [25]	1917	273.15–318.15	drop weight	3	—
Harkins and Brown [26]	1919	293.15	capillary rise	1	—
Richards and Palitzsch [27]	1919	293.15	drop weight	1	—
Harkins et al. [28]	1920	293.15	capillary rise	1	—
Maass and Hatcher [29]	1920	273.15–291.15	capillary rise	2	—
Stocker [30]	1920	290.15–296.95	capillary rise	15	0.14
Richards and Carver [31]	1921	293.15	capillary rise	2	—
Sugden [32]	1921	293.15	capillary rise	1	0.22
Bircumshaw [33]	1922	298.15	drop weight	1	—
Ferguson [34]	1922	288.15	pull on sphere	1	0.023
Sugden [35]	1922	293.15	maximum bubble pressure	1	—
Butler [36]	1923	283.15–343.15	capillary rise	7	—
Ferguson [37]	1923	290.35–293.95	capillary rise	5	—
Richards et al. [38]	1924	273.15–333.80	capillary rise	3	—
Ferguson and Vogel [39]	1926	287.15–297.05	Wilhelmy plate	3	—
Harkins and Gilbert [40]	1926	298.15	pendant drop	1	—
Rehbinder [41]	1926	283.15–373.15	maximum bubble pressure	5	—
Moser [42]	1927	273.15–373.10	du Noüy ring	79	—
Warren [43]	1927	273.15–363.15	capillary rise	19	0.05
Harkins and Jordan [44]	1930	298.15	du Noüy ring	16	0.2
Rehbinder and Taubmann [45]	1930	283.15–363.15	maximum bubble pressure	6	—
Butler and Wightman [46]	1932	298.15	maximum bubble pressure	1	0.05

Table S1. – continued

author(s)	year	temp. range <i>T/K</i>	methods	no. data	<i>u(γ)</i> mN·m ⁻¹
Butler and Lees [47]	1932	298.15	capillary rise	1	0.10
Ernst et al. [48]	1932	298.15	capillary rise	1	–
Ferguson and Kennedy [49]	1932	288.15–298.15	maximum bubble pressure	2	–
Speakman [50]	1933	292.15	capillary rise	1	–
Butler et al. [51]	1934	298.15	capillary rise	1	0.05
Belton [52]	1935	298.15	maximum bubble pressure	1	–
Sabinina and Terpugow [53]	1935	283.15–323.15	capillary rise	5	–
Trieschmann [54]	1935	295.15	maximum bubble pressure	1	0.15
Ernst et al. [55]	1936	298.15	capillary rise	1	–
Valentiner and Hohls [56]	1938	293.15–323.15	du Noüy ring	4	–
Cockett and Ferguson [57]	1939	290.45–346.25	capillary rise	29	–
Neros and Eversole [58]	1941	288.15–323.15	capillary rise	2	–
Smith and Sorg [59]	1941	298.15	pendant drop	1	–
Addison [60]	1943	286.15–333.15	vibrating-jet	4	–
Glagoleva [61]	1947	293.15–303.15	maximum bubble pressure	3	–
Grant et al. [62]	1948	273.15–373.15	differential bubbling	10	–
Urazovskij and Tchetaev [63]	1949	299.15–323.15	capillary rise	9	–
Douglas [64]	1950	298.15	pendant drop	2	0.1
Vierk and Fredenhagen [65]	1950	293.15	maximum bubble pressure	1	–
Voljak [66]	1950	273.15–643.15	capillary rise	40	–
Hacker [67]	1951	250.65–300.65	capillary rise	50	0.06
Teitelbaum et al. [68]	1951	273.15–333.15	capillary rise	11	–
Fox and Chrisman [69]	1952	293.15	du Noüy ring	1	0.035
Voronkov [70]	1952	293.15	maximum bubble pressure	1	0.05
Heiks et al. [71]	1954	374.95–497.55	capillary rise	10	–
Sutherland [72]	1954	292.25–294.55	oscillating jet	11	0.3
		293.15	du Noüy ring	1	0.1
Kovalenko et al. [73]	1956	298.15–313.15	maximum bubble pressure	2	–
Hoffmann et al. [74]	1957	288.15–293.15	du Noüy ring	2	0.2
Howard and McAllister [75]	1957	288.15–343.35	capillary rise	7	0.3
Murphy et al. [76]	1957	298.15	du Noüy ring	6	0.05
Hirata and Kanai [77]	1958	293.15	capillary rise	1	–
Krishna and Venkateswarlu [78]	1958	293.15	drop weight	1	–
Ling and Van Winkle [79]	1958	303.15–373.15	capillary rise	2	0.35
Ling and Van Winkle [80]	1958	303.15–373.15	capillary rise	2	0.5
Uchida and Matsumoto [81]	1958	303.15–363.15	capillary rise	7	–
Crawford and Van Winkle [82]	1959	373.15	capillary rise	1	0.5
Padday	1960	293.15	Wilhelmy plate	2	0.03
Padday	1960	293.15	du Noüy ring	1	0.1
Hasaba et al. [84]	1961	273.15–353.15	capillary rise	10	–
Slowinski and Masterton [85]	1961	300.15	Wilhelmy plate	1	0.05
Bordi and Vannel [86]	1962	284.15–317.65	capillary rise	16	–
Catchpole and Ellis [87]	1963	373.15	capillary rise	1	0.1
Nagy [88]	1963	298.15	drop weight	1	–
Drost-Hansen [89]	1965	298.15	capillary rise	1	–
Kowalska et al. [90]	1965	293.15	capillary rise	1	0.3
Murto et al. [91]	1967	293.15–303.15	maximum bubble pressure	3	–
Padday [92]	1968	293.15–313.15	Wilhelmy plate	3	–

Table S1. – continued

author(s)	year	temp. range T/K	methods	no. data	$u(\gamma)$ mN·m ⁻¹
Gittens [93]	1969	278.17–318.15	capillary rise	86	—
		278.17–318.15	drop volume	118	—
Kawanishi et al. [94]	1970	298.35	Wilhelmy plate	1	—
Konobeev and Lyapin [95]	1970	293.15–333.15	capillary rise	2	—
Pike and Bonnet [96]	1970	303.15	Wilhelmy plate	1	—
Wright and Akhtar [97]	1970	303.15	capillary rise	1	0.08
Nakanishi et al. [98]	1971	303.15	capillary rise	1	0.3
Bonnet and Pike [99]	1972	293.15–313.15	Wilhelmy plate	3	0.25
Cini et al. [100]	1972	276.57–322.99	du Noüy ring	66	—
		283.69–320.16	Wilhelmy plate	13	—
Granzhan and Kirillova [101]	1972	298.15–353.15	maximum bubble pressure	4	—
Wagenbreth [102]	1972	273.15–313.15	capillary rise	5	—
Vargaftik et al. [103]	1973	273.55–643.54	capillary rise	147	—
Padday et al. [104]	1975	292.87–296.92	rod	20	0.3
Taylor and Mingins [105]	1975	293.15	Wilhelmy plate	1	0.04
Thakur and Hickman [106]	1975	372.50–372.96	Wilhelmy plate	9	—
Ramakrishnan et al. [107]	1976	293.15	maximum bubble pressure	2	0.07
Toryanik and Pogrebnyak [108]	1976	298.15–343.15	maximum bubble pressure	6	0.35
Artamonov et al. [109]	1977	298.15	drop weight	1	—
Tornberg [110]	1977	293.15–323.15	drop volume	5	0.33
Padday [111]	1979	298.90	cone	1	—
Patterson and Ross [112]	1979	293.15	pendant drop	1	—
Ross and Patterson [113]	1979	293.15	pendant drop	1	—
Furlong and Hartland [114]	1980	293.25–294.65	vertical cylinder	8	0.1
Won et al. [115]	1981	298.15	du Noüy ring	1	—
Pallas and Pethica [116]	1983	298.15	capillary rise	1	0.1
		298.15	Wilhelmy plate	1	0.02
Rotenberg et al. [117]	1983	293.15	sessile drop	1	—
		293.15	pendant drop	1	—
Drummond et al. [118]	1985	298.15	du Noüy ring	1	—
Cheong and Carr [119]	1987	298.15	du Noüy ring	1	1.4
Gaonkar and Neuman [120]	1987	293.15–298.15	Wilhelmy plate	2	0.04
Owens et al. [121]	1987	293.15	Wilhelmy plate	1	0.05
Braslau et al. [122]	1988	293.15	capillary waves	1	—
Jennings and Pallas [123]	1988	298.15	drop shape	1	0.3
Kalbassi and Biddulph [124]	1988	373.15	bubble detachment pressures	1	0.3
Ramkumar and Kudchadker [125]	1989	303.15–343.15	capillary rise	5	0.14
Cheng et al. [126]	1990	294.15	Wilhelmy plate	1	0.15
		294.15	ADSA-P	1	0.06
Floriano and Angell [127]	1990	245.95–333.15	capillary rise	63	—
Pallas and Harrison [128]	1990	293.15–298.15	drop shape	2	0.036
Wallenberger and Lyzenga [129]	1990	297.65	microwaves	1	1.6
Hansen and Rødsrud [130]	1991	294.15	pendant drop	2	0.2
Noordmans and Busscher [131]	1991	295.15–297.15	ADSA-P	10	1.3
Holcomb and Zollweg [132]	1992	293.15–303.15	capillary rise	3	0.9
		293.15–303.15	differential bubble pressure	3	0.1
Hansen [133]	1993	298.15	pendant drop	2	0.1
		298.15	sessile drop	3	0.1
		298.15	sessile bubbles	3	0.1

Table S1. – continued

author(s)	year	temp. range <i>T/K</i>	methods	no. data	$u(\gamma)$ mN·m ⁻¹
Stückrad et al. [134]	1993	293.15	du Noüy ring	1	0.1
		293.15	oscillating drop	1	—
Gunde et al. [135]	1995	293.15	pendant drop	20	—
Kinart et al. [136]	1995	298.15	drop weight	1	0.07
Krotov et al. [137]	1995	293.15	touching drops	1	—
Rao and Enhorning [138]	1995	293.15	maximum bubble pressure	1	—
Trinh and Ohsaka [139]	1995	252.24–296.84	driven shape oscillation frequency + drop volume	15	0.7
Vázquez et al. [140]	1995	293.15–323.15	Wilhelmy plate	7	0.25
Semmler et al. [141]	1996	298.15	LASDA	1	0.06
Slauenwhite and Johnson [142]	1996	293.15	spinning single bubbles	1	—
Zhang et al. [143]	1996	293.15	sphere	2	—
Álvarez et al. [144]	1998	298.15–323.15	Wilhelmy plate	6	0.02
Hawrylak et al. [145]	1998	298.15	du Noüy ring	1	0.2
Tsierkezos and Molinou [146]	1998	283.15–323.15	du Noüy ring	5	0.1
Glinski et al. [147]	1999	293.15	Wilhelmy plate	1	—
Horozov and Arnaudov [148]	1999	297.15	fast formed drop	1	—
Nath [149]	1999	298.15	n/a	1	—
Gente and La Mesa [150]	2000	298.15	du Noüy ring	1	0.2
Lee et al. [151]	2000	298.15–323.15	capillary rise	6	—
Feenstra et al. [152]	2001	273.85–296.65	maximum bubble pressure	4	—
Glinski et al. [153]	2001	293.15	Wilhelmy plate	1	0.2
Gunde et al. [154]	2001	293.15	pendant drop	2	—
Maham and Mather [155]	2001	298.15–328.15	capillary rise	4	—
Yuan and Herold [156]	2001	296.65	drop weight	1	0.18
Kalies et al. [157]	2002	293.15	Wilhelmy plate	1	0.5
Kulankara and Herold [158]	2002	295.45	drop weight	1	2.0
Kumar et al. [159]	2003	298.15–318.15	Wilhelmy plate	3	0.03
Santos et al. [160]	2003	303.15	du Noüy ring	1	0.01
Alakoç et al. [161]	2004	298.15	oscillating capillary jet	6	3.5
Gómez-Díaz and Navaza [162]	2004	293.15–323.15	Wilhelmy plate	5	0.03
Hadrlová et al. [163]	2004	298.15	du Noüy ring	1	—
Hyvärinen et al. [164]	2004	297.20–298.15	Wilhelmy plate	2	1.4
Kijevcanin et al. [165]	2004	303.15	du Noüy ring	1	0.01
Li et al. [166]	2004	293.15	drop volume	1	0.04
Maham et al. [167]	2004	298.15–328.15	Wilhelmy plate	4	0.01
Sönmez and Cebeci [168]	2004	298.15	drop weight	1	—
Belda et al. [169]	2005	298.15	stalagmometer	1	—
Fujii [170]	2005	296.15–301.15	oscillating drop	9	3.5
Ali et al. [171]	2006	283.15–303.15	pendant drop	5	—
Dixmier [172]	2006	283.15–373.15	capillary pressure difference	5	—
Granados et al. [173]	2006	298.15	drop volume	1	0.03
Liu et al. [174]	2006	298.15	pendant drop	1	—
Richter and Vollhardt [175]	2006	295.15	Wilhelmy plate	1	—
Romero and Paéz [176]	2006	298.15	capillary rise	1	0.3
Enders et al. [177]	2007	288.15–328.15	pendant drop	5	0.2
Kilaru et al. [178]	2007	293.15	du Noüy ring	1	0.05
Lisa and Lisa [179]	2007	290.15–310.15	drop volume	3	0.05
Markarian and Terzyan [180]	2007	298.15–328.15	maximum bubble pressure	5	0.1

Table S1. – continued

author(s)	year	temp. range <i>T/K</i>	methods	no. data	$u(\gamma)$ mN·m ⁻¹
Romero et al. [181]	2007	283.15–308.15	capillary rise	6	0.3
Tuckermann [182]	2007	293.15	du Noüy ring	1	–
Fujimura and Iino [183]	2008	298.15	surface-wave resonance	1	0.14
García-Abuín et al. [184]	2008	293.15–323.15	Wilhelmy plate	5	0.07
Johnson et al. [185]	2008	303.15	du Noüy ring	1	0.01
Liu et al. [186]	2008	298.15	pendant drop	1	–
Williams et al. [187]	2008	278.15–298.15	du Noüy ring	5	3.0
Batchelor et al. [188]	2009	293.15	Wilhelmy plate	1	0.1
Rilo et al. [189]	2009	298.15	drop volume	1	–
Romero et al. [190]	2009	288.15–308.15	drop volume	5	0.01
Bermúdez-Salguero et al. [191]	2010	298.15	pendant drop	1	0.04
Mohsen-Nia et al. [192]	2010	293.15–323.15	maximum bubble pressure	6	0.08
Russo and Hoffmann [193]	2010	298.10–300.20	maximum bubble pressure	20	0.4
Sadeghi et al. [194]	2010	303.15–343.15	pendant drop	15	0.17
Zhang et al. [195]	2010	298.15	du Noüy ring	1	0.1
Rafati et al. [196]	2011	283.15–308.15	du Noüy ring	3	0.2
Russo and Hoffmann [197]	2011	296.80–321.30	maximum bubble pressure	8	0.4
Wang et al. [198]	2011	298.15	Wilhelmy plate	1	0.15
Han et al. [199]	2012	303.15–333.15	pendant drop	4	0.04
Khattab et al. [200]	2012	293.15–323.15	drop number	7	0.1
Quijada-Maldonado et al. [201]	2012	298.15	du Noüy ring	1	0.12
Begum et al. [202]	2013	303.15	Wilhelmy plate	1	0.51
Fu et al. [203]	2013	298.20–308.20	Wilhelmy plate	2	0.1
Hrubý et al. [204]	2014	248.27–273.10	capillary rise	26	0.15
Lago et al. [205]	2014	298.15	Wilhelmy plate	1	0.1
Ren et al. [206]	2014	298.15	Wilhelmy plate	1	0.1
Singh et al. [207]	2014	298.15	drop volume	1	0.2
Ameta and Singh [208]	2015	298.15–308.15	pendant drop	15	0.01
Cárdenas et al. [209]	2015	298.15	maximum bubble pressure	1	0.01
Chang et al. [210]	2015	298.15–328.15	Wilhelmy plate	4	0.2
Vinš et al. [211]	2015	247.23–303.36	capillary rise	64	0.15
This work	2015	272.64–343.48	Wilhelmy plate	33	0.03

Table S2. The present experimental air-liquid surface tension data for water at the pressure of 0.1 MPa. Mean values \bar{T} and $\bar{\gamma}$ of the experimental readings of the temperature and surface tension, respectively, and the number of the averaged data readings ^a

\bar{T} /K	$\bar{\gamma}$ /mN·m ⁻¹	Number of data
272.64	75.78	20
273.04	75.71	75
275.06	75.41	54
276.72	75.17	24
280.09	74.69	45
280.34	74.65	47
283.27	74.23	78
286.54	73.74	50
288.33	73.46	53
290.57	73.12	64
293.04	72.74	85
294.13	72.58	52
295.71	72.33	78
297.40	72.07	79
298.42	71.91	47
299.35	71.77	106
300.30	71.62	68
301.78	71.39	51
302.43	71.28	46
303.41	71.13	61
304.45	70.96	62
305.69	70.76	84
306.63	70.61	58
307.99	70.39	66
310.60	69.97	69
313.69	69.47	38
316.02	69.07	92
318.49	68.66	83
323.62	67.79	85
328.10	67.03	81
333.49	66.10	76
337.83	65.34	47
343.49	64.33	15

^a Standard uncertainties u ($k = 1$) for the state variables: $u(T) = 0.02$ K, $u(p) = 0.001$ MPa. Combined expanded uncertainty U_c with confidence level 0.95 ($k = 2$) for a single measurement of the property: $U_c(\gamma) = 0.03$ mN·m⁻¹.

Table S3. The present recommended data $\gamma_R(T)$ for the surface tension of water at selected temperatures, T , with corresponding reference interval ($\gamma_{\min}, \gamma_{\max}$), its center γ_C and half-width $\Delta\gamma$

T/K	$\gamma_R/\text{mN}\cdot\text{m}^{-1}$	$\gamma_{\min}/\text{mN}\cdot\text{m}^{-1}$	$\gamma_{\max}/\text{mN}\cdot\text{m}^{-1}$	$(\gamma_C \pm \Delta\gamma)/\text{mN}\cdot\text{m}^{-1}$
248.15	79.04	78.98	79.11	79.05 ± 0.06
253.15	78.40	78.35	78.46	78.40 ± 0.06
258.15	77.75	77.69	77.80	77.75 ± 0.05
263.15	77.07	77.03	77.12	77.07 ± 0.05
268.15	76.39	76.34	76.43	76.39 ± 0.04
273.15	75.69	75.65	75.72	75.69 ± 0.04
278.15	74.97	74.93	75.00	74.97 ± 0.04
283.15	74.24	74.21	74.27	74.24 ± 0.03
288.15	73.50	73.46	73.52	73.49 ± 0.03
289.15	73.35	73.31	73.37	73.34 ± 0.03
290.15	73.19	73.15	73.22	73.19 ± 0.03
291.15	73.04	73.00	73.07	73.03 ± 0.03
292.15	72.89	72.85	72.92	72.88 ± 0.03
293.15	72.74	72.69	72.76	72.73 ± 0.03
294.15	72.58	72.54	72.61	72.57 ± 0.03
295.15	72.43	72.38	72.45	72.42 ± 0.03
296.15	72.28	72.23	72.30	72.26 ± 0.03
297.15	72.12	72.07	72.14	72.11 ± 0.03
298.15	71.97	71.92	71.99	71.95 ± 0.04
299.15	71.81	71.76	71.83	71.80 ± 0.04
300.15	71.65	71.60	71.68	71.64 ± 0.04
301.15	71.50	71.45	71.52	71.48 ± 0.04
302.15	71.34	71.29	71.36	71.33 ± 0.04
303.15	71.18	71.13	71.21	71.17 ± 0.04
304.15	71.02	70.97	71.05	71.01 ± 0.04
305.15	70.86	70.81	70.89	70.85 ± 0.04
306.15	70.70	70.65	70.73	70.69 ± 0.04
307.15	70.54	70.49	70.57	70.53 ± 0.04
308.15	70.38	70.32	70.41	70.37 ± 0.04
313.15	69.57	69.51	69.60	69.56 ± 0.05
323.15	67.90	67.84	67.95	67.89 ± 0.05
333.15	66.19	66.12	66.24	66.18 ± 0.06
343.15	64.42	64.35	64.48	64.41 ± 0.07
353.15	62.60	62.53	62.67	62.60 ± 0.07
363.15	60.74	60.66	60.81	60.73 ± 0.07
373.15	58.83	58.75	58.90	58.82 ± 0.08
383.15	56.87	56.79	56.95	56.87 ± 0.08
393.15	54.88	54.80	54.95	54.87 ± 0.08
403.15	52.84	52.76	52.91	52.84 ± 0.08
413.15	50.76	50.69	50.84	50.76 ± 0.07
423.15	48.64	48.58	48.72	48.65 ± 0.07
433.15	46.49	46.43	46.57	46.50 ± 0.07
443.15	44.31	44.26	44.39	44.32 ± 0.07
453.15	42.10	42.05	42.17	42.11 ± 0.06
463.15	39.86	39.81	39.93	39.87 ± 0.06
473.15	37.59	37.55	37.66	37.60 ± 0.05
483.15	35.30	35.27	35.36	35.32 ± 0.05

Table S3. – continued

T/K	$\gamma_R/\text{mN}\cdot\text{m}^{-1}$	$\gamma_{\min}/\text{mN}\cdot\text{m}^{-1}$	$\gamma_{\max}/\text{mN}\cdot\text{m}^{-1}$	$(\gamma_C \pm \Delta\gamma)/\text{mN}\cdot\text{m}^{-1}$
493.15	32.99	32.97	33.05	33.01 ± 0.04
503.15	30.67	30.65	30.72	30.69 ± 0.03
513.15	28.33	28.32	28.39	28.35 ± 0.03
523.15	25.99	25.98	26.04	26.01 ± 0.03
533.15	23.64	23.63	23.69	23.66 ± 0.03
543.15	21.30	21.28	21.35	21.31 ± 0.03
553.15	18.96	18.94	19.01	18.97 ± 0.03
563.15	16.64	16.61	16.68	16.65 ± 0.04
573.15	14.34	14.31	14.38	14.35 ± 0.04
583.15	12.08	12.05	12.12	12.08 ± 0.03
593.15	9.86	9.83	9.90	9.86 ± 0.04
603.15	7.70	7.67	7.74	7.71 ± 0.04
613.15	5.63	5.60	5.67	5.63 ± 0.03
623.15	3.67	3.65	3.70	3.68 ± 0.03
633.15	1.88	1.87	1.91	1.89 ± 0.02
643.15	0.39	0.39	0.40	0.39 ± 0.01

Table S4. Literature sources of the a posteriori primary data on the surface tension for water with method, number of data in the data set, their root-mean-square (rmsd) and absolute systematic deviation (bias) from the recommended data, and the quantitative estimate of the data accuracy $u(\gamma)$ given by their author. The literature sources are ordered according to the root-mean-square deviation of the data from the present standard reference data

author(s)	year	method	no. data	bias $\text{mN}\cdot\text{m}^{-1}$	rmsd $\text{mN}\cdot\text{m}^{-1}$	$u(\gamma)$ $\text{mN}\cdot\text{m}^{-1}$
Voronkov [70]	1952	maximum bubble pressure	1	0.003	0.003	0.05
Pallas and Pethica [116]	1983	Wilhelmy plate	1	0.004	0.004	0.02
Kalies et al. [157]	2002	Wilhelmy plate	1	0.005	0.005	0.50
Nath [149]	1999	n/a	1	0.005	0.005	—
Butler and Wightman [46]	1932	maximum bubble pressure	1	0.007	0.007	0.05
Butler et al. [51]	1934	capillary rise	1	0.007	0.007	0.05
Ernst et al. [48]	1932	capillary rise	1	0.007	0.007	—
Ling and Van Winkle [80]	1958	capillary rise	2	0.009	0.010	0.50
Padday [111]	1979	cone	1	0.011	0.011	—
Richards and Carver [31]	1921	capillary rise	2	0.007	0.012	—
Ali et al. [171]	2006	pendant drop	5	0.009	0.013	—
Lee et al. [151]	2000	capillary rise	6	0.011	0.013	—
Magini [14]	1911	capillary rise	1	0.014	0.014	—
Ameta and Singh [208]	2015	pendant drop	15	0.006	0.014	0.01
Enders et al. [177]	2007	pendant drop	5	0.011	0.017	0.20
Glinski et al. [147]	1999	Wilhelmy plate	1	0.018	0.018	—
Glinski et al. [153]	2001	Wilhelmy plate	1	0.018	0.018	0.20
Voljak [66], $T < T_b$	1950	capillary rise	11	0.019	0.022	—
Ernst et al. [55]	1936	capillary rise	1	0.023	0.023	—
Teitelbaum et al. [68]	1951	capillary rise	11	0.003	0.024	—
Hadrlová et al. [163]	2004	Wilhelmy plate	1	0.025	0.025	—
Lago et al. [205]	2014	Wilhelmy plate	1	0.025	0.025	0.10
Urazovskij and Tchetaev [63]	1949	capillary rise	9	0.023	0.026	—
this work	2015	Wilhelmy plate	33	0.021	0.027	0.03
Maham and Mather [155]	2001	capillary rise	4	0.027	0.030	—
Padday [92]	1968	Wilhelmy plate	3	0.025	0.032	—
Dixmier [172]	2006	capillary pressure diff.	5	0.020	0.032	—
Ling and Van Winkle [79]	1958	capillary rise	2	0.026	0.032	0.35
Belton [52]	1935	maximum bubble pressure	1	0.033	0.033	—
Kawanishi et al. [94]	1970	Wilhelmy plate	1	0.034	0.034	—
Hrubý et al. [204]	2014	capillary rise	26	0.000	0.034	0.15
Padday and Russell [83]	1960	Wilhelmy plate	2	0.008	0.036	0.03
Rotenberg et al. [117]	1983	pendant drop	2	0.035	0.036	—
Kumar et al. [159]	2003	Wilhelmy plate	3	0.022	0.038	0.03
Vázquez et al. [140]	1995	Wilhelmy plate	7	0.015	0.043	0.25
Álvarez et al. [144]	1998	Wilhelmy plate	7	0.015	0.043	0.02
Nakanishi et al. [98]	1971	capillary rise	1	0.044	0.044	0.30
Kowalska et al. [90]	1965	capillary rise	1	0.046	0.046	0.30
Wallenberger and Lyzenga [129]	1990	microwave	1	0.046	0.046	1.60
Sugden [32]	1921	capillary rise	1	0.047	0.047	0.22
Ferguson [18]	1922	pull on sphere	1	0.053	0.053	0.02
Harkins and Gilbert [40]	1926	pendant drop	1	0.053	0.053	—
Harkins and Brown [26]	1919	capillary rise	1	0.053	0.053	—

Table S4. – continued

author(s)	year	method	no. data	$ \text{bias} $ $\text{mN}\cdot\text{m}^{-1}$	rmsd $\text{mN}\cdot\text{m}^{-1}$	$u(\gamma)$ $\text{mN}\cdot\text{m}^{-1}$
Harkins et al. [28]	1920	capillary rise	1	0.053	0.053	–
Owens et al. [121]	1987	Wilhelmy plate	1	0.054	0.054	0.05
Crawford and Van Winkle [82]	1959	capillary rise	1	0.055	0.055	0.50
Humphreys and Mohler [6]	1895	capillary rise	41	0.012	0.056	–
Maham and Mather [155]	2004	Wilhelmy plate	4	0.044	0.056	0.01
Wright and Akhtar [97]	1970	capillary rise	1	0.056	0.056	0.08
Mohsen-Nia et al. [192]	2010	maximum bubble pressure	6	0.052	0.056	0.08
Richter and Vollhardt [175]	2006	Wilhelmy plate	1	0.061	0.061	–
García-Abuín et al. [184]	2008	Wilhelmy plate	5	0.012	0.061	0.07
Vinš et al. [211]	2015	capillary rise	64	0.012	0.062	0.15
Cini et al. [100]	1972	Wilhelmy plate	14	0.053	0.063	–
Pallas and Pethica [116]	1983	capillary rise	1	0.064	0.064	0.10
Han et al. [199]	2012	pendant drop	4	0.051	0.068	0.04
Furlong and Hartland [114]	1980	vertical cylinder	8	0.033	0.069	0.10
Vargaftik et al. [103], $T < T_b$	1973	capillary rise	32	0.011	0.074	–
Rehbinder [41]	1926	maximum bubble pressure	5	0.027	0.075	–
Kovalenko et al. [73]	1956	maximum bubble pressure	2	0.056	0.075	–
Toryanik and Pogrebnyak [108]	1976	maximum bubble pressure	6	0.054	0.077	0.35
Russo and Hoffmann [193]	2010	maximum bubble pressure	20	0.059	0.083	0.40
Vargaftik et al. [103], $T > T_b$	1973	capillary rise	115	0.018	0.089	–
Voljak [66], $T > T_b$	1950	capillary rise	29	0.118	0.162	–

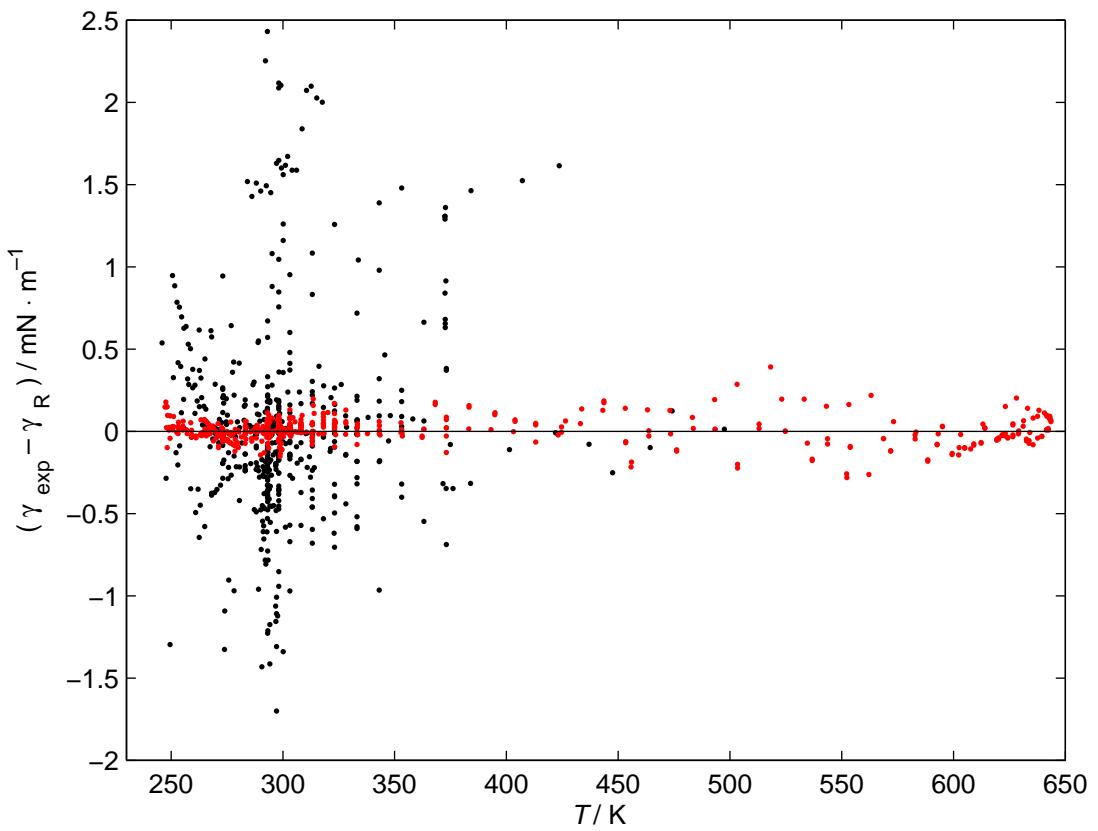


Figure S1. Deviations $\gamma_{\text{exp}} - \gamma_{\text{R}}$ of the experimental water surface tension values γ_{exp} from the recommended values $\gamma_{\text{R}}(T)$ as a function of the temperature T . Red dot, primary data; Black dot, secondary data. The deviations are depicted for all 1055 density data points included in the generation of the recommended water surface tension values.

References

- [1] Buliginsky. Untersuchung über die Capillarität einiger Salzlösungen bei verschiedenen Concentrationen. *Ann. Phys.* **1886**, *134*, 440–454.
- [2] Rodenbeck, H. Ueber Capillaritätsbestimmungen von Flüssigkeitsgemischen. *Beibl. Ann. Phys.* **1880**, *4*, 104–108.
- [3] Volkmann, P. Ueber die Cohäsion von Salzlösungen. *Ann. Phys. Chem.* **1882**, *17*, 354–390.
- [4] Traube, J. Ueber die Bestimmung der Capillaritätsconstanten einiger wässriger und alkoholischer Lösungen durch Beobachtung der Steighöhen im capillaren Rohre. *J. Prakt. Chem.* **1885**, *31*, 177–218.
- [5] Ramsay, W.; Shields, J. The Molecular Complexity of Liquids. *J. Chem. Soc.* **1893**, *63–64*, 1089–1109.
- [6] Humphreys, W. J.; Mohler, J. F. Surface tension of water at temperatures below zero degree centigrade. *Phys. Rev.* **1895**, *2*, 387–391.
- [7] Volkmann, P. Beiträge zur Feststellung der wahren Oberflächenspannung des reinen Wassers für Temperaturen zwischen 0 und 40 °C. *Ann. Phys.* **1895**, *56*, 457–491.
- [8] Sentis, H. Dissertation, Paris **1897**. Data taken from [212].
- [9] Sohet, A. Variations des hauteurs capillaires et des tensions superficielles de l'eau de l'alcool et des solutions d'eau et d'alcool avec la température. *Mem. Soc. R. Sci. Liege* **1898**, *20*, 1–56.
- [10] Grabowsky, W. Dissertation, Universität Königsberg **1904**. Data taken from [212].
- [11] Hartley, H.; Thomas, N. G.; Applebey, M. P. Some Physico-Chemical Properties of Mixtures of Pyridine and Water. *J. Chem. Soc.* **1908**, *93*, 538–560.
- [12] Grunmach, L. Bestimmung der Oberflächenspannung und anderer physikalischer Konstanten von Essigsäure-Wassermischungen. *Ann. Phys.* **1909**, *28*, 217–258.
- [13] Drucker, K.; Moles, E. Glaslöslichkeit in wässrigen Lösungen von Glycerin und Isobuttersäure. *Z. Phys. Chem.* **1910**, *75*, 405–436.
- [14] Magini, R. Sulle misure di tensione superficiale, *Atti R. Accad. Lincei* **1911**, *20*, 30–37.
- [15] Morgan, J. L. R.; McAfee, A. The weight of a falling drop and the laws of tate, IX. The drop weights of the associated liquids, water, ethyl alcohol, methyl alcohol and acetic acid; and the surface tensions and capillary constants calculated from them. *J. Am. Chem. Soc.* **1911**, *33*, 1275–1290.
- [16] Grunmach, L. Experimentelle Bestimmung der Oberflächenspannung von Alkohol-Wasser-mischungen nach der Kapillarwellenmethode. *Ann. Phys.* **1912**, *38*, 1018–1026.
- [17] Morgan, J. L. R.; Neidle, M. The weight of a falling drop and the laws of tate, XVIII. The drop weights, surface tensions and capillary constants of aqueous solutions of ethyl, methyl, and amyl alcohols, and of acetic and formic acid. *J. Am. Chem. Soc.* **1913**, *35*, 1856–1865.
- [18] Ferguson, A. On the Surface-tensions of Liquids in contact with different Gases. *Philos. Mag.* **1914**, *28*, 403–412.
- [19] Richards, T. W.; Coombs, L. B. The surface tension of water, methyl, ethyl and isobutyl alcohols, ethyl butyrate, benzene and toluene. *J. Am. Chem. Soc.* **1915**, *37*, 1656–1676.
- [20] Sentis, M. H. Tension superficielle de quelques liquides. *Ann. Universite de Grenoble* **1915**, *27*, 593–624.

- [21] Morgan, J. L. R.; Davis, C. E. The properties of mixed liquids. I. Sulfuric acid-water mixtures. *J. Am. Chem. Soc.* **1916**, *38*, 555–568.
- [22] Morgan, J. L. R.; Egloff, G. The properties of mixed liquids. II. Phenol-water and triethylamine-water mixtures. *J. Am. Chem. Soc.* **1916**, *38*, 844–857.
- [23] Harkins, W. D.; Brown, F. E.; Davies, E. C. H. The structure of the surfaces of liquids, and solubility as related to the work done by the attraction of two liquid surfaces as they approach each other. *J. Am. Chem. Soc.* **1917**, *39*, 356–364.
- [24] Jaeger, F. M. Über die Temperaturabhängigkeit der molekularen freien Oberflächenenergie von Flüssigkeiten im Temperaturbereich von -80 bis +1650 °C. *Z. Anorg. Allg. Chem.* **1917**, *101*, 1–214.
- [25] Morgan, J. L. R.; Scarlett, A. J. The properties of mixed liquids. IV. The law of mixtures II. *J. Am. Chem. Soc.* **1917**, *39*, 2275–2293.
- [26] Harkins, W. D.; Brown, F. E. The determination of surface tension (free surface energy), and the weight of falling drops: the surface tension of water and benzene by capillary height method. *J. Am. Chem. Soc.* **1919**, *41*, 499–524.
- [27] Richards, T. W.; Palitzsch, S. Compressibility of aqueous solutions, especially of urethane, and the polymerization of water. *J. Am. Chem. Soc.* **1919**, *41*, 59–69.
- [28] Harkins, W. D.; Clark, G. L.; Roberts, L. E. The orientation of molecules in surfaces, surface energy, adsorption, and surface catalysis. V. The adhesional work between organic liquids and water. *J. Am. Chem. Soc.* **1920**, *42*, 700–712.
- [29] Maass, O.; Hatcher, W. H. The properties of pure hydrogen peroxide I. *J. Am. Chem. Soc.* **1920**, *42*, 2548–2569.
- [30] Stocker, H. Die Oberflächenspannung schwingender Flüssigkeitsstrahlen, untersucht an Wasser und wässriger Salzlösungen. *Z. Phys. Chem., Stoechiom. Verwandtschaftsl.* **1920**, *94*, 149–180.
- [31] Richards, T. W.; Carver, E. K. A critical study of the capillary rise method of determining surface tension, with data for water, benzene, toluene, chloroform, carbon tetrachloride, ether and dimethyl aniline. *J. Am. Chem. Soc.* **1921**, *43*, 827–847.
- [32] Sugden, S. The Determination of Surface Tension from the Rise in Capillary Tubes. *J. Chem. Soc.* **1921**, *119*, 1483–1492.
- [33] Bircumshaw, L. L. The Surface Tension of Mixtures of Alcohol and Water at 25 °. *J. Chem. Soc.* **1922**, *121*, 887–891.
- [34] Ferguson, A. Studies in capillarity, I. Some general considerations and a discussion of methods for the measurements of interfacial tensions. *Trans. Faraday Soc.* **1922**, *32*, 370–383.
- [35] Sugden, S. The Determination of Surface Tension from the Maximum Pressure in Bubbles. *J. Chem. Soc.* **1922**, *121*, 858–866.
- [36] Butler, R. R. Surface Phenomena in Sucrose Solutions. *J. Chem. Soc.* **1923**, *123*, 2060–2065.
- [37] Ferguson, A. On the measurement of the surface tension of a small quantity of liquid. *Proc. Phys. Soc., London* **1923**, *36*, 37–44.
- [38] Richards, T. W.; Speyers, C. L.; Carver, E. K. The determination of surface tension with very small volumes of liquid, and the surface tensions of octanes and xylenes at several temperatures. *J. Am. Chem. Soc.* **1924**, *46*, 1196–1211.

- [39] Ferguson, A.; Vogel, I. On the "hyperbola" method for the measurements of surface tensions. *Proc. Phys. Soc., London* **1926**, *38*, 193–203.
- [40] Harkins, W. D.; Gilbert, E. C. The structure of films of water on salt solutions. II. The surface tension of calcium chloride solutions at 25 °. *J. Am. Chem. Soc.* **1926**, *48*, 604–607.
- [41] Rehbinder, P. Wasser als oberflächenaktivität Stoff. Oberflächenaktivität und Adsorptionsskräfte II. *Z. Phys. Chem.* **1926**, *121*, 103–126.
- [42] Moser, H. Der Absolutwert der Oberflächenspannung des reinen Wassers nach der Bügelmethode und seine Abhängigkeit von der Temperatur. *Ann. Phys.* **1927**, *82*, 993–1012.
- [43] Warren, E. L. The Surface-Tension Balance. *Philos. Mag.* **1927**, *4*, 358–385.
- [44] Harkins, W. D.; Jordan, H. F. A method for the determination of surface and interfacial tension from the maximum pull on a ring. *J. Am. Chem. Soc.* **1930**, *52*, 1751–1772.
- [45] Rehbinder, P.; Taubmann, A. Grenzflächenaktivität und Orientierung polarer Moleküle in Abhängigkeit von der Temperatur und der Natur der Trennungsfläche. VI. Grenzflächeneigenschaften aromatischer Amine und ihrer Salze. *Z. Phys. Chem., Abt. A* **1930**, *147*, 188–205.
- [46] Butler, J. A. V.; Wightman, A. Adsorption at the Surface of Solutions. Part I. The Surface Composition of Water-Alcohol Solutions. *J. Chem. Soc.* **1932**, 2089–2097.
- [47] Butler, J. A. V.; Lees, A. D. Absorption at the Surface of Solutions. Part II. The Effect of Lithium Chloride on the Surface of Water-Alcohol Solutions. *J. Chem. Soc.* **1932**, *134*, 2097–2104.
- [48] Ernst, R. C.; Litkenhous, E. E.; Spanier, J. W. The physical properties of the ternary system acetone-n-butyl alcohol–water. *J. Phys. Chem.* **1932**, *36*, 842–854.
- [49] Ferguson, A.; Kennedy, S. J. Notes on surface-tension measurement. *Proc. Phys. Soc.* **1932**, *44*, 511–520.
- [50] Speakman, J. C. The Measurement of Surface and Interfacial Tensions. *J. Chem. Soc.* **1932**, 1449–1453.
- [51] Butler, J. A. V.; Wightman, A.; MacLennan, W. H. Adsorption at the Surface of Solutions. Part III. The Surface Structure of Solutions of the Lower Aliphatic Alcohols. *J. Chem. Soc.* **1934**, 528–532.
- [52] Belton, J. W. The surface tensions of ternary solutions. Part I. The surface tensions of aqueous solutions of (a) sodium and potassium chlorides, (b) sodium chloride and hydrochloric acid. *Trans. Faraday Soc.* **1935**, *31*, 1413–1419.
- [53] Sabinina, L.; Terpugow, L. Die Oberflächenspannung des Systems Schwefelsäure–Wasser. *Z. Phys. Chem., Abt. A* **1935**, *173*, 237–241.
- [54] Trieschmann, H.-G. Oberflächenspannung und Solvatation. *Z. Phys. Chem., Abt. B* **1935**, *29*, 328–334.
- [55] Ernst, R. C.; Watkins, C. H.; Ruwe, H. H. The physical properties of the ternary system ethyl alcohol-glycerin-water. *J. Phys. Chem.* **1936**, *40*, 627–635.
- [56] Valentiner, S.; Hohls, H. W. Oberflächenspannungen von Alkohol-Wasser- Mischungen. *Z. Phys.* **1938**, *108*, 101–106.
- [57] Cockett, A. H.; Ferguson, A. The Surface Tension of Water and Heavy Water. *Philos. Mag.* **1939**, *28*, 685–693.

- [58] Neros, C. A.; Eversole, W. G. The surface tension of aqueous perchloric acid at 15 °, 25 °, and 50 °C. *J. Phys. Chem.* **1941**, *45*, 388–395.
- [59] Smith, G. W.; Sorg, L. V. The measurement of boundary tension by the pendant-drop method. I. The aliphatic alcohols. *J. Phys. Chem.* **1941**, *45*, 671–681.
- [60] Addison, C. C. The Properties of Freshly Formed Surfaces. Part I. The Application of the Vibrating-jet Technique to Surface-tension Measurements on Mobile Liquids. *J. Chem. Soc.* **1943**, 535–541.
- [61] Glagoleva, A. A. Surface tension of the binary system HCOOH–H₂O and CH₃COOH –H₂O. *Zh. Obshch. Khim.* **1947**, *17*, 1044–1047.
- [62] Grant, W. E.; Darch, W. J.; Bowden, S. T.; Jones, W. J. The surface tension and viscosity of solutions of uranyl salts. *J. Phys. Colloid Chem.* **1948**, *52*, 1227–1236.
- [63] Urazovskij, S. S.; Tchetaev, P. M. Novyj variant kapilljarnogo metoda dlja izmerenija malykh izmenenij poverchnostnogo natjazenija i ego primenenie. *Kolloidn. Zh.* **1949**, *11*, 359–362.
- [64] Douglas, H. W. A Pendent-Drop Apparatus for Surface and Interfacial Tensions. *J. Sci. Instrum.* **1950**, *27*, 67–69.
- [65] Vierk, A.-L.; Fredenhagen, K. Experimentelle Untersuchungen an den Zweistoffsystemen: Wasser-Acetonitril, Wasser-Dioxan, Äthanol-Acetonitril und Cyclohexan-Dioxan. *Z. Anorg. Chem.*, **1950**, *251*, 283–296.
- [66] Voljak, L. D. Issledovanie temperaturnoj zavisimosti poverchnostnogo natjazhenija vody. *Dokl. Akad. Nauk SSSR* **1950**, *74*, 307–310.
- [67] Hacker, P. T. Experimental values of the surface tension of supercooled water. *Natl. Advis. Comm. Aeronaut., Tech. Notes* **1951**, 1–20.
- [68] Teitelbaum, B. Y.; Gortalova, T. A.; Sidorova, E. E. Politermitcheskoe issledovanie poverchnostnogo natjazhenija vodnych rastvorov nizhshich spirtov. *Zh. Fiz. Khim.* **1951**, *25*, 911–919.
- [69] Fox, H. W.; Chrisman, C. H. The ring method of measuring surface tension for liquids of high density and low surface tension. *J. Phys. Chem.* **1952**, *56*, 284–288.
- [70] Voronkov, M. G. Parachor i struktura organitcheskich soedinenij. III. Odnojadernye aromaticheskie uglevodorydy. *Zh. Fiz. Khim.* **1952**, *26*, 813–821.
- [71] Heiks, J. R.; Barnett, M. K.; Jones, L. V.; Orban, E. The density, surface tension and viscosity of deuterium oxide at elevated temperatures. *J. Phys. Chem.* **1954**, *58*, 488–491.
- [72] Sutherland, K. L. The oscillating jet method for the measurement of surface tension. *Aust. J. Chem.* **1954**, *7*, 319–328.
- [73] Kovalenko, K. N.; Trifonov, N. A.; Tissen, D. C. Fiziko-khimitcheskoe issledovanie sistemy voda-uksusnyj angidrid-dioksan. *Zh. Obshch. Khim.* **1956**, *26*, 2404–2410.
- [74] Hoffmann, W.; Schoeneck, H.; Wanninger, W. Die Oberflächenspannung aräometrischer Prüfungsflüssigkeiten im Temperaturbereich von 15 °C bis 25 °C. *Z. Phys. Chem.* **1957**, *11*, 56–64.
- [75] Howard, K. S.; McAllister, R. A. Surface Tension of Aceton-water Solutions Up to Their Normal Boiling Points. *AICHE J.* **1957**, *3*, 325–329.
- [76] Murphy, N. F.; Lastovica, J. E.; Fallis, J. G. Correlation of Interfacial Tension of Two-Phase Three-Component Systems. *Ind. Eng. Chem.* **1957**, *49*, 1035–1042.

- [77] Hirata, M.; Kanai, T. Physical Properties of Binary Systems. *Kagaku Kogaku* **1958**, *22*, 155–157.
- [78] Krishna, P. M.; Venkateswarlu, D. Modification of drop weight method for measuring surface and interfacial tension of liquids. *J. Indian Chem. Soc.* **1958**, *35*, 804–806.
- [79] Ling, T. D.; Van Winkle, M. Properties of Binary Mixture as a Function of Composition. *Ind. Eng. Chem.* **1958**, *3*, 88–95.
- [80] Ling, T. D.; Van Winkle, M. Interfacial Tension at the Boiling Point and Vapor Viscosity near the Dew Point of Binary Mixtures. *Ind. Eng. Chem.* **1958**, *3*, 82–88.
- [81] Uchida, S. I.; Matsumoto, K. Some Physical Properties of Methanol-Water Solutions at Various Temperatures up to the Boiling Point. *Kagaku Kogaku* **1958**, *22*, 570–572.
- [82] Crawford, H. R.; Van Winkle, M. Surface Tension and Density of Ternary Systems at Their Normal Boiling Point. *Ind. Eng. Chem.* **1959**, *51*, 601–606.
- [83] Padday, J. F.; Russell, D. R. The measurement of the surface tension of pure liquids and solutions. *J. Colloid Sci.* **1960**, *15*, 503–511.
- [84] Hasaba, S.; Uemura, T.; Narita, H. Some thermal and physical properties of lithium bromide-water solutions for the designing of absorption refrigerating machines. *The refrigeration* **1961**, *36*, 622–629.
- [85] Slowinski, E. J.; Masterton, W. L. A simple absolute method for the measurement of surface tension. *J. Phys. Chem.* **1961**, *65*, 1067–1068.
- [86] Bordi, S.; Vannel, F. Proprietà superficiali e variazioni strutturali dell’acqua. *Ann. Chim.* **1962**, *52*, 80–93.
- [87] Catchpole, J. P.; Ellis, S. R. M. A Boiling Point Tensiometer. *J. Chem. Eng. Data* **1963**, *8*, 418–419.
- [88] Nagy, L. G. Thermodynamic investigation of the interfacial properties of liquid mixtures. *Period. Polytech., Chem. Eng.* **1963**, *7*, 75–91.
- [89] Drost-Hansen, W. Aqueous Method of Study and Structural Properties. *Ind. Eng. Chem.* **1965** *57*, 18–37.
- [90] Kowalska, E.; Kowalski, W.; Ślączka, A. Oznaczanie napięcia powierzchniowego roztworów dwuskładnikowych na podstawie pomiaru prędkości rozchodzenia się ultradźwięku, *Roczn. Chem.* **1965**, *39*, 449–454.
- [91] Murto, J.; Kivinen, A.; Kivimaa, S.; Laakso, R. Fluoroalcohols. Part 4. Densities, partial molar volumes, refractive indices, viscosities, dielectric constants, and surface tensions of 1,1,1,3,3-hexafluoro-2-propanol–water mixtures, *Suom. Kemistil. B* **1967**, *40*, 250–257.
- [92] Padday, J. F. Effect of Temperature on the Wettability of Low-Energy Surfaces. *J. Colloid Interface Sci.* **1968**, *28*, 557–564.
- [93] Gittens, G. J. Variation of Surface Tension of Water with Temperature. *J. Colloid Interface Sci.* **1969**, *30*, 406–412.
- [94] Kawanishi, T.; Seimiya, T.; Sasaki, T. Corrections for Surface Tension Measured by Wilhelmy Method. *J. Colloid Interface Sci.* **1970**, *32*, 622–627.
- [95] Konobeev, B. I.; Lyapin, V. V. Plotnost, vjazkost i poverchnostnoe natjazhenie nekotorych binarnych sistem. *Zh. Prikl. Khim.* **1970**, *43*, 803–811.

- [96] Pike, F. P.; Bonnet, J. C. The End-Correction in the Wilhelmy Technique for Surface Tension Measurements. *J. Colloid Interface Sci.*, **1970**, *34*, 597–605.
- [97] Wright, E. H. M.; Akhtar, B. A. Soluble Surface Films of Short-chain Monocarboxylic Acids on Organic and Aqueous Substrates. *J. Chem. Soc. B* **1970**, *70*, 151–157.
- [98] Nakanishi, K.; Matsumoto, T.; Hayatsu, M. Surface Tension of Aqueous Solutions of Some Glycols. *J. Chem. Eng. Data* **1971**, *16*, 44–45.
- [99] Bonnet, J. C.; Pike, F. P. Surface Properties of Nine Liquids. *J. Chem. Eng. Data* **1972**, *17*, 145–150.
- [100] Cini, R.; Loglio, G.; Ficalbi, A. Temperature Dependence of the Surface Tension of Water by the Equilibrium Ring Method. *J. Colloid Interface Sci.* **1972**, *41*, 287–297.
- [101] Granzhan, V. A.; Kirillova, O. G. Fiziko-khimicheskij analiz binarnych sistem amidov s vodoj. *Tr. GIAP* **1972**, *13*, 5–10.
- [102] Wagenbreth, H. Dichte und Oberflächen Spannung von Äthanol-Wasser-Mischungen für die Internationalen Alkoholtafeln der OIML. *PTB-Mitt.* **1972**, *82*, 299–303.
- [103] Vargaftik, N. B.; Voljak, L. D.; Volkov, B. N. Issledovanije poverchnostnogo natjazenia pri temperaturach, blizkikh k krititcheskoj. *Teploenergetika* **1973**, *8*, 80–84.
- [104] Padday, J. F.; Pitt, A. R.; Pashley, R. M. Menisci at a Free Liquid Surface: Surface Tension from the Maximum Pull on a Rod. *J. Chem. Soc., Faraday Trans. 1* **1975**, *71*, 1919–1931.
- [105] Taylor, J. A. G.; Mingins, J. Properties of the Non-polar Oil/Water Interface, Part 1.—Procedures for the Accurate Measurement of the Interfacial Pressure of an Insoluble Monolayer, *J. Chem. Soc., Faraday Trans. 1* **1975**, *71*, 1161–1171.
- [106] Thakur, D. K.; Hickman, K. Surface Tension of Water at 100 °C. *J. Colloid Interface Sci.* **1975**, *50*, 525–531.
- [107] Ramakrishnan, S.; Mailliet, K.; Hartland, S. Measurement of surface and interfacial tension from the maximum pressure in sessile and pendant bubbles and drops. *Proc. Indian Acad. Sci. A* **1976**, *83*, 107–118.
- [108] Toryanik, A. I.; Pogrebnyak, V. G. Poverchnostnoe natyazhenie vodnych rastvorov acetona. *J. Struct. Chem.* **1976**, *17*, 536–537.
- [109] Artamonov, A. F.; Bektursynov, B.; Gorjaev, M. I. Polutchenie i svojstva α -monoglyceridov naftenovych kislot. *Izv. Akad. Nauk Kaz. SSR, Ser. Khim.* **1977**, *27*, 85–88.
- [110] Tornberg, E. A. Surface Tension Apparatus According to the Drop Volume Principle. *J. Colloid Interface Sci.* **1977**, *60*, 50–53.
- [111] Padday, J. F. Menisci Formed by a Cone at a Free Liquid Surface. *J. Chem. Soc., Faraday Trans. 1* **1979**, *75*, 2827–2838.
- [112] Patterson, R. E.; Ross, S. The Pendent-Drop Method to Determine Surface or Interfacial Tensions. *Surf. Sci.* **1979**, *81*, 451–463.
- [113] Ross, S.; Patterson, R. E. Surface and Interfacial Tensions of Conjugate Solution in Ternary Systems. *J. Chem. Eng. Data* **1979**, *24*, 111–115.
- [114] Furlong, D. N.; Hartland, S. Wall Effects in Measurement of Surface Tension Using a Vertical Cylinder, Part 2.—Experimental. *J. Chem. Soc., Faraday Trans. 1* **1980**, *76*, 467–472.
- [115] Won, Y. S.; Chung, D. K.; Mills, A. F. Density, Viscosity, Surface Tension, and Carbon Dioxide Solubility and Diffusivity of Methanol, Ethanol, Aqueous Propanol, and Aqueous Ethylene Glycol at 25 °C. *J. Chem. Eng. Data* **1981**, *26*, 140–141.

- [116] Pallas, N. R.; Pethica, B. A. The Surface Tension of Water. *Colloids Surf.* **1983**, *6*, 221–227.
- [117] Rotenberg, Y.; Boruvka, L.; Neumann, A. W. Determination of Surface Tension and Contact Angle from the Shapes of Axisymmetric Fluid Interfaces. *J. Colloid Interface Sci.* **1983**, *93*, 169–183.
- [118] Drummond, C. J.; Warr, G. G.; Grieser, F.; Ninham, B. W.; Evans, D. F. Surface Properties and Micellar Interfacial Microenvironment of *n*-Dodecyl β -D-Maltoside. *J. Phys. Chem.* **1985**, *89*, 2103–2109.
- [119] Cheong, W. J.; Carr, P. W. The surface tension of mixtures of methanol, acetonitrile, tetrahydrofuran, isopropanol, tertiary butanol and dimethylsulfoxide with water at 25 °C. *J. Liq. Chromatogr.* **1987**, *10*, 561–581.
- [120] Gaonkar, A. G.; Neuman, R. D. The Uncertainty in Absolute Values of Surface Tension of Water. *Colloids Surf.* **1987**, *27*, 1–14.
- [121] Owens, N. F.; Johnston, D. S.; Gingell, D.; Chapman, D. Surface properties of a long-chain 10:12 diynoic acid monolayer at air-liquid interfaces. *Thin Solid Films* **1987**, *155*, 255–266.
- [122] Braslau, A.; Pershan, P. S.; Swislow, G.; Ocko, B. M.; Als-Nielsen, J. Capillary waves on the surface of simple liquids measured by x-ray reflectivity. *Phys. Rev. [Sect.] A* **1988**, *38*, 2457–2470.
- [123] Jennings jr., J. W.; Pallas, N. R. An Efficient Method for the Determination of Interfacial Tensions from Drop Profiles. *Langmuir* **1988**, *4*, 959–967.
- [124] Kalbassi, M. A.; Biddulph, M. W. Surface Tension of Mixtures at Their Boiling Points. *J. Chem. Eng. Data* **1988**, *33*, 473–476.
- [125] Ramkumar, D. H. S.; Kudchadker, A. P. Mixture Properties of the Water + γ -Butyrolactone + Tetrahydrofuran System. 2. Viscosities and Surface Tensions of γ -Butyrolactone + Water at 303.15–343.15 K and γ -Butyrolactone + Tetrahydrofuran at 278.15–298.15 K. *J. Chem. Eng. Data* **1989**, *34*, 463–465.
- [126] Cheng, P.; Li, D.; Boruvka, L.; Rotenberg, Y.; Neumann, A. W. Automation of Axisymmetric Drop Shape Analysis for Measurements of Interfacial Tension and Contact Angle. *Colloids Surf.* **1990**, *43*, 151–167.
- [127] Florianot, M. A.; Angell, C. A. Surface Tension and Molar Surface Free Energy and Entropy of Water to –27.2 °C. *J. Phys. Chem.* **1990**, *94*, 4199–4202.
- [128] Pallas, N. R.; Harrison, Y. An Automated Drop Shape Apparatus and the Surface Tension of Pure Water. *Colloids Surf.* **1990**, *43*, 169–194.
- [129] Wallenberger, A. P.; Lyzenga, D. R. Measurement of the Surface Tension of Water Using Microwave Backscatter from Gravity-Capillary Waves. *IEEE Trans. Geosci. Electron.* **1990**, *28*, 1012–1016.
- [130] Hansen, F. K.; Rødsrud, G. Surface Tension by Pendant Drop. *J. Colloid Interface Sci.* **1991**, *141*, 1–9.
- [131] Noordmans, J.; Busscher, H. J. The influence of droplet volume and contact angle on liquid surface tension measurements by axisymmetric drop shape analysis-profile (ADSA-P). *Colloid Surf.* **1001**, *58*, 239–249.
- [132] Holcomb, C. D.; Zollweg, J. A. Improved Differential Bubble Pressure Surface Tensiometer. *J. Colloid Interface Sci.* **1992**, *154*, 51–65.
- [133] Hansen, F. K. Surface Tension by Image Analysis: Fast and Automatic Measurements of Pendant and Sessile Drops and Bubbles. *J. Colloid Interface Sci.* **1993**, *160*, 209–217.

- [134] Stückrad, B.; Hiller, W. J.; Kowalewski, T. A. Measurement of dynamic surface tension by the oscillating droplet method. *Exp. Fluids* **1993**, *15*, 332–340.
- [135] Gunde, R.; Hartland, S.; Mäder, R. Sphere Tensiometry: A New Approach to Simultaneous and Independent Determination of Surface Tension and Contact Angle. *J. Colloid Interface Sci.* **1995**, *176*, 17–30.
- [136] Kinart, C. M.; Kinart, W. J.; Bald, A.; Szejgis, A. Study of the intermolecular interactions in liquid N,N-dimethylacetamide–water mixtures. *Phys. Chem. Liq.* **1995**, *30*, 151–157.
- [137] Krotov, V. V.; Prokhorov, V. A.; Pavlov, S. Yu.; Rusanov, A. I. Method of touching drops: a new method for measuring the surface and interfacial tension of liquids. *Colloids Surf., A* **1995**, *104*, 165–168.
- [138] Rao, P.; Enhorning, G. Surface tension determined with a micromethod. *Colloids Surf., B* **1995**, *4*, 159–163.
- [139] Trinh, E. H.; Ohsaka, K. Measurement of Density, Sound Velocity, Surface Tension, and Viscosity of Freely Suspended Supercooled Liquids. *Int. J. Thermophys.* **1995**, *16*, 545–555.
- [140] Vázquez, G.; Alvarez, E.; Navaza, J. M. Surface Tension of Alcohol + Water from 20 to 50 °C. *J. Chem. Eng. Data* **1995**, *40*, 611–614.
- [141] Semmler, A.; Ferstl, R.; Kohler, H.-H. New Laser Technique for Automatic Interfacial Tension Measurements: Laser Scanning Drop Shape Analysis (LASDA). *Langmuir* **1996**, *12*, 4165–4172.
- [142] Slauenwhite, D. E.; Johnson, B. D. Effect of organic matter on bubble surface tension. *J. Geophys. Res.* **1996**, *101*, 3769–3774.
- [143] Zhang, L.; Ren, L.; Hartland, S. More Convenient and Suitable Methods for Sphere Tensiometry. *J. Colloid Interface Sci.* **1996**, *180*, 493–503.
- [144] Álvarez, E.; Rendo, R.; Sanjurjo, B.; Sánchez-Vilas, M.; Navaza, J. M. Surface Tension of Binary Mixtures of Water *N*-Methyldiethanolamine and Ternary Mixtures of This Amine and Water with Monoethanolamine, Diethanolamine, and 2-Amino-2-methyl-1-propanol from 25 to 50 °C. *J. Chem. Eng. Data* **1998**, *43*, 1027–1029.
- [145] Hawrylak, B.; Andrecyk, S.; Gabriel, C.-E.; Gracie, K.; Palepu, R. Viscosity, Surface Tension, and Refractive Index Measurements of Mixtures of Isomeric Butanediols with Water. *J. Solution Chem.* **1998**, *27*, 827–841.
- [146] Tsierkezos, N. G.; Molinou, I. E. Thermodynamic Properties of Water + Ethylene Glycol at 283.15, 293.15, 303.15, and 313.15 K. *J. Chem. Eng. Data* **1998**, *43*, 989–993.
- [147] Glinski, J.; Chavepeyer, G.; Platten, J. K. Surface properties of diluted aqueous solutions of 1,2-pentanediol. *J. Chem. Phys.* **1999**, *111*, 3233–3236.
- [148] Horozov, T.; Arnaudov, L. A Novel Fast Technique for Measuring Dynamic Surface and Interfacial Tension of Surfactant Solutions at Constant Interfacial Area. *J. Colloid Interface Sci.* **1999**, *219*, 99–109.
- [149] Nath, S. Surface Tension of Nonideal Binary Liquid Mixtures as a Function of Composition. *J. Colloid Interface Sci.* **1999**, *209*, 116–122.
- [150] Gente, G.; La Mesa, C. Water-Trifluoroethanol Mixtures: Some Physicochemical Properties. *J. Solution Chem.* **2000**, *29*, 1159–1172.
- [151] Lee, J. W.; Park, S. B.; Lee, H. Densities, Surface Tensions, and Refractive Indices of the Water + 1,3-Propanediol System. *J. Chem. Eng. Data* **2000**, *45*, 166–168.

- [152] Feenstra, P. A.; Judd, R. L.; Weaver, D. S. A Practical Device for Surface tension Measurements in Volatile Fluids. *HVACR Res.* **2001**, *7*, 3–14.
- [153] Glinski, J.; Chavepeyer, G.; Platten, J. K. Surface properties of diluted aqueous solutions of 3-picoline. *Colloids Surf., A* **2001**, *178*, 207–212.
- [154] Gunde, R.; Kumar, A.; Lehnert-Batar, S.; Mäder, R.; Windhab, E. J. Measurements of Surface and Interfacial Tension for Maximum Volume of a Pendant Drop. *J. Colloid Interface Sci.* **2001**, *244*, 113–122.
- [155] Maham, Y.; Mather, A. E. Surface thermodynamics of aqueous solutions of alkylethanolamines. *Fluid Phase Equilib.* **2001**, *182*, 325–336.
- [156] Yuan, Z.; Herold, K. E. Surface tension of pure water and aqueous lithium bromide with 2-ethyl-hexanol. *Appl. Therm. Eng.* **2001**, *21*, 881–897.
- [157] Kalies, G.; Bräuer, P.; Schmidt, A.; Messow, U. Calculation and Prediction of Adsorption Excesses on the Ternary Liquid Mixture/Air Interface from Surface Tension Measurements. *J. Colloid Interface Sci.* **2002**, *247*, 1–11.
- [158] Kulankara, S.; Herold, K. E. Surface tension of aqueous lithium bromide with heat/mass transfer enhancement additives: the effect of additive vapor transport. *Int. J. Refrig.* **2002**, *25*, 383–389.
- [159] Kumar, A.; Mohandas, V. P.; Ghosh, P. K. Experimental Surface Tensions and Derived Surface Properties of Binary Mixtures of Water + Alkoxyethanols (C_1E_m , $m = 1, 2, 3$) and Water + Ethylene Glycol Dimethyl Ether ($C_1E_1C_1$) at (298.15, 308.15, and 318.15) K. *J. Chem. Eng. Data* **2003**, *48*, 1318–1322.
- [160] Santos, B. M. S.; Ferreira, A. G. M.; Fonseca, I. M. A. Surface and interfacial tensions of the systems water + *n*-butyl acetate + methanol and water + *n*-pentyl acetate + methanol at 303.15 K. *Fluid Phase Equilib.* **2003**, *208*, 1–21.
- [161] Alakoç, U.; Megaridis, C. M.; McNallan, M.; Wallace, D. B. Dynamic surface tension measurements with submillisecond resolution using a capillary-jet instability technique. *J. Colloid Interface Sci.* **2004**, *276*, 379–391.
- [162] Gómez-Díaz, D.; Navaza, J. M. Surface Behavior of Aqueous Solutions of Pyrrolidine and Piperidine. *J. Chem. Eng. Data* **2004**, *49*, 1406–1409.
- [163] Hadrlová, K.; Hovorka, Š.; Bartovská, L. Concentration Dependence of Surface Tension for Very Dilute Aqueous Solutions for Organic Nonelectrolytes. *J. Chem. Eng. Data* **2004**, *49*, 1003–1007.
- [164] Hyvärinen, A. P.; Lihavainen, H.; Hautio, K.; Raatikainen, T.; Viisanen, Y.; Laaksonen, A. Surface Tensions and Densities of Sulfuric Acid + Dimethylamine + Water Solutions. *J. Chem. Eng. Data* **2004**, *49*, 917–922.
- [165] Kijevcanin, M. L.; Ribeiro, I. S. A.; Ferreira, A. G. M.; Fonseca, I. M. A. Water + esters + methanol: experimental data, correlation and prediction of surface and interfacial tensions at 303.15 K and atmospheric pressure. *Fluid Phase Equilib.* **2004**, *218*, 141–148.
- [166] Li, X. X.; Liu, Y. X.; Wei, X. H. Density, Viscosity, and Surface Tension at 293.15 K and Liquid-Liquid Equilibria from 301.15 K to 363.15 K under Atmospheric Pressure for the Binary Mixture of Diethylene Glycol Diethyl Ether + Water. *J. Chem. Eng. Data* **2004**, *49*, 1043–1045.
- [167] Maham, Y.; Chevillard, A.; Mather, A. E. Surface Thermodynamics of Aqueous Solutions of Morpholine and Methylmorpholine. *J. Chem. Eng. Data* **2004**, *49*, 411–415.

- [168] Sönmez, İ.; Cebeci, Y. Investigation of relationship between critical surface tension of wetting and oil agglomeration recovery of barite. *Colloids Surf., A* **2004**, *234*, 27–33.
- [169] Belda, R.; Herraez, J. V.; Diez, O. A study of the refractive index and surface tension synergy of the binary water/ethanol: influence of concentration. *Phys. Chem. Liq.* **2005**, *43*, 91–101.
- [170] Fujii, H.; Matsumoto, T.; Ueda, T.; Nogi, K. A new method for simultaneous measurement of surface tension and viscosity. *J. Mater. Sci.* **2005**, *40*, 2161–2166.
- [171] Ali, K.; Haq, A. U.; Bilal, S.; Siddiqi, S. Concentration and temperature dependence of surface parameters of some aqueous salt solutions. *Colloids Surf., A* **2006**, *272*, 105–110.
- [172] Dixmier, M. Compressibility, the measurement of surface tension, and particle size in molecular or nuclear matter. *J. Colloid Interface Sci.* **2006**, *294*, 391–399.
- [173] Granados, K.; Gracia-Fadrique, J.; Amigo, A.; Bravo, R. Refractive Index, Surface Tension, and Density of Aqueous Mixtures of Carboxylic Acids at 298.15 K. *J. Chem. Eng. Data* **2006**, *51*, 1356–1360.
- [174] Liu, W.; Zhao, T.; Zhang, Y.; Wang, H.; Yu, M. The Physical Properties of Aqueous Solutions of the Ionic Liquid [BMIM][BF₄]. *J. Solution Chem.* **2006**, *35*, 1337–1346.
- [175] Richter, L.; Vollhardt, D. Force Measuring Method for Determination of Surface Tension of Liquids: A Comparison. *Tenside Deterg.* **2006**, *43*, 256–261.
- [176] Romero, C. M.; Paéz, M. S. Surface tension of aqueous solutions of alcohol and polyols at 298.15 K. *Phys. Chem. Liq.* **2006**, *44*, 61–65.
- [177] Enders, S.; Kahl, H.; Winkelmann, J. Surface Tension of the Ternary System Water + Acetone + Toluene. *J. Chem. Eng. Data* **2007**, *52*, 1072–1079.
- [178] Kilaru, P.; Baker, G. A.; Scovazzo, P. Density and Surface Tension Measurements of Imidazolium-, Quaternary Phosphonium-, and Ammonium-Based Room-Temperature Ionic Liquids: Data and Correlations. *J. Chem. Eng. Data* **2007**, *52*, 2306–2314.
- [179] Lisa, G.; Lisa, C. Prediction of excess thermodynamic properties from experimental refractive index of binary mixtures. 1. Water - propionic acid mixtures at 290.15, 300.15 and 310.15 K. *Rev. Roum. Chim.* **2007**, *52*, 647–653.
- [180] Markarian, S. A.; Terzyan, A. M. Surface Tension and Refractive Index of Dialkylsulfoxide + Water Mixtures at Several Temperatures. *J. Chem. Eng. Data* **2007**, *52*, 1704–1709.
- [181] Romero, C. M.; Páez, M. S.; Miranda, J. A.; Hernández, D. J.; Oviedo, L. E. Effect of temperature on the surface tension of diluted aqueous solutions of 1,2-hexanediol, 1,5-hexanediol, 1,6-hexanediol and 2,5-hexanediol. *Fluid Phase Equilib.* **2007**, *258*, 67–72.
- [182] Tuckermann, R. Surface tension of aqueous solutions of water-soluble organic and inorganic compounds. *Atmos. Environ.* **2007**, *41*, 6265–6275.
- [183] Fujimura, Y.; Iino, M. The surface tension of water under high magnetic fields. *J. Appl. Phys.* **2008**, *103*, 124903.
- [184] García-Abuín, A.; Gómez-Díaz, D.; Navaza, J. M.; Vidal-Tato, I. Surface Tension of Aqueous Solutions of Short *N*-Alkyl-2-pyrrolidinones. *J. Chem. Eng. Data* **2008**, *53*, 2671–2674.
- [185] Johnson, I.; Costa, H. F.; Ferreira, A. G. M.; Fonseca, I. M. A. Density, Viscosity, and Surface and Interfacial Tensions of Mixtures of Water+*n*-Butyl Acetate+1-Propanol at 303.15 K and Atmospheric Pressure. *Int. J. Thermophys.* **2008**, *29*, 619–633.

- [186] Liu, W.; Cheng, L.; Zhang, Y.; Wang, H.; Yu, M. The physical properties of aqueous solution of room-temperature ionic liquids based on imidazolium: Database and evaluation. *J. Mol. Liq.* **2008**, *140*, 68–72.
- [187] Williams, D. L.; Jupe, C. L.; Kuklenz, K. D.; Flaherty, T. J. An Inexpensive, Digital Instrument for Surface Tension, Interfacial Tension, and Density Determination. *Ind. Eng. Chem. Res.* **2008**, *47*, 4286–4289.
- [188] Batchelor, T.; Cunder, J.; Fadeev, A. Y. Wetting study of imidazolium ionic liquids. *J. Colloid Interface Sci.* **2009**, *330*, 415–420.
- [189] Rilo, E.; Pico, J.; García-Garabal, S.; Varela, L. M.; Cabeza, O. Density and surface tensions in binary mixtures of C_nMIM-BF₄ ionic liquids with water and ethanol. *Fluid Phase Equilib.* **2009**, *285*, 83–89.
- [190] Romero, C. M.; Jiménez, E.; Suárez, F. Effect of temperature on the behavior of surface properties of alcohols in aqueous solutions. *J. Chem. Thermodyn.* **2009**, *41*, 513–516.
- [191] Bermúdez-Salguero, C.; Gracia-Fadrique, J.; Amigo, A. Surface Tension Data of Aqueous Binary Mixtures of Methyl, Ethyl, Propyl, and Butyl Acetates at 298.15 K. *J. Chem. Eng. Data* **2010**, *55*, 2905–2908.
- [192] Mohsen-Nia, M.; Rasa, H.; Naghibi, S. F. Experimental and theoretical study of surface tension of *n*-pentane, *n*-heptane, and some of their mixtures at different temperatures. *J. Chem. Thermodyn.* **2010**, *42*, 110–113.
- [193] Russo, J. W.; Hoffmann, M. M. Influence of Typical Impurities on the Surface Tension Measurements of Binary Mixtures of Water and the Ionic Liquids 1-Butyl-3-Methylimidazolium Tetrafluoroborate and Chloride. *J. Chem. Eng. Data* **2010**, *55*, 5900–5905.
- [194] Sadeghi, M.; Taghikhani, V.; Ghotbi, C. Measurement and Correlation of Surface Tension for Single Aqueous Electrolyte Solutions. *Int. J. Thermophys.* **2010**, *31*, 852–859.
- [195] Zhang, X. J.; Wang, J. Y.; Hu, Y. Q. Interfacial Tension of *n*-Alkane and Ionic Liquid Systems. *J. Chem. Eng. Data* **2010**, *55*, 4687–4690.
- [196] Rafati, A. A.; Bagheri, A.; Najafi, M. Surface tension of non-ideal binary and ternary liquid mixtures at various temperatures and $p = 81.5$ kPa. *J. Chem. Thermodyn.* **2011**, *43*, 248–254.
- [197] Russo, J. W.; Hoffmann, M. M. Measurements of Surface Tension and Chemical Shift on Several Binary Mixtures of Water and Ionic Liquids and Their Comparison for Assessing Aggregation. *J. Chem. Eng. Data* **2011**, *56*, 3703–3710.
- [198] Wang, J. Y.; Jiang, H. C.; Liu, Y. M.; Hu, Y. Q. Density and surface tension of pure 1-ethyl-3-methylimidazolium L-lactate ionic liquid and its binary mixtures with water. *J. Chem. Thermodyn.* **2011**, *43*, 800–804.
- [199] Han, J.; Jin, J.; Eimer, D. A.; Melaaen, M. C. Density of Water (1) + Monoethanolamine (2) + CO₂ (3) from (298.15 to 413.15) K and Surface Tension of Water (1) + Monoethanolamine (2) from (303.15 to 333.15) K. *J. Chem. Eng. Data* **2012**, *57*, 1095–1103.
- [200] Khattab, I. S.; Bandarkar, F.; Fakhree, M. A. A.; Jouyban, A. Density, viscosity, and surface tension of water+ethanol mixtures from 293 to 323 K. *Korean J. Chem. Eng.* **2012**, *29*, 812–817.
- [201] Quijada-Maldonado, E.; van der Boogaart, S.; Lijbers, J. H.; Meindersma, G. W.; de Haan, A. B. Experimental densities, dynamic viscosities and surface tensions of the ionic liquids series 1-ethyl-3-methylimidazolium acetate and dicyanamide and their binary and ternary mixtures with water and ethanol at $T = (298.15 \text{ to } 343.15 \text{ K})$. *J. Chem. Thermodyn.* **2012**, *51*, 51–58.

- [202] Begum, S. K.; Clarke, R. J.; Ahmed, M. Sh.; Begum, S.; Saleh, M. A. Volumetric, viscosimetric and surface properties of aqueous solutions of triethylene glycol, tetraethylene glycol, and tetraethylene glycol dimethyl ether. *J. Mol. Liq.* **2013**, *177*, 11–18.
- [203] Fu, D.; Wei, L.; Liu, S. Experiment and model for the surface tension of carbonated MEAMDEA aqueous solutions. *Fluid Phase Equilib.* **2013**, *337*, 83–88.
- [204] Hrubý, J.; Vinš, V.; Mareš, R.; Hykl, J.; Kalová, J. Surface Tension of Supercooled Water: No Inflection Point down to $-25\text{ }^{\circ}\text{C}$. *J. Phys. Chem. Lett.* **2014**, *5*, 425–428.
- [205] Lago, S.; Rodriguez-Cabo, B.; Arce, A.; Soto, A. Water/oil/[P_{6,6,6,14}][NTf₂] phase equilibria. *J. Chem. Thermodyn.* **2014**, *75*, 63–68.
- [206] Ren, N. N.; Gong, Y. H.; Lu, Y. Z.; Meng, H.; Li, Ch. X. Surface Tension Measurements for Seven Imidazolium-Based Dialkylphosphate Ionic Liquids and Their Binary Mixtures with Water (Methanol or Ethanol) at 298.15 K and 1 atm. *J. Chem. Eng. Data* **2014**, *59*, 189–196.
- [207] Singh, M. L.; Tripathi, S. C.; Lokhande, M.; Gandhi, P. M.; Gaikar, V. G. Density, Viscosity, and Interfacial Tension of Binary Mixture of Tri-iso-amyl Phosphate (TiAP) and n-Dodecane: Effect of Compositions and Gamma Absorbed Doses. *J. Chem. Eng. Data* **2014**, *59*, 1130–1139.
- [208] Ameta, R. K.; Singh, M. Surface tension, viscosity, apparent molal volume, activation viscous flow energy and entropic changes of water+ alkali metal phosphates at $T = (298.15, 303.15, 308.15)\text{ K}$, *J. Mol. Liq.* **2015**, *203*, 29–38.
- [209] Cárdenas, H.; Cartes, M.; Mejía, A. Atmospheric densities and interfacial tensions for 1-alkanol (1-butanol to 1-octanol) + water and ether (MTBE, ETBE, DIPE, TAME and THP) + water demixed mixtures. *Fluid Phase Equilib.* **2015**, *396*, 88–97.
- [210] Chang, Ch. W.; Hsiung, T. L.; Lui, Ch. P.; Tu, Ch. H. Densities, surface tensions, and isobaric vapor-liquid equilibria for the mixtures of 2-propanol, water, and 1,2-propanediol. *Fluid Phase Equilib.* **2015**, *389*, 28–40.
- [211] Vinš, V.; Fransen, M.; Hykl, J.; Hrubý, J. Surface Tension of Supercooled Water Determined by Using a Counterpressure Capillary Rise Method. *J. Phys. Chem. B* **2015**, *119*, 5567–5575.
- [212] Wohlfarth, C.; Wohlfarth, B. Surface Tension of Pure Liquids and Binary Liquid Mixtures. **1997**.