

Supporting Information

Ternary platinum-copper-nickel nanoparticles anchored to hierarchical carbon supports as free-standing hydrogen evolution electrodes

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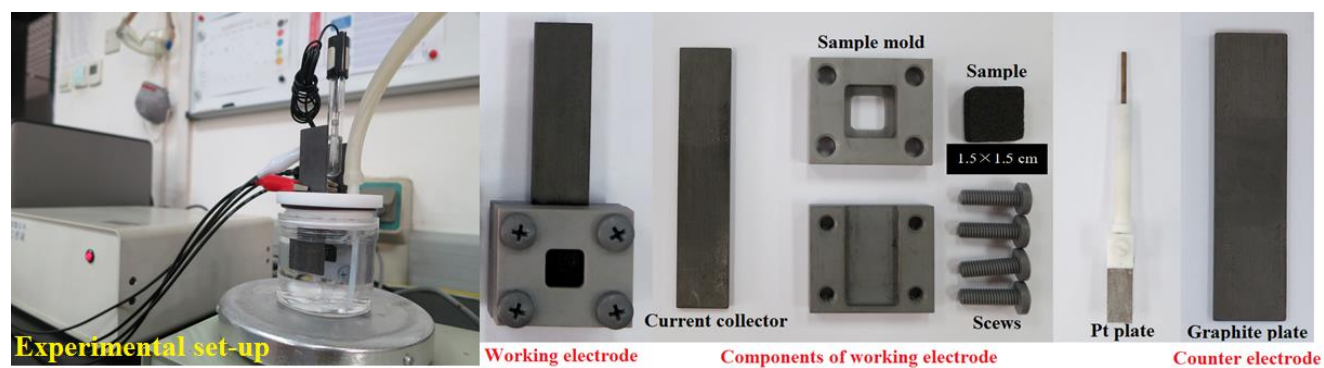


Figure S1 Experimental set-up of electrochemical measurements.

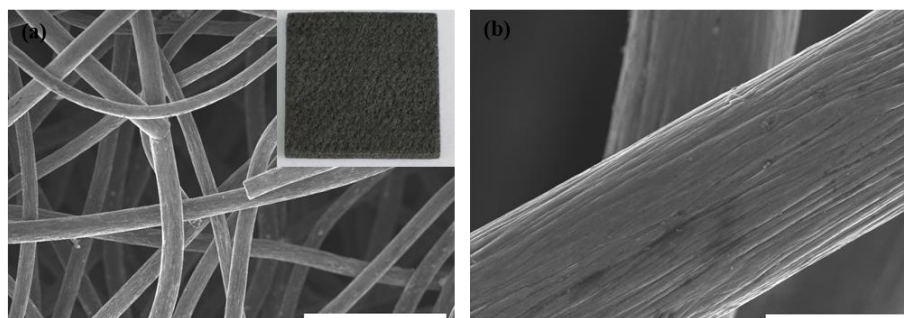


Figure S2 FESEM micrographs of pristine carbon felt (Inset in (a) is an optical image of a piece of $5 \times 5 \times 0.5$ cm carbon felt sheet.) Scale bars, 100 μm (a), 10 μm (b)

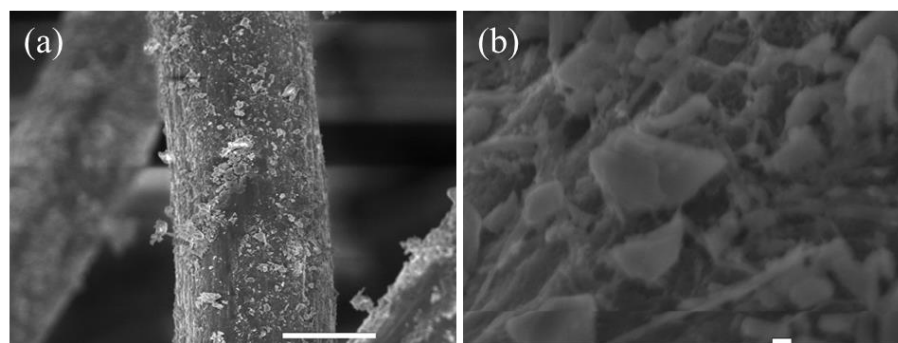


Figure S3 FESEM micrographs of nickel-copper oxalates decorated carbon felt. Scale bars, 10 μm (a), 100 nm (b).

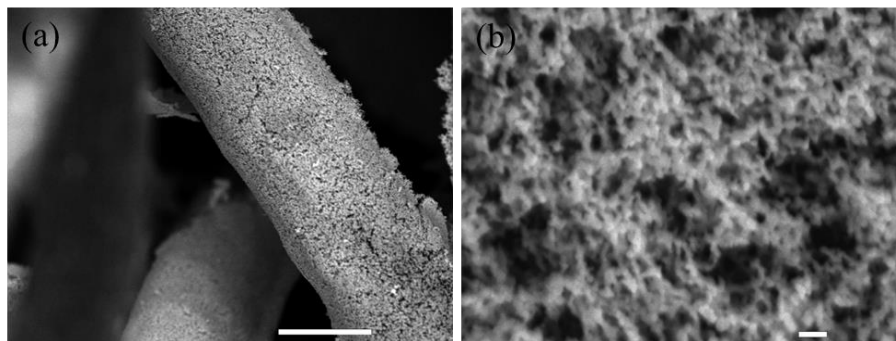


Figure S4 FESEM micrographs of metallic nickel-copper decorated carbon felt. Scale bars, 10 μm (a), 100 nm (b).

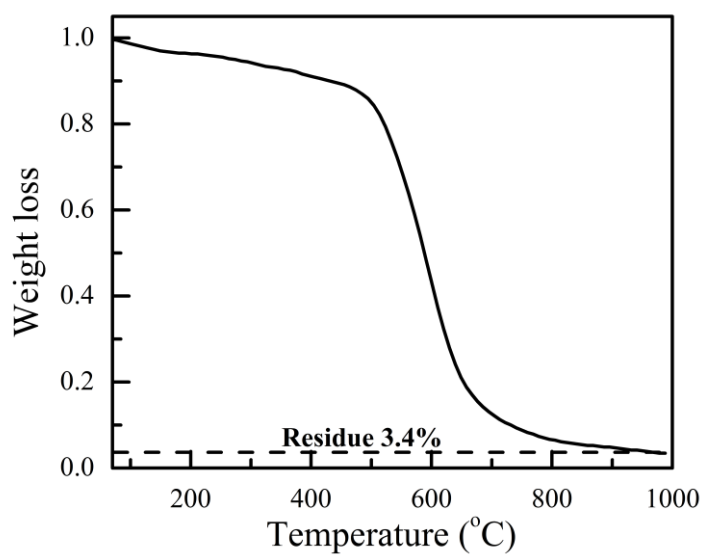


Figure S5 Thermogravimetric curves of $\text{Pt}_{42}\text{Cu}_{57}\text{Ni}_1/\text{CNF}@CF$.

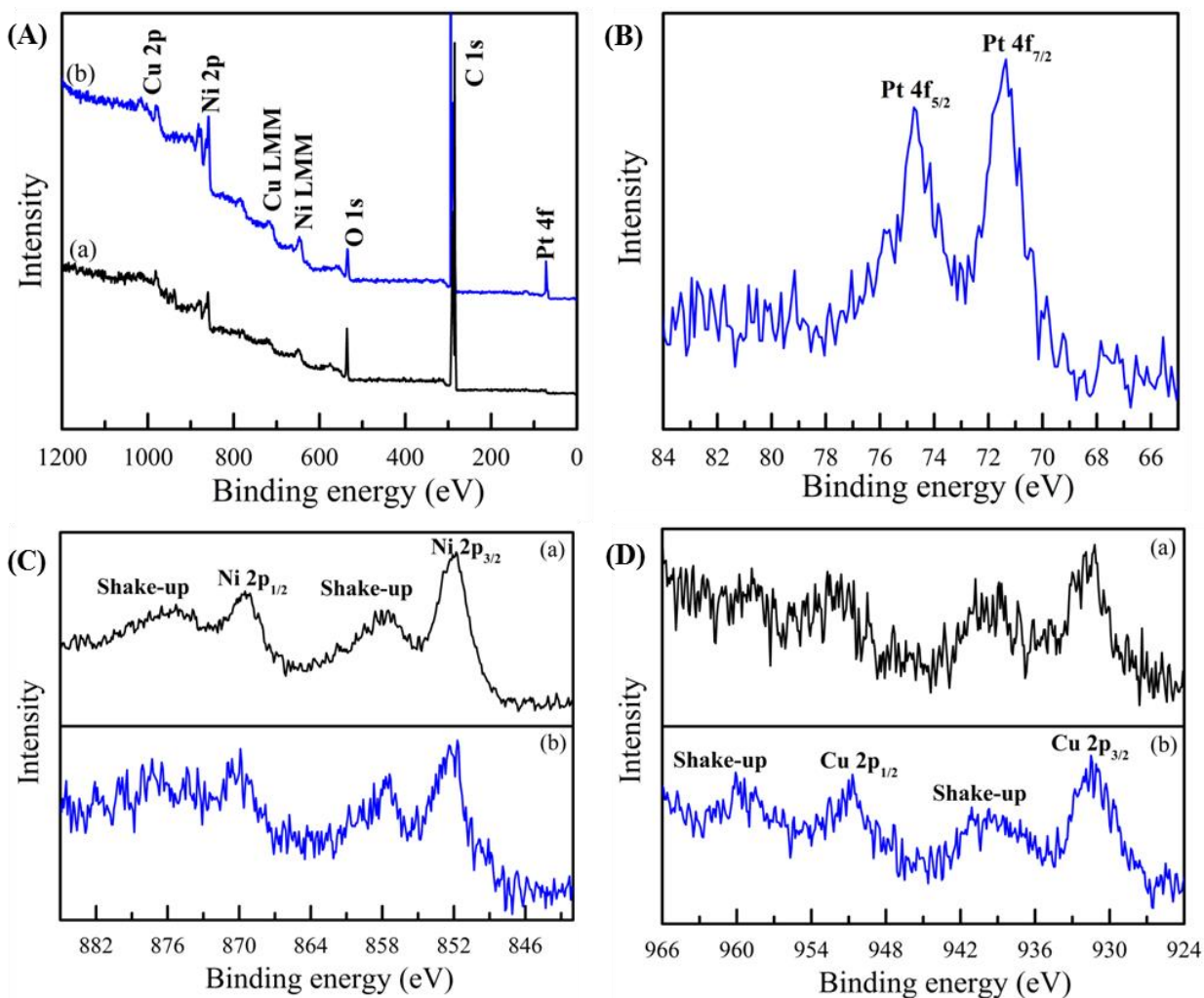


Figure S6 (A) Survey spectra, XPS profiles of (B) Pt 4f, (C) Ni 2p and (D) Cu 2p in (a) $\text{Ni}_{65}\text{Cu}_{35}/\text{CNF}@\text{CF}$ and (b) $\text{Pt}_{13}\text{Cu}_{73}\text{Ni}_{14}/\text{CNF}@\text{CF}$

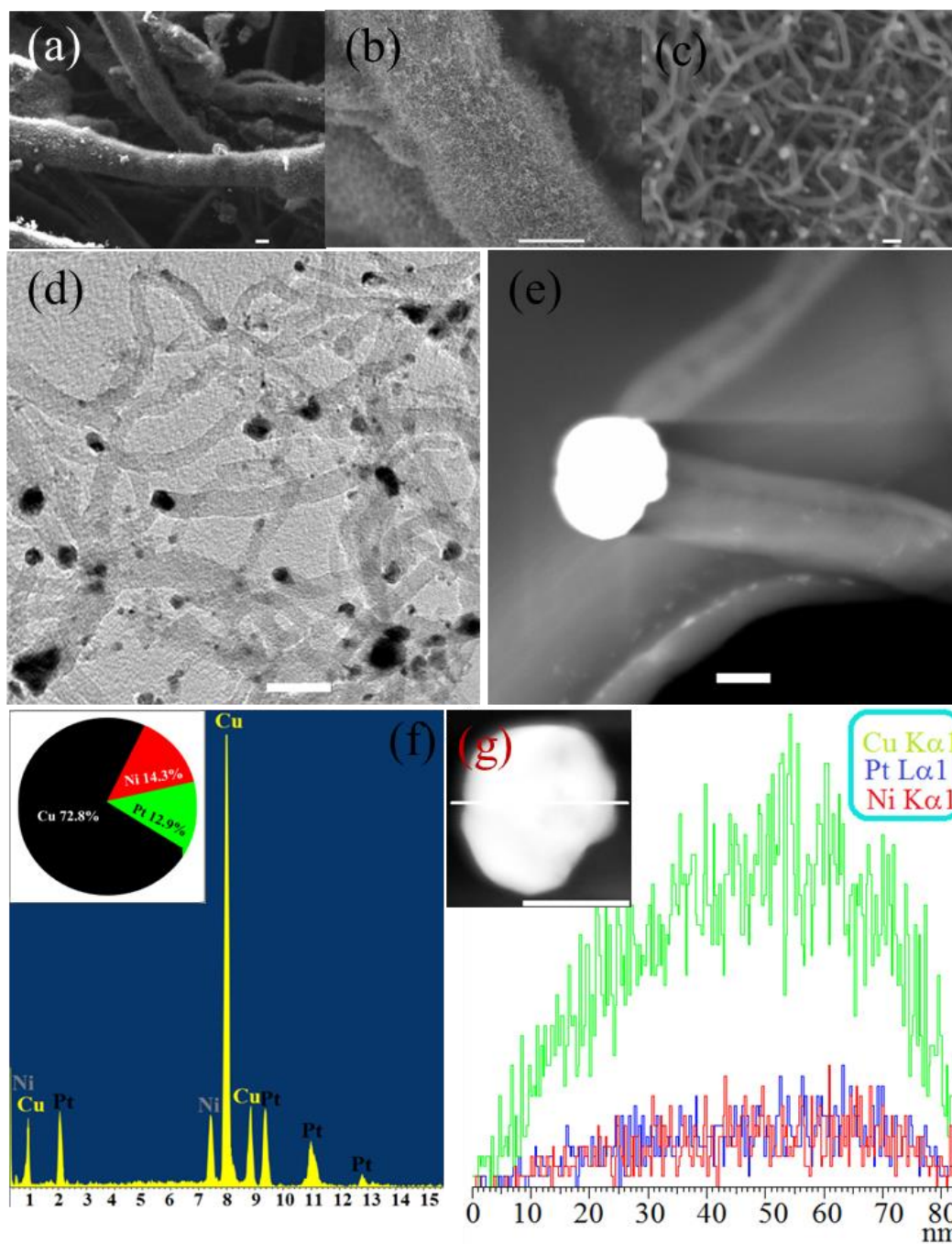


Figure S7 (a)-(c) FESEM micrographs of Pt₁₃Cu₇₃Ni₁₄/CNF@CF, (d) TEM, (e) HAADF-STEM micrographs of isolated Pt₁₃Cu₇₃Ni₁₄/CNF@CF, (f) EDS profile of Pt₁₃Cu₇₃Ni₁₄ NPs (inset in (f), atomic composition of Pt₁₃Cu₇₃Ni₁₄ NPs), (g) line scan and elemental distribution of a Pt₁₃Cu₇₃Ni₁₄ NP. Scale bars, 10 μm (a, b), 100 nm (c), 200 nm (d), 50 nm (e) and (g)

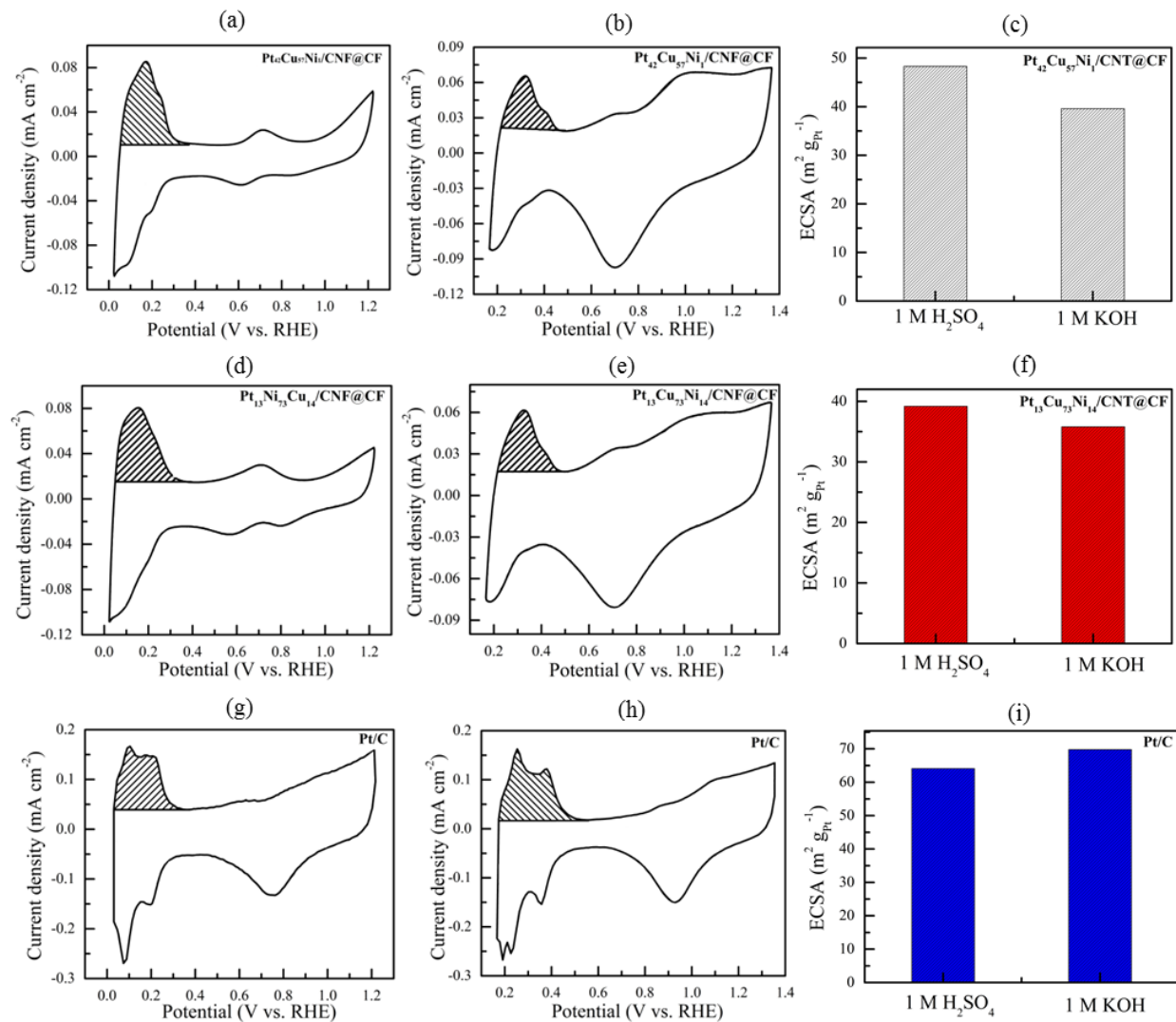


Figure S8 (a) and (b) Cyclic voltammetry curves of $\text{Pt}_{42}\text{Cu}_{57}\text{Ni}_1/\text{CNF}@CF$ in 1 M H_2SO_4 and 1 M KOH with a scan rate of 50 mV s^{-1} , respectively, and (c) ECSA values of $\text{Pt}_{42}\text{Cu}_{57}\text{Ni}_1/\text{CNF}@CF$ in 1 M H_2SO_4 and 1 M KOH; (d) and (e) cyclic voltammetry curves of $\text{Pt}_{13}\text{Cu}_{73}\text{Ni}_{14}/\text{CNF}@CF$ in 1 M H_2SO_4 and 1 M KOH with a scan rate of 50 mV s^{-1} , respectively, and (f) ECSA values of $\text{Pt}_{13}\text{Cu}_{73}\text{Ni}_{14}/\text{CNF}@CF$ in 1 M H_2SO_4 and 1 M KOH; (g) and (h) cyclic voltammetry curves of Pt/C in 1 M H_2SO_4 and 1 M KOH with a scan rate of 50 mV s^{-1} , respectively, and (i) ECSA values of Pt/C in 1 M H_2SO_4 and 1 M KOH.

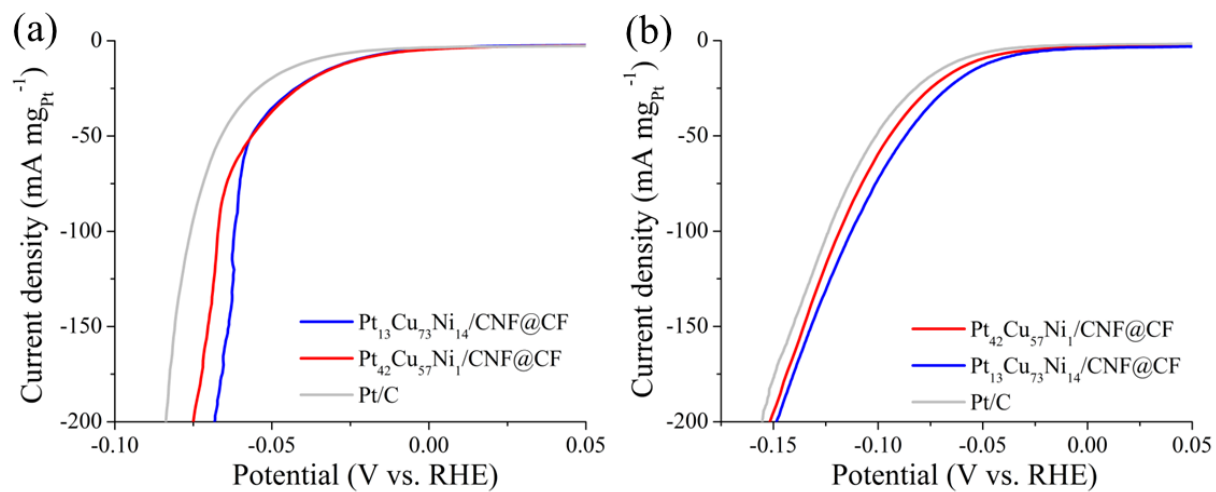


Figure S9 (a) and (b) Pt-mass normalized polarization curves of catalysts in 1 M H₂SO₄ and 1 M KOH, respectively.

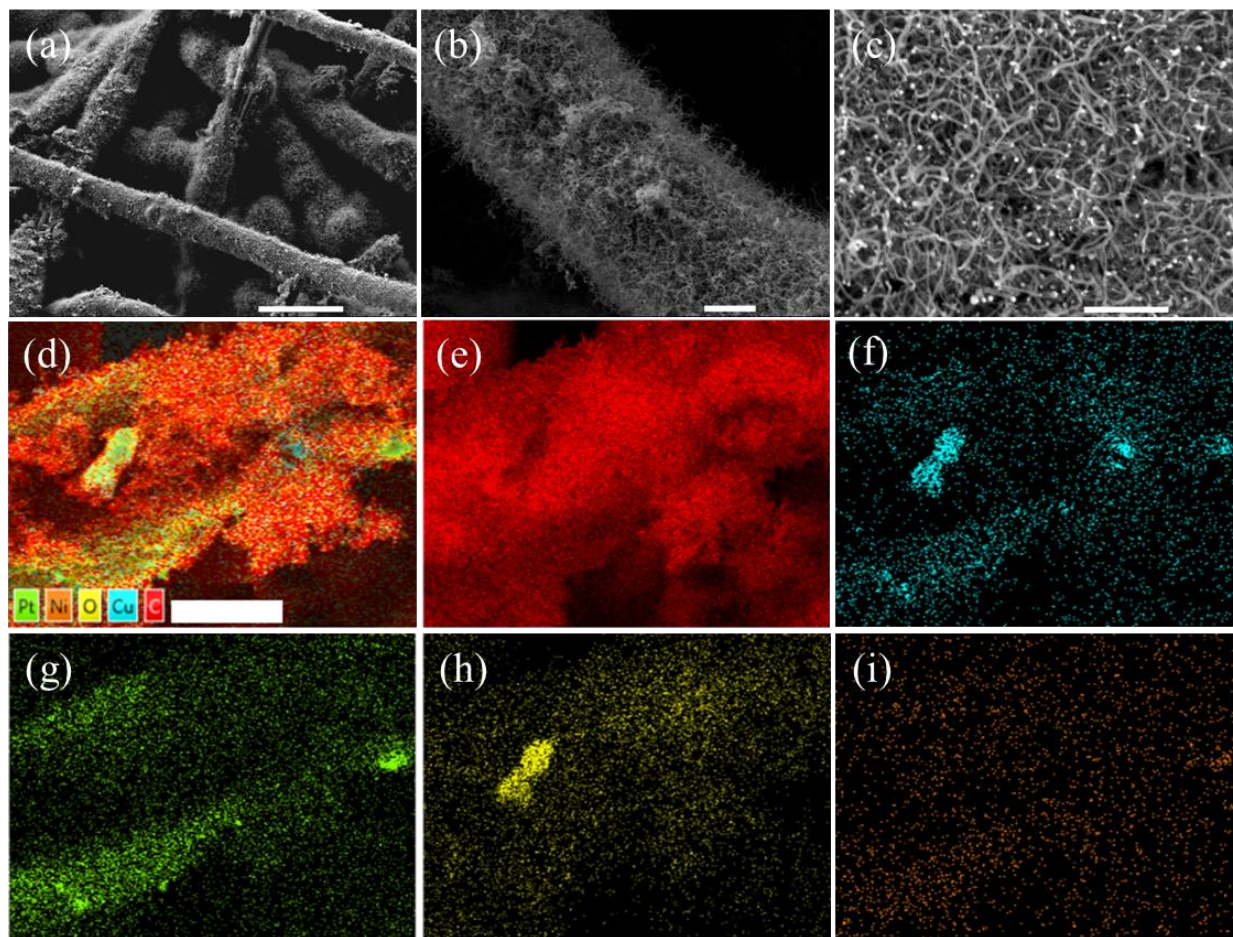


Figure S10 (a)-(c) FESEM micrographs of spent $\text{Pt}_{13}\text{Cu}_{73}\text{Ni}_{14}/\text{CNF}@\text{CF}$ in 1 M H_2SO_4 , (d) overall EDS micrograph, (e)-(i) elemental mapping micrographs of carbon, copper, platinum, oxygen, and nickel, respectively. Scale bar, 10 μm .

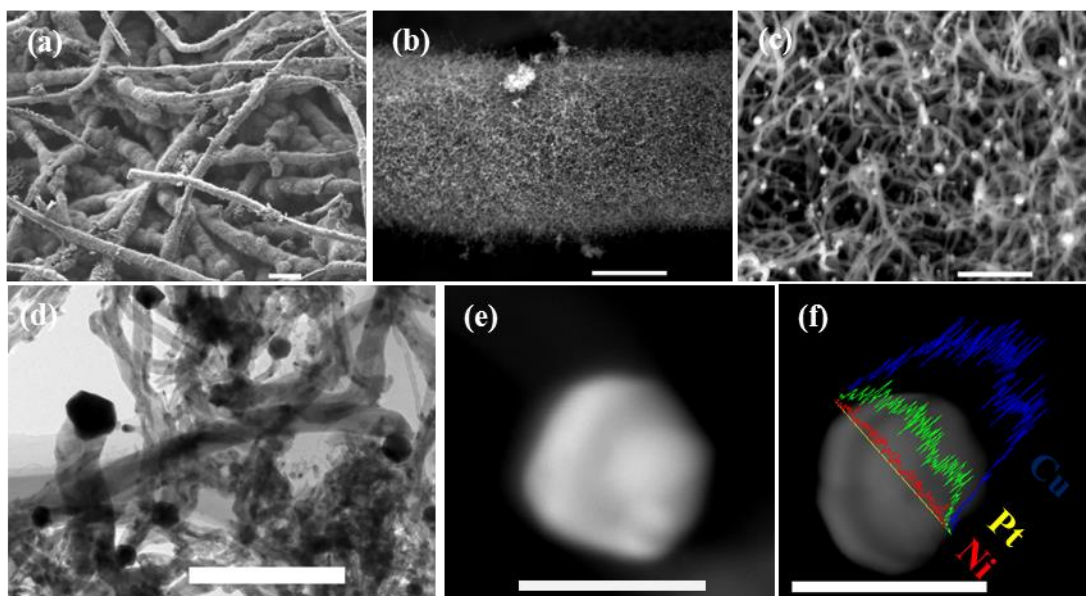


Figure S11 (a)-(c) FESEM micrographs of spent $\text{Pt}_{42}\text{Cu}_{57}\text{Ni}_1/\text{CNF@CF}$ in 1 M H_2SO_4 , (d) TEM micrograph of isolated $\text{Pt}_{42}\text{Cu}_{57}\text{Ni}_1/\text{CNF@CF}$, (e) high-resolution HAADF-STEM micrograph of a typical $\text{Pt}_{42}\text{Cu}_{57}\text{Ni}_1$ NP, (f) line scan and elemental distribution of a $\text{Pt}_{42}\text{Cu}_{57}\text{Ni}_1$ NP. Scale bars, 100 μm (a), 10 μm (b), 1 μm (c) and (d), 100 nm (e) and (f)