

# Supporting information for:

## Deuteration of Perylene Enhances Photochemical Upconversion Efficiency

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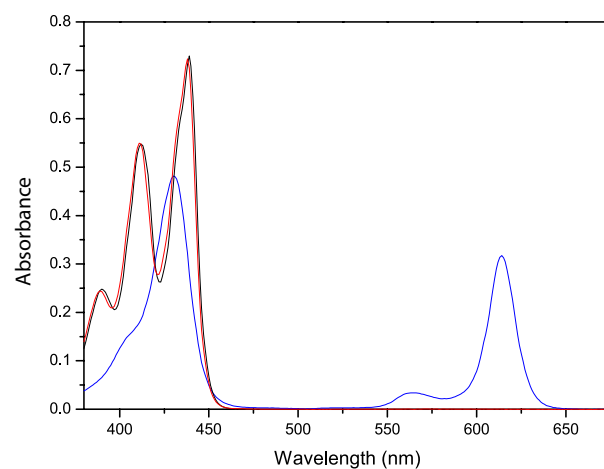


Figure S1: Absorption spectra of 0.01 mM TPTBPt (blue), 0.1 mM perylene (black), and 0.1 mM deuterated perylene (red) in a 1 mm cuvette.

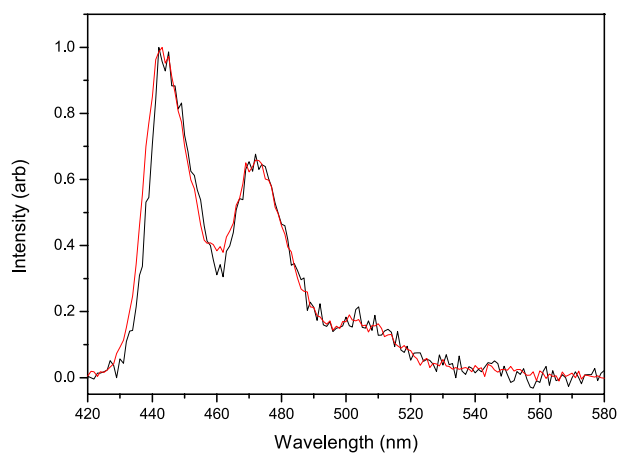


Figure S2: Fluorescence emission spectra of perylene (black) and deuterated perylene (red) with 410 nm excitation.

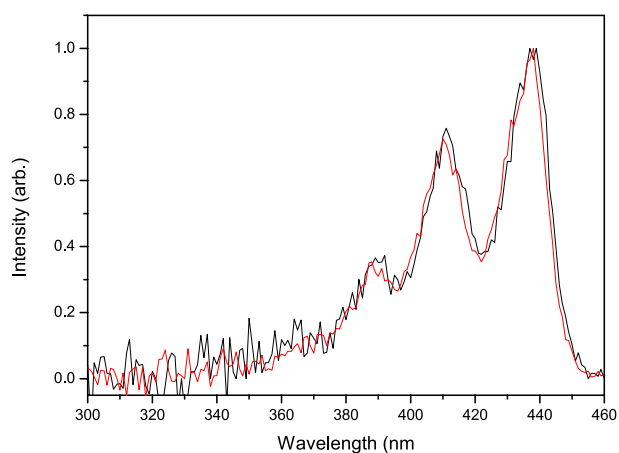


Figure S3: Fluorescence excitation spectra of perylene (black) and deuterated perylene (red) observed at 470 nm.

**Table S1: Scaling factors used to normalise action spectra at 470 nm direct response peak**

Pump Power (mW)	0	2.4	3.3	4.5	4.5	6.4	9.8	16.2	16.2	29.9	159.7
Scaling factor (%)	1.59	1.29	1.59	2.54	3.86	2.37	3.23	3.43	3.97	3.51	1.34

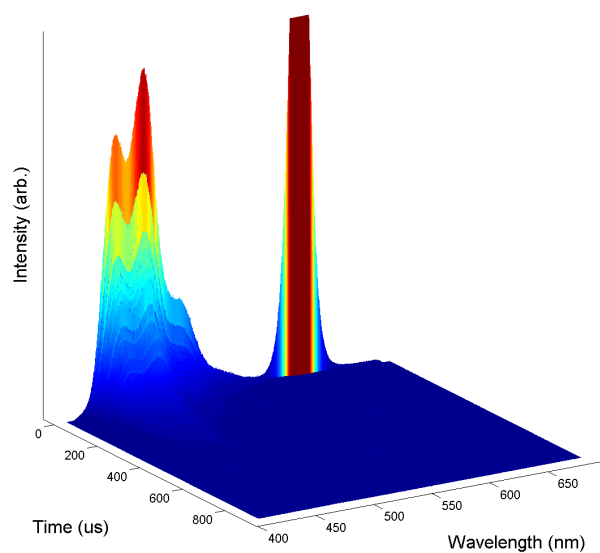


Figure S4: Typical time-resolved photoluminescence spectrum of P0 solution. Laser excitation at 615 nm is observed in the first time slice only.

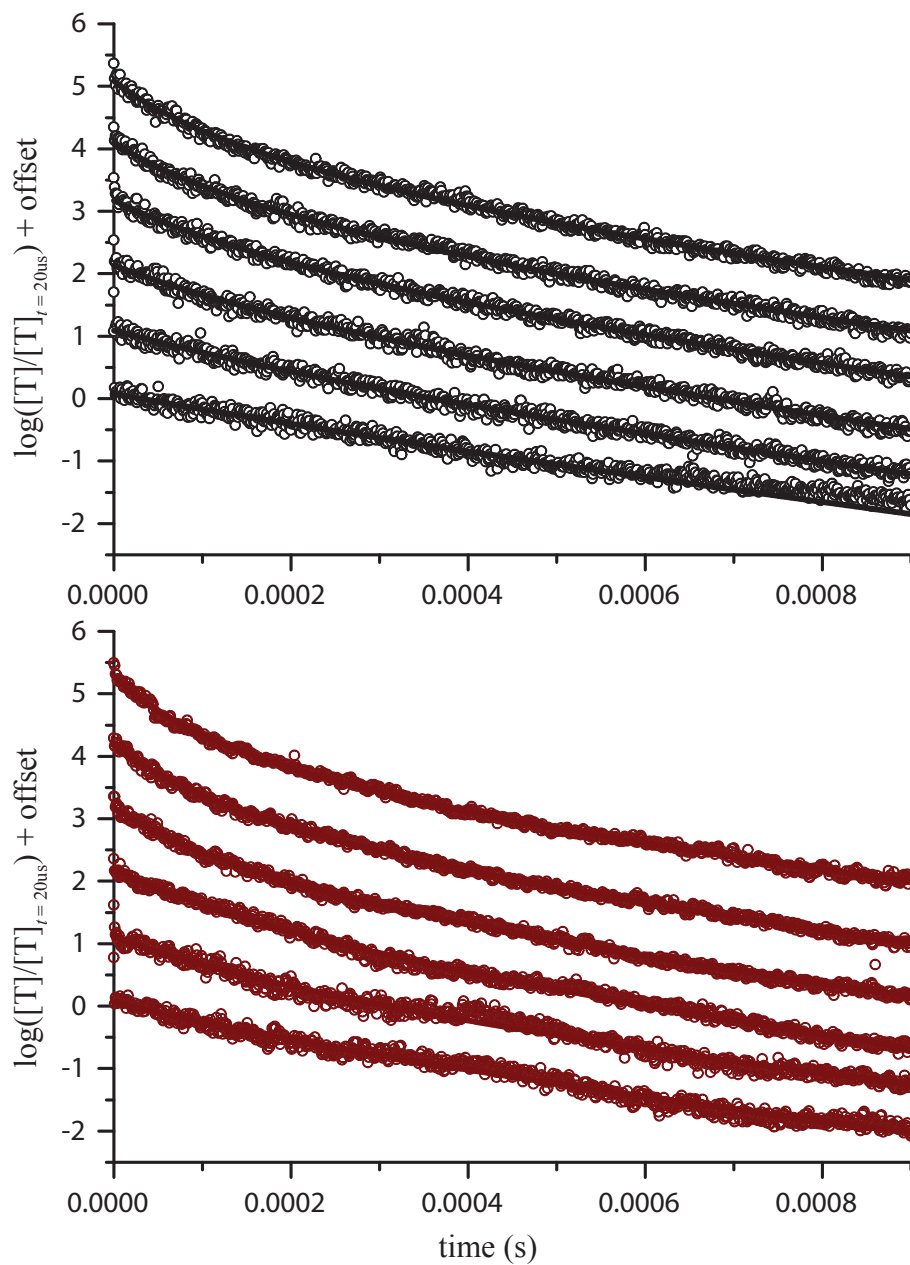


Figure S5: Transient log-normalized triplet concentrations (from square-root of luminescence) for P0 (top) and P98 (bottom). Sequentially higher pulse energies are offset for clarity, revealing kinetics which deviate more significantly from exponential decay. Solid lines are fits to a mixed first and second order kinetics model.